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Poly(ethylene terephthalate)-carbon nanofiber nano composite for fiber spinning; properties and combustion behavior

Jenny Alongi,^{1*} Alberto Frache²

^{1*}Viale Teresa Michel 5, 15121 Alessandria, Italia; tel/fax: +39 0131 229337/+39 0131 229399; e-mail: jenny.alongi@polito.it; alberto.frache@polito.it
 ²Dipartimento di Scienza dei Materiali e Ingegneria Chimica, Politecnico di Torino, Corso Duca degli Abruzzi 24, 10129 Torino, Italy.

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Abstract: Carbon nanofiber (CNF)-polyethylene terephthalate (PET) blends were previously prepared by melt blending and, subsequently, melt spun in order to obtain nanostructured fibers characterized by high flame retardant properties and resistance to the combustion. The morphological analysis showed that CNFs are homogeneously distributed and finely dispersed within PET matrix. The mechanical properties in tensile testing of the fibers change in the presence of CNFs: the elongation at break increases, whereas the tenacity and the tensile strength decrease. The combustion tests by cone calorimetry reveal a relevant decrease of heat release rate, total heat evolved and total smokes released by the nanocomposites as compared to neat PET.

Keywords: carbon nanofibers; CNF; PET; nanocomposites;

Introduction

In the last twenty years, there has been a high level of interest in using filler particles having at least one dimension of nano-scale size (nanofiller) in order to reach substantial improvements with low filler loading (<5wt.-%) [1-4]. One of the most promising applications involves the improvement in flame retardancy properties of polymers added with nanofillers. The possibility to reduce flammability of numerous polymers using nanoclays have been shown by many authors and several mechanisms are proposed [5-17]. Indeed, nanocomposite systems are a new attractive alternative to conventional commercial flame retardants. Nowadays, the most common approach to nanocomposite fire retardants is based on the use of layered silicates characterized by a large aspect ratio. Carbon nanotubes and carbon nanofibers are suitable candidate as flame retardant: Kashiwagi and coworkers [13, 14] demonstrated that multi-walled carbon nanotubes can be used for reducing the flammability properties of polypropylene, and that it is possible also to use single walled carbon nanotubes for poly(methyl methacrylate) for the same purpose [15]. These fillers are also used as reinforcement. The in situ formation of a continuous surface protective charred layer embedding carbon nanotubes is playing a protective and remarkable role for the relevant reduction of total heat released as well as heat release rate during the combustion of carbon nanotubes polymer nanocomposite material [16-18]. Despite similarities between carbon nanofibers (CNFs) and carbon nanotubes, there are no published studies on CNF use for preparing polymer nanocomposites for the reduction of flammability. Ma et al. [19] reported on fibers spun from polyester/carbon nanofiber composites. In this study, different grades of CNFs have been melt blended with poly(ethylene terephthalate) and then spun into fibers; the authors compared compounding behavior using either twin-screw extruder, or two-roll mill, or high shear mixer, in order to establish the dispersion level of carbon nanofibers and the related mechanical properties. They observe that the tensile strength and modulus were not significantly increased by the addition of nanofibers, whereas compressive strength and torsional moduli of composite fibers were considerably higher than for neat PET fibers.

In this work, different grades of CNFs have been melt compounded with poly (ethylene terephthalate) and then spun into fibers. Morphology, thermal stability, mechanical and flame retardancy properties have been investigated.

Results and discussion

PET-CNF nanocomposites

-Morphology

Carbon nanofibers, CNFPS and CNFLHT, are cylindrical nanostructures of graphene arranged layers, characterized by a complicate bundling due to their length (Figure 1). Apparently, relevant differences in terms of purity grade are not visible.



Fig. 1. SEM micrographs of CNFPS (a) and CNFLHT (b) carbon nanofibers.



Fig. 2. FESEM micrographs at different magnifications of CNFLHT carbon nanofibers.

