Effect of relative humidity on foliar absorption of P and Rb by Chrysanthemum and Pilea*

SZCZEPAN MARCZYŃSKI 1, and H. B. TUKEY, Jr. 2

¹ Faculty of Horticulture, Warsaw Agricultural University, ul. Nowoursynowska 166, 02-766 Warszawa, Poland ² University of Washington, Seattle, WA, USA

(Received: December 21, 1984)

Abstract

Air humidity differentially affected the foliar absorption of ³⁵P and ⁸⁶Rb by Chrysanthemum morifolium 'Giant # 4 Indianapolis White' and Pilea cardierei. As the relative humidity increased from 47 to 80% in most cases, there was in increase in uptake. Further raising of humidity to 92% increased uptake of Rb and P by Chrysanthemum leaves only when guttation did not occure. Air humidity influenced most strongly the uptake of calcium phosphate probably due to its low solubility.

INTRODUCTION

Foliar absorption of substances is dependent upon environmental factors such as relative humidity. Generally there is an increase in uptake with an increase in relative humidity (Babiker and Duncan, 1975; Marczyński and Jankiewicz, 1978a; b; Prasad et al., 1967; Sachs et al., 1967) but exceptions have been found (Morton, 1966; Teubner et al., 1957; Westwood and Batjer, 1960). Therefore this research was done to clarify the relationship between relative humidity and foliar uptake of P and Rb.

^{*} This work was partially supported by the nursery industry through a grant from the Horticultural Research Institute of the American Association of Nurserymen, and by the Thomas and Franches Reilly Grant of the Ra-Pid-Gro Corporation, Dansville, NY. The assistance of Robert Spaulding and E. S. Jackson is gratefully acknowledged.

MATERIALS AND METHODS

Methods were similar to those described earlier (Marczyński and Tukey, 1986). Rooted cuttings of Chrysanthemum morifolium 'Giant # 4 Indianapolis White' and Pilea cardierei were grown in a greenhouse in a Hoagland's nutrient solution containing half-strength P and K until 2-3 new leaves had developed. One day before treatment the plants were placed in growth chambers with a day-night temperature of $22^{\circ}\text{C}\pm1^{\circ}\text{C}$, light intensity of 2×10^{4} lx $\pm1\times10^{8}$ lx, and 16-hour day length, conditions which were maintained throughout the experiments. In experiments requiring relative humidities above $80^{\circ}/_{\circ}$, plants were placed in transparent plastic tents built within the growth chamber. A humidifier and two fans were placed in the tent to increase humidity and to maintain air circulation; the floor was also kept constantly wet.

The upper or the lower surface of the third leaf from te apex was treated in the middle of the day with 20 μ l of 10 mM rubidium phosphate, pH 7.5, which was double labelled with \$^{33}P\$ and \$^{36}Rb\$. In one experiment, rubidium phosphate, pH 5.0 was also applied and in another calcium (pH 5.0) and ammonium phosphate (pH 8.0) were used. At the end of the treatment period, usually 48 h, a l. 35 cm diameter disk encompassing the treated area of the leaf was removed with a cork borer. The remainder of the plant was ashed and assayed for radioactivity as described previously (Marczyński and Tukey, 1986). In some of the experiments, the disks punched from Pilea leaves were washed, and radioactivity in the disks after washing was determined. Total uptake was the radioactivity remaining in the disks after washing plus radioactivity translocated from the treated spot to other plants parts.

Each treatment had 5-10 replications. After angular transformation, data were subjected to analysis of variance; Tukey's W-procedure was used for comparing treatment means (Steel and Torrie, 1960). Plant water potential was measured with a Plant Water Status Console Model 3005, Soilmoisture Equipment Company.

