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TITLE: β -lactoglobulin as a molecular carrier of linoleate: characterisation and effects on intestinal epithelial cells in vitro

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11

25 **ABSTRACT**

26

27 The dairy protein β -lactoglobulin (β lg) is known to bind hydrophobic ligands
28 such as fatty acids. In the present work, we investigated the biological activity *in vitro*
29 of linoleate once complexed to bovine β lg. Binding of linoleate (C18:2) to bovine β lg
30 was achieved by heating at 60 °C for 30 min at pH 7.4, resulting in a linoleate/ β lg
31 molar binding stoichiometry of 1.1, 2.1 and 3.4. Two types of binding sites were
32 determined by ITC titrations. Binding of linoleate induced the formation of covalent
33 dimers and trimers of β lg. The LD₅₀ on Caco-2 cells after 24 hours was 58 μ M
34 linoleate. However cell viability was unaffected when 200 μ M linoleate was presented
35 to the Caco-2 cells as part of the β lg complex. The Caco-2 cells did not increase
36 mRNA transcript levels of long chain fatty acid transport genes, *FATP4* and *FABPpm*,
37 or increase levels of the cAMP signal, in response to the presence of 50 μ M linoleate
38 alone or as part of the β lg complex. Therefore, it is proposed that β lg can act as a
39 molecular carrier and alter the bioaccessibility of linoleate/linoleic acid.

40

41 Key words: β -lactoglobulin; Sodium Linoleate/Linoleic Acid; Stoichiometry; Caco-2;
42 Cytotoxicity

43

44 INTRODUCTION

45

46 Linoleic acid (LA, *cis,cis*-9,12-octadecadienoic acid, n-6, 18:2) is an essential
47 long-chain fatty acid (LCFA). World Health Organization (WHO) and Food and
48 Agriculture Organization of the United Nations (FAO) recommend an adequate intake
49 of LA of 2 % of total energy.¹ LA is a precursor to long chain metabolites such as γ -
50 linolenic acid, arachidonic acid and eicosapentaenoic acid.^{2,3} Some of these
51 polyunsaturated fatty acids such as LA or γ -linolenic acid have been shown to have
52 anti-inflammatory properties.⁴ Replacement of saturated fat with LA is advised to
53 improve serum lipoprotein profiles and reduce the risk of developing cardiovascular
54 coronary artery disease.⁵ LA is also cytotoxic at high concentrations to cancerous cells
55 *in vitro*.^{6,7} LCFA are taken up by intestinal epithelial cells by both active transport via
56 specific fatty acid (FA) transporters and passive diffusion.⁸ However, uptake and
57 bioavailability of fatty acids may be altered depending on the food matrix.^{9,10}

58 Bovine β -lactoglobulin (β lg) is the major whey protein in bovine milk but
59 absent in human milk. It is a globular protein with a monomeric molecular weight of
60 18.4 kDa, consisting of 162 amino acids.¹¹ Despite intensive studies on biological,
61 chemical and physical properties of this protein, its biological function still remains
62 unknown.¹¹⁻¹⁴ Structurally, β lg belongs to the lipocalin family¹⁵, of which most are
63 able to bind small hydrophobic molecules, such as FA, hydrophobic vitamins or
64 curcumin.¹¹⁻¹⁹ β lg may be involved in transport, through the gastric tract, of
65 hydrophobic substances naturally present in bovine milk, though clear evidence for
66 this is lacking.^{13,14} All members of the lipocalin family contain a β -barrel, shaped into
67 a flattened calyx, composed of eight antiparallel β -strands. It has been suggested that
68 β lg binds hydrophobic ligands in its internal calyx.^{20,21} The existence of binding sites
69 in a crevice near the α -helix on the external surface of the β -barrel has also been
70 reported.^{11,22-26} However, the binding sites and stoichiometry of several ligands have
71 been controversial.^{12,21,25}

72 β lg and LA are derived from food sources, β lg from milk and LA from many
73 edible oils and fats, such as safflower oil, grape seed oil or corn oil and indeed milk
74 fat.²⁷⁻²⁹ On a daily basis, people in the Western world consume significant quantities
75 of both β lg and LA. Little information is known on the interaction between the water-
76 soluble form of LA, linoleate, and β lg. Indeed how this interaction impacts on protein
77 structure and on LA biological properties. This study investigated the β lg-linoleate

78 complex formation in aqueous solution, the binding properties and the effect on
79 protein structure. To investigate the bioavailability of the FA in the formed
80 complexes, compared to FA alone, cytotoxicity was measured on intestinal epithelial
81 cells *in vitro*. The active transport of FA in the cells was studied by two different
82 methods. Intracellular cyclic adenosine 3',5'-monophosphate (cAMP) levels in viable
83 Caco-2 cells were measured in the presence of linoleate alone or within β lg-linoleate
84 complex, as an indication of active FA transport using cAMP signal transduction.
85 Messenger RNA transcript levels of the LCFA transporter genes, Fatty Acid Binding
86 Protein (*FABPpm*) and Fatty Acid Transport Protein 4 (*FATP4*), in the presence of
87 linoleate alone or within β lg-linoleate complex were also measured.

88

89 **MATERIALS AND METHODS**

90

91 *Materials*

92 β -lactoglobulin (96 % purity) was obtained from Davisco Foods International,
93 Inc. (Eden Prairie, Minnesota) and sodium linoleate (purity \geq 98 %) from Sigma-
94 Aldrich (St. Louis, MO). All other chemicals and solutions were purchased from
95 Sigma-Aldrich unless stated otherwise.

96

97 *Isothermal Titration Calorimetry (ITC)*

98 ITC was used to determine the interaction parameters between β lg and
99 linoleate. ITC experiments were performed on a VP-ITC microcalorimeter (Microcal,
100 Northampton MA). Solutions of β lg (0.163 mM) and linoleate (9.64 mM) in
101 phosphate buffered saline (PBS; 0.01 M phosphate buffer, 2.7 mM KCl, 137 mM
102 NaCl, pH 7.4) were degassed under vacuum before titration experiments.
103 Measurements were performed at 60 °C. The reference cell was filled with PBS, and
104 the sample cell (1.425 mL) was filled with β lg solution. β lg was titrated at 60 °C with
105 29 successive 10 μ L injections of linoleate. The injection time was 20 s, and the time
106 between injections was fixed at 600 s to allow thermodynamic equilibrium. During
107 titrations, the solution in the sample cell was stirred at 310 rpm to ensure complete
108 mixing of the solution. The control measurement was obtained by titrating sodium
109 linoleate into the buffer. The first injection peak was ignored for the analysis. Data
110 were analysed using MicroCal ORIGIN version 7.0 provided by the manufacturer: the

111 integrated area of each peak was plotted versus the linoleate/ β lg molar ratio,
112 providing binding constants.

