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7-20-2016

## A critical review of high entropy alloys (HEAs) and related concepts

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D.B. Miracle and O.N. Senkov, "A critical review of high entropy alloys (HEAs) and related concepts" in "Beyond Nickel-Based Superalloys II", Chair: Dr Howard J. Stone, University of Cambridge, United Kingdom Co-Chairs: Prof Bernard P. Bewlay, General Electric Global Research, USA Prof Lesley A. Cornish, University of the Witwatersrand, South Africa Eds, ECI Symposium Series, (2016). http://dc.engconfintl.org/superalloys\_ii/33

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# A CRITICAL REVIEW OF HIGH ENTROPY ALLOYS AND RELATED Seyond Nickel-Based Superalloys

### Cambridge, UK 20 July 2016

A CRITICAL REVIEW OF HIGH ENTROPY ALLOYS AND RELATED CONCEPTS DBM and O.N. Senkov, *Acta Mater.*, OVERVIEW, In Review.

**New Strategies and Tests to Accelerate Discovery and Development of Multi-PRINCIPAL ELEMENT STRUCTURAL ALLOYS** DBM, B.S. Majumdar, K. Wertz and S. Gorsse, *Scripta Mater.*, Accepted.

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### INTRODUCTION

History, Definitions, Hypotheses, Visualizations

### THERMODYNAMICS

Enthalpy and Entropy of Solid Solution and Intermetallic Phases

### TAXONOMY

Elemental Constituents, Alloy Families

### MICROSTRUCTURES

Definitions, Observed and Calculated Phases, Assessment

### PROPERTIES

Functional, Mechanical, Assessment

### **APPLICATIONS & DESIGN**

Functional, Structural

### **FUTURE WORK**

**Basic and Applied** 

### **MAJOR ACCOMPLISHMENTS**

Ideas, Achievements, Closing Remarks





## **MANUSCRIPT APPROACH**



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### Each section is written in an independent style

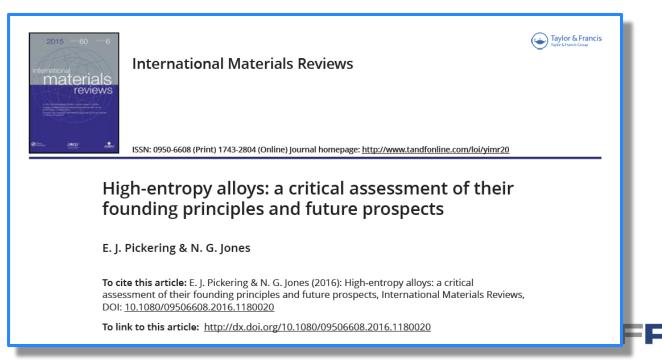
- Extensively cross-referenced to other sections, separate summaries

### Only major results are shown here

- Emphasize Mechanical Properties and Major Accomplishments

### Critical review of main hypotheses recently published

- Scope is different but conclusions are the same





### **EARLY HEA HYPOTHESES** *It's Time to Learn and Move Ahead!*



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# If it's published often enough, it's accepted as fact, even if there are no data to support it!

- …more than expected 1-phase, solid solution microstructures with 'simple' crystal structures...
- ✓ ...unusually strong and ductile...
- ✓ ...sluggish diffusion...
- ✓ ...violates Gibbs phase rule...

**NONE OF THESE STATEMENTS ARE CORRECT** 

THERE ARE GREAT REASONS TO STUDY HEAS, BUT NOT THESE

The problem lies not with the new ideas, but in lettinggo of the old ones.John Maynard Keynes



### **INTRODUCTION** *Definitions and Use of Terms*



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# Intended or not, HEAs are associated with controlling configurational entropy to produce 1-phase solid solutions

 This stirs unproductive controversy that distracts from the major objective of exploring vast composition space by placing limits on possibilities

### We apply a careful use of terms to avoid these implications

- We use the term, "HEA" when configurational entropy or the intent to produce 1-phase, solid solution microstructures are important
- We use multi-principle element alloys (MPEAs) or complex concentrated alloys (CCAs) to evoke vast composition space with no restrictions concerning entropy or the phases present
- This is a practical approach to clarify discussion, to focus efforts, and to avoid an unproductive controversy, and is not intended to detract from contributions of pioneers in the field

