

# CHARACTERIZING DROPLET BREAKUP RATES OF SHEAR-THINNING DISPERSED PHASE IN MICROREACTORS

Voon-Loong Wong<sup>a,b,\*,1</sup>, Katerina Loizou<sup>b,2</sup>, Phei-Li Lau<sup>a</sup>, Richard S. Graham<sup>c</sup>, and  
Buddhika N. Hewakandamby<sup>c</sup>

<sup>a</sup>Department of Chemical and Environmental Engineering, University of Nottingham  
Malaysia Campus, Semenyih 43500, Malaysia

<sup>b</sup>Department of Chemical and Environmental Engineering, University of Nottingham,  
Nottingham NG7 2RD, United Kingdom

<sup>c</sup>School of Mathematical Sciences, University of Nottingham, Nottingham NG7 2RD,  
United Kingdom

## ABSTRACT

A two-phase flow predictive model with the integration of conservative level-set method (LSM) and Carreau-Yasuda constitutive equation was developed herein. The LSM was chosen as a potential interface capturing scheme for elucidating the interfacial phenomena including insight into the mechanism of shear-thinning droplets. In present paper, the dynamics of shear-dependent droplet emergence, growth, detachment and translocation in a Newtonian microsystem were examined via computational fluid dynamics (CFD) analysis. Dilute sodium carboxymethylcellulose (Na-CMC) solution was treated as dispersed phase ( $70 \text{ mPa}\cdot\text{s} < \eta_o < 10.2644 \text{ Pa}\cdot\text{s}$ ) whereas the olive oil ( $68 \text{ mPa}\cdot\text{s}$ ) was designated as continuous phase. Visualisation experiments were carried out and these laboratory data were used to validate the simulation results. Detailed 2D simulations were presented to examine systematically the impact of fluid properties on the droplet breakup rate at predefined flow rate ratio,  $Q$  of 0.05. The results yielded an inflection point in the dependence of droplet breakup rate on Na-CMC concentration was found in between the dilute and semi-dilute concentration regimes. This inflection point displays a non-monotonic profile which is mainly caused by the considerable viscosity effect of Na-CMC polymer when its concentration increases above a critical

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\* Corresponding author; [v.wong@hw.ac.uk](mailto:v.wong@hw.ac.uk) (Voon-Loong Wong)

<sup>1</sup>Current address: School of Engineering and Physical Sciences, Heriot-Watt University Malaysia Campus, 62200 Putrajaya Wilayah Persekutuan, Malaysia.

<sup>2</sup>Current address: Department of Chemical Engineering, University College London, London, United Kingdom.

30 value ( $C > C^* \sim 0.40$  wt%). This striking behaviour highlights the importance of  
31 rheological effects in flows with a shear-dependent fluid under various flow  
32 conditions. The viscous effect of Na-CMC fluids substantially affects the manipulation  
33 over the droplet pinch-off time and production rate. Thus, it necessitate the control of  
34 the shear rate by adjusting the flow conditions and aspect ratio of microchannels.

35 **Keywords:** non-Newtonian; microfluidics; level-set; droplet breakup; T-junction  
36 geometry.

37

## 38 **1. Introduction**

39 Two-phase flow is a term covering the motion of two different interacting fluids that  
40 are in different phases such as liquid-liquid and liquid-vapor. In two-phase flow  
41 microfluidics, an emulsion contains a mixture of two immiscible liquids as one phase  
42 being dispersed throughout the other phase in small droplets. Most common  
43 emulsions include direct emulsions, oil droplets in an immiscible and continuous  
44 water phase, or inverted emulsions, water droplets in an immiscible and continuous  
45 oil phase. Emulsions are typically made by fissioning droplets with shear or impact  
46 and the resulting suspensions possess a wide size distribution of drop sizes  
47 (Umbanhowar, Prasad, & Weitz, 2000). At low enough Reynolds ( $Re$ ) number, a  
48 laminar flow regime is assumed and these droplets are translocate through  
49 microfluidic structures having dimensions most easily measured in microns. The  
50 manipulation of droplets in a confined microfluidic system has been highlighted as  
51 one of the earliest tools used in the fields of biomedical sciences. The characteristics  
52 of droplets become reliable tool for performing biological operations such as analyte  
53 encapsulation, sampling, metering, dilution, reaction and detection (Huebner et al.,  
54 2008; Niu & deMello, 2012; Tawfik & Griffiths, 1998; Theberge Ashleigh et al., 2010).

55 Droplets can be generated via a number of methods in microfluidic devices, including  
56 breakup in co-flowing stream(Cramer, Fischer, & Windhab, 2004; Moon, Cheong, &  
57 Choi, 2014; Utada, Fernandez-Nieves, Stone, & Weitz, 2007), breakup in cross-  
58 flowing stream (Garstecki, Fuerstman, Stone, & Whitesides, 2006; Qiu, Silva,  
59 Tonkovich, & Arora, 2010; Xu, Li, Tan, & Luo, 2008), hydrodynamics flow-focusing  
60 (Anna & Mayer, 2006; Peng, Yang, Guo, Liu, & Zhao, 2011), and microchannel

61 emulsification (Kobayashi, Nakajima, & Mukataka, 2003; van der Zwan, Schroën, &  
62 Boom, 2009; Yobas, Martens, Ong, & Ranganathan, 2006). Cross-flowing in a T-  
63 junction is one of the easiest microfluidic methods of generating highly  
64 monodispersed droplets. The formation of droplet at a T-junction, at which viscous  
65 shear-stresses induced by continuous stream of the horizontal channel overcome  
66 surface tension at the liquid-liquid interface and pull off droplets of the dispersed  
67 phase from the vertical channel. This is mainly due to the instabilities of free surface  
68 between the phases are sufficiently large. Thus, the size and frequency of the  
69 droplets can be accurately manipulated by modifying the relative pressures of the  
70 two immiscible liquid in order to enable the production of a wide range of vesicle  
71 shapes and patterns (Thorsen, Roberts, Arnold, & Quake, 2001). Besides, opposed  
72 flowing (Shui, van den Berg, & Eijkel, 2009) and perpendicular flowing (Leshansky &  
73 Pismen, 2009) are another operation modes producing monodispersed droplet  
74 formation in a T-junction microchannel.

75 A numerical modelling approach to the multiphase flow problem provides a detailed  
76 and comprehensive description of the formation of microdroplets since a number of  
77 statistical information can be extracted from a predictive model. As the dimension of  
78 the interest gets smaller, the surface-based interfacial tension and the viscosity  
79 become more significant in controlling critical flow behavior of multiphase flow in  
80 microscale, especially when handling fluids that have a complex microstructure  
81 leading to non-Newtonian phenomena. Additionally, the non-Newtonian flow curve  
82 presents a nonlinear relationship between shear-stress and the rate of deformation.  
83 For instance, previous research efforts have been much devoted to the experimental  
84 analysis of the dynamics and relevant hydrodynamics of viscoelastic droplets  
85 (Arratia, Cramer, Gollub, & Durian, 2009; Husny & Cooper-White, 2006; Steinhaus,  
86 Shen, & Sureshkumar, 2007) and few studies have focused on other shear-  
87 dependent fluids such as purely viscous fluids and time-dependent fluids (Chhabra &  
88 Richardson, 2008). Hitherto, there have been far fewer attempts to develop a  
89 predictive numerical model for the relevant physics of non-Newtonian droplets  
90 generation in a Newtonian bulk phases. However, there is no unique constitutive  
91 model that can represent the different characteristic behaviours of non-Newtonian  
92 fluids.

93 In microfluidics, the geometry of droplet interface is usually complex and it can  
94 undergo large deformations or even topology changes such as fission and fusion in  
95 microchannel. There are two approaches, namely interface tracking (Hou,  
96 Lowengrub, & Shelley, 2001; Tryggvason et al., 2001) and interface capturing  
97 (Bonometti & Magnaudet, 2007), to represent the flow problem of droplet interface  
98 evolution or moving boundaries either explicitly or implicitly to the incompressible  
99 Navier-Stokes equation discretised on a fixed grid. Interface tracking of the moving  
100 boundary in multiphase system is an explicit representative that requires the  
101 computational meshes to track the evolving interface for each time-step. In contrast,  
102 the interface capturing approaches is an implicit representative that uses a phase  
103 function discretised on the fixed grid to represent the interface (Bonometti &  
104 Magnaudet, 2007). In present paper, conservative level-set method (LSM) is adopted  
105 as it is a simple and robust scheme of interface capturing approaches for tracking  
106 moving interfaces and shapes (Osher & Sethian, 1988). It permits numerical  
107 computations of such objects involving curves and surfaces to be performed on a  
108 fixed Cartesian grid without having to parameterize them (Olsson & Kreiss, 2005;  
109 Olsson, Kreiss, & Zahedi, 2007; Osher & Sethian, 1988). In the LSM, the surface  
110 tension force is conventionally modelled as a distributed body force though  
111 concentrated in a band around the interfaces. The variation of surface tension force  
112 across the interface can be difficulties in the application of others common interface  
113 capturing methods, including the volume-of-fluid (Rider & Kothe, 1998) (VOF) and  
114 lattice-Boltzmann method (LBM) (Takada, Misawa, Tomiyama, & Fujiwara, 2000).  
115 However, the LSM can resolve to the challenges of mass conservation and the  
116 treatment of discontinuities across the flexible interface (Olsson & Kreiss, 2005;  
117 Olsson et al., 2007).

118 The present paper demonstrates systematic sets of numerical simulations for the  
119 microdroplet generation of a shear-thinning Na-CMC droplets in Newtonian flow at a  
120 microfluidic T junction using a developed predictive computational model. The  
121 present model is adopted with the integration of conservative level-set approach and  
122 non-Newtonian constitutive law. Fundamental principles and application of  
123 microfluidic systems were presented due to the selection and interpretation of the  
124 subsequent numerical analysis. Numerical simulations of the Na-CMC microdroplets  
125 formation in an olive oil-based continuous phase were carried out. The present study

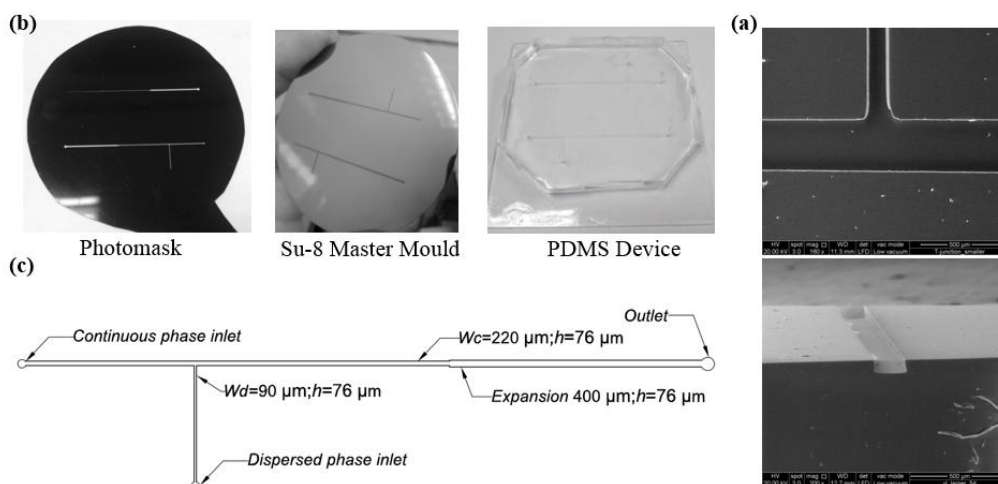
126 reveals the interesting phenomena of shear-thinnings droplet formation during the  
 127 systematic variation in flow rates, interfacial tension, and surface wettability. As a  
 128 result, the rheological characteristics of Na-CMC solution are strongly depending on  
 129 its concentration. Thus, the impact of these rheological characteristics of Na-CMC  
 130 can be of great interest to provide an insightful understanding to the relevant physics  
 131 of non-Newtonian droplet formation process in microfluidic flow.

