

Spring 4-11-2016

# Assessment of the CO<sub>2</sub> capture potential from irreplaceable industrial sources

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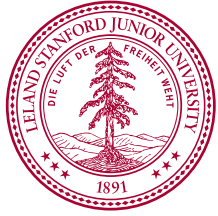
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## Recommended Citation

Peter Psarras and Jennifer Wilcox, "Assessment of the CO<sub>2</sub> capture potential from irreplaceable industrial sources" in "CO<sub>2</sub> Summit II: Technologies and Opportunities", Holly Krutka, Tri-State Generation & Transmission Association Inc. Frank Zhu, UOP/Honeywell Eds, ECI Symposium Series, (2016). [http://dc.engconfintl.org/co2\\_summit2/9](http://dc.engconfintl.org/co2_summit2/9)

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# Assessment of the CO<sub>2</sub> capture potential from irreplaceable industrial sources

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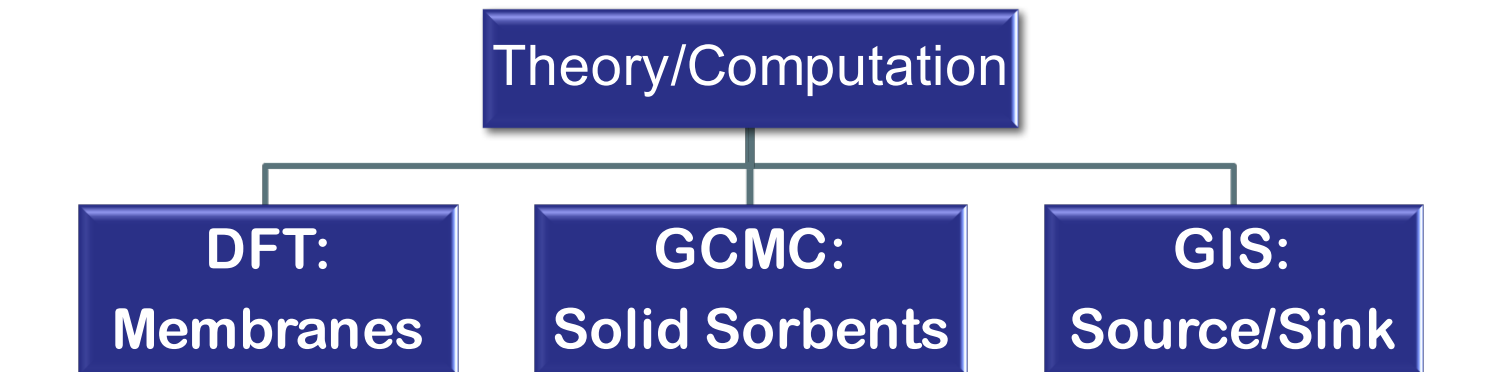
CO<sub>2</sub> Summit II: Technologies and Opportunities  
April 10-14, 2016 Santa Ana Pueblo, NM, USA

# Clean Energy Conversions Lab

[cec-lab.stanford.edu](http://cec-lab.stanford.edu)

Mission Statement:

*To design and understand trace metal and carbon dioxide transformation and/or capture on surfaces to prevent their release into the atmosphere.*

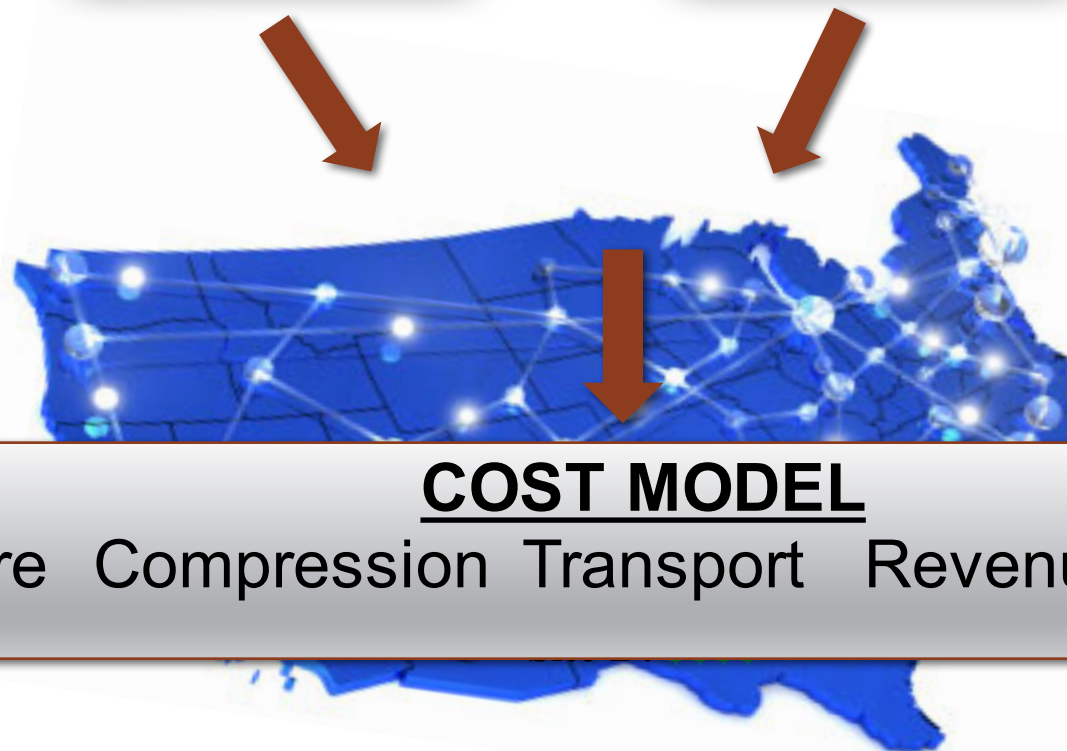


## Motivation – Phase II

Industrial Sources (Output)



Utilization Opportunities (Demand)

