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Ultra-High Temperature Ceramics: Materials For Extreme Environmental Applications II

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## The Next Steps for Ultra-High Temperature Ceramics

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## **The Next Steps for UHTC Materials**

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Ultra-High Temperature Ceramics for Extreme Environment Applications II

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Background Why UHTCs? Current research & development focus

Systems Needs/Designers Phobias – are UHTCs getting a bad reputation? How to integrate into structure? Catastrophic Failure Long-term applications - oxidation

What needs to be done? New Processing Technologies Testing in Relevant Environments



## Why UHTCs?



What does a UHTC need to do:

- Carry engineering load at RT  $\sqrt{}$
- Carry load at high use temperature
- Respond to thermally generated stresses (coatings)
- Survive thermochemical environment  $\sqrt{}$

High Melting Temperature is a major criteria, but not the only one Melting temperature of oxide phases formed Potential eutectic formation

Thermal Stress –  $\mathbf{R}' = \sigma \mathbf{k}/(\alpha \mathbf{E})$ Increasing strength helps, but only to certain extent

Applications are not just function of temperature

- Materials needs for long flight time reusable vehicles are different that those for expendable weapons systems



## **Random Thoughts**



In many aerospace systems, designers will add weight and complexity by using metals and active fuel cooling to avoid using advanced ceramics and composites.

- Direct function of the conservatism engrained in industry and the system integrator being contracted to build a system and not demonstrate a materials technology

- Unfamiliarity with designing with brittle materials - safety factor.

- Advantages of weight savings and uncooled temperature capability not high enough to overcome risk aversion

Using monolithic ceramics and CMC requires a different design approach, not just dropping in a ceramic part to replace a metallic component

- Rocket nozzle examples – learning how to use a brittle material in a highly transient heating environment

Need for subscale materials/component testing in realistic environments is imperative

Onus is on US to develop materials that will be used by designers

## High Temp Materials Selection Thermochemistry is Driver



- Metallurgical Thermo tools employed

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- Lines divide stability regions of metal vs condensed metal oxide

- Each element shown is best representative of Group (eg., Ta-Nb-V)

- Each element is also representative of compounds (HfC, HfN, HfB<sub>2</sub>, HfO<sub>2</sub>, etc.)

- So what ???? How is this useful ????



## Current research vs future directions



- Increase strength SPS (engineers don't like lack of toughness)
- Increase thermal conductivity (HfB<sub>2</sub> is already very high)
- Decrease modulus (graphite second phases, but eutectic issue)
- Decrease thermal expansion (intrinsic property)
- Design around thermal stresses rocket nozzles (ONR & AFRL)
- Understand and improve oxidation behavior why SiC?
- Develop UHTC-matrix composites hot pressing, SPS, HIP, all produce bulk monolithic materials.
  - Modulus/CTE mismatch w/ C fibers a problem
  - Current polymeric precursors expensive and air sensitive
  - Organnometallics
  - Melt infiltration refractory alloys reactive with fibers
  - Densification/conversion of matrix
  - Alternative Processing routes (CVC, in-situ reinforcement, cermets)



## **Hypersonics**



## High Operational Velocities ⇒ High Temps & Heating Rates









### Pressure-Time Traces of Graphite & HfB<sub>2</sub> Throats Showing Non-eroding Behavior of Ceramic Nozzle

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### **Thermal Stress-Based Failure Mode:**

**Thinner Liners - Lower Thermally Generated Stresses** 



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NAVSEA

**Backface Temperature** Small Nozzle Example  $T_{o} = 4400^{\circ}F$ 500  $P_{o} = 1500 \text{ psi}$  $\phi = 0.198$  inch 400 \_400°F Rise Time = 0.15 sec Time = 5.0 sec300 200 0.025 HfB<sub>2</sub> 0.050 HfB<sub>2</sub> 0.250 HfB<sub>2</sub> 100 0.500 Foam 0.500 Foam 0.500 Foam 0 **Carbon Foam Insulator Stress** HfB<sub>2</sub> Liner Stress 4 125 0.250 HfB, Compressive Stress (ksi) 100 0.500 Foam 3 Tensile Stress (ksi) T=300°F 75 0.050 HfB, RT Allowable - 2 ksi 2 0.500 Foam T=1300°F 50 0.050 HfB<sub>2</sub> 0.250 HfB<sub>2</sub> RT Allowable - 44 ksi 0.025 HfB<sub>2</sub> 0.500 Foam 0.500 Foam 0.500 Foam T=3940°F T=3920°F T=3140°F 0.025 HfB<sub>2</sub> 25 0.500 Foam T=2270°F 0 C-foam provides insulation and allows expansion of liner





Similar Axial Stress Fields Have Been Observed In Previous Analysis Efforts Maybe the models are too conservative in predicting thermal stress failures?