132

## 133 2. Experimental Methodology

### 134 2.1 Microfluidics Device Fabrication

135 A channel dimension of  $220\ \mu\text{m}$  ( $w_c$ )  $\times$   $90\ \mu\text{m}$  ( $w_d$ )  $\times$   $76\ \mu\text{m}$  ( $h$ ) was used in present  
 136 validation studies. It was fabricated in-house by photolithography and soft  
 137 lithography technique. Fig. 1 illustrates the dimensions of a T-junction employed in  
 138 the numerical and experimental studies. Prior to the start of fabrication process, an  
 139 out sourced positive photolithographic mask (clear lines with a black background)  
 140 was used to transfer of the photo-lithographically pattern onto the negative mould.  
 141 The photolithographic mask with the desired layout of microchannel structure was  
 142 designed using standard computer assisted design (CAD) program. The  
 143 polydimethylsiloxane (PDMS) mould was fabricated by moulding a mixture of PDMS  
 144 liquid pre-polymer, a 10:1 mixture of Sylgard 184 silicone elastomer and curing agent  
 145 (Dow Corning, USA), onto the SU-8 master mould with SU-8 2025 (MicroChem  
 146 Corporation, Newton, MA) as the photoresist that ultimately becomes the pattern on  
 147 the silicon wafer. The glass slide with a cured PDMS thin layer and the surface of the  
 148 PDMS mould with the micropattern indent then brought into conformal contact before  
 149 flow experiment was conducted.



150

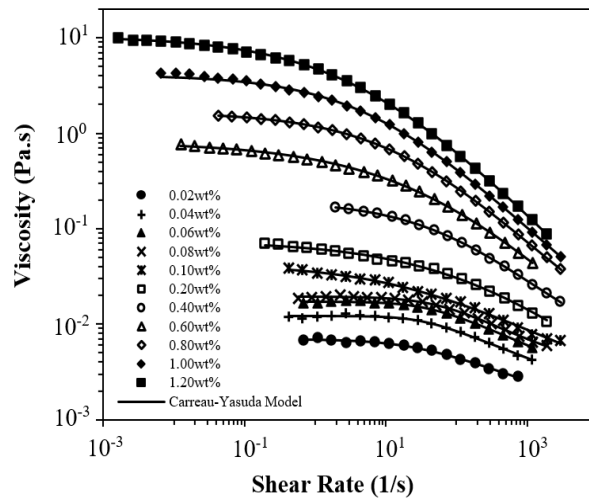
151 **Fig. 1:** Illustration of (a) microfluidics T-junction composed of rectangular channels;  
152 (b) schematic diagram of microfluidics T-junction composed of rectangular channel;  
153 (c) Scanning electron microscope (SEM) image of fabricated T-junction of PDMS  
154 microchannel and cross section of PDMS microchannel.  
155

## 156 **2.2 Fluid Characterisation**

157 A calibrated BS/U-tube viscometer was used to perform the kinematic viscosity  
158 measurement for the transparent Newtonian olive oil (Sigma Aldrich). The capillary  
159 diameter of BS/U-tube viscometer with  $0.50 \text{ mm} \pm 0.01 \text{ mm}$  was used. The  
160 viscometers were mounted upright in a beaker (2000 mL) of water at controlled room  
161 temperature ( $20 \pm 2^\circ\text{C}$ ). Each sample solutions were allowed to attain the room  
162 temperature for 10 minutes. The viscosity measurements for Newtonian solution  
163 were conducted three times, and average values were taken for analysis. For shear-  
164 thinnings Na-CMC ( $[\text{C}_6\text{H}_7\text{O}_2(\text{OH})\text{CH}_2\text{COONa}]_n$ , Sigma Aldrich) aqueous solution, the  
165 rheological measurements were performed on controlled stress rheometer (MCR  
166 302, Anton Paar) equipped with a cone-and-plate geometry (cone plate with  
167 diameter of 50 mm; angle 0.04 radian) at controlled temperature of  $20^\circ\text{C}$ . The  
168 samples were carefully loaded onto the measuring plate of the rheometer and left to  
169 idle for 10 minutes prior to viscosity measurement. Fig. 2 illustrates the shear  
170 viscosity against shear rate plot of Na-CMC solutions at various concentrations  
171 ranging from 0.02 wt% to 1.20 wt% has been plotted over a log-log scale that covers  
172 nearly six orders of magnitude of shear rate (Wong, Loizou, Lau, Graham, &  
173 Hewakandamby, 2017). In present work, power-law model is not potentially selected  
174 to describe the behaviour of shear-thinnings working fluid of Na-CMC aqueous  
175 solution as it poses limitations on its range of applicability over a wide range of shear  
176 rate. In order to circumvent the drawback of the power-law model, alternative  
177 approaches such as Carreau-Yasuda model utilize viscosity functions that have finite  
178 values both at very low and high shear rate (Chhabra & Richardson, 2008).  
179 Additionally, all the measurement data were well-fitted to the well-known Carreau-  
180 Yasuda model (Chhabra & Richardson, 2008) for the shear-thinnings behaviour of  
181 Na-CMC:

$$182 \quad \eta(\dot{\gamma}) = \eta_\infty + (\eta_o - \eta_\infty) [1 + (\lambda_{CY} \dot{\gamma})^a]^{\frac{n-1}{a}} \quad (1)$$

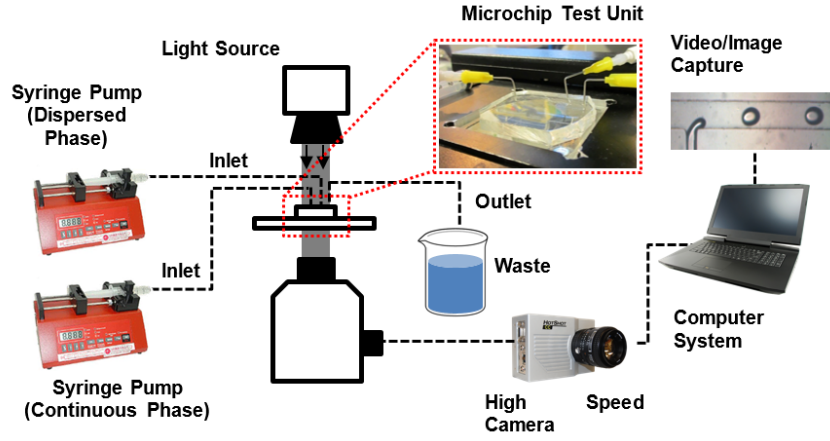
183 where  $\eta_o$  is zero shear viscosity,  $\eta_\infty$  is infinite shear viscosity,  $\lambda_{CY}$  is the relaxation  
 184 time,  $\dot{\gamma}$  is shear rate,  $n$  is power-law exponent and  $a$  is the fitting parameter for  
 185 Carreau-Yasuda model.



186  
 187 **Fig. 2:** Shear viscosity plotted against shear rate for a series of Na-CMC shear-  
 188 thinning solutions with different concentrations (Wong et al., 2017).  
 189

### 190 2.3 Emulsification Setup

191 The continuous (olive oil) and dispersed phase liquids (Na-CMC aqueous) were  
 192 dispensed separately into the reservoirs of microchannel; each fluid was driven  
 193 through microchannel at the desired continuous ( $Q_c$ ) and dispersed flow rates ( $Q_d$ )  
 194 using syringe pumps (AL-1000, Florida and NE-1000, Netherlands), respectively.  
 195 The processes before and after droplet formation in microfluidics device were  
 196 recorded using a high speed camera (MIC Hotshot 1280 cc) connected to an  
 197 epifluorescence microscope (Olympus IX51, Japan). The experimental setup with  
 198 flow visualisation is illustrated in Fig. 3. After stabilizing the system for predetermined  
 199 time intervals (20 minutes), videos were recorded at 500 frames per second (fps)  
 200 after each flow rates of either continuous or dispersed phase were adjusted. The  
 201 average effective droplet diameter of 30 droplets under experimental condition was  
 202 measured through an image processing routine using MATLAB. All the collected data  
 203 will be validated prior to the parametric analysis.



**Fig. 3:** Schematic diagram of experimental setup for flow visualization.

### 3. CFD Modelling and Simulation

#### 3.1 Theory Model

A predictive numerical model was developed to track the fluid-fluid interfaces between two immiscible fluid phases of different density and viscosity. This applied the conservative LSM from CFD module using COMSOL Multiphysics. The mathematical model used in the computational fluidic dynamic simulation utilized a numerical time-stepping procedure to obtain the model behaviour over time. The governing equations for momentum and conservation laws of mass was considered, which was shown in the following forms with the assumption that the fluid is incompressible:

$$\rho \frac{\partial \mathbf{u}}{\partial t} + \rho (\mathbf{u} \cdot \nabla) \mathbf{u} = -\nabla p + \nabla \cdot \eta (\nabla \mathbf{u} + (\nabla \mathbf{u})^T) + F_{st} \quad (2)$$

$$\nabla \cdot \mathbf{u} = 0 \quad (3)$$

where  $\rho$ ,  $\eta$ , and  $F_{st}$  denote the density, dynamic viscosity, and the surface tension force respectively,  $p$  denotes pressure while  $I$  is the identity matrix. Naturally, the Navier-Stokes equation (Equation (2)) is solved on the fixed grid to control the motion of multiphase system. The density and the viscosity of the two fluids at any point can be calculated using the two equations given below:

$$\rho = \rho_1 + (\rho_2 - \rho_1) \tilde{\phi} \quad (4)$$

$$\eta = \eta_1 + (\eta_2 - \eta_1) \tilde{\phi} \quad (5)$$

where  $\rho_1$  and  $\rho_2$  are the densities of continuous phase and dispersed phase, and  $\eta_1$  and  $\eta_2$  are the viscosities of continuous phase and dispersed phase. A smeared out



229 approach is used where the discretisation of the Heaviside function ( $H_{sm}(\phi)$ ) can be  
 230 useful as it is better suited to numerical computations (Deshpande & Zimmerman,  
 231 2006):

$$232 \quad \tilde{\phi} = H_{sm}(\phi) = \begin{cases} 0, & \text{if } \phi < -\varepsilon \\ \frac{1}{2} + \frac{\phi}{2\varepsilon} + \frac{1}{2\pi} \sin\left(\frac{\pi\phi}{\varepsilon}\right), & \text{if } -\varepsilon \leq \phi \leq \varepsilon \\ 1, & \text{if } \phi > \varepsilon \end{cases} \quad (6)$$

233  
 234 where  $\varepsilon$  denotes the interface thickness. The  $F_{st}$  term acting on the interface  
 235 between two fluid phases is determined by following equation:

$$236 \quad F_{st} = \sigma k \mathbf{n}_\Gamma \delta_{sm} \quad (7)$$

237 where  $\sigma$  denotes surface tension,  $k$  denotes local interfacial curvature,  $\mathbf{n}_\Gamma$  is the unit  
 238 normal vector to the interface pointing into the droplet, and the  $\delta_{sm}$  denotes the  
 239 smeared out Dirac delta function ( $\delta_{sm}$ ) concentrated at the interface between two  
 240 fluids. These above parameters can be calculated by Equation 8, 9, and 10,  
 241 respectively:

$$242 \quad k = -\nabla \cdot \mathbf{n}_\Gamma \quad (8)$$

$$243 \quad \mathbf{n}_\Gamma = \frac{\nabla \phi}{|\nabla \phi|} \quad (9)$$

$$244 \quad \delta_{sm}(\phi) = \begin{cases} 0, & \text{if } \phi < -\varepsilon \\ \frac{1}{2\varepsilon} + \frac{1}{2\varepsilon} \cos\left(\frac{\pi\phi}{\varepsilon}\right), & \text{if } -\varepsilon \leq \phi \leq \varepsilon \\ 0, & \text{if } \phi > \varepsilon \end{cases} \quad (10)$$

245 To retain the level-set function ( $\phi$ ), a re-initialization procedure is required for the  
 246 finite element approximation of the level-set equation. A re-initialized and  
 247 conservative level-set method is used to describe and convect the fluid interface.