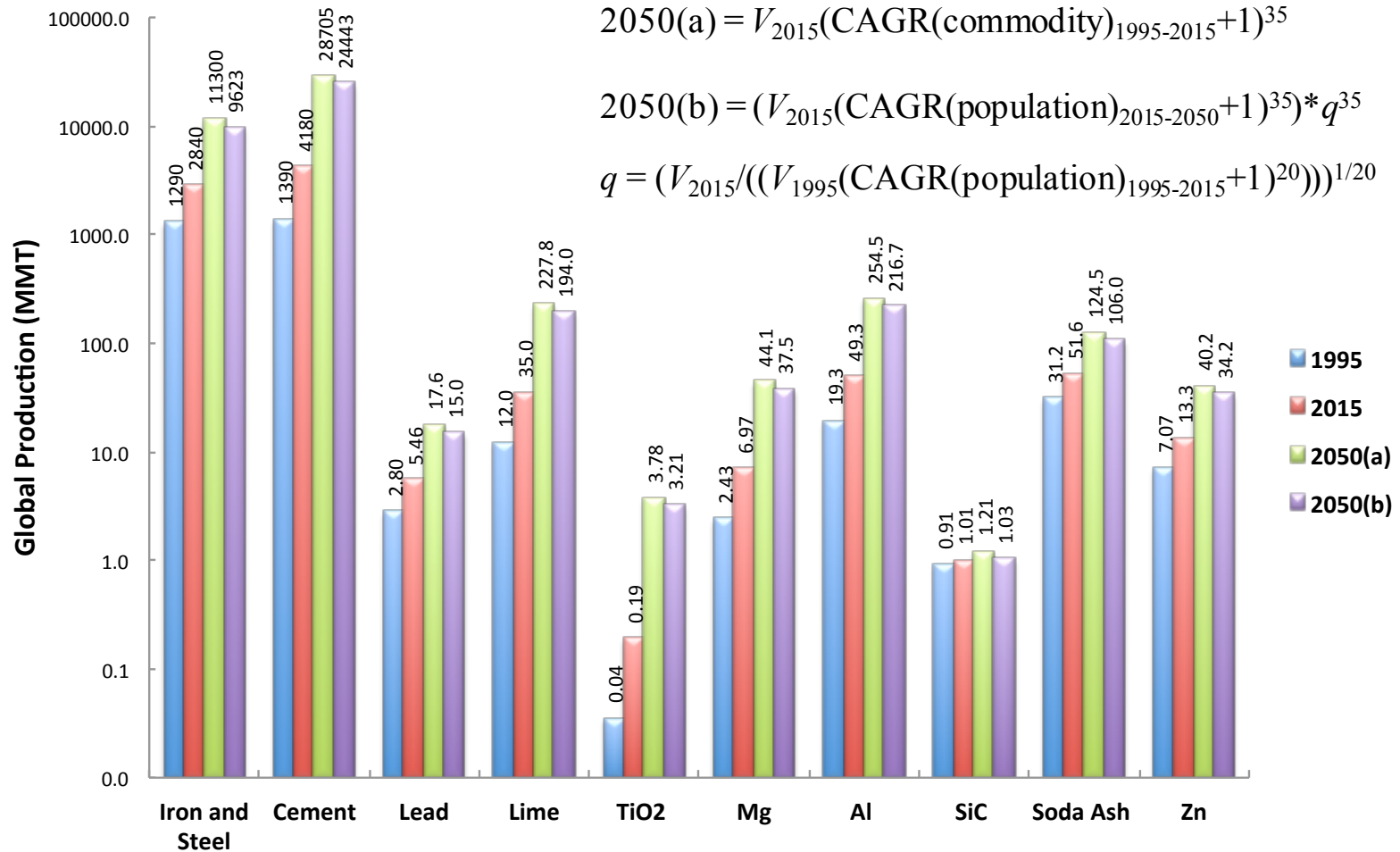


### **COST MODEL**

Capture   Compression   Transport   Revenue   Credit

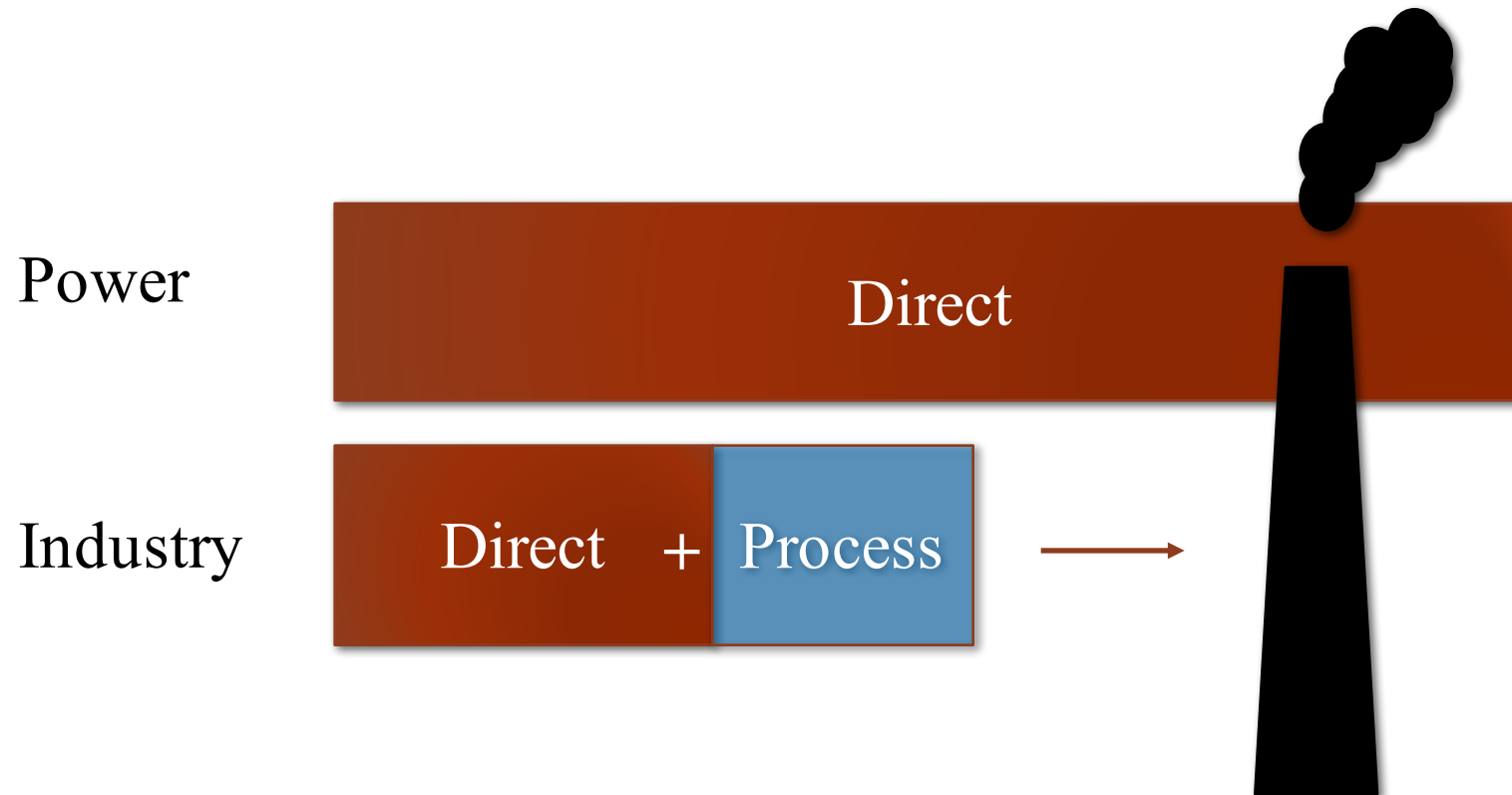
# Irreplaceable Industrial Processes (IIP)

