



**High-pressure Raman spectroscopy of Ca(Mg,Co)Si<sub>2</sub>O<sub>6</sub> and Ca(Mg,Co)Ge<sub>2</sub>O<sub>6</sub> clinopyroxenes**

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## High-pressure Raman spectroscopy of Ca(Mg,Co)Si<sub>2</sub>O<sub>6</sub> and Ca(Mg,Co)Ge<sub>2</sub>O<sub>6</sub> clinopyroxenes

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### Abstract

In situ high-pressure Raman spectra were collected on four pyroxenes, with composition CaCoSi<sub>2</sub>O<sub>6</sub>, CaMgSi<sub>2</sub>O<sub>6</sub>, CaCoGe<sub>2</sub>O<sub>6</sub> and CaMgGe<sub>2</sub>O<sub>6</sub> and, up to  $P = 7.6$  and  $8.3$  GPa, for silicates and germanates, respectively. The peak wavenumbers  $\nu_i$  increase almost linearly with pressure; the slope  $d\nu_i/dP$  is more pronounced for the modes at higher wavenumbers, and higher in germanates than in silicates. No phase transition or change in the compressional behaviour was observed within the  $P$ -range investigated. The strong dependence of the peak position with pressure of the high energy stretching modes is due to the high sensitivity of the vibrational frequencies probed by Raman spectroscopy to subtle changes in the tetrahedral deformation, which are overlooked by single crystal X-ray diffraction.

**Keywords:** micro-Raman spectroscopy, high pressure, pyroxenes, germanates, silicates.

### Introduction

Pyroxenes, being major phases in Earth mantle, have been the object of several in situ high-pressure investigations: vibrational and elastic properties were measured often on the basis of crystal structure refinements <sup>[1-13]</sup>. The phase transitions discovered in the high-pressure experiments gave hints to the geophysical modelling of the velocity of the seismic waves at the

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3 mantle boundaries, and proved to be crucial to interpret the evolution of the structural deformations  
4 in silicates [2, 14-21].

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6 Raman spectroscopy is one of the most suitable techniques for the investigation of the  
7 vibrational behaviour of minerals at high pressure. Phase transitions are readily revealed, in some  
8 cases even before the same transitions are confirmed by in situ high-pressure X-ray diffraction [7, 22-  
9 24]. Another advantage is that Raman peaks are related to specific structural features, providing  
10 clues on structural deformations alternative to the more time-consuming in situ single-crystal X-ray  
11 diffraction investigations [25, 26]. The potential of Raman spectroscopy in unravelling subtle  
12 structural features was recently improved by detailed quantum mechanical analysis in pyroxenes:  
13 the careful description of the vibrational dynamics, now available for pyroxenes, may relate specific  
14 vibrational modes to the structural changes [26, 27].

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16 The rather flexible structure of pyroxenes allows large deformation with pressure and  
17 temperature. For instance, in the holotypic diopside ( $\text{CaMgSi}_2\text{O}_6$ ), the pyroxene structure can be  
18 deformed without any transformation at pressure beyond 50 GPa, and only at 53 GPa it transforms  
19 into an ilmenite-like structure [29]. Moreover, in natural and synthetic pyroxenes widespread solid  
20 solutions are possible.

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22 The structural unit formula of pyroxenes is  $\text{M}_2\text{M}_1\text{T}_2\text{O}_6$ : the M2 site, hosted in a distorted 6-  
23 8 fold coordinated polyhedron, can be populated by Ca, Na, Mg, Fe, Li, Co, Mn, Zn; the M1 site, in  
24 an almost regular octahedron, by Mg, Mn, Fe, Cr, Ti, Zn, Co, Ni, Sc, In, Ga, Al; the T tetrahedral  
25 site by Si, Ge, and, in part, by Al.

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27 Most studies concerned Si pyroxenes. Ge pyroxenes, and more in general Ge-silicates were  
28 the object of extensive investigation only in the pioneering studies on phase transitions in the Earth  
29 mantle [30, 31]. At the time, the high-pressure conditions expected in the mantle could not be  
30 achieved experimentally, and germanates were used as model systems instead of the isomorphic  
31 silicates, as phase transitions occur at lower pressure in germanates than in silicates.

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33 Recently, there has been a reappraisal for studies aiming to the systematic analysis of the  
34 evolution of structural parameters and physical properties in pyroxenes. Being Ge the only cation  
35 that can fully replace Si in pyroxenes, any investigation on Ge-pyroxenes will give important hints  
36 to model the behaviour of the T tetrahedron in pyroxenes. In addition, as a by-product, several  
37 innovative phase transitions were observed in Ge pyroxenes [32-34]. In situ high-pressure studies are  
38 however scarce and most limited to X-ray diffraction structure refinements and elastic behaviour of  
39 Ge-pyroxenes. To our knowledge, the only study reporting Raman spectra of Ge-pyroxenes at high  
40 pressure was performed by Hofer et al. [35] for  $\text{LiScGe}_2\text{O}_6$  and  $\text{NaScGe}_2\text{O}_6$  and no attempt was made  
41 to compare the results on Ge-pyroxenes with the Si-counterparts.

