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## Isotopic methods for non-destructive assessment of carbon dynamics in shrublands under long-term climate change manipulation

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1 **Isotopic methods for non-destructive assessment of carbon dynamics in shrublands**  
2 **under long-term climate change manipulation**

3

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7 **Running head:** Isotope techniques for ecosystem C science

8 ***Tweet:** The pros and cons of carbon assessment methods using isotopes across climate change*  
9 *experiments in shrublands.*

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30

31 **Summary**

32

33 1. Long-term climate change experiments are extremely valuable for studying ecosystem  
34 responses to environmental change. Examination of the vegetation and the soil should be non-  
35 destructive to guarantee long-term research. In this paper, we review novel field methods using  
36 isotope techniques for assessing carbon dynamics in the plant-soil-air continuum, based on  
37 recent field experience and examples from a European climate change manipulation network.

38 2. Eight European semi-natural shrubland ecosystems were exposed to controlled warming and  
39 drought manipulations. One field site was additionally exposed to elevated atmospheric CO<sub>2</sub>.  
40 We evaluate the isotope methods that were used across the network to evaluate carbon fluxes  
41 and ecosystem responses: 1) analysis of the naturally rare isotopes of carbon (<sup>13</sup>C and <sup>14</sup>C) and  
42 nitrogen (<sup>15</sup>N); 2) tracing changes in isotopic signatures in ecosystem compartments *in-situ*, by  
43 using pulse labelling with <sup>13</sup>CO<sub>2</sub>, soil injections of <sup>13</sup>C- and <sup>15</sup>N-enriched substrates, or  
44 continuous labelling with <sup>13</sup>C-depleted CO<sub>2</sub> by Free Air Carbon dioxide Enrichment (FACE);  
45 and 3) manipulation and tracing the isotopic composition of soil substrates (<sup>14</sup>C) in lab-based  
46 studies.

47 3. Questions related to long-term carbon turnover processes were investigated by natural <sup>14</sup>C  
48 signals, specifically <sup>14</sup>C signature of soil respiration gave insights into the decomposition of old  
49 soil carbon sources. Contrastingly, the stable isotopes <sup>13</sup>C and <sup>15</sup>N were used for shorter-term  
50 processes, as the residence time in a certain compartment of the stable isotope label signal is  
51 limited. <sup>13</sup>C-labelling techniques exert a minimal physical disturbance, however, the dilution of  
52 the applied isotopic signal can be challenging, and the contamination of the field site with  
53 released excess <sup>13</sup>C can be a problem for subsequent natural abundance (<sup>14</sup>C and <sup>13</sup>C) or label  
54 studies.

55 4. Based on the experience with the experimental work, we provide recommendations for the  
56 application of the reviewed methods to study carbon fluxes in the plant-soil-air continuum in  
57 long-term, large-scale climate change experiments.

58

59 **Key-words:** warming; drought; bomb-C; FACE; pulse-labelling; stable isotopes; <sup>14</sup>C

60	<b>Inhalt</b>	
61	Isotopic methods for non-destructive assessment of carbon dynamics in shrublands under long-term	
62	climate change manipulation.....	1
63	Summary .....	2
64	Introduction.....	3
65	Methodologies using natural abundance of carbon isotopes.....	6
66	1 Ecosystem processes reflected by stable isotope fractionation ( <sup>13</sup> C and <sup>15</sup> N) .....	6
67	2 Bomb- <sup>14</sup> C technique to assess sources of soil respiration .....	8
68	Methods using <i>in-situ</i> <sup>13</sup> C labelling to study rhizodeposition utilisation.....	10
69	1 <sup>13</sup> C-CO <sub>2</sub> pulse labelling .....	11
70	2 Free Air Carbon dioxide Enrichment (FACE)-labelling.....	13
71	3 In situ injection of <sup>13</sup> C-enriched substrate solutions .....	14
72	Use of labelled carbon-compounds to analyse carbon mineralisation by soil microorganisms.....	16
73	Conclusions and recommendations .....	17
74	Acknowledgements .....	18
75	References .....	18
76	Table.....	30
77	Box 1. Isotopic signal of plant leaf responses to precipitation .....	32
78	Box 2. Impact of warming and drought on the <sup>14</sup> C signature of soil respiration .....	34
79	Box 3. Analysing rhizodeposit utilisation by microbes in the field.....	36
80	Box 4. Exponential decay kinetics for <sup>14</sup> CO <sub>2</sub> evolution during microbial <sup>14</sup> C substrate mineralisation	
81	.....	38

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84

## 85 **Introduction**

86

87 Global climate change scenarios predict that increased greenhouse gas (e.g. CO<sub>2</sub>, CH<sub>4</sub> and  
88 N<sub>2</sub>O) concentrations in the atmosphere will alter the periodicity and magnitude of drought  
89 events and will increase mean global temperatures by approximately 0.2 °C per decade (IPCC  
90 2013). For the European continent this will manifest as drier summers in the South and  
91 increased precipitation in the North (IPCC 2013). Elucidating the consequences of such  
92 atmospheric changes for biogenic carbon fluxes is one of the main challenges for the scientific

93 community. Some models have predicted a positive feedback to climate change, resulting from  
94 higher increases in respiratory fluxes from ecosystems (e.g. carbon release through soil  
95 respiration) than in net primary productivity, which would lead to further increases in  
96 atmospheric CO<sub>2</sub> (Friedlingstein et al. 2006; Denman et al. 2007). To assess the likelihood of  
97 positive feedback, experimental studies that analyse the long-term adaptations of ecosystem  
98 carbon fluxes to climate change are critically needed. However, climate change experiments  
99 are often conducted at short or medium time scales due to funding constraints, or due to the  
100 limited life-span of the experimental plots, as repeated removal of samples often leads to  
101 disturbances and experimental artefacts in the studied system. Hence, there is a necessity for  
102 the maintenance of long-term experiments using non-destructive methods.

103 Carbon fluxes through the plant-soil-air continuum play a central role in soil carbon cycling.  
104 Drought and warming alter the quantity and composition of carbon inputs to the soil by  
105 changing plant available carbon and nitrogen sources (de Graaff et al. 2007; Leakey et al. 2009).  
106 Microbial growth in soil is generally constrained by available carbon, therefore qualitative and  
107 quantitative changes in rhizodeposition are likely to alter the activity of heterotrophic  
108 microorganisms and the rates of soil organic matter (SOM) mineralisation (Zak et al. 2000;  
109 Phillips et al. 2006). Consequently, aboveground to belowground fluxes might largely  
110 determine carbon emissions from ecosystems under the different climate change scenarios  
111 (Chapin et al. 2009).

112 Stable carbon isotope studies can give important insights into carbon fluxes through the plant-  
113 soil-air continuum with the minimal disturbance to the system. The isotopic carbon composition  
114 of compartments in this continuum is a result of the different isotope fractionation processes  
115 along the pathway from CO<sub>2</sub> fixation by plants to carbon allocation to soil (reviewed in  
116 Brüggemann et al. 2011). Thus, the analysis of the natural abundance of carbon isotopes in  
117 these compartments can give information about some processes related to photosynthesis and  
118 carbon losses through plant or soil respiration (Brüggemann et al. 2011). In addition, *in-situ*  
119 pulse labelling with the heavy stable carbon isotope (<sup>13</sup>C) is a powerful tool to analyse short-  
120 term dynamics of carbon allocation to the soil with high resolution (Högberg et al. 2008; Epron  
121 et al. 2012; Reinsch & Ambus 2013). The recent development of techniques for <sup>13</sup>C analyses in  
122 specific compounds such as phospholipid fatty acids (PLFAs), amino-sugars, RNA and DNA,  
123 constitutes a remarkable advance in the studying of carbon cycling. These analyses allow for  
124 the examination of rhizodeposit utilisation by microbes or trophic interactions between  
125 functional groups (Ostle et al. 2003; Jin & Evans 2010; Ruess & Chamberlain 2010). The

126 application of these isotopic methods can therefore provide unique information about  
127 aboveground-belowground linkages and their alterations in response to climate changes.

128 In order to investigate long-term effects of climate change on shrubland ecosystems, an  
129 experimental network was established across Europe (the INCREASE network). Studying the  
130 response of shrublands to climate change is important, since they are representative ecosystems  
131 in Mediterranean and North European countries, where they play an important ecological role  
132 in preserving biodiversity (Verdú 2000; Wessel et al. 2004). In addition, land area covered by  
133 shrublands has dramatically decreased in temperate Europe during the past century, due to land  
134 use changes, increased pollution and eutrophication, and climate change (Fagúndez 2013). In  
135 Mediterranean regions, however, shrublands have increased their extension due to land  
136 abandonment (Fagúndez 2013).

137 Within the climate change network, common non-destructive methods were used across sites  
138 to ensure the comparison of treatment effects across different climatic regions (Beier et al. 2004;  
139 Mikkelsen et al. 2008). Evaluating the impact of climate change treatments on shrubland carbon  
140 dynamics was one of the main objectives of this experimental network, and thus a range of  
141 methodologies to quantify and trace distinct carbon pools and their fluxes have been applied  
142 since 1999. Priority was given to those techniques that minimise disturbances to vegetation and  
143 soil to guarantee long-term research.

