Molecular Modeling of Novel Fuel Cell Membranes

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Acknowledgment

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conversion devices. The fuel cells of interest contain polymer electrolyte membranes (PEMs) that inhibit the conduction of electrons and facilitate the transport of 2H₂ → 4H* + 4e protons. Nafion® is the most widely used Cathode: $O_2 + 4H^+ + 4e^- \rightarrow 2H_2O$ BTraz) has been show to perform better than membrane for fuel cell applications. However, alternatives are desired because traditional acidic PEMs such as Nafion and Nafion is expensive, allows significant Ph-SPEEKK. We used molecular dynamics amounts of methanol crossover, and to study the PEM morphology and the functions poorly at low humidity or high transport of water, hydronium, and methanol temperature. An acid-base blend membrane in Ph-SPEEKK/PSf-BTraz blend membranes.

Our aim is to understand the fundamental science behind the enhanced properties of SPEEKK) and basic polysulfone tethered the blend membrane.

Previous Research

composed of both acidic sulfonated

poly(ether ether ketone ketone) (Ph-

with 5-amino-benzotriazole (PSf-

Introduction

Fuel cells are clean and efficient energy

- · There is no membrane that meets the performance, durability, and cost requirements for widespread adoption of fuel cell technology.
- Nafion®, the most widely used PEM, sets the standard for comparative analysis.
- Aromatic membranes based on the poly(ether ether keetone) (PEEK) family show promise (Devanathan, Idupulapati, & Dupuis, 2012)



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- · Experimental studies cannot adequately describe nonoscale morphology and transport mechanisms (Devanathan, 2008).
- · Multiscale modeling helps us understand the interplay between microscopic and macroscopic phenomena.
- · Multiscale modeling reduces research and development costs, increases product quality, and leads to the development of new materials (Horstemeyer, 2009).
- Present work used ab initio molecular dynamics (AIMD) and molecular dynamics (MD). MD results are presented here.

- · The sulfonate groups are farther apart in the blend membrane at similar hydration levels.
- · Water and methanol diffuse slower in the blend membrane.
- · The superior performance of the blend membrane may be due to reduced methanol crossover.
- · Further studies are needed to relate the structure to transport properties.

Multiscale Modeling







SPEEK/PSf-BTraz Blend Membrane

• Experiment shows that this acid-base blend membrane has higher proton

conductivity and lower methanol crossover vs. Ph-SPEEKK (Li et al., 2010).

Sulfur (SO3-) coordination number (MD) in Blend vs. Ph-SPEEKK

Summary

CSU The California State University







Ph-SPEEKK

8 9

Distance (Å



Distance (Å)



Chemical structure

seed in this material are those of the authors and do not necessarily reflect the views of the S.D. Bechtel. Jr. Foundation or the National Science Foundation. This project has also been made possible with support of the National Marine Sanctuary Foundation. The STAR program is administered by the Cal Poly This material is based upon work supported by the S.D. Bechtel, Jr. Foundation and by the National Science Foundation under Grant No. 0952013. Any opinions, for Excellence in Science and Mathematics Education (CESaME) on behalf of the California State University (CSU). PNNL-SA-89661