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# Enhanced oxidation resistance of Ti-rich Mo-Si-B alloys by pack-cementation process

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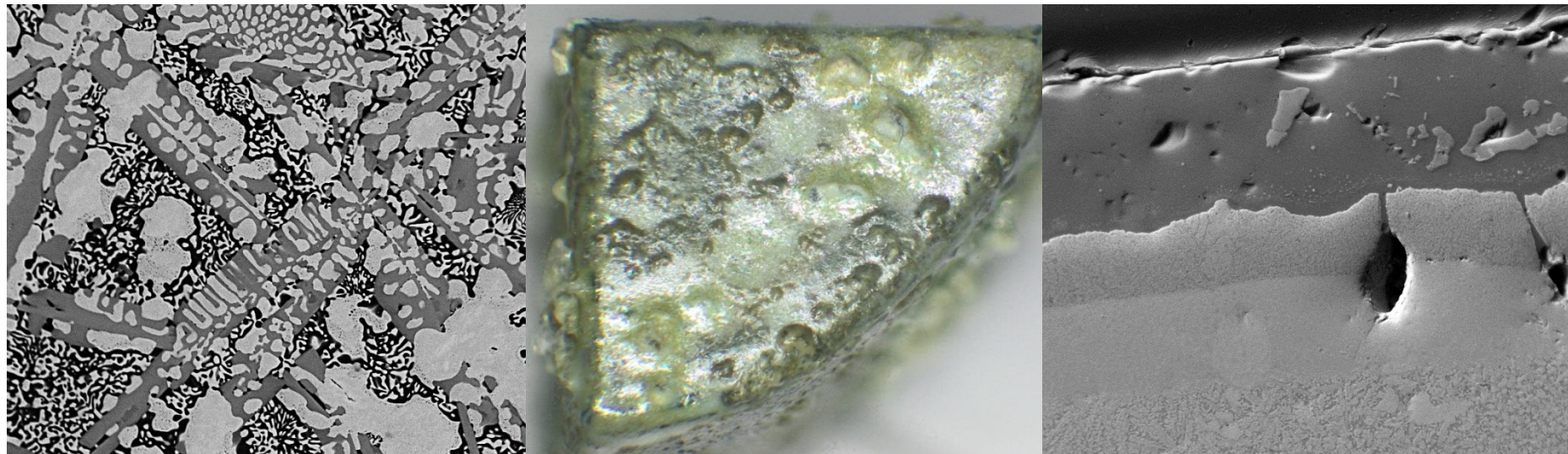
# Enhanced oxidation resistance of Ti-rich Mo-Si-B alloys by pack-cementation process

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# Motivation

- Mo-Si-B alloys
  - Possess oxidation resistance and good creep resistance at 1200 °C
  - Suffer from relatively high density (9.6 g/cm<sup>3</sup>)
  
- Potential alternatives are Ti-rich Mo-Si-B alloys
  - Reduced density (7.8 g/cm<sup>3</sup>)
  - Higher creep resistance
  - Lower oxidation resistance (formation of mixed SiO<sub>2</sub>·TiO<sub>2</sub> duplex oxide scale)
  
- Improve oxidation behavior of Ti-rich Mo-Si-B alloys



- Pack-cementation previously successfully applied to Mo-Si-B based alloys to enhance oxidation resistance



- Transfer the process to Mo-Si-B-Ti

# Material and experimental procedure

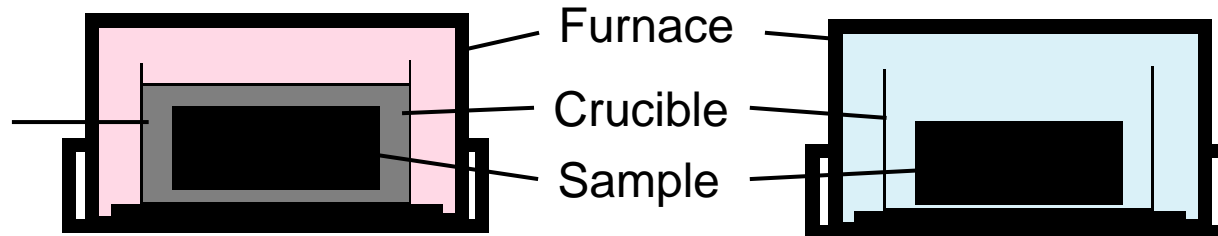
- Substrate to be coated: Mo-12.5Si-8.5B-27.5Ti (in at.%)
  - Prepared by arc-melting in Ar-atmosphere
  - Heat-treated at 1600 °C for 100 h

