

9-16-2016

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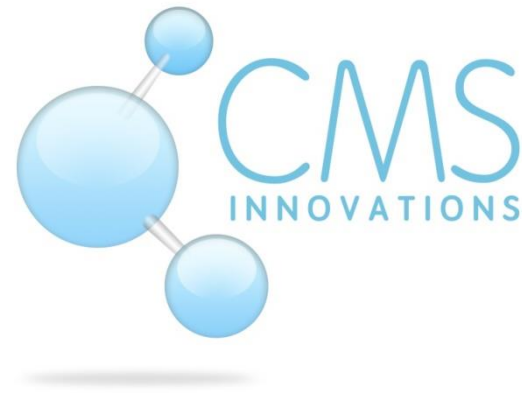
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Recommended Citation

Hans G. L. Coster, Jia Shin Ho, Lee Nuang Sim, and Anthony G. Fane, "In-situ monitoring of RO membranes using electrical impedance spectroscopy: Threshold fluxes and fouling" in "Advanced Membrane Technology VII", Isabel C. Escobar, Professor, University of Kentucky, USA Jamie Hestekin, Associate Professor, University of Arkansas, USA Eds, ECI Symposium Series, (2016). http://dc.engconfintl.org/membrane_technology_vii/43

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***In-situ* Monitoring of RO Membranes using Electrical Impedance Spectroscopy: Threshold fluxes and Fouling**



Hans G. L. Coster



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School of Chemical and
Biomolecular Engineering**

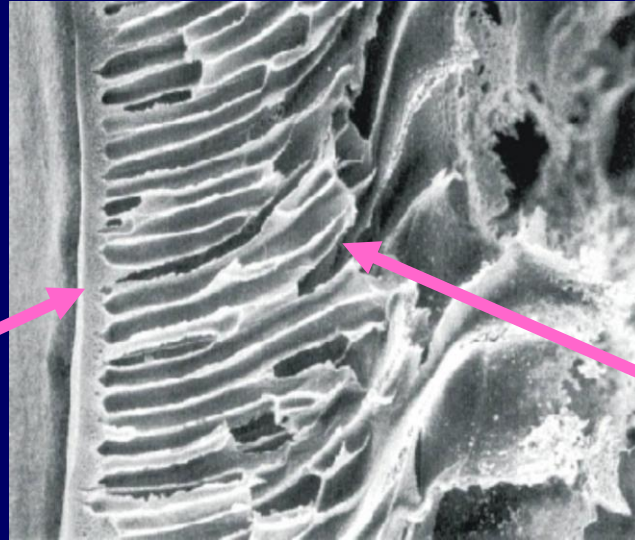
In collaboration with



**Singapore Membrane Technology
Centre**

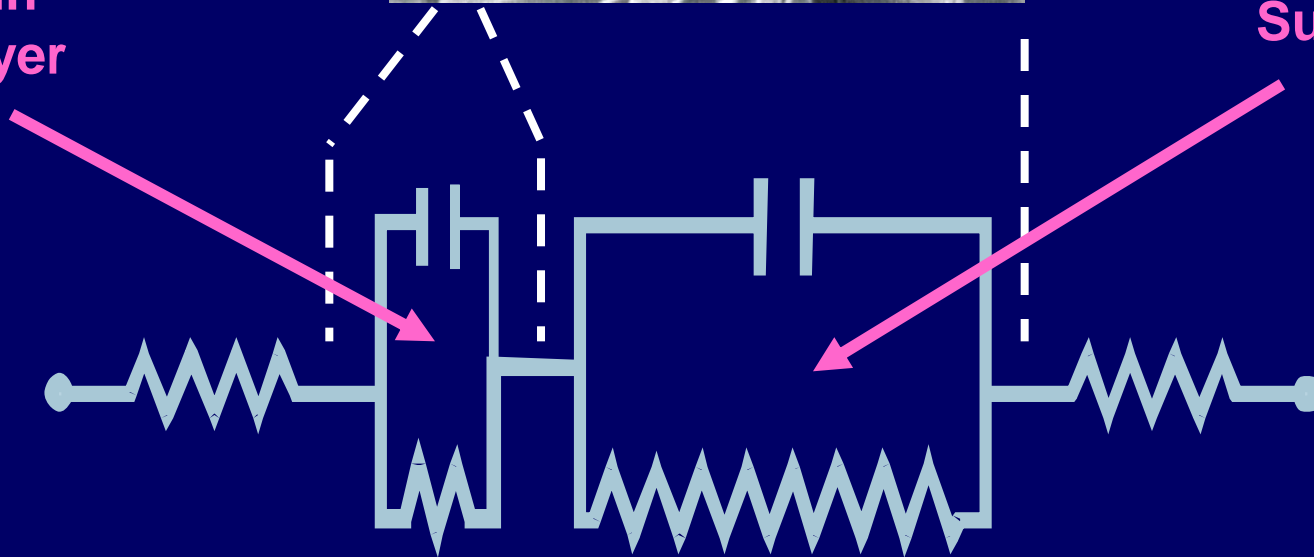


Membrane Dielectric Structure



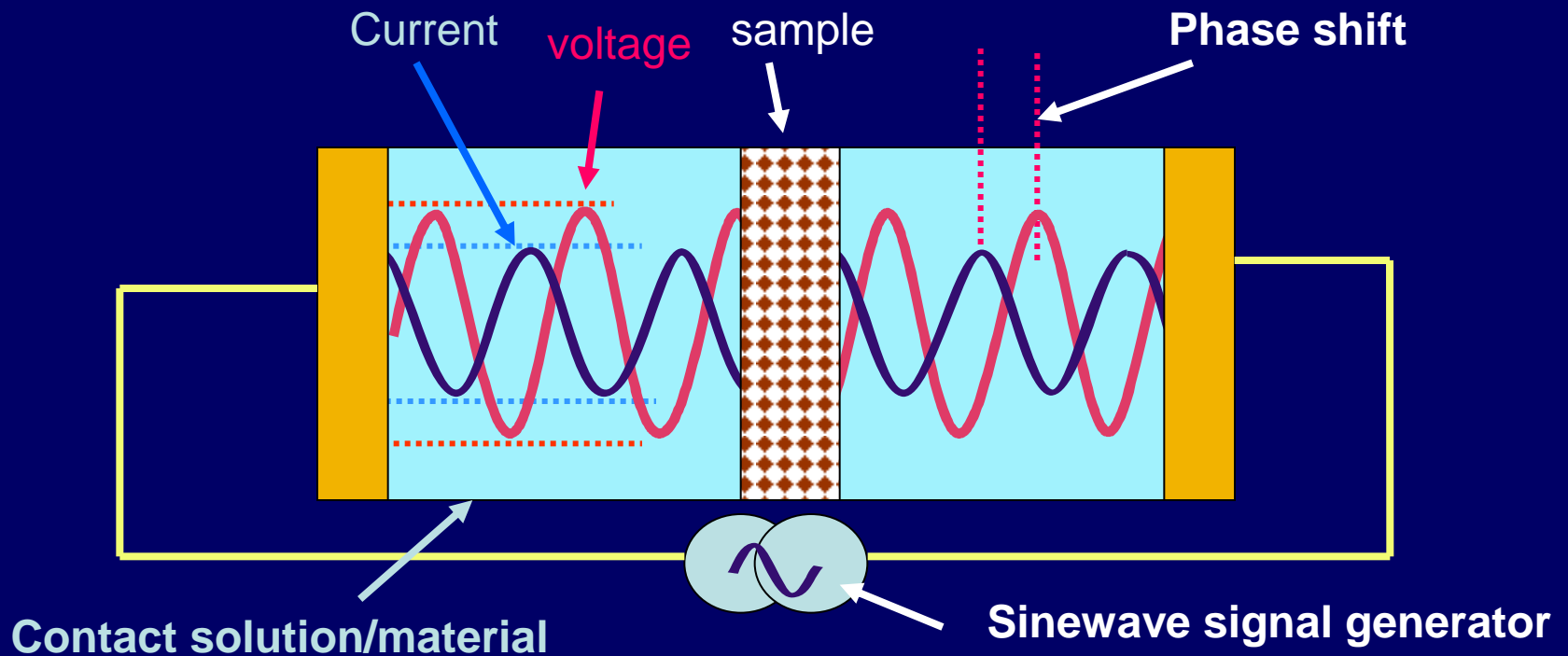
Skin Layer

Sub layer

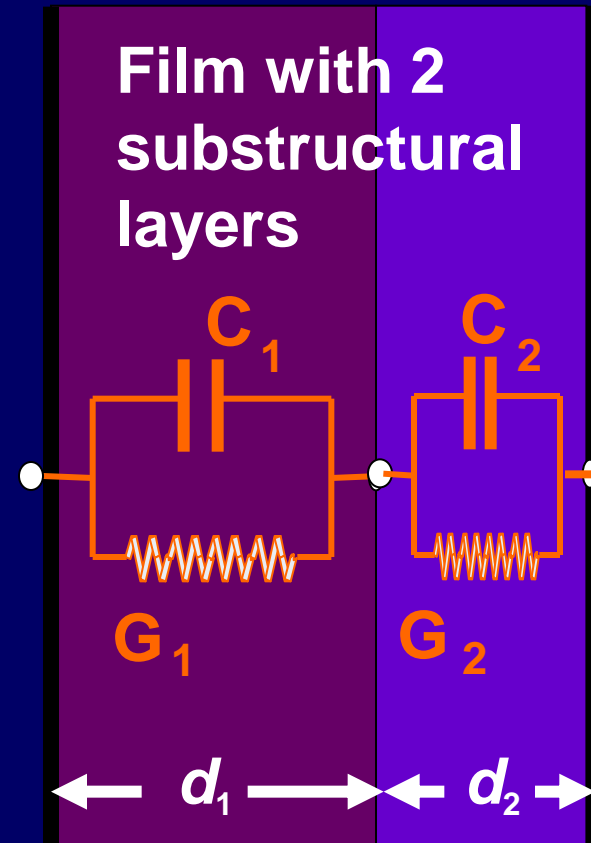
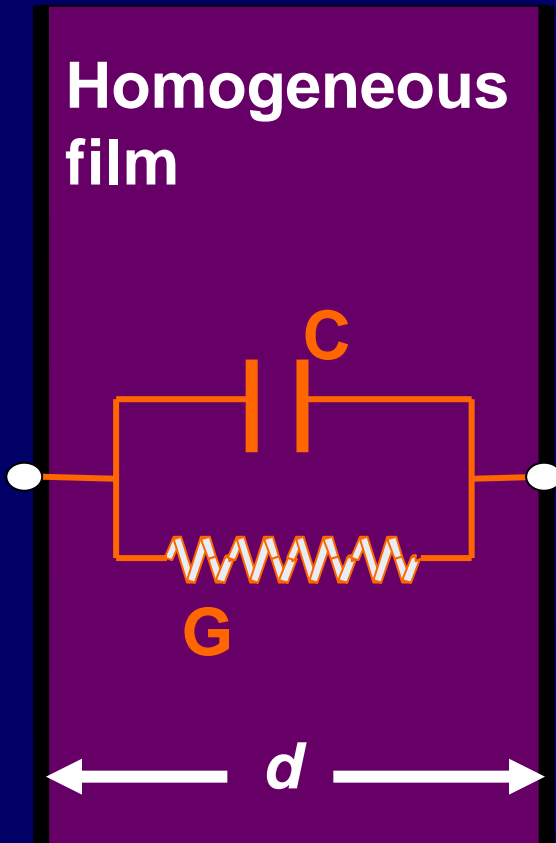


What is involved in Impedance Spectroscopy ?

