# Relationships between Substrate Promiscuity and Chiral Selectivity of Esterases from Phylogenetically and Environmentally Diverse **Microorganisms**

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1 Communication

- 2 **Relationships between substrate promiscuity and**
- **3** chiral selectivity of esterases from phylogenetically

# 4 and environmentally diverse microorganisms

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28 Abstract: Substrate specificity and selectivity of a biocatalyst are determined by the protein 29 sequence and structure of its active site. Finding versatile biocatalysts acting against multiple 30 substrates while at the same time being chiral selective is of interest for the pharmaceutical and 31 chemical industry. However, the relationships between these two properties in natural microbial 32 enzymes remain underexplored. Here, we performed an experimental analysis of substrate 33 promiscuity and chiral selectivity in a set of 145 purified esterases from phylogenetically and 34 environmentally diverse microorganisms, which were assayed against 96 diverse esters, 20 of 35 which were enantiomers. Our results revealed a negative correlation between substrate promiscuity and chiral selectivity in the evaluated enzymes. Esterases displaying prominent 36 37 substrate promiscuity and large catalytic environments are characterized by low chiral selectivity, a 38 feature that has limited commercial value. Although, low level of substrate promiscuity does not 39 guarantee high chiral selectivity, the probability that esterases with smaller active sites possess 40 chiral selectivity factors of interest for industry (>25) is significantly higher than for promiscuous 41 enzymes. Together, the present study unambiguously demonstrates that promiscuous and 42 selective esterases appear to be rare in nature and that substrate promiscuity can be used as an 43 indicator of the chiral selectivity level of esterases, and vice versa.

- 44 **Keywords:** esterase; metagenomics; promiscuity; selectivity
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#### 46 **1. Introduction**

47 Presently, there is a great necessity of suitable biocatalysts with high process performance, as 48 greener alternatives to chemical synthesis [1,2]. It is expected that up to 40% of bulk chemical 49 synthesis processes could be substituted by enzymatic catalysis by 2020 [1]. Along with 50 requirements of a technical nature such as process development and optimization, it is however 51 widely recognized that the establishment of enzymatic processes is mainly a problem of finding, 52 optimizing or designing new and/or better performing enzymes. Nature is a rich reservoir from 53 where enzymes can be isolated [3,4], because they are continuously evolving as a consequence of 54 natural selection. Promiscuous enzymes are effective for converting multiple substrates, thus, they 55 are industrially relevant [4-6]. Enzymes need to also be robust and, preferably, chiral selective to 56 reduce raw material costs in the synthesis of pure chiral compounds [1,2,4]. That is, they need to be 57 able to cleave preferentially only one chiral ester when offered a racemic mixture. Is it possible to 58 find versatile enzymes displaying prominent substrate range and stringent chiral selectivity? 59 Evaluating this possibility was the starting point of the present study.

60 In this study we are interested in investigating as model enzymes serine ester hydrolases, 61 hereafter referred to as esterases, from the structural superfamily of  $\alpha/\beta$ -hydrolases. The activity of 62 these esterases relies mainly on a catalytic triad usually formed by Ser, Asp/Glu and His [8]. This 63 enzyme class was selected for a number of reasons: it is widely distributed in the environment, it has 64 important physiological functions, it includes hydrolases that are among the most important 65 industrial biocatalysts, and extensive biochemical knowledge has been accumulated [4,5,7].

