



## Research Article

# Microscale Research on Effective Geosequestration of CO<sub>2</sub> in Coal Reservoir: A Natural Analogue Study in Haishiwan Coalfield, China

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A natural analogue study in CO<sub>2</sub>-rich coalfield (Haishiwan, China) provides a strong support for safe, reliable, and long-term storage by analyzing the mechanism of CO<sub>2</sub> migration, entrapment, and storage in coal reservoir. Thus, effects of geological tectonism on reservoir properties were investigated. Simultaneously, coal and oil shale samples before and after supercritical CO<sub>2</sub> (SCCO<sub>2</sub>) treatment via geochemical reactor were collected to analyze changes in pore structure, functional group distributions, and SCCO<sub>2</sub> extraction. Observations from in situ properties of coal seam indicate that there is a positive relationship with CH<sub>4</sub> contents and F19 fault whereas CO<sub>2</sub> and carbonate contents decrease as the distance from F19 increases. Analysis of pore properties reveals that SCCO<sub>2</sub> enlarges the development of coal pore and facilitates the diffusion and seepage channel of coal reservoir, while no changes in larger pores are found in oil shale, which may restrain fluids from passing through. Then, oxygen-containing functional groups are mobilized by SCCO<sub>2</sub> from oil shale, associated with a decrease in sorption sites. The sealing capacity of cap rock (oil shale) and geological tectonism (F19 fault), as the major contributors to CO<sub>2</sub> enrichment and accumulation, provides insights into the suitable selection of CCGS site for long geological time.

## 1. Introduction

Atmospheric CO<sub>2</sub> originates mainly from fossil fuel combustion, which is responsible for about 70% of the greenhouse effect [1, 2]. China's CO<sub>2</sub> emissions in 2016 continue to be number one in the world with 27.3% of total [3]. To reduce anthropogenic CO<sub>2</sub> emissions, CCGS is put forward as a promising approach by CO<sub>2</sub> capture, transport, and injection into a storage site, such as depleted oil and gas reservoirs, saline aquifers, and unmineable coal seams [4–6]. As one of the most common potential disposal sites, CO<sub>2</sub> geological storage in coal mines has a significant meaning related to CO<sub>2</sub>-enhanced coalbed methane (ECBM) recovery [7]. However, storage in coal seam faces tremendous challenges: (1) the potential geological risks related to seismicity and

ground movement in unstable coal reservoirs; (2) the health, safety, and environmental risks causing CO<sub>2</sub> leakage to cap rock, tectonic zone, and fractures [8, 9]. Therefore, special attention should be given on the effective and safe storage of CO<sub>2</sub> in suitable sites, with the purpose of CO<sub>2</sub> leakage control, which serves as a primary risk factor.

In fact, only several pilot CCGS projects have been conducted and it is apparent that CO<sub>2</sub> storage in coal is not widely accepted in China [10]. A knowledge on CO<sub>2</sub> behavior of long-term storage is limited for the field experiments in coalfields [11]. Previous studies have been demonstrated that CO<sub>2</sub> in Yaojie coalfield can be naturally stored over long geological period under suitable conditions [12]. To some extent, the presence of CO<sub>2</sub> in Yaojie coalfield can be analogous to long-term storage of CO<sub>2</sub>, which may yield insights

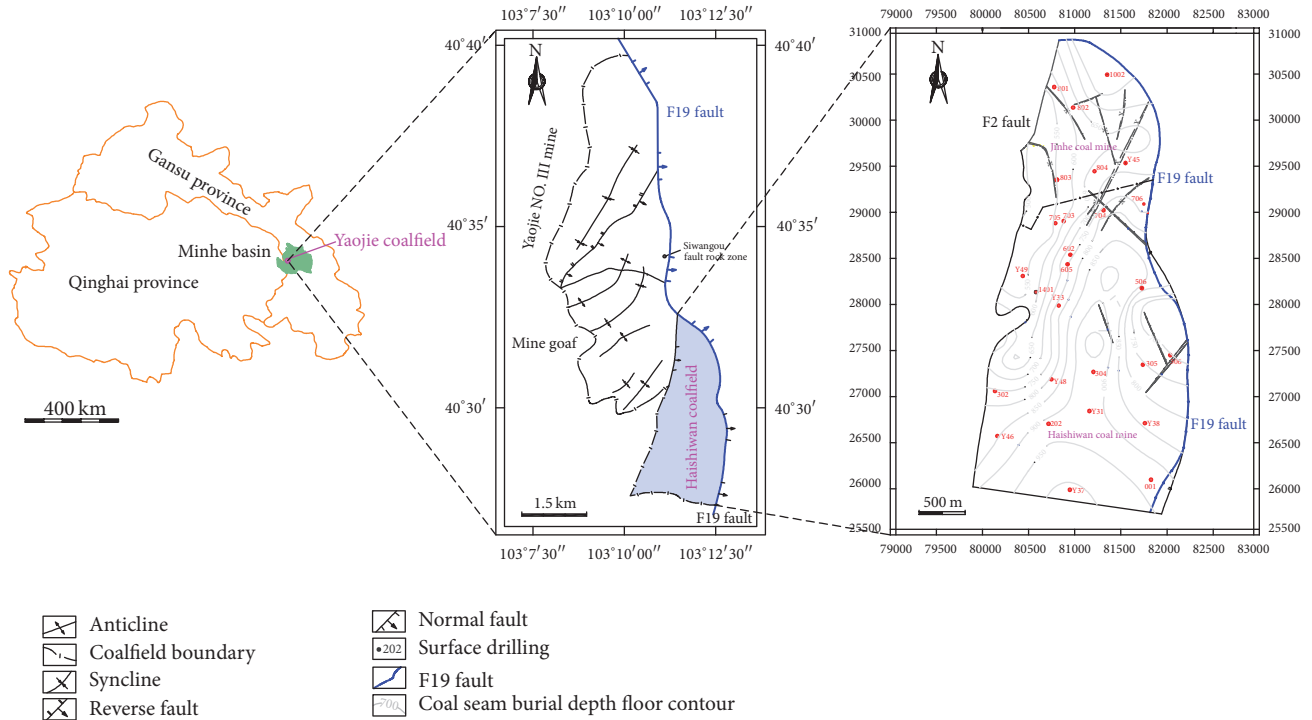


FIGURE 1: Geographical location of Yaojie coalfield and exploration of surface borehole in Haishiwan coalfield.

into the mechanisms of  $\text{CO}_2$  migration, entrapment, and storage in coal reservoirs. Therefore, the suitability of  $\text{CO}_2$  injection sites and other activities in this area need to be fully considered from the macroscopic and microscopic views.