144 Here, we review isotope methods that have been applied across this climate change  
145 experimental network to study ecosystem carbon dynamics in the plant-soil-air continuum. In  
146 particular, we focus on methodologies that: 1) analyse the abundance of naturally rare isotopes  
147 of carbon ( $^{13}\text{C}$  and  $^{14}\text{C}$ ) and nitrogen ( $^{15}\text{N}$ ) in the different ecosystem compartments to evaluate  
148 their responses to climate change; and 2) trace experimentally-induced changes in the isotopic  
149 signatures to assess rhizodeposition utilisation by soil biota, by using either  $^{13}\text{CO}_2$  pulse  
150 labelling, continuous labelling with  $^{13}\text{C}$ -depleted  $\text{CO}_2$  from Free Air Carbon dioxide  
151 Enrichment (FACE), or injections of  $^{13}\text{C}$ - and  $^{15}\text{N}$ -enriched substrates in the field, and finally  
152 3) manipulate and trace the isotopic composition of C-compounds to analyse C mineralisation  
153 by soil microorganisms in laboratory studies. Along-side the methods, data from the field  
154 studies are presented as accompanying illustrative boxes, and practical recommendations for  
155 the applications of these methodologies at large-scale climate change experiments are outlined  
156 in Table 1. The combination of the isotope methods with methods for *in-situ* quantification of

157 aboveground, root and fungal mycelia biomass will increase our understanding of climate  
158 change effects on carbon dynamics with the least possible disturbance.

159

## 160 **The experimental climate change network INCREASE**

161 The experimental network for the study of climate change impacts on European shrublands  
162 (INCREASE, ‘An Integrated Network on Climate Research Activities on Shrubland  
163 Ecosystems’) was established in 1998. The network is comprised of eight shrublands situated  
164 across a natural temperature gradient of mean annual temperature from c. 8 °C in the North to  
165 c. 16 °C in the South, and a rainfall gradient ranging from 510 mm to 1741 mm from East to  
166 West (see Fig. 1 in Reinsch et al. 2017). These sites represent Continental, Atlantic and  
167 Mediterranean shrublands. At each site, whole-ecosystem warming and drought treatments  
168 were applied in triplicates of 20 m<sup>2</sup> plots, by using automated retractable curtain constructions  
169 (see Beier et al. 2004 and Mikkelsen et al. 2008 for a full description). At one of the Danish  
170 sites (DK-BRA), a FACE treatment was installed, and combinations of the climate treatments  
171 were established and resulted in a plot size of 9 m<sup>2</sup> (Mikkelsen et al. 2008). Climatic conditions  
172 at the plot level (air temperature, humidity, soil temperature and moisture) were recorded in  
173 half-hour or hourly intervals, and main carbon pools and fluxes have been periodically  
174 monitored. Most frequent measurements include aboveground plant biomass and composition  
175 (Kröel-Dulay et al. 2015), litter production, soil respiration and net ecosystem carbon exchange  
176 (Beier et al. 2008; 2009; Lellei-Kovács et al. 2016). Measurements of litter decomposition, soil  
177 nitrogen mineralisation (Emmett et al. 2004) and soil microbial biomass and activity (Sowerby  
178 et al. 2005) have also been conducted with different periodicity.

179

## 180 **Methodologies using natural abundance of carbon isotopes**

181

### 182 **1 Ecosystem processes reflected by stable isotope fractionation (<sup>13</sup>C and <sup>15</sup>N)**

183

184 The relative abundance of the rare and heavy stable isotopes of nitrogen (<sup>15</sup>N) and carbon (<sup>13</sup>C)  
185 compared to the most abundant stable isotope, <sup>14</sup>N and <sup>12</sup>C respectively, is a signature that  
186 reflects the isotopic discrimination associated with gain and loss processes of a given entity.  
187 These signatures are expressed as the delta (δ) notation (e.g. δ<sup>13</sup>C and δ<sup>15</sup>N in ‰), which is the  
188 deviation of the rare isotope abundance in the sample compared to a reference material (Brand

189 & Coplen 2014; Muccio & Jackson 2009). The naturally occurring background level is termed  
190 'natural abundance' of the given rare stable isotope (Berglund & Wieser 2011). Most natural  
191 processes (chemical, physical or enzymatically catalysed) discriminate against heavy isotopes  
192 (e.g.  $^{13}\text{C}$ ,  $^{15}\text{N}$ ,  $^{18}\text{O}$ ), which in open systems results in an isotopically depleted product with  
193 comparably smaller concentration of the heavy isotopes than its corresponding substrate  
194 (Robinson 2001; Fry 2006). If the dominant process rate changes, or if the substrate is  
195 exhausted, then the  $\delta$  value of the product (such as the plant leaf) may significantly change, due  
196 to the underlying fractionation. Delta notation is appropriate when dealing with natural samples  
197 that are not labelled with excess heavy isotopes (see next sections), and variations at the level  
198 of natural abundance are useful for evaluating natural discrimination processes. Importantly,  
199 natural abundance cannot be studied if 'overlain' by a study of labelling with heavy isotopes.

200 Decreases in soil water availability due to drought can alter the isotope signature of both carbon  
201 and nitrogen in the aboveground plant biomass. During drought stress, leaves reduce stomatal  
202 opening to preserve water. As this happens, the space that confines the air as an immediate  
203 source of  $\text{CO}_2$  for photosynthesis (the sub-stomatal cavity) becomes a more closed system due  
204 to the restriction of the renewal of  $\text{CO}_2$ , and as a result a higher proportion of the heavy  $^{13}\text{C}$  in  
205  $\text{CO}_2$  is fixed by Rubisco (C3 plants; Tcherkez et al. 2011). Hereby the discrimination against  
206 the heavy  $^{13}\text{C}$  isotope is decreased. As a consequence, in plants with a C3 photosynthetic  
207 pathway a  $^{13}\text{C}$  enrichment in the leaf occurs during drought stress (Cernusak et al. 2013).  
208 Indeed, the  $^{13}\text{C}$  enrichment at the leaf level is related to an increased intrinsic water use  
209 efficiency (WUEi), the ratio of assimilation to stomatal conductance (Farquhar & Richards  
210 1984; Donovan & Ehleringer 1994). Changes in soil water availability may also alter the leaf  
211 nitrogen isotope signature by changing the nitrogen availability with soil depth, and thereby the  
212  $^{15}\text{N}$  signature of the plant nitrogen source (Lloret et al. 2004). Since  $\delta^{15}\text{N}$  is often analysed at  
213 the same solid sample as  $\delta^{13}\text{C}$  by IRMS (Isotope Ratio Mass Spectrometer), interpretation of  
214  $\delta^{15}\text{N}$  can be a useful complement to understand the ecosystem processes. Nitrate and  
215 ammonium sources of different origin or at different soil depths can vary in  $\delta^{15}\text{N}$  signature (Xue  
216 et al. 2009). Hence, if a drought event changes the vertical nitrogen availability in the soil, the  
217 plant nitrogen source can shift to a different soil depth possibly causing a change in  $\delta^{15}\text{N}$   
218 signature in the leaves. In general, an increase in  $\delta^{15}\text{N}$  signature in the leaves indicates a  
219 progressive N saturation and/or N losses in the surrounding system because all major pathways  
220 of N loss (denitrification, ammonia volatilization and nitrate leaching) cause  $\delta^{15}\text{N}$  enrichment  
221 of the remaining nitrogen (Peñuelas et al 2000). Interpretation of changes in leaf  $\delta^{15}\text{N}$ , however,



222 is not straightforward since leaf  $\delta^{15}\text{N}$  signatures might largely depend on mycorrhizal  
223 associations and shifts in nitrogen sources between organic and inorganic compounds (the  
224 increase in plant  $\delta^{15}\text{N}$  values with aridity may also result from increasing reliance on recycled  
225 organic N sources as opposed to new inputs) under a drought or warming could influence the  
226 leaf  $\delta^{15}\text{N}$  as well (Michelsen et al. 1998; Pardo et al. 2006; Andresen et al. 2009).

227 Across the field sites, the effects of warming and drought on plant  $^{13}\text{C}$  and  $^{15}\text{N}$  natural  
228 abundance was monitored over four years, starting two years after onset of the climate  
229 manipulations. Current year shoots or leaves were analysed for  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  immediately after  
230 each artificially prolonged drought. Plant material was dried at  $70^\circ\text{C}$  and ground to a fine  
231 powder before analysis for natural abundance values of  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  using isotope ratio mass  
232 spectrometry (IRMS). We expected to find higher  $\delta^{13}\text{C}$  values: i) in drought treated plants  
233 (compared to control plots) and, ii) in plants growing at drier locations across the precipitation  
234 gradient (within a given plant species). Furthermore, we expected iii) the  $\delta^{15}\text{N}$  to change in  
235 response to drought, as the nitrogen source (depth) is changed (at one location, within-species).  
236 Some significant effects of the drought treatment were observed on plant tissue  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$   
237 (Box 1). Differences between years (effect of time) were more pronounced than the effect of  
238 the drought treatment for *Populus alba*  $\delta^{13}\text{C}$  (HU), *Erica multiflora*  $\delta^{15}\text{N}$  (SP) and *Globularium*  
239 *alypum*  $\delta^{15}\text{N}$  and  $\delta^{13}\text{C}$  (SP). Only *Calluna vulgaris* showed a significant response to the drought  
240 treatment for  $\delta^{13}\text{C}$  as hypothesized (Box 1A). For *C. vulgaris*, which was growing at several  
241 locations (UK-CL, NL and DK-MOLS), the  $\delta^{13}\text{C}$  was higher at drier locations, when compared  
242 along the precipitation gradient, and also higher in the drought treatment, at the NL or UK-CL  
243 sites (Box 1B). Finally, we found no response of leaf  $\delta^{15}\text{N}$  to drought or warming, however, *P.*  
244 *alba* had a much depleted  $\delta^{15}\text{N}$  relative to the other species. We attribute these differences to  
245 species specific utilization of different nitrogen sources (perhaps more dependent on nitrate at  
246 the HU site), or different mycorrhizal associations with higher rates of isotopic fractionation  
247 (e.g. ericoid mycorrhiza in *E. multiflora* and *C. vulgaris*; Michelsen et al. 1998; Pardo et al.  
248 2006).

