

5-26-2016

Segregation of equal-sized particles of different densities in a vertically vibrated fluidized bed

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E. Cano-Pleite, A. Acosta-Iborra, F. Hernández-Jiménez, T. Tsuji, and C. R. Müller, "Segregation of equal-sized particles of different densities in a vertically vibrated fluidized bed" in "Fluidization XV", Jamal Chaouki, Ecole Polytechnique de Montreal, Canada Franco Berruti, Wewstern University, Canada Xiaotao Bi, UBC, Canada Ray Cocco, PSRI Inc. USA Eds, ECI Symposium Series, (2016). http://dc.engconfintl.org/fluidization_xv/185

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Segregation of equal-sized particles of different densities in a vertically vibrated fluidized bed

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- Many operations in the chemical and energy-conversion industries rely on the fluidization of heterogeneous materials.
- During fluidization, particles of different densities can segregate even if they are of the same size.
- In mechanically vibrated fluidized beds, the oscillatory movement of the bed vessel affects the dynamics of the dense and bubble phases.
- This work aims to experimentally characterize density driven segregation in a vibrated fluidized bed.
- Experiments comprise a mixture of spherical particles of two different densities and similar diameter.

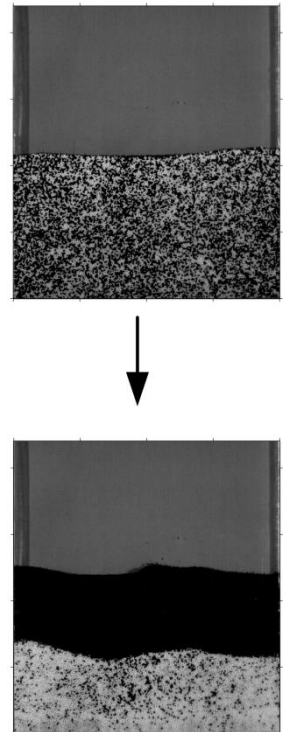
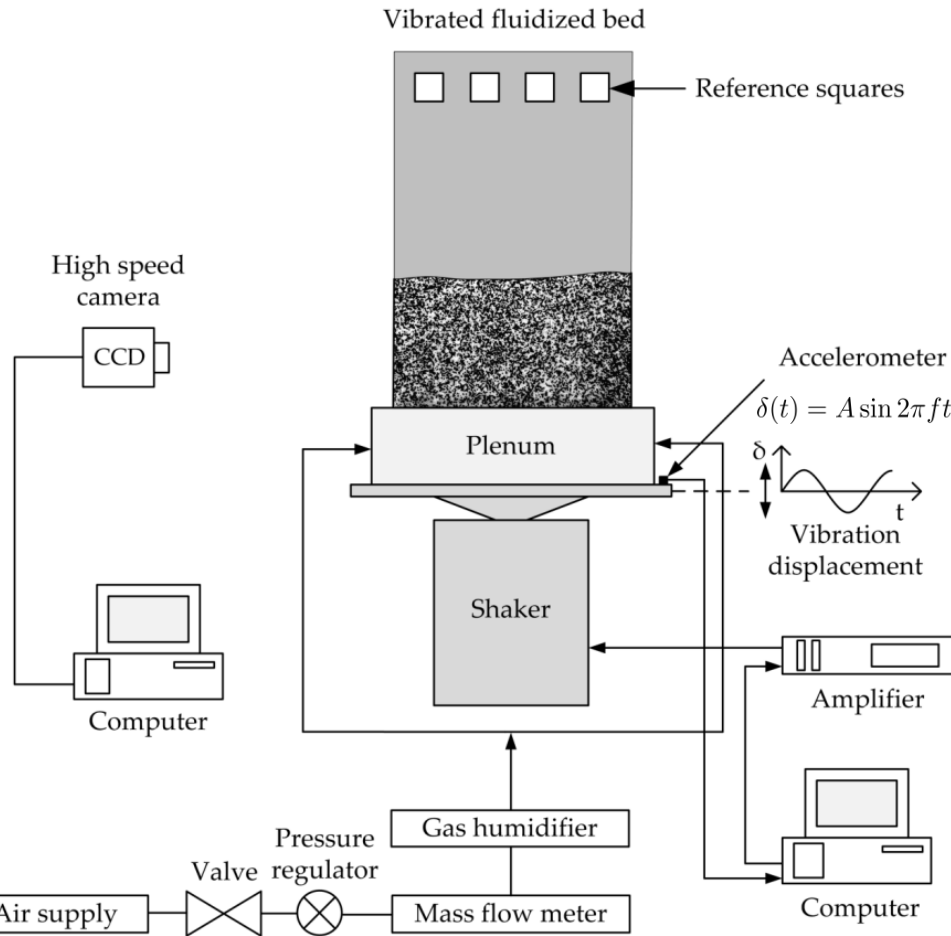