248 The following equation describes the convection of re-initialized level-set function:

$$249 \quad \frac{\partial \phi}{\partial t} + \mathbf{u} \cdot \nabla \phi = \gamma \nabla \cdot \left[ \varepsilon \nabla \phi - \phi(1-\phi) \frac{\nabla \phi}{|\nabla \phi|} \right] \quad (11)$$

250 where  $\gamma$  and  $\varepsilon$  are numerical stabilization parameters, where the former denotes re-  
 251 initialization parameter and latter parameter determines the thickness of the  
 252 interface. Equation (11) is coupled to the governing equations (Equation (2) and (3))  
 253 in present numerical model. The  $\gamma$  approximates the maximum speed occurring in  
 254 the computational domain. The  $\varepsilon$  assumed as the maximum mesh size in  
 255 subdomains in the neighbourhood of the interface. After the grid convergence

256 analysis, the parameters  $\gamma$  and  $\varepsilon$  with the value of 0.065 m/s and  $5.8 \times 10^{-6}$  m were  
 257 calculated based on the maximum flow velocity in microchannel and optimum mesh  
 258 size, respectively.

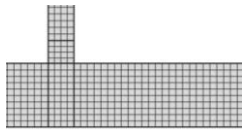
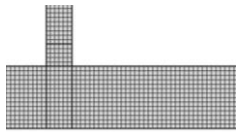
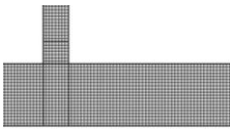
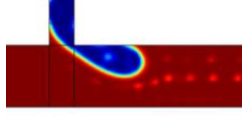
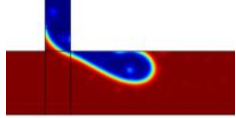
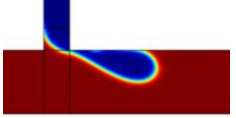
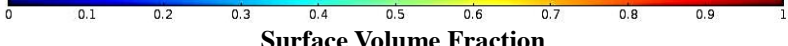
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### 260 3.2 Domain Discretisation and Grid Convergence Analysis

261 A T-shaped geometry with prescribed dimension of  $220 \mu\text{m} \times 90 \mu\text{m}$  was created  
 262 and meshed with quadrilaterals elements. An entrance thickness ( $h$ ) of  $73.5 \mu\text{m}$  was  
 263 prescribed in numerical system define the depth of the microchannel. The typical  
 264 finite element mesh for structured mesh of two-dimension (2D) mapped mesh for a  
 265 2D model was selected. Mesh refinement analysis were performed to quantify the  
 266 dependency of simulation results on the grid size and achieve an optimal grid  
 267 resolution. Meshes of varying degrees of resolution were set up for the T-junction  
 268 domain with the same grid size of near-wall region. Table 1 illustrates three  
 269 examples of mesh geometry with prescribed dimensions in COMSOL Multiphysics.  
 270 Mesh can be arranged as to be clustered near the wall for optimum grid resolution in  
 271 order to resolve the boundary layer flow in future work.

272

273 **Table 1:** Comparison of a T-shaped geometry with coarser and finer mesh size.

No. of Mesh Elements	976 (Coarse)	2072	4024 (Finer)
2D Structured Mapped Mesh			
Two-phase Flow at $t=0.05\text{s}$ and $Q=0.05$ ( $Q_c:2.0\text{ml/hr}$ ; $Q_d:0.1\text{ml/hr}$ )			
			

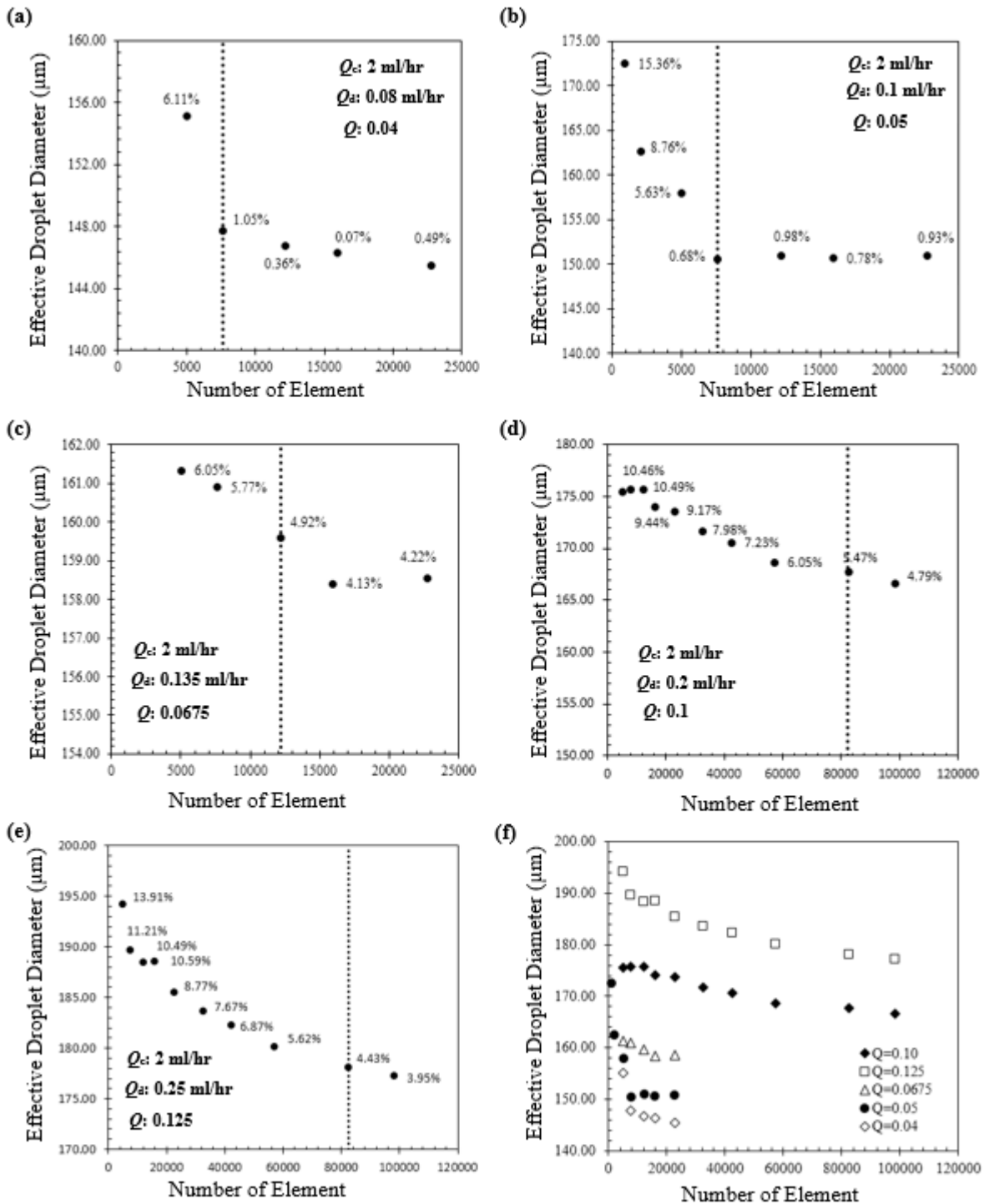
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275 The present model is set up for transient analysis which provides the time domain  
 276 response of a system subjected to time-dependent loads. The effect of mesh size  
 277 was examined by increasing the number of mesh elements from initial number of  
 278 elements of 976 (coarsest) to 98184 (finest). Finer grid size are generated by  
 279 simultaneously increasing the number of nodes in all direction to obtain as close to a  
 280 uniform refinement. The number of elements is increased by a mean factor of  $\sqrt{2}$  for  
 281 each refinement settings. Thus, the total number of nodes for each refinement is

282 approximately doubled over the previous grid size. Preliminary test of grid  
 283 convergence analysis was solely carried out at flow rate ratio,  $Q$  of 0.05, which is a  
 284 quotient of a flow rate of the flow for dispersed phase and continuous phase. An  
 285 acceptable relative error and error percentage of 0.15% and  $\leq 1\%$  between the last  
 286 two finer grids (12166 and 15963) was obtained, respectively. Fig. 4(b) illustrates the  
 287 grid convergence analysis at  $Q$  of 0.05. An optimal simulations results was achieved  
 288 at 7644 number of elements. The relative error of measurements shows the error  
 289 deviation in relation to the effective droplet diameter between each mesh resolutions.  
 290 While the error percentage of measurements shows the error deviation in relation to  
 291 the effective droplet diameter between numerical and experimental data. In the  
 292 present work, all the parametric studies used, as key output, the effective droplet  
 293 diameter. Therefore, an integration operator was added to find the area  
 294 corresponding to the dispersed phase, where  $\tilde{\phi} \geq 0.5$ , in order to calculate the  
 295 effective droplet diameter by the following equation:

$$296 \quad d_{\text{eff}} = 2 \cdot \sqrt{\frac{1}{\pi} \int_{\Omega} (\tilde{\phi} > 0.5) d\Omega} \quad (12)$$

297 This is the diameter of a spherical droplet that has equivalent volume of the formed  
 298 droplet. The extensive studies of grid convergence analysis was also performed on  
 299 different flow rate ratio. Fig. 4 illustrates the mesh dependence profile with error  
 300 percentage of droplet size measurement at different flow rate ratio  $Q$ . For a constant  
 301  $Q_c$  at 2.00 ml/hr, the effective droplet diameter was measured with the variation in  $Q_d$   
 302 (0.08 ml/hr to 0.125 ml/hr) for various number of elements. As seen in Fig. 4, the  
 303 results of convergence are found to achieve more rapidly and effectively in cases at  
 304 lower flow rate ratio ( $Q \leq 0.0675$ ). Higher flow rate ratios are limited to a certain  
 305 range in numerical model due to difficulties of numerical dissipation in the advection  
 306 step of fluid simulation. Moreover, further refinement is required to sufficiently resolve  
 307 the higher velocity profiles. Thus, the subsequent parametric studies were mainly  
 308 based on  $Q$  of 0.05 at optimal mesh size. Particularly, the numerical simulations  
 309 were performed at a time-step size of  $2.57 \times 10^{-5}$  seconds calculated using the  
 310 Courant-Friedrichs-Lewy (CFL) conditions. It shows a relation between the  
 311 computational cell size, the transient time-step size, and the fluid velocity within the  
 312 cell. A Courant number of 0.25 is selected in the present study considering as robust  
 313 value to maintain the stability of calculations.



314

315 **Fig. 4:** Grid convergence analysis for different flow-rate ratio profile: (a)  $Q=0.04$ ; (b)  
 316  $Q=0.05$ ; (c)  $Q=0.0675$ ; (d)  $Q=0.1$ ; (e)  $Q=0.125$ ; (f) combination of mesh refinement  
 317 profile.