249

## 250 **2 Bomb- $^{14}\text{C}$ technique to asses sources of soil respiration**

251

252 The analysis of natural  $^{13}\text{C}$  abundance in  $\text{CO}_2$  can be used to distinguish between autotrophic  
253 and heterotrophic components of soil respiration in some ecosystems where the sources of  
254 respired substrates (i.e. recent photosynthates vs. SOM) have contrasting  $\delta^{13}\text{C}$  values. Changes

255 in  $\delta^{13}\text{C}$  values can be observed after vegetation changes, e.g. where C4 plants are introduced in  
256 areas previously dominated by C3 vegetation, or vice versa, the so-called C3-C4 shift. In these  
257 situations, the  $\delta^{13}\text{C}$  signature of plants and soil carbon may differ by up to 10 ‰ (Hanson et al.  
258 2000). The partitioning of soil respiration between historical SOM and recent photosynthates  
259 can then be calculated using a linear isotope mixing models (Robinson & Scrimgeour 1995;  
260 Hanson et al. 2000). However, such vegetation shifts are rare in natural ecosystems, and  
261 therefore the use of  $^{13}\text{C}$  natural abundance to differentiate between autotrophic and  
262 heterotrophic respiration is limited (but see Millard et al. 2008).

263 As an alternative to stable carbon isotopes ( $^{12}\text{C}$ ,  $^{13}\text{C}$ ), the natural radioactive  $^{14}\text{C}$  abundance can  
264 be used to identify sources of soil respiration. Radiocarbon signatures of recent and older carbon  
265 are different as a result of the nuclear bomb tests in the atmosphere during the 1950/60s. These  
266 tests led to an increase in the  $^{14}\text{C}$  content in the atmospheric  $\text{CO}_2$  in the Northern hemisphere,  
267 which reached its maximum in 1963 ('bomb peak'). Ever since the subsequent atmospheric  
268 nuclear test moratorium, the 'bomb- $^{14}\text{C}$ ' content has decreased due to the dilution with fossil  
269 fuel-derived  $\text{CO}_2$  in the atmosphere and its incorporation in ocean and terrestrial carbon pools  
270 (Trumbore 2009). Through that incorporation, the radiocarbon analysis of ecosystem fluxes  
271 provides information about the age of decomposed carbon substrates and can be used to  
272 differentiate carbon sources within ecosystems: recently plant-assimilated carbon (autotrophic  
273 component of soil respiration) should have a similar radiocarbon signature as the current  
274 atmosphere, while the radiocarbon content of older carbon released through SOM  
275 mineralisation (heterotrophic component) reflects the year of fixation of that carbon, again a  
276 mixing model solution. Several studies have successfully achieved the separation of sources of  
277 respiration across ecosystems using the 'bomb- $^{14}\text{C}$ ' method (Gaudinski et al. 2000; Cisneros-  
278 Dozal et al. 2006; Czimczik et al. 2006; Schuur & Trumbore 2006; Subke et al. 2011). In these  
279 studies, analysis of the  $^{14}\text{C}$ - $\text{CO}_2$  signatures of roots and SOM was performed under controlled  
280 conditions and collated with analyses of field gas efflux (the mixed pool). Radiocarbon analysis  
281 of soil or ecosystem respiration has been used to evaluate the response of a range of ecosystems  
282 to different factors of climate change, such as increasing temperatures, decreasing rainfall or  
283 permafrost thaw (Borken et al. 2006; Muhr & Borken 2009; Muhr et al. 2009; Schuur et al.  
284 2009).

285 We tested the effect of experimental warming and drought on the natural abundance of  $^{14}\text{C}$  in  
286 respired soil  $\text{CO}_2$  at early stages of the climate manipulations at the Peaknaze field site (UK-  
287 PK). Our hypothesis was that drought increased heterotrophic respiration more than warming

288 in this seasonally waterlogged soil, due to a greater responsiveness of old soil carbon to drought  
289 relative to temperature as a driver (Bol et al. 2003; Domínguez et al. 2015). Therefore, we  
290 expected the greatest  $^{14}\text{C}$ -enrichment in the field-collected soil respiration samples from the  
291 drought plots. Soil efflux samples were collected in the late experimental drought period  
292 (September 2011), using a molecular sieve sampling system (Bol et al. 1995; Hardie et al. 2005;  
293 Hardie et al. 2009) attached to closed dark respiration chambers placed on the soil overnight.  
294  $\text{CO}_2$  was subsequently recovered from the molecular sieve traps for  $^{14}\text{C}$  analysis by Accelerator  
295 Mass Spectrometry (AMS; Box 2). Soil and root samples were collected to conduct separate  
296 incubations to obtain the  $^{14}\text{C}$ -signatures of the heterotrophic and autotrophic respiration,  
297 respectively. These incubations were performed in leak-tight glass jars with a connection to the  
298 molecular sieve sampling system. The results revealed a high heterogeneity of the  $^{14}\text{C}$  signature  
299 of the soil efflux with no significant effect of the warming treatment, and a trend towards the  
300 release of older carbon from the drought plots (although not statistically significant). By  
301 comparison with the known record of post-bomb atmospheric  $^{14}\text{C}$ - $\text{CO}_2$  concentration (Box 2),  
302 the carbon being released from the plots was estimated to have been fixed between six and eight  
303 years earlier (M. Dominguez, unpublished).

304

### 305 **Methods using *in-situ* $^{13}\text{C}$ labelling to study rhizodeposition utilisation**

306

307 *In-situ* pulse labelling with the stable carbon isotope ( $^{13}\text{C}$ ) is an efficient method for evaluating  
308 the time lag between carbon assimilation and  $\text{CO}_2$  release from soil (Kuzyakov & Gavrichkova  
309 2010). It can be applied to investigate a wide range of processes in the plant-soil-atmosphere  
310 continuum (Högberg et al. 2008; Brüggemann et al. 2011; Epron et al. 2012). With this method,  
311 a concentrated pulse of  $^{13}\text{C}$ -enriched or depleted substrate in the form of  $\text{CO}_2$  or of a carbon-  
312 containing organic substrate, is released into the undisturbed ecosystem. Subsequently  
313 assimilation or heterotrophic consumption will transfer the labelled carbon, and the  $^{13}\text{C}$  content  
314 of the product and organism will reflect the rate and the quantity of carbon transfer from one  
315 pool to another (Studer et al. 2014). Analysis of  $^{13}\text{C}$  in specific compounds such as PLFAs is a  
316 specific tool to assess the utilisation of different carbon sources by different functional groups  
317 of soil biota. The development of  $^{13}\text{C}$  labelling has increased the recognition of the central role  
318 that rhizodeposition plays in soil carbon cycling (Ostle et al. 2003; Jin & Evans 2010; Kuzyakov  
319 & Gavrichkova 2010; Dijkstra et al. 2013).

320

## 321 **<sup>13</sup>C-CO<sub>2</sub> pulse labelling**

322

323 In <sup>13</sup>C-CO<sub>2</sub> pulse labelling experiments, <sup>13</sup>C enriched CO<sub>2</sub> is released in closed, intact plant-  
324 soil systems during daylight hours, typically for 1.5 to 6 hours, where it is assimilated by the  
325 green plant biomass. Plant and soil samples are taken from unlabelled and labelled systems at  
326 different time intervals, with a higher sampling frequency within the first 48 hours after the  
327 labelling. The allocation of <sup>13</sup>C to belowground pools (roots, exudates, microbiota) is  
328 subsequently analysed, which allows the determination of the fraction of recently fixed carbon  
329 actively utilized by e.g. different microbial functional groups. Using <sup>13</sup>C-CO<sub>2</sub> pulse labelling,  
330 several authors demonstrated that the flux of recently photosynthesized carbon to soil microbes  
331 occurs very fast, often within a few hours of <sup>13</sup>CO<sub>2</sub> uptake (Rangel-Castro et al. 2005). The  
332 maximum incorporation of <sup>13</sup>C into microbial RNA occurs within four to eight days after the  
333 pulse (Ostle et al. 2003). Fungi typically show the greatest utilisation of plant-derived carbon  
334 within the first 48 hours after plant labelling. Lower <sup>13</sup>C enrichment in bacterial biomarker  
335 PLFAs indicates a delay in the utilisation of plant-derived carbon by bacteria, or a greater  
336 dependence of bacteria on carbon sources different from recently-fixed carbon. Gram positive  
337 bacteria and, in particular actinomycetes, are known to rely less on plant-derived carbon than  
338 Gram negative bacteria (Butler et al. 2004; Treonis et al. 2004; Balasooriya et al. 2008; 2013;  
339 2014). The levels of allocation of belowground fixed carbon and the subsequent utilisation by  
340 microbes might be affected by a range of factors such as the seasonality of plant activity.  
341 Usually, more carbon is allocated belowground towards the end of the growing season  
342 (Högberg et al. 2010; Balasooriya et al. 2013), under exposure to elevated atmospheric CO<sub>2</sub>  
343 concentrations (Denef et al. 2007; Jin & Evans 2010; Reinsch et al. 2013), under drought  
344 conditions (Fuchslueger et al. 2014) or in plants grown on fertile soils (Denef et al. 2009;  
345 Paterson et al. 2011).

