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Biotechnologies for marine oil spill cleanup: indissoluble ties with microorganisms

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1 Keywords

Oil bioremediation; Slow release particles; Oil spill snorkel; Hydrostatic pressure; Deep
sea

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5 Abstract

6 The ubiquitous exploitation of petroleum hydrocarbons (HCs) has been accompanied by

7 accidental spills and chronic pollution in marine ecosystems, including the deep ocean.

8 Physico-chemical technologies are available for oil spill cleanup, but HCs must

9 ultimately be mineralized by microorganisms. How environmental factors drive the

10 assembly and activity of hydrocarbon-degrading microbial communities remains

11 unknown, limiting our capacity to integrate microorganism-based cleanup strategies with

12 current physico-chemical remediation technologies. Here, we summarize recent findings

13 about microbial physiology, metabolism and ecology and describe how microbes can be

14 exploited to create improved biotechnological solutions to cleanup marine surface and

15 deep waters, sediments and beaches.

16

18 Oil Spills in the Oceans and the Role Played by Microorganisms

The first large marine oil spill occurred in 1907 with the sinking of the Thomas W. 19 20 Lawson, which released 7,400 tons of paraffin oil off the coast of the United Kingdom. 21 Since then, estimates indicate that more than seven million tons of oil have been 22 released into the environment from over 140 large spills [1], with the Deepwater Horizon (DWH) disaster releasing more than 700,000 tons of crude oil in the Gulf of Mexico [2]. 23 24 While accidents like DWH are widely publicized, more than 90% of oil pollution comes 25 from non-accidental natural and anthropogenic sources [3], including run-off from land-26 based sources and routine ship operations, such as deballasting and tank washing [4].

27

Oil spills can have extensive environmental and economic impact. For instance, 28 estimates suggest that more than 250,000 seabirds were killed during the Exxon-Valdez 29 30 oil spill in 1989 and that the DWH disaster cost more than \$61 billion [1]. Crude oil that is released into the environment undergoes chemical, physical and biological 31 32 modifications known as weathering. The fate of weathered crude oil hydrocarbons (HCs) depends on microorganisms that have developed strategies to increase their 33 bioavailability and degradation pathways that allow them to use HCs as carbon and 34 energy sources. Hazen et al. [5] observed that the half-lives of n-alkanes that leaked 35 into the deep sea during the DHW oil spill were shorter than expected (1-6 days). Even 36 so, slicks of oil heavily enriched with polycyclic aromatic hydrocarbons (PAHs) reached 37 38 shores and were detected for at least a year [6], requiring cleanup interventions.

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40 Current Response Techniques to Oil Spills in Marine Environments

41 Current emergency responses include mechanical containment and recovery of spilled oil, addition of chemical dispersants, and physical cleanup of shorelines. When an oil 42 spill occurs in a marine environment, the magnitude of the spill is assessed to determine 43 the most suitable response action. Figure 1 highlights the different cleanup approaches 44 45 currently used for waters, shorelines and sediments. The primary combat strategy is mechanically containing and recovering oil by using different types of booms, barriers 46 47 and skimmers, as well as the application of natural or synthetic sorbent materials. 48 Containment booms are typically used in surface waters as barriers to control the oil from spreading and impacting shorelines or other marine resources like aguaculture 49 50 facilities. With the aid of different types of booms (fences, curtains or inflatable booms), oil is concentrated into thicker layers, facilitating subsequent removal with skimmers. 51 Different types of skimmers (weir, oleophilic and suction skimmers) have been 52 53 developed that are suitable for specific oil types and in the presence of ice or debris. 54 Sorbents are either natural or synthetic-like polymeric materials used for very small spills to adsorb the spilled oil either on the surface or internally through swelling. They 55 are also used to remove final traces of oil after collection with skimmers, or in areas that 56 cannot be reached by skimmers. 57

58

59 Dispersants (Box 1) are mixtures of surfactants and solvents that reduce spilled oil into 60 small droplets (smaller than 100 µm in size) by the action of waves and currents; they 61 are mainly used in open and deep waters [7,8] (Figure 1). Oil droplets are spread from 62 the seawater surface into the **water column**, thus preventing the oil slick from reaching 63 the shoreline. Oil droplets have a higher surface-to-volume ratio than oil slicks have,

which allows more of the oil's surface to be exposed to microorganisms. Dispersants are effective when applied immediately following a spill, before the lightest HC components volatilize, and their performance depends on water salinity, temperature and wave action. Considering the toxicity of the dispersants' components [7,8], including to microorganisms [9], the next technological goal in oil spill remediation is replacing chemical surfactants with non-toxic **biosurfactants** (see Glossary).

70

A series of physical methods have been developed for different types of oil 71 contamination. When oil slicks reach sandy beaches or rocky coastlines, they are 72 73 manually collected or sprayed with water to return them back to the water, where they 74 are adsorbed by sorbent materials (Figure 1). Another technique for oil removal is *in situ* 75 burning (ISB), the controlled burning of oil at the spill's location to reduce the amount of 76 oil on the surface. ISB, which should be coupled to toxicity assessments of burned residues [10], has been used on several occasions including the DWH accident, where 77 78 the amount of burned oil was higher than the amount of mechanically collected oil despite estimate uncertainties [11]. In deep-sea spills, capping the head of the leaking 79 oil well is the first action to be taken and is complemented by the other approaches 80 81 (Figure 1).

82

83 **Bioremediation and natural attenuation**

Accurate knowledge of the metabolic potential of microorganisms and of the environmental factors that shape their interactions, viability and degradation activity is required to optimize **bioremediation** intervention strategies. HC-degrading marine

microorganisms can be divided into specialists and generalists according to their ability
to grow respectively on a narrow range of HCs or on a wider set of carbon sources [12].
They include strains of *Bacteria*, *Archaea* and *Fungi* (Table 1, Table S1).
Hydrocarbonoclastic-bacteria are the most studied in terms of their physiology, ecology
and biotechnology because of their ubiquity and prevalence.

