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Ion Transport across Grain Boundaries in Fast Lithium Ion Conducting Glass Ceramics

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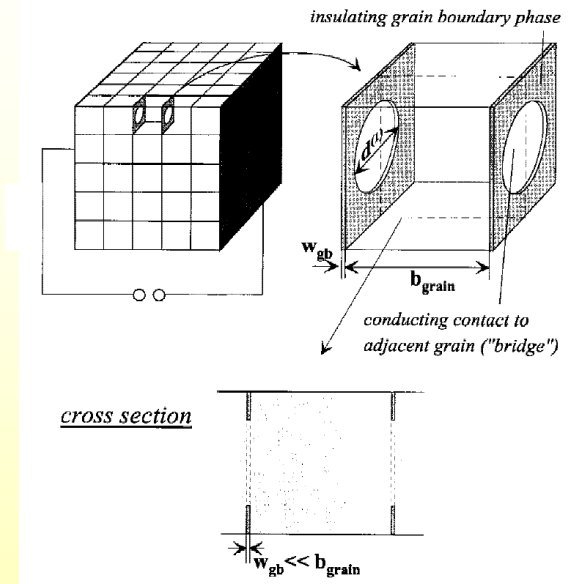
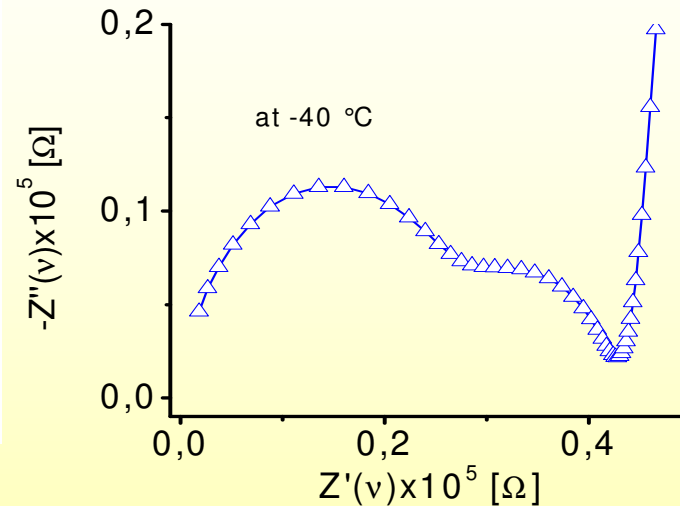
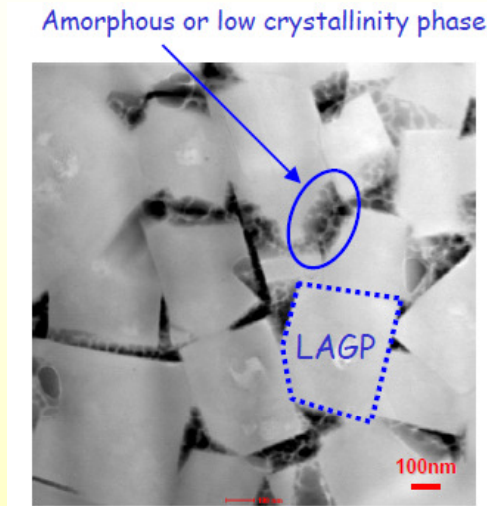
Bernhard Roling, Katharina I. Gries, Michael Gellert, Kerstin Volz, Fabio Rosciano, and Chihiro Yada, "Ion Transport across Grain Boundaries in Fast Lithium Ion Conducting Glass Ceramics" in "Functional Glasses: Properties And Applications for Energy and Information", H. Jain, Lehigh Univ.; C. Pantano, The Pennsylvania State Univ.; S. Ito, Tokyo Institute of Technology; K. Bange, Schott Glass (ret.); D. Morse, Corning Eds, ECI Symposium Series, (2013). http://dc.engconfintl.org/functional_glasses/7

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Ion Transport across Grain Boundaries in Fast Lithium Ion Conducting Glass Ceramics



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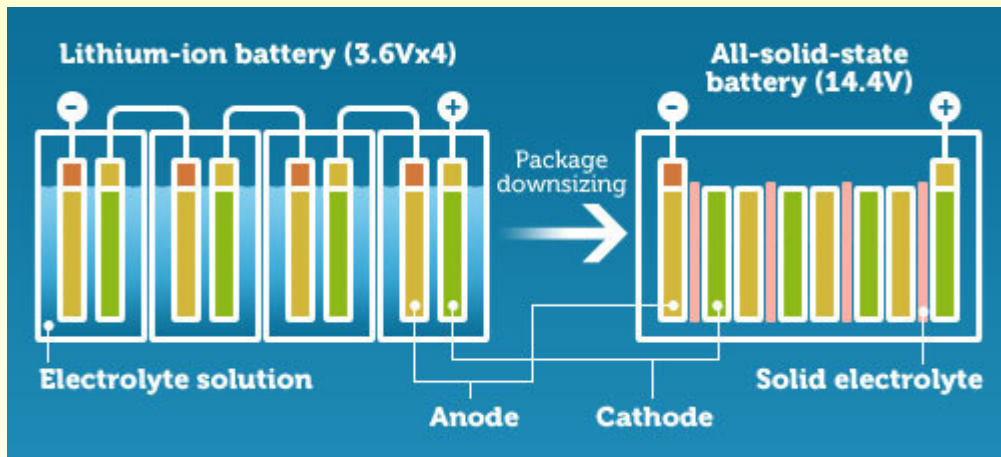
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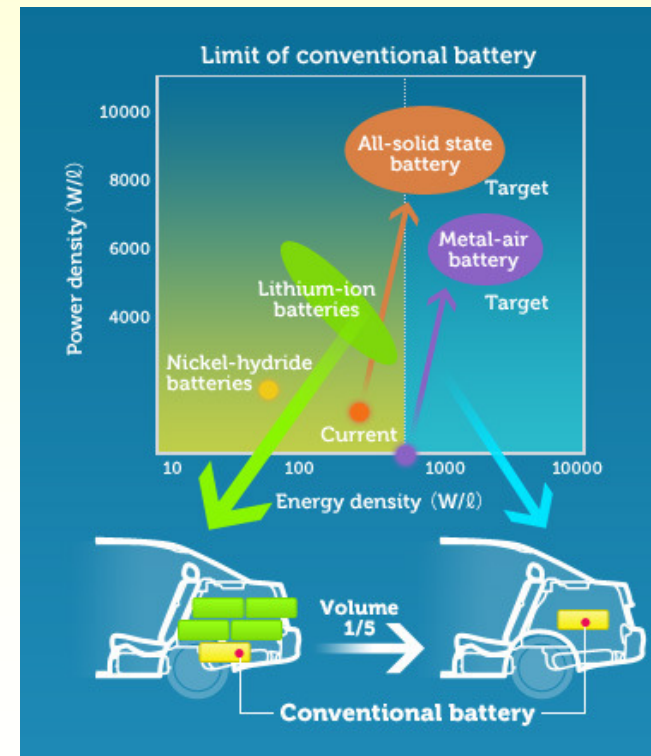
Solid-State Batteries



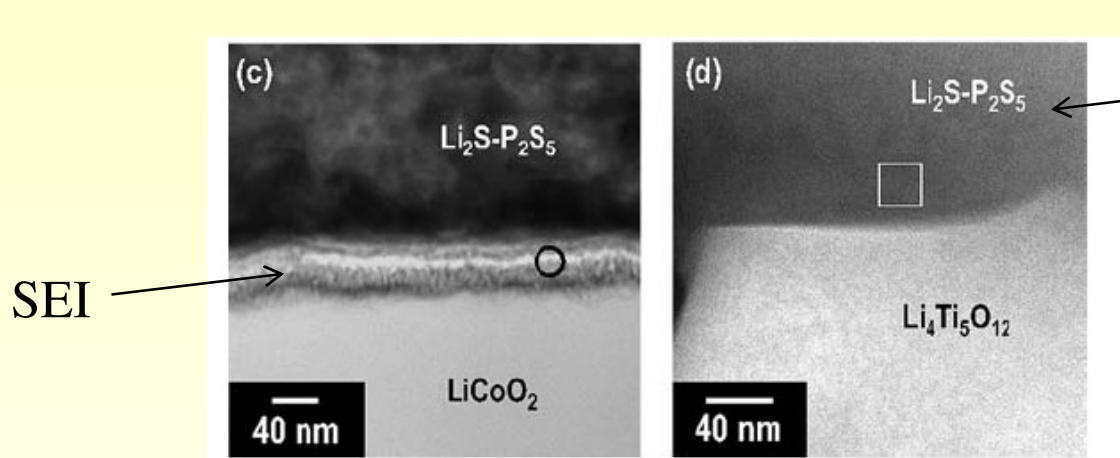
No separation of individual cells

→ More compact packaging

http://www.toyota-global.com/innovation/environmental_technology/next_generation_secondary_batteries.html



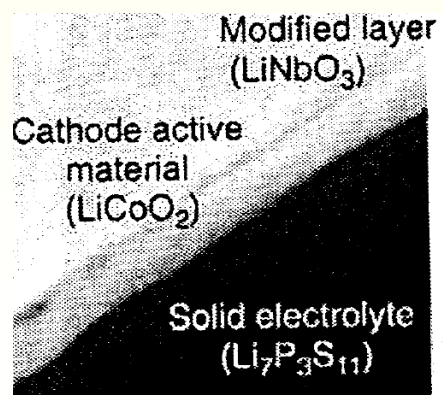
Solid-State Batteries



Room-temperature ionic conductivity
 $\sigma \approx 10^{-3} \text{ S/cm}$

Tatsumisago and coworkers,
 J. Mater. Chem. 21 (2011) 118.

