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## The 2009 edition of the GEISA spectroscopic database

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### 1. Introduction<sup>1</sup>

Spectroscopic remote sensing is an indispensable tool of modern meteorology. It is used to investigate climate change and provide an improved understanding of the different phenomena driving an atmospheric system in order to predict its past and future evolution. In particular, spectrally highly resolved radiances measured by powerful observational techniques such as ground-, aircraft-, balloon-, or satellite-based sensors enable global monitoring of atmospheres, provide a wealth of information about its actual state. The corresponding sensors have been improved significantly in recent years. Currently, there are many satellitebased instruments recording high quality spectra in order to understand the atmospheric state in great detail. Planetary examples include the recent Mars Express (http://www. esa.int/SPECIALS/Mars Express/index.html), Venus Express (http://www.esa.int/esaMI/Venus\_Express/) and Cassini-Huygens missions (http://www.esa.int/SPECIALS/Cassini-Huygens/index.html), studying the terrestrial planets and Jupiter, Saturn and Titan, respectively. Numerous spacebased missions continually provide a very large number of spectral observations which produce new revelations in planetology.

For remote sensing of astronomical objects, an essential prerequisite is high accuracy forward radiative transfer modeling. This in turn requires extensive knowledge of both the fundamental spectroscopic parameters of atmospheric constituents and the equations governing the propagation of

#### ABSTRACT

The updated 2009 edition of the spectroscopic database GEISA (Gestion et Etude des Informations Spectroscopiques Atmosphériques; Management and Study of Atmospheric Spectroscopic Information) is described in this paper. GEISA is a computer-accessible system comprising three independent sub-databases devoted, respectively, to: line parameters, infrared and ultraviolet/visible absorption cross-sections, microphysical and optical properties of atmospheric aerosols. In this edition, 50 molecules are involved in the line parameters sub-database, including 111 isotopologues, for a total of 3,807,997 entries, in the spectral range from  $10^{-6}$  to 35,877.031 cm<sup>-1</sup>.

The successful performances of the new generation of hyperspectral sounders depend ultimately on the accuracy to which the spectroscopic parameters of the optically active atmospheric gases are known, since they constitute an essential input to the forward radiative transfer models that are used to interpret their observations. Currently, GEISA is involved in activities related to the assessment of the capabilities of IASI (Infrared Atmospheric Sounding Interferometer; http://smsc.cnes.fr/IASI/index.htm) on board the METOP European satellite through the GEISA/IASI database derived from GEISA. Since the Metop-A (http://www.eumetsat.int) launch (19 October 2006), GEISA is the reference spectroscopic database for the validation of the level-1 IASI data. Also, GEISA is involved in planetary research, i.e., modeling of Titan's atmosphere, in the comparison with observations performed by Voyager, or by ground-based telescopes, and by the instruments on board the Cassini–Huggens mission.

GEISA, continuously developed and maintained at LMD (Laboratoire de Météorologie Dynamique, France) since 1976, is implemented on the IPSL/CNRS (France) "Ether" Products and Services Centre WEB site (http://ether.ipsl.jussieu.fr), where all archived spectroscopic data can be handled through general and user friendly associated management software facilities. More than 350 researchers are registered for on line use of GEISA. © 2011 Elsevier Ltd. All rights reserved.

> radiation through the atmosphere. Numerous physical phenomenon that influence the radiative transfer of a planet can be discerned and often measured from the variation of specific spectral features. As a consequence, spectroscopy is at the root of modern planetology, enabling us to determine the physical properties of planets remotely. Generally, forward models used in such studies are generated from line-by-line codes. Their accuracy is affected in many ways, and uncertainty in the spectroscopic information is one of the greatest impacts.

> During second half of the 20th century, the synergy between the simultaneous development of new technologies (high speed processing with computers, high-resolution laboratory facilities, quantum-mechanical treatment in theoretical spectroscopy, etc.), provided the means to interpret a multitude of long-path atmospheric transmissions by performing radiance calculations for numerous scenarios. As a result, the first standardized spectroscopic database, the so-called "AFGL tape", oriented towards the Earth's atmosphere, was initiated in 1973, at Air Force Geophysics Laboratory USA, by McClatchey et al. [1] and Garing and McClatchey [2]. This early database was limited to the strongest infrared absorbers ( $H_2O$ ,  $CO_2$ ,  $O_3$ ,  $N_2O$ , CO,  $CH_4$ , and  $O_2$ ) in the terrestrial atmosphere. It contained approximately 100,000 transitions.

> In 1976, the ARA group at LMD (http://ara.abct.lmd. polytechnique.fr) initiated a similar effort with the development of GEISA [3–8]. The initial emphasis of GEISA and HITRAN varied somewhat because HITRAN was focused on the terrestrial atmosphere while GEISA was oriented towards planetary atmospheres (in particular to support the Voyager mission to the giant planets). The GEISA archive included the

<sup>&</sup>lt;sup>1</sup> Acronyms used in the text are documented in Appendix A.

same seven atmospheric absorbers as in HITRAN, with a dozen additional species such as: NH<sub>3</sub>, PH<sub>3</sub>, C<sub>2</sub>H<sub>4</sub>, GeH<sub>4</sub>, C<sub>3</sub>H<sub>8</sub>, C<sub>2</sub>H<sub>2</sub>, HC<sub>3</sub>N, HCOOH, C<sub>3</sub>H<sub>4</sub>, as well as molecules such as: NO, SO<sub>2</sub>, NO<sub>2</sub>. There are other notable differences:

- a specific major initial task of GEISA has been to develop software so that users of the database could easily perform various kinds of extractions for their own applications in atmospheric physics and molecular spectroscopy;
- since the very first edition of GEISA, any isotopologue of a species having symmetry properties different from that of the main isotopologue (e.g., CH<sub>3</sub>D and CH<sub>4</sub>, C<sub>2</sub>HD and C<sub>2</sub>H<sub>2</sub>) was entered as an independent molecular species; its line intensities were given for a 100% sample rather than scaling by standard isotopic abundances (as for <sup>12</sup>CH<sub>4</sub> and <sup>13</sup>CH<sub>4</sub>).

The ARA group has continued to develop and maintain GEISA for over three decades, responding to incorporate new species and improve the completeness and accuracies of the spectroscopic parameters. Since quality of its reference information strongly impacts applications of planetary radiative transfer, there is an acute and constant demand for validated, operational and interactive public spectroscopic databases that are comprehensive and trustworthy. In its present structure, GEISA is a computer accessible database system, which, as described previously [5-9], delivers the necessary data to interpret the terrestrial and planetary atmospheric observations. GEISA comprises three independent sub-databases devoted, respectively, to (a) line parameters, (b) infrared and ultraviolet absorption crosssections, and (c) microphysical and optical properties of atmospheric aerosols. It is used on-line by more than 300 laboratories for studies in atmospheric physics, astronomy and astrophysics, and planetology.

The role of molecular spectroscopy in modern atmospheric research has entered a new phase with the advent of highly sophisticated spectroscopic instruments and computers. The launch of high spectral resolution vertical infrared sounders like AIRS (http://www-airs.jpl.nasa.gov/) on board EOS (http://eospso.gsfc.nasa.gov/)-Aqua (http:// aqua.nasa.gov/) since May 2002, or IASI (http://smsc.cnes. fr/IASI/index.htm) on board the European polar satellite Metop-A (http://www.eumetsat.int/Home/Main/Satellites/ Metop/index.htm?l=en; http://www.esa.int/export/esaLP/ LPMetop.html) since October 2006, have opened promising perspectives for remote sensing applications as the improvement of temperature and water vapor profile retrieval, cloud and surface characteristics retrieval, or retrievals of greenhouse gases (CO2 and CH4 for example) and of various chemical species. The January 2009 launch of the GOSAT satellite (http://www.gosat.nies.go.jp/index\_e. html) is another noteworthy event. The main aim of this mission is to measure the column amounts and profiles of the concentration of CO<sub>2</sub> and CH<sub>4</sub> over the globe.

Since the launch of Metop-A, GEISA has been declared as the reference basis by the international working group (ISSWG) in charge of the IASI hyperspectral sounder, through the GEISA/IASI database [10] which was derived from GEISA, as a sub-set for selected molecules, within the 599–3001 cm<sup>-1</sup> spectral range. GEISA/IASI is currently and routinely used for the validation of the level-1 IASI data, using the 4A radiative transfer model [11,12]; 4A/LMD; 4A/OP co-developed by LMD and Noveltis, http://www.noveltis.fr/, with the support of CNES).

The contents of each of the three sections of GEISA in its 2009 edition (hereafter GEISA-09) will be described in this paper. Recommendations on the quality of spectroscopic line parameters required (from the conclusions of experts involved in atmospheric and planetary science) will also be summarized.

GEISA is freely accessible from Ether, the CNRS/CNES/ IPSL Products and Services Center, website (http://ether. ipsl.jussieu.fr/).

It should be noted that other well known spectroscopic data compilations are available including:

- HITRAN (former "AFGL tape") for atmospheric and planetary remote sensing (see Ref. [13] for 2004 and 2008 Editions);
- MIPAS [14] specifically tied to satellite experiments in the Earth's atmosphere;
- BEAMCAT, for millimeter and submillimeter wave propagation in the Earth's atmosphere [15];
- the JPL Catalog of microwave to sub-millimeter transitions [16] which contains, for the most part, rotational transitions of a few hundred molecules which can or may be observed in the atmospheres of Earth or other planets to molecules occurring in the Inter StellarMedium (ISM) or in CircumStellar Envelopes (CSE) of late type stars. A small, but probably increasing number of entries contain infrared transitions;
- the CDMS Catalog [17] which also contains mostly rotational transitions of molecules important for the ISM or CDEs. Naturally, some of the molecules are also of relevance for Earth's atmosphere or that of other planets. Furthermore, a number of entries deal with infrared transitions of such molecules. Selected examples are low-lying vibrational modes of C<sub>3</sub> and C<sub>3</sub>O<sub>2</sub> or selected bands of CH<sup>+</sup>, C<sub>2</sub>H, or CH<sub>3</sub>CCH.

#### 2. Line parameters GEISA-09 sub-database description

#### 2.1. General overview

In the significant 2009 update described below, the GEISA-09 sub-database of line parameters archives, at the reference temperature of 296 K, the spectral properties of 50 molecular species (111 isotopologues) corresponding to a total of 3,807,997 entries in the spectral range from  $10^{-6}$  to 35,877.031 cm<sup>-1</sup> ( $10^{10}$  to 0.28 µm). This represents an increase of 8 molecular species, 14 isotopologues and 2,139,626 entries since the GEISA-03 [8,9] edition. This 28% increase in entries is mainly due to: extension of spectral ranges (i.e., CO<sub>2</sub>, N<sub>2</sub>O, etc.), addition of new vibrational bands and isotopologues (i.e., SO<sub>2</sub>, etc.), more sophisticated theoretical and/or experimental determination of the spectroscopic parameters (i.e., CO<sub>2</sub>, HNO<sub>3</sub>, H<sub>2</sub>CO, C<sub>2</sub>H<sub>2</sub>, HCN, C<sub>4</sub>H<sub>2</sub>, SF<sub>6</sub>, etc.), and new archived molecular species (see Table 1 for details). The newly archived molecular species

#### Table 1

Contents of the GEISA-09 sub-database on line parameters. Details per molecule of the 2009 evolution of GEISA content since its 2003 edition. Reference temperature is 296 K.

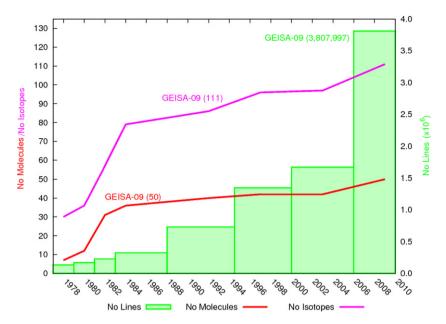
Mol.	ID	GEISA-03				GEISA-09				Refs.
		Spectral range (cm <sup>-1</sup> )	# lines	Intensit	y exponent	Spectral range (cm <sup>-1</sup> )	# lines	Intensit	y exponent	-
				Max.	Min.			Max.	Min.	
H <sub>2</sub> 0	1	0.007-22,656.465	58,726	-18	-31	0.007-25,232.004	67,789	-18	-33	[20-31]
$CO_2$	2	436.123-9648.007	76,826	-18	-41	5.891-12,784.053	413,619	-18	-42	[32–69]
03	3	0.026-4060.783	319,248	-19	-30	0.026-6395.379	389,378	-19	-30	[70–97]
$N_2O$	4	0.838-5131.249	26,681	-17	-27	0.838-7796.633	50,633	-17	-25	[98-102]
CO	5	3.414-8464.882	13,515	-18	-77	3.414-8464.882	13,515	-18	-77	No update
$CH_4$	6	0.010-9199.285	216,196	-18	-33	0.001-9199.284	240,991	-18	-38	[103–126]
02	7	$10^{-6}$ -15,927.806	6290	-23	-50	10 <sup>-6</sup> -15,927.230	6428	-23	-50	[127–137]
NO	8	$3 \times 10^{-5}$ -9273.214	99,123	-19	-84	10 <sup>-6</sup> 9273.214	105,079	-19	-94	[133–139]
SO <sub>2</sub>	9	0.017-4092.948	38,853	-19	-27	0.017-4092.948	68,728	-19	-28	[140–166]
$NO_2$	10	0.498-3074.366	104,224	-18	-27	0.498-3074.152	104,223	-18	-27	[167–168]
$NH_3$	11	0.058-5294.502	29,082	-18	-38	0.058-5294.501	29,082	-18	- 38	[169–170]
$PH_3$	12	17.805-2478.765	11,740	-18	-27	17.805-3601.652	20,421	-18	-27	[171–178]
$HNO_3$	13	0.035-1769.982	171,504	-19	-26	0.012-1769.982	669,988	-19	-27	[179–195]
OH	14	0.005-35,877.030	42,866	-16	-84	0.005-35,877.031	42,866	-16	-84	No update
HF	15	41.110-11,535.570	107	-16	-25	41.111-11,535.570	107	-16	-25	No update
HCl	16	20.240-13,457.841	533	-18	-25	20.240-13,457.841	533	-18	-25	No update
HBR	17	16.231-9758.565	1294	-18	-32	16.231-9758.564	1294	-18	-32	No update
HI	18	12.509-8487.305	806	-19	-29	12.509-8487.305	806	-19	-29	No update
CLO	19	0.015-1207.639	7230	-20	-29	0.015-1207.639	7230	-20	-29	No update
OCS	20	0.381-4118.004	24,922	-17	-27	0.381-4199.671	33,809	-17	-27	[196–210]
H <sub>2</sub> CO	21	$3 \times 10^{-6}$ –2998.527	2701	-19	-37	$3 \times 10^{-6}$ -3099.958	37,050	-19	-37	[211-216]
$C_2H_6$	22	725.603-2977.926	14,981	-20	-27	706.601-3000.486	27,644	-20	-29	[217–236]
CH <sub>3</sub> D	23	7.760-3306.810	35,518	-22	-29	7.7602-6510.326	49,237	-22	-29	[237–245]
$C_2H_2$	24	604.774-4225.435	3115	-17	-26	604.774-9889.038	11,340	-17	-27	[246-255]
$C_2H_4$	25	701.203-3242.172	12,978	-19	-25	701.203-3242.172	18,378	-19	-36	[256–260]
GEH <sub>4</sub>	26	1937.37-2224.570	824	-18	-21	1937.371-224.570	824	-18	-21	No update
HCN	27	2.870-18,407.973	2550	-18	-27	0.006-17,581.010	82,042	-18	-33	[261-292]
$C_3H_8$	28	700.015-799.930	8983	-21	-23	700.015-799.930	8983	-21	-23	[293-298]
$C_2N_2$	29	203.955-2181.690	2577	-19	-23	203.955-2181.690	2577	-19	-23	[299-301]
$C_4H_2$	30	190.588-654.425	1405	-19	-23	191.635-730.235	119,480	-18	-23	[302-315]
HC₃N	31	474.293-690.860	2027	-19	-23	463.604-759.989	179,347	-19	-23	[316-319]
HOCI	32	0.0236-3799.682	17,862	-19	-27	0.0236-3799.682	17,862	-19	-27	No update
$N_2$	33	1992.63-2625.497	120	-27	-33	1992.63-2625.497	120	-27	-33	[320-322]
CH₃Cl	34	674.143-3172.927	18,344	-19	-31	674.143-3172.927	18,344	-19	-31	[323-324]
$H_2O_2$	35	0.043-1499.486	100,781	-19	-28	0.043-1730.371	126,983	-19	-28	[325-327]
$H_2S$	36	2.985-4256.547	20,788	-18	-25	2.985-4256.547	20,788	-18	-25	[328-331]
HCOOH	37	1060.96-1161.251	3388	-19	-21	10.018-1889.334	62,684	-19	-25	[332-342]
COF <sub>2</sub>	38	725.005-2001.348	83,750	-19	-23	725.005-2001.348	83,750	-19	-23	No update
SF <sub>6</sub>	39	940.424-952.238	11,520	-19	-21	588.488-975.787	92,398	-19	-23	[343-348]
$C_3H_4$	40	290.274-359.995	3390	-20	-22	288.913-673.479	19,001	-19	-23	[349-357]
HO <sub>2</sub>	41	0.173-3675.818	38,804	-19	-25	0.173-3675.819	38,804	-19	-25	No update
ClONO <sub>2</sub>	42	763.641-792.488	32,199	-21	-24	0.636-797.741	356,899	-21	-27	[358-362]
CH₃BR		-				794.403-1705.612	36,911	-20	-26	[363-377]
CH₃OH	44	-				0.019-1407.206	19,897	-19	-34	[378-389]
NO <sup>+</sup>		-				1634.83-2530.462	1206	-18	-80	[390-391]
HNC		-				0.217-4814.904	5619	-17	-24	[392-401]
C <sub>6</sub> H <sub>6</sub>	47	-				642.427-705.262	9797	-20	-23	[402-405]
$C_2HD$		-				416.785-3421.864	15,512	-22	-28	[406-409]
CF <sub>4</sub>	49	-				594.581-1312.647	60,033	-19	-23	[410-420]
CH <sub>3</sub> CN		-				890.052-1650.000	171,172	-19	-37	[428-432]
			Total # lines:				Total # lines:			(
			1,668,371				3,807,997			

Note: "No update" in the Refs. column indicates that the contents in GEISA-09 and GEISA-03 are identical.

are: CH<sub>3</sub>Br, CH<sub>3</sub>OH, NO<sup>+</sup>, HNC, C<sub>6</sub>H<sub>6</sub>, C<sub>2</sub>HD, CF<sub>4</sub>, and CH<sub>3</sub>CN. As stated above, the molecules included in GEISA-09 (and since the database creation), are constituents, not only of the atmospheres of Earth (major permanent and trace molecules), but also of other planets (such as: C<sub>2</sub>H<sub>4</sub>, GeH<sub>4</sub>, C<sub>3</sub>H<sub>8</sub>, C<sub>2</sub>N<sub>2</sub>, C<sub>4</sub>H<sub>2</sub>, HC<sub>3</sub>N, H<sub>2</sub>S, HCOOH and C<sub>3</sub>H<sub>4</sub>, mainly for

giant planets). The evolution of the GEISA line parameters sub-database, since 1975, is presented in Fig. 1.

The parameters of each spectral line or molecular vibrational-rotational transition are stored in the new "standard format" for GEISA and GEISA/IASI as described in Ref. [10], with some newly introduced technical



**Fig. 1.** Evolution of the GEISA line parameters sub-database since 1975. The year identifications are on the X-axis. The evolution of the number of lines is displayed as histograms (green color) with corresponding scale given on the right Y-axis. The total number of molecules and isotopes, included in each atlas, are color coded, as red and purple curves, respectively, with corresponding scale on the left Y-axis.

modifications, i.e., extended format field for quantum identifications for the lower and upper states of the transition, for self-broadened half-width and air-induced pressure shift of the line transition, corresponding, respectively, to fields of format symbols:  $E_i$  (i=1-4), M and N; former fields identified as P (accuracy indices for wavenumber, intensity and half-width) and Q (indices for lookup of references for wavenumber, intensity and half-width) have been removed. See Appendix B for details.

Evolution of the contents of GEISA-09 since the 2003 edition of GEISA (hereafter GEISA-03) is summarized in Table 1. Individual GEISA-09 molecule names and their corresponding identification codes (ID codes defined for the GEISA management software) are in the first two columns of the Table. The following columns give successively for GEISA-03 and for GEISA-09: the spectral range  $(cm^{-1})$ , the number of entries, the exponents of the maximum and minimum intensity values (expressed in  $\text{cm}^{-1}/(\text{molecule cm}^{-2})$  at 296 K), for each molecule, and finally the 2009 update references. The spectroscopic line parameters of 31 of the 42 molecules included in GEISA-03 have been updated. The parameters of 11 molecules, i.e., CO, OH, HF, HCl, HBr, HI, ClO, GeH<sub>4</sub>, HOCl, COF<sub>2</sub>, and HO<sub>2</sub>, are kept unchanged as in GEISA-03. The details of the GEISA-09 sub-database on line parameters are given in Table 2. The items listed in columns 3–6 for each molecular species, given in column 1, are: the number of lines, the intensity average in cm molecule<sup>-1</sup> (different expression for  $cm^{-1}/(molecule cm^{-2}))$ , the average half-width at halfmaximum (HWHM in  $cm^{-1} atm^{-1}$ ), the present isotopologue identification codes (see Table 2 of Ref. [7] for isotopic species code identifications and complementary information, in Appendix C, for new isotopic and molecular species in GEISA-09); for each isotopic species listed in column 6, are given in columns 7–11: the number of lines with associated minimum and maximum wavenumbers  $(cm^{-1})$  and intensities (in cm molecule<sup>-1</sup>).

Table 3 summarizes the differences between the GEISA-09 and the HITRAN 2008 (hereafter HITRAN-08) [13] databases in terms of the number of lines, bands, and isotopologues. An example of quantitative comparison between H<sub>2</sub>O intensity values in GEISA-09 and HITRAN-08 is given in Fig. 2. In the spectral range  $1400-2100 \text{ cm}^{-1}$ , 5626 transitions with common quantum identification in both databases and with intensity values larger than  $10^{-23}$  cm<sup>-1</sup>/(molecule cm<sup>-2</sup>), are involved in this comparison. One can notice that 8% of the strong lines (intensities greater or equal  $10^{-20}$  cm<sup>-1</sup>/(molecule cm<sup>-2</sup>)) exhibit differences greater that 5%. Evaluations of impact on atmospheric radiative transfer modeling, using HITRAN or GEISA, are presented, for instance, in Jacquinet-Husson et al. [8], Matricardi [18] Newman [19]. It may be noted that the previous updates of HITRAN and GEISA databases have been finalized at nearly the same time, and include very similar data sources for many molecules. Because of its origin, certain molecules, mainly related with planetary atmospheres (especially those of the giant planets) are specific to GEISA, such as: GeH<sub>4</sub>, C<sub>3</sub>H<sub>8</sub>, C<sub>2</sub>N<sub>2</sub>, C<sub>4</sub>H<sub>2</sub>, C<sub>3</sub>H<sub>4</sub>, HC<sub>3</sub>N, HNC, C<sub>6</sub>H<sub>6</sub>, and C<sub>2</sub>HD. On the other hand, species HOBr and O are HITRAN specific, and in HITRAN CH<sub>3</sub>D and C<sub>2</sub>HD are considered isotopologues of methane and acetylene, but they are independent molecules in GEISA (see Sections 1 and 2). In Table 3, molecular species formulae are listed in column 1 and their identification codes for database managements in column 2. For each molecular species and for each data base, the number of bands, isotopologues and lines, are given in columns 3, 4 and 5, respectively. The related minimum and maximum of the spectral range

## Table 2

The GEISA-09 sub-database on line parameters. Spectral and intensity ranges per molecule and isotopologue.

Mol.	ID	# lines	Intensity average (cm molecule <sup>-1</sup> )	HWHM average (cm <sup>-1</sup> atm <sup>-1</sup> )	Isot. ID	# lines	Minimum wavenumber (cm <sup>-1</sup> )	Maximum wavenumber (cm <sup>-1</sup> )	Minimum intensity (cm molecule <sup>-1</sup> )	Maximum intensity (cm molecule <sup>-1</sup> )
H <sub>2</sub> O	1	67,789	$1.088 \times 10^{-21}$	0.0699	161 81 171 162 182 172	41,147 8360 5468 11,980 659 175	0.401 6.785 6.471 0.007 1173.772 1234.235	25,224,909 14,362,151 13,909,783 13,900,444 3824,717 1598,7655	$\begin{array}{c} 9.400\times10^{-33}\\ 2.005\times10^{-28}\\ 2.671\times10^{-31}\\ 1.240\times10^{-32}\\ 2.033\times10^{-27}\\ 2.033\times10^{-27}\\ \end{array}$	$\begin{array}{c} 2.654 \times 10^{-18} \\ 5.390 \times 10^{-21} \\ 9.830 \times 10^{-22} \\ 2.700 \times 10^{-22} \\ 5.083 \times 10^{-26} \\ 9.319 \times 10^{-27} \end{array}$
CO <sub>2</sub>	2	413,619	$2.724 \times 10^{-22}$	0.0704	626 636 628 627 638 637 828 728 838	165,181 66,657 110,136 19,064 39,007 2741 10,045 493 295	345.936 433.190 5.891 10.600 449.686 580.856 484.297 626.438 2115.684	12,784.052 12,462.048 11,422.648 8270.099 6744.160 6768.643 8162.742 5031.885 2276.481	$\begin{array}{c} 1.000 \times 10^{-30} \\ 1.009 \times 10^{-28} \\ 4.870 \times 10^{-42} \end{array}$	$\begin{array}{c} 3.520\times 10^{-18}\\ 3.740\times 10^{-20}\\ 6.870\times 10^{-21}\\ 1.260\times 10^{-21}\\ 7.810\times 10^{-23}\\ 1.400\times 10^{-23}\\ 1.330\times 10^{-23}\\ 2.500\times 10^{-24}\\ 3.289\times 10^{-24} \end{array}$
03	3	38,9378	$5.247 \times 10^{-23}$	0.0698	666 668 686 667 676	233,132 44,302 24,886 58,171 28,887	0.026 0.921 1.177 0.289 0.213	6395.379 2767.874 2739.289 820.380 822.795	$\begin{array}{c} 3.904 \times 10^{-29} \\ 4.692 \times 10^{-28} \\ 9.970 \times 10^{-29} \\ 5.135 \times 10^{-31} \\ 1.433 \times 10^{-31} \end{array}$	$\begin{array}{l} 4.060\times10^{-20}\\ 7.760\times10^{-23}\\ 7.560\times10^{-23}\\ 5.356\times10^{-25}\\ 5.827\times10^{-25}\end{array}$
N <sub>2</sub> O	4	50,633	$1.254 \times 10^{-21}$	0.0750	446 456 546 448 447 458 548 556	34,468 4466 4841 4412 1778 105 108 455	0.838 5.028 4.8580 541.342 549.367 2121.770 2144.997 1226.536	7796.633 5088.906 4992.236 4672.579 4429.961 2203.983 2226.290 3415.768	$\begin{array}{c} 1.016 \times 10^{-29} \\ 5.220 \times 10^{-26} \\ 4.720 \times 10^{-26} \\ 1.614 \times 10^{-25} \\ 1.614 \times 10^{-25} \\ 1.673 \times 10^{-25} \\ 1.675 \times 10^{-25} \\ 1.675 \times 10^{-25} \\ 1.642 \times 10^{-25} \end{array}$	$\begin{array}{c} 1.003\times 10^{-18}\\ 3.423\times 10^{-21}\\ 3.513\times 10^{-21}\\ 1.930\times 10^{-21}\\ 4.017\times 10^{-22}\\ 6.637\times 10^{-24}\\ 7.631\times 10^{-24}\\ 1.210\times 10^{-23}\\ \end{array}$
CO	5	13,515	$7.543 \times 10^{-22}$	0.0467	26 36 27 28 37 38	5908 4768 748 770 580 741	3.53010 3.414 3.714 3.629 1807.871 3.462	8464.882 8180.219 6338.061 6266.577 6196.551 6123.294	$\begin{array}{c} 7.880 \times 10^{-78} \\ 3.610 \times 10^{-73} \\ 8.190 \times 10^{-40} \\ 7.610 \times 10^{-39} \\ 1.030 \times 10^{-36} \\ 2.580 \times 10^{-40} \end{array}$	$\begin{array}{l} 4.460\times10^{-19}\\ 4.690\times10^{-21}\\ 1.600\times10^{-22}\\ 8.320\times10^{-22}\\ 1.680\times10^{-24}\\ 8.700\times10^{-24} \end{array}$
CH4	6	240,991	$8.225 \times 10^{-23}$	0.0521	211 311	212,115 28,876	0.010 0.032	9155.326 6069.084	$\begin{array}{c} 1.117 \times 10^{-39} \\ 4.936 \times 10^{-34} \end{array}$	$\begin{array}{c} 2.099 \times 10^{-19} \\ 2.317 \times 10^{-21} \end{array}$
02	7	6428	$3.885  imes 10^{-26}$	0.0430	66 67 68	1431 4326 671	0.000 0.000 1.572	15,927.230 14,536.515 15,851.213	$\begin{array}{l} 9.808 \times 10^{-51} \\ 8.513 \times 10^{-51} \\ 1.186 \times 10^{-35} \end{array}$	$\begin{array}{l} 8.762 \times 10^{-24} \\ 3.439 \times 10^{-27} \\ 1.727 \times 10^{-26} \end{array}$
NO	8	105,079	$4.625 \times 10^{-23}$	0.0477	46 48 56	100,902 679 699	0.000 1601.909 1609.585	9273.214 2038.846 2060.462	$\begin{array}{c} 1.451 \times 10^{-95} \\ 4.190 \times 10^{-28} \\ 4.430 \times 10^{-28} \end{array}$	$\begin{array}{c} 1.188 \times 10^{-20} \\ 1.390 \times 10^{-22} \\ 2.550 \times 10^{-22} \end{array}$
SO <sub>2</sub>	9	68,728	$5.850  imes 10^{-22}$	0.0090	626 646	57,963 10,765	0.017 1060.196	4092.948 2500.400	$\begin{array}{c} 1.020 \times 10^{-28} \\ 4.980 \times 10^{-24} \end{array}$	$\begin{array}{l} 4.851 \times 10^{-20} \\ 4.493 \times 10^{-23} \end{array}$
NO <sub>2</sub> NH <sub>3</sub>			$\begin{array}{l} 5.980 \times 10^{-22} \\ 1.639 \times 10^{-21} \end{array}$	0.0742 0.0827	646 411 511	104,223 27,992 1090	0.498 0.0582 0.375	3074.153 5293.578 5179.786	$\begin{array}{l} 4.240\times10^{-28}\\ 8.086\times10^{-39}\\ 5.460\times10^{-29}\end{array}$	$\begin{array}{l} 1.302\times10^{-19}\\ 4.585\times10^{-19}\\ 1.992\times10^{-21} \end{array}$
$PH_3$	12	20,421	$1.367\times10^{-21}$	0.0648	131	20,421	17.805	3601.652	$1.849\times10^{-28}$	$\textbf{2.520}\times 10^{-19}$
$HNO_3$	13	669,988	$1.768\times10^{-22}$	0.1048	146	669,988	0.012	1769.982	$3.590\times10^{-28}$	$3.130\times10^{-20}$
ОН	14	42,866	$2.806  imes 10^{-20}$	0.0440	61 62 81	42,711 90 65	0.005 0.010 0.053	35,877.030 1.824 6.325	$\begin{array}{l} 1.500 \times 10^{-85} \\ 2.090 \times 10^{-31} \\ 1.200 \times 10^{-30} \end{array}$	$\begin{array}{l} 6.450 \times 10^{-17} \\ 5.780 \times 10^{-29} \\ 1.200 \times 10^{-26} \end{array}$
HF	15	107	$\textbf{6.773}\times \textbf{10}^{-19}$	0.0407	19	107	41.111	11,535.570	$1.110 \times 10^{-26}$	$1.440\times10^{-17}$
HCl	16	533	$3.189\times10^{-20}$	0.0403	15 17	284 249	20.270 20.240	13,457.841 10,994.721	$\begin{array}{c} 1.090 \times 10^{-26} \\ 1.010 \times 10^{-26} \end{array}$	$\begin{array}{l} 5.030 \times 10^{-19} \\ 1.610 \times 10^{-19} \end{array}$
HBr	17	1294	$4.769 \times 10^{-21}$	0.0429	11 19	642 652	16.231 16.236	9757.189 9758.565	$\begin{array}{c} 1.528 \times 10^{-32} \\ 9.450 \times 10^{-33} \end{array}$	$\begin{array}{c} 1.178 \times 10^{-19} \\ 1.211 \times 10^{-19} \end{array}$
HI	18	806	$1.361\times10^{-21}$	0.0500	17	806	12.5094	8487.305	$1.644\times10^{-30}$	$3.423\times10^{-20}$
C10	19	7230	$1.605 \times 10^{-22}$	0.0873	56 76	3599 3631	0.028 0.015	1207.639 1199.840	$\begin{array}{c} 1.520 \times 10^{-29} \\ 5.090 \times 10^{-30} \end{array}$	$\begin{array}{l} 3.240 \times 10^{-21} \\ 1.030 \times 10^{-21} \end{array}$