92

93 Natural attenuation is determined by the complex metabolic networks built up by 94 microorganisms and their interactions with other organisms such as algae [12-14]. In seawater, one of the main factors limiting the rate of HC degradation is low HC solubility 95 96 [13]. By producing biosurfactants and modifying cell membrane hydrophobicity, microorganisms increase and modulate HC bioavailability according to environmental 97 conditions, such as in the case of *Alcanivorax* spp. [15]. Oxygen availability determines 98 the pathway of HC breakdown. In the water column, under aerobic conditions, microbial 99 100 enzymes activate HC molecules by incorporating oxygen atoms, generating 101 corresponding alcohols, which are further oxidized to carboxylic acids that are degraded 102 through β -oxidation [16]. The most studied aerobic enzymes are alkane hydroxylases, encoded by the *alkB*, *p450* and *almA* genes [16,17]. Under anaerobic conditions typical 103 in sediments, HC catabolism is activated by the addition of fumarate to the secondary 104 105 carbon and catalysed by alkyl-succinate synthases [16].

106

107 The entire complexity of the metabolic pathways involved in HC catabolism under *in situ* 108 conditions is not yet understood, especially in relation to the environmental conditions, 109 although the increasing availability of **meta-omics** data is clarifying the picture [18–22].

Marine microbial communities responding to HC inputs are subjected to ecological successions generally starting with aliphatic HC-degraders (*e.g.*, *Alcanivorax* spp.) and the subsequent enrichment of other microorganisms able to metabolize aromatic HCs (*e.g.*, *Cycloclasticus* spp.) as observed in seawater [23], sediments [24] and beaches [21]. The response of the autochthonous microbial community after an oil spill depends upon the community's initial functional diversity. This response is also modulated by the local environmental conditions.

117 A metagenomic study of chronically polluted sediments across a transect from the Mediterranean Sea to the Red Sea reconstructed the occurring metabolic pathways, 118 119 demonstrating that biodegradation capabilities expand according to the increasing 120 yearly average sediment temperature across the transect despite a decrease in bacterial richness [18]. Although some common and specialized HC-degrading bacteria 121 122 were not detected by 16S rRNA gene sequencing, the metabolic networks accounting 123 for both aerobic and anaerobic HC degradation pathways indicated that the chronically polluted marine sediments harbour a higher catabolic diversification than do freshly 124 polluted samples [18]. This suggests that microbial communities exposed to chronic 125 126 pollution possess a wide repertoire of catabolic abilities that allow them to respond more 127 promptly to oil spills [18]. HC cleanup may occur through transient enrichment of specialized microorganisms that degrade the different HC classes; their taxa succession 128 129 patterns have been described [23]. The combination of degradation capabilities and pathways of different microbial community members is required for PAH degradation as 130 indicated by genomes assembled from metagenomic datasets following stable isotope 131 probing of the surface and deep plume waters of the DWH spill [14]. This suggests that 132

key degradative pathways are distributed across the different members of the DWH
community that cooperatively and co-ordinately react to initiate PAHs degradation [14].

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136 Biotechnologies for water and coastal pollution

Bioremediation technologies compatible with natural biogeochemical cycles comprise 137 biosurfactant amendments, biostimulation and bioaugmentation. Biosurfactants 138 139 favour oil solubilisation and oil droplet formation in water (Figure 2), making HCs 140 available to non-biosurfactant-producing microorganisms. Bacteria, yeast and fungi mainly produce anionic or neutral biosurfactants. Renowned producers include 141 142 Acinetobacter, Bacillus and Pseudomonas, with Alcanivorax predominating in oilcontaminated marine surface waters worldwide, owing to its capacity to produce large 143 amounts of glycolipid biosurfactants [25]. Compared to dispersants, biosurfactants have 144 low to no toxicity, high biodegradability, surface and emulsification activities and high 145 146 stability under extreme temperature, pH and salinity conditions [26,27]. Biosurfactants positively impact HC biodegradation [28,29] and have been applied to oil recovery in 147 reservoirs, to oil transportation in pipelines and to production of emulsified fuels [30]. 148

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Biostimulation is among the most effective approaches to enhance the degrading activity of indigenous bacterial communities since it improves the C/N/P ratio that is unbalanced after oil spills [27,31–33]. Nutrient delivery represents the main limitation of biostimulation due to rapid leaching along the water column [34]. Nutrient **microencapsulation** within slow-release particles (SRPs) eventually combined with biosurfactants and oil-degrading microorganisms may improve biostimulation efficiency

(Figure 2). SRPs are produced with low-toxicity polymers, e.g., polyurethane-polyurea 156 157 copolymers, alginates and chitosans [35,36]. A recent strategy based on microinjecting, cryo-crosslinking and coating with ethyl cellulose films allowed the synthesis of alginate-158 beads with increased load capacity and slow release of N/P fertilizers [36]. Even though 159 the addition of nutrients can promote the growth of heterotrophs, thus creating 160 161 competition with HC-degraders [37], increased total petroleum HC removal was 162 reported by using alginate-encapsulated diesel-degrading bacteria as compared to free-163 cells [38]. Recent metagenomic data indicated that biostimulation with different nitrogen sources favoured different HC catabolic pathways without substantially affecting the 164 165 taxonomic structure of the bacterial community [39].

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Indigenous microbial populations may not have the full metabolic ability to process the 167 complex mixture of spilled HCs or specialized HC-degraders might be underrepresented 168 169 in the existing community. Bioaugmentation with site-allochthonous microorganisms might enhance HC degradation [40] although the long-term presence of such non-170 indigenous microorganisms has been questioned [41]. Autochthonous bioaugmentation 171 172 (ABA) uses indigenous microorganisms enriched from specific habitats of the contaminated site (e.g., surface or deep-sea water) (Figure 2). This approach has been 173 174 successfully implemented in a study aimed at simulate an oil spill event, showing that 175 the use of pre-adapted HC degrading bacteria in combination to biostimulation provided the best results in term of HC removal [42]. A further alternative is bioaugmentation with 176 mobile genetic elements [43], which aims at horizontal transfer of remediation genes 177

from an exogenous inoculant to indigenous microorganisms by using catabolic plasmidsor transposons.