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Dimensions: 0.3 x 0.01 x 0.5 m

	$d_{p,L}$ (mm)	ρ_L (kg/m ³)	$d_{p,D}$ (mm)	ρ_D (kg/m ³)
Mixture 1	1–1.3	2500	1–1.2	6000
Mixture 2	1–1.3	2500	1–1.2	4100

White (dense) particles and black (light) particles are initially mixed.

Different combinations of vibration amplitude, frequency and superficial gas velocity. Central cases:

- $f = 15$ Hz, $A = 4$ mm, $U_o/U_{mf,D} = 0.91$ for Mixture 1.
- $f = 15$ Hz, $A = 4$ mm, $U_o/U_{mf,D} = 1.16$ for Mixture 2.

$$\Lambda = \frac{A(2\pi f)^2}{g}$$

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Digital Image Analysis was used to characterize the rate and extent of segregation with time

Segregation index based on Lacey's mixing index:

$$SI = 1 - \frac{\sigma_0^2 - \sigma_c^2}{\sigma_0^2 - \sigma_R^2}$$

$$\sigma_R^2 = \bar{c}(1 - \bar{c})/n_p \quad \text{perfectly random mixture}$$

$$\sigma_0^2 = \bar{c}(1 - \bar{c}) \quad \text{completely segregated mixture}$$

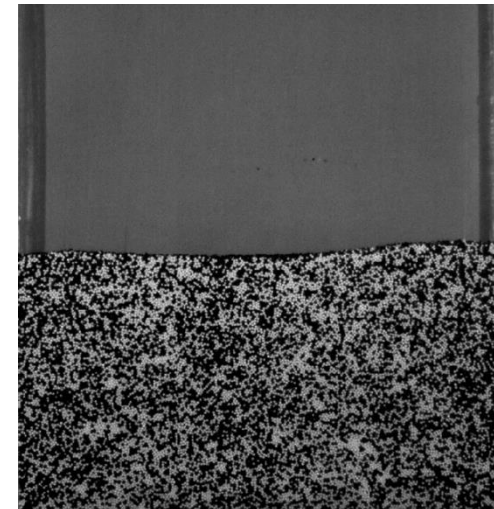
Variance of the concentration of white particles

$$\sigma_c^2 = \frac{1}{N_c - 1} \sum_{k=1}^{N_c} (c_k - \bar{c})^2$$

Bed divided in N_c cells of $N_x \times N_y$ pixels

$$c_k = \frac{1}{N_x N_y} \sum_{x_k=1}^{N_x} \sum_{y_k=1}^{N_y} \frac{G_{x_k, y_k} - G_{min}}{G_{max} - G_{min}} \phi_k$$

$$\bar{c} = \frac{1}{N_c} \sum_{k=1}^{N_c} c_k$$



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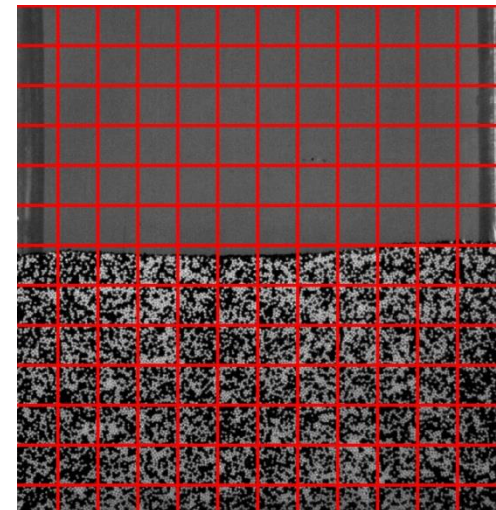
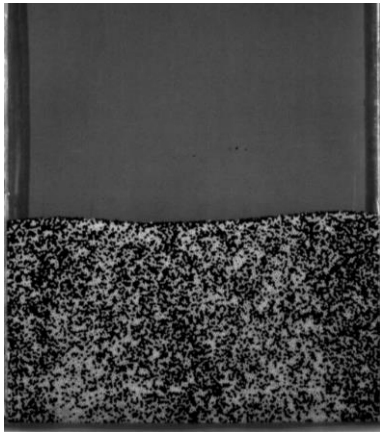