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3 In this work, we report the high-pressure Raman measurements on two end member Ge-  
4 pyroxenes,  $\text{CaMgGe}_2\text{O}_6$  and  $\text{CaCoGe}_2\text{O}_6$ , and of their Si-counterparts,  $\text{CaMgSi}_2\text{O}_6$  and  $\text{CaCoSi}_2\text{O}_6$ ,  
5 comparing the pressure dependence of the Raman modes in Mg and Co germanate and silicate  
6 pyroxenes.  
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### 9 10 11 **Experimental methods**

12 An ETH-type diamond anvil cell (DAC) <sup>[36]</sup> was used for the high-pressure Raman  
13 experiments. Stainless-steel T301 foil, 250  $\mu\text{m}$  thick, pre-indented to a thickness of about 100  $\mu\text{m}$ ,  
14 with a 300  $\mu\text{m}$  hole obtained by electro-spark erosion, was used as a gasket. Type-II diamonds were  
15 used as anvils (culet  $\varnothing$  600  $\mu\text{m}$ ). Natural diopside <sup>[37]</sup>, with composition  $\text{CaMg}_{0.93}\text{Fe}_{0.07}\text{Si}_2\text{O}_6$ ,  
16 synthetic  $\text{CaCoSi}_2\text{O}_6$  <sup>[38, 39]</sup>, and  $\text{CaCoGe}_2\text{O}_6$  <sup>[40]</sup> single crystals were loaded in the DAC for the  
17 experiments.  $\text{CaMgGe}_2\text{O}_6$  was available only as a polycrystalline material <sup>[40]</sup>, and a few crystallites  
18 (average size 8-10  $\mu\text{m}$ ) were loaded in the *P*-chamber. Therefore, the absolute intensities of the  
19 Raman modes in  $\text{CaMgGe}_2\text{O}_6$  could be affected by the orientation of the crystallites.  
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26 Silicate and germanate pyroxenes were loaded in two separate runs, so that  $\text{CaCoSi}_2\text{O}_6$  and  
27  $\text{CaMgSi}_2\text{O}_6$ , and  $\text{CaMgGe}_2\text{O}_6$  and  $\text{CaCoGe}_2\text{O}_6$  experienced the same pressure, respectively. The  
28 crystals were placed in the gasket hole together with some ruby chips for pressure measurements by  
29 the ruby-fluorescence method (precision of  $\pm 0.05$  GPa according to Mao et al. <sup>[41]</sup>). Methanol:  
30 ethanol = 4:1 mixture was used as hydrostatic pressure-transmitting medium <sup>[42]</sup>. Raman spectra  
31 were collected in the pressure range 0.0001-7.6 GPa for silicates and 0.0001-8.27 for germanates,  
32 using an Olympus BX40 microscope attached to a Jobin-Yvon Horiba LabRam confocal Raman  
33 spectrometer, equipped with a charge-coupled detector (CCD). The samples were excited with the  
34 473.1 nm blue light of a diode pumped Nd:YAG laser. The laser beam was focused on the sample  
35 on a spot of about 1.5  $\mu\text{m}$  diameter (50x ultra long working distance objective, NA = 0.55) and the  
36 confocal aperture was set at 150  $\mu\text{m}$ . The spectra were collected in backscattered geometry, in the  
37 spectral range 100-1000  $\text{cm}^{-1}$ , with 600 s counting time and three accumulations. The spectrometer  
38 was calibrated using the emission lines of a spectroscopic Zn lamp. No special care was taken to  
39 enhance or minimize possible polarization effects. Raman peak profiles were integrated using  
40 pseudo-Voigt functions.  
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## Results and discussion

The crystal structures of the studied pyroxenes are described in the  $C2/c$  space group and are quite similar, with small differences in the bond distances and polyhedral distortion<sup>[40]</sup>. Pyroxenes are made of tetrahedral chains, generated four times by symmetry in the unit cell, connected by a ribbon of edge sharing M1 octahedra and by the M2 cations in a distorted 8-fold configuration (Fig. 1). Factor-group analysis at  $\Gamma$  point ( $\mathbf{k}=0$ ) shows that  $C2/c$  pyroxenes, with 20 atoms in the reduced primitive unit cell ( $Z = 2$ ), have 30 ( $14 A_g + 16 B_g$ ) Raman active modes<sup>[43, 26]</sup>.