113

114 *Preparation of β lg-linoleate complexes*

115 β lg-linoleate complex were prepared by heating a solution of β lg and sodium
116 linoleate according to Lišková et al. (2011)³⁰ with the following modifications.
117 Briefly, 0.163 mM β lg was dissolved in Phosphate Buffer Saline (pH 7.4) and sodium
118 linoleate was added to reach final linoleate/ β lg molar ratios of 5, 7.5 and 10. Solutions
119 were heated for 30 min at 60 °C, then immediately cooled on ice. Samples were
120 extensively dialysed against distilled water prior to freeze-drying. A control of sodium
121 linoleate was dialysed using the same conditions. No FA was detectable by gas
122 chromatography in the control.

123

124 *Determination of the FA content by gas chromatography (GC)*

125 The FA content of the complexes was determined by GC following a protocol
126 adapted from Palmquist and Jenkins (2003)³¹. Briefly, the internal standard
127 tridecanoic acid (C13:0) was added to ~4 mg of complexes. FA were converted to
128 fatty acid methyl esters (FAME) by the addition of 1.5 mL 10 % methanolic HCl and
129 1 mL hexane. The samples were vortexed and heated to 90 °C for 2 h. After cooling
130 on ice, 1 mL hexane and 3 mL 10 % K₂CO₃ were added and samples were vortexed.
131 After phase separation, the heptane phase (upper phase) containing the FAME were
132 analysed as previously described by Coakley et al. (2003)³², using a CP-SELECT CB
133 column for FAME (100 m, 0.25 mm, 0,25 μ m film thickness, Varian BV,
134 Middelburg, the Netherlands), a Varian 3400 GLC (Varian, Walnut Creek, CA) and a
135 flame ionization detector .

136

137 *Gel Permeation HPLC*

138 The concentration of monomers and aggregates were determined by gel-
139 permeation-HPLC (GP-HPLC) using a TSK G SW guard column (7.5×7.5 mm,
140 Tosoh Bioscience GmbH, Stuttgart, Germany) and a TSK G2000 SW column
141 (7.5×600 mm, Tosoh Bioscience GmbH) connected to an HPLC system, consisting of
142 a Waters 2695 Separations Module, a Waters 2487 Dual λ Absorbance Detector and
143 an Empower Pro software (Waters, Milford, MA) to acquire and analyse data. 0.05
144 mg of protein was injected using a solution of 30 % acetonitrile (LabScan Analytical

145 Sciences, Dublin, Ireland) (v/v) and 0.1 % (w/v) trifluoroacetic acid in Milli-Q[®] water
146 (Millipore, Carrigtwohill, Ireland) as an eluent, at a flow rate of 0.5 mL/min. The use
147 of acetonitrile ensured that native β lg was eluted in monomeric form. The method was
148 calibrated using a set of protein molecular-weight standards (Sigma-Aldrich).

149 The proportions of monomers (including native and unfolded) of β lg were
150 deduced from GP-HPLC data by integration of the peaks area. The proportion of β lg
151 oligomers in samples was calculated by subtraction of the concentration of monomer
152 from the initial protein concentration, determined by HPLC.

153

154 *Polyacrylamide gel electrophoresis*

155 Samples were analysed by sodium dodecyl sulphate polyacrylamide gel
156 electrophoresis (SDS-PAGE) in order to determine the nature of the oligomers
157 interaction. Mini-PROTEAN TGX precast Gels (4-20 % resolving gel, Bio-Rad
158 Laboratories Inc., Hercules, CA) were used on a Mini Protean II system (Bio-Rad)
159 according to the manufacturer's instructions. Samples were prepared under reducing
160 (with β -mercaptoethanol) and non-reducing conditions. Protein was visualized by
161 staining with Coomassie blue (Bio-Safe Coomassie Stain G-250, Bio-Rad). An
162 Amersham Low Molecular Weight Calibration kit (14.4 to 97 kg/mol, GE Healthcare
163 UK Limited, UK) was used as molecular weight standards.

164

165 *Cell Culture*

166 The Caco-2 cell line was purchased from the European Collection of Cell
167 Cultures (collection reference: ECACC 86010202) and was derived from human
168 colonic adenocarcinoma cells. When fully differentiated, Caco-2 cells can mimic the
169 enterocytes of the intestine.

170 Cells cultures were maintained in a humidified 37 °C incubator with a 5 %
171 CO₂ in air atmosphere. Cells were routinely grown in 75 cm² plastic flasks in
172 Dulbecco's modified Eagle medium (DMEM) containing 4.5 g/L glucose and 0.584
173 g/L L-glutamine. Media for subculture was supplemented with 10 % (v/v) foetal
174 bovine serum (FBS), 100 U/mL penicillin and 100 mg/mL streptomycin. At 80 %
175 confluency, cells were trypsinated with 0.25 % trypsin/EDTA, diluted 1:6 in media
176 and reseeded. Media was changed three times a week. All cells used in these studies
177 were between passage number 25 and 40.

178

179 *Cytotoxicity assay*

180 Cytotoxicity of test samples on Caco-2 cell proliferation was determined by
181 MTS assay, using CellTiter 96 Aqueous One Solution Cell Proliferation Assay
182 according to the manufacturer's instructions (Promega Corporation, Madison,
183 Wisconsin). Briefly, Caco-2 cells were seeded in 96-well plates, at a cell density of
184 2×10^4 cells/well, using serum-free media (DMEM only supplemented with 100 U/mL
185 penicillin and 100 mg/mL streptomycin). After 24 h, cells were treated with different
186 concentrations of linoleate (0 to 150 μ M), β lg (0 to 150 μ M) or linoleate: β lg
187 complexes (complex with a molar ratio of 1, 2 and 3, which contained 0 to 150 μ M of
188 linoleate) in serum-free media for 24 h. The One Solution Cell Proliferation reagent
189 (20 μ L) was then added to each well for a further 3 h. Viability was defined as the
190 ratio of absorbance of treated cells to untreated cells (cells exposed to serum-free
191 Media only) at 490 nm. Each cell exposure was repeated by six and intra-plate
192 variation was accounted for by repeating the exposures on 3 different days (n=18).
193 The Lethal Dose 50 (LD₅₀) values were determined using Graph-Pad Prism software
194 3.03 (GraphPad Software Inc., La Jolla CA). The sigmoidal dose-response with
195 variable slope was used to fit the measured curves and calculate LD₅₀.