## (CCA,MPEA) = HEA w/o (SS, 1-phase, entropy)



### **THERMODYNAMICS** *All Terms Can Be Important*



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- Four primary thermodynamic terms must be considered in the competition between SS and IM phases
  - Each of these four terms can be significant
- The competition between phases is controlled by relatively small differences between these four larger values Usually (*but not always!*), ( $H^{SS}-H^{IM}$ ) > 0 and ( $S^{SS}-S^{IM}$ ) > 0
- No single value or pair of values consistently dominate phase selection
- These considerations make it difficult to accurately predict phases by focusing on a single thermodynamic term

$$\Delta G^{\text{SS-IM}} = G^{\text{SS}} - G^{\text{IM}} = (H^{\text{SS}} - H^{\text{IM}}) - T(S^{\text{SS}} - S^{\text{IM}})$$





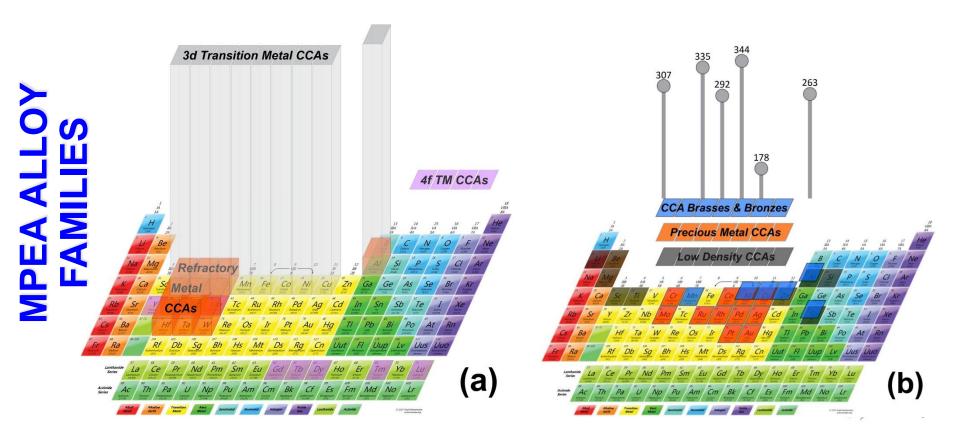




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A remarkable focus is found on elements used in MPEAs, and in groupings of elements

A broader use of element groupings is now underway





### MECHANICAL PROPERTIES 3d Transition Metal MPEAs



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MPEAs are often claimed to be unusually strong and ductile, but available data don't support these claims

- 3d transition metal MPEAs have essentially the same properties as austenitic stainless steels and nickel alloys
  - Generally have RT  $\sigma_{y}$  below 300 MPa and  $\sigma_{uts}$  below 700 MPa
  - Significant strengthening from work-hardening and grain refinement are not expected to be effective above about  $T_m/2$  (about 500C)

# Austenitic stainless steels and nickel alloys are 3d transition metal CCAs

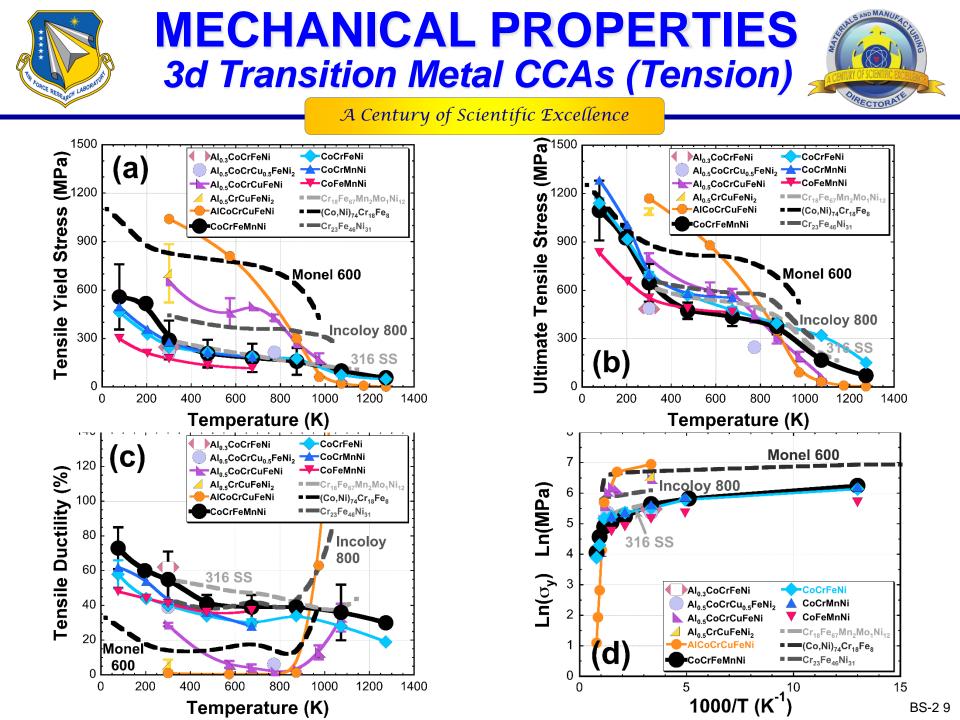
 All austenitic stainless steels and nickel alloys have Cr-Fe-Ni as principal elements and can also have significant additions of Co, Cu, Mn, Mo