Recent quantum mechanical calculation of the vibrational modes in diopside and orthoenstatite<sup>[27, 28]</sup> showed that the vibrational modes are a complex mixture of bending and stretching-like vibrations involving different atoms: in general, almost pure stretching modes may be found only at wavenumbers higher than  $1000\text{ cm}^{-1}$ . In terms of prevailing mode vibration, in the wavenumber range below  $500\text{ cm}^{-1}$ , modes associated with cation translations as well as longer-wavelength lattice modes are found; the different response of the vibrational modes to changes in pressure, temperature and composition depends on whether they are involved in structural changes of the M1 or M2 polyhedra. The peaks in the mid wavenumber region, between  $500$  and  $800\text{ cm}^{-1}$ , are associated to inter-tetrahedral stretching and bending modes of the tetrahedral chain, with a significant influence of M1 and M2 cation bonding. The modes at higher wavenumbers are associated to Si–O stretching modes of the non-bridging oxygens, i.e. those not shared between tetrahedra along the chain.

The spectra of the investigated silicates show few major peaks between  $300$  and  $400\text{ cm}^{-1}$ , mostly related to stretching and bending in the M2 and M1 polyhedra, a single feature at  $\sim 670\text{ cm}^{-1}$  and another intense peak at about  $1010\text{ cm}^{-1}$ , related to bending and stretching of the tetrahedra, respectively. In germanates, we find similar features albeit downshifted in energy<sup>[41]</sup>. The strong peaks at  $\sim 670$  and  $\sim 1010\text{ cm}^{-1}$  in silicates have their counterpart at  $\sim 550\text{ cm}^{-1}$  and  $\sim 850\text{ cm}^{-1}$  in germanates, respectively<sup>[40]</sup>. In germanates, the peak corresponding to that at about  $1010\text{ cm}^{-1}$  in silicates is accompanied by a second peak at lower intensity.

The weak Raman features that can hardly be followed throughout the high-pressure runs were not considered in this study, thus reducing the number of described peaks with respect to those reported by Chopelas and Serghiou<sup>[7]</sup>. Only the peak positions of six strong peaks could be followed both in  $\text{CaCoSi}_2\text{O}_6$  and in  $\text{CaMgSi}_2\text{O}_6$ . In  $\text{CaMgGe}_2\text{O}_6$  and  $\text{CaCoGe}_2\text{O}_6$ , 16 and 11 peaks could be clearly identified. Among them, the peak positions of 13 and 9 peaks, respectively, were followed with pressure.

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3 The relative intensity of the peaks varies considerably with pressure, but, as the  
4 measurement spot may change during measurements, it is not possible to discriminate the effect of  
5 pressure or orientation on the relative intensities.  
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8 We found that the wavenumbers  $\nu_i$  of the Raman features increase linearly with increasing  
9 pressure: the derived slopes  $d\nu_i/dP$  are reported in Table 1. The deviation from linear behaviour  
10 observed in diopside, in response to the non-linear volume and structural changes with pressure <sup>[7]</sup>  
11 was not detected in our experiments, for the limited  $P$ -range of investigation.  
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14 With increasing pressure, the peak positions vary smoothly and no sudden change in the  
15 slope  $d\nu_i/dP$  (which may be indicative of a change in the compressional behaviour), nor appearance  
16 of extra peaks (which may be indicative of a phase transition) was observed (Fig. 1-2). A change of  
17 the compressional behaviour was found in diopside at pressure higher than those of this study, both  
18 by Raman spectroscopy and X-ray diffraction <sup>[5,7]</sup>; moreover, at  $P > 53$  GPa a reversible transition  
19 to the ilmenite structure occurs <sup>[44]</sup>.  
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25 The smooth increase in wavenumber with pressure found for all peaks corresponds to the  
26 decreased unit-cell volume. The low energy modes between 150 and 200  $\text{cm}^{-1}$  in germanates show  
27 small variations. Intermediate modes show  $d\nu_i/dP$  between 2.5 and 3.5  $\text{cm}^{-1}/\text{GPa}$ . The tetrahedral  
28 stretching modes show higher  $P$ -induced variations, between 3.9 and 5.3  $\text{cm}^{-1}/\text{GPa}$ . In diopside, the  
29 volumes of the M1 and M2 polyhedra decrease by 7.8 and 8.0%, respectively, in the range 0.0001  
30 to 10.16 GPa, whereas that of Si tetrahedron decreases by only 2.7% <sup>[12]</sup>. Even if small, the decrease  
31 in Si-O bonds distances with pressure has a significant effect.  
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37 The high effect on the Raman peak positions of small changes in interatomic distances in  
38 tetrahedra, hardly detected in the structure refinements based on high-pressure single-crystal X-ray  
39 diffraction data, was observed also in the orthopyroxene enstatite <sup>[28]</sup>.  
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42 The same is found for pyroxenes of the series  $\text{CaMgSi}_2\text{O}_6$  -  $\text{CaCoSi}_2\text{O}_6$ , where the structural  
43 changes in the tetrahedron are negligible by single-crystal X-ray diffraction <sup>[45]</sup>, and the tetrahedral  
44 bond lengths appear to be unchanged in all the series. The strong changes in higher wavenumber  
45 modes have been related only to small differences in the tetrahedral distortion <sup>[46]</sup>.  
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48 In the description of the high pressure behaviour of the studied pyroxenes, we will compare  
49 separately the differences between Co and Mg pyroxenes with the same cation at the tetrahedral  
50 site, and those between Ge and Si pyroxenes with the same cation at the M1 site.  
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53 Between Co and Mg pyroxenes, the difference in  $d\nu_i/dP$  of the corresponding Raman peaks  
54 is modest. Usually, Mg end-members show a slightly higher  $d\nu_i/dP$  for peaks at lower wavenumber  
55 (Tab. 1-2). For the Si-O stretching modes, which were ascribed mainly to the stretching of Si-O  
56 with non-bridging oxygen sites [27], the difference between Co and Mg pyroxenes is not  
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3 significant, whereas in germanates the two Ge-O stretching peaks display evident differences  
4 between Co- and Mg-members. The main peak in germanates shows a  $dv_i/dP$  in the Mg end-  
5 member, higher than in the Co end-member ( $5.3 \text{ cm}^{-1}/\text{GPa}$  compared to  $4.9 \text{ cm}^{-1}/\text{GPa}$ ). The opposite  
6 is true for the secondary peak, with the  $809 \text{ cm}^{-1}$  peak of Co pyroxene showing a  $dv_i/dP$  higher than  
7 that observed for the  $828 \text{ cm}^{-1}$  peak of the Mg pyroxene (*i.e.*,  $5.0 \text{ cm}^{-1}/\text{GPa}$  vs.  $3.9 \text{ cm}^{-1}/\text{GPa}$ ,  
8 respectively).