From the macroscopic view, a group of geological factors including occurrence of coal seam, subsurface condition, cap rock types, and regional tectonic condition should be analyzed and evaluated [13–15]. The microscopic view refers to the interactions between  $\text{CO}_2$  and rocks (coal or cap rock) and the effects of  $\text{CO}_2$  on the physical (structure of pore or fracture) and chemical (molecules) properties [16]. For coal, the interactions with  $\text{CO}_2$  are generally divided into three categories: (1) swelling effect, caused by  $\text{CO}_2$  absorbing into coal matrix, leading to a decrease in the fracture aperture and permeability of coal reservoir [17, 18]; (2) the transformation and rearrangement of coal structure after  $\text{SCCO}_2$  treatment [19, 20]; (3)  $\text{SCCO}_2$  extraction of organic matter and inorganic material in coal, accompanied by the existence of water [21]. Although the above findings have been published in the literature over the past decade in the area of CCGS, a knowledge gap exists on the effects of  $\text{SCCO}_2$  on cap rocks, that is, oil shale in this study. In Haishiwan coalfield, the roof of coal seam is enclosed by a mass of oil shale, which serves as a better sealant for  $\text{SCCO}_2$  pressure [12]. Actually, cap rock, one of the most important factors, dominates the sealing capacity of coal reservoirs [22]. Hence, an effort should be made to study the changes of oil shale structure after  $\text{SCCO}_2$  treatment.

The main purpose of this study is to investigate the feasibility of effective and safe geological sequestration of  $\text{CO}_2$  in coal seams underling cap rock (oil shale), which takes Haishiwan coal seam in China with high  $\text{CO}_2$  content as

a natural analogue. A knowledge of the mechanisms of  $\text{CO}_2$  migration, entrapment, and storage in Haishiwan coalfield contributes to a better understanding of long-term storage of  $\text{CO}_2$  in the field of CCGS. Therefore, the structure in this paper is as follows. Firstly, geological factors related to tectonism and its influence on in situ properties of coal seam were investigated and analyzed. Subsequently, samples of coal and oil shale were treated by  $\text{SCCO}_2$  geochemical reactor to perform comparative tests, including pore properties, functional group distributions, and Gas Chromatography-Mass Spectrum (GC-MS) analysis. Then, the mechanisms of  $\text{CO}_2$  migration, entrapment, and storage were discussed to demonstrate how to effectively and safely store  $\text{CO}_2$  in coalfield.

## 2. Regional Geology

**2.1. Geological Setting of Yaojie Coalfield.** As shown in Figure 1, Yaojie coalfield, as a secondary tectonic unit in Minhe basin, is located in the west margin of Gansu and Qinghai provinces. There are many well-developed faults and constructions in Minhe, which is a Mesozoic–Cenozoic mountain depression basin expanded by the Qilian orogenic belt. The tectonic evolution of Minhe basin is composed of Indosinian movement, Yanshan movement, and Himalayan movement, resulting in the basin formation and transformation. Yaojie coalfield belongs to middle Jurassic coal-bearing series in Mesozoic. The strike, dip, and area are 2.6 km, 11 km, and 28.6 km<sup>2</sup>, respectively. The coal bearing basin extended in a NNW direction with the F19 fault zone located to the east of it. Yaojie coalfield, belonging to the Yaojie Coal-Electricity Corporation, consists of Zhangergou coal mine, Yaojie NO. III coal mine, Jinhe coal mine, and Haishiwan coal

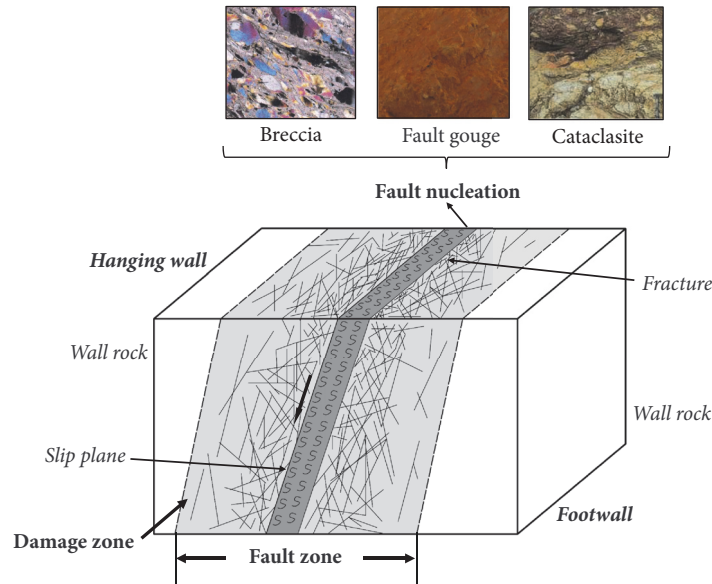


FIGURE 2: Structure diagram of ductile shear fault.

mine. Haishiwan coalfield, as a concealed coalfield, is located in the south of Yaojie coal fields in Minhe basin. The main coal seam containing  $\text{CO}_2$  is No. 2 coal seam; the average thickness of which is 20 m with thicker in the west and thin in the east. The roof of No. 2 coal seam is carbon mudstone, siltstone, and oil shale. The floor of No. 2 coal seam is carbon mudstone.

**2.2. F19 Faults.** There is no direct evidence exactly to show the formation time of  $\text{CO}_2$  fluids were affected by F19 faults, but the isotopic age chlorite thermal alteration and the fission track age of the calcite veins of ultrabasic have proved that  $\text{CO}_2$  hydrothermal activity was concentrated in this period [12]. In Neogene period with Himalayan movement, F19 fault was a tensional activity period, the permeability and fluid potential of which increased with the possibility of the upward migration of  $\text{CO}_2$  fluids [23]. Therefore,  $\text{CO}_2$  fluids were mainly formed in Neogene period. Whereafter, fault tensional activity quitted or transformed into pressure shear state, making the vertical migration of  $\text{CO}_2$  fluids slow down to stagnation. Up to the present, the vertical closure and principal compressive stress of NE direction suggest F19 fault has been still in a closed state. For this area, ductile shear fault is classified as damage zone and fault nucleation zone, which is presented in Figure 2. The fault nucleation was filled with breccia, fault gouge, and cataclasite, and the old faults have been blocked by quartz, calcite, and zeolite, which could result in a poor permeability in F19 fault. However, there are amount of minor faults and fractures in damage zone, indicating that good permeability in this area may be the main lateral migration channel of  $\text{CO}_2$  fluids.

**2.3. Sources of  $\text{CO}_2$ .** It has been proven that  $\text{CO}_2$  of Earth's crust derived from stratal carbonate pyrolysis generating inorganic  $\text{CO}_2$ , which are influenced by contact metamorphism of carbonates and dynamic metamorphism of fault activities [24]. According to geological survey in Geological

Institute (Gansu, China), magmatic intrusion has never happened in the region since the Paleozoic [25]. Therefore, the possibility of contact metamorphism can be excluded during inorganic  $\text{CO}_2$  generating process. However, due to the existence of F19 fault, the fault zone tends to suffer intense extrusion and shear, leading to the hydrothermal activities with structural and tectonic characteristics [26]. Given that carbonate rocks in active fault zone may result in dynamometamorphism under the effect of ductile-brittle shearing action, element differentiation may occur and  $\text{CO}_2$  will be generated after migration [27]. This is the main formation mechanism of inorganic  $\text{CO}_2$  in this region.

### 3. Experimental Section

**3.1. Sample Preparation.** Coal samples were sampled from the air-return way of coalface in Haishiwan coal mine, while oil shale samples were derived from the roof of No. 2 coal seam. The samples were immediately sealed and sent to the laboratory as soon as possible to prevent oxidation. Subsequently, the samples were crushed and screened to the appropriate quantity and sizes for each test. Before tests, samples were dried at  $80^\circ\text{C}$  for 48 h using turbo molecular pump vacuum.

