#### ORIGINAL ARTICLE



# Synthesis, characterisation, and in vitro cellular uptake kinetics of nanoprecipitated poly(2-methacryloyloxyethyl phosphorylcholine)-*b*-poly(2-(diisopropylamino)ethyl methacrylate) (MPC-DPA) polymeric nanoparticle micelles for nanomedicine applications

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Abstract Nanoscience offers the potential for great advances in medical technology and therapies in the form of nanomedicine. As such, developing controllable, predictable, and effective, nanoparticle-based therapeutic systems remains a significant challenge. Many polymerbased nanoparticle systems have been reported to date, but few harness materials with accepted biocompatibility. Phosphorylcholine (PC) based biomimetic materials have a long history of successful translation into effective commercial medical technologies. This study investigated the synthesis, characterisation, nanoprecipitation, and in vitro cellular uptake kinetics of PC-based polymeric nanoparticle micelles (PNM) formed by the biocompatible and pH responsive block copolymer poly(2-methacryloyloxyethyl phosphorylcholine)-b-poly(2-(diisopropylamino)ethyl methacrylate) (MPC-DPA). Atom transfer radical polymerisation (ATRP), and gel permeation chromatography (GPC) were used to synthesise and characterise the welldefined MPC<sub>100</sub>-DPA<sub>100</sub> polymer, revealing organic GPC, using evaporative light scatter detection, to be more accurate than aqueous GPC for this application. Subsequent nanoprecipitation investigations utilising photon correlation spectroscopy (PCS) revealed PNM size increased with polymer concentration, and conferred Cryostability. PNM diameters ranged from circa 64-69 nm, and increased upon hydrophobic compound loading, circa 65-71 nm, with loading efficiencies of circa 60 % achieved, whilst remaining monodisperse. In vitro studies demonstrated that the PNM were of low cellular toxicity, with colony formation and MTT assays, utilising V79 and 3T3 cells, yielding comparable results. Investigation of the in vitro cellular uptake kinetics revealed rapid, 1 h, cellular uptake of MPC<sub>100</sub>-DPA<sub>100</sub> PNM delivered fluorescent probes, with fluorescence persistence for 48 h. This paper presents the first report of these novel findings, which highlight the potential of the system for nanomedicine application development.

**Keywords** Nanomedicine · Phosphorylcholine · MPC-DPA · Micelle · Nanoparticle · Drug delivery

### Introduction

Recent estimates place the annual global incidence of, and mortality from, human cancers at circa 14 and 8 million cases per annum and growing (Torre et al. 2015). This is clearly a significant societal burden and challenge, which must be overcome for both humanitarian and financial reasons. The advent of nanotechnology-enabled medical therapies, termed nanomedicines (Farokhzad and Langer 2006; Sahoo et al. 2007), offers the potential for great advances in the development of novel anti-cancer treatments (Kawasaki and Player 2005), with one of the main areas of research being stimuli-responsive polymeric nanoparticle-based nanomedicines (Crucho 2015). As part of that research and development momentum, elucidating the cellular uptake kinetics of nanoparticles is critical for developing a greater understanding of nanoparticulatebased intracellular delivery systems (NIDS) that utilise



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endocytosis for therapeutic delivery via cell membrane transport (Sahay et al. 2010) in an active and applied manner. Phosphorylcholine (PC) based polymers (Hayward and Chapman 1984) have a successful record of innovative biomaterials development and application (Lewis 2000), including biocompatible pH responsive polymeric NIDS (Salvage et al. 2005). As such a number of studies have explored the in vitro cellular uptake of pH responsive poly(2-methacryloyloxyethyl phosphorylcholine)-bpoly(2-(diisopropylamino)ethyl methacrylate) (MPC-DPA) block copolymer nanoparticles (Lomas et al. 2008; Massignani et al. 2010) which have demonstrated the potential for therapeutic delivery (Colley et al. 2014), but have been primarily concerned with polymersome, vesiclebased, systems of circa 100 nm or greater diameter. However, to date, the smaller diameter micelle-based MPC-DPA nanoparticle systems have attracted less attention, with studies focused primarily on achieving delivery (Licciardi et al. 2008; Yu et al. 2013), without fully considering the kinetic aspects of the cellular uptake reported. The small size of micelles has advantages, as it may facilitate particle accessibility to systemic locations not accessible to larger sized delivery vehicles (Singh and Lillard 2009). Since the first reports of MPC-DPA polymers (Ma et al. 2003) and nanoparticles (Salvage et al. 2005; Licciardi et al. 2005), the systems have remained considerably less well characterised than their extensively studied alternatives, such as PEG-based nanosystems (Torchilin 2002; Shiraishi et al. 2009; Locatelli and Franchini 2012; Logie et al. 2014; Laskar et al. 2015). MPCbased polymers are thought to possess improved antifouling properties (Lewis 2000) in comparison to PEG-based polymers due to the greater level of hydration reported for MPC, circa 23 water molecules per MPC unit (Goda et al. 2006; Morisaku et al. 2008), in contrast to circa 3 per PEG unit (Maxfield and Shepherd 1975; Shikata et al. 2006). The premise of these biomimetic NIDS being the ability to avoid reticuloendothelial system (RES) clearance (Gref et al. 1994), more so if within the 30-100 nm particle diameter size range (Garcia et al. 2014), and thus attain sufficiently long systemic circulatory times such that they can harness the enhanced permeability and retention (EPR) effect (Maeda 2001) in order accumulate at tumour sites due to the abnormally leaky vasculature (Takakura et al. 1998; Ho and Shoichet 2013). In the case of pH-responsive MPC-DPA (Ma et al. 2003; Salvage et al. 2005), the PNM would then release the therapeutic cargo in response to the lowered pH environment found in endosomes (Sharma and Sharma 1997) and tumour tissue (Liu et al. 2014).

It has been previously reported that MPC-DPA diblock copolymers form monodisperse polymeric nanoparticle micelles (PNM) via nanoprecipitation (Salvage et al. 2015). This current paper expands upon the previous work,



examining the synthesis and characterisation of the diblock copolymer MPC<sub>100</sub>-DPA<sub>100</sub>, followed by a further exploration of the polymer with regard to the nanoprecipitation parameters of polymer concentration, solvent to non-solvent volume ratio, model drug compound loading, and the effect of these on particle size and polydispersity, followed by an investigation of in vitro cellular uptake and delivery. This paper reports for the first time novel data and findings regarding MPC<sub>100</sub>-DPA<sub>100</sub> PNM formation, stability, loading, and intracellular uptake kinetics, which aids to further the understanding of applied endocytosis, for these PC-based PNM systems to be harnessed for effective intracellular delivery applications, and potentially nanomedicine.

### Materials and methods

# Synthesis and characterisation of MPC<sub>100</sub>-DPA<sub>100</sub> block copolymer

The MPC<sub>100</sub>-DPA<sub>100</sub> diblock copolymer was synthesised via atom transfer radical polymerisation (ATRP) in methanol (MeOH) at ambient room temperature, having adapted a published ATRP method (Ma et al. 2003) by utilising 2-(4-morpholino)ethyl 2-bromosiobutvrate (MEBr) as the initiator, at [MPC]:[MEBr]:[CuBr]:[Bpy]:[DPA] = 100:1:1:2:100 relative molar ratios. The MEBr was synthesised and purified according to a literature protocol (Bories-Azeau et al. 2004), and confirmed by <sup>1</sup>H nuclear magnetic resonance (NMR) analysis. The resultant MPC-DPA copolymer was characterised for molecular weight (Mw and Mn) and polydispersity (Mw/ Mn) via <sup>1</sup>H NMR and gel permeation chromatography (GPC). <sup>1</sup>H NMR was undertaken with a Bruker Ascend<sup>TM</sup> 400 MHz spectrometer, 16 scans per spectrum, using chloroform-d (CDCl<sub>3</sub>) and methanol-d<sub>4</sub> (CD<sub>3</sub>OD) NMR solvents (Sigma, UK) at a 3:1 (v/v) ratio (Pearson et al. 2013); the CDCl<sub>3</sub> contained 1 % (v/v) tetramethylsilane (TMS) as an internal standard (Li et al. 2004). GPC analysis was undertaken using both organic and aqueous mobile phase and column types to evaluate their comparative suitability and accuracy. The organic GPC system utilised Phenogel<sup>TM</sup> 5  $\mu$ m 10<sup>4</sup> Å and 5  $\mu$ m 10<sup>3</sup> Å columns  $(300 \text{ mm} \times 7.8 \text{ mm})$  (Phenomenex, UK) in series, and a Polymer Labs (PL-ELS-2100 Ice) evaporative light scatter (ELS) detector (45 °C nebuliser and evaporator,  $1.1 \, \mathrm{l} \, \mathrm{min}^{-1}$  nitrogen flow). The organic mobile phase eluent was chloroform (CHCl<sub>3</sub>):MeOH (3:1 v/v) (Fisher, UK), containing 0.1 % acetic acid (AcAc) and 0.05 % trifluoroacetic acid (TFA) (Sigma, UK), with a flow rate of  $1.0 \text{ ml min}^{-1}$  at 45 °C, and a sample injection volume of 5  $\mu$ l at 1 mg ml<sup>-1</sup>. The aqueous GPC system utilised

BioSep<sup>TM</sup> SEC-s3000 5  $\mu$ m and SEC-s2000 5  $\mu$ m columns (300 mm x 7.8 mm) (Phenomenex, UK) in series, and a Shodex (RI-101) refractive index (RI) detector. The aqueous mobile phase eluent was 0.5 M acetate buffer (pH 2) (Sigma, UK), with a flow rate of 0.5 ml min<sup>-1</sup> at 40 °C, and a sample injection volume of 30  $\mu$ l at 10 mg ml<sup>-1</sup>. A series of near monodisperse polyethylene glycol (PEG) and polyethylene oxide (PEO) GPC standards (Crawford Scientific, UK) were used for GPC calibration.