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181 Biotechnological approaches to remediate oil-contaminated marine sediments

Marine sediments represent an important sink for petroleum HCs after accidental spills. 182 A number of different chemical, physical, and microbiological processes contribute to 183 184 the transport of petroleum from a positively buoyant state in the water column down to 185 the seafloor, such as weathering, adsorption onto settling particulate matter (including so-called marine oil snow [44], Figure 2), and the addition of chemical dispersants. In 186 187 the case of the DWH spill, 1.8-14% of the oil reached the seafloor as estimated using hopanes as a biomarker tracer [45] or 0.5-9% as estimated using radiocarbon 188 distributions [46]. Upon sedimentation, oil penetrates the upper sediment layers 189 (typically 1-30 cm, depending on site conditions). In the sediments it may persist due to 190 191 the prevailing anoxic conditions (Figure 3A) that drastically limit the occurrence of oxidative biodegradation processes, and the low bioavailability resulting from the strong 192 193 sorption onto hydrophobic sedimentary materials.

194

In situ bioremediation is typically regarded as one of the most effective and sustainable strategies to cleanup contaminated sediments. Different approaches have been proposed to stimulate naturally occurring microbial communities that degrade petroleum HCs in marine sediments [47,48]. These typically involve the subsurface addition of degradation rate-limiting nutrients, electron acceptors and (bio)surfactants. The successful stimulation of the indigenous microbial community dwelling in oil polluted 201 sediment was achieved by adding nitrogen and phosphorous, ultimately resulting in a 202 higher HC removal compared to the non-biostimulated control [31]. Similarly, the activity 203 of indigenous oil-degraders can be boosted by the addition of biosurfactants [42]. The interplay between the bioavailability of electron acceptors (*e.g.*, oxygen, nitrate, sulfate, 204 Fe^{3+/}Mn⁴⁺) and HCs is probably the most critical factor affecting the efficacy of sediment 205 206 bioremediation systems. Under aerobic conditions, petroleum HC biodegradation occurs 207 rapidly and therefore different engineered approaches have been proposed to deliver 208 oxygen to the sediments. Among them, a modular slurry system, which performs in situ 209 aeration of the contaminated sediments, while minimizing the risk of spreading the 210 contamination away from the treatment zone, has recently been developed (Figure 3B) [49]. Although the system was highly effective in stimulating the metabolism of aerobic 211 HC-degrading bacteria and in reducing sediment toxicity, its application turned out to be 212 highly labour- and energy-intensive. Other methods include supplying oxygen-releasing 213 214 compounds (e.g., calcium peroxide-based chemicals) to the contaminated sediment. However, rapid abiotic oxygen consumption by reactions with reduced chemical species 215 $(e.g., Fe^{2+}, S^{2-})$ and the difficulties in controlling the rate of oxygen release over time are 216 217 some limitations of these approaches.

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Recently, an innovative bioelectrochemical system termed an "oil-spill snorkel" was proposed to accelerate oil HC biodegradation in marine sediments [50]. The system consists of a conductive graphite rod (the "snorkel") positioned for electrochemically connecting two spatially segregated redox zones: the anoxic contaminated sediment and the oxic (O₂-containing) overlying water (Figure 3C). The portion of the snorkel

224 positioned in the anoxic sediment serves as an electron acceptor (*i.e.*, an anode), 225 sinking electrons deriving directly from the microbially catalyzed anaerobic oxidation of HCs and from the chemical and/or biochemical oxidation of reduced species (*i.e.*, S²⁻, 226 Fe²⁺) occurring in the bulk of the sediment. Upon transfer to the buried portion of the 227 snorkel, the electrons move to the upper portion (*i.e.*, the cathode), driven by the 228 229 existing redox gradient, where they combine with oxygen and protons to form water as a 230 by-product. Besides serving as a virtually inexhaustible respiratory electron acceptor in 231 the anaerobic oxidation of petroleum HCs, the snorkel was suggested to indirectly stimulate HC biodegradation by sulfate-reducing bacteria via the scavenging of toxic 232 233 sulfide diffusing from the bulk of the sediment [51]. This finding has major practical 234 implications as it suggests that the radius of influence of the oil-spill snorkel may extend far from where the rod is positioned. Although the feasibility of the oil-spill snorkel has 235 236 been demonstrated only at the laboratory scale, the technology has a very low energetic 237 and environmental footprint (e.g., no energy input or maintenance required) and could be ideally applied for long-term remediation of contaminated sediments in remote open 238 239 sea areas.

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241 Oil spills in the deep sea

The DWH spill (Gulf of Mexico, April 2010) was the first oil spill originating in the deep sea (1544 m below the surface level, bsl). Hydrostatic pressure (HP) and temperature gradients, alongside the large injection of dispersants, fractionated petroleum HCs along the water column, with the slow buoyant migration of light or dispersed HCs forming multiple plumes at 800-1300 m bsl (about 8-13 MPa and 5°C) [52,53]. Bacteria

247 detected in the oil plume were subjected to microbial succession, with HC composition and quantity proposed to account for such a change [5,54,55]. While the 248 249 main physiological drivers for the microbial community shift remain uncertain [56], there is a consensus about the initial enrichment of Oceanospirillales and pseudomonads 250 (May 2010) followed by a shift in dominance to Colwellia, Cycloclasticus, 251 252 Pseudoalteromonas and methylotrophs (until August 2010). Ammonium and dissolved 253 inorganic nitrogen (DIN) were unrelated to plume samples [5,54]. Genes related to the 254 transport of iron or nitrogen- and phosphorous-based compounds [57] or their metabolism (e.g., nitrate and nitrite reduction, [58]) were highly transcribed, but their 255 256 correlation with plume samples or microbial succession is uncertain [5,20,54,57,58]. The gene expression for sulphur cycling and sulphite reduction increased, with sulphate 257 reduction the postulated activity, despite O₂ levels that remained relatively high [5,58]. 258

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260 Microbial succession was more difficult to assess in the deep-sea sediments because of the patchy oil distribution. Marine snow pulses resulting in oil deposition after June 2010 261 fuelled O₂ respiration on the surface of the seafloor (enriching Roseobacter, 262 Verrucomicrobiaceae and Bacteroidetes until October 2010). Anoxic microniches were 263 favoured and Deltaproteobacteria and other anaerobes developed [59]. DIN, ammonium 264 and total petroleum HCs were correlated in surface sediments and denitrification-related 265 genes were highly transcribed [60]. Deltaproteobacteria were also enriched in 266 subseafloor samples [61]. One year after the DWH spill, surface sediments around the 267 wellhead area were enriched with Actinobacteria, Firmicutes, Chloroflexi and 268 methylotrophic bacteria [62]. However, a time-dependent survey suggested that the 269

270 microbial community of the sediments had returned to pre-spill conditions, with a 271 response to petroleum only individuated at a finer taxonomic level (as with the obligate 272 polycyclic aromatic HC-degrader *Cycloclasticus* [59]). Continued sedimentation 273 imposed anaerobic conditions on the remaining HCs, with total PAHs in 3-cm deep 274 sediments slowly approaching pre-spill levels [63].

