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| Affinity and selective aroma traits Tiziana Mariarita Granato ^a DeFENS, University of Milan, Italy ^b Dipartimento di Agraria, University of Naph | ity of plant proteins fo ^a , Pasquale Ferranti ^b , Stefar ^{15 Federico II, Portici, Italy} | or red wine componer nia Iametti ^{a,} *, Francesta Be | ns relevant to color and |
| ARTICLE INFO | ABSTRACT | | |
| Keywords: Red wine Fining Plant proteins Wine aroma Wine color Anthocyanidins | The effects of fining wit wine and wine aged for controls. Color traits and and of aroma-related con ficiency, although with d zymatic hydrolyzates dii cyanidins complexes, and ponents upon fining were thocycanidin components wine ageing. | h various plant proteins were assessed i twelve and twenty-four months, and im fining efficiency were considered, along mounds or either varietal or fermentativ listinct fast sics, and haf similar effects of freedoh these ability to interact with a virtual some arome components of fermere e any motied or absent. Effects of all th were most enviceable in young red w | on Aglianico red wine, using both the young cluding wine unfined or fined with gelatin as with the content of various types of phenolics: e origin. All agents had comparable fining ef- n wine color. Individual plant proteins and en- some anthocyanins, with specific proantho- ntative origin. Changes in varietal aroma com- te fining agents tested in this study on the an- ine, and decreased markedly with increasing |
| 1. Introduction Flavonoids are important com | ponents of grapes, essential to the | 1999; Vincenzi, Panighel, Ga the potential impact of fining pounds of varietal origin (Lub Ugliano, Gambuti, Genovese, | zzola, Flamini, & Curioni, 2015) and on agents of various nature on aroma com- bers, Charpentier, & Feuillat, 1996; Moio, & Piombino, 2004; Nasi, Ferranti, Amato |

rationaux are responsible for the color and astringency cred wines as well as for the yellow hue of oxidized white wines, the are also in-volved in the development of haze and precipitary adone with other technological problems, such as clogging of fill action membranes and adsorption on surfaces (Smith, Mcrae, & Binden, 2015). The reactions nator role in changes neversion of grape anadsorption on surfaces (Smith Mcrae, & Bindon, of anthocyanins and proanthocyanidins play a n taking place during wine ageing, that include on thocyanins to other pigments responsible and tonality, and for decreased astring for cha ency (Ca sa & Harbertson 2014).

and tominy, that not declease intrinsitive (clusted a finite resol, 2014). The main purpose of using a protein-based fining agent is to soften red wines, and is reportedly related to the efficiency of proteins in complexing the phenolic communic in the wine and in their removal through precipitation (Gonzalex, Neves, Favre, & Gil, 2014; Sarni-Manchado, Deleris, Avallone, Cheynier, & Moutounet, 1999). Similar studies have been carried out on the relevance of the inter-action among phenolics and aroun components (Dufour & Bayonove,

& Chianese, 2008; Volley, Lamer, Dubois, & Feuillat, 1990).

Commonly used protein fining agents include gelatin, casein, egg albumin, and proteins of plant origin able to replace the animal ones. In response to winemakers' interest in replacing animal-derived fining agents with plant-based products, plant proteins were investigated as possible wine fining agents (Lefebvre et al., 2000). Following studies included a variety of plant-derived proteins in both intact and partially hydrolyzed form, (Gambuti, Rinaldi, & Moio, 2012; Gazzola, Vincenzi, Marangon, Pasini, & Curioni, 2017; Marchal, Marchal-Delahaut, Lallement, & Jeandet, 2002; Maury, Sami-Manchado, Lefebvre, Cheynier, & Moutounet, 2003; Noriega-Dominguez, Duran, Virseda, & Marin-Arroyo, 2010; Simonato, Mainente, Selvatico, Violoni, & Pasini, 2013; Tschiersch, Nikfardjam, Schmidt, & Schwack, 2010). Fining effi-ciency of all the tested plant proteins and of enzymatically prepared protein hydrolysates was related to their capability to precipitate con-densed species, which was found - in most cases - to relate to the pro-

Abbreviations: LC-MS, liquid e , tatography-mass spectrometry; ESI, electrospray ionization; TFA, trifluoroacetic acid; TIC, total ion current; OPCs, oligomeric proanthocyanidim complexes.

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https://doi.org/10.1016/j.foodchem.2018.02.085 Received 21 April 2017; Received in revised form 15 February 2018; Accepted 15 February 2018 Available online tex 0308-8146/ © 2017.

tein surface hydrophobicity (Granato, Nasi, Ferranti, Iametti, & Bonomi, 2014; Granato et al., 2010; Le Bourvellec & Renard, 2013). Molecular features of the target compounds (size, hydrophobicity, degree of polymerization and formation of adducts) and their concentration were major factors in the effectiveness of the various proteins. All these factors relate to the characteristics of the original grapes and to the winemaking conditions (Castillo-Sanchez et al., 2008; Gonzalez-Neves et al., 2014; Karamanidou, Kallithraka, & Hatzidimitrio, 2011; Le Bourvellec & Renard, 2013; Smith et al., 2015). Other recent studies also reported on the impact of combined treatments (e.g., separate and sequential use of bentonite and proteins) on some relevant traits in young red wines (Ben Aziz, Mouls, Fulcrand, Douieb, & Hajjaj, 2017), or on the influence of other wine components – such as sugars and polyols – on the fining process (Maury, Sarni-Manchado, Poinsaut, Cheynier, & Moutounet, 2016).

Here we evaluated the effect of plant-derived proteins on the same Aglanico red wine either right after winemaking or after aging for twelve or twenty-four months. High-resolution separative techniques and mass-spectrometry based approaches were combined to compare fining-related changes in the qualitative and quantitative anthocyanidin profiles and – in the case of young wine – in the profile of volatiles. This experimental plan allowed the effects of the interaction between fining proteins and some of the relevant compounds in both young and aged wine to be evaluated, with the final goal of addressing the nature and role of the possible molecular determinants of specific interactions between the fining agents and molecules relevant to sensory traits.

In fining trials were included also two preparations obtained from partial enzymatic hydrolysis of pea proteins, in order to assess whether the modification of their size and of the number/accessibility of hydrophobic sites on their surface (Bonomi, Mora, Pagani, & Iametti, 2004; Granato et al., 2010; Nakai & Li-Chan, 1988), could increase their fining efficiency while minimizing possible negative effect on wine quality, as suggested by previous studies on partially hydrolized cereal proteins (Marchal et al., 2002; Maury et al., 2003; Tschiersch et al., 2010).

2. Materials and methods

2.1. Experimental wines

The young red wine used in this study (Aglia produced in 2012 by Cantine Tora (located in uso, Be vento, a DOC area in the Campania Region), from A local vineyards. The same wine was used for after twelve and twenty-four months of agoing apes grown in tests carried out ad the following chemical characteristics, as assessed sta OIV methods rd (International Organisation of Vine and 15.10% (v/v); titratable acidity, 6.21ine, 20): alcohol content ric acid); volatile 3.31; fre acidity, 0.29 g/L (as acetic acid); sulfur dioxide, 18 mg/L; total sulfur dioxide, 30 mg/L.