- Injection of sinusoidal AC currents through the membrane
- Measurement of the current and voltage across the membrane
- Measurement of the phase shift between the voltage & current



Electrical representation of films



$$C_1 = 0.006 \text{ F/m}^2$$

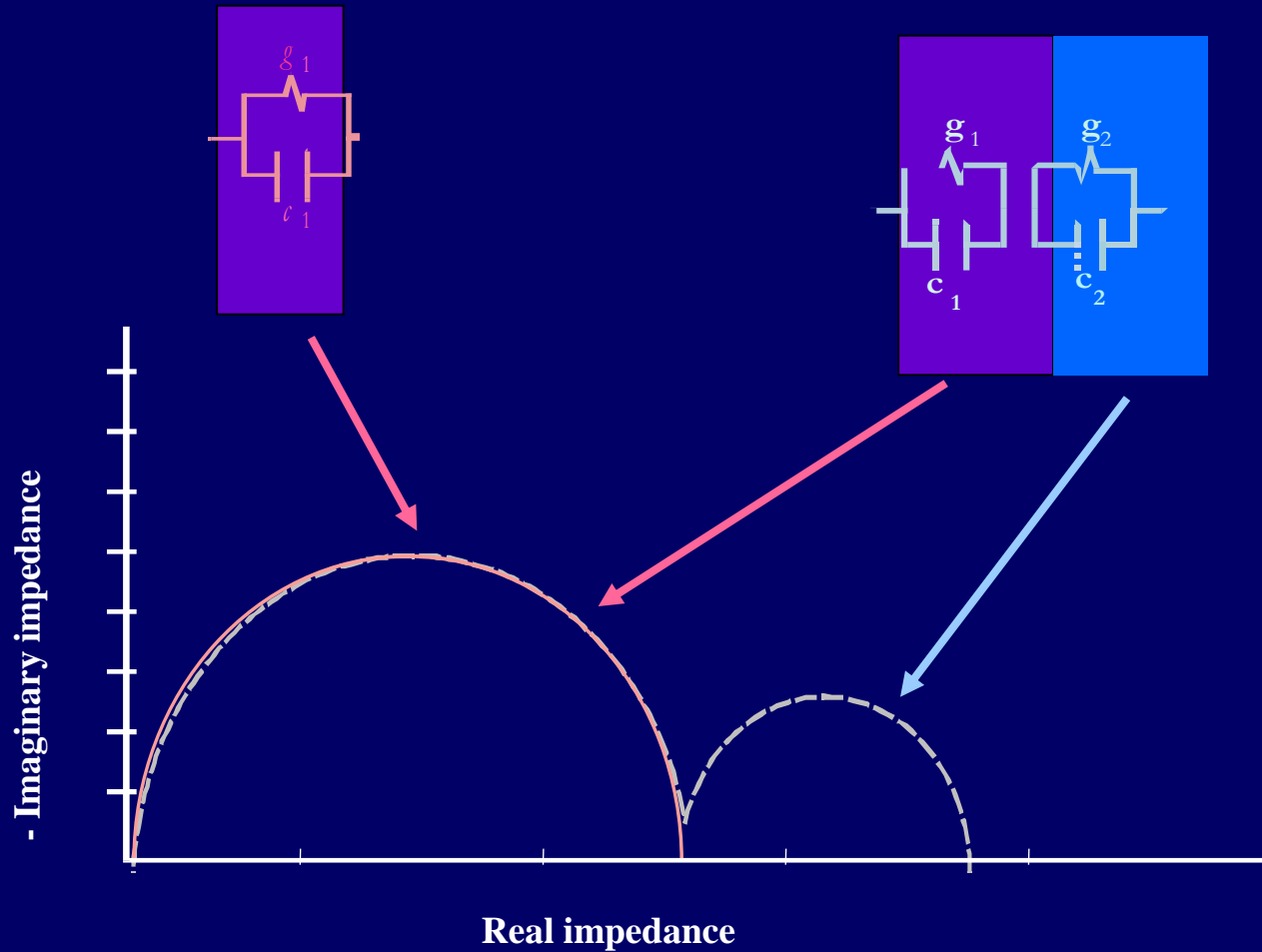
$$C_2 = 0.059 \text{ F/m}^2$$

$$G_1 = 0.003 \text{ S/m}^2$$

$$G_2 = 4.35 \text{ S/m}^2$$

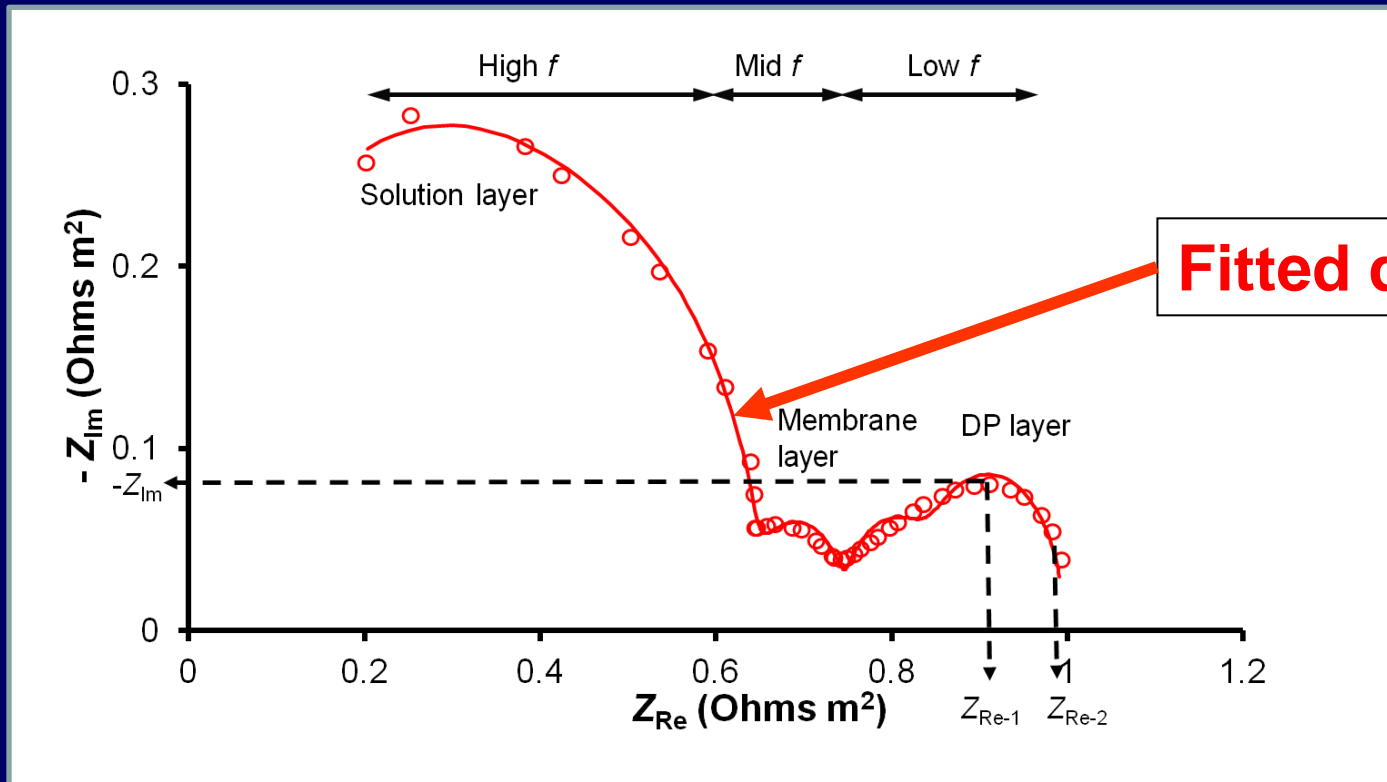
Nyquist Plots

Help to discern various processes and layers with different time constants



additional arcs appear for each element with a different time constant

EIS Membrane characteristics



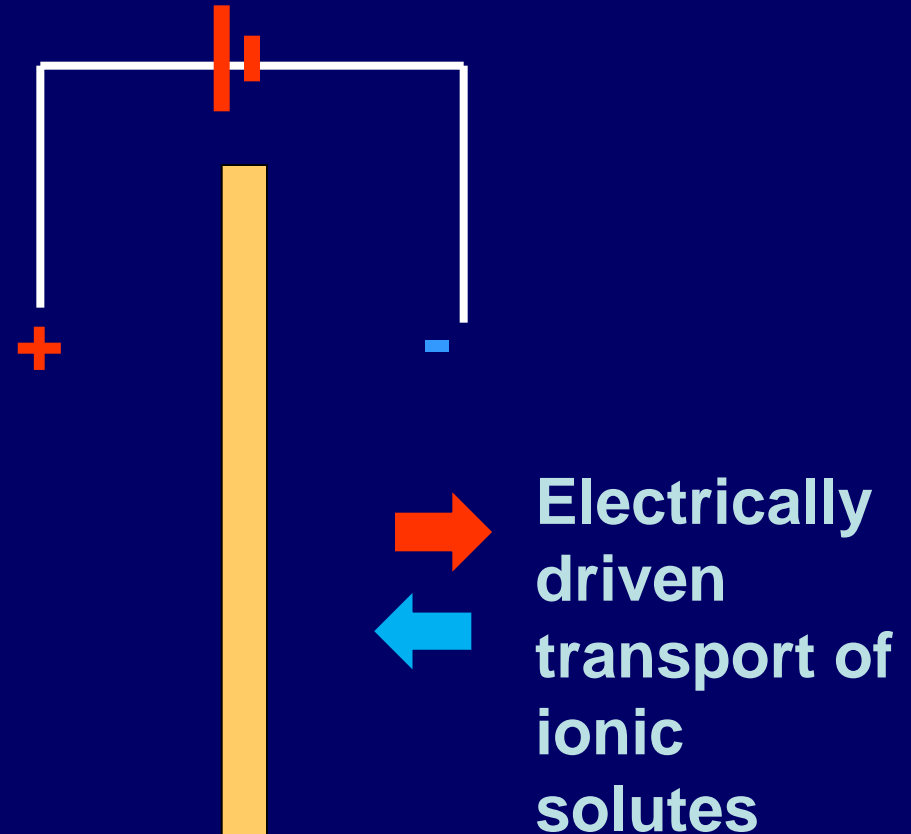
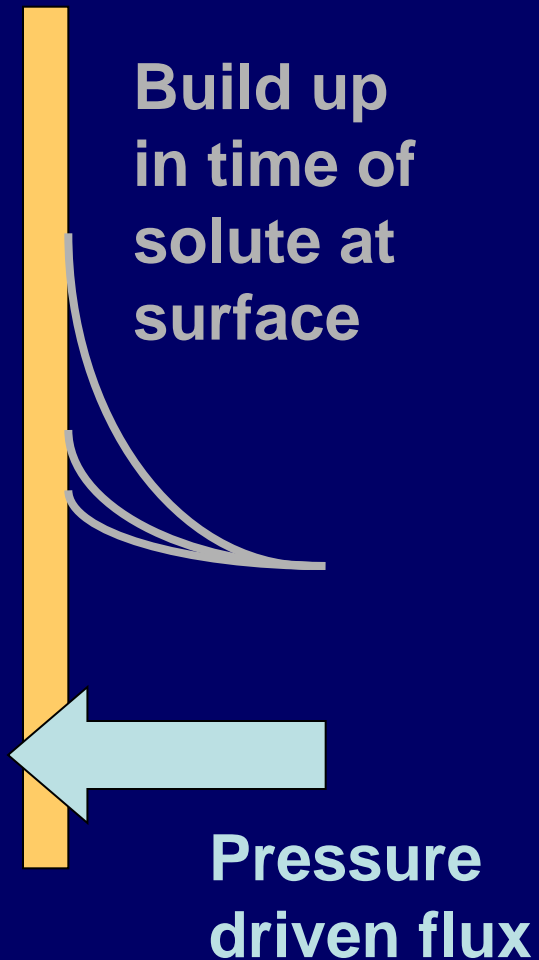