# Preparation of MPC<sub>100</sub>-DPA<sub>100</sub> polymeric nanoparticle micelle systems

The MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems were prepared using a previously reported nanoprecipitation method (Salvage et al. 2015), utilising methanol (MeOH) (Fisher, UK) and phosphate buffer saline (PBS) (Oxoid, UK) as the solvent and the non-solvent phases, respectively.

# *Polymer concentration, freeze-thaw, and freeze-dry rehydrate effect*

Solutions of the MPC<sub>100</sub>-DPA<sub>100</sub> copolymer (40 mg ml<sup>-1</sup>) were prepared in MeOH, and aliquots (100, 200, 300, 400, and 500 µl) of these solutions were added drop-wise to 9.9, 9.8, 9.7, 9.6, and 9.5 ml of PBS, pH 7.4, respectively. The resultant samples were 0.22 µm pore size syringe filtered (Millipore, UK), each sample was divided into 3 equal volumes, and these sub-divided into 3 groups for separate testing. The final prepared copolymer concentrations were 0.4, 0.8, 1.2, 1.6, and 2.0 mg ml<sup>-1</sup>, respectively. The particle size and polydispersity of group 1 (Control) samples were determined by photon correlation spectroscopy (PCS) without further treatment. To assess the effect of a freeze-thaw (FT) process, group 2 samples were frozen at -20 °C for 24 h, and then allowed to thaw for 24 h, before examination of particle size and polydispersity using PCS. To assess the effect of a freeze-dry (FD) and rehydration process, group 3 samples were frozen at -20 °C for 24 h, lyophilised for 18 h using a Christ Alpha 2-4 freeze dryer, and then rehydrated with 18.2 MΩ-cm deionised water (Elga Purelab<sup>®</sup>) for 24 h, before examination using PCS.

### Loading with the fluorescent probe Nile Red

Solutions of the MPC<sub>100</sub>-DPA<sub>100</sub> copolymer (40 mg ml<sup>-1</sup>) were prepared in MeOH with and without the hydrophobic fluorescent probe Nile Red (NR) (Sigma, UK) at 2 mg ml<sup>-1</sup>, based on a previously published polymer to NR weight loading ratio of 10:0.5 (Salvage et al. 2015). Aliquots (100  $\mu$ l) of these were added drop-wise, to 9.9 ml of PBS, pH 7.4, and the resultant samples 0.22  $\mu$ m pore

size syringe filtered. The final prepared copolymer concentration was  $0.4 \text{ mg ml}^{-1}$ .

# Coumarin-6 solubility in Methanol

To determine the solubility of the fluorescent probe Coumarin-6 (Cm-6) (Sigma, UK) in MeOH, a standard curve was constructed (Tan et al. 1976). Briefly, the lambda max  $(\lambda_{max})$  absorbance of Cm-6 in MeOH (20 µg ml<sup>-1</sup>) was determined (457 nm) using a Lambda 25 UV/Vis spectrophotometer (PerkinElmer, UK), 200–700 nm scan, 1 nm spectral bandwidth. The absorbance of Cm-6 in MeOH solutions (1, 5, 10, 15 µg ml<sup>-1</sup>) at 457 nm were recorded and a standard curve constructed. Finally, Cm-6 in MeOH (1 mg ml<sup>-1</sup>) was prepared, 0.22 µm pore size syringe filtered to remove undissolved Cm-6, diluted 1 in 20 with MeOH, the absorbance measured at 457 nm, and Cm-6 solubility in MeOH calculated from the standard curve.

# Loading with the fluorescent probe Coumarin-6

Solutions of the MPC<sub>100</sub>-DPA<sub>100</sub> copolymer (40 mg ml<sup>-1</sup>) were prepared in MeOH with and without the hydrophobic fluorescent probe Cm-6 at 0.3 mg ml<sup>-1</sup>, the Cm-6 solubility in MeOH was determined from the standard curve. Aliquots (100, 200, 300, 400, and 500  $\mu$ l) of these solutions were added drop-wise to 9.9, 9.8, 9.7, 9.6, and 9.5 ml of PBS (pH 7.4), respectively, the resultant samples 0.22  $\mu$ m pore size syringe filtered, and the absorbance measured at 457 nm. The final prepared copolymer concentrations were 0.4, 0.8, 1.2, 1.6, and 2.0 mg ml<sup>-1</sup>, respectively. The Cm-6 loading capacity of the resultant MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems was determined from the Cm-6 in MeOH standard curve, corrected for blank PNM, and Cm-6 control absorbance.

# Particle size characterisation

Particle size characterisation of the prepared  $MPC_{100}$ -DPA<sub>100</sub> PNM systems was undertaken using photon correlation spectroscopy (PCS), and cryo-scanning electron microscopy (Cryo-SEM).

# Photon correlation spectroscopy

The PCS particle size measurements were undertaken using a Malvern Instruments Zetasizer Nano ZS90 equipped with a 633 nm HeNe laser and 90° detector collection angle. PCS measurements were carried out in triplicate, at 25 °C, with a thermal equilibration time of 1 min per degree of temperature change, plus 5 min, from room temperature, employed before PCS measurement commenced, the sample temperature was maintained by a



Peltier thermal cuvette mounting stage. Each PCS size measurement had a duration of 120 s, consisting of 12 subrun analyses (Pearson et al. 2013) by the Malvern PCS software, recording the fluctuations in the scattered light intensity, to determine the intensity based hydrodynamic diameter  $(D_h)$  and system polydispersity (Pd). The  $D_h$ being calculated with the Stokes-Einstein equation  $(D_{\rm h} = {\rm k}T/3\pi\eta d)$ , where  $D_{\rm h} = {\rm hydrodynamic}$  diameter, d = diffusion coefficient, k = Boltzmann'sconstant, T = absolute temperature, and  $\eta =$  viscosity (Kaszuba et al. 2008). Samples were syringe filtered (0.22  $\mu$ m) prior to PCS measurement to ensure sample quality. Results determined, via the monomodal Cumulants analysis, to be polydisperse (Pd >0.1) were further deconvolved using the Malvern software non-negative least squares (NNLS) based algorithm (Kaszuba et al. 2008), to resolve optimal particle size distributions (PSD) (Woodbury et al. 2006; Ansari and Nyeo 2012), with 50 and 200 nm monodisperse polystyrene nanoparticle standards (Duke Scientific, USA) employed to illustrate the principle of Cumulants versus NNLS analysis.

#### Cryo-scanning electron microscopy

The MPC<sub>100</sub>-DPA<sub>100</sub> PNM system (0.4 mg ml<sup>-1</sup>) in PBS, pH 7.4, was examined under Cryo-SEM using a PP3000T cryogenic sample preparation system (Quorum Technologies, UK), together with a SIGMA field emission gun scanning electron microscope (FEG-SEM) (Zeiss, UK). The sample was flash frozen in slushy nitrogen, circa -207° C, on a Cryo-SEM sample holder, to avoid the Leidenfrost effect (Song et al. 2010), using the Quorum Prepdek<sup>TM</sup> workstation. The frozen sample was transferred under vacuum to the Cryo-SEM exchange chamber, maintained at  $-130^{\circ}$  C with liquid nitrogen, and the sample freeze-fractured. The sample was sublimed at  $-90^{\circ}$ C for 5 min to remove frozen water and reveal the PNM, a sputter coating of platinum (5 nm) applied, and the sample transferred to the Cryo-SEM stage within the FEG-SEM chamber. The Cryo-SEM stage and anti-contaminator were cooled to -130 and -170 °C, respectively, via capillary fed, liquid nitrogen cooled, nitrogen gas. The sample was FEG-SEM imaged at 2 kV accelerating voltage, using a 30 µm aperture, 8.5 mm working distance, and an Everhart-Thornley secondary electron detector.

#### In vitro cellular cytotoxicity testing

V79 fibroblast cell (ATCC, USA) and 3T3 fibroblast cell (ATCC, USA) stocks were maintained in Dulbecco's Modified Essential Medium (DMEM) (PAA, UK) supplemented with 10 % foetal bovine serum (FBS) (PAA, UK) at 37 °C in 5 % CO<sub>2</sub>, with passage at 70–80 % confluency.