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276 Understanding cooperation, competition and succession of microorganisms after an oil 277 spill in the deep sea is key to predicting the success of the oil cleanup. Considering the difficulty of *in situ* studies at extreme depths, testing physiological response of isolates 278 279 to the specific environmental conditions, such as temperature or HP, is a necessary 280 complementary effort to explain changes in bacterial community composition and activity. However, the impact of HP on the HC degradation physiology after DWH was 281 neglected and recent studies show that HP can affect the metabolism of oil-degrading 282 283 bacteria. It has been recently proposed that the impaired metabolic response of Alcanivorax spp., which are ubiquitous and typically the first microorganisms identified 284 after surface oil spills, to HP explains the lack of their detection in the DWH deep oil 285 plume [64-66]. The DWH deep-sea plume [67] and sediment [61] revealed a low 286 Alcanivorax abundance, which was unrelated to hydrocarbon concentrations [61], 287 contrary to the abundance of other Oceanospirillales [5]. Consequently, the contribution 288 289 of Alcanivorax to oil degradation in the plume was considered negligible [61]. Meanwhile, results from oiled beach sands [68], oil mousses collected on surface 290 291 waters [62] and plume samples [14,67] showed a significant Alcanivorax abundance after cultivation under atmospheric pressure. Alcanivorax isolates were also obtained in 292

enrichment cultures from HC-free, deep-sea water samples collected from depths up to5000 m below sea level [69].

295 We call these observations the 'Alcanivorax paradox', i.e., the lack of response of ubiguitous HC-degrading bacteria to HCs in the deep sea. All of the isolation studies on 296 the deep sea neglected to consider HP as a contributing variable despite it being a 297 298 unique feature of deep-water environments. Recent data showed that a HP of 10 MPa 299 (comparable to that in the DWH plume) inhibits growth of three Alcanivorax species (A. 300 borkumensis, A. dieselolei and A. jadensis) on dodecane as a sole carbon source and that even a HP of 5 MPa causes a significant reduction in cell replication [64-66]. Under 301 302 HP transcription, multimeric protein complexes such as ATPase and ribosomes are 303 increased as compared to those under surface-water-resembling conditions. The respiratory chain shifts from cytochrome oxidases to reductases, alongside an 304 enhanced expression of genes of the Na⁺-translocating NADH reductase complex 305 (RNF-NQR). Translation, energy generation and electron transport are typical targets of 306 307 microbial piezoadaptation (Box 2). Under HP, the A. borkumensis SK2 strain synthesizes the osmolyte ectoine [65,66], which has been suggested to be a pressure-308 309 responsive compound (i.e., piezolyte). Synthesis of ectoine is energy intensive and 310 does not provide apparent advantages to Alcanivorax growth under HP [65,66].

311

312 Concluding remarks and future perspectives

313 Due to increasing demand for energy worldwide, the exploration of novel oil fields has 314 increased. These efforts increase the risk of exposure of marine life and the marine 315 environment to HCs [8]. Marine microorganisms are the ultimate HC degraders and play

316 a key role in cleanup events [5]. However, they can be exploited more efficiently if the 317 black box of their response to multifactorial environmental and pollution parameters is opened and fundamental questions on their metabolism, physiology and ecology are 318 319 answered (see Outstanding Questions). Meta-omics data are largely contributing to elucidate how environmental factors drive the assembly and function of HC-degrading 320 321 communities. More players than those currently known are revealed and they cooperate 322 to add their 'incomplete' metabolic pathways to HC degradation [14]. Metabolic 323 complementation and synthrophy could be as actual strategies for effective oil HC 324 cleanup in marine ecosystems. This suggests that biostimulation and bioaugmentation 325 approaches, including autochthonous bioaugmentation, should be carefully rethought 326 [31]. Furthermore, the recent advancements in understanding metabolism, physiology and adaptation of HC-degrading microorganisms are contributing to explain their 327 328 successes and failures in various polluted environments. For instance, the impaired 329 response of Alcanivorax to HP and its associations to different polluted compartments following the DWH disaster clarifies its absence in the deep-sea oil plume [65]. 330 Similarly, recent developments in microbial bioelectrochemistry are contributing to the 331 332 design of novel solutions for HC cleanup through the channelling of electron flows, such as that exploited by the "oil-spill snorkel," to compensate for the absence of oxygen in 333 334 contaminated sediments [50]. For certain, novel discoveries in microbial ecology and 335 physiology associated with HC pollution in marine ecosystems will enlighten and lead to 336 more efficient and sustainable microbial biotechnology approaches to ridding the oceans of oil. 337

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Figure Legends

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Figure 1. Current (Non-Microbial) Technologies for Emergency Responses to Oil 341 Spills in Marine Environments. On the sea surface, the oil spots accidentally released 342 343 by oil tankers or offshore platforms can be chemically dispersed or physically contained by plastic booms and partially recovered with skimmers or they can undergo in situ 344 burning. These processes may enhance accumulation of solid residues (e.g., tar balls) 345 that can sink. The volatile fractions move to the atmosphere where the HCs can be 346 transported far away and fall back to the surface again. HCs can reach the coast and 347 contaminate beaches where they can be removed with oil-sorbent materials. On rocky 348 349 beaches, combinations of high-pressure water spray and application of sorbent 350 materials are used. A further possibility is the removal of the contaminated sand and exsitu treatment in specialized centres. HC contamination can occur at the seafloor due to 351 losses from oil wells and shipwrecks or during drilling operations. Oil droplets can then 352 353 move across the water column towards the seafloor, spread horizontally to form HC plumes in the water column or reach the sea surface and eventually the coastal zone. 354 355 Dispersants are used to decrease the size of oil droplets and increase the surface-to-356 volume ratio. Oil from sunken ships can be removed through suction from intake platforms positioned on the sea surface, while wellhead capping is necessary to stop oil 357 blowout from deep-sea wells. Natural attenuation and dispersion phenomena in the 358 water column and volatilization in the atmosphere additionally contribute to HC removal. 359 360

