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RAPID COMMUNICATION

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Non-doped, blue-emitting, color-stable, organic light-emitting diode based on 2,2':6',2''-ternaphthalene

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Abstract A novel non-doped blue-emitting material, 2,2':6',2"-ternaphthalene (NNN), which is based on three naphthalene units forming a rodlike molecule, has been synthesized and characterized. Organic light-emitting diodes based on NNN as emitter exhibit blue emission with Commission Internationale de l'Eclairage color coordinates of x = 0.18 and y = 0.11. In addition, the morphology and crystallographic structure of the NNN thin films has been investigated. NNN deposited on poly[3,4-(ethylenedioxy) thiophene]:poly (styrene sulfonate) on Indium Tin Oxide-covered glass forms crystalline structures of high crystallographic quality. The molecules are aligned almost perpendicular to the substrate surface and exhibit a tilt angle of 23° .

1 Introduction

Since the first reports of organic electroluminescent (EL) devices in the late 1980s [1-4], there has been significant interest in the field of organic light-emitting diodes

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Institute of Solid State Physics, Graz University of Technology, Petersgasse 16, 8010 Graz, Austria (OLEDs) driven by questions related to fundamental research as well as by proposed industrial applications. An example is solid-state lightning [5-7], which holds great promise as an energy-efficient light source. Another example is the use of OLEDs for full color displays [8, 9], which requires red, green and blue emitters of high color purity. As described in a recent review [10], the development of red and green emitters of sufficient quality did not pose too many problems but the performance of blue emitters often suffers from the wide band gap of the used materials. The intrinsically wide band gap leads to difficulties in the band alignment with respect to the electronic levels of other layers in the OLED and thus to limitations in the charge carrier injection and further to a reduced EL efficiency. Another problem related to the issue of a high band gap is the consequence that the lowest unoccupied molecular orbital (LUMO) is close to the vacuum energy level, which reduces the lifetimes of blue OLEDs because the material is susceptible to oxidation. Nevertheless, there are many reports on blue-emitting OLEDs [4, 11-20] and the optimization of device structure as well as the investigation of new blue emitters is still in progress.

In this paper, we present a new naphthalene-based small molecule [2,2':6',2''-ternaphthalene (NNN), see Fig. 1a], which can be used as a blue-emitting active layer in an OLED. We present the electrical and optical device characteristics, as well as the morphological and structural analysis of the evaporated thin films.

2 Experimental processing

2.1 Synthesis of 2,2':6',2"-ternaphthalene (NNN)

The details of NNN synthesis are reported elsewhere [21].



Fig. 2 a Electroluminescence spectra of a NNN OLED at various drive current levels. b Current voltage characteristics of the device. c Contour plot of the emission intensity depending on wavelength and current. d CIE color map showing the emission color of the device at

various current levels. **e** Image of the operating device: the emission area is defined by patterning the top Al contact (shadow mask deposition) and the width of the broader line (*right*) is 0.5 mm

2.2 Device preparation

The first step in preparing the OLED device was to spin coat poly[3,4-(ethylenedioxy) thiophene]:poly(styrene sulfonate) (PEDOT:PSS) on Indium Tin Oxide (ITO)-covered glass as an anode contact. Thereafter, samples have been inserted into the NNN deposition chamber and they have been preheated to the growth temperature for 30 min to clean the surface from adsorbents and to avoid temperature gradients during the deposition process. Next, NNN was deposited by hot wall epitaxy [22] for 60 min at a substrate temperature of 80 °C and at a base pressure of 2×10^{-6} mbar. Prior to deposition, NNN has been purified by a thermal sublimation procedure. The final step was the deposition of 100 nm of Al as cathode through a shadow

mask. Figure 1b shows a sketch of the fabricated OLED device.

2.3 Electrical and optical device characterization

For electro-optical characterization, the devices were mounted in an electrically shielded box equipped with an optical window. The measurements were performed at room temperature under ambient atmosphere. The collected EL was focused into an optical fiber coupled to a Avantes AvaSpec 2048 spectrometer with a resolution of 2.4 nm. Simultaneous to the spectral analysis, the electrical characterization was performed with a Keithley 6487 current–voltage source meter. Images of the operating OLEDs have been acquired with a digital optical camera attached to a Nikon SMZ-2T microscope.

2.4 Morphological investigation

Atomic force microscopy (AFM) studies of the deposited organic films were performed using a Digital Instruments Dimension 3100 in tapping mode. The images have been acquired at scan speeds of $4-6 \mu m/s$ using SiC tips exhibiting a cone angle of 40° . Nominal values for resonance frequency and tip radius are 325 kHz and 10 nm, respectively. Data analysis was performed with the software Gwyddion [23].

2.5 Structural investigation

X-Ray diffraction (XRD) measurements were carried out on a Philips X'pert X-ray diffractometer using Cr K α radiation ($\lambda = 2.29$ Å) and a secondary graphite monochromator. Specular scans were performed in Bragg– Brentano configuration by varying the z-component of the scattering vector q [24]. Consequently it is possible to detect lattice planes which are parallel to the sample surface.

The monoclinic unit cell parameters of NNN, which have been used to evaluate the measurement data, are given by a = 8.15 Å, b = 5.98 Å, c = 19.45 Å and $\beta = 94.58^{\circ}$ [25].