#### Cell colony formation assay

The in vitro cell colony formation assay used a method fully described previously (Salvage et al. 2015), which had been successfully employed to test MPC-DPA systems. Briefly, MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems were prepared at  $0.4 \text{ mg ml}^{-1}$ , with and without the hydrophobic fluorescent probes NR and Cm-6 loaded. The 0.4 mg ml<sup>-1</sup> MPC-DPA PNM systems were syringe filter sterilised (0.22 µm) (Millipore, UK) before preparing a series of halving dilutions in DMEM (2.5 % FBS) to provide polymer concentrations of 200, 100, 50, 25, 12.5, and 6.25  $\mu$ g ml<sup>-1</sup>. The samples were incubated, in triplicate, with V79 cells for 5 days at 37 °C, in 5 % CO<sub>2</sub>, before fixing with 3.7 % w/v formaldehyde (Sigma, UK) for 30 min, and staining with 10 % v/v Giemsa stain (Sigma, UK) for 30 min. The colonies visible in each well were counted, and the cell survival percentage, for each dilution of each sample type, determined by comparison with control cells grown in cell medium only. Control samples of MeOH, MeOH + Cm-6, MeOH + NR, and DMEM (2.5 % FBS) at the concentrations used for the polymer sample preparation were also tested. Triplicate repeats of the assay were performed for data robustness.

#### MTT cell viability assay: Stage 1-comparability

The in vitro 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) (Sigma, UK) cell viability assay was based on previously published methods (Mosmann 1983; Sieuwerts et al. 1995) and carried out in 2 stages of assessment. The first stage to establish the comparability of the MTT assay to the cell colony formation assay for testing MPC-DPA PNM systems using both V79 cells and 3T3 cells. The second stage to further evaluate the MTT assay using 3T3 cells for MPC-DPA testing; primarily with regard to sample volume and concentration. For Stage 1 of the MTT assay using V79 cells, 24 well tissue culture (TC) plates (Nunc, UK) were seeded at a density of 10,000 cells per well (Benoit et al. 1997) in 1 ml of DMEM (10 % FBS) and incubated at 37 °C, in 5 % CO2 for 24 h to facilitate cell attachment. For Stage 1 of the MTT assay using 3T3 cells, 24 well TC plates were seeded at density of 15,000 cells per well (Lee et al. 2011) in 1 ml of DMEM (10 % FBS) and incubated as per the V79 cells. MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems were prepared at 0.4 mg ml<sup>-1</sup>, with and without Cm-6 loaded. The 0.4 mg ml<sup>-1</sup> MPC-DPA PNM systems were filter sterilised (0.22 µm) before preparing a series of halving dilutions in DMEM (10 % FBS) to provide polymer concentrations of 200, 100, 50, 25, 12.5, and 6.25  $\mu$ g ml<sup>-1</sup>. The samples were incubated, in triplicate, with the TC plated V79 and 3T3

cells for 24 h at 37 °C, in 5 % CO<sub>2</sub>. Control samples of MeOH, MeOH + Cm-6, and DMEM (10 % FBS) at the concentrations used for the polymer sample preparation were also tested. The cells were subsequently washed with PBS, and incubated with 1 ml MTT in DMEM  $(0.5 \text{ mg ml}^{-1})$  for 4 h at 37° C in 5 % CO<sub>2</sub>. The MTT medium was then removed, the cells washed with PBS, and 1 ml dimethyl sulfoxide (DMSO) (Sigma, UK) added to each well to dissolve the formazan resulting from the interaction between the living test cells and MTT. 100 µl of the DMSO formazan solution was transferred from each sample well to 96 well clear bottom assay plates (Nunc, UK) and the absorbance measured at 540 nm using a Labsystems Multiskan Ascent 354 microplate reader. The absorbance values of the test samples were compared to those of the control samples to calculate cell viability. Triplicate repeats of the assay were performed for data robustness.

### MTT cell viability assay: Stage 2-evaluation

For Stage 2, evaluation of the MTT assay, 3T3 cells were seeded in 48 well TC plates (Nunc, UK) at a density of 7500 cells per well in 500 µl of DMEM (10 % FBS) and incubated at 37 °C, in 5 % CO2 for 24 h. MPC100-DPA100 PNM systems, with and without Cm-6 loaded, were prepared at 2.0 mg ml<sup>-1</sup>. The 2.0 mg ml<sup>-1</sup> MPC-DPA PNM systems were filter sterilised (0.22 µm) before preparing a series of halving dilutions in DMEM (10 % FBS) to provide polymer concentrations of 1000, 500, 250, 125 and 62.5, and 31.25  $\mu$ g ml<sup>-1</sup>. Following the 24 h for cell attachment, the DMEM was replaced with 500 µl of the test sample dilutions in triplicate, and the cells incubated for a further 24 h at 37 °C, in 5 % CO<sub>2</sub>. Control samples of MeOH, MeOH + Cm-6, and DMEM (10 % FBS) at the concentrations used for the polymer sample preparation were also tested. After 24 h incubation, the cells were washed with PBS, and incubated with 0.5 ml MTT in DMEM (0.5 mg ml<sup>-1</sup>) for 4 h at 37 °C in 5 % CO<sub>2</sub>. After MTT incubation, the wells were washed with PBS, before 200 µl acidic (pH 1.8) isopropanol (IPA) (Sigma, UK) was added to each well to dissolve the formazan. The IPA dissolved formazan (100 µl) was transferred to 96 well assay plates and the absorbance measured at 540 nm. The absorbance values of the test samples were compared to those of the control samples to calculate cell viability. Triplicate repeats of the assay were performed for data robustness.

#### In vitro cellular uptake kinetics

V79 and 3T3 cells were maintained as per the in vitro cellular cytotoxicity testing.

Intracellular uptake kinetics of Nile Red loaded MPC-DPA PNM

V79 cells were seeded into 12 well TC plates (Nunc, UK) at 5000 cells per well, in triplicate, in 1 ml of DMEM (10 % FBS), and incubated at 37 °C, in 5 % CO<sub>2</sub> for 24 h. 3T3 cells were maintained, prepared, and incubated as per the V79 cells, in separate 12 well TC plates.  $MPC_{100}$ DPA<sub>100</sub> PNM systems loaded with NR were prepared and diluted with DMEM (10 % FBS), to produce a final polymer concentration of 200  $\mu$ g ml<sup>-1</sup>. The prepared PNM with NR system was filter sterilised (0.22 µm), and used to replace the DMEM on the V79 and 3T3 cells. The test cells, with NR loaded PNM, were then incubated for a 24 h and 120 h at 37 °C, in 5 % CO<sub>2</sub>. At each incubation time point (24 and 120 h), a set of wells were washed twice with 1 ml PBS, before applying fresh DMEM (10 % FBS) to each well. The cells were then observed for NR fluorescence (Greenspan et al. 1985) with a Leica SP5 confocal laser scanning microscopy (CLSM) (Leica, UK), using an excitation wavelength of 514 nm, and an emission collection wavelength band of 524-674 nm

# Intracellular uptake kinetics of Coumarin-6 loaded MPC-DPA PNM

3T3 cells were seeded into 12 well TC plates at 5000 cells per well, in triplicate, in 1 ml of DMEM (10 % FBS), and incubated at 37 °C, in 5 % CO<sub>2</sub> for 24 h. MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems loaded with Cm-6 were prepared and diluted with DMEM (10 % FBS) to produce a final polymer concentration of 1 mg ml<sup>-1</sup>. The prepared PNM with Cm-6 system, was filter sterilised (0.22  $\mu$ m), and used to replace the DMEM on the 3T3 cells. The cells were then incubated (37 °C, 5 % CO<sub>2</sub>) for a 1, 4, and 24 h to facilitate PNM uptake. At each time point (1, 4, 24 h), a set of cells were washed twice with 1 ml PBS, fresh DMEM (10 % FBS) applied to each well, and the cells observed for Cm-6 fluorescence (Kristoffersen et al. 2014) with CLSM using an excitation wavelength of 488 nm, and an emission collection wavelength band of 500-610 nm, to investigate the cellular uptake kinetics of the Cm-6 loaded MPC<sub>100</sub>-DPA<sub>100</sub> PNM. Control samples of Cm-6 in DMEM (10 % FBS), at the concentrations used for the polymer sample preparation were also tested.

