#### A Novel Membrane Reactor for the Continuous Production of Biodiesel

Marc A. Dubé, André Tremblay and Peigang Cao Dept. of Chemical Engineering University of Ottawa, CANADA



#### Outline

- Background
- Challenges to biodiesel quality and profitability
- A solution to the challenges!
- Results, concluding remarks



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#### Background

 Biodiesel is a fuel comprised of mono-alkyl esters of long chain fatty acids derived from lipid feedstock (e.g. vegetable oils and/or animal fats).

Lipid (TG) + alcohol

 Fatty acid alkyl ester (FAAE or biodiesel) + glycerin

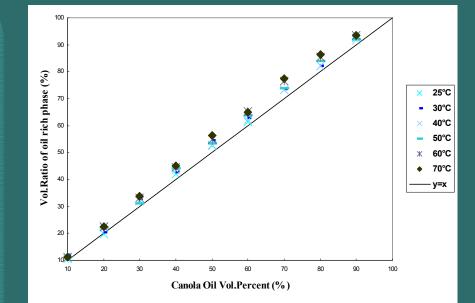
TG + alcohol — DG + alcohol — MG + alcohol — DG + FAAE MG + FAAE FAAE + Glycerin



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1. Mass transfer limited reaction rate: lipids and methanol or ethanol are immiscible under normal reaction conditions.

- Improve mixing (energy costs)
- Mutual solvent (separation costs)
- Other alcohol (cost)





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- 2. Incomplete conversion due to reversibility of transesterification reaction: leads to low quality biodiesel (presence of TG, DG, MG) and loss of reactants.
  - Force equilibrium to products by adding excess alcohol (costs)
  - Use of extreme reaction conditions (energy costs)
  - Multiple water washes (cost, generation of waste)
  - Unreactables?







"Reaction completeness is the most critical fuel quality parameter." (Van Gerpen 2005, *Fuel Proc. Tech.*, 86:1097-1107)

- 3. High cost of virgin feedstock leads to use of low cost feedstock which is high in water and/or FFA:
  - difficult downstream purification (soaps)
  - poor cold flow properties
  - 2-step acid/base or 1-step acid reaction (see Zhang et al. 2003 *Biores. Tech.* 89:1-16, 90:229-240)





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4. Traditional downstream purification such as water washing to remove excess alcohol and catalyst may generate large amounts of toxic waste water and incur high energy costs.
– Heterogeneous catalysts?



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#### 5. Most commercial processes are run in batch mode.

 Continuous reaction can lead to reduced down-time, higher throughput, more stable operation.



### **Objectives**

- Develop a continuous reaction process for the production of biodiesel and overcome the challenges due to mass transfer limitations, incomplete conversion, use of high FFA feedstock and downstream purification.
- Investigate different factors affecting the production process and the biodiesel purity.



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#### A promising solution: membrane reactor

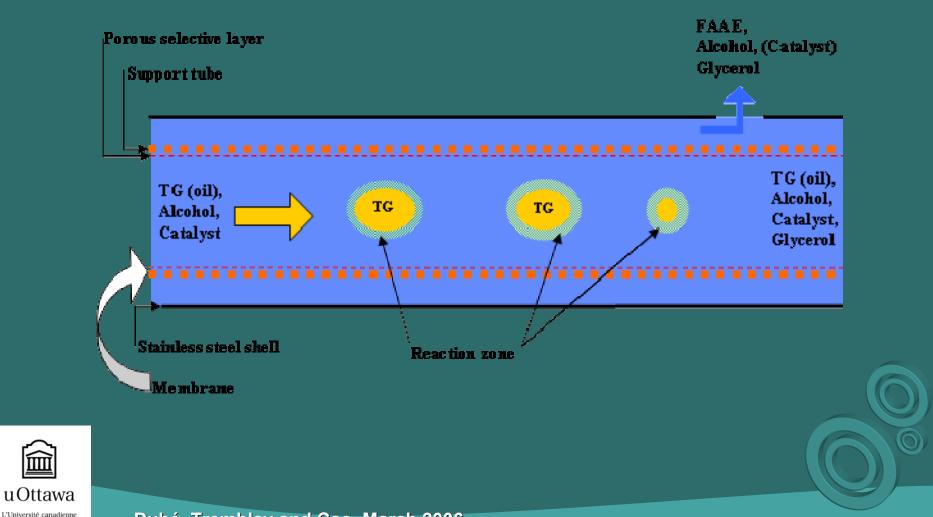
- Definition: a device for simultaneously carrying • out a reaction and membrane-based separation in the same physical enclosure.
- Theory: due to the immiscibility of lipid feedstock and alcohol, lipids form droplets which are excluded from passing through the membrane pores. The micro-porous inorganic membrane selectively permeates FAAE, alcohol and glycerol while retaining the emulsified oil droplets.