RESULTS

Greater amounts of both ³³P and ⁸⁶Rb were absorbed through the abaxial leaf surfaces of *Pilea* than the adaxial surfaces. For example, 80-90% of the ³³P and 60-70% of the ⁸⁶Rb applied to the abaxial leaf surface were absorbed as compared with 3-5% of the ³³P and 20-25% of the ⁸⁶Rb through the adaxial surfaces (Table 1). Less than 1% of the ³³P and 20% of the ⁸⁶Rb applied to the adaxial surface of *Pilea* leaves was translocated from the treated area to other plant parts within 48 h

Table 1

Effect of relative humidity on absorption of ³³P and ⁸⁶Rb with 10 mM rubidium phosphate by Pilea leaves for 48 h

		3:	3 P		⁸⁶ Rb			
Air humidity	absorbed total in		absorbed and trans- located in %		absorbed total in		absorbed and trans- located in %	
	adaxial surface	abaxial surface	adaxial surface	abaxial surface	adaxial surface	abaxial surface	adaxial surface	abaxial surface
78% RH growth chamber 80% RH	4.85	83.79	0.77	12.24	25.25	58.14	21.50	42.32
plastic tent 91 % RH	4.31	86.00	0.57	14.97	20.57	65.20	16.50	47.05
plastic tent	3.25	90.03	1.00	14.71	19.56	69.85	11.33	51.48

No significant differences were found between means within the same column at P = 0.01.

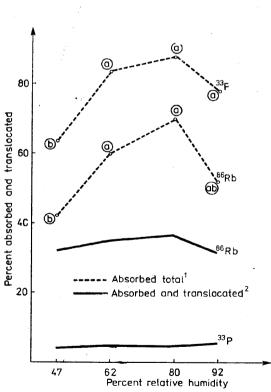


Fig. 1. Effect of relative humidity on absorption and translocation of ^{33}P and ^{66}Rb with 10 mM rubidium phosphate by abaxial surface of *Pilea* leaves. 1 Means on each curve designated by the same letters do not differ significantly at P = 0.01.

 2 There were no significant differences between means at P=0.01

as compared with $14^{0}/_{0}$ of the 33 P and $51^{0}/_{0}$ of the 86 Rb applied to the abaxial surfaces. These results are similar to those of previous experiments (Marczyński and Tukey, 1986).

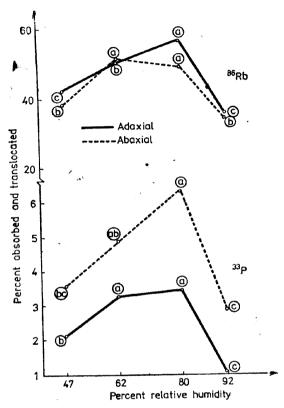


Fig. 2. Effect of relative humidity on absorption and translocation of ^{33}P and ^{86}Rb with 10 mM rubidium phosphate by *Chrysanthemum* leaves. Means on each curve designated by the same letters do not differ significantly at P=0.01

As the relative humidity increased from 47 to 62%, there was an increase in total uptake (radioactivity in treated areas after washing, plus radioactivity translocated to other plant parts) of SBR and SBR and SBR and SBR was higher with the 62% condition but there were no significant differences in contents of SBR and SBR outside the treated spots (Fig. 1). When the relative humidity was raised from 62 to 80%, in most cases the increase in uptake was nonsignificant (Figs. 1, 2). Further raising of humidity to 92% decreased uptake and translocation of SBR and SBR by Chrysanthemum leaves (Fig. 2). A similar trend was found with the lower surface of Pilea leaves, but differences were not statistically significant (Fig. 1).

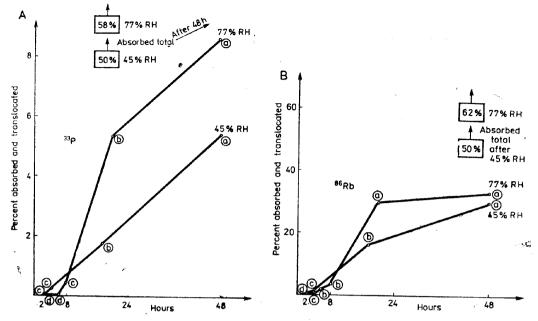


Fig. 3. Effect of relative humidity on absorption and translocation of \$3P and \$6Rb with 10 mM rubidium phosphate by abaxial surface of Pilea leaves. A — absorption and translocation of \$3P, B — absorption and translocation of \$6Rb. Means on each curve designated by the same letters do not differ significantly at P = 0.05. There was a significant difference between humidities at P = 0.01. Drying time at 45%0 RH — 80 min and at 77%0 RH — 170 min