## ■ Pack-cementation process / two-step process

1) Pack-coating at 1000 °C for 40 h in Ar-atmosphere

2) Conditioning at 1400 °C for 10 h in air

Powder mixture:  
 35 wt.% Si + B  
 2.5 wt.% NaF  
 62.5 wt.% Al<sub>2</sub>O<sub>3</sub>

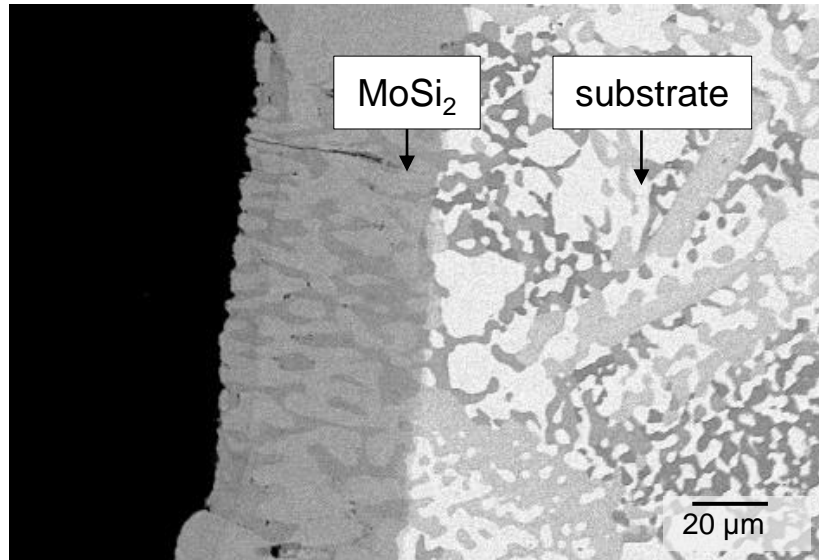


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- Pack-cementation process / two-step process
  - 1) Pack-coating at 1000 °C for 40 h in Ar-atmosphere
  - 2) Conditioning at 1400 °C for 10 h in air
  
- Investigation
  - Cyclic oxidation experiments: between T (800, 1100, 1200 °C) and 25 °C
    - Oxidation kinetics up to 1000 h
  - Microstructure examination
    - Characterize layer development

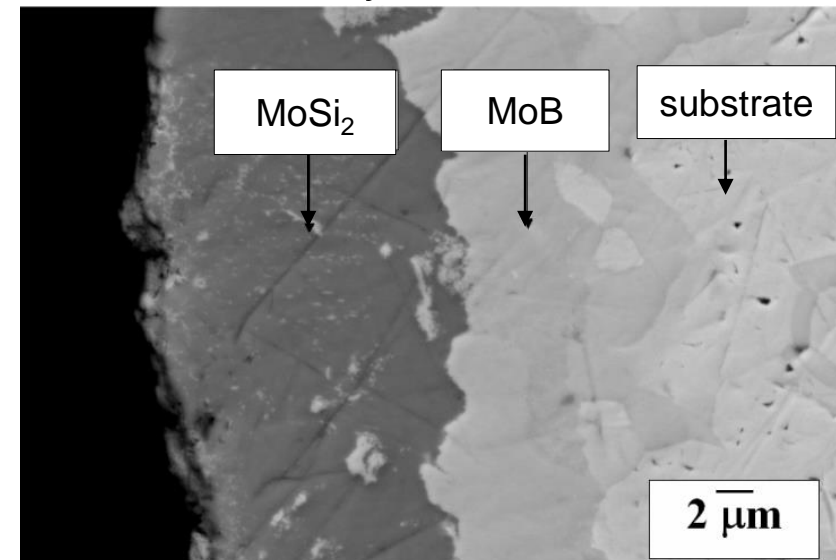
# Microstructure: layer after step 1

Si+B pack-cementation on  
Mo-12.5Si-8.5B-27.5Ti



- Layer consists of  $\text{MoSi}_2$  with a Ti content of 13 at.%
- No equilibrium between  $\text{MoSi}_2$  layer and substrate