Table 2 (continued)

Mol.	ID	# lines	Intensity average (cm molecule <sup>-1</sup> )	HWHM average (cm <sup>-1</sup> atm <sup>-1</sup> )	lsot. ID	# lines	Minimum wavenumber (cm <sup>-1</sup> )	Maximum wavenumber (cm <sup>-1</sup> )	Minimum intensity (cm molecule <sup>-1</sup> )	Maximum intensity (cm molecule <sup>-1</sup> )
OCS	20	33,809	$3.436 \times 10^{-21}$	0.0894	622 624 632 623 822 634	19,130 6665 3243 2788 1626 357	0.406 0.396 0.404 509.007 0.380 1972.188	4199.671 4165.233 4055.090 4163.069 4045.602 2032.039	$\begin{array}{c} 8.550 \times 10^{-20} \\ 6.400 \times 10^{-27} \\ 1.720 \times 10^{-27} \\ 4.678 \times 10^{-26} \\ 2.620 \times 10^{-28} \\ 1.010 \times 10^{-23} \end{array}$	$\begin{array}{c} 1.220\times10^{-18}\\ 4.720\times10^{-20}\\ 1.200\times10^{-20}\\ 8.430\times10^{-21}\\ 2.090\times10^{-21}\\ 5.240\times10^{-22} \end{array}$
H₂CO	21	37,050	$1.175 \times 10^{-21}$	0.1079	126 128 136	36,120 367 563	0.000 0.034 0.037	3099.941 47.486 72.744	$\begin{array}{c} 1.224\times 10^{-38} \\ 1.392\times 10^{-30} \\ 2.424\times 10^{-30} \end{array}$	$7.436 \times 10^{-20}$ $1.332 \times 10^{-22}$ $7.548 \times 10^{-22}$
C <sub>2</sub> H <sub>6</sub>	22	28,439	$1.600 \times 10^{-22}$	0.0670	226 236	22,402 6037	706.601 725.603	3000.486 918.717	$5.422 \times 10^{-29}$ $1.320 \times 10^{-28}$	$3.210 \times 10^{-20}$ $1.770 \times 10^{-23}$
CH₃D	23	49,237	$1.903\times10^{-25}$	0.0542	212 312	45,024 4213	7.760 959.394	6510.326 1694.123	$\begin{array}{c} 5.677 \times 10^{-30} \\ 2.768 \times 10^{-29} \end{array}$	$\begin{array}{c} 5.714 \times 10^{-23} \\ 1.398 \times 10^{-25} \end{array}$
$C_2H_2$	24	11,340	$3.877 \times 10^{-21}$	0.0720	221 231	11,055 285	604.774 613.536	9889.0377 6588.935	$\begin{array}{c} 4.425 \times 10^{-28} \\ 3.820 \times 10^{-26} \end{array}$	$\begin{array}{c} 1.187 \times 10^{-18} \\ 4.942 \times 10^{-19} \end{array}$
$C_2H_4$	25	18,378	$1.081 \times 10^{-21}$	0.0861	211 311	18,097 281	701.203 2947.832	3177.173 3180.238	$\begin{array}{c} 2.764 \times 10^{-37} \\ 5.061 \times 10^{-24} \end{array}$	$\begin{array}{c} 8.412 \times 10^{-20} \\ 1.618 \times 10^{-21} \end{array}$
GeH <sub>4</sub>	26	824	$\textbf{4.978}\times 10^{-20}$	0.1000	411	824	1937.371	2224.570	$1.960\times10^{-22}$	$3.680  imes 10^{-19}$
HCN	27	82,042	$4.201 \times 10^{-22}$	0.1002	124 125 134 224	79,957 791 791 503	0.006 2.870 2.880 2.415	17,581.009 3550.842 3532.252 2725.192	$\begin{array}{c} 8.057 \times 10^{-34} \\ 5.156 \times 10^{-32} \\ 1.431 \times 10^{-31} \\ 1.801 \times 10^{-30} \end{array}$	$\begin{array}{l} 7.010\times10^{-19}\\ 2.468\times10^{-21}\\ 3.785\times10^{-21}\\ 7.317\times10^{-23} \end{array}$
C₃H <sub>8</sub>	28	8983	$4.139\times10^{-23}$	0.0800	221	8983	700.015	799.930	$1.583\times10^{-24}$	$1.810\times10^{-22}$
$C_2N_2$	29	2577	$1.885  imes 10^{-21}$	0.1023	224	2577	203.955	2181.690	$3.130  imes 10^{-24}$	$1.200  imes 10^{-20}$
$C_4H_2$	30	119,480	$2.530  imes 10^{-22}$	0.0999	211	119,480	191.635	730.235	$3.024\times10^{-24}$	$1.435  imes 10^{-19}$
HC₃N	31	179,347	$6.982  imes 10^{-23}$	0.0998	124	179,347	463.604	759.989	$1.052 \times 10^{-24}$	$4.040\times10^{-20}$
HOCI	32	17,862	$1.867\times10^{-21}$	0.0689	165 167	9293 8569	0.023 0.349	3799.249 3799.682	$\begin{array}{c} 1.650 \times 10^{-27} \\ 7.220 \times 10^{-28} \end{array}$	$\begin{array}{l} 3.590 \times 10^{-20} \\ 1.140 \times 10^{-20} \end{array}$
N <sub>2</sub>	33	120	$5.605 \times 10^{-29}$	0.0343	44	120	1992.628	2625.497	$1.590\times10^{-34}$	$3.548\times10^{-28}$
CH₃Cl	34	18,344	$4.370\times10^{-22}$	0.0951	215 217	10,039 8305	679.050 674.143	3172.927 3161.830	$\begin{array}{l} 9.051 \times 10^{-32} \\ 4.192 \times 10^{-26} \end{array}$	$\begin{array}{c} 1.128 \times 10^{-20} \\ 3.542 \times 10^{-21} \end{array}$
$H_2O_2$	35	126,983	$4.622\times10^{-22}$	0.0999	166	126,983	0.043	1730.371	$5.064\times10^{-29}$	$5.582\times10^{-20}$
H <sub>2</sub> S	36	20,788	$2.992 \times 10^{-22}$	0.0740	121 131	12,330 3564	2.985 5.601	4256.547 4098.234	$1.450 \times 10^{-26}$ $2.020 \times 10^{-26}$ $2.020 = 10^{-26}$	$1.360 \times 10^{-19}$ $5.990 \times 10^{-21}$
	27	62,684	$1.231 \times 10^{-21}$	0.1010	141	4894	5.615 10.018	4171.176	$2.020 \times 10^{-26}$ $3.966 \times 10^{-26}$	$1.080 \times 10^{-21}$ $5.068 \times 10^{-20}$
COF <sub>2</sub>		83,750	$1.231 \times 10^{-21}$ $2.105 \times 10^{-21}$	0.0845	261 269	62,684 83,750	725.005	1889.334 2001.348	$3.968 \times 10^{-24}$	$3.940 \times 10^{-20}$
			$2.103 \times 10^{-22}$ $5.117 \times 10^{-22}$	0.0845	269 29	-	588.488	2001.348 975.787	$4.740 \times 10^{-24}$ $1.000 \times 10^{-24}$	$3.940 \times 10^{-20}$ $1.453 \times 10^{-20}$
SF <sub>6</sub>			$5.117 \times 10^{-22}$ $6.338 \times 10^{-22}$			92,398			$4.230 \times 10^{-24}$	
C <sub>3</sub> H <sub>4</sub>		19,001 38,804	$6.338 \times 10^{-22}$ $6.847 \times 10^{-22}$	( – ) 0.1070	341 166	19,001 38,804	288.912 0.173	636.482 3675.818	$4.230 \times 10^{-26}$ $1.000 \times 10^{-26}$	$\begin{array}{l} 1.550 \times 10^{-20} \\ 2.744 \times 10^{-20} \end{array}$
HO2 ClONO2			$7.958 \times 10^{-24}$	0.1404	564 764	206,861 150,038	0.636	797.741 790.805	$7.547 \times 10^{-28}$ $7.519 \times 10^{-28}$	$2.744 \times 10^{-22}$ $3.850 \times 10^{-22}$ $1.260 \times 10^{-22}$
CH₃br	43	36,911	$1.293\times10^{-22}$	0.0939	79 81	18,692 18,219	794.403 795.083	1705.612 1696.896	$9.970 \times 10^{-27}$ $1.000 \times 10^{-26}$	$2.580 \times 10^{-21}$ $2.530 \times 10^{-21}$
CH₃oh	44	19,897	$9.181  imes 10^{-22}$	0.1000	216	19,897	0.019	1407.205	$8.826 \times 10^{-35}$	$3.771  imes 10^{-20}$
NO+		1206	$2.168 \times 10^{-21}$	0.0600	46	1206	1634.831	2530.462	$6.121 \times 10^{-81}$	$1.186 \times 10^{-19}$
HNC		5619	$4.201  imes 10^{-22}$	0.1002	142	5619	0.217	4814.904	$1.001 \times 10^{-25}$	$1.164  imes 10^{-18}$
C <sub>6</sub> H <sub>6</sub>		9797	$8.394 \times 10^{-22}$	0.1014	266	9797	642.427	705.262	$4.070  imes 10^{-24}$	$9.490 \times 10^{-21}$
C <sub>2</sub> HD		15,512	$4.843 \times 10^{-25}$	0.0680	122	15,512	416.785	3385.564	$5.194 \times 10^{-29}$	$3.219 \times 10^{-23}$
CF4		60,033	$1.377 \times 10^{-21}$	0.5000	291	60,033	594.581	1312.646	$7.912 \times 10^{-24}$	$4.717 \times 10^{-20}$
CH <sub>3</sub> CN Total		17,172 3,807,99	$\textbf{2.688}\times10^{-22}$	0.0792	234	17,172	890.052	1650.000	$1.200 \times 10^{-38}$	$3.824 \times 10^{-20}$

Note: (-) Missing data.

 Table 3

 Summary of differences for molecular species cataloged in the line parameter portion of GEISA-09 (G) and HITRAN 2008 (H) [13].

Mol.	Mol	ID	# ba	nds	# is	ot	# lines		Spectral coverage (cm <sup>-1</sup> )			
	G	Н	G	Н	G	Н	G	Н	Minimum w	avenumber (cm <sup>-1</sup> )	Maximum way	venumber (cm <sup>-1</sup> )
									G	Н	G	Н
H <sub>2</sub> O	1	1	245	373	6	6	67,789	69,201	0.007	0.007	25,232.004	25,232.004
$CO_2$	2	2	3747	2832		9	413,619	314,919	5.891	0.736	12,784.052	12,784.052
O <sub>3</sub>	3	3	162	218	5	5	389,378	409,686	0.026	0.026	6395.379	5786.118
$N_2O$	4	4	369	351	8	5	50,633	47,843	0.838	0.838	7796.633	7796.633
CO	5	5	104	47	6	6	13,515	4477	3.414	3.462	8464.882	8464.881
CH <sub>4</sub>	6	6	138(	3) 138	2	2 <b>(</b> §)	240,991 <b>(</b> §)	240,854 <b>(</b> §)	0.001	0.001	9155.326	9155.326
02	7	7	19	19	3	3	6428	6428	0.000	0.000	15,927.230	15,927.230
NO	8	8	293	293	3	3	105,079	105,079	0.000	0.000	9273.214	9273.214
SO <sub>2</sub>	9	9	17	13	2	2	68,728	58,250	0.017	0.017	4092.948	4092.948
$NO_2$	10	10	11	11	1	1	104,223	104,223	0.498	0.498	3074.153	3074.153
$NH_3$	11	11	78	78	2	2	29,082	29,084	0.058	0.058	5293.578	5293.578
PH <sub>3</sub>	12	28	19	18	1	1	20,423	20,099	17.805	770.877	3600.701	3600.701
HNO <sub>3</sub>	13	12	26	18	1	1	669,988	487,254	0.012	0.012	1769.982	1769.982
ОН	14	13	245	221	3	3	42,866	31,976	0.005	0.003	35,877.030	19,267.804
HF	15	14	6	6	1	1	107	107	41.111	41.111	11,535.570	11,535.570
HCl	16	15	17	17	2	2	533	613	20.240	20.240	13,457.841	13,458.024
HBr	17	16	16	16	2	2	1293	1293	16.232	16.231	9758.312	9758.312
HI	18	17	9	9	1	1	806	806	12.509	12.509	8487.305	8487.305
ClO	19	18	12	16	2	2	7230	11,501	0.015	0.015	1207.639	1207.639
OCS	20	19	192	164	6	5	33,809	29,361	0.381	0.381	4199.671	4199.671
H <sub>2</sub> CO	21	20	17	17	3	3	37,050	37,050	0.000	0.000	3099.958	3099.958
C <sub>2</sub> H <sub>6</sub>	22	27	6	6	2	2	28,439	28,439	706.601	706.601	3000.486	3000.486
CH <sub>3</sub> D(§)	23	(§)	26	26	2	2 2(§)	49,237 <b>(</b> §)	49,237 <b>(</b> §)	7.760	7.760	6510.326	6510.326
$C_2H_2$	24	26	118	118	2	2(8)	11,340	11,340	604.774	604.774	9889.038	9889.038
$C_2H_2$ $C_2H_4$	25	38	12	12	2	2	18,378	18,378	701.203	701.203	3177.173	3177.173
GeH <sub>4</sub>	26	ABS	1	ABS	1	ABS	824	ABS	1937.371	ABS	2224.570	ABS
HCN	20	23	775	30	4	3	82,042	4253	0.006	0.015	17,581.009	3423.927
C <sub>3</sub> H <sub>8</sub>	28	ABS	1	ABS	1	ABS	8983	ABS	700.015	ABS	799.930	ABS
$C_{2}N_{2}$	28	ABS	7	ABS	1	ABS	2577	ABS	203.955	ABS	2181.690	ABS
$C_2 N_2$ $C_4 H_2$	29 30	ABS	, 1509		1	ABS	119,480	ABS	191.635	ABS	730.2352	ABS
HC <sub>3</sub> N	31	ABS	3302		1	ABS	179,347	ABS	463.604	ABS	755.696	ABS
HOCI	32	21	6	8	2	льз 2	17,862	16,276	403.004 0.024	1.081	3799.682	3799.682
	32 33	21	1	° 1		2	17,862	120				
N <sub>2</sub>		22 24	1 14		1	2			1992.628	1992.628	2625.497	2625.497
CH₃Cl	34	24 25		83	2		18,344	196,171	674.143	0.873	3172.927	3172.927
H <sub>2</sub> O <sub>2</sub>	35		130	130	1	1	126,983	126,983	0.043	0.043	1730.371	1730.371
H <sub>2</sub> S	36	31	30	30	3	3	20,788	20,788	2.985	2.985	4256.546	4256.547
HCOOH	37	32	8	8	1	1	62,684	62,684	10.018	10.018	1889.334	1889.334
$COF_2$	38	29	7	7	1	1	83,750	70,601	725.005	725.005	2001.348	2001.348
$SF_6(*)$	39	30	6	3	1	1	92,398	2,889,065(*)	588.488	580.000	975.788	996.000
$C_3H_4$	40	ABS	22	ABS	1	ABS	19,001	ABS	288.912	ABS	673.479	ABS
HO <sub>2</sub>	41	33	4	4	1	1	38,804	38,804	0.173	0.173	3675.818	3675.818
ClONO <sub>2</sub> (*)	42	35	7	3	2	2	356,899	32,199 <b>(*)</b>	0.636	763.641	797.741	797.741
CH <sub>3</sub> Br	43	40	6	6	2	2	36,911	36,911	794.403	794.403	1705.612	1705.612
CH <sub>3</sub> OH	44	39	16	16	1	1	19,897	19,897	0.019	0.019	1407.205	1407.205
NO <sup>+</sup>	45	36	6	6	1	1	1206	1206	1634.831	1634.831	2530.462	2530.462
HNC	46	ABS	84	ABS	1	ABS	5619	ABS	0.217	ABS	4814.904	ABS
C <sub>6</sub> H <sub>6</sub>	47	ABS	1	ABS	1	ABS	9797	ABS	642.427	ABS	705.262	ABS
$C_2HD$	48	ABS	348	ABS	1	ABS	15,512	ABS	416.785	ABS	3385.564	ABS
CF <sub>4</sub>	49	42	5	5	1	1	60,033	60,033	594.581	594.581	1312.647	1312.647
CH₃CN	50	41	2	2	1	1	17,172	3572	890.052	890.052	1650.000	945.655
0	ABS	34	ABS	1	ABS	1	ABS	2	ABS	68.716	ABS	158.303
HOBr	ABS	37	ABS	1	ABS	2	ABS	4358	ABS	0.155	ABS	315.908

Note: ABS stands for a molecular species not included in the actual database (HITRAN or GEISA)

(§) CH<sub>3</sub>D considered as an individual molecule in GEISA; but as an isotopologue of CH<sub>4</sub> in HITRAN.

(§) For HITRAN, column 5, sub-column "H", includes:

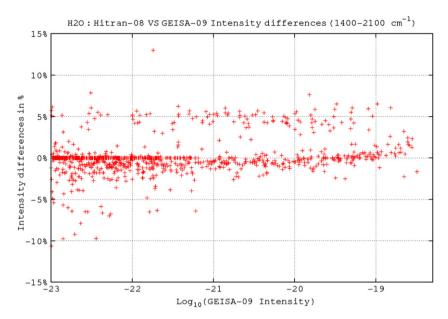
• for CH<sub>4</sub> (Mol. "6"), total # lines of isotopologues numbered "1" and "2" (coded "211" and "311" [7], respectively, in GEISA);

• for CH<sub>3</sub>D (Mol. "23"), total # lines of CH4 isotopologues numbered "3" and "4" (coded "212" and "312" [7], respectively, in GEISA).

(\*) Molecule included in HITRAN 2008 supplemental line list.

(in cm<sup>-1</sup>) are in the last four columns. The parameters for molecules  $SF_{6}$ , ClONO<sub>2</sub> and  $CF_{4}$  have been archived in the supplemental line list of HITRAN-08 whereas they are kept

in the main list of GEISA-09. The format of HITRAN-08 line parameters [13] is different from that of GEISA. Among the GEISA management software capabilities a program has



**Fig. 2.** Quantitative comparison between  $H_2O$  intensity values in GEISA-09 and HITRAN-08 [13]. Line intensity differences (GEISA-09 value minus HITRAN-08 value), in percent of GEISA-09 value, are on the Y-axis; the base 10 logarithm of GEISA-09 line intensity is on the X-axis. Data points are indicated by a read +.

been created which makes it possible to convert routinely the format of one database into the other database format, in their actual public release (GEISA-09 to HITRAN-08 format, or HITRAN-08 to GEISA-09 format). This has proven to be an efficient added capability for both database users especially for the purpose of easier identification and evaluation of the impact of spectroscopic content in specific applications (such as planetary atmosphere radiative transfer modeling).

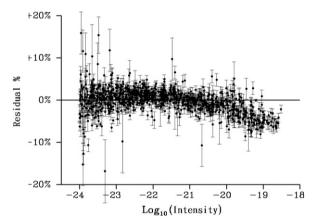
#### 2.2. Description of updates per individual molecules

#### 2.2.1. H<sub>2</sub>O (molecule 1)

The water molecule is of great interest both in terrestrial and in planetary studies, not the least because of its interference during ground-based observations.

The GEISA-09  $H_2O$  update involves spectroscopic parameters from three different origins, i.e. in the spectral region 500–7973 cm<sup>-1</sup>, the JPL data of Toth are available with their related description and references from the mark4sun website at http://mark4sun.jpl.nasa.gov/spec data.html. These data represent a total of 36,849 lines.

In the 10–2000 cm<sup>-1</sup> spectral region, for the normal isotopologue  $H_2^{16}$ O, updated line parameters are computed using the results of Coudert et al. [20]. This update covers line position and line intensity analyses of data up to the second triad as well as line strength (or line intensity) measurements for  $v_2$  band transitions. Using the spectroscopic parameters from this reference and the theoretical approach of Lanquetin et al. [21], a line list of 5624 entries was generated with a line intensity cutoff of  $10^{-27}$  cm<sup>-1</sup>/ (molecule cm<sup>-2</sup>). This calculation along with the line measurements of Ref. [20] revealed that experimental line intensity values for transitions belonging to the  $v_2$  band in the 1000–2000 cm<sup>-1</sup> range were underestimated in previous measurements of Toth in 1998 [22], for the strongest



**Fig. 3.** Residuals of observed H<sub>2</sub>O line intensities of Toth Ref. [22] minus those from Coudert et al. [20] (archived in GEISA-09). The X-axis is the base 10 logarithm of the observed line intensity in cm<sup>-1</sup>/ (molecule cm<sup>-2</sup>). The Y-axis corresponds to the residual in percent of the observed line intensities [22]. Each data point is indicated by a dot, error bars are also drawn. For clarity, the figure only displays the 967 strongest transitions from Ref. [22], belonging to the v<sub>2</sub> band, with an intensity larger than  $10^{-24}$  cm<sup>-1</sup>/(molecule cm<sup>-2</sup>).

transitions in this region. Fig. 3 shows the differences between the new intensity values from Coudert et al. [20] and those reported in Ref. [22]. In agreement with Ref. [20], this figure emphasizes that discrepancies of about -5% arise for strong transitions with an intensity on the order of  $10^{-19}$  cm<sup>-1</sup>/(molecule cm<sup>-2</sup>).

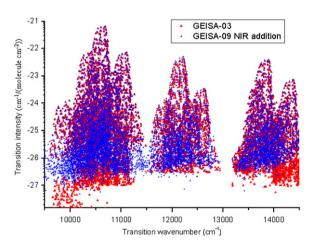
For the spectral range 9500–14,500 cm<sup>-1</sup>, line positions and intensities were taken from Tolchenov and Tennyson [23]. These data, representing 12,027 entries, came from a refit of room temperature Fourier transform absorption spectra of pure, natural abundance-water vapor by Schermaul et al. [24,25] recorded at path lengths from 5 up to 800 m. These parameters have demonstrated [23] to give a more consistent representation of the underlying spectrum than previous studies. In this spectral region, line broadening and shifting due to  $N_2$  and  $O_2$ pressure effects are included from calculations which use a semi-empirical approach based on impact theory modified by introducing additional parameters to extend the use of empirical data [26]. This method was further developed by using anharmonic wavefunctions in the estimates of the line parameters. The main feature is the use of a complete set of high accuracy vibration-rotation dipole transition moments calculated for all possible transitions using wavefunctions determined from variational nuclear motion calculations and an ab initio dipole moment surface [27]. Full details of this approach are described in Ref. [28], the results of H<sub>2</sub>O line parameters calculation and comparison with experimental data are presented in Refs. [29-31].

Fig. 4 exhibits the differences between H<sub>2</sub>O lines, in the frequency range 9500–14,500 cm<sup>-1</sup>, archived in GEISA-03, and those added in GEISA-09. The exponents of the intensity values (expressed in cm<sup>-1</sup>/(molecule cm<sup>-2</sup>)) at 296 K) are on the *Y*-axis and the transition wavenumbers on the *X*-axis. Data included in each GEISA edition are identified by different colors: red triangles for GEISA-03 and blue crosses for GEISA-09.

In practice, the resulting total file the GEISA-09 update has been processed as follows: as a first step Toth's data were retained and replaced, for the main  $H_2^{16}$ O isotopologue, by Coudert's data for lines with similar quantum identification; as a second step the file was finalized by adding the new data for the 9500–14,500 cm<sup>-1</sup> spectral region. The GEISA-09 H<sub>2</sub>O archive comprises 67,789 entries against 58,726 in GEISA-03.

### 2.2.2. CO<sub>2</sub> (molecule 2)

Carbon dioxide, like water, is an ubiquitous species observed in most of the solar system planets. To accommodate planetary applications, the GEISA-09 line list



**Fig. 4.**  $H_2O$  intensity versus wavenumber for transitions present in GEISA-03 ( $\bigwedge$  in red) and for those added in GEISA-09 (+ in blue), in the NIR spectral region.

update has been processed with 412,831 new transitions from seven isotopologues ( ${}^{12}C^{16}O_2$ ,  ${}^{13}C^{16}O_2$ ,  ${}^{16}O^{12}C^{18}O$ ,  ${}^{16}O^{12}C^{17}O$ ,  ${}^{16}O^{13}C^{18}O$ ,  ${}^{16}O^{13}C^{17}O$  and  ${}^{12}C^{18}O_2$ ) between 5.9 and 12,784.0 cm<sup>-1</sup>. It has to be noted that 788 transitions, of the two other isotopologues species:  ${}^{13}C^{18}O_2$  and  ${}^{17}O^{12}C^{18}O$ , have been retained from GEISA-03 in the final GEISA-09 CO<sub>2</sub> line list. The increase in the number of transitions (from 76,826 to 413,619) compared to the GEISA-03 list (see Table 2) arises from lowering the minimum intensity to  $10^{-30}$  cm<sup>-1</sup>/(molecule cm<sup>-2</sup>) at 296 K, in the seven isotopologues update entries, and merging the two compilations: the CDSD-296 databank [32] and partly the JPL near-infrared line list [33], as explained below.