361 Figure 2. Biotechnological Approaches to the Remediation of Hydrocarbon (HC)-362 Polluted Marine Environments. The activity of HC-degrading microorganisms can be enhanced by nutrients and biosurfactants, provided alone or in combination. Besides 363 the HC-degraders, microbes playing key roles such as biosurfactant producers are 364 pivotal to set up a successful intervention. Bioaugmentation (BA) of HC-degrading 365 366 microbes selected under laboratory conditions can be used to enhance HC degradation 367 rates. A strategy to overcome the reported lack of adaptability of allochthonous 368 microbes consists of autochthonous bioaugmentation (ABA) using isolates obtained from the matrix to be treated. Biostimulation is also used to enhance the activity of the 369 370 native and/or the augmented microbial communities and biosurfactants can be added to increase HC bioavailability. By exploiting suitable bio-carriers, nutrients, biosurfactants 371 and degrading microbes can be mixed to formulate slow-release particles (SRP) that 372 allow continuous and homogeneous release of the active ingredients in the target 373 374 environments. SRP applied to the sediments may contain alternative oxidants (e.g., NO₃⁻). Microalgae are involved in the water column cleanup following an oil spill. 375 Supporting their growth would enhance marine snow formation, which results in HC 376 377 precipitation to the seafloor. This might limit the oil-impacted seafloor area before deep-378 sea currents drive the oil plume far from the HC-spill origin and provide nutrients to 379 microbial HC-degrading populations on the seafloor. All biotechnological solutions must 380 take into account the environmental conditions of the system, including temperature and 381 hydrostatic pressure.

383 Figure 3. Natural and Engineered Biodegradation Processes in Marine Sediments 384 Contaminated by Petroleum Hydrocarbons. (A) Steady-state stratification of hydrocarbon-fuelled respiratory metabolisms based on the availability of terminal 385 electron acceptors. In the most superficial sediment layer (ranging from a few 386 millimetres to several centimetres depending on site characteristics), microorganisms 387 388 respire the oxygen that diffuses from the overlaying seawater. Below the oxic zone, nitrate, Mn⁴⁺ and Fe³⁺, if present, are used for anaerobic respiration. In the lower 389 390 sedimentary layer, sulfate, which is not a limiting factor in marine environments, becomes the dominant respiratory electron acceptor. (B) Schematic representation of 391 392 the "oil-spill snorkel", a conductive graphite rod half-buried in the contaminated 393 sediment that creates an electrochemical connection between the anoxic sediment and the oxic overlaying seawater. In principle, the "snorkel" may accelerate HC oxidation by 394 both serving as a direct electron acceptor in the respiratory metabolism of electro-active 395 396 bacteria growing at its surface and by stimulating, in the bulk of the sediment, the metabolism of sulphate-reducing bacteria via sulphide scavenging. (C) Schematic 397 representation of a modular slurry system (MSS) designed for stimulating aerobic 398 biodegradation processes in otherwise anoxic sediments. The system allows in situ 399 aeration of the sediment, temperature control, and also possibly the delivery of nutrients 400 401 and/or other biostimulating agents to increase, for instance, the bioavailability of 402 sediment-bound contaminants.

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Table1. Main cultivated marine hydrocarbon-degrading bacteria and their phylogenetic, physiological and ecological

features.

A/AN	Genome	Microorganism	Phylogeny	Target substrate	Habitat/Ecology	Physiology	Refs
A	Available	Alcanivorax borkumensis	<i>γ-proteobacteria,</i> Alcanivoracaceae	<i>n</i> -alkanes	Seawater, sediment, beach sand, coastal salt marsh	Biosurfactant producer, OHCB	[70]
A	Available	Alcanivorax dieselolei	y-proteobacteria, Alcanivoracaceae	<i>n</i> -alkanes	Seawater, sediment	Resistance to mild pressure increase, OHCB	[69]
A	Available	Marinobacter hydrocarbonoclasticus:	γ-proteobacteria, Alteromonadaceae	<i>n</i> -alkanes, PAHs	Seawater, sediment	Biofilm producer; oil surface colonizers	[71]
Þ	Available	Cycloclasticus pugetii	ƴ-proteobacteria, Piscirickettsiaceae	PAHs	Sediment	Highly efficient transport systems for the capture of nutrients and oligo- elements	[72]
A	Available	Oleispira antarctica	ƴ-proteobacteria, Oceanospirillaceae	<i>n</i> -alkanes	Seawater	cold-adapted OHCB	[73]

A: aerobic; AN: anaerobic; OHCB: obligate hydrocarbonoclastic bacteria; PAH: polycyclic aromatic hydrocarbons.

1 Text Boxes

2 **BOX 1.** The use and impacts of dispersants following oil spills.

There are consolidated approaches for dispersant administration on oil-polluted surface 3 waters. In general, a dispersant administration protocol takes into account the trade-off 4 5 between the efficiency of dispersion into fine droplets and the increased toxicity of 6 water-accommodated oil fractions to marine life. It has been recently highlighted that dispersants may alter and steer the diversity and activity of microbial communities with 7 potential, not yet understood effects on oil HC degradation [7]. There are no 8 formulations specifically designed for deep-sea releases. It is believed that in this case 9 no solvents are needed as the light HC oil components are all there, and mixing is 10 11 favoured by natural gas that is released from the crude oil as it ascends through the water column. During the DWH oil spill, about seven million litres of the Corexit 12 dispersant were injected into the surface of the Gulf of Mexico and deep-sea waters 13 [74]. Key Corexit components such as the anionic surfactant dioctyl sodium 14 sulfosuccinate were not biodegraded in situ [75], while addition of Corexit in ex situ tests 15 inhibited HC biodegradation and shifted microbial communities towards potential 16 17 dispersant degraders (e.g., Colwellia; [7]).

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BOX 2. Effects of hydrostatic pressure on microbial cells.