Each dielectric element or transport process will have a characteristic electrical time-constant.

These various elements/processes can be readily distinguished in a Nyquist plot of the imaginary vs the real impedance.

Data for RO with NaCl 2000ppm and silica 200 ppm, crossflow 0.15 m/s: from Ho, Sim, Gu, Webster, Fane & Coster (2015) .

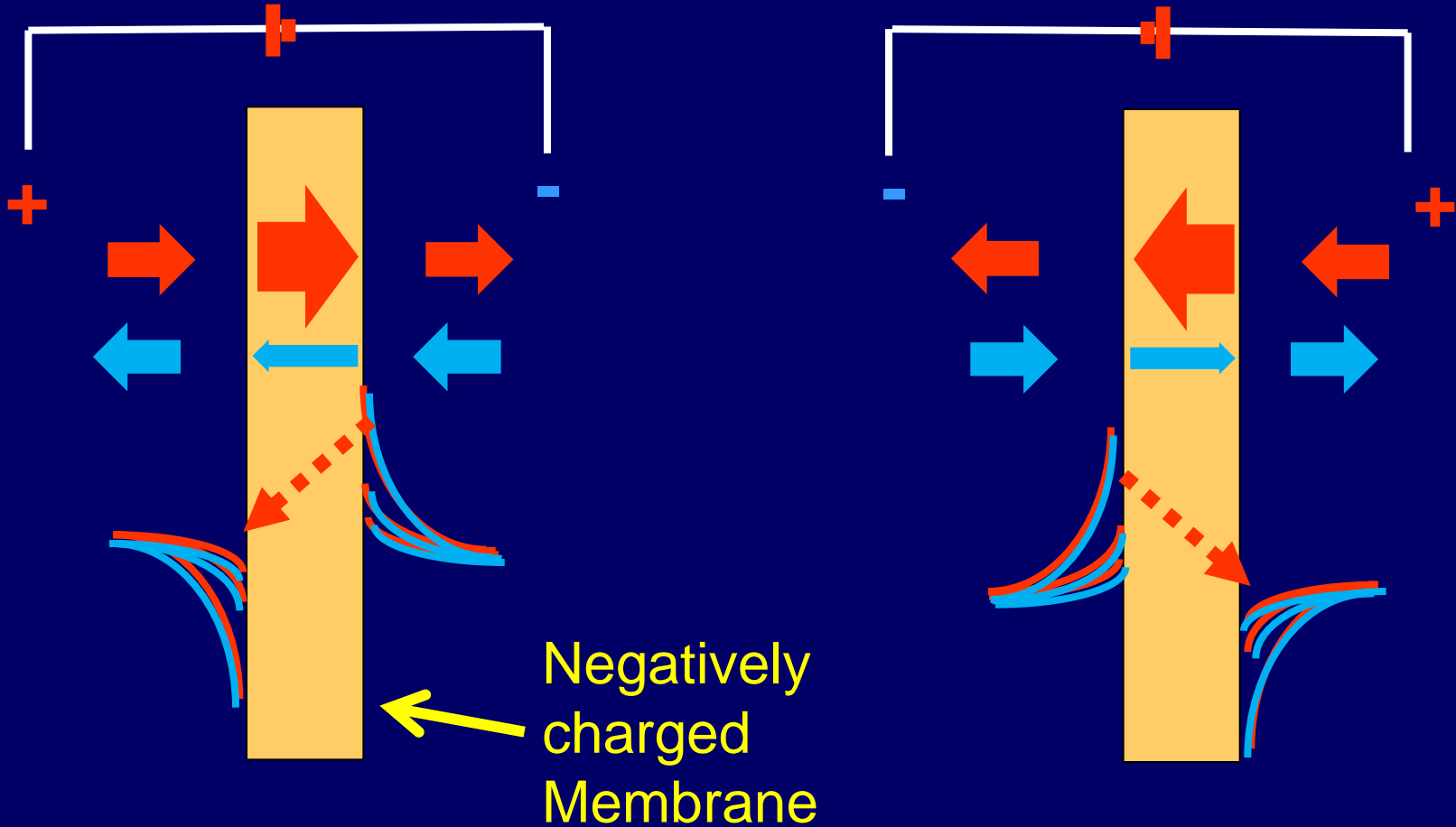
Non-linear Effects: Concentration polarization

membrane



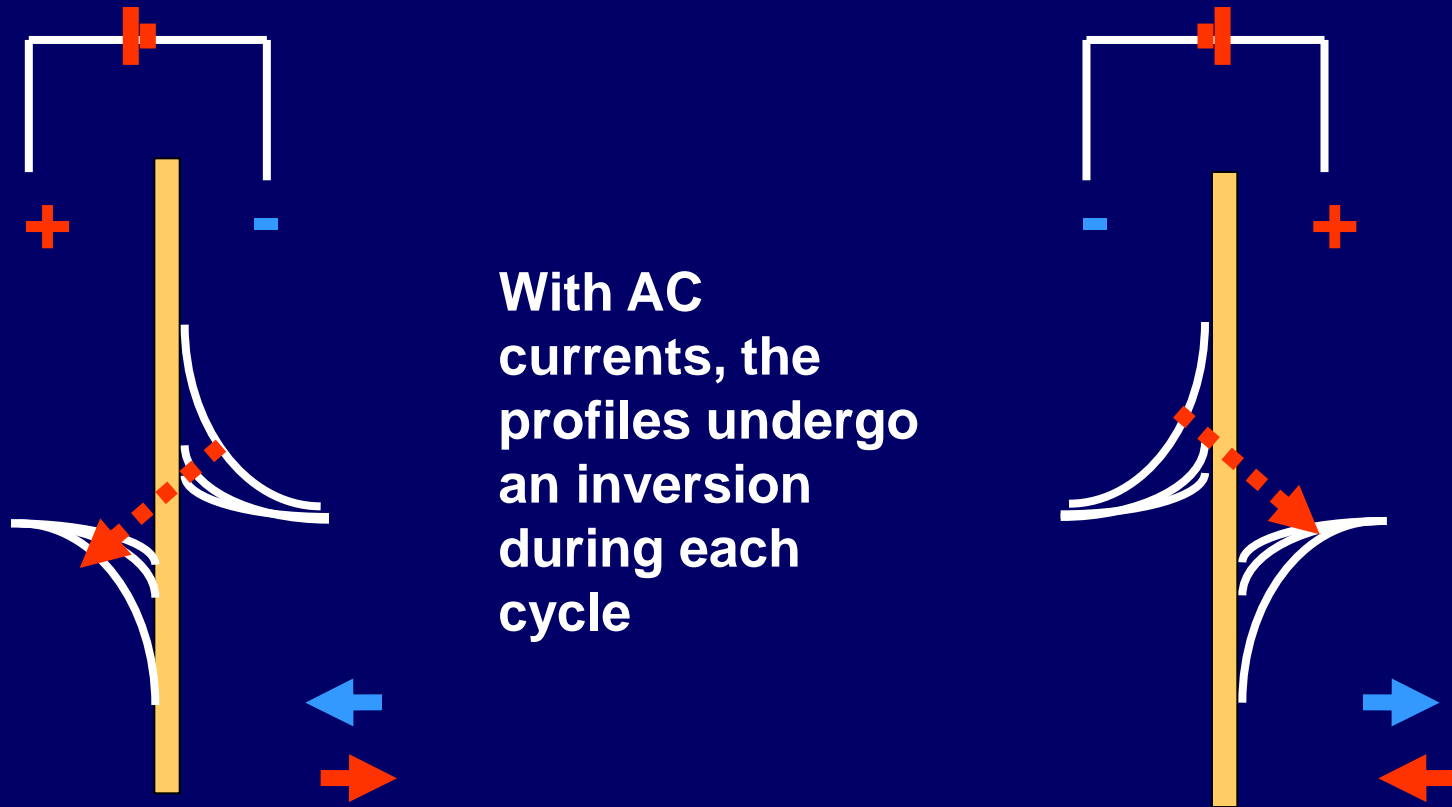
Ion concentration
profiles will depend on
the transport numbers

Electrical Diffusion Polarization



The electric potential due to the back diffusion is in the same direction as the driving potential