From the statistical investigation of FESEM images of the type shown in Figure 2, it has been found that in the case of CNFPS (Figure 3a), the major part of fibers have a

diameter in the range 40-80 nm (ca.80%), whereas in the case of CNFLHT 80% fibers fall in the range 40-100 nm (Figure 3b).



Fig. 3. Statistical data of CNFPS (a) and CNFLHT (b) carbon nanofibers.



Fig. 4. SEM micrographs of PET-CNFPS (a, b) and CNFLHT (c, d) nanocomposites.

When these fibers are introduced in the polymer matrix, a homogeneous and uniform distribution, as well as a high level dispersion are evident, as shown in Figure 4.

-Thermal stability

Thermal stability of PET changes in presence of CNFs, as shown in Table 1. In nitrogen, the CNFs anticipate the weight loss, whereas in air they delay the degradation. Indeed, the T_{onset} at 5% in the nanocomposites is lower in nitrogen and higher in air, respectively, as compared to PET; whereas, in either atmosphere, the T_{max} decreases for the presence of both the carbon nanofibers. In nitrogen volatilization occurs in a single step whereas a carbonization phenomenon is present in air with formation of a char residue (T_{max1}) which is oxidized at higher temperature (T_{max2}).

	Formulation	T _{onset 5%} [°C]	T _{max1} [°C]	T _{max2} [°C]	Residue at 800 °C [%]
Nitrogen					
	PET	419	461	-	9.5
	PET-CNFPS	411	455	-	15.8
	PET-CNFLHT	415	456	-	18.9
Air					
	PET	402	457	577	1.0
	PET-CNFPS	410	457	568	0.4
	PET-CNFLHT	404	457	555	0.6

 Tab. 1. TGA data of PET-CNF nanocomposites.

DSC data of Table 2 and in Figure 5a show that the crystallization temperature (T_c) increases in the presence of CNFs acting as nucleating agents. It is also seen that the nanocomposites show a slightly lower melting temperature (T_m) as compared to neat PET (Figure 5b). Finally, the crystallinity (X_c) of PET is increased from 30 up to 34% for both types of CNFs (Table 2).

Tab. 2. DSC data of PET-CNF nanocomposites.

Formulation	T_{c}	Δ	T_m	Xc
	[°C]	[°C]	[°C]	[%]
PET	200	-	255	30
PET-CNFPS	213	+13	249	34
PET-CNFLHT	215	+15	251	34

-Combustion behavior

The cone calorimetry data are shown in Table 3 and represented graphically in Figure 6. Regarding TTI, both formulations give a decrease of the ignition time (22%). The pkHRR decreases from 523 to 340 and 318 kW/m² for PET-CNFPS and PET-CNFLHT (35 and 39%, respectively). An average 15% reduction in smoke production (TSR and SEA) is observed for CNFLHT based nanocomposite whereas it is negligible in the case of CNFPS.

The flammability behavior of the samples prepared is completed by LOI test: LOI value increases from 21 up to 23 for both formulations (+10%) (Table 3).



Fig. 5. DSC 2nd (a) and 3rd cycle (b) curves of PET-CNF nanocomposites.

Sample	TTI±σ	Δ [%]	pkHRR±σ [k\///m ²]	Δ [%]	TSR±σ [m ² /m ²]	Δ [%]	SEA±σ	Δ	LOI [%]
PET	209±4	-	523±60	-	2937±252	-	403±26	-	21
PET- CNFPS	163±15	-22	340±55	-35	2782±109	-5	380±20	-6	23
PET- CNFLHT	164±5	-22	318±21	-39	2520±62	-14	333±16	-17	23

Tab. 3. Combustion data of PET-CNF nanocomposites.