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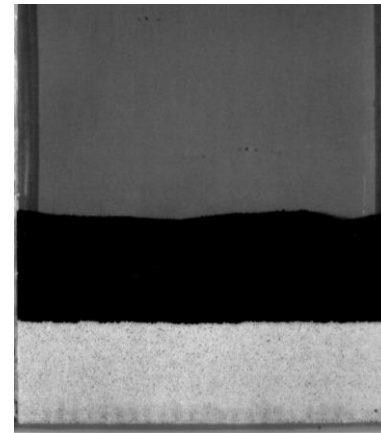
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Gas → Segregation



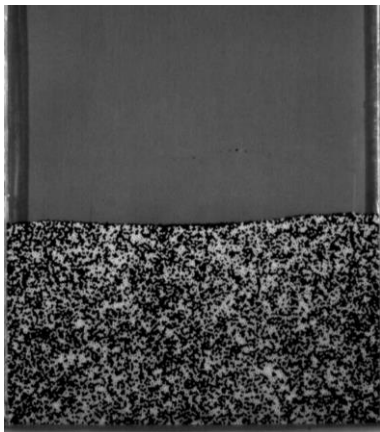
No vibration
 $U/U_{mf,D}=1.07$

Vibration → Vibration-induced mixing



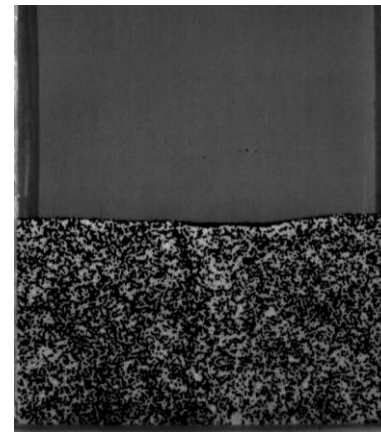
$f=15$ Hz
 $A=4$ mm
No gas

Low vibration strength + gas → Vibration induced segregation



$f=15$ Hz
 $A=4$ mm
 $U/U_{mf,D}=0.91$

High strength vibration + gas → Low segregation extent



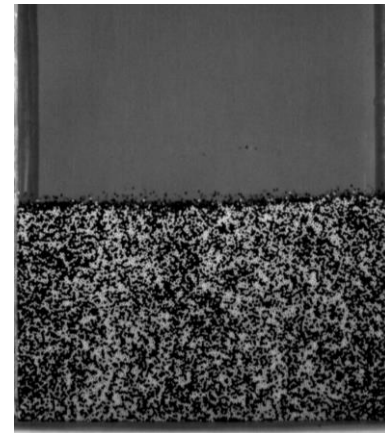
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 $A=4$ mm
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Gas → Segregation



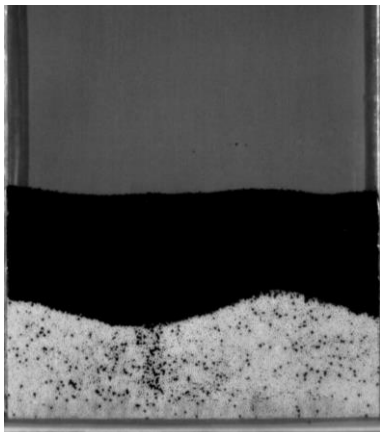
No vibration
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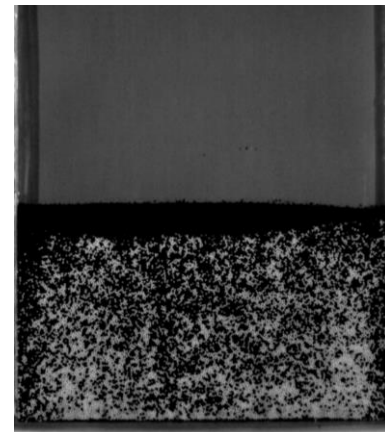
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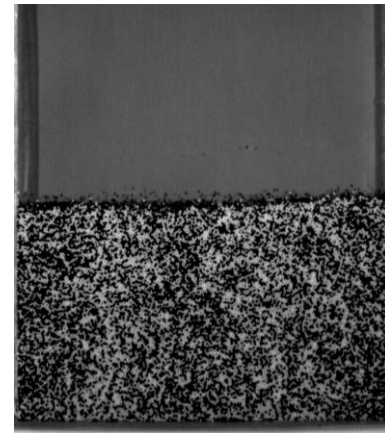
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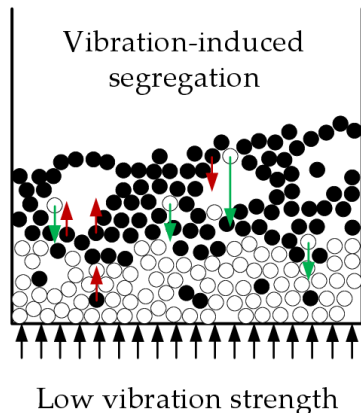
No vibration
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Vibration → Vibration-induced mixing