Si+B pack-cementation on  
Ti-free Mo-alloy



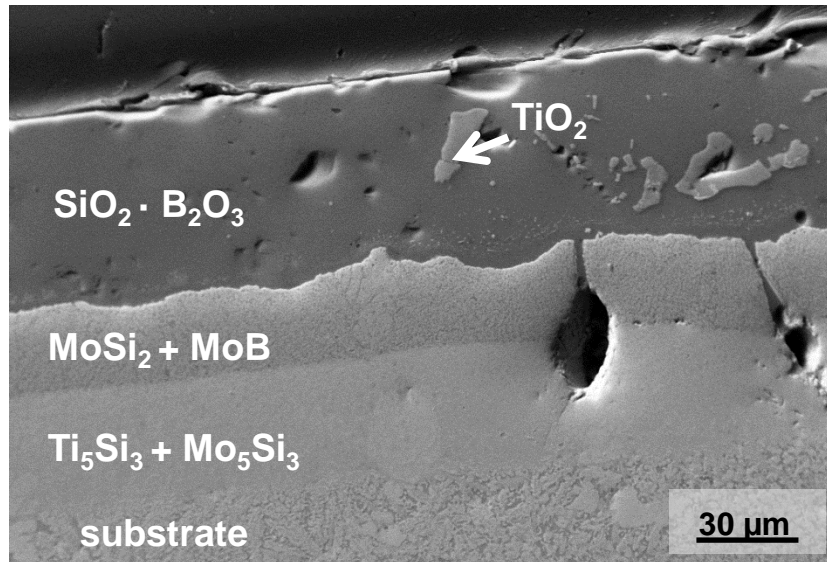
R. Sakidja et al., Scripta Mater. 53, 723 (2005)

- Outer layer composed of  $\text{MoSi}_2$  while inner layer is MoB
- MoB layer after cementation favors formation of  $\text{Mo}_5\text{SiB}_2$  – acts as a diffusion barrier

