

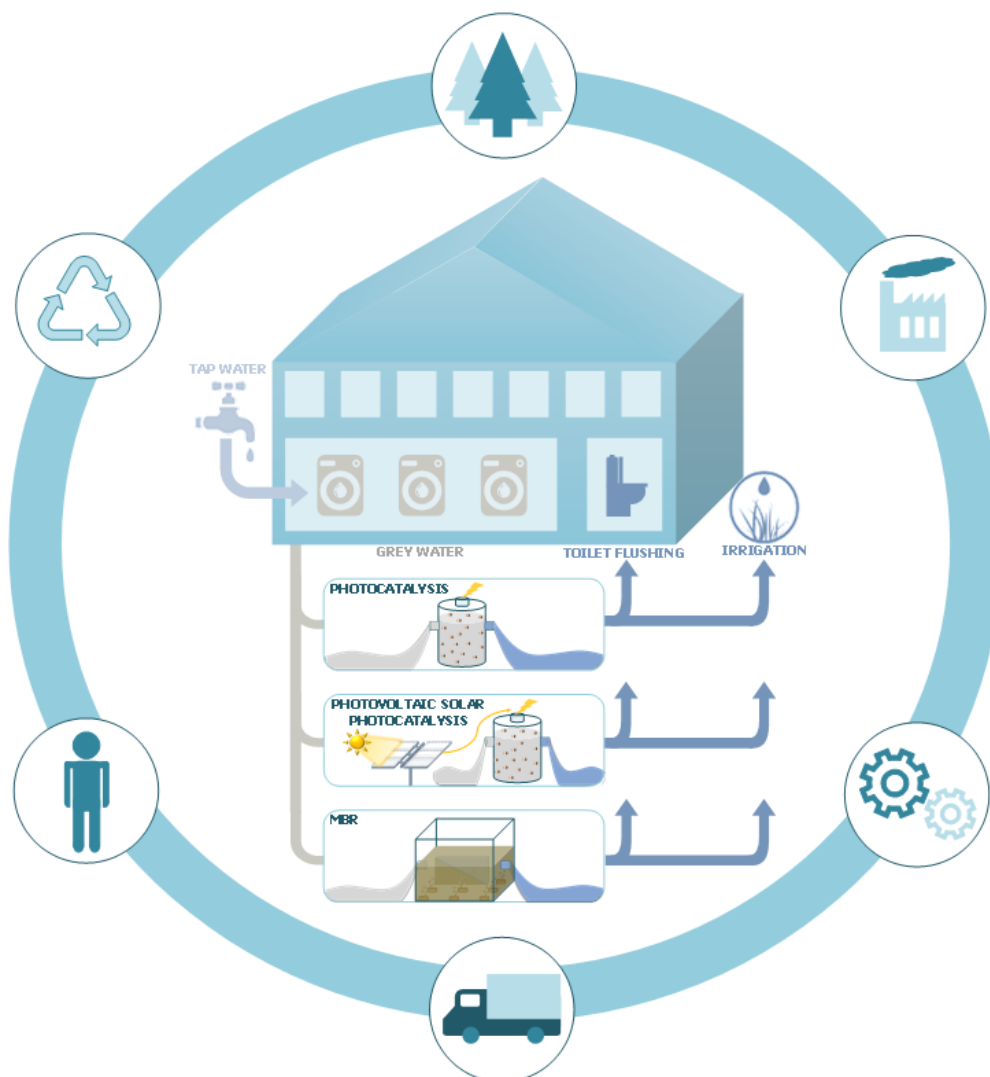
LCA of Greywater Management within a Water Circular Economy Restorative Thinking Framework