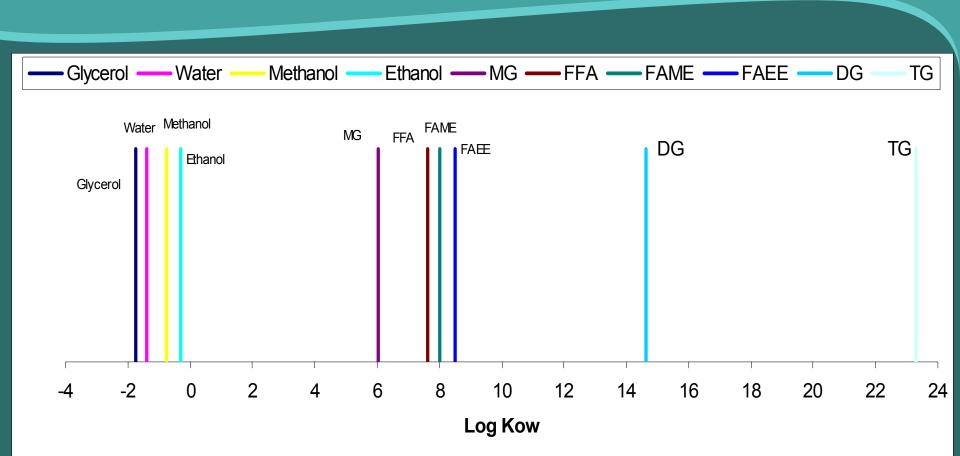
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## Principle

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## Inorganic membranes

- Inorganic to resist methanol, FAAE, catalyst
- Can be bundled to increase surface area



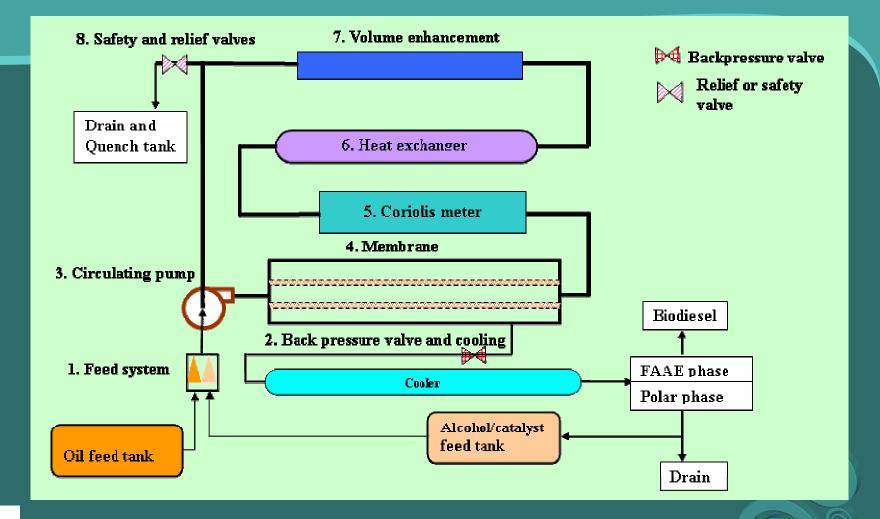






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#### **Prototype reactor**





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### Experimental

#### Reagents:

- Lipid feedstock: virgin canola oil, virgin palm oil, yellow grease, waste canola oil
- Alcohols: methanol, ethanol, alcohol blends, denatured alcohol
- Catalysts: sodium hydroxide, sulphuric acid

#### Reaction conditions:

- Alcohol:lipid molar ratios from 50:1 to ~6:1
- Temperatures from 55 to 70°C
- Catalyst concentrations from 0.5 2 wt.% (base), from 1 5 wt.% (acid)
- Circulating flow rates from 90 to 180 kg/min
- Reactor residence times from 1 to 3 h
- Characterization: HPLC, GC



### Experimental

#### **Procedure:** 0

- Alcohol/catalyst and lipid feedstock are charged to the reactor loop and heated to the reaction temperature.
- Circulation pump is used to provide a cross-flow through the membrane reactor loop.
- Semi-continuous or continuous feeding of alcohol/catalyst and/or lipid feedstock at a controlled molar ratio ensures a trans-membrane pressure causing the permeation of FAAE/alcohol/glycerol/catalyst while retaining the lipid feedstock inside the membrane reactor loop.