2.2. Wine fining trials

Fining agents for experimental activities included commercially available protein extracts from so bean and pea, lentil flour, and wheat gluten proteins (all from Proteiti Gianni, Milan, Italy). These extracts had a protein conten ranging from 82 in the case of legume proteins to 91% in the case of glutor, all expressed on a dry weight basis. Water content was in the 6-8% range (w/w). The study also included two experimental preparations obtained from enzymatic hydrolysis of pea proteins (PH1 and PH2), that were prepared by treating pea protein isolates with two different food grade enzymes. PH1 was prepared by

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using a trypsin-like protease (Amano N, PRZ1250448N, Amano Enzyme Europe Limited, Chipping Norton, UK), whereas a chymotrypsin-like enzyme (Europa Protease 2, Europa Bioproducts Ltd, Wicken, UK) was used to prepare PH2. In both cases, the hydrolysis was carried out at 37 °C for 2 h, and arrested by lowering the pH to 2.5 with HCI. Insoluble materials were recovered, by nortrifugation (10 min at 3000xg), washed with water, and lyophilize. MS characterization of PH1 and PH2 confirmed hydrolysis util to the enzymes, with disapearance of large-sized proteins and a similar age distribution of the hydrolytic products. Peptide size was centered around 20000Da in both PH1 and PH2, but different polynephtics were present in PH1 and PH2 as a consequence of the different polynephtics.

both PH1 and PH2, but different periods were present in PH1 and PH2 as a consequence of the different specificity of the enzymes. Fining tests were carried ut by adding protein fining agents in plastic-stoppered 100-mL cylinders, that were completely filled with about 120 mL of wine to ammunize an exposure and ensuing oxidation. Plant-derived proteins were need ut the maximum dosage commonly employed for animal proteins (20 g/hl for gelatin and egg proteins), as done in previous strutus (Sneado et al., 2014). An untreated sample was used as a neartive bontrol, and a cold-water soluble gelatin (PUL-VICLAR S, Enarth LPU) was included in this study for comparative purposes. Gelatin was used at the same concentrations (20 g/hl) used for plant proteins.

2.3. Clarification kinetics and spectrophotometric measurements

Tarbidity are measured in Nephelometric Turbidity Units (NTU) using adurbidimeter (LP 2000, Hanna Instruments). Optical density, unoidy, any the volume of lees were measured in duplicate 1, 4, 10, 20, 32, 105–58 and 168 h after the addition of fining agents. Kinetics were studied at room temperature ($20^{\circ}C\pm 2^{\circ}C$). Absorbance at 420, 520, apd 620 nm was determined in an UV-VIS spectrophotometer (mod. 1601, Shimadzu). All analyses were carried out in triplicate.

4. Structural characterization of phenolic compounds after wine fining

Phenolic compounds were isolated by loading 5 mL of wine on a Sep-Pak C18 cartridge (Waters, Milford, MA, USA) previously condi-tioned by sequential washing with 5 mL of methanol and 5 mL of water. The cartridge was washed three times with 10 mL of water and the samples eluted with 70% aqueous ethanol containing 0.1% TFA. Samples were stored at -18 °C until used. LC/MS analysis was carried Samples were stored at -15 C min used. E./ MS analysis was carlied out by means of a LC/MS single quadrupole instrument (HP1100-MSD, Agilent Technologies, Santa Clara, CA, USA) by using a C18 column (Vydac, Hesperia CA, USA; 2.1 * 250 mm). The eluant was 0.1% (v/v) TFA in HELC-grade water (solvent A) and 0.1% (v/v) TFA in ace-tonitrile (solvent B). Oligomeric proanthocyanidins complexes (OPCs) were separated at a constant flow-rate of $0.2 \, \text{mL/min}$, with a linear gradient of solvent B in the following proportions (v/v): 4 min, 0% B; 4 -14 min, 0-18% B; 14–22 min, 18–28% B, 22–24 min, 28% B; 24 -26 min, 28-60% B; 26-27 min; 60-80% B; and 27-30 min, 80-100% B. The total run time was 30 min. Detection was carried out at 280 and 520 nm. For LC/ESI-MS analysis, proanthocyanidins were characterized according to the conditions used for wine model solutions (Granato et al., 2010). To increase sensitivity of ESI-MS measurements, the samples were assayed twice, scanning in the positive ion mode from m/z 100 to 1000, and from m/z 1000 to 2000, at a scan rate of 4.90 s per scan and 0.1 s inter-scan delay. The source temperature was 180 °C. The capillary voltage was 3.6 kV and the cone voltage was maintained either at A0 or 25 V, according to different experiments. N₂ was used as both drift d previous studies (Granato et al., 2014).

2.5. Determination of volatile compounds in wines after fining

Analysis of volatile compounds (including either varietal molecules, such as terpenes, and volatile phenols or non-varietal compounds, such as acids, esters, aldehydes, lactones, etc.) was performed by solid phase micro-extraction (SPME) and gas chromatography-mass spectrometry (GC/MS, Nai, et al., 2008). SPME holders and fibers were from Supelo (Aldrich, Bornem, Belgio). SPME fibers were immersed in the headspace of the samples (120 mL) of wine until equilibrium was reached. The internal standard 1-octanol (0.500 mg/L) was added to the wine before the extraction. Thermal desorption of the analytes from the fiber inside the GC injection port was carried out in the split mode (1/10) at a desorption temperature of 250 °C during 1 min. For GC/MS analysis, all samples were analyzed with an HP 7890 gas chromatograph coupled to a 5975C quadrupole HP mass spectrometer. The gas chromatograph was equipped with an HP-5 capillary colume (30 m \times 0.32 mm ID), with He as the carrier gas. For analysis of aroma compounds, the GC oven temperature was increased from 40°C (held for 7 min) to 180 °C, at 5 °C/min. The mass spectrometer operated in eletron mode (EI, 70 eV) and scans covered the 45–350 m/z range. In other cases, a SIM method was used (for terpene compounds m/z 93, 12, 136). The identification of odorous components was done by referring to the National Institute of Standards and Technology (NIST) library and by comparison with suitable standards, as reported elsey where (Nasi et al., 2008). Quantitative determinations are obtained by means of calibration curves in the concentration ranges typical of wines for each compound. Seven concentration levels and five replicates per level were used – in the range of verified linearity – for calibration terposes.

2.6. Statistical analysis

Analysis of variance (ANOVA) was performed by utilizing a graphics XV version 15.1.02 (StatPoint Inc., Warrenton, V/ US Samples were used as factor. When the factor effect was found is significant ($p \le 0.05$), significant differences among the verse means were determined using Fisher's Least Significant Dincre (LSD) tet.

3. Results and discussion

3.1. Fining efficiency of plant proteins

Fig. 1 shows the time course of wine dentification for the various proteins. All of the plant proteins tested nere are nooluble in wine and they have to be eliminated by decanting or filtering after fining. Indeed, turbidity of wines increased after a dition or plant proteins but slowly stabilized to values close to that of wine treated with gelatin. After 7 days, all treated wines had turbidity ranging between 5.7 and 22.9 NTU.

Lentil flour, PH1 and soy patiellies usere – in this order – the most efficient fining agents in the young of wine, giving final NTU values of 5.7 ±0.2, 9.9 ±0.3, and 10.31 ±0.7, respectively, all lower than the final NTU value in the obstrop un-fined wine (15.5 ± 0.2), at difference with other plant-based fining gents. The hydrolyzed pea proteins (PH1 and PH2) had a slower fining rate than other matrices, probably because of the longer thin userof d for flocculate formation from smaller polypeptides, but at the end of the treatment their fining efficiency was better than that of the corresponding intact proteins (final NTU values; 9.9 ± 0.3 for PH1; 17.4 ± 1.2 for PH2; 18.5 ± 0.8 for intact pea proteins) and comparable to that of gelatin (final NTU, 14.4 ± 1.0). At contrast, gluten gave a marked decrease in turbidity in the first 24 h but, at



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e course of *viglianico* red wine clarification. All treatments were carried out at **nu.**

end of the process (i.e., after seven days), gluten gave the highest a VrU observed in this study (22.90 \pm 2.1). It has to be noted that the results reported here were obtained by

It has be noted that the results reported here were obtained by bing similar amounts of the various fining agents, thus allowing a straßytforward comparison of their relative efficiencies and of the time course of the fining process. Further studies will be needed to assess phether the fining efficiency or the time course of fining may change when varying the amounts of individual fining agents.

3.2. Interaction of plant proteins with proanthocyanidins

The removal of proanthocyanidins and of their derivatives by various plant proteins with respect to controls was evaluated through a quantitative LC-ESI MS analysis, that gave the results summarized in Fig. 2. Lentil and soy proteins – closely followed by PH1 – showed the highest ability to interact with proanthocyanidins in red wine, with a decrease in residual proanthocyanidin ranging from 40% (for monomeric compounds) to 70% (for proanthocyanidin trimers).



