

1 **The role of *Cymodocea nodosa* on the dynamics of**
2 **trace elements in different marine environmental**
3 **compartments at the Mar Menor Lagoon (Spain)**

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13 **Abstract**

14 During mining activities historically developed at Sierra Minera (Cartagena-La
15 Unión, Spain), high amounts of trace elements were discharged to the Mar
16 Menor coastal lagoon mainly through El Beal Wadi. The objective of this study
17 is to establish the role played by the *Cymodocea nodosa* in the coastal marine
18 dynamics of trace elements at the mouth of the wadi. To this end, the content of
19 nine trace elements (As, Cd, Cr, Cu, Fe, Mn, Ni, Pb and Zn) in different marine
20 environmental compartments (i.e. marine and coastal sediments, *C. nodosa*
21 tissues collected from live seagrass and *C. nodosa* beach cast litter) at two
22 different locations were determined by inductively coupled plasma atomic
23 emission spectrometry. The results showed that the seagrass *C. nodosa* could
24 mobilise part of the elements present in marine sediments and water, thereby
25 causing their re-accumulation in the coastal sediments through the *C. nodosa*
26 beach cast litter.

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31 **Keywords**

32 Mining waste; Marine chemistry; Trace metals mobilisation; Analytical
33 chemistry; Seagrass leaf litter beds; Mar Menor coastal lagoon

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36 **1. Introduction**

37 The Mar Menor is one of the biggest hypersaline coastal lagoons (salinity
38 ranging between 38 and 51 psu) in the Mediterranean Sea. The lagoon is
39 relatively shallow (i.e. mean depth of 3.5 m) with a surface area of 135 km² and
40 a water volume of approximately 580 million m³ (María-Cervantes et al., 2009;
41 Marín-Guirao et al., 2005c; Pérez-Ruzafa et al., 2005). It is an area with high
42 ecological value and is currently protected by different multi-statement laws and
43 conventions. It was declared a national protected area (Jefe de Estado S. R. M.
44 Juan Carlos I, 1989) and, since 2004, a RAMSAR international site. It has also
45 been declared a *Site of Community Importance (SCI)* and a *Special Area of*
46 *Conservation* under the Habitats Directive integrated in the Nature 2000
47 Network (EU habitats Directive). Moreover, it was established a *Special*
48 *Protected Area of Mediterranean Interest* by Barcelona Convention 2001 (Da
49 Cruz and Consejería de Agricultura, 2003; Gutiérrez and Giménez, 2009).

50 Despite these regulations, the lagoon is exposed to important environmental
51 threats related to anthropogenic activities. One of these threats is the mining
52 waste from the mining district of Sierra Minera (Cartagena-La Unión, Murcia,
53 Spain) that arrives to the lagoon, even to date. The mining district of Sierra
54 Minera was a very important mining nucleus for more than 2500 years located
55 close to the southern shore of the Mar Menor Lagoon. For hundreds of years,
56 but mainly during the 30 years of mining activity developed in this zone in the
57 20th century, an important amount of mining wastes was directly discharged to
58 the Mar Menor coastal lagoon mainly through two important temporally water

59 courses (i.e. El Beal and Ponce Wadis)(Conesa and Jiménez-Cárceles, 2007;
60 García-García, 2004; Navarro et al., 2008; Pérez-Ruzafa et al., 1987;
61 Simonneau, 1973). Presently (i.e. 30 years later), sediments with high amounts
62 of toxic elements (mainly Pb, Zn, Fe, Cu and Cd) arrive to the lagoon, even to
63 date, through the typical torrential rains occurring at this region (Conesa and
64 Jiménez-Cárceles, 2007; Gonzalez-Fernandez et al., 2011; Gutiérrez and
65 Giménez, 2009; Marín-Guirao et al., 2007).

66 Several studies in the last decade have pointed out elevated concentrations of
67 toxic metal elements in this area (i.e. mining district and the coastal lagoon).
68 Most of the studies have been performed across the mining area and at the
69 different wadis that drain these mining zones. The results confirmed that a large
70 proportion of mining wastes, which remain accumulated in this zone, could be
71 dispersed through the surrounded areas because of the weathering processes
72 (i.e. torrential rainfalls and strong winds), thereby releasing great amounts of
73 trace elements through the stream (García-Lorenzo *et al.*, 2014, 2012; García
74 *et al.*, 2007; González-Alcaraz *et al.*, 2013; Gonzalez-Fernandez *et al.*, 2011;
75 Navarro *et al.*, 2008). These mining wastes, which arrive to the Mar Menor
76 Lagoon through the wadis, transport high amounts of trace elements to water,
77 and this has caused an important impact on the entire coastal lagoon, as mining
78 wastes could be redistributed from the estuary zones to other nearby zones
79 because of variations between different environmental factors such as marine
80 currents, winds, and the existence of temperature and salinity gradients
81 (Dassenakis *et al.*, 2010; García and Muñoz-Vera, 2015; Tsakovski *et al.*,
82 2012). However, the most affected areas of the lagoon are the mouths of the
83 wadis, mainly El Beal Wadi. Studies performed at the mouth of this wadi have

84 revealed concentration values of trace elements in sediments and the water
85 column, and these values were found to exceed some of the European limits
86 established in different soil regulations (Gonzalez-Fernandez *et al.*, 2011;
87 Tsakovski *et al.*, 2009).

88 The presence of high amounts of trace elements could affect adversely the
89 entire marine environment of the lagoon because of their availability for the
90 different organisms living there. Specifically, seagrasses present significant
91 capacity to incorporate trace elements in their tissues, which might be available
92 to higher trophic levels in the marine food chain (Di Leonardo *et al.*, 2017;
93 Lyngby and Brix, 1989). Although some of these elements are considered
94 essential (i.e. Fe, Zn, Mn, Cu and Ni), others can cause adverse effects in the
95 ecosystem, such as the nonessential elements (i.e. Pb, As, Cr and Cd), if they
96 are present at elevated concentrations. In case of the Mar Menor Lagoon, the
97 marine flora and fauna are continuously exposed to high concentrations of trace
98 elements. Some studies have evaluated the toxicity of these elements for
99 different marine organisms and the marine phanerogam *Cymodocea nodosa*
100 (Ucria) Aschers., which is the characteristic seagrass of this zone. Most studies
101 reported a relation between the elemental concentrations found in the marine
102 fauna and those found in the sediment and the different tissues of the *C.*
103 *nodosa*, thus ensuring the transfer of the elements to the food chain (Conesa *et*
104 *al.*, 2011; María-Cervantes, 2009; Marín-Guirao *et al.*, 2008, 2005a, 2005b,
105 2005c; Sanchiz *et al.*, 2000).

106 Furthermore, trace elements accumulated in the seagrass could be mobilised or
107 transferred to the marine environment after shedding of the oldest leaves that
108 remain deposited in the seagrass beds. Despite this issue, there are few studies

109 on the impact of detached leaves of seagrass in the marine environment.
110 Hosokawa et al., (2016) evaluated the effects of the leaf litter of the seagrass
111 *Zostera marina* L. (eelgrass) accumulated in the seagrass beds in the mouth of
112 Tokyo Bay (Japan) and observed an exchange of trace elements (Cu, Zn, Pb
113 and Cd) between leaves during decomposition and in the sediments of the
114 seagrass bed. Di Leonardo et al., (2017) also evaluated the trace element
115 levels (As, Cd, Co, Cr, Cu, Hg, Mn, Mo, Ni, Pb, V and Zn) and organic carbon
116 storage capacity of *Posidonia oceanica* (L.) Delile leaves and its impact on
117 sediments after shedding in the seagrass bed in the Augusta Bay (Italy). They
118 reported that *P. oceanica* banquettes have an important role in the mobilisation
119 of trace elements in the marine environment. Moreover, the oldest seagrass
120 leaves detached from the seagrass might be dragged by the marine currents
121 and the swell to the shore, where leaves could remain accumulated. This
122 seagrass beach cast litter, which contributes to the beach stability, might
123 support high densities of faunal populations and constitutes an important source
124 of nutrients for the organisms (Ochieng and Erfemeijer, 1999). Thus, the trace
125 elements accumulated in the seagrass could be available to the faunal
126 community and the vegetation and the sediment of the coastal environment.