AC Diffusion Polarization Effects

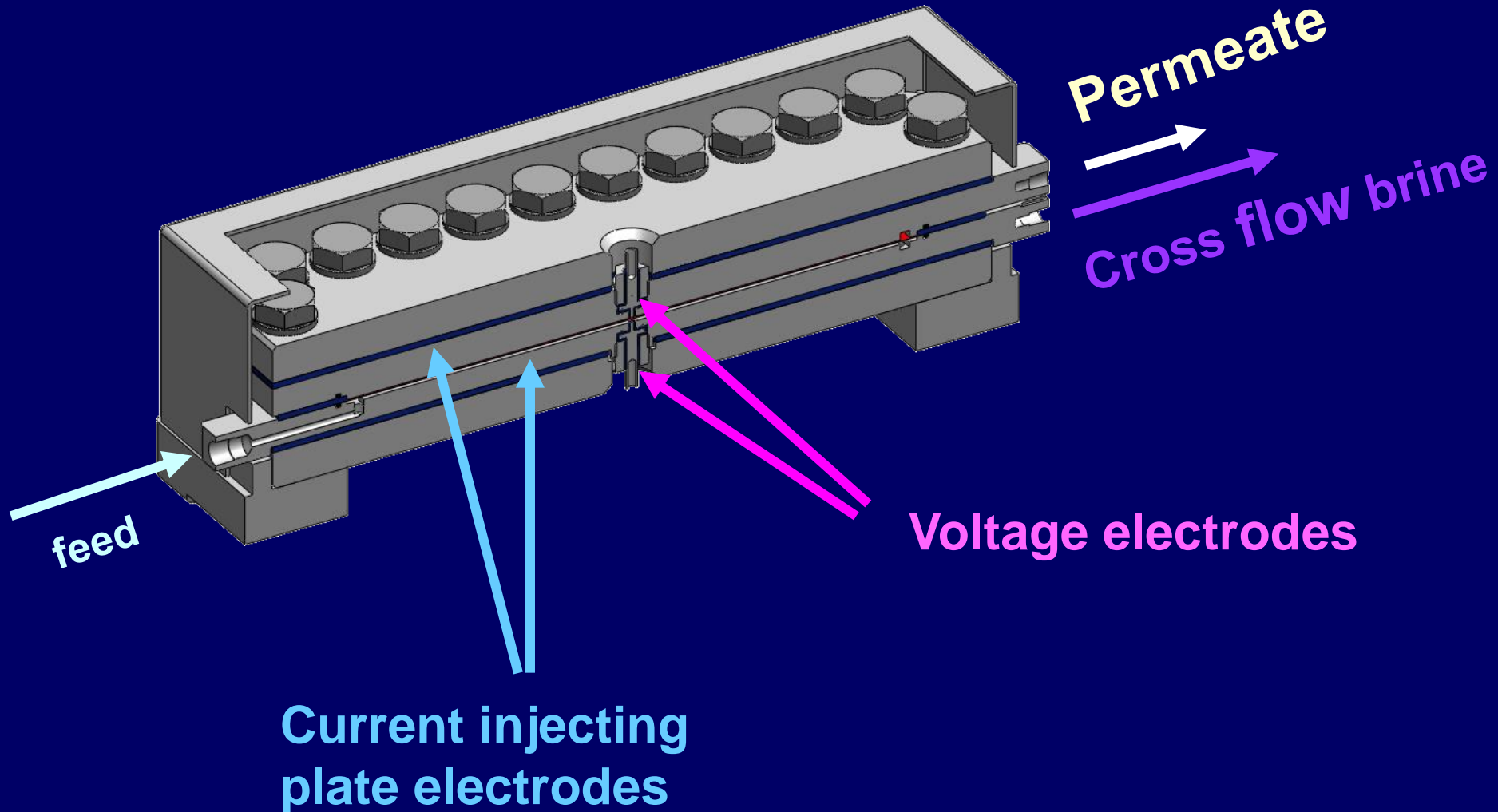


At very high frequencies there is insufficient time for concentration polarization to manifest

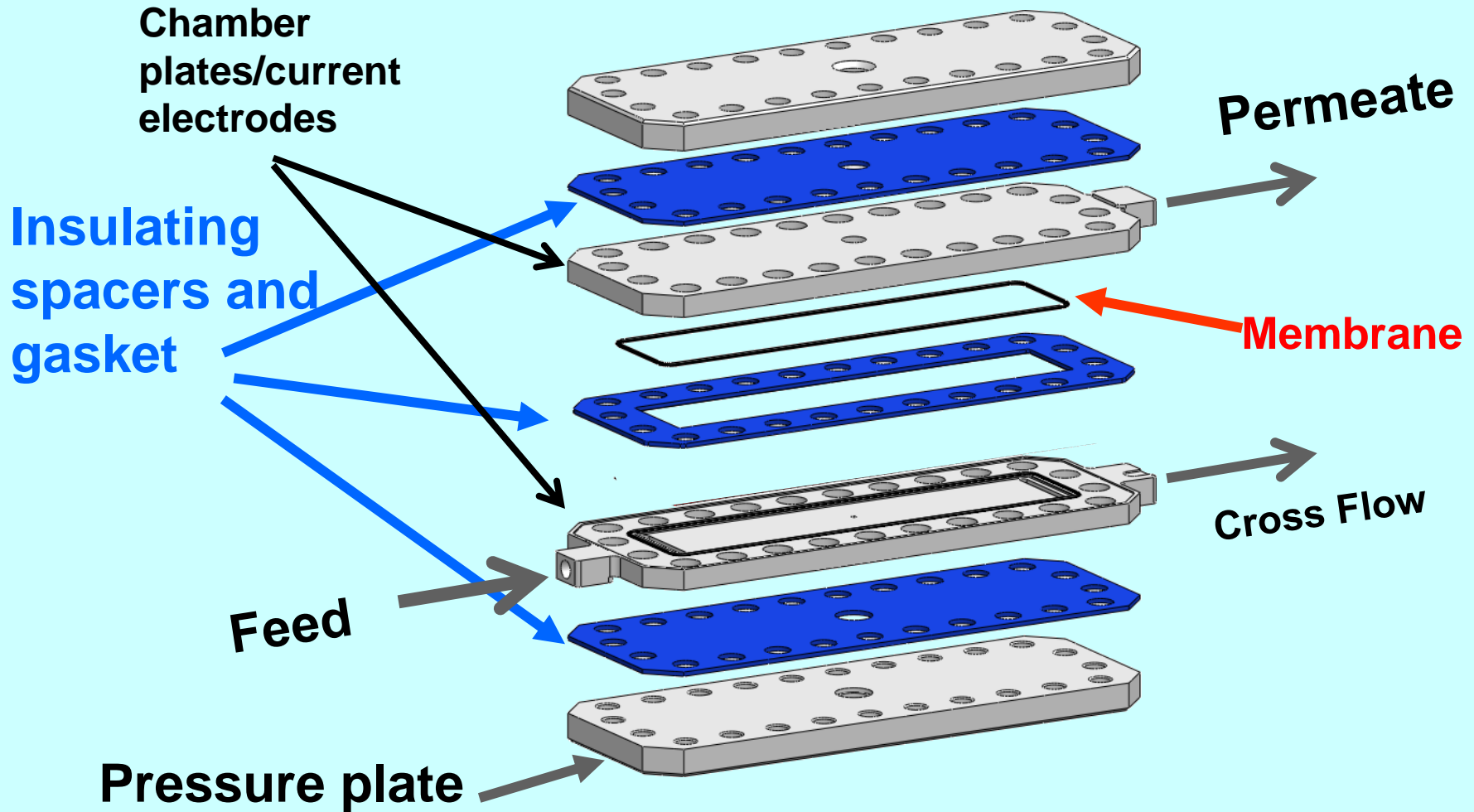
At low frequencies the concentration polarization will be much larger than at high frequencies of the AC.

