

University of Nebraska - Lincoln
DigitalCommons@University of Nebraska - Lincoln

Mechanical & Materials Engineering Faculty
Publications

Mechanical & Materials Engineering, Department
of

2008

Control of the nanoscale crystallinity and phase separation in polymer solar cells

Chih-Wei Chu

University of California, Los Angeles

Hoichang Yang

Rensselaer Polytechnic Institute

Wei-Jen Hou

University of California, Los Angeles

Jinsong Huang


University of Nebraska-Lincoln, jhuang@unc.edu

Gang Li

University of California, Los Angeles

See next page for additional authors

Follow this and additional works at: <http://digitalcommons.unl.edu/mechengfacpub>

 Part of the [Mechanics of Materials Commons](#), [Nanoscience and Nanotechnology Commons](#), [Other Engineering Science and Materials Commons](#), and the [Other Mechanical Engineering Commons](#)

Chu, Chih-Wei; Yang, Hoichang; Hou, Wei-Jen; Huang, Jinsong; Li, Gang; and Yang, Yang, "Control of the nanoscale crystallinity and phase separation in polymer solar cells" (2008). *Mechanical & Materials Engineering Faculty Publications*. 181.
<http://digitalcommons.unl.edu/mechengfacpub/181>

This Article is brought to you for free and open access by the Mechanical & Materials Engineering, Department of at DigitalCommons@University of Nebraska - Lincoln. It has been accepted for inclusion in Mechanical & Materials Engineering Faculty Publications by an authorized administrator of DigitalCommons@University of Nebraska - Lincoln.

Authors

Chih-Wei Chu, Hoichang Yang, Wei-Jen Hou, Jinsong Huang, Gang Li, and Yang Yang

Control of the nanoscale crystallinity and phase separation in polymer solar cells

Chih-Wei Chu,^{1,a)} Hoichang Yang,³ Wei-Jen Hou,¹ Jinsong Huang,¹ Gang Li,^{1,2,b)} and Yang Yang^{1,c)}

¹Department of Materials Science and Engineering, University of California, Los Angeles, California 90095, USA

²Solarmer Energy, Inc. El Monte, California 91731, USA

³Rensselaer Nanotechnology Center, Rensselaer Polytechnic Institute, Troy, New York 12180, USA

(Received 8 January 2008; accepted 31 January 2008; published online 11 March 2008)

Grazing-incidence x-ray diffraction and atomic force microscopy were performed on bulk heterojunction regioregular poly(3-hexylthiophene) (RR-P3HT) [6,6]-phenyl-C₇₁-butyric acid methyl esters spin-cast films with different film processing conditions to correlate the crystalline nanostructure of P3HT with the corresponding solar cell performance. The increase in long wavelength absorption for solvent annealed films is related to highly conjugated crystal structure of RR-P3HT phase-separated in the active layer. Upon thermal annealing, the solvent annealed 50-nm-thick device shows high solar cell performance with fill factor up to 73% and power conversion efficiency of 3.80%. © 2008 American Institute of Physics. [DOI: 10.1063/1.2891884]

Bulk heterojunction (BHJ) solar cells¹ based on regioregular poly(3-hexylthiophene) (RR-P3HT) and fullerene have achieved 4–5% power conversion efficiency (PCE).^{2–10} The incident photon conversion efficiency (IPCE) is approaching that of that of their inorganic semiconductor counterparts.^{3,4,6,8} The major limitation is the absorption of the polymers in the solar spectrum. Further progress in materials as well as device architecture⁹ is required.