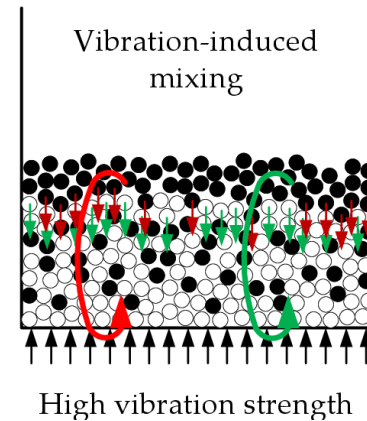
$f=15$ Hz
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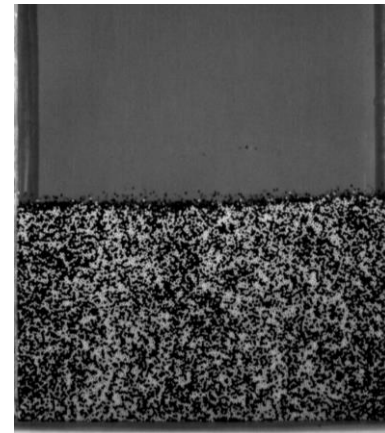
$f=20$ Hz
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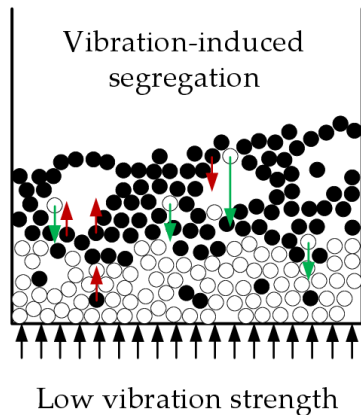
No vibration
 $U/U_{mf,D}=1.07$

Vibration → Vibration-induced mixing



$f=15$ Hz
 $A=4$ mm
 No gas

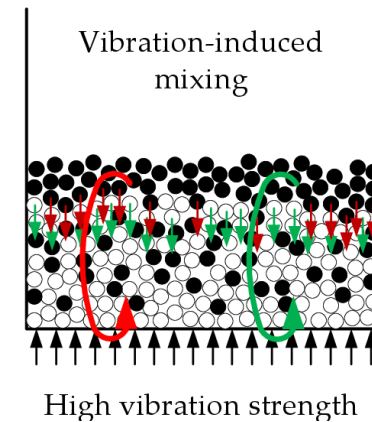
Low vibration strength + gas → Vibration induced segregation