127 Nonetheless, to the best of our knowledge, none of the studies performed in the
128 Mar Menor coastal lagoon have considered the role of the *C. nodosa* beach
129 cast litter in the mobilisation of trace elements or its impact on the coastal zone.
130 Therefore, the goal of the present work is to establish the role played by the *C.*
131 *nodosa* beach cast litter on the coastal marine dynamics of trace elements in
132 one of the most affected areas of the Mar Menor Lagoon (Spain). To this end,
133 the content of nine trace elements (i.e. As, Cd, Cr, Cu, Fe, Mn, Ni, Pb and Zn)

134 was determined in different marine compartments (marine and coastal
135 sediments, the seagrass *C. nodosa* and *C. nodosa* beach cast litter) at the
136 mouth of El Beal Wadi (Mar Menor Lagoon).

137

138 **2. Materials and methods**

139 2.1 Study location and sample collection

140 The study area was sited at two different localities at the Mar Menor coastal
141 lagoon, both separated by a distance of 17 km (Fig. 1): (i) El Beal locality (B)
142 sited at the mouth of El Beal Wadi (37° 39' 57.4" N, 0° 48' 47.2" W), which is
143 directly affected by the mining waste of Sierra Minera (Cartagena-La Unión,
144 Murcia, Spain). According to previous literature, this is one of the most
145 contaminated locations at the lagoon (Simonneau, 1973). (ii) San Pedro locality
146 (SP) sited at north of the lagoon, far away from the mining waste discharge
147 point (37° 49' 4.6" N, 0° 46' 28.9" W) (Fig.1). Both localities have similar
148 hydrological conditions (e.g. water temperature, nitrate concentration, spatial
149 and temporal dynamics of chlorophyll) with the exception of toxic metal
150 contamination by mining wastes (Pérez-Ruzafa et al., 2005; Umgieser et al.,
151 2014). With regard to anthropogenic factors, both localities have in their vicinity
152 a marina at a distance of approximately 1 km, and none of them are bathing
153 area for tourists; hence, there is no withdrawal of the seaweed litter washed
154 ashore.

155 Two zones were studied for each locality: marine and coastal. In each zone, two
156 sites separated from each other by a distance of between 300 and 350 meters
157 and randomly chosen were sampled. Marine sites were characterised by the

158 presence of a *C. nodosa* meadow few meters away from the coastline and an
159 average depth of 60 cm. On the other hand, coastal sites were located in front
160 of the marine sites where accumulations of seagrass beach cast litter were
161 deposited in the shore by wave action. For each site, three replicates of
162 sediments (marine and coastal) and three samples of seagrass (live *C. nodosa*
163 and beach cast litter) were collected using a square box (22.5 x 22.5 cm). In
164 case of the marine zone, the entire shoots of *C. nodosa* (including leaves,
165 rhizomes and roots) and the sediment surrounding the seagrass rhizomes were
166 collected. For coastal samples, both the *C. nodosa* beach cast litter washed
167 ashore and the sediments under those rests were collected. *C. nodosa* beach
168 cast litter was composed of (mainly) leaves and some amounts of both
169 rhizomes and roots. In all cases, coastal and marine sediment samples were
170 collected from the surface (approximately 5 cm depth) of the sediments
171 surrounding the *C. nodosa* samples collected and the marine samples were
172 collected at a depth of between 0.5 to 1 m.

173

174 2.2 Reagents

175 High-purity water with a resistivity > 18 M Ω cm obtained from a Milli-Q water
176 Direct-Q3 purification system (Millipore Inc., Paris, France) was used
177 throughout this work. For sample digestion, 65% w w⁻¹ HNO₃, 30% w v⁻¹ H₂O₂
178 (Panreac, Castellar del Valles, Spain) and 65% w w⁻¹ HClO₄ (Carlo Erba
179 Reagents, Sabadell, Spain) were used. Calibration standard solutions were
180 prepared from 1000 mg L⁻¹ multi-elemental ICP reference solution (ICP-IV;
181 Merck, Darmstadt, Germany) and As mono-elemental solution (Merck,
182 Darmstadt, Germany).

183

184 2.3 Sample preparation

185 Sediment samples were air-dried for 2 days and then oven-dried at 60 °C until
186 constant weight. Once mollusc and pebbles were removed, sediments were
187 grinded in a glass mortar and finally sieved (0.5 mm) and stored in properly
188 labelled polypropylene bottles.

189 All *C. nodosa* samples were first rinsed with tap water and then scraped with a
190 glass slide to remove the leaves fouling and then, rinsed again with deionised
191 water. Finally, samples were cut, oven-dried at 60 °C until constant weight and
192 then stored in properly labelled polypropylene bottles.

193 Prior to the analysis, samples were digested using a microwave oven (model
194 Start D, Milestone S.r.l., Sorisole, Italy) at conditions recommended by the
195 manufacturer (Programmes HPR-EN-33 and HPR-AG-02 for sediments and
196 seagrasses, respectively). Digestion of samples collected at each sampling
197 point was done in triplicate. Hence, at the end of the digestion process, for each
198 sampling point, 9 solutions were available for each sediment or seagrass
199 sample. For the digestion of sediments, 6 ml of HNO₃ 65% w w⁻¹, 1 ml of HClO₄
200 65% w w⁻¹ and 1 mL of H₂O₂ 30% w v⁻¹ were added to 0.25 g of sample in
201 Teflon vessels. Samples were heated for 30 minutes at maximum power. After
202 the digestion process, the solutions obtained were totally clear, and no
203 sediments or deposits were observed.

204 For the digestion of seagrass samples, 6 ml of HNO₃ 65% w w⁻¹ and 2 mL of
205 H₂O₂ 30% w v⁻¹ were added to 0.5 g of sample. The heating programme used
206 was same as that of the sediment digestion procedure.

207 After digestion, samples were brought to a final weight of 25 g with ultrapure
208 water. For analysis of most concentrated elements (i.e. Fe, Pb and Zn), these
209 solutions were diluted 1:10 with ultrapure water prior to the analysis, while the
210 rest of the elements (i.e. Mn, As, Cd, Cr, Cu and Ni) were analysed directly.

211

212 2.4 Trace metal analysis

213 2.4.1 Analysis by ICP-AES

214 The concentration of trace elements (i.e. As, Cd, Cr, Cu, Fe, Mn, Ni, Pb and Zn)
215 was determined by inductively coupled plasma atomic emission spectrometry
216 (ICP-AES; model 720, Agilent, Santa Clara, CA, USA). Details of operating
217 conditions and emission lines selected are presented in Table 1.

218 Internal standardisation (IS) was employed as a calibration strategy to perform
219 the ICP-AES determination to minimise the signal variations during the analysis
220 as a consequence of the complex matrix samples (mainly acid and easily
221 ionised elements). (Todolí and Mermet, 1999; Todolí *et al.*, 2002) Yttrium (Table
222 1) was used as the internal standard throughout the work (Ivaldi and Tyson,
223 1996; Shirdam *et al.*, 2008; Zachariadis and Vogiatzis, 2010). Two Yttrium
224 emission lines, ionic and atomic, were selected to correct for the possible
225 differences in ionisation and/or excitation mechanisms for each line. Hence, the
226 internal standard emission line employed for As, Cr, and Cu determination was
227 the atomic one (Y 410.237 nm), whereas the remaining elements were
228 determined using the ionic one (Y 324.228 nm).

229

230 *2.4.2 Method Validation*

231 The analyte methodology developed for metal analysis in sediment and
232 seagrass samples was validated by recovery because of the lack of a proper
233 certified material. To this end and considering the concentrations previously
234 reported in the literature (Conesa et al., 2011; García-García, 2004; Marín-
235 Guirao et al., 2005c), sediment and seagrass samples were spiked at different
236 concentrations (i.e. major elements 100 mg kg^{-1} , minor elements 0.5 mg kg^{-1})
237 with a multi-element ICP reference solution before the mineralisation step. In all
238 cases, recovery values were quantitative ($100 \pm 30\%$). The method detection
239 limits (see Table 1) determined for the elements analysed were calculated
240 attending to the 3σ criterion (Miller and Miller, 2002).

241

242 *2.4.3 Statistical analysis*

243 The content of nine trace elements (i.e. As, Cd, Cr, Cu, Fe, Mn, Ni, Pb and Zn)
244 in the seagrass and sediment samples was analysed separately using a nested
245 3-factor mixed ANOVA model to determine differences among two littoral zones
246 (coastal and marine) at two localities using site replication factor as it is shown
247 by the sampling design (site within locality by zone).