346 In the climate change network, several pulse-labelling experiments have been conducted in  
347 combination with <sup>13</sup>C-PLFA analyses to study rhizodeposit utilisation by microbes. At the  
348 Clocaenog site (UK-CL) we aimed to study the utilisation of rhizodeposits along a soil moisture  
349 gradient, by applying a <sup>13</sup>C-CO<sub>2</sub> pulse during the late growing-season (August 2011). We used  
350 transparent domes of 50 cm diameter and 100 cm height, enclosing individual *C.*  
351 *vulgaris* plants. We applied repeated pulses of <sup>13</sup>C-CO<sub>2</sub> (99 atom% <sup>13</sup>C = 99% <sup>13</sup>C + 1% <sup>12</sup>C)  
352 over eight hours (Box 3). The domes were sealed to a frame which was inserted into the ground  
353 at least ten days before the pulse, and had several sealed septa to collect gas samples to estimate  
354 the concentration of the <sup>13</sup>C-labelled CO<sub>2</sub>. Plant leaves and soil from the rooting zone were

355 collected at different times after the labelling, using a higher sampling frequency during the  
356 first hours after the pulse. Soils were freeze-dried, sieved to  $\leq 5$  mm and PLFAs were extracted.  
357 Fatty acid methyl esters (FAMES) were analysed by gas chromatography combustion-isotope  
358 ratio mass spectrometry (GC-c-IRMS). The main challenge was the low recovery of  $^{13}\text{C}$  label  
359 in the belowground compartment, especially in individual FAMES. Despite the applied  $^{13}\text{C}$   
360 concentration of 99 atom%, the apparent low photosynthetic rates combined with the excessive  
361 dilution of the  $^{13}\text{C}$  label in the large carbon pools of unlabelled woody branches and root- and  
362 microbial biomass resulted in an overall low level of  $^{13}\text{C}$  enrichment in the FAMES (Box 3).  
363 Similar patterns have also been observed in other pulse labelling experiments (Griffith et al.  
364 2004).

365 Three pulse-labelling events were conducted at the Brandbjerg site (DK-BRA,) between 2010  
366 and 2013 (Box 3). The Brandbjerg experiment consists of drought and warming manipulations  
367 in combination with ambient and elevated levels of  $\text{CO}_2$  concentration. The developed  
368 experimental setup for pulse-labelling aimed i) to be easily deployable in remote areas, ii) to  
369 distribute labelled  $^{13}\text{C}\text{-CO}_2$  to as many plots at the same time as possible to ensure similar and  
370 constant conditions for  $\text{CO}_2$  uptake by the vegetation, and iii) to ensure constant  $\text{CO}_2$   
371 concentration available to the vegetation throughout the labelling period. Therefore, a mobile  
372 flow-through system suitable for continuous  $^{13}\text{C}\text{-CO}_2$  delivery was developed (Box 3): A gas-  
373 tight vinyl balloon (~3 m diameter) was filled with  $\text{CO}_2$  free synthetic air and mixed with  $^{13}\text{C}\text{-}$   
374  $\text{CO}_2$  (50 or 99 atom%) that supplied the transparent chambers enclosing the vegetation of  
375 interest with air over the duration of the experiments ranging from 4 to 7.5 hours. Air was  
376 pumped continuously through gas tight tubing via electric diaphragm pumps (Reinsch & Ambus  
377 2013). The first experiment was conducted at the end of the growing season (October 2010),  
378 when we observed the highest allocation of carbon belowground as measured by  $^{13}\text{C}$  in soil  
379 respiration (Reinsch et al. 2014). The second experiment was conducted in the spring (May  
380 2011) and showed a major allocation of carbon to aboveground structures under elevated  
381 atmospheric  $\text{CO}_2$  concentration, but carbon allocation to belowground structures was higher in  
382 drought plots than in untreated control plots. The allocation of recently-assimilated carbon  
383 under warming conditions was similar to that under ambient conditions. The last experiment,  
384 conducted in early season 2013 (June), was performed during a period with impeded  
385 photosynthetic activity and indicated that labelling performance is poor when vegetation is  
386 recovering from harsh winter conditions with bare frost or severe drought conditions (Box 3).

387 Thus, it is important that the vegetation of interest displays green, photosynthetically active  
388 structures to facilitate CO<sub>2</sub> uptake and sufficient labelling of ecosystem carbon pools.

389 Our studies illustrate the complexity of controlling *in-situ* pulse-labelling experiments in  
390 ecosystems dominated by woody plants, which is even more challenging with <sup>13</sup>C-CO<sub>2</sub> than  
391 with <sup>14</sup>C-CO<sub>2</sub> because of their respective atmospheric backgrounds and detection limits (Epron  
392 et al. 2012). Ideally, <sup>13</sup>C doses for *in-situ* use should be carefully tested in trials, considering  
393 the nature of the studied vegetation and the compounds to be analysed. If e.g. specific  
394 compounds of the soil microbial biomass are the main interest, then strong isotopic doses should  
395 be applied, and it is advisable to deploy the <sup>13</sup>C pulse when plants naturally allocate carbon  
396 belowground e.g. when preparing for winter. The <sup>13</sup>C signal can be increased by using highly  
397 labelled <sup>13</sup>C-CO<sub>2</sub> (99 atom %). However, the usage of a highly enriched CO<sub>2</sub> can potentially  
398 lead to blurry signals and has to be applied with caution (Watzinger 2015). Furthermore, <sup>13</sup>C-  
399 CO<sub>2</sub> concentration inside the labelling chamber should be as close as possible to ambient  
400 values, because unrealistic high CO<sub>2</sub> concentration will change plant CO<sub>2</sub> uptake. Repeated  
401 moderated <sup>13</sup>C-CO<sub>2</sub> applications during longer exposure times might be more appropriate, but  
402 inside closed transparent chambers, temperature and humidity may increase if the labelling  
403 period is prolonged, which also affects photosynthetic processes (Epron et al. 2012). Losses of  
404 <sup>13</sup>C due to physical diffusion and adsorption/desorption into the chamber and tubing material  
405 should also be considered. In particular, the back-diffusion of the <sup>13</sup>CO<sub>2</sub> from the soil to the  
406 atmosphere which entered the soil pores during the labelling might confound the interpretation  
407 of measured belowground respiration (Subke et al. 2009; Selsted et al. 2011). However, when  
408 applied properly, the insights into terrestrial carbon allocation will be detailed and novel (Box  
409 3).

410

## 411 **2 Free Air Carbon dioxide Enrichment (FACE)-labelling**

412

413 An alternative method for <sup>13</sup>C pulse-labelling of vegetation and whole-ecosystems is to utilize  
414 the <sup>13</sup>C-depleted CO<sub>2</sub> in already planned or ongoing FACE experiments. The FACE technique  
415 has through decades been used within cropping systems (Kimball 2016), grasslands (Hovenden  
416 et al. 2014; Reich et al. 2014; Mueller et al. 2016) and forests (Terrer et al. 2016) experiments,  
417 with the primary goal of assessing potential carbon dynamics and enhancement of plant growth  
418 (Andresen et al. 2016). As a side effect, the change in carbon isotopic composition of vegetation  
419 exposed to FACE-treatment can be used to trace freshly assimilated carbon into soil microbial

420 biomass, fauna and organic carbon pools. This approach was used at the Brandbjerg site (DK-  
421 BRA). The CO<sub>2</sub> used to elevate concentrations of atmospheric CO<sub>2</sub> to 510 ppm had δ<sup>13</sup>C values  
422 ranging from -3.0 to -36.7 ‰ throughout 8 years of experimental treatment, with an overall  
423 mean of -26.1 ‰. The source of the CO<sub>2</sub> supplied by Air Liquide (Air Liquide, Denmark) was  
424 most often a brewery surplus CO<sub>2</sub> as a chemically obtained side product. The FACE mixing of  
425 the added CO<sub>2</sub> with ambient CO<sub>2</sub> in the moving air mass resulted in a <sup>13</sup>C depletion ranging  
426 from -6.7 to -15.6 ‰. On average, this equals a depletion of CO<sub>2</sub> in FACE plots of -4.8 ‰  
427 relative to the atmospheric -8 ‰ average. Ecosystem carbon pools became depleted  
428 accordingly, and the FACE <sup>13</sup>C depletion acted as a long-term persistent isotope labelling. As  
429 a result, soil fauna (Enchytraeids) sampled from each of the climate-treated plots was  
430 significantly depleted in δ<sup>13</sup>C by -0.5 to -2.0 ‰ in the CO<sub>2</sub> treatments (Andresen et al. 2011).  
431 This was due to translocated <sup>13</sup>C substrate through the food web, starting with plant assimilation  
432 of <sup>13</sup>C-depleted CO<sub>2</sub>, followed by plant root exudation and microbial utilization of the <sup>13</sup>C  
433 depleted substrate and eventual digestion of microbes by enchytraeids. Hereby the freshly  
434 supplied carbon source was recognized to be transferred in the natural setting, within a given  
435 time scale. Also microbial biomass and PLFAs had different baseline of <sup>13</sup>C content in ambient  
436 (not-treated) plots compared to CO<sub>2</sub> treated plots (Andresen et al. 2014). This was used for the  
437 calculation of <sup>13</sup>C enrichments for each PLFA biomarker individually, also illustrating the  
438 pathway of newly-assimilated carbon into microbial biomass. A drawback of the <sup>13</sup>C-FACE  
439 label is again contamination of the surroundings, as even short and small un-planned draft winds  
440 can carry the depleted label onto ‘ambient’ plots, and these will most likely be ‘contaminated’  
441 with <sup>13</sup>C (though not markedly exposed to high CO<sub>2</sub> concentrations) after some years of FACE  
442 activity. Therefore, one needs to collect reference material for the ‘natural abundance’ level  
443 well away from the FACE experiment.