The current version of the CDSD-296 databank is an extension and development of its previous 2003 version [34] which was used in GEISA/IASI [10] and GEISA-03. For the four most abundant isotopologues <sup>12</sup>C<sup>16</sup>O<sub>2</sub>, <sup>13</sup>C<sup>16</sup>O<sub>2</sub>, <sup>16</sup>O<sup>12</sup>C<sup>18</sup>O and <sup>16</sup>O<sup>12</sup>C<sup>17</sup>O, the line positions and line intensities are calculated using new sets of effective Hamiltonian and effective dipole moment constants. These new constants are determined by including extensive new measurements in the fitting (see [35–59] and references therein); in particular, the data obtained at JPL and at the Joseph Fourier University (Grenoble, France) resulted in better accuracy and completeness for the near infrared calculations. Using Fourier transform spectroscopy experiments the first team has performed very precise measurements of both line positions and line intensities of nine isotopologues of carbon dioxide in the 4300–7000 cm $^{-1}$  region [33,38,48,52,57]. The second team used highly sensitive CW-CRDS experiments and measured line positions and line intensities of a large number of lines including very weak lines as low as  $10^{-29} \text{ cm}^{-1}/(\text{molecule cm}^{-2})$  of several isotopologues in the 5851–7045 cm<sup>-1</sup> region [35,40,46,51,55,56,58,59]. The parameters obtained by including these weak lines belonging to high *I* values or to hot band transitions considerably improved the extrapolation properties of elaborated models of effective Hamiltonian and effective dipole moment operators. The theoretical approach used for global modeling of high resolution spectra of carbon dioxide is presented in Refs. [60-63]. Extension of the wavenumber region for the rare isotopologues was done using the sets of the effective dipole moment parameters belonging to the most abundant isotopologues. In order to meet the needs of the modern infrared sensors the intensity cutoff was lowered to  $10^{-30}$  cm<sup>-1</sup>/(molecule cm<sup>-2</sup>) at 296 K. Because of this a large number of additional weak bands and weak lines corresponding to high values of the angular momentum quantum number of the strong bands became available in the new version of CDSD-296. The accuracy of the line parameters of these weak lines strongly relies on the extrapolation abilities of the models used. It was shown in Ref. [64] that the effective operator models, used for the generation of CDSD-296, provide reliable extrapolation properties.

On average, the residuals between CDSD calculated line positions and those observed are two times larger than measurement uncertainties. The CDSD calculated line intensities are practically always within their measurement uncertainties for all isotopologues. Air- and self-broadening parameters were calculated using the equations from Rothman et al. [65], but the air-induced pressure shift parameter was set to zero throughout. The current atmospheric version of the databank is available on the web site of the IAO: ftp.iao.ru/pub/CDSD-2008/296.

Finally, it was determined that some of the intensities in the near infrared line list from Toth et al. were more accurate than the reanalyzed values and that the newer pressure broadening coefficients (widths and shifts) in the Toth et al. studies [66,67] better represented the measured spectra. Therefore, this line list, consisting initially of 28,530 entries, has been retained for the GEISA-09 update too, adopting the following process for its inclusion: first, 15,788 lines whose intensities were lower than  $10^{-26}$  cm<sup>-1</sup>/(molecule cm<sup>-2</sup>) at 296 K for the two main isotopologues  ${}^{12}C^{16}O_2$  and  ${}^{13}C^{16}O_2$  and for all the isotopologues with intensities between  $10^{-29}$  and  $10^{-30}$  cm<sup>-1</sup>/ (molecule cm<sup>-2</sup>) at 296 K (Brown private communication) were discarded; second, the 12,742 remaining lines were merged with the CDSD data, replacing them when the quantum identification was the same.

With this change, the choice of Toth et al. [66,67] for air-broadening, self-broadening and air-induced pressure shift of the line parameters is included in GEISA-09. These broadening parameters were replaced, for  $^{12}C^{16}O_2$  lines with the same quantum numbers, by the results from the latest work of Predoi-Cross et al. [68] for the temperature dependences of air-broadened CO<sub>2</sub> widths, temperature dependence of air-induced pressure shift and temperature dependence of the self-broadened half-widths. The parameters from Predoi-Cross et al. were implemented for the entire line list when available. Since the database was completed, a new effort to predict air-broadened pressure shifts has been undertaken by Hartmann [69] which will be considered for future database updates.

#### 2.2.3. O<sub>3</sub> (molecule 3)

An update of the line positions and intensities has been made for the three main isotopologues of ozone,  ${}^{16}O_3$ ,  ${}^{16}O^{16}O^{18}O$ , and  ${}^{16}O^{18}O^{16}O$ . For the main isotopologue  ${}^{16}O_3$ , the list of the 27 newly included bands (spectral range from 1632 to 4845  $cm^{-1}$ ) in GEISA-09 is given in the first column of Table 4 with associated spectral interval  $(cm^{-1})$ , number of lines and sum of line intensities, listed under columns 2-4, respectively. Table 5 lists the 28 updated bands with a similar display. These data cover the spectral range from 1613 to 4845 cm<sup>-1</sup>. The line list is given with an intensity cutoff of  $2 \times 10^{-26}$  cm<sup>-1</sup>/(molecule cm<sup>-2</sup>) at 296 K for 100% <sup>16</sup>O<sub>3</sub> abundance. These results are based on the analyses of the absorption spectra recorded in the GSMA laboratory using the FTS of the Champagne-Ardennes University (Reims, France) [70]. The calculations of the line positions were made using the Hamiltonian parameters for the lower states (000), (100) and (001)from Ref. [71], for the (010) state from Ref. [72] and for the (020) state from Ref. [73].

The line positions of three bands associated with the  $(0\ 3\ 1)$  upper state  $(3v_2+v_3-2v_2,\ 3v_2+v_3-v_2,\ and\ 3v_2+v_3)$  have been calculated using Hamiltonian parameters of Ref. [74]. The transition moment parameters of the  $v_2+v_3$  band [75] were used for calculation of line intensities for the  $3v_2+v_3-v_2$  band. The line intensities of two other bands were calculated with the transition moment parameters of Ref. [74].

Table 4

New ozone bands	$({}^{16}O_3)$	in the	GEISA-09	edition.
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Band	Spectral region (cm <sup>-1</sup> )	Number of lines	Sum of line intensities $(10^{-22} \text{ cm}^{-1}/(\text{molecule cm}^{-2}))$
031-020	1632–1711	1109	1.747
022-020	1921-2067	1046	0.740
121-020	1984-2079	1817	14.342
130-001	1991-2061	3	0.005
130-100	2040-2102	10	0.026
201-010	2281-2325	11	0.004
031-010	2333-2407	742	0.477
022-010	2603-2769	1629	1.740
131-020	2666-2741	899	0.834
031-000	3032-3111	689	0.420
130-000	3133-3249	384	0.126
022-000	3256-3511	1826	1.234
121-000	3286-3480	1764	7.481
131-010	3369-3440	910	0.694
113-100	3506-3566	466	0.197
014-001	3525-3605	992	1.316
113-010	3864-3968	1466	4.398
014-010	3875-3968	183	0.076
320-010	3888-4000	279	0.175
202-000	4034-4207	1387	1.108
131-000	4065-4145	714	0.460
301-000	4179-4264	1213	2.489
221-000	4444-4525	1066	1.041
014-000	4522-4700	1998	1.638
113-000	4562-4668	1599	8.814
320-000	4586-4700	587	0.435
212-000	4700-4845	924	0.415

Table 5	
Updated ozone bands ( <sup>16</sup> O <sub>3</sub> ) in the GEISA-09 e	edition.

Band	Spectral region (cm <sup>-1</sup> )	Number of lines	Sum of line Intensities $(10^{-21} \text{ cm}^{-1}/(\text{molecule cm}^{-2}))$
111-100	1613–1849	1271	0.269
012-001	1616-1826	1581	0.645
111-001	1629–1854	1557	0.131
012-100	1637-1706	85	0.004
210-100	1701-2051	1663	0.198
210-001	1719-2066	388	0.015
003-100	1848-2104	1920	1.183
003-001	1867-2098	2847	1.313
102-100	1869-2071	2206	0.429
012-010	1872-2120	3794	3.221
201-100	1888-2243	2831	10.979
201-001	1896-2289	2165	0.331
102-001	1901-2086	2965	15.787
111-010	1918-2220	3520	43.121
210-010	2005-2353	3050	0.844
300-001	2012-2313	1804	0.921
300-100	2021-2288	2508	0.475
003-010	2254-2396	1809	1.199
102-010	2270-2407	479	0.040
130-010	2424-2552	487	0.019
012-000	2590-3025	3886	3.293
111-000	2626-3050	3604	25.087
121-010	2678-2774	1851	1.658
210-000	2704-3156	3327	0.812
003-000	2907-3202	4512	141.143
201-000	2919-3273	2706	7.910
102-000	2925-3196	4646	13.774
300-000	2955-3398	2445	0.472

The line positions of six bands associated with the upper states (0 2 2) and (1 2 1) have been calculated using the Hamiltonian parameters for the upper states from Ref. [76]. The calculations of the line intensities of the  $2v_2+2v_3$  and  $v_1+2v_2+v_3$ ,  $2v_2+2v_3-v_2$  and  $v_1+2v_2+v_3-v_2$ ,  $2v_2+2v_3-2v_2$  and  $v_1+2v_2+v_3-2v_2$  bands were made with the transition moment parameters from Refs. [76–78], respectively.

The line positions of four bands of Table 4 and of all bands of Table 5 (except the band  $v_1+2v_2+v_3-v_2$ ) associated with the upper states  $\{(0 1 2), (1 1 1), (2 1 0), (2$ (003), (102), (201), (130), (300)} have been calculated using the Hamiltonian parameters for the upper states from Ref. [79]. The transition moment parameters for the cold bands (2590-3400 cm<sup>-1</sup> spectral range) of these states are given in Ref. [79]. The calculations of the main part of the hot bands line intensities have been done with the transition moments given in Refs. [75,78,80]. The dipole moment transitions of the  $2v_1 + v_3 - v_2$ ,  $v_1 + 2v_3 - v_2$ and  $3v_3 - v_1$  bands can be found at the web sites of the S&MPO system [81], similarly in Russia: http://smpo. iao.ru/1446x915/en/tran/par/1/8-2/; http://smpo.iao.ru/ 1446x915/en/tran/par/1/8-3/, or in France: http://ozone. univ-reims.fr/1446x915/en/tran/par/1/8-2/; http://ozone. univ-reims.fr/1446x915/en/tran/par/1/8-3/.

Three bands of the (1 3 1) upper state have been calculated with the Hamiltonian parameters [82] and the transition moment parameters [82,76,80] for the cold and hot bands, respectively.

The line positions of the eight bands associated with the upper states {(014), (113), (320)} and the line intensities of cold bands have been calculated using the Hamiltonian and the transition moments parameters from Ref. [83]. The transition moment parameters from Refs. [84,85] were used to calculate the line intensities of the  $v_1+v_2+3v_3-v_1$ ,  $v_2+4v_3-v_3$ , and  $v_1+v_2+3v_3-v_2$ hot bands. Estimates of the transitions moments of the  $4v_3$  and  $3v_1+v_2$  bands [86] were used for the calculations of the line intensities of the  $v_2+4v_3-v_2$ , and  $3v_1+2v_2-v_2$  hot bands.

The calculations for the  $2v_1+2v_3$ ,  $3v_1+v_3$ ,  $2v_1+2v_2+v_3$  and  $2v_1+v_2+2v_3$  bands are based on the results from Refs. [87–90].

Table 6 lists 9 bands in the 5935–6394 cm<sup>-1</sup> spectral region. These results were obtained by using CW–CRDS technique [91,92]. The spectra were recorded in Laboratoire de Spectrométrie Physique at the Joseph Fourier University (Grenoble, France). The analysis and theoretical modeling of these data have been reported in Refs. [92,93]. Note that the 6017–6131 and 6318–6394 cm<sup>-1</sup> spectral ranges are dominated by a band labeled as  $2v_1+2v_2+3v_3$ . See Refs. [92,93] for more details.

The spectral interval  $1854-2768 \text{ cm}^{-1}$  has been updated for the two isotopologues  ${}^{16}O^{16}O^{18}O$  and  ${}^{16}O^{18}O^{16}O$ , this region relates to the bands:  $2\nu_3$ ,  $\nu_1+\nu_2+\nu_3-\nu_2$ ,  $\nu_1+\nu_3$ ,  $2\nu_1$ , and  $\nu_1+\nu_2+\nu_3$ . Bands  $2\nu_3$ ,  $\nu_1+\nu_2+\nu_3-\nu_2$ ,  $2\nu_1$ ,  $\nu_1+\nu_2+\nu_3$  of  ${}^{16}O^{16}O^{18}O$  as well as bands  $\nu_1+\nu_2+\nu_3$  and  $\nu_1+\nu_2+\nu_3-\nu_2$  of  ${}^{16}O^{18}O^{16}O$  have been

Table 6	
New GEISA-09 ozone bands ( <sup>16</sup> O <sub>3</sub> ) from CW-CRDS spectra [91	.921.

Band	Spectral region (cm <sup>-1</sup> )	Number of lines	Sum of line Intensities $(10^{-24} \text{ cm}^{-1})/(\text{molecule cm}^{-2}))$
034-000	5935-6083	610	1.178
105-000	5971-6071	1006	2.456
124-000	6004-6363	1933	4.566
223 <sub>I</sub> -000	6017-6131	1578	13.188
510-000	6030-6139	272	0.401
025-000	6225-6311	913	7.656
430-000	6295-6395	75	0.298
501-000	6301-6366	685	6.335
22311-000	6318-6394	717	6.758

included in the GEISA-09 database for the first time. The calculations of all bands of both isotopologues were made using the Hamiltonian parameters for the lower states for the (000) and (010) states from Ref. [94]. Hamiltonian parameters of the upper vibrational states correspond to Ref. [95] for <sup>16</sup>O<sup>16</sup>O<sup>18</sup>O and to Ref. [96] for <sup>16</sup>O<sup>18</sup>O<sup>16</sup>O. The transition moment parameters of both species, given by Barbe and De Backer-Barilly [97] have been obtained from studies of Fourier-transform ozone spectra enriched in oxygen-18. The broadening parameters (both air and self) for all B-type and A-type bands were derived, respectively, from those of  $v_1$  and  $v_3$  bands. These values are originated from the S&MPO system [81] (see Ref. [54] at http://smpo.iao.ru/1280x795/en/refs/9/ therein). These parameters are different from those reported in HITRAN-08 [13].

The temperature dependence coefficient n=0.76 of the air pressure broadening has been attributed to all the transitions. The absolute intensities are obtained from direct experimental measurements for each band (no indirect normalization).

Line lists are given with a cutoff intensity of  $1 \times 10^{-24} \text{ cm}^{-1}/(\text{molecule cm}^{-2})$  at 296 K for 100% abundances of  ${}^{16}\text{O}{}^{16}\text{O}{}^{18}\text{O}$  and  ${}^{16}\text{O}{}^{16}\text{O}{}^{16}$ O. It has to be noted that all the ozone data in GEISA-09 are given in natural abundance of isotopologues.

#### 2.2.4. N<sub>2</sub>O (molecule 4)

The N<sub>2</sub>O line list has been almost completely revised. Only the rotational part: 0.83–45.263 cm<sup>-1</sup> has been kept from GEISA-03 (451 entries). All 50,182 lines of Toth's data [98–101] from the website: http://mark4sun.jpl.nasa.gov/ n2o.html, which cover the spectral range 525.462272– 7796.633112 cm<sup>-1</sup>, have been included in GEISA-09 line list. As a consequence, three new isotopologues have been added:  ${}^{15}N_{1}^{16}O$ ,  ${}^{14}N_{15}N_{18}O$  and  ${}^{15}N_{14}N_{18}O$ , representing a total of 668 entries. The source of the N<sub>2</sub>O broadening parameters are from Ref. [101] which gives N<sub>2</sub> and air widths and shifts of N<sub>2</sub>O. These data were used to generate the parameters used for the website http://mark4sun.jpl. nasa.gov in 2004 under science data.

The  $N_2O$  GEISA-09 archive now comprises 50,633 entries and eight isotopologues. This represents an increase of 23,952 entries over GEISA-03 (26,681 entries).

The JPL catalog contains new entries for the rotational transitions of N<sub>2</sub>O in its  $v_2$ =0, 1, and 2 vibrational states as well as for the singly substituted isotopologues in their

ground vibrational states. These entries are based in particular on [102]. These entries will be considered for the next update of GEISA.

#### 2.2.5. *CH*<sub>4</sub> (molecule 6)

Many of the infrared methane line parameters of  ${}^{12}\text{CH}_4$ were updated between 0 and 3300 cm<sup>-1</sup>, but little changes were made for the  ${}^{13}\text{CH}_4$  parameters. As noted in Sections 1 and 2, entries for methane isotopologues  ${}^{12}\text{CH}_3\text{D}$  and  ${}^{13}\text{CH}_3\text{D}$  are included in GEISA as an independent molecule, CH<sub>3</sub>D, numbered "23" (see Table 2); related updates are described below. At the longer wavelengths, a minimum intensity limit of  $10^{-29}$  cm/molecule at 296 K was applied out of planetary considerations, but the weak lines were still not included in the near-IR regions. Misaligned fields in the near-IR quantum numbers were corrected, but only a few new assignments (and thus lower state energies) were entered to existing entries. Significant changes were made for air-broadening coefficients between 5800 and 6180 cm<sup>-1</sup>.

Below 3300 cm<sup>-1</sup>, new calculated <sup>12</sup>CH<sub>4</sub> line positions and intensities were obtained from the global analysis by Albert et al. [103] of the three lowest polyads (ground state, dyad from 900 to 1900  $\text{cm}^{-1}$  and pentad from 1900 to  $3400 \text{ cm}^{-1}$ ). In the far-IR, the intensities of ground state-ground transitions were adjusted by 16% based on Wishnow et al. [104], but no change was required for the dyad-dyad ( $v_2 - v_2$ ,  $v_2 - v_4$ ,  $v_4 - v_4$ ) hotbands. Some predicted pentad ( $2v_4$ ,  $v_2 + v_4$ ,  $v_1$ ,  $v_3$  and  $2v_2$ ) positions were recomputed using semi-empirical upper state energy levels obtained by adding observed positions to calculated lower state energies. The hot band parameters between 900 and  $3500 \text{ cm}^{-1}$  and of the Octad (3200 and  $4900 \text{ cm}^{-1}$ ) were taken from GEISA-03 rather than the global study because the prior database had better accuracies for the strongest features in the interval; a minimum intensity limit for hot bands was set to  $10^{-27} \text{ cm}^{-1}$ /(molecule cm<sup>-2</sup>) at 296 K.

The line list for methane near  $6000 \text{ cm}^{-1}$  was somewhat improved using new measurements of intensities, empirical lower state energies and broadening parameters of the stronger features. First, the intensities and widths for the 5860–6180 cm<sup>-1</sup> region were replaced by results from Frankenberg et al. [105]. This also included implementation of the empirical lower state energies of Margolis [106,107] which were missing in GEISA-03. In addition, lower state values from Gao et al. [108] were added. However, several thousand weak lines  $(<10^{-24} \text{ cm}^{-1})$  molecule cm<sup>-2</sup>) are still missing between 5500 and 6180 cm<sup>-1</sup>. There are a number of recently published and ongoing studies which will help to improve the near infrared (4800–7700 cm<sup>-1</sup>) line parameters [109–118].

For broadening, relatively few (< 3000) direct measurements of widths and pressure shifts are available for methane transitions so that default values for self- and air-broadened widths, air-induced pressure shifts and temperature dependences are applied (similar to those used in earlier versions of GEISA see [119,120]. For the 7.5 um region of the Dvad, new measurements of  $\sim$  500 transitions from Smith et al. [121] were inserted for selfand air-broadening widths, shifts and temperature dependence of widths. For the 3.3 µm region of the Pentad,  $\sim$ 3800 theoretically predicted broadening coefficients (air-widths, pressure shifts and temperature dependences) from Antony et al. [122] and  $\sim$  500 prior measurements [119] were inserted for  $v_3$ . At 2.3  $\mu$ m (the Octad), the self- and air-broadening parameters of Predoi-Cross et al. [123,124] were retained in the list carried over from the GEISA-03 database.

In the 1.66  $\mu$ m region (the Tetradecad) over 480 airbroadened widths and shifts and some temperature dependence were inserted between 5560 to 5860 cm<sup>-1</sup> [111], while the scaled N<sub>2</sub>-broadening reported by Frankenberg et al. [105] were used from 5860 to 6184 cm<sup>-1</sup>. Otherwise, defaults constants of 0.75 below 5860 cm<sup>-1</sup> or 0.85 above 5860 cm<sup>-1</sup> were set for the temperature dependence.

Lastly, the current methane database is customized to interpret atmospheric remote sensing of the Earth. Further near-IR analyses will be needed for planetary and stellar applications. (e.g., [125]). Calculations of partition functions [126] and much weaker transitions can be found at http://www.iao.ru/mirs or http://icb. u-bourgogne.fr/JSP/TIPS.jsp. However, extrapolations to higher values of quanta provide less accurate parameters, particularly for the intensities.

## 2.2.6. O<sub>2</sub> (molecule 7)

Line parameters for the oxygen A-band  $(b^1\Sigma_g^+ \leftarrow X^3\Sigma_g^-)$  were revised for  ${}^{16}O_2$  and  ${}^{16}O^{18}O$ , and those of  ${}^{16}O^{17}O$  were added in the 0.76 µm region. The line positions, intensities, air- and self-broadened half-widths and air-induced pressure shifts were taken from the work of Robichaud et al. [127–130] who performed CW-CRDS of the P branch. The positions now have accuracies of 0.00006 cm<sup>-1</sup> or better for  ${}^{16}O_2$  and  ${}^{16}O^{18}O$  and 0.00050 cm<sup>-1</sup> for  ${}^{16}O^{17}O$  through calibration against atomic potassium calibration standards [131]. The differences between the old and new positions are on average 0.0007 cm<sup>-1</sup> [127] for  ${}^{16}O_2$  and 0.002 cm<sup>-1</sup> [129] for  ${}^{16}O^{17}O$ , but much larger for  ${}^{16}O^{18}O$  (up to 0.20 cm<sup>-1</sup>) because the latter were based on 60-year-old results [132].

Line intensities changed only slightly for the first two isotopologues: -0.8% for  $^{16}O_2$ , +1% for  $^{16}O^{18}O$ , but  $\pm 5\%$  for  $^{16}O^{17}O$  (depending on the rotational quanta). The

accuracies are thought to be  $\pm$  1% or better for the first two species, but more study is needed for  ${}^{16}\text{O}{}^{17}\text{O}$ .

For all three species, the widths are computed via an expression from Yang et al. [133]

$$HWHM = A + \frac{B}{1 + c_1 J' + c_2 J'^2 + c_3 J'^4}$$
(1)

by using the <sup>16</sup>O<sub>2</sub> constants from Table 6 of Robichaud et al. [128] based on retrievals done with Galatry (not Voigt) profiles. For the widths, the values at high quantum numbers (J > 22), previously in error by more than 40% near J=30, are now thought to be accurate to  $\pm 2\%$ .

Pressure-induced shifts are still rather uncertain  $(\pm 0.003 \text{ cm}^{-1})$  with different studies in poor agreement (e.g., [129,134,135]). For the interim, the measured A-band pressure shifts of Robichaud et al. [127] for the P branch and the averages of shifts from Predoi-Cross et al. [134,135] for the R branch were inserted, along with the temperature dependence of widths from Brown and Plymate [136].

Finally, it should be emphasized that even with these improvements, the line parameters are not sufficient to reproduce atmospheric observations at 13,100 cm<sup>-1</sup> because Voigt line shapes are inadequate. The combined analyses of Tran and Hartmann [137], Predoi-Cross et al. [134,135] and Robichaud et al. [128–130] have demonstrated the need to consider line mixing, Galatry and/or speed dependence line shapes in order to model the oxygen A-band properly.

It has to be noted that the revised O<sub>2</sub> GEISA-09 line list does not derive from the GEISA-03 line list, but from the HITRAN 2004 [13] one. The major difference between the two line lists is for the intensity values, especially in the 1.27  $\mu$ m spectral region (Ref. [32] of Ref. [8]). As an unfortunate consequence of this alternate line list inclusion (due to a final mis-manipulation among different generated test files for data validation studies), the two lines closest to the  $\gamma$  band head (at 15,927.701 and 15,927.805 cm<sup>-1</sup>), present in GEISA-03, are now detrimentally missing in GEISA-09, propagating a technical error occurring in HITRAN 2004. This must be fixed in the next GEISA edition.

### 2.2.7. NO (molecule 8)

The GEISA-03 NO line list has been totally replaced by a new one provided by Goldman [138]. The new line list is partially based on the work described by Goldman et al. [139], and is equivalent to the updated NO in HITRAN-08 [13]. The updates mainly consist of: including, for the first time, Einstein-*A* coefficients to replace, in format field *M*, the former GEISA-03 transition probabilities (see Table 11) and the implementation of hyperfine splitting for the microwave and far infrared lines. Magnetic dipole satellite transitions between spin components of the electronic ground state have also been added, and are further identified by the letter "*m*" in the first field for the upper state quantum numbers. When lines with resolved hyperfine structure were not available from Ref. [139], they were taken from the JPL catalog [16].

These updates have increased the total number of NO transitions in GEISA from 99,123 to 105,079 (293 bands).

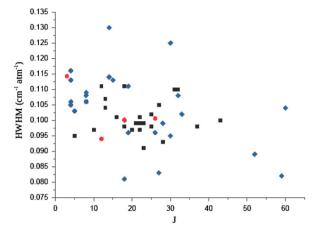
## 2.2.8. SO<sub>2</sub> (molecule 9)

Sulfur dioxide, SO<sub>2</sub>, is well known to be both of astrophysical and planetary importance. SO<sub>2</sub> is an important constituent on Venus [140-144] and Io [145-148], where it actively participates in the photochemistry of their atmospheres. It has also been observed in comets [149-152]. In the terrestrial atmosphere,  $SO_2$  is a trace species produced by both anthropogenic and natural sources; mainly present in the troposphere [153–155], it is a primary pollutant emitted by fuel combustion and responsible for the production of acid rain. The most important natural sources of SO<sub>2</sub> are the oxidation of sulfur compounds from oceans and marshes and from volcanic eruptions and outgassing. Most volcanic SO<sub>2</sub> emissions remain in the troposphere where the lifetime of the species strongly depends on the meteorological conditions. After major volcano eruptions, SO<sub>2</sub> is also present in the stratosphere [156–157] in high concentrations, where it is converted into sulfate aerosols which affect both stratospheric chemistry and climate. SO<sub>2</sub> has been detected on Io in the microwave and more recently at 19  $\mu$ m [158]; this is of high relevance to the studies of exchanges between the atmosphere and the surface of the satellite.

The GEISA-03 database provided SO<sub>2</sub> parameters in seven different spectral regions, which correspond to transitions in the microwave region and the 19.3, 8.6, 7.3, 4, 3.7 and 2.5  $\mu$ m spectral regions. However in the 19.3, 8.6 and 7.3  $\mu$ m spectral regions new studies [159–162] have been performed improving the corresponding spectral parameters. These three spectral regions are important for SO<sub>2</sub> measurements in atmospheres. The 7.3  $\mu$ m which is the strongest SO<sub>2</sub> infrared region unfortunately cannot be used for ground measurements of SO<sub>2</sub> since it is severely overlapped with the strong v<sub>2</sub> band of water vapor. On the other hand, the v<sub>1</sub> band, although about nine times weaker corresponds to a rather clear atmospheric window. Finally the rather weak 19.3  $\mu$ m region can be used for retrieving SO<sub>2</sub> in the atmosphere of planets [158].

Based on the new studies an improved line list including line positions, intensities, transition assignments and lower state energy levels has been generated. It includes not only, for the main isotopologue  ${}^{32}SO_2$ , the cold bands  $v_2$ ,  $v_1$  and  $v_{3}$ , but also the corresponding hot bands  $2v_{2} - v_{2}$ ,  $3v_{2} - 2v_{2}$ ,  $v_1+v_2-v_2$  and  $v_3+v_2-v_2$ , as well as the  $v_1$ ,  $v_3$ ,  $v_1 + v_2 - v_2$ ,  $v_2 + v_3 - v_2$ ,  $v_1 + v_3$  bands of <sup>34</sup>SO<sub>2</sub>, from the results of a series of papers [163-165] devoted to the high resolution study of the absorption of the <sup>34</sup>SO<sub>2</sub> species in the infrared. The resulting newly archived <sup>34</sup>SO<sub>2</sub> spectral line parameters are much better than the previous ones, related only to the  $v_1 + v_3$  band. The accuracy for line positions is estimated to be better than 0.001 cm<sup>-1</sup>. For line intensities the accuracy is estimated to be of the order of 2-3%degrading up to about 15% for high I or  $K_a$  transitions. Finally, the GEISA-09 SO<sub>2</sub> line list comprises a total of 68,728 lines among which 43,941 are new or updated entries. From GEISA-03 have been kept 24,787 entries of the main isotopologue <sup>32</sup>SO<sub>2</sub>, in two distinct spectral regions, i.e., 0.017394–256.241135 cm<sup>-1</sup> (9622 rotational lines) and 2433.192300–4092.948220 cm<sup>-1</sup> (15,165 lines).

As far as the pressure broadening coefficients are concerned the situation is different for air-broadening and self-broadening coefficients.



**Fig. 5.**  ${}^{32}SO_2$  Lorentz air-broadened half-width parameters (at 296 K) ( Microwave,  $\checkmark v_3$  band,  $\blacksquare v_1$  band) versus the quantum number *J* of the lower level of the transition.

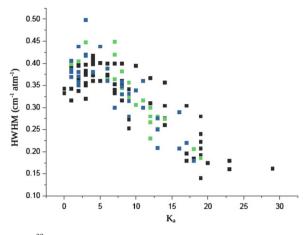
For the air-broadened half-width coefficients (HWHM) it turns out that it was only possible to estimate an average value for this parameter. In fact no variation of this parameter with respect to the lower quantum numbers J or  $K_a$  of the transitions could be determined. As an example, Fig. 5 presents the measured parameters (Y-axis) with respect to the lower quantum numbers I of the transitions (X-axis), for the microwave and the  $v_3$  and  $v_1$  bands spectral regions. It appears not possible to derive any clear variation (the same is true when these parameters are plotted versus the quantum number  $K_a$ ) so only an average value of 0.1025 cm<sup>-1</sup> atm<sup>-1</sup> could be determined. This value has been used for all the updated lines of isotopologue <sup>32</sup>SO<sub>2</sub>; in the case of isotopologue  $^{34}$ SO<sub>2</sub>, the value 0.1000 cm<sup>-1</sup> atm<sup>-1</sup> has been attributed to the lines of  $v_1 + v_3$  The GEISA-09 missing value  $-0.9999 \text{ cm}^{-1} \text{ atm}^{-1}$  has been given to the lines of  $v_2 + v_3 - v_2$  for both isotopologues. Related with the entries remaining from GEISA-03, the average value is  $0.1000 \text{ cm}^{-1} \text{ atm}^{-1}$  for the 9622 rotational lines and  $0.1100 \text{ cm}^{-1} \text{ atm}^{-1}$  for the other 15,165 lines.