Following China National Standard GB/T 212-2008, proximate analyses of moisture, ash, and volatile matter were performed using a 5E-MAG6600 proximate analyzer (Changsha Kaiyuan Instruments, China). The mean maximum reflectance of vitrinite ( $R_{o,max}$ , %) was determined using a Zeiss microscope-photometer (German) following China National Standard GB/T 6948-2008. The basic properties of the coal and oil shale samples are listed in Table 1.

**3.2.  $\text{SCCO}_2$  High-Pressure Geochemical Reactor.** To better understand the changes in physical and chemical structure, the samples (coal and oil shale) before and after  $\text{ScCO}_2$  treatment were conducted by  $\text{SCCO}_2$  high-pressure geochemical

TABLE 1: Basic properties of the coal and oil shale samples.

Sample	$M_{ad}$ (%)	$A_d$ (%)	$V_{ad}$ (%)	$R_{o,max}$ (%)
Coal	1.84	5.91	28.45	0.96
Oil shale	1.31	5.51	26.38	0.62

Note.  $M_{ad}$  means the moisture content (air-dried basis),  $A_d$  means the ash content (dried basis),  $V_{ad}$  means the volatile matter content (air-dried basis), and  $R_{o,max}$  means the maximum vitrinite reflectance.

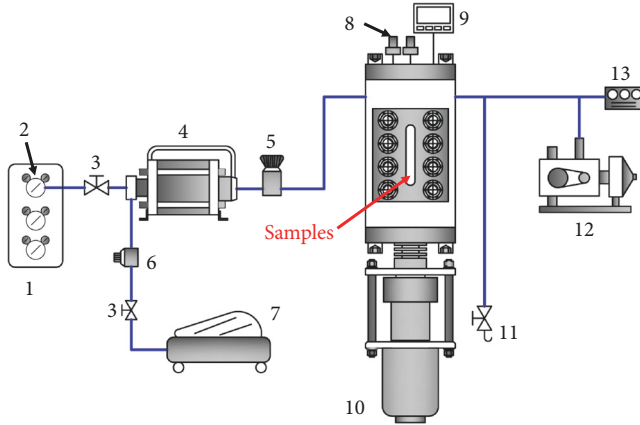


FIGURE 3: Schematic of SCCO<sub>2</sub> high-pressure geochemical reactor simulating CO<sub>2</sub> injection. (1) Gas supply; (2) reducing valve; (3) valve; (4) gas booster pump; (5) reducing valve; (6) pressure regulating valve; (7) air compressor; (8) temperature and pressure transducers; (9) temperature and pressure indicator; (10) SCCO<sub>2</sub> high-pressure geochemical reactor; (11) safety valve; (12) vacuum pump; (13) vacuum gauge.

reactor, which can simulate CO<sub>2</sub> injection in the reservoir (Figure 3). Firstly, samples were degassed in the vacuum pump system (60°C and 4 Pa for 48 h). Secondly, CO<sub>2</sub> was injected to pipeline while air compressor started to provide power. 10 min later, gas booster pump was opened and then high-pressure CO<sub>2</sub> was filled in the reactor. Eventually, the pressure and temperature of reaction were set at 10 MPa and 50°C, respectively, and the reaction time is 360 h. Actually, This condition could not lead to chemical degradation of the coal matrix [28]. To some degree, the pressure-temperature conditions in this study were used to simulate CO<sub>2</sub> sequestration scenarios in coal beds (1 km depth), though it could not be representative of CO<sub>2</sub> sequestration in deep coal beds based on field [28].

**3.3. Experimental Methods.** Pore structure analysis of macropore was performed using A PoreMaster 33 automated mercury intrusion porosimeter (Quantachrome Instruments, United States), which could effectively measure pore diameters of 6 to 100 μm over a pressure range from 0.14 to 215 MPa. The mesopore and micropore analysis were obtained using an AUTOSORB-1 (Quantachrome Instruments, United States). With these two methods, specific surface area (SSA), total pore volume (TPV), pore size distributions (PSDs), and porosity of micropores, mesopores, and macropores can be precisely analysed.

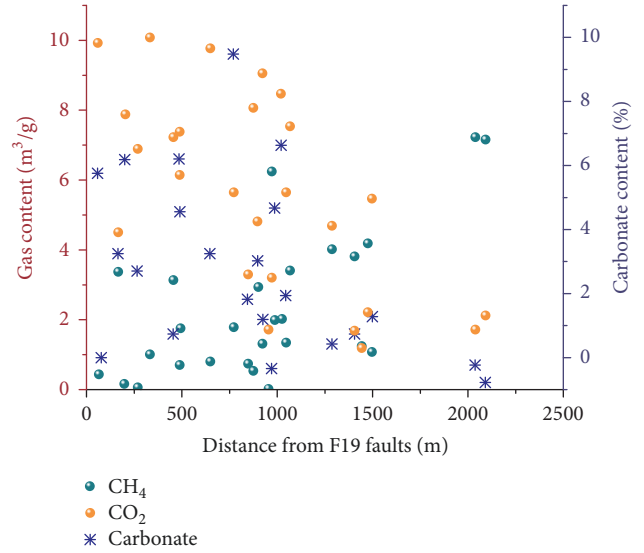


FIGURE 4: Changes of CO<sub>2</sub> and carbonate distributions as the distance from F19 increases.

GC-MS analysis was conducted by a Hewlett-Packard 6890/5973 GC/MS associated with a capillary column applied with HP-5MS (cross-link 5% PH MEsiloxane, 30 m × 0.25 mm i.d., 0.25 mm film thickness), a quadrupole analyzer, and operated in electron-impact (70 eV) mode. The data can be performed automatically by Chemstation software. The compounds were characterized by comparing mass spectra according to NIST05 library data.

FTIR spectra of the samples were measured using a device that combined a VERTEX 80 Fourier transform infrared spectrometer with a HYPERION 2000 infrared microscope (Bruker, Germany) at a resolution of 0.06 cm<sup>-1</sup> and spectral region of 8000–350 cm<sup>-1</sup>.

## 4. Results

**4.1. Relationships with F19 Fault, CO<sub>2</sub>, and Carbonate Distributions.** Figure 1 exhibits the locations of 28 surface boreholes for the No. 2 coal seam during geological exploration period in Haishiwan coalfield. Gas (CO<sub>2</sub>, CH<sub>4</sub>) and carbonate mineral contents were obtained from surface borehole coal samples, which were shown in Figure 4.

The on-site CO<sub>2</sub> contents are between 0.3 m<sup>3</sup>/g and 10 m<sup>3</sup>/g, whereas CH<sub>4</sub> contents are between 0.03 m<sup>3</sup>/g and 7.23 m<sup>3</sup>/g. Then, a negative correlation was found between CO<sub>2</sub> contents and the distance from F19 faults, while it seems



TABLE 2: Pore structure analyses of the oil shale and coal sample.