Hydrostatic pressure (HP) is an intrinsic feature of the deep sea. HP linearly increases with depth (about 0.1 MPa every 10 m of seawater column) and affects cell structures and functions, with a genome-based response proposed for piezophilic adaptation. Le Châtelier's principle predicts that processes entailing volume reduction are favoured

24 under HP. HP can affect chemical reactions (both for volume change at equilibrium and for activation) and macromolecule structures, particularly the weak non-covalent bonds 25 and protein complexes with several subunits. This is the case for ribosome assembly or 26 27 for nucleic acid/protein complexes with volume increases [76-78], where effects are generally observed at sub-lethal pressures. Another recognized HP effect is on cell 28 membranes, whose relative abundance in mono- (and to a lower extent poly-) 29 30 unsaturated fatty acids, necessary to maintain membrane fluidity and cell homeostasis, 31 can be affected. A high degree of fatty acid unsaturation does not favour fatty acid acyl 32 group packing due to the physical encumbrance of the lateral chains, with positive 33 effects on the cell membrane's homeoviscous properties and curvature elastic stress [79]. Increased unsaturation of fatty acids on the membrane may result from another 34 typical deep-sea condition, *i.e.*, low temperature, whose effects are additive to HP [80]. 35 HP affects membrane transport systems and transmembrane enzymes, such as 36 ATPases and cytochromes, due to direct effects on enzyme folding or the lipid 37 environment on the membrane. Adaptation to HP potentially involves energy generation 38 in the cell, likely to counteract HP-related stressing effects. The model piezophile 39 40 *Photobacterium profundum* SS9 possesses two complete operons for the F₀F₁ ATPase and multiple cytochrome sets, supporting the pivotal importance of electron and proton 41 transport under high HP [81]. Similarly, Shewanella piezophila possesses two 42 43 respiratory chains, active either under low or high HP [82]. The capacity to offset HP impact on cell turgor pressure has been little studied. This would involve the intracellular 44 accumulation of piezolytes, as occurring for osmolytes and salinity. While their 45 mechanism of action remains unclear, HP increase is consistent with the accumulation 46

of N-trimethylamine oxide (TMAO) [83], β-hydroxybutyrate [84] and ectoine [65]. In *Alcanivorax borkumensis*, intracellular ectoine accumulation and gene upregulation
under HP were correlated with decreased cell damage and higher cell number, but
culture activity was not increased. Experiments on the synergistic effects of osmotic and
HP increase suggest that ectoine water-reclamation capacity might also explain its
function at high HP [66].

55 Glossary Box

56 **Bioaugmentation:** The addition of active microorganisms with specialized metabolic 57 capacities to enhance a metabolic process, such as HC degradation in a polluted site.

Bioremediation: The exploitation of the catabolic abilities of microorganisms that use pollutants (such as HCs) as a source for carbon and energy for their metabolism and growth. Two general approaches can be defined in which the polluted material is left on site (*in-situ* bioremediation) or is removed and treated away from the polluted site (*exsitu* bioremediation). It can be performed by indigenous microbial communities or by allochthonous microorganisms added to the polluted system through bioaugmentation.

Biostimulation: The modification of the environmental conditions of a polluted matrix to favour the metabolism of microorganisms capable of bioremediation, generally consisting of the addition of nutrients, but also of oxygen or other electron acceptors and eventually electron donors (for instance in the case of reductive dechlorination), or of the addition of substances enhancing HC bioavailability (*e.g.,* biosurfactants).

69 **Biosurfactants:** Amphipatic surface-active macromolecules produced by microorganisms to increase the availability of HCs. They can be grouped as low-70 71 molecular-weight molecules with good solubilization properties (lipopeptides, 72 phospholipids, but most commonly glycolipids e.g., rhamnolipids, trehalose lipids, sophorolipids) or high molecular weight compounds with coating properties preventing 73 oil-droplet coalescence (e.g., proteins, polysaccharides, lipids and their complexes). 74

Meta-omics: Molecular analyses based on the comprehensive study of enzymes and
 proteins (metaproteomics), RNA and DNA sequences (metatransciptomics and
 metagenomics) or metabolites (metabolomics) in a given environmental sample.

Micro-encapsulation: The incorporation of different chemical components (such as nutrients, enzymes, microorganisms, etc.) within a coating to form small capsules in the size range of micrometers to a few millimetres, to protect the content from early degradation and favour its localized and time-dependent release.

Microbial succession: The temporal change in dominant species within a microbial community in response to the changing environmental conditions, such as the changes in the type and concentrations of carbon sources.

Mobile genetic elements: Genetic elements such as plasmids or transposons that contain genetic modules allowing their movement within the genome of an organism or between genomes of different organisms. They may carry catabolic genes that enable the cells that express them to degrade specific compounds.

Natural attenuation: The process of HC biodegradation following an oil spill, mediated 89 90 by the indigenous microbial community. This process relies upon the presence of 91 microorganisms with HC-catabolic capacities in the ecosystem prior to the spill. The availability of HC substrates enriches the microorganisms that are capable of exploiting 92 HCs for growth. Natural attenuation is made possible by consortia of microorganisms 93 94 with complementary functions. These microorganisms include HC degraders, biosurfactant producers, and nutrient providers involved in N and P cycles or 95 siderophore producers. 96

97 Water column: the term conceptually represents all the different layers of water present 98 in a water body, such as the oceans, moving from the surface to the sediment on the 99 bottom. Along the water column, different layers can be identified and their boundaries 100 are defined by specific water depth.

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Competing interests

The authors declare no competing financial interests.

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Biotechnologies for marine oil spill cleanup: indissoluble ties with microorganisms

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Trends Box (Max 900 characters w/spaces. Actual: 898)

Trends Box

The cleanup of oil spills in marine environments ultimately relies on microbial metabolism of hydrocarbons (HC), which complements the current chemico-physical techniques used in emergency response.

Consolidated biotechnologies include microbial communities biostimulation, biosurfactant supplementation and bioaugmentation HC-degrading microbial cells.

The effectiveness of biotechnologies is limited by our understanding of the microbial ecology of polluted marine systems. We lack knowledge on how environmental factors, such as hydrostatic pressure, temperature and dispersant toxicity, affect microbial successions.

The recent availability of meta-omics data and the improved understanding of microbial metabolism are leading to novel biotechnologies for marine oil spill cleanup, such as slow-release particles for efficient biostimulation and bioelectrochemical approaches for sediment cleanup.