The situation is quite different for the self-broadening parameters. It was possible indeed to observe a clear variation of these parameters with respect to the  $K_a$  quantum number of the lower state of the transitions (see illustration in Fig. 6 for the  $v_1$ ,  $v_2$ , and  $v_3$  bands, with a display similar to Fig. 5). On the other hand no variation with respect to the quantum number *J* could be observed.

Based upon these results, it was decided to include in the database the following values for the self-broadened half-width coefficients:

HWHM<sub>self</sub>=0.4 cm<sup>-1</sup>/atm for  $K_a ≤ 5$ HWHM<sub>self</sub>=0.156 cm<sup>-1</sup>/atm for  $K_a ≥ 21$ HWHM<sub>self</sub> is calculated through a linear interpolation for  $6 ≤ K_a ≤ 20$ 

For the updated transitions, these new parameters have been used for all the lines of main isotopologue  ${}^{32}SO_2$  except for those of band  $v_2 + v_3 - v_2$  for which the GEISA-09 missing value  $-9.9999 \text{ cm}^{-1} \text{ atm}^{-1}$  has been



**Fig. 6.** <sup>32</sup>SO<sub>2</sub> Lorentz self-broadened half-width parameters ( $v_2$ ,  $v_3$  v<sub>3</sub> band,  $v_1$  band) versus the quantum number  $K_a$  of the lower level of the transition.

given to both isotopologues, as well as to 83 lines of isotopologue  ${}^{34}SO_2$  in the spectral region 1165.402–1379.267 cm<sup>-1</sup>.

For the GEISA-03 retained transitions, the missing value has been attributed to the rotational transitions and the default value  $0.39 \text{ cm}^{-1} \text{ atm}^{-1}$  to all the other ones

As a consequence, an accuracy of 10–15% for the newly updated air-broadening and self-broadening parameters seems reasonable.

Finally, a "standard" default value of 0.75 has been used for the temperature dependence coefficient n of the air-broadening halfwidth.

It is worthwhile mentioning that the CDMS catalog provides an entry for  $v_2$  which is based on extensive rotational transitions in its  $v_2=0$  and 1 states [166] along with previous IR data. This entry may be the basis for a GEISA entry in the next update.

#### 2.2.9. NO<sub>2</sub> (molecule 10)

The study by Perrin et al. [167] provided accurate line positions and absolute intensities for several NO<sub>2</sub> bands, including the  $v_2$  and  $v_3$  fundamentals and their associated hot bands. Benner et al. [168] obtained precise line positions and relative intensities for the  $v_3$  band including accurate determinations of position differences for a large number of spin-splittings. In addition, air-broadened halfwidth and air-induced pressure shift coefficients and their variations with temperature were also determined for over 1000 transitions. These two studies [167,168] were combined to form an updated NO<sub>2</sub> line list at 6 µm. The positions and absolute line intensities are retained to values from Ref. [167] for the  $v_3$  band transitions, while the measured values of half-width, pressure-induced shift and the temperature dependence exponents of half-width coefficients were inserted line-by-line.

For all other transitions the values calculated using the empirical expressions of Ref. [168] were applied for the half-width, pressure shift and their temperature dependences. Values for higher  $K_a$  quantum numbers were constrained to the highest measured  $K_a$  ( $K_a$ =9 for half-width and  $K_a$ =7 for pressure-induced shift coefficients).

No pattern was discerned for the air-broadening temperature dependence exponents, and a simple linear equation in m (m=N'' for P and O branch transitions and N'' + 1 for R-branch transitions) was fit to the measurements. For selected widths, the RMS deviation was 2.5%. In GEISA-03, the air-broadened half-width coefficients of all transitions were set to a default value of  $0.067 \text{ cm}^{-1} \text{ atm}^{-1}$  at 296 K, the self-broadened halfwidth coefficients to  $0.095 \text{ cm}^{-1} \text{ atm}^{-1}$  at 296 K, air induced pressure-shift coefficients were set to zero and the temperature dependence exponents of air-broadened half-width coefficients were set to a default value of one. In the new database at 6 um, only the self-broadened half-width coefficients remain as default values  $(0.095 \text{ cm}^{-1} \text{ atm}^{-1} \text{ at } 296 \text{ K})$ , as was done in [8,9]. The NO<sub>2</sub> line lists are similar in GEISA-09 and HITRAN-08 [13].

#### 2.2.10. NH<sub>3</sub> (molecule 11)

The line parameters given in GEISA-03 for the spectral interval 0.058–5294.502 cm<sup>-1</sup> from Kleiner and Brown [169] and described in Kleiner et al. [170] have been slightly revised in GEISA-09, on the basis of an updated line list issued soon after the final completion of GEISA-03 final process. The NH<sub>3</sub> line lists are similar in GEISA-09 and HITRAN-08 [13] except for duplication of two lines in HITRAN, at 4561.037254 cm<sup>-1</sup> and 4568.372254 cm<sup>-1</sup>.

## 2.2.11. *PH*<sub>3</sub> (molecule 12)

Phosphine has been detected in the atmosphere of both Jupiter and Saturn [171,172] and is a significant absorber in the 5  $\mu$ m window in Jupiter where it was used to probe the deeper atmosphere [173]. Features of PH<sub>3</sub> near 3425 cm<sup>-1</sup> are clearly seen in ground-based spectra of Saturn [174,175], and line parameters for these bands are needed for the interpretation of data recorded by VIMS on the Cassini spacecraft [176].

Based on the work of Butler et al. [177], 9 new bands have been added in the region from 2724 to  $3602 \text{ cm}^{-1}$ , representing an increase of 8359 entries since GEISA-03 PH<sub>3</sub> archive. The collision-broadened parameters of the 770 to 2472 cm<sup>-1</sup> spectral range have been updated using the results in Ref. [177]. Over 8000 line positions and intensities of phosphine, between 2724.477 and 3601.652 cm<sup>-1</sup>, were measured at 0.0115 cm<sup>-1</sup> resolution.

Quantum assignments were made to most of the eight interacting vibrational states:  $3v_2$  (2940.8 cm<sup>-1</sup>),  $2v_2+v_4$  (3085.6 cm<sup>-1</sup>),  $v_2+2v_4$  (3214.9 cm<sup>-1</sup>),  $v_1+v_2$  (3307.6 cm<sup>-1</sup>),  $v_2+v_3$  (3310.5 cm<sup>-1</sup>),  $3v_4$  ( $\sim$  3345 cm<sup>-1</sup>),  $v_1+v_4$  (3426.9 cm<sup>-1</sup>), and  $v_3+v_4$  (3432.9 cm<sup>-1</sup>). However, a recent global study of PH<sub>3</sub> by Nikitin et al. [178] demonstrated the complexities of modeling this region and revealed the need to investigate the consistencies between band intensities at 5 and 3 µm.

#### 2.2.12. HNO<sub>3</sub> (molecule 13)

A very important improvement has been brought to the entire list of lines of HNO<sub>3</sub>. The entire GEISA-03 content (171,504 entries in the spectral range  $0.035141-1769.982240 \text{ cm}^{-1}$ ) has been replaced with data originating from two different sources, i.e., from Perrin [179], in the

spectral range 0.011922–769.982240 cm<sup>-1</sup>, and from Petkie [180] in the spectral range 0.155640–527.247646 cm<sup>-1</sup>.

In Perrin's work, an improved set of line positions, line intensities and line broadening parameters was generated in the infrared spectral region, using new and accurate experimental results concerning line positions and line intensities as well as sophisticated theoretical methods. The present update was performed in two steps, described in Refs. [181,182], respectively.

The first study [181] was performed in the 820– 1770 cm<sup>-1</sup> spectral range covered by the MIPAS instrument and the results of this first update are summarized in Table 5 of Ref. [181]. The line positions have been improved for the  $v_5$  and  $2v_9$  cold bands and  $v_5+v_9-v_9$ hot band around 11.2 µm and for the  $v_8+v_9$  and  $v_6+v_7$ bands around 8.3 µm (see details in Refs. [181,183] and in Refs. therein). In addition, the line intensities were updated in the 11.3, 8.3 and 7.6 µm spectral ranges by making use of the cross-sections measurements performed in Ref. [184].

The results of the second update are described in Table 1 of Ref. [182]. The intensities for the  $v_6$  and  $v_8$  bands centered at 646.826 and 763.154 cm<sup>-1</sup>, respectively, were decreased by about 20–30% as compared to GEISA-03. Near 11.3 µm approximate parameters for the  $v_5+v_7-v_7$  and  $v_5+v_6-v_6$  hot bands have been added to the line list for the first time. Also a complete update of the air-broadening parameters was performed in the 11 µm region following recent line-broadening calculations [185]. It should be noted that the air-broadening parameters implemented in the narrow Q branches of the  $v_8$  and  $v_5+v_9-v_9$  bands at 763.154 and 885.425 cm<sup>-1</sup>, respectively, account empirically for line mixing effects as evidenced by laboratory measurements.

The validation of these updates in the new line list was performed during several ground based, balloon borne or satellite measurements of atmospheric HNO<sub>3</sub> [181,185,186].

Future studies should concentrate to the improvements of HNO<sub>3</sub> line parameters in the 7.6  $\mu$ m region. Indeed this region which corresponds to the v<sub>3</sub> and v<sub>4</sub> bands located at 1325.7354 and 1303.5182 cm<sup>-1</sup>, respectively, needs major updates in term of line positions and intensities. Also, the previous studies in this region [188] did not consider resonances due to several dark states which perturb the 3<sup>1</sup> and 4<sup>1</sup> energy levels.

The cataloged spectral parameters of nitric acid have been updated in the millimeter/sub-millimeter-wave and 22 µm far-infrared regions. The calculated line parameters are based on the spectroscopic constants derived from the analyses of millimeter and sub-millimeter wave rotational spectra found in Refs. [189–191]. All predictions were calculated using the SPCAT program package [192]; (http://spec.jpl.nasa.gov/ftp/pub/calpgm/spinv.pdf) for a temperature of 296 K, an isotopic abundance of 0.989, a rotational partition function of 27,343, and a vibrational partition function of 1.304 [193].

In the mm/sub-mm-wave region, the pure rotational transitions from the vibrational states with band origins below  $1000 \text{ cm}^{-1}$  have been included in this update. These vibrational states account for about 97% of the

thermally populated molecules at 296 K. This includes transitions in the ground state,  $v_9=1$ ,  $v_7=1$ ,  $v_6=1$ ,  $v_8=1$ , and the interacting  $v_5=1/v_9=2$  dyad. The details of the analyses and measurements can be found in Refs. [189,190] and the set of references contained therein.

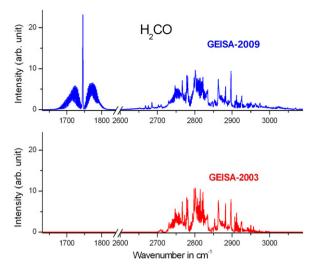
In the 22 µm far-infrared spectral region, line parameters for the fundamental  $v_9$  band as well as the two hot band  $v_9 - v_9$  and  $v_5 - v_9$ , have been updated. Line positions for the bands were calculated from the rotational analyses in Refs. [189,190] and the band origins determined in Refs. [183,194]. The high-resolution far-infrared spectrum in Ref. [195] was used both as a stringent test of the predicted far-infrared transition frequencies and to determine the relative intensities of the hot bands referenced to the intensity of the fundamental  $v_9$  band determined in Ref. [193]. Details of the far-infrared simulation can be found in Ref. [191].

The new HNO<sub>3</sub> GEISA-09 line list has been processed as the following: starting from Perrin's line list [179], Petkie's data have been included [180], replacing the Perrin's values for transitions with same quantum identifications. The final GEISA-09 HNO<sub>3</sub> line list comprises 669,988 entries in the spectral range 0.011922–1769.982240 cm<sup>-1</sup>.

#### 2.2.13. OCS (molecule 20)

Substantial revisions involving five isotopologues  ${}^{16}O^{12}C^{32}S$ ,  ${}^{16}O^{12}C^{34}S$ ,  ${}^{16}O^{13}C^{32}S$ ,  ${}^{16}O^{12}C^{33}S$ , and  ${}^{18}O^{12}C^{32}S$ provide new parameters for some 50 bands between 3800 and 4200 cm<sup>-1</sup>; 13 allowed and two forbidden bands arise from the ground state while the remainders are hot bands. The number of transitions increases from  $\sim$ 1100 transitions (for  $2v_3$  of five isotopologues and the  $v_2+2v_3-v_2$  of <sup>16</sup>O<sup>12</sup>C<sup>32</sup>S and <sup>16</sup>O<sup>12</sup>C<sup>34</sup>S) to 10,425 lines. Most of the line positions are calculated using the effective rovibrational energy constants based on a global analysis [196–200] whose line position accuracy was reported to be  $5 \times 10^{-5}$  cm<sup>-1</sup> [196]. The calculated line intensities are taken from analyses of new FTIR measurements [201-202] performed at JPL to support Venus studies. Sung et al. [201] reported line intensities of the  $2v_3$  band at 4101.387 cm<sup>-1</sup>,  $v_1+2v_2+v_3$ at 3937.427 cm<sup>-1</sup>, and  $4v_2 + v_3$  at 4141.212 cm<sup>-1</sup> of  $^{16}O^{12}C^{32}S$ . The new band strengths are in good agreement (1.3%) with the prior studies by Bermejo et al. [203] and Naim et al. [196]. Intensities of all the other bands are determined by Toth et al. [202] with many bands being measured for the first time, and their uncertainties range from 1% to 6% depending on bands. The line intensities vary through five orders of magnitude, but very weak unassigned features are omitted from the database pending further analysis.

The air- and self-broadened half-widths are computed, respectively, using Refs. [204–207]. The self-broadened temperature dependence exponents of  $v_1$  from Bouanich et al. [207] are also applied for the broadened half-width coefficients in this region. For the transitions whose J values are greater than 65 and 75, their air- and self-broadened half-widths coefficients at 296 K are set to 0.12 and 0.0817 cm<sup>-1</sup>/atm, respectively [208]. Air-induced pressure shift coefficients for  $2v_3$  band of OCS reported by Domenech et al. [209] are inserted for the first time. In a separate parameter file, the air-broadened OCS



**Fig. 7.** Overview of the H<sub>2</sub>CO line parameters in the 5.7 μm and 3.6 μm spectral regions. Lower and upper traces describe the status in GEISA-03 [8,9] and in GEISA-09, respectively.

half-width coefficients are replaced by CO<sub>2</sub>-broadened half-width coefficients, using the measurements of Bouanich et al. [210] in the  $v_1$  band of OCS. This second database is intended to support remote sensing of Venus at 2.4  $\mu$ m and is archived in a GEISA-09 complementary files line list.

### 2.2.14. H<sub>2</sub>CO (molecule 21)

Formaldehyde ( $H_2CO$ ) in the atmosphere can be retrieved in the 5.7 µm region by MIPAS aboard the ENVISAT satellite [211] and by the ACE-FTS instrument on board the Canadian satellite SCISAT-1 at 3.6 µm [212]. For this reason the major update for  $H_2CO$  in the infrared region which consists of the complete replacement of the line positions and line intensity parameters near 3.6 µm and the addition of a line list in the 5.7 µm region [213].

The line positions were generated using the models and the parameters described in details in Refs. [214,215] for the 5.7 and 3.6  $\mu$ m, respectively. The 5.7  $\mu$ m corresponds to the  $v_2$  band together with three dark bands. In the 3.6  $\mu$ m region the lines belong to the  $v_1$  and  $v_5$  bands together with nine dark bands. In addition, a consistent set of line intensity parameters was generated for both the 5.7 and 3.6  $\mu$ m spectral regions [213] from analyzing high-resolution Fourier transform spectra recorded in the 1600–3200 cm<sup>-1</sup> spectral range.

The calculated band intensities derived for the 5.7 and  $3.6 \,\mu\text{m}$  bands are in excellent agreement with the values achieved recently by medium resolution band intensity measurements.

Compared to the GEISA-03 database which contains only 1161 lines near  $3.6 \,\mu$ m, the quality of the line parameters in GEISA-09 is significantly improved in terms of both the positions and intensities. Details giving the description of the new database which involves 3713 and 31,796 transitions at 5.7 and 3.6  $\mu$ m, respectively, are given in Table 9 of Ref. [213]. A subsequent and complementary study dealing with measurements and calculations of formaldehyde pressure induced self- and  $N_2$ -broadened half-width coefficients is in progress [216].

Fig. 7 illustrates the extended  $H_2CO$  line parameter information included in GEISA-09. Comparative absorptions as synthetic spectra (intensity, in similar arbitrary unit, along the Y-axis, versus wavenumber along the X-axis), are displayed in the 5.7 µm and 3.6 µm spectral regions, corresponding, respectively, to GEISA-03 (red curve) and GEISA-09 (blue curve) archives.

#### 2.2.15. $C_2H_6$ (molecule 22)

The GEISA-03 line list for the 12 µm region of ethane contained data for the  $v_9$  fundamental band of  ${}^{12}C_2H_6$ , from a 1992 analysis by Daunt et al. [217], and the  $v_{12}$ fundamental band of <sup>13</sup>CH<sub>3</sub><sup>12</sup>CH<sub>3</sub>, from a high-resolution work by Weber et al. [218,219]. In the updated 2009 edition, only the line list for the  $v_{12}$  band of  ${}^{13}CH_3^{12}CH_3$ has been kept; the data for the  $v_9$  band of  ${}^{12}C_2H_6$  has been replaced with a new list which includes a total of 21,607 lines belonging to the  $v_9$ ,  $3v_4$ ,  $v_9+v_4-v_4$ , and  $v_9+2v_4-2v_4$  bands ( $v_4$  is the torsional mode near 289.3 cm<sup>-1</sup>). It was generated by Vander Auwera et al. [220] using a spectrum of the  $v_9$  band recorded at the PNNL [221], results from a global analysis of data involving the four lowest vibrational states of ethane [222] and measurements of pressure-broadening parameters [223,224]. Details can be found in [220]. As a result, the sum of the line intensities and wavenumber coverage in the 12  $\mu$ m region are increased from 5.881  $\times$  10<sup>-19</sup> to  $1.011\times 10^{-18}~cm^{-1}/(molecule~cm^{-2})$  at 296 K (natural abundance) and from 725.6–918.7  $cm^{-1}$  to 706.6– 961.2 cm<sup>-1</sup>, respectively. As shown by Nixon et al. [225] and Coustenis et al. [226] based on Cassini CIRS (http://cirs.gsfc.nasa.gov/) data, the new list for the 12  $\mu$ m spectral region of <sup>12</sup>C<sub>2</sub>H<sub>6</sub> constitutes a significant improvement over the previously available data, leading to the first measurement of  ${}^{12}C/{}^{13}C$  isotopic ratio of C<sub>2</sub>H<sub>6</sub> in the atmosphere of Titan. It can be mentioned that C<sub>2</sub>H<sub>6</sub> acts as the main catalyst in photosensitized

dissociation in Titan's stratosphere, as shown by Wilson and Atreya [227].

Note that the quantum number notation for representing rotation-torsion states has evolved since the GEISA-03 edition. In [220], the levels are identified by *J*, the quantum number for the total angular momentum of the molecule, *K*, the quantum number for its component along the symmetry axis,  $\ell$ , the quantum number associated with the vibrational angular momentum of the degenerate mode v<sub>9</sub>, and  $\sigma$ =0-3 which labels the torsional sublevels. In the new line list archived in GEISA-09, the latter is replaced by the symmetry species A<sub>1s</sub>, A<sub>2s</sub>, A<sub>3s</sub>, A<sub>4s</sub>, E<sub>1s</sub>, E<sub>2s</sub>, E<sub>3s</sub>, E<sub>4s</sub> and G<sub>s</sub> in the G<sub>36</sub><sup>+</sup> extended permutation-inversion group. Because all the allowed species are s-species, the letter 's' is omitted: for instance,  $E_{1s}$  symmetry is given as 'E1' and  $A_{1s}+A_{2s}$  is given as 'A12'. The symmetry of the vibration–rotation–torsion levels of  ${}^{12}C_2H_6$  corresponding to the excitation of  $v_9$  and  $v_4$  is given in Table 7. This new notation is common to GEISA-09 and HITRAN-08 [13]. In GEISA-09, the former notation has been kept for the not updated  $v_{12}$  fundamental band of  ${}^{13}CH_3^{12}CH_3$ .

Vander Auwera et al. [220] determined absolute line intensity information by matching to low-resolution cross-sections. They indicated that the best match between high-resolution spectra of the  $v_9$  band of pure ethane and spectra calculated at the same experimental conditions using the generated line list could be obtained

Table 7

Symmetry in the  $G_{36}^+$  extended permutation-inversion group of vibration-rotation-torsional levels of  ${}^{12}C_2H_6$  involving the excitation of the  $v_9$  bending and  $v_4$  torsional modes of vibration.

(a) $v_9 = even$ ,	v <sub>4</sub> =even					
K	J	$\sigma =$	0	$\sigma = 1$	$\sigma = 2$	$\sigma = 3$
0 $6n \pm 1$ $6n \pm 2$ 6n+3 $6n \neq 0$	Even Odd	E <sub>1s</sub>	(10)	$G_{s}$ (16) $E_{3s}+E_{4s}$ (8)	$E_{3s} (2)$ $E_{4s} (6)$ $G_s (16)$ $E_{3s}+E_{4s} (8)$	$E_{1s}(4)$ $A_{1s}+A_{2s}(16)$
(b) $v_9 = even$ ,	$v_4 = odd$	13			-32 + -43 (-7	
K	J	$\sigma =$	:0	$\sigma = 1$	<i>σ</i> =2	σ=3
0 $6n \pm 1$ $6n \pm 2$ 6n + 3 $6n \neq 0$	Even Odd	A <sub>4s</sub> E <sub>2s</sub>	(6) (10) (4) $+A_{4s}$ (16)	$G_{s}$ (16) $E_{3s}+E_{4s}$ (8)	$E_{3s} (2)$ $E_{4s} (6)$ $G_s (16)$ $E_{3s}+E_{4s} (8)$	$\begin{array}{l} E_{2s}\left(4\right)\\ A_{3s}\!+\!A_{4s}\left(16\right)\end{array}$
(c) $v_9 = odd$ , v	-					
G	Κ	J	$\sigma = 0$	$\sigma = 1$	$\sigma = 2$	$\sigma = 3$
0 $6n \pm 1$ $6n \pm 2$ 6n+3 $6n \neq 0$	≥ 0 < 0	Even Odd Even Odd	$\begin{array}{l} A_{3s}\left(6\right)\\ A_{4s}\left(10\right)\\ A_{4s}\left(10\right)\\ A_{3s}\left(6\right)\\ E_{2s}\left(4\right)\\ A_{3s}+A_{4s}\left(16\right)\end{array}$	$G_{s}(16)$ $E_{3s}+E_{4s}(8)$	$\begin{array}{c} E_{3s}\left(2\right)\\ E_{4s}\left(6\right)\\ E_{4s}\left(6\right)\\ E_{3s}\left(2\right)\\ G_{s}\left(16\right)\\ E_{3s}+E_{4s}\left(8\right)\end{array}$	E <sub>2s</sub> (4) A <sub>3s</sub> +A <sub>4s</sub> (16)
(d) $v_9 = odd$ , v	v <sub>4</sub> =odd					
G	Κ	J	$\sigma = 0$	$\sigma = 1$	$\sigma = 2$	$\sigma = 3$
0	≥ 0 < 0	Even Odd Even Odd	$\begin{array}{c} A_{1s} \left( 6 \right) \\ A_{2s} \left( 10 \right) \\ A_{2s} \left( 10 \right) \\ A_{1s} \left( 6 \right) \end{array}$		$\begin{array}{c} E_{3s}\left(2\right)\\ E_{4s}\left(6\right)\\ E_{4s}\left(6\right)\\ E_{3s}\left(2\right)\end{array}$	
$6n \pm 1$ $6n \pm 2$ 6n+3 $6n \neq 0$		Jua	$E_{1s}(4)$ $A_{1s}+A_{2s}(16)$	$G_{s}$ (16) $E_{3s}+E_{4s}$ (8)	$G_{s}(16)$ $E_{3s}+E_{4s}(8)$	E <sub>1s</sub> (4) A <sub>1s</sub> +A <sub>2s</sub> (16)

*J* and *K* are, respectively, the quantum numbers associated to the total angular momentum of the molecule and its projection along the molecule top 3-fold symmetry axis,  $\sigma = 0-3$  is the torsional index and  $G = K - \ell \ge 0$  with  $\ell = \pm 1$  the vibrational angular momentum quantum number associated to v<sub>3</sub>.  $n \ge 0$ . The nuclear spin statistical weights are given in parentheses [218]. Empty cells correspond to non-existing levels.

provided the line intensities were reduced by about 9%. Very recently, Devi et al. [229] carried out extensive measurements of spectral line parameters in the  $v_9$  band of  ${}^{12}C_2H_6$ . They showed that their high-resolution line intensities were 10–15% lower than those in the line list of Ref. [220]. These observations seem to indicate that the line intensities in GEISA-09, and also in HITRAN-08 [13], are probably 10–15% too high. Note that the Devi et al [229] work also provides improved characterization of the temperature dependence of the N<sub>2</sub>- and self-broadening parameters, which will be applied in future editions of databases.

In the 3.3  $\mu$ m region, the  $v_7$  fundamental band of <sup>12</sup>C<sub>2</sub>H<sub>6</sub> exhibits a number of strong unresolved Qbranches (<sup>p</sup>Q<sub>4</sub> to <sup>r</sup>Q<sub>4</sub>), observed between 2973 and 3001 cm<sup>-1</sup>. GEISA contained a list of 421 lines belonging to the <sup>p</sup>Q<sub>3</sub> branch observed near 2976 cm<sup>-1</sup>, generated by Pine and Rinsland [230]. To complement this rather limited information, the line positions and intensities determined for the other strong Q-branches by Goldman et al. [231,232] have been added to this edition, even though the data are now quite dated and only allow a rather approximate modeling of the observed structure of the branches. The other line parameters have been set to the same values as for the  $v_9$  band. The quantum number labeling of all the levels and the symmetry of torsionally split levels are also defined as for the  $v_9$  band. The symmetry of levels involving unresolved torsional components is expressed using the species of the D<sub>3d</sub> group, i.e.,  $A_{1g}$  (8),  $A_{1u}$  (8),  $A_{2g}$  (16),  $A_{2u}$  (16),  $E_g$  (20) and  $E_u$  (20) (the numbers between parentheses are the nuclear spin statistical weights [228]). Recent studies [233,234-236] are becoming available for infrared ethane. For the important atmospheric region at 3.3 µm, Harrison et al. [233] measured absorption cross-sections for both pure ethane and mixtures with synthetic dry air at a number of temperatures and pressures appropriate for atmospheric conditions. These data were later converted to line-byline parameters by Toon [234] with lower state energies estimated from the recorded spectra [233]. Two theoretical analyses interpreted high resolution spectra and modeled the observed positions. Di Lauro et al. [235] predicted positions and relative intensities of seven bands at 7 µm while Lattanzi et al. [236] extended assignments in four bands at 3.3  $\mu$ m and modeled those line positions. Both studies are reporting new databases applicable to atmospheric remote sensing with the caveat that the intensities must be studied further and that unassigned observed features are not yet included.

## 2.2.16. CH<sub>3</sub>D (molecule 23)

This GEISA independent molecule (see the section Introduction and Table 2) involves methane isotopologues  ${}^{12}CH_3D$  and  ${}^{13}CH_3D$ . For the GEISA-09 modifications, nine new infrared bands were added at three different wavelengths (8, 2.9 and 1.56 µm). In addition, a far-IR prediction (version 1) from the CDMS database [17], based on frequencies reported by Lattanzi et al. [237], was included. The  ${}^{13}CH_3D$  species was added to the database for the first time because the isotopologue  ${}^{13}CH_3D$  was recently detected in Titan's stratosphere [238], using

Cassini/CIRS infrared spectrum near 8.7  $\mu$ m. Fitting simultaneously the  $\nu_6$  band of both  ${}^{13}$ CH<sub>3</sub>D and  ${}^{12}$ CH<sub>3</sub>D and the  $\nu_4$  band of CH<sub>4</sub>, this detection allowed a precise determination of the D/H ratio in methane and yielded a  ${}^{12}$ C/ ${}^{13}$ C ratio in  ${}^{13}$ CH<sub>3</sub>D consistent with that measured in normal methane.

A prediction of the <sup>13</sup>CH<sub>3</sub>D triad ( $v_6$ ,  $v_3$  and  $v_5$ ) between 952 and 1694 cm<sup>-1</sup> was based on the line positions and energy levels analysis by Ulenikov et al. [239]. The intensities were calculated using the transition dipole moment parameters of the <sup>12</sup>CH<sub>3</sub>D from Brown et al. [240]. The calculations were limited to J=K=18 as they are the maximum quantum numbers covered by the experimental rovibrational term values published in Ref. [239].