In this manuscript, grazing-incidence x-ray diffraction (GIXRD) highlights that controlled film processing can tune polymer nanostructure in RR-P3HT/fullerene blend films. High efficiency device was achieved with an optically thin layer (~50 nm thickness). It is possible to further improve polymer solar cells efficiency via multiple device stacking or tandem structure.^{11,12}

For GIXRD, RR-P3HT (Rieke Metals, Inc)/[6,6]-phenyl-C₇₁-butyric acid methyl esters ([70]PCBM)¹³ (Solenne B. V.) (1:1 in weight) films were spin cast (room temperature) on poly(3,4-ethylenedioxythiophene):poly(styrene sulfonate) (PEDOT:PSS)-coated glass substrates, from solutions with 20 mg/ml of both components in either 1,2 dichlorobenzene (DCB) (boiling point, $T_b = 180^\circ\text{C}$) or chlorobenzene (CB) ($T_b = 131^\circ\text{C}$). Synchrotron-based GIXRD procedure is described elsewhere.¹⁴

Figure 1(a) shows 2D GIXRD patterns of P3HT/[70]PCBM films from solutions in DCB (for 1 and 2) and CB (for 3 and 4). For the sample 2, DCB is completely evaporated during spin casting at 1000 rpm for 90 s, while a relatively short time of 30 s provides solvent-annealing effect on the sample 1 still containing small amounts of the solvent residue. The film thickness is ~150–160 nm. In the 2D GIXRD patterns of the two samples, intense reflections

of (*h*00) and (010) crystal planes of RR-P3HT are indicated along the q_z and q_{xy} axis, respectively, corresponding to π -conjugated planes in RR-P3HT crystals, where hexyl side chains of RR-P3HT have an edge-on structure with respect to the substrate. In contrast, fast solvent evaporation conditions by relatively volatile CB induce less-ordered crystalline structure of RR-P3HT in the sample 3 (1000 rpm, 90 s) and 4 (3000 rpm, 90 s), as confirmed in AFM topography showing less crystalline RR-P3HT [Fig. 1(d)]. In addition, samples 3 and 4 contain high portion of kinetically favorable face-on structure of RR-P3HT crystals (i.e., the side chain parallel to the substrate), as confirmed by intense reflection of (010) crystal planes (with a layer spacing of 3.81 Å) along the q_{xy} axis in the 2D GIXRD patterns.^{14,15} From one-dimensional (1D) out-of-plane X-profiles for the P3HT/[70]PCBM films examined [Fig. 1(b)], we found that sample 1 with the longest crystal growth time has the highest crystallinity, although the edge-on molecular ordering becomes broader by crystal growth of RR-P3HT in a pseudosolidlike state [Fig. 1(c)]. An increase in polymer interlayer spacing is clearly observed—16.3 Å in 1, 16.5 Å in 2, 16.8 Å in 3, and 16.9 Å in 4. As a result, the controlled evaporation rate of high boiling solvent is one strategy to induce highly ordered and crystalline RR-P3HT in the blend film. This provides freedom to make high efficient solar cells with wide various film thickness.

The polymer solar cells in this study were constructed on indium tin oxide (ITO) coated glass substrates modified by PEDOT:PSS. In order to systematically control solvent evaporation rate, RR-P3HT/[70]PCBM (20:20 mg/ml) was dissolved in 1,2,4-trichlorobenzene (TCB) with a high T_b of 214.4 °C. The active layer was deposited by spin casting at 3000 rpm (~50 nm, measured by Dektak profilometer). The slow growth process is carried out by covering active layer in a petri dish immediately after spin casting for 50 s. By spin casting at 3000 rpm for 180 s, the film was completely dry and hereby still referred as “fast growth.” Thermal annealing is at 110 °C for 15 min. The absorption measurement was performed using Varian Cary 50 UV-visible spectrophotometer. 25 nm of Ca and 100 nm of Al were thermally evapo-

^{a)}Present address: Research Center for Applied Sciences, Academia Sinica, Nankang, Taipei, Taiwan 11529.

^{b)}Author to whom correspondence should be addressed. Electronic mail: gangli@solarmer.com.

^{c)}Author to whom correspondence should be addressed. Electronic mail: yangy@ucla.edu.