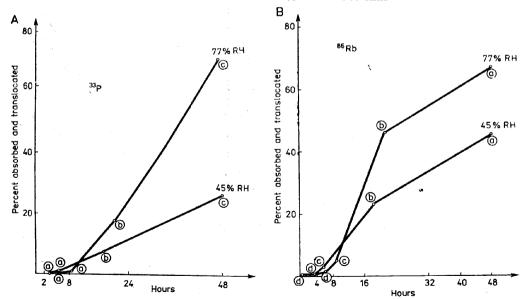


Fig. 4. Effect of relative humidity on absorption and translocation of \$\$^3P\$ and \$\$^6Rb\$ with 10 mM rubidium phosphate by adaxial surface of Chrysanthemum leaves. A — absorption and translocation of $8P , B — absorption and translocation of $8Rb . Means on an each curve designated by the same letter do not differ significantly at P = 0.05. There was a significant difference between humidities at P = 0.01. Drying time at $45^{9}/_{0}$ RH — 140 min and at $77^{9}/_{0}$ RH — 290 min

The decrease in uptake and translocation at RH of 92% was surprising. Perhaps the treating droplet dried more slowly which resulted in slower increase of radioisotope solution concentration and slower absorption from the droplet. Another possibility could be the change of the CO, level in the plastic tent used to attain 92% RH. Some other factors are also not excluded. To check this, two experiments were done with or without plastic tents. In the first one, absorption was compared at the highest (77%) and lowest (45%) relative humidities which could be maintained in the growth chambers without use of a plastic tent. Two to 10 h after application, relative humidity had no significant effect on uptake and translocation of 33P and 86Rb by either Pilea or Chrysanthemum leaves, although a constant trend to greater absorption and translocation at lower air humidity was observed (Figs. 3, 4). After this initial period, there was an increase in absorption and translocation by both plants at 77% RH, as compared with 45% RH, up to 48 h after treatment (Figs. 3, 4).

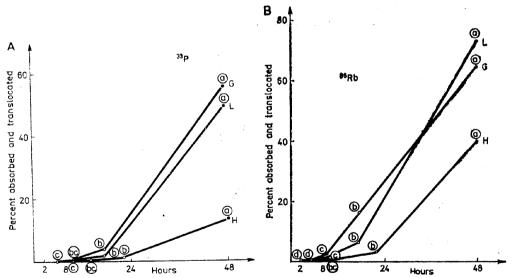


Fig. 5. Effect of relative humidity on absorption and translocation of ^{33}P and ^{86}Rb with 10 mM rubidium phosphate by adaxial surface of Chrysanthemum leaves. A — absorption and translocation of ^{33}P , B — absorption and translocation of ^{86}Rb . Means on each curve designated by the same letter do not differ significantly at P=0.05. There was a significant difference between L and H at P=0.01 and no significant difference between L and G at P=0.05. G — growth chamber with 78% RH, L — plastic tent with 80% RH and H — plastic tent with 91% RH. Drying time at G=5 h, at L — 6 h and at H — 13 h

In the second trial, absorption was compared at the highest (91%) and lowest (80%) relative humidities which could be maintained in two plastic tents built within one growth chamber. To check whether the plastic tent itself had an influence on uptake, plants were also treated in a second growth chamber without a plastic tent, but with temperature, light intensity and relative humidity (78%) very close to that within one of the plastic tents. There was no effect of humidity on uptake by the lower and upper surfaces of Pilea leaves, during the 48-h absorption period (Table 1). But in Chrysanthemum, less 86Rb and 33P was absorbed and translocated from the treated spot in the higher humidity (91%) than in lower humidity (80%) (Fig. 5), similarly as was noted before (Fig. 2). Further, there were no differences in CO₂ levels in the growth chambers and in the plastic tent placed within the growth chamber. Similarly, there was no difference in uptake and translocation of **P and 68Rb by both Pilea and Chrysanthemum leaves in the growth chamber and in the plastic tent within the second growth chamber, both of which were at approx, the same relative humidity (78-89%).

At night in the 91% relative humidity, droplets of guttation fluid were seen on the whole surface of Pilea leaves and on the leaf edges of Chrysanthemum. During guttation at night the water potential of Chrysanthemum was -1.3 bars and in the same humidity during the day was -2.0 bars. For Pilea plants it was respectively -1.1 and -2.6 bars. Because the plastic tent did not have an influence on uptake (Fig. 5), and because drying of the droplet could have an influence on uptake only during the first 10 to 20 h (Figs. 3, 4), it was supposed that perhaps the decreases in uptake and translocation at high humidities were in some way related to guttation. To check this, plants were treated within plastic tents with relative humidities of 79% and 92% as before, but with constant light which eliminated guttation. Further, the plants were treated with phosphates with different solubilities.