Fig. 7 HAADF-STEM images of the interface between the LiCoO_2 or $\text{Li}_4\text{Ti}_5\text{O}_{12}$ active material and the solid electrolyte in the composite electrodes pressed at room temperature ((a) and (b)), and pressed at 210 °C ((c) and (d)).



Coating of cathode with protective layer,
 e.g. LiNbO_3 (thickness in the range of 10 nm)

Solid-State Li-S-Batteries with $\text{Li}_2\text{S-P}_2\text{S}_5$ Glass as Electrolyte

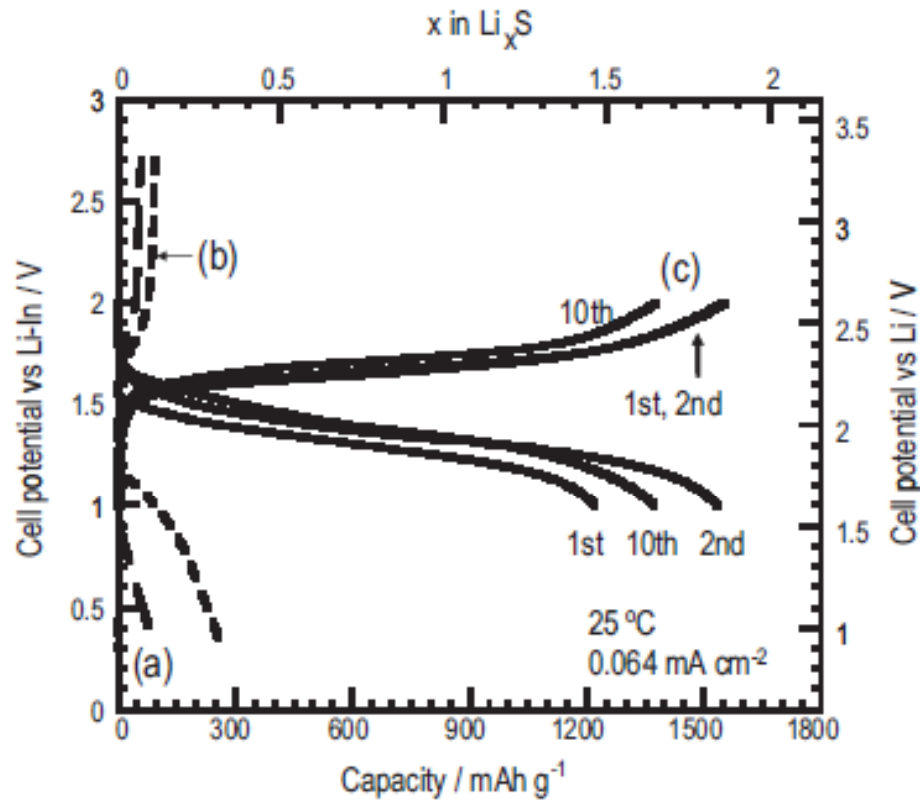
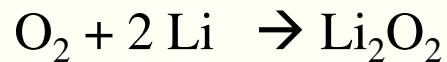
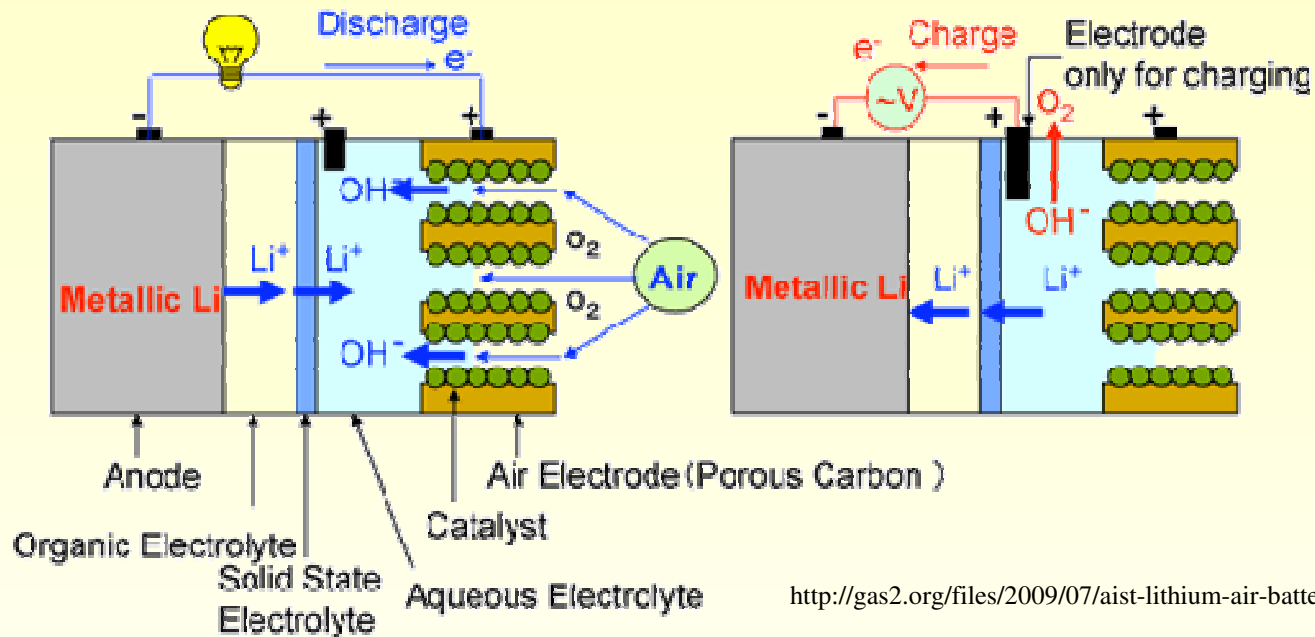


Fig. 3. Charge-discharge curves of all-solid-state cells of Li-In/80 Li_2S -20 P_2S_5 glass-ceramic/S using (a) (S+AB+SE), (b) (S-AB+SE), and (c) S-AB-SE electrodes as the working electrode.

Tatsumisago and coworkers,
Electrochim. Acta 56 (2011) 6055.

At low charge/discharge rates, the battery capacity is close to the theoretical capacity.

Lithium-Air Batteries



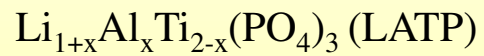
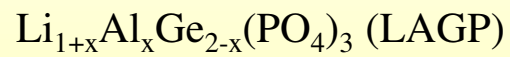
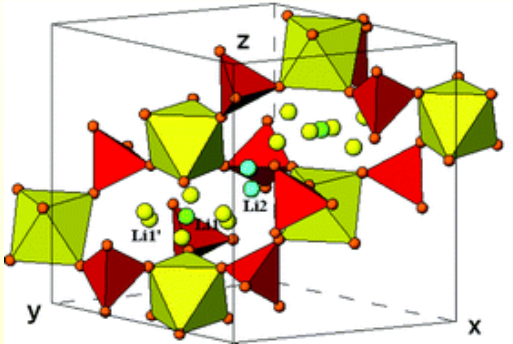
$$\text{EMF} \approx 3.1 \text{ V}$$

$$\text{theoretical energy density} \approx 3,6 \frac{\text{kWh}}{\text{kg Li}_2\text{O}_2}$$

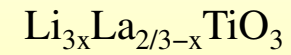
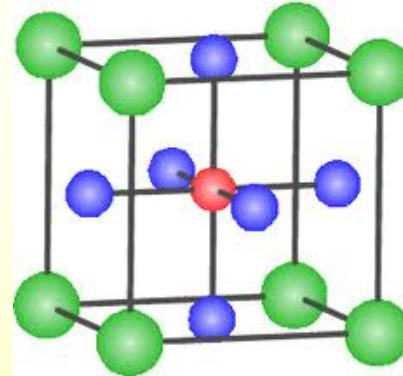
(comparable to mechanical energy from
1 kg gasoline)

Crystalline Fast Lithium-Ion Conductors

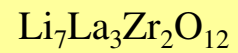
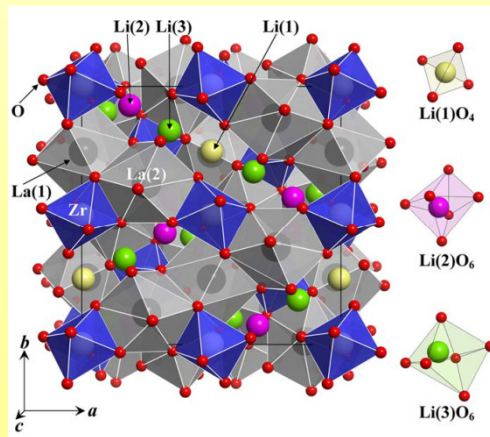
NASICON