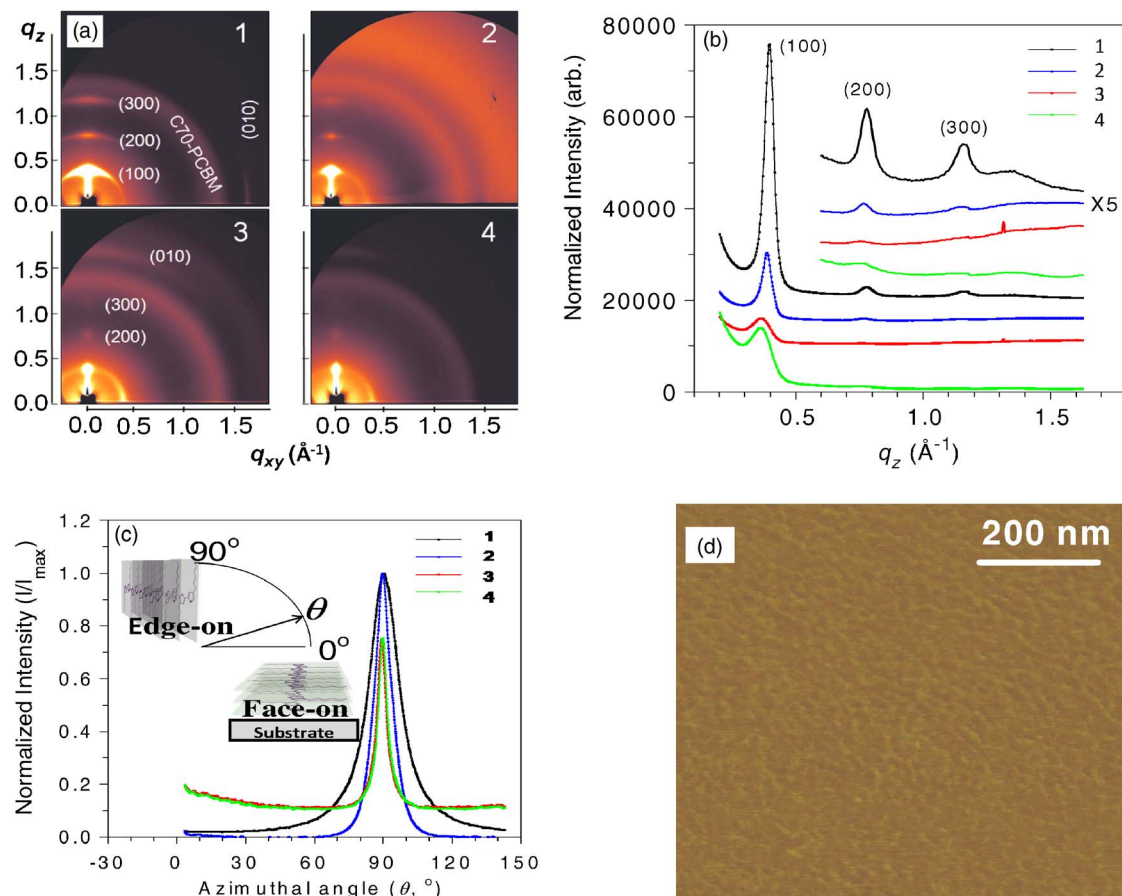


FIG. 1. (Color online) (a) 2D GIXRD patterns of four RR P3HT/[C70]PCBM (1:1 ratio) films [(1) DCB, 1000 rpm 30 s; (2) DCB, 1000 rpm 90 s; (3) CB, 1000 rpm 90 s; (4) CB, 3000 rpm, 90 s]. (b) 1D out-of-plane x-ray profiles. (c) Azimuthal scan profiles obtained at $q(100)$ peak positions. The inset in (c) represents the correlation between the measured azimuthal angle and molecular orientation of RR P3HT. (d) Tapping mode AFM phase image of film 4.

rated through a shadow mask under a vacuum of 10^{-6} Torr. Testing procedures follow Ref. 4.

