High surface area mesoporous silica for hydrogen sulfide effective removal

Gomaa Abdelgawad Mohammed Ali^{1,2,*}, Ahmed Barhoum ^{3,4,5}, Vinod Kumar Gupta ⁶, Amr Ahmed Nada ^{5,7}, Heba El–Maghrabi ^{5,8}, Ramesh Kanthasamy ⁹, Essam Ramadan Shaaban ¹⁰, Hamed Algarni ^{11,12}, Kwok Feng Chong ^{1,*}

¹Faculty of Industrial Sciences & Technology, Universiti Malaysia Pahang, Gambang, 26300 Kuantan, Malaysia; ²Chemistry Department, Faculty of Science, Al–Azhar University, Assiut, 71524, Egypt; ³Department of Materials and Chemistry, Vrije Universiteit Brussel (VUB), Pleinlaan 2, 1050 Brussels, Belgium; ⁴Chemistry Department, Faculty of Science, Helwan University, Helwan, Cairo 11795, Egypt; ⁵Institut Européen des Membranes, Université Montpellier 2, CC 047, Place Eugène Bataillon, 34095, Montpellier Cedex 5, France ; ⁶Department of Applied Chemistry, University of Johannesburg, Johannesburg, South Africa; ⁷Department of Analysis and Evaluation, Egyptian Petroleum Research Institute, Cairo, Nasr city P.B. 11727, Egypt; ⁸Catalysis Department, Refining Division, Egyptian Petroleum Research Institute, Cairo, Nasr city P.B. 11727, Egypt; ⁹Faculty of Chemical & Natural Resources Engineering, Universiti Malaysia Pahang, Gambang, 26300, Kuantan, Malaysia; ¹⁰Physics Department, Faculty of Science, Al–Azhar University, Assiut, 71524, Egypt; ¹¹Research Center for Advanced Materials Science (RCAMS), King Khalid University, Abha 61413, P. O. Box 9004, Saudi Arabia; ¹² Department of Physics, Faculty of Sciences, King Khalid University, P.O. Box 9004, Abha, Saudi Arabia

Abstract: Background: Removal of sulfur-containing compounds from the aqueous environment is necessary as these compounds pose potential risks to human health, hygienic management and bring great economic losses due to fouling of resin bed and corrosion of process equipment.

Objective: This work aims to study the H2S removal efficiency using high surface area mesoporous silica (MCM-41).

Method: In this study, mesoporous silica (MCM–41) with a high surface area of 1270 m²/g and high porosity of 69% was prepared by sol-gel technique.

Results: The obtained MCM–41 has exhibited a superior performance in adsorbing H2S from wastewater with a maximum adsorption capacity of 52.14 mg/g. The adsorption isotherm and kinetics of the current adsorption process are best represented by Freundlich isotherm and pseudo-second-order models, respectively.

Conclusion: Therefore, MCM-41 is an excellent adsorbent for wastewater treatment applications.

Keywords: Mesoporous Silica, Sol-Gel, Hydrogen Sulfide, Adsorption, Wastewater treatment.