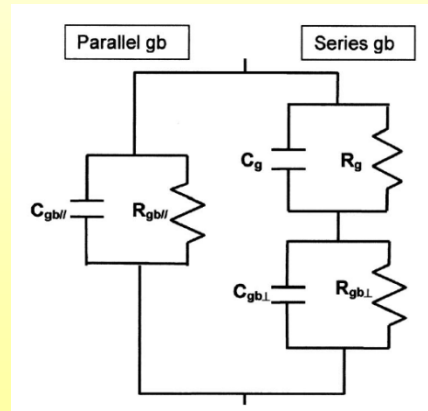
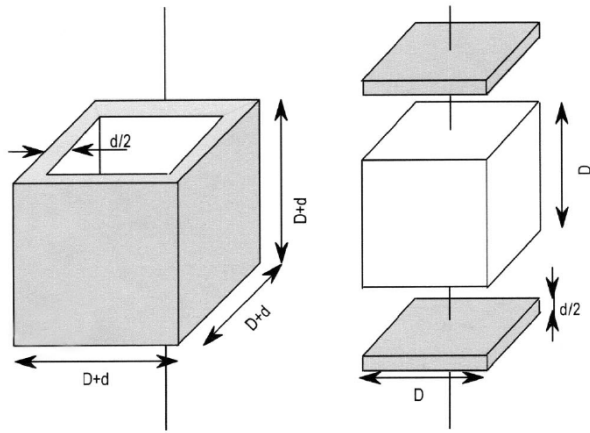
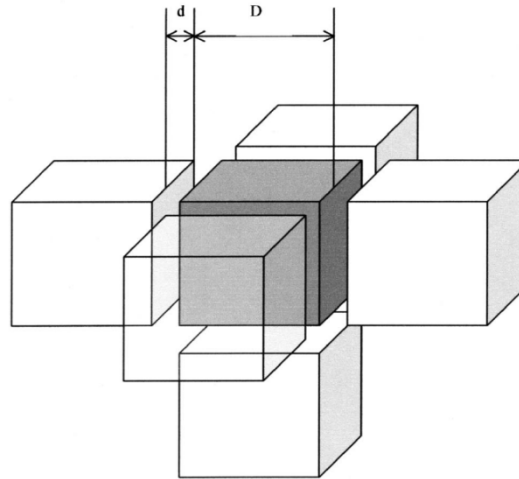
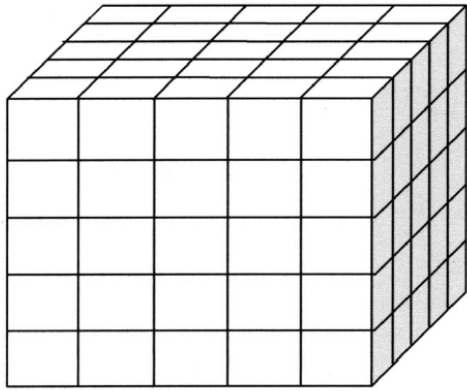
Perovskite



Garnet



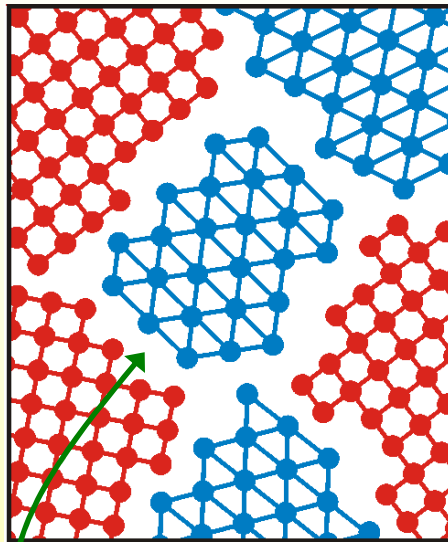
Brick Layer Model for Grain Boundary Ion Transport



Parallel gb conduction is only then relevant, if

- $\sigma_{gb} \gg \sigma_g$ or
- D is comparable to d

Parallel Grain Boundary Conduction in Lithium Ion Conductors

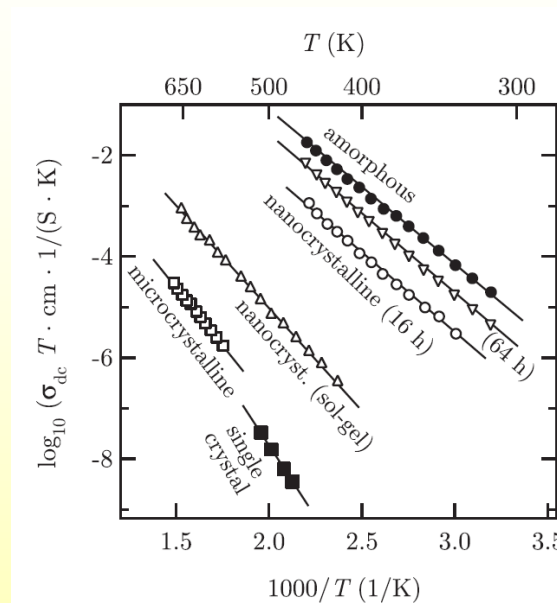


**Li₂O - B₂O₃
nanocomposite**

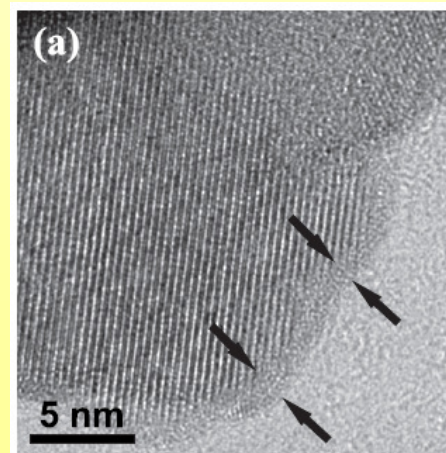
Fast Li⁺ ion conduction
at interfaces

P. Heitjans,
S. Indris,
Phys. Cond. Mat.
15 (2003) R1257.

LiNbO₃

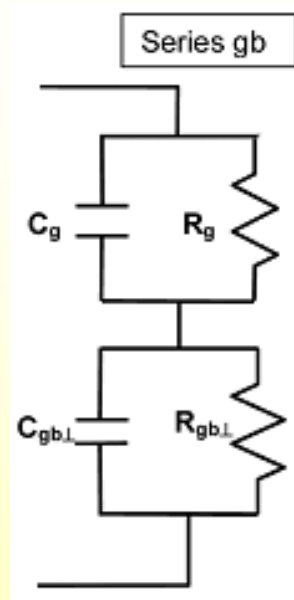
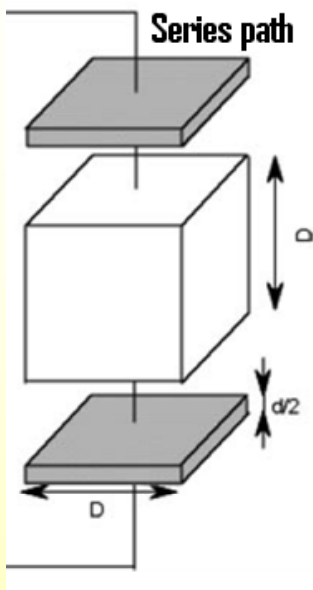


P. Heitjans,
M. Masoud,
A. Feldhoff,
M. Wilkening,
Faraday Discuss.
134 (2007) 67.



Large amount of
amorphous LiNbO₃
at the boundaries
of the nanograins

If parallel grain boundary conduction is negligible:

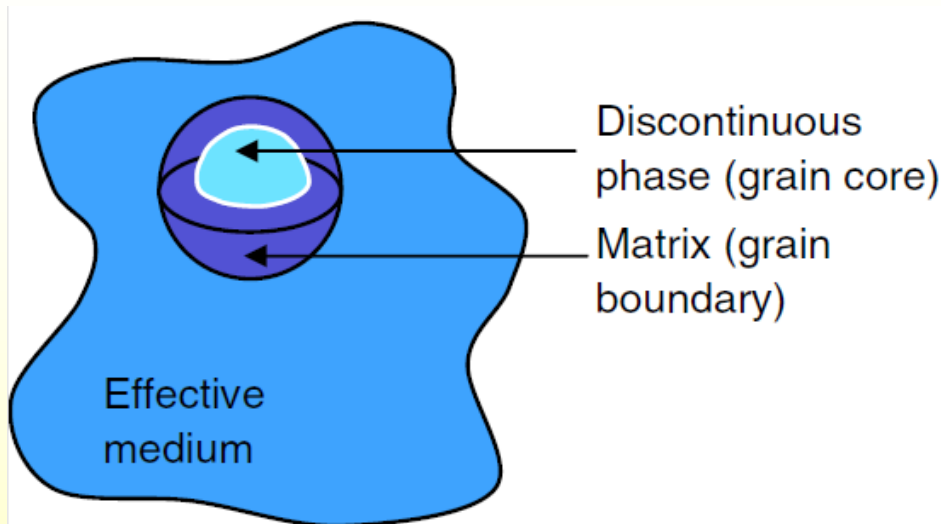


Simple results for ratios of capacitances and resistances:

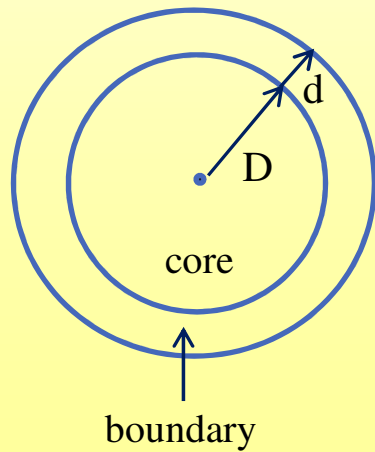
$$\frac{R_g}{R_{gb\perp}} = \frac{\sigma_{gb}}{\sigma_g} \cdot \frac{D}{d}$$

$$\frac{C_g}{C_{gb\perp}} = \frac{\epsilon_g}{\epsilon_{gb}} \cdot \frac{d}{D}$$

Nano-Grain Composite Model



Kidner et al.,
 J. Am. Ceram. Soc.
 91 (2008) 1733.