1. Must produce CO<sub>2</sub> as a reactionary byproduct
2. Product is “essential”

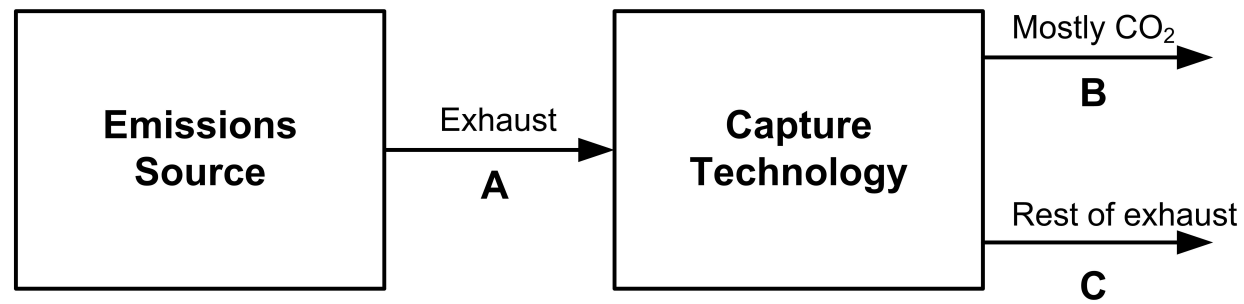


## Irreplaceable Industrial Processes (IIP)

1. Must produce CO<sub>2</sub> as a reactionary byproduct
2. Product is “essential”
3. There exists no \*reasonable carbon-free route to product or analogous product



