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## Mechanically agitated calorimetric cells working under pressure at macro and micro scale: application to gas hydrates

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## Abstract

Originally applied to fields related to oil and gas production and flow assurance, high pressure differential scanning calorimetry (HP-DSC) has now been involved in several new studies such as carbon dioxide sequestration by  $CO_2/CH_4$  exchange in naturally occurring gas hydrates or  $CO_2$  hydrate reversible formation/dissociation for refrigeration loops. However, the technique still has some limitations, which are linked to the fact that the gas hydrate formation occurs at the gas/liquid interface, and because the hydrate nucleation can be rather difficult in small volumes especially in quiescent conditions. It leads to several problems such as inefficient gas dissolution, long induction times, formation of a hydrate crust covering the gas/liquid interface, low hydrate to water conversion, etc. As a result, it is very difficult to determine accurately the heat capacities and the kinetics of formation/dissociation of several systems involving gas hydrates.

This study presents two prototypes of calorimetric cells equipped with an in-situ mechanical agitation system, which allow performing experiments under pressure (150 bar maximum for the cells used in this work). The first system presented, called MIXCEL®, was developed for macro-calorimetry analysis (experiments carried out with a BT 2.15 Calvet Calorimeter from SETARAM Instrumentation). Very recently, we have developed a novel prototype of micro-calorimetric agitated cell (called MICROMIXCEL®) for microDSC analyses (experiments carried out using a microDSC7 evo from SETARAM Instrumentation). Both technical aspects, and results obtained at macro and micro scales with gas hydrate systems are presented and discussed.

For MIXCEL®, the cell body, in which the sample can be inserted, is a 17mm outer diameter stainless steel cylinder with an internal volume of approximatively 7 cm<sup>3</sup>. Mechanical stirring in the measuring cell is provided by a screwable stirrer. It is interchangeable and several stirrer designs can be used. For this study, a three-staged impeller composed by three 2-blades curved turbines was used, which can rotate up to 500 RPM. The link between the wellhead and the measuring cell is provided by two concentric tubes allowing the stirring rod to rotate freely in the tubes. On the upper part of the system, a magnetic coupling allows linking a DC motor (ATEX standards) to the shaft. The fluids (i.e. gases) are injected in or out of the cell by means of two nozzles located on the sides of the upper part.

For MICROMIXCEL®, the measuring cell is a HP cell made of stainless steel with a 0,33 cm<sup>3</sup> internal volume. A mechanical stirrer, which is a custom type anchor, is screwed to the shaft and can rotate inside the cell at several hundreds of RPM. The measuring cell is connected to the upper part of the system by HP tubing. The agitator shaft is located in this tubing. Two nozzles are inserted on the upper part to allow for gases inlet and outlet. Note that, in

contrast with the MIXCEL®, only one HP tube is present around the shaft (no concentric tubes) for the inlet and outlet gas circulation. A small DC motor equipped with a hand-made magnetic coupling allows the rotation of the agitator.

For the two agitated cells, and for several complex systems such as ice slurries or clathrate compounds (i.e. CO<sub>2</sub> hydrates, mixed THF-CO<sub>2</sub> hydrates), we have demonstrated by performing experiments in comparison with the same cells without agitation, that the effect of the in-situ agitator reduces the crystallization metastabilty, increases the water to hydrate conversion and lowers efficiently the time for analysis. We have showed the potential of such technologies for thermodynamic studies by obtaining very easily hydrate phase equilibrium data under pressure. We have also performed a detailed study of the cyclopentane (CP) hydrate, as this system is very hard to analyze with non-agitated calorimetry techniques due to very low water to hydrate conversion. Interestingly, we have obtained, both at macro and micro scales, a very robust value of the phase change enthalpy of this hydrate.

The use of such novel calorimetric cells thus opens a wide range of possibilities for complex systems, such as gas hydrates, which must be analyzed under both pressurized and agitated conditions.