Figure 2 shows the effect of growth process on the absorbance for the thin films of RR-P3HT:[70]PCBM (1:1 wt %), before and after thermal annealing at 110°C . Although the fast-grown film is more homogeneous and smooth, the spectral features of P3HT subsides dramatically, which is attributed to [70]PCBM induced destruction of the RR-P3HT chain packing (ordering).¹⁶ On the other hand, the absorption spectra of solvent annealed film has very pronounced vibronic peaks, indicating strong interchain interaction among RR-P3HT chains as well as phase segregation. Thermal annealing significantly enhances the absorption of fast-grown film in long-wavelength region, although cannot match that in solvent annealed films. This indicates that the latter is more efficient in enhancing RR-P3HT chain packing and crystallinity. The IPCE data follow the same trend [Fig 2(b)]. The IPCE of the slow-growth device with thermal annealing is 54% and the plateau is also broader in the red region. This agrees with the improved light harvesting and carrier mobility of RR-P3HT.

The illuminated ($100\text{ mW}/\text{cm}^2$) J - V characteristics are shown in Fig. 3(a). The fast-growth film without thermal treatment shows $J_{\text{SC}}=3.21\text{ mA}/\text{cm}^2$, $V_{\text{OC}}=0.72\text{ V}$, and fill factor (FF)=29.81%, which translates to a PCE of 0.69%. Slow-growth process dramatic improvement is observed in both J_{SC} and FF, which results in PCE=3.32%. Both enhanced absorption and carrier mobility enhancement due to the significant polymer ordering in the film are believed to

responsible to the improvements. The best device performance is achieved with slow-growth film combined with thermal annealing at 110°C ($J_{\text{SC}}=8.59\text{ mA}/\text{cm}^2$, $V_{\text{OC}}=0.62\text{ V}$, FF=71.3%, and PCE=3.80%). Thermally annealed fast-grown device gets major boost in J_{SC} (from 3.21 to 7.56 mA/cm^2) and FF (from 29.8% to 53.3%). As expected, the resulting 2.49% PCE is significantly lower than that of purely solvent annealed device.

The V_{OC} , J_{SC} , and FF of the device with both solvent and thermal annealing as function of the incident light intensity (P_{in}) are plotted in Fig. 3(b). FF reaches a maximum of 73.1% at an incident intensity of $20\text{ mW}/\text{cm}^2$. The FF remains at 71.3% at AM1.5G $100\text{ mW}/\text{cm}^2$ illumination. These values are among the highest reported numbers in organic solar cells so far. The improved FF provides hint of improved transport property, which could be attributed to the fact that the charge became easier to extract due to the lower thickness of active layer. Table I summarizes the solar cell

TABLE I. Summary of device performance of ITO/PEDOT:PSS/P3HT:PCBM (film thickness $\sim 50\text{ nm}$)/Ca/Al polymer solar cells with different fabrication conditions.

Condition	V_{oc} (V)	J_{sc} (mA/cm^2)	FF (%)	R_s (Ω/cm^2)	PCE (%)
Fast grown, annealed	0.62	7.56	53.27	5.1	2.49
Fast grown, as-cast	0.72	3.21	29.81	10.1	0.69
Slow grown, annealed	0.62	8.59	71.3	1.7	3.80
Slow grown, as-cast	0.61	8.33	66.39	4.4	3.32