# Intracellular fluorescence persistence of MPC-DPA PNM delivered Cm-6

V79 cells were seeded into 12 well TC plates at 5000 cells per well, in triplicate, in 1 ml of DMEM (10 % FBS), and incubated at 37 °C, in 5 % CO<sub>2</sub> for 24 h. 3T3 cells were prepared as per the V79 cells, in separate 12 well TC



plates. MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems loaded with Cm-6 were prepared and diluted with DMEM (10 % FBS) to produce a final polymer concentration of 200  $\mu$ g ml<sup>-1</sup>. The prepared PNM with Cm-6 system, was filter sterilised  $(0.22 \text{ }\mu\text{m})$ , and used to replace the DMEM on the V79 and 3T3 cells. The test cells, with Cm-6 loaded PNM, were then incubated for a further 24 h at 37 °C, in 5 % CO<sub>2</sub>. Following incubation, the wells were washed twice with 1 ml PBS, and fresh DMEM (10 % FBS) applied to each well. The cells were then observed for Cm-6 fluorescence with CLSM, using an excitation wavelength of 488 nm, an emission collection wavelength band and of 500–610 nm, this was designated time point zero (t = 0). Following removal of the Cm-6 loaded PNM systems from the cells at the initial (t = 0) CLSM observation of Cm-6 fluorescence, incubation (37 °C, 5 % CO<sub>2</sub>) of the V79 and 3T3 cells, in DMEM (10 % FBS), was continued for a further 24 h (t = 24), and 48 h (t = 48). CLSM observation was undertaken at t = 24 and t = 48, to investigate of PNM delivered intracellular persistence Cm-6 fluorescence.

### **Results and discussion**

# Synthesis and characterisation of MPC<sub>100</sub>-DPA<sub>100</sub> block copolymer

Synthesis of the MPC<sub>100</sub>-DPA<sub>100</sub> diblock copolymer was undertaken via ATRP, and characterised with <sup>1</sup>H NMR and GPC. Figure 1a-d displays the <sup>1</sup>H NMR spectra, in CDCl<sub>3</sub>:CD<sub>3</sub>OD (3:1 v/v), the chemical structure, and NMR peak assignments, of the monomers MPC (a), and DPA (b), the ARTP initiator MEBr (c), and the resultant synthesised block copolymer MPC-DPA (d). The NMR peak assignments for MPC-DPA in Fig. 1d were consistent with previous reports of MPC-DPA synthesis (Pearson et al. 2013; Ruiz-Perez et al. 2015), indicating the ATRP synthesis was successful using the MEBr initiator, and MeOH solvent, at ambient room temperature. Regarding the choice of initiator, early studies reported successful utilisation of oligo(ethylene glycol) 2-bromoisobutyrate (OEGBr) for ATRP synthesis of homopolymers and block copolymers, such as PMPC (Lobb et al. 2001) and POEGMA (Wang and Armes 2000), including the first report of MPC-DPA copolymer synthesis (Ma et al. 2003). However, subsequent polymerisation studies utilising OEG and ME initiators (Robinson et al. 2002) indicated that OEG could form one of the copolymer blocks, directly affecting the resultant polymer properties, whilst in contrast, the ME group did not, and instead provided a convenient NMR label. Thus in later reports of MPC-DPA copolymer synthesis (Du et al. 2005) OEGBr, which could have acted as a



third block in the copolymer, was replaced with MEBr due to its small size and lack of effect on the MPC-DPA properties, and hence the utilisation of MEBr in this current work. Whilst the synthesis reported herein was successful, it has been reported elsewhere that MeOH may produce transesterification of methacrylate monomers during ATRP (Bories-Azeau and Armes 2002; Connell et al. 2012), and as such future studies might consider using ethanol (EtOH) as an alternative solvent to MeOH, although this may have an effect on the ATRP kinetics. As per the method of ATRP polymer synthesis utilised (Ma et al. 2003), and modified for this study, MPC<sub>100</sub> polymerisation was monitored via <sup>1</sup>H NMR until circa 98 % complete. The residual MPC monomer was preferentially consumed and polymerised after DPA monomer addition due to the higher reactivity of the MPC monomer, thus attaining a degree of polymerisation (DP) of 100 for MPC. The subsequent DP achieved for DPA, and the Mn(NMR) for the MPC-DPA copolymer, was confirmed with <sup>1</sup>H NMR by comparing the integrals of the methylene group protons in the MPC block at chemical shift 4.0 ppm, with the methine group protons in the DPA block at chemical shift 3.0 ppm, and methylene group protons in the DPA block at chemical shift 2.6 ppm, seen in Fig. 1d as peak assignments f, l, and j, respectively. This indicated a DP of 100 was attained for DPA, with the Mn(NMR) of 50,860 for the synthesised MPC<sub>100</sub>-DPA<sub>100</sub> block copolymer being equal to the target Mn (g  $mol^{-1}$ ).

The GPC elution profile data, for organic and aqueous GPC analysis of  $MPC_{100}$ -DPA<sub>100</sub> (Fig. 2) confirmed that appropriate choice of GPC system is critical for accurate GPC analysis of polymers to be achieved, as it can be seen that a single polymer can produce different elution profiles in different mobile phases. The molecular weight determinations and polydispersity are summarised in Table 1.

The aqueous GPC resulted in a broad elution profile (Fig. 2), similar to other reports of MPC polymer aqueous GPC analysis (Yamada et al. 2015), and aqueous Mw(GPC), Mn(GPC), and Mw/Mn (polydispersity) values of 63,170, 38,950, and 1.62, respectively. In contrast, the organic GPC produced a sharp elution profile peak, and organic Mw(GPC), Mn(GPC), and Mw/Mn (polydispersity) values of 58,150, 56,080, and 1.04, respectively. Additionally, the aqueous GPC used an RI detector, which are more susceptible to back pressure and temperature drift effects, and of lower sensitivity, than an ELS detector (Petritis et al. 2002), as employed herein for the organic GPC, and reflected by the reduced sample injection volumes and concentrations required. However, whilst the organic GPC appeared better suited for accurate analysis of the MPC-DPA polymer, than aqueous GPC, the organic Mn(GPC) value attained (56,080) differed from the Mn(NMR) value (50,860). This effect has been described in other polymer synthesis studies (Stenzel et al. 2004),



Fig. 1 <sup>1</sup>H NMR spectre, peak assignments, and chemical structure for **a** MPC, **b** DPA, **c** MEBr, and **d** MPC<sub>100</sub>-DPA<sub>100</sub>, in CDCl<sub>3</sub>:CD<sub>3</sub>OD (3:1 v/v)



Fig. 2 GPC elution profiles for  $MPC_{100}$ - $DPA_{100}$  illustrating the effect of mobile phase type (organic vs. aqueous)

with NMR to GPC Mn differences of over 100 % evident in some cases (Lin et al. 2014). Indeed, the aqueous GPC to NMR variance herein was circa 30 %, and other reports suggest that consistent MPC-DPA Mn(GPC) determination has also proven challenging previously (Du et al. 2005; Lomas et al. 2010; Madsen et al. 2011, 2013; Yu et al. 2013), with apparent NMR to GPC differences of over 50 % in some reports (Du et al. 2005; Madsen et al. 2013; Yu et al. 2013), but in this current study it was relatively low at circa 10 % for the organic GPC analysis. The variance observed may be attributed to differences in the properties of the GPC standards compared to the test polymer, and the choice of mobile phase additives, particularly with MPC where intra- and inter-molecular electrostatic interactions are thought to produce hydrodynamic sizes that differ from those predicted (Mahon and Zhu 2008; Yamada et al. 2015), which is also applicable to GPC analysis of other polymer types (Lee and Chen 2010). Additionally, previous GPC analysis of MPC-DPA polymers has often utilised poly(methyl methacrylate) (PMMA) as the GPC calibration standards (Du et al. 2005; Lomas et al. 2010; Madsen et al. 2011, 2013). However, PMMA is hydrophobic and has low solubility in polar solvents such as MeOH. Moreover, polymer characterisation studies have often added varying amounts of lithium bromide (LiBr) to



Table 1 Molecular weight (Mn and Mw) and polydispersity (Mw/ Mn) of block copolymer  $MPC_{100}\text{-}DPA_{100}$  determined via  $^1\text{H}$  NMR and GPC

| MPC <sub>100</sub> -DPA <sub>100</sub> | Mn     | Mw     | Mw/Mr |
|----------------------------------------|--------|--------|-------|
| Target (g mol <sup>-1</sup> )          | 50,860 | _      | _     |
| <sup>1</sup> H NMR                     | 50,860 | _      | _     |
| GPC (Organic)                          | 56,080 | 58,150 | 1.04  |
| GPC (Aqueous)                          | 38,950 | 63,170 | 1.62  |
|                                        |        |        |       |

the mobile phase (Stenzel et al.; 2004, Du et al. 2005; Lomas et al. 2010; Madsen et al. 2011, 2013; Penfold et al. 2015) to modify, shift, and sharpen the GPC elution profile (Coppola et al. 1972; Hann 1977; Dubin et al. 1977). Although this is an accepted GPC practice, it runs the risk of under or over compensation, with detector signal suppression being a potential consequence of the latter. For this study, non-volatile LiBr was not compatible with the ELS detector, and instead AcAc and TFA were added to the CHCl<sub>3</sub>:MeOH organic mobile phase to modify the pH for improved DPA compatibility. Additionally, PEG and PEO standards were employed, which, being amphiphilic, were soluble in both the organic and aqueous mobile phases, and thus more reflective of the hydrodynamic properties and hydrophilic nature of the MPC-DPA polymer, which may therefore account for the apparent improvement in GPC accuracy seen herein.