Sample	Measurement of micropores with CO <sub>2</sub> adsorption		Measurement of mesopores with N <sub>2</sub> adsorption		Measurement of macropores with MIP		Porosity (%)
	DFT-TPV (mL/g)	DFT-SSA (m <sup>2</sup> /g)	BJH-TPV (mL/g)	BET-SSA (m <sup>2</sup> /g)	MIP-TPV (mL/g)	MIP-SSA (m <sup>2</sup> /g)	
Oil shale	0.023	89.17	0.0018	3.43	0.0076	1.95	2.31
Coal	0.076	187.51	0.0054	6.92	0.017	2.23	5.82

Note. The sample weight is on an air-dried basis. SSA refers to specific surface area; TPV refers to total pore volume.

to be a positive correlation between CH<sub>4</sub> contents and F19 faults. Because CO<sub>2</sub> molecule has stronger capacity to diffuse and adsorb than CH<sub>4</sub> molecule, the highest content of CO<sub>2</sub> may be found near F19 faults. That is, it can be regarded as the input of CO<sub>2</sub>. It is summarized that CH<sub>4</sub> in the coal seam will be recovered after CO<sub>2</sub> sequestration through CO<sub>2</sub> injection.

The distribution of carbonate mineral ranges from 0.2% to 9.7%. The data in Figure 4 suggests that carbonate contents decrease with increasing distance from F19 fault. According to the literature, carbonate, as a tracing mineral of inorganic CO<sub>2</sub> migration, is related to the existence of water, pH value, pressure, and former mineral [29]. The distribution of carbonate mineral indicates that under a certain condition CO<sub>2</sub> is likely to transform into carbonate minerals near the F19 fault zone.

**4.2. Pore Characteristics.** Studies on the pore structure of oil shales and coals indicate the development and connectivity of smaller pores (micropores) and larger pores (macropores) directly relate to the desorption, diffusion, and seepage, respectively [30, 31]. The International Union of Pure and Applied Chemistry (IUPAC) classified pores into micropores (<2 nm), mesopores (2–50 nm), and macropores (>50 nm), which is widely recognized by the scholars [32].

For mesopores, N<sub>2</sub> adsorption at 77 K can be regarded as the standard method to determine TPV and SSA using the Barrett–Joyner–Halenda (BJH) and Brunauer–Emmett–Teller (BET) models, respectively [32–34]. For micropores, CO<sub>2</sub> adsorption at 273 K may overcome the disadvantages that N<sub>2</sub> adsorption is inaccessible to ultramicropores (<0.7 nm), which can be interpreted by the density functional theory (DFT) analyses for TPV and SSA [35, 36]. In this study, physisorption method (N<sub>2</sub>/CO<sub>2</sub> adsorption) can be automatically analyzed by the ASiQwin computer software from Quantachrome (United States). However, physisorption method is not suitable for larger macropores (>300 nm) analysis [37]. Thus, mercury intrusion porosimetry (MIP) is more effective for TPV and SSA of macropores through PoreMaster from Quantachrome (United States).

As shown in Table 2, in general, micropores volume and surface area of oil shale and coal samples occupy a large proportion in total pore volume. However, there is less development of macropores, mesopores, and micropores in oil shale, compared to coal sample. Especially for MIP, the porosity of coal is 5.82%, which seems to be more than twice as large as oil shale of 2.31%. It may indicate that

coal has a better migration channel than oil shale, due to a more complex and developed pore structure. Also, the SSA of micropores dominate in coal, compared with oil shales, suggesting that the most fluid molecules (CH<sub>4</sub> or CO<sub>2</sub>) can be inclined to adsorb and store in micropores.

Generally, the PSD of micropore and lower pore size range of mesopore can be characterized by N<sub>2</sub> adsorption using DFT method [38, 39]. MIP with a function of Incremental pore volume and pore diameter can reflect the PSD of larger pore size range mesopore and macropore. These methods show a good representation from micropore to macropore, which almost cover the whole range of PSDs.

Figure 5 depicts the PSDs of untreated and SCCO<sub>2</sub>-treated oil shale. No obvious changes were observed in the PSD of MIP, especially in the PSD of macropores. This indicates that SCCO<sub>2</sub> has little influence on the larger pores (mesopore and macropore) of oil shale. The inset in Figure 5 highlights that smaller pores (micropore and mesopore) decrease obviously after SCCO<sub>2</sub> treatment, suggesting that SCCO<sub>2</sub> may alter the inherent structure of oil shale by reducing the TPV [40]. The decrease in micropores and mesopores may be attributed to the adsorption-induced deformation in pores after SCCO<sub>2</sub> treatment [41]. Figure 6 shows the PSDs of coal before and after SCCO<sub>2</sub> treatment. Unlike the PSD of oil shale, the PSDs of coal in Figure 6 exhibit a bipolar distribution, which contains more micropores and macropores, and mesopores are not developed in coal. This may indicate that coal has a strong ability for gas desorption, diffusion, and seepage, compared with oil shale. Furthermore, the PSD of MIP after SCCO<sub>2</sub> treatment shows more macropores and mesopores than untreated coal sample, except for the pore diameter range around 100 nm. For DFT method, two well-defined peaks and some of disorderly peaks are found in micropores and mesopores, suggesting the well-developed micropores in coal structure and better adsorbed ability. With SCCO<sub>2</sub> treatment, micropores and mesopores show an increase in the PSDs of DFT method. Eventually, SCCO<sub>2</sub> promotes the development of micropores, mesopores, and macropores in coal, which may facilitate the movement of fluids from the coal matrix via desorption and diffusion and transport in the fracture via seepage.

The cumulative pore volume curves of the untreated and SCCO<sub>2</sub>-treated oil shale analyzed by MIP are displayed in Figure 7. In general, the concave shape is shown in the ejection curves, which may result from a mass of open pores and semiclosed pores. This may contribute to the connectivity

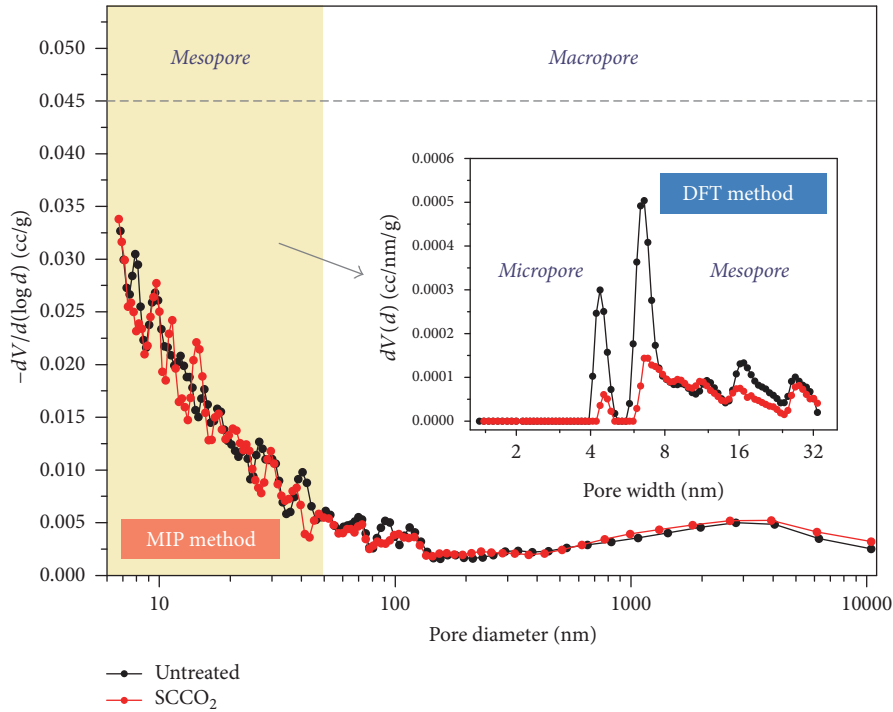


FIGURE 5: Pore size distribution of untreated and  $\text{SCCO}_2$ -treated oil shale from the MIP and DFT method.