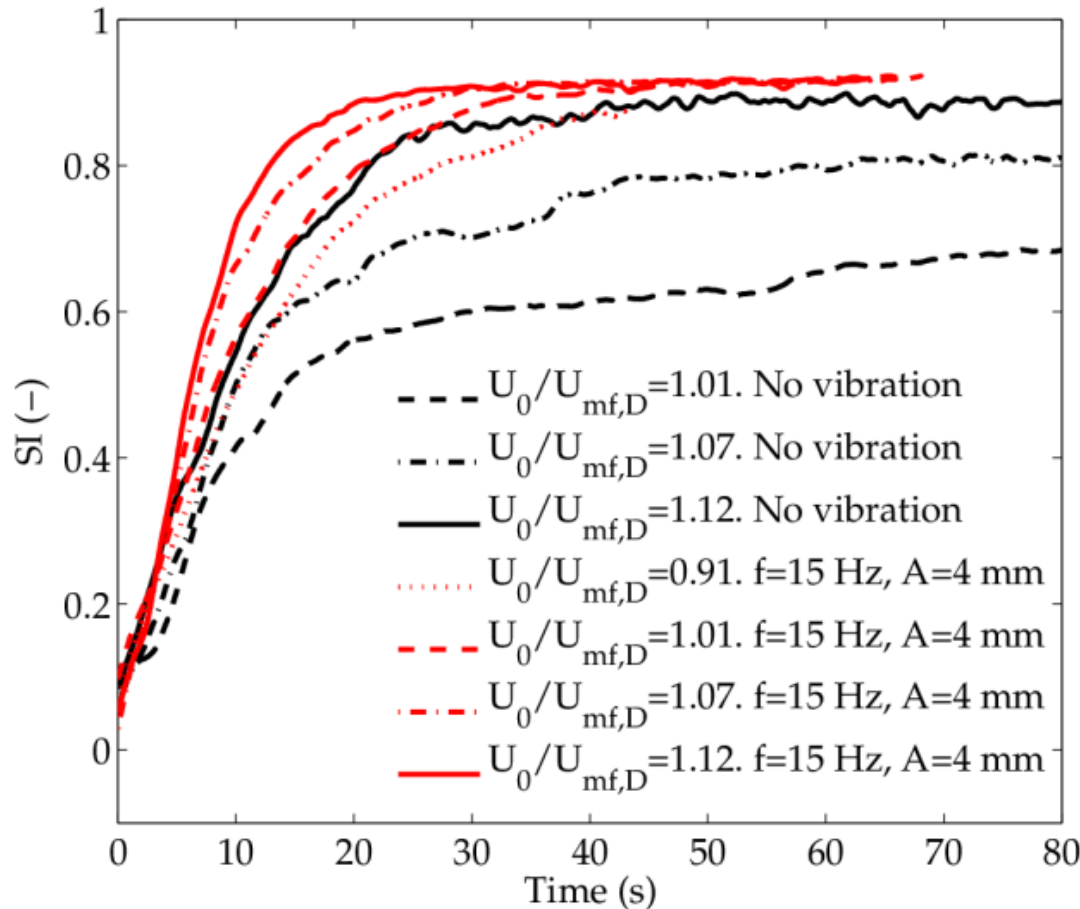


Transitional segregation

Mixture 2
 $f=15$ Hz
 $A=4$ mm
 $U/U_{mf,D}=1.1$

High strength vibration + gas → Low segregation extent