248 The following factors were considered: (1) L: Locality, a fixed factor, with two
249 levels (B: El Beal and SP: San Pedro); (2) Z: Zone, fixed and orthogonal to the
250 previous one, with two levels (C: coastal; M: marine); (3) S: Site, a random
251 factor and nested in the interaction term Locality x Zone, with two levels
252 (Underwood, 1997). From each site, three samples were collected ($n=3$). The
253 model of source of variation was as follows:

254 $X = \mu + L + Z + S(L \times Z) + (L \times Z) + Residual$

255 Prior to ANOVA, heterogeneity of variance was tested with Cochran's C-test
256 (Sokal and Rohlf, 1981). Data were transformed with the square root of $x+1$ if
257 variances were significantly different from $P=0.05$, and $\log(x+1)$ transformed if
258 variance was still heterogeneous. Where variances remained heterogeneous,
259 untransformed data were analysed, as ANOVA is a robust statistical test and is
260 relatively unaffected by heterogeneity of variances, particularly in balanced
261 experiments (Underwood, 1993, 1994). However, in such cases, special care
262 was taken in the interpretation of results. Furthermore, in such cases, to reduce
263 type I error, the level of significance was reduced to <0.01 . *A posteriori* multiple
264 comparison was carried out with the Student–Newman–Keuls (SNK) procedure
265 (Underwood, 1997).

266

267 **3. Results**

268 The concentrations obtained for all elements in the sediment and seagrass
269 samples at the two localities tested are summarised in Table 2 and Fig 2. As
270 expected, samples from the El Beal (B) locality present higher metal
271 concentrations than samples from the San Pedro (SP) locality. Results also
272 indicate that coastal samples (sediments and seagrass) show higher
273 concentration levels than marine ones.

274 Regarding metal concentration values in sediment samples, Fe, Pb, Zn and Mn
275 are the most concentrated elements, independent of the locality and zone.
276 Nevertheless, the lowest concentration of elements depends on the locality
277 considered. Thus, irrespective of the type of sediment (i.e. coastal or marine),

278 Cd is the element with the lowest concentration value reported at the El Beal
279 locality. With regard to the San Pedro locality, As, Ni and Cd are not detected in
280 any sediment sample, while Cr, with a concentration of approximately 3 mg kg⁻¹
281 in both sediments (marine and coastal), is found at the lowest concentration
282 detected. Chromium concentration levels in sediments from the San Pedro
283 locality are significantly lower than those found in the El Beal locality (i.e.
284 sevenfold to tenfold). Although natural phenomena could affect metal levels in
285 sediment samples (i.e. year and season of sampling, sampling location,
286 meteorological conditions, etc.) (García and Muñoz-Vera, 2015), the values
287 obtained in the present work for El Beal samples are comparable with those
288 previously reported by other authors (Table 3). To evaluate the current degree
289 of contamination of this area, after the mines were closed 30 years ago, the
290 concentrations determined in the coastal sediments from the El Beal locality
291 have been compared with the current limits established by different regulations.
292 Table 4 reports the concentration limits established for different types of soils
293 according to regulations by some European countries and the concentration
294 limits established in the region of Murcia (where the study is performed) that
295 were considered (European Parliament, 2008; Faz et al., 2009). When
296 comparing data in Tables 2 and 4, it can be derived that, despite potential
297 differences in soil characteristics, Pb, Zn, Cu and Cd levels in the coastal
298 sediment at the El Beal locality exceed, in terms of not only the limits
299 established by the different European countries (García-García, 2004; María-
300 Cervantes, 2009) but also both the background levels and those established for
301 contaminated soils in the Mar Menor area (Faz *et al.*, 2009). These results
302 clearly indicate the high pollution level by toxic metals at the El Beal locality.

303 In the case of seagrass samples, similar to that observed for sediment samples,
304 the elements that presented higher concentration values are Fe, Pb, Zn and
305 Mn, whereas As, Cu and Ni are found at trace levels. Interestingly, As is
306 detected only in samples from the El Beal locality. The remaining elements (i.e.
307 Cr and Cd) are not detected in any seagrass sample. In this case, the results
308 obtained for *C. nodosa* samples are also comparable with those reported in
309 previous studies (see Table 3); however, there are no concentration values for
310 *C. nodosa* beach cast litter, as it has not been previously studied in the Mar
311 Menor Lagoon.

312 To evaluate the differences in the elemental concentration pattern of sediment
313 and seagrass samples from El Beal and San Pedro, a nested 3-factor mixed
314 ANOVA model was applied. For sediments, the ANOVA test was applied to
315 those elements present in all samples investigated (Fe, Pb, Zn, Mn, Cu and Cr).
316 Arsenic, nickel and cadmium were discarded, as they are not present in the
317 sediments from San Pedro. Similarly, the ANOVA test for seagrass samples
318 was applied only to elements present in the El Beal and San Pedro samples
319 (i.e. Fe, Pb, Zn, Mn and Cu). Table 5 shows the results obtained in the ANOVA
320 analysis for both sediment and seagrass samples at the El Beal and San Pedro
321 localities. Fe concentration values in the sediment samples are consistently
322 higher at the El Beal locality (L), but they showed no difference between zones
323 (coastal and marine, Z). With regard to Pb, Zn, Cu and Cr, concentration values
324 in the coastal sediments are 330, 520, 20 and 10 times higher, respectively, in
325 the El Beal locality than in the San Pedro locality. Moreover, the concentration
326 pattern of these elements in the sediment samples shows significant differences
327 in the interaction (LxZ). While in the San Pedro locality, there are no differences

328 between zones, in the El Beal locality, the values obtained on the coastal
329 samples are higher than those obtained in the marine zone (SNK: $p < 0.01$).
330 Finally, Mn concentration values in sediment samples from the El Beal locality
331 are significantly higher than those in the San Pedro locality. Nevertheless,
332 concentration values in marine sediment samples are significantly higher than
333 those in the coastal ones from the El Beal locality (SNK: $p < 0.01$), unlike what is
334 observed for the other elements. On the contrary, there are no significant
335 differences between zones in the San Pedro locality. For seagrass samples, Fe
336 concentration values do not show significant differences, between neither zones
337 nor localities. For the rest of the elements (Pb, Zn, Mn, and Cu), the
338 concentration values in *C. nodosa* beach cast litter are significantly higher in the
339 El Beal locality than in the San Pedro locality, and these values are also higher
340 than those in *C. nodosa* samples for each locality (SNK: $p < 0.01$).

341

342 **4. Discussion**

343 The localities selected in this study (El Beal and San Pedro) have similar
344 characteristics (i.e. hydrological, external factors, etc.); hence, it is feasible to
345 evaluate the influence of *C. nodosa* on the mobilisation of trace elements in one
346 of the most contaminated areas of the Mar Menor Lagoon regarding an area not
347 directly affected by mining activity.

348 To this end, it is important to analyse and evaluate the elemental composition of
349 the sediments in which the *C. nodosa* has grown. Given that mining wastes
350 were discharged to the lagoon through El Beal Wadi, it was totally expected that
351 the metal concentration values obtained were significantly higher in the El Beal

352 locality than in the San Pedro locality (Dassenakis et al., 2010; García-García,
353 2004; Navarro et al., 2008). Moreover, the concentration values obtained for
354 each element analysed are in good agreement with the original composition of
355 the ore mine, as Fe, Pb, Zn and Mn showed higher concentration levels than
356 As, Cu, Ni, which appeared as trace elements in the ore mine (García-García,
357 2004; María-Cervantes, 2009; Marín-Guirao et al., 2005a). Differences in the
358 metal level could also be attributed to soil characteristics. The discharge of
359 mining wastes through El Beal Wadi has changed the original granulometry and
360 composition (SiO_2) of lagoon sediments at this locality. It has been found that
361 sediments at the El Beal locality contain clay, which favours the retention of
362 elements in the soil dissolved fraction (Conesa et al., 2011; Marín-Guirao et al.,
363 2008, 2005a, 2005b; Zaaboub et al., 2014). Differences in the composition of
364 both coastal and marine sediments could be explained considering that
365 elements in marine sediments are partially dissolved with time into the lagoon.
366 Umgiesser et al. (2014) estimated that the water renewal time of the lagoon is
367 longer than 1 year because of its choked nature. On the other hand, it is
368 essential to consider the formation of soluble metal sulphides that occurs
369 because of either the high concentration of organic matter present in the Mar
370 Menor Lagoon or the mining wastes rich in sulphide ions (Chapman *et al.*,
371 1983). The specific pattern shown by Mn in the marine samples collected from
372 the El Beal locality can be explained considering that Mn is present in three
373 different forms, i.e. in its insoluble forms (Mn^{3+} and Mn^{4+}), in the superficial
374 aerobic layer and in its soluble form (Mn^{2+}) in a deeper layer where manganese
375 is retained by oxidation–reduction cycling, diffusing between these two layers,
376 which will depend on the reactivity of this element and the abiotic factors of the

377 media (i.e. pH, redox potential, etc.) (Anschutz et al., 2005; Delfino and Lee,
378 1968; Grygo-Szymanko et al., 2015; Hierro et al., 2014; Jara-Marini et al.,
379 2015).