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13 Comparing Ge- and Si-pyroxenes, we observe that the  $P$ -induced variation of the most  
14 intense and resolved stretching modes is significantly higher in germanates than in silicates (Table  
15 1-2, Fig. 3), whereas for lower wavenumber peaks we did not find significant differences.

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18 The higher  $P$ -induced variations of Mg- vs. Co-pyroxenes in intermediate wavenumber  
19 modes and of Ge- vs. Si-pyroxenes in higher wavenumber modes are likely related to the different  
20 compressional behaviour of the polyhedron hosting Co or Mg and of the tetrahedron hosting Ge or  
21 Si: we can expect that higher polyhedral compression is correlated to higher variation of Raman  
22 modes (ascribable to the polyhedral vibrations). Any interpretation would, therefore, need in situ  
23 high-pressure structural data, which, among the investigated phases, are available only for diopside  
24 [1,12]. Therefore, some assumptions will be done here, from the behaviour of pyroxenes with  
25 compositions similar to those of this study.

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28 About the higher  $dv_i/dP$  of intermediate wavenumber modes of Mg- vs. Co-pyroxenes, a  
29 first observation is that the unit-cell volume of the  $\text{CaMgSi}_2\text{O}_6$  pyroxene is more compressible than  
30 in pyroxenes where Mg is replaced by a transition metal as Fe, Ni, Mn or Zn [35]. This was observed  
31 by comparing the unit-cell volume vs. the volume compression at 10 GPa in a series of pyroxenes;  
32 transition metal pyroxenes plot in a common trend, all with lower compressibility than  $\text{CaMgSi}_2\text{O}_6$   
33 pyroxene [35]. In addition, the M1 polyhedron is more compressible when it is populated by Mg  
34 rather than by Fe: the M1 volume compressibility is  $7.7(2) \cdot 10^{-3}$  and  $6.8(3) \cdot 10^{-3} \text{ GPa}^{-1}$  in  
35  $\text{CaMgSi}_2\text{O}_6$  and  $\text{CaFeSi}_2\text{O}_6$ , respectively [12,47]. From our observations on Raman modes, we may  
36 assume that Co behaves as Fe and other transition metals, and that the M1 polyhedron is less  
37 compressible when occupied by Co than when it is populated by Mg.