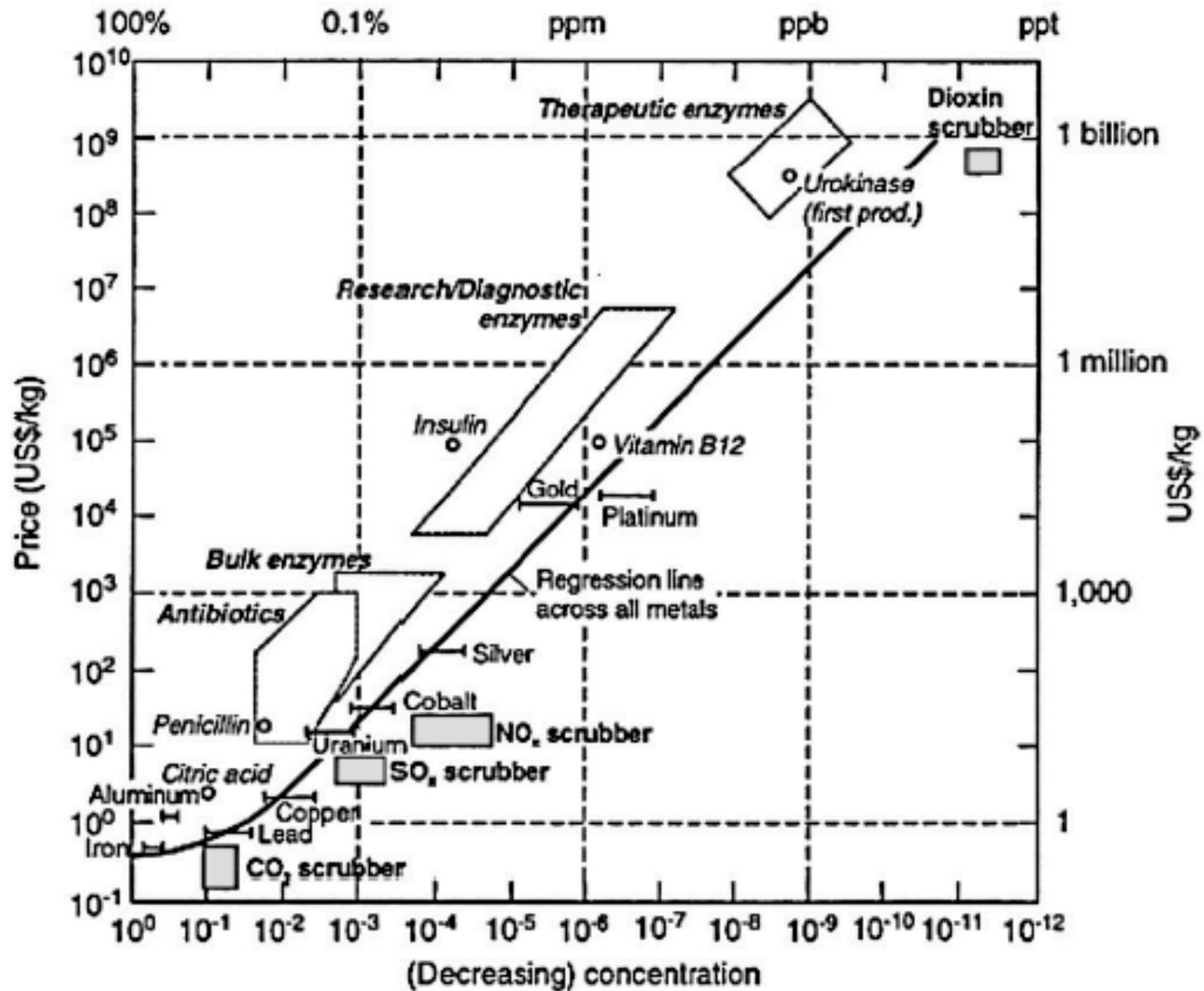
## Minimum Work - Separations



$$W_{\min} = RT \left[ n_B^{CO_2} \ln(y_B^{CO_2}) + n_B^{B-CO_2} \ln(y_B^{B-CO_2}) \right] + RT \left[ n_C^{CO_2} \ln(y_C^{CO_2}) + n_C^{C-CO_2} \ln(y_C^{C-CO_2}) \right] - RT \left[ n_A^{CO_2} \ln(y_A^{CO_2}) + n_A^{A-CO_2} \ln(y_A^{A-CO_2}) \right]$$

Wilcox, Carbon Capture, Springer, 2012

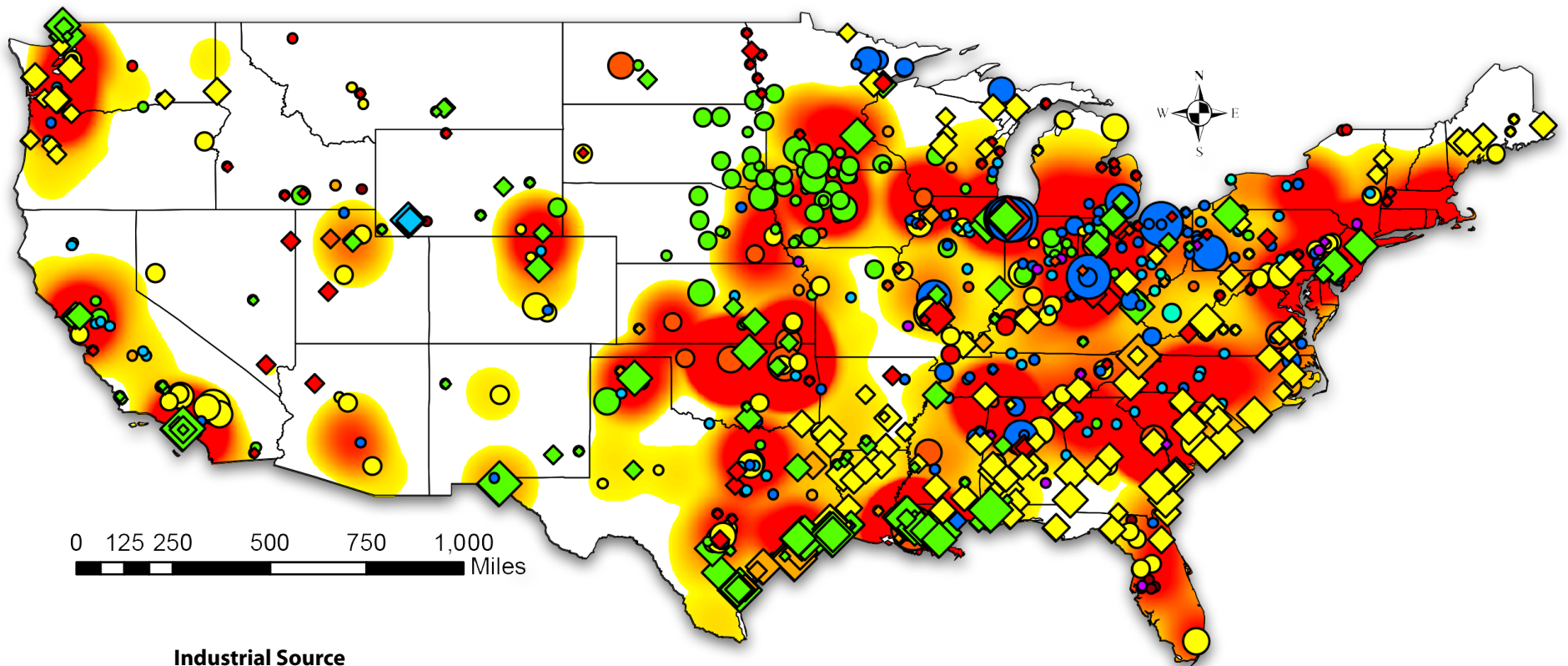
# Sherwood Cost Estimation



Grubler A (1998) Technology and Global Change (Cambridge Univ Press, Cambridge, UK).



# U.S. Source Distribution



0 125 250 500 750 1,000 Miles

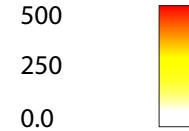
### Industrial Source

- |                |   |                  |   |
|----------------|---|------------------|---|
| Aluminum       | ● | Lime             | ◆ |
| Ammonia        | ● | Magnesium        | ◆ |
| Carbonate Use  | ● | Petrochemicals   | ◆ |
| Cement         | ● | Pulp and Paper   | ◆ |
| Ethanol        | ● | Refining         | ◆ |
| Ferroalloy     | ● | Silicon Carbide  | ◆ |
| Glass          | ● | Soda Ash         | ◆ |
| Iron and Steel | ● | Titanium Dioxide | ◆ |
| Lead           | ● | Zinc             | ◆ |

