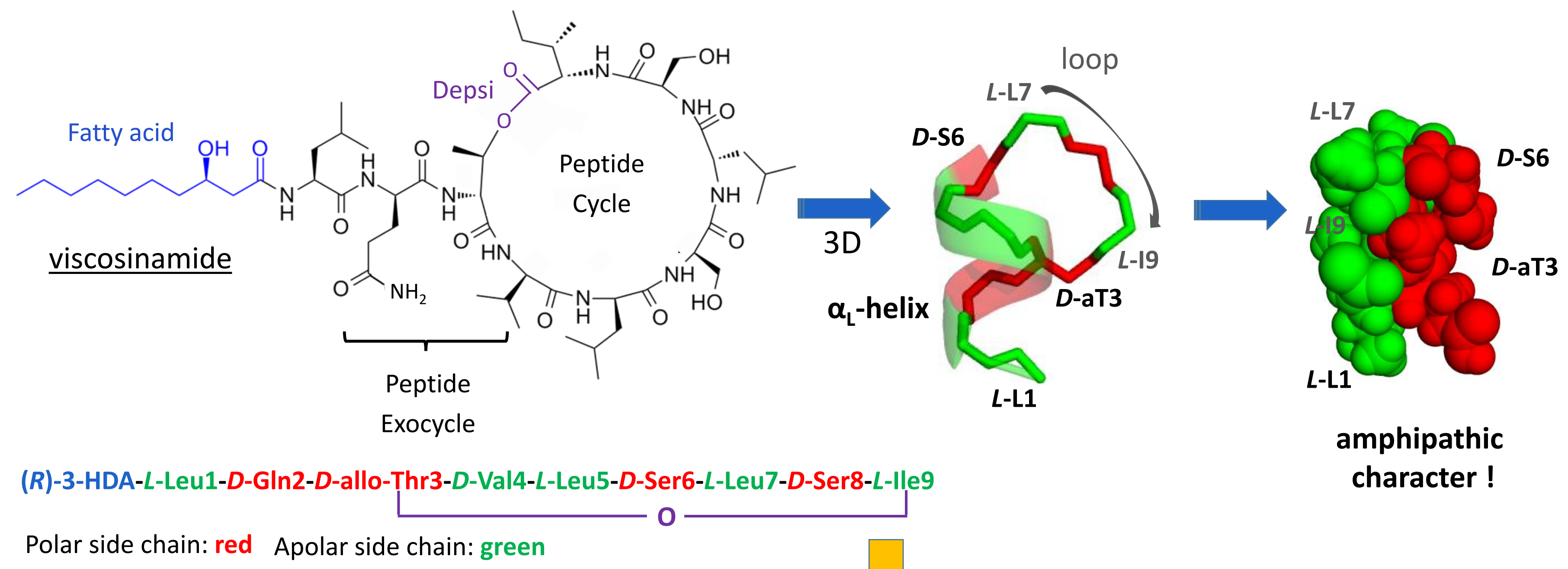


What CLPs are? How do they look like?

- Cyclic lipodepsipeptides (CLPs) are secondary metabolites of *Pseudomonas* and *Bacillus* bacterial species produced via non ribosomal pathways [1]
- They are consisted of a fatty acid moiety linked to the N-terminus of a peptide chain which is cyclized by an ester (or depsi) bond formation between its C-terminus and an OH capped side chain of a Ser or Thr
- Peculiar primary structural features: D-amino acids + alternation of polar and apolar amino acid side chains
- Tertiary structure: backbone conformation assessed: alpha-helix + loop → amphipathic character [2]