Biotechnologies for marine oil spill cleanup: indissoluble ties with microorganisms

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Outstanding Question Box (Max 2000 characters w/spaces. Actual: 1664)

1 Outstanding Question Box

The continuous increasing understanding of microbial physiology, metabolism and 2 ecology is driving the development of novel, microbially driven biotechnological 3 applications to the oil hydrocarbon (HC) cleanup in marine environments. Among those 4 5 described in this review, the application of slow-release particles (SRPs), the "oil-spill 6 snorkel" and modular slurry systems are potentially promising biotechnologies for the remediation of polluted water and sediments based on the activity of microorganisms or 7 their products. However, a series of questions and unresolved issues regarding both 8 basic and applied science remain. These guestions are: 9

10 1) How do we improve the control of component release by SRPs?

How do we select novel, improved and versatile biosurfactants or modify existing
 ones for applications as components of biodispersants in a cost-effective way?
 What is the fine structure and dynamics of the full functional and metabolic
 cooperation between microbial components of HC-degrading microbial consortia?
 How do neglected factors such as hydrostatic pressure affect the response of

16 autochthonous natural microorganisms in polluted marine compartments?

17 5) How do combinations of environmental stressors act on HC-degraders?

18 6) Is there a biogeographically driven distribution of marine microbial HC degraders?

19 7) What is the prevalence and functional importance of unculturable HC degraders that

20 are emerging from recent meta-omics reconstructions?

8) What is the balance and diversification of electron flow in the "oil-spill snorkel"?

9) Is it advantageous to combine different biotechnologies, for instance SRPs and the
"oil-spill snorkel"?

2

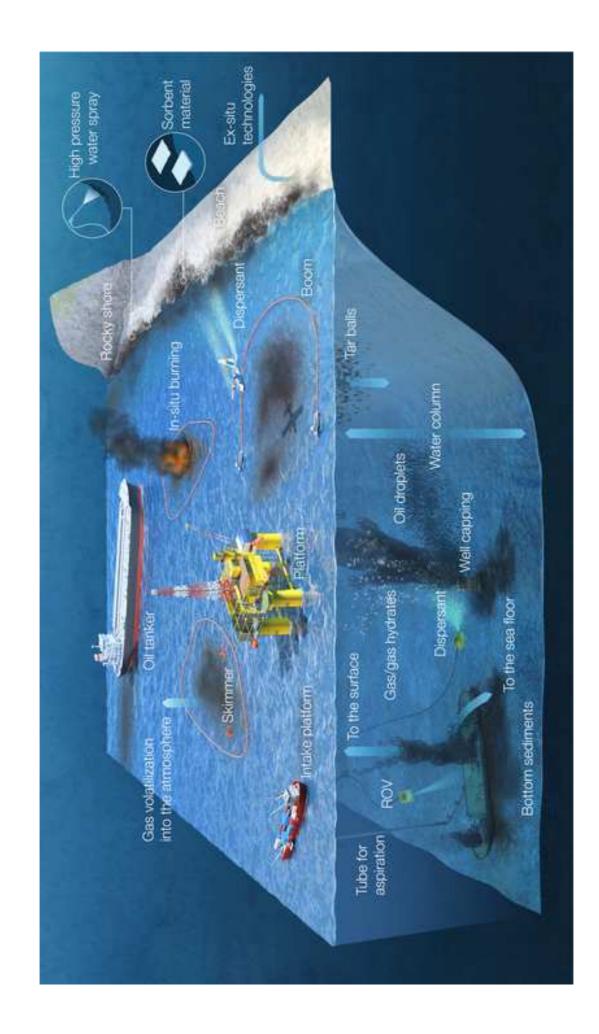
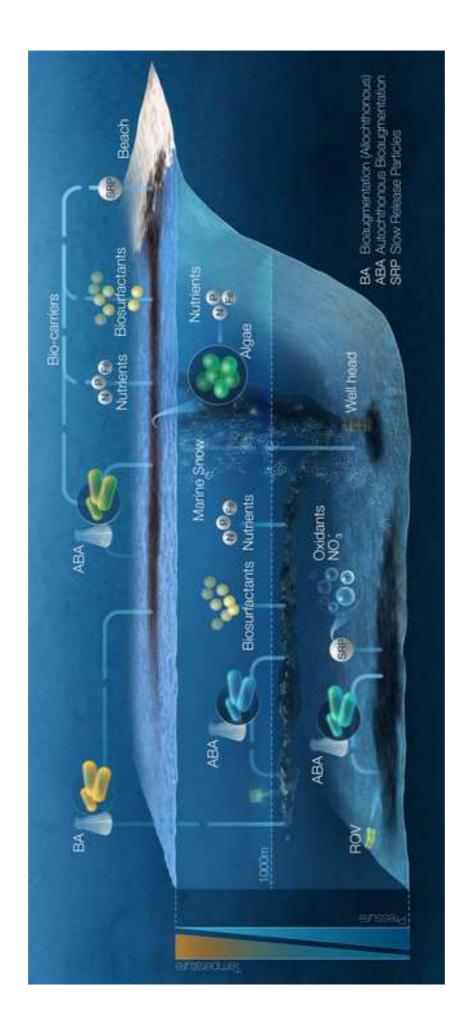
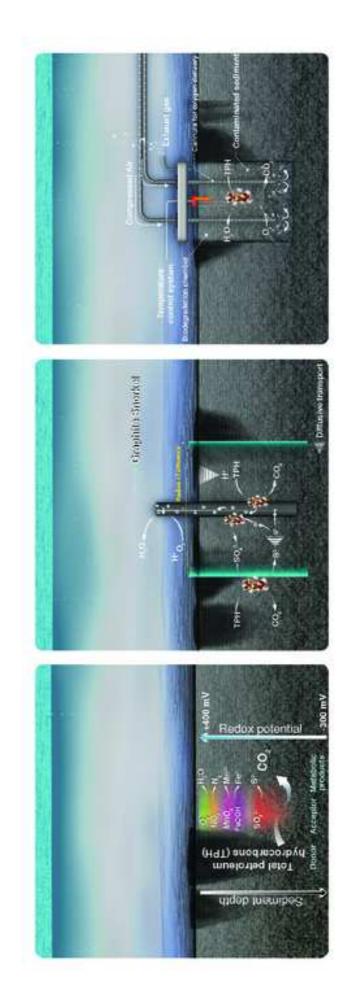