Fig. 2. Percentage loss of individual low molecular weight proanthocyanidins in finet Aglianico red wine, as assessed by LC-RSI MS analysis. Results are expressed in com parison to unfined wine. Values in the same group with identical superscripts are not significantly different (LSD; $p \le 0.05$).

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A different efficiency in the removal of monomeric and dimeric proanthocyanidins was observed for soy and gluten proteins in different wines. Gluten proteins had the lowest impact on proanthocyanidins in red wine, although they interacted extensively with proanthocyanidins in white wine (Granato et al., 2014). Whether this originates from limited competition among various protein-adsorbed species (in white wine) rather than from some co-operative binding (in red wine) remains to be assessed. The nature and amount of surface hydrophobic regions in the various proteins also may come into play (Granato et al., 2010), as soy proteins showed a behavior opposite to that of gluten proteins when comparing red and white wines. A different affinity for proanthocyanidins was observed for pea pro-

A different affinity for proanthocyanidins was observed for pea proteins and their hydrolysis products. Pea proteins were the least effective removing agent, giving a 3–5% decrease in catechin- epicatechin and 15–22% decreases in OPC dimers and trimers. In contrast, PH1 showed a much higher ability to interact with OPCs than the native proteins, giving a decrease ranging from 30% (for monomeric species) to more than 70% (for dimers and trimers). Also, PH1 was noticeably more effective than PH2. These differences are likely related to the different molecular size, hydrophobicity, and conformation of pea proteins and of their hydrolytic fragments. In particular, hydrolysis with chimotrypin-like enzymes (acting on hydrophobic regions, and used for preparation of PH2) gave products with a lower fining ability than that observed for PH1 (prepared by using a trypsin-like enzyme, and thus retaining in intact form its hydrophobic sequences).

These observations confirm that formation of insoluble (and, therefore, removable) protein-OPC complexes is not governed simply by hydrophobic forces, as the formation of insoluble complexes – and the ensuing clarifying effects – requires multiple and simultaneous types of interactions, each of them with distinct thermodynamic aspects relate to protein/protein and protein/phenolics affinity, and are intertwined with the kinetics of individual steps in the formation of insoluble protein aggregates incorporating the target species. In this frame, one possible explanation of the better behavior of hydrolyzed pea proteins could impair accessibility to phenolics binding sites, that are expected to be hydrophobic in nature (Granato et al., 2010). and are likely burne thus drophobic in nature (Granato et al., 2010). and are likely burne thus disclosed the native ones was that the large native proteins could impair accessibility to phenolics binding sites, that are expected to be the elective hydrolysis with enzymes that do not act on hydrophobic residues (as done for PH1) could improve the onessibility of hydrophobic binding sites, and favor cross-interaction among proteins and the eventual precipitation of insoluble comblexe.

3.3. Effects of plant proteins on anthocyanins and chromatic characteristic

Many factors responsible for the color to be done could be affected by fining treatments, with particular reference to anthocyanins and to additional components that cause both a violet shift in color (bathochromic effect) and an increase in color intensity (hyperchromic effect) (Zhang, He, Zhou, Liu, & Yuan, 2015). Changes in the red wine color related to the fining treatments ore analyzed by both color measurements and UV/Vis spectrophotometry at 420, 520, and 620 nm (to include the blue component on young red wines). Although all fining agents decreased the color related by changes in color intensity index and tonality during whe reatments (Figs. 1S and 2S). Absorhance changes at specific wavelengths were expected to provide information about the influence of fining proteins on color. In general, all treatments slightly lowered the intensity of the three color components. The effect on the yellow and red components was slightly more pronounced than that on the blue one (Fig. 2S), that reportedly relates to Food Chemistry xxx (2018) xxx-xxx

the association between pigments and co-pigments, and involves other substances, mainly derivatives of the flavonol and flavone subgroups (Boulton, 2001). However, none of the observed variations in tonality was statistically significant.

The anthocyanin pattern in the various samples was outlined by carrying out a thorough anthocyanin profung, by means of a combined liquid chromatography/electrospray onization mass spectrometry (LC-ESI MS) approach. Forcy toou eartho yanin derivatives were identified and quantified, as listed in Table 1.

The anthocyanin derivatives identified in this study included adducts with pyruvic acid and acctaldence and derivatives formed by copigmentation. Copigmentation invoices complexation phenomena, generally at low bond energy hydrogen bonds and hydrophobic interactions), either between the various forms of anthocyanins or between anthocyanins and other – analy coloress – phenolic compounds, that results in changes in intrinsity of specific color traits. Among the copigmentation-relevant species, directed in this study and reportedly relevant to wine color cibranget al., 2015), the vinyl-phenol derivatives could have resulted from the decarboxylation of p-coumaric acid by yeast decarboxyl sees and by the consequent reaction with malvidin, either as a monoglucoside or as acylated monoglucosides (p-coumarylglucoside). Further rearmsgements could involve carbon 4 of the anthocyanin and the oxygen on carbon 5, leading to the formation of a new oxygen hetencyclic species, which is colorly uson oxidation. Afother results of pigments identified in Asilanico variants

Arother your of pigments identified in *Aglianico* young wine resured from the addition of pyruvic acid onto anthocyanins. Compared we have not provide the second second second second second quantum second second second second second second second quantum second sec

By comparing anthocyanin-related compounds in fined and control wines, it is evident that all protein fining agents were found to be more of less able to interact with anthocyanins, although the effects of interction were dissimilar for individual compounds. Indeed, the residual amount of some compounds was higher in the treated wine than in wine undergoing spontaneous settling. This was observed for delphinidin, peonidin, and malvidin-30-glucoside upon treatment with gluten and pea proteins, and for several co-pigmented anthocyanins in wine finde with lentil proteins. A possible explanation could be related to the capability of these fining agents to interact with pigments preventing them from become bound to the solids that were removed by the racking step after the spontaneous settling of reference wine.

The effects of the various fining agents on individual anthocyanins may be exemplified by taking into account changes involving some of the most abundant and most relevant species among those listed in Table 1. Taking malvidin-3-O-glucoside as an example, the data in Table 1 make it evident that the efficiency of removal decreased in the order: PH1 > SI = LE > gelatin > PH2, whereas GL and PI were ineffective. However, only soybean isolates, PH1 and gelatin were effective in removing relevant (and comparatively abundant) malvidin-related species, such as (ep/oct-mv3-O-glu, m-vi)nguiaaicol, and w-3-O-glu-8-ethyl-(ep/)cat. Lentil proteins were ineffective in removing these complex malvidin-related species, in spite of lentil proteins having the highest fining efficacy in terms of lowering NTU values after prolonged fining (see Fig. 1 and related comments).

3.4. Effects of fining agents on volatile composition

Our analysis of the impact of the use of plant proteins as fining agents in red wines was completed by characterizing the volatile compounds profile in the fined products by LC/MS (Nasi et al., 2008). As reported in the Supplementary Table 1S, *Aglianico* has a complex varietal aroma composition. Terpenes such as α -pinene, β -pinene,

Table 1 Mean values of concentration (mg/)) and standard deviations (n = 2) for anthocyanic phenolics belonging to different chemical families (monomeric anthocyanins, anthocyanin-flavan-3-ol adducts mediated by acetaldehyde, pyranoanthocyanins and hy-droxyphenyl-pyranoanthocyanins) as identified by HPLC-MS in young Aglianico wine either unfined or fined with various plant proteins (SI, soy protein isolate; LE, lentil flour; GL, gluten; PL, pea protein hydrolysate1; PH2, pea pro-tein hydrolysate2; GE, gelatin). RT, retention time; n.d.: not detected.