# 3d TM MPEA strengths do not compete with precipitation hardened stainless steels or nickel superalloys

–  $\sigma_y$  <100 MPa at about 1100 K, but superalloy sheet requires  $\sigma_y$  >200 MPa and blades require  $\sigma_y$  >600 MPa



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### **REFRACTORY METAL CCAs** Structural Properties Summary



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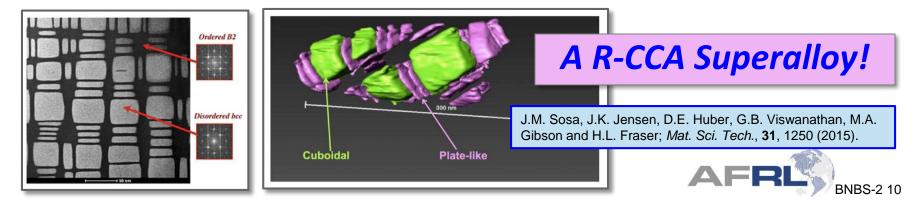
# Refractory metal CCAs were created to compete with nickel superalloys

- Refractory metal CCAs are still in relatively early stages of exploration (only 26 alloys in this assessment)
- A wide range in elemental properties (density, modulus, environmental resistance) suggest a wide range in alloy properties

### Densities range from 5.59 to 13.75 g cm<sup>-3</sup>

# Most alloys have one or two BCC phases, one or more Laves phases, and one alloy reports a B2 phase

- The full extent of the BCC solid solution phase field has not been explored





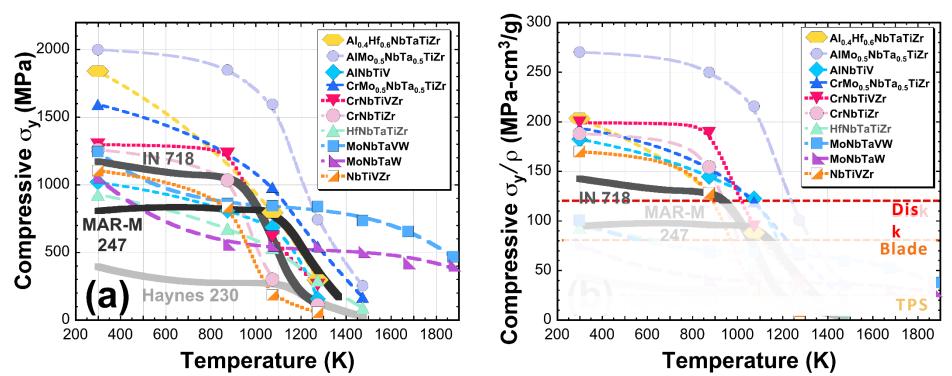
### **REFRACTORY METAL CCAs** Structural Properties Summary



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# Most mechanical properties are measured in compression with only a few tensile studies

- Several alloys have yield and specific yield strengths that show potential to increase stresses and/or temperatures compared to superalloys
- The strongest alloys have ≥10% RT compressive ductility, offering the possibility of useful tensile ductility









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### **Continued exploration of principal element combinations**

- New alloy bases
- Control the type, volume fraction, size, morphology and distribution of second phases to design precipitation strengthened microstructures
- Extent of solid solution phases

### Tensile properties of alloys with compressive ductility ≥10% Characterize oxidation behavior and explore alloying to improve environmental stability

- Simple, hi-throughput oxidation screening tests are recommended

# Effect of deformation processing on microstructure, properties

– Cold-rolling has been demonstrated in one alloy





# RAPIDLY CHARACTERIZE

New Strategies and Tests to Accelerate Discovery and Development of Multi-Principal Element Structural Alloys DBM, B.S. Majumdar, K. Wertz and S. Gorsse, *Scripta Mater.*, Accepted.