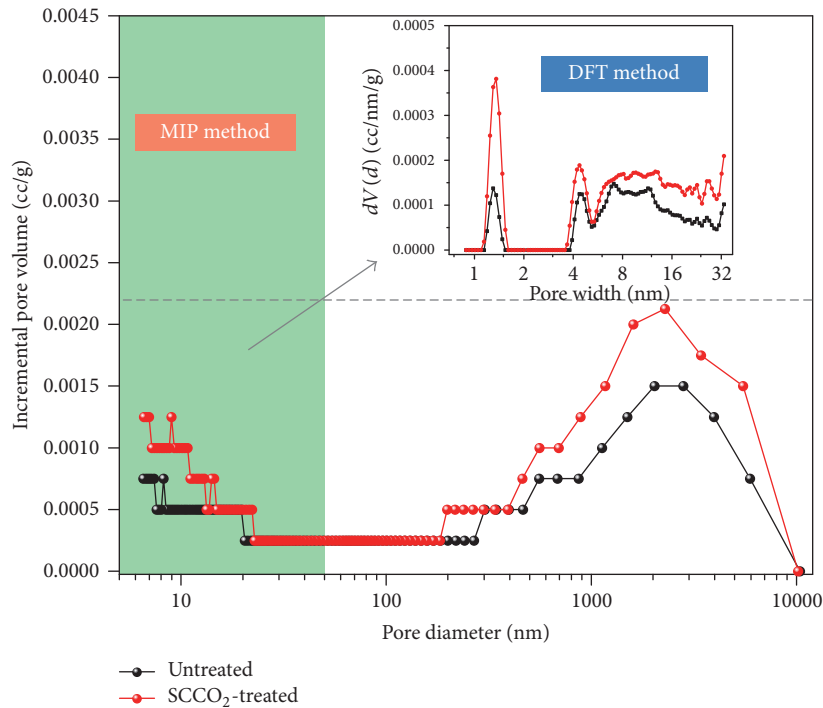


FIGURE 6: Pore size distribution of untreated and  $\text{SCCO}_2$ -treated coal from the MIP and DFT method.

of pores. Then, the injection volume of untreated and  $\text{SCCO}_2$  treated oil shale shows similar trends with the same increases, which means the same amount of mercury was injected into pores before and after  $\text{SCCO}_2$  treatment. Also, the changes in the ejection volume for oil shale show a decrease of 6.28%

(0.00175 to 0.00164 cc/g), indicating a decrease in effective pores after  $\text{SCCO}_2$  treatment. Therefore, the hysteresis loop becomes narrow with  $\text{SCCO}_2$  treatment, which may result from the decrease in semiclosed pores. The reason may be attributed to adsorption-induced deformation and physical

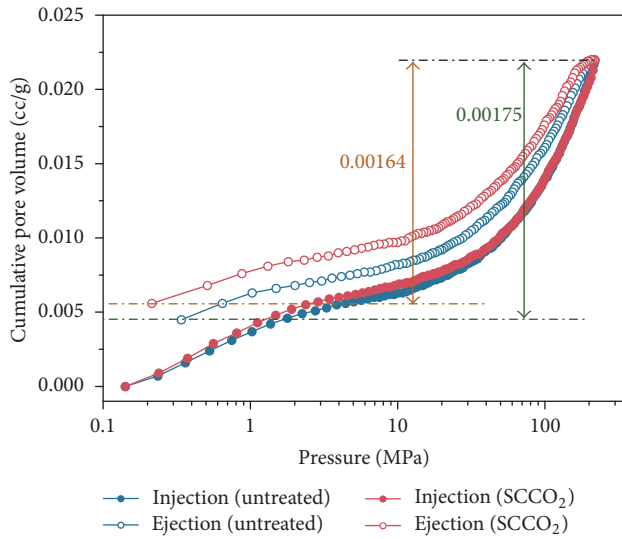


FIGURE 7: Changes in cumulative pore volume as a function of pressure for oil shale before and after SCCO<sub>2</sub> treatment by applying MIP.

constriction caused by the SCCO<sub>2</sub> dissolving and mobilizing hydrocarbons [42, 43]. Taken in conjunction with the little changes in macropores and a decrease in micropores and mesopores of oil shale as mentioned above, the results can be summarized as the decrease in the characteristics and connectivity of the pore structure related to desorption, diffusion, and seepage.

**4.3. Functional Group Distribution.** Infrared spectrum analysis has been widely used to study the macromolecular structure of the reservoir rocks, which is derived from the important information regarding the functional groups of organic compounds. Due to the specific molecular vibration spectrum of functional groups, the molecular structure of oil shale can be confirmed by the position and intensity of absorption peaks in the infrared spectrogram. To better understand the sealing ability of cap rocks, changes in the chemical structure of the original and SCCO<sub>2</sub> untreated oil shale above the coal reservoir were analyzed using FT-IR.

Generally, the infrared spectrum is split into the functional group region (spectral peak range: 4000–1500 cm<sup>-1</sup>) and the fingerprint region (spectral peak range: 1500–400 cm<sup>-1</sup>), which are shown in Figure 8. After SCCO<sub>2</sub> treatment, the absorption bands of oil shale change significantly in the region of -CH<sub>2</sub>, -CH<sub>3</sub> symmetrical or antisymmetric stretching vibrations (2935–2918, 2858–2847, and 2882–2862 cm<sup>-1</sup>), and -CH<sub>3</sub> antisymmetric deformation vibrations (1460–1435 cm<sup>-1</sup>) in naphthenic or aliphatic bonds. Also, the spectra are characterized by a broad hydroxyl region (3697–3610 cm<sup>-1</sup>), carbonyl in aldehyde, ketone, and ester (1736–1722 cm<sup>-1</sup>), C–O in epoxy compounds or ethers (1330–1060 cm<sup>-1</sup>), and bands in the carboxyl region (1715–1690 cm<sup>-1</sup>). Moreover, the spectra contain C=C stretching vibrations of aromatic or

fused rings (1635–1595 cm<sup>-1</sup>), Si–O stretching vibrations (1060–1020 cm<sup>-1</sup>), -SH with a peak at 475 cm<sup>-1</sup>, and -S–S– with a peak at 540 cm<sup>-1</sup>.

As mentioned above, the spectra, corresponding to functional groups, are grouped into three categories: (1) methyl, methylene in naphthenic, or aliphatic bonds; (2) oxygen-containing functional groups, such as hydroxyl, carbonyl, and carboxyl; (3) other functional groups. Generally, oil shale, similar to coal, has a dual pore structure that includes both matrix pores and fractures and can adsorb gas or fluids in the surface area. Thus, the first category may be thought as aliphatic and aromatic hydrocarbons in the pore structure or matrix of oil shale. The second category refers to oxygen-containing functional groups on the external surface of oil shale. The third category may be related to the silicate minerals (Si–O), organic sulfur compounds (-SH), and so on. Therefore, SCCO<sub>2</sub> may extract some hydrocarbons or other containing functional groups compounds in the matrix pores or fractures, resulting in the changes on the surface site and body structure of oil shale.