196

197 *Real-Time Cell Analyzer (RTCA)*

198 Cell growth was monitored in real time using the Real Time Cell Analyzer
199 (RTCA) SP Instrument, the xCELLigence system (Roche Diagnostic Limited, West
200 Sussex, UK). The RTCA system measures the impedance of the bottom of the well
201 which is a function of cell number and cell morphology. As the cell numbers increase
202 the impedance increases. Correspondingly if the cell morphology changes (cells swell
203 or shrink), the impedance will also be affected. The RTCA software generates a cell
204 index value based on the level of impedance. Caco-2 cells in serum-free media were
205 seeded in 16 E-Plates (Roche Diagnostic Limited) at a cell density of 1×10^4 cells/well.
206 After 24 h in a humidified 37 °C incubator with a 5 % CO₂ in air atmosphere, cells
207 were treated for 48 h with different concentration of linoleate (0 to 100 μ M), β lg (0 to
208 100 μ M) or linoleate- β lg complexes (molar ratio of 3 linoleate/ β lg containing 0 to 100
209 μ M linoleate) in serum-free media. Data were analysed using the RTCA software 1.2
210 (Roche Diagnostic): Cell Index was plotted versus time and Effective Concentration
211 50 (EC₅₀) was determined by regression analysis of the cell index data versus the

212 concentration of the compound after 48 h. Using the software, the time dependent
213 EC₅₀ curves were determined by calculating the EC₅₀ values at 20 time points within
214 48 h and these EC₅₀ values were plotted versus time.

215

216 *Cyclic AMP Assay*

217 A cAMP assay based on homogeneous time-resolved fluorescence, the cAMP
218 HiRange kit (Cisbio Bioassays, Codolet, France), was performed according to
219 manufacturer's instructions. Caco-2 cells were seeded into 96-well-half-area plates
220 (Cruinn Diagnostics, Ireland) at a density of 1×10^5 cells/well in serum-free media.
221 After an overnight incubation in a humidified 37 °C incubator with 5 % CO₂ in air
222 atmosphere, media was aspirated and 25 µL serum-free media/IBMX (3-Isobutyl-1-
223 methylxanthine) was then added to all wells and the plate was pre-incubated for 30
224 min at 37 °C, 5 % CO₂. Cells were incubated with 25 µL of linoleate (0 to 50 µM),
225 βlg (0 to 50 µM) or linoleate-βlg complexes (molar ratio of 3 linoleate/βlg containing
226 0 to 50 µM linoleate) at room temperature for 30 min with gentle shaking. Forskolin
227 (25 µL of 1 µM) was used as a positive control. Accumulation of the intracellular
228 cAMP was measured using the cAMP HiRange kit following the manufacturer's
229 instruction. Fluorescence was read using a FLUOstar Omega multi-mode microplate
230 reader and analysed with the FLUOstar Omega software (BMG LABTECH GmbH,
231 Ortenberg, Germany). Intracellular levels of cAMP (in the nM range) were
232 determined by interpolating fluorescence readings from a cAMP standard curve
233 generated in the same assay.

234

235 *Messenger RNA levels of FABPpm and FATP4*

236 For quantifying *FABPpm* and *FATP4* mRNA levels, real time PCR (RT-PCR)
237 was performed in a LightCycler 480 instrument (Roche Diagnostic Limited) based on
238 the principles of absolute quantification.

239 Caco-2 cells in serum-free media were seeded in 6-well plates at a density of
240 5.7×10^5 cells/well overnight in a humidified 37 °C incubator with 5 % CO₂ in air
241 atmosphere. Cells were then exposed to linoleate (0 to 50 µM) or linoleate-βlg
242 complexes (molar ratio of 3 linoleate/βlg containing 0 to 50 µM linoleate) for 4 h in
243 serum-free media. A control was done with βlg (0 to 50 µM). Supernatant was
244 removed and total RNA was extracted from the cells using the QIAGEN miRNeasy

245 Mini kit (QIAGEN Limited, West Sussex, UK). Quality and quantity of total RNA
246 was measured by glyoxyl gel electrophoresis and spectrophotometrically using the
247 NanoDrop 1000 (Thermo Fisher Scientific, Wilmington NC). First strand cDNA was
248 generated from 1 µg total RNA using the Bioline cDNA synthesis kit (Bioline,
249 London, UK).

250 All primers were designed across intron/exon boundaries and synthesised by
251 Eurofins MWG Operon (Ebersberg, Germany). Primers for human *FABPpm* were
252 designed using DNASTAR Lasergene 8 software (DNASTAR, Madison WI, USA) and
253 based on the GenBank sequence (accession number NM002080). *FABPpm* forward
254 primer sequence was 5'-CCGGAACAGTGGGAAGGAAATAGC-3' and the reverse
255 primer sequence was 5'-TTGAGGGGAGGGTTGGAATACAT-3'. The annealing
256 temperature for amplification was 57 °C. The forward primer sequence used for
257 human *FATP4* was 5'-CAGGGCGCCAACAACAAGAAGATT-3' and the reverse
258 primer sequence was 5'-GCAAAGCGCTCCAGGTCACAGT-3', both designed from
259 the accession number NM002080. The annealing temperature for amplification was
260 58 °C.

261 Plasmid standards for *FABPpm* and *FATP4* were created by cloning an
262 amplified PCR product into the pCR4-TOPO vector using the TOPO-TA cloning
263 system (Invitrogen, Life Technologies, Carlsbad CA) according to the manufacturers'
264 instructions. The cloned amplicon was (a) identified by PCR amplification, using gene
265 specific primer pairs, and/or digestion of plasmid DNA using the *EcoRI* restriction
266 enzyme and (b) confirmed by sequencing (Beckman Coulter Genomics, Essex, UK).
267 For RT-PCR standards, plasmid DNA was linearized and quantified using the
268 Nanodrop 1000. Standard curve preparation involved creating a series of dilutions
269 from 10⁹ to 10² copies/µL.

270 For each 10 µl Lightcycler reaction, 1 µL of test cDNA or serially-diluted
271 standard was used. The LightCycler 480 SYBR Green I Master kit (Roche
272 Diagnostics Limited) was used for quantification according to the manufacturer's
273 instructions using 0.5 µM of both the forward and reverse primers. All cDNA samples
274 were tested in duplicates. Data were analysed using the LightCycler 480 Software
275 (Roche Diagnostic Limited).

276

277 *Statistical analysis*

278 Results were compared using Minitab 15 statistical Software (Minitab
279 Limited, Coventry, UK) and the ANOVA system with a Fisher's least significant
280 difference comparison. Experiments were performed at least in triplicate.