These results confirm the capability of carbon nanofibers to enhance the fire stability of PET. In literature, usually the employ of carbon nanotubes as flame retardant additives has been well documented. Indeed, their highly elongated shape and thus the high aspect ratio play a crucial factor in the reduction of the flammability of some polymers such as polypropylene [21, 22], poly methyl methacrylate [23] and poly ethylene vinyl acetate [24]. Analogously, carbon nanofibers can behave similarly to carbon nanotubes as flame retardants promoting the *in situ* formation of a continuous network structured protective layer from the tubes. This layer acts as physical barrier to oxygen and heat transfer for the polymer leading to a significant reduction in the heat release rate [5].



Fig. 6. HRR curves of PET-CNF nanocomposites.

PET-CNF fibers

The mechanical characteristics of the fibers from CNFPS described in Table 4 show an increase of the elongation at break of 22% and a decrease of the tensile strength and tenacity of 53 and 35%, respectively if compared to neat PET.

Tab. 4. Mechanical	properties of PET fibers.
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Mechanical properties	PET-CNFPS05	Reference	Δ[%]
Elongation at break [%]	34.4	28.1	+22
Breaking load [cN]	452.1	691.0	-34
Tenacity [cN/tex]	26.1	40.0	-35

The combustion behavior of PET-CNFPS05 textile is compared in Table 5 and Figure 7 with data obtained from the commercial benchmark formulation PET-FR05. TTI decreases for the presence of both fillers: for PET-CNFPS05 the variation is of -20%, while for PET-FR05 is -16%. Regarding the HRR, the related peak decreases in a drastic manner in presence of both CNFPS05 and FR05 (45 and 44%): indeed, the pkHRR decreases from 510 to 282 and 288 kW/m² for PET-CNFPS05 and PET-FR05, respectively. Regarding TSR, a decrease of 12 and 40 % has been measured for PET-FR05 and PET-CNFPS05, respectively. Unfortunately, the optical density of smoke identified by SEA value increases in presence of CNFs when compared with neat PET.

	TTI±σ	Δ	pkHRR±σ	Δ	Residue± <i>a</i>	Δ	TSR±σ	Δ	SEA±σ	Δ
Sample	[s]	[%]	[kW/m ²]	[%]	[%]	[%]	$[m^2/m^2]$	[%]	$[Kg/m^2]$	[%]
PET	128±10	-	510±50	-	17±2	-	545±54	-	355±36	-
PET- FR05	108±5	-16	288±31	-44	9±1	-47	481±1	-12	665±4	+48
PET- CNFPS05	102±2	-20	282±28	-45	14±6	-18	329±30	-40	476±11	+34

Tab. 5. Combustion data of PET fibers.



Fig. 7. HRR curves of PET fiber.

Experimental part

Materials

Poly(ethylene terephthalate) was provided by Sinterama S.p.A., whereas CNFs were purchased from Pyrograf[®] Products Inc. Two types of CNFs having different purity grade (of carbonaceous residue amount) were used as received: CNFPS and CNFLHT. Table 6 summarizes their characteristics.

Tab. 6. Structures and codes of the carbon fibers used.

Trade name	Code	Characteristics	Nanofiber grade
Pyrograf III PR-24-XT- PS	CNFPS	Pirolytically stripped carbon fibers	Medium purity
Pyrograf III PR-24-XT- LHT	CNFLHT	Heat treated carbon fibers	High purity

Fire retardant commercial PET textile STOPFIRE[®] supplied by Sinterama S.p.A. was used as a benchmark. The code used within this work is PET-FR05.

Preparation of PET-CNF nanocomposites

Melt blending for preparation of 5wt.-% CNFPS- and CNFLHT- PET masterbatches was carried out using a twin screw extruder (Leistritz ZSE 18 HP 40D) which temperature and screw profile is described in Figure 8. Before melt blending, materials were dried in a vacuum oven (8 h at 120 °C).



Fig. 8. Temperature and screw profile of the extruder.

Preparation of PET-CNF specimens

The specimens for cone calorimetry (100X100X6 mm) and LOI test (80X6.5X3 mm) were prepared by injection molding.