380 When the elemental concentration pattern for both sediments and *C. nodosa* is
381 compared, it is clearly observed that the metal content in the live *C. nodosa*
382 reflects the composition of the sediment bed where it has grown. This means
383 that *C. nodosa* samples collected in the El Beal locality have higher metal levels
384 due to the mining wastes. In general, bioconcentration factors (defined as the
385 ratio between the concentration of metal in the seagrass and that in the
386 sediments) (Mountouris et al., 2002) for *C. nodosa* samples are lower at the El
387 Beal locality than in the San Pedro locality. Thus, for instance, Fe
388 bioconcentration factor was 0.03 and 0.16 for *C. nodosa* samples from El Beal
389 and San Pedro, respectively. These differences in bioconcentration factors are
390 explained considering the high metal levels for the marine sediments at El Beal.
391 In the case of Fe, concentration levels in *C. nodosa* at the El Beal locality are
392 tenfold higher than those at the San Pedro locality, whereas levels in sediments
393 are 65-fold higher. Similar conclusions could be obtained for the remaining
394 elements studied.

395 Differences in elemental composition between the seagrass beach cast litter
396 and *C. nodosa* can be explained by considering different factors. First, trace
397 elements accumulated into the leaves of live seagrass could be leaked from
398 them to the water (Brix and Lyngby, 1982). Second, the elemental concentration
399 in seagrass beach cast litter depends on the proportion of leaves and roots
400 dragged to the shore. It is well known that inside the live *C. nodosa*, the
401 elements undergo translocation processes, which involve the transport of the

402 trace elements to different organs of the plant according to the necessities
403 thereof (Malea and Haritonidis, 1999). Indeed, two groups of elements can be
404 distinguished attending to their function in the seagrass. The first group
405 comprises elements linked to plant physiological functions, i.e. Fe, Zn, Mn, Cu
406 and Ni, while the second one comprises elements that are toxic for plants, i.e.
407 As and Pb. Furthermore, previous studies have demonstrated that As, Cu, Zn,
408 Ni and Mn are accumulated into *C. nodosa* leaves and Fe and Pb into roots
409 (Llagostera *et al.*, 2011; Malea and Haritonidis, 1999; Malea and Kevrekidis,
410 2013). Third, the higher elemental content shown in the beach cast litter could
411 also be explained considering metal bioadsorption ability of the detached leaves
412 (Sánchez *et al.*, 1999) and the organic matter loss by the meteorological
413 conditions and the coastal fauna, which contribute to increase in the elemental
414 concentration by biomass unit of beach cast litter (Di Leonardo *et al.*, 2017;
415 Ochieng and Erfteimeijer, 1999). Finally, when the seagrass beach cast litter
416 begins to decompose, the exchange of elements from these detached leaves to
417 the sediments contributes to the increase in metal concentration in the coastal
418 sediment. Muñoz-Vera *et al.* (2016, 2015) reported similar conclusions in
419 different bioaccumulation studies with two different jellyfish species. They
420 highlighted the hazard associated with the deposition of these organisms in
421 soils and landfills, as the bioaccumulated metals may produce toxic effects
422 because of pollution on soil (marine and coastal sediments), water, and food
423 chain. Considering experimental data and previous considerations, Figure 3
424 shows a representation of the hypothetical role of the *C. nodosa* seagrass in the
425 mobilisation of trace elements in the marine environment of the Mar Menor
426 Lagoon. Trace elements accumulated in the marine sediments (1) are partially

427 dissolved in water and partially mobilised by the seagrass, which accumulates
428 some of these elements in roots (Fe and As) (2) and others in leaves (Cu, Ni,
429 Pb, Zn and Mn) (3). When seagrass is dragged to the shore by the marine
430 currents (4), it begins to decompose; consequently, elements present in the
431 beach cast litter are reaccumulated in the coastal sediments, and they could be
432 available for the coastal fauna (5). Experimental data in this work show that
433 metal mobilisation by *C. nodosa* beach cast litter occurs in both the El Beal and
434 San Pedro localities, but this effect is more significant for the area directly
435 affected by the mining wastes. Although metal concentration in the *C. nodosa*
436 beach cast litter is much lower than that in the coastal sediments and this does
437 not suppose an important impact on the coastal sediments from the mouth of El
438 Beal Wadi, it is an important variable to consider, as it could be a source of
439 contamination if it is deposited in unpolluted surrounding areas.

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441 **5. Conclusion**

442 The results of the study show that, as expected, El Beal Wadi, which has been
443 affected by mining waste, presents higher concentrations of trace elements than
444 the unaffected zone. Moreover, results obtained also show that, irrespective of
445 the locality considered, the samples from the coastal zones present higher
446 concentrations than samples from the marine ones. This fact could be a
447 consequence of the interaction between different marine environmental
448 compartments, as a portion of the trace elements accumulated in the marine
449 sediments could be dissolved in the water and another portion could be
450 mobilised by the *C. nodosa*, which could keep them retained in leaves.

451 Consequently, when the detached leaves of this seagrass are accumulated in
452 the shore, their interaction with the coastal fauna and the meteorological
453 conditions could promote the mobilisation of the trace elements retained in the
454 leaves to the coastal sediments. Thus, it is essential to consider the possibility
455 of removing the seagrass beach cast litter to minimise the impact of elevated
456 trace element concentration levels in surrounding unpolluted zones.

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460 support in field sampling.

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474 **Table 1.** ICP-AES operating conditions, emission lines (nm) and detection limits
 475 for each element (mg kg⁻¹ dry weight). (I)/ (II): Line type, atomic and ionic,
 476 respectively.

		ICP-AES	
Plasma forward power/ W		1400	
Argon flow rate/ L min⁻¹			
Plasma		15	
Auxiliary		1.5	
Nebulizer (Q_g)		0.7	
Sample uptake rate (Q_L)/ mL min⁻¹		1.5	
Sample introduction			
Nebulizer		Meinhard, type K	
Spray chamber		Standard cyclonic	
View mode		Axial	
Integration time /s		1	
Readings/replicates		3	
Element	Emission line*/nm	Method detection limits/ mg kg⁻¹	
		Sediment	Live <i>C. nodosa</i>
As	193.696 (I)	20	10
Cd	226.502 (II)	5	2
Cr	357.868 (I)	1	0.5
Cu	327.395 (I)	2	1
Fe	238.204 (II)	20	10
Mn	257.610 (II)	8	4
Ni	216.555 (II)	6	3
Pb	220.353 (II)	0.8	0.4
Zn	202.548 (II)	6	3
Y	324.228 (II)	-	-
Y	410.237 (I)	-	-

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* (I)/ (II): Line type, atomic and ionic, respectively.

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Table 2. Elemental concentrations in samples mg kg⁻¹ dry weight; Mean ± confidence interval; n=18; 95% confidence level). LoD: Limit of Detection.

Element	Sediment				<i>C. nodosa</i>			
	Coastal		Marine		Coastal		Marine	
	Beal	San Pedro	Beal	San Pedro	Beal	San Pedro	Beal	San Pedro
Fe	141000 ± 2000	2800 ± 180	104000 ± 12000	1610 ± 80	3000 ± 200	570 ± 7	2600 ± 1400	250 ± 80
Pb	8700 ± 70	26 ± 2	2530 ± 30	27 ± 3	4350 ± 14	264 ± 4	810 ± 60	8 ± 3
Zn	8300 ± 800	16 ± 6	3700 ± 300	9 ± 3	1030 ± 8	322.6 ± 0.9	279.3 ± 1.2	44 ± 9
Mn	2450 ± 70	214 ± 3	3500 ± 40	210 ± 30	3160 ± 70	1040 ± 50	447 ± 9	100 ± 30
As	485 ± 19	< LoD	203 ± 7	< LoD	16.4 ± 0.8	< LoD	19 ± 9	< LoD
Cu	155 ± 4	7.0 ± 0.8	25.2 ± 0.7	4.88 ± 0.13	43.0 ± 0.4	20.26 ± 0.03	12 ± 2	8.0 ± 0.4
Cr	30.3 ± 0.5	3.2 ± 0.3	22.8 ± 0.4	3.45 ± 0.03	< LoD	< LoD	< LoD	< LoD
Ni	29.25 ± 0.11	< LoD	21.7 ± 1.4	< LoD	6.77 ± 0.08	3.758 ± 0.003	< LoD	< LoD
Cd	19 ± 3	< LoD	8.6 ± 1.1	< LoD	< LoD	< LoD	< LoD	< LoD

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Table 3. Elemental concentration published for the different samples from the Beal wadi (mg kg⁻¹ dry weight).