281

282 **RESULTS**

283

284 Linoleate/ β lg complexes were produced at 60 °C. The heating condition was
285 below the denaturation temperature of β lg (~70 °C at neutral pH), in a temperature
286 range where β lg is in the R-state.^{33,34} β lg R-state is characterized by small changes in
287 β lg tertiary structure, a slight expansion in its volume and an increase in accessible
288 surface area compared to native β lg.³⁵ The salt of LA was used because it is fully
289 soluble in water up to its critical micelle concentration (CMC: 2 mM³⁶). Using the
290 water-soluble form of LA permits direct contact between β lg and linoleate allowing
291 complex formation whilst avoiding the use of ethanolic solutions to solubilise the
292 FA.³⁷

293

294 *Interaction between linoleate and β lg*

295 The thermodynamic parameters of linoleate binding to β lg were investigated
296 by ITC. The changes in the enthalpy during the binding of β lg titrated with a solution
297 of sodium linoleate was investigated at pH 7.4 and at 60 °C. The heat exchange from
298 the interaction of β lg with linoleate is shown in Figure 1. Each peak represents the
299 heat exchange within the system after an injection and indicates that linoleate-binding
300 to β lg is an exothermic process. Figure 1B depicts peak integration corrected by
301 control (titration of sodium linoleate in buffer). The energy released during the
302 titration decreased as the molar ratio increased. The curve levelled off at a
303 linoleate/ β lg ratio of ~3. The saturation point of β lg by linoleate was above a molar
304 ratio of 13. For linoleate/ β lg molar ratios of 3 to 13, the released heat did not plateau.
305 Instead the heat decreased by 1.5 kcal/mol of linoleate. The binding titration curve
306 was best fitted according to a "two set of binding sites model", which yielded the
307 thermodynamic constants for the two sites: $k_{a1} = (2.70 \pm 2.03) \times 10^5 \text{ M}^{-1}$, $n_1 = 0.62 \pm 0.004$;
308 $k_{a2} = (5.91 \pm 3.85) \times 10^3 \text{ M}^{-1}$, $n_2 = 5.75 \pm 0.51$, where k_a is the association constant and n the
309 stoichiometry. The return to thermodynamic equilibrium was very slow (>1500 s). A

310 longer equilibrium time did not affect the results (data not shown). The slow return to
311 thermodynamic equilibrium could be due to the occurrence of other structural events,
312 i.e. formation of oligomers (see below). No heat change was observed during direct
313 injection of sodium linoleate in buffer solution (control sample).

314

315 *βlg-linoleate complex formation: stoichiometry and oligomerisation of the protein*

316 βlg-linoleate complexes were prepared by heating 0.16 mM βlg at 60 °C for 30
317 min at pH 7.4 in the presence of 5, 7.5 and 10 molar equivalent of linoleate. To
318 remove excess of unbound FA, samples were extensively dialysed prior to freeze-
319 drying.

320 Final stoichiometry of linoleate/βlg was determined from freeze-dried
321 complexes using GC. A comparison of molar ratios of linoleate/βlg before the
322 reaction, after dialysis and freeze-drying is shown in Figure 2A. The amount of
323 linoleate bound to βlg increased by increasing the initial ratio of linoleate/βlg. For an
324 initial linoleate/βlg molar ratio of 5, 7.5 and 10, the amount of linoleate bound to one
325 protein after dialysis was 1.05 ± 0.10 , 2.14 ± 0.06 and 3.35 ± 0.47 moles, respectively.
326 Consequently, complexes with a linoleate/βlg molar ratio of 1, 2 and 3 were formed.

327 As shown by GP-HPLC analysis (Figure 2B and 2C), the presence of linoleate
328 induced oligomerisation of the protein. The amount of oligomers increased
329 significantly with the molar ratios of linoleate/βlg. In the absence of linoleate
330 (control), the amount of oligomers in βlg samples was 12 %. In the presence of
331 linoleate (3 linoleate/βlg), the amount of oligomers reached up to 45 % of total protein
332 concentration. The oligomers were mainly dimers and trimers of βlg as shown in
333 Figure 2C and confirmed by SDS-PAGE experiment (see below). However, no
334 significant change in the protein secondary structures was associated with oligomer
335 formation, as indicated by FTIR (data not shown).

336 These results were confirmed by SDS-PAGE. Under non-reducing conditions,
337 the SDS-PAGE analysis of native βlg and heated βlg (at 60 °C) (Figure 2C) showed a
338 major band corresponding to the βlg monomer with small amount of dimers and
339 trimers. In the presence of linoleate, the intensity of the bands corresponding to βlg
340 dimers and trimers intensified. Under reducing conditions, no difference was observed
341 between native βlg, heated βlg and linoleate/βlg complexes with a molar ratio of 1, 2
342 and 3 linoleate/βlg (Figure 2C). This indicated that dimers and trimers were
343 covalently bound by disulfide links.

344

345 *Cytotoxicity on Caco-2 cells*

346 To elucidate the effect of β lg:linoleate on human epithelial cell viability, the
347 human colonic adenocarcinoma cells, Caco-2, were exposed to β lg, linoleate or
348 complexes for 24 h (Figure 3A). LD₅₀, the concentration required to decrease the cell
349 viability by 50 %, was then calculated. β lg was not toxic to Caco-2 cells at the
350 concentrations tested (0 to 150 μ M), as measured by MTS assay. In contrast, the LD₅₀
351 of linoleate was 58.04 \pm 4.21 μ M. Linoleate: β lg complexes, where corresponding
352 linoleate concentration varied from 0 to 150 μ M, had no cytotoxic effect on Caco-2
353 cells after 24 h incubation.

354 Toxic effects of β lg, linoleate or complexes in real time over 48 h were studied
355 using RTCA (Figure 3B). The cell index, which is function of the impedance at the
356 bottom of the well, was measured. Cell index relates to cell viability and/or cell
357 morphology. Results showed a decrease of cell index by 1.95 when the concentration
358 of β lg increased from 0 to 100 μ M, compared to the control cells without compound.
359 However the parallel MTS assay showed no change in cell viability (Figure 3A).
360 Taken together, this allowed the authors to conclude that β lg alters cell morphology
361 rather than cell viability. Interestingly, at the low concentrations of 5, 10 and 25 μ M
362 linoleate, an increase of cell index was observed for linoleate alone compared to
363 control (Figure 3B-linoleate). At concentrations of 50, 75 and 100 μ M, linoleate cell
364 index decreased by 2.54 (for 100 μ M linoleate) compared to control. However, the
365 response of Caco-2 cells to β lg-linoleate complexes differed to linoleate alone at
366 equivalent molar concentrations. A decrease in cell index was observed after 3 h of
367 exposure to 100 μ M of linoleate whereas the 3 linoleate/ β lg complex (100 μ M
368 linoleate) required 12.5 h exposure to decreased the cell index (cf. arrows on Figures
369 3B-linoleate and 3B-cplx). β lg-linoleate complexes containing 5, 10, 25 and 50 μ M
370 linoleate increased Caco-2 cell index after 48 h. β lg-linoleate complexes containing
371 100 μ M of linoleate in the complex decreased the cell index by 0.49 after 48 h. EC₅₀,
372 the concentration required to obtain 50 % of the maximum effect, was calculated at
373 different time points. After 48 h, EC₅₀ was 35 μ M for linoleate alone and 98 μ M when
374 linoleate was part of a 3 linoleate/ β lg complex. A time dependent EC₅₀ was calculated
375 to indicate EC₅₀ changes as a function of time (Figure 3C). At 13 h, the EC₅₀ was
376 reached with 74 μ M linoleate alone. In contrast, it took 30 h for the EC₅₀ to reach 78
377 μ M linoleate when 3 linoleate was complexed to β lg.