Impedance Monitoring Cross-flow Module

Cross section of module

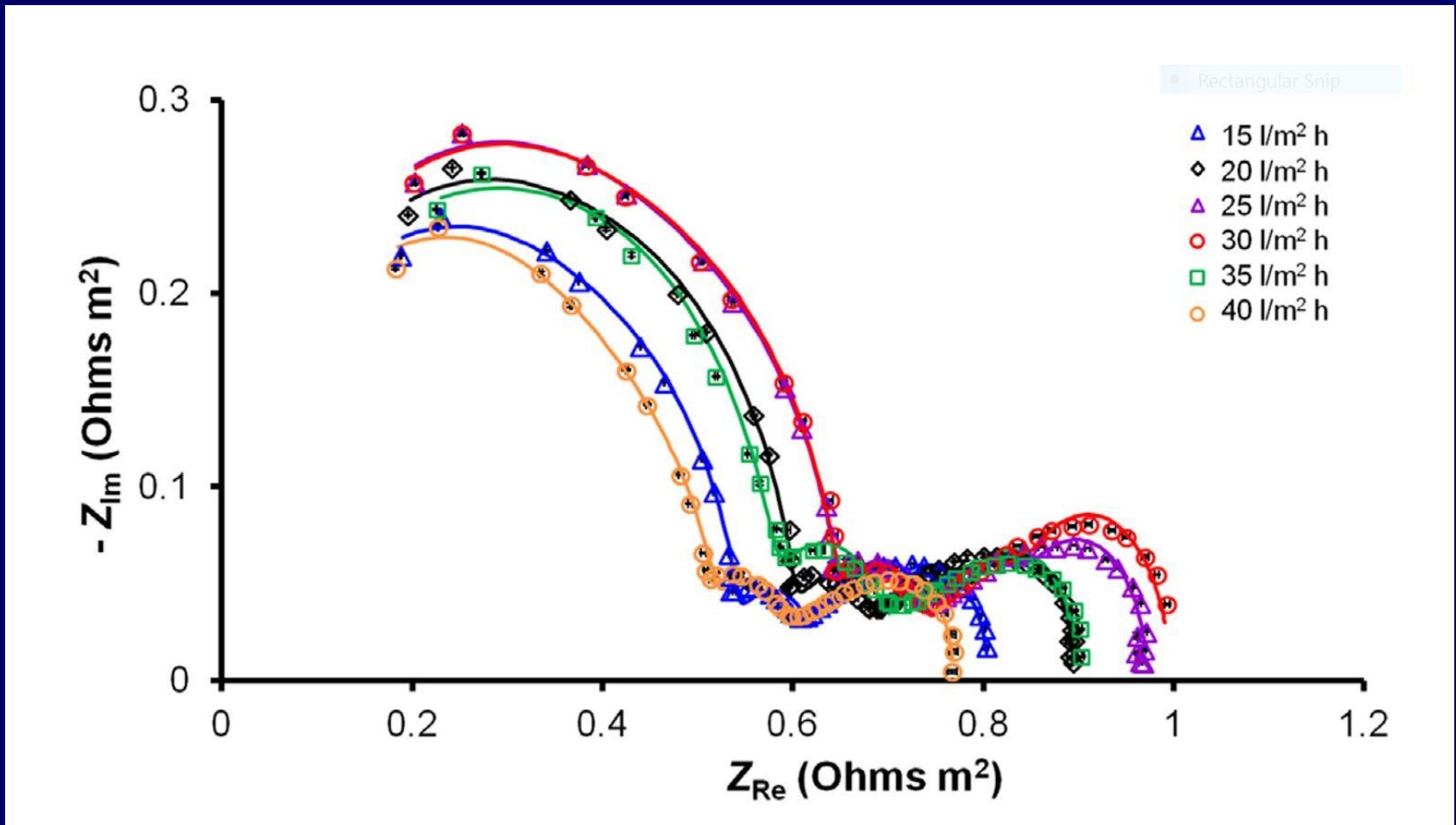


An Impedance Cross-flow Module for monitoring membrane fouling *in situ*



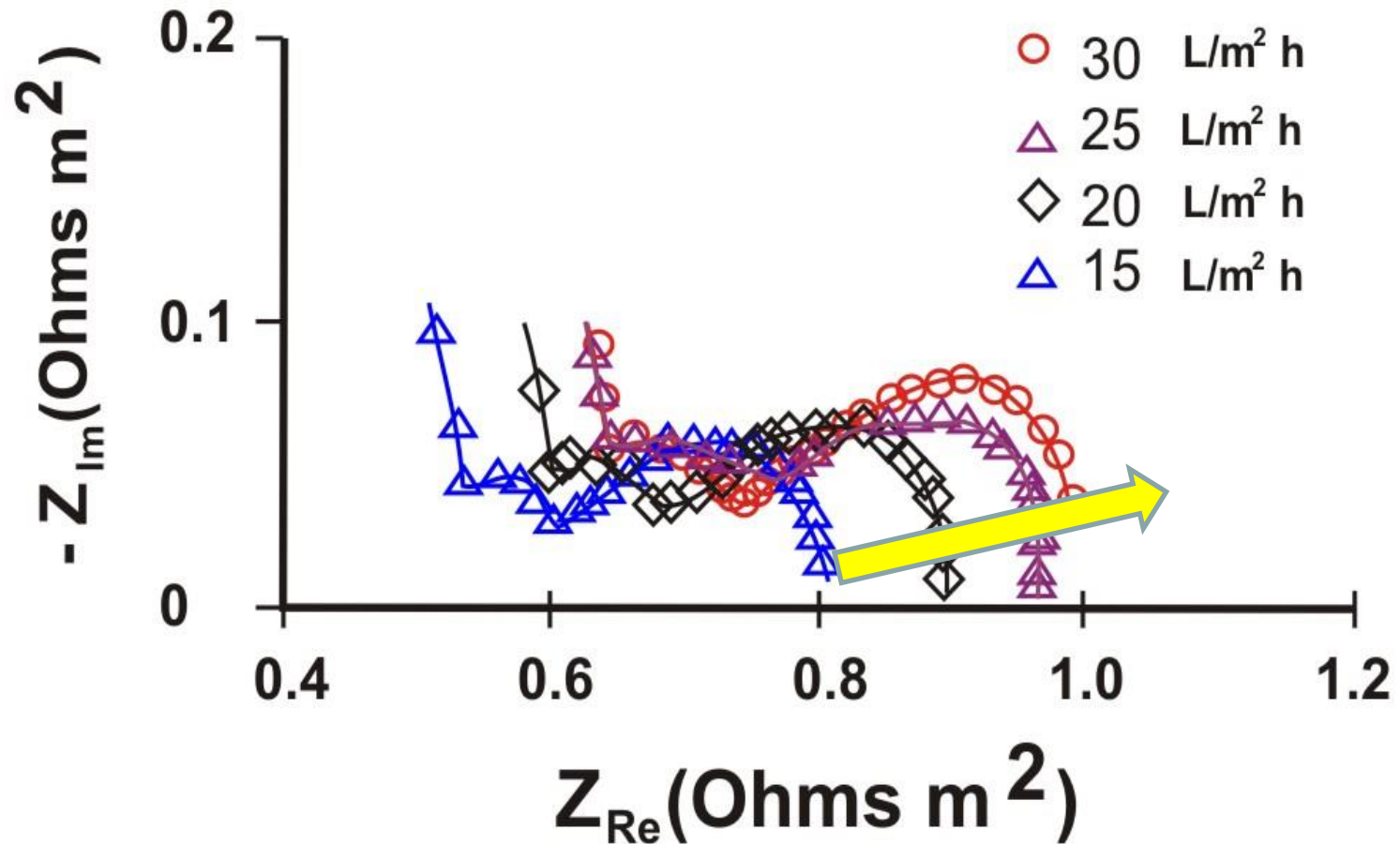
Exploded view of internal plates and gaskets.

EIS: Effect of Flux

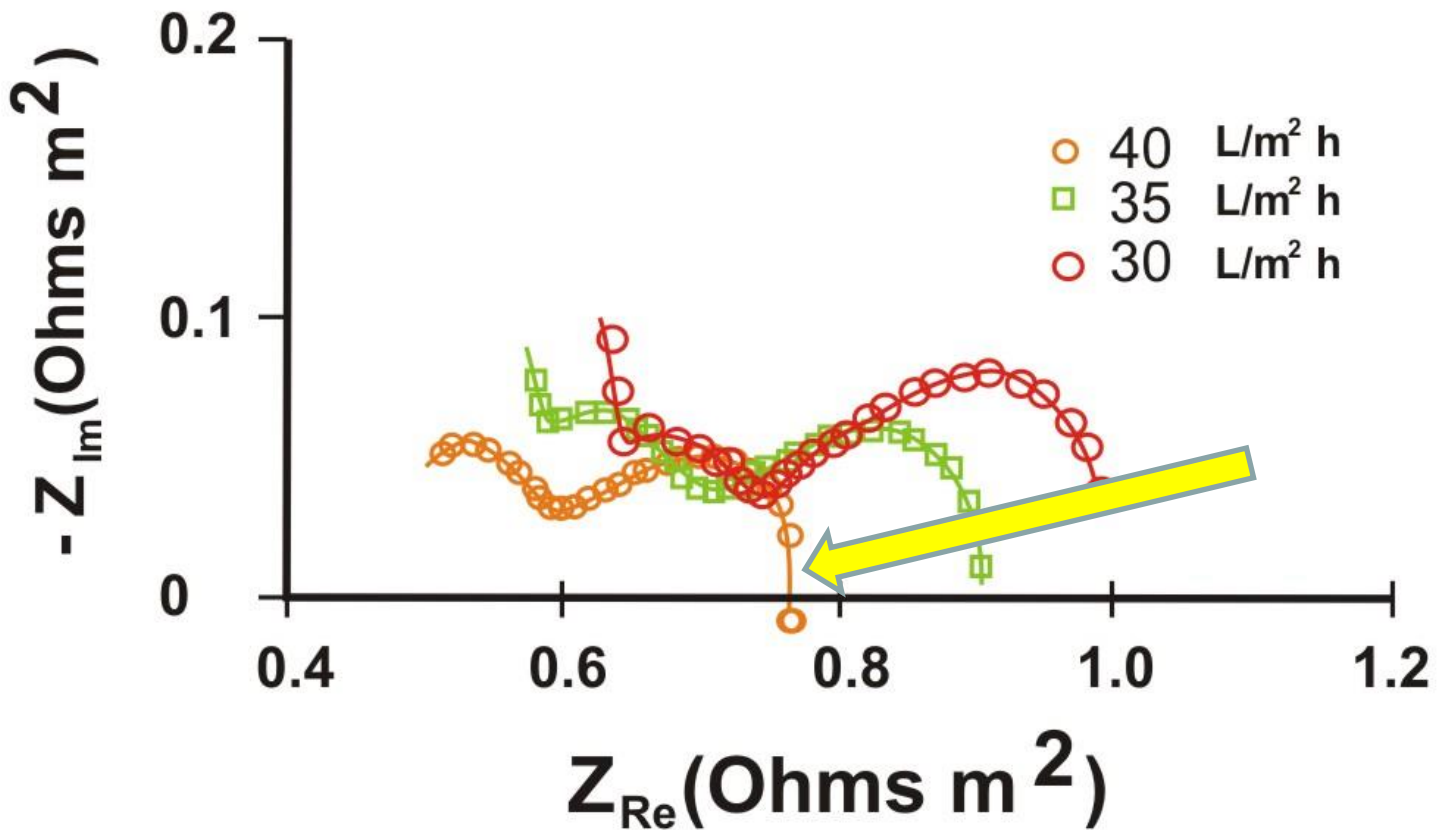


RO feed: 200 ppm silica with 2000 ppm NaCl; crossflow velocity; 0.15 m/s.