| | RT, min | m/z | Unfined | Fined wine | | | | | | |
|--|-----------|------|------------------|------------------|------------------|------------------|------------------|-----------------|------------------|------------------|
| | | | | SI | LE | GL | PI | PH1 | PH2 | GE |
| de-3-O-glu | 19.8 | 465 | 5.47 ± 0.09 | 2.12 ± 0.08 | 2.29 ± 0.03 | 5.74 ± 0.05 | 6.61 ± 0.03 | 2.37 ± 0.02 | 4.92 0.08 | 4.70 ± 0.01 |
| cy-3-O-glu | 20.5 | 449 | 0.40 ± 0.04 | 0.19 ± 0.00 | 0.20 ± 0.03 | 0.44 ± 0.01 | 0.37 ± 0.01 | 0.23 ± 0.02 | 0.24 ± 0.02 | 0.34 ± 0.04 |
| pe-3-O-glu | 21.90 | 463 | 4.69 ± 0.22 | 0.31 ± 0.15 | 3.01 ± 0.00 | 4.82 ± 0.19 | 5.41 ± 0.03 | 2.81 ± 0.05 | 3.69 ± 0.13 | 4.04 ± 0.09 |
| mv-3-O-glu | 21.90 | 493 | 77.08 ± 1.09 | 41.03 ± 0.35 | 41.91 ± 0.03 | 72.68 ± 0.15 | 87.18 ± 0.55 | 36.68 - 0.16 | 67.63 ± 0.25 | 60.87 ± 0.26 |
| pt-3-O-glu | 20.90 | 479 | 11.06 ± 0.45 | 5.59 ± 0.48 | 5.48 ± 0.28 | 11.39 ± 0.22 | 12.85 ± 0.05 | 5.44 ± 0.00 | 9.18 ± 0.00 | 10.15 ± 0.68 |
| Monoglucosides residual, % of unfined v | vine | | | 50 | 54 | 96 | 114 | 48 | 87 | 81 |
| pt-3-O-(6-O p-coumaryl)-glu | 26.90 | 625 | 0.91 ± 0.08 | 0.51 ± 0.03 | 0.98 ± 0.01 | 0.73 ± 0.04 | 0.62 ± 0.00 | 0.53 ± 0.02 | 0.53 ± 0.05 | 0.59 ± 0.04 |
| de-3-O-(6-O p-coumaryl)-glu | 25.70 | 611 | 0.43 ± 0.11 | 0.17 ± 0.03 | 0.63 ± 0.01 | 0.56 ± 0.08 | 0.43 ± 0.01 | 0.23 ± 0.00 | 0.23 ± 0.12 | 0.28 ± 0.04 |
| cy-3-O-(6-O p-coumaryl)-glu | 26.70 | 595 | 0.35 ± 0.08 | n.d. | 0.31 ± 0.01 | 0.27 ± 0.00 | 0.12 ± 0.01 | 0.10 ± 0.01 | 0.10 ± 0.00 | 0.12 ± 0.00 |
| pe-3-O-(6-O p-coumaryl)-glu | 28.11 | 609 | 1.69 ± 0.38 | 1.13 ± 0.25 | 1.68 ± 0.40 | 1.49 ± 0.05 | 1.16 ± 0.10 | 1.02 ± 0.16 | 1.02 ± 0.06 | 1.13 ± 0.08 |
| mv-3-O-(6-O p-coumaryl)-glu | 28.16 | 639 | 11.03 ± 1.25 | 6.91 ± 0.11 | 12.34 ± 0.70 | 10.99 ± 0.80 | 7.87 ± 0.95 | 6.50 ± 0.31 | 6.50 ± 0.45 | 6.57 ± 0.22 |
| p-cumaroyl glucosides residual, % of uni | ined wine | | | 60 | 58 | 97 | 111 | 60 | 71 | 87 |
| pt-3-O-(6-O p-caffeoyl)-glu | 28.16 | 641 | 1.06 ± 0.10 | 0.67 ± 0.00 | 1.14 ± 0.35 | 0.98 ± 0.21 | 0.81 ± 0.00 | 0.73 ± 0.40 | 0.73 ± 0.09 | 0.59 ± 0.13 |
| pe-3-O-(6-O p-caffeoyl)-glu | 27.90 | 625 | 0.17 ± 0.09 | 0.11 ± 0.05 | 0.17 ± 0.00 | 0.14 ± 0.00 | 0.11 ± 0.13 | 0.08 ± 0.07 | 0.08 ± 0.05 | 0.07 ± 0.00 |
| mv-3-O-(6-O p caffeoyl)-glu | 26.40 | 655 | 0.18 ± 0.03 | 0.15 ± 0.02 | 0.22 ± 0.01 | 0.24 ± 0.01 | 0.17 ± 0.03 | 0.13 ± 0.01 | 0.13 ± 0.02 | 0.16 ± 0.00 |
| cy-3-O-(6-O p-caffeoyl)-glu | 28.00 | 611 | 0.10 ± 0.01 | 0.12 ± 0.01 | 0.14 ± 0.01 | a.d. | 0.10 ± 0.00 | 0.16 ± 0.02 | 0.16 ± 0.03 | 0.14 ± 0.01 |
| p-caffeoyl glucosides residual, % of unfit | ned wine | | | 69 | 73 | 90 | 111 | 64 | 78 | 84 |
| mv-4-vinylphenol | 20.44 | 447 | 0.54 ± 0.22 | 0.16 ± 0.09 | 0.41 ± 0.01 | 0.24 ± 0.13 | 0.18 ± 0.06 | 0.13 ± 0.00 | 0.13 ± 0.04 | 0.19 ± 0.00 |
| mv-3-O-glu-4-vinylphenol | 27.28 | 609 | 0.32 ± 0.05 | 0.10 ± 0.00 | 0.27 ± 0.02 | 0.21 ± 0.01 | 0.07 ± 0.00 | 0.10 ± 0.01 | 0.10 ± 0.01 | 0.14 ± 0.00 |
| pe-3-O-glu-4-vinylguaiacol | 29.01 | 609 | 0.32 ± 0.00 | 0.22 ± 0.08 | 0.31 ± 0.00 | 0.26 ± 0.01 | 0.19 ± 0.03 | 0.20 ± 0.00 | 0.20 ± 0.00 | 0.21 ± 0.01 |
| mv-3-O-glu-4-vinylguaiacol | 28.16 | 639 | 11.10 ± 0.98 | 4.47 ± 0.94 | 12.15 ± 0.46 | 10.90 ± 1.09 | 7.87 ± 0.10 | 6.42 ± 0.55 | 6.42 ± 0.55 | 6.63 ± 0.23 |
| mv-3-O-glu pyruvic acid (vitisin A) | 25.89 | 561 | 0.24 ± 0.09 | 0.15 ± 0.01 | 0.10 ± 0.05 | 0.24 ± 0.05 | 0.22 ± 0.00 | 0.14 ± 0.06 | 0.14 ± 0.00 | 0.24 ± 0.10 |
| pe-3-O-glu pyruvic acid | 25.16 | 531 | 2.61 ± 0.17 | 1.52 ± 0.52 | 2.25 ± 0.12 | 2.02 ± 0.24 | 2.05 ± 0.21 | 1.40 ± 0.13 | 1.40 ± 0.14 | 1.41 ± 0.41 |
| de-3-O-glu pyruvic acid | 19.90 | 533 | 1.86 ± 0.35 | 1.29 ± 0.10 | 2.06 ± 0.44 | 1.72 ± 0.12 | 1.44 ± 0.22 | 1.22 ± 0.15 | 1.22 ± 0.01 | 1.41 ± 0.14 |
| mv-3-O-coumaroyl- glu pyruvic acid | 25.50 | 707 | 1.11 ± 0.41 | 0.16 ± 0.01 | 1.08 ± 0.12 | 0.95 ± 0.11 | 0.83 ± 0.09 | 0.22 ± 0.09 | 0.22 ± 0.00 | 0.19 ± 0.00 |
| (epi)cat-mv-3-O-glu | 19.80 | 781 | 2.44±0.52 | 0.75 ± 0.09 | 2.75 ± 0.10 | 2.66 ± 0.12 | 2.28 ± 0.21 | 0.65 ± 0.01 | 0.65 ± 0.15 | 0.83 ± 0.29 |
| (epi)cat-pe-3-O-glu | 19.80 | 751 | 0.25 0.08 | 0.04 ± 0.01 | 0.26 ± 0.03 | 0.23 ± 0.00 | 0.20 ± 0.00 | 0.06 ± 0.01 | 0.06 ± 0.01 | 0.05 ± 0.01 |
| (epi)cat-mv-3-O-couglu | 25.03 | 927 | 0.27 ± 0.05 | 0.06 ± 0.01 | 0.31 ± 0.01 | 0.28 ± 0.00 | 0.27 ± 0.01 | 0.05 ± 0.00 | 0.05 ± 0.02 | 0.05 ± 0.00 |
| di(epi)cat-mv-3-O-glu | 20.82 | 1069 | 0.15 ± 0.02 | 0.01 ± 0.01 | 0.14 ± 0.01 | 0.14 ± 0.00 | 0.12 ± 0.05 | 0.02 ± 0.00 | 0.02 ± 0.00 | 0.01 ± 0.01 |
| mv-3-O-glu-8-ethyl-(epi)cat | 23.97 | 809 | 4.92 ± 0.52 | 2.01 ± 0.11 | 5.93 ± 0.21 | 4.73 ± 0.35 | 4.14 ± 0.25 | 2.03 ± 0.44 | 2.03 ± 0.12 | 1.99 ± 0.45 |
| mv-3-O-glu-8-ethyl-(epi)cat | 24.58 | 809 | 8.27 ± 0.78 | 3.57 ± 0.23 | 8.90 ± 0.41 | 6.56 ± 0.35 | 6.14 ± 0.32 | 3.19 ± 0.35 | 3.19 ± 0.12 | 1.17 ± 0.09 |
| mv-3-O-glu-8-ethyl-(epi)cat | 25.04 | 809 | 2.82 ± 0.21 | 1.02 ± 0.04 | 3.26 ± 0.19 | 3.06 ± 0.15 | 1.48 ± 0.08 | 1.02 ± 0.12 | 1.02 ± 0.35 | 3.02 ± 0.17 |
| mv-3-O-glu-o-ethyl(epigallo)gallocat | 23/24 | 821 | 0.55 ± 0.01 | 0.10 ± 0.01 | 0.60 ± 0.05 | 0.46 ± 0.12 | 0.56 ± 0.22 | 0.13 ± 0.01 | 0.13 ± 0.05 | 0.10 ± 0.01 |
| mv-3-O-glu-o-ethyl(epigallo)galloeat | 25.00 | 821 | 0.37 ± 0.05 | 0.03 ± 0.01 | 0.31 ± 0.01 | 0.29 ± 0.01 | 0.32 ± 0.00 | 0.06 ± 0.00 | 0.06 ± 0.00 | 0.05 ± 0.01 |
| pe-3-O-coumaroyl-glu-8-ethyl-(epi)cat | 27.45 | 925 | 0.25 ± 0.06 | 0.04 ± 0.00 | 0.24 ± 0.01 | 0.17 ± 0.04 | 0.16 ± 0.03 | 0.08 ± 0.00 | 0.08 ± 0.03 | 0.07 ± 0.01 |
| mv-3-O-coumaroyl-glu-8-ethyl-(epi)cat | 27.45 | 955 | 2.63 ± 0.25 | 0.88 ± 0.30 | 2.26 ± 0.23 | 2.21 ± 0.16 | 1.80 ± 0.32 | 0.87 ± 0.15 | 0.87 ± 0.10 | 0.97 ± 0.26 |
| mv-3-O-glu-4-vinyl(epi)cat | 23.24 | 805 | 1.45 ± 0.35 | 0.71 ± 0.25 | 1.36 ± 0.15 | 1.17 ± 0.12 | 1.20 ± 0.15 | 0.71 ± 0.09 | 0.71 ± 0.05 | 0.68 ± 0.13 |
| mv-3-O-glu-4-vinyl(epi)cat | 25.00 | 805 | 0.87 ± 0.20 | 0.45 ± 0.05 | 0.88 ± 0.18 | 0.80 ± 0.15 | 0.78 ± 0.09 | 0.44 ± 0.02 | 0.44 ± 0.12 | 0.42 ± 0.19 |
| mv-3-O-glu-4-vinyl-di(epi)cat | 21.14 | 1093 | 0.01 ± 0.00 | 0.01 ± 0.00 | 0.02 ± 0.00 | n.d. | 0.01 ± 0.00 | n.d. | n.d. | n.d. |
| mv-3-O-glu-4-vinyl-di(epi)cat | 24.31 | 1093 | 0.09 ± 0.01 | 0.02 ± 0.01 | 0.09 ± 0.00 | 0.08 ± 0.01 | 0.09 ± 0.00 | 0.02 ± 0.00 | 0.02 ± 0.00 | 0.01 ± 0.01 |
| mv-3-O-cou-glu-4-vinyl-(epi)cat | 28.60 | 951 | 0.77 ± 0.10 | 0.31 ± 0.01 | 0.76 ± 0.00 | 0.61 ± 0.01 | 0.40 ± 0.02 | 0.30 ± 0.01 | 0.30 ± 0.02 | 0.34 ± 0.09 |
| mv-3-O-cou-glu-4-vinyl-(epi)cat | 29.30 | 951 | 0.19 ± 0.09 | 0.11 ± 0.01 | 0.26 ± 0.01 | 0.18 ± 0.01 | 0.18 ± 0.01 | 0.10 ± 0.00 | 0.10 ± 0.09 | 0.16 ± 0.02 |
| mv-3-O-ac-glu-4-vinyl-(epi)cat | 25.00 | 847 | 0.11 ± 0.01 | 0.05 ± 0.00 | 0.11 ± 0.01 | 0.13 ± 0.01 | 0.11 ± 0.01 | 0.08 ± 0.00 | 0.08 ± 0.00 | 0.03 ± 0.00 |
| mv-3-O-ac-glu-4-vinyl-(epi)cat | 26.29 | 847 | 0.41 ± 0.14 | 0.12 ± 0.01 | 0.37 ± 0.09 | 0.33 ± 0.01 | 0.30 ± 0.20 | 0.12 ± 0.11 | 0.12 ± 0.15 | 0.18 ± 0.01 |
| mv-3-O-(6-O-ac)-glu | 25.43 | 535 | 9.25 ± 0.90 | 6.11 ± 0.10 | 9.43 ± 0.45 | 8.51 ± 0.50 | 6.83 ± 0.20 | 5.31 ± 0.12 | 5.31 ± 0.15 | 6.15 ± 0.29 |
| mv-3-O-glu acetaldehyde | 23.15 | 517 | 1.91 ± 0.21 | 1.97 ± 0.20 | 1.93 ± 0.04 | 1.54 ± 0.12 | 1.36 ± 0.10 | 1.08 ± 0.10 | 1.08 ± 0.12 | 1.83 ± 0.09 |
| Total anthocyanin content | | | 171.41 | 85.66 | 129.95 | 161.80 | 166.01 | 83.43 | 121.67 | 118.51 |