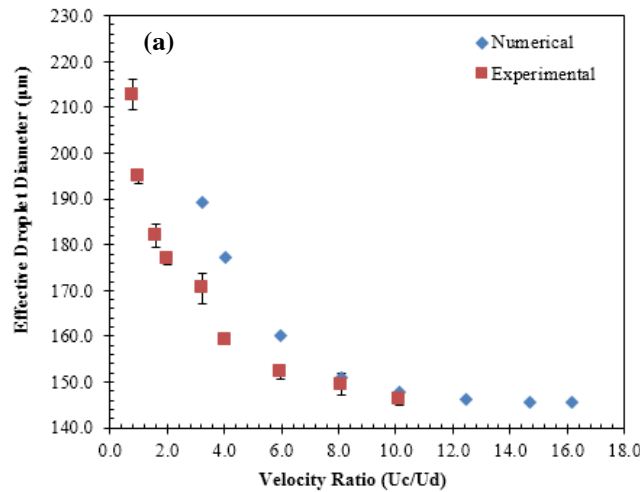
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### 320 3.3 Numerical Model Validation with Experimental Justification

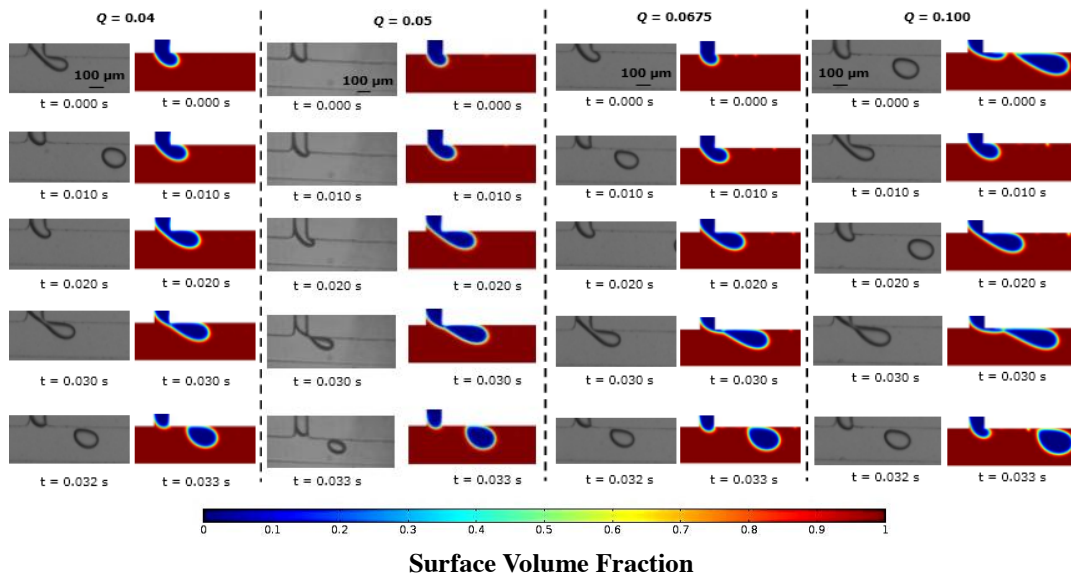
321 Mesh convergence analysis was studied to quantify the dependency of simulation  
 322 results on mesh size and achieve an optimal grid resolution of 7644. A preliminary

323 validation of numerical simulations of the formation of deionized water droplets in  
 324 olive in a T-junction microchannel was carried out and gave qualitative agreement  
 325 with laboratory experimental data to predict with reasonable accuracy in the range of  
 326 velocity applied. The results of convergence is found to be achieved more rapidly at  
 327 lower flow rate ratio. Fig. 5 illustrates the variation in effective droplet diameter  
 328 between the experimental and numerical studies.



329

(b)



330

331 **Fig. 5:** Comparison of (a) effective droplet diameter between numerical and  
 332 experimental result in the range of velocity ratio applied. Error bars shown indicates  
 333 the standard deviation in droplet size measurement of 30 droplets under fixed  
 334 experimental condition. Dashed box shown denotes the (b) droplet breakup  
 335 phenomena in the range between  $Q$  of 0.04 and 0.1.

336

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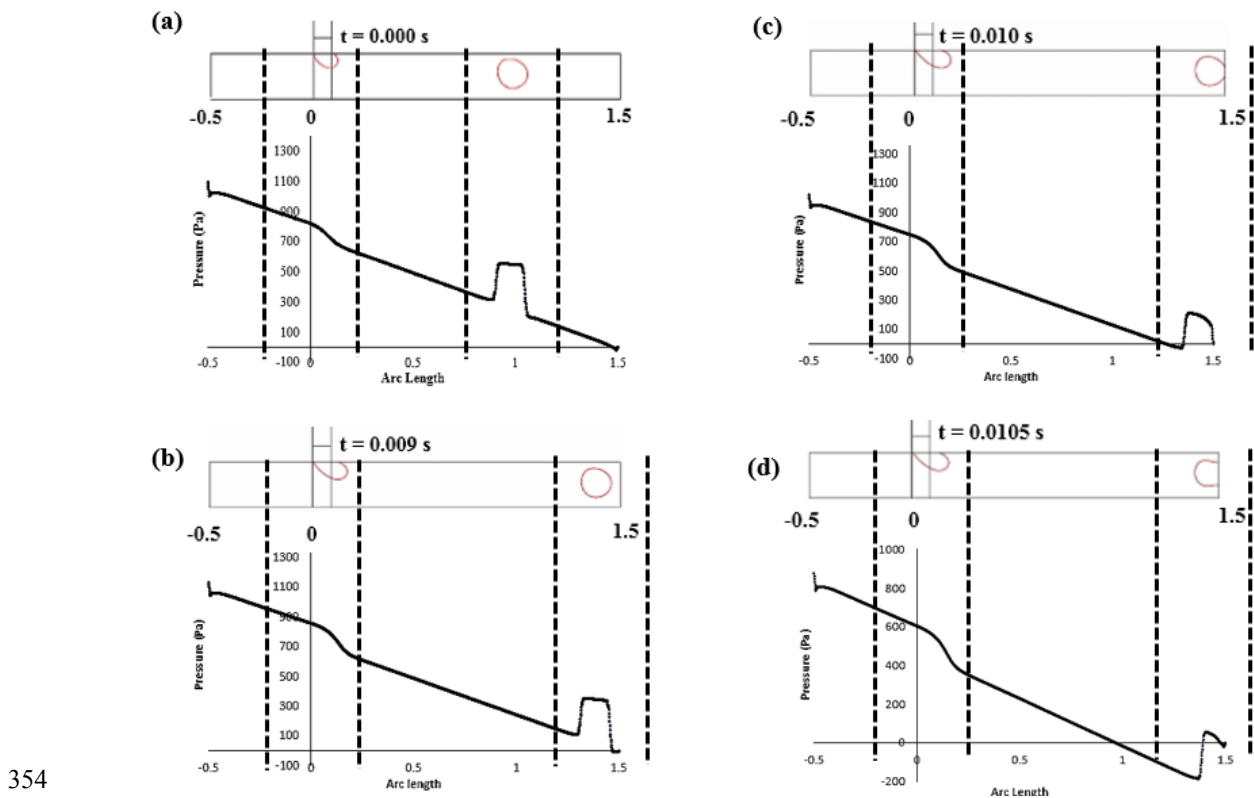
338 The data was recorded at  $5 \times 10^{-3}$  seconds intervals to capture the growth and  
 339 detachment phenomenon of the fluid-fluid interfaces. A contact angle of  $180^\circ$  that

340 represents the complete repulsion of Na-CMC droplets by the PDMS with  
 341 hydrophobic channel wall surface was applied in the computation. As can be seen in  
 342 Fig. 5, the numerical simulation of detachment process was shown in similar manner  
 343 with the experimental data at  $Q < 0.0675$ . Nevertheless, such an agreement was  
 344 physically unreasonable at higher  $Q$  as jet breakup phenomena is found to be more  
 345 significant for numerical simulation over time. This might be due to the numerical  
 346 dissipation increases the viscosity of working fluid and causes it appear more  
 347 viscous than intended at higher  $Q$ . Thus, the  $Q$  of 0.05, as equivalent to the velocity  
 348 ratio ( $u_c/u_d$ ) of 8, was adopted in present parametric studies subsequently.

349

### 350 3.4 Simulated Laplace Pressure of Droplet Interface Profile

351 First, the static pressure is decreasing linearly along the cross-sectional 2D plane of  
 352 the microchannel. Fig. 6 illustrates the pressure drop distribution along the channel a  
 353 droplet containing with 0.02 wt% Na-CMC solution.



354

355 **Fig. 6:** The pressure distribution of a generated 0.20wt% Na-CMC droplet interface  
 356 along the T-junction microchannel at  $Q$  of 0.05.

357

358 In the presence of curved interfaces, the curvature induce a pressure jump, which is  
 359 known as Laplace pressure. The Laplace pressure jumps was determined in the

360 path across the droplet body while crossing the front and rear interfaces. As the  
361 dimension of the interest gets smaller, surfaces tension becomes dominant over  
362 gravitational forces and others physical forces such as viscous and inertial forces.  
363 The surface tension becomes an important surface energy parameter that controls  
364 the stability of interfaces between the two phases when the droplets are forming.  
365 Due to the existence of surface tension effect in a case of a liquid droplet, the  
366 Laplace law implies a greater pressure inside the droplet than a continuous phase.  
367 As the radius of the droplet become smaller, the pressure becomes larger on the  
368 concave side of liquid interface. A Young-Laplace equation is usually used to  
369 determine the pressure difference across a fluid interface as a function of curvature.  
370 Moreover, the magnitude of this pressure differential can be expressed in term of  
371 surface tension:

$$372 \Delta P_L = \sigma \left( \frac{1}{R_1} + \frac{1}{R_2} \right) \quad (13)$$

373 where  $\sigma$  is the surface tension of the respective liquid interface and  $R_1$  and  $R_2$  are the  
374 two principal curvature radii of the interface. Additionally, the effect of viscous  
375 dissipation inside the droplets may also prominently affects the pressure drop  
376 distribution when the viscous effect is considerably increased.

