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Ultra-High Temperature Mechanical Properties of ZrB₂-Based Ceramics

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- Dr. Eric Neuman
 - Dissertation research
- Drs. Jeremy Watts, Harlan Brown-Shaklee, and Eric Neuman
 - Building UHT mechanical testing system
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MISSOURI **SATHigh Melting Temperature Materials**

법운 30 ⊢ nitrides NbC specific gravity 0 silicides orides ZrC 3500 € sulphides silicates Ta,C catoldes ide HfN Re HfB, ThO, 5 ပ္ TaB. ure, TaN 0s ZrB, Melting Tempera ŦſŇ ReW Re,W UO, HfÓ, TIB, 2 WC W.C 3000 S.V. Ushakov and A. 2000 WB. melting temperature °C Navrotsky, "Experimental ZrB vc Re₂Ti, SIC ZrO, Mn,B, Approaches to UN ThN ThC, ThC Ta.B. CaO Thermodynamics above Mo BeO UB, LaB, OsTa, PrC, 1500°C," Journal of the Ta,Si, 2500 ← •Ru Sc.O, WPt Hf.Si. CeS American Ceramic llr. 腦 r,Nb Y,O, B,C Nb,N AlN Nb Society, 95(5) 1463-1482 W,Si, ♦lr,Ti UC, La,O, ThS HfMo, Cr,O, Hf,Si, Sr,SiO, NdB NbN VN Zr,Si, ThB, Mo,B CeB, PrS B Hf GdC, Gd.O. TaSi, OsTi Mo.A V,C HfS V,Si, Ca,SiO, IrTa Ni,C Th,N, MgAl_sO, Al_sO, Pr,S, Mo,Si, RuZr Mg,SiO, CrB La,C, Ta,N AI,Mo MoSi 2000 < Borides Elements Carbides Oxides Silicates Silicides Sulfides InterMe Nitrides

(2012).





Motivation and Purpose

Intrinsic properties

- Measuring is simple conceptually, but difficult in practice
- Thermal conductivity
 - Impurity effects
 - Lack of single crystals
- Heat capacity
 - Historic data are inaccurate

Systematically study the mechanical behavior of ZrB₂-based ceramics at ultra-high temperatures





Historic Strength Studies



Development of Oxidation-Resistant Diborides: Mechanical Properties" AFML-TR-68-190 Part II, Vol IV.



- Limited studies of ZrB₂ at elevated temperatures
- ZrB₂ strength decreases as grain size increases
- Diffusional creep limits strength at the highest temps





- More studies than pure ZrB₂, but few over 1600°C
- ZrB₂-SiC eutectic at 2270°C limits upper use temperature



- Disilicide additions can improve strength compared to SiC
- No studies of ZrB₂ with silicides above 1500°C
- Use limited by T_{melt} (MoSi₂ 2030°C; TaSi₂ ~2200°C)



MISSOURI SET

UHT Mechanical Test System

- Instron 33R4204 load frame
- Custom-built environmental chamber
 - Inert atm or mild vacuum
 - $\sim 10^{-14}$ atm pO₂ using Ar
- Induction heating system
 - Capable of 2600°C
 - Heating rate of 100's°C/min
- Graphite load train and test fixture
- Testing limited by phase stability
 Ex: ZrB₂-C eutectic at 2370°C





*Neuman, et al., Am. Ceram. Soc. Bull., 92[1] 36-38 (2013)





Microstructures

ZrB₂



99.2% dense

Grain size: 19.7 ± 13.0 µm

Held at 2150°C for 1 hr

 Grow grains and reduce creep above 1800°C

Minimal entrapped porosity No residual carbon

ZrB₂-30SiC



>99.9% dense

ZrB₂ grain size: $1.9 \pm 0.9 \mu m$ SiC cluster size: $6.1 \pm 4.4 \mu m$ Max SiC cluster size: $59.1 \mu m$ Micrcracking threshold ~ $15 \mu m$





- Air: Strength 300-400 MPa up to 1200°C, ~200 MPa above (oxidation)
- Inert: Strength stabilizes at ~200 MPa up to 2300°C (creep?)
- Critical flaw size consistent with grain size at elevated temperatures
 - Grain growth during tests at 2000°C and above

Neuman, et al., J. Am. Ceram. Soc., 96[1] 47-50 (2013)





- Air: σ increases ≤1000°C (flaw healing), decreases >1200°C (oxidation)
- Inert: σ >550 MPa up to 1800°C, decreases to 2200 MPa at 2200°C
- Both: K_{IC} decreases steadily from ~4 MPa•m^{1/2} at RT to ~3.5 MPa•m^{1/2}

Neuman, et al., *J. Euro. Ceram. Soc.*, 33[15-16] 2889-2899 (2013) Neuman, et al., *J. Euro. Ceram. Soc.*, 35[2] 463-476 (2014)





- Below 1000°C, SiC clusters are the critical flaw
- Oxidation damage controls strength at ≥1200°C (3 pt. bending of bar halves)
- Observed fracture origins are consistent with size of calculated critical flaws
- Improve strength by reducing SiC cluster size or reducing oxidation damage

XISSOURI ZrB₂-SiC Fracture in Argon

- ~20% intergranular fracture at RT
- Fraction steadily increases with temperature
 - Trend contrary to historic reports
 - SiC cluster size grows >1800°C
- Lower temp. tests do not predict UHT strength







- Fe, Co, U containing phase present following hot pressing
 - Appears wetting to both ZrB₂ and SiC
 - Possibly a silicide phase Reduction of SiC by Fe and Co



MISSOURI SCICAN We Increase Use Temperature?



What are we looking for?

- Chemical stability at 2000°C+
- Second phase additions
 - Improve properties
 - Enhance densification
- Higher eutectic temps
 - ZrB₂ SiC at 2270°C
 - ZrB₂ ZrC at 2660°C
 - ZrB₂ ZrC_{0.88} at 2830°C

- HfB₂ - HfC_{0.9} at 3140°C







- >99% dense
- ZrB_2 grain size 4.9 ± 3.0 µm
- ZrC cluster size 1.8 ± 1.5 μm
- Max ZrC cluster 9.8 µm
- 9.5 ± 0.4 vol% ZrC
- 0.10 ± 0.02 vol% C
- 0.05 ± 0.02 vol% porosity





- Strength decreases from ~700 MPa at 800°C to ~300 MPa at 1600°C
 - Maintains ~300 MPa up to 2300°C
- ~4.5 MPam^{1/2} at RT with no discernible trend

- Minimum about temperature for relaxation of residual stresses

Neuman, et al., submitted



- Two regimes of failure behavior
 - 1400°C and below strength limited by machining damage
 - 1600°C and above controlled by subcritical crack growth
- Both related to original flaw population induced by machining



MISSOURI Can We Get Higher Strengths?

- Materials
 - Higher eutectic temperatures
 - HfB₂-HfC, (Hf,Ta)C, others?
- Microstructures
 - Fully dense, no microcracks
 - Grain size?? (creep vs σ)
 - No grain growth at UHT??
- Impurities
 - Eliminate undesired impurities (transition metals, oxides, etc.)
 - Effects of doping or pinning??

We have only scratched the surface for UHT structure-property relationship studies