Food Chemistry xxx (2018) xxx-xxx ethyl -----20 40 60 80 100 mg/L -----****** soy pea lentil gluter 0.01 0.02 0.00 0.03 0.04 gelati PH 1 PH 2 773 0 2 6 8 phenyl 田 ethyl 0.1 0.2 0.3 0.0 mg/l



limonene, 3-carene, and geraniol were detectable in Aglianico wine, along with β-damascenone, a norisoprenoid compound associated to rosewood nuances (Karamanidou et al., 2011). Varietal and fermen-tative aroma-relevant molecules detected in *Aglianico* wine and their

individual sensory attributes and odor thresholds are listed in Table 1S. All the varietal components in the untreated wine were retained upon fining with all the proteins tested in this study. An example of the TIC tracings of untreated wine compared to one find with lentil proteins, the most effective fining agent (at least in terms of residual turbidity of the treated wine, see Fig. 1 and related comments), is presented as Supplementary material (Fig. 3S). Peaks with the largest areas in the TIC chromatograms corresponded to fer-

with the largest areas in the Tric timonialogiants correspondent to ter-mentation-derived compounds such as various esters, faitty acids, and alcohols. Quantitative data for the compounds most affected by fining with the various fining agents are presented in Figs. 3 (for ethyl esters) and 4 (for phenylethyl alcohol and its acetate ester, and for isoamyl ac-

etate and benzaldehyde). In general, gelatin was the fining agent causing the largest decrease

in the three ethyl esters considered in Fig. 3. As for the components considered in Fig. 4, benzaldehyde and phenylethyl acetate were the components most affected by the fining process, that had marginal ef-fects on the levels of phenyethyl alcohol and of isoamyl acetate. The

decrease in benzaldehyde upon fining ranged from 75 to 55%, with gelatin being most effective in its removal. Benzaldehyde is related to bitter almond notes, that are often considered a non-desirable trait.