Titan and Saturn observations [241,242] also revealed the need for additional parameters at 2.9  $\mu$ m. Six new <sup>12</sup>CH<sub>3</sub>D vibrational bands (v<sub>2</sub>+v<sub>3</sub>, v<sub>2</sub>+v<sub>5</sub>, v<sub>2</sub>+v<sub>6</sub>, v<sub>3</sub>+2v<sub>6</sub> and 3v<sub>6</sub>) were included for the first time from the analysis of positions and line intensities of by Nikitin et al. [243]. Finally, the Boussin et al. [244] empirical line list in the 3v<sub>2</sub> region at 1.56  $\mu$ m was included. The self- and airbroadened widths were generally applied using empirical formula obtained from <sup>12</sup>CH<sub>3</sub>D triad measurements [245]; however, self- and air-broadened widths and shifts observed by Boussin et al. [244] were used for 3v<sub>2</sub>. For temperature dependence of widths, CH<sub>4</sub> values averaged by *J* [119] were used as a rough estimate. Additional laboratory and theoretical studies are needed to complete and improve the new mid- and near-IR parameters.

#### 2.2.17. C<sub>2</sub>H<sub>2</sub> (molecule 24)

Acetylene has been identified in some of the giant planets and Titan since the mid-1940s, and recently has been quantified by the Galileo (http://nssdc.gsfc.nasa.gov/ planetary/galileo.html) and Cassini-Huygens missions. Up to now, the data available in GEISA for acetylene isotopologues, namely <sup>12</sup>C<sub>2</sub>H<sub>2</sub> and <sup>12</sup>C<sup>13</sup>CH<sub>2</sub>, were limited to the lower energy region of the spectrum, up to  $3 \,\mu m$ (note that the C<sub>2</sub>HD molecule has a different code, i.e., "48", than the one of  $C_2H_2$ , i.e., "24"; see Table 1). This new edition sees the extension of data into the near infrared range for these two isotopologues, with the inclusion of a list of line parameters generated by El Hachtouki and Vander Auwera [246] and Jacquemart et al. [247,248]. In the 1.5 µm region, corresponding to the simultaneous excitation of the symmetric and antisymmetric C–H stretching modes  $v_1$  and  $v_3$ , respectively, the line list was created following the high-resolution intensity study [246]. The identification of the lines, their positions and lower state energies are from Kou et al. [249], and the line intensities are calculated using the parameters of Table 7 of [246]. Note that there is a mistake in [246]: the isotopic abundance used for <sup>12</sup>C<sup>13</sup>CH<sub>2</sub> is a factor 2 too small; it should read 0.02176 instead of 0.01088. As a result, the vibrational transition dipole moments of <sup>12</sup>C<sup>13</sup>CH<sub>2</sub> listed in Tables 6 and 7, and in Fig. 7 of [246] are a factor 2 too large. The list included in GEISA-09 contains the corrected values. Also, a large update has been performed for the <sup>12</sup>C<sub>2</sub>H<sub>2</sub> isotopologue and led to new data in nine spectral regions, namely, in the regions around 3.8, 3, 2.5, 2.2, 1.9, 1.7, 1.5, 1.4, 1.3, 1.2, and 1  $\mu$ m. The new line lists are described in details in Refs. [247,248]. Corrections of the 2.5 and 3.8  $\mu$ m spectral regions of  ${}^{12}C_2H_2$  have also been performed [250,251] and are described in Ref. [247]. Table 8 summarizes the number of new bands (cold/hot in column 2) and transitions (column 3) of the spectral regions added in GEISA-09, together with the intensity ranges (in cm<sup>-1</sup>/ (molecule cm<sup>-2</sup>)) and spectral domains involved (in  $\mu$ m in the first column and in cm<sup>-1</sup> in the fourth column). It has to be noted that no transitions are unidentified so that the lower state energies are calculated using the spectroscopic constants from Kabbadj et al. [252].

Fig. 8 illustrates the noticeable improvements brought to GEISA-09. The information is displayed as the following: spectral regions corresponding to existing (or in project)  $C_2H_2$  spectroscopic data are identified by vertical rectangles perpendicular to 3 horizontal axes, with spectral region mean value given at the center of each figure; on the 3 horizontal axes are given, downwards: the spectral range extend in two units, µm (top axis) and

#### Table 8

Summary of the bands and transitions added for the  ${}^{12}C_2H_2$  molecule in the GEISA-09 database. Reference temperature is 296 K.

Spectral	Number of	Number of	Spectral	Intensity range
region	bands <sup>a</sup>	transitions <sup>a</sup>	domain	(cm <sup>-1</sup> /
(µm)	(cold/hot)	(cold/hot)	(cm <sup>-1</sup> )	(molecule cm <sup>-2</sup> ))
3.8 <sup>b</sup> 3 <sup>c</sup> 2.5 <sup>b</sup> 2.2 <sup>c</sup> 1.9 <sup>c</sup> 1.7 <sup>c</sup> 1.5 <sup>e</sup>	2/3 0/18 4/5 4/4 7/0 2/4 2/2	90/331 77 <sup>e</sup> /1971 450/720 254/392 539/0 175/350 129/224 2021/142	2499-2769 3139-3398 3762-4226 4421-4798 5032-5567 5692-6032 6448-6685	$\begin{array}{c} 10^{-21} - 10^{-25} \\ 10^{-20} - 10^{-26} \\ 10^{-21} - 10^{-27} \\ 10^{-22} - 10^{-25} \\ 10^{-24} - 10^{-26} \\ 10^{-23} - 10^{-26} \\ 10^{-20} - 10^{-24} \\ 10^{-23} - 10^{-28} \end{array}$
$1.5^{c}$	4/16	200/1443	6277-6865	$ \begin{array}{r} 10^{-23} - 10^{-28} \\ 10^{-22} - 10^{-25} \\ 10^{-25} - 10^{-24} \\ 10^{-26} - 10^{-23} \\ 10^{-25} - 10^{-22} \end{array} $
$1.4^{c}$	4/0	347/0	7042-7476	
$1.3^{d}$	1/0	51/0	7671-7791	
$1.2^{d}$	2/0	132/0	8407-8612	
$1.0^{d}$	3/1	193/108	9516-9890	

<sup>a</sup> A <sup>12</sup>C<sup>13</sup>CH<sub>2</sub> data are not mentioned.

<sup>b</sup> New data from Refs. [246,249].

<sup>c</sup> New data from Refs. [247,248].

<sup>d</sup> New data from Ref. [248].

<sup>e</sup> New data from Ref. [246].

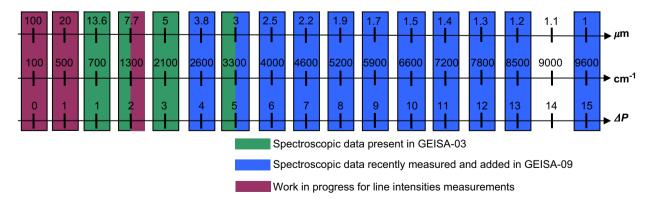
cm<sup>-1</sup> (middle axis); the  $\Delta P$  polyad series values (bottom third axis; see [247] and Refs. therein for definition) associated with each of the spectral regions mean values identified along the above two axes.

These data improve and summarize the current experimental spectroscopic knowledge on acetylene. Several of the spectral regions involved are of atmospheric, planetary, astrophysical, or meteorological interest, e.g., at 3, 2.2, 1.5, and 1  $\mu$ m. The study of the region at 7.7  $\mu$ m, very useful for several applications, is in progress [253]. In this spectral region, intensity measurements were undertaken because the knowledge of C<sub>2</sub>H<sub>2</sub> line intensities is important for several applications, especially for astrophysical interest. For example, the acetylene molecule has been observed in the circumstellar envelopes of carbonrich stars. Using IRS on board the SST telescope, Matsuura et al. [254] detected acetylene bands at 7 and 14 um in carbon-rich asymptotic giant branch stars in the Large Magellanic Cloud. Around 7 µm, GEISA-09 only contains line positions and intensities that Vander Auwera calculated from his absolute intensity measurements in the  $(v_4+v_5)^0_+$  band [255], for the rotational quantum number J up to 35. But intensities measured in [255] for some lines of the  $(v_4+v_5)^2$  band are not reported in the databases. The temperature of interest for applications being around 500 K [254], the knowledge of intensities in the remaining hot bands is also important. In Ref. [254], Matsuura et al. could not reproduce the shapes that they observed in their IRS-SST spectra around 7 µm because of the lack of data available in the databases.

The GEISA-09  $C_2H_2$  line list involves 11,340 entries and 118 vibrational transitions against 3115 entries and 29 vibrational transitions in GEISA-03.

#### 2.2.18. C<sub>2</sub>H<sub>4</sub> (molecule 25)

Ethylene has been identified in the atmospheres of some of the giant planets and Titan. The spectroscopic information available for this molecule in GEISA-03 dates back to the 1997 edition [7]. It includes the 10 and 3.3  $\mu$ m spectral regions of the main isotopologue and the 3.3  $\mu$ m region of <sup>12</sup>C<sup>13</sup>CH<sub>4</sub> [7]. The 10  $\mu$ m region of <sup>12</sup>C<sub>2</sub>H<sub>4</sub> involves the v<sub>10</sub>, v<sub>7</sub>, v<sub>4</sub> and v<sub>12</sub> bands observed near 826, 949, 1027 and 1442 cm<sup>-1</sup>, respectively. The first



**Fig. 8.** Improvement of data available in GEISA-09 for the  ${}^{12}C_2H_2$  isotopologue of acetylene. *P* is the pseudo-quantum number defined for acetylene as:  $P=5v_1+3v_2+5v_3+v_4+v_5$ , where  $v_1$ ,  $v_2$ ,  $v_3$ ,  $v_4$ , and  $v_5$  are the quantum numbers associated with the normal modes of vibration of the molecule in the ground electronic state. Note that the thickness of each box does not represent the frequency span.

three bands are already in GEISA [8], while the  $v_{12}$  band is absent. Recently, Rotger et al. [256] carried out an experimental and theoretical study of line positions and intensities in the  $v_{12}$  band of  ${}^{12}C_2H_4$ . 1240 line positions and 871 intensities, measured in a set of Fourier transform spectra recorded at ULB, were fitted using the tensorial formalism developed at ICB with global root mean square deviations of  $1.6 \times 10^{-4}$  cm<sup>-1</sup> and 1.88%, respectively [256]. Using the refined model thus obtained, the positions, intensities and lower state energies of 5400 lines in the  $v_{12}$  band were calculated. These lines correspond to transitions from levels with  $I \le 40$ , and lower and upper state rotational energies up to 1380 and 1510 cm<sup>-1</sup>, respectively. This initial list of line parameters was complemented with the self- and air-broadening parameters, and the temperature dependence of the airbroadening parameter based on literature [257-260] (see [256] for details). This  $v_{12}$  band line list (5400 entries), whose content is summarized in Table 9, has been added to the present 2009 edition of GEISA.

#### 2.2.19. HCN (molecule 27)

In planetary atmospheres, HCN is an important nitrile, of astrobiological interest in many cases and whose abundance and its variations has been thoroughly studied (see for example Coustenis et al. [226] in the case of Titan). A major improvement has been accomplished on the entire list of lines of HCN. The entire GEISA-03 content (2550 entries in the spectral range 2.870484–18,407.972700 cm<sup>-1</sup>) has been replaced, in GEISA-09, with new data originating from two different sources, i.e., from Harris [261] and from Maki [262]. The new line list comprises 82,042 entries in the spectral range 0.00636–17,581.009367 cm<sup>-1</sup>.

Harris's [261] data are related to the main isotopologue  $H^{12}C^{14}N$ . Among a total of 108,402 entries 28,624 have been implemented in a supplemental line list because they did not have upper vibrational state identifications. The HCN archive has been obtained from a combination of experimental and theoretical data. The theoretical data were taken exclusively from the line list of Harris et al. [263]. Experimental data were used in preference to the *ab initio* data where they were available. The line list covers the spectral region 0.011561– 17,943 cm<sup>-1</sup>. Hot bands, with a lower vibrational state of 3 quanta of bend, are given for many of the lower energy transitions. Data are included for transitions up to the (5001) stretching combination bands. The HCN line list was constructed in the following stages:

- Construction of a list of laboratory determined energy *levels*: The available laboratory line measurements [264–270]; for line positions were gathered. From these line position data, a list of HCN energy levels was determined. This was done by using a technique that deviates only slightly from that of Harris et al. [271]; the rotational constants are used to compute energy levels up to an angular momentum quantum number of 60.
- Construction of a list of laboratory determined line positions: Using the laboratory determined energy levels it is straight forward to compute a list of line frequencies for dipole allowed transitions. The well known selection rules for dipole transitions require a change in symmetry and allow a change in angular momentum of 0,  $\pm$  1. When applied to HCN the allowed transitions form two groups. The first has a change in parity of the vibrational angular momentum with no change in total angular momentum. The second group has no change in the parity of the vibrational angular momentum, but a change of plus or minus one in total angular momentum. For all the dipole allowed transitions between laboratory determined energy levels, line positions were computed by subtracting the lower state energy from upper state energy.
- Construction of a list of laboratory determined line intensities: A list of line intensities were computed from laboratory data [272,273,268–270]. These data are usually given in the form of band strengths or dipole moments that are often supplemented with Herman-Wallis factors. From this data, the line intensities of individual lines were computed by using the relevant Höln–London factor and the equation given by Maki et al. [272].
- Construction of laboratory determined line list: Experimentally measured line intensities were inserted into the list of laboratory determined energy levels. In this way, a HCN line list is created that is based upon laboratory measurements.
- Augmentation of the laboratory determined line list with ab initio line intensities: Many of the intensities for the

Table 9

Summary of the content of the line list for the  $v_{12}$  band of  ${}^{12}C_2H_4$ . The intensities are given at 296 K for an isotopologue abundance of 0.9773.

	Value	Value	
F-min (cm <sup>-1</sup> )	1380.0239	Int-min ( $cm^{-1}/(molecule cm^{-2})$ )	$2.764  imes 10^{-37}$
F-max (cm <sup>-1</sup> )	1509.9819	Int-max $(cm^{-1}/(molecule cm^{-2}))$	$6.948 \times 10^{-21}$
J''max	40	Int-sum $(cm^{-1}/(molecule cm^{-2}))$	$1.549 \times 10^{-18}$
K <sup>"</sup> <sub>a max</sub>	20	HWHM <sub>air</sub> min (cm <sup>-1</sup> atm <sup>-1</sup> )	0.0813
# lines	5400	HWHM <sub>air</sub> max $(cm^{-1} atm^{-1})$	0.0989
n	0.82	$HWHM_{self}$ (cm <sup>-1</sup> atm <sup>-1</sup> )	0.125

'min' and 'max' represent the minimum and maximum values of the corresponding quantity, respectively; '*F* is the wavenumber; J'' and  $K_a''$  are rotational quantum numbers; '# lines' is the total number of lines; '*n*' is the temperature dependence exponent of the air-broadening parameter; 'Int' is the line intensity; 'Int-sum' is the sum of all the line intensities; 'HWHM<sub>air</sub>' and 'HWHM<sub>self</sub>' are, respectively, the air- and self-broadening parameters.

dipole allowed bands have not been measured. The resulting gap in the laboratory determined line list may only be filled by *ab initio* data.

Many of the transitions in the *ab initio* line list of Harris et al. [263] have been assigned an approximate vibrational quantum number. It has been therefore possible to insert the line strengths from the Harris et al. [263] line list into the GEISA-09 line lists, creating a more complete list of lines.

• Augmentation with ab initio data and truncation: The upper and lower energy levels for many strong room temperature lines have not been determined. In order to account for these strong lines, the GEISA-09 line lists for HCN were augmented with purely *ab initio* line position and intensity data from Harris et al. [263]. Finally, to reduce the size of the final line list, a minimum line intensity of 10<sup>-30</sup> cm<sup>-1</sup>/(molecule cm<sup>-2</sup>) was chosen. Lines with intensities below this level were removed from the final line lists.

Maki's data [262] include the isotopologues: H<sup>12</sup>C<sup>14</sup>N, H<sup>12</sup>C<sup>15</sup>N, H<sup>13</sup>C<sup>14</sup>N and a new isotopologue species for GEISA-09, i.e., D<sup>12</sup>C<sup>14</sup>N and comprise 5 files in the spectral ranges: 0.014975–175.672283 cm<sup>-1</sup> (408 entries); 533.819433-895.585448 cm<sup>-1</sup> (981 entries); 1241.392310-1591.111005 cm<sup>-1</sup> (709 entries); 2428.365681-3609.137515 cm<sup>-1</sup> (1710 entries) and 452.016228–2725.191923 cm<sup>-1</sup>) (452 entries) and 452.016228-2725.191923 cm<sup>-1</sup> (452 entries) for DCN. Note that DCN is considered as an isotopologue of HCN and not as an independent molecule, because it has the same symmetry as HCN (see Sections 1 and 2).

The origin of the spectroscopic parameters is as the following:

- The values of line positions and their uncertainties were based on a large body of data that included many very accurate microwave and mm-wave measurements [264–281] and also several infrared measurements [265,266,282,283]. For each isotopologue all the wavenumber data were included in a least-squares analysis that made it possible to calculate all the transition wavenumbers, and their uncertainties, given in the GEISA-09 line list. These uncertainties are twice the standard deviation.
- The intensities of the far-infrared transitions are • assumed to be well represented by the dipole moment measured for each vibrational state. The best dipole moment measurements are those given by Tomasevich [284] and by DeLeon and Muenter [285] and Ebenstein and Muenter [286]. The dipole moment is very large and any Coriolis-type mixing of intensity with other vibrational states would probably have a very small effect because the vibrational transition moments are small compared to the dipole moment. For that reason it was assumed that the intensities of the far-infrared transitions could be calculated by using the same dipole moment for all values of *J*. The intensities for the  $v_2$ transitions for H<sup>12</sup>C<sup>14</sup>N, H<sup>12</sup>C<sup>15</sup>N, and H<sup>13</sup>C<sup>14</sup>N were taken from the work of Devi et al. [287]. The same intensity constants were used for the hot bands that

accompany  $v_2$ . For transitions that involve  $v_2 > 1$ , the effects of 1-type resonance were included as described by Maki et al. [272]. The intensities of the  $2v_2$  band and hot band are based on the measurements of Devi et al. [288] and Maki et al. [272,289]. For these transitions the effects of 1-type resonance were taken into account [272,289]. The intensities of the  $v_1$  transition for HCN, H<sup>13</sup>CN, and HC<sup>15</sup>N were taken from the work of Devi et al. [290]. The hot bands were assumed to require the same intensity constants, as was verified by the agreement with the measurements of Devi et al. [290]. For the  $v_1 - v_2$  transitions near 2600 cm<sup>-1</sup> the intensity constants were taken from the measurements of Maki et al. [272]. The intensities of the  $v_2 + v_3$  band near 2800 cm<sup>-1</sup> came from the work of Maki et al. [289] and the intensities of the  $2v_2 + v_3$  transitions near  $3520 \text{ cm}^{-1}$  were taken from that same work. In all cases the GEISA-09 archived line intensities of HCN could be used to calculate the intensities for other conditions such as temperature or isotopic composition.

• Except for the regions  $2428-2720 \text{ cm}^{-1}$  and 3089-3450 cm<sup>-1</sup>, the air-broadened half-width and airinduced pressure shift coefficients of HCN, and their temperature dependences, were based on the data given by Devi et al. [287] for the  $v_2$  band of HCN. Except for some transitions that did not include states with  $v_1 > 0$ , the line parameters for the regions 2428–  $2720 \text{ cm}^{-1}$  and  $3089-3450 \text{ cm}^{-1}$  were based on the measurements by Rinsland et al. [291]. Their earlier work on the  $2v_2$  and  $v_1$  bands of HCN indicated that, aside from the wavenumbers of the transitions and the intensities, there is very little vibrational dependence of the Lorentz pressure-broadened half-width coefficients for HCN. The only parameters that seemed to be dependent on the vibrational state were the airinduced shift coefficients and their temperature dependences. In GEISA-09 those parameters were assumed to have the same values given by Rinsland et al. [291] for all transitions with  $v_1 = 1$  in the upper state. The air-induced pressure shift parameters for all transitions with  $v_1 = 0$  were assumed to be the same as those measured by Devi et al. [287] for  $v_2$ . Devi et al. [287,288] believed that the parameters were the same, within experimental error, for both  $v_2$  and for  $2v_2$  and probably would be the same for the ground state as well. Since GEISA-09 includes transitions involving much higher rotational states, to J=60, than those measured by Malathy Devi et al. [287,288,290] and Rinsland et al. [291] (J < 34), the trends in the various line shape parameters were extrapolated beyond reasonable bounds and the uncertainties in the parameters were increased to attempt to encompass reasonable values. All of the broadening and shift parameters for H<sup>13</sup>C<sup>14</sup>N and H<sup>12</sup>C<sup>15</sup>N were assumed to be the same as for the most common isotopologue, H<sup>12</sup>C<sup>14</sup>N. Within experimental error, this assumption was based on a number of measurements made on the  $2v_2$  band of H<sup>12</sup>C<sup>14</sup>N [288].

The air induced pressure shifts for DCN were given values that were 70% of those for HCN in agreement with

a private communication from Smith [292]. That estimate was not based on any real measurements of DCN but rather was based on the trend shown by HCl and DCl. The other parameters for DCN were the same as for HCN, but again that was not based on any measurements.

In a supplementary file (Table S1), the uncertainties in the various parameters are summarized and given along with many other parameters used for processing Maki's new HCN data.

The HCN GEISA-09 line list has been processed as follows: starting from Harris's line list [261], Maki's data [262] have been included, replacing those of Harris's with the same quantum number identifications.

The HCN GEISA-09 line list includes 82,042 entries (775 transitions) against 2550 entries (41 transitions) in GEISA-03.

## 2.2.20. C<sub>3</sub>H<sub>8</sub> (molecule 28)

The intensities of the v<sub>26</sub> band transitions were corrected in GEISA-03 line list which includes only the cold band. A PNNL spectrum at 298 K and 0.11 cm<sup>-1</sup> resolution vields an intensity of  $4.27 \times 10^{-19} \text{ cm}^{-1}/(\text{molecule cm}^{-2})$ for the whole band, including cold and hot bands [293], a value  $(4.33 \times 10^{-19} \text{ cm}^{-1}/(\text{molecule cm}^{-2}))$  that agrees with the earlier measurement of Giver et al. [294] at lower resolution. The vibrational partition function at 296 K is 2.71, so that the intensity of the fundamental cold band should be about  $4.27/2.71 = 1.58 \times 10^{-19} \text{ cm}^{-1}/2.71$ (molecule  $cm^{-2}$ ). In the GEISA-03 line list, an intensity of  $3.76 \times 10^{-19} \text{ cm}^{-1}$  (molecule cm<sup>-2</sup>) had been set for the cold band, based on some low-resolution spectra that include the hot bands, which is incorrect. The GEISA-03 intensities have thus been multiplied by a factor of 1.58/ 3.76=0.420 in the GEISA-09 edition (see Nixon et al. [295]). This scaling factor yields an intensity of  $1.58 \times 10^{-19}$  cm<sup>-1</sup>/  $(molecule cm^{-2})$  at 296 K for the fundamental band, as estimated from the integrated intensity of the whole region in the PNNL spectrum,  $4.27 \times 10^{-19}$  cm<sup>-1</sup>/(molecule cm<sup>-2</sup>) including hot bands, divided by the vibrational partition function at 296 K (2.71).

For all bands, a Lorentz half-width coefficient of  $0.12 \text{ cm}^{-1} \text{ atm}^{-1}$  at 296 K and a temperature dependence exponent of 0.50 was assumed for all transitions, following N<sub>2</sub>-broadening measurements by Nadler and Jennings [296] and Hillman et al. [297]. Propane has been identified in the atmospheres of some of the giant planets and of Titan. New propane data from Flaud et al. [298] including hot bands will be archived in the next version of the GEISA database. The total number of entries (8983) has not been altered since GEISA-03.

### 2.2.21. C<sub>2</sub>N<sub>2</sub> (molecule 29)

A mistake was found in the relative intensities of the hot sub-bands of the  $v_5$  band listed in GEISA-03 (and in previous versions). More precisely, the intensities of the  $(02)^2 \leftarrow (02)^1$ ,  $(03)^1 \leftarrow (02)^2$  and  $(03)^3 \leftarrow (02)^2$  sub-bands were two times too big; they have been corrected accordingly in GEISA-09. Following this correction, the total band intensity has been updated by multiplying all line intensities by 0.95, a factor that yields the best agreement with Grecu et al. [299] absolute intensity measurements in the  $(01)^1 \leftarrow (00)^0$  cold band, as listed in Table 3 (data for 8 mbar of N<sub>2</sub> pressure) of Ref. [299]. Note that this determination slightly disagrees with the older measurement of the integrated band intensity by Kim and King [300], which would yield intensities 15% larger. For the Lorentz broadening parameter (HWHM), we used the expression "0.12–0.00035m" at 296 K, derived from a fit of the data points in Fig. 5 of Grecu et al. [301]. We arbitrarily assumed a temperature dependence exponent of 0.75 for all transitions listed in GEISA-09. This updated line list will be used for planetary studies in the case of Titan. The total number of entries (2577) has not been altered since GEISA-03.

## 2.2.22. $C_4H_2$ (molecule 30)

The diacetylene line list (issued 1982, 1986 [3,4]) still included in GEISA-03 (1405 entries: 5 bands) has been replaced in GEISA-09 by a new line list (119.480 entries: 1509 bands) based on experimental and theoretical studies by Jolly et al. [302]. The lines included belong to the  $v_8$  and  $v_9$  bands in the range between 581–730 cm<sup>-1</sup> and 191–257 cm<sup>-1</sup>, respectively. The number of lines has been increased from 1405 to 119,480. Due to low energy vibrational modes, the vibrational partition function of  $C_4H_2$  is large ( $Q_v = 61$  at 300 K). This means that only 28 % of the molecules are in the ground state at room temperature. In the previous GEISA-03 line list, hot band transitions from three different excited levels were present in the  $v_9$  band complex but none in the strong  $v_8$ band complex. The new line list includes hot band transitions with lower vibrational levels up to about  $1300 \text{ cm}^{-1}$  for the v<sub>8</sub> band complex and up to about 900 cm<sup>-1</sup> for the weaker  $v_9$  band. This was necessary to allow for the inclusion of the contributions of all the hot band transitions with a non negligible intensity at room temperature. The minimum intensity of the lines is  $3 \times 10^{-24}$  cm<sup>-1</sup>/(molecule cm<sup>-2</sup>) at 296 K. It was also necessary to extend the quantum identification, in particular the vibrational quantum numbers of both upper and lower levels. All v values for the nine vibration modes of C<sub>4</sub>H<sub>2</sub> have been included in the assignment together with the four  $\ell$  values corresponding to all bending modes (v<sub>1</sub>,  $v_2, v_3, v_4, v_5, v_6, v_7, v_8, v_9, \ell_6, \ell_7, \ell_8, \ell_9$ ). The new line list is based on a global analysis study as described by Fayt et al. [303]. High resolution data from Arié and Johns [304] were fitted together with other experimental data in the infrared [305,306] and in the microwave domain [307]. Since no new intensity measurements were available, band intensity measurements by Koops et al. [308] were chosen to infer the absolute intensities of the lines.

The improvement of the data is very important in particular for the study of planetary atmospheres. Diacetylene was first detected in Titan's atmosphere by the IRIS [309] spectrograph on board the Voyager spacecraft [310] and is now under close scrutiny by the CIRS spectrometer on board Cassini [226,311,312]. Using the new line list Jolly et al. [302] were able to obtain a new fit of the diacetylene contribution in Titan's atmosphere recorded by Cassini–CIRS resulting in very precise abundance determination and the first detection of the <sup>13</sup>C isotopologues of diacetylene in Titan's atmosphere. Recently detection of the main isotopologue occurred in the atmospheres of both Uranus and Neptune using the Spitzer space telescope [313,314]. Outside the solar system, detection of diacetylene was achieved in the post-AGB object CRL2688 and in the proto-planetary nebulae CRL618 [315]. All detections so far where obtained thanks to the strong v<sub>8</sub> bending mode centered at 628 cm<sup>-1</sup>, but the weaker v<sub>9</sub> bending mode at 220 cm<sup>-1</sup> was also detected by IRIS and CIRS in Titan's atmosphere.

#### 2.2.23. HC<sub>3</sub>N (molecule 31)

A line list for cyanoacetylene has been included in GEISA since its first editions [3,4] thanks to Goldman's data [316]. It was already modified in 1990 following a new analysis by Arié et al. [317]. In GEISA-09 a completely new line list, based on experimental and theoretical studies by Jolly et al. [318], replaces the previous version. Lines included belong to the  $v_5$  and  $v_6$  band in the range between 463 and 760 cm<sup>-1</sup>. The number of entries has increased from 2027 (20 bands) in the 1990 version to 179,347 (3302 bands) in GEISA-09. This considerable increase was necessary to take into account all hot band transitions with a minimum intensity of  $10^{-24} \text{ cm}^{-1}$ /(molecule cm<sup>-2</sup>) at 296 K. Transitions with lower vibrational levels up to about  $1500 \text{ cm}^{-1}$  had to be included to take into account all the intensity of the bands. Only few lines belonging to hot band transitions where included [318] in the previous line list. To obtain this new line list, a global analysis was performed fitting simultaneously high resolution data from Arié et al. [317] together with all available experimental data including microwave and infrared measurements. As a result, positions and relative intensities of lines belonging to 123 excited sub-states could be obtained. As for  $C_4H_2$ , the assignment code needed to be modified to take into account levels with high vibrational quanta numbers (v,  $\ell$ ). HC<sub>3</sub>N possess four stretching and three bending modes. A complete vibrational assignment includes all seven v values and three  $\ell$  values ( $v_1$ ,  $v_2$ ,  $v_3$ ,  $v_4$ ,  $v_5$ ,  $v_6$ ,  $v_7$ ,  $\ell_5$ ,  $\ell_6$ ,  $\ell_7$ ). The absolute intensities of the lines have were derived from a new measurement of the integrated band intensities of  $v_5$  and  $v_6$  at 0.5 cm<sup>-1</sup> resolution as described in Jolly et al. [318].