if $D \gg d$

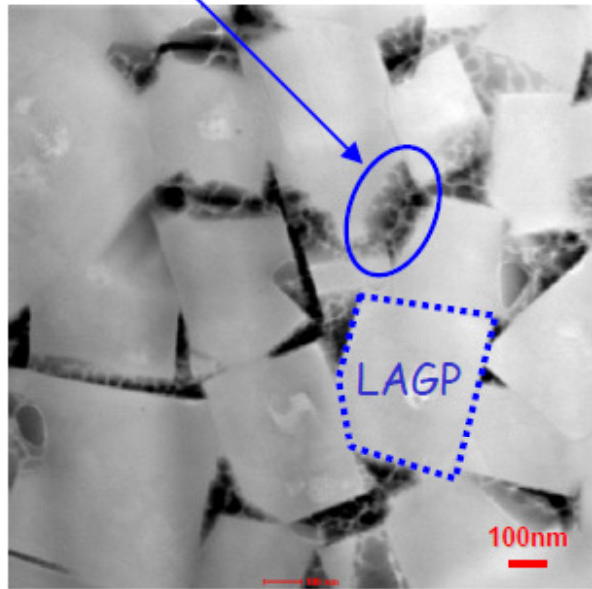
$$\frac{C_{gb}}{C_{gc}} = \frac{\epsilon_{gb}}{\epsilon_{gc}} \cdot \frac{D}{d}$$

$$\frac{R_{gb}}{R_{gc}} = \frac{\sigma_{gc}}{\sigma_{gb}} \cdot \frac{d}{D}$$

in agreement with
 brick layer model

TEM Images of $\text{Li}_{1.5}\text{Al}_{0.5}\text{Ge}_{1.5}(\text{PO}_4)_3$ (LAGP)

Amorphous or low crystallinity phase

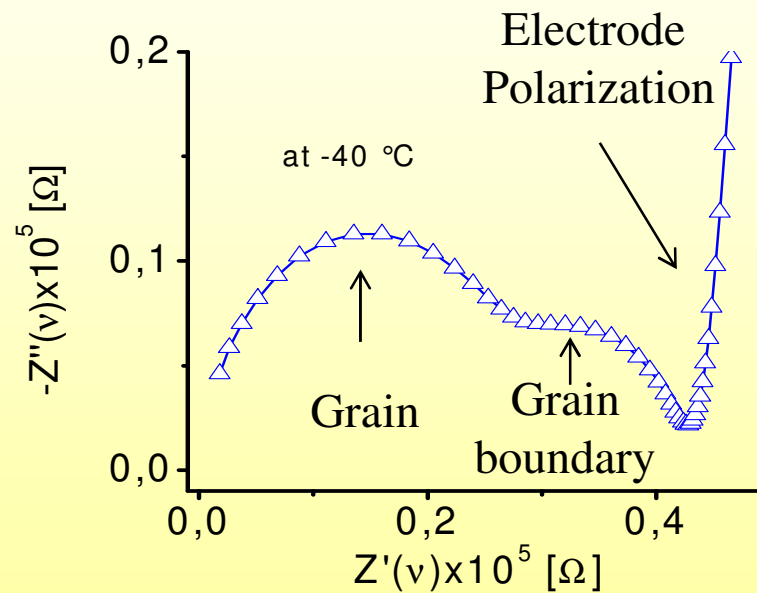
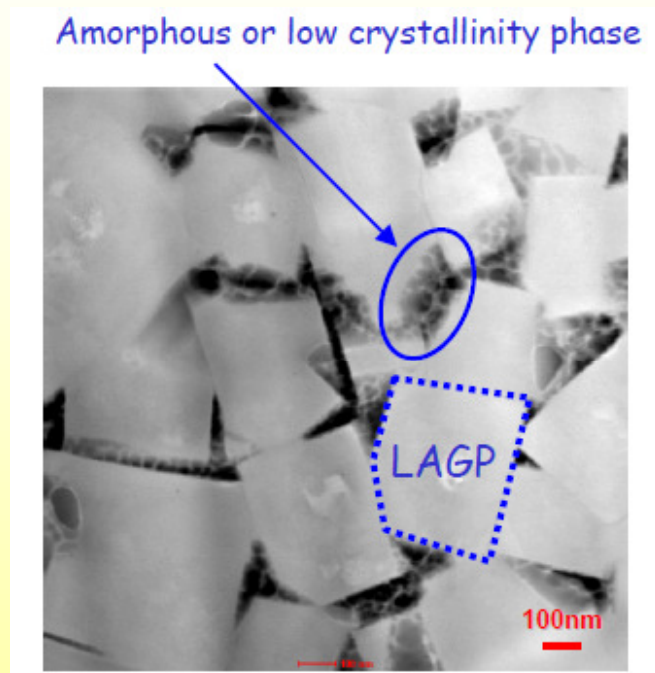


C. R. Mariappan, C. Yada,
F. Rosciano, B. Roling,
J. Power Sources 196 (2011) 6455.

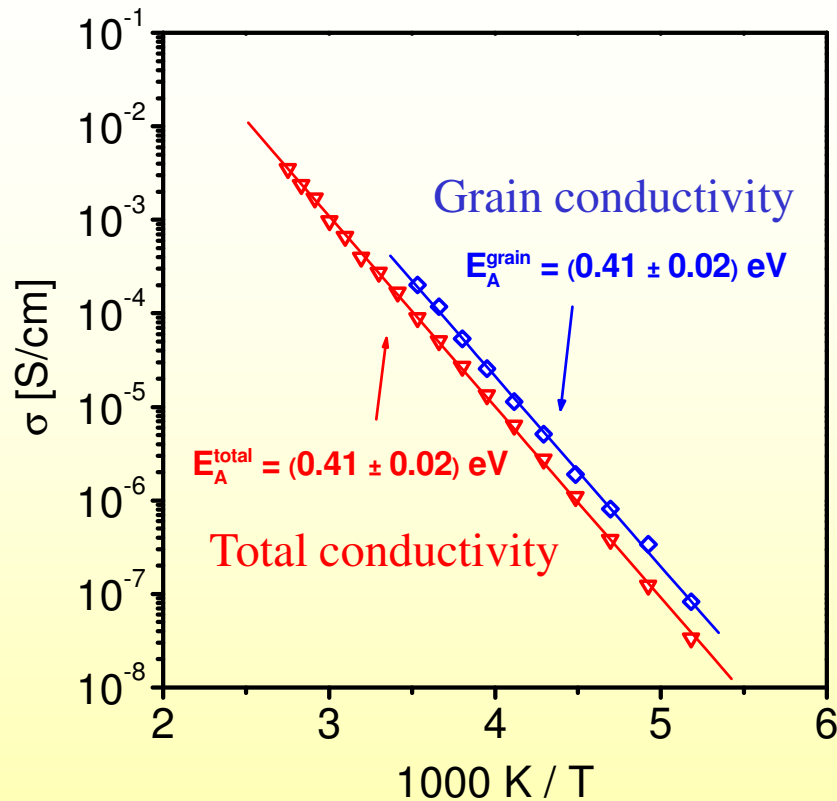
(i) Existence of impurity phases (e.g. AlPO_4) and amorphous phases with low ionic conductivity

(ii) Contact area between grains is lower than assumed in the BLM

Impedance Spectrum of $\text{Li}_{1.5}\text{Al}_{0.5}\text{Ge}_{1.5}(\text{PO}_4)_3$ (LAGP)



Grain Conductivity and Total Conductivity of LAGP



Grain conductivity

$$\sigma_g = \frac{1}{R_g} \left(\frac{d}{A} \right)$$

Total conductivity

$$\sigma_t = \frac{1}{(R_g + R_{gb})} \left(\frac{d}{A} \right)$$

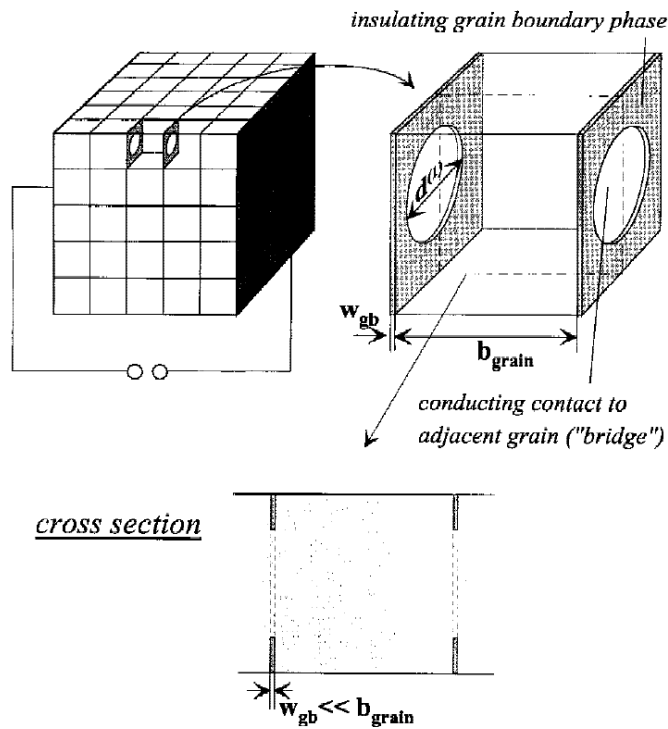
Grain and grain boundary resistance
are almost identical.



Purely geometrical current
constriction due to limited
grain contact area

Activation energies are identical.