Figure 1





Biotechnologies for marine oil spill cleanup: indissoluble ties with microorganisms 3

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ÂN ÂN			A	> Z	A Y	A ≺	> ≺	A Y	A Y	A/AN Genome
	Pseudomonas stutzeri	Pseudomonas pachastrellae	Oleiphilus messinensis	Oleibacter marinus	Oleispira antarctica	Cycloclasticus pugetii	Marinobacter hydrocarbonoclasticus:	Alcanivorax dieselolei	Alcanivorax borkumensis	e Microorganism
	<i>γ-proteobacteria, Pseudomonadaceae</i>	<i>γ-proteobacteria,</i> Pseudomonadaceae	<i>y-</i> proteobacteria, Oleiphilaceae	<i>y-</i> proteobacteria, Oceanospirillaceae	<i>γ-proteobacteria,</i> Oceanospirillaceae	γ-proteobacteria, Piscirickettsiaceae	<i>y-proteobacteria, Alteromonadaceae</i>	<i>y-</i> proteobacteria, Alcanivoracaceae	<i>y-proteobacteria, Alcanivoracaceae</i>	Phylogeny
	<i>n</i> -alkanes, PAHs, BTEX	<i>n</i> -alkanes, PAHs	<i>n</i> -alkanes	<i>n</i> -alkanes	<i>n</i> -alkanes	PAHs	<i>n</i> -alkanes, PAHs	<i>n</i> -alkanes	<i>n</i> -alkanes	Target substrate
)	Seawater, marsh and marine sediments, beach sand	Sediment, beach sand	Seawater, sediment	Seawater	Seawater	Sediment	Seawater, sediment	Seawater, sediment	Seawater, sediment, beach sand, coastal salt marsh	Habitat/Ecology
Key role in N	Biofilm producer	Bioemulsification activity	Biofilm producer on oil droplets, OHCB	Adapted to tropical marine environments	Cold-adapted OHCB	Highly efficient transport systems for the capture of nutrients and oligo-elements	Biofilm producer; oil surface colonizers	Resistance to mild pressure increase, OHCB	Biosurfactant producer, OHCB	Physiology
150 001	[19–21]	[17-19]	[16]	[14-15]	[13]	[11, 12]	[8-10]	[5-7]	[1-4];	Refs

Supplementary Table 1. A comprehensive list of main cultivated marine hydrocarbon-degrading microorganisms and

their phylogenetic, physiological and ecological features.

AN	AN	AN	A	A	⊳	A	Þ	A	A/AN
z	z	~	Z	Z	×	Y	~	~	Genome
Desulfococcus oleovorans	Desulfosarcina- Desulfococcus cluster strains	Desulfatibacillum alkenivorans	Sphingopixis sp.	Rhodobacter sp. SS12.29; Rhodococcus sp. ice-oil-488 s	Dietzia maris	Acinetobacter venetianus	Alteromonas naphthalenivorans	Thalassolituus oleivorans	Microorganism
δ -proteobacteria, Desulfobacteraceae	δ -proteobacteria Desulfobacteraceae	<i>δ-proteobacteria, Desulfobacteraceae</i>	α-proteobacteria, Sphingomonadaceae	α-proteobacteria, Rhodobacteraceae	Actinobacteria, Dietziaceae	γ-proteobacteria, Moraxellaceae	γ-proteobacteria, Alteromonadaceae	γ-proteobacteria, Oceanospirillaceae	Phylogeny
<i>n</i> -alkanes, aromatic HCs	Short chain <i>n</i> -alkanes	<i>n</i> -alkanes	PAHs	PAHs	<i>n</i> -alkanes, PAHs	<i>n</i> -alkanes	PAHs	<i>n</i> -alkanes	Target substrate
Sediment	Sediments of marine HC seeps	Sediment	Seawater	Seawater	Seawater, deep sea hydrothermal field	Surface water, sediment.	Seawater, tidal flat sediment	Surface seawaters, sediments, coastal and estuarine areas	Habitat/Ecology
Sulfate-reducing bacteria	Propane and butane degraders; sulfate-reducing bacteria	High metabolic versatility for anaerobic alkane utilization	Key role in reducing the accumulation of metabolites resulting from PAH degradation	Key role in reducing the accumulation of metabolites resulting from PAH degradation	Biosurfactant producer	Biosurfactant producer	<i>r</i> -strategist, fast growth in nitrogen-deficient seawater	ОНСВ	Physiology
[33,34]	[31,32]	[30]	[29]	[29]	[27,28]	[26]	[25]	[24]	Refs

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A/AN	Genome	Microorganism	Phylogeny	substrate	Habitat/Ecology	Physiology	Refs
A	А	Bacillus pumilus	Bacilli, Bacillaceae	<i>n</i> -alkanes, PAHs	Sediment	Resistance to heavy metals	[20,35]
A	Z	Bacillus stratosphericus	Bacilli, Bacillaceae	PAHs, BTEX	Seawater	High metabolic versatility, biosurfactant producer	[36]
AN	Y	Archaeoglobus fulgidus	Euryarchaeota, Archaeoglobaceae	<i>n</i> -alkanes	Shallow marine hydrothermal system	Extremophile	[37]
AN	Y	Thermococcus sibiricus	Euryarchaeota, Thermococcaceae	<i>n</i> -alkanes	Oil reservoir	High metabolic versatility	[38]
AN	¥	Ferroglobus placidus	Euryarchaeota, Archaeoglobaceae	Aromatic HCs	Shallow marine hydrothermal system	Hyperthermophilic	[39,40]
A	z	Dothideomycetes- related taxa	Fungi	PAHs	Beach sediment, tarballs, salt marshes	1	[41,42]
A: aer	obic; AN: a	AN: anaerobic; OHCB: oblig	A: aerobic; AN: anaerobic; OHCB: obligate hydrocarbonoclastic bacteria, PAH: polycyclic aromatic hydrocarbons; Y/N: Y	bacteria, P,	AH: polycyclic aror	romatic hydrocarbor	ns; Y/N: Y,

genome available for at least one strain/ N, genome not available (Genome availability checked on NCBI database on 27 November 2016).

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