# Microstructure

## Layer after conditioning (step 2)

Mo-12.5Si-8.5B-27.5Ti

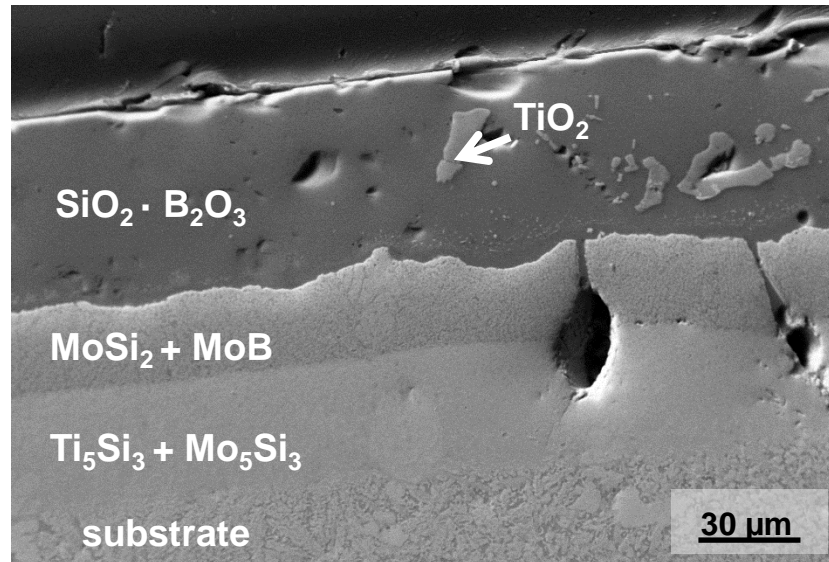


- Three separate layers
- Oxide scale and two diffusion layers

# Microstructure

## Layer after conditioning (step 2)

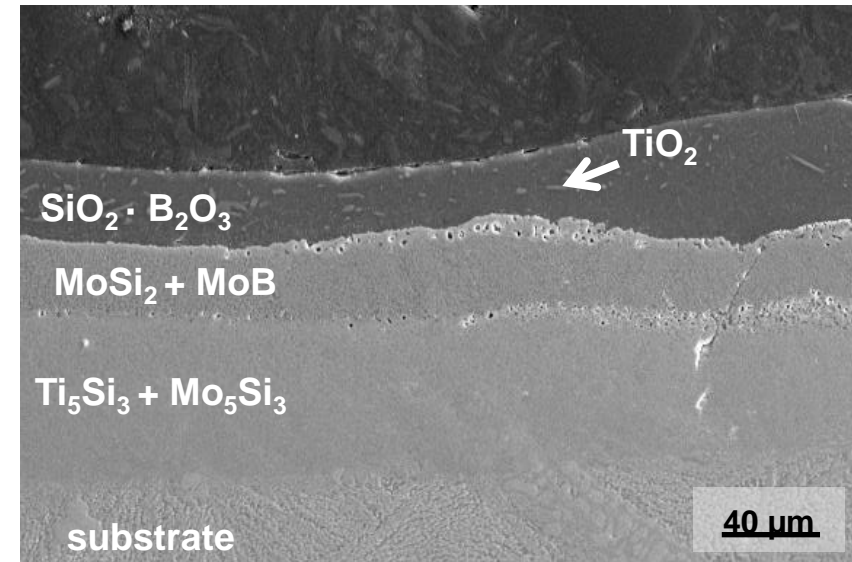
Mo-12.5Si-8.5B-27.5Ti



- Three separate layers
- Oxide scale and two diffusion layers

## Layer after cyclic oxidation test

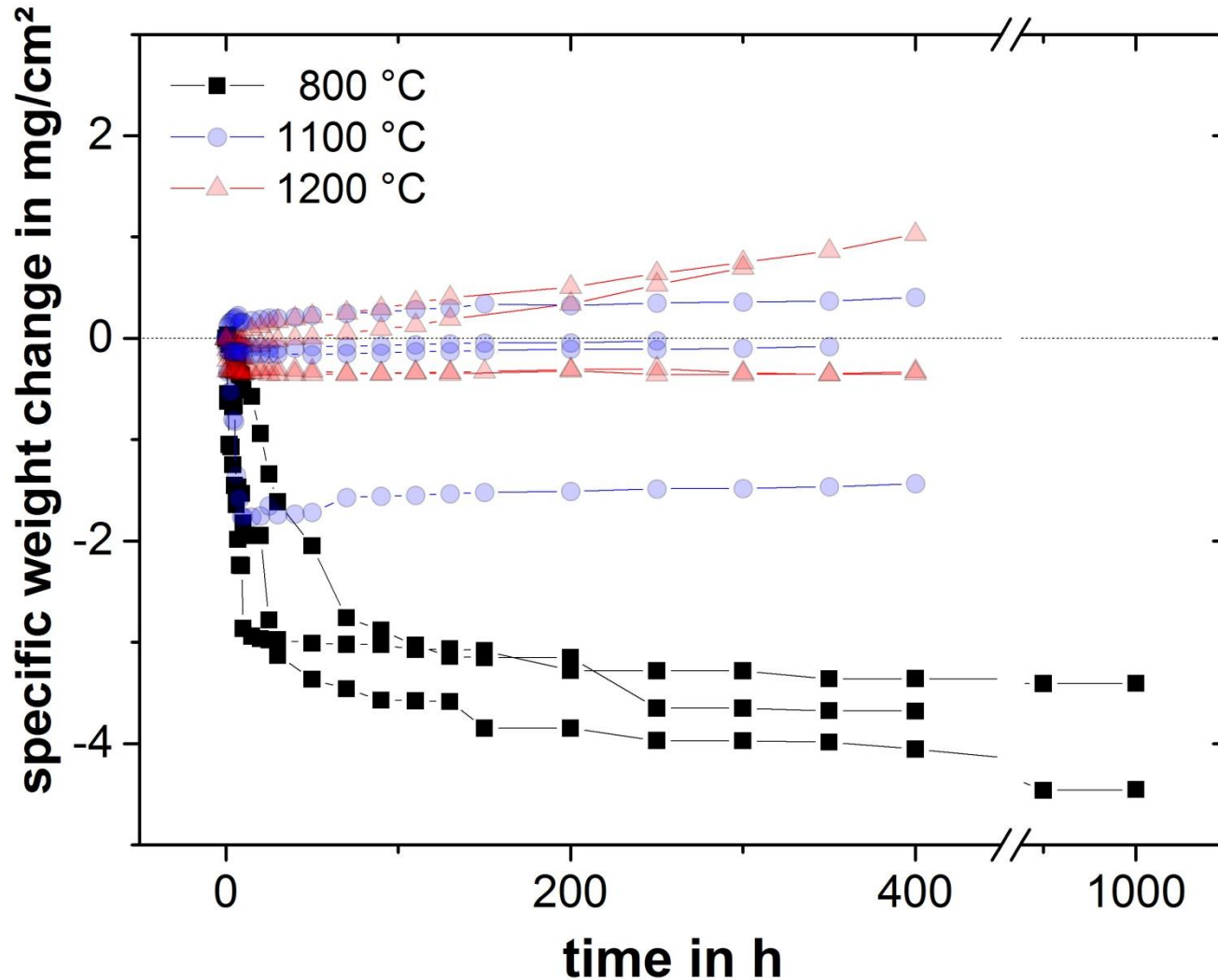
1100 °C for 500 h



- MoSi<sub>2</sub> layer still present
- No spalling
- Coating intact after thermal cycling