66 Just focusing on those from uncultivated microorganisms discovered through metagenomic 67 approaches, esterases with prominent chiral selectivity have been identified and their use in the 68 kinetic resolution of a number of esters reported. Recent examples include those preferably 69 hydrolyzing one of the chiral esters in racemic mixtures of ibuprofen esters [9,10], ketoprofen esters 70 [11-14], solketal of phenylalkyl carboxylic esters [15], esters acids, 71 1,1,1-trifluoro-2-phenylbut-3-yn-2-yl acetate and 3,7-dimethyl-1,6-octadien-3-yl acetate [16,17], 72 methyl 3-phenylglycidate [18], 1-phenylethyl acetate [19,20], ofloxacin butyl ester [21], 1-octin-3-ol, 73 trimethylsilylbutinol, 3-chlor-1-phenyl-1-propanol, cis/trans-1,2-cyclohexanediol and 74 isopropylidenglycerol acetate [22], glycidyl butyrate [23], methyl-mandelate, 75 glycidyl-4-nitrobenzoate, methyl-3-bromo-2-methyl propionate, methyl lactate, menthyl acetate, 76 neomenthyl acetate, pantolactone, and methyl 3-hydroxybutyrate [22,24,25], 1-octin-3-ol, 77 3-chlor-1-phenyl-1-propanol, and trimethylsilylbutinol [22], methyl-3-hydroxy-2-methylpropionate 78 [26], and esters of secondary alcohols [27,28], to cite some. The advances in metagenomics 79 techniques and screening methods have allowed the discovery of these and other selective esterases 80 [29]. These studies exemplify that esterases with selective character occur naturally, and that their 81 chiral preference depend on structural factors in the proximity of the active-site. However, whether 82 the selective character of these esterases, and many others, is linked to a broad or a narrow substrate 83 spectrum has not been investigated, due to limited substrate sets employed.

Here we investigate the relationships between the level of substrate promiscuity and chiral selectivity of a large set of 145 phylogenetically and environmentally diverse microbial esterases, whose specific activity against 96 distinct esters that included 20 chiral esters have been recently reported [5]. We provide unambiguous experimental evidences suggesting a negative association between substrate specificity and chiral selectivity in native esterases.

### 89 2. Results and Discussions

#### 90 2.1. Relationships between substrate promiscuity and chiral selectivity

We have recently described an extensive analysis of the substrate spectra of 145 phylogenetically and environmentally diverse microbial esterases [5]. Experimental data on substrate conversion (i.e., units g<sup>-1</sup> or U g<sup>-1</sup>) followed for 24 h, at pH 8.0 and 30°C was reported for 96 distinct esters. They included esters with variation in size of acyl and alcohol groups and with 95 growing residues (aromatic, aliphatic, branched and unbranched), halogenated esters, sugar esters, 96 lactones, an alkyl di-ester, and 20 chiral esters (including (*R*) and (*S*) enantiomers of menthyl acetate, 97 N-benzyl-proline ethyl ester, methyl mandelate, ethyl 4-chloro-3-hydroxybutyrate, methyl 98 3-hydroxybutyrate, methyl 3-hydroxyvalerate, neomenthyl acetate, methyl and ethyl lactate, and 99 pantolactone). By meaning of the partitioning coefficient (log P value), which reflects electronic and 91 steric effects and hydrophobic and hydrophilic characteristics, the 96 esters do show a broad 92 chemical and structural variability [5]. This chemical variability characterized also the chiral esters 93 tested (Figure 1).



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Figure 1 Representative chemical structures of 20 chiral esters used to evaluate chiral selectivity.

105 To find the relationships between substrate promiscuity and chiral selectivity we calculated the 106 chiral selectivity factor for each of the 145 esterases and the 20 chiral esters tested (i.e. two 107 enantiomers per pair) that were included in the 96-ester library (Figure 1). Selectivity factor was 108 calculated as the ratio of specific activity (U  $g^{-1}$ ) of the preferred over the non-preferred chiral ester 109 when both esters were tested separately (see Materials and Methods). These calculations were 110 extracted from datasets reported previously [5]. It should be mentioned that these apparent values 111 may not correspond to true selectivity or enantiomeric factors calculated when the enzyme is 112 confronted to a racemic mixture, because the rates of hydrolysis of the enantiomers were measured 113 separately; nevertheless, recent studies have clearly demonstrated that apparent and true selectivity 114 values closely match each other [15]. These values were plotted against the number of esters 115 hydrolyzed by each of the esterases (Figure 2), previously reported for each of the esterases [5].