Finite-Element Calculations by Fleig and Maier



J. Fleig, J. Maier, J. Am. Ceram. Soc
82 (1999) 3485.

$$\frac{R_{gb}}{R_g} \approx \frac{1}{\sqrt{4\alpha}}$$

Fraction of contacted area

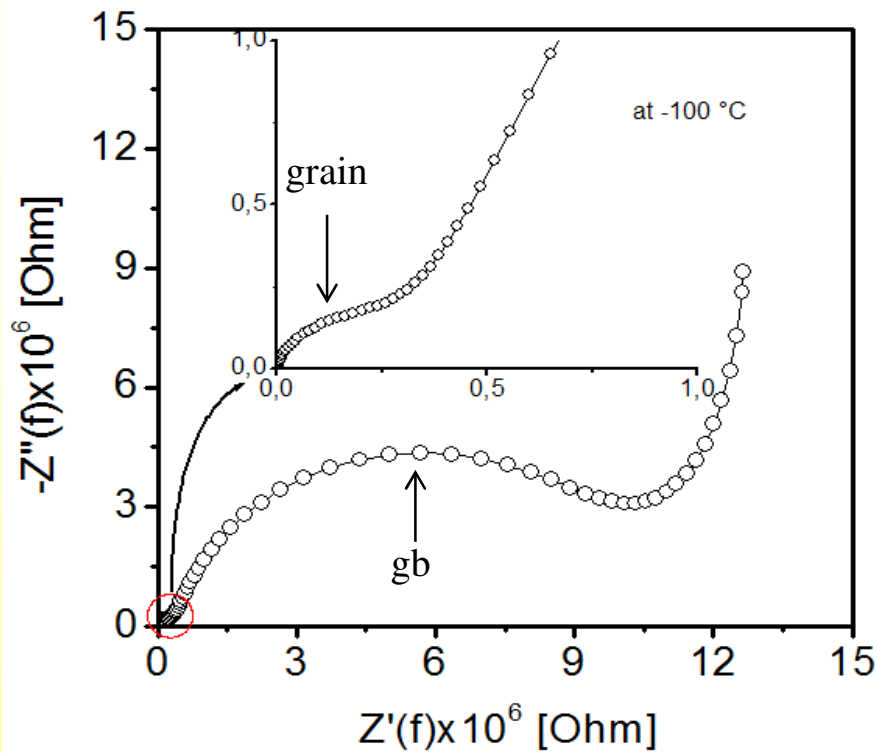
$$\alpha = A_{\text{contact}} / A_{\text{grain}}$$

For LAGP:

$$R_{gb} \approx R_g \quad \longrightarrow \quad \alpha \approx 0.25$$

in reasonable agreement
with TEM results

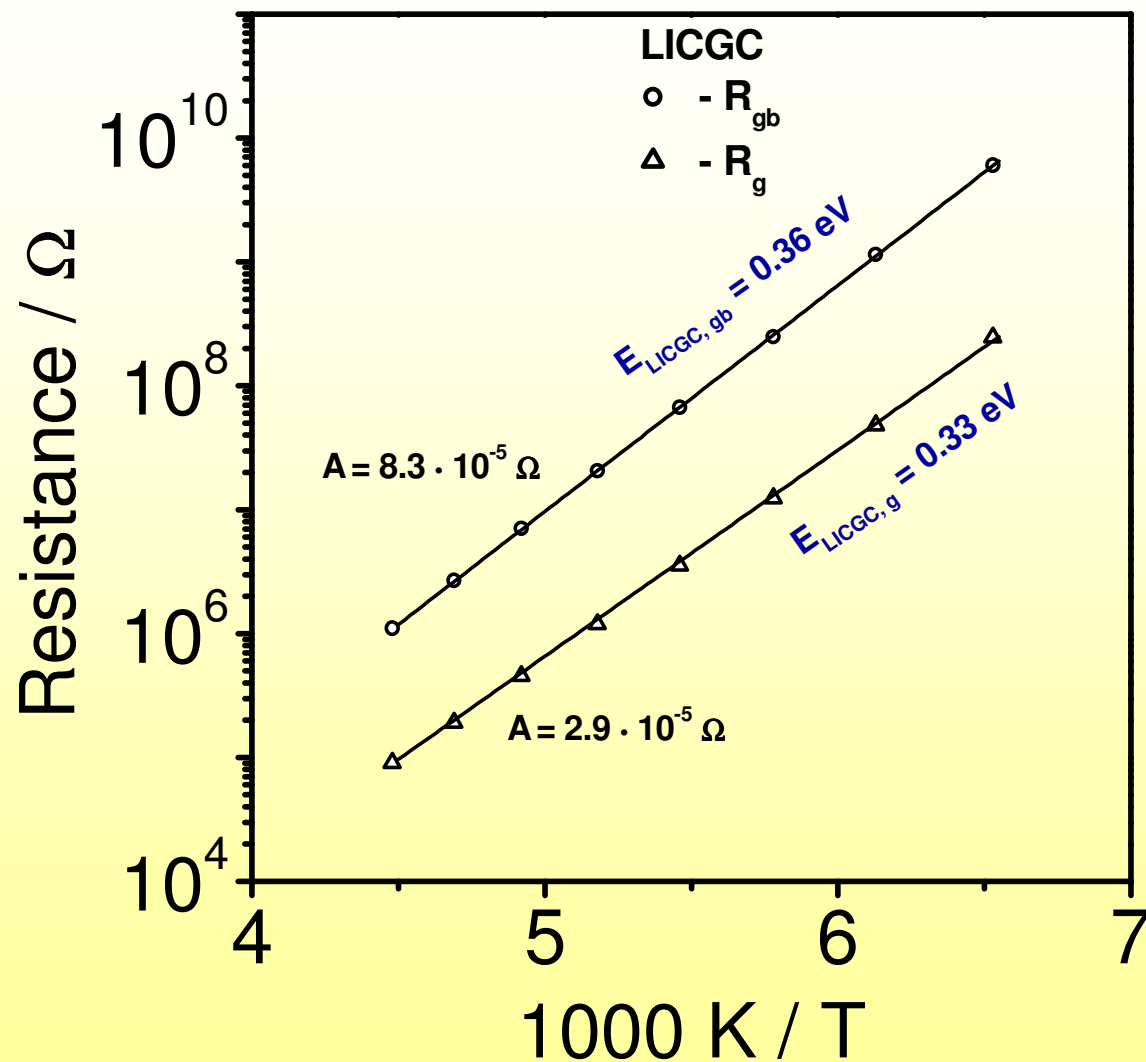
Impedance Spectroscopy on Ohara Glass Ceramic (commercial)



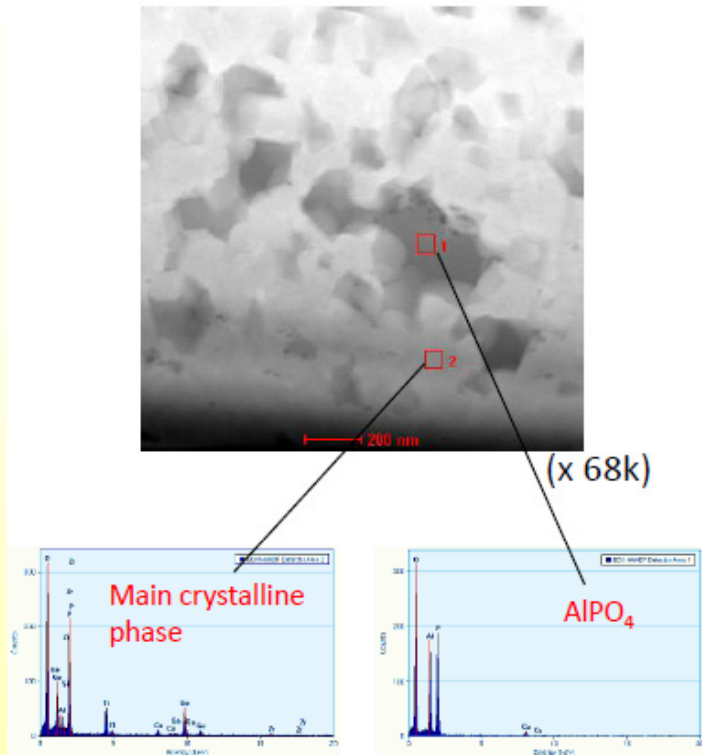
$\text{Li}_{1+x}\text{Al}_x\text{Ti}_{2-x}(\text{PO}_4)_3$
doped with various
other oxides

Grain boundary resistance
is more than one order of magnitude
higher than grain resistance

Arrhenius Plot of Grain and Grain Boundary Resistance



Ohara Glass Ceramic



Grain conductivity at room temperature:

$$\sigma_{\text{grain}} \approx 10^{-3} \text{ S/cm}$$

$$E_{\text{A}}^{\text{grain}} = 0.33 \text{ eV}$$

Grain boundary conductivity at room temperature:

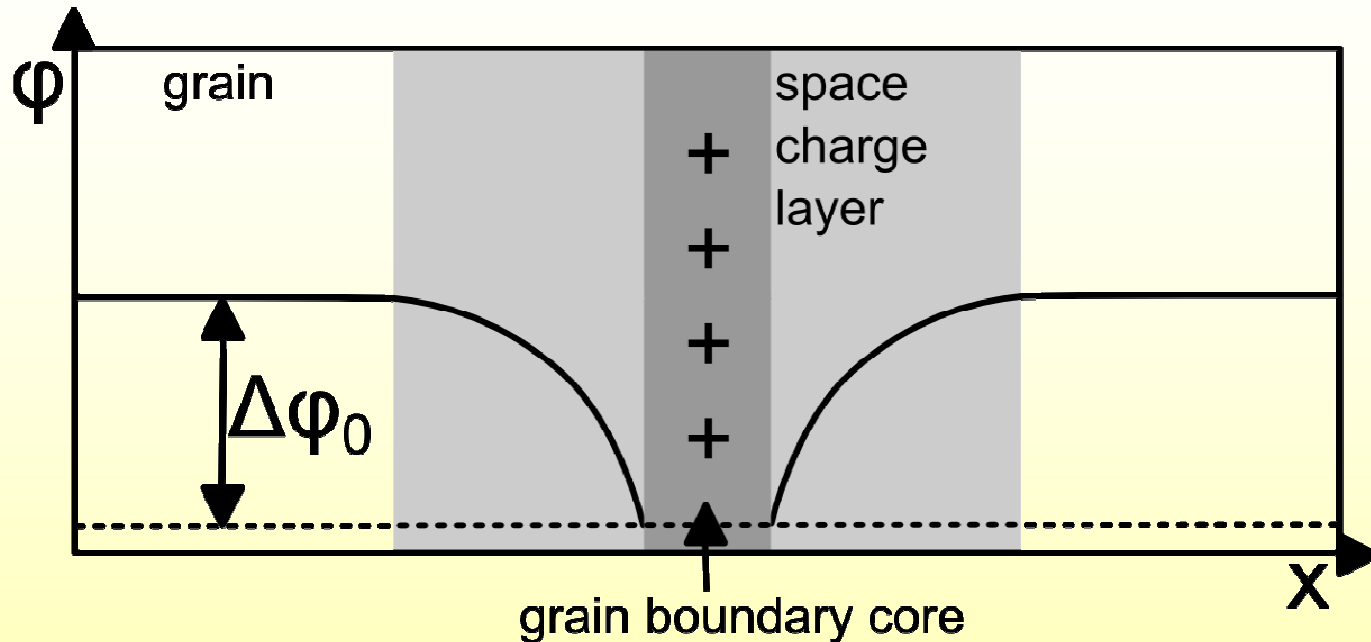
$$\sigma_{\text{grain boundary}} \approx 10^{-4} \text{ S/cm}$$

$$E_{\text{A}}^{\text{grain boundary}} = 0.36 \text{ eV}$$

Origin of higher activation energy of grain boundary conductivity?