In summary, it is clear that appropriate and accurate GPC analysis of polymers presents unique challenges, and indeed, other studies have suggested GPC should only be seen as an Mw and Mn estimate (Yusa et al. 2005). Moreover, some reports have concluded that not all polymers are suitable for GPC analysis and instead have relied solely on NMR for Mn determination (Chu et al. 2009; Lee and Chen 2010) due to polymer hydrodynamic size, rather than molecular weight, influencing the GPC retention time (Yamada et al. 2015). However, in this current work the GPC analysis was effective, the data was comparable with NMR, and correlated well with the target molecular weight, the resultant organic GPC peak profile for MPC<sub>100</sub>-DPA<sub>100</sub> (Fig. 2) was narrow and sharp, and the GPC(Mw/Mn) polydispersity (1.04) was low, indicating that the synthesised MPC<sub>100</sub>-DPA<sub>100</sub> polymer was very well defined and of very good purity.

# Preparation of MPC<sub>100</sub>-DPA<sub>100</sub> polymeric nanoparticle micelle systems

*Polymer concentration, freeze-thaw, and freeze-dry rehydrate effect* 

Nanoprecipitation offers the potential for the controlled and predictive assembly of biocompatible nanoparticles for





**Fig. 3** Particle diameter (*outer bar*) and polydispersity (*inner bar*), measured with PCS, of MPC<sub>100</sub>-DPA<sub>100</sub> micelles before (Control), and after, freeze–thaw (FT), and freeze-dry (FD) treatments (Mean  $\pm$  SD, n = 3)

drug delivery based nanomedicine applications (Lepeltier et al. 2014), and the previous nanoprecipitation study of MPC<sub>100</sub>-DPA<sub>100</sub> PNM demonstrated the potential for monodisperse nanoparticle formation (Salvage et al. 2015). Therefore, further studies investigating the effect of increasing polymer concentration, and the associated increase in MeOH volume, used during nanoprecipitation of MPC-DPA PNM were undertaken, together with the effects of post-nanoprecipitation FT and FD treatments. It can be seen from Fig. 3 that increasing polymer concentration, and associated MeOH volume relative to PBS, had an apparent effect on the overall  $D_{\rm h}$  of the Control (CT) systems. The CT samples were analysed post-nanoprecipitation, and without further treatment, and displayed increasing  $D_{\rm h}$  values ranging from circa 64 nm to 69 nm relative to increasing polymer concentration, which was in agreement with reports of nanoprecipitated particle sizes shifting with changes in polymer concentration and solvent volumes (Schubert and Muller-Goymann 2003; Bilati et al. 2005). However, whilst these earlier studies were indicative of possible particle size trends, it must also be taken into consideration that the nanoparticles formed are not directly comparable to the dynamic and fluid micelle systems self-assembled by MPC-DPA polymers, where micelle assembly is spontaneously driven by the hydrophobic effect and electrostatic forces (Tanford 1974; Kronberg et al. 1994), a multifaceted interaction that is not fully understood (Meyer et al. 2006). However, subsequent reports of nanoprecipitation being used to form micelles from block copolymers (Aliabadi et al. 2007) were consistent with the findings herein, and previously (Salvage et al. 2015), in that the solvent to non-solvent ratio, and the polymer concentration can affect the resultant micelle size, polydispersity, and loading capacity. Moreover, given that choice of solvent influences MPC-DPA PNM size (Salvage

et al. 2015), and MPC-DPA polymers can form a range particle morphologies (Pearson et al. 2013), further study to help elucidate the effects seen would be useful. It can also be seen (Fig. 3) that the Pd values for the CT samples, remained stable and <0.1, thus indicating system monodispersity was maintained. The presence of MeOH in the systems is unlikely to have influenced the PCS, as previous work demonstrated that MPC-DPA PNM size. determined by PCS, remained stable and unchanged following dialysis and dilution to remove solvents (Salvage et al. 2015). The FT samples were subjected to a freezethaw process post-nanoprecipitation, and it can been seen from Fig. 3 that whilst this had little effect on the overall  $D_{\rm h}$  of the samples, the Pd was affected, with all of the FT treated samples displaying an increased Pd, which was greater for the lower (0.4 mg  $ml^{-1}$ ) polymer concentration samples compared to the higher  $(2.0 \text{ mg ml}^{-1})$  concentration samples, relative to the untreated CT samples. The FD samples were subjected to a freeze-dry rehydration process, post-nanoprecipitation, and in contrast to the CT and FT samples, this produced distinct changes in the composition of the systems. As seen in Fig. 3, all of the FD samples displayed increased D<sub>h</sub> and Pd values, relative to the untreated CT and FT treated samples, and irrespective of polymer concentrations tested, which was in agreement with reports that other polymer-based nanoparticle systems suffer stability issues upon freeze-drying (Cao and Jiang 2012; Logie et al. 2014).

When evaluating the results it is important to consider the different physical states each of the sample groups experienced. The CT samples were untreated, and thus remained in a liquid aqueous phase, and hence the  $D_{\rm h}$  and Pd remained stable and monodisperse. The FT samples were frozen (-20 °C), and whilst this could help maintain system stability, ice formation upon freezing may have disrupted the uniformity and integrity of the micelles in suspension, and thus account for the increased Pd observed. Faster, snap freezing, to avoid ice crystal formation, may address the observed increase in Pd for the FT treated samples (Fig. 2), indeed the Cryo-SEM slushy nitrogen (-207 °C) frozen sample appeared relatively monodisperse (Fig. 10). The FD samples were rehydrated from a lyophilised state, which may, to some degree, be considered similar to the manner in which MPC-DPA polymersomes are formed via film rehydration (Pegoraro et al. 2014), and thus resulted in systems with an increased  $D_{\rm h}$ and Pd, indicative of a polydisperse sample containing a mixed PSD (Woodbury et al. 2006; Ansari and Nyeo 2012).

The 50 and 200 nm polystyrene standard sample, shown in Fig. 4a illustrates how Cumulants analysis fits a Gaussian distribution to the PCS intensity data, and reports a single average intensity based diameter size ( $D_h$ ) figure for the system, which whilst appropriate for monodisperse systems, does not truly reflect the population size distributions that may be contained within a polydisperse sample. For monodisperse samples, NNLS provides no additional information, but for polydisperse systems, NNLS can help to obtain optimal PSD information from the intensity based monomodal Cumulants data (Ansari and Nyeo 2012), and will thus provide a more realistic indication of sample consistency. This is clearly illustrated by the 50 and 200 nm standard peaks resolved with NNLS from a mixed polydisperse sample seen in Fig. 4a. Subsequent NNLS analysis of the FT samples (Fig. 4g-k) revealed that the observed increase in Pd relative to decreasing polymer concentration (Fig. 3), was due to the presence of two particle populations, seen as two peaks (Fig. 4g, h) at the lower polymer concentrations (0.4 and  $0.8 \text{ mg ml}^{-1}$ ), with a transitional phase between polydisperse to monodisperse when the polymer concentration was  $1.2 \text{ mg ml}^{-1}$  (Fig. 4i). Similarly, when the polydisperse FD treated sample data was deconvolved with NNLS (Fig. 41-p), it was clear that two discrete particle populations had also formed in the FD systems, the diameters of which were consistent with MPC-DPA mixed micelle and polymersome systems previously reported (Pearson et al. 2013). It should be noted, that when considering PCS sizing data, the size of the intensity distribution peaks should not be treated as an indication of the number contribution to overall sample composition, as the Rayleigh approximation indicates that the intensity of scattered light is proportional to particle diameter to the power of six  $(D^6)$ (Lucyanna et al. 2011), and thus the significance of larger particles is over represented.

In summary, the data (Figs. 3 and 4) suggest that the MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems prepared with increasing polymer concentrations were stable with regards to  $D_h$  and Pd, and additionally, afforded a level of Cryo-protection and system stability, which was consistent with other reported studies of FT and FD polymeric nanoparticle systems (Date et al. 2010; Logie et al. 2014), and may be useful for improved storage and distribution of systems applied, for example, as stimuli (pH) responsive nanomedicines (Crucho 2015).

### Loading with the fluorescent probes Nile Red and Coumarin-6

The MPC<sub>100</sub>-DPA<sub>100</sub> PNM system was loaded with the hydrophobic dyes NR and Cm-6 via nanoprecipitation, and resulted in clearly visible increases (Fig. 5) in aqueous solubility for both NR (Fig. 5a, b), and Cm-6 (Fig. 5c, d). Loading micelles with hydrophobic compounds, such as NR and Cm-6, can result in particle size changes (Sharma et al. 2008), when the compounds preferentially partition





Fig. 4 PCS spectre ( $25^{\circ}$  C) of MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems (PBS, pH 7.4) formed via nanoprecipitation, illustrating the effect of polymer concentration, freeze-thaw, and freeze-dry rehydrate treatments on particle population size and distribution

into the hydrophobic environment of the micelle cores (Marzio et al. 2008), and can also result in improved micelle stability (Kataoka et al. 2001). In this study loading with either NR or Cm-6, at a polymer concentration of 0.4 mg ml<sup>-1</sup>, produced an apparent (circa 1–2 nm)

increase in the  $D_h$  of the PNM system (Fig. 6), with the Pd values remaining monodisperse, which was consistent with previous reports of both MPC-DPA (Licciardi et al. 2006) and other polymer based (Aliabadi et al. 2007; Sezgin et al. 2006) micelle size alterations upon hydrophobic loading.