CLP bioactivity

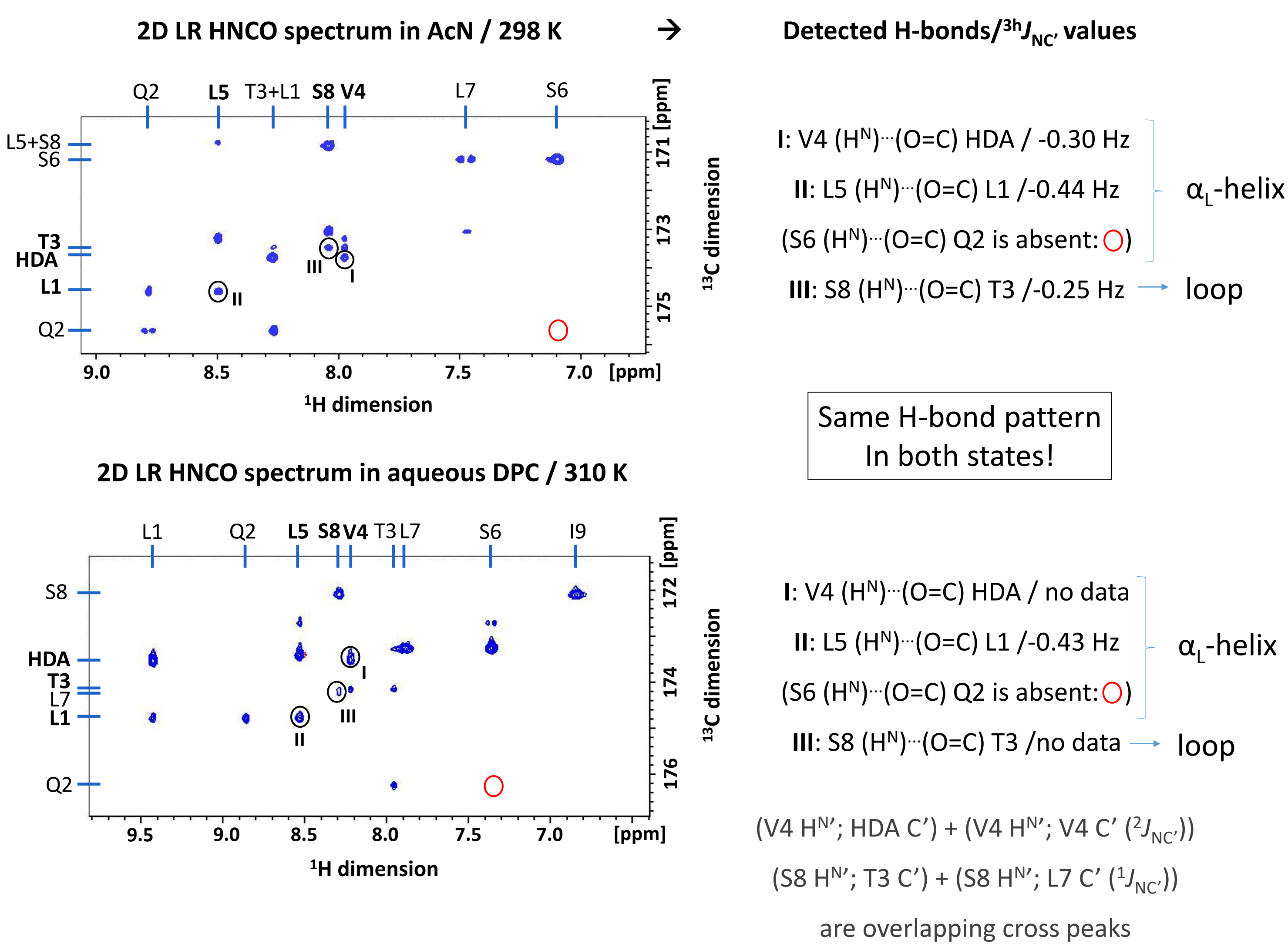
- Bacterial swarming (motility), biofilm formation [1]
- Stimulation of the plant immune system → crop protection [3]
- In vitro* testing → activity against bacteria, viruses, fungi (non-exhaustive) [1]
- Novel antibiotics: daptomycin (marketed as **CUBICIN®**) [4]
- Anticancer effects below cytotoxic level (xantholysin, MD0066, viscosin) [5]

Structure – function/mode of action relationships not well understood!