Elements	Sediments		<i>C. nodosa</i>
	Coastal	Marine	
Pb	7100 ± 3900 ^a	950 ^c	126 ^e
	4990 ± 261 ^b	4673 ± 310 ^d	300 ^f
Zn	-	-	551 ± 3 ^d
	8700 ± 8300 ^a	650 ^c	78 ^e
As	5120 ± 263 ^b	3858 ± 870 ^d	350 ^f
	-	-	269 ± 2 ^d
Cu	312 ± 154 ^a	-	-
	310 ± 24 ^b	-	-
Cd	-	-	-
	107 ± 49 ^a	35 ^c	-
Cd	101 ± 7 ^b	59.0 ± 0.1 ^d	-
	-	-	16.7 ± 0.6 ^d
Cd	22 ± 23 ^a	15 ^c	0.2 ^e
	13 ± 1 ^b	16.0 ± 0.3 ^d	0.2 ^f
	-	-	0.2 ± 0.3 ^d

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a: María-Cervantes, 2009; b: Conesa et al., 2011; c: García-García, 2004; d: Marín-Guirao et al., 2008; e: Marín-Guirao et al.,

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2005c; f: Marín-Guirao et al., 2005b.

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Table 4. Concentration values established for soils by different European countries (mg kg⁻¹).

Element	Europe				Spain	
	Netherland ^a	Denmark ^a	Turkey ^b		Reference levels in Mar Menor lagoon ^c	Background levels in Mar Menor lagoon ^c
			Contaminated	Extrem. contaminated		
Fe	-	-	-	-	-	-
Pb	> 530	> 400	150	600	77	44
Zn	> 720	> 1000	500	3000	51	39
Mn	-	-	-	-	-	-
As	> 55	> 20	-	-	-	-
Cu	> 190	> 500	100	500	49	30
Cr	-	-	-	-	-	-
Ni	-	-	-	-	-	-
Cd	> 12	> 5	5	20	0.2	0.2

a: María-Cervantes, 2009; b: García-García, 2004; c: Faz *et al.*, 2009.

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500 **Table 5.** Summary of three-way multifactorial variance analysis (ANOVA) of different elements concentration value on sediments
 501 and *C. nodosa* samples. Factor Locality (L) was fixed, factor Zone (Z) was fixed and Factor Site (Si) was random within the
 502 interaction Locality x Zone. Df: degrees of freedom; MS: Mean Square; F: value of F statistic. ns: non significant; *Significant at
 503 $p < 0.05$; ** $p < 0.01$; *** $p < 0.001$; ns: no significant differences; NT: no transformation performed.

Source of variation		Fe sediment				Fe <i>Cymodocea nodosa</i>		
df	MS	F	P	F versus	MS	F	P	
L	1	3242346.81	180.59	0.000***	Si (LxZ)	3347482.81	5.41	0.081 ns
Z	1	3607681.60	4.50	0.101 ns	Si (LxZ)	842003.60	0.14	0.73 ns
Si (LxZ)	4	3486049.14	27.88	0.000***	Res	6164513.44	78.94	0.000***
LxZ	1	9848886.73	3.95	0.118 ns	Si (LxZ)	16222.31	0	0.962 ns
Res	16	7340955.37				78089.69		
Tot	23							
Cochran's Test		C = 0.4344	n.s.			C = 0.7287	< 0.01	
Transformation		none				none		

Source of variation		Pb sediment				Pb <i>Cymodocea nodosa</i>		
df	MS	F	P	F versus	MS	F	P	
L	1	158.98	430.29	0.000***	Si (LxZ)	5863661	238.53	0.000***
Z	1	1.98	42.68	0.003**	Si (LxZ)	1605811	951.03	0.000***
Si (LxZ)	4	0.05	2.49	0.085 ns	Res	11074	0.14	0.966 ns
LxZ	1	2.53	54.6	0.002**	Si (LxZ)	6170331	460.20	0.000***
Res	16	0.02				80740		
Tot	23							
Cochran's Test		C = 0.4719	n.s.			C = 0.642	< 0.01	

Transformation		Ln (x+1)				none		
Source of variation		Zn sediment				Zn <i>Cymodocea nodosa</i>		
df	MS	F	P	F versus	MS	F	P	
L	1	9212523	32.38	0.005**	Si (LxZ)	1333214	568.43	0.000***
Z	1	2631272	15.47	0.017*	Si (LxZ)	1589024	3061.30	0.000***
Si (LxZ)	4	2755236	57.34	0.000***	Res	219	0.09	0.986 ns
LxZ	1	2397626	15.39	0.017 ns	Si (LxZ)	333291	642.08	0.000***
Res	16	48053			6089			
Tot	23							
Cochran's Test		C = 0.3960	n.s.		C = 0.642 < 0.01			
Transformation		none				none		

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Source of variation		Mn sediment				Mn <i>Cymodocea nodosa</i>		
df	MS	F	P	F versus	MS	F	P	
L	1	5681407	860.17	0.000***	Si (LxZ)	9192968	303.59	0.000***
Z	1	1650748	67.20	0.001***	Si (LxZ)	20012975	660.90	0.000***
Si (LxZ)	4	24557	0.42	0.792 ns	Res	30280	0.44	0.776 ns
LxZ	1	1668924	67.96	0.001***	Si (LxZ)	4757789	157.12	0.000***
Res	16	58489			68466			
Tot	23							
Cochran's Test		C = 0.4634	n.s.		C = 0,6363 < 0.01			
Transformation		none				none		

Source of variation		Cu sediment				Cu <i>Cymodocea nodosa</i>		
df	MS	F	P	F versus	MS	F	P	
L	1	30.06	390.36	0.000***	Si (LxZ)	1041.79	57.89	0.002*

Z	1	6.43	190.43	0.000***	Si (LxZ)	2851.64	158.46	0.000***
Si (LxZ)	4	0.03	2.36	0.097 ns	Res	18	3.75	0.025 ns
LxZ	1	3.34	99.09	0.001***	Si (LxZ)	550.8	30.61	0.005 ns
Res	16	0.01				4.8		
Tot	23							
Cochran's Test		C = 0.5103	n.s.			C = 0.7287	< 0.01	
Transformation		Ln (x+1)				none		

Source of variation		Cr sediment			
df		MS	F	P	F versus
L	1	3222.24	2.19	0.000***	Si (LxZ)
Z	1	80.21	48.84	0.002**	Si (LxZ)
Si (LxZ)	4	1.61	0.42	0.7889 ns	Res
LxZ	1	89.99	55.92	0.001***	Si (LxZ)
Res	16	3.79			
Tot	23				
Cochran's Test		C = 0.4455			
Transformation		none	n. s.		