378

379 *Cellular Response to β lg, linoleate and β lg-linoleate complexes*

380 Cyclic AMP is a ubiquitous intercellular/intracellular messenger which may be
381 involved in active FA uptake by cells.³⁸ Cyclic AMP levels were measured in Caco-2
382 cells exposed to linoleate, β lg, or complexes at non-toxic concentrations (50 μ M
383 linoleate). No changes were detected in cAMP levels by incubation with linoleate (0
384 to 50 μ M), β lg (0 to 50 μ M) or linoleate- β lg complexes (0 to 50 μ M linoleate) as
385 measured by a FRET-based time-resolved fluorescence assay. Messenger RNA
386 transcript levels of the FA transporter genes, *FATP4* and *FABPpm*, in Caco-2 cells
387 were investigated as an indication of active transport of linoleate across the cell
388 membrane. Messenger RNA levels of *FATP4* and *FABPpm* were not significantly
389 increased in Caco-2 cells after 4 h incubation with linoleate (0 to 50 μ M) or linoleate-
390 β lg complexes with a concentration of 0 to 50 μ M linoleate (Supplementary Figure 1).

391

392 **DISCUSSION**

393

394 β lg and linoleate formed complexes that protected Caco-2 cells from the
395 cytotoxic effects of linoleate (Figure 3). Intracellular cAMP levels (Figure 4), mRNA
396 *FATP4* and *FABPpm* levels were unaffected by the presence of linoleate either alone
397 or in a protein complex. SDS-PAGE and HPLC analysis of the complexes revealed
398 the formation of intermolecular disulfide bonds between protein molecules, which
399 increased with higher molar ratios of linoleate/ β lg (Figures 2B and 2C). These
400 observations, combined with those from ITC (Figure 1), allowed us to suggest a
401 binding mechanism between β lg and linoleate which modifies the cytotoxic effect of
402 the FA.

403

404 Linoleate interacted with β lg via two different binding sites with respective
405 affinity constants of 2.7×10^5 and 5.9×10^3 M^{-1} . These association constants are similar
406 to those reported for other hydrophobic ligands.^{39,40} Spector and Fletcher (1970)⁴⁰
407 reported two binding sites with association constant in the order of 10^5 M^{-1} and 10^3 M^{-1}
408 for the binding of β lg to palmitate, oleate, stearate and laurate. Concomitantly to
409 linoleate binding, we also observed the formation of covalent protein oligomers, i.e.
410 dimers and trimers, that could explain the stoichiometry value ($n = 0.62$) determined

411 from ITC experiments. This value could result from a mixture of complexes such as
412 $(\beta\text{lg})_2\text{-(linoleate)}_1$ ($n = 0.5$) and $(\beta\text{lg})_3\text{-(linoleate)}_2$ ($n = 0.67$). This hypothesis is
413 consistent with the simple shape of the ITC peaks, with slow return to equilibrium
414 attributed to the induced oligomerisation step. Recently, a crystallographic structure
415 of native βlg /linoleate complex, showing the fatty acid located at the protein calyx
416 (stoichiometry = 1) was published.⁴¹ Our results suggest that heating of βlg and
417 linoleate mixture may lead to the formation of other types of complexes.

418 βlg has a weak aptitude to aggregation below the temperature of
419 denaturation.⁴² At 60 °C, the presence of linoleate increased the formation of
420 disulfide-linked dimers and trimers without formation of larger aggregates. This work
421 confirms previous work in our laboratory³⁰ where binding of sodium oleate to βlg at
422 60 °C decreased monomeric βlg and increased the formation of dimers and trimers.
423 Protein aggregation in the presence of lipids has also been reported for other protein
424 systems.⁴³ βlg oligomerisation into covalent dimers and trimers may be triggered by
425 slight structural changes induced by linoleate binding to βlg monomers, as suggested
426 by previous studies.^{44,45}

427 From the presented results, we propose a hypothetical binding mechanism where
428 the interaction of the FA with the protein and the oligomerisation of βlg take place in
429 a single step:

- 430 - At 60 °C, native βlg monomers reversibly unfold to form non-native R-state
431 monomers.³³
- 432 - Negatively charged linoleate molecules interact with positively charged regions at
433 the surface of βlg monomers, as suggested from the exothermic signal of ITC
434 experiments. Consequently, the formation of a linoleate: βlg complex would
435 favour additional hydrophobic interactions between proteins.
- 436 - Two or three βlg molecules were then non covalently “cross-linked” by one or
437 two linoleate molecules bound to a high affinity binding site this favours the
438 formation of βlg oligomers by intermolecular disulfide bonds, making the
439 unfolding irreversible after cooling.

440 This assumption is in agreement with the single peak structure showing a slow return
441 to thermodynamic equilibrium of ITC results. It probably indicates that several
442 physico-chemical phenomena can contribute simultaneously to the measured signal
443 including: (i) binding of linoleate to βlg , (ii) conformational changes of βlg following

444 binding of linoleate molecules, (iii) oligomerisation of the protein β lg and (iv)
445 counterion release.

446 The protection provided by β lg:linoleate complexes to Caco-2 cells from
447 linoleate differs to β lg:Conjugated Linoleic Acid (CLA) complex. Although the
448 different CLA isomers display varying effects on biological functions,^{46,47} a 2.46
449 c9,t11-CLA/ β lg molar ratio complex resulted in a 30 % increase in cytotoxicity after
450 48 h of exposure to c9,t11-CLA at a concentration of 100 μ M, compared to c9,t11-
451 CLA alone.¹⁷ HAMLET/BAMLET (Human/Bovine Alpha-lactalbumin Made LETHal
452 to Tumor cells), a complex formed of oleic acid (OA) and α -lactalbumin (α la), is
453 more cytotoxic than OA on its own.⁴⁸ OA and LA by themselves exhibit cytotoxic
454 effects on various cell lines.^{48,49} However, binding these FA to proteins such as β lg,
455 modify their cytotoxic effect compared to the FA on its own. Indeed OA/ α la complex
456 is ~40 % more cytotoxic to human larynx carcinoma cells compared to free OA.⁴⁸
457 However, recent studies based on direct measurement of OA content in the incubation
458 mixture would argue that OA alone or involved in a complex have comparable
459 cytotoxicity effects on various cells, with the protein alone having no effect.^{30,50}
460 Frapin et al. (1993)³⁹ showed that the structural constraints imposed by the double
461 bonds of FA only weakly affects the interaction of FA with β lg. The reduced
462 cytotoxic effect observed with linoleate/ β lg complexes may relate to the solubility of
463 the FA. OA has a poor solubility in aqueous solution, its CMC is between 20 and 69
464 μ M at pH 8.3 at the temperature and salt concentration tested by Knyazeva et al.
465 (2008).⁴⁸ Therefore in the absence of protein, the amount of OA available to the cells
466 would be low. The binding of OA to β lg or other proteins such as α la (HAMLET,
467 BAMLET) increased the solubility of OA (Joseph J. Kehoe, personal communication)
468 and possibly its bioavailability. The solubility of FA increases with the number of
469 C=C double bonds in the aliphatic chain.⁵¹ Consequently, the solubility of LA (C18:2)
470 is higher than that of OA (C18:1). Under the experimental conditions used by Collin
471 et al. (2010)³⁶, sodium linoleate has a CMC of 2 mM. Hence, the binding of linoleate
472 to β lg is unlikely to alter solubility but potentially alter linoleate uptake by altering
473 levels of free FA.