Upon fining, the levels of phenyl ethyl acetate dropped to 10% of what



Fig. 3. Percentage loss of ethyl esters in Aglianico wine after vario obtained by means of static headspace-GC/MS analysis. Values i identical superscripts are not significantly different (LSD; $p \le 0.05$)

g. 4. Percentage loss of various aroma compounds in Aglianico wine after various ning treatments, obtained by means of static headspace-GC/MS analysis. Values in the me panel with identical superscripts are not significantly different (LSD; $p \le 0.05$). Fig. 4. Pero

process may have a limited impact on the aroma traits related to these

specific components. On the basis of the chemical and biochemical characteristics of the proteins used here (and of the chemical and physical properties of individual aroma components) it is not possible at the moment to provide a straightforward rationale for some aroma-relevant molecules more being affected by fining or less than others. Again, as pointed out before in our comments on the fining efficacy, it may be possible that using different amounts of individual fining agents – rather than the identical amount used in this study for comparative purposes - will affect the aromatic profile of the product. These aspects, including the ki-netics of the various interactions discussed so far and their dependence on the amount of the various fining agents, will be the subject of further studies.

3.5. Interaction of plant proteins with anthocyanins in aged red wines

The same plant proteins tested as fining agents on young Aglianico wine were used – at the same concentration – to treat the same wine after twelve and twenty-four months of ageing. The approach used to understand the positive or negative effects of fining on wine quality was focused on profiling the products of reactions involving antho-

cyanins derivatives, and allowed to evaluate the presence, abundance, and sensitivity to fining of acetaldehyde-mediated condensation products and of pyranoanthocyanins. All these species are reportedly important for improvement and stabilization of wine color, as they have shown to be resistant to pH variations, to SO₂ bleaching, and to further anthocyanin oxidation (Escribano-Ballon, Alvarez-Garcia, Rivas-Gonzalo, Heredia, & Santos-Buelga, 2001). Table 2 reports the effect of fining on the anthocyanin profile in

Table 2 reports the effect of fining on the anthocyanin profile in Aglianico red wines treated after twelve and twenty-four months of ageing, and makes it possible a comparison among changes related to both wine aging and to fining wines of different age. Both one- and two-year aged wine showed a decrease in total anthocyanin content, along with the age-related formation of co-pigmented products. The observed decrease in total anthocyanin content of aged wines with respect to the young one is well explained by a combination of reactions with various other compounds in the wine, as well as by breakdown reactions. This is exemplified in Table 2, among others, by the ageing-related increase of caffeoyl and vinylguaiacol derivatives of peonidin, with a concomitant decrease in the levels of the original monoglucoside. Table 25 in the accompanying Supplementary materials offers a comprehensive vision of the most relevant data presented in Tables 1 and 2.

There are some aspects of the interactions among fining proteins and anthocyanins that are highlighted when comparing the effects of the various fining agents on wines of different age. By analyzing the data presented in Table 1 and 2 (and the comprehensive comparison in Table 25) it is evident that the anthocyanin derivatives progressively formed upon aging are removed more efficaciously by the fining process than the corresponding non-modified glucosides. Whether this behavior relates to a decrease in their polarity or to ageing-related changes in the complex wine matrix remains to be asseed.

Another point arising from the comparison among the effects of fining wines of different age is that even the fining agents that provec most effective in anthocyanin removal from young Aglianico rung (namely, soybean, PHI, and gelatin) were quite ineffective when hus on one-year aged wine. Conversely, all fining agents removed more than 85% of total anthocyanins from two-year aged wine. This surgests that both the chemical nature of the involved compounds such their concentration play a role in the formation of insolution proteining gregates incorporating the target anthocyanins, and - ultimately - in their removal.

In an attempt to relate our analytical data to aits, spec trophotometric measurements were used to ca uta parative s observed for scrutiny of fined and unfined wine of the s young wine, fining of aged wines resulted in tensity, but the tonality remained unaffected, a eased color in-ed by the ratio between absorbance parameters at v shown). In other words, fining-dependency anin profiles of aged wine discussed statistically significant fining-dependency ious w lengths (data not modifi ions in the antho not accompanied by wine tonality, regardless of the type of fining ager used.

4. Conclusions

A first conclusion that may be drawn from this work is that proteins of plant origin are at least as effective – at the same addition rate – as

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animal-derived proteins in fining of red wine, even when winemaking involves grapes with complex varietal aroma composition. Varietal aroma compounds in young Aglianico wine were not affected by the fining treatment. The impact of fining with plant proteins on the aroma components of fermentative origin in young vine was much less marked than that of gelatin, used here as a reference animal protein. Also, the effects of fining on fermentative norma components was much lower in this study than what proteions reported for white wine (Granato et al., 2014). This should art researcher in this area as for the feasibility of extending or gen calizing unservations made on model systems to real ones.

model systems to real ones. Also noteworthy is the observation that fining with plant proteins did not alter the color traits of the treated wine, at least in terms of tonality, despite the ability of ume of the plant proteins studied here (most notably, soybean, leafth, and une of the pea protein hydrolysate) to selectively remove specific anthoryanins components. Whether the observed selectivity has an impact on organoleptic traits other than color was beyond the scope of this study, and remains to be assessed, possibly under actual whemaking conditions. This type of studies will also allow to excurate the combination of treatments and agents that could be best suited or improving quality traits - or for removing undesired components - impriven wine. This struct also demonstrates that simple biotechnological inter-

This struct bloc demonstrates that simple biotechnological interventions such as mitted proteolysis of some of the plant proteins used here) muchave a positive impact on their behavior as fining agents. In this fame, hit suddy points out that some of the proteolyzed pea proteins used in this study have a ligand specificity quite different – from both aquality five and quantitative standpoint – from that observed in the consystending native proteins. The different fining efficacy and opecificity reported here for hydrolyzates prepared with different enzymes underscores the necessity of careful selection of the appropriate hydrolysis conditions, and suggests the possibility of exploiting some satures of plant protein hydrolysates in further studies aimed at developing a possible targeted action on wine-relevant chemical components.

Future studies based on the molecular analyses reported here will necessarily involve an appropriate evaluation of the sensory impact of the fining process on young red wine, and could be completed by a thorough analysis of the chemical and sensory changes occurring upon ageing of the red wine fined at an early stage. From the applicative standpoint, further studies also may wish to address how the time course and efficiency of fining (and of the related molecular changes, as explored in this study) may relate to the amount of individual fining agents and of their derivatives. These studies may be also helpful in improving our current understanding of the nature and relevance of the various intermolecular interactions that are relevant to the fining process as a whole, also when different types of grapes and different winemaking procedures are involved.