Fig. 5 Samples loaded with NR ( $\mathbf{a}$ ,  $\mathbf{b}$ ) and Cm-6 ( $\mathbf{c}$ ,  $\mathbf{d}$ ), via nanoprecipitation (MeOH) into PBS (pH 7.4), illustrating the improved solubility of NR ( $\mathbf{b}$ ) and Cm-6 ( $\mathbf{d}$ ) in the presence of MPC<sub>100</sub>-DPA<sub>100</sub> PNM



**Fig. 6** Particle diameter (*outer bar*) and polydispersity (*inner bar*), measured with PCS (25° C), of MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems (PBS, pH 7.4), formed via nanoprecipitation (MeOH-PBS), illustrating the effect of loading with NR and Cm-6 on the  $D_{\rm h}$  and Pd (Mean  $\pm$  SD, n = 3)

### Coumarin-6 solubility in Methanol

The solubility of Cm-6 in MeOH was determined by constructing a standard curve (Tan et al. 1976) which is shown in Fig. 7. The  $\lambda_{max}$  measured for Cm-6 in MeOH was 457 nm, and was consistent with supplier (Acros, UK) specifications. The coefficient of determination ( $R^2$ ) value for the standard curve (Fig. 7) was 0.9995, suggesting the data possessed a good linear fit, and the Cm-6 solubility in MeOH was calculated to be 0.3 mg ml<sup>-1</sup>, which was of the same order as Cm-6 solubility in CHCl<sub>3</sub> (Davda and Labhasetwar 2002), and significantly higher than the supplier (Sigma, UK) reported solubility in water of 0.25 µg ml<sup>-1</sup>.



Fig. 7 Standard curve of Cm-6 solubility in MeOH at 457 nm



**Fig. 8** Particle diameter (*outer bar*) and polydispersity (*inner bar*), measured with PCS (25° C), of MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems (PBS, pH 7.4), formed via nanoprecipitation (MeOH-PBS), illustrating the effect of increasing polymer concentration, and Cm-6 loading, on the  $D_{\rm h}$  and Pd (Mean  $\pm$  SD, n = 3)

#### Loading with the fluorescent probe Coumarin-6

Having determined the solubility of Cm-6 in MeOH a series of MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems were prepared, via nanoprecipitation (MeOH-PBS), with and without Cm-6 loaded, across a range of polymer concentrations  $(0.4-2.0 \text{ mg ml}^{-1})$  to determine the effect on  $D_h$  and Pd of the systems, and the loading efficiency of the systems. It can be seen from Fig. 8, that the effect noted and discussed previously in Fig. 6, whereby  $D_h$  increased by circa 1–2 nm upon Cm-6 loading, was also evident for the higher polymer concentration samples. Thus, Cm-6 loading of the PNM produced  $D_h$  values ranging from circa 65 to 71 nm relative to increasing polymer concentration, whilst the Pd values again remained monodisperse, indicating good system stability.

The level of Cm-6 loading achieved with the  $MPC_{100}$ -DPA<sub>100</sub> PNM systems was determined from the standard curve (Fig. 7) and it can be seen from Fig. 9 that Cm-6





**Fig. 9** Cm-6 loading efficiency (%) (*outer bar*) and Cm-6 loading weight (%) (*inner bar*) achieved for MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems (PBS, pH 7.4), formed via nanoprecipitation (MeOH-PBS), illustrating the effect of increasing polymer concentration (Mean  $\pm$  SD, n = 3)

loading efficiency (LE %) ranged from circa 57–64 % relative to increasing polymer concentration, whilst the Cm-6 loading weight (LW %) ranged from 0.43 to 0.48 % (circa 4–5  $\mu$ g Cm-6 per mg of polymer) relative to increasing polymer concentration, which should be viewed in context of the theoretical maximum weight loading of 0.75 % (7.5  $\mu$ g Cm-6 per mg of polymer), due to the amount of Cm-6 utilised.

 $LE\% = [achieved loading/theoretical loading] \times 100$ 

LW  $\% = [achieved loading/weight of PNM] \times 100$ 

Due to the principles of the nanoprecipitation method employed (Salvage et al. 2015), the weight loading was restricted by the solubility of Cm-6 in MeOH, and thus despite achieving circa 60 % LE, which was comparable to previous reports of LE for MPC-DPA based polymersomes (Colley et al. 2014), the LW % achieved (circa 0.5 %) was relatively low in contrast compared to other MPC-DPA micelle reports (Giacomelli et al. 2006; Licciardi et al. 2006, 2008). The LW % (drug:polymer) achievable in this study was however restricted, and thus lower than previous reports, because the starting w/w % ratio (drug:polymer) herein was 1:133, whilst comparable previous work (Giacomelli et al. 2006; Licciardi et al. 2006, 2008) had a w/w % preparation ratio of circa 1:10 (drug:polymer), and thus LW % achieved was also tenfold higher at circa 5 %. Alternative MPC containing polymer micelles (Chu et al. 2009) have also reported higher LW % being achieved, but again the preparation ratio of drug to polymer was higher at 1:5. However, on a pro-rata basis (circa 0.5 vs. 5 %) the MPC<sub>100</sub>-DPA<sub>100</sub> PNM system tested herein performed comparably to other MPC-DPA systems. Also, given the desire to increase the efficacy of therapeutic compounds, whilst reducing the amount of drug required, this lower LW



may be of advantage. It would be interesting to reduce the starting polymer concentration by tenfold, down to 40  $\mu$ g ml<sup>-1</sup>, with 3  $\mu$ g ml<sup>-1</sup> Cm-6 (1:13 drug:polymer), to examine whether LE was maintained, and thus LW increased. However, that may also result in a reduced number of PNM in the system, which would act to attenuate attempts to increase LW %. Additionally, it has been shown that the drug to polymer ratio can indeed have a marked effect on LE % and LW % achieved, together with influencing the resultant micelle size (Sezgin et al. 2006). This would be in agreement with previous reports of  $MPC_{100}$ - $DPA_{100}$  PNM (Salvage et al. 2015), where micelle size was increased by over 100 % at a high polymer to drug ratio. Despite the apparently low LW % achieved, compared to other MPC-DPA reports, when the current data is compared to reports of alternative nanoparticle systems loaded with Cm-6 (Trapani et al. 2009) and micelles loaded with cyclosporine A (Aliabadi et al. 2007), the results of this study compare favourably and indeed exceed the previous LW and LE levels reported. It may also be interesting to investigate nanoprecipitation using alternative solvents to determine whether LE and LW can be increased, however it is noted that previous reports of solvent effects on MPC polymers (Lewis et al. 2000; Edmondson et al. 2010; Salvage et al. 2015) indicate that this may also alter the size and morphology of the resultant nanoparticles.

In summary, the MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems have been successfully loaded with hydrophobic model compounds in the form of the fluorophores NR (Fig. 6) and Cm-6 (Fig. 8), with the  $D_h$  and Pd of the PNM systems remaining stable, and good loading efficiency achieved (Fig. 9), indicating further their potential application as polymeric nanomedicine vectors (Khemtong et al. 2009).

### Particle size characterisation

Particle size characterisation of the MPC<sub>100</sub>-DPA<sub>100</sub> micelles was undertaken using PCS, and Cryo-SEM. The data from the PCS analysis of the PNM systems (Figs. 6 and 8) indicated that the PSD were monodisperse (Pd < 0.1) and displayed  $D_{\rm h}$  values of circa 64–69 nm (PNM) and 65-71 nm (PNM + NR/Cm-6). The Cryo-SEM image data (Fig. 10) was in agreement with the PCS data (Fig. 6), displaying a uniform PSD and morphology, with PNM diameters of circa 65 nm. The observed morphology (Fig. 10) of these MPC-DPA PNM was consistent with the physical shape and appearance of other Cryo-SEM imaged micelle systems (Issman and Talmon 2012; Basak and Bandyopadhyay 2013). A PCS investigation of the effects of freezing, and freeze drying, upon the PNM systems revealed a marked effect on the PSD of the MPC<sub>100</sub>- $DPA_{100}$  system, as seen in Fig. 4, which was relative to polymer concentration and treatment, as discussed earlier.