**4.4. GC-MS Analysis.** An investigation of the potential of SCCO<sub>2</sub> extraction was carried out for coal sample. The organic compounds, such as hydrocarbons or other substances, were presented clearly by the GC-MS detector. The partial results are exhibited in Figure 9.

According to the analysis of Chemstation software, SCCO<sub>2</sub> is capable of mobilizing small hydrocarbon molecules from the coal matrix. The hydrocarbons in the coal sample extract consisted largely of the small molecular weight *n*-alkanes, aromatic hydrocarbons (C<sub>6</sub>H<sub>6</sub>), and, most notably, *n*C<sub>5</sub>, *n*C<sub>6</sub>, *n*C<sub>7</sub>, *n*C<sub>8</sub>, and *n*C<sub>9</sub>. These results are consistent with the research of Kolak and Burruss [21]. That is, SCCO<sub>2</sub> mobilized comparable amounts of hydrocarbons from coal sample. According to the previous studies, the SCCO<sub>2</sub> extraction of hydrocarbons is classified as aliphatic hydrocarbons and polycyclic hydrocarbons.

It has been demonstrated that hydrocarbons, existing in the macropores, mesopores, and micropores of coal structure, may generate high activation energy barriers within the coal matrix, which may hinder gas or fluids from transporting through physical constrictions. Thus, gas molecules (such as CO<sub>2</sub>) have insufficient capacity to pass through coal beds. However, once it interacts with SCCO<sub>2</sub>, hydrocarbons and mineral matter may be extracted by SCCO<sub>2</sub>, contributing to reopen pores and fractures. Given the results of pore structure and GC-MS analysis, important information need to be put forward that SCCO<sub>2</sub> may facilitate coal permeability and fluid transport, resulting in the development of pore structure and migration channel of coal.

## 5. Discussion

**5.1. Contributions of Coalbed to CO<sub>2</sub> Migration for Geological Sequestration.** The measured CO<sub>2</sub> pressures in No. 2 coal seam are more than the critical pressure (7.38 MPa), and reservoir temperature of No. 2 coal seam is over critical temperature (34°C) according to the calculations, which have

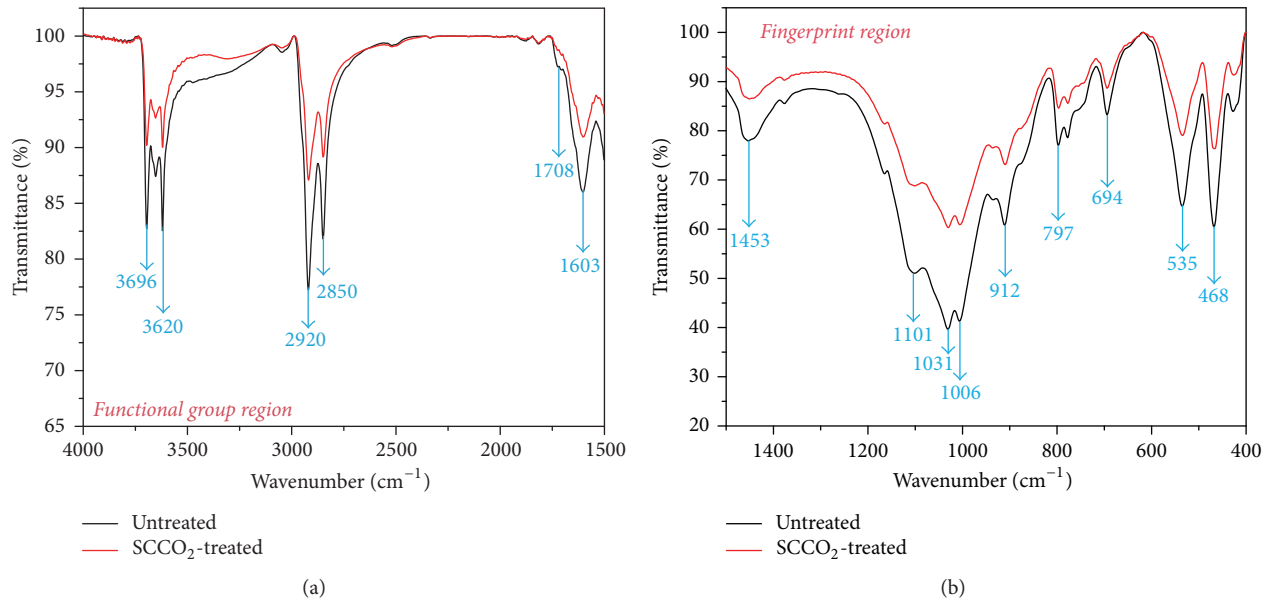


FIGURE 8: Infrared spectrogram of functional group region (4000–1500  $\text{cm}^{-1}$ ) and the fingerprint region (1500–400  $\text{cm}^{-1}$ ) for oil shale before and after  $\text{SCCO}_2$  treatment.

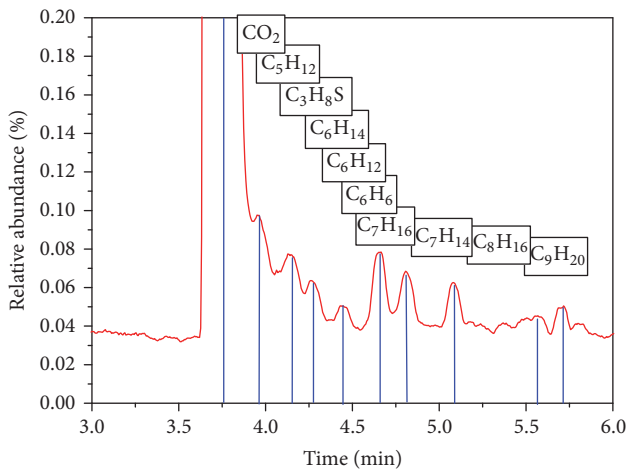


FIGURE 9: Relationship between relative abundance and retention time for  $\text{SCCO}_2$  extraction from coal through GC-MS analysis.

been demonstrated by previous studies [44]. The investigation has confirmed that  $\text{SCCO}_2$  can occur in the coal seam, even in the roof of coal seam (under cap rock) [45]. Because the source of inorganic  $\text{CO}_2$  is derived from crust and flows along F19 faults, which will be discussed in detail in Section 5.2,  $\text{CO}_2$  may migrate and accumulate to  $\text{SCCO}_2$  in the coal seam. In this process, coal fractures may initially narrow and a decrease of permeability may occur in coal reservoir. However, the results of coal pore properties and GC-MS analysis from Figures 5–9 indicate that  $\text{SCCO}_2$  enlarges the fracture width and promotes pore development by extracting hydrocarbons and mineral matter. The reasons are as follows: the migration channel, that is, fracture aperture, is a complex factor, which is related to

external stress,  $\text{CO}_2$  injection pressure, adsorption swelling effect, and  $\text{SC-CO}_2$  extraction [46]. After a long time of evolution, an irreversible change may occur in the channel for gas diffusion and seepage, contributing to the permeability of coal reservoir [16]. Eventually, the influence of  $\text{SCCO}_2$  on physical and chemical of coal may occur accompanied with the changes of pore structure,  $\text{CH}_4$  displacement, and carbonate evolution after a period of  $\text{CO}_2$  injection.