Appendix A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at https://doi.org/10.1016/j.foodchem.2018.02.085.

| | age | unnned | 51 | LE | GL | М | PHI | PHZ | GE |
|---|----------|--------------------------------|--------------------------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|
| cy-3-O-glu | 1v | 0.24 + 0.12 | 0.67 ± 0.01 | 0.23 + 0.02 | 0.44 + 0.11 | 0.63 + 0.01 | 0.25 + 0.06 | 0.81 + 0.09 | 0.14 + 0.02 |
| | 2v | 0.30 ± 0.01 | 0.16 ± 0.01 | n.d. | n.d. | n.d. | 0.15 ± 0.00 | 0.12± 0.04 | n.d. |
| pe-3-O-glu | 1v | 2.36 ± 0.36 | 2.34 ± 0.30 | 1.49 ± 0.21 | 1.93 ± 0.19 | 3.41 ± 0.25 | 2.07 ± 0.26 | 2.76 ± 0.14 | 1.58 ± 0.20 |
| 1 | 2v | 0.65 + 0.03 | 0.70 ± 0.01 | 0.34 ± 0.01 | n.d. | nd | 0.50 + 0.05 | 0.44 + 0.03 | 0.29 + 0.01 |
| my-3-O-glu | 1v | 10.65 ± 0.40 | 8.61 ± 0.20 | 7.81 ± 0.22 | 7.22 ± 0.22 | 9.71 ± 0.39 | 8.99+0.32 | 715 ± 0.25 | 8.46 ± 0.19 |
| | 24 | 9.64 + 0.51 | 246 ± 0.21 | 2 61 + 0 29 | 2 30 + 0.41 | 2 53 + 0 20 | 2 20 + 0 10 | 329 ± 0.15 | 2 15 + 0 20 |
| nt.3-O-alu | 1v | 0.82 ± 0.12 | 0.78 ± 0.15 | 0.62 ± 0.01 | 0.83 ± 0.21 | 1.02 ± 0.26 | 0.72+0.05 | 1.04 ± 0.09 | 0.69 ± 0.09 |
| pr 5 0 Sia | 20 | 0.84 ± 0.01 | 0.20 ± 0.02 | 0.10+0.01 | n.d | 0.18 ± 0.02 | 0.20 + 0.02 | 0.20+0.04 | nd |
| de 2 O glu | 2y 1v | 0.04 ± 0.01 | 0.20 ± 0.05 | 0.19±0.01 | n.u. | n.d | 0.30 1 0.02 | 0.20 ± 0.04 | nd |
| ue-o-o-gu | 20 | 0.27±0.01 | n.d. | nd. | n.d. | in the | n.d. | n.d. | n.d. |
| pe-3-0-(6-0 p-coumary)-elu | 2y 1v | 0.27 ± 0.01 0.70 ± 0.10 | 0.46 ± 0.04 | 0.44 ± 0.16 | 0.49 ± 0.08 | 0.52 ± 0.10 | 0.54 ± 0.09 | 0.39±0.02 | 0.49 ± 0.12 |
| pe o o (o o p country) gru | 20 | 0.16±0.02 | n d | nd | nd | 0.02 1 0.10 | nd | nd | nd |
| mu 2.0 (6.0 m sourcemi) alu | 2.9 | 2.40 + 0.26 | 1.42 . 0.10 | 1.58 . 0.07 | 1.64 . 0.10 | 100 0 20 | 1.07 . 0.12 | 1.20 . 0.10 | 1.65 . 0.04 |
| iiiv-3-0-(0-0 p-countaryi)-giu | 1 y 2 | 2.40 ± 0.20 | 1.43 ± 0.19 0.17 + 0.01 | 1.36 ± 0.07 | 1.04 ± 0.12 | 1.03 ± 0.20 | 1.87±0.12 | 1.20 ± 0.18 | 1.03 ± 0.04 |
| h no (co h -h- | 2y | 0.52 ± 0.07 | 0.17 ± 0.01 | 0.16±0.01 | 0.16 ± 0.01 | 0.23 0.00 | n.a. | n.a. | 0.1/±0.00 |
| de-3-0-(6-0 p-coumaryi)-giu | 19 | n.d. | n.a. | n.d. | n.a. | n.a. | n.a. | n.d. | n.d. |
| | 29 | 0.10±0.02 | n.a. | n.a. | na | n.a. | n.a. | n.a. | n.a. |
| cy-3-O-(6-O p-coumaryi)-giu | 19 | n.d. | n.a. | n.d. | n.c. | n.a. | n.d. | n.a. | n.d. |
| | 2y | 0.1/±0.01 | n.a. | | n.a. | 0.17 ± 0.0 | 0.14 ± 0.01 | 0.12 ± 0.0 | n.a. |
| pe-3-O-(6-O p-catteoyl)-giu | 1y | 1.79 ± 0.16 | 0.92 ± 0.15 | 1.17 ± 0.07 | 1.16 ± 0.02 | 1.36 ± 0.21 | 1.35 ± 0.12 | 0.81 ± 0.01 | 1.41 ± 0.05 |
| | 2y | 2.27 ± 0.27 | 0.70 ± 0.11 | 1.05 ± 0.11 | 1.12 ± 0.19 | n.d. | 0.96 ± 0.07 | 0.92 ± 0.09 | n.d. |
| mv-3-O-(6-O p-caffeoyl)-glu | 1y | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |
| | 2y | 0.14 ± 0.01 | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |
| pe-3-O-glu-4-vinylguaiacol | 1y | 0.97 ± 0.10 | 0.55 ± 0.09 | 0.68 ± 0.16 | 0.73 ± 0.12 | 0.72 ± 0.07 | 0.68 ± 0.17 | 0.49 ± 0.06 | 0.68 ± 0.19 |
| | 2y | 2.08 ± 0.24 | 0.78 ± 0.15 | 1.08 ± 0.24 | 1.32 ± 0.14 | 0.15 ± 0.01 | 1.29 ± 0.14 | 1.11 ± 0.12 | n.d. |
| mv-3-O-glu-4-vinylguaiacol | 1y | 0.14 ± 0.12 | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | 0.12 ± 0.12 |
| | 2y | 0.20 ± 0.03 | 0.11 ± 0.01 | n.d. | 0.13 ± 0.0 | 0.20 ± 0.02 | n.d. | n.d. | 1.67 ± 0.19 |
| pe-3-O-glu pyruvate | 1y | 0.26 ± 0.03 | 0.16 ± 0.00 | 0.16 ± 0.02 | 0.19 ± 0.01 | 0.22 ± 0.01 | 0.21 ± 0.03 | 0.14 ± 0.00 | 0.20 ± 0.02 |
| | 2у | 1.79±0.20 | 0.17 ± 0.01 | 2.29 ± 0.23 | 0.25 ± 0.01 | 0.32 ± 0.03 | 0.24 ± 0.01 | 0.23 ± 0.01 | 0.21 ± 0.02 |
| de-3-O-glu pyruvate | ly | 0.34 ± 0.09 | n.d. | n.d. | 0.23 ± 0.03 | n.d. | n.d. | n.d. | n.d. |
| | 2y | 1.49 ± 0.19 | 0.25 ± 0.01 | 0.30 ± 0.02 | 0.24 ± 0.02 | 0.43 ± 0.03 | 0.25 ± 0.01 | 0.27 ± 0.02 | 0.23 ± 0.02 |
| mv-3-O-cou-glu pyruvate | Ly I | 0.47 ± 0.07 | 0.26 ± 0.01 | 0.22 ± 0.02 | 0.27 ± 0.03 | 0.45 ± 0.04 | 0.50 ± 0.00 | 0.21 ± 0.01 | 0.28 ± 0.02 |
| | 2y | 0.20 ± 0.01 | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |
| mv-3-0-(6-0-ac)-glu | 1. | 2.75 ± 0.46 | 1.62 ± 0.14 | 1.86 ± 0.17 | 1.81 ± 0.13 | 2.18 ± 0.32 | 2.32 ± 0.22 | 1.56 ± 0.19 | 2.01 ± 0.14 |
| | 2v | 1.17 ± 0.14 | 0.44 ± 0.05 | 0.42 ± 0.02 | 0.48 ± 0.04 | 0.54 ± 0.09 | 0.05 ± 0.02 | 0.45 ± 0.04 | 0.43 ± 0.01 |
| mv-3-O-glu acetaldehvde | 1v | 0.70 ± 0.08 | 0.42 ± 0.09 | 0.55 ± 0.02 | 0.55 ± 0.12 | 0.50 + 0.04 | 0.48 ± 0.01 | 0.38 + 0.03 | 0.47 ± 0.04 |
| | 2v | n.d. | n d | n.d. | n.d. | nd | n.d. | n.d. | n.d. |
| (epi)cat-my-3-O-glu | 1v | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |
| | 2v | 0.70 ± 0.03 | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |
| di(epi)cat-my≤3-O-glu | 1v | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |
| | 2y | 0.01 ± 0.00 | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |
| my-3-O-glu-8-ethyl-(epi)cat | 1v | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |
| | 2v | 0.18 ± 0.00 | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |
| my-3-O-glu-8-ethyl-(epi)cat | 1v | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |
| | 2v | 0.36 ± 0.01 | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |
| my-3-O-glu-8-ethyl-(epi)cat | 1v | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |
| | 2y | 0.06 ± 0.02 | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |
| my-3-O-glu-o-ethyl(epigallo)gallocat | 1v | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |
| 0.0000000000000000000000000000000000000 | 2v | 0.16 ± 0.01 | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |
| my-3-O-coumaroyl-glu-8-ethyl-(epi)cat | 1v | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |
| · · · · · · · · · · · · · · · · · · · | 2v | 0.14 ± 0.01 | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |
| my-3-O-glu-4-vinyl(epi)cat | 1v | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. |