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507 **Figure captions**

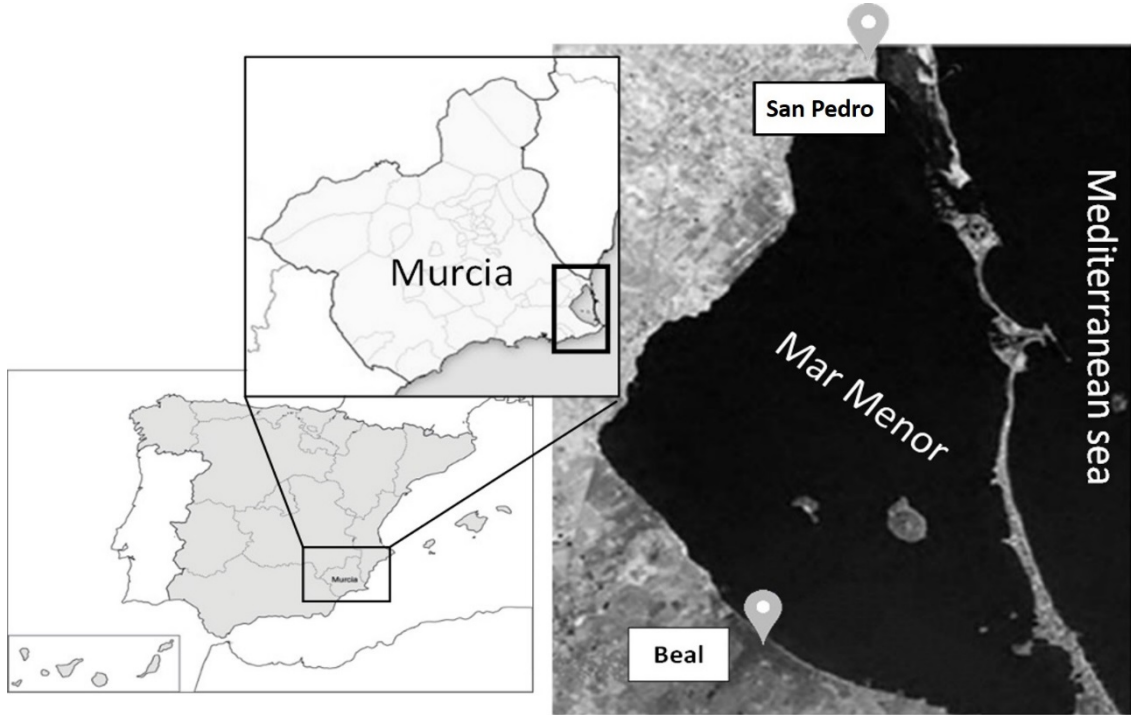
508 **Figure 1.** Location of the studied area with the Sierra minera of Cartagena – La Unión, Mar
509 Menor coastal lagoon and the two sampling localities (Beal: mouth of the Beal wadi and San
510 Pedro: San Pedro del Pinatar).

511 **Figure 2.** Trace elements concentrations in *Cymodocea nodosa* (A) and sediments (B) samples
512 from the two locations (El Beal (B) and San Pedro (SP)) and for marine (Grey) and coastal
513 (White) samples. Elemental concentrations in mg kg^{-1} dry weight versus the location.

514 **Figure 3.** Hypothetical scheme of *Cymodocea nodosa* role on the dynamic of trace elements in
515 the Mar Menor lagoon.

516

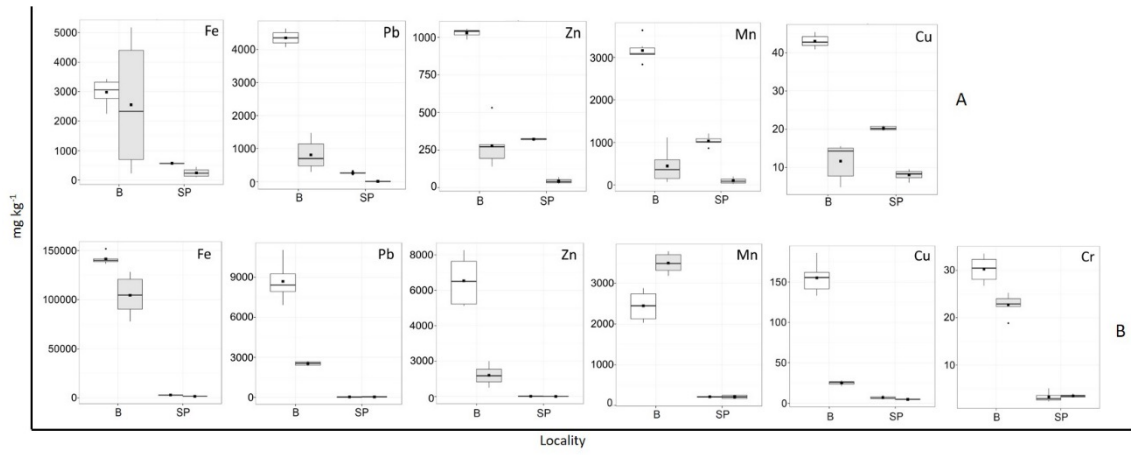
517 Figure 1



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520 Figure 2

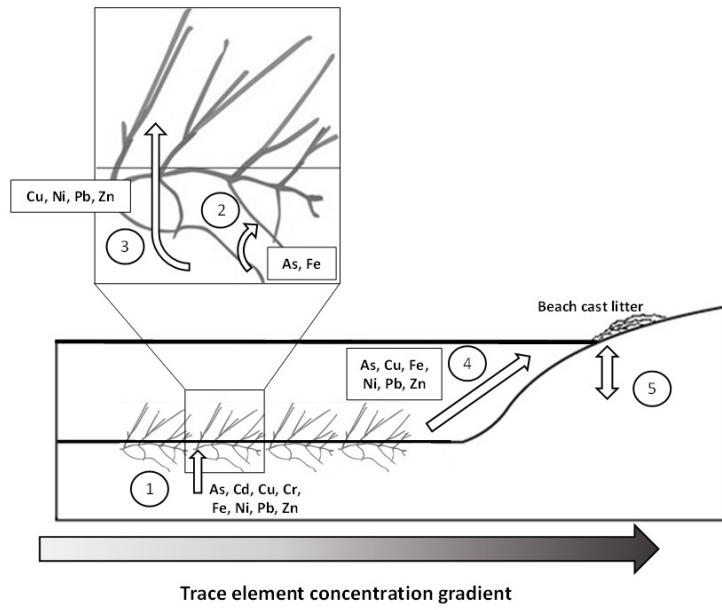


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523 Figure 3

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527 **References**

- 528 Anschutz, P., Dedieu, K., Desmazes, F., Chaillou, G., 2005. Speciation,
529 oxidation state, and reactivity of particulate manganese in marine
530 sediments. *Chem. Geol.* 218, 265–279.
531 <https://doi.org/10.1016/j.chemgeo.2005.01.008>
- 532 Brix, H., Lyngby, J.E., 1982. The distribution of cadmium, copper, lead, and zinc
533 in eelgrass (*Zostera marina* L.). *Sci. Total Environ.* 24, 51–63.
534 [https://doi.org/10.1016/0048-9697\(82\)90057-2](https://doi.org/10.1016/0048-9697(82)90057-2)
- 535 Chapman, B.M., Jones, D.R., Jung, R.F., 1983. Processes controlling metal ion
536 attenuation in acid mine drainage streams. *Geochim. Cosmochim. Acta* 47,
537 1957–1973. [https://doi.org/10.1016/0016-7037\(83\)90213-2](https://doi.org/10.1016/0016-7037(83)90213-2)
- 538 Conesa, H.M., Jiménez-Cárceles, F.J., 2007. The Mar Menor lagoon (SE
539 Spain): A singular natural ecosystem threatened by human activities. *Mar.*
540 *Pollut. Bull.* 54, 839–849. <https://doi.org/10.1016/j.marpolbul.2007.05.007>
- 541 Conesa, H.M., María-Cervantes, A., Álvarez-Rogel, J., González-Alcaraz, M.N.,
542 2011. Influence of soil properties on trace element availability and plant
543 accumulation in a Mediterranean salt marsh polluted by mining wastes:
544 Implications for phytomanagement. *Sci. Total Environ.* 409, 4470–4479.
545 <https://doi.org/10.1016/j.scitotenv.2011.07.049>
- 546 Da Cruz, H., Consejería de Agricultura, A. y M.A., 2003. Programa de gestión
547 integrada del litoral del Mar Menor y su zona de influencia.
548 <http://servicios.laverdad.es/servicios/textos/EstudiodeViabilidad.pdf> (Last
549 access July 2018).

550 Dassenakis, M., Garcia, G., Diamantopoulou, E., Girona, J.D., Garcia-Marin,
551 E.M., Filippi, G., Fioraki, V., 2010. The impact of mining activities on the
552 hypersaline Mar Menor lagoon. *Desalin. Water Treat.* 13, 282–289.
553 <https://doi.org/10.5004/dwt.2010.1036>

554 Delfino, J.J., Lee, G.F., 1968. Chemistry of Manganese in Lake Mendota,
555 Wisconsin. *Environ. Sci. Technol.* 2, 1094–1100.
556 <https://doi.org/10.1021/es60023a004>

557 Di Leonardo, R., Mazzola, A., Cundy, A.B., Tramati, C.D., Vizzini, S., 2017.
558 Trace element storage capacity of sediments in dead *Posidonia oceanica*
559 mat from a chronically contaminated marine ecosystem. *Environ. Toxicol.*
560 *Chem.* 36, 49–58. <https://doi.org/10.1002/etc.3539>

561 European Parliament, 2008. Directiva 2008/98/CE of the European Parliament
562 and of the council of 19 November 2008 on waste.