### 5.2. Effect of Oil Shale on Seal Capacity of $\text{CO}_2$ Sequestration.

The findings of FTIR analysis in this study confirm that the amount of oxygen-containing functional groups were extracted from oil shale during  $\text{SCCO}_2$  injection. Generally, oxygen-containing functional groups dominate the adsorption capacity through the sorption sites [47]. The decrease of oxygen-containing functional groups in oil shale may weaken the capacity of  $\text{CO}_2$  adsorption, causing the decrease in the  $\text{CO}_2$  sorption sites. The direct effect may cause less free  $\text{CO}_2$  changing into adsorbed state. With migrating towards oil shale, more free  $\text{CO}_2$  may gather beneath the cap rock. Section 5.1 has confirmed that  $\text{SCCO}_2$  may facilitate the diffusion and seepage channel of coal reservoir. In this case, plenty of hydrocarbons after  $\text{SCCO}_2$  extraction can move in the channel of coal bed, accompanied by free  $\text{CO}_2$  flow. After a long injecting time, hydrocarbons will gather below oil shale. Studies indicate that hydrocarbons may restrict physical access (fractures and larger pores) to oil shale matrix [48]. A large amount of hydrocarbons and other mineral matters, existing in the oil shale structure, may have a negative effect on the permeability, which in turn can hinder  $\text{CO}_2$  escaping from cap rock.

### 5.3. Implications of Tectonic Deformation on $\text{CO}_2$ Entrapment.

Original coal seam is known to be a typical dual-porosity



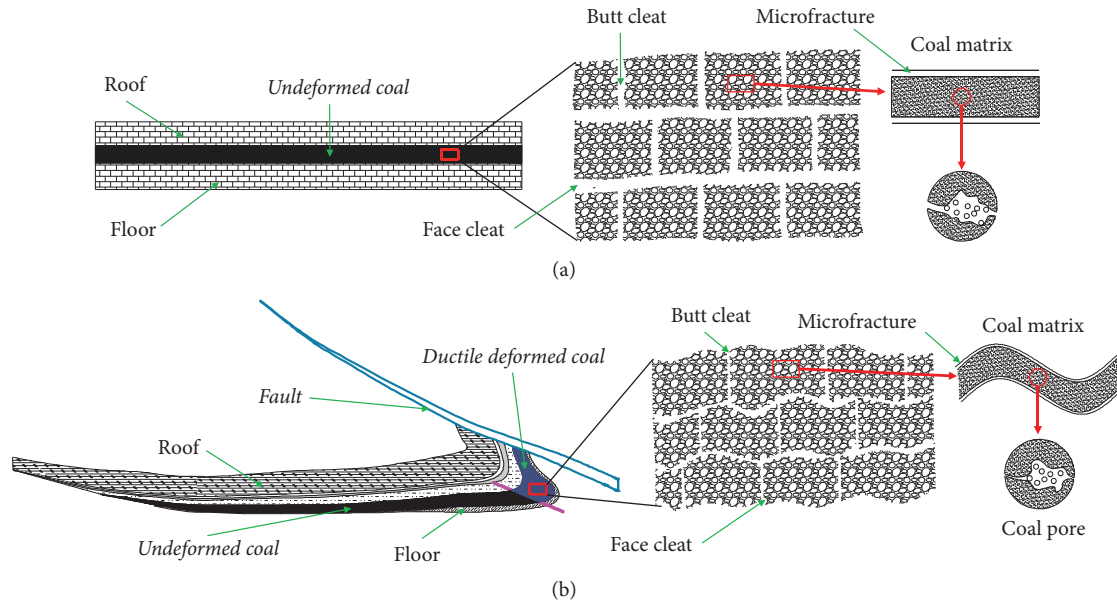


FIGURE 10: Influence of tectonic deformation on CO<sub>2</sub> entrapment in Haishiwan coalfield.

system that contains amount of coal matrix blocks divided by face cleat and butt cleat, as shown in Figure 10(a). Coal matrix, as a part of the blocks, consists of hierarchical pores with different morphology. However, because of the long-term geological formation movement, some areas may be subjected to structural changes. Taking Haishiwan coalfield as an example, undeformed coal and ductile deformed coal can be distinguished clearly in the field, which is shown in Figure 10(b). With fault tectonism, the migration of tectonic deformation may occur locally in coal seam, leading to the existence of coal reservoir and surrounding rocks. Moreover, a great quantity of winding and rub can not only produce more mylonite or other broken coals but also seal original cleats, micro-fractures, and seepage pores for CO<sub>2</sub> transport [49]. Therefore, poor connectivity and low permeability in the area of ductile deformed coal seam may block CO<sub>2</sub> migration along the bedding, which is liable to preserve more CO<sub>2</sub> fluids. This phenomenon serves as a role for CO<sub>2</sub> entrapment although geological sequestration in coal depends on multiple factors.

**5.4. How to Effectively and Safely Store CO<sub>2</sub> in Haishiwan Coalfield.** From a sequestration viewpoint, effective and safe storing in coal reservoir refers to the stable injection, migration, and storage of CO<sub>2</sub> in the appropriate spots for avoiding CO<sub>2</sub> leakage to cause Environmental Safety and Health (ES&H) issues. In Haishiwan coalfield, the favourable conditions of gas sources, migration and conduction of reservoir, sealing ability of cap rock, traps, and storage can bring about a steady environment of CO<sub>2</sub> gas reservoir.

According to the source and occurrence regularity of CO<sub>2</sub> in Yaojie coalfield as mentioned above, high concentration of CO<sub>2</sub> in No. 2 coal seam is derived from the indirect accumulation mode with gas production, gas guidance, gas migration, and gas storage through the brittle-ductile shear

dynamic metamorphism of the active F19 fault zone [50]. Based on the analysis of F19 fault tectonic movement and CO<sub>2</sub> distribution in No. 2 coal seam, F19 fault played a negative role on gas source and gas storage and leads to gas entrapment during the process of CO<sub>2</sub> accumulation.