Cyanoacetylene is a molecule of great interest for planetary atmospheres and in particular for Titan's atmosphere where it has been detected by IRIS [309] during the Voyager mission [310]. The presence of HC<sub>3</sub>N was confirmed by the ISO space telescope [311] and has been observed in details since 2004 by the CIRS spectrometer on board the Cassini spacecraft. The quality of the new observations by CIRS improves greatly in terms of spectral and spatial resolution on the previous observations. Recently, Jennings et al. [319] used the new line list proposed by Jolly et al. [318] to obtain a good fit of the  $HC_3N$  feature at 663 cm<sup>-1</sup> in Titan's spectrum. The contribution of hot bands where clearly observed as a large shoulder on the high energy side of the main feature. The quality of the fit enabled small contributions due to <sup>13</sup>C isotopologues of HC<sub>3</sub>N to be observed, for the first time in the solar system. The contribution of hot bands a cold environment such as Titan' atmosphere is not surprising given that the partition function equals to 1.69 at 200 K, which means that about 40 % of the molecules are still in an excited state.

## 2.2.24. N<sub>2</sub> (molecule 33)

The whole of the line parameters of  $N_2$  has been replaced by a new line list provided by Goldman [320]. Improvements to the line parameters mainly include intensities and half-widths. The new intensities are obtained by the use of two works: the work by Goldman et al. [321], where a semi-empirical Herman-Wallis formulation of the vibration-rotation effects on the intensities associated with a final scaling based on observed spectra, and the work by Li and Le Roy [322] based on ab initio methods. Values derived by both, Goldman et al. [321] and Li and Le Roy, methods are very similar. However, the *ab initio* matrix elements of Ref. [322] have been adopted for the GEISA-09 line list, because it can be expected that the Herman-Wallis formulation of Goldman et al. yields less accurate values with increasing *I*. Presently, the GEISA-09 N<sub>2</sub> line list is restricted to only the (1-0) N<sub>2</sub> band. It should be noted that Li and Le Roy method makes it possible to derive additional line parameters for other bands that may be of atmospheric importance. The absolute accuracy of the Li and Le Roy intensities is estimated to be about 1% by the authors; these new values are still being validated. As described in Ref. [321], the new half-widths are based on available experimental and theoretical studies. As stated in Ref. [321], further extensions are expected in the near future. The total number of entries (120) has not been altered since GEISA-03.

#### 2.2.25. CH<sub>3</sub>Cl (molecule 34)

The GEISA-03 line list for CH<sub>3</sub>Cl, which was based on Ref. [323], has been revised using data from Ref. [324]. In particular, previously unassigned vibrational transitions have been identified as belong to  $2v_3$  and the self-broadened half-widths have also been revised for both isotopologues. The total number of entries (18,344) has not been altered since GEISA-03.

#### 2.2.26. H<sub>2</sub>O<sub>2</sub> (molecule 35)

The H<sub>2</sub>O<sub>2</sub> (hydrogen peroxide) data previously archived in GEISA-03 (100,781 entries; 2 bands), for the  $v_6$  band in the 7.9- $\mu$ m region, have been completely replaced, leading to improved line positions and intensities in GEISA-09 (126,983 entries; 130 bands). Indeed, this new list is more complete as it includes several hot torsional-vibration sub-bands of the  $v_6$  band (up to the n=2 torsional quantum number), instead of only the two main torsional components of the  $v_6$  band (in the n=0,  $\tau = 1$  and n = 0,  $\tau = 3$  torsional quantum numbers). In addition the new line positions are more accurate since the vibration-torsion-rotation coupling the energy levels from the  $6^1$  state with those from the  $2^1$ ,  $3^1$  and ground vibrational states were accounted for. The line intensities are also more accurate as these parameters are based on new line intensity measurements and on a sophisticated theoretical treatment which account for the torsional effects. The sources of the new data are Perrin et al. [325] and Klee et al. [326].

## 2.2.27. H<sub>2</sub>S (molecule 36)

No new line list has been implemented in GEISA-09 for hydrogen sulfide, but updates occurred for air- and selfbroadened pressure half-widths. Where available, measured values have been adopted – those from Sumpf et al. [327], Kissel et al. [328,329] and Waschull et al. [330], for air-broadened half-widths and those from Refs. [327,330] and from Sumpf [331] for self-broadened half-widths. Otherwise, default values of 0.074 and 0.1580 have been assigned to air- and self-broadened half-widths, respectively. These values have been obtained as averages of the ones previous quoted in the above reference. The total number of entries (20,788) has not been altered since GEISA-03.

## 2.2.28. HCOOH (molecule 37)

GEISA-09 contains a complete replacement and enhancement of the spectroscopic information provided for formic acid (62,684 entries; 8 bands). Indeed, until GEISA-03, only parameters for 3388 lines of the v<sub>6</sub> band of *trans*-H<sup>12</sup>C<sup>16</sup>O<sup>16</sup>OH near 9 µm were available. They originated from the work of Goldman and Gillis [332]. The sum of the line intensities was equal to  $1.757 \times 10^{-17}$  cm<sup>2</sup>/ (molecule cm<sup>-1</sup>) at 296 K, determined using a Fourier transform laboratory spectrum recorded at the University of Denver.

GEISA-09 provides spectroscopic information for *trans*- $H^{12}C^{16}O^{16}OH$  in three spectral regions: the pure rotation spectrum in the far infrared, the  $v_6$  and  $v_8$  bands near 9  $\mu$ m, and the  $v_3$  band around 5.6  $\mu$ m.

Far-infrared Fourier transform spectra of the pure rotation spectrum of formic acid were recorded in the range from 20 to 130 cm<sup>-1</sup> and analyzed by Vander Auwera [333]. To provide an accurate set of parameters describing the rotational structure of the ground state of trans-H<sup>12</sup>C<sup>16</sup>O<sup>16</sup>OH, 592 far-infrared line positions were fitted together with 372 microwave lines [334-336]. The resulting constants and known dipole moment [337] were then used to calculate the positions, intensities and lower state energies of 6808 *a*- and *b*-type pure rotation lines observed between 10 and 100 cm<sup>-1</sup>, originating from  $J/K_a$ levels ranging from 0/0 to 70/17, corresponding to  $\Delta K_a = 0, \pm 1$  and  $\Delta K_c = \pm 1, \pm 3$ , and being stronger than  $4.0 \times 10^{-26} \text{ cm}^{-1}$ /(molecule cm<sup>-2</sup>) at 296 K. The line positions have been substantiated by a study of Winnewisser et al. [338]. Note that the intensities listed in GEISA-09 are a factor 4 larger than those listed in Table II of [333], because of the oversight of the nuclear spin degeneracy of the hydrogen in the latter. To complement these data, the self- and air-broadening parameters, and temperature dependence exponent of the air-broadening parameter of all the lines were set to the same values as applied to the  $v_6$  and  $v_8$  bands (see here below).

The 9  $\mu$ m spectral region was updated according to the recent work by Vander Auwera et al. [339]. They reported absolute line intensities measurements for the v<sub>6</sub> and v<sub>8</sub> bands using Fourier transform spectroscopy, taking the existing dimer (HCOOH)<sub>2</sub> into account in the analysis. They showed that the intensities reported by Goldman and Gillis [332], and therefore in GEISA-03, were a factor of about 2 lower than the average of the other existing

laboratory measurements, and than theoretical calculations. Relying on results of that work, Perrin and Vander Auwera [340] generated a new set of 49,625 line positions, intensities and lower state energies covering the range from 940.20 to 1244.41 cm<sup>-1</sup>. To complete these data, the self- and air-broadening parameters, and the temperature dependence exponent of the air-broadening parameter of all the lines were set to  $0.32 \text{ cm}^{-1} \text{ atm}^{-1}$ [339],  $0.101 \text{ cm}^{-1} \text{ atm}^{-1}$  [341] at 296 K, and n=0.75, respectively. With a sum of the line intensities equal to  $3.51 \times 10^{-17} \text{ cm}^{-1}/(\text{molecule cm}^{-2})$  at 296 K and a threefold increase of the wavenumber coverage, this new list was shown to provide a significantly improved modeling of the v<sub>6</sub> spectral region of formic acid [340].

Using high-resolution Fourier transform spectra of trans-HCOOH recorded at 5.6 µm, Perrin et al. [342] carried out an extensive analysis of the strong v<sub>3</sub> fundamental band at 1776.83 cm<sup>-1</sup>, starting from results of a previous analysis [337]. As pointed out in the literature, the  $v_3$  band is significantly perturbed by resonances with numerous dark bands. Perrin et al. [342] were able to assign series belonging to the  $v_5 + v_7$ ,  $v_5 + v_9$ ,  $v_6 + v_7$  and v<sub>6</sub>+v<sub>9</sub> dark bands, located at 1843.48, 1792.63, 1737.96 and 1726.40 cm<sup>-1</sup>, respectively. The model used to calculate energy levels accounted partly for the observed resonances, and reproduced most of the observed line positions, within experimental uncertainties. Absolute line intensities were also determined in that work with an accuracy estimated to 15% [342]. From these results, the first database for the 5.6 µm region of the formic acid spectrum was built. It includes 6251 lines belonging to the  $v_3$ ,  $v_5 + v_7$ ,  $v_5 + v_9$ ,  $v_6 + v_7$  and  $v_6 + v_9$  bands of *trans*- $H^{12}C^{16}O^{16}OH$  with  $J \le 66$ ,  $K_a \le 18$ , and lower and upper states energies up to 2700 and 3600 cm<sup>-1</sup>, respectively. Table 6 of Ref. [342] details the contents of the line list.

#### 2.2.29. SF<sub>6</sub> (molecule 39)

Sulfur hexafluoride is a strong greenhouse gas whose concentration in the atmosphere should be monitored and limited, according to the Kyoto protocol [343]. The spectrum of  $SF_6$  is, however, poorly characterized, (at least for atmospheric purposes). The main reason is that this molecule is heavy, which has two important consequences for its spectrum: (i) there are low-lying bending vibrational modes producing a lot of hot bands and (ii) the spectrum is very dense so that even at high resolution there is virtually no isolated line, each line being a cluster of many overlapping transitions. The second point renders the determination of line intensities and, thus, of dipole moment derivatives, very difficult.

Although at lot of work remains to be done on this molecule, many vibrational bands have been investigated in the past years at ICB [344]. A new line list for the v<sub>3</sub> stretching and the v<sub>4</sub> bending fundamental regions has been produced. The only partial knowledge of the inactive v<sub>6</sub> lowest fundamental still prevents a full hot band analysis, especially for v<sub>3</sub>+v<sub>6</sub>-v<sub>6</sub>. However, the lower spectral density in the v<sub>4</sub> region has allowed the detailed investigation of v<sub>4</sub>+v<sub>6</sub>-v<sub>6</sub> [345]. In the case of v<sub>3</sub> itself, which is the strongest absorption band, a very detailed line position analysis exists, based on various

high-precision experimental data (FTIR but also saturated absorption and IR–IR double resonance). In this case, the resulting accuracy for line positions is estimated to be better than 0.001 cm<sup>-1</sup> up to J=100. For the v<sub>4</sub> fundamental, the accuracy for line positions is around 0.001 cm<sup>-1</sup> up to J=100 and for the v<sub>4</sub>+v<sub>6</sub>-v<sub>6</sub> hot band it is *ca.* 0.002 cm<sup>-1</sup> up to J=65. The J values given above correspond to the highest values for the assigned lines. The accuracy may decrease quickly when extrapolating to higher J values, although this is difficult to estimate in a quantitative manner.

As mentioned above, the question of line intensities in the case of  $SF_6$  is a difficult problem. To generate the present list, we used the best-known dipole moment derivative values found in the literature [346,347]. We checked with the previous list for v<sub>3</sub> from GEISA-03 that we obtain exactly the same intensities in this case. However, we globally estimate the line intensity accuracy to be no better than 20 %, in the absence of precise intensity measurements on isolated lines.

Analyses and calculations have been performed with the HTDS software [348]. The whole original ICB line list contains 30,106,484 entries. It has been reduced by applying intensity cutoff in suitable for SF<sub>6</sub> impact signatures in most atmospheric radiative transfer calculation. The applied intensity cutoff, in  $\text{cm}^{-1}/(\text{molecule cm}^{-2})$  at 296 K, had the value  $10^{-24}$  for band v<sub>3</sub> (46,031 lines retained among 2,826,164 in the original list) and  $10^{-23}$ for bands  $v_4$  (10,986 lines retained among 2,657,543) and  $v_4 + v_6 - v_6$  (35,381 lines retained among 24,622,777). As a consequence, the new line list for SF<sub>6</sub> in GEISA-09 (spectral range 580 to 996  $cm^{-1}$ ) contains a total number of entries reduced to 92,398 lines (6 bands). For the whole line list, a default value of  $0.50 \text{ cm}^{-1} \text{ atm}^{-1}$  has been given to the air-broadening pressure half-widths and of 0.65 to the associated temperature dependence coefficient n.

## 2.2.30. C<sub>3</sub>H<sub>4</sub> (molecule 40)

Line parameters for two  $CH_3C_2H$  ( $C_3H_4$ ) bands (the  $v_{10}$  at 331 cm<sup>-1</sup> and the  $v_9$  at 639 cm<sup>-1</sup>) were provided by Graner (private communication), based on constants of Pekkala et al. [349] for the frequency calculations and Blanquet et al. [350] for intensities of individual lines. For the  $v_{10}$  band, the study of a first spectrum at a resolution of 0.0056 cm<sup>-1</sup> by Horneman et al. [351] was followed by the analysis of a 0.002 cm<sup>-1</sup> resolution spectrum by Graner and Wagner [352]. The description of the  $v_{10}$  was accomplished and, in addition, two main hot bands were also provided [352,349]. In the 16-µm region, the  $v_9$  fundamental band was recorded at 0.003 cm<sup>-1</sup> resolution and a full analysis was completed by Pekkala et al. [349] and Pekkala [353].

The extraction of intensities from these high resolution spectra was not an easy task. As a consequence, global intensities from the literature were used to predict individual line intensities, as explained by Horneman et al. [351].

This dataset was first applied to Titan in Coustenis et al. [354]; see Fig. 11a therein. Both propyne bands were detected on Titan and the more accurate spectroscopic parameters are presented GEISA-09 which is updated for the first time since the 1992 edition. These parameters allow for a better determination of the molecule abundance since it can now be separated from the nearby  $C_4H_2$  band [312]. The  $C_3H_4$  GEISA-09 line list includes 19,001 entries (22 transitions) against 3390 entries (1 transition) in GEISA-03.

The CDMS catalog contains entries for v=0,  $v_{10}=1$ ,  $v_{10}$  and  $v_9$  based on [355,356]. A future update of GEISA shall consider these entries or may even be based on [357].

## 2.2.31. ClONO<sub>2</sub> (molecule 42)

The rotational transitions from 0 to  $45 \text{ cm}^{-1}$  for the ground and  $v_9=1$  vibrational states have been included in the GEISA database for the first time.

The predicted transitions for each isotopologue are based on the spectroscopic constants derived from the analyses of millimeter and submillimeter wave rotational spectra in Refs. [358–360]. All predictions were calculated using the SPCAT program package ([192]; http://spec.jpl.nasa.gov/ftp/ pub/calpgm/spinv.pdf) for a temperature of 296 K.

From Ref. [361], isotopic abundances of 0.74957 and 0.23969 and rotational partition functions of 4,788,362 and 4,910,202 were used in the predictions for the 35 and 37 chlorine isotopologues, respectively. A vibrational partition function of 4.02 [362] was used that includes the  $v_9$  vibrational mode that has a band origin near 121 cm<sup>-1</sup>. The rotational spectra from the ground and  $v_9=1$  states account for about 39% of the thermally populated states. Future updates will include the addition of pure rotational spectra from higher lying vibrational states and the infrared simulation of the  $v_6$  fundamental band and the first two associated hot bands for each isotopologue in the 22 µm region. Due to the low lying  $v_9$  mode, there will be significant hot band intensity contributions to each infrared band.

The  $CIONO_2$  GEISA-09 line list includes 356,899 entries (7 bands) against 32,199 entries (3 bands) in GEISA-03.

#### 2.2.32. CH<sub>3</sub>Br (molecule 43)

CH<sub>3</sub>Br contributes significantly to ozone depletion since it is dissociated by UV radiation producing Br radicals who catalyze the destruction of ozone [363]. This molecule is the major contributor to bromine in the stratosphere and the main organobromide in the lower atmosphere. The bromine atoms are 50 to 60 times more destructive of ozone than the chlorine atoms coming from the chlorofluorocarbons compounds (CFCs) [364].

Methyl bromide spectroscopic line parameters are present for the first time in GEISA-09 (36,911 entries; 6 bands). Two line lists of both isotopologues have been generated: one around 10 µm for the v<sub>6</sub> band [365], and the other around 7 µm for the interacting v<sub>2</sub> and v<sub>5</sub> bands [365–367]. In natural abundances, methyl bromide is composed of 50.54% of  $CH_3^{79}Br$  and 49.46% of  $CH_3^{81}Br$ . Note that the broadening coefficients and its temperature dependence obtained in Refs. [365,366] around 10 µm have been used for the 7µm spectral region. Air-broadening coefficients have been deduced from nitrogenbroadening coefficients using a constant scaling as for the H<sub>2</sub>O molecule, for which air-broadening coefficients could be obtained by multiplying N<sub>2</sub>-broadening coefficients by the value 0.9 as suggested in Refs. [369–371]. Since CH<sub>3</sub>Br is chemically and structurally related to CH<sub>3</sub>Cl, we proposed to use the ratio HWHM<sub>air</sub>/  $HWHM_{N_2} = 0.96$  in order to convert the N<sub>2</sub>-broadening to the air-broadening. This value comes from the ratio  $HWHM_{N_{\rm o}}/HWHM_{O_{\rm o}}\!=\!1.25$  obtained by averaging measurements of CH<sub>3</sub>Cl from Refs. [372,373]. Note that this result is quite similar to what has been proposed for ozone in Ref. [374]. This procedure is approximate since  $HWHM_{N_2}/HWHM_{O_2}$  varies from line to line. However it is expected to be accurate within a few percent which is sufficient in view of the experimental uncertainties and the accuracy of the calculations. Also, the air-width temperature dependence has been added in both spectral regions, using the same values as those obtained for the  $N_2$ -width temperature dependence (see Eq. (5) and text in Ref. [366]). Accuracies and details of the line parameter calculation can be found in Refs. [366,368]. In the absence of experimental or theoretical results for air-pressure shifts for CH<sub>3</sub>Br, the GEISA standard missing value of -9.999999 cm<sup>-1</sup> atm<sup>-1</sup> has been used for all transitions. Note also that line mixing effects have been observed and modeled in the strong Q-branches between 220 and 300 K [375,376]. Line mixing parameters (for direct calculation or Rosenkranz profile [377]) are available upon request to the authors.

#### 2.2.33. CH<sub>3</sub>OH (molecule 44)

The importance of methanol microwave, millimeter wave, sub-millimeter wave and terahertz spectroscopy to space science and astrophysics can be traced back to several decades ago when methanol was first discovered in interstellar clouds and star forming regions [378]. The rich variety of torsion-rotational methanol transitions falling in the frequency bands accessible to most radio and sub-millimeter wave telescopes and notably the new Herschel (http://www.esa.int/SPECIALS/Herschel/SEMB M00YUFF\_0.html), ALMA (http://www.eso.org/sci/facil ities/alma/) and SOPHIA (http://www.sofia.usra.edu/) observatories, leads to a dense and detailed interstellar spectrum and demands an accurate knowledge of the methanol energy levels so that the interstellar "methanol weeds" can be removed. The infrared spectroscopy of methanol has also acquired renewed importance in wide areas of application in recent years, such as the recent observations of the 10 µm feature in forest fire [379], the influence of biogenic emissions on upper-tropospheric methanol as revealed from space [380], observations in the terrestrial atmosphere [381], the 3 µm features in several comets and the icy mantles of interstellar dust grains [382-385]. These applications require reliable simulation of the absorption band profiles at any prescribed conditions of temperature and density. Achieving reliable calculations in turn requires detailed understanding of the vibration-torsion-rotation structures of the bands, in terms of both the line positions and intensities.

A methanol line list (19,897 entries; 16 bands) is included for the first time in the GEISA database; it consists of two regions, 0.019265–33.336958 cm<sup>-1</sup> and 911.608420–1407.205540 cm<sup>-1</sup>. The first region is based on a global analysis of the first two torsional states of  $v_{12}$ =0, and 1 and  $J_{max}$ =20 [386] which led to a prediction

list to  $J_{max}$ =26 at a frequency cutoff of 1 THz [387]. Line strengths in that list were calculated using permanent dipole moment values of  $\mu_a = 2.999 \times 10^{-30}$  C m (0.899 D) and  $\mu_{\rm b} = -4.803 \times 10^{-30} \,{\rm Cm}$  (-1.44 D). The list was originally designed at that time to assist the radio astronomy community. More recently, an expanded global analysis with  $v_{12}=0, 1, 2$  and  $J_{max}=30$  has been published [388]. The second region was built on extensive Fourier transform spectroscopic analyses of methanol spectra in the 10 µm region ([389] and references therein). Due to strong vibration-torsion and rotational interactions, the transitions observed in the 10 µm region arise not only from the  $v_8$  CO-stretch fundamental band, but also from  $v_8$  hot bands and nearby vibrations such as  $v_5$ ,  $v_6$  and  $v_7$ entering in the region with different  $v_{12}$  torsional combinations. Within the limits of the isolated vibrationtorsion-rotation band model, the predicted positions and intensities unfortunately did not reproduce the spectrum within experimental uncertainties for  $v_8$  and  $v_8 + v_{12}$ . In addition to strong and medium intensity transitions of the  $v_8$  and  $v_8 + v_{12}$  bands, there are many additional transitions appearing with visible intensity in the spectral window; these were identified as belonging to the  $v_8 + 2v_{12} - 2v_{12}$ ,  $v_7$ —ground,  $v_7 + v_{12} - v_{12}$ ,  $v_6 - v_{12}$ ,  $v_6 - 2v_{12}$ ,  $v_6 + v_{12} - v_{12}$ ,  $v_5 - 2v_{12}$ ,  $3v_{12}$ —ground and 4v<sub>12</sub>—ground bands. Many of these transitions are perturbation-induced, gaining intensity via anharmonic and Coriolis interactions with the strong  $v_8$  vibration in the region. Thus, with an isolated-band approach, these transitions cannot be modeled in either position or intensity. Therefore, it has been chosen simply to include empirical positions and intensities of these features whenever available in our database.

Arriving at the ultimate 10 µm region database, several steps were taken to ensure that the contents reflected the best knowledge of the molecule at the present time (i.e., with observed positions and intensities substituted for predictions whenever available). More specifically, (i) line positions (for 95% of the transitions) were replaced with observed values from the NRC FT spectra except for the congested Q-branch region, in which Q transitions were recomputed from the corresponding observed R- and P-transitions using averaged upper-state term values; (ii) intensities were replaced with measured intensity retrievals from the highest density Kitt Peak spectrum (1.95 Torr, 10 cm). With the predicted database as the input, over 13,500 new intensities were retrieved between 970 and 1085 cm<sup>-1</sup>, including a few lines not currently assigned. Weak lines in the prediction that could not be discerned in the new effort were added to the database with a "default intensity" in order to maintain a complete record of known assignment; the very low intensity value of  $10^{-26}$  cm<sup>-1</sup>/ (molecule  $cm^{-2}$ ) was chosen so that these unmeasured transitions would not contribute extra absorption in the radiative transfer calculations for most applications.

The lower state transition energy is referenced to 128.1069 cm<sup>-1</sup> for the K=0 *a* level, the temperature dependence of width have been attributed the 0.75 default value and the air pressure shift of the line transition has the GEISA-09 standard missing value (i.e., -9.999999). The vibrational index are: Ground,  $v_{12}$ ,  $2v_{12}$ ,  $3v_{12}$ ,  $4v_{12}$ ,  $v_8$ ,

 $v_8 + v_{12}$ ,  $v_8 + 2v_{12}$ ,  $v_7$ ,  $v_7 + v_{12}$ ,  $v_6$ ,  $v_6 + v_{12}$ ,  $v_5$  for upper and lower states and the torsional symmetry A, E<sub>1</sub> or E<sub>2</sub>; the overall rotational angular momentum *J* involves component *K* along the molecular *a*-axis. Resolved *K*-doublets of A symmetry have an additional  $\pm$  to distinguish the A<sup>+</sup> or A<sup>-</sup> component of the doublet.

#### 2.2.34. NO<sup>+</sup> (molecule 45)

The data on the nitric oxide molecular ion  $\mathrm{NO}^+$  are debuting in GEISA-09.

The line positions of the GEISA-09 NO<sup>+</sup> line list are issued of a paper by Lopez Puertas et al. [391] and associated subsequent publications: using high resolution (0.035 cm<sup>-1</sup> unapodized) spectra of the Earth's atmosphere, recorded by the MIPAS experiment, line positions of rovibrational NO<sup>+</sup> transitions have been obtained with an unprecedented accuracy. As a consequence, comparisons with the HITRAN 1996 line list [390] have shown that the spectral line positions of the NO<sup>+</sup> (1–0) and of the NO<sup>+</sup> (2–1) rovibrational bands are shifted by about 0.15 cm<sup>-1</sup> and approximately 0.05–0.1 cm<sup>-1</sup>, respectively.

For  $J'' \leq 40$ , the archived line positions have been derived from a set of Hamiltonian constants for NO<sup>+</sup> obtained from a fit of the MIPAS data together with the existing microwave and infrared data (see [391] for details). Accurate frequencies for high-*J* values cannot be predicted with the use of those constants; consequently, for *J*'' greater than 40, line positions have been derived from the former HITRAN 1996 archive [390].

It has to be noted that the newly generated NO<sup>+</sup> GEISA-09 line list by Flaud [391] (1206 entries; 6 bands, in the spectral range 1634.831–2530.462 cm<sup>-1</sup>), is similar with the HITRAN-08 one [13]; this includes, in particular, the intensities, kept from Werner and Rosmus [390], and the default value 0.06 cm<sup>-1</sup> chosen for the air-broadened half-widths, as well.

It is clear that new high resolution spectra of the NO<sup>+</sup> species are needed in order to improve its spectral parameters.

## 2.2.35. HNC (molecule 46)

Although HCN and HNC actually lie on a single potential energy surface, they are separated by a significant barrier [392]. Within GEISA-09 they are treated as separate species and HNC is a new molecular species for this new edition of GEISA. HNC is the less stable isomer but is known to be overabundant compared to HCN in the interstellar medium (e.g., [393]). Furthermore the partition function of HNC increases much more rapidly with temperature than that of HCN meaning that at temperatures of about 2500 K, the equilibrium abundance of HNC should be about 20% of HCN [394]. The spectrum of HNC has been identified in carbon stars Harris et al. [395].

The GEISA-09 HNC line list was compiled by Harris [261] for the main isotopologue  $H^{14}N^{12}C$ . Among an initial total of 9117 entries, in the spectral range 0.216955–12,594.316928 cm<sup>-1</sup>, 3498 are included only in a supplemental line list because they did not have upper vibrational states identification. Consequently, the final GEISA-09 archived HNC data comprises 5619 lines (84 bands) in the spectral range 0.216955–4814.904168 cm<sup>-1</sup>.

As for HCN, the GEISA-09 HNC line list was constructed from a combination of experimental and theoretical data. The theoretical data are taken exclusively from the line list of Harris et al. [263]. Experimental data are used in preference to the *ab initio* data when they are available. The GEISA-09 HNC line list is less extensive than that for HCN; it is also less accurate since there is substantially less laboratory data to base it on. The spectral region covered for HNC is 0.217–12,594 cm<sup>-1</sup>. Hot bands with a lower vibrational state of 2 quanta of bend, are given for most of the transitions.

The GEISA-09 HNC line list was constructed in the following stages:

- Construction of a list of laboratory determined energy *levels*: The laboratory line frequency measurements of Northrup et al. [396] were used to determine a set of experimental HNC energy levels. This was done by using a technique that deviates only slightly from that of Harris et al. [271]; the rotational constants are used to compute energy levels up to an angular momentum quantum number of 60.
- Construction of a list of laboratory determined line positions: Using the laboratory determined energy levels it is straight forward to compute a list of line positions for dipole moments allowed transitions. The well known selection rules for dipole moments transitions require a change in symmetry and allow a change in angular momentum of 0, +1. When applied to HNC the allowed transitions form two groups. The first has a change in parity of the vibrational angular momentum with no change in total angular momentum. The second group has no change in the parity of the vibrational angular momentum, but a change of plus or minus one in total angular momentum. For all the dipole moments allowed transitions between laboratory determined energy levels, line positions were computed by subtracting lower state energy from upper state energy.
- Construction of a list of laboratory determined line intensities: A list of line intensities was computed from laboratory data given by Nezu et al. [397]. These data are given in the form of band dipoles that are supplemented with Hermann–Wallis factors. From these data, the intensities of individual lines were computed by using the relevant Höln–London factor and the equation given by Maki et al. [272].
- Construction of laboratory determined line list: Experimentally measured line intensities were inserted into the list of laboratory determined energy levels. In this way, an HNC line list based upon laboratory measurements was created.
- Augmentation of the laboratory determined line list with ab initio line intensities: Many of the intensities for the dipole allowed bands have not been measured. The resulting gap in the laboratory determined line list may only be filled by *ab initio* data.