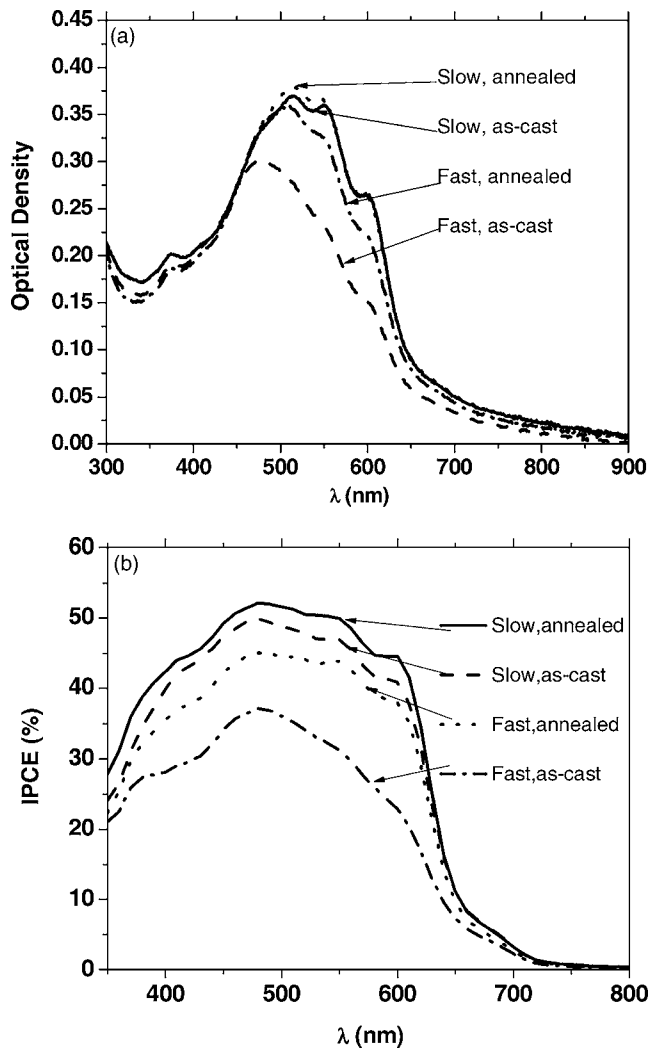


FIG. 2. (a) UV-visible spectra for films of P3HT:[70]PCBM (in 1:1 wt/wt ratio), for both slow grown and fast grown films from TCB as solvent, before (dash line) and after (solid line) annealing. The films were spun cast at 3000 rpm for 50 s (film thickness ~ 50 nm) and the annealing was done at 110 °C for 15 min inside glove box. (b) IPCE of RR-P3HT:[70]PCBM solar cells with two types of active layers: fast grown and slow grown.

performance of all devices discussed in this study.

The performance improvement through solvent annealing can be explained by film morphology. While RR-P3HT self-organizes to fibrillar highly crystalline domains with periodicity close to twice the exciton diffusion length,¹⁶ it leads to a percolated network of both crystalline RR-P3HT and [70]PCBM, i.e., a quasi-optimized morphology for polymer solar cell. PCBM is believed to play a minor role because solvent annealing is at room temperature.

In conclusion, we have shown that film processing conditions can control polymer nanoscale crystallinity and phase separation using GIXRD and AFM techniques. Efficient polymer solar cells (3.80% under AM 1.5G illumination at 100 mW/cm²) based on thin P3HT:[70]PCBM heterojunction with low optical density have been demonstrated. The enhancement of the J_{SC} and FF of the solvent annealed device with thermal treatment compared to the fast-growth device without thermal treatment is attributed to increase absorption in red region and improved charge transfer properties.