### **Conventional Characterization**

### 1<sup>st</sup> Tier

-Tensile strength, ductility

### 2<sup>nd</sup> Tier

- -Fracture toughness
- -Fatigue
- -Creep

### **Final Characterizations**

- -Environmental resistance
- -'...ilities'

### Strengths

 Enables intuitive 'pre-selection' when a knowledge base already exists for a small number of candidate alloy systems.

### Weaknesses

 Starts with difficult-to-measure properties that depend on both composition and microstructure, making it hard to reject alloys quickly.

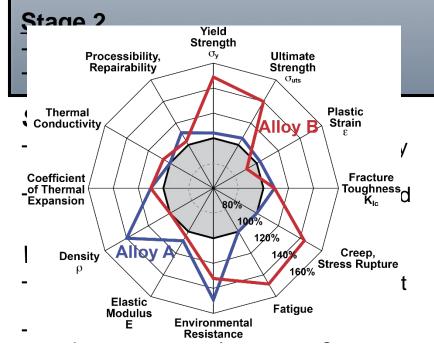
### **New Characterization Strategy**

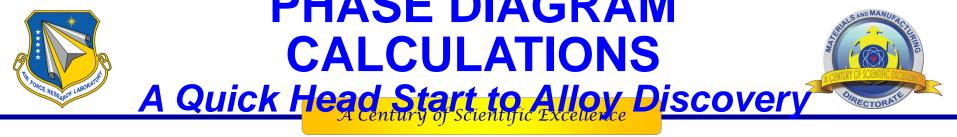
### Stage 0

- Hi throughput phase diagram calculations

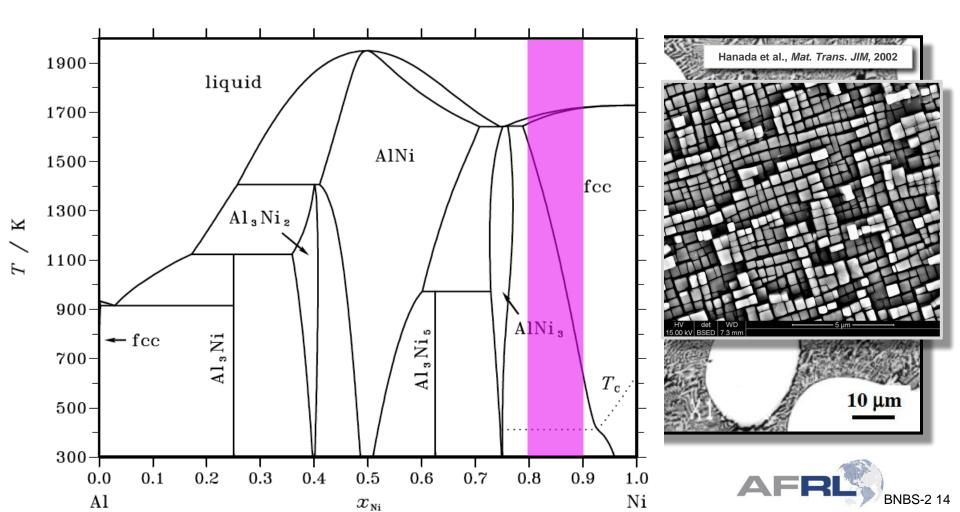
### Stage 1

- Microstructure-independent properties
- Environmental resistance
- Modulus, density, thermal properties





Specific phase diagram features enable two-phase microstructures that may give a good balance of properties



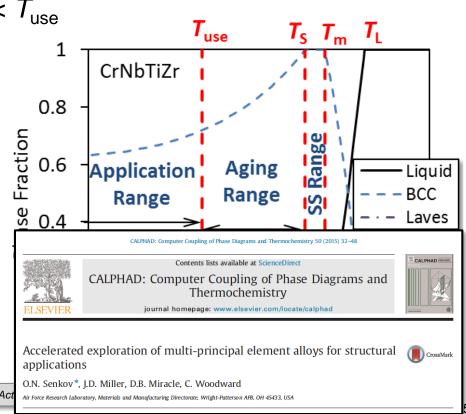


Specific phase diagram features enable two-phase microstructures that may give a good balance of properties