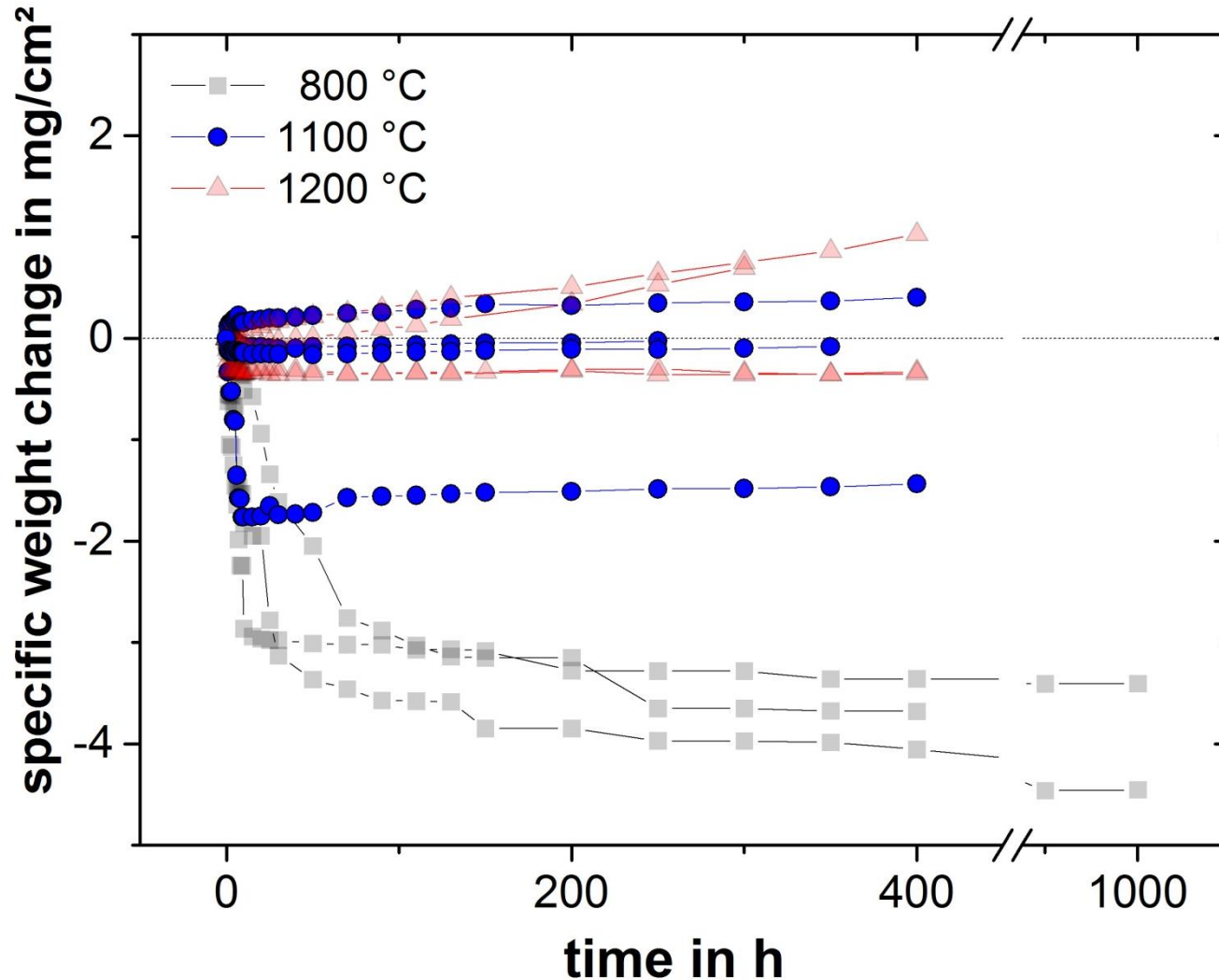


# Oxidation behavior: Effect of temperature



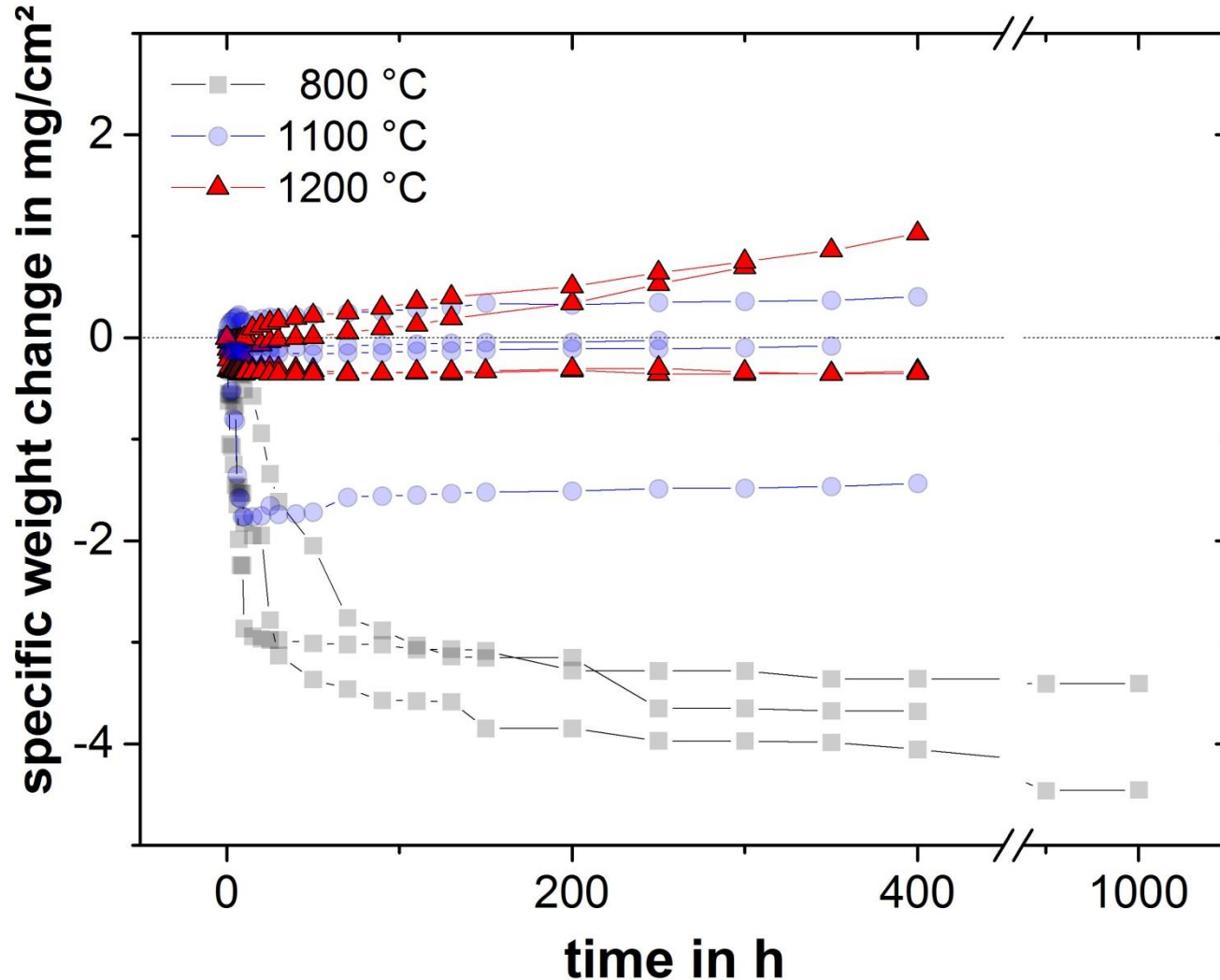
■ Weight change

# Oxidation behavior: Effect of temperature



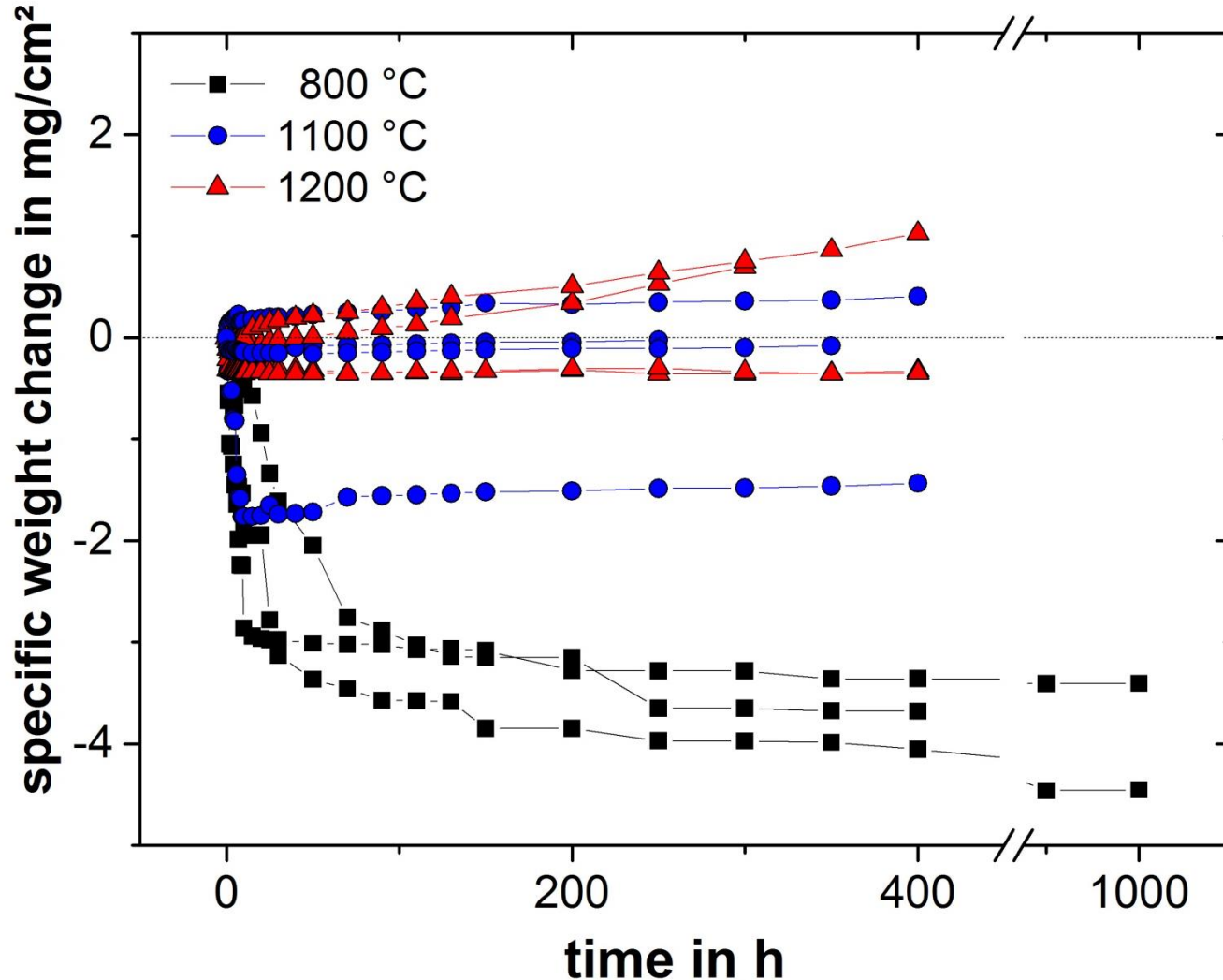
- Weight change
- Self-healing