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40 Structural observations could also explain the very similar  $P$ -induced variation of the higher  
41 wavenumber stretching modes in  $\text{CaCoSi}_2\text{O}_6$  and  $\text{CaMgSi}_2\text{O}_6$ , which could be ascribed to a very  
42 similar tetrahedral compression for the two phases. Again, we may compare  $\text{CaFeSi}_2\text{O}_6$  and  
43  $\text{CaMgSi}_2\text{O}_6$ : between 0.0001 and 10 GPa, the tetrahedral volume decreases by the same amount of  
44  $0.063 \text{ \AA}^3$ . This is predictable, as the tetrahedral site is fully populated by Si, and we may assume  
45 that also in  $\text{CaCoSi}_2\text{O}_6$  the Si-tetrahedron experiences the same tetrahedral compression. On the  
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3 other side, it is not clear the origin of the difference observed in the higher wavenumber Ge-O  
4 stretching modes (Fig. 3).  
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6 The differences between silicates and germanates at higher wavenumber likely involve  
7 differences in the tetrahedral compression. In GeO<sub>2</sub>, structurally analogous to  $\alpha$ -quartz, the structure  
8 is compressed mainly by a deformation of the tetrahedral framework, with very little intra-  
9 tetrahedral compression. Yet, in the comparative paper by Glinnemann et al. [48], the reported  
10 average tetrahedral bond distances show a higher compression in germanates than in silicates,  
11 4.1(1.2) · 10<sup>-4</sup> vs. 2.1(4) · 10<sup>-4</sup> GPa<sup>-1</sup>. In pyroxenes, we do not have a similar example, *i.e.* a couple of  
12 pyroxenes different only for the Si vs. Ge substitution. A close example for comparison is that of  
13 C2/c NaAlSi<sub>2</sub>O<sub>6</sub> and NaScGe<sub>2</sub>O<sub>6</sub>, where good quality data are available [35, 49]. Assuming that, as we  
14 found, the effect of the M1 polyhedron is negligible on the higher wavenumber modes, we may  
15 compare the different compressibility of the average T-O bond lengths. The values of  
16 compressibility in NaAlSi<sub>2</sub>O<sub>6</sub> and NaScGe<sub>2</sub>O<sub>6</sub>, are 8.8(7) · 10<sup>-4</sup> and 11.2(5) · 10<sup>-4</sup> GPa<sup>-1</sup>, respectively,  
17 which support the indication of a higher tetrahedral compression in Ge-pyroxenes.  
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### 30 Conclusions

31 The high-pressure Raman spectra of the pyroxenes CaMgGe<sub>2</sub>O<sub>6</sub> and CaCoGe<sub>2</sub>O<sub>6</sub>, and  
32 CaMgSi<sub>2</sub>O<sub>6</sub> and CaCoSi<sub>2</sub>O<sub>6</sub> did not show evidence either of a phase transition or a change in  
33 compressional behaviour at least up to 8 GPa. The structures apparently keep the C2/c space group  
34 within the investigated *P*-range. A phase transition in the Ca-germanate members could be  
35 expected, as diopside experiences a phase transition at pressure in excess of 10 GPa [44], and a lower  
36 transition pressure is expected in isotopic germanates; such a transition was not observed in this  
37 study.  
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43 A slightly higher compression of Mg- with respect to Co-pyroxenes was observed in lower  
44 energy modes (irrespective of the tetrahedral population), likely related to a different compression  
45 of the M1 polyhedron. Such a difference almost disappears for higher energy modes, which are  
46 most or completely related to tetrahedral bond stretching.  
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50 A higher compression of the tetrahedral Ge-O bonds than the Si-O counterparts is here  
51 inferred, on the basis of the significantly higher *P*-induced variation of the corresponding modes at  
52 higher wavenumbers. The tetrahedral compression is hardly observed by in situ X-ray single crystal  
53 diffraction, especially in the *P*-range 0.0001-10 GPa, whereas Raman spectroscopy appears as a  
54 suitable technique to model slight changes in intra-tetrahedral configurations.  
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Table 1. Pressure-dependence of the Raman peak positions in silicate pyroxenes. Slope  $dv_i/dP$  and intercept from a linear fit are given; strongest peaks in bold. In parentheses: *e.s.d.* of the linear best fit.

CaMgSi <sub>2</sub> O <sub>6</sub>			CaCoSi <sub>2</sub> O <sub>6</sub>			Chopelas and Serghiou [7]	
$\nu$ (cm <sup>-1</sup> )	Intercept (cm <sup>-1</sup> )	$\partial\nu/\partial P$ (cm <sup>-1</sup> /GPa)	$\nu$ (cm <sup>-1</sup> )	Intercept (cm <sup>-1</sup> )	$\partial\nu/\partial P$ (cm <sup>-1</sup> /GPa)	Intercept (cm <sup>-1</sup> )	$\partial\nu/\partial P$ (cm <sup>-1</sup> /GPa)
<b>323</b>	<b>324(1)</b>	<b>2.8(2)</b>	<b>310</b>	<b>312(1)</b>	<b>2.4(2)</b>	324.6	3.36
<b>356</b>	<b>357(1)</b>	<b>2.7(2)</b>	326	328(2)	4.3(5)	356.4	2.98
367	366(1)	3.1(2)	<b>345</b>	<b>346(1)</b>	<b>2.8(2)</b>	365.2	4.01
<b>389</b>	<b>390(1)</b>	<b>3.3(2)</b>	<b>375</b>	<b>375(1)</b>	<b>2.4(2)</b>	390.2	5.01
<b>667</b>	<b>665(1)</b>	<b>2.9(2)</b>	<b>663</b>	<b>662(1)</b>	<b>2.8(2)</b>	666.7	3.30
<b>1013</b>	<b>1012(1)</b>	<b>4.4(2)</b>	<b>1011</b>	<b>1011(1)</b>	<b>4.4(2)</b>	1015.9	4.14