### Phase diagram criteria from <u>CAL</u>culated <u>PHA</u>se <u>Diagram</u> (CALPHAD) calculations

- $-T_{\rm m} > T_{\rm use}$ , no phase transformation  $< T_{\rm use}$
- At least 1 SS phase and no more than 1 IM phase at  $T_{\rm use}$
- One phase must dissolve above  $T_{use}$
- fcc, bcc, hcp or ordered derivative crystal structures
- These are selective criteria

Calculation of 135,000 phase diagrams has identified ~100 candidate high temperature structural alloy systems





## EXPERIMENTS A Centur Stagle fiel Excellence



## **STAGE 0**

**Calculated Phase Diagrams** 

## **STAGE 1**

**Structure-Insensitive Properties** 

## **STAGE 2**

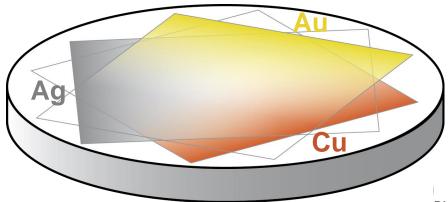
**Structure-Sensitive Properties** 

# Materials libraries with controlled composition gradients

Density, modulus, T<sub>m</sub>, environmental resistance

# Environmental resistance is probably the most impactful test

- Possibility to reject many candidates early in the evaluation process
- Fast and conceptually straightforward
- Has not yet been demonstrated





## EXPERIMENTS A Centus tage fi 2 Excellence



## **STAGE 0**

**Calculated Phase Diagrams** 

## **STAGE 1**

**Structure-Insensitive Properties** 

## **STAGE 2**

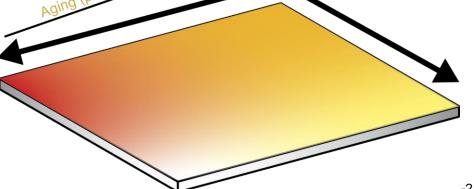
**Structure-Sensitive Properties** 

### Fixed-composition materials libraries with controlled microstructure gradients

- Strength, tensile ductility, creep, fracture, fatigue
- Libraries are conceptually simple

### Stage 2 unmet charenges

- Strength can be as see see hardness
- There are no hi thrup put tests for ductility, creep, fatigue, toughness
- This is a major barrier





### MAJOR ACCOMPLISHMENTS Four HEA Core Effects



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Present data & analyses do not support an observable effect of configurational entropy on preferred formation of 1-phase alloys, solid solution phases or simple crystal structures

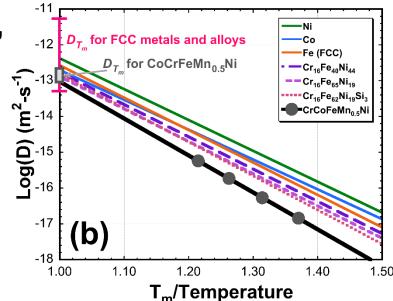
- Two direct experiments and two computational studies do not support the 'high entropy' hypothesis
- Six issues bias observations toward 1-phase/ SS/ 'simple' microstructures
- A simple 'structure in structure out' (SISO) analysis links observed alloy microstructures to the structures of the constituent elements

# Limited data show that MPEA diffusion is not unusually 'sluggish'

 In the same range as FCC elements and conventional alloys

### Insufficient data to evaluate the 'lattice distortion' hypothesis

The 'cocktail effect' is a colorful phrase for 'non-linearity' of elemental combinations in general





## **MAJOR ACCOMPLISHMENTS**



- Introduced and established the concept of exploring the vast compositional realm of multi-principle element alloys.
- Establish importance not dominance of configurational entropy as a control variable in phase selection.
- Define/explore 7 new alloy families and 122 new base alloys, emphasizing 3d transition metal and refractory metal alloys.
- Expand the known 3d transition metal FCC phase field and discover up to 10 other extended 1-phase fields.
- Establish composition/ microstructure/ properties relations for 3d transition metal MPEAs
- Expand initial ideas to include multi-phase microstructures, functional materials and high-throughput evaluations.
- High throughput calculations identify over 200 alloy bases as potential high temperature structural materials.