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E



## Understanding of Oxidation at NAVSEA **Relevant Temperatures**



### (1627C not high enough!)

Ref: Evaluation of Ultra High Temperature Ceramics for Propulsion Application S.R. Levine, et al., NASA Glenn Res. Ctr, ACerS PacRim Conf., Seattle, WA, 01 Nov 2002





# Borides are not the only UHTC Materials



Boria has high Pv, SiC active oxidation, and HfO₂ does not sinter below 1800 °C, so oxidation behavior **improves** with temperature

### Furnace Testing of HfC – 1500°C

Pesting Issues at Low T – Sinterability of HfO<sub>2</sub>









## Arc-jet tests help understanding thermochemical environment







HfN<sub>.95</sub>

But sample geometry (radius of curvature), enthalpy, catalycity, and pressure all affect performance

HfN<sub>.75</sub>







### SHARP-B2 UHTC flight test [Sannes, 9 and Johnson, 10]

Very progressive & expensive

 Failure attributed to UHTC processing

### Slide from D. Glass, NASA-Langley



Pre-test

Post-test





- ZrB<sub>2</sub>-SiC nose tip during arc-jet test [Marino, 11].
  - Failure attributed to titanium screw [12].

Second assembly withstood two test at 3 MW/m<sup>2</sup> for 108 sec. Non-critical damage observed at base of UHTC tip. [Scatteia, 12]

Affecting future of Italian UHTC work?

Even in constrained 1-D heating environment for oxidation testing, <u>low thermal shock resistance</u> and <u>low fracture toughness</u> of <u>monolithic ceramics</u> can result in catastrophic failure



## Improve oxidation behavior



- Eliminate presence of Si detrimental above 1700 ℃
- Test in relevant environments
- Dopants to control oxygen diffusion through oxide scale
- Decrease porosity of scale
  - Substoichiometry & Doping Wuchina & Opeka
  - Liquid phase sintering Fahrenholtz & Hilmas

HfCxOy Interphase





Current NSWCCD Research Oxidation and Mechanical Properties in the Hf-Ta-N System



*Hf, Ta:* <u>Linear</u> Oxidation Kinetics at *High Temperature (HT)* 

- *Hf-20Ta:* <u>*Parabolic*</u> *Kinetics to 2000℃ with strongly adherent scale (Marnoch, 1965)*
- HfC-25TaC: <u>Linear</u> Kinetics at HT (Holcomb, 1988)
- *HfB*<sub>2</sub>-20TaB<sub>2</sub>: <u>Linear</u> Kinetics at HT (Talmy, unpublished)
- (Hf,Ta)N<sub>x</sub> Ceramics: Not Investigated To-Date; Initial Navy Expts Conducted with HfTa alloys





## UHTC-matrix composite – ca 1997 and Hf-HfN-HfB<sub>2</sub> cermet

Hf acetate + C, CVI, C/C late stage PIP or gas infiltration, nano HfC





X-51 Inlet leading edge was follow-on to work initiated in late 1990s through SBIR



BEI Hf-17N-30B

### Chemical Vapor Composites CVD with Particulate or fiber reinforcement





### **Chemical Vapor Composites**

Powders and fibers are mixed with reactant vapors and deposited on a reusable mold, producing a composite in the exact shape of the component



### LOW COST

- · Single step
- · High deposition rate
- · Net shape

### HIGH QUALITY

- · Strong tough
- Lightweight
- · High temperature

### VERSATILE

- · Metal composites
- · Ceramic composites
- Transition joints

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## The Intersection of Basic Science and Engineering



Ab-initio calculations for discovery of new materials and correlation of chemical & physical models with materials by design approach for improving properties

High-fidelity analysis to understand multi-scale effects on macroscopic properties

Chemical Vapor Infiltration Organnometallic Precursors Mean Free Path Modulus Mismatch – fiber breakage

**PIP Processing** 

Preceramic Polymer Development Particulate Additions – filtration Increase Yield & Reduce Cycles

Other Techniques Vapor Phase Conversion UHTC fibers

## New Materials for Hypersonic Vehicles Include Eutectics

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F F I C E Hard-Hard and Hard-Soft Ceramic Eutectics? A B

## New Materials for Propulsion Systems Include Eutectics



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at 2000°C

B+8

Microstructure and Properties of Ceramic-Metal Eutectics?

Eutectic Temps & Intersolubilities Vary Significantly: HfC-W: 2890 ℃,

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HfN-W: 2800 ℃ (P<sub>N2</sub>-Dependent), HfB<sub>2</sub>-W: 2280 ℃

HfC-Ta: 2550 ℃ (large intersolubilities) HfN-Ta: 2800 ℃ (some intersolubility) HfB<sub>2</sub>-Ta: Reaction to (TaHf)B

Composition Control and Properties of "Metal" Phase a Major Issue for Cermet Eutectics











## **Notional Eutectic Production** via Additive Manufacturing



**Pre-Alloyed** 

## From CAD to part using laser or e-beam(?) metal deposition: Metals were low-hanging fruit – Refractory Eutectic Ceramics and/or Coatings



are a bit tougher





## Conclusions



UHTCs are necessary for future hypersonic flight/propulsion systems due to higher use temperatures

Oxidation must be understood and controlled

Monolithic Ceramics will not likely be used in flight hardware

Flaw Sensitivity (Attachment issues) Thermal Shock failure

Current work - joining and reinforced UHTCs

Do good research, but think like an engineer, not a scientist!