3 Results and discussion

The OLEDs have been prepared and analyzed as explained in Sect. 2. As shown in Fig. 2a, c electroluminescence was 733

collected and spectrally analyzed in the range of 300-900 nm. An emission pattern is observable in the blue spectral range of the visible regime, with its main peaks at 383 and 405 nm and an additional shoulder at 427 nm representing the 0-0, 0-1 and 0-2 vibronic transitions, respectively. The onset of the spectral emission at 368 nm gives an estimated band gap for NNN of ≈ 3.37 eV. Spectroscopic measurements have been obtained at drive currents ranging from 0-30 mA and there is no significant shift in the positions of the emission peaks. Therefore, it can be stated that the emission color is stable in the investigated current regime. This point is further underlined by the calculation of the color coordinates of the emission according to the 1931 Commission Internationale de l'Eclairage (CIE), as depicted in Fig. 2d. Values of x = 0.183 ± 0.002 and $y = 0.109 \pm 0.007$ have been obtained.

Figure 2b shows the current–voltage characteristics of the NNN OLED, which exhibits a clear current onset at 3.8 V and a negligible leakage current. The current onset is roughly 0.4 V above the band gap of NNN, which indicates small losses. The device homogeneity is demonstrated by the fabrication of structured Al top contacts. The lightemitting area, which is depicted in Fig. 2e, consists of a $\approx 0.5 \times 5.5$ mm² rectangle with a signature attached to it, representing the logo of our institute with a line width of 100 µm. The image is evidence of a uniform light intensity and a homogeneous current transport, without significant indications of pinholes.

Further investigations have been carried out in relation to the morphological quality of the fabricated organic thin films. AFM images of the ITO, PEDOT and NNN surfaces are depicted in Fig. 3. The films have been prepared using the same method employed for the fabrication of the OLED device. The ITO surface is very smooth, exhibiting a rms

 $Z_{MAX} = 13 \text{ nm}$ $Z_{MAX} = 13 \text{ nm}$ $Z_{MAX} = 72 \text{ nm}$







Fig. 4 a Specular X-ray investigation of NNN deposited at 80 °C on top of glass–ITO–PEDOT. All observed Bragg peaks can be explained by the presence of NNN crystals and indium oxide. The *red line* represents a powder spectrum of In_2O_3 taken from the

literature [37]. **b** Side view of a NNN crystal; the *gray rectangle* shows the *a*–*c* plane of the monoclinic unit cell. **c** Top view of the NNN crystal, with the *gray rectangle* showing the *a*–*b* plane of the unit cell

roughness of 1.18 nm. PEDOT layers have been deposited on top to improve band alignment and injection as reported for a similar small molecule-based OLED (*para*-hexaphenyl) [15]. The PEDOT surface exhibits a rms roughness of 7.7 nm. On top of PEDOT, the active NNN layer has been deposited and, interestingly, we do not observe a closed film using AFM but NNN molecules appear to be organized in individual crystallites with an average height of 46 nm and lateral dimensions in the order of 0.5–1 μ m. It follows, that the low leakage current of the device cannot be related to the active layer but the PEDOT layer is responsible for this observation, as PEDOT acts as an electron blocking layer.

To further study the molecular order of the crystalline NNN thin films, X-ray diffraction measurements have been performed. The peak positions in the specular scan shown in Fig. 4a can be explained by NNN crystals with their (001) net-planes parallel to the substrate surface and the respective crystal structure has been reported elsewhere [25]. The (00l) series is indexed in the plot and the (001)reflex is observed at 4.51°. Peaks are present up to the eighth order, which reflects the high crystallographic quality of the grown thin films. (001) Contact planes of crystallites based on rodlike organic molecules prepared on isotropic substrates have been also reported for e.g., parahexaphenyl [26], pentacene [27] and α -sexithiophene [28]. In addition, it is possible to detect peaks which can be explained by the presence of indium oxide (In_2O_3) . It is visible, that the measured ITO peak positions are found at smaller angles as predicted by the In₂O₃ powder spectrum (red line). This can be understood, because ITO is based on doping of an In_2O_3 lattice (replacing In^{3+} by Sn^{4+} ions) [29] and the substitutional incorporation of Sn leads to an increased lattice constant [30, 31].

A real space model of the molecular orientation is presented in Fig. 4b, c which can be derived from the crystallographic unit cell and the crystal contact plane. The side view (cut through the a–c plane of the NNN crystal) shows that the molecules are aligned almost perpendicular to the substrate. The long molecular axis is tilted with respect to the substrate surface by a value of 67°. The cut through the a–b plane of the NNN crystal (top view) shows that the molecular planes (defined by common planes of naphthalene units) of adjacent molecules are rotated by 54°. The so-called herringbone packing is commonly observed in rodshaped all-aromatic planar molecules like *para*-hexaphenyl [26], α -sexithiophene [32] or pentacene [33].

It has to be stated, that an upright standing orientation of a rodlike chromophore is not optimized for OLED emission as reported by Yanagi and Okamoto [34] for the example of *para*-hexaphenyl. This is caused by the alignment of the optical transition dipole, which is typically parallel to the long molecular axis [34, 35, 36].

4 Conclusions

OLEDs have been prepared on the basis of 2,2':6',2''-ternaphthalene (NNN) as a non-doped emitter. Electro-optical characterization showed homogeneous deep-blue emission, exhibiting CIE color coordinates of x = 0.18 and y = 0.11, which is close to the National Television System Committee (NTSC) blue color standard (x = 0.14 and y = 0.08) [15]. In addition, color stability, low losses and a low leakage current of the device could be demonstrated. Interestingly, the NNN thin film is not composed of a closed layer, but of individual (001)-oriented crystals consisting of standing molecules (tilt angle 23° with respect to the layer normal) packed in a herringbone structure.

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