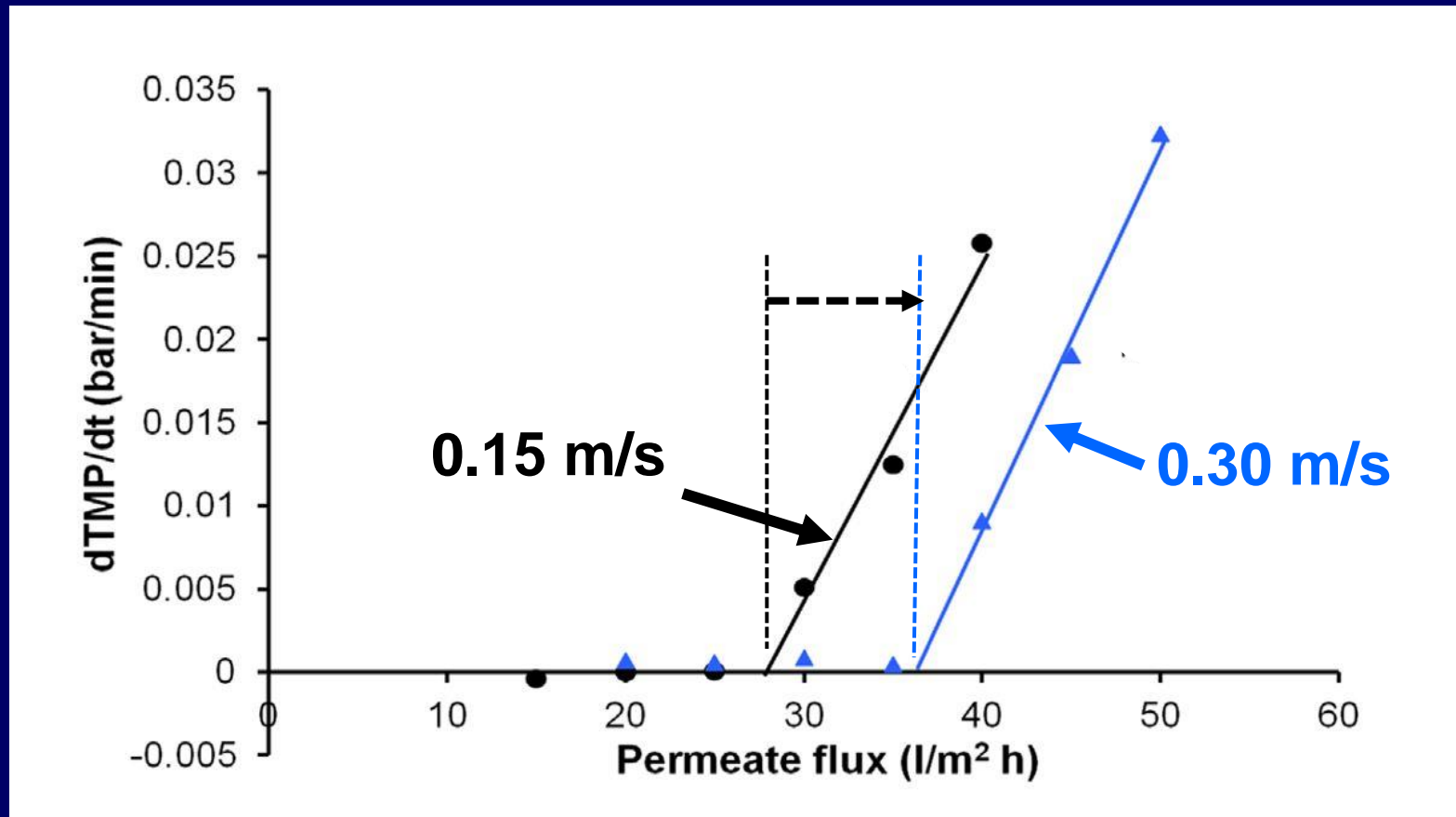
EIS Nyquist plots vs Flux at low fluxes



EIS Nyquist plots at higher fluxes

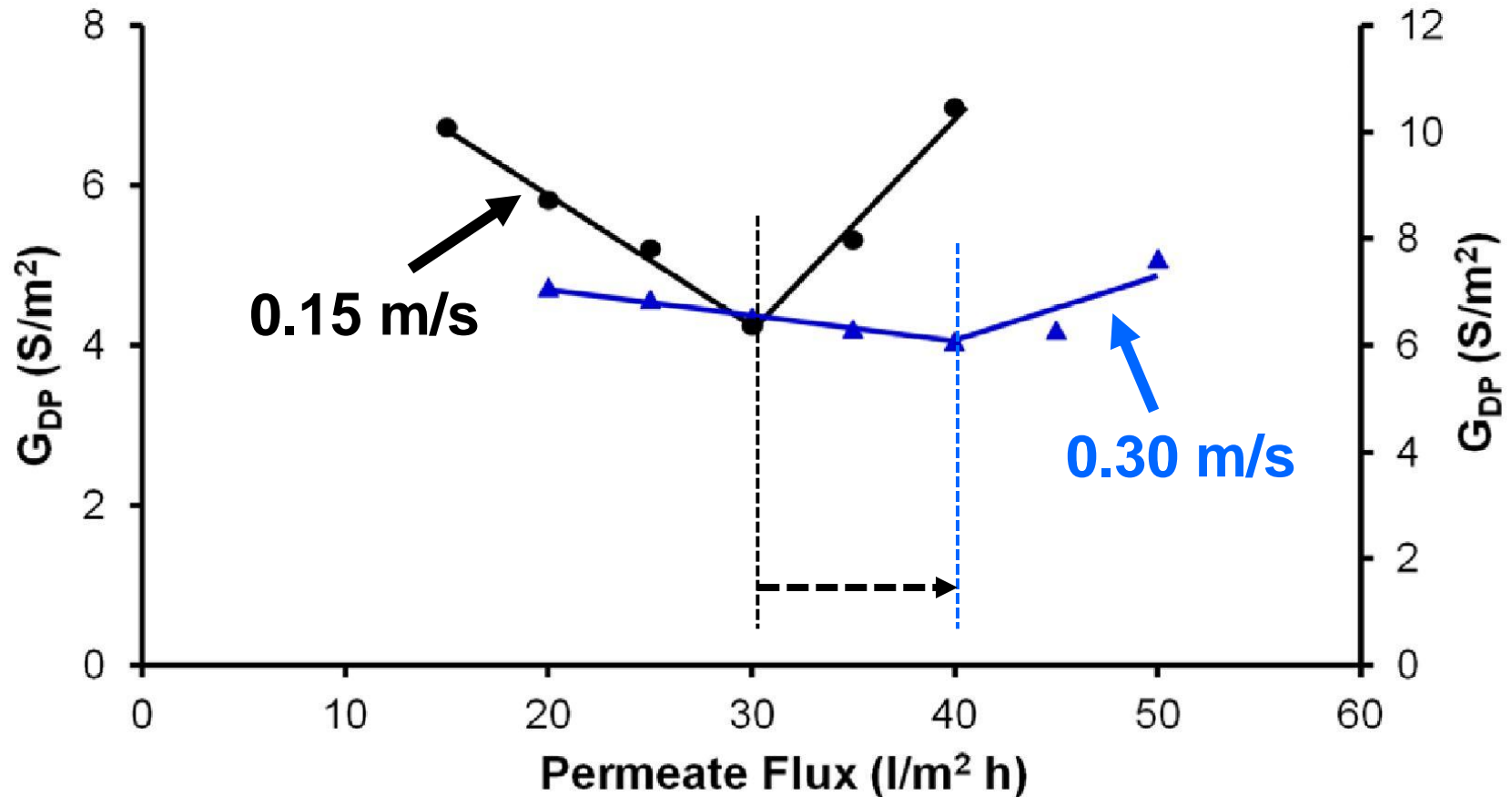


Fouling: A Threshold Phenomena



RO feed: 200 ppm silica with 2000 ppm NaCl.

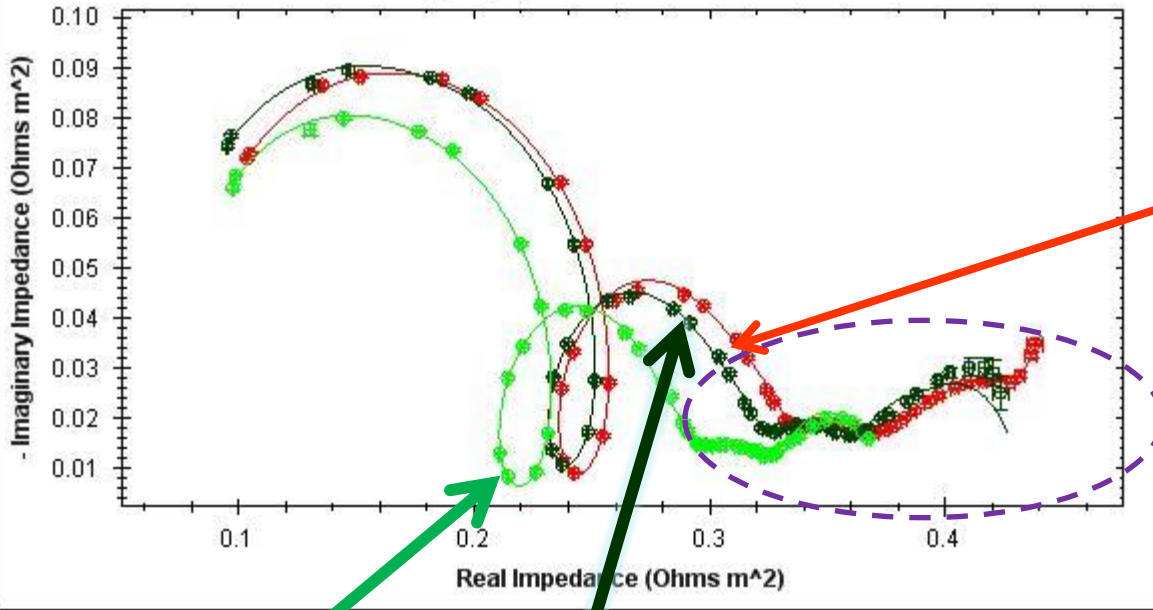
EIS detection of the Threshold



RO feed: 200 ppm silica with 2000 ppm NaCl.