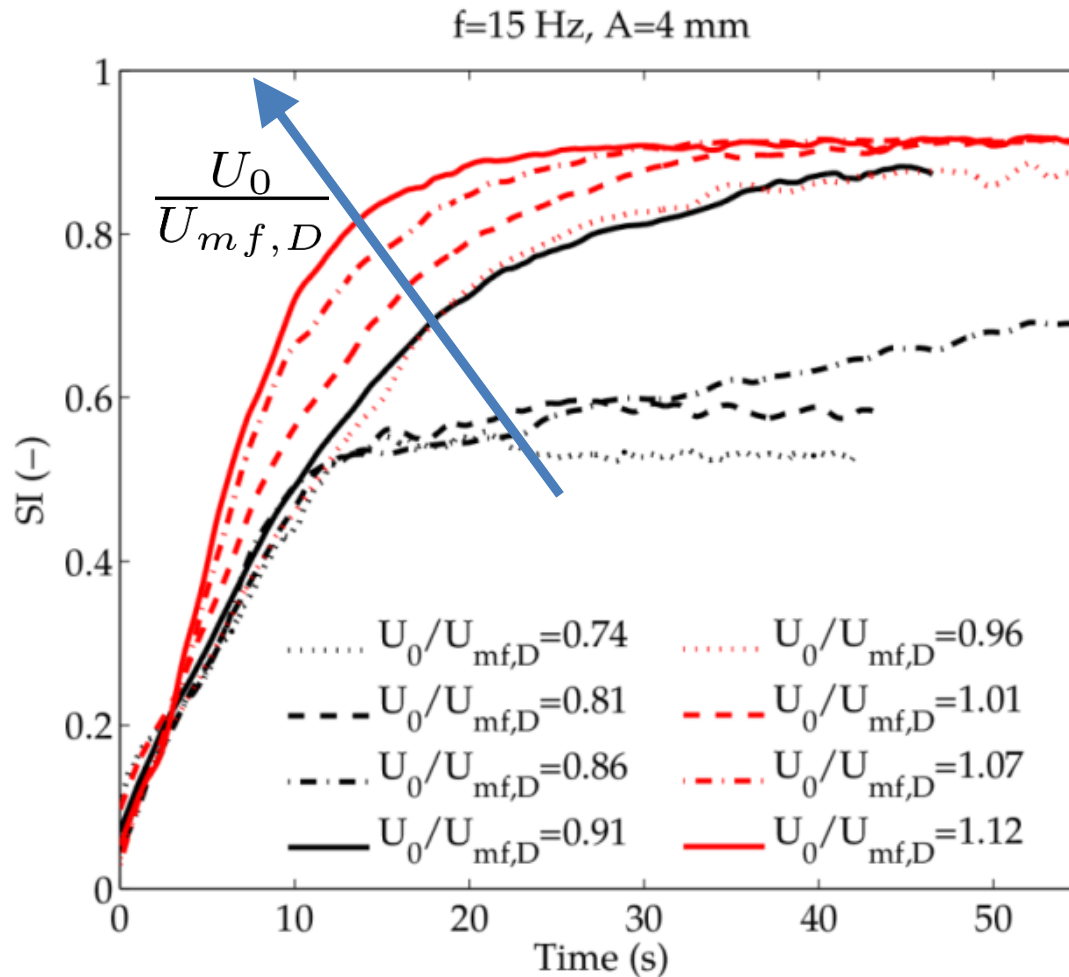




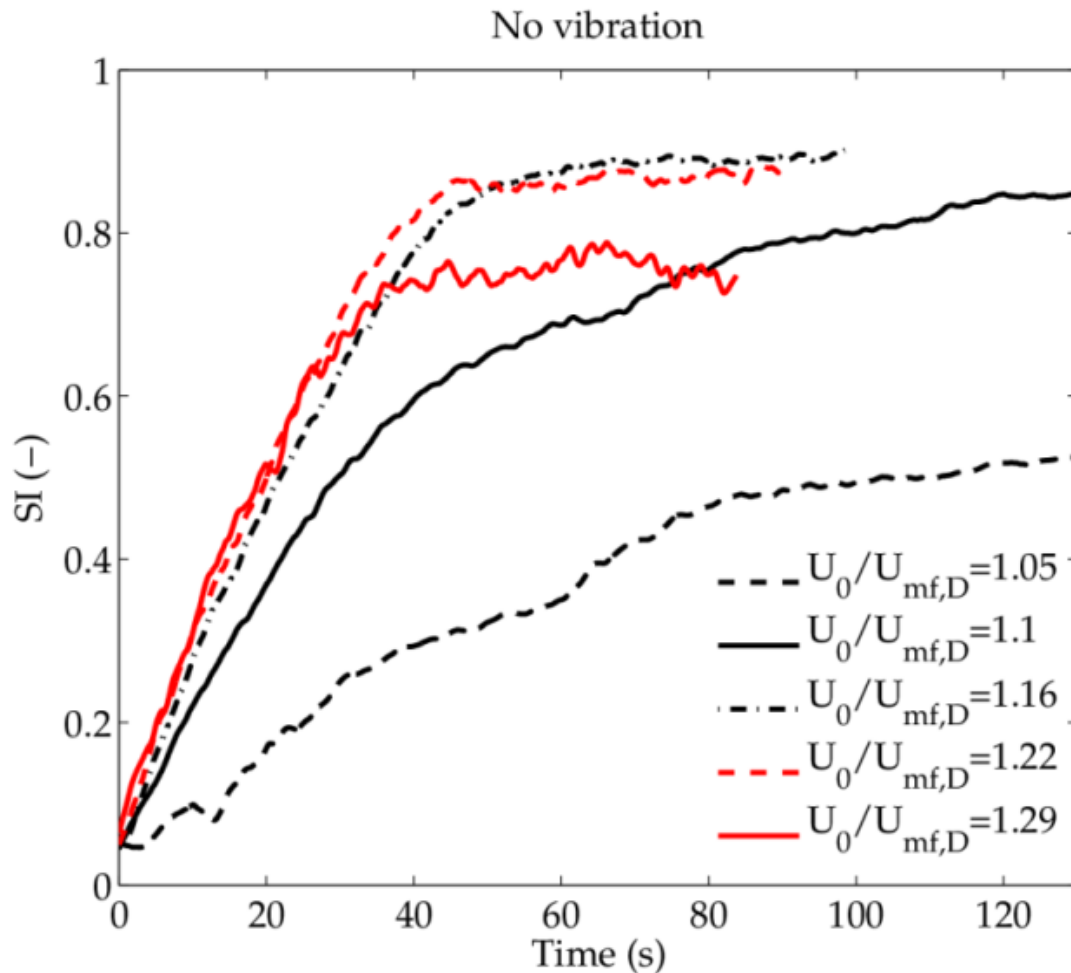
- Vibration allows segregation in cases in which $U_0 < U_{mf,D}$