 Table 2

 Mean values of concentration (mg/l) and standard deviations (n = 2) of anthocyanic phenolic compounds identified by HPLC-MS in Aglianico wine of the given vintage either unfined or fined with various plant proteins (SI, soy protein isolate; IE, lentil flour; GL, gluten; PI, pea protein hydrolysate 1; PH2, pea protein hydrolysate 2; GE, gelatin; n.d., not detected).

References

- Bakker, J., Timberlake, C.F., 1997. Isolation, identification, and characterization of new color-stable anthocyanins occurring in some red wines. Journal of Agricultural and Food Chemistry 45, 35–43.
 Ben Aziz, M., Mouis, L., Fulcrand, H., Douieb, H., Hajigi, H., 2017. Phenolic compounds of Moroccan red press wines: Influence of fining agents and micro-oxygenation treat-ments. JWT-Food Science Technology 79, 143–150.
 Bonomi, F., Mora, G., Pagari, M.A., Ianetti, S., 2004. Probing structural features of water-insoluble proteins by front-face fluorescence. Analytical Biochemistry 329, 104 –111.

- ments. LWT-Food Science Technology 78, 142-150.
 noomi, F., Mora, G., Pagani, M.A., Ianetti, S., 2004. Probing structural features of water-insoluble proteins by front-face fluorescence. Analytical Biochemistry 329, 104–111.
 oution, R., 2001. The copigmentation of anthocyanins and its role in the color of red wine: A critical review. American Journal Enology Viticulture 52, 67–87.
 assasa, L.F., Harbertson, J.F., 2014. Extraction, evolution, and sensory impact of phenolic componeds during red wine meareration. Annual Review of Food Science and Technology 5, 83–109.
 assas, L.R., Harbertson, J.F., 2014. Extraction, evolution, and sensory impact of phenolic componeds during red wine meareration. Annual Review of Food Science and Technology 5, 83–109.
 assas, L.R., Mejuto, X.C., 2008. Phenolic compounds and colour stability of vinhae wines: Influence of winemaking protocol and fining agents. Food Chemistry 106, 18–26.
 tuber, C., Bavo, C.L., 1999. Interaction between wine polyphenols and around relations of the accellabelyd-semidiated condensatical. M. Riva-Gonzulo, J.C., Heredia, E.J., Santos-Buelga, C., 2001. Color and stability of pigments derived from the accellabelyd-semidiated condensation. Between mathidina-O-glucoside and (+)-cate-ethin. Journal of Agricultural and Food Chemistry 49, 1213–1217.
 ambuti, A., Rindai, A., Moio, L., 2012. Use of patital, a potein extracted from potato, as alternative to animal proteins in fining of red wine. European Food Research and Technology 235, 753–765.
 azaola, D., Vincenzi, S., Marragon, M., Pasini, G., Curioni, A., 2017. Grape seed extract: The first protein-based fining agent endogenoom, F., 2014. Hing white wine with plant proteins: Effects of fining on prounthocyanidism and arona components. European Food Research and Technology 238, Cavre, G., Gil, G., 2014. Effect of fining on the colour and pigment composition of young red wines. Food Chemistry 157, 385–392.
 matto, T.M., Piano, F., Nasi, A., Ferranti, P.

- 10.4 (A 1090-11) (A 2014) (A 2

Food Chemistry xxx (2018) xxx-xxx

- ces de la
- Karamanidou, A., Kallithraka, S., Hatzidimitrio, E., 2011. Fining of red wines: Effects: their analytical and sensory parameters. Journal International des Sciences de Vigne et du Vin 45, 47–60.
 Lubbers, S., Charpentier, C., Feullat, M., 1996. Study of the binding of aroma compoun by bentonite in must, wine and model systems. Viti 35, 59–62.
 Marchale, R., Marchal-Bichahut, L., Lallement, A., Lendor, J., 2002. Wheat gluten used a clarifying agent of red wines. Journal of Agricultural andProod Chemistry 50, 1 –134.
 Manny, C., Sarni-Manchado, P., Lefebrox, S., Chaynier, and Montounet, M., 2003. Internet of polyascharides and glyterol on groumbocymbiling precisition of red wine American Dolyascharides and Glyterol on groumbocymbilin precipitation by prote fining agents. Food Hydrocollods 60, 598–60.
 Montounet, M., 2004. Influence of charing acoustic precisition or concentration of selectoric fining, agents, M., Genyare J., Pomsun, P., 2004. Influence of charing and fining, A., Gambard, A., Genyare J., Pombrino, P., 2004. Influence of charing fining agents. Food Hydrocollods 60, 598–60. ms. Vitis 35, 59–62. ., Jeander P., 2002. Wheat gluten used as Agricultural and Food Chemistry 50, 177 Cheynier, M. Moutounet, M., 2003. In-
- M., 2003. In-n of red wines.
- iet, M., 2016. In-
- , P., 2004. Influence of clari-Enology Viticulture 55, 7–12. Nakai, S., Li-Chan, E., 1988. Hydr s in food systems. CRC Press, Boca c inter
- Raton, FL. si, A., Ferra nese, L., 2008. Identification of free and bound and authenticity markers of non-aromatic grapes e of mass spectrometric techniques. Food Cheminti, P. combined and w es through a
- istry 110, 762-P., Marin-Arroyo, M.R., 2010. Non-an-ues. Journal International des Sciences de iega-l imal la Vi
- [19] A. Avallone, S., Cheynier, V., Moutounet, M., 1999, and on wine condensed tannins precipitated by proteins Boy, American Journal Enology Viticulture 50, 81–86.
 Ivatica, P., Violoni, M., Pasini, G., 2013. Assessment of the strated from commercial corn gluten and sensory analysis Food Science Technology 54, 549–556.
 Most Science Technology 54, 549–556.
 on, K.A., 2015. Impact of winemaking practices on the con-on of tannins in red wine. Australian Journal of Grape and 14.

- confirstitution of tamins in red wine. Australian Journal of Urape and Anne Dysearchor, 601–614. Heres'D (e., Ni Járdijam, M.P., Schmidt, O., Schwack, W., 2010. Degree of hydrolysis Neuro very ability provietins used as finiting agents and its influence on polyphenol re-mines. Ircov red wine. European Food Research and Technology 231, 65–74. The approximation of the control of the state of the state of the state effect of proteins and benomine fining on the wine aronna loss. Journal of Agri-entals, an angleta, A., Gazzala, D., Hamini, R., Curinoni, A., 2015. Study of combined effect of proteins and benomine fining on the wine aronna loss. Journal of Agri-rostinantis on the behavior of aronna compounds in a model wine. Journal of Agri-cultural and Food Chemistry 38, 248–251. rg, B., He, F., Zhou, P.P., Liu, Y., Duan, C.Q., 2015. Copfementation between mat-vidina-30-glarcode and hydroxytoniamatic actids in red wine model solutions Invest-gations with experimental and theoretical methods. Food Research International 78, 313–3200. gations v 313-320

11

http://web8.elsevierproofcentral.com/en/index.html?token=SPS4f1de3b6... 2/26/2018