Space charge layers? Mechanical stresses?

Space Charge Model



- Charged gb core with oppositely charged space charge layer
- Space charge layer results in electrostatic barrier for ion transport;

Important: Single barrier

Well established in the field of oxide ion conductors

Measurements with DC Bias on CeO₂ doped with 1% Y₂O₃



Figure 1. HRTEM of grain boundary in 1.0 mol % Y₂O₃-doped CeO₂. The moiré rings are also visible.

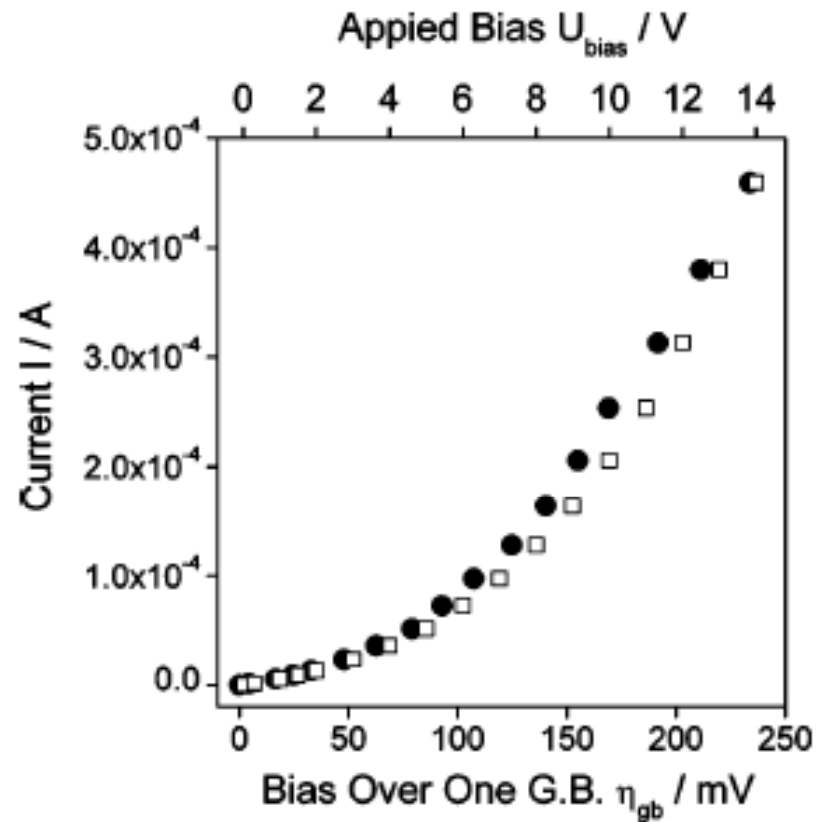


Figure 3. Current I as a function of bias over one grain boundary η_{gb} (filled circles). For comparison the current I -applied bias U_{bias} relation is also plotted (open squares).

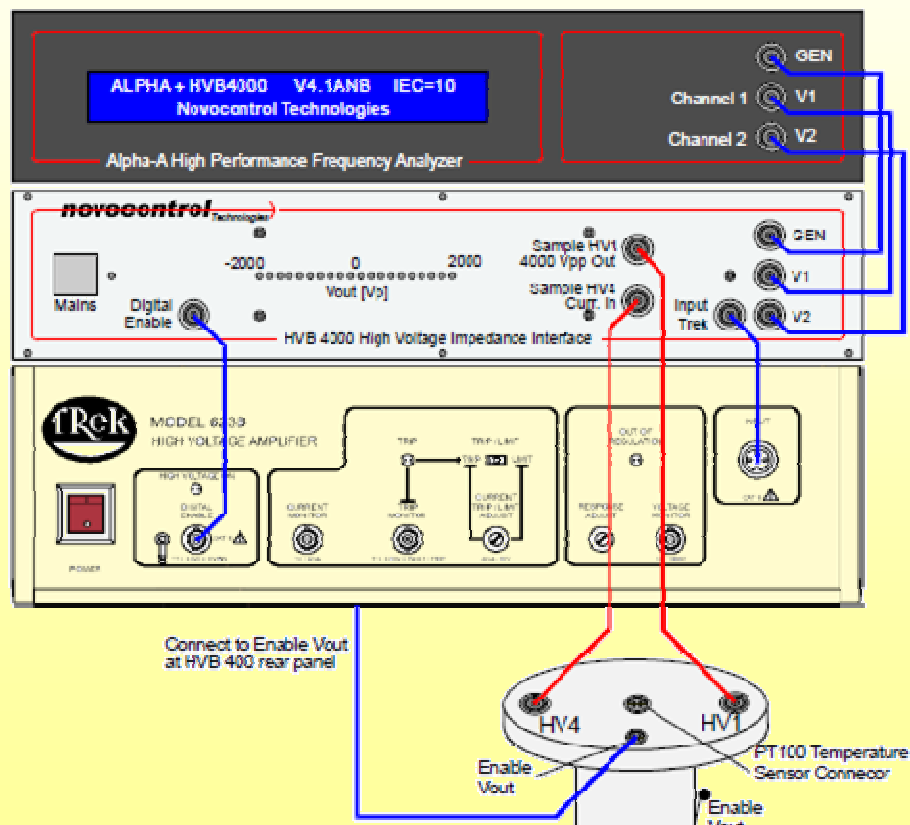
Guo, Waser et al.,
Electrochem. Solid State Lett.
8 (2005) J1 and 8 (2005) E67.

**Height of
space charge barrier:
about 0.4 eV**

High-Voltage Measurement System

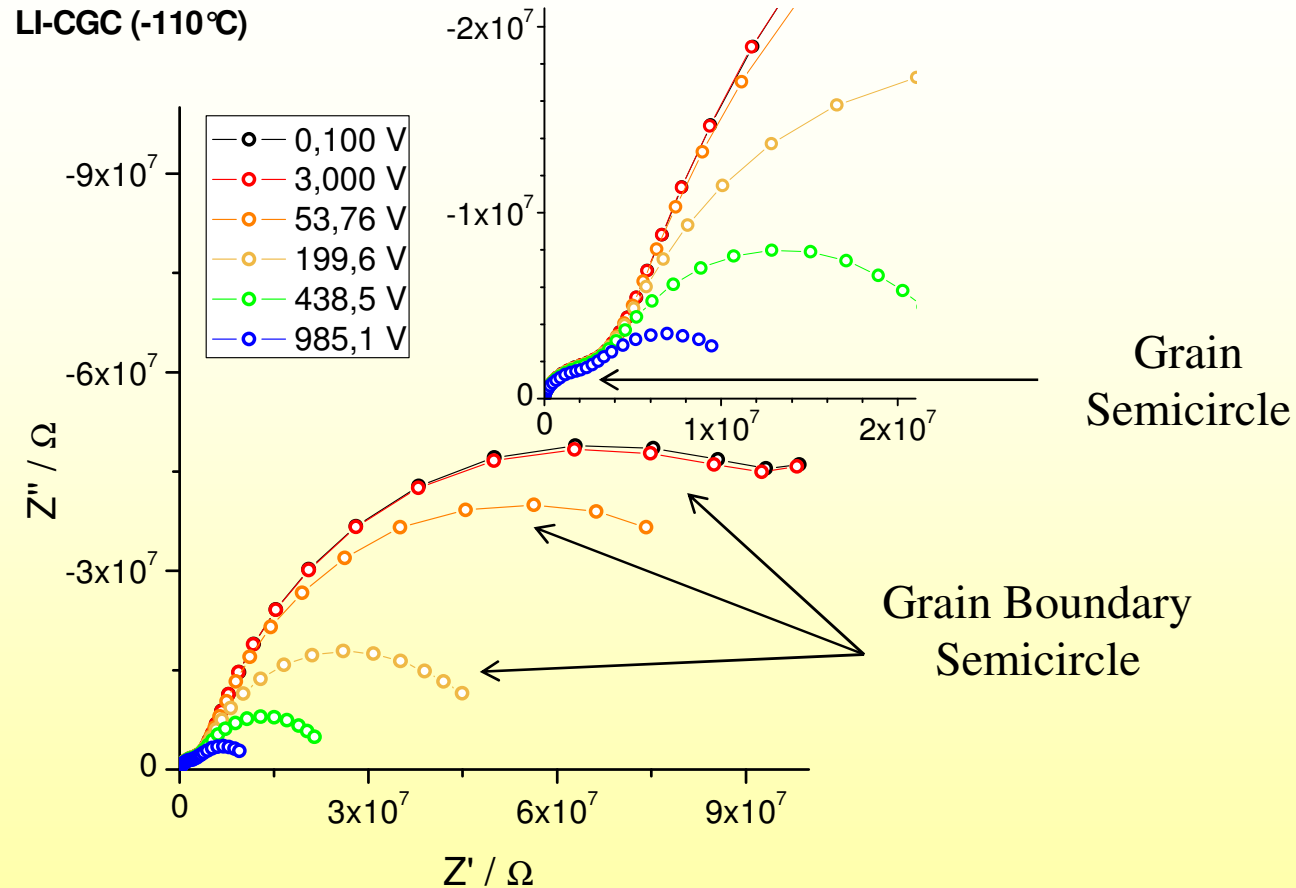
Novocontrol Alpha-AK High Performance Frequency Analyser, equipped with:

- High-Voltage Amplifier Trek model 623B
- Novocontrol HVB4000 High-Voltage Impedance Interface



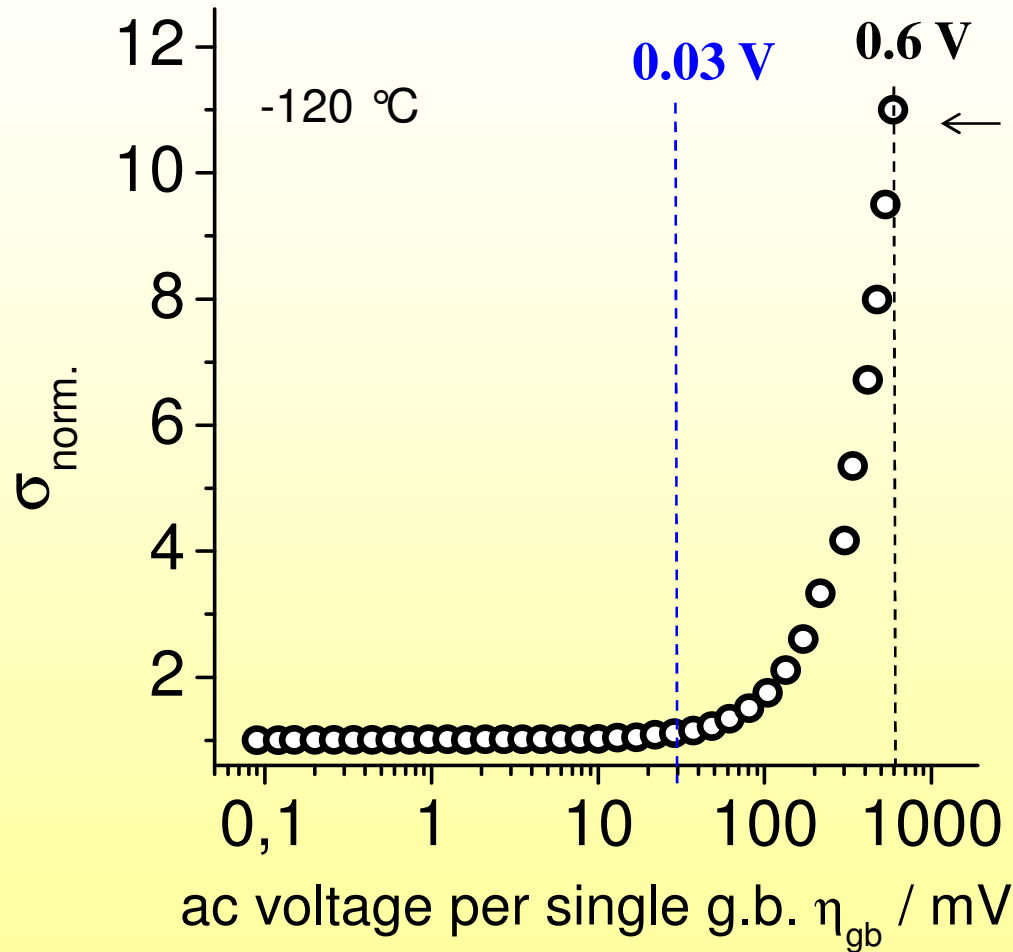
- Frequency range: 3 μ Hz - 10 kHz
- Maximum amplitude of ac voltage: 2 kV
- Current resolution: 5 fA

Nonlinear Impedance Spectroscopy with High AC Voltages



- **Almost 1 V ac voltage per single grain boundary can be applied without any irreversible changes of the grain boundary properties**
- **Grain boundary resistance decreases with increasing voltage**

Grain Boundary Conductivity vs. AC Voltage per Single Grain Boundary



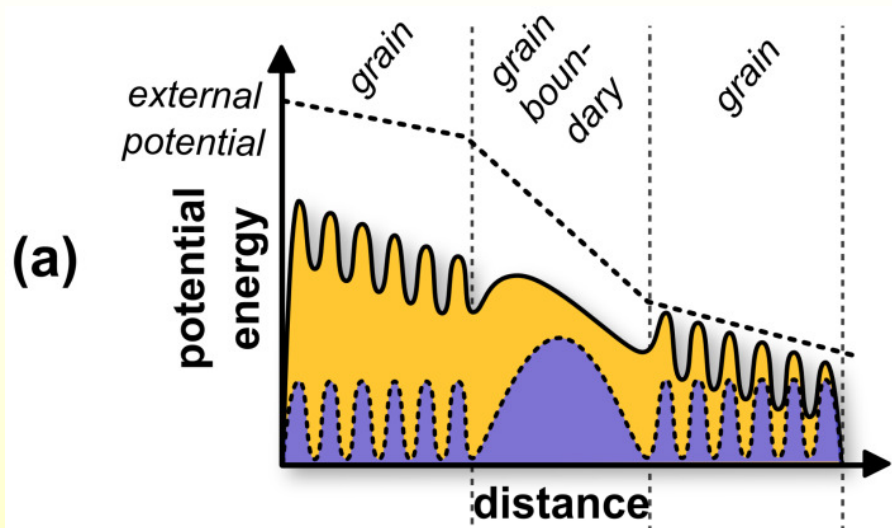
← Grain boundary resistance is still slightly higher than grain resistance

If a **single higher barrier** existed at each grain boundary

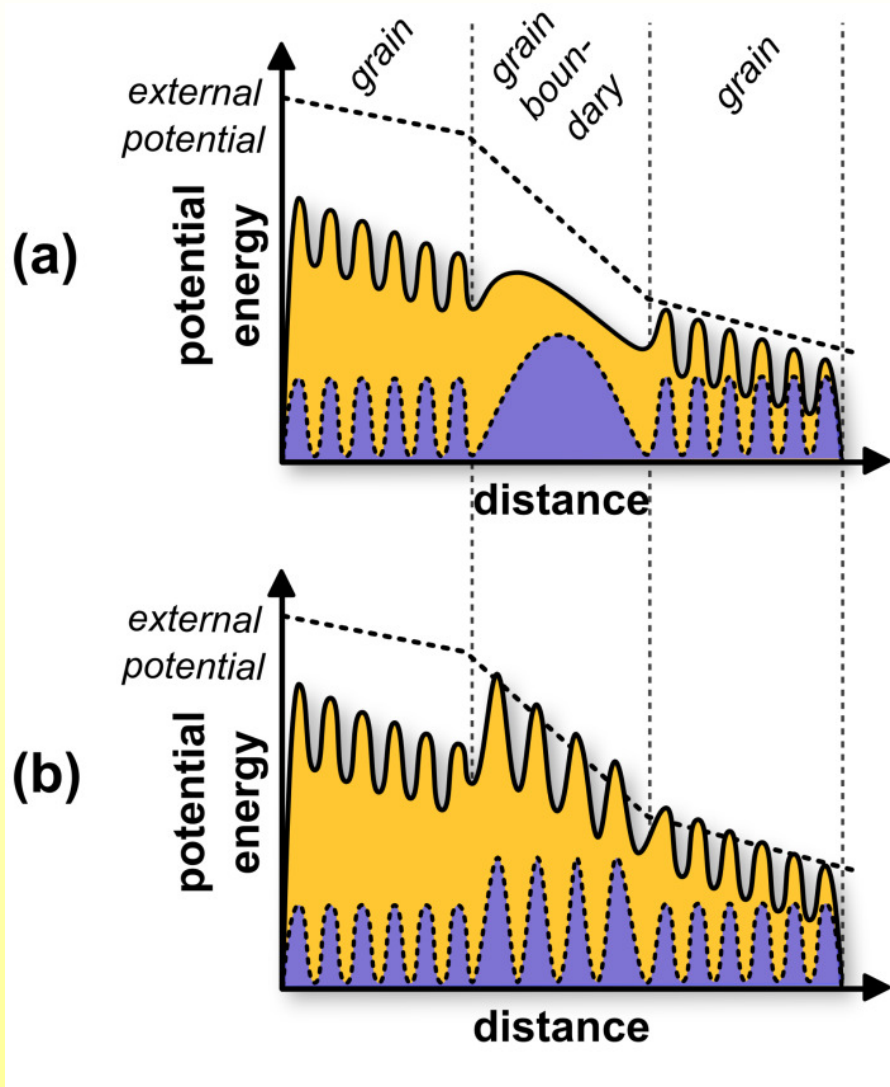


An external voltage of about 30 mV should be sufficient for decreasing the grain boundary resistance to the level of the grain resistance