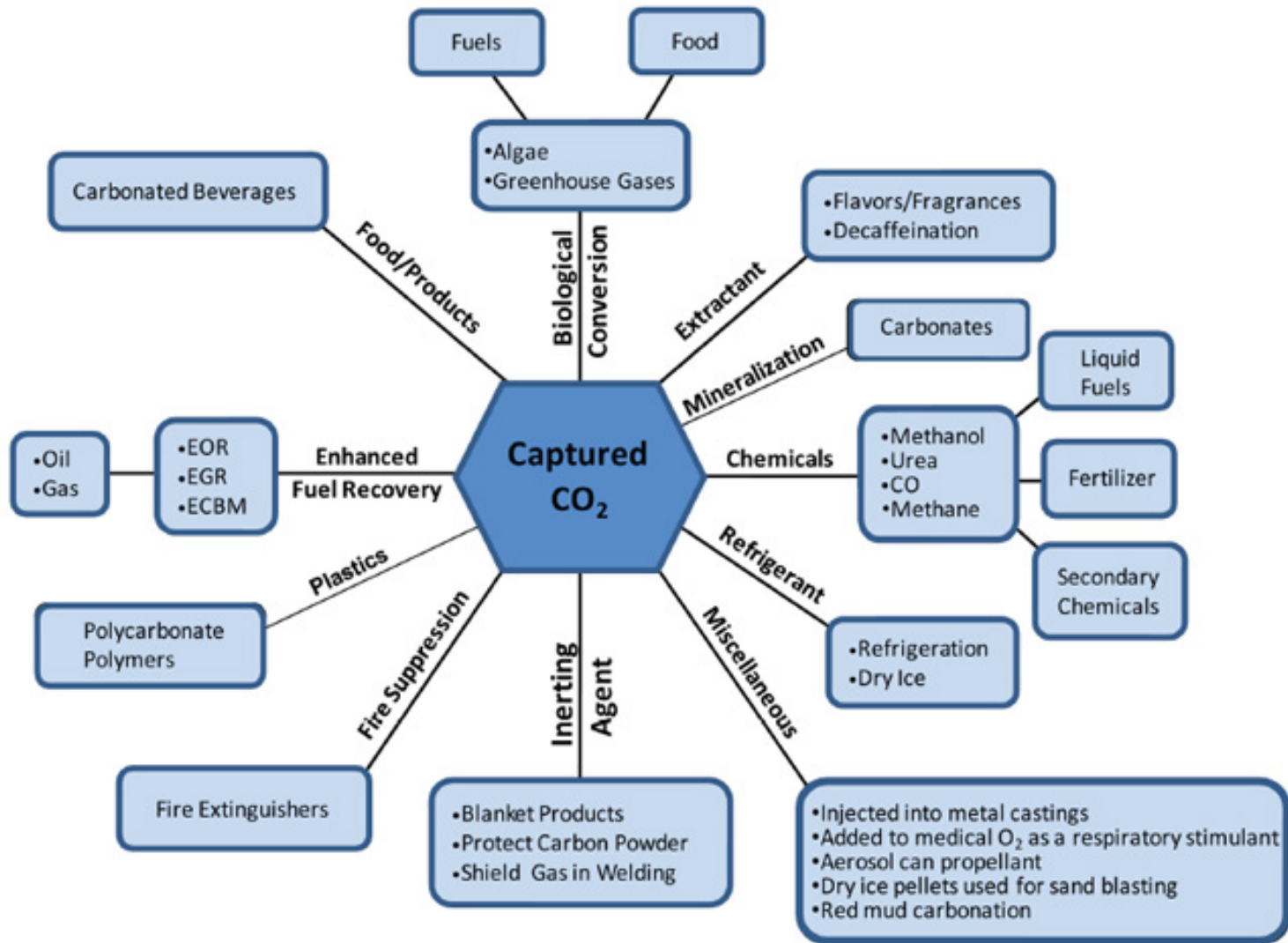
### Source Output (kt CO<sub>2</sub> per year)

- |                |   |   |
|----------------|---|---|
| < 500          | ● | ◆ |
| 150.01 – 1250  | ● | ◆ |
| 1250.01 – 3000 | ● | ◆ |
| 3000.01 – 6250 | ● | ◆ |
| > 6250         | ● | ◆ |