- Vibration reduces the segregation time and increases the segregation extent.

- U_0 increases the segregation index and reduces the segregation time.



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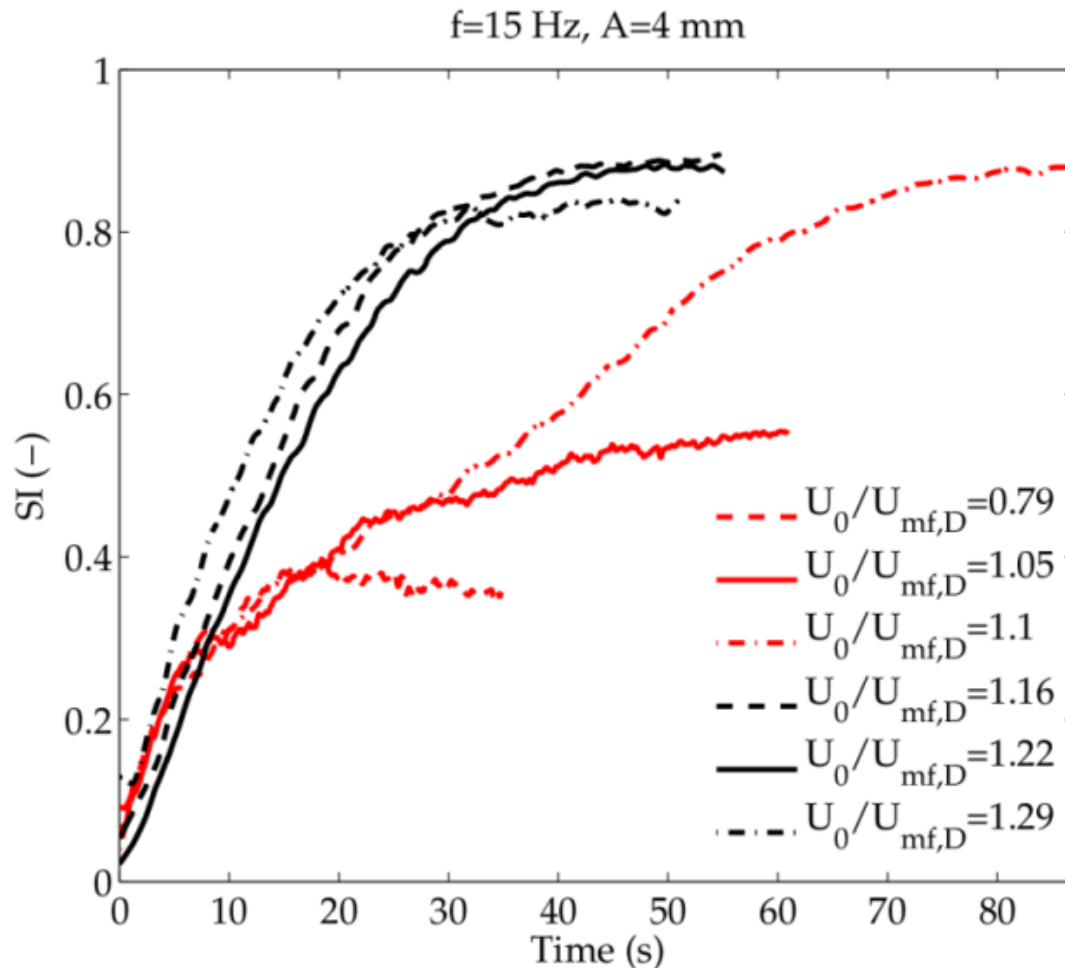


- Vibration allows segregation in cases in which $U_0 < U_{mf,D}$

- Vibration reduces the segregation time and extent.