444

### 445 **3 In situ injection of <sup>13</sup>C-enriched substrate solutions**

446

447 As a much more localized approach to a specific area, *in-situ* addition of <sup>13</sup>C- and <sup>15</sup>N-enriched  
448 substrates directly below the soil surface can be used to assess the competition for the substrate  
449 between i) plants and soil microbes, ii) microbial groups, and iii) the effects of the climate  
450 change treatments upon the competition for carbon or nitrogen substrates. Much research has  
451 focused on the sharing of nitrogen sources between plant and microbes (Kuzyakov & Xu 2013)  
452 using *in-situ* soil injections of <sup>15</sup>N labelled inorganic nitrogen (ammonium and nitrate) or

453 organic nitrogen (amino acids) (Bardgett et al., 2003; Sorensen et al., 2008). Once amino acids  
454 with dual labelled compounds ( $^{15}\text{N}$  and  $^{13}\text{C}$ ) were available for experimental use, double-  
455 labelled substrate was used to explore e.g. plant uptake of intact amino acids (Näsholm et al.  
456 2009; Rasmussen et al. 2010), and microbial utilization of carbon substrates (Dungait et al.  
457 2013; Rinnan & Baath 2009).

458 In a labelling experiment at the DK-BRA site, amino acid injections into the soil were  
459 conducted to analyse the impact of the climate treatments on the uptake of free amino acid  
460 nitrogen by plants and soil microbes. Dual-labelled glycine ( $^{13}\text{C}_2^{15}\text{N}$ -glycine: 99 atom%  $^{13}\text{C}$  -  
461 of both carbon atoms - and 99 atom%  $^{15}\text{N}$ ) was added to  $20 \times 20 \text{ cm}^2$  sub-plots (Andresen et al.  
462 2009). Each sub-plot received 0.1 L of re-demineralised water labelled with 0.027 g glycine,  
463 corresponding to  $687 \text{ mg glycine m}^{-2}$  ( $223 \text{ mg C m}^{-2}$  or  $0.016 \text{ mg glycine g}^{-1}$  dry weight soil).  
464 The label was injected into the soil just below the soil surface with a syringe moved among 16  
465 evenly spaced points of a template, placed on top of the vegetation (Andresen et al. 2009). One  
466 day (c. 24 h) after labelling with glycine, soil cores were sampled from the soil surface to 15  
467 cm depth, for determining the relative uptake of the amino acid in plant roots (IRMS solid  
468 sample) and soil microbes. As in many other soil labelling experiments, the largest label  
469 recovery (measured by  $^{15}\text{N}$  recovery since respiratory losses of  $^{13}\text{C}$  remain unknown) was found  
470 in the total microbial biomass compared to total plant biomass (Kuzyakov & Xu 2013). A  
471 subsample of fresh soil was extracted with re-demineralised water, and another set of  
472 subsamples was first vacuum-incubated with chloroform for 24 hours to release microbial  
473 carbon and nitrogen (Joergensen & Mueller 1996; Brookes et al. 1985), before extraction with  
474 re-demineralised water. A third subsample of soil was freeze dried (lyophilized) and later used  
475 for PLFA extractions. The  $^{13}\text{C}$  enrichment in marker PLFAs thus indicated the activity (vitality)  
476 of the specific microbial group (Watzinger 2015). We found that bacteria opportunistically  
477 utilised the freshly added glycine substrate, i.e. incorporated  $^{13}\text{C}$ , whereas fungi showed only  
478 minor or no glycine derived  $^{13}\text{C}$ -enrichment (Andresen et al. 2014). In comparison,  $^{13}\text{C}$  traced  
479 into the microbial community via the  $^{13}\text{C}$ - $\text{CO}_2$  pulse label at the same site (DK-BRA) also  
480 reached the bacterial community first. Bacteria showed high  $^{13}\text{C}$  enrichment compared to fungal  
481 groups (Reinsch et al. 2014). This suggests that *in-situ* injection of  $^{13}\text{C}$  substrates might be a  
482 plausible alternative to mimic rhizodeposition effects. With the direct addition of  $^{13}\text{C}$  label to  
483 the soil a strong labelling of the microbial community was more easily achieved than with the  
484 indirect  $^{13}\text{C}$  labelling of microbes via plant assimilated  $^{13}\text{C}$ - $\text{CO}_2$  (Box 3).

485



486 **Use of labelled carbon-compounds to analyse carbon mineralisation by soil**  
487 **microorganisms**  
488

489 Since soil microorganisms have an important role in controlling the availability of nutrients via  
490 mineralisation of SOM, our understanding of how microbial functioning in the ecosystem is  
491 altered by global change must be improved (Grayston et al. 1997). Microbial catabolic diversity  
492 of a soil is directly related to the carbon decomposition function within a soil and potentially  
493 provides a sensitive and ecologically relevant measure of the microbial community structure  
494 (Garland & Mills 1991). Subsequently, multiple assays have been developed to generate  
495 community level physiological profiles (CLPP) that can act as fingerprints of microbial  
496 function. Three approaches for measuring CLPP in soils are reported in the literature: (i) Biolog  
497 (Garland & Mills 1991); (ii) a substrate-induced respiration (SIR) technique (Degens & Harris  
498 1997); and (iii) MicroResp (Campbell et al. 2003). These methods are all based on quantifying  
499 CO<sub>2</sub> respired during the mineralisation of organic carbon compounds that vary in size, charge  
500 and structural complexity. The first approach, Biolog MicroPlate™ (Biolog), assesses the  
501 catabolic diversity of soil organisms using a microtitre plate by incubating a soil culture in the  
502 presence of nutrients and 95 different carbon substrates; respired CO<sub>2</sub> is used to reduce a  
503 tetrazolium violet salt, which results in a colour change that can be quantified colorimetrically  
504 (Garland & Mills 1991). This approach, however, has been criticized for bias towards fast  
505 growing organisms that thrive in culture (Preston-Mafham et al. 2002). In response to the  
506 criticism of the Biolog method, Degens & Harris (1997) developed a method based on SIR  
507 where individual substrates are added to intact soil and evolved CO<sub>2</sub> is sampled and quantified.  
508 Finally, Campbell et al. (2003) combined aspects of both methods (MicroResp™) where the  
509 response to carbon substrate addition to soil is measured colorimetrically using a cresol red  
510 indicator dye in a microtitre plate format.

511 Community level physiological profiling of soils samples collected from all treatments across  
512 the network was conducted to determine the catabolic utilisation profile, turnover and pool  
513 allocation of low molecular weight (LMW) carbon compounds by using a selection of <sup>14</sup>C-  
514 labelled substrates. This method enabled the attribution of respired CO<sub>2</sub> to specific metabolic  
515 processes that facilitates the quantification and qualification of microbial mineralisation  
516 kinetics of substrates varying in structural complexity and recalcitrance. The kinetics of  
517 microbial <sup>14</sup>C-CO<sub>2</sub> evolution can be described using a first order exponential decay model (Box  
518 4). The number of terms used in the exponential decay model can be used to explain how  
519 microbial kinetics relates time, substrate complexity and carbon pool allocation to, for example,

520 rapidly cycled labile soil solution carbon, microbial structural carbon and recalcitrant  
521 extracellular soil organic carbon (Kuzyakov & Demin 1998; Nguyen & Guckert 2001; Boddy  
522 et al. 2007). Attribution of modelled carbon pool sizes and turnover rates to biological function  
523 are not only time and substrate dependent. Therefore, soil physical, biological and chemical  
524 interactions may be miss attributed to biological function. Indeed, the lack of knowledge and  
525 techniques available to examine the interaction between discrete carbon pools (Glanville et al,  
526 2016). Using the half-life of  $^{14}\text{C}$  labelled carbon in soil solution we were able to examine the  
527 environmental gradient of the warming treatment across the climate change network and  
528 identified that temperature becomes rate limiting for microbial uptake of carbon from the soil  
529 solution pool at  $< 10.5\text{ }^{\circ}\text{C}$ . We also showed that experimentally manipulated warming simply  
530 speeds up the catabolic utilisation of labile LMW carbon in a predictable pattern (Box 4).