From the 145 esterases, 40 did not show appreciable activity under assay conditions for any of the chiral esters tested. From those being active against at least one of the chiral esters (105 in total), 80 esterases were characterized by selectivity factors below a threshold of 25. Although esterases with stringent selectivity are preferred, it is commonly considered that enzymes with selectivity factor of 25 or above begin to have commercial value [31]. On the other hand, we found 25 chiral

121 selective esterases, as jugded by a selectivity factor above 25 (Figure 2; Figure 3). Ten of them

- 122 showed stringent selectivity, that is they were capable of hydrolyzing only one of the enantiomer 123 (Figure 3). Twelve of them were characterized by selectivity factors ranging from 25.9 to 59.3, and
- 124 three did show prominent selectivity factors ranging from 219 to 686 (Figure 2; Figure 3).

125 As shown in Figure 2, we found a negative association between the level of substrate 126 promiscuity, by meaning of the number of esters hydrolyzed, and the chiral selectivity factor. More 127 in details, according to criteria previously established [5] we considered an esterase specific if it 128 hydrolyses 9 esters or fewer, as moderate-to-highly promiscuous if it hydrolyses between 10 and 42 129 esters, and as prominently promiscuous if it hydrolyses 42 or more esters. None of the 25 hydrolases 130 which showed a selectivity factor  $\geq$  25 were prominently promiscuous. Rather, they were capable of 131 accepting 36 or fewer substrates. However, not all hydrolases converting 36 or fewer esters and 132 acting against chiral esters were selective according to the 25-selective factor threshold. Indeed, only 133 25 out of 85 in total (or 29%) were selective, with different selectivity factors and chiral preferences 134 (Figure 3). This is most likely due to the fact that the ability to selectively hydrolyze an enantiomer in 135 a racemate may depend on the topology of the catalytic environment [5].



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137 Figure 2 Chiral selectivity factor vs number of esters hydrolyzed per each of the 145 hydrolases 138 tested. Selectivity factor was calculated per each pair of enantiomers as the ratio of specific activity 139  $(U g^{-1})$  of the preferred over the non-preferred chiral ester when each of the chiral esters was tested 140 separately. Chiral esters are color coded. The value 100 was arbitrarily given to represent those 141 esterases capable of hydrolyzing, under our assay conditions, only one of the enantiomers (100% 142 selective) and those with selectivity factors higher than 100. These data are based on the data 143 reported previously [5], using conditions described in Materials and Methods. The level of 144 promiscuity, according to criteria previously established [5], is marked under a shadowed grey 145 background. The 25-selectivity factor threshold at which an esterase started to have commercial 146 value is indicated by a horizontal dashed gray line.



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148 Figure 3 Chiral preferences of 25 hydrolases which were found to be selective for at least one chiral 149 ester according to the 25-selective factor threshold. The figure illustrates the specific activity (U g<sup>1</sup>; 150 represented by the size of the circles) of each esterase per each of the 20 chiral esters tested. The ID 151 code for each esterase (for full description see ref. [5]) is shown on the top; the number of esters (out 152 of 96 tested) hydrolyzed by each esterase is shown in brackets. The Figure was created with the R 153 language console from data previously reported [5]. The list of the 20 chiral esters tested is shown on 154 the left, with (R)-enantiomer in blue and (S)-enantiomer in red color. The protocol established and 155 used to identify the esters hydrolyzed by each esterase is described in Materials and Methods.