Table 2. Pressure-dependence of the Raman modes in germanate pyroxenes. Slope  $dv_i/dP$  and intercept from a linear fit are given; strongest peaks in bold. In parentheses: *e.s.d.* of the linear best fit.

CaMgGe <sub>2</sub> O <sub>6</sub>			CaCoGe <sub>2</sub> O <sub>6</sub>		
$\nu$ (cm <sup>-1</sup> )	Intercept (cm <sup>-1</sup> )	$\partial\nu/\partial P$ (cm <sup>-1</sup> /GPa)	$\nu$ (cm <sup>-1</sup> )	Intercept (cm <sup>-1</sup> )	$\partial\nu/\partial P$ (cm <sup>-1</sup> /GPa)
150	149.0(5)	2.0(1)	155	154.9(5)	0.4(1)
166	168(2)	2.0(6)	165	166.5(7)	1.1(2)
191	191.7(3)	0.8(1)	187		
			<b>262</b>	<b>263(1)</b>	<b>2.6(1)</b>
<b>275</b>	<b>275(1)</b>	<b>0.9(3)</b>	285		
<b>297</b>	<b>294.4(6)</b>	<b>3.1(1)</b>	298		
317	318(1)	2.7(4)	326	324.5(5)	3.4(1)
<b>342</b>	<b>343(1)</b>	<b>2.2(2)</b>	345		
400	408(2)	4.3(5)	388		
433	431.7(5)	2.9(1)	<b>423</b>	<b>421.2(5)</b>	<b>2.5(2)</b>
<b>553</b>	<b>551.1(5)</b>	<b>3.5(1)</b>	<b>547</b>	<b>545.7(5)</b>	<b>3.4(1)</b>
726	721(1)	2.9(2)	721		
755			755	754.0(6)	4.7(2)
<b>828</b>	<b>830(2)</b>	<b>3.9(3)</b>	809	806(2)	5.0(4)
<b>857</b>	<b>854.7(5)</b>	<b>5.3(1)</b>	<b>843</b>	<b>840.9(5)</b>	<b>4.8(1)</b>

### Captions for figures

Fig. 1. Raman spectra of  $\text{CaMgSi}_2\text{O}_6$  (a) and  $\text{CaCoSi}_2\text{O}_6$  (b) at different pressures. The spectra are cut at about  $800\text{ cm}^{-1}$ , excluding the region of the ethanol and methanol peaks; (c) example of deconvolution of the highest energy peak of the silicates, shifted from  $1010\text{ cm}^{-1}$  at ambient pressure to  $1037\text{ cm}^{-1}$  at 5.58 GPa.

Fig. 2. Raman spectra of  $\text{CaMgGe}_2\text{O}_6$  (a) and  $\text{CaCoGe}_2\text{O}_6$  (b) at different pressures.

Fig. 3: Raman peak position changes with pressure increase ( $\text{cm}^{-1}/\text{GPa}$ ) of the strong peaks at  $\sim 660\text{-}1010\text{ cm}^{-1}$  ( $\text{CaCoSi}_2\text{O}_6$  and  $\text{CaMgSi}_2\text{O}_6$ ) and at  $\sim 550\text{-}850\text{ cm}^{-1}$  ( $\text{CaCoGe}_2\text{O}_6$  and  $\text{CaMgGe}_2\text{O}_6$ ). In silicates the peaks at high wavenumber are superimposed