Fig. 10 Cryo-SEM image of MPC<sub>100</sub>-DPA<sub>100</sub> polymeric nanoparticle micelles displaying diameters of circa 65–70 nm. *Scale* bar = 200 nm



Fig. 11 V79 cell colony formation in vitro cytotoxicity assay of the MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems (PBS, pH 7.4) with and without NR and Cm-6 loaded. Percent survival derived from comparison to cell medium control wells. Samples incubated for 5 days at 37 °C in 5 %  $CO_2$  (Mean  $\pm$  SD, n = 3)

#### In vitro cellular cytotoxicity testing

#### Cell colony formation assay

Cell colony formation assays are a well-established technique for determining cell survival (Franken et al. 2006), a variant utilising V79 cells has previously been used to test MPC-DPA PNM (Salvage et al. 2005, 2015), and in this study was employed further for the investigation of MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems with the addition of loaded hydrophobic compounds (NR and Cm-6). The data seen in Fig. 11 was consistent with previous MPC-DPA data, in that the PNM alone were essentially non-toxic at the concentrations tested (6.25–200 µg ml<sup>-1</sup>). However, it was found that the NR loaded MPC-DPA PNM samples, and MeOH-NR samples, displayed reduced cell survival rates of circa 50 and 80 %, respectively, at 200 µg ml<sup>-1</sup>. The data (Fig. 11) would suggest that the reduced cell survival was due to the NR, as the MPC-DPA, and MeOH controls did not produce a reduction. The use of MeOH in nanoprecipitation, and the associated risks of MeOH toxicity, have been discussed in depth previously (Salvage et al. 2015). Additionally, the data at 200  $\mu$ g ml<sup>-1</sup> also suggested that loading NR into MPC-DPA PNM, increased the apparent NR toxicity, possibly due to the more effective intracellular delivery of the NR by the PNM, the occurrence of which is supported by Fig. 15. Such an affect would be consistent with previous reports of enhanced delivery utilising MPC-DPA polymersome based nanoparticle systems (Pegoraro et al. 2013). In contrast, the samples containing Cm-6 displayed a much greater level of cell toxicity, with an apparent reduction to circa 80 % survival at 25  $\mu$ g ml<sup>-1</sup>, and 0 % survival from 50–200  $\mu$ g ml<sup>-1</sup> concentration, with cellular uptake of Cm-6 supported by Fig. 17. Previous work on MPC-DPA PNM dilution stability (Salvage et al. 2015), suggests that the PNM were above their critical micelle concentration (CMC) at the levels employed and tested for this study. There have been other reports of NR (Lin et al. 2009; Snipstad et al. 2014; Drozdek and Bazylinska 2015) and Cm-6 (Davda and Labhasetwar 2002; Sun et al. 2010; Rivolta et al. 2011; Tang et al. 2011; Zhang et al. 2014; Drozdek and Bazylinska 2015) nanoparticle mediated intracellular delivery, often without toxicity issues, however the extended period of the Cm-6 exposure involved in the V79 colony formation assay of 5 days (120 h), may have served to augment the apparent NR and Cm-6 toxicity seen herein. The V79 colony formation assay, whilst useful as an initial gauge of potential cytotoxicity is limited by its design, in that the low seeding density employed is suboptimal for many cell lines, and can result in cell death or impaired cell growth (Rubin et al. 1995), hence an MTT assay is often employed instead (Nikzad and Hashemi 2014).

#### MTT cell viability assay: Stage 1-comparability

It has been suggested that Coumarin compounds have the potential for therapeutic applications (Kontogiorgis et al. 2012), and thus further work for this study focused on the Cm-6 systems, with the MTT assay (Mosmann 1983; Sieuwerts et al. 1995) employed to facilitate comparative testing between V79 cells and 3T3 cells, both of which are well established and widely used in vitro research cell lines. The V79 MTT assay data (Fig. 12) was reflective of, and consist with, the V79 colony assay data (Fig. 11) where the MPC-DPA PNM, and MeOH, controls, were essentially non-toxic, whilst the Cm-6 loaded MPC-DPA PNM and MeOH-Cm-6 samples produced an apparent reduction in cell viability at polymer test concentrations of 50–200  $\mu$ g ml<sup>-1</sup>. The Cm-6 associated reduction in V79





**Fig. 12** V79 cell MTT in vitro cytotoxicity assay of the MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems (PBS, pH 7.4) with and without Cm-6 loaded. Percent viability derived from comparison to cell medium control wells. Samples incubated for 24 h at 37 °C in 5 % CO<sub>2</sub> (Mean  $\pm$  SD, n = 3)



**Fig. 13** 3T3 cell MTT in vitro cytotoxicity assay of the MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems (PBS, pH 7.4) with and without Cm-6 loaded. Percent viability derived from comparison to cell medium control wells. Samples incubated for 24 h at 37 °C in 5 % CO<sub>2</sub> (Mean  $\pm$  SD, n = 3)

cell viability to circa 70-80 % seen in Fig. 12, was less marked than the reduction to 0 % V79 cell survival seen in Fig. 11. However, the MTT assay involved a much shorter sample incubation time of 24 h, compared to 120 h for the colony assay, which may have attenuated the apparent Cm-6 cellular toxicity towards the V79 cells seen in Fig. 11. In contrast to the V79 MTT data (Fig. 12), when the samples were tested with 3T3 cells using the MTT assay (Fig. 13) the apparent toxicity of Cm-6 was much less evident. The control samples of MPC-DPA PNM, and MeOH, were again essentially non-toxic, and consistent with the results seen in Figs. 11 and 12, whilst the only discernible reduction in cell viability for the Cm-6 containing samples, to circa 70 %, was at the highest polymer concentration level (200  $\mu$ g ml<sup>-1</sup>) tested. This apparent difference between V79 and 3T3 cell sensitivity to Cm-6, is consistent with other reports of in vitro response variability between V79 and 3T3 cells (Sivak et al. 1982) and in this instance may be due to the reported differences in kinase C levels





**Fig. 14** 3T3 cell MTT in vitro cytotoxicity assay of the MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems (PBS, pH 7.4) with and without Cm-6 loaded. Percent viability derived from comparison to cell medium control wells. Samples incubated for 24 h at 37 °C in 5 % CO<sub>2</sub> (Mean  $\pm$  SD, n = 3)

between V79 and 3T3 cells (Adams and Gullick 1989), as kinase C has been linked to endocytosis regulation (Le et al. 2002), and thus PNM uptake levels. Additionally, it has also been reported that the toxicity of Coumarin compounds varies between species (Lake and Grasso 1996) sometimes profoundly (Born et al. 2000), and given that V79 cells are of hamster origin, whilst 3T3 cells are of mouse, this may also have influenced the results herein. However, when Figs. 11, 12, and 13 are comparable to the colony formation assay for effective detection and quantification of cellular toxicity, and the data was consistent with the CLSM data for the Cm-6 exposed cells seen in Figs. 16 and 17.

#### MTT cell viability assay: Stage 2-evaluation

Given the apparent sensitivity of V79 cells to Cm-6 (Figs. 11 and 12), further Stage 2 MTT assay evaluation focused on 3T3 cells to facilitate testing of the PNM systems at higher preparation concentrations, smaller sample volumes, and with an alternative solvent for the formazan. The reduced scale being desirable for efficiency, and both DMSO and IPA have been reported as MTT assay solvents (Mosmann 1983; Sieuwerts et al. 1995). The results of the Stage 2 MTT testing of the MPC-DPA PNM systems, with and without Cm-6 loaded, can be seen in Fig. 14. The data were consistent with the lower concentration, larger volumes tested (Fig. 13), in that polymer concentrations below 250  $\mu$ g ml<sup>-1</sup> were essentially non-toxic. However there was an apparent increase in the toxicity, with cell viability reduced to circa 60 %, for the MPC-DPA PNM, with Cm-6 loaded, at the higher polymer concentration of 500  $\mu$ g ml<sup>-1</sup>, shown in Fig. 14. This was similar to the effect seen in Figs. 11 and 12, and as discussed previously may be due to enhanced intracellular delivery of the Cm-6 Fig. 15 Confocal laser scanning microscopy images of V79 and 3T3 cells incubated with MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems (PBS, pH 7.4), loaded with NR, for 24 and 120 h, displaying intracellular fluorescence indicative of PNM uptake. *Scale bar* = 25  $\mu$ m



by the MPC-DPA PNM system. At the highest concentration tested, 1 mg ml $^{-1}$ , all samples produced a reduction in 3T3 cell viability, ranging from circa 60-70 %, but remained above IC<sub>50</sub> levels (Sieuwerts et al. 1995), this was possibly the result of initial DMEM dilution with the test samples (Salvage et al. 2015), as other MPC-DPA based polymers have demonstrated low cellular toxicity at up to  $5 \text{ mg ml}^{-1}$  in polymersome form (Lomas et al. 2008). Overall the data in Fig. 14 was consistent with Fig. 13, indicating the use of smaller test volumes, and acidified IPA in place of DMSO, was acceptable and effective. With regard to the direct comparison of the two assay types, MTT determines cell viability by testing inhibition of cell growth and inducement of cell death, whilst the colony assay determines cell survival based on the ability of a cell to grow into a colony. Additionally, it is also important to consider that the colony formation assay is more prone to user variability due to the subjective nature of visually counting colonies compared with the colorimetric MTT assay, and hence it is not unusual to find papers employing both techniques to characterise novel materials (Gao et al. 2012), indeed there are papers dedicated to exploring the comparisons (Nikzad and Hashemi 2014, Fotakis and Timbrell 2006), however in this current study the colony formation assay and the MTT assay data were in general agreement.