When guttation did not occur (under constant light) absorption and translocation of both ³³P and ⁵⁶Rb were the same or greater at 92% than at 79% RH (Table 2) independent of the phosphate used. Phosphorus absorption and translocation were particularly increased with increased humidity when *Pilea* and *Chrysanthemum* were treated with calcium phosphate (Table 2), the solubility of which is only 0.5 g/100 g H₂O at 25°C (L i n k e, 1965). Absorption of ammonium phosphate (solubility 41 g/100 g H₂O at 25°C) by *Pilea* leaves was not affected by humidity. Absorption and translocation by *Chrysanthemum* leaves were higher at 92% than at 79% RH but differences were much less than with calcium phosphate (Table 2).

T a ble 2

Effect of relative humidity and pH on absorption and translocation of ³³P and ⁸⁶Rb with 10 mM phosphates by *Chrysanthemum* adaxial and *Pilea* abaxial leaf surface for 48 h. Plants were kept under constant light

	Air humidity	Chrysanthemum				Pilea	
		³³ P ⁸⁶ Rb		³³ P		86Rb	
Phosphate		translo		abcarbad total	absorbed and translocated	absorbed total	absorbed and translocated
Rb-phosphate	79% RH	5.2 c	23.4 b	66.5 bc	6.5 b	42.3 a	31.6 a
pH-5.0	92% RH	13.3 a	53.1 a	82.0 a	13.3 a	42.2 a	37.2 a
Rb-phosphate	79% RH	2.5 a	46.3 a	55.8 a	3.1 c	39.1 a	30.1 a
pH-7.5	92% RH	2.0 d	46.1 a	78.0 ab	4.1 c	45.2 a	32.0 a
Ca-phosphate	79% RH	1.3 d		6.7 d	0.9 d		
pH-5.0	92% RH	7.6 bc		54.1 c	11.6 a		
NH ₄ -phosphate	79% RH	8.4 b		77.3 ab	12.1 a		
	92% RH	12.4 a		80.1 a	12.2 a		·

Means within the same column followed by a different letter are significantly different at P = 0.01.

DISCUSSION

The influence of air humidity on leaf functions including absorption is complex depends on several factors.

In our experiments, rubidium phosphate at pH 7.5 showed a trend to lower absorption in the higher humidity during the first 6-10 h, probably due to slow drying of the droplet. It was reported that prolonged drying of the droplet in higher air humidity increases absorption by extending the time of hydratation (Bukovac, 1974; Pallas and Williams, 1962). With compounds with high water retention, such as rubidium phosphate at pH 7.5 (Reed and Tukey, 1978) it is probable that some water can be retained by deposit on the leaf surface during a few hours after visible drying even in low air humidity (i.e. 45%). Acceleration of drying increases the concentration of the treated compound and increases absorption which agrees with work of others (Greene and Bukovac, 1971; Middleton and Sanderson, 1965). In line with this are reports that rewetting a deposit of magnesium (Oland and Opland, 1956), phosporus (Thorne, 1958), strontium and cesium (Morton, 1966) decreased or did not change leaf absorption, probably due to an effect of concentration. But in some experiments, rewetting increased leaf absorption of strontium (Ambler, 1964), 3-CP (Bukovac, 1965), and Dalapon (Prasad et al., 1962), possibly because these compounds have a lower resistance to complete

drying. In our experiments, rubidium phosphate at pH 7.5 in $77^{0}/_{0}$ RH, within 10-20 h after application, exceed absorption occured in $45^{0}/_{0}$ RH (Figs. 3, 4), probably due to complete drying of the droplet in the lower humidity. In $77^{0}/_{0}$ RH the compound did not dry completely at least during 20 h and its absorption increased greatly in time.