377

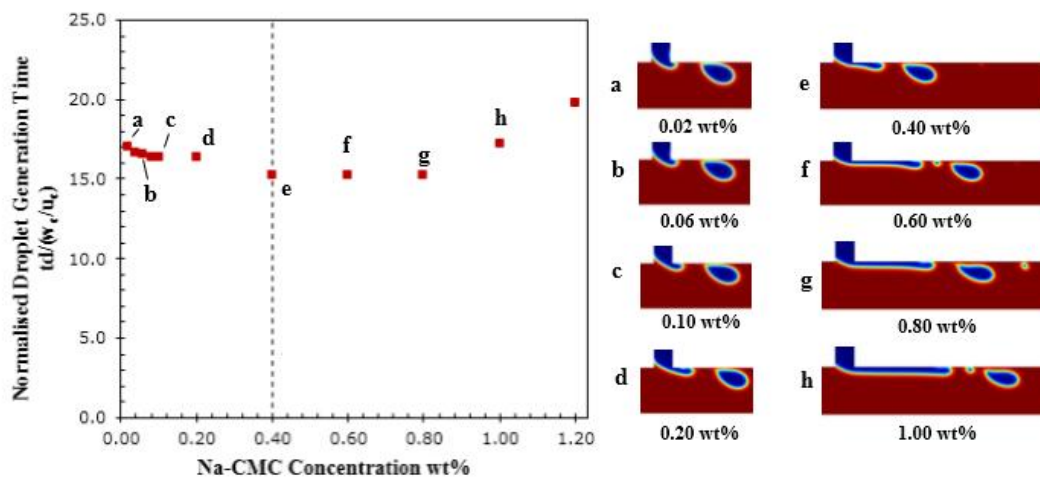
## 378 **4. Results and Discussion**

### 379 **4.1 Effect of Na-CMC Viscosity on Droplet Breakup Time and Production Rate**

380 The present investigation was designed to determine the Na-CMC viscosity effect on  
381 the droplet breakup time and production rate. The viscosity of polymer solution is a  
382 function of concentration and the molecular weight of dissolved polymer. While the  
383 concentration of Na-CMC solution is increased in a series of simulations from 0.02  
384 wt% to 1.20 wt% at a constant  $Q$  of 0.05, the evolution of droplet breakup time can  
385 be discerned in two distinct non-monotonic behaviour. Fig. 7(a) illustrates the Na-  
386 CMC concentration effect on the normalized droplet breakup time whereas the  
387 normalized production rate is shown in Fig. 7(b). The variation of viscosity of a shear  
388 thinning drop occurs during the pinch off process. Initially, the breakup time  
389 decreases as Na-CMC concentration increases from 0.02 wt% to 0.40 wt%. The  
390 shear-thinning effect of Na-CMC solution increases with the concentration. The  
391 greater shear-thinning effect may exhibit a decrease in polymer viscosity upon the

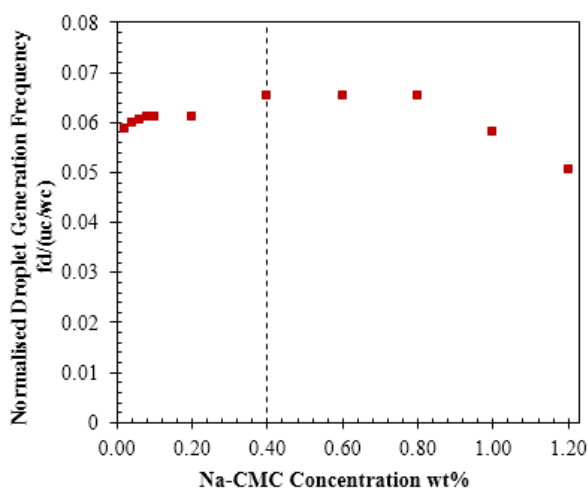
392 application of shear near the channel wall due to the inertial force. At concentrations  
 393 well above 0.40 wt%, the Na-CMC droplet breakup time increases may to the  
 394 viscous force becomes significantly prevailing over the inertial force induced by the  
 395 continuous phase on the forming interface. In general, the largest shear rate occurs  
 396 at the corner edge of the T-junction and thus the shear-induced destabilization of the  
 397 dispersed thread causes the breakup of thread leading to the formation of droplets.  
 398 As can be seen in Fig. 8, the distribution of non-Newtonian wall shear-rate  
 399 decreases as compared to the Newtonian wall shear rate profile. While in the lower  
 400 wall shear rate range, the Na-CMC solutions exhibited a significant increase in  
 401 viscosity. This means the low shear rate viscosity occurs at the high concentration of  
 402 Na-CMC content. Consequently, the droplets pinch-off in larger viscosity fluids for  
 403 which the inertial effects are unamplified.

404 (a)



405

406 (b)

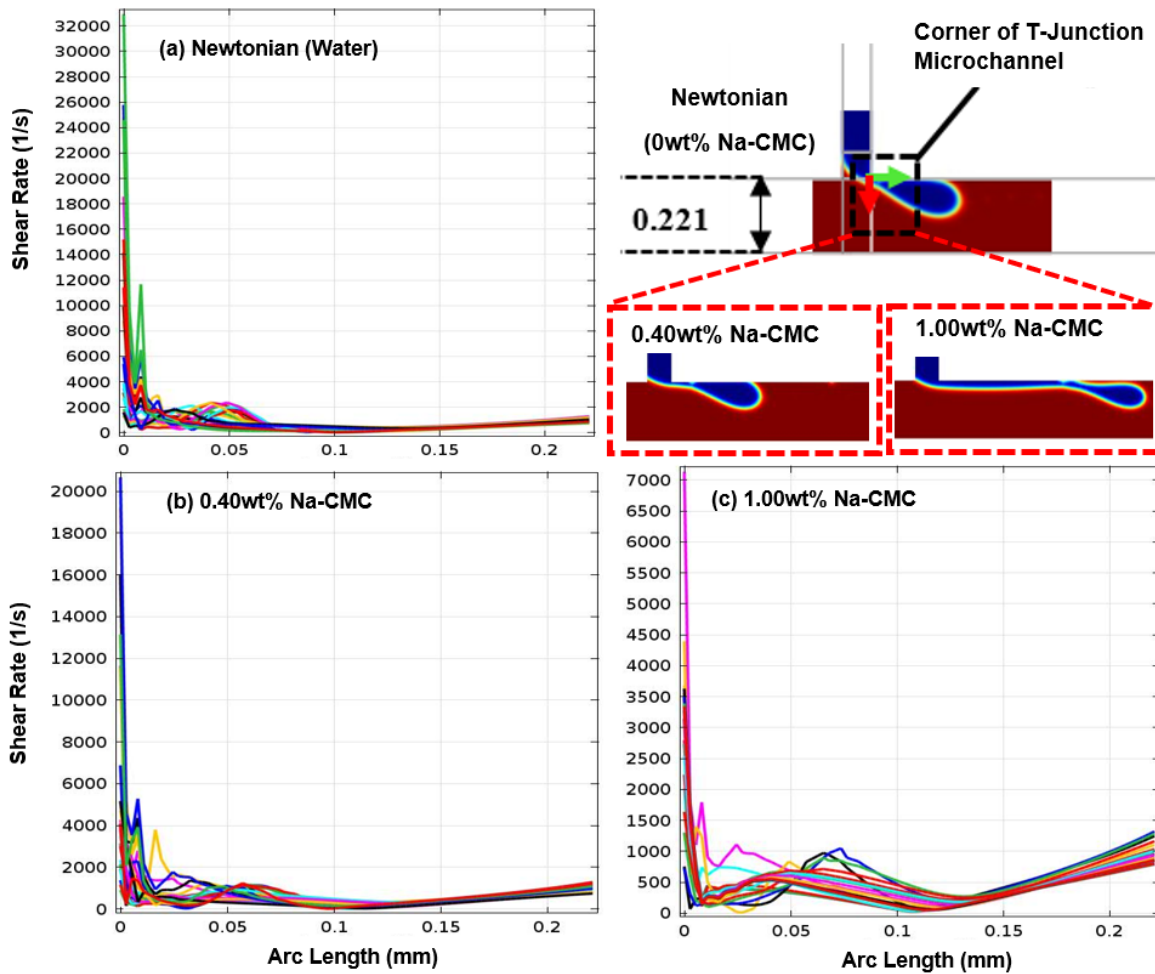


407

408 **Fig. 7:** Effect of Na-CMC concentration on (a) droplet breakup time (b) droplet  
 409 production rate (for system:  $Q_d/Q_c=0.05$ ).



410  
411 At dilute Na-CMC concentrations below 0.40 wt%, the viscosity effect is not  
412 considerable as the direct intermolecular interactions are negligible. As the Na-CMC  
413 concentration increases, the shear-thinning effect becomes more significant and  
414 causes rapid pinch-off due to the high shear stress exerted by the continuous phase  
415 near to the wall. The viscosity of the shear thinning drop is reduced when the drop  
416 begins to neck response to the increased shear rate in that region. As the neck  
417 continues to thin, the region of lower viscosity grows to encompass almost the entire  
418 drop occurring within the neck and just outside it where the outflows from the neck  
419 occur (Malcolm R. Davidson & Cooper-White, 2006; M. R. Davidson, Cooper-White,  
420 & Tirtaatmadja, 2004). Nevertheless, extending breakup occurs and lower production  
421 rate was found for the Na-CMC solution concentration above 0.40 wt%. This is  
422 mainly caused by the development of entanglement coupling between the polymer  
423 chains, which begins manipulating the fluid characteristics of Na-CMC solution. The  
424 elongation of the necking into a thin filament connect between the primary droplet  
425 and the upper dispersed phase thread, suggest that an increase in polymer  
426 concentration which plays an important role in resisting drop pinch-off with the  
427 formation of thinning filament. Thus, the viscous fluid thread is then stretched by the  
428 mainstream flow to the downstream region and the breakup event is delayed  
429 substantially.



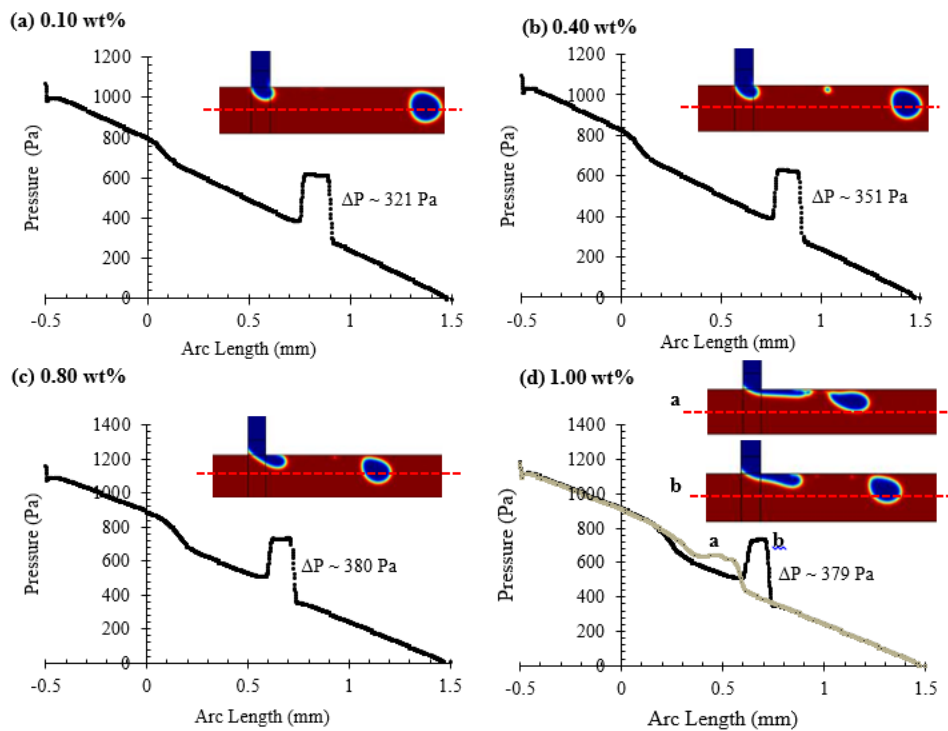
430

431 **Fig. 8:** Shear rate profile of fluid flow at the edge of T-junction microchannel along  
 432 the arc length of continuous phase for (a) Newtonian system; (b) 0.40wt% Na-CMC  
 433 non-Newtonian system; and (c) 1.0 wt% Na-CMC non-Newtonian system (for  
 434 system:  $Q_d/Q_c=0.05$ ).