Many of the transitions in the *ab initio* line list of Ref. [263] have been assigned an approximate vibrational quantum number. We were therefore able to insert the line strengths from Ref. [263] line list into the GEISA-09 final HNC file, creating a more complete list of lines.

• Augmentation with ab initio data and truncation: The upper and lower energy levels for many strong room temperature lines have not been determined. In order to account for these strong lines the HNC GEISA-09 line list was augmented with purely *ab initio* line frequency and intensity data from Ref. [263]. Finally, 3498 entries have been suppressed from the original file because of lack of upper vibrational quanta identification.

Recent reports on emission spectra of HNC [398,399] not only provide many more experimental lines, but also used in the analyses comparatively recent pure rotational data [400,401]. These results will be used in updated CDMS entries and may also be used in future GEISA updates.

#### 2.2.36. C<sub>6</sub>H<sub>6</sub> (molecule 47)

Benzene  $(C_6H_6)$  (benzol, benzin) is an aromatic hydrocarbon produced in the Earth's atmosphere and is found in air due to emissions from the burning of coal and oil and also from gas stations, and from motor vehicle exhaust. Benzene is also of importance for astronomical studies. The high abundances of N<sub>2</sub> and CH<sub>4</sub> in the atmosphere of Titan, Saturn's largest moon lead to high abundances of nitrogen and carbon compounds, and its atmosphere and smog-like haze are of particular interest because of its similarity to the atmosphere that may have existed on Earth before life began. Polycyclic Aromatic Hydrocarbons (PAHs) are important interstellar species, and their precursor benzene (C<sub>6</sub>H<sub>6</sub>) has been detected in our solar system, in particular on Titan. Benzene was identified on Titan through ISO and Cassini/CIRS data [312]. It has also been measured in the upper atmospheres of Jupiter at midlatitudes and Saturn (disk average) [402].

Benzene is introduced in GEISA-09 for the first time. Line parameters for the  $v_4$  band of benzene near 678 cm<sup>-1</sup> were provided by Dang-Nhu (private communication) and generated from the molecular constants and band strength compiled in Dang-Nhu and Plíva [403].

Two approaches were used to determine the absolute intensities. Dang-Nhu et al. [404] made a line-by-line study, using a very high resolution tunable diode laser which yielded 30 individual intensities, from which a vibrational strength was derived (see also [403]). At the same time, a study at medium resolution  $(1 \text{ cm}^{-1})$  performed on spectra recorded at LISA by Raulin et al. [405] provided the integrated band intensity of benzene in the spectral region which was related to the previous one through the vibrational partition function.

This dataset (9797 lines) was first applied to modeling of the Titan spectrum in Coustenis et al. [354,318]; see Figs. 5, 6, 8, 9 and 11a therein.

### 2.2.37. C<sub>2</sub>HD (molecule 48)

The line list of monodeuterated acetylene is new in the GEISA database. The need for a line list of deuterated acetylene arose following the recent detection of this isotopologue in the atmosphere of Titan by Coustenis et al. [406]. The line list has been assembled by a joint effort of several laboratories [407]. It is based on new band intensity measurements performed at a resolution of  $0.5 \text{ cm}^{-1}$  in France (LISA) and a new analysis done in Belgium (ULB, UCL) of the high resolution spectra of C<sub>2</sub>HD recorded in Italy (University of Bologna) [407]. The new global fit was obtained by using the computer package developed in UCL and dedicated to both energy and intensity treatments [408,409]. Included lines belong to both bending modes  $v_4$  and  $v_5$  which could be detected on Titan thanks to their strong Q-branch at 519 and 678 cm<sup>-1</sup>, respectively. Lines belonging to both strong stretching modes  $v_1$  and  $v_3$  centered at 3335.6 and 2583.6  $\text{cm}^{-1}$ , respectively, are also present in the new line list. A total of 15,512 lines (348 bands) are present in the list with a minimum intensity of 1.6  $10^{-25}$  cm<sup>-1</sup>/ (molecule cm<sup>-2</sup>) at 296 K. All five vibrational modes and both  $\ell$  values are used in the vibrational assignment of the upper and the lower level of each transition ( $v_1$ ,  $v_2$ ,  $v_3$ ,  $v_4$ ,  $v_5$ ,  $\ell_4$ ,  $\ell_5$ ). The line broadening parameters of C<sub>2</sub>HD have been assumed to be equal to those of the most abundant C<sub>2</sub>H<sub>2</sub> isotopologue.

The study of deuterated acetylene in planetary atmospheres is of great importance and in particular the determination of D/H isotopic ratios. The recent detection of <sup>12</sup>C<sub>2</sub>HD in Titan allowed a value of D/H [406] to be determined. This could be compared to the values obtain for CH<sub>4</sub> (CH<sub>3</sub>D) and H<sub>2</sub> (HD) as C<sub>2</sub>HD is the third deuterated molecule to be detected in Titan's atmosphere.

## 2.2.38. CF<sub>4</sub> (molecule 49)

In the previous editions of GEISA, tetrafluorocarbon (CF<sub>4</sub>) was referred to as CFC-14 and was only included in the cross-sections part [410], with no line list. It is, however, a strong greenhouse gas of both anthropogenic and natural origin [411,412]. Its concentration is increasing in the atmosphere [413,414]. Although it has been identified and measured from balloon-borne measurements [415], its spectroscopy remains only very patchily investigated, for much the same reasons as for SF<sub>6</sub> (presence of many hot bands, dense spectrum with clustered lines). Its infrared spectrum is dominated by the strong v<sub>3</sub> stretching fundamental band at 1282 cm<sup>-1</sup> [415], this band being strongly coupled with the first overtone of the v<sub>4</sub> bending mode.

The v<sub>4</sub> (around 15.8 µm) and  $2v_4/v_3$  regions (around 7.3 µm) have been recently reinvestigated, thanks to several new Fourier transform infrared spectra recorded at a resolution of 0.003 cm<sup>-1</sup>. Just as in the previous work of Gabard et al. [416], a simultaneous analysis of the ground state, v<sub>4</sub>, v<sub>3</sub>,  $2v_4$  and  $v_3 - v_3$  bands was performed, thanks to the XTDS and SPVIEW programs [417] developed by the ICB group. Compared to Ref. [416], the present work extends the analysis to much higher *J* values (70 instead of 40 for v<sub>4</sub> and 63 instead of 32 for the  $2v_4/v_3$  dyad). As for absorption intensities, it was possible to go a bit further than for SF<sub>6</sub>. By calculating synthetic spectra for exactly the same physical conditions as for the experiment, it was possible to fit the v<sub>4</sub> and v<sub>3</sub> dipole-moment derivatives. The results compare very

well to the literature values of Papoušek et al. [418]. The details of this new analysis will be given in a forthcoming paper [419].

This analysis allowed to generate the first reliable line list for  $^{12}$ CF<sub>4</sub> that covers the spectral ranges 600 to 670 cm<sup>-1</sup> (v<sub>4</sub>) and 1276 to 1290 cm<sup>-1</sup> (2v<sub>4</sub>/v<sub>3</sub>). Tetrafluorocarbon becomes GEISA-09 molecule number 49. The precision for line positions is estimated to be around 0.001 cm<sup>-1</sup>, up to *J*=60. The intensity accuracy, however, may not be better than 20%, especially for the high-*J* regions. Line-broadening coefficients were taken from Ref. [420]. The newly archived CF<sub>4</sub> line list comprises 60,033 entries (5 bands) in the spectral range 594-1312 cm<sup>-1</sup>.

## 2.2.39. CH<sub>3</sub>CN (molecule 50)

CH<sub>3</sub>CN (methyl cyanide, acetonitrile, ethanenitrile), a molecule of astronomical and atmospheric importance, is a new entry in GEISA-09. Line parameters of CH<sub>3</sub>CN are needed for planetary studies because this species has been observed, by heterodyne millimeter wave spectroscopy from the ground [421], on Titan. The dissociation of N<sub>2</sub> leads to the formation of nitriles such as HCN, HC<sub>3</sub>N and  $C_2N_2$ , identified for the first time by the Voyager probes in the earlier 1980s. One of the goals of the Cassini-Huygens mission, investigating the Saturn system between 2004 and 2008, was to map all the photochemical compounds, hydrocarbons and nitriles, in order to better understand the photochemical cycle of Titan and its coupling with the dynamics and the production of organic aerosols [312,422]. CH<sub>3</sub>CN spectra have been observed in comets [423], and in interstellar molecular clouds [424], as well.

CH<sub>3</sub>CN is also a gas present in the Earth's atmosphere with a lifetime of several months, mainly emitted through forest fires and then probably deposited in the oceans. Since 1993, this molecule has been classified as an atmospheric pollutant and is the object of a number of varied chemical, biological and atmospheric [425–428] studies.

The GEISA-09 CH<sub>3</sub>CN line list consists of spectroscopic parameters for two different regions.

*Region* 1: as the result of a multispectrum nonlinear least squares fitting technique applied to measure accurate zero-pressure line center positions, Lorentz self- and N<sub>2</sub>-broadening coefficients and self- and N<sub>2</sub>-pressureinduced shift coefficients, 3571 features have been archived in the  $v_4$  parallel band region between 890 and 946 cm<sup>-1</sup>. Published line positions and intensities from Rinsland et al. [429] have been supplemented by unpublished measurements from the same dataset, as well as selected values from preliminary Hamiltonian calculations. Only lines with intensities greater than  $10^{-24}$  $(cm^{-1}/(molecule cm^{-2}))$  at 296 K are included. The spectral region from 918.5 to 920.3  $cm^{-1}$  (containing the Q branch and the P1 and P2 manifolds) proved too dense to measure directly and so these parameters are represented by 326 calculated transitions of v<sub>4</sub>. Some 2243 lines are given without quantum identifications; many are thought to be hot band lines involving yet unanalyzed upper state levels of  $v_4 + v_8$ . The lower state energy of these unidentified lines is set to the GEISA-09 standard missing value, i.e., -0.9999. It should be noted that a number of hotband lines are not included in the list; this is most noticeable at the hot band Q branch near  $924 \text{ cm}^{-1}$ . Measured self-broadening coefficients were available, and identified lines with the same K quantum number and the same or very close *m* were assigned approximately the same or interpolated values. The total number of lines with self-broadening assigned in this manner is 2185. For the lines lacking measured or estimated Lorentz half-width coefficients for air- and self-broadening, default values of 0.14 and 1.5 cm<sup>-1</sup> atm<sup>-1</sup> at 296 K were used, respectively (obtained as an approximate average of measured values). The measured N<sub>2</sub> shifts [429], where available, were inserted for air shifts. Unmeasured pressure shifts are set to zero, the approximate average of the measured values. There are no measurements of the temperature dependence of the Lorentz half-width in air and only one in  $N_2$  [430], so the default *n* is set to the single measured  $N_2$  value of 0.72.

Region 2: an excerpt in the spectral range of 970-1650 cm<sup>-1</sup> of an empirical "pseudo-line-list" (total extent 870–1650 cm<sup>-1</sup>), where the  $v_7$  band around 1050 cm<sup>-1</sup> and the  $\nu_3,\ \nu_6,\ \nu_7\!+\!\nu_8$  bands around  $1450\,cm^{-1}$  are located. This represents a total of 13,601 entries. A pseudo-line list, typically derived by fitting equally spaced "pseudo-lines" to laboratory spectra, is not intended to supplant any proper quantum-mechanically based line list. However, it provides a convenient means for radiative transfer calculations in case quantummechanically derived line lists are unavailable or unreliable. In the process of building up the GEISA-09 CH<sub>3</sub>CN line list, the mixing of quantum-mechanically derived lines and pseudo-lines has been avoided, as one cannot expect to get realistic results in a radiative transfer calculation if the quantum-mechanically derived lines have not been taken into account during the derivation of the pseudo-lines. The pseudo-line-list for CH<sub>3</sub>CN has been successfully used to identify and quantify CH<sub>3</sub>CN in the Earth's atmosphere from balloon-borne solar occultation Fourier-Transform infrared measurements [428] and to attempt its detection on Titan from Cassini CIRS infrared data.

The CH<sub>3</sub>CN pseudo-line-list [431] was created based on 29 laboratory spectra taken at PNNL. The measurements and the absorption cross-sections, including assignments of major bands, are described by Rinsland et al. [432]. The cross-sections were converted back into transmittance spectra from knowledge of the cell length and gas concentrations. The resulting laboratory transmittance spectra were then simultaneously fitted by iteratively adjusting the strengths and ground-state energies of the pseudo-lines. At each line frequency, an effective strength and ground-state energy was derived by simultaneous non-linear least squares fitting to the 29 spectra. The air-broadened half-width was calculated from the ground-state energy using a simple parameterization that results in air-broadened half-widths between 0.04 and 0.08  $\text{cm}^{-1}$ /atm and gives the most appropriate fit to the narrowest features in the considered frequency region. The self-broadened half-width, the temperature dependency of the air-broadened half-width, and the pressure shift were chosen to be values that are typical for heavy molecules.

Due to the resolution of the laboratory spectra of  $0.1125 \text{ cm}^{-1}$  and their spectral point spacing of  $0.0603 \text{ cm}^{-1}$ , a pseudo-line spacing of  $0.05 \text{ cm}^{-1}$  was considered to be appropriate. Note that when the pseudo-line-list is used in radiative transfer calculations, it is recommended that the Doppler-width of the lines is set to the value of the pseudo-line spacing. Otherwise calculations for low pressures will lead to unrealistic spikes at the positions of the individual pseudo-lines.

## 2.2.40. Non updated molecules

Since GEISA-03 no update occurs for the following molecules: CO (molecule 5), OH (molecule 14), HF (molecule 15), HCl (molecule 16), HBr (molecule 17), HI (molecule 18), ClO (molecule 19), GeH<sub>4</sub> (molecule 26), HOCl (molecule 32), COF<sub>2</sub> (molecule 38) and HO<sub>2</sub> (molecule 38). For HO<sub>2</sub>, a technical error in the GEISA-03 rotational quantum number identification is corrected in GEISA-09.

# 3. GEISA-09 infrared absorption cross-sections sub-database

The infrared absorption cross-section sub-database of GEISA-03 contents has been extensively described in Refs. [8,9] (see Table 2.5 of Ref. [9]). In the spectral range from 200 cm<sup>-1</sup> to 2000 cm<sup>-1</sup>, 35 molecular species have been already archived, i.e., CFC-13, CFC-113, CFC-114, CFC-115, CFC-11, CFC-12, CFC-14, HCFC-22, HCFC-123, HCFC-124, HFC-125, HFC-134a, HCFC-141b, HCFC-142b, HFC-152a, HCFC-225ca, HCFC-225cb, HFC-32, HFC-143a, HFC-134, N<sub>2</sub>O<sub>5</sub>, SF<sub>6</sub>, ClONO<sub>2</sub>; HFC-143, HCFC-21, CCl<sub>4</sub>, C<sub>2</sub>F<sub>6</sub>, C<sub>2</sub>H<sub>2</sub>, C<sub>2</sub>H<sub>4</sub>, C<sub>2</sub>H<sub>6</sub>, C<sub>3</sub>H<sub>8</sub>, C<sub>4</sub>H<sub>8</sub>, HNO<sub>4</sub>, SF<sub>5</sub>CF<sub>3</sub>, HCH-365mfc. Note that in this list, many of the molecular species are identified by their common name (i.e., CFC-14 for CF<sub>4</sub>).

# 3.1. Complementary data for species already implemented in GEISA-03

## 3.1.1. SF<sub>5</sub>CF<sub>3</sub> (trifluoromethyl sulfur pentafluoride)

SF<sub>5</sub>CF<sub>3</sub> IR absorption cross-sections from M. Hurley were implemented in GEISA-03. The absorption crosssections measured by Rinsland et al. [433], at five temperatures between 213 and 323 K in the infrared bands of SF<sub>5</sub>CF<sub>3</sub> are newly added to GEISA-09. The spectra were recorded at a resolution of 0.112 cm<sup>-1</sup> using a commercial Fourier transform infrared spectrometer and a 20 cm temperature-controlled sample cell. The full spectral range of the measurements was  $520-6500 \text{ cm}^{-1}$ , with only weak bands observed beyond 1400 cm<sup>-1</sup>. Absorption of thermal radiation in the 8-12 µm atmospheric window region being important for climate change, the measured integrated cross-sections of the significant absorption bands in that spectral region have been added to the GEISA archive as summarized in Table 10. It has to be noted that the SF<sub>5</sub>CF<sub>3</sub> atmospheric growth has closely paralleled the rise of SF<sub>6</sub> during the past three decades, with an estimated radiative forcing of 0.57 W m<sup>-2</sup> ppb<sup>-1</sup>, slightly higher than for SF<sub>6</sub> [434].

#### 3.2. Molecular species added since GEISA-03 edition

#### 3.2.1. $C_6H_6$ (benzene)

To provide a database for both Earth's and planetary atmosphere studies (as an example, benzene has recently been detected in the atmosphere of Titan as the first PAH of this kind [312]), integrated band intensities of benzene at temperatures of 278, 298, and 323 K, in the spectral range 600–6500 cm $^{-1}$  by Rinsland et al. [435], have been added to GEISA-09 IR cross-sections archive. These data derived from pressure broadened (1 atm N<sub>2</sub>) laboratory spectra of benzene vapor (in natural abundance) recorded at PNNL with a 0.112 cm<sup>-1</sup> resolution Bruker-66 V Fourier transform spectrometer configured to operate in the mid-infrared. Using very high precision capacitance nanometers, over nine sample pressures were recorded for each of the three temperatures. Hard-mounted into the spectrometer, a temperature-stabilized static cell (19.94 cm path length), was used for support of the samples introduced into it. Two-hundred fifty-six interferograms were averaged for each sample spectrum. A composite spectrum was calculated for each cell temperature from the individual absorbance spectra recorded at that temperature. The average uncertainty (NIST type-A) is, respectively: 0.40%, 0.38% and 0.54% for the 278, 298, and 323 K spectra. The number density for the three composite spectra was normalized to 296 K. The spectra give the absorption IR cross-sections ( $cm^2$  molecule<sup>-1</sup>, naperian units) of benzene as a function of wavenumber, as summarized in Table 10 for GEISA-09 contents.

## 3.2.2. CH<sub>3</sub>CN (acetonitrile, – methyl cyanide)

Infrared cross-sections were measured at the Pacific Northwest National Laboratory by Rinsland et al. [432]. These 29 spectra covered 600 and 6500 cm<sup>-1</sup> with a resolution of  $0.1125 \text{ cm}^{-1}$  and were measured at three different temperatures (276 K, 299 K, and 324 K). They were recorded with different CH<sub>3</sub>CN volume mixing ratios at 1 atm pressure using N<sub>2</sub> as pressure broadening gas. Table 10 summarizes the related GEISA-09 contents.

#### 3.2.3. C<sub>2</sub>H<sub>3</sub>NO<sub>5</sub> (peroxyacetyl nitrate, – PAN)

PAN is an interesting molecule, linking carbon and nitrogen chemistry, which has recently gained a new importance for remote sensing. The terrestrial spectroscopic signature of PAN in the thermal infra-red was first observed in Los Angeles smog [436] but the ability to observe PAN concentrations more widely has been revolutionized by recent detections in high resolution spectra obtained in balloon-borne [437] and space-borne experiments [438,439]. New spectroscopic data for PAN, in the form of cross-sections, have therefore been included in the GEISA database for the first time, based on the measurements of Allen et al. [440,441]. The cross-sections cover the spectral range between 560 and 1400 cm<sup>-1</sup> at three temperature (295, 273, 250 K), and between 1686 and 2000  $\text{cm}^{-1}$  at two temperatures (295 and 250 K). The data include all bands from  $v_4$  to  $v_{19}$ , except for  $v_{16}$ centered at  $1653 \text{ cm}^{-1}$  which is detected in the original measurements at 295 K but is not included here because of weakness of the band and residual water vapor

Table 10		

Summary of GEISA-09 update and additions for infrared cross-sections.

Molecule	Temperature (K)	Spectral range (cm <sup>-1</sup> )	Foreign broadening pressure <sup>a</sup> (Pa)	Refs.
Trifluoromethyl sulfur pentafluoride, $SF_5CF_3$	213 243 278 298 323	600-2600	101,324.72	[433,434]
Benzene, C <sub>6</sub> H <sub>6</sub>	278 298 323	600–6500	101,324.72	[435]
Acetonitrile, CH₃CN	276.1 298.7 324.1	624–4574	101,324.72	[429,432]
PAN, C <sub>2</sub> H <sub>3</sub> NO <sub>5</sub>	250 273 295	560–2000 560–1400 560–2000	0	[436–443]
Acetone, (CH <sub>3</sub> ) <sub>2</sub> CO	214.0 223.4 223.5 223.5 223.6 233.4 233.4 233.4 233.5 253.3 253.3 253.3 253.3 253.4 253.8 272.3 272.3 272.3 272.3 272.3 272.6 297.4 297.5 297.5 297.8	600-1800	$\begin{array}{c} 0\\ 0\\ 20,811.60\\ 80,260.10\\ 50,902.50\\ 19,985.00\\ 50,022.50\\ 80,113.40\\ 0\\ 50,089.20\\ 79,886.80\\ 20,051.70\\ 0\\ 20,025.00\\ 49,915.90\\ 92,765.70\\ 0\\ 20,718.30\\ 50,062.50\\ 93,325.60\\ 0\\ \end{array}$	[444-447]

<sup>a</sup> Pressure=0.0 Pa: spectra measured for pure gas.

contamination. The band assignments are based on those reported in Gaffney et al. [442] and Bruckmann and Wilner [443]. The five main bands are  $v_4$ ,  $v_5$ ,  $v_9$ ,  $v_{10}$  and  $v_{16}$  centered at 1842, 1741, 1302, 1161.5 and 791.5 cm<sup>-1</sup>, respectively; a small shift of 1 cm<sup>-1</sup> was observed in the peak of the  $v_4$  band at 1842 cm<sup>-1</sup> with temperature [441]. Uncertainties in the cross-sections were estimated to be 5% at 250 K [441] rising to 7% at 295 K [440]. See Table 10 for details.

## 3.2.4. (CH<sub>3</sub>)<sub>2</sub>CO (acetone)

Acetone is a fundamental molecule in volatile organic chemistry which evaporates rapidly, even from water and soil. Once in the atmosphere, it is degraded by UV light with a 22-day half-life. Acetone dissipates slowly in soil, animals, or waterways since it is sometimes consumed by microorganisms, but it is a significant groundwater contaminant due to its high solubility in water. Acetone may pose a significant risk of oxygen depletion in aquatic systems due to the microbial activity consuming it.

The spectroscopic signature of acetone in spectra of the terrestrial atmosphere has been reported first in the  $v_{19}/v_{23}$  band complex centered at 530 cm<sup>-1</sup> (citation in

[444]) and in the v<sub>17</sub> band at 1220 cm<sup>-1</sup> [445,437] and in the  $v_{16}$  Q-branch at 1365 cm<sup>-1</sup> [439]. Hence it has been important to include new spectroscopic data for acetone, under the form of cross-sections, in the GEISA database for the first time, based on the measurements of Waterfall [446]. The cross-sections, at spectral resolution of  $0.03 \text{ cm}^{-1}$ , cover the spectral range between 600 and  $1800 \text{ cm}^{-1}$  around six temperature series (214, 223, 233, 253, 272 and 297 K); see Table 10 for precise details. The data include the  $v_{18}$  band centered at 830 cm<sup>-1</sup>, the  $v_{17}$  at 1218 cm<sup>-1</sup>, the  $v_{16}/v_5$  bands close to 1360 cm<sup>-1</sup> overlapping with the unresolved bands of  $v_{15}$ ,  $-v_4$  and  $v_{21}$ centered between 1430 and 1460 cm<sup>-1</sup>, and the  $v_3$  band centered at 1738 cm<sup>-1</sup>; band assignments are taken from Wang et al. [447]. The  $v_7$  band, centered at 777 cm<sup>-1</sup>, and the  $v_{22}/v_6$  near 1093 cm<sup>-1</sup> are only very weakly present in the measured cross-sections. The main cross-section influence is for the strongest bands observed between 1200 and 1800 cm<sup>-1</sup> for which uncertainties range from 5% (7% for the center of the  $v_3$ , 1738 cm<sup>-1</sup> band) at the strongest parts of the band to 10% towards the edges. For the 830 cm<sup>-1</sup> band, errors are approximately 12% at band center rising to greater than 20% at the band edges.

## 4. GEISA-09 absorption cross-sections sub-database in UV/visible regions

For molecules absorbing in the UV and visible spectral regions, except for  $O_2$ ,  $H_2O$ , NO and OH, there are no individual line lists available, since most molecules and radicals show rather broad-band absorption features due to repulsive potential curves or surfaces in the upper electronic states of the relevant transitions, or due to predissociation. Therefore, absorption cross-sections at all relevant temperatures (and in some cases, pressures), have to be used.

The requirements for the accuracy of molecular absorption cross-sections in this spectral range are most important for strong absorbers, especially for  $O_3$ , but it is also important to highlight the need of very accurate spectral calibration (hence reference data derived from FTS are usually recommended) and for a high dynamic range and absolute consistency between different spectral regions (e.g., between the ultraviolet and visible bands for  $O_3$ , or the consistency with respect to the mid-infrared bands). The latter requirements are especially important when the retrieved atmospheric data are used for photochemical modeling or for chemical budget calculations, where uncharacterized biases between different spectral regions may lead to difficult problems.

In the GEISA database, only UV-visible reference spectra of such molecules that have already been detected in the Earth's or planetary atmospheres are presented, i.e., contrary to other databases which focus on photochemical data (e.g., the NASA-JPL, NIST or IUPAC recommendations) or offer a complete coverage of all absorption spectra (e.g., the MPI Mainz and SoftCon databases). It is also important to note that, in contrast to the HITRAN database, for the sake of coherence of future studies, and to facilitate the comparison with previous work, the GEISA database contains different sets of relevant absorption cross-sections, i.e., multiple data sources and data sets for the same species; however recommendations are made for each molecule.

The GEISA database contains UV–visible absorption cross-sections for the following molecules and radicals: NO<sub>2</sub>, CS<sub>2</sub>, O<sub>3</sub>, SO<sub>2</sub>, O<sub>2</sub>–O<sub>2</sub> (O<sub>4</sub>), OCIO, H<sub>2</sub>CO, OBrO, BrO, NO<sub>3</sub>, HONO, IO, OIO, and aromatic hydrocarbons (i.e., C<sub>6</sub>H<sub>6</sub>, C<sub>7</sub>H<sub>8</sub> and the three isomers of (C<sub>6</sub>H<sub>4</sub>(CH<sub>3</sub>)<sub>2</sub> as well). Here below, each molecule and the relevant reference data in the new GEISA-09 edition will be discussed individually.

## 4.1. NO<sub>2</sub> (nitrogen dioxide)

For NO<sub>2</sub>, there is quite a variety of different laboratory measurements of ultraviolet–visible absorption cross-sections. For atmospheric applications, the currently recommended data set by Orphal [448] is the one of Vandaele et al. [449], but it is important to stress that also the data of Voigt et al. [450], Yoshino et al. [451], Harder et al. [452] and Frost et al. [453] are of high quality and show excellent agreement with each other. The cross-section of Harder et al. may contain a slight contamination by HONO, however. For applications where a very high signal-to-noise ratio is required or in spectral regions where the

previously mentioned  $NO_2$  absorption cross-sections are limited, the data recorded with GOME by Burrows et al. [454] or with SCIAMACHY by Bogumil et al. [455] are recommended (again, it is important to note that these data are limited by the spectral resolution of the instruments). It has to be recalled that, in general, the crosssections recorded by an FTS have a wavelength calibration of better than 0.01 nm [448] which is an important advantage for atmospheric applications, in particular when retrieving several absorbers simultaneously.

### 4.2. CS<sub>2</sub> (carbon disulfide)

Small amounts of carbon disulfide  $CS_2$  are released by volcanic eruptions and marshes. The absorption cross-sections, recorded with an FTS at 294 K covering the 290–350 nm spectral range are from Vandaele et al. [456].

### 4.3. O<sub>3</sub> (ozone)

As for NO<sub>2</sub>, there exist many laboratory measurements of UV-visible absorption cross-sections at atmospheric temperatures (see [448]). However, only a few of them cover the entire spectral range from the ultraviolet to the near-infrared. Therefore, it is difficult to recommend one single data set that would be best suited for all applications. For the Huggins bands (300-360 nm), the recommended reference data are those of Brion et al. [457] and those of Bass and Paur [458], since both data sets cover most relevant temperatures (note however that the data of Brion et al. are not available below 218 K) and have been recorded at high resolution. While the data of Bass and Paur were used as some kind of standard during the past 20 years, more recent studies tend to recommend the data of Brion et al. for atmospheric remote-sensing applications (since they show better wavelength calibration, wavelength sampling, less noise and less inconsistencies concerning the temperature dependence of the cross-sections). For applications where absorption crosssections over a broader spectral range are needed (in particular in the visible and near-infrared, i.e., the Chappuis and Wulf bands, the O<sub>3</sub> cross-sections recorded with GOME [454] or with SCIAMACHY [455] are recommended. These absorption cross-sections show also a very high signal-to-noise ratio, but are partly limited by the spectral resolution of the instruments. If O<sub>3</sub> cross-sections at very high spectral resolution are needed, then the data of Voigt et al. [459] are recommended.