# 156 2.2. Occurrence of multi selective esterases

157 Figure 3 summarizes the chiral preference of esterases fitting to the 25-selective factor threshold 158 for each of the 20 chiral esters tested. As can be seen in Figure 3 esterases showed different level of 159 promiscuity and preference for (R) or (S) methyl acetate, menthyl mandelate, methyl 160 3-hydroxybutyrate, N-benzyl-proline ethyl ester, ethyl-4-chloro-3-hydroxybutyrate, and (m)ethyl 161 lactate. Esterases selective for (R) or (S)-pantolactone, N-benzyl-proline ethyl ester and neomenthyl 162 acetate were the less abundant, suggesting these chiral esters are less preferred substrates. As 163 shown in Figure 3, we also found that 5 out of 25 esterases fitting to the 25-selective threshold did 164 show stringent selectivity or selectivity factor higher than 25 for several chiral esters differing in 165 chemical and structural nature (Figure 3). They include one being selective for methyl 166 3-hydroxybutyrate (*R*-selective) and ethyl lactate (S-preference) (EH31), for one 167 ethyl-4-chloro-3-hydroxybutyrate (R-selective), methyl 3-hydroxybutyrate (R-selective) and ethyl 168 lactate (S-selective) (EH47), one for methyl mandelate (S-selective), methyl 3-hydroxyvalerate 169 (S-selective) and methyl lactate (S-selective) (EH71), and two for menthyl acetate (R-selective) and 170 methyl lactate (S-selective) (EH72 and EH78). The other 20 esterases did show the capacity to 171 preferentially hydrolyze only one enantiomer (Figure 3). This suggests that multi selective esterases 172 may have a lower abundancy.

# 173 3. Materials and Methods

#### 174 3.1. Source of chemicals, enzymes and datasets

All chemicals for which activity data are reported were of the purest grade available and were purchased as reported [5]. The present study used datasets of hydrolytic activity (U g<sup>-1</sup>) for 145 esterases assayed at 550 nm using 96-structurally diverse esters in 384-well plates. Reactions were followed for 24 h, at pH 8.0 and 30°C. Datasets are available elsewhere [5].

#### 179 3.2. Selectivity factor calculation

180 The chiral selectivity factor is defined as the ratio of the specific activity [30] for each 181 enantiomer, measured separately as described previously [5]. Briefly, reaction mixture contains 5 182 mM N-(2-hydroxyethyl)piperazine-N'-(3-propanesulfonic acid buffer, pH 8.0, 4.5% (v/v) acetonitrile 183 or dimethyl sulfoxide, 0.45 mM Phenol Red (used as a pH indicator), a concentration of each of the 184 esters of 1.14 mg ml<sup>-1</sup>, and 2 µg of proteins. Reactions were allowed to proceed kinetically at 30°C 185 and hydrolytic activity (U g<sup>-1</sup>) calculated followed for 24 h [5]. Selectivity factor was calculated 186 considering the preferred over the non-preferred chiral ester, whatever the preferred (*R*) or (*S*) ester.

# 187 4. Conclusions

188 Herein we show the value of the systematic investigation of enzyme activity to deepen our 189 understanding of the relationships between substrate promiscuity and chiral selectivity. By 190 comparing the number of esters that 145 diverse esterases hydrolyze as an indicator of the substrate 191 promiscuity level, and their selectivity factors as an indicator of enantio-selectivity, we found 192 unambiguous evidences that esterases with broad substrate spectra do commonly show low 193 selectivity for chiral molecules. In this study, the proportion of esterases with both prominent 194 promiscuity and selectivity approaches zero percent. By contrast, the proportion of esterases with 195 low to moderate promiscuity but prominent selectivity was as high as 29%. This suggests that the 196 substrate promiscuity may be used as an indicator of the selective character of esterases. 197 Promiscuous esterases acting against multiple substrates, while at the same time being 198 enantio-selective appear to be rare in nature, or at least in the habitats from where the esterases 199 herein described were isolated [5]. As these enzymes are of interest for application purposes [1-6,32], 200 protein engineering and rational design may be needed to obtain esterases being promiscuous and 201 selective for industrial applications. We anticipate that the possibility to transform a promiscuous 202 but not selective esterase into an efficient enantio-selective biocatalyst would require less 203 engineering effort because increasing the selectivity for an enantiomer may involve a reduced 204 number of contacts close to the active sites [for a recent example see ref. 33]. Opposite, increasing the 205 substrate spectra of a selective non-promiscuous esterase would require large rearrangement of the 206 catalytic environment which may, at the same time, result in significant reduction or even loss of 207 enantioselectivity. This is because non-promiscuous esterases are characterized by catalytic 208 environments that are highly exposed and have small volumes, while an esterase for being 209 promiscuous requires a large active site volume and lower relative solvent accessible surface area 210 [5], that are difficult to be designed through few mutations.

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