### Sink Demand (kt CO<sub>2</sub>)

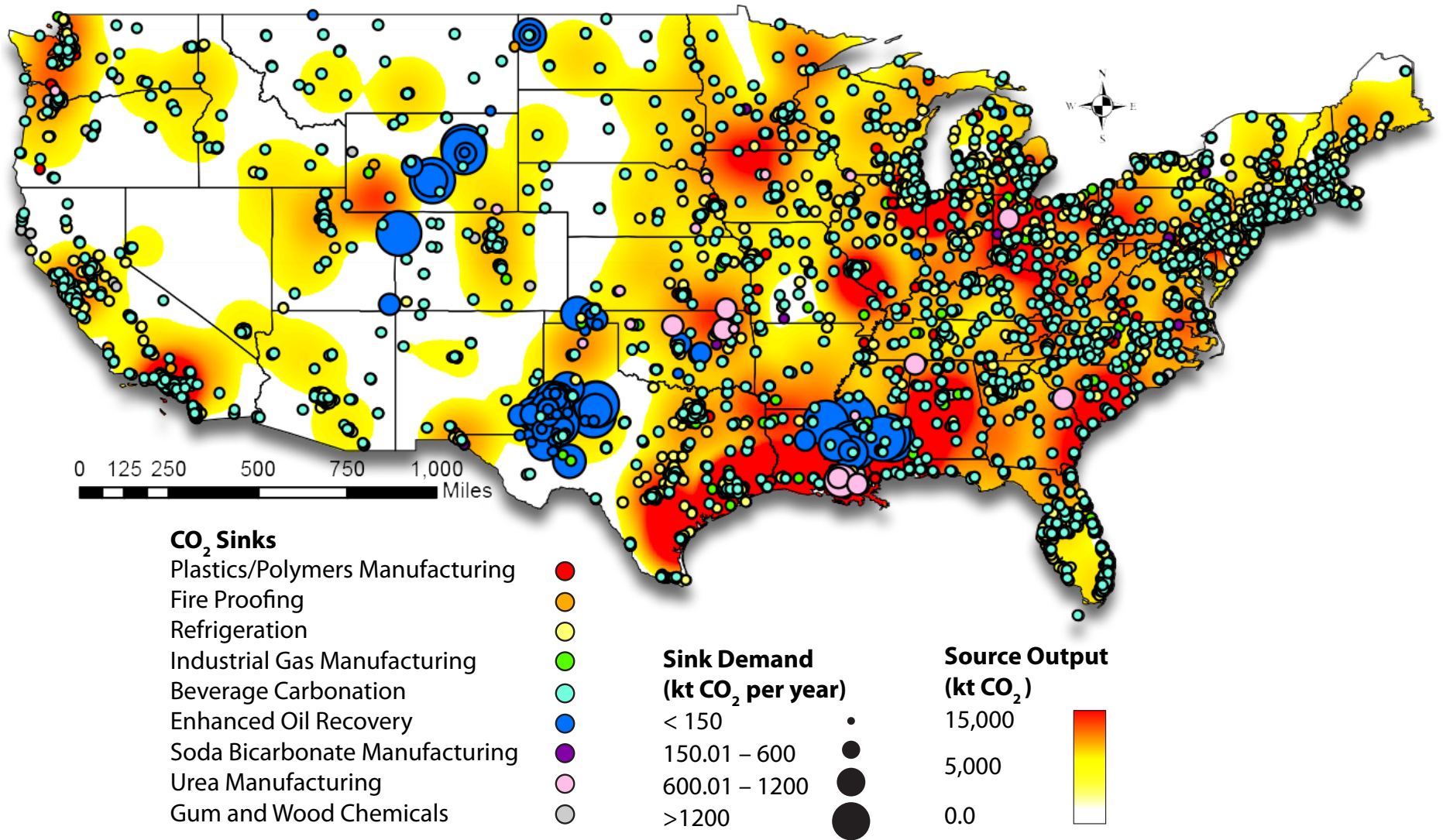


# CO<sub>2</sub> sinks – reuse opportunities



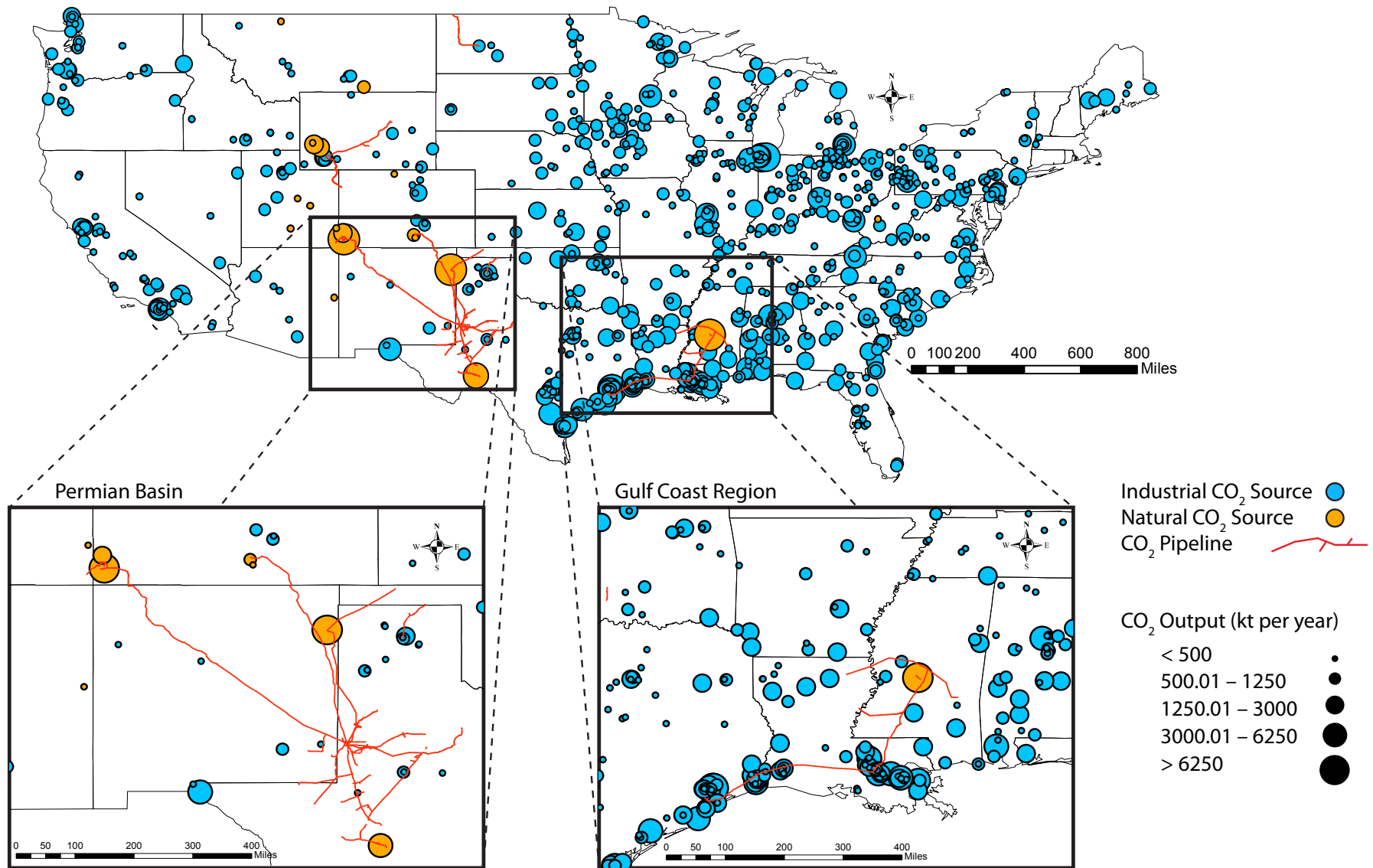
Reference : <http://www.netl.doe.gov/research/coal/carbon-storage/research-and-development/co2-utilization>

# US (Irreplaceable) Industrial Sources of CO<sub>2</sub>



Psarras et. al, *A Pathway Toward Removing CO<sub>2</sub> Emissions from the Industrial Sector*, in preparation