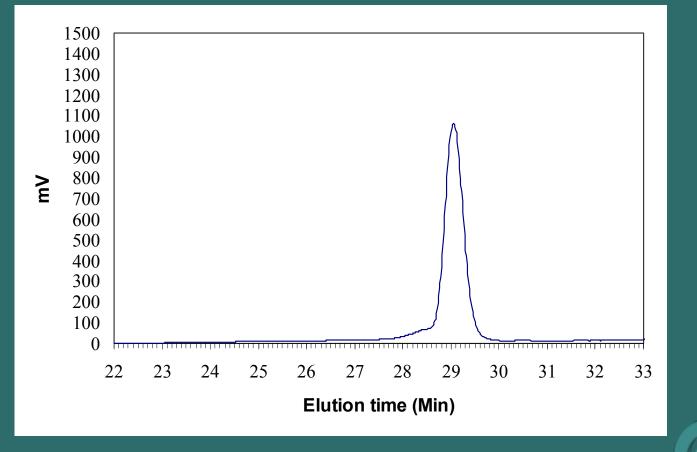


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- Due to immiscibility of lipid feedstock and alcohol, emulsion formed in circulating loop of membrane reactor.
- Formation of large lipid droplets (diameters = 20 to 1800 microns) prevents permeation of lipid through membrane pores (pore diameter <1.4 microns).</li>
- FAAE, alcohol and catalyst being miscible, pass through membrane pores. Some glycerol also passes through membrane pores.
- RESULT: NO lipids (TG, DG, MG) in the permeate stream! High conversions not required.
- Free and total glycerin contents of biodiesel easily meet international standards for purity.



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- Low quality feedstock (e.g. yellow grease, FFA content ~17wt%) reacted with base catalyst.
- Formed soaps appeared to be retained in membrane reactor resulting in straightforward purification of FAAE in the permeate.
- Membrane also retained particulate and unreactable matter, thus eliminating presence of stable emulsion phase on washing biodiesel.

Soaps



Washed permeate

Washed retentate



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- Permeate readily dephased at room temperature:
  - permitted recycling of polar phase from permeate stream
  - allowed for lowering of overall alcohol:lipid molar ratio to ~6:1.



Homogeneous permeate

Cool to ~25°C





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#### **ASTM results**

 Free glycerin: 0.002 mass% (max. = 0.020)
 Total glycerin: 0.037 mass % (max. = 0.240)

- TG < 0.001 mass%



#### Permeate ~25°C

Water wash



#### **Biodiesel**



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#### Membrane reactor advantages

- 1. No bound glycerin
- 2. Enhanced reaction rate
- 3. Easy separation of products
- 4. High FFA feedstock handling
- 5. Continuous flow
- 6. Blocks most impurities
- 7. High purity biodiesel!



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## **Concluding remarks**

- Membrane reactor for continuous production of biodiesel from various lipid and alcohol feedstock has been developed: results in high purity biodiesel which meets standards such as EN14214 and ASTM D6751.
- Ability of reactor to produce biodiesel from variety of lipid feedstock, some containing high amounts FFA, has been demonstrated.
- The membrane reactor was particularly useful in retaining TG, DG, MG, soaps and most of the glycerol. These findings indicate excellent potential for the commercial use of this reactor.



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## **Concluding remarks**

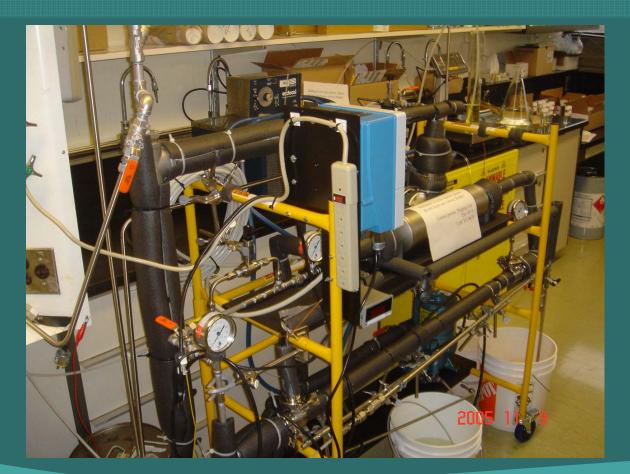
- Petrodiesel production: use of distillation to purify leaving behind tars
- Conventional biodiesel production: unreactables remain (distillation is not cost effective)
- Membrane reactor offers a radical change in quality: unreactables removed



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### Acknowledgements

#### Natural Science and Engineering Research Council (NSERC) Canada.





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