Spinning of PET-CNF fibers

Spinning was performed diluting the PET-CNF compound to 0.5wt.-% inorganic content at 285 °C and 3200 m/min producing PET-CNF fibers. Filament number of fibers was 48 and the count was 173.48 Dtex. The fibers have been then textured (drawn and bulked) to obtain textile thread (textured yarn) used for the flammability tests. Density of textile fabrics was 152 g/m².

Scanning Electron Microscopy (SEM)

Observations were carried out using a Leica Electron Optics 435 VP SEM with an acceleration voltage of 15kV and a 20 mm working distance. Specimens were mounted on aluminum stubs with double sided adhesive tape. The metallization with gold was performed in rarefied argon atmosphere (20Pa) using an Emitech K550 Sputter Coater, with a current of 20 mA for 180 s.

In order to investigate the average diameter of CNFs taken in consideration, a statistical investigation using the software IMAGEJ was performed on a series of 100 fibers selected in SEM micrographs.

Field Emission Scanning Electron (FESEM)

Examinations were performed with a Zeiss Supra_ 40 on the surface of fragile fracture from samples fractured after cooling by immersion in liquid nitrogen. The samples were coated with chromium.

ThermoGravimetric Analysis (TGA)

A TA instruments Q500 thermo-balance provided with a platinum pan was used. The measures are conducted in nitrogen or air atmosphere: $(60 \text{cm}^3/\text{ min})$ in the range 50-800°C with heating rate of 10 °C/ min. The experimental error was ±0.5% by weight. The data collected were T_{onset} (5% of weight loss), T_{max} (maximum rate of weight loss) and the final residue.

Differential Scanning Calorimetry (DSC)

Thermal analysis was carried out on a Model QA1000; TA Instrument DSC. Three successive cycles were performed following the scheme:

1st cycle: heating 25-280 °C at 20 °C/min and isothermal at 280 °C for 3 min.

2nd cycle: cooling 280-25 °C at 10 °C/min and isothermal at 25 °C for 3 min.

3rd cycle: heating conditions as 1st cycle.

Crystallinity (X_c) value was calculated following equations (1) and (2) [20]:

 $X_{c} = \Delta H_{c}^{*} 100 / \Delta H_{m}^{0} = \Delta H_{c}^{*} 100 / 140$ for PET

(1)

 $X_{c} = \Delta H_{c}^{*} 100 / \Delta H_{m}^{0} (1-\Phi) = \Delta H_{c}^{*} 100 / \Delta H_{m}^{0} (1-0.05), \text{ for PET-CNF nanocomposites}$ (2)

Oxygen consumption calorimetry (Cone calorimeter test)

A cone calorimeter (Fire Testing Technology, FTT) was used under a heat flux of 35 kW/m² according to ISO 5660-1. The parameters measured were time to ignition (TTI, s), heat release rate (HRR, kW/m²) and the relative peak (pkHRR, kW/m²), total heat release (THR, kW/m²), residue (%), total smoke release (TSR, m²/m²) and specific extinction area (SEA, Kg/m²). For each formulation, the test was repeated three times and standard deviation (σ) was reported for each measurement.

Limiting oxygen index (LOI)

LOI value was measured by oxygen index apparatus (FIRE) according to the ISO 4589-2 standard.

Mechanical test on fibers

The mechanical properties were measured directly on spun fibers using a Textechno, Statimat II model dynamometer according to the EN ISO 2062 standard.

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References

[1] Pinnavaia, T.J.; Beall, G.W. editors. *Polymer-clay nanocomposites*, John Wiley & Sons; 2001 New York.

[2] Utracki, L.A.; Sepehr, M.; Boccaleri, E. Polym. Adv. Technol. 2007, 18, 1.

[3] Ray, S.S.; Okamoto, M. *Prog. Polym. Sci.* **2003**, *28*, 1539 and references quoted therein.

[4] Alexandre, M.; Dubois, P. Mater. Sci. Technol. R 2000, 28, 1.