## RECOMMENDATIONS



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### Continue to explore and develop MPEAs for structural use

- Expand the range of alloy families, possibilities have barely been scratched
- Include both SS and precipitation strengthened microstructures
- Expand thermo-mechanical processing efforts to control microstructure

# Apply strategy linking high-throughput computations & experiments to develop high temperature MPEAs

- High throughput experiments for structural materials requires new tools

### Establish basic scientific concepts

- Strengthening models
- Lattice distortion effects







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All but 2 allotropic transformations convert to BCC at T<sub>m</sub> Crystal structures at RT and at T<sub>m</sub>

Element	Atomic Number	Used in # of alloys	Structure at RT*	Structure at T <sub>m</sub> *	Element	Atomic Number	Used in # of alloys	Structure at RT*	Structure at T <sub>m</sub> *
Н	1				Rh	45	1	A1 (FCC)	
Li	3	7	A2 (BCC)		Pd	46	3	A1 (FCC)	
Be	4	1	A3 (HCP)	A2 (BCC)	Ag	47	1	A1 (FCC)	
В	5	3	– (hR105)		Cd	48		A3 (HCP)	
С	6	7	A3 (HCP)		In	49	. –	A6 (tl2)	
N	7				Sn	50	17	A5 (tl4)	
0	8				Sb	51		A7 (hR2)	
Na	11		A2 (BCC)		Те	52		A8 (hP3)	
Mg	12	8	A3 (HCP)		Cs	55		A2 (BCC)	
Al	13	263	A1 (FCC)		Ba	56		A2 (BCC)	
Si	14	15	A4 (cubic) <sup>¶</sup>		La	57		A3 (DHCP)	A1 (FCC)
P	15		- (oC8)		Ce	58		A3 (HCP)	A2 (BCC)
S	16		– (oF128)		Pr	59	4	A3 (DHCP)	A2 (BCC)
K	19		A2 (BCC)		Nd	60	1	A3 (DHCP)	A2 (BCC)
Ca	20	4	A1 (FCC)	A2 (BCC)	Sm	62		C19 (hR3)	A2 (BCC)
Sc Ti	21 22	1		A2 (BCC)	Eu Ca	63	2	A2 (BCC)	
V	22 23	101 55	A3 (HCP)	A2 (BCC)	Gd Tb	64 65	2 2		A2 (BCC)
v Cr	23 24	307	A2 (BCC)			66	2		A2 (BCC)
Mn	24 25	307 79	A2 (BCC) A12 (cl58)	A2 (BCC)	Dy Ho	67	Ζ	A3 (HCP) A3 (HCP)	A2 (BCC)
Fe	25 26	79 335	A12 (CISO) A2 (BCC)	A2 (BCC) A2 (BCC)	Er	68		A3 (HCP)	
Co	20	292	A2 (BCC) A3 (HCP)	A2 (BCC) A1 (FCC)	Tm	69	1	A3 (HCP)	
Ni	28	292 344	A1 (FCC)	AT (FCC)	Yb	70	I	A3 (FCC)	A2 (BCC)
Cu	29	178	A1 (FCC)		Lu	71	2	A3 (HCP)	A2 (DCC)
Zn	30	7	A3 (HCP)		Hf	72	5	A3 (HCP)	A2 (BCC)
Ga	31	'	A11 (oC8)		Та	73	13	A2 (BCC)	A2 (000)
Ge	32	1	A4 (cubic) <sup>¶</sup>		W	74	2	A2 (BCC)	
As	33		A7 (hR2)		Re	75	2	A3 (HCP)	
Se	34		A8 (hP3)		Os	76		A3 (HCP)	
Rb	37		A2 (BCC)		lr	77		A1 (FCC)	
Sr	38		A1 (FCC)	A2 (BCC)	Pt	78		A1 (FCC)	
Y	39	6	A3 (HCP)	A2 (BCC)	Au	79	2	A1 (FCC)	
Żr	40	33	A3 (HCP)	A2 (BCC)	Hg	80	-	A10 (hR1)	
Nb	41	29	A2 (BCC)		TI	81		A3 (HCP)	A2 (BCC)
Мо	42	49	A2 (BCC)		Pb	82		A1 (FCC)	- (
Ru	44	1	A3 (HCP)		Bi	83		A7 (mC4)	

Table 3