Some workers reported that solutions containing surfactants were absorbed independently of air humidity (Thompson et al., 1958; Westwood and Batjer, 1960). It is also possible that adding compounds with high solubility may increase absorption. Few workers have reported on the relation between air humidity, solubility, and hygroscopicity of different compounds which Reed and Tukey (1978) have shown to be most important and is demonstrated in this work. Rubidium phosphate at pH 7.5 and ammonium phosphate, which are both soluble in water, were absorbed and translocated in similar quantities as both 79% and 92% RH. Calcium phosphate with its low water retention, was absorbed 6-10 times more at higher humidity than at lower.

It is widely thought that air humidity influences leaf absorption mainly by its effect on cuticle hydration (Ambler, 1964; Babiker and Duncan, 1975; Schönherr, 1976) or by increasing cell water content (Thorne, 1958). But as shown here, the relationship between solubility and air humidity is even more important. Absorption of certain compounds is strongly dependent on air humidity and the ability to form a concentrated film of solution on the leaf surface.

Leaf morphology, including trichomes and cuticle types can influence the air humidity at the surface (Reed and Tukey, 1978) and in this way could modify the influence of the surrounding air humidity on leaf absorption.

It is hard to explain the lower absorption and translocation by Chrysanthemum leaves in relative humidity above 90% as compared with 80% (Figs. 2, 5). This difference was noticed only when guttation appeared. Perhaps high root pressure during guttation, which has been demonstrated by high water potential, inhibited absorption. This is supported by the work of Pallas and Williams (1962) who noticed decreased absorption of Pallas and Williams (1962) who noticed decreased absorption of Pallas and Williams (1962) who noticed decreased absorption of pressure. In addition, it is postulated that guttation is an active process (Häusermann and Wyssling, 1973; Schrept, 1965) which can decrease reserves of ATP and possibly inhibit absorption of ions. In our experiments, the guttation solution did not contain any radioactivity, as was reported by others (Ziegler and Lüttge, 1968), so decreases in absorption were not caused by loss of activity in the guttation fluid.

The Pilea plants did not show any influence of guttation or water

potential on P and Rb absorption (Table 1). This could be related to plant specificity as it was reported that *Prospis jutiflora* (Middleton and Sanderson, 1965), a xerophyte, absorbed the same amount of 2, 4, 5-T between $35^0/_0$ and $100^0/_0$ RH.

Humidity can have a very diverse influence on the absorption by leaves. It influences the physico-chemical state of the compound adsorbed on the leaf surface, the physico-chemical state of the leaf surface itself and leaf functions like stomata opening as well as the physiological conditions of the whole plant — its transpiration, gas exchange, root pressure, water conditions of the cells etc. Therefore the effect of increased humidity on absorption differs markedly with different compounds, different external conditions other than humidity and with plant species. This paper show examples of that diversity.

The very intriguing phenomenon found in this paper — the lowering of absorption in higher humidity seems to have a connection with guttation, but further research is required to explain it.

REFERENCES

- Ambler J. E., 1964. Translocation of strontium from leaves of bean and corn plants. Radial. Bot. 4: 259-265.
- Babiker A. G. T., Duncan H. J., 1975. Penetration of bean leaves by asulam as influenced by adjuvants and humidity. Pestic. Sci. 6: 655-664.
- Bukovac M. J., 1965. Some factors affecting the absorption of 3-chlorophenoxy-α-propionic acid by leaves of the peach. Proc. Amer. Soc. Hort. Sci. 87: 131-138.
- Bukovac M. J., 1974. Foliar penetration of plant growt substance with special reference to tree fruits. Acta. Hort. 34: 69-78.
- Greene D. W., Bukovac M. J., 1971. Factors influencing the penetration of naphtaleneacetamide into leaves of pear (*Pyrus communis* L.). J. Amer. Soc. Hort. Sci. 96: 240-246.
- Häusermann E., Frey-Wyssling A., 1973. Phosphatase-aktivitat in hydathoden. Protoplasma 57: 371-380.
- Linke W. E., 1965. Solubilities of inorganic and metal-organic compounds. Vol. II. 4th ed. Amer. Chemical Soc., Washington, D.C.
- Marczyński S., Jankiewicz L. S., 1978a. Uptake and translocation of labelled iodide ion in the privet (*Ligustrum vulgare* L.) as related to its defoliating activity. Acta Agrobot. 31: 11-21.
- Marczyński S., Jankiewicz L. S., 1978b. The effect of controlled temperature and humidity on the effectiveness of chemical defoliation of *Ligustrum vulgare* L. and *Spiraea bumalda* Burv. shrubs. Acta Agrobot. 31: 181-194.
- Marczyński S., Tukey H. B., Jr, 1986. Effect of temperature and light on foliar absorption of P and Rb by Chrysanthemum and Pilea. Acta Agrobot. 39: 235-247.
- Middleton L. J., Sanderson J., 1965. The uptake of inorganic ions by plant leaves. J. Exp. Bot. 16: 197-215.