474 The cytotoxicity of linoleate was concentration dependent, in agreement with
475 that observed with LA.⁷ Norman et al. (1988)⁵² showed that sodium linoleate was
476 more cytotoxic to the epithelial mouse cells, Ehrlich Ascites Tumor, than emulsified
477 LA. This effect may be explained by the higher solubility of sodium linoleate in

478 aqueous bioassays with greater access to the Caco-2 cells. Prior to uptake, LCFA
479 enters a low pH microclimate at the enterocyte surface. As this local pH is below their
480 pK_a , protonation of LCFA will occur with LCFA entering in the FA form rather than
481 the salt form.⁵³ To date, it is not well understood how intestinal cells metabolize LA.
482 A previous study showed that FA cytotoxic effect was initiated by mitochondrial
483 apoptotic pathway with cytochrome C release, indicating that uptake of LA is
484 essential for its cytotoxic effect.⁶ FA cytotoxicity may also occur by an alteration of
485 the cellular n-6 to n-3 polyunsaturated FA ratio adversely affecting membrane
486 permeability and fluidity.⁸

487 LCFA are hydrophobic and so uptake by enterocytes was thought to occur by
488 diffusion. However, recent studies suggest the involvement of a protein-transfer
489 mechanisms, with transport of LCFA reaching saturation at high concentrations in
490 Caco-2 cells.^{8,38,54} It is likely that an efficient LCFA uptake by cells requires both
491 passive and facilitated transfer, possibly using a cAMP pathway.³⁸ However, in our
492 study, no change in intracellular cAMP levels was observed by viable intestinal cells
493 exposed to different concentrations of linoleate and β lg-linoleate complexes (0 to 50
494 μ M linoleate). Nevertheless, the requirement of cAMP in a facilitated LCFA uptake
495 or metabolism is controversial and appears to depend on the FA and cell type used.
496 Bovine oocytes treated with 100 μ M LA for 6 or 24 h decreased intracellular cAMP
497 levels.⁵⁵ A perfusion of 1 mmol/L plasma of eicosapentaenoic acid (EPA; C20:5 n-3)
498 during 150 min decreased cAMP level by 0.27 nmol/g tumor on MCF-7 human breast
499 cancer xenografts perfused in situ in nude rats.⁵⁶ In contrast, C6 glioma cells
500 incubated with 100 μ M EPA for 48h, increased cAMP levels by ~250 %.⁵⁷ The FA
501 transporters, FABPpm and FATP4 have been involved in the uptake of LCFA by
502 intestinal cells.^{54,58} Messenger RNA transcript levels of *FATP4* and *FABPpm* were not
503 increased upon exposure to linoleate which suggests that either (a) there is sufficient
504 quantities of FATP4 and FABPpm transporter proteins to transport linoleate or (b)
505 these transporters are not involved in linoleate transport, at the concentrations tested.³⁹

506 This study has demonstrated that β lg can bind at least three linoleate per β lg
507 monomer at two different sets of binding sites. According to cell proliferation assays,
508 linoleate can inhibit the viability of Caco-2 cells, but β lg-linoleate complexes appear
509 to protect cells from the cytotoxicity effect of linoleate. This effect could be due to the
510 relatively high solubility of linoleate. Caco-2 exposure to linoleate or β lg-linoleate

511 complexes did not modify intracellular cAMP levels or mRNA transcript levels of the
512 LCFA transporter genes, *FABPpm* and *FATP4*.

513

514 **ABBREVIATIONS USED**

515

516 α la, α -lactalbumin; β lg, β -lactoglobulin; cAMP, cyclic adenosine 3',5'-
517 monophosphate; CLA, conjugated linoleic acid; CMC, critical micelle concentration;
518 DMEM, Dulbecco's modified Eagle medium; EC₅₀, effective concentration 50; FA,
519 fatty acids; FABPpm, fatty acid binding protein; FAME, fatty acid methyl ester;
520 FATP4, fatty acid transport protein 4; FBS, foetal bovine serum; GC, gas
521 chromatography; GP-HPLC, gel permeation high performance liquid
522 chromatography; HAMLET/BAMLET, human/bovine α -lactalbumin made lethal to
523 tumor cells; ITC, isothermal titration calorimetry; k_a , association constant; LA,
524 linoleic acid; LCFA, long chain fatty acid; LD₅₀, lethal dose 50; n, reaction
525 stoichiometry; OA, oleic acid; RTCA, real time cell analyzer; RT-PCR, real time
526 polymerase chain reaction.

527

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529

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534

535 **SUPPORTING INFORMATION AVAILABLE**

536

537 Supporting Information Available: Supplementary Figure 1, levels of *FABPpm*
538 and *FATP4* mRNA transcripts in Caco-2 cells after 4h exposure to linoleate. This
539 material is available free of charge via the Internet at <http://pubs.acs.org>.