531

## 532 **Conclusions and recommendations**

533

534 Stable isotope studies provide insightful information about carbon (and nitrogen) fluxes through  
535 the plant-soil-atmosphere continuum with minimal disturbance to the system, and contribute to  
536 advance our understanding of climate change impacts on aboveground-belowground linkages.  
537 However, their application is not exempt from difficulties and disadvantages. To keep a high  
538 caution and avoid mistakes, our collective recommendations for applying the described  
539 methods are provided and addressed in Table 1.

540 *In-situ* pulse-labelling studies are powerful to analyse short-term carbon fluxes in the plant-soil  
541 system, but there are major seasonality constraints to the distribution of the label throughout  
542 the ecosystem compartments, *i.e.* the seasonality of carbon allocation belowground due to  
543 changing plant activity, or the plant health status which determines the amount of tracer entering  
544 the system. A significant challenge was the achievement of sufficient  $^{13}\text{C}$  enrichment in  
545 microbial biomass in *in-situ*  $^{13}\text{C}$ - $\text{CO}_2$  pulse-labelling studies, where the pools of background  
546 carbon in the studied compartments were high and hence diluted the  $^{13}\text{C}$  signal. This was less  
547 of a problem for  $^{14}\text{C}$  studies, due to more sensitive analysis methods.

548 Importantly, field plots previously ‘contaminated’ by highly enriched isotope labelling should  
549 be considered potentially inoperable for further scientific isotope studies using the natural  
550 abundance approach. However, plant and soil structures remain largely undisturbed. In outlook  
551 for setting up a large-scale climate manipulation, areas that have not been previously used for  
552 experimental work with isotopes should consequently be selected. In particular, the ‘bomb-C’

553 method is very sensitive to the contamination of soil or plant samples with <sup>14</sup>C-enriched  
554 material, and thus its application should be limited to sites and facilities where no <sup>14</sup>C-labelling  
555 work has been conducted. Additionally, it should be noted that any history of fertilization might  
556 also alter the natural isotope abundance of ecosystem compartments (in particular <sup>15</sup>N  
557 signatures), potentially confounding experimental results. Furthermore, military training  
558 grounds, public recreational activity and vicinity of traffic are known to potentially contaminate  
559 the soil with ‘artificial compounds’ which might interfere with delicate measurements on HPLC  
560 and GC-MS systems.

561 Incubation studies with isotope labelled carbon compounds *in-situ* or *in-vitro* are relatively fast  
562 to conduct and produce insightful data. The rapid utilisation of labelled substrates by soil  
563 microorganisms, occurring immediately upon application, is a controlling factor for the timing  
564 of the experimental work both at field and lab conditions. Hence, a sampling scheme needs to  
565 be carefully planned before experiment initiation. Furthermore, pre-obtained knowledge of site  
566 specific plant and microbial activity, substrate affinity and natural substrate concentrations is  
567 important for planning any tracer application experiment (Table 1).

568

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580

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899

900 **Table**

901 **Table 1.** Suggestions and advice to consider when applying isotopic methods for the study of  
 902 carbon fluxes in the plant-soil system.

Method	Expenses (cost)	Advice (do's and don'ts)	Before you start	Data analysis hint	Time spent
<b>Bomb-C (natural <sup>14</sup>C abundance)</b>	High (AMS analysis); Equipment for CO <sub>2</sub> sampling is relatively cheap (closed chambers, carbon free pump, batteries, and molecular sieve system). An IRGA is also required.	<ul style="list-style-type: none"> <li>- Avoid materials and labs with possible <sup>14</sup>C contamination</li> <li>- If soil CO<sub>2</sub> is to be analysed in the field, long incubation times are required to get sufficient CO<sub>2</sub> for AMS analysis (typically &gt;1 ml).</li> <li>- Think carefully about the soil depths to be analysed, and take the sample consistently. <sup>14</sup>C signatures might vary strongly along few cm in the soil.</li> <li>- If bulk soil <sup>14</sup>C is to be analysed, try to remove the roots as much as possible, because of their contrasted <sup>14</sup>C signature.</li> </ul>	<ul style="list-style-type: none"> <li>- Discuss your experimental setup and objectives with the Radiocarbon facility staff.</li> <li>- If you are not sure about potential <sup>14</sup>C contamination in your lab, use another lab or make a swipe test.</li> <li>- Plan carefully the minimum number of samples required, as AMS analysis are expensive.</li> <li>- Make previous trials to assess the incubation times required to get a sufficient CO<sub>2</sub> sample</li> <li>- Go through the whole process of sample preparation with a trial sample.</li> </ul>	<ul style="list-style-type: none"> <li>- Discuss your results with the Radiocarbon facility staff.</li> </ul>	<ul style="list-style-type: none"> <li>- Processing time depends on the type of sample, although is usually low; determination by AMS may take several months depending on the facility.</li> </ul>
<b>In situ <sup>13</sup>C pulse-labelling</b>	<sup>13</sup> C- enriched compounds used for labelling and as standards are usually expensive; <sup>13</sup> C determination by IRMS is much cheaper than <sup>14</sup> C by AMS, although more expensive than <sup>14</sup> C by liquid scintillation ( <sup>14</sup> C labelling).	<ul style="list-style-type: none"> <li>- Consider the target pools to be analysed and plan your doses consequently. Take the potential dilution of the label by the unlabelled root system or soil carbon pool into account.</li> <li>- Think about the trade-off between faster and stronger or weaker and longer <sup>13</sup>C pulses.</li> <li>- If your study requires a high <sup>13</sup>C enrichment, mind the potential risk of contaminating the site.</li> <li>- Avoid above ambient CO<sub>2</sub> concentrations in the chamber.</li> <li>- If you need to monitor CO<sub>2</sub> during your pulse, remember that IRGAs are rather insensitive to <sup>13</sup>CO<sub>2</sub>.</li> <li>- High sampling frequency immediately after the pulse application is recommended.</li> </ul>	<ul style="list-style-type: none"> <li>- Test your chamber and tubing materials for adsorption / desorption effects, and ensure these are without carbon content (use PTFE (Teflon) tape, not gluing paper-based).</li> <li>- Make a previous trial if possible and go through the whole process of sample preparation.</li> </ul>	Report the label addition per area: g <sup>13</sup> C m <sup>-2</sup> .	<ul style="list-style-type: none"> <li>- Pulse labelling experiments are usually short, but intensive. High sampling frequency after the pulse is very time-consuming.</li> <li>- Sample processing depends on the type of sample and number of replicates.</li> <li>- Experiments requiring root washing or microbial compound extraction are time consuming.</li> </ul>
<b>Natural abundance of isotopes (<sup>13</sup>C and <sup>15</sup>N)</b>	Rather cheap IRMS analysis.	<ul style="list-style-type: none"> <li>- Make sure the history of sampling site is known (previous labelling experiments?)</li> </ul>	<ul style="list-style-type: none"> <li>- Avoid sample contamination.</li> <li>- Be aware that FACE can dilute the isotopic signal, most CO<sub>2</sub> enriched systems use <sup>13</sup>C depleted sources,</li> </ul>	<ul style="list-style-type: none"> <li>- Make sure the experiment is feasible with sufficient δ<sup>13</sup>C shift and fractionation expected to be strong enough to</li> </ul>	<ul style="list-style-type: none"> <li>- Sampling time and grinding / weighing of sample.</li> <li>- Analysis usually done at dedicated</li> </ul>