- Morton H. L., 1966. Influence of temperature and humidity on foliar absorption, translocation and metabolism of 2, 4, 5-T mesquite seedlings. Weeds 14: 136-141.
- Oland K., Opland T. B., 1956. Uptake of magnesium by apple leaves. Physiol. Plant. 9: 401-411.
- Pallas J. E., Jr., Williams G. G., 1962. Foliar absorption and translocation of \$2P and 2,4-D as affected by soil. Bot. Gaz. 123: 175-180.
- Prasad R., Foy C. L., Crafts A. S., 1962. Role of relative humidity and solution additives on the foliar absorption and translocation of radio-labeled 2,2-dichloro propionic acid (Dalapon). Pl. Physiol. (Suppl.). 377 XIII.
- Prasad R., Foy C. L., Crafts A. S., 1967. Effects of relative humidity on absorption and translocation of foliarly applied Dalapon. Weeds 15: 149-156.
- Reed D. W., Tukey, Jr. H. B., 1978. Effect of pH on foliar absorption of phosphorus compounds by *Chrysanthemum*. J. Amer. Soc. Hort. Sci. 103: 337-340.
- Sachs R. M., Shia Y., Maire R. G., 1967. Penetration, translocation and metabolism of ¹⁴C-Alar (B-9), a plant growth retardant. Pl. Physiol. 42: 50.
- Schönherr J., 1976. Water permeability of isolated cuticle membranes: The effect of pH and cations on diffusion, hydrodynamic permeability and size of pores in the cutin matrix. Planta 128: 113-126.
- Schrept E., 1965. Licht-under elektronenmikroskopische beobachtungen and der trichom-hydrathoden von Cicer arietinum. Z. Pflanzenphysiol. 52: 245-254.
- Steel G. D., Torrie J. H., 1960. Principles and procedures of statistics. McCraw-Hill Book Co., Inc. New York.
- Teubner F. G., Wittwer S. H., Long W. G., Tukey H. B., 1957. Some factors affecting absorption and transport of foliar applied nutrients as revealed by radioactive isotopes. Mich. Ag. Exp. St. Quart. Bul. 39: 398-415.
- Thorne G., 1958. Factors affecting uptake of radioactive phosphorus by leaves and its translocation to other parts of the plant. Annals of Botany 22: 382-398.
- Thompson A. H., Rogers B. L., Harley C. P., 1958. Some factors affecting chemical-thinning of Golden Delicious apples with naphthaleneacetic acid plus Tween 20. Proc. Amer. Soc. Hort. Sci. 72: 45-51.
- Westwood M. N., Batjer L. P., 1960. Effects of environment and chemical additives on absorption of naphthaleneacetic acid by apple leaves. Proc. Amer. Soc. Hort. Sci. 76: 16-29.
- Ziegler H., Lüttge V., 1968. The exudation of ¹⁴C-Flurenol by guttation. Z. Pflanzenhr. Pflanzenpathol. Pflanzenschutzber. 64: 115.

Wpływ względnej wilgotności powietrza na wnikanie P i Rb do liści Chrysanthemum i Pilea

Streszczenie

Wilgotność powietrza w różnym stopniu wpływała na wnikanie ³²P i ⁸⁶Rb do liści *Chrysanthemum* i *Pilea* cadierei. Najczęściej wnikanie zwiększało się, gdy względna wilgotność powietrza wzrastała z 47 do 80%. Jeśli nie wystąpiła gutacja dalsze podnoszenie względnej wilgotności powietrza, do 92%, powodowało zwiększanie pobierania Rb i P. Wilgotność powietrza wpływała najsilniej na pobieranie fosforanu wapnia, przypuszczalnie w związku z jego małą rozpuszczalnością.