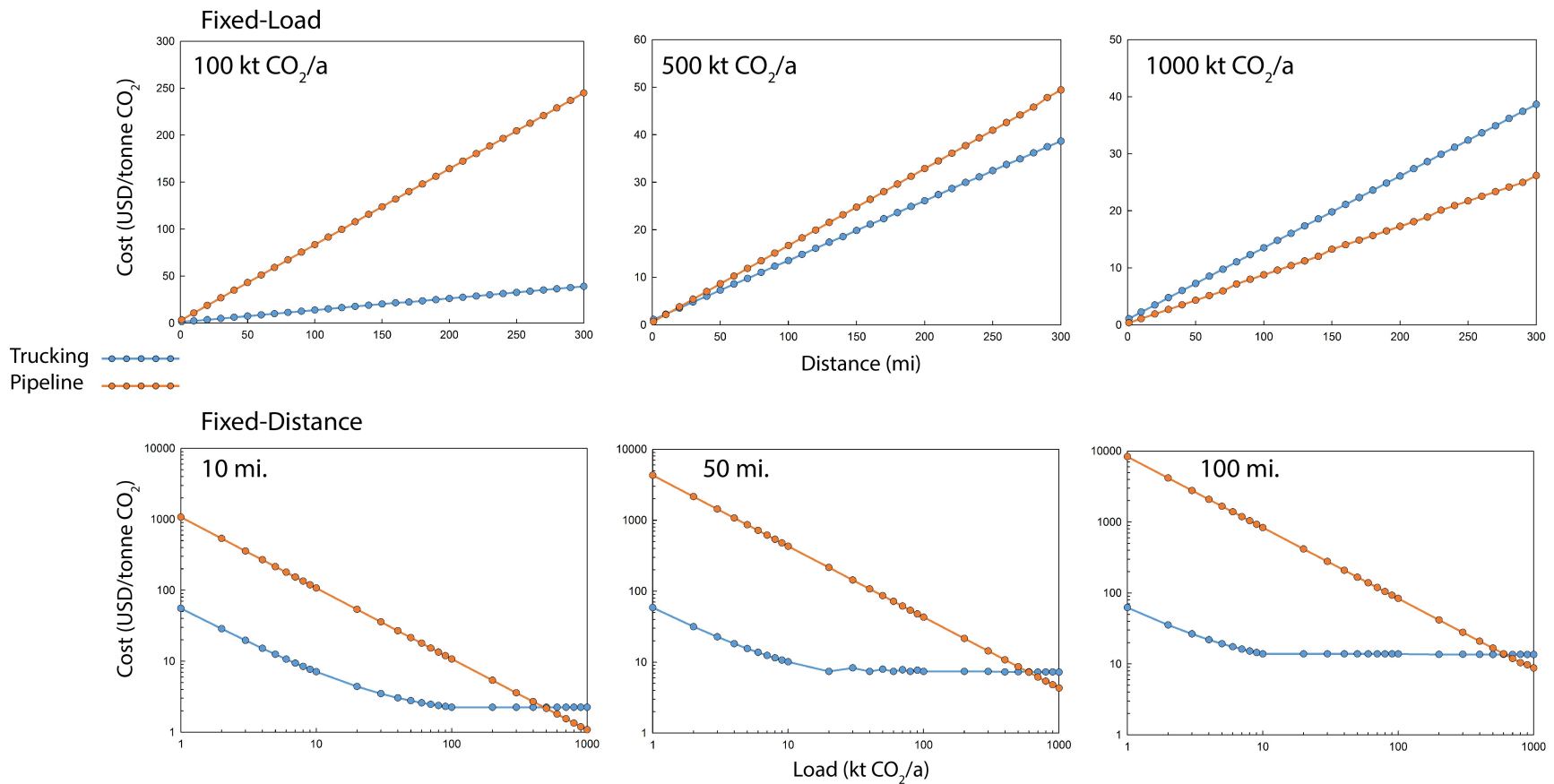
# CO<sub>2</sub> Utilization – EOR – Naturally Sourced



# Cost Model: Capture + Compression + Transport

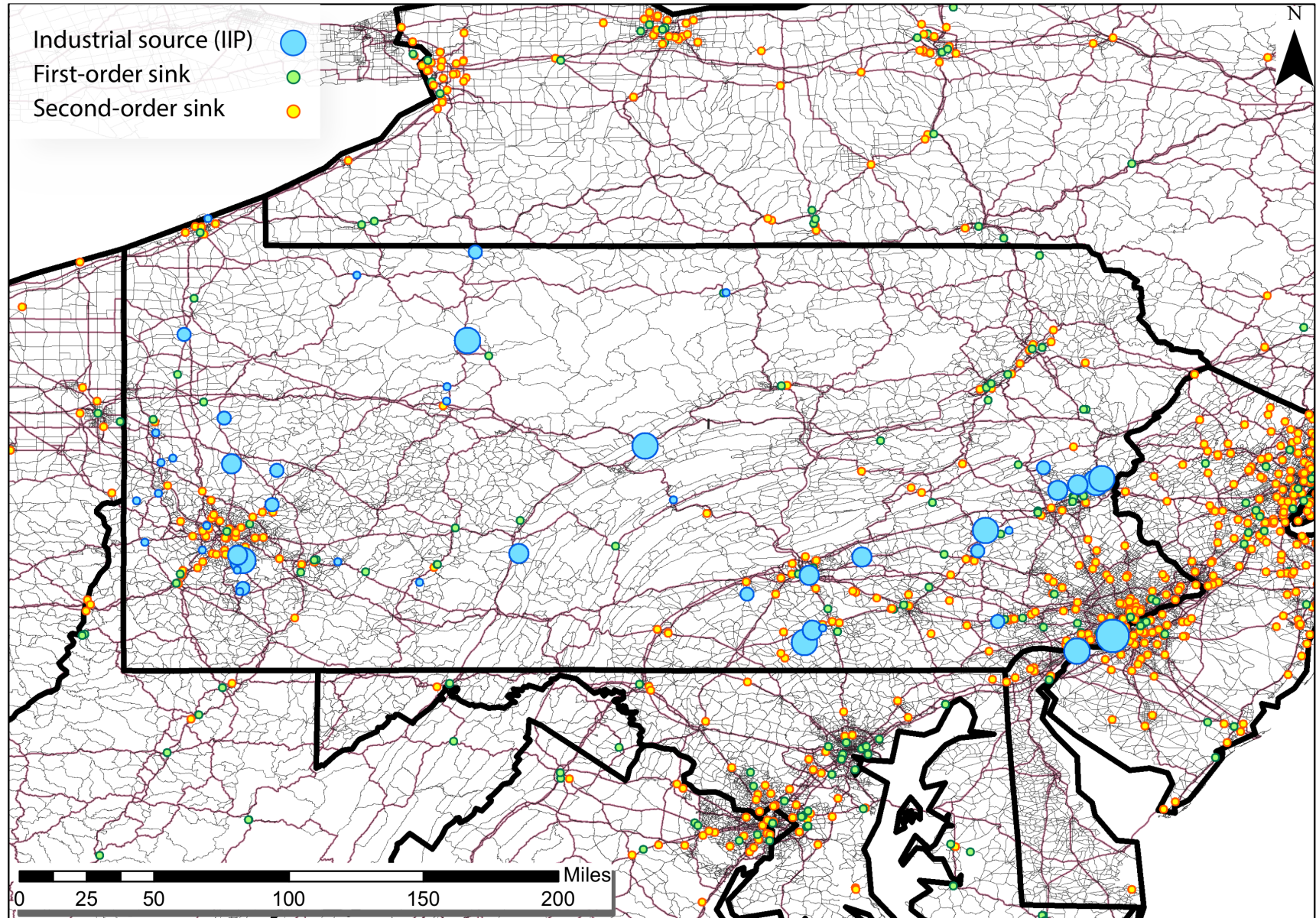
Mode	P (MPa)	T (K)	Compression		
			Power (kWh/tonne) <sup>a</sup>	Cooling Power (kWh/tonne)	Total Power (kWh/tonne)
pipeline	10	308.15	116.4	39.9	156.3
	15	308.15	119.4	39.9	159.2
tanker	1.7	243.15	90.3	16.5	106.8