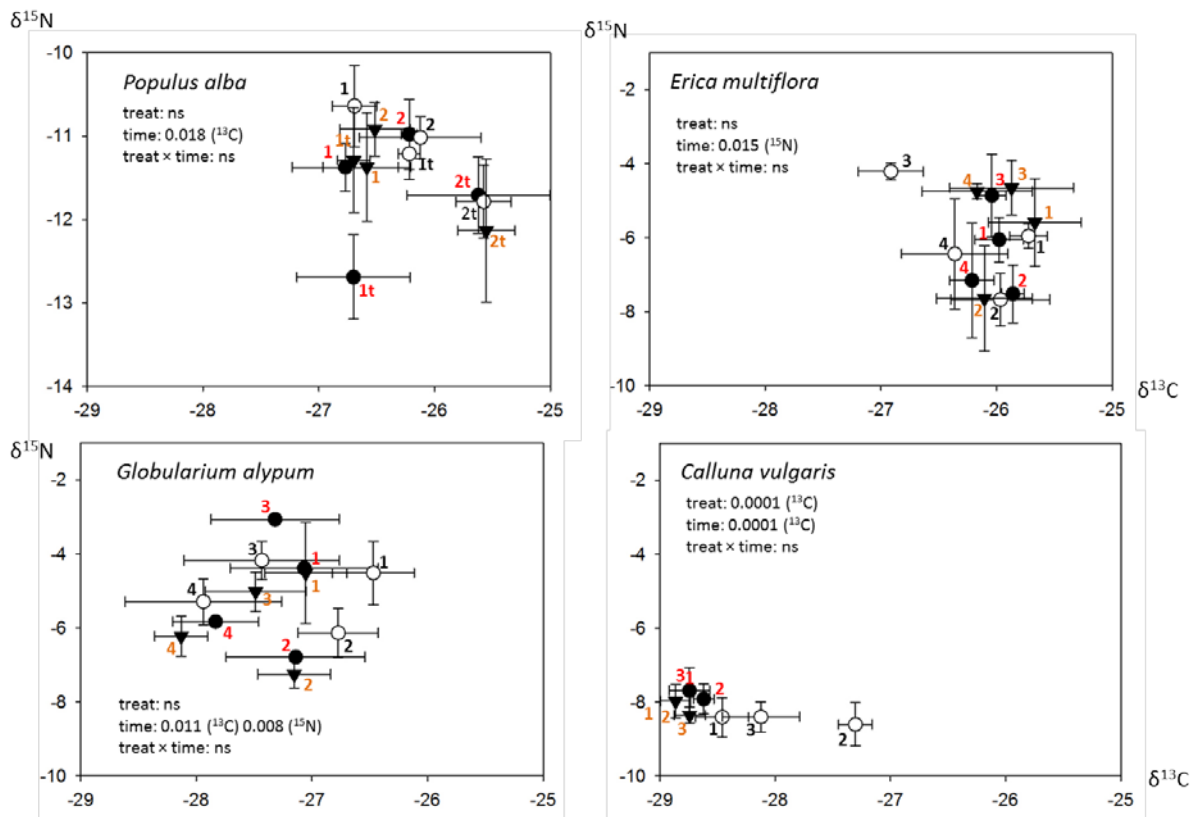
			because this is cheaper.	measure, base this on known discrimination at certain points in the carbon and nutrient cycle.	natural abundance facility.
<b><sup>14</sup>C-substrates</b>	Analysis of the trapped <sup>14</sup> C-CO <sub>2</sub> is relatively cheap.	- High risk of contaminating lab equipment.	- You need to work in a dedicated <sup>14</sup> C lab safely away from the natural abundance facility.		- Continue sampling until decline in emission is level, this ensures better model fit.
<b><sup>13</sup>C-injection <i>in situ</i></b>	IRMS of dry matter plant material and soil cheap. The GC-c-MS of PLFAs for determination of <sup>13</sup> C-enrichment of individual PLFAs requires a specialised lab.	- Contamination risk of <sup>13</sup> C leaching is present, but smaller to our judgement than from <sup>13</sup> C-CO <sub>2</sub> experiments. - Do not use areas dedicated to natural abundance work.	- Labelling intended for soil microbial components is more intense from <sup>13</sup> C liquid substrate <i>in-situ</i> injection than from <sup>13</sup> C-CO <sub>2</sub> pulse labelling.		- Soil sampling is destructive, consider to have several parallel plots to harvest an undisturbed plot at each sampling event. - Sample handling from field work until the extraction takes a few days so plan only one sampling event per week if possible.

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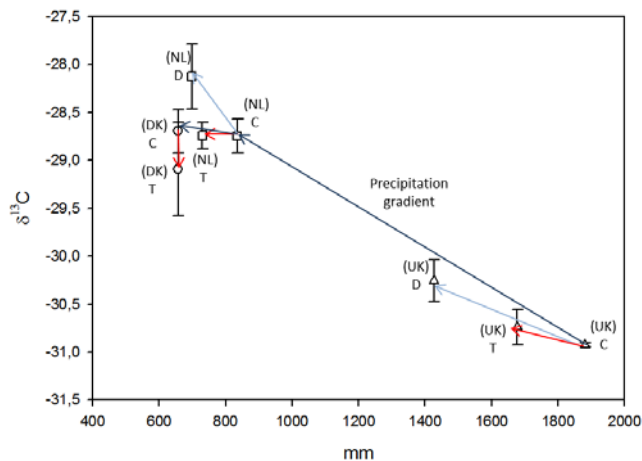
904 **Box 1. Isotopic signal of plant leaf responses to precipitation**

905 **A**



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907 **B**



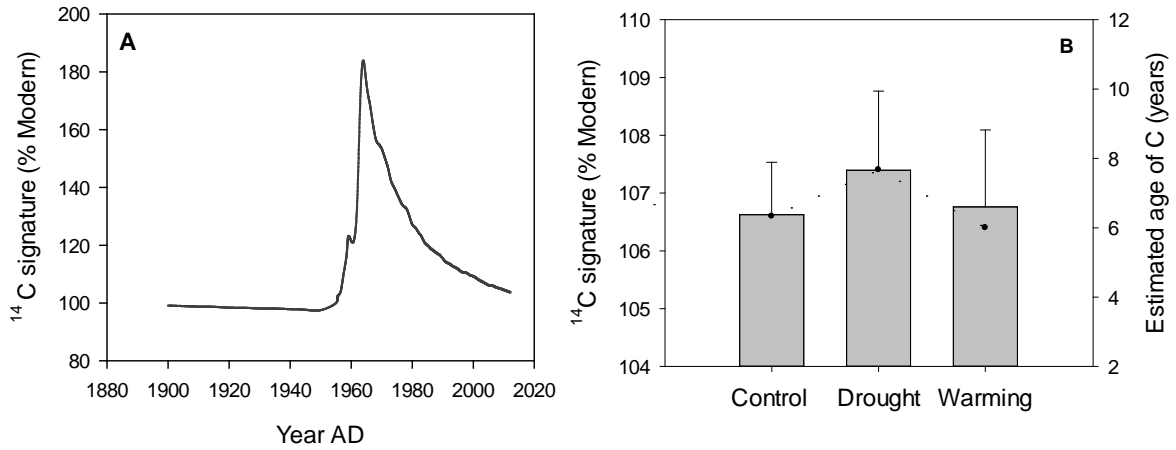
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909 Stable isotopes in aboveground plant material:  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  from isotopic ratio mass  
 910 spectrometry (IRMS) analysis of solid samples. **A:** Leaves and twigs (t) from *P. alba* (HU), *E.*  
 911 *multiflora* L. (SP), *G. alypum* L. (SP) and *C. vulgaris* (NL); filled circle ● is control, open circle  
 912 ○ is drought treatment, ▼ is warming treatment. P-values indicate effects of treatment, year,  
 913 and the interaction of these factors on  $^{13}\text{C}$  or  $^{15}\text{N}$ , analysed by two-way ANOVA; ns is non-

914 significant effect. Number indicates year (2001=1, 2002=2, 2003=3 or 2004=4). Species (site)  
915 differences and annual differences are stronger than treatment effects. **B:**  $\delta^{13}\text{C}$  of *C. vulgaris*  
916 versus annual precipitation of the previous year. Within each site, leaf  $\delta^{13}\text{C}$  was higher in the  
917 drought treatment in comparison to control. Across sites, plants from drier sites (lower  
918 precipitation) show higher leaf  $\delta^{13}\text{C}$  values. The response to the drought treatment is the same  
919 as moving to a drier site.

920

921 **Box 2. Impact of warming and drought on the  $^{14}\text{C}$  signature of soil**  
 922 **respiration**  
 923



924

925 Closed dark respiration chambers and a molecular sieve sampling system was used in the field  
 926 (UK-PK) to collect  $\text{CO}_2$  of the soil efflux for the analysis of its  $^{14}\text{C}$  signature by Accelerator  
 927 Mass Spectrometry (AMS). **A:** Recent and older carbon sources have contrasting radiocarbon  
 928 ( $^{14}\text{C}$ ) signatures as a result of the nuclear bomb tests in the atmosphere during the 1950/60s.  
 929 These tests led to a global increase in the  $^{14}\text{C}$  content in the atmospheric  $\text{CO}_2$ , which reached  
 930 its maximum in ~AD1963. The unit for  $^{14}\text{C}$  signature (% Modern) is a measurement of the  
 931 deviation of the  $^{14}\text{C}/^{12}\text{C}$  ratio of a sample from the "Modern" standard, which is defined as 95%  
 932 of the radiocarbon concentration (in AD 1950) of a reference material (NBS Oxalic Acid I,  
 933 SRM 4990B), adjusted to a  $\delta^{13}\text{C}$  reference value of  $-19\text{‰}$ . **B:** The  $^{14}\text{C}$  signature of the soil  
 934 efflux measured at the site (bars, left axis) was highly heterogeneous (ranging from 105.49 to  
 935 110.13 % Modern; values of  $> 100\%$  Modern suggest that a substantial component (and  
 936 potentially all) of the carbon was trapped by photosynthesis during the post-bomb era i.e. since  
 937 ~AD 1955), with no significant effect of the warming treatment, and a trend towards the release

938 of older carbon in the drought plots. On average, the carbon being released from the plots had  
939 been fixed from the atmosphere between six and eight years earlier (line, right axis). Incubations  
940 confirmed that the carbon respired by roots (mostly of *V. myrtillus L.*) was recently fixed  
941 (similar  $^{14}\text{C}$  signature as the atmosphere at the time of sampling), while the carbon released  
942 from root-free soil samples (heterotrophic component) showed a variable range of ages, with  
943 substantial components of pre-bomb carbon (carbon fixed before AD 1955). **C:** Detail of a  
944 closed static chamber used to collect  $\text{CO}_2$  from the soil efflux. Air is pumped in a closed loop  
945 from the chamber through a quartz glass cartridge containing a zeolite molecular sieve, which  
946 traps the  $\text{CO}_2$  allowing it to be returned to the laboratory, recovered by heating, and analysed  
947 by AMS.