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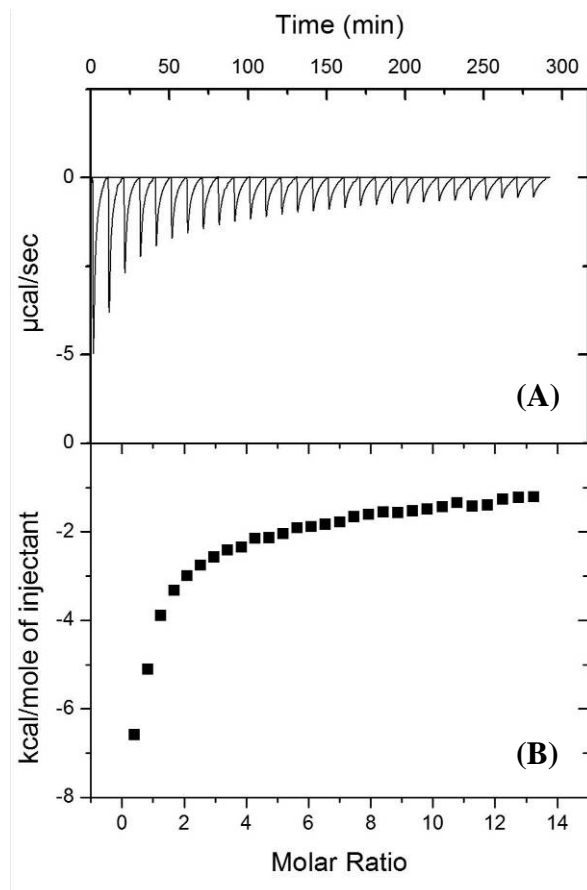
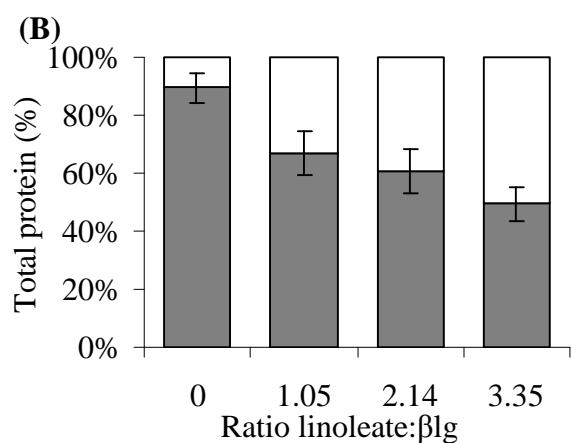
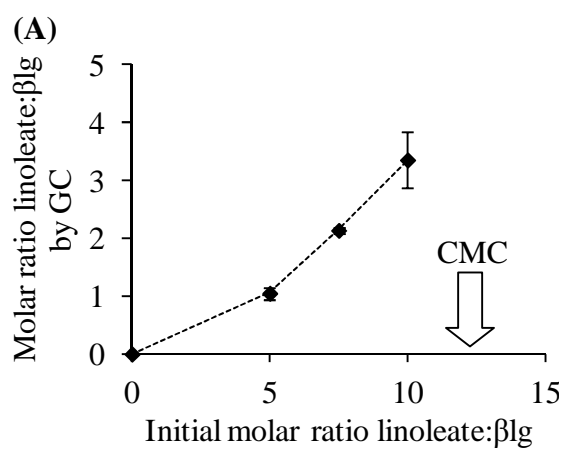


Figure 1: Microcalorimetric titration of βlg with linoleate in PBS buffer (pH 7.4) at 60 °C. (A) raw heat signal for the titration of βlg (0.16 mM) with 10 μL increments of 9.64 mM linoleate. (B) area under each peak integrated and plotted against the linoleate/ βlg molar ratio.



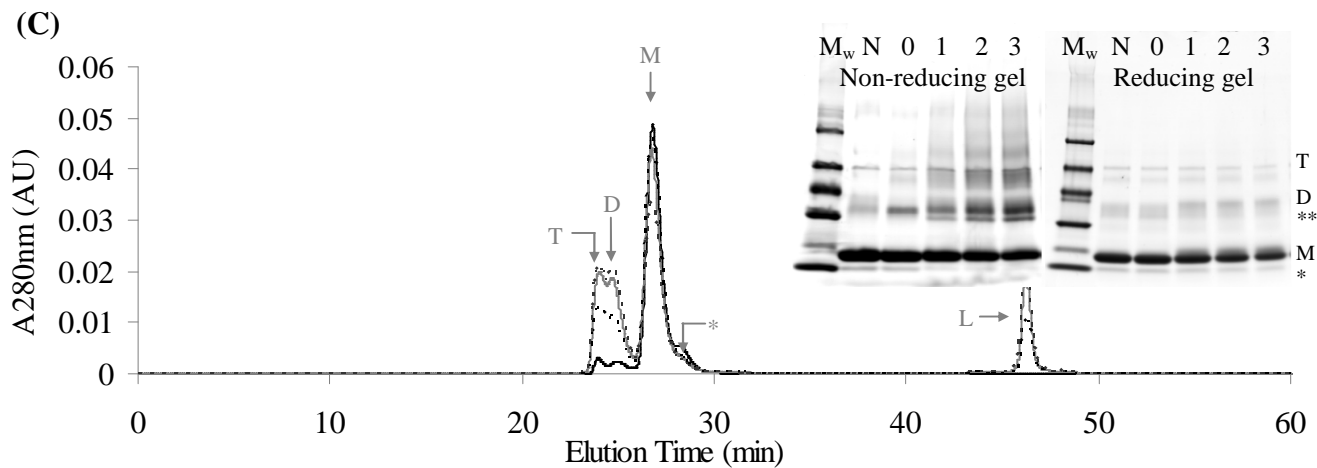


Figure 2: β lg-linoleate complexes formations: 0.16 mM β -lactoglobulin in the absence or presence of linoleate were heat treated at 60 °C for 30 min, extensively dialysed and freeze dried. (A) Correlation of the molar ratios of linoleate/ β lg added to the starting solutions with the molar ratios of linoleate/ β lg that were detected by GC analysis in the β lg/linoleate samples after extensive dialysis and freeze-drying. The arrow represent the CMC of the sodium linoleate.³⁶ Complexes with a linoleate/ β lg molar ratio of 1.05 ± 0.10 , 2.14 ± 0.06 and 3.35 ± 0.47 moles were formed, referred as complexes with a linoleate/ β lg molar ratio of 1, 2 and 3. (B) Protein composition observed by GP-HPLC (shown if Figure 2C). The molar ratio of linoleate/ β lg in dialysed solutions is indicated on the x-axis, the total protein content on the y-axis as follows: grey area, monomers; white area, oligomers. (C) Composition of solutions of β lg in presence or absence of linoleate observed by GP-HPLC chromatograms (30 % acetonitrile and 0.1 % TFA). Black line, β lg heated without sLA; dotted black line, complex of 1 linoleate/ β lg molar ratio; Light grey line, complex of 2 linoleate/ β lg molar ratio; dotted grey line, complex of 3 linoleate/ β lg molar ratio. The inserts show SDS-PAGE profile of β lg-linoleate complexes under non-reducing (left gel) and reducing (right gel) conditions. M_w , molecular weight markers (14.4, 20.1, 30, 45, 66, 97 kDa); N, native β lg; 0, β lg control treated to complex formation conditions; lane 1, 2 and 3, complex with a final molar ratio of 1, 2 and 3 linoleate/ β lg, respectively. T, β lg trimer; D, β lg dimer; M, β lg monomer; *, α la monomer; L, linoleate. A band with slightly lower molecular weight than β lg dimers was observed in the presence of linoleate under non reducing conditions (**). This maybe due to the formation of heterodimers of β lg and α la, as the electrophoretic band corresponding to α la monomers decreased in presence of linoleate.