#### 4.4. SO<sub>2</sub> (sulfur dioxide)

SO<sub>2</sub> presents three main regions of absorption in the near ultraviolet domain. The strongest band lies in the 45,000 cm<sup>-1</sup> (220 nm) region and corresponds to the  $\hat{C}^1B_2-X^1A_1$  electronic transition. A strong absorption structure extends between 29,000 and 40,000 cm<sup>-1</sup>, which can be ascribed to at least two electronic transitions. Underlying the structured bands of the  $A^1A_2-X^1A_1$  [460], the 'continuous' absorption has been attributed to the  $B^1B_1-X^1A_1$  transition, which has been predicted by theory [461] and measured by Brand et al. [462]. The

 $A^1A_2-X^1A_1$  transition is forbidden but is observed because of strong vibrational interactions through the v<sub>3</sub> vibration mode and is strongly perturbed by the <sup>1</sup>B<sub>1</sub> state. The allowed transition B<sup>1</sup>B<sub>1</sub>-X<sup>1</sup>A<sub>1</sub> is so perturbed that no rotational or vibrational analysis is possible. It forms a continuum due to the density of weak absorptions. A weak absorption feature arises in the 25,000– 29,000 cm<sup>-1</sup> region (345–400 nm). It has been assigned to the a<sup>3</sup>B<sub>1</sub>-X<sup>1</sup>A<sub>1</sub> electronic transition and is a spinforbidden transition.

In the previous edition of GEISA [8,9] the UV/vis data set for SO<sub>2</sub> consisted in cross-sections recorded with the SCIAMACHY spectrometer [455], covering five temperatures between 203 and 293 K and interesting for planetary science application. The 2009 update consists in recently obtained absorption cross-sections, at high resolution and at high temperatures, in support to planetary applications [463,464]. They were recorded in the 24,000-44,000 cm<sup>-1</sup> spectral range (227–420 nm) with a Fourier Transform spectrometer at a resolution of 2 cm<sup>-1</sup> (0.45 cm MOPD and boxcar apodization). Pure SO<sub>2</sub> samples were used and measurements were performed at room temperature (298 K) as well as at 318, 338 and 358 K. Temperature effects were investigated and were found in favorable agreement with existing studies in the literature. Comparison of the absorption cross-sections at room temperature [465,466] shows good agreement in intensity with most of the literature data, but shows that most of the latter suffer from inaccurate wavelength scale definition. Moreover, literature data are often given only on restricted spectral intervals, whereas this new data set offers the considerable advantage of covering the large spectral interval extending from 24,000 to 44,000  $cm^{-1}$ , at the four temperatures investigated. These data are also available in digital form from the website of the Belgian Institute for Space Aeronomy (http://www.aeronomie.be/ spectrolab/).

#### 4.5. $O_2-O_2(O_4)$ (the so-called oxygen "dimer")

These broad features are mainly used for air mass determination in atmospheric remote-sensing applications. It is rather difficult to recommend one particular set of data since the differences between the available cross-sections are still not well understood. Therefore, the data of Greenblatt et al. [467], of Newnham and Ballard [468], and of Vandaele et al. [456] are all available in the archive.

## 4.6. OClO (chlorine dioxide)

OCIO is involved in polar stratospheric chemistry, linking the catalytic cycles of CIO and BrO, and has been observed in ultraviolet–visible spectra from ground, airborne platforms and satellites. It is important to monitor stratospheric OCIO in order to validate the quantitative understanding of ozone destruction in polar winter. As in the previous edition of GEISA, the UV–visible absorption cross-sections of Kromminga et al. [469] that were recorded at different temperatures using high-resolution Fourier-transform spectroscopy are recommended. For the sake of coherence with previous studies, GEISA also contains the OCIO cross-sections of Wahner et al. [470].

#### 4.7. H<sub>2</sub>CO (formaldehyde, also called CH<sub>2</sub>O or HCHO)

Formaldehyde is another important source of OH radicals in the troposphere, and one of the smallest organic molecules in the atmosphere. Gratien et al. [471] have demonstrated that the high-resolution  $H_2CO$  absorption cross-sections of Meller and Moortgat [472] are in excellent agreement with the available infrared cross-sections. For applications requiring a very high signal-to-noise ratio, the data recorded with SCIAMACHY [455] may also be of interest.

#### 4.8. OBrO

Only cross-sections recorded by an FTS were selected. For OBrO (385–616 nm spectral range) cross-sections are available only at room temperature [473].

#### 4.9. BrO (bromine monoxide)

BrO is observed in the stratosphere but also in the marine troposphere and in volcanic plumes. There are two sets of data which have been recorded using high-resolution Fourier-transform spectroscopy and cover all relevant atmospheric temperatures: Wilmouth et al. [474] and Fleischmann et al. [475]; both show very good agreement. As for OCIO, for the sake of coherence with previous studies, GEISA also contains the BrO absorption crosssections of Wahner et al. [476] that were used as reference spectra, before the new data became available.

## 4.10. NO<sub>3</sub> (nitrogen trioxide; the nitrate radical)

For NO<sub>3</sub>, the main night-time oxidant in the troposphere, but also strongly occurring in the stratosphere, the recommended data set for all atmospheric temperatures is the one of Yokelson et al. [477]; note however that there exists a room-temperature spectrum that was recorded using high-resolution Fourier-transform spectroscopy [448]. The latter paper also provides an accurate theoretical model for the temperature dependence of the strong peak at 662 nm, which is based on the molecular symmetry and structure of the radical.

#### 4.11. HONO (nitrous acid)

Nitrous acid is an atmospheric species that has received a lot of attention in the past decades, since it is a source of OH radicals in the troposphere, while its sources are still not well understood. A recent study by Gratien et al. [478] has shown that the HONO absorption cross-sections of Bongartz et al. [479] and of Stutz et al. [480] are in very good agreement with each other and with the available infrared cross-sections. Therefore, both data sets are recommended.

## 4.12. CHOCHO (glyoxal)

Glyoxal is a small organic molecule involved in tropospheric chemistry and aerosol formation. It has only recently been measured for the first time in the Earth's atmosphere using optical methods. Its sources are still not fully understood, especially since some CHOCHO is also observed over the Pacific Ocean. The recommended absorption cross-sections for CHOCHO are those of Volkamer et al. [481] recorded using high-resolution Fouriertransform spectroscopy.

## 4.13. IO (iodine monoxide)

IO has been observed only in the marine troposphere, and an upper limit of less than 1 pptv has been established for stratospheric IO. The reference data in GEISA are the cross-sections of Spietz et al. [482] which have an excellent signal-to-noise ratio, rather high resolution, and are in good agreement with other studies and with photochemical models of IO chemistry following flash photolysis of suitable precursors.

## 4.14. OIO (iodine dioxide)

As for IO, the OIO radical has been observed only in the marine troposphere. Its atmospheric relevance has been established only as late as 1996 when it was observed for the first time in flash-photolysis experiments by Himmelmann et al. [483]. The reference data in GEISA are the absorption cross-sections of Gomez-Martin et al. [484].

#### 4.15. Aromatic hydrocarbons

UV absorption cross-sections (cm<sup>2</sup> molecule<sup>-1</sup>) of five gaseous aromatic hydrocarbons have been measured with a FTS Bruker IFS120M at the resolution of 1 cm<sup>-1</sup> (0.9 cm MOPD and boxcar apodization) over the 30,000– 42,000 cm<sup>-1</sup> spectral range (238–333 nm). The molecules, benzene (C<sub>6</sub>H<sub>6</sub>), toluene or methylbenzene (C<sub>7</sub>H<sub>8</sub>), and the three isomers of dimethyl-benzene (C<sub>6</sub>H<sub>4</sub>(CH<sub>3</sub>)<sub>2</sub>) also called meta-, ortho-, and para-xylene, were chosen for their importance in the chemistry of tropospheric ozone [485], in urban air quality problems [486] and in astronomical studies [317,318,402,487].

The recordings were carried out under different pressure and temperature conditions with pure samples. The complete dataset is composed of absorption cross-sections for: (i) benzene at 253, 263, 273, 283 and 293 K, (ii) toluene at 263, 273, 283 and 293 K, and (iii) the three isomers of xylene at 273, 283 and 293 K. Wavenumbers are given by increments of  $0.2 \text{ cm}^{-1}$ . Systematic and nonsystematic errors are given separately, a value of 8% being estimated for the former and individual values being reported in a separate column for the latter. The experimental set-up and the procedure of analysis are given in details in [488].

Comparisons with recent studies in the same UV region [488–491] show that large discrepancies are present in some cases which are largely attributed to the experimental difficulties and to a resolution effect.

Compared to these studies, a better spectral resolution, an accurate wavelength scale, and several atmospheric temperatures are provided. A linear parameterization for the temperature effect is also proposed for benzene and toluene in support of remote sensing atmospheric studies both on Earth and on other planets. These data are also available in digital form from the website of the Belgian Institute for Space Aeronomy (http://www.aeronomie.be/ spectrolab/).

# 5. GEISA-09 sub-database on microphysical and optical properties of atmospheric aerosols

Besides the molecular species which define the gaseous infrared opacity in the Earth's atmosphere, aerosol particles also contribute to this opacity. Consequently, a GEISA aerosols sub-database has been constructed. It gathers the micro-physical and optical properties from four published aerosol data catalogs, i.e., Massie [492–494], Rublev [495], Hess et al. [496], Köpke et al. [497], the overall content of which deals with the archive of complex refractive indices and possibly computed optical related properties, for selected basic aerosol components. Softwares for data management and userselected aerosol mixtures elaboration are available as well.

The GEISA-09 aerosols sub-database contains data on microphysical and optical properties of basic aerosol components. The following 4 sub-databases are included:

# 5.1. A database on refractive indices of basic atmospheric aerosol components

This database by Massie [492–494] comprises an extensive archive of complex refractive indices, determined both in situ and in laboratory, from spectral transmission and reflection measurements (over 40 references), of various aerosol components, i.e.,

- Solid substances (0.33–50,000 cm<sup>-1</sup>)
- Water ice (0.–22,570 cm<sup>-1</sup>)
- Water droplets (0.33–15,000 cm<sup>-1</sup>)
- Water soluble components (250–50,000 cm<sup>-1</sup>)
- H<sub>2</sub>SO<sub>4</sub> solutions (0.–50,000 cm<sup>-1</sup>)
- HNO<sub>3</sub> solutions (0.–16,382 cm<sup>-1</sup>)
- Thin films (482–7000 cm<sup>-1</sup>)
- Ternary  $H_2SO_4/HNO_3/H_2O$  solution droplets (2.000–12.1126 cm<sup>-1</sup>).

#### 5.2. The aerosols database from LITMS

The first part of the archive [495] consists in complex indices of refraction of aerosol components, which have been used for the computation of archived aerosol integrated optical properties (extinction coefficient, single scattering albedo, asymmetry factor). In the second part, the so-called AERCOMP (FORTRAN code) software package, allowing the determination of optical properties for user-defined aerosol mixtures, has been included with its associated files for basic aerosol constituent optical properties and related scattering phase functions, as well.

#### 5.3. The database and associated software package OPAC

The first part of this archive [496] is a data set of microphysical properties and the associated optical properties of:

- *Ten basic aerosol components*: Insoluble, soot, water soluble, two sea salt modes (various kinds of salt contained in seawater), three mineral modes (mixtures of quartz and clay minerals), mineral transported, sulfate droplets.
- *Six water clouds*: Stratus (continental and maritime), cumulus (continental (clean and polluted) and maritime), fog, and three kinds of cirrus ice clouds, both in the solar and terrestrial spectral range.

The second part is a FORTRAN program making it possible to extract data from the above archive and allowing for the calculation of any user-defined mixtures of these components. A set of computed typical mixtures is archived, as well.

#### 5.4. The Global Aerosol Data Set GADS

Global fields of all optical parameters necessary for an estimate of the radiative forcing by aerosol particles and to quantify the resulting climate effects are not available from measurements due to the multiple influence parameters. Therefore, using the OPAC aerosol archive, GADS (Köpke et al. [497]), provides the related global aerosol distribution as climatologically averaged values both for the winter (December through February) and summer (June through August) seasons on a global grid with a resolution of  $5^{\circ} \times 5^{\circ}$  longitude and latitude, independently of the components selected in OPAC. More details on the archived files structure is given in Ref. [498].

# 5.5. GEISA interactive web distribution through Ether Products and Services Centre

The scientific input into GEISA is maintained at LMD (Ecole Polytechnique): http://ara.abct.lmd.polytechnique.fr) which involves selection and collection of new or enhanced spectroscopic data in cooperation with spectroscopy laboratories, both theory and experiment, and experts; processing of the data; software development and maintenance for the data base management and products extraction.

The GEISA on line web access and its associated maintenance are responsibility of the Ether Products and Services Centre (http://ether.ipsl.jussieu.fr), at IPSL, where the database is implemented. Ether is especially involved in distribution and generation of products of interest to the Atmospheric Chemistry Research community.

The GEISA web site is freely accessible, via the GEISA logo, through the welcome page of the Ether

web site, which offers the following GEISA interrogation facilities:

- Very detailed information on the available spectroscopic data, i.e., spectroscopic parameters of the individual lines and cross-sections (IR and UV/vis), and optical and microphysical properties of atmospheric aerosols as well.
- FTP access to file data for a quick download of the database, partly or fully.
- Interactive access to the individual line spectroscopic parameters, making possible sharper search and extraction of data of interest. In this purpose, six options are available for display, histograms visualization and extractions of user's selected data information.

## 6. Concluding comments

The 2009 edition of GEISA exhibits important updates in spectroscopic parameters and significant addition of archived molecular species (line transitions and crosssections sub-databases), with an associated extension of spectral ranges (especially towards near IR regions). Some specific results of this effort are especially valuable for various current research programs aiming at a better knowledge of the Earth's and planetary atmospheres, as well as climate and environmental evolution understanding. Examples of such updates, among the ones detailed above are:

- (a) Within the frame work of the IASI METOP program, and ISSWG associated GEISA/IASI efforts, among the 14 molecular species selected for operational meteorological soundings, i.e., H<sub>2</sub>O, CO<sub>2</sub>, O<sub>3</sub>, N<sub>2</sub>O, CO, CH<sub>4</sub>, O<sub>2</sub>, NO, SO<sub>2</sub>,  $NO_2$ ,  $HNO_3$ , OCS,  $C_2H_2$ , and  $N_2$ , the spectroscopic parameters of eleven of them have been updated in the GEISA/IASI spectral range (599–3001  $\text{cm}^{-1}$ ); three of them (CO, O<sub>2</sub>, OCS) are unchanged in this specific spectral region (however O<sub>2</sub> and OCS are updated elsewhere in the database). Recently, in conjunction with the 2009 update, to aid the atmospheric chemistry and climate monitoring capabilities of IASI soundings, 6 additional molecular species have been included in the GEISA/IASI archive, i.e., HCN, NH<sub>3</sub>, HCOOH, C<sub>2</sub>H<sub>4</sub>, CH<sub>3</sub>OH and H<sub>2</sub>CO (this list is not final or complete) as well as PAN cross-sections. All these molecular species have been updated in GEISA-09 and PAN is a new crosssection entry. HCN is almost totally new in GEISA-09.
- (b) Related with spectroscopy requirements for space studies of outer planets and Titan, as documented in Ref. [8], important subsequent updates have been done in GEISA-09. Data on complementary new molecular species, to those already archived in GEISA, have been included in GEISA-09, i.e., C<sub>6</sub>H<sub>6</sub> (individual lines and cross-sections as well) and C<sub>2</sub>HD; among the updated molecules are especially: C<sub>4</sub>H<sub>2</sub>, HC<sub>3</sub>N, C<sub>2</sub>H<sub>6</sub>, C<sub>2</sub>H<sub>2</sub>, C<sub>2</sub>H<sub>4</sub>, HCN, C<sub>3</sub>H<sub>4</sub>, CH<sub>3</sub>CN. Also HNC is introduced as a new molecular species. HCN and HNC molecules are of great astronomical interest. They have been observed in many galactic and extragalactic objects, ranging from circum-stellar masers through interstellar clouds to planetary atmospheres.

Even regularly updated and evolving spectroscopic databases, such as GEISA, still have their limitations and shortcomings; these have to be continuously corrected or improved upon to meet the requirements of a diverse group of users. Detailed examples, but a non-exhaustive list of these requirements was given in Ref. [8]. Among those still not met we can underline:

- (a) From the "Summary Report on the Second IASI International Conference" [498]; http://smsc.cnes.fr/IASI, among the conclusions on IASI related RTM spectroscopy problems still to be solved, high priority was given to the investigation of areas mainly related to H<sub>2</sub>O, CO<sub>2</sub> and CH<sub>4</sub>, summarized as follows:
  - For H<sub>2</sub>O (highest priority): Review of the accuracy of line widths (could be more important than the intensities); review of the continuum in the short wave window region (i.e., band 3; spectral range 2000–2760 cm<sup>-1</sup>); measurements of widths and shifts should be made (with temperature dependence if possible).
  - For CO<sub>2</sub>: The inconsistency between CO<sub>2</sub> v<sub>2</sub> and CO<sub>2</sub> v<sub>3</sub> bands bias by improving the CO<sub>2</sub> spectroscopy in the v<sub>3</sub> band.
  - For CH<sub>4</sub>: Improvement of the methane spectroscopy, introducing line mixing.

It has to be noted that, since this report and GEISA-09 have been issued, Toth et al. [499] revisited  $H_2^{16}$ O line strengths in the  $v_2$  and  $2v_2 - v_2$  bands at 6  $\mu$ m. These results will be considered for the next GEISA Edition.

(b) In the frame work of space studies of outer planets and Titan: besides the 2009 updates described above, the description of data that remain to be obtained and implemented in GEISA, as given in Ref. [8], is still available.

Finally, in terms of outstanding spectroscopy issues, among the most important actions already underway, but which must be reinforced and maintained, is the necessary validation of archived spectroscopic data. This and other activities will be performed in conjunction with the recently started VAMDC European project (http://www. vamdc.eu). This project aims at building an interoperable e-Infrastructure for the exchange of atomic and molecular data [500].

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## Appendix A. List of acronyms

	4A	Atlas Automatisé des Absorptions								
		Atmosphériques								
	4A	Automatized Atmospheric Absorption Atlas								
	4A/OP	4A/O Perational release								
	ACE	Atmospheric Chemistry Experiment								
		P Aerosol Composite								
	AFGL	Air Force Geophysics Laboratory								
	AGB	Asymptotic Giant Branch								
	AIRS	Advanced InfraRed Sounder								
	ALMA	Atacama Large Millimeter/submillimeter Array								
	ARA	Atmospheric Radiation Analysis								
	BIRA/IASE	3 Institut d'Aéronomie Spatiale de Belgique/								
	Belgian Institute for Space Aeronomy									
	BEAMCAT	۲ BErnese Atmospheric Meta Catalog Access								
		Tool								
	CDMS	Cologne Database for Molecular Spectroscopy								
	CDSD	Carbon Dioxide Spectroscopic Databank								
	CIRS	Composite InfraRed Spectrometer								
	CNRS	Centre National de la Recherche Scientifique								
		(France)								
	CNES	Centre National d'Etudes Spatiales (France)								
	CSE	Circum Stellar Envelope								
	CW-CRD	S Continuous Wave-Cavity Ring Down								
		Spectroscopy								
	DU	Denver University								
	ENVISAT	ENVIronmental SATellite								
	EPS	European Polar System								
	EOS-aqua	Earth Observing System-water								
	EU	European Union								
	EUMETSA	T European Organisation for the Exploitation of								
		Meteorological Satellites								
	FT	Fourier Transformed								
	FTIR	Fourier Transformed InfraRed spectroscopy								
	FTS	Fourier Transform Spectrometer								
	GADS	Global Aerosol Data Set								
	GEISA	Gestion et Etude des Informations Spectroscopi-								
		ques Atmosphériques; Management and study								
		of Atmospheric Spectroscopic Information								
	GOME	Global Ozone Monitoring Experiment								

- GSMA Groupe de Spectroscopie Moléculaire et Atmosphérique (France) HITRAN HIgh-resolution TRANsmission molecular
- absorption data base
- HTDS Highly spherical Top Data System
- HWHM Line Half-width at Half Maximum (line broadening)
- IASI Infrared Atmospheric Sounder Interferometer
- ICB Institut Carnot de Bourgogne
- INSU Institut National des Sciences de l'Univers (France)
- IPSL Institut Pierre Simon Laplace
- ISM Inter-Stellar Medium
- IAO Institute of Atmospheric Optics (Russia)
- IR InfraRed IRS InfraRed Spectrograph
- ISO Infrared Space Observatory
- ISSWG IASI Sounding Science Working Group
- IUPAC International Union of Pure and Applied Chemistry
- JPL Jet Propulsion Laboratory (USA)
- LADIR Laboratoire de Dynamique, Interaction et réactivité (France)
- LESIA Laboratoire d'Etudes Spatiales et d'Instrumentation en Astrophysique (France)
- LISA Laboratoire Inter-Universitaire des Systèmes Atmosphériques (France)
- LITMS Laboratory for Information Technologies and Mathematical Simulation (Russia)
- LMD Laboratoire de Météorologie Dynamique (France)
- MIPAS Michelson Interferometer for Passive Atmospheric Sounding
- Metop Meteorological operational satellite
- MOPD Maximum Optical Path Difference
- NASA National Aeronautics and Space Administration (USA)
- NCAR National Center for Atmospheric research (USA) NIR Near InfraRed
- NRC National Research Center (Canada)
- NIST National Institute of Standards and Technologies
- OPAC Optical Properties of Aerosols and Clouds
- PAH Polycyclic aromatic hydrocarbon
- PAN PeroxyAcetyl Nitrate
- PNNL Pacific Northwest National Laboratory (USA)
- RTM Radiative Transfer Modeling
- SCIAMACHY SCanning Imaging Absorption spectroMeter for Atmospheric ChartograpHY
- SCISAT-1 Scientific Satellite-1
- SOPHIA Stratospheric Observatory for Infrared Astronomy SPCAT Spare Parts Catalog Software
- S&MPO Spectroscopy & molecular properties of Ozone SST Spitzer Space Telescope
- UCL Université catholique de Louvain (Belgium)
- ULB Université Libre de Bruxelles (Belgium)
- UV Ultra Violet
- VAMDC Virtual Atomic and Molecular Data Centre
- VIMS Visible and Infrared Mapping Spectrometer VIS Visible

Appendix B. Description of the format used for the line parameters archive in the 2009 edition of GEISA (http:// ether.ipsl.jussieu.fr/etherTypo/?id=1306)

The GEISA-09 individual line list sub-database includes 31 spectroscopic line parameters corresponding to 252 characters record per entry, as described in Table 11: spectroscopic parameters symbolic field names are in the first line, and the associated field lengths and FORTRAN format descriptors in lines 2 and 3, respectively. Standard missing values, as adopted for each parameter and for the whole database, are detailed in line 4. Those values are mainly negative; blank characters correspond to missing transition quantum number identifications and to internal GEISA code as well; value "0" have been attributed to non identified field L (HITRAN isotopologue number). The description of each field is given at the bottom of the table.

Some modifications have been brought to the GEISA-03 [7–9] (http://ether.ipsl.jussieu.fr/etherTypo/?id=1072) format, i.e.,

- Lengths of fields *E*1, *E*2, *E*3, *E*4, *N*, *O*, *N'*, *O'* have been extended.
- Fields *P* and *Q*, related with specific HITRAN internal information, have been suppressed. Field *P*, uncertainty codes for wavenumber, intensity and halfwidth, has been replaced in GEISA by effective values of the errors. Field *Q*, HITRAN indices for lookup of references for wavenumber, intensity and halfwidth has no correspondence in GEISA which does not include reference numbers among its line parameters.

It has to be noticed that:

- Fields *K* and *L* are HITRAN-08 [13] specific, for the users' easier possible interface between the two databases. These fields information content, i.e., molecule number (*K*) and isotope number (*L*) as in HITRAN, makes it possible to apply a biunivoque correspondence with GEISA related fields (*I*) and (*G*), respectively. It has to be noted that field (*L*) corresponds, in HITRAN, to the isotopologue fractional abundance code; in GEISA, field (*G*) represents the code associated with the isotope chemical formula; these specificities are handled in the software making it possible to convert the format of a one of the database into the one of the other.
- Value in field *M* is given only if directly provided by the author of the spectroscopic line file. This field has been newly appended in GEISA-09.

Detailed description of fields *E*1 and *E*2, specific of each molecule, is given on the GEISA distribution WEB site at, http://ether.ipsl.jussieu.fr, for database interactive software use facilities. The shift of the positions of certain already existing fields (such as field *R*), as consequence of modifications in the format of GEISA-09 since the GEISA-03 one, has to be noted.

#### Table 11

Fields of the format for GEISA-09 individual line list sub-database.

Parameter	Α		В		С	D	i	E1	E2	E3	E4	F	G
Field length FORTRAN descriptor Undefined values	12 F12.6 - 0.99999	99	11 1PD11.4 -9.9999D-	01	6 0PF6.4 0.9999	10 F10.4 0.99		25 A25	25 A25	15 A15	15 A15	4 F4.2 0.99	3 I3 - 99
Parameter	Ι	J	K	L	Μ		Ν	0		R	A'	B'	
Field length FORTRAN descriptor Undefined values	3 13 99	3 A3	2 12 -9	1 I1 0	10 1PE10.3 – 9.999E	-01	7 0PF7.4 9.9999	9 F9.6 9.99	99999	6 F6.4 - 0.9999	10 F10.6 0.999999	11 1PD1 - 9.9	1.4 9999D – 01
Parameter	С	F	0′		R'	N'	S	S'		Т	T'	U	U′
Field length FORTRAN descriptor Undefined values	6 0PF6.4 0.9999	4 F4.2 - 0.99	9 F9.6 9.999	9999	6 F6.4 0.9999	7 F7.4 9.9999	4 F4.2 - 0.99	4 F4.2 - 0.		8 F8.6 - 0.999999	8 F8.6 0.999999	4 F4.2 - 0.99	4 F4.2 -0.99

A: Wavenumber of the line v (cm<sup>-1</sup>).

*B*: Intensity of the line  $I(\text{cm}^{-1}/(\text{molecule cm}^{-2}))$ .

*C*: Air broadening pressure half-width  $\alpha$  (HWHM)(cm<sup>-1</sup> atm<sup>-1</sup>).

*D*: Energy of the lower transition level E' (cm<sup>-1</sup>).

 $E_i$  (i=1,2,3,4): Transition quantum identifications for the lower and upper state transition (unitless).

*F*: Temperature dependence coefficient *n* of the air broadening half-width (unitless).

G: Identification code for isotopologue as in GEISA (unitless).

*I*: Identification code for molecule as in GEISA (unitless).

J: Internal GEISA code for the data identification (unitless).

K: Molecule number as in HITRAN [13] (unitless).

*L*: Isotopologue number (1=most abundant, 2=second, etc.) as in HITRAN [13] (unitless).

*M*: Einstein *A*-coefficient  $(s^{-1})$ .

*N*: Self-broadening pressure half-width (HWHMself) ( $cm^{-1} atm^{-1}$ ).

O: Air pressure shift of the line transition  $(cm^{-1} atm^{-1})$ .

*R*: Temperature dependence coefficient of the air pressure shift (unitless).

A': Estimated accuracy on the line position  $(cm^{-1})$ .

*B*': Estimated accuracy on the intensity of the line  $(cm^{-1}/(molecule cm^{-1}))$ .

*F*: Estimated accuracy on the temperature dependence coefficient of the air broadening half-width (unitless)

*O*': Estimated accuracy on the air pressure shift of the line transition  $(cm^{-1} atm^{-1})$ .

*R*': Estimated accuracy on the temperature dependence coefficient of the air pressure shift (unitless).

*N*': Estimated accuracy on the self-broadened ( $cm^{-1} atm^{-1}$ ).

S: Temperature dependence coefficient of the self-broadening half-width (unitless).

S': Estimated accuracy on the temperature dependence coefficient of the self-broadening half-width (unitless).

*T*: Self-pressure shift of the line transition  $(cm^{-1} atm^{-1})$ .

*T*: Estimated accuracy on the self-pressure shift  $(cm^{-1} atm^{-1})$ .

*U*: Temperature dependence coefficient of the self-pressure shift (unitless).

U': Estimated accuracy on the temperature dependence coefficient of the self-pressure shift (unitless).

 Table 12

 Complementary description of codes for new isotopologues or molecules since former GEISA editions

Molecule	Molecule code	Isotopologue code	Formula
N <sub>2</sub> O	4	458	14N14N18O
		548	<sup>15</sup> N <sup>14</sup> N <sup>18</sup> O
		556	<sup>15</sup> N <sup>15</sup> N <sup>16</sup> O
CH₃D	23	312	<sup>13</sup> CH <sub>3</sub> D
HCN		224	D <sup>12</sup> C <sup>14</sup> N
CH₃Br	43	79	<sup>12</sup> CH <sub>3</sub> <sup>79</sup> br
		81	<sup>12</sup> CH <sub>3</sub> <sup>81</sup> br
CH₃OH	44	216	<sup>12</sup> CH <sub>3</sub> <sup>16</sup> OH
NO <sup>+</sup>	45	46	$^{14}N^{16}O^{+}$
HNC	46	142	H <sup>14</sup> N <sup>12</sup> C
C <sub>6</sub> H <sub>6</sub>	47	266	<sup>12</sup> C <sub>6</sub> H <sub>6</sub>
$C_2HD$	48	122	<sup>12</sup> C <sub>2</sub> HD
CF <sub>4</sub>	49	291	<sup>12</sup> C <sup>15</sup> F <sub>4</sub>
CH₃CN	50	234	$^{12}\text{CH}_3^{12}\text{C}^{14}\text{N}$

## Appendix C. New molecules and isotopologues in GEISA-09

Description of new molecule and isotopologue codes in GEISA-09 is given in Table 12. The molecule names and associated codes are in the two first columns; for each molecule, the isotopologue codes and the corresponding detailed formula are in columns 3 and 4, respectively.

#### Appendix D. Supplementary materials

Supplementary data associated with this article can be found in the online version at doi:10.1016/j.jqsrt.2011. 06.004.

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