<sup>a</sup> Estimated from compression work calculated per tonne CO<sub>2</sub> generated

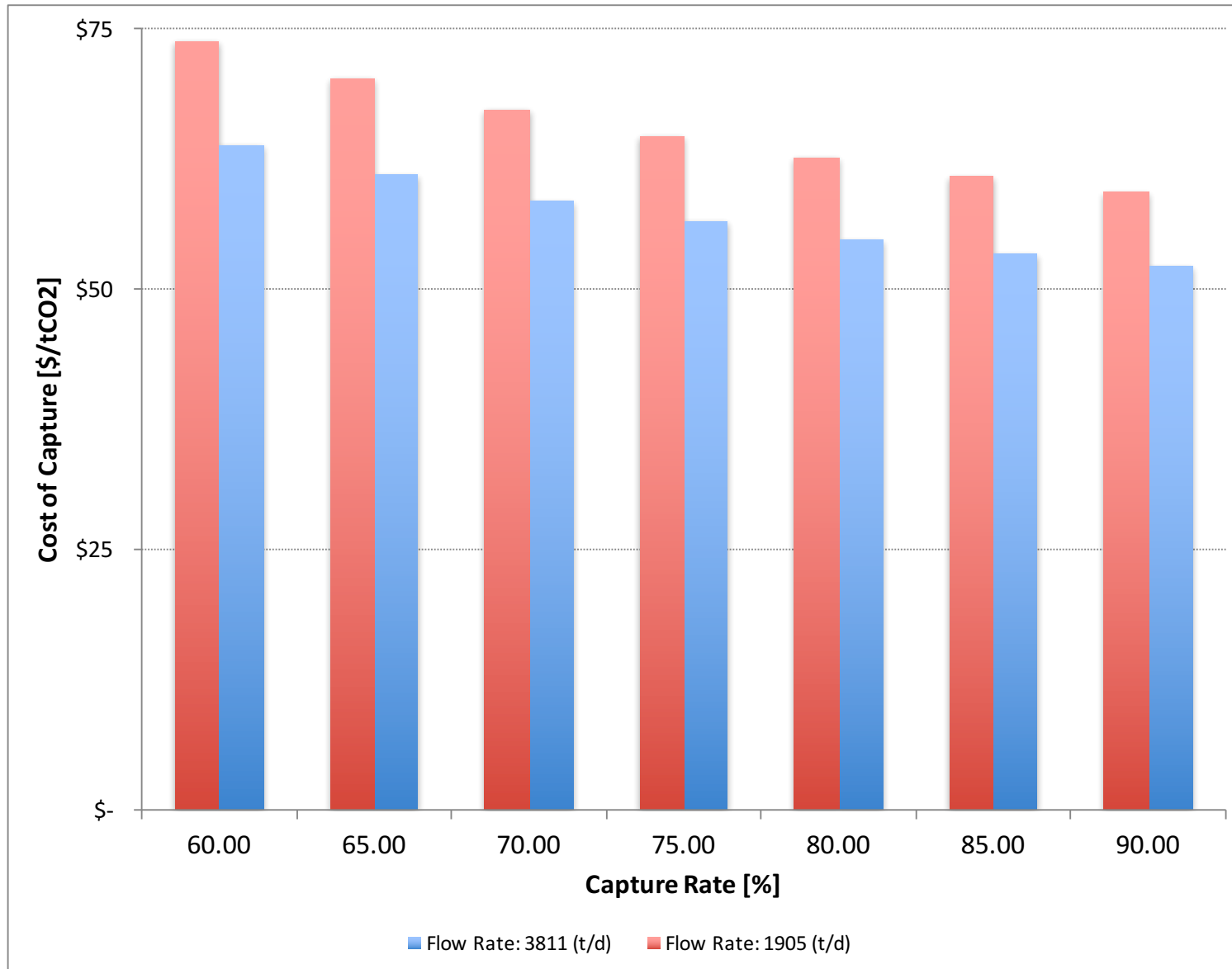




# Regional Level: Pennsylvania



# Future Directions – Adjustment to Cost Model





## Summary

A pathway forward to carbon neutrality in the Industrial sector should involve

- a) Development of a reliable cost-model
- b) Identification of “low-hanging fruit” for implementation of capture technologies. Will help to serve as a driver for learning and public acceptance
- c) Work to advance **CO<sub>2</sub> reuse opportunities**
- d) Work to increase the CO<sub>2</sub> mol % of flue streams via process redesign

### Acknowledgements

#### **Energy Resources Engineering:**

Jennifer Wilcox, Praveen Bains, Panunya, Charoensawadpong, Mark Carrington

#### **Graduate School of Business:**

Steven Comello and Stephan Reichelstein

David Medeiros – Stanford Geospatial Reference/Instruction Specialist