- U_0 increases the segregation index and reduces the segregation time.

- Large gas velocities decrease the segregation extent.



- Vibration allows segregation in cases in which $U_0 < U_{mf,D}$

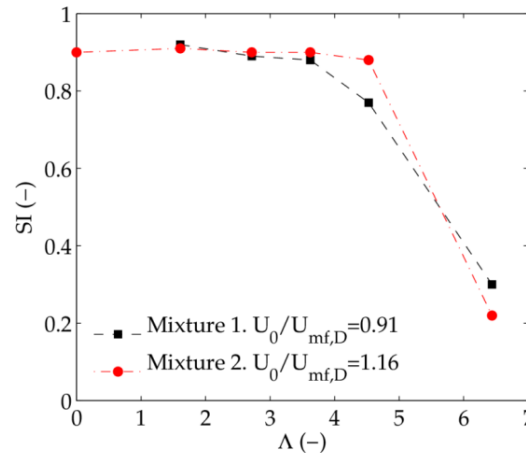
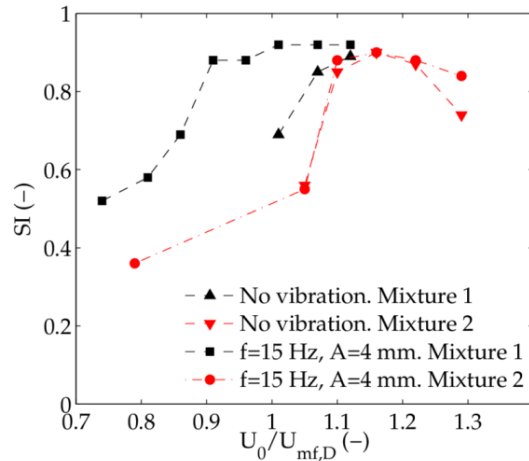
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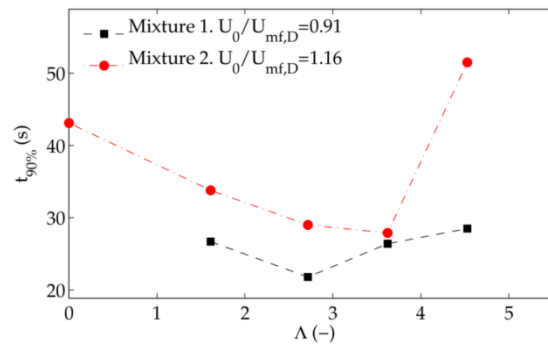
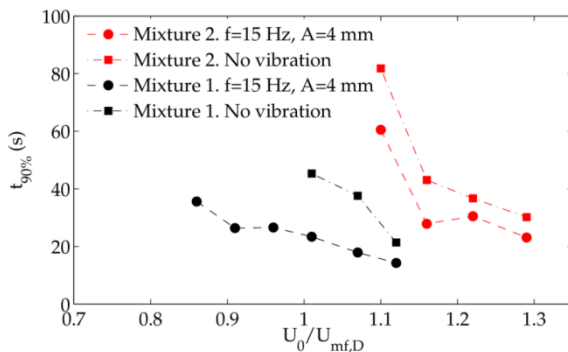
- **Transitional segregation**

Segregation extent at large t



- Vibration extends the range of large SI.
- Large vibration strengths decrease the final SI.
- Optimum combination of Λ and $U_0/U_{mf,D}$.

Time to reach 90% of the maximum segregation



- The segregation time decreases when increasing $U_0/U_{mf,D}$.
- Mixture 1 segregates faster than Mixture 2.

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- A segregation index was developed to quantify segregation from DIA data.
- The gas superficial velocity and the vibration of the bed vessel possess counteracting effects on segregation.
- Gas superficial velocity enhances segregation.
- Low vibration strengths cooperate with gas and enhances segregation.
- High vibration strengths promote mixing.
- There is an optimum combination of Λ and $U_0/U_{mf,D}$ to maximize segregation.



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Thank you for your attention

This work has been partially funded by the Universidad Carlos III de Madrid (Ayudas a la movilidad 2015) and by the Spanish Ministry of Economy and Competiveness (project ENE2015/00188/001)