435

436 Fig. 9 illustrates the Laplace pressure profile of a Na-CMC droplet interface  
 437 curvature for different polymer concentrations at the middle plane of the  
 438 microchannel. At dilute Na-CMC concentration below 0.40 wt%, droplets are self-  
 439 propelled some distances from the wall mainly subjected to the gradient of the  
 440 surface energy and higher velocity flow stream of the main channel after the sharp  
 441 breakup occurs at the corner of T-junction. For the larger concentration of Na-CMC  
 442 ( $C > 0.40$  wt%) dispersed fluid, the presence of instabilities promotes the breakup of  
 443 jets. After the formation of the primary breakup droplets, the generated droplet  
 444 moves near the wall of microchannel as it is initially exposed to the flow projected  
 445 from the side branch at the moment of breakup. After a certain distance, it will flow  
 446 near the center of the bulk phase. The droplet with a radius in an emulsion will exert  
 447 greater pressure on the inner concave interface than on the convex side. When the

448 droplet deformation occurs, the Laplace pressure of the deformed droplet is a  
 449 function of the radius along the droplet surface (see section 3.4). Thus, as the  
 450 dispersed thread is elongated, a smaller droplet will be formed causing smaller  
 451 radius of curvature that result in a larger inward force and expected to experience a  
 452 greater pressure (see Fig. 9).  
 453



454  
 455 **Fig. 9:** A qualitative plot of the Laplace pressure profile of a generated Na-CMC  
 456 droplet interface curvature at concentration of 0.10 wt%, 0.40 wt%, 0.80 wt%, and  
 457 1.00 wt% along the middle plane of microchannel (for system:  $Q_d/Q_c=0.05$ ).  
 458

#### 459 4.2 Effect of Surface Wettability on Droplet Breakup Time and Production Rate

460 The effect of shear-thinning nature on droplet breakup process has not been  
 461 extensively studied in the preceding analysis. In a T-shaped microchannel, the  
 462 continuous and dispersed phase were dispensed at the prescribed flow rates  
 463 through the microchannel with a hydrophobic wall surface ( $\theta > 90^\circ$ ). Additionally, the  
 464 continuous phase needs to wet the surface of the channel walls preferentially in  
 465 order to repel the dispersed phase droplets away from the wall. The surface  
 466 wettability of microchannel are of utmost importance for the stability of the droplets  
 467 formation process in a microfluidic device. The effect of surface wettability on the  
 468 total droplet formation time, which is the transition between a growing and a  
 469 detached droplet, was discussed. Fig. 10 illustrates surface wettability effect on the

470 normalized droplet formation rate. Results revealed that the droplet detachment  
471 occurs more rapidly at shorter times as the  $\theta$  increases. The inertial force, shear  
472 force and surface wettability are competing effects influencing the dynamics of the  
473 droplet breakup process. The wettability driving the fluid toward the surface and the  
474 contact area between the droplets and solid surface increases for the smaller  $\theta$ . The  
475 smaller  $\theta$  tends to reduce the droplet deformation and delay the breakup process. As  
476  $\theta$  increases, the thread is no longer flowing close to the wall as its adhesion strength  
477 to the wall reduces, causing less resistance to flow. Therefore, the inertial force  
478 driving the flow of thread is prevailing. The time for the formation of droplet is also  
479 essentially dependent of the strength of shear force acting on the interface, because  
480 it dominates over the adhesion force for larger  $\theta$ . Bashir *et al.* (2011) reported that  
481 the degree of confinement promotes the breakup for the larger  $\theta$ ; whereas the  
482 degree of confinement suppresses the breakup for smaller  $\theta$ .

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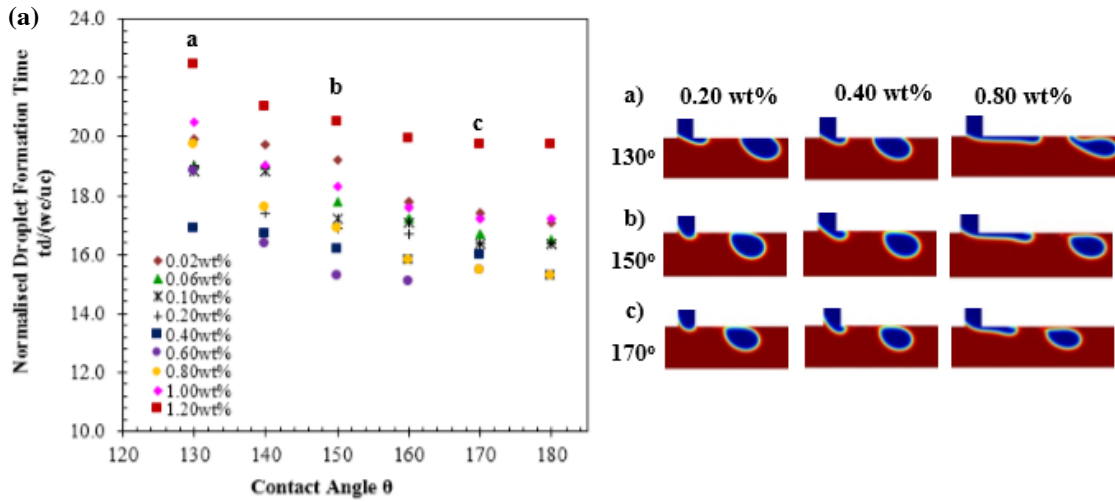
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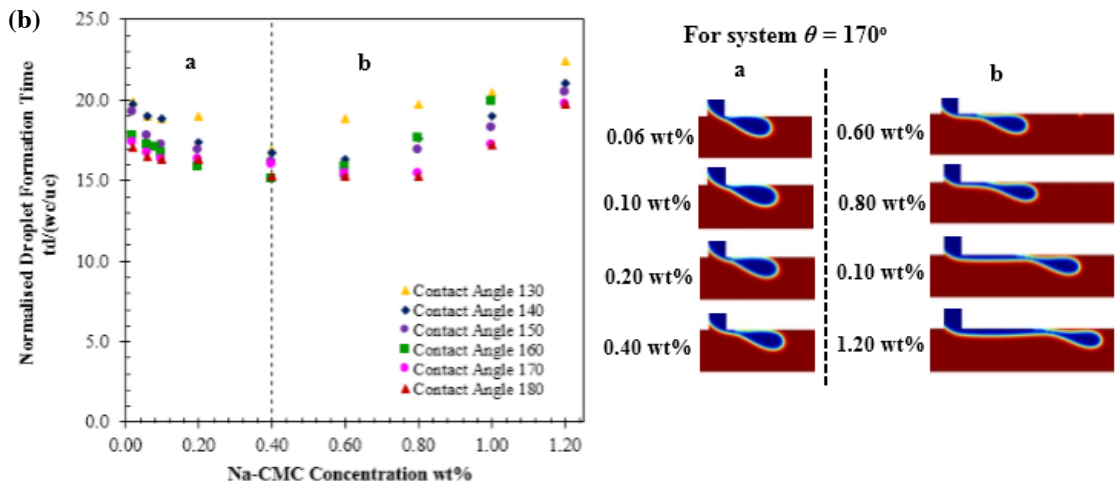
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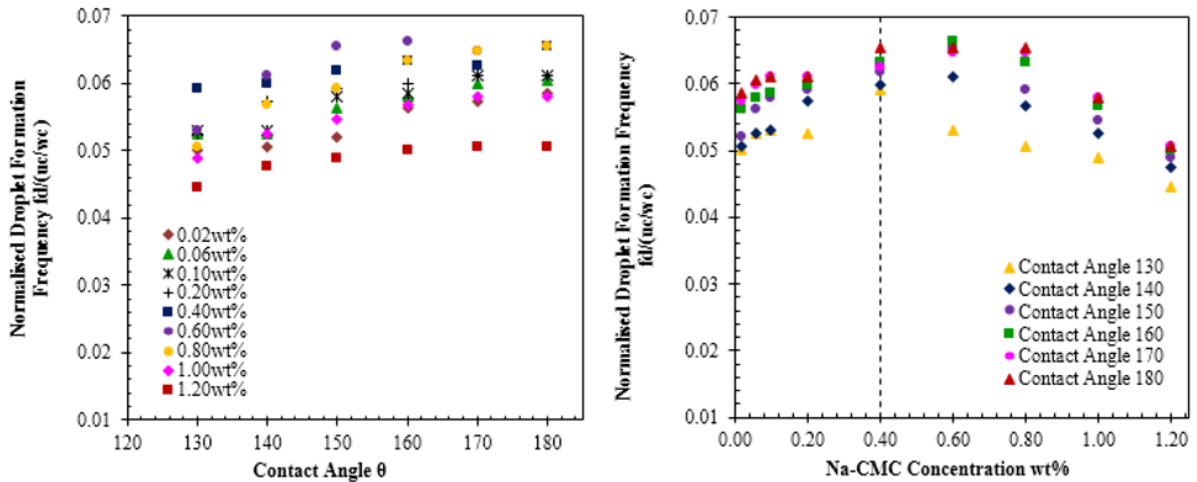
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502

503 **Fig. 10:** (a) Effect of contact angle on the normalized droplet formation time of Na-  
 504 CMC solutions. (b) Variation in normalized droplet formation rate of shear-thinning  
 505 droplets for a. dilute ( $C < 0.40$  wt%) and b. semi-dilute ( $C > 0.40$  wt%) Na-CMC  
 506 concentrations regime at fixed  $\theta$  (for system:  $Q_d/Q_c=0.05$ ).

507

508 For a fixed value of  $\theta$ , the result reveals that the droplet formation time decreases as  
509 the concentration of the Na-CMC increases from 0.02 wt% to 0.40 wt% (see Figure  
510 10). Interestingly, a similar phenomenon does not happen for the larger  
511 concentration ( $C > 0.40$  wt%). This is due to the fact that the concentration of the  
512 polymer solutions beyond the critical overlap concentration exert a significant  
513 influence on retarding the droplet breakup time. As previously mentioned, the direct  
514 intermolecular interactions can be neglected for low concentrations. Thus, the rapid  
515 pinch-off of shear-thinning droplets occurring was due to the high shear stress in  
516 these low concentrations regimes. When the Na-CMC concentration is increasing,  
517 the rheological behaviour of Na-CMC solution may be governed by the development  
518 of entanglement coupling between the chains and contribute significantly to the  
519 increasing of viscosity and the formation of thinning filament.