563 Faz, A., Martínez, S., Acosta, J.A., 2009. Aportaciones a los niveles de fondo y
564 referencia de metales pesados en suelos naturales de la Región de
565 Murcia.
566 [http://www.murcianatural.carm.es/c/document_library/get_file?uuid=aab20](http://www.murcianatural.carm.es/c/document_library/get_file?uuid=aab20a9f-a490-46ac-84a6-e07ae8da3a45&groupId=14)
567 [a9f-a490-46ac-84a6-e07ae8da3a45&groupId=14](http://www.murcianatural.carm.es/c/document_library/get_file?uuid=aab20a9f-a490-46ac-84a6-e07ae8da3a45&groupId=14) (Last access July 2018).

568 García-García, C., 2004. Impacto y riesgo ambiental de los residuos minero-
569 metalúrgicos de la Sierra de Cartagena-La Unión (Murcia-España). Ph.
570 Thesis, Universidad Politécnica de Cartagena (Murcia, Spain).

571 García-Lorenzo, M.L., Pérez-Sirvent, C., Martínez-Sánchez, M.J., Molina-Ruiz,
572 J., 2012. Trace elements contamination in an abandoned mining site in a
573 semiarid zone. *J. Geochemical Explor.* 113, 23–35.

574 <https://doi.org/10.1016/j.gexplo.2011.07.001>

575 García-Lorenzo, M.L., Pérez-Sirvent, C., Molina-Ruiz, J., Martínez-Sánchez,
576 M.J., 2014. Mobility indices for the assessment of metal contamination in
577 soils affected by old mining activities. *J. Geochemical Explor.* 147, 117–
578 129. <https://doi.org/10.1016/j.gexplo.2014.06.012>

579 García, G., Manteca, J.I., Peñas, J.M., 2007. Leaching and transport of Zn
580 through soil profiles in a seasonal river of a mining area in SE Spain. *Glob.*
581 *NEST J.* 9, 214–223.

582 García, G., Muñoz-Vera, A., 2015. Characterization and evolution of the
583 sediments of a Mediterranean coastal lagoon located next to a former
584 mining area. *Mar. Pollut. Bull.* 100, 249–263.
585 <https://doi.org/10.1016/j.marpolbul.2015.08.042>

586 González-Alcaraz, M.N., Conesa, H.M., Álvarez-Rogel, J., 2013. When liming
587 and revegetation contribute to the mobilisation of metals: Learning lessons
588 for the phytomanagement of metal-polluted wetlands. *J. Environ. Manage.*
589 116, 72–80. <https://doi.org/10.1016/j.jenvman.2012.11.044>

590 Gonzalez-Fernandez, O., Jurado-Roldan, A.M., Queralt, I., 2011. Geochemical
591 and mineralogical features of overbank and stream sediments of the Beal
592 Wadi (Cartagena-La union mining district, SE Spain): Relation to former
593 lead-zinc mining activities and its environmental risk. *Water. Air. Soil Pollut.*
594 215, 55–65. <https://doi.org/10.1007/s11270-010-0458-1>

595 Grygo-Szymanko, E., Tobiasz, A., Walas, S., 2015. Speciation analysis and
596 fractionation of manganese – a review. *TrAC Trends Anal. Chem.* 80, 112–
597 114. <https://doi.org/10.1016/j.trac.2015.09.010>

598 Gutiérrez, A., Giménez, M., 2009. La ordenación, planificación y gestión del
599 litoral en la Región de Murcia, in: García, M. (Coord.), Sanz, J. (Dir.),
600 Estudios Sobre La Ordenación, Planificación y Gestión Del Litoral: Hacia
601 Un Modelo Integrado y Sostenible, pp. 299–312.

602 Hierro, A., Olías, M., Cánovas, C.R., Martín, J.E., Bolivar, J.P., 2014. Trace
603 metal partitioning over a tidal cycle in an estuary affected by acid mine
604 drainage (Tinto estuary, SW Spain). *Sci. Total Environ.* 497–498, 18–28.
605 <https://doi.org/10.1016/j.scitotenv.2014.07.070>

606 Hosokawa, S., Konuma, S., Nakamura, Y., 2016. Accumulation of trace metal
607 elements (Cu, Zn, Cd and Pb) in surface sediment via decomposed
608 seagrass leaves : A mesocosm experiment using *Zostera marina* L . *PLoS*
609 *One* 11, 1–18. <https://doi.org/10.1371/journal.pone.0157983>

610 Ivaldi, J.C., Tyson, J.F., 1996. Real-time internal standardization with an axially-
611 viewed inductively coupled plasma for optical emission spectrometry.
612 *Spectrochim. Acta - Part B At. Spectrosc.* 51, 1443–1450.

613 Jara-Marini, M.E., García-Camarena, R., Gómez-Álvarez, A., García-Rico, L.,
614 2015. Fractionation and risk assessment of Fe and Mn in surface
615 sediments from coastal sites of Sonora, Mexico (Gulf of California).
616 *Environ. Monit. Assess.* 187:468, 1–12. [https://doi.org/10.1007/s10661-](https://doi.org/10.1007/s10661-015-4683-3)
617 [015-4683-3](https://doi.org/10.1007/s10661-015-4683-3)

618 Jefe de Estado S. R. M. Juan Carlos I, 1989. Ley 4/1989, de marzo, de
619 Conservación de los Espacios naturales y de la Flora y Fauna silvestres.

620 Llagostera, I., Pérez, M., Romero, J., 2011. Trace metal content in the seagrass
621 *Cymodocea nodosa*: Differential accumulation in plant organs. *Aquat. Bot.*

622 95, 124–128. <https://doi.org/10.1016/j.aquabot.2011.04.005>

623 Lyngby, J.E., Brix, H., 1989. Heavy metals in eelgrass (*Zostera marina* L .)
624 during growth and decomposition. *Hydrobiologia* 176/177, 189–196.
625 <https://doi.org/10.1007/BF00026554>

626 Malea, P., Haritonidis, S., 1999. *Cymodocea nodosa* (Ucria) Aschers. as a
627 Bioindicator of Metals in Thermaikos Gulf, Greece, during Monthly
628 Samplings. *Bot. Mar.* 42, 419–430. <https://doi.org/10.1515/BOT.1999.048>

629 Malea, P., Kevrekidis, T., 2013. Trace element (Al, As, B, Ba, Cr, Mo, Ni, Se,
630 Sr, Tl, U and V) distribution and seasonality in compartments of the
631 seagrass *Cymodocea nodosa*. *Sci. Total Environ.* 463–464, 611–623.
632 <https://doi.org/10.1016/j.scitotenv.2013.06.074>

633 María-Cervantes, A., 2009. Aproximación a los riesgos derivados de la
634 presencia de residuos mineros en los saladares del entorno del Mar
635 Menor: Dinámica de metales pesados y de arsénico y su acumulación en
636 plantas y moluscos. Ph. Thesis, Universidad Politécnica de Cartagena
637 (Murcia, Spain).

638 María-Cervantes, A., Jiménez-Cárceles, F.J., Álvarez-Rogel, J., 2009. As, Cd,
639 Cu, Mn, Pb, and Zn contents in sediments and mollusks (*hexaplex*
640 *trunculus* and *tapes decussatus*) from coastal zones of a mediterranean
641 lagoon (mar menor, SE Spain) affected by mining wastes. *Water. Air. Soil*
642 *Pollut.* 200, 289–304. <https://doi.org/10.1007/s11270-008-9913-7>

643 Marín-Guirao, L., Atucha, A.M., Barba, J.L., López, E.M., García Fernández,
644 A.J., 2005a. Effects of mining wastes on a seagrass ecosystem: Metal
645 accumulation and bioavailability, seagrass dynamics and associated

646 community structure. *Mar. Environ. Res.* 60, 317–337.
647 <https://doi.org/10.1016/j.marenvres.2004.11.002>

648 Marín-Guirao, L., Cesar, A., Marín, A., Lloret, J., Vita, R., 2005b. Establishing
649 the ecological quality status of soft-bottom mining-impacted coastal water
650 bodies in the scope of the Water Framework Directive. *Mar. Pollut. Bull.* 50,
651 374–387. <https://doi.org/10.1016/j.marpolbul.2004.11.019>

652 Marín-Guirao, L., Cesar, A., Marín, A., Vita, R., 2005c. Assessment of sediment
653 metal contamination in the Mar Menor coastal lagoon (SE Spain): Metal
654 distribution, toxicity, bioaccumulation and benthic community structure.
655 *Ciencias Mar.* 31, 413–428.