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949 **Box 3. Analysing rhizodeposit utilisation by microbes in the field**

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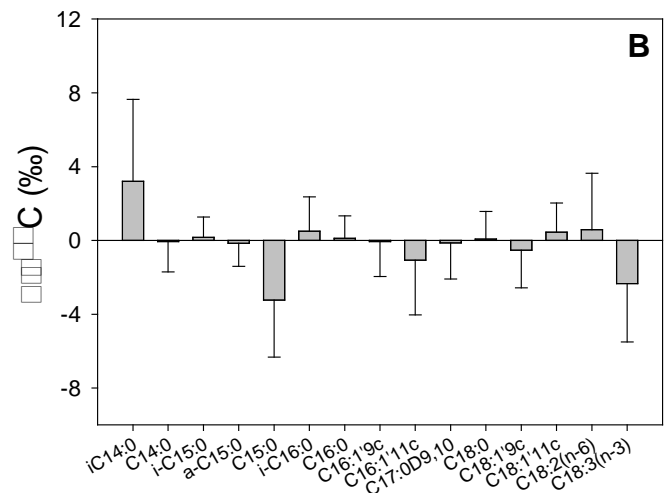
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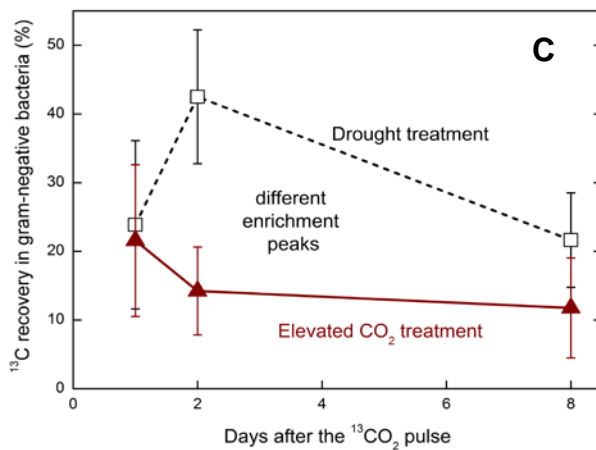
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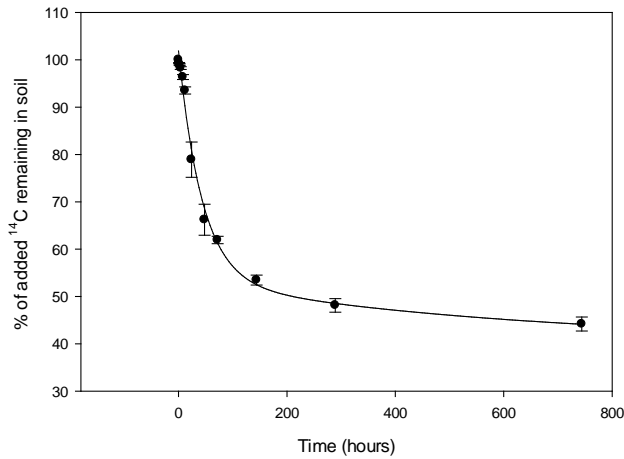
965 *In-situ* pulse-labelling experiments adding  $^{13}\text{C}$ - $\text{CO}_2$  to closed transparent chambers were used  
 966 to study the translocation of the recently-fixed carbon belowground and the rhizodeposition  
 967 utilization by microbes, e.g. by measuring  $^{13}\text{C}$  incorporation into microbial biomarkers. **A:** At  
 968 the Clocaenog site (UK-CL) this technique was applied along a peat layer gradient. Repeated  
 969 pulses of  $^{13}\text{C}$ - $\text{CO}_2$  were applied during eight hours to *C. vulgaris* using sealed domes attached  
 970 to a core inserted into the ground. **B:** The incorporation of  $^{13}\text{C}$  into soil microbial PLFAs was  
 971 analysed. Despite a high applied dose of  $^{13}\text{C}$  (99 atom %), the dilution of the tracer within the  
 972 large pool of unlabelled root biomass was remarkable, and as a consequence most of the  
 973 analysed PLFAs showed no  $^{13}\text{C}$  enrichment. **C:**  $^{13}\text{C}$  recovery in Gram negative bacteria after a  
 974  $^{13}\text{C}$ - $\text{CO}_2$  pulse at the Brandbjerg site (DK-BRA). The enrichment pattern in PLFAs attributed  
 975 to Gram negative bacteria in soils exposed to drought and elevated  $\text{CO}_2$  concentration (+120

976 ppm) for 8 years show different carbon utilization patterns and magnitudes under imposed  
977 climatic conditions implying changed carbon cycle dynamics. **D:** Flow-through pulse-labelling  
978 equipment showing the gas reservoir containing  $^{13}\text{C-CO}_2$  for up to eight hours of labelling  
979 connected to transparent Plexiglas chambers via tubing.

980

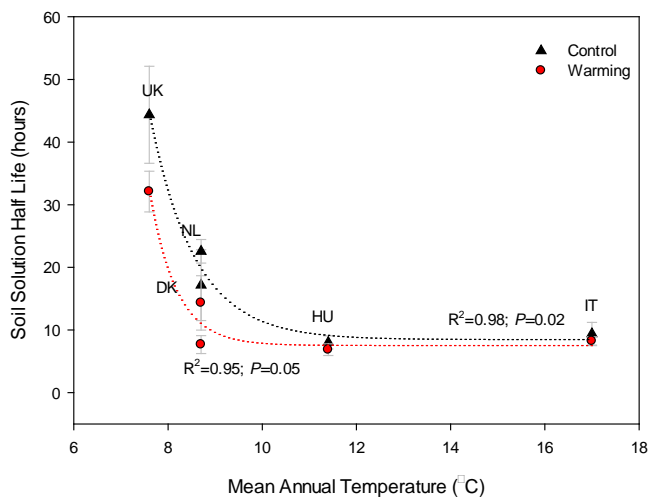
981 **Box 4. Exponential decay kinetics for  $^{14}\text{CO}_2$  evolution during microbial  $^{14}\text{C}$**   
982 **substrate mineralisation**

983 **A**



984

985 **B**



986

987 The catabolic utilisation profile, turnover and pool allocation of low molecular weight (LMW)  
988 carbon substrates was determined in soils collected across the experimental network. A  
989 selection of sixteen  $^{14}\text{C}$  labelled amino acids and sugars varying in structural complexity and  
990 recalcitrance were used in a multiple substrate induced respiration (SIR) assay on soil. Evolved  
991  $\text{CO}_2$  was collected using NaOH traps and absorbed  $^{14}\text{CO}_2$  was measured with a scintillation  
992 counter (Perkin Elmer). The rate at which radiolabelled substrates were metabolised by the  
993 microbial community was used to determine microbial uptake kinetics and turnover in the

994 absence of plants (no autotrophic input). A double term first order decay model with an  
995 asymptote described our data best;

$$996 \quad f = y_0 + a_1 e^{-k_1 t} + a_2 e^{-k_2 t} \quad \text{eqn. 1.}$$

997 where  $f$  describes the amount of  $^{14}\text{C}$ -labelled substrate or metabolites remaining in the soil at  
998 time  $t$ , the asymptote  $y_0$  explains the  $^{14}\text{C}$  labelled material adsorbed to unavailable soil  
999 complexes or metabolites partitioned into unavailable recalcitrant material, the exponential  
1000 coefficient  $k_1$  describes the initial rapid uptake and respiration of  $^{14}\text{C}$  labelled substrate by the  
1001 microbial community immediately following carbon substrate application, and  $k_2$  describes a  
1002 slower secondary mineralisation phase which we ascribed to immobilisation in microbial  
1003 biomass and transformation of organic metabolites (microbial turnover). The terms  $a_1$  and  $a_2$   
1004 relate to the proportion of  $^{14}\text{C}$  labelled substrate that is associated with each exponential  
1005 coefficient at time ( $t$ ).

1006 The mean residence time (MRT) or substrate half-life ( $t_{1/2}$ ) can be calculated according to

$$1007 \quad t_{1/2} = \frac{\ln(2)}{k_n} \quad \text{eqn. 2.}$$

1008 The turnover of each pool can then be calculated as the inverse of the MRT (1/MRT).

1009

1010 **A:** For substrate mineralisation the equation:  $y = 40.3766 + 48.3216^{0.0230x} + 13.1812^{0.0017x}$  fitted  
1011 the data with an  $r^2$  of 0.99. Using the coefficients ( $k_n$ ) from the fitted equation, the half-life of  
1012 the substrate can be calculated using eqn. 2. Substrate half-life was in the first phase (soil  
1013 solution uptake) 30 h, and the second slower phase (microbial turnover) 408 h. Approximately  
1014 40 % of the substrate was immobilised in the soil, 48.3 % respired during the first phase, and  
1015 13.2 % respired during the slower second phase. **B:** Half-life of the substrate in the soil solution  
1016 versus mean annual temperature, in control (triangle) and warming (circle) treatments, data  
1017 points are mean  $\pm$  SE ( $n=3$ ). Warming treatment and relative warmer site, simply increases the  
1018 catabolic utilisation of labile LMW-carbon until a threshold mean annual temperature of 11.5  
1019  $^{\circ}\text{C}$ .

1020