Membrane Signatures in early stages of Filtration

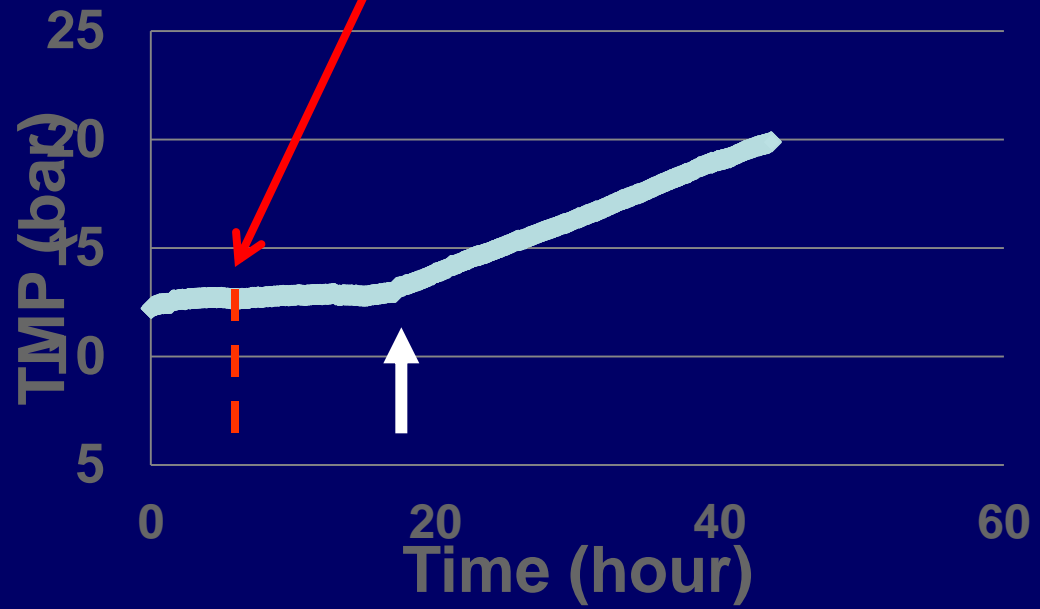
Average Nyquist with Model Plot



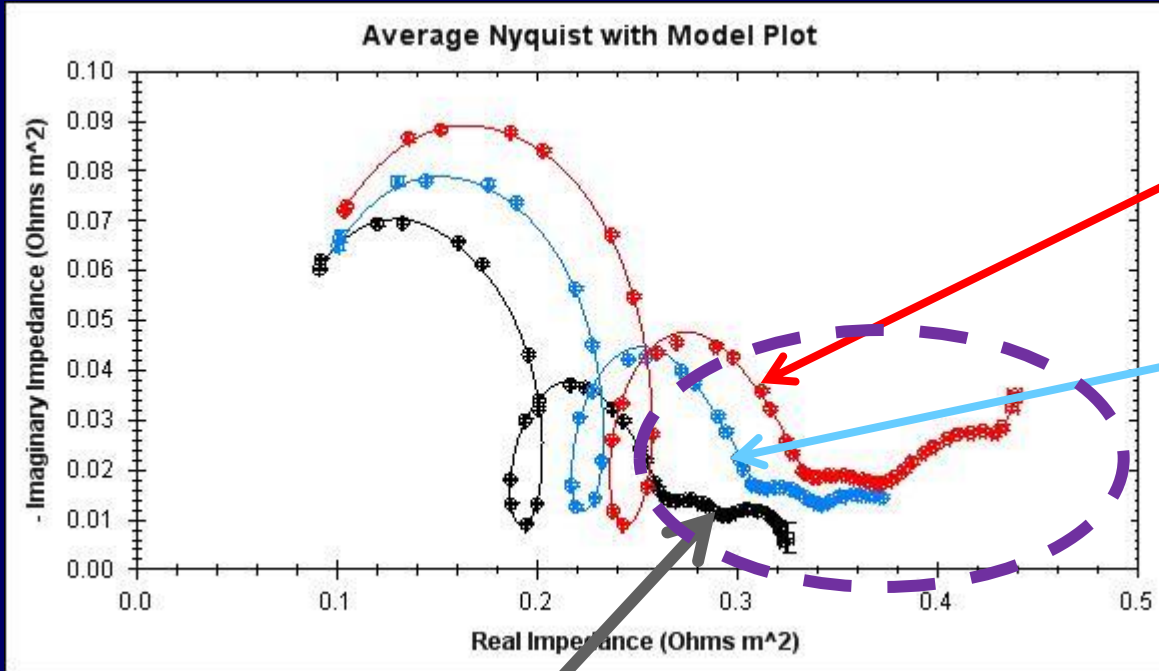
5 hrs silica

Saline –
no silica

2 hrs Silica



Signatures of Membrane Fouling



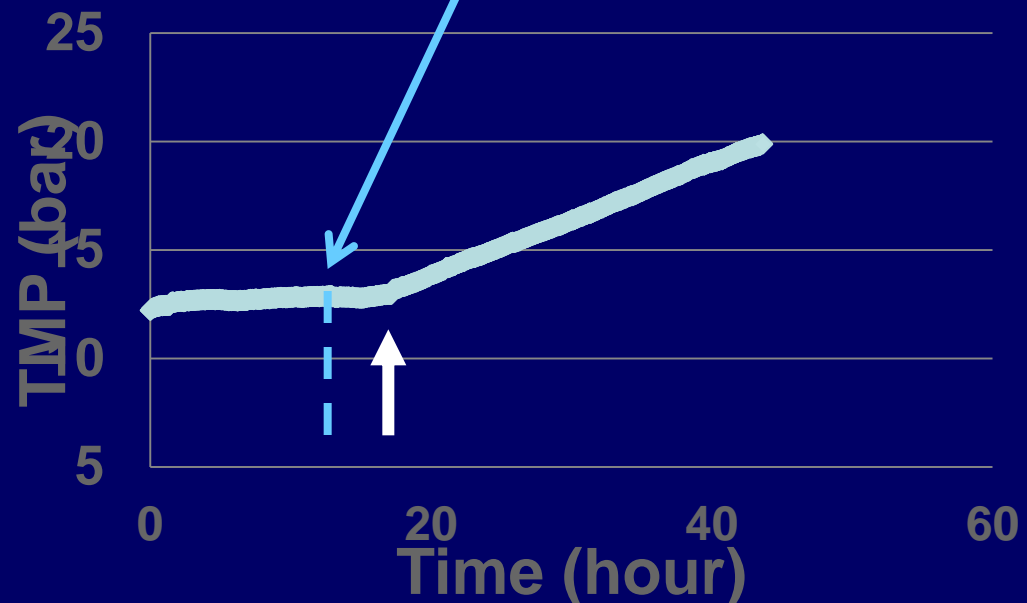
5 hrs silica

15 hrs silica

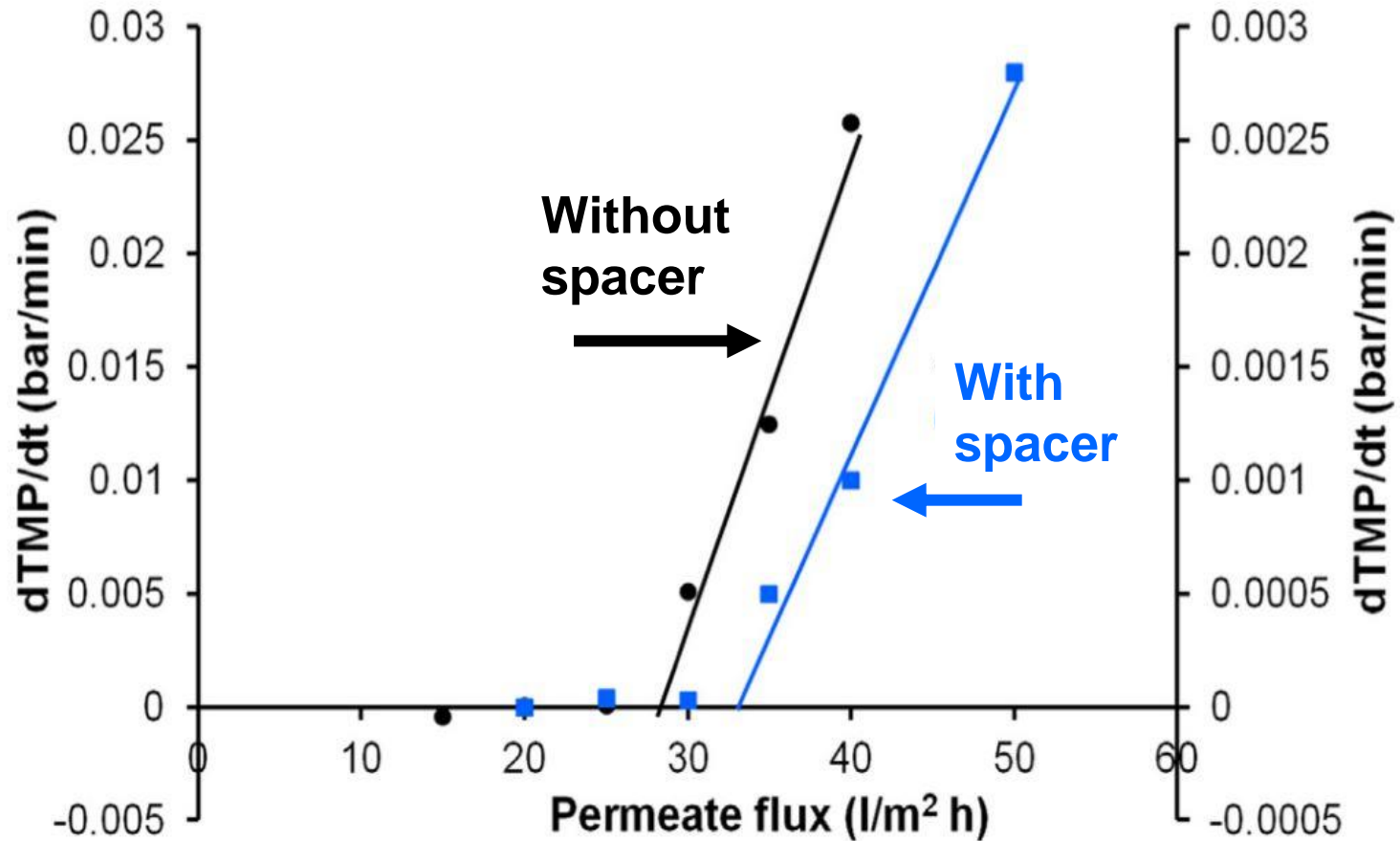
48 hrs silica

Cake enhanced concentration polarization

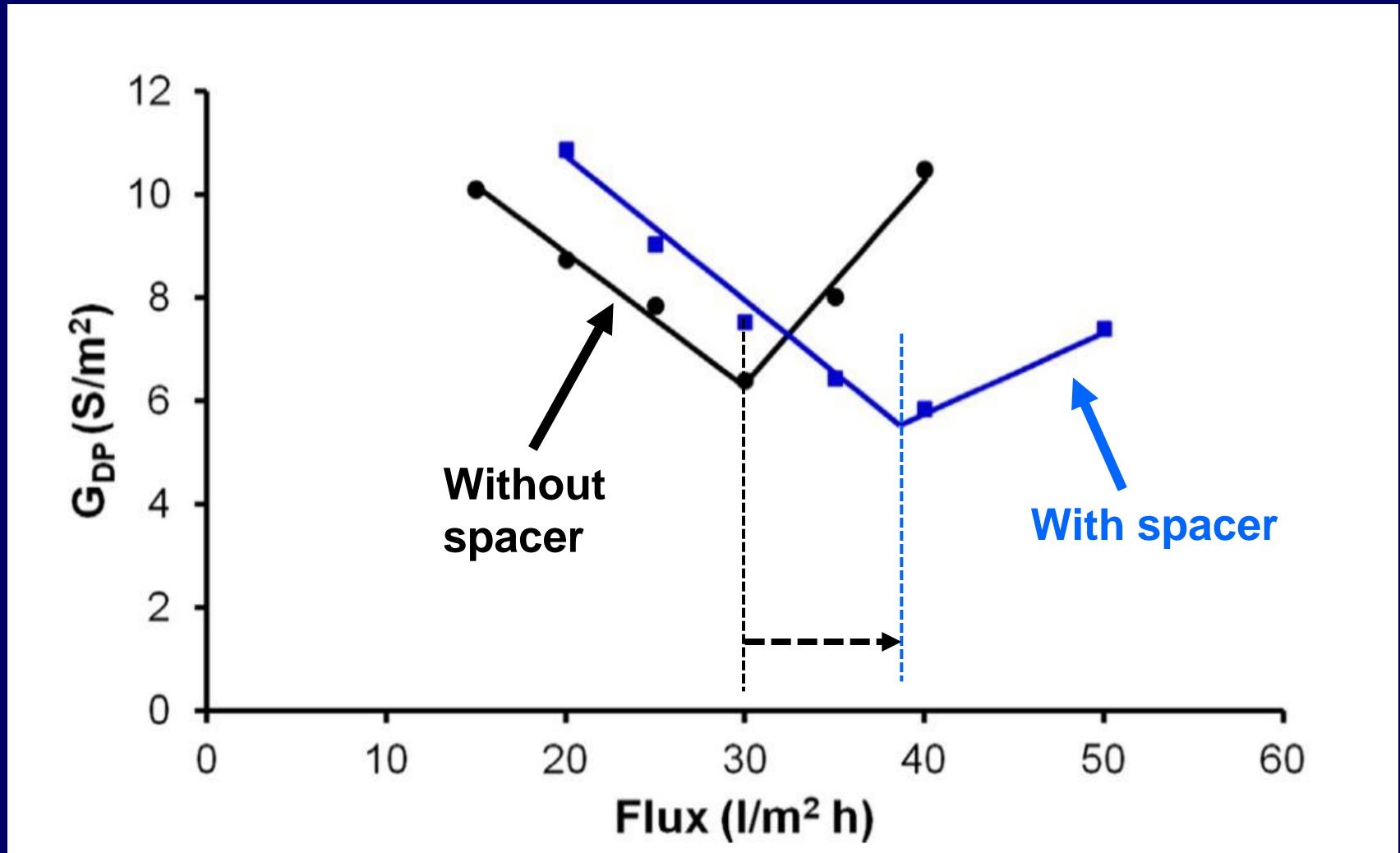
The EIS signatures changed well ahead of fouling revealed by TMP