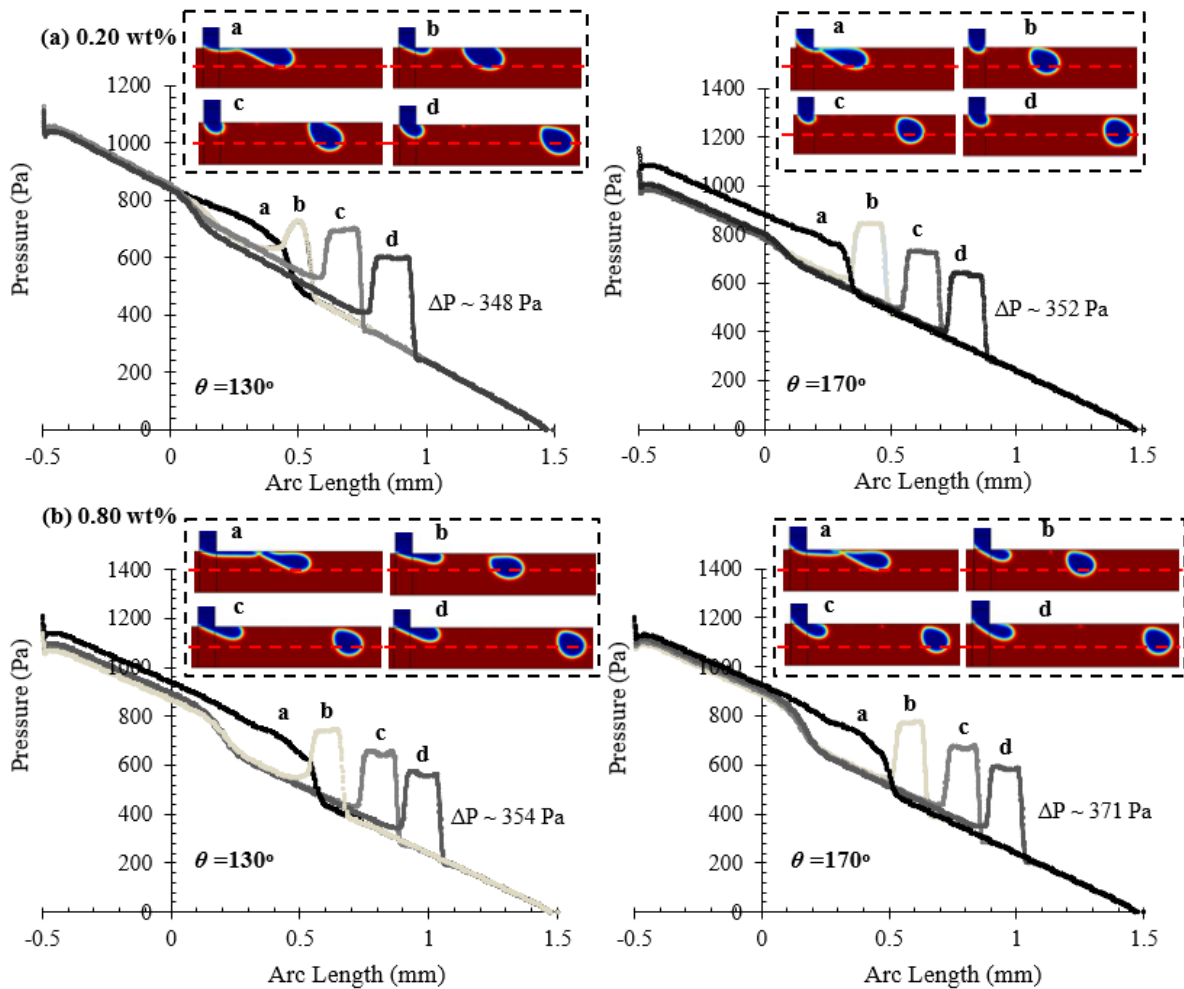
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521 Hong and Cooper-White (2009) studied the effect of non-Newtonian Carbopol  
522 dispersions on the droplet detachment behaviour in the bulk phase of silicone oil.  
523 The Carbopol droplet pinch-off time increases with increasing the viscosity of  
524 dispersions<sup>37</sup>. Besides, Arratia *et al.*(2009) also reported that the time for polymer  
525 polyacrylamide (PAA) droplet breakup was retarded as the polymer molecular weight  
526 is increased at fixed  $Q$  (Hong & Cooper-White, 2009). Husny and Cooper-White  
527 (2006) also claimed that pinch-off occurred rapidly without any significant filament  
528 formation during necking for Newtonian droplet formation; but, this rapid necking  
529 event was retarded with the formation of a thinning filament for polyethylene oxide  
530 (PEO) solutions (Husny & Cooper-White, 2006). As can be seen in Fig. 10, a similar  
531 observation was found for the higher Na-CMC concentration as the formation of a  
532 thin and stable filament between the droplet and thread is more apparent when  $C >$   
533  $0.40$  wt%. The dynamics of the thinning filament are governed by the shear stress  
534 and viscous pressure, by which the filament is elongated drastically and resists the  
535 droplet pinch-off. Thus, the delayed pinch-off may be due to the retardation effect of  
536 the fluid elasticity, which it can be increased by increasing the polymer concentration.  
537 However, the entire phenomena was observed from our predictive model with no  
538 elastic stress applied in our present study. Hence, that shear-thinning alone is also  
539 sufficient to induce a filament and that elasticity is not necessary.

540

541 In contrast, results revealed that the production rate of the Na-CMC droplet  
542 increases with enhanced surface wettability. The droplet generation frequency  
543 increases with dispersed phase concentration over the range of 0.02 wt% to 0.40  
544 wt% when the contact angle was held constant at each wetting condition. However,  
545 the rate of droplet production begins to decrease with increasing the concentration of  
546 Na-CMC dispersed phase concentration larger than 0.40 wt%, as noted in the  
547 previous section.

548 A comparison of pressure profiles of Na-CMC droplet at different contributions were  
549 illustrated in Fig. 11. At low and dilute concentrations of the Na-CMC polymer, the  
550 pressure drop of a stabilized curved surface of Na-CMC droplet is increasing along  
551 the bulk phase with increasing the wetting conditions. The weak adhesion and strong  
552 cohesion of the liquid in the bulk phase was found for the larger contact angles. The  
553 unfavourability of channel surface to the dispersed thread plays and leads a major  
554 role in the transport of droplet and lead to a larger shear force to be exerted on the  
555 drop surface. Rapid deformation of the droplet occurs due to the adequate shearing  
556 force induced by the continuous phase. Thus, the droplet mass was shown to  
557 decrease when the initial contact angle is larger and leads to a larger pressure drop.  
558 The higher concentration of Na-CMC dispersed phase contributes the same  
559 phenomenon. For the Na-CMC concentration above 0.40 wt%, lower contact angles  
560 promote and dampen greater spreading and dampens the breakup process. The  
561 spreading of the elongated thread is eventually broken up into smaller droplets when  
562 the surface energy is overcome. Neglecting the impact of fluid property on surface  
563 wettability, the much larger concentration of fluid samples withdrawn from the  
564 dispersed thread can create much greater pressure drop at the same contact angle.



565

566 **Fig. 11:** A qualitative plot of the Laplace pressure profile of a generated Na-CMC  
 567 droplet interface curvature at concentration of 0.20 wt% and 0.80 wt%, along the  
 568 middle plane of microchannel at  $\theta = 130^\circ$  and  $170^\circ$  (for system:  $Q_d/Q_c=0.05$ ).

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### 4.3 Effect of Interfacial Tensions on Droplet Breakup Time and Production

572

#### Rate

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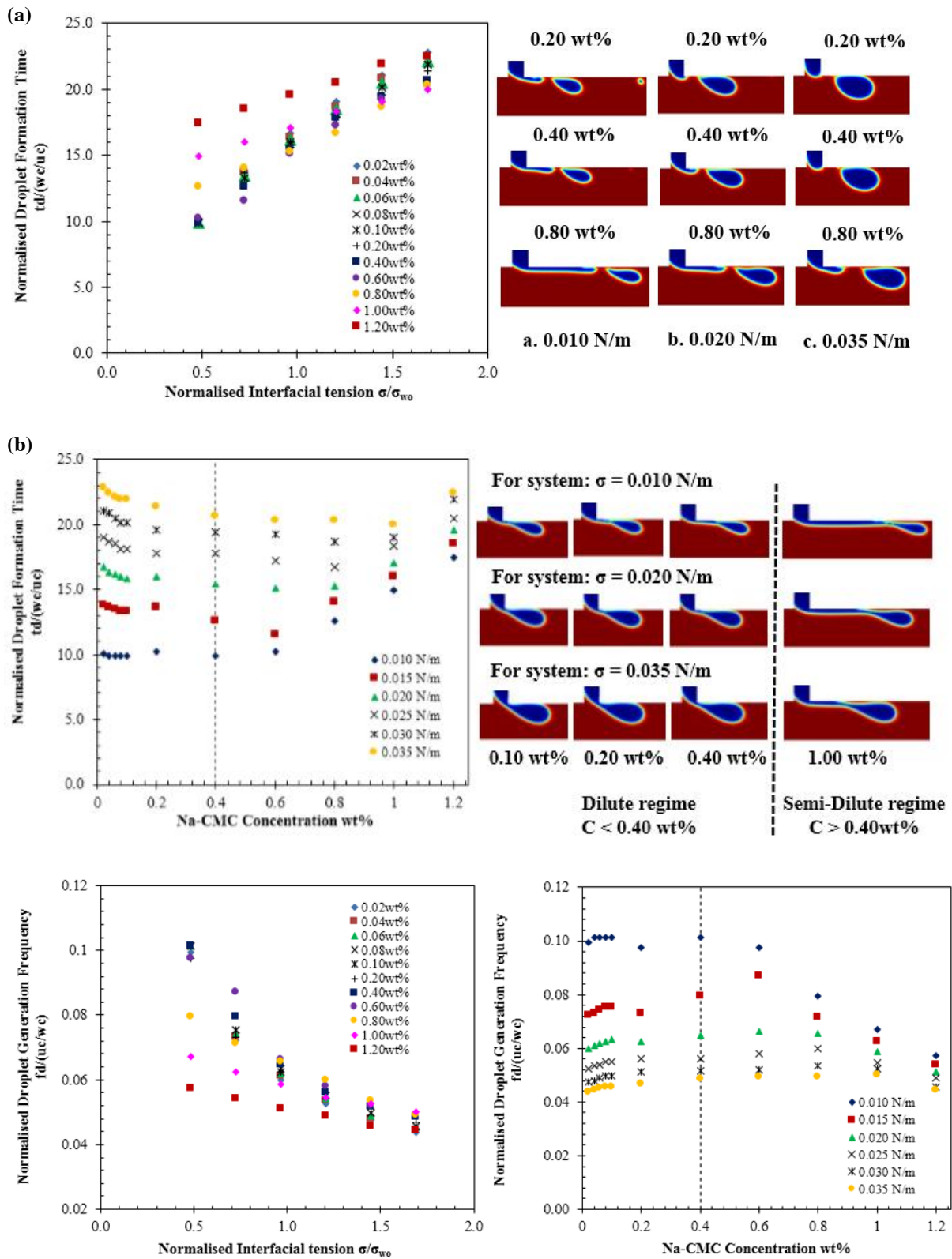
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The effect of interfacial tension on the time taken for the droplet growth, deformation and detachment were also investigated. The impact of interfacial tension on the Na-CMC droplet breakup time is illustrated in Fig. 12(a). As a result, the droplet breakup time increases with increasing interfacial tension. As interfacial tension increases, the retraction of the interface induced by surface tension forces becomes greater due to the relatively high surface free energy. This will tend to hinder the droplet formation process as the Na-CMC droplets take longer time approach to thermodynamics equilibrium. At the low concentration regime ( $C < 0.40$  wt%), the droplet is pinched-off sharply at the corner of T-junction. In addition, this breakup



582 regime is driven primarily by the build-up of pressure upstream which is mainly due  
 583 to the high degree of confinement of the droplet in bulk phase.



584

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 586

587  
 588  
 589

**Fig. 12:** (a) Effect of interfacial tension on the normalized droplet formation time of Na-CMC solutions (for system:  $Q_d/Q_c=0.05$ ). (b) Variation in normalized droplet

590 formation rate of Na-CMC droplets for various concentrations at each interfacial  
591 tension ( $\sigma$ ).

592 The droplet breakup time decreases when Na-CMC concentration is increased from  
593 0.02 wt% to 0.40 wt% (see Figure 12(b)). In contrast to this phenomenon, the break-  
594 up time increases as Na-CMC concentration is larger than 0.40 wt%. For dilute Na-  
595 CMC concentrations below 0.40 wt%, the interfacial forces are more dominant than  
596 viscous forces due to the insufficient polymer chain overlap leading to earlier  
597 occurrence of pinch-off. While the Na-CMC concentrations above 0.40 wt%,  
598 interfacial forces are less prevalent in strength and the larger viscosities give rise to  
599 longer breakup time which can be attributed to the higher magnitude of the dispersed  
600 thread pressure. A similar observation has been reported by Zhang and Basaran  
601 (1995) who studied the high viscous pendant drops (Zhang & Basaran, 1995).  
602 Tirtaatmadja *et al.* (2006) also claimed that the polymer molecules can be highly  
603 extended during their approach to pinch region and this contributes to the formation  
604 of filament. The filament can be further extended by the stretching force at a  
605 constant rate until full extension of polymer coil is achieved (Tirtaatmadja, McKinley,  
606 & Cooper-White, 2006). In general, high stretching of polymer chains is associated  
607 with high elasticity.