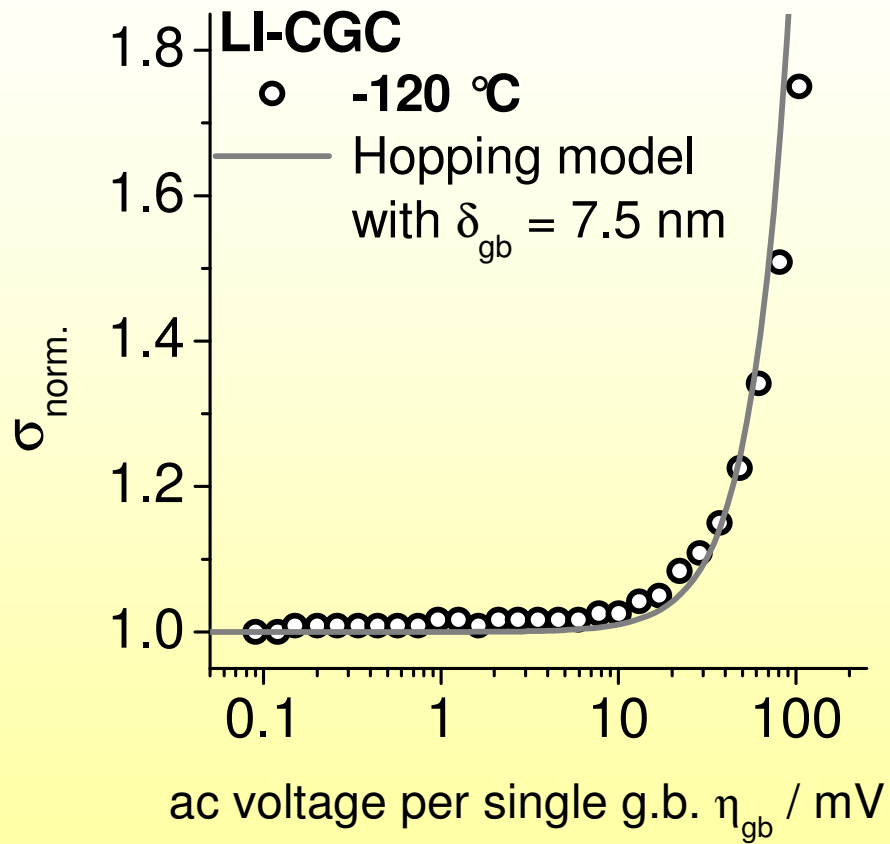
Influence of High Voltages on Ion Transport Across Grain Boundary



Influence of High Voltages on Ion Transport Across Grain Boundary



Estimation of Grain Boundary Thickness

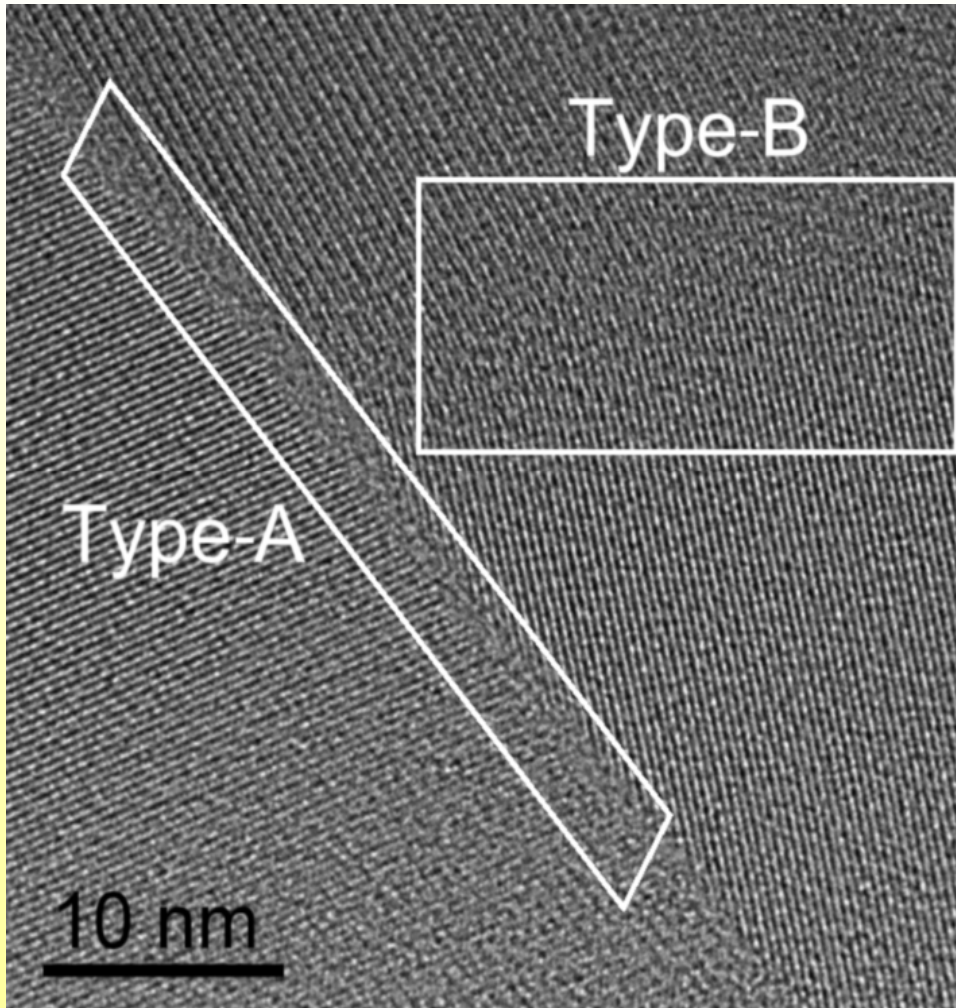


Assumption:
Field dependence of
gb conductivity follows
universal behaviour
of thin ion-conducting
layers.



Thickness of
grain boundary:
about 7.5 nm

HR-TEM Images of Grain Boundaries



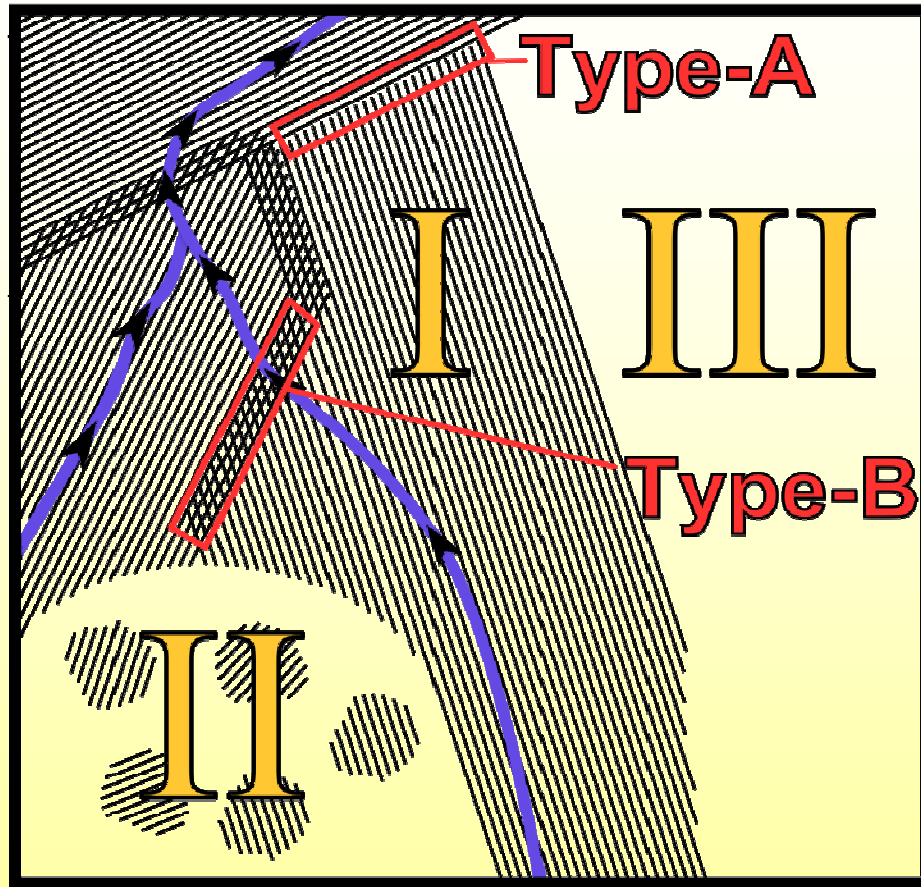
Type-B Grain Boundary:

- Layer between grains with similar lattice orientation:
- High degree of crystallinity
- Mechanical stresses may lead to slightly higher activation energy
- Thickness: about 5-10 nm

Type-A Grain Boundary:

Amorphous layer between grains with strongly dissimilar lattice orientation
→ Highly resistive

Model for Grain Boundary Transport



Blue lines: Ion transport pathways

Conclusions

$\text{Li}_{1.5}\text{Al}_{0.5}\text{Ge}_{1.5}(\text{PO}_4)_3$ (LAGP)

- Grain and grain boundary resistance exhibit the same activation energy.
→ **Purely geometrical current constriction; fraction of contacted area: about 25%**

Ohara Glass Ceramic ($\text{Li}_{1+x}\text{Al}_x\text{Ti}_{2-x}(\text{PO}_4)_3$ doped with other oxides)

- Grain boundary resistance exhibits a slightly higher activation energy than the grain resistance.
- Nonlinear impedance spectra provide strong indication that the grain boundary resistance is **not** caused by a **single (space charge) barrier**, but by **several serial barriers**
- Fit of nonlinear impedance data suggests that the **thickness of the grain boundaries is in the range 5-10 nm.**
- HR-TEM images reveal type-A (amorphous) and **type-B** (high degree of crystallinity) **grain boundaries.**
- **Thickness of type-B** grain boundaries is, in fact, in the range 5-10 nm. **Mechanical stresses** may lead to slightly higher activation energy.

Many thanks
for
your attention!