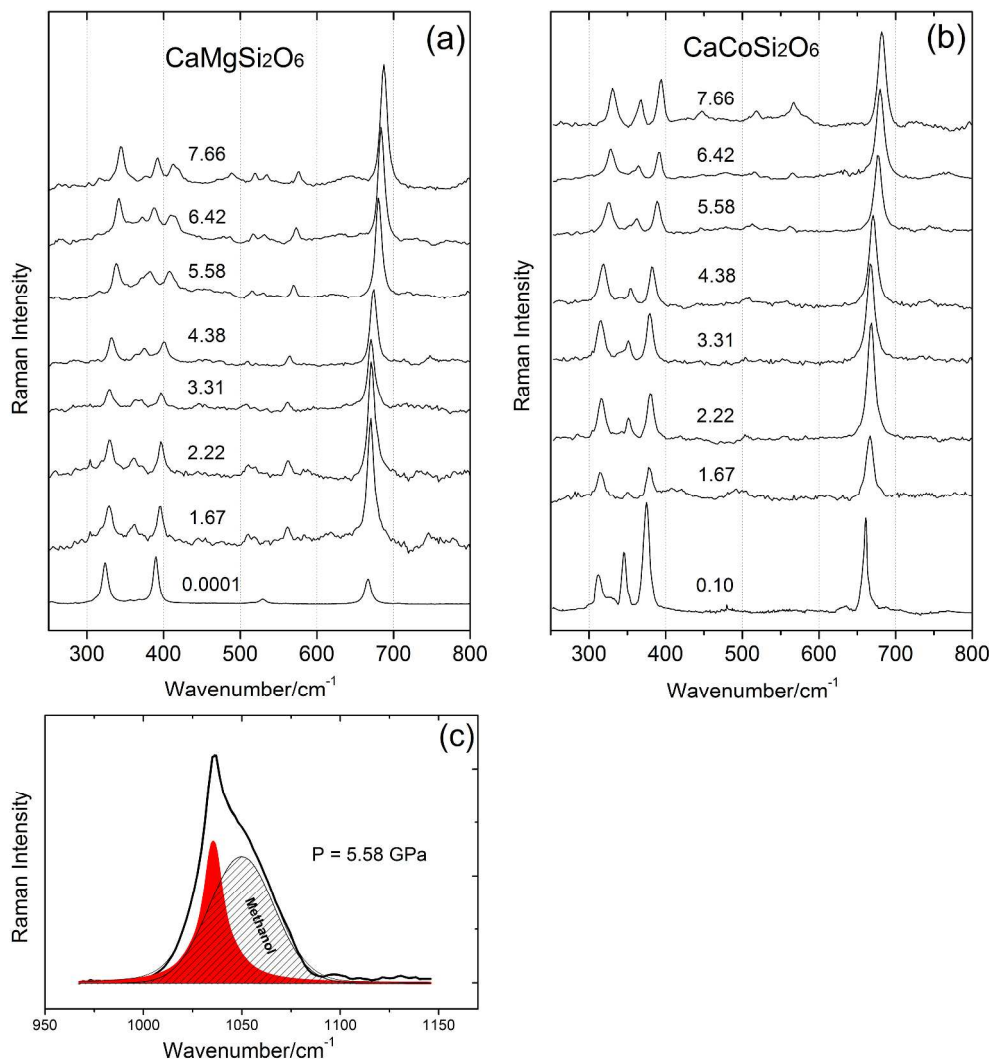


Fig. 1. Raman spectra of  $\text{CaMgSi}_2\text{O}_6$  (a) and  $\text{CaCoSi}_2\text{O}_6$  (b) at different pressures. The spectra are cut at about  $800 \text{ cm}^{-1}$ , excluding the region of the ethanol and methanol peaks; (c) example of deconvolution of the highest energy peak of the silicates, shifted from  $1010 \text{ cm}^{-1}$  at ambient pressure to  $1037 \text{ cm}^{-1}$  at 5.58 GPa.

Fig.1

465x508mm (300 x 300 DPI)