656 Marín-Guirao, L., Lloret, J., Marín, A., 2008. Carbon and nitrogen stable
657 isotopes and metal concentration in food webs from a mining-impacted
658 coastal lagoon. *Sci. Total Environ.* 393, 118–130.
659 <https://doi.org/10.1016/j.scitotenv.2007.12.023>

660 Marín-Guirao, L., Lloret, J., Marín, A., García, G., García Fernández, A.J., 2007.
661 Pulse-discharges of mining wastes into a coastal lagoon: Water chemistry
662 and toxicity. *Chem. Ecol.* 23, 217–231.
663 <https://doi.org/10.1080/02757540701339422>

664 Miller, J.N., Miller, J.C., 2002. *Statistics and chemometrics for analytical*
665 *chemistry*, 4th ed. Prentice Hall Pearson Education Limited Harlow,
666 England.

667 Mountouris, A., Voutsas, E., Tassios, D., 2002. Bioconcentration of heavy metals
668 in aquatic environments: the importance of bioavailability. *Mar. Pollut. Bull.*
669 44, 1136–1141.

- 670 Muñoz-Vera, A., García, G., García-Sánchez, A., 2015. Metal bioaccumulation
671 pattern by *Cotylorhiza tuberculata* (Cnidaria, Scyphozoa) in the Mar Menor
672 coastal lagoon (SE Spain). *Environ. Sci. Pollut. Res.* 22, 19157–19169.
673 <https://doi.org/10.1007/s11356-015-5119-x>
- 674 Muñoz-Vera, A., Peñas Castejón, J.M., García, G., 2016. Patterns of trace
675 element bioaccumulation in jellyfish *Rhizostoma pulmo* (Cnidaria,
676 Scyphozoa) in a Mediterranean coastal lagoon from SE Spain. *Mar. Pollut.*
677 *Bull.* 110, 143–154. <https://doi.org/10.1016/j.marpolbul.2016.06.069>
- 678 Navarro, M.C., Pérez-Sirvent, C., Martínez-Sánchez, M.J., Vidal, J., Tovar, P.J.,
679 Bech, J., 2008. Abandoned mine sites as a source of contamination by
680 heavy metals: A case study in a semi-arid zone. *J. Geochemical Explor.* 96,
681 183–193. <https://doi.org/10.1016/j.gexplo.2007.04.011>
- 682 Ochieng, C.A., Erftemeijer, P.L.A., 1999. Accumulation of seagrass beach cast
683 along the Kenyan coast : a quantitative assessment. *Aquat. Bot.* 65, 221–
684 238. [https://doi.org/https://doi.org/10.1016/S0304-3770\(99\)00042-X](https://doi.org/https://doi.org/10.1016/S0304-3770(99)00042-X)
- 685 Pérez-Ruzafa, Á., Fernandez, A.I., Marcos, C., Gilabert, J., Quispe, J.I., García-
686 Charton, J.A., 2005. Spatial and temporal variations of hydrological
687 conditions, nutrients and chlorophyll in a Mediterranean coastal lagoon
688 (Mar Menor, Spain). *Hydrobiologia* 550, 11–27.
689 <https://doi.org/10.1007/s10750-005-4356-2>
- 690 Pérez-Ruzafa, Á., Marcos, C., Pérez-Ruzafa, I.M., Ros, J.D., 1987. Evolución
691 de las características ambientales y de los poblamientos del mar menor.
692 *Anales De Biología*, 12, 53-65.
693 <http://revistas.um.es/analesbio/article/view/35381>

- 694 Sánchez, A., Ballester, A., Blázquez, M.L., González, F., Muñoz, J., Hammami,
695 A., 1999. Biosorption of copper and zinc by *Cymodocea nodosa*. *FEMS*
696 *Microbiol. Rev.* 23, 527–536. [https://doi.org/10.1016/S0168-](https://doi.org/10.1016/S0168-6445(99)00019-4)
697 6445(99)00019-4
- 698 Sanchiz, C., García-Carrascosa, A.M., Pastor, A., 2000. Heavy metal contents
699 in soft-bottom marine macrophytes and sediments along the Mediterranean
700 coast of Spain. *Mar. Ecol.* 21, 1–16. [https://doi.org/10.1046/j.1439-](https://doi.org/10.1046/j.1439-0485.2000.00642.x)
701 0485.2000.00642.x
- 702 Shirdam, R., Modarres-Tehrani, Z., Dastgoshadeh, F., 2008. Microwave
703 assisted digestion of soil, sludge and sediment for determination of heavy
704 metals with ICP-OES and FAAS. *Rasayan J. Chem.* 1, 757–765.
- 705 Simonneau, J., 1973. Évolution sédimentologique et géochimique récente du
706 remplissage. Université Paul Sabatier (Toulouse).
- 707 Sokal, R.R., Rohlf, F.J., 1981. *Biometry: The principles and practice of*
708 *statistics in biological research*, 2nd ed. W. H. Freedman and Co. New
709 York, E. E. U. U.
- 710 Todolí, J.-L., Mermet, J.-M., 1999. Acid interferences in atomic spectrometry:
711 analyte signal effects and subsequent reduction. *Spectrochim. Acta Part B*
712 *At. Spectrosc.* 54, 895–929.
- 713 Todolí, J.L., Gras, L., Hernandis, V., Mora, J., 2002. Elemental matrix effects in
714 ICP-AES. *J. Anal. At. Spectrom.* 17, 142–169.
715 <https://doi.org/10.1039/b009570m>
- 716 Tsakovski, S., Kudlak, B., Simeonov, V., Wolska, L., Garcia, G., Dassenakis,

717 M., Namieśnik, J., 2009. N-way modelling of sediment monitoring data from
718 Mar Menor lagoon, Spain. *Talanta* 80, 935–941.
719 <https://doi.org/10.1016/j.talanta.2009.08.015>

720 Tsakovski, S., Kudłak, B., Simeonov, V., Wolska, L., Garcia, G., Namieśnik, J.,
721 2012. Relationship between heavy metal distribution in sediment samples
722 and their ecotoxicity by the use of the Hasse diagram technique. *Anal.*
723 *Chim. Acta* 719, 16–23. <https://doi.org/10.1016/j.aca.2011.12.052>

724 Umgiesser, G., Ferrarin, C., Cucco, A., De Pascalis, F., Bellafiore, D., Ghezzi,
725 M., Bajo, M., 2014. Comparative hydrodynamics of 10 Mediterranean
726 lagoons by means of numerical modeling. *J. Geophys. Res. Ocean.* 119,
727 2212–2226.

728 Underwood, A.J., 1997. *Experiments in Ecology. Their logical design and*
729 *interpretation using analysis of variance*, in: *Experimental Design and Data*
730 *Analysis for Biologists*. Cambridge University Press, Cambridge.

731 Underwood, A.J., 1994. On beyond BACI: sampling designs that might reliably
732 detect environmental disturbances. *Ecol. Appl.* 4, 3–15.
733 <https://doi.org/10.2307/1942110>

734 Underwood, A.J., 1993. The mechanics of spatially replicated sampling
735 programmes to detect environmental impacts in a variable world. *Austral*
736 *Ecol.* 18, 99–116. <https://doi.org/10.1111/j.1442-9993.1993.tb00437.x>

737 Zaaboub, N., Oueslati, W., Helali, M.A., Abdeljaouad, S., Huertas, F.J., Lopez
738 Galindo, A., 2014. Trace elements in different marine sediment fractions of
739 the gulf of tunis (central mediterranean sea). *Chem. Speciat. Bioavailab.*
740 26, 1–12. <https://doi.org/10.3184/095422914X13884279095945>

741 Zachariadis, G.A., Vogiatzis, C., 2010. An Overview of the Use of Yttrium for
742 Internal Standardization in Inductively Coupled Plasma–Atomic Emission
743 Spectrometry. *Appl. Spectrosc. Rev.* 45, 220–239.
744 <https://doi.org/10.1080/05704921003719122>
745