[5] Gilman, J.W.; Kashiwagi, T. SAMPE J **1997**, 33, 40.

[6] Gilman, J.W. Appl. Clay Sci. 1999, 15, 31.

[7] Gilman, J.W.; Jackson, C.L.; Morgan, A.B.; Harris, R.H.; Manias, E.; Giannelis, E.P.; Wuthenow, M.; Hilton, D.; Phillips, S.H. *Chem. Mater.* **2000**, *12*, 1866.

[8] Zanetti, M., Camino, G., Müllhaupt, R. Polym. Degrad. Stab. 2001, 74, 413.

[9] Zhu, J.; Uhl, F.M.; Morgan, A.B.; Wilkie, C.A. Chem. Mater. 2001, 13, 4649.

[10] Alexandre, M.; Beyer, G.; Henrist, C.; Cloots, R.; Rulmont, A.; Jerome, R.; Dubois, P. *Macromol. Rapid Commun.* **2001**, *22*, 943.

[11] Zhu, J.; Start, P.; Mauritz, A.; Wilkie, C.A. Polym. Degrad. Stab. 2002, 77, 253.

[12] Morgan, A.B.; Harris, R.H.; Kashiwagi, T.; Chyall, L.J.; Gilman, J.W. *Fire Mater* **2002**, *26*, 247.

[13] Kashiwagi, T.; Harris, R.H.; Zhang, X.; Briber, R.M.; Cipriano, B.H.; Raghavan, S.R.; Awad, W.H.; Shields, J.R. *Polymer* **2004**, *45*, 881.

[14] Kashiwagi, T.; Grulke, E.; Hilding, J.; Groth, K.; Harris, R.H.; Butler, K.; Shields, J.; Kharchenko T.; Douglas, J. *Polymer* **2004**, *45*, 4227.

[15] Kashiwagi, T.; Du, F.; Winey, K.I.; Groth, K.M.; Shields, J.R.; Bellayer, S.P.; Kim, H.; Douglas, J.F. *Polymer* **2005**, *46*, 471.

[16] Kashiwagi, T.; Fagan, J.; Douglas, J.F.; Yamamoto, K.; Heckert, A.N.; Leigh, S.D.; Obrzut, J.; Du, F.; Gibson, S.L.; Mu, M.; Winey, K.I.; Haggenmueller, R. *Polymer* **2007**, *48*, 4855.

[17] Bocchini, S.; Frache, A.; Camino, G.; Claes, M. Eur Polym J. 2007, 43, 3222.

[18] Bocchini, S.; Annibale, E.; Frache, A.; Camino, G. *e-polymer* **2008**, no. 20.

[19] Ma, H.; Zeng, J.; Realff, M.L.; Kumar, S.; Schiraldi, D.A. *Compos. Sci. Technol.* **2003**, *63*, 1617.

[20] Tjong, S.C.; Bao, S.P. J. Polym. Sci. Part B: Polym. Phys. 2004, 42, 2878.

[21] Kashiwagi, T.; Harris, R.H.; Zhang, X.; Briber, R.M.; Cipriano, B.H.; Raghavan, S.R.; Awad, W.H.; Shields, J.R. *Polymer* **2004**, *45*, 881.

[22] Kashiwagi, T.; Grulke, E.; Hilding, J.; Harris, R.; Awad, W.H; Douglas, J.F. *Macromol Rapid Commun* **2002**, *23*, 761.

[23] Kashiwagi, T.; Du, F.; Winey, K.I.; Groth, K.M.; Shields, J.R.; Bellayer, S.P.; Kim, H.; Douglas, J.F. *Polymer* **2005**, *46*, 471.

[24] Kashiwagi, T.; Grulke, E.; Hilding, J.; Groth, K.M.; Harris, R.; Butler, K.; Shields, J.R; Kharchenko, S.; Douglas, J. *Polymer* **2004**, *45*, 4227.