# Oxidation behavior: Effect of temperature



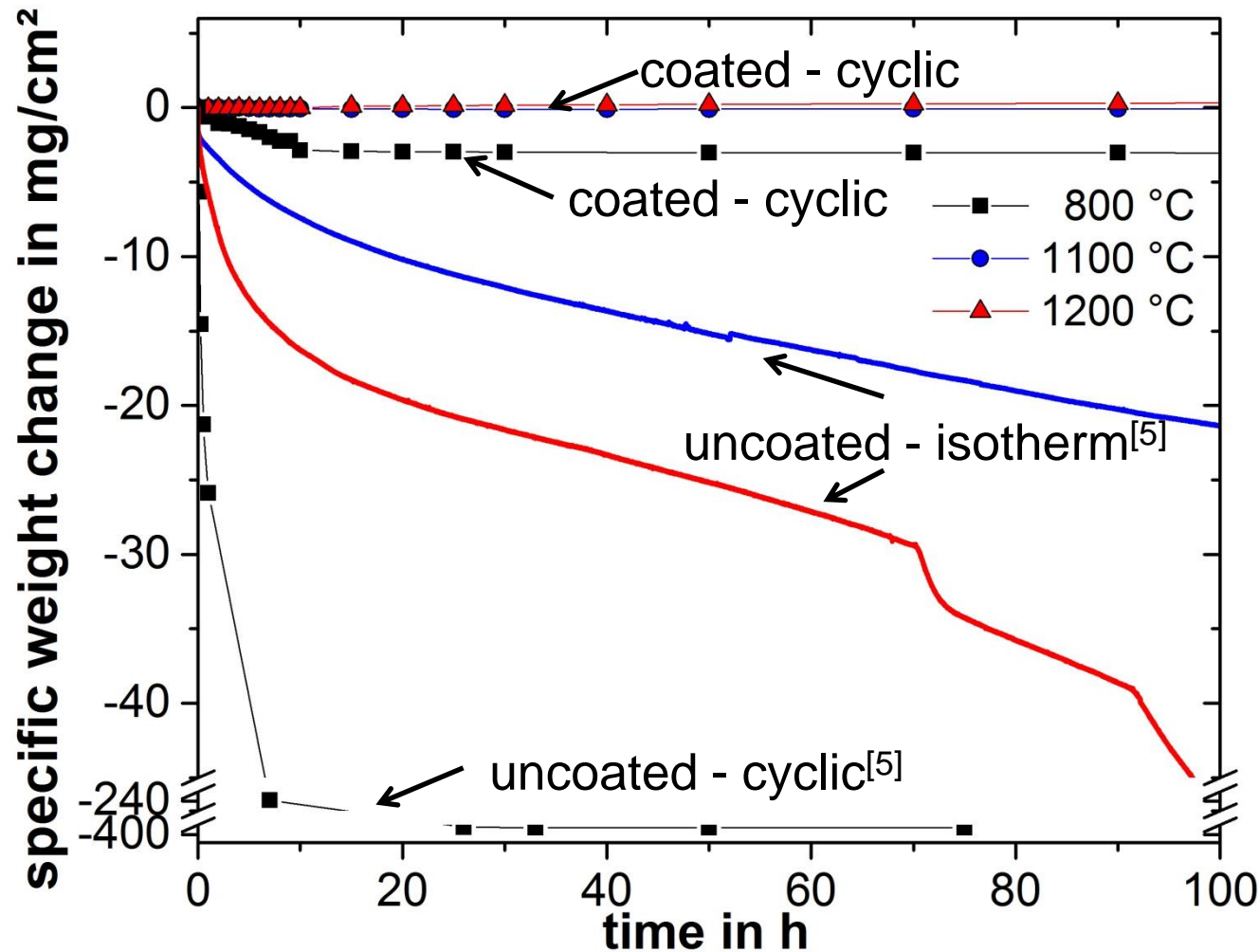
- Weight change
- Self-healing
- Temperature effect

# Oxidation behavior: Summary



- $T_{\text{oxidation}} \uparrow$
- less initial weight loss
- earlier transition to mass gain

# Oxidation behavior: benefit of coating



Significant improvement of oxidation resistance

[5] M. A. Azim et al., *Oxid. Met.* 80, 231 (2013)

# Summary

- Same Pack-cementation process used for ternary Mo-Si-B can be also applied to Ti containing alloy, but it has a benefit of density
- Increasing temperatures
  - initial mass loss decrease
  - earlier turnover to mass gain
- Formation of a three layer coating during conditioning
  - consisting of borosilicate,  $\text{MoSi}_2$  and  $\text{Ti}_5\text{Si}_3$
- Layers are intact after thermal cycling and no spalling

# Thank you for listening

## ACKNOWLEDGMENTS



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