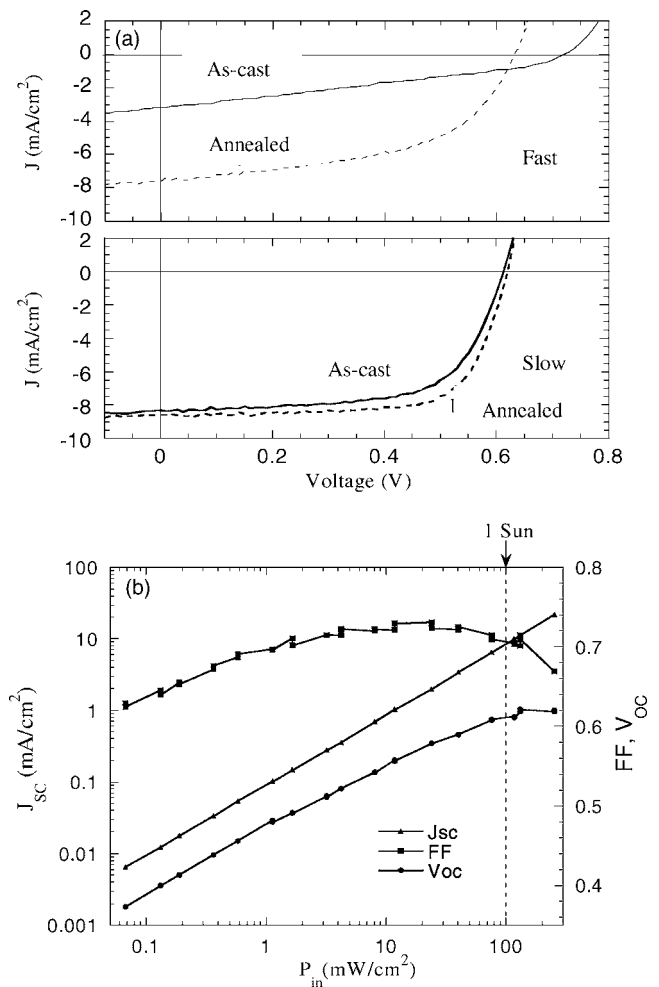


FIG. 3. (a) J - V characteristics of the fast (top) and slow (bottom) grown devices under AM 1.5G illumination with intensity of 100 mW/cm². (b) J_{SC} (triangles), FF (squares), and V_{OC} (circles), as a function of the incident power density (P_{in}) for the slow grown devices with annealing at 110 °C.

- ¹G. Yu, J. Gao, J. C. Hummelen, F. Wudl, and A. J. Heeger, *Science* **270**, 1789 (1995).
- ²Y. Kim, S. A. Choulis, J. Nelson, D. D. C. Bradley, S. Cook, and J. R. Durrant, *Appl. Phys. Lett.* **86**, 063502 (2005).
- ³F. Padinger, R. S. Rittberger, and N. S. Sariciftci, *Adv. Funct. Mater.* **13**, 85 (2003).
- ⁴G. Li, V. Shrotriya, J. Huang, Y. Yao, T. Moriarty, K. Emery, and Y. Yang, *Nat. Mater.* **4**, 864 (2005).
- ⁵G. Li, V. Shrotriya, Y. Yao, and Y. Yang, *J. Appl. Phys.* **98**, 043704 (2005).
- ⁶W. Ma, C. Yang, X. Gong, K. Lee, and A. J. Heeger, *Adv. Funct. Mater.* **15**, 1617 (2005).
- ⁷M. Nakazono, T. Kawai, and K. Yoshino, *Chem. Mater.* **6**, 864 (1994).
- ⁸P. Schilinsky, C. Waldauf, and C. J. Brabec, *Appl. Phys. Lett.* **81**, 3885 (2002).
- ⁹G. Li, C.-W. Chu, V. Shrotriya, J. Huang, and Y. Yang, *Appl. Phys. Lett.* **88**, 253503 (2006).
- ¹⁰J. Huang, G. Li, and Y. Yang, *Appl. Phys. Lett.* **87**, 112105 (2005).
- ¹¹V. Shrotriya, E. H. Wu, G. Li, Y. Yao, and Y. Yang, *Appl. Phys. Lett.* **88**, 064104 (2006).
- ¹²K. Kawano, N. Ito, T. Nishimori, and J. Sakai, *Appl. Phys. Lett.* **88**, 073514 (2006).
- ¹³M. M. Wienk, J. M. Kroon, W. J. H. Verhees, J. Knol, J. C. Hummelen, P. A. van Hal, and R. A. J. Janssen, *Angew. Chem.* **42**, 3371 (2003).
- ¹⁴H. Yang, S. W. LeFevre, C. Y. Ryu, and Z. Bao, *Appl. Phys. Lett.* **90**, 172116 (2007).
- ¹⁵V. Shrotriya, G. Li, Y. Yao, Y. Yang, T. Moriarty, and K. Emery, *Adv. Funct. Mater.* **16**, 2016 (2006).
- ¹⁶G. Li, Y. Yao, H. Yang, V. Shrotriya, G. Yang, and Y. Yang, *Adv. Funct. Mater.* **17**, 1636 (2007).