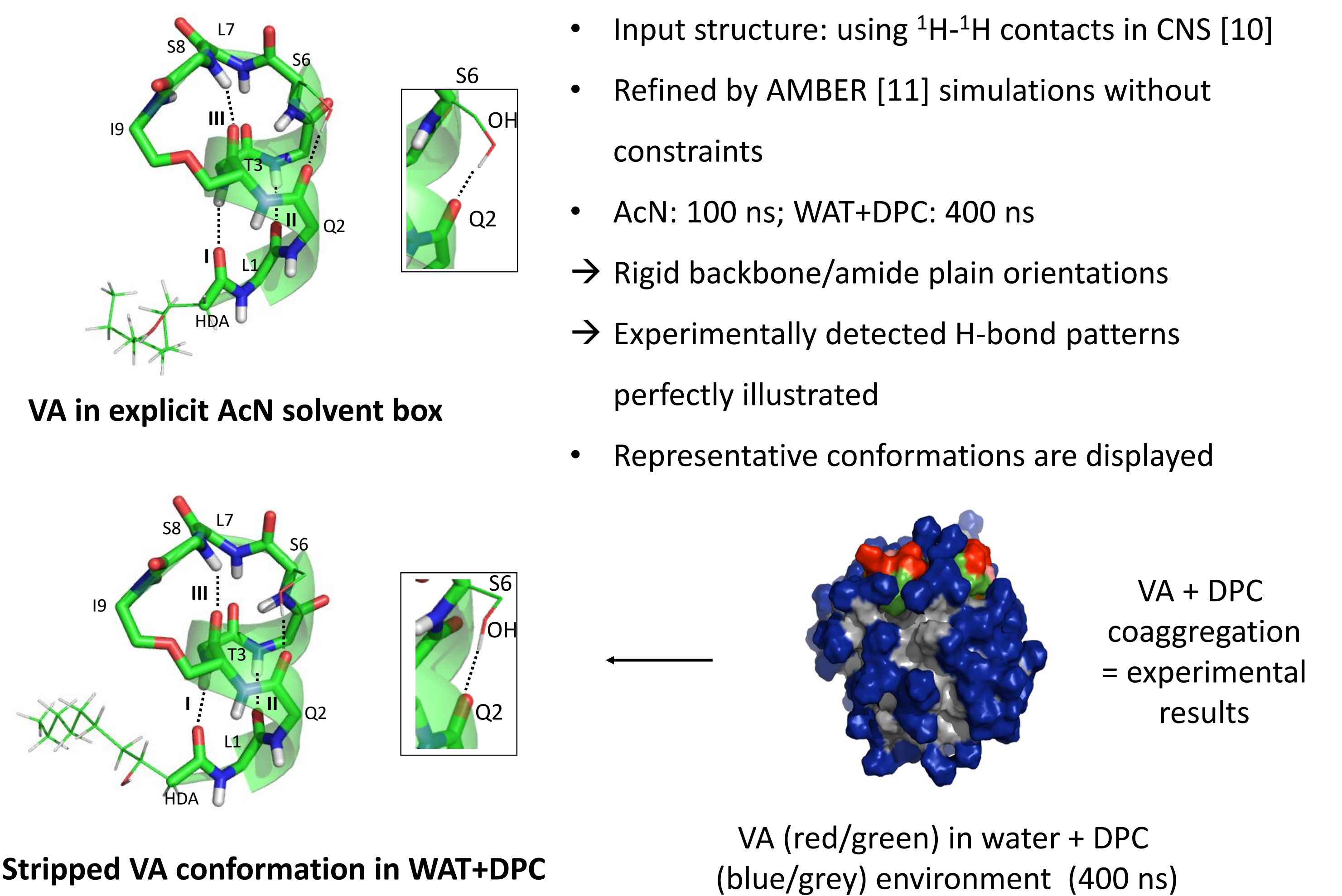
- Goals**
- More detailed structural information is needed (than ¹H-¹H distance restraints) →
 - Direct evaluation of amide plane orientations and H-bond pattern
 - I) Intramolecular: peptide conformation in monomeric state vs in membrane-mimicking environment [6]
 - II) Pore formation in low polarity medium [7] → structural characterization of a self-assembly
- How?**
- Growing *Pseudomonas* DR54 in minimal salt medium → ¹³C-, ¹⁵N-labelled viscosinamide (VA): first ever isotope labelled CLP
 - J-correlation spectroscopic methods: ³J_{HNHA} [φ] and ³J_{NC'} [r,Θ] → H-bonds
 - Complementary *in silico* studies: AMBER molecular dynamic simulations

Results I) Conformational rigidity of VA

- In polar solvent i.e. AcN-d3: VA is in *monomeric state*
- In aqueous DPC-d38 solution: VA is *coaggregated* with the DPC molecules (DOSY)
- VA adopts the 'same' conformation in both states!**
- Experimental assessment:



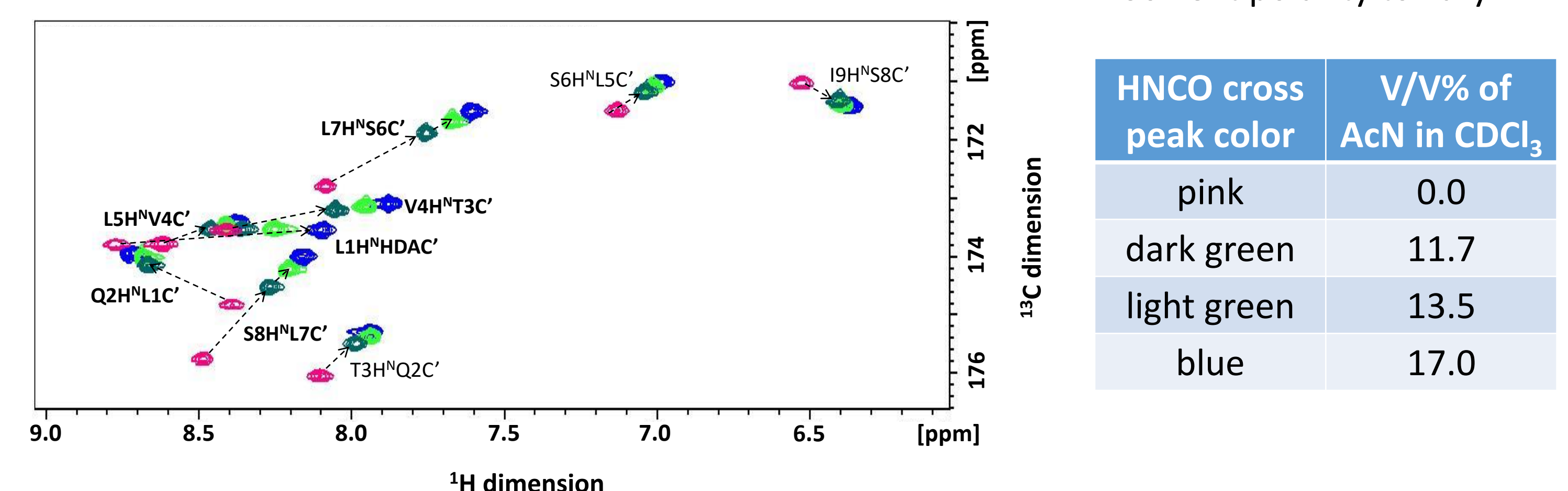
In silico assessment:



Results II) Intermolecular interactions in VA self-assembly

- In low polarity solvent i.e. chloroform-d: the amphipathic VA self-assembles (→ NMR spectral line broadenings + DOSY)
- Plausible model for CLP self-assembly [12]**
- LR HNCO did not indicate the intermolecular H-bonds due to the *fast* (>1/³J_{NC'}) exchange between the monomeric and assembled states
- Population averaged amide group chemical shifts → let's influence it!

HNCO cross peaks of VA dissolved in CDCl₃ + AcN-d3 mixtures / 278 K



Conclusion and future prospects

- Viscosinamide displays identical conformation in its monomeric state (dissolved in AcN) and in its coaggregated state with real cell membrane-mimicking DPC micelles
- The structural assessment detailed the orientation of the amide planes and of the intramolecular H-bond pattern using J-correlation NMR methods and AMBER molecular dynamic simulations
- The protocol is planned to be applied for larger CLPs e.g. xantholysin, tolaasin
- Interpeptide interactions have been indirectly shown for the self-assembly of viscosinamide. In the future the full structural elucidation of such supramolecular organization will be performed using isotope-filtered NOESY

References

[1] Raaijmakers, J.M. et al., *Mol. Plant. Microb. Interact.* 2006, **19** (7), 699–710 [2] Jourdan, F.; Lazzaroni, S. et al., *Proteins: Struct., Funct., Genet.*, 2003, **52**, 534–543 [3] Khong, N.G.; Randoux, B. et al., *Commun. Agric. Appl. Bio. Sci.*, 2013, **78** (3), 479–487 [4] Fishbach, M.A.; Walsh, C.T.; *Chem. Rev.*, 2006, **106**, 3466–3496 [5] Pascual, J.; Garcia-López, M. et al., *Syst. Appl. Microbiol.*, 2014, **37**, 412–416 [6] Warschawski, D.E. et al., *Biochim. Biophys. Acta - Biomembranes* 2011, **1808**, 1957–1974 [7] Lee, T., Hall, K., Aguilar, M., *Curr. Top. Med. Chem.*, 2016, **16**, 25–39 [8] Vuister, G.; Bax, A. J. Am. Chem. Soc. 1993, **115**, 7112–7111 [9] Cordier, F. et al., *Nat. Protoc.*, 2008, **3**, 235–241 [10] Brunger, A. T.; *Nat. Protoc.*, 2007, **2**, 2728–2733 [11] Salomon-Ferrer, R.; Gotz, A.W. et al., *J. Chem. Theory Comput.*, 2013, **9**, 3878–3888 [12] Sinnaeve, D.; Hendrickx, P.M.S. et al., *Chem. Eur. J.*, 2009, **15**, 12653–12662