Large amount of inorganic CO<sub>2</sub>, produced by the dynamic metamorphism of the active F19 fault zone, could move upwards along F19 fault and accumulate in the border of No. 2 coal seam and F19 fault. Driven by pressure gradient, CO<sub>2</sub> flow through the bedding orientation of No. 2 coal seam [51, 52]. With the continuous injection of CO<sub>2</sub>, the pressure of coal reservoir increased significantly, generating SCCO<sub>2</sub> condition. The interaction between coal and SCCO<sub>2</sub> may contribute to the development of pore structure and fracture system of coal seam, which may provide conditions for the further migration. Simultaneously, CO<sub>2</sub> migration may provide a vector by which Polycyclic Aromatic Hydrocarbons (PAHs) and other hydrocarbons are extracted from coal matrix and transported to the direction of cap rock (oil shale) [18, 21]. These may cause the decrease in the permeability of oil shale by these substances obstructing seepage channel, which contributes to the increase in sealing capacity of oil shale [45]. Besides, oil shale and tight sandstone in the roof and floor of No. 2 coal seam have a low permeability. Also, the coal seam near cap rock may has a low permeability due to the swelling effect of CO<sub>2</sub> adsorption in coal matrix. The above factors may result in the CO<sub>2</sub> accumulation below the cap rock (the roof of No. 2 coal seam), which is exhibited in Figure 11. When CO<sub>2</sub> fluids transport from injection point to coal reservoir, competitive adsorption between CO<sub>2</sub> and CH<sub>4</sub> molecules occurs, leading to CH<sub>4</sub> desorption and migration. However, the sealing ability of cap rock (oil shale) and geological tectonism (F19 fault) contributes to gas enrichment and accumulation during a long geological time. Therefore, on-site conditions, that is, the existence of cap rock and

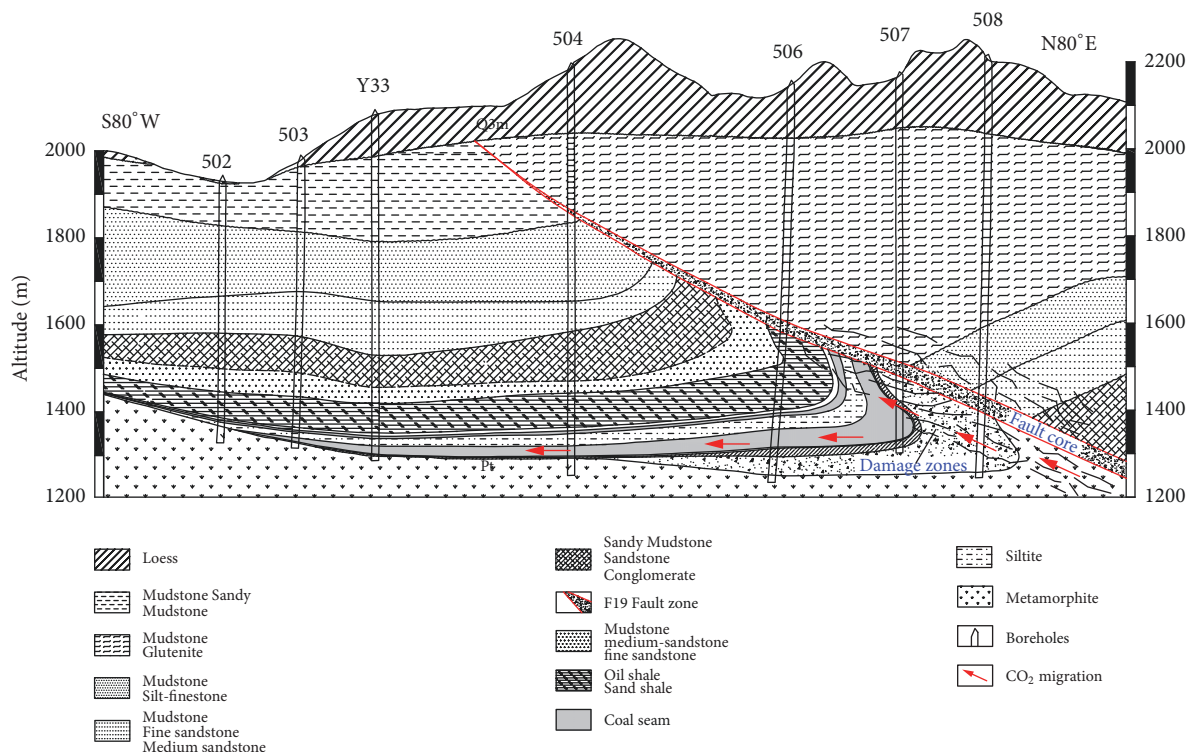


FIGURE 11: The mode of CO<sub>2</sub> enrichment and accumulation in Haishiwan coalfield.

geological tectonism, need to be considered, which have a positive implication for safe and effective CO<sub>2</sub> geological sequestration. The stable regional tectonic conditions and cap rock with low permeability are key factors for Haishiwan coalfield, providing an insight into the suitable selection of CCGS site.

## 6. Conclusions

Natural analogue studies of Haishiwan coalfield demonstrated that CO<sub>2</sub> can be naturally stored in deep coal seams for long period on condition that cap rock (oil shale) with good sealing capacity and stable regional tectonism, which contribute to gas entrapment and accumulation during a long geological period. Simultaneously, No. 2 coal seam, influenced by consecutive CO<sub>2</sub> injection and existence of SCCO<sub>2</sub>, may generate amount of fractures and will facilitate fluids migration. To study the suitable selection of CCGS site, in situ conditions and experiments were investigated and analyzed through geological tectonism on reservoir properties, pore structure, functional group distributions, and SCCO<sub>2</sub> extraction. Major findings are summarized as follows.

(1) CH<sub>4</sub> contents increase and CO<sub>2</sub> contents decrease with increasing distance from F19 faults, indicating that CH<sub>4</sub> can be recovered after CO<sub>2</sub> injection. Also, higher carbonate content is closer to F19 faults, which may be derived from the transformation of CO<sub>2</sub> near tectonism.

(2) Changes in smaller pores and seepage-flow pores (mesopores and macropores) are facilitated by SCCO<sub>2</sub>, but macropores in oil shale show no significant change and a

slight decrease in micropores and mesopores was found after SCCO<sub>2</sub> treatment. Thus, transport in coal matrix and seepage ability of coal seam will be promoted. Meanwhile, changes in pore structure properties of oil shale can lead to a low permeability and connectivity, which hinder the movement of fluid desorption, diffusion, and seepage.

(3) For FTIR analysis, amount of matters containing hydrocarbons or functional groups compounds in oil shale matrix are mobilized by SCCO<sub>2</sub>, causing the changes on the surface site and body structure. Then, aliphatic hydrocarbons and polycyclic hydrocarbons obtained from SCCO<sub>2</sub> extraction enhance the development of pore structure of coal, generating more migration channel.

## Conflicts of Interest

The authors declare no competing financial interests.

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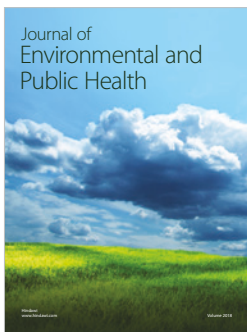
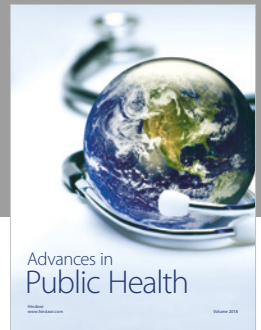
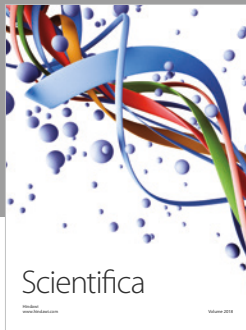
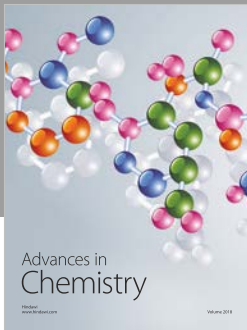
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