#### In vitro cellular uptake kinetics

## Intracellular uptake kinetics of Nile Red loaded MPC-DPA PNM

Upon PNM delivery of NR to V79 and 3T3 cells, there was visible NR fluorescence evident with CLSM in both the



cells types after 24 h exposure, as seen in Fig. 15, with concentrated areas of NR florescence within the cellular cytoplasm, indicative of successful PNM uptake and NR delivery. This was consistent with, and an improvement upon, the previous findings of intracellular uptake after 48 h (Salvage et al. 2015). Due to the hydrophobic and lipophilic nature of NR, it is probable that the NR may have undergone intracellular lipid interactions (Diaz et al. 2008), with the NR fluorescence primarily contained within the cell cytoplasm, and generally absent from the nucleus, which the merged images in Fig. 15 appear to support, and is consistent with other reports of intracellular dye localisation (Drozdek and Bazylinska 2015, Stefancíkova et al. 2014). The V79 and 3T3 cells were viable and NR fluorescently stained after both the 24 and 120 h exposure periods, with the 120 h exposure being representative of the 5 days colony assay incubations period, and consistent with the cell survival data seen in Fig. 11, although it was noted that the 3T3 cells appeared to be more active, with pseudopodia visible, in a similar manner to Cm-6 exposed 3T3 cells seen in Fig. 17.

# Intracellular uptake kinetics of Cm-6 loaded MPC-DPA PNM

Cm-6 uptake by 3T3 cells was confirmed by the intracellular fluorescence observed using CLSM, as shown in Fig. 16. The control Cm-6 samples displayed very little fluorescence at 1 h, with that being detectable located primarily outside of the cells either in the surrounding aqueous medium, or on the cell surface due to the hydrophobic nature of Cm-6. After 4 h incubation the extracellular control Cm-6 accumulation and fluorescence appeared to have increased, but there was no evidence of intracellular uptake, which was consistent with similar reports (Zhang et al. 2014), and after 24 h it became apparent that the control Cm-6 was adsorbed to the outer cell surface (Fig. 16) and not internalised, with only minimal Cm-6 fluorescence visible, and consistent with other free Cm-6 cell exposure studies (Drozdek and Bazylinska 2015). In contrast the 3T3 cells exposed to MPC-DPA PNM delivered Cm-6 displayed clear intracellular Cm-6 fluorescence after 1 h exposure, and increasing at 4 and 24 h. This enhanced and rapid 1 h intracellular delivery of MPC-DPA micelles was consistent with previous reports for both MPC-DPA polymersome (Lomas et al. 2008), and other polymeric micelle system (Zhang et al. 2014; Tu et al. 2011) cellular uptake. It is thought that class B cell surface scavenger receptor medicated endocytosis is responsible for PC-based nanoparticle uptake (Boullier et al. 2005; Colley et al. 2014). Interestingly, it has also been suggested that very rapid uptake, occurring within minutes of exposure, may be the result of cell contact



mediated diffusion rather than endocytosis (Snipstad et al. 2014). Examination of the CLSM images (Fig. 16) suggest that, after 1 h incubation, the PNM delivered Cm-6 was initially contained within the cell cortex of the cytoplasm, and did not reach the perinuclear region of the cytoplasm until 4 h incubation had elapsed. This was consistent with other reports of dye localisation following intracellular delivery with micelles (Sun et al. 2010; Stefancíkova et al. 2014), via endocytosis (Sharma and Sharma 1997). After 24 h incubation with PNM delivered Cm-6, the 3T3 cells displayed intense intracellular Cm-6 fluorescence (Fig. 16), but it had remained principally outside of the cell nucleus. The nuclear envelope is a double layered bilayer membrane, as opposed to a single bilayer outer cell membrane (Smover and Jaspersen 2014), which may have prevented Cm-6 nuclear penetration (Zhao et al. 2011). The increasing intensity of Cm-6 fluorescence observed (Fig. 16) over the 1-24 h incubation period, may also have been the result of a time-dependant release of Cm-6 from the MPC-DPA PNM core and intracellular endosomal environment, where Cm-6 fluorescence quenching may have occurred (Bae et al. 2003). Overall, the Fig. 16 Brightfield images indicated the 3T3 cells were healthy, and the fluorescence images suggested that successful, and rapid, intracellular MPC-DPA PNM mediated delivery of Cm-6 had been achieved.

# Intracellular fluorescence persistence of MPC-DPA PNM delivered Cm-6

Following confirmation of MPC-DPA PNM mediated Cm-6 intracellular delivery (Fig. 16), subsequent investigation examined the persistence of the Cm-6 fluorescence in both V79 and 3T3 cells. It can be seen from Fig. 17 that as before, after 24 h incubation, designated t = 0, there was clear PNM delivered intracellular Cm-6 fluorescence in the V79 and 3T3 cells. It was noted that at t = 0 whilst the morphology of the 3T3 cells appeared healthy and normal, the V79 cells, in contrast, displayed irregular cell morphologies, which may have indicated sensitivity to the Cm-6, and would be consistent with the apparent Cm-6 toxicity towards V79 cells seen in Figs. 11 and 12, and discussed earlier. Following t = 0 the cells were washed, and then incubated in fresh cell medium for a further 24 and 48 h, t = 24 and t = 48, respectively, and it can be seen in Fig. 17 that intracellular Cm-6 fluorescence persisted in the V79 and 3T3 cells through to t = 48. Although Cm-6 fluorescence persisted over time, there was also an associated reduction in fluorescence intensity as time elapsed, possibly resulting from exocytosis of the Cm-6 from the cells (Sun et al. 2010; Strobel et al. 2015), or a dilution effect of cellular division, which would be consistent with the apparent recovery of normal cell morphology by the Fig. 16 Confocal laser scanning microscopy images of 3T3 cells incubated for 1, 4, and 24 h with MPC<sub>100</sub>-DPA<sub>100</sub> Cm-6 loaded PNM systems (PBS, pH 7.4), and control Cm-6 systems (PBS, pH 7.4), illustrating both the rapid (1 h) cellular uptake, and enhanced levels of Cm-6 uptake, achieved by the PNM Cm-6 system. *Scale bar* = 25  $\mu$ m



V79 cells seen at t = 48. The 3T3 cells appeared more tolerant of the Cm-6, with normal cell morphology and extending pseudopodia visible at t = 48 indicative of active cellular processes, which was in agreement with the lower toxicity seen in Fig. 13.

In summary, the results of the in vitro cellular toxicity assays (Figs. 11, 12, 13, 14) and the intracellular uptake studies (Figs. 15, 16, 17) indicate that  $MPC_{100}$ -DPA<sub>100</sub> PNM systems were of minimal cellular toxicity, and capable of rapid intracellular delivery, and therefore a good





**Fig. 17** Confocal laser scanning microscopy images of V79 and 3T3 cells incubated for 24 h with MPC<sub>100</sub>-DPA<sub>100</sub> Cm-6 loaded PNM systems (PBS, pH 7.4), displaying the fluorescence seen after incubation (t = 0), and subsequent fluorescence persistence over 24 (t = 24) and 48 (t = 48) hours. *Scale bars* = 125 and 25 µm for fluorescence and zoom images respectively

candidate for live cell imaging applications, providing an appropriate choice of cellularly tolerated fluorophore is made, and thus potentially, theranostic nanomedicine (Cole and Holland 2015) applications.



### Conclusions

In conclusion, this paper reports the synthesis, characterisation, in vitro cellular uptake kinetics, and nanoprecipitation of PC-based polymeric nanoparticle micelles (PNM) from the well-defined, and low polydispersity, diblock copolymer MPC<sub>100</sub>-DPA<sub>100</sub>, as confirmed by <sup>1</sup>H NMR and GPC analysis. It had previously been demonstrated that MPC<sub>100</sub>-DPA<sub>100</sub> PNM systems were biocompatible, dilution, time, and thermally stable, pH stimuli responsive, and formed monodisperse nanoparticle micelles via nanoprecipitation (Salvage et al. 2005, 2015). Herein, this study has expanded upon the prior art to present in this paper the first report of considerable further novel data, including an improved GPC protocol, the apparent Cryo-stability properties of the MPC-DPA block copolymer relative to concentration, the concentration stable and controllable nanoprecipitation of monodisperse PNM with diameters of circa 65-70 nm, Cryo-SEM imaging of the PNM, the successful loading of the hydrophobic fluorophore Cm-6 as a model drug compound into the PNM, with loading efficiencies of circa 60 %, together with the rapid (1 h) intracellular uptake of MPC<sub>100</sub>-DPA<sub>100</sub> PNM delivered Cm-6, with the resultant fluorescence persisting for 48 h. These novel data and findings further strengthen the potential for MPC-DPA PNM systems to be developed for a range of nanoparticle-based applications including hydrophobic therapeutic delivery, diagnostics, and combined multifaceted theranostics, and thus of nanomedicine interest.

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