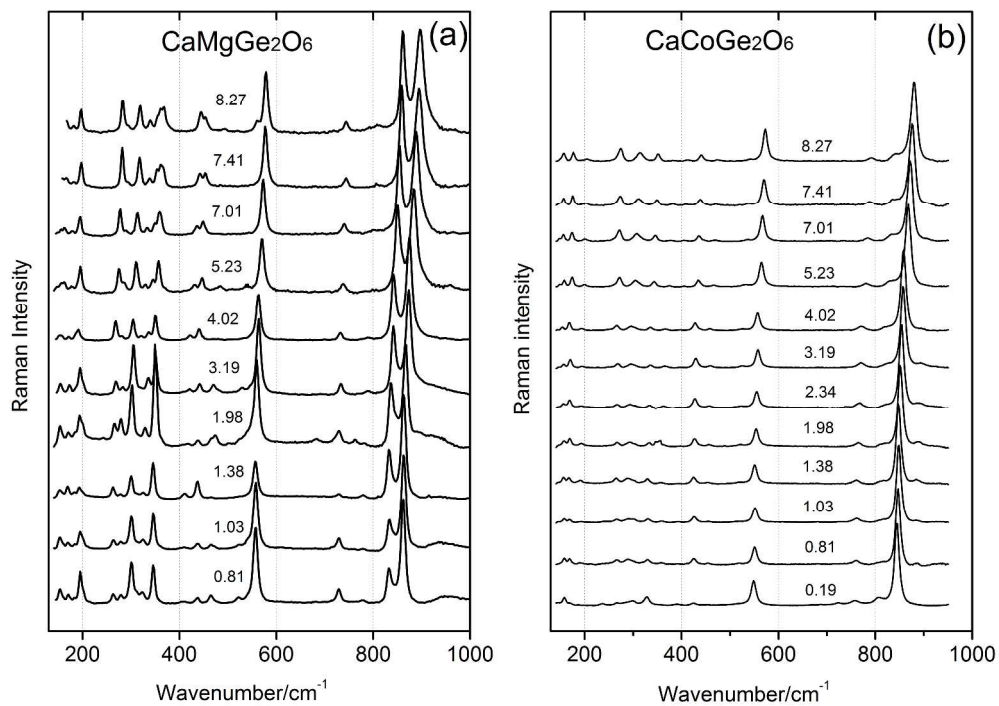


Fig. 2. Raman spectra of CaMgGe<sub>2</sub>O<sub>6</sub> (a) and CaCoGe<sub>2</sub>O<sub>6</sub> (b) at different pressures.

Fig.2

578x413mm (300 x 300 DPI)

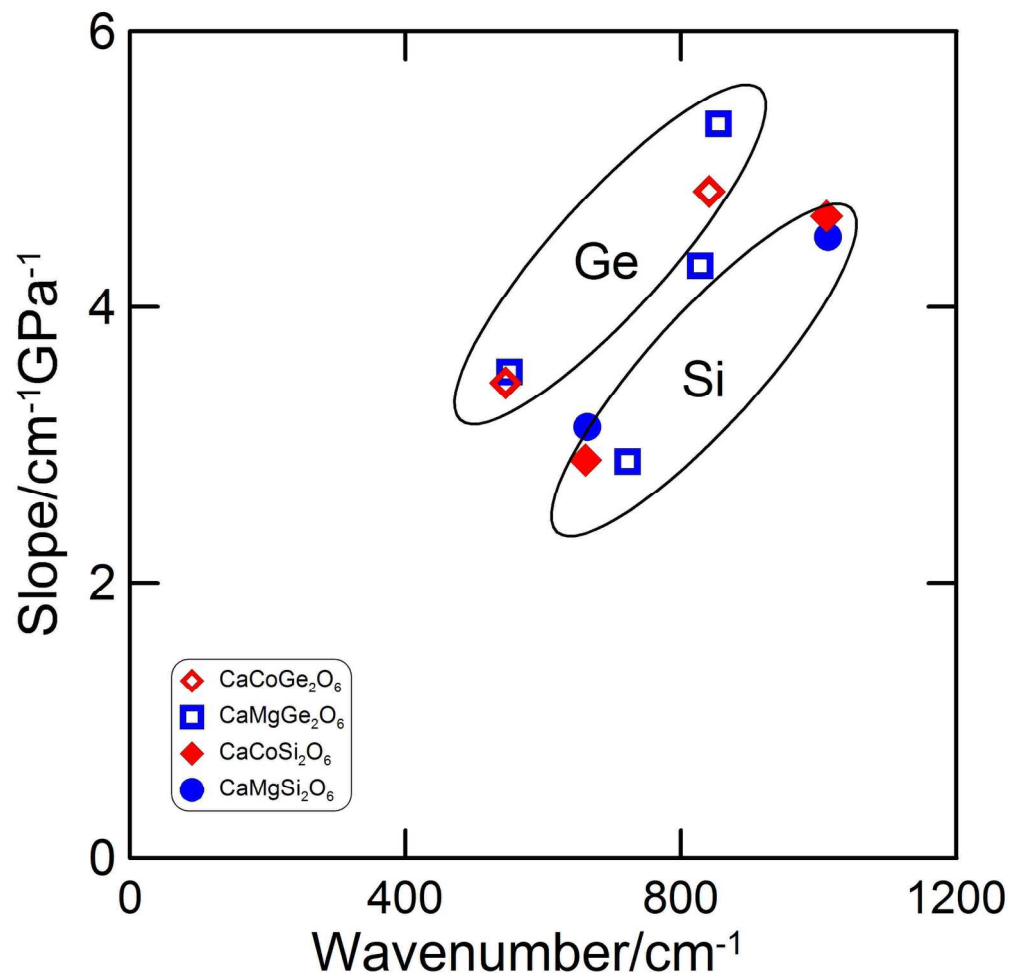


Fig. 3: Raman peak position changes with pressure increase ( $\text{cm}^{-1}/\text{GPa}$ ) of the strong peaks at  $\sim 660\text{-}1010$   $\text{cm}^{-1}$  ( $\text{CaCoSi}_2\text{O}_6$  and  $\text{CaMgSi}_2\text{O}_6$ ) and at  $\sim 550\text{-}850$   $\text{cm}^{-1}$  ( $\text{CaCoGe}_2\text{O}_6$  and  $\text{CaMgGe}_2\text{O}_6$ ). In silicates the peaks at high wavenumber are superimposed.

Fig.3

180x176mm (300 x 300 DPI)