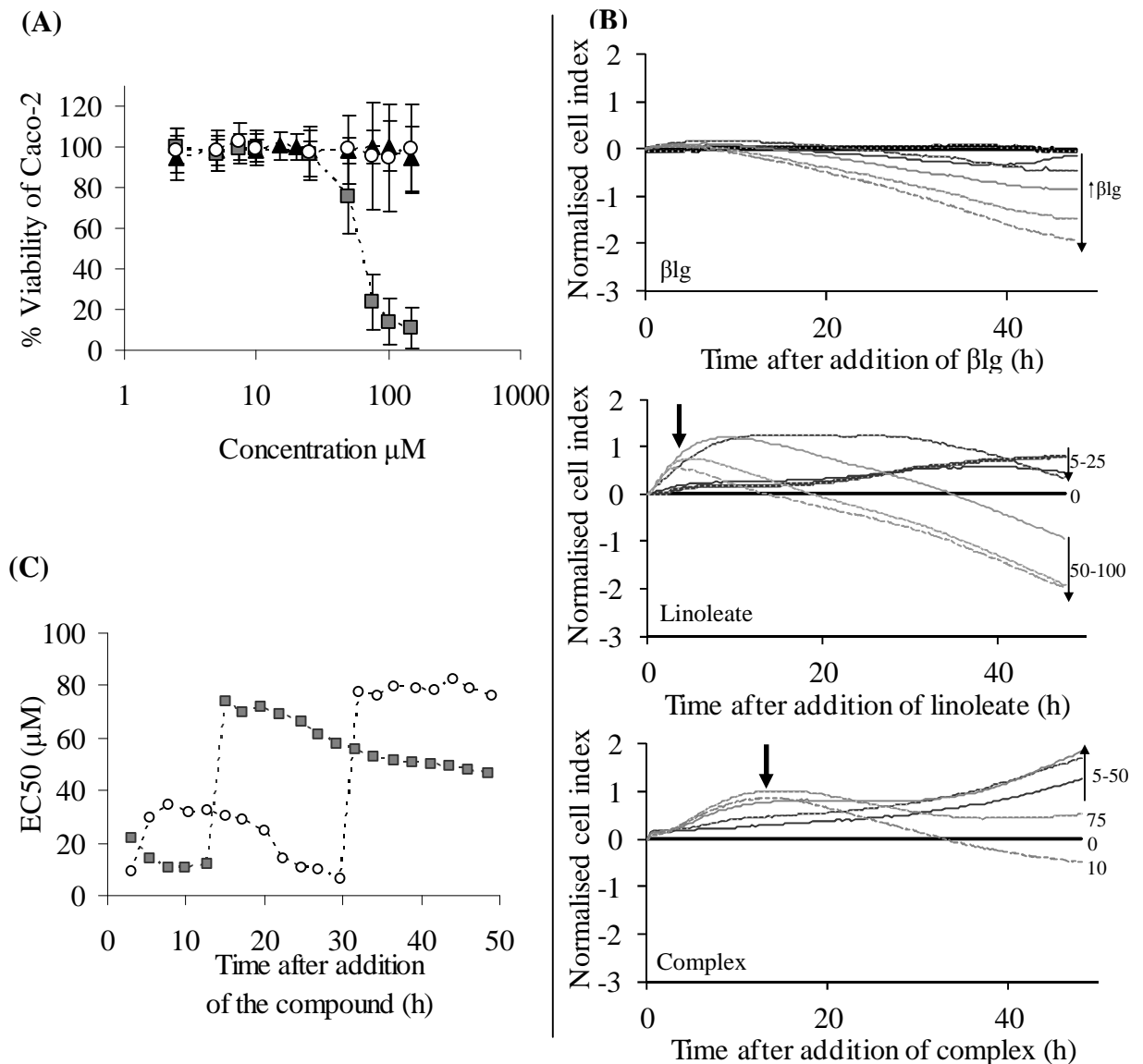


Figure 3: Cytotoxicity of βlg , linoleate and βlg -linoleate complexes on Caco-2 cells. Cytotoxicity activity was assessed using (A) MTS assay and (B, C) RTCA system. (A) % viability after 24 h on 2×10^4 Caco-2 cells compared to control cells. βlg concentration - \blacktriangle -, linoleate concentration - \blacksquare - and linoleate concentration in the βlg -linoleate complexes - \circ - are given on the x-axis (0 to 150 μM). (B) Normalised cell index (difference between the cell index and the cell index without compound) over time in hours, 1×10^4 Caco-2 cells were exposed to 0 μM —, 5 μM ~~~, 10 μM —, 25 μM ----, 50 μM —, 75 μM ----, 100 μM -- of βlg (B- βlg), linoleate (B-linoleate) and linoleate in the 3 linoleate- βlg complex (B-complex). The large arrows on B-linoleate and B-complex indicate the start of the normalised cell index decrease for the highest concentration. (C) Time dependence EC_{50} is based on the Figure B,

EC₅₀ is calculated over a 48 h exposure to linoleate -□- or 3 linoleate-βlg complex -◇-.

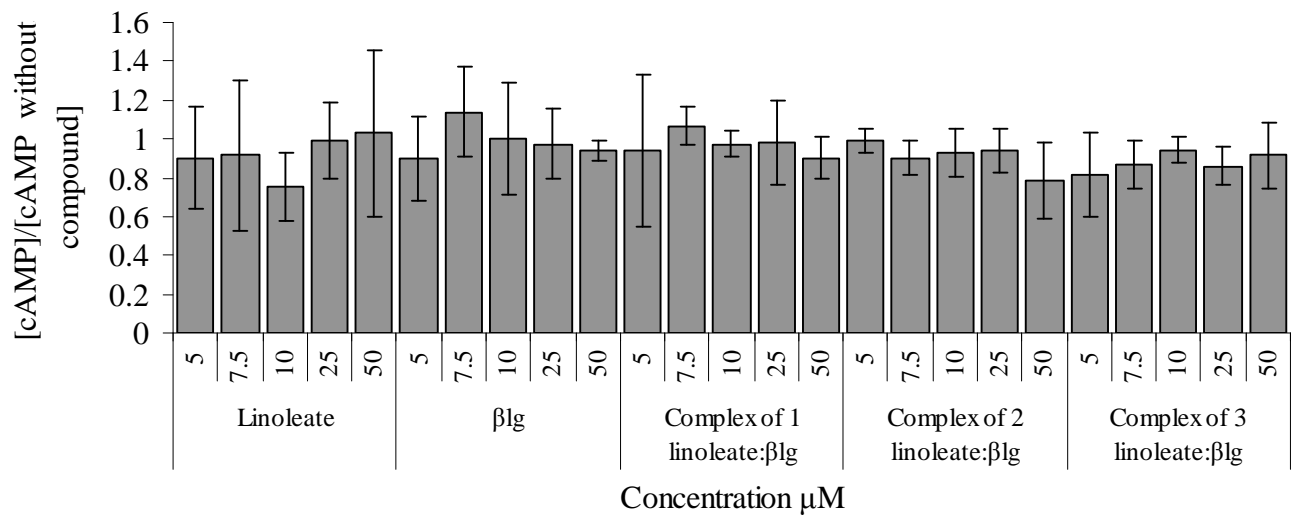


Figure 4: Relative intracellular cAMP levels in 1×10^5 Caco-2 cells treated with different concentrations of linoleate, βlg and βlg-linoleate complexes. cAMP without compound is defined as cAMP levels in Caco-2 cultured in medium without compound. No significant difference between samples and concentrations were found.

TABLE OF CONTENT GRAPHIC