608

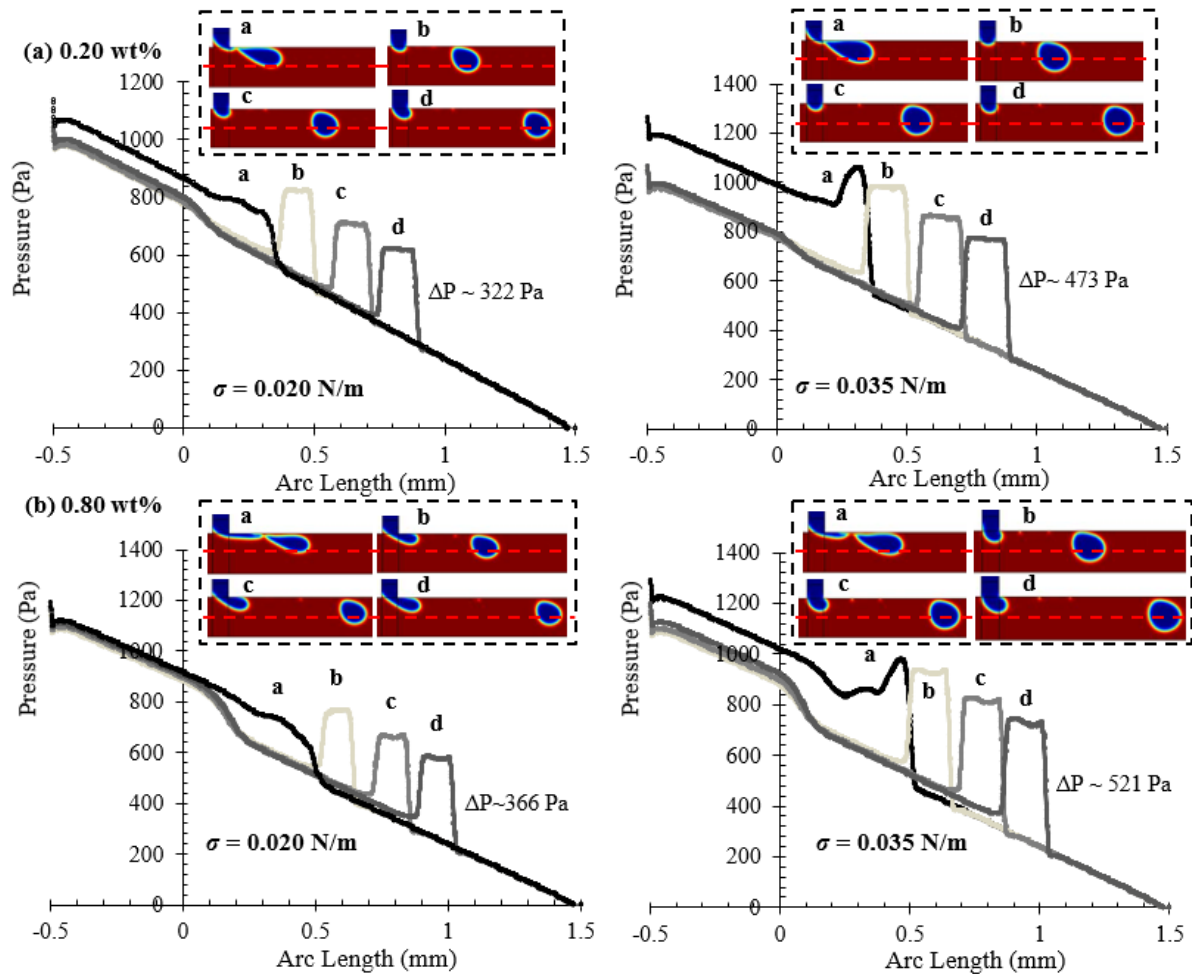
609 For simplicity the viscous force is disregarded, the jetting phenomena and the  
610 prolonged thinning of the fluid filament at the rear is more substantial for the shear-  
611 thinning droplet with lower interfacial tension at dilute concentration regimes. Less  
612 energy is required to disrupt an interface with low magnitude of interfacial tension.  
613 Thus, low interfacial tension liquid thread tends to breakup rapidly. Nevertheless, at  
614 low  $Q$ , jetting occurs when the inertial forces induced by continuous phase exceed  
615 interfacial tension forces. As noted in the previous observation, the length of filament  
616 gets longer when the Na-CMC polymer concentration is increased. This presumably  
617 prevents the neck of the dispersed thread from pinching off. Thus, the existence of a  
618 thin polymeric filament will tend to decelerate the breakup process, especially for  $C >$   
619 0.40 wt%. Previous studies have reported that the formation of filament was due to  
620 the elasticity effect. Nevertheless, a similar behaviour was also found for the working  
621 solution which is purely viscous and shear-thinning characteristics at the larger  
622 concentrations.

623

624 The breakup time for a droplet is increased leading to a decreased production rate at  
625 fixed  $Q$  for larger magnitude of interfacial tension forces. This exhibits the similar  
626 observations to the studies made by Peng *et al.* (2011) and Bashir *et al.* (2011) who  
627 studied water-oil emulsions in flow-focusing and cross-flowing microfluidics devices,  
628 respectively. The droplet production rate at each fixed interfacial tension has been  
629 investigated. In lower concentration regimes ( $C < 0.40$  wt%), the earlier occurrence  
630 of droplet breakup is observed. In contrast, the production rate decreases with  
631 increasing the Na-CMC concentration in the semi-dilute concentration regime ( $C >$   
632  $0.40$  wt%), it takes a longer time for the dispersed phase viscous force to be  
633 overcome by the opposing inertial force and shear stress induced by the continuous  
634 phase.

635

636 The pressure-jump profiles of Na-CMC droplet at different interfacial tensions were  
637 illustrated in Fig. 13. At lower concentration of Na-CMC solutions, the larger  
638 interfacial tension contributes to a larger pressure drop profile. As interfacial tension  
639 increases, the strength of attractive force increases and causes the liquid surface to  
640 contract toward the interior phase and thus the repulsive collisional forces is reduced  
641 in order to resist the contraction. Thus, the breakup process of droplet is hindered  
642 and larger droplet is generated. Similarly, fluids with the higher concentration  
643 produce larger size droplets at the larger surface tensions. Nevertheless, the  
644 generation of smaller droplets did not reveal a larger pressure-drop profiles when  
645 compare to the previous consequences. It is postulated that the effect of surface  
646 tension enables higher surface energy which is caused by the gradient of  
647 confinement. Forcing the detached droplet translocate through a confined region will  
648 increase the pressure within the droplet at the location where it is almost fully  
649 occupied. This causes the increment of the curvature effect that requires the external  
650 fluid to apply the extra pressure.



651

652 **Fig. 13:** A qualitative plot of the Laplace pressure profile of a generated Na-CMC  
 653 droplet interface curvature at concentration of 0.20 wt% and 0.80 wt% along the  
 654 middle plane of microchannel at  $\sigma = 130^\circ$  and  $170^\circ$  (for system:  $Q_d/Q_c=0.05$ ).  
 655

## 656 5. Conclusions

657 The essential role of viscosity, surface wettability, and interfacial tension on  
 658 emulsification process of Na-CMC shear-thinning droplet has been highlighted in the  
 659 present simulation using conservative level-set numerical method. The evolution of  
 660 the breakup time and the droplet production rate is effectively governed by the  
 661 physical properties of working fluid. Olive oil was selected as the continuous phase  
 662 while the Na-CMC polymer was used as the non-Newtonian dispersed phase fluid.  
 663 The present simulation data revealed that droplet breakup time and production rate  
 664 have a striking non-monotonic relationship with the Na-CMC polymer concentration  
 665 due to the considerable rheological shear-thinning nature of Na-CMC polymer  
 666 solution. As the concentration increases, the polymer concentration crosses over  
 667 from the dilute to semi-dilute regime. While  $C < 0.40$  wt%, the droplet breakup time

668 decreases when the Na-CMC concentration is increased at fixed  $\theta$  and  $\sigma$ . While in  
669 semi-dilute regime, droplet breakup time increases when if Na-CMC concentration is  
670 increased. This phenomenon is mainly due to the dispersed phase viscous forces  
671 dominating over the breakup dynamics and relevant hydrodynamics. As Na-CMC  
672 concentration increases, a laminar elongated dispersed thread is formed connecting  
673 to the primary droplet due to the high viscous pressure, and thus the droplet breakup  
674 point moves progressively downstream of the outlet channel. The presence of high  
675 concentration of polymer molecules leads to a prolonged fluid thread and retardation  
676 of pinch-off development. In present parametric analysis, there are many features  
677 that were previously attributed to elastic effects that still remain a defining challenge  
678 for the highly shear-thinning and viscous Na-CMC polymer solution. This illustrates  
679 the potential of integrating the elastic stress model with present numerical method, in  
680 order to investigate the fluid elasticity effect on the growth of droplet with shear-  
681 thinning characteristics.

682

## 683 **Nomenclature**

684  $a$  Fitting parameter in Carreau-Yasuda model

685  $d_{eff}$  Effective droplet diameter ( $\mu\text{m}$ )

686  $F_{st}$  Surface tension force acting on the interface ( $\text{N}/\text{m}^3$ )

687  $F_D$  Cross flow drag force ( $\text{N}/\text{m}^3$ )

688  $h$  Depth of the channel ( $\mu\text{m}$ )

689  $I$  Identity matrix

690  $k$  Curvature of fluid-fluid interface

691  $n$  Power-law exponent for Carreau-Yasuda Model

692  $\mathbf{n}_r$  Unit normal vector at the interface

693  $p$  Pressure ( $\text{N}/\text{m}^2$ )

694  $Q$  Flow rate ratio

695  $Q_c$  Flow rate of the continuous phase ( $\text{ml}/\text{hr}$ )

696	$Q_d$	Flow rate of the dispersed phase (ml/hr)
697	$t$	Time-step (s)
698	$u$	Velocity component in x-direction (m/s)
699	$v$	Velocity component in y-direction (y-direction) (m/s)
700	$\mathbf{u}$	Velocity field
701	$R$	Curvature radii of the interface (m)

702

### 703 **Greek Symbols**

704	$\eta$	Dynamic viscosity of fluid (Pa.s)
705	$\eta_d$	Dynamic viscosity of the dispersed phase (Pa.s)
706	$\eta_c$	Dynamic viscosity of the continuous phase (Pa.s)
707	$\eta_o$	Zero shear viscosity (Pa.s)
708	$\eta_\infty$	Infinite shear viscosity (Pa.s)
709	$\lambda_\eta$	Viscosity ratio ( $\eta_d/\eta_c$ )
710	$\lambda_{CY}$	Relaxation Time in Carreau-Yasuda Model(s)
711	$\rho$	Fluid density (kg/m <sup>3</sup> )
712	$\tau$	Shear stress (Pa)
713	$\dot{\gamma}$	Shear rate (1/s)
714	$\gamma$	Reinitialization parameter (m/s)
715	$\varepsilon$	Thickness of the interface (m)
716	$\sigma$	Interfacial tension (mN/m)
717	$\phi$	Level set function
718	$\delta_{sm}$	Dirac delta function concentrated at interface

719  $\Omega$  Computational domain

720  $\partial\Omega$  Domain boundary

## 721 **Acknowledgement**

722 This research project was financially supported by Malaysia Intercampus Doctoral  
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