Effect of Spacers



Effect of Spacers on Gdp

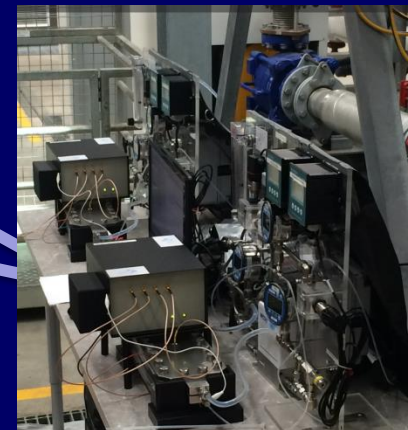
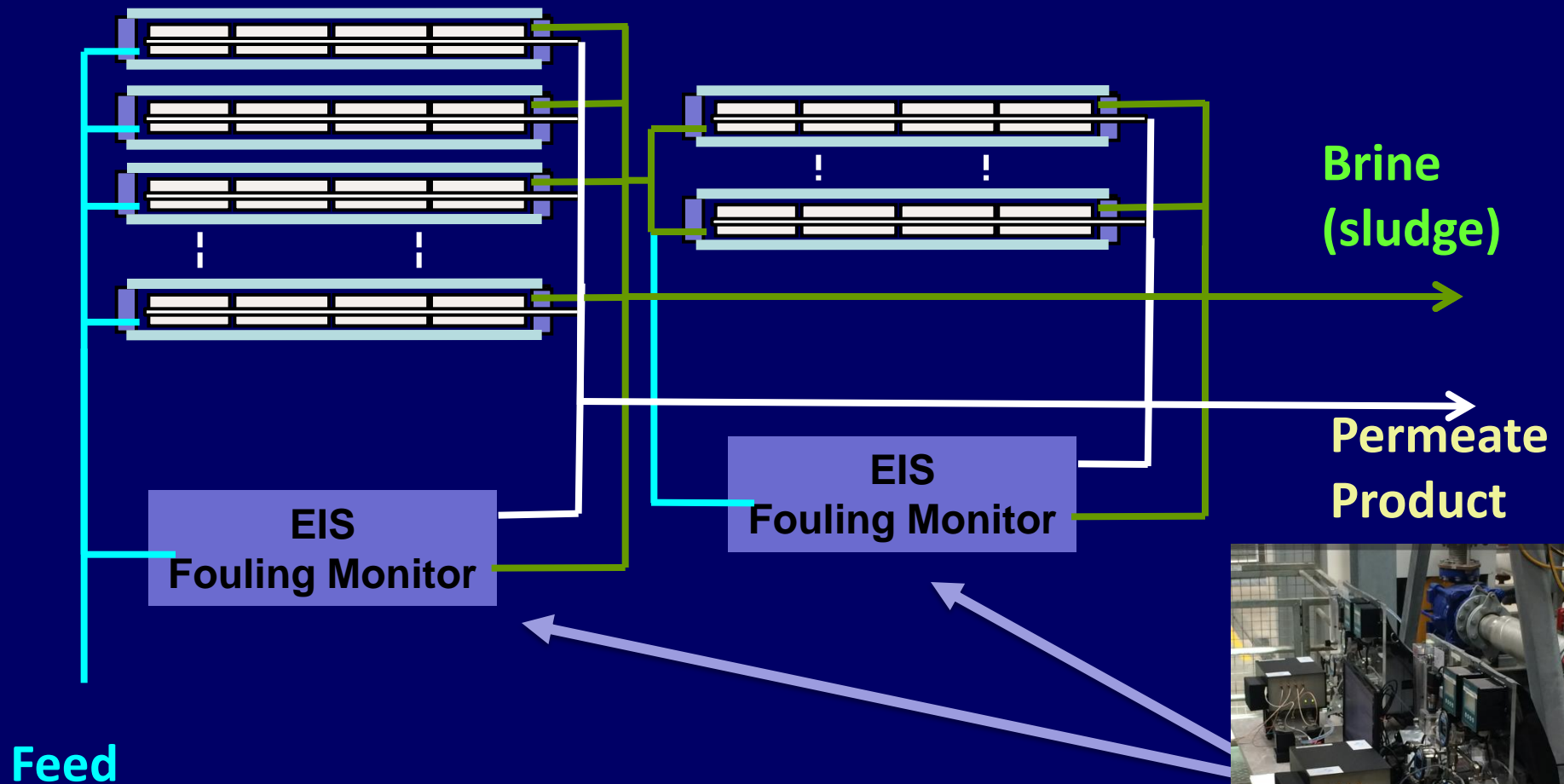


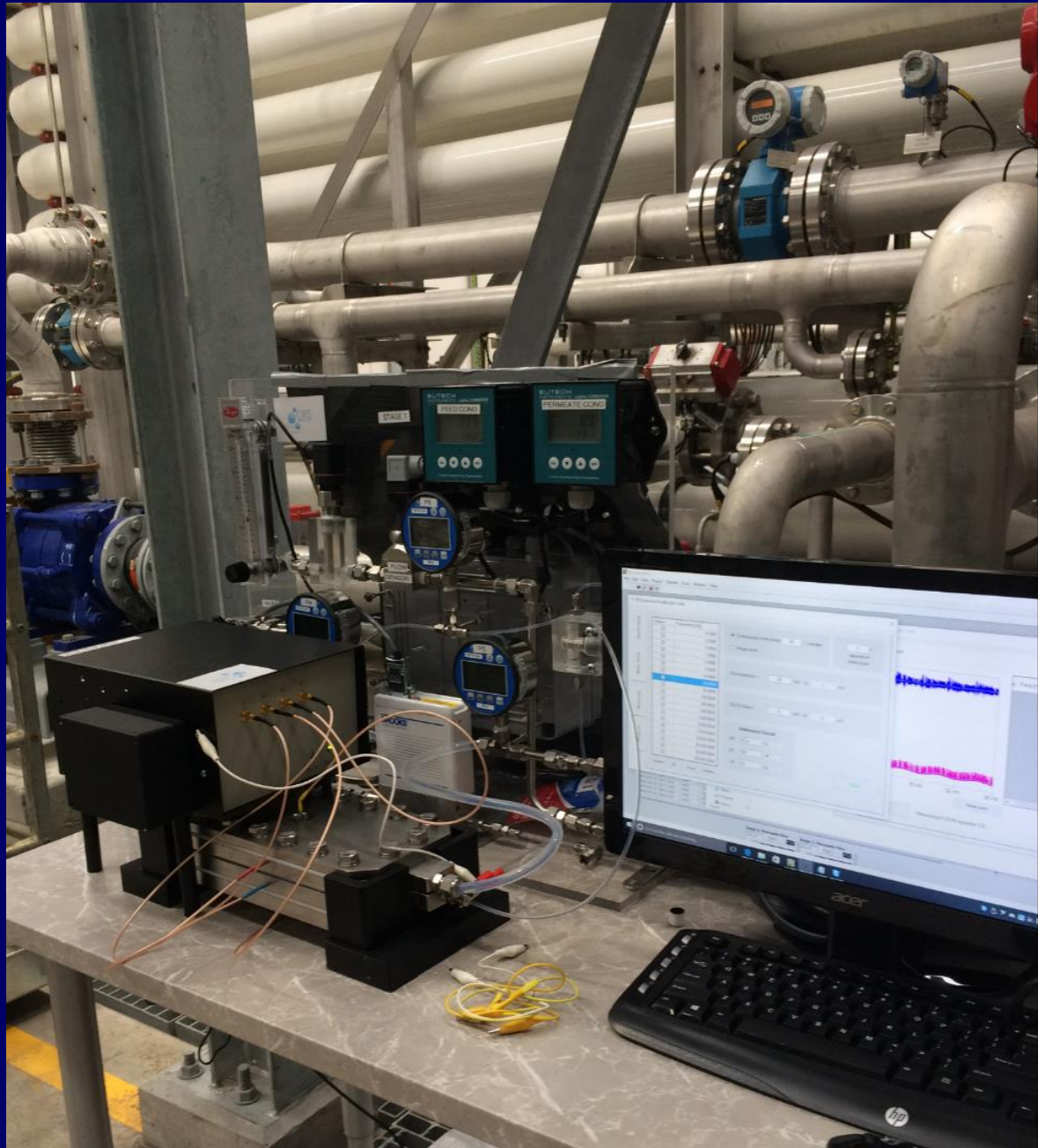
RO feed: 200 ppm silica with 2000 ppm NaCl; crossflow velocity; 0.15 m/s

Silica fouling: Suggested mechanism

- **Slow built up of the silica layer on the membrane surface; Electrical conductance in the concentration polarization layer drops- G_{dp} decreases.**
- **Impact of the cake enhanced concentration polarization (CECP) effect; The increased concentration polarization of NaCl at the membrane surface increases the conductance of the concentration polarization layer and G_{dp} .**
- **More NaCl permeates through the membrane which shows up in a decrease in rejection.**

Locating a “Canary” Membrane Fouling Monitor in Water Treatment Plants





Collaborators

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