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GRAPHICAL ABSTRACT



12 **ABSTRACT**

13 Greywater reuse is an attractive option for the sustainable management of water under
14 water scarcity circumstances, within a water circular economy restorative thinking
15 framework. Its successful deployment relies on the availability of low cost and
16 environmentally friendly technologies. The life cycle assessment (LCA) approach
17 provides the appropriate methodological tool for the evaluation of alternative
18 treatments based on environmental decision criteria and, therefore, it is highly useful
19 during the process conceptual design. This methodology should be employed in the
20 early design phase to select those technologies with lower environmental impact. This
21 work reports the comparative LCA of three scenarios for greywater reuse:
22 photocatalysis, photovoltaic solar-driven photocatalysis and membrane biological
23 reactor, in order to help the selection of the most environmentally friendly technology.
24 The study has been focused on the removal of the surfactant sodium
25 dodecylbenzenesulfonate (SDBS), which is used in the formulation of detergents and
26 personal care products and, thus, widely present in greywater. LCA was applied using
27 the Environmental Sustainability Assessment (ESA) methodology to obtain two main
28 environmental indicators in order to simplify the decision making process: natural
29 resources (NRs) and environmental burdens (EBs). Energy consumption is the main
30 contributor to both indicators owing to the high energy consumption of the light source
31 for the photocatalytic greywater treatment. In order to reduce its environmental
32 burdens, the most desirable scenario would be the use of solar light for the
33 photocatalytic transformation. However, while the technological challenge of direct use
34 of solar light is approached, the environmental suitability of the photovoltaic solar
35 energy driven photocatalysis technology to greywater reuse has been demonstrated,
36 as it involves the smallest environmental impact among the three studied alternatives.

37

38 **Keywords**

39 Life cycle assessment (LCA); Light emitting diodes (LEDs); Membrane biological
40 reactor (MBR); Photocatalysis; Photovoltaic solar energy.

41

42 **1. Introduction**

43 The economic, environmental, and social impact of past water resources development
44 and the present water scarcity lead to a new paradigm in water resource management.

45 Therefore, the application of sustainable water supply solutions is essential (Ortiz et
46 al., 2015; Wilcox et al., 2016). In this scenario, the implementation of a circular
47 economy strategy results in a promising approach. This concept, has been already
48 introduced in several environmental policy initiatives of the European Commission
49 (EC) (European Commission, 2017a). The circular economy restorative thinking
50 demands that wastewater should be considered a valuable non-conventional resource
51 used to sustain scarce life-essential resources (Abu-Ghunmi et al., 2016). Therefore,
52 the development of wastewater recycling systems has gained attention over the last
53 years (Guo et al., 2014; Holloway et al., 2016; Wilcox et al., 2016). However, limited
54 awareness of potential benefits among stakeholders and the general public, and lack
55 of a supportive and coherent framework for water reuse are the major barriers currently
56 preventing a wider spreading of this practice in the EU. For these reasons the EC is
57 working on legislative or other instruments to boost water reuse when it is cost-efficient
58 and safe for health and the environment (European Commission, 2017b).

59 One of the most interesting alternatives is the on-site treatment and reuse of greywater
60 in households, hotels, and sport centers (Fountoulakis et al., 2016; Gabarró et al.,
61 2013; March et al., 2004; Merz et al., 2007; Sánchez et al., 2010). Greywater is

62 domestic wastewater originated in washing machines, kitchen sinks, baths, and hand
63 basins. Spanish law allows its recycling under several circumstances (Real Decreto
64 1620/2007, 2007). Hence, it is adequate for toilet flushing, irrigation, laundry, fire
65 extinguishing, groundwater discharge or car and window washing (Ghunmi et al., 2011;
66 Liberman et al., 2016; Santasmasas et al., 2013). This kind of water contains
67 surfactants, which are compounds commonly used in the formulation of detergents and
68 personal care products that represent an environmental hazard due to their low
69 biodegradability and their ability to provoke foams (Suárez-Ojeda et al., 2007). One of
70 the most representative surfactants is the sodium dodecylbenzenesulfonate (SDBS)
71 (Dominguez et al., 2016; Sanchez et al., 2010; Sanchez et al., 2011). Several methods
72 have been considered for greywater treatment in literature including biological,
73 chemical, and physico-chemical processes (Ghunmi et al., 2011). However, most of
74 these techniques are ineffective for the total removal of surfactants or they can only
75 transport these contaminants to a different phase resulting in a concentrated waste
76 volume (Dhouib et al., 2005). One of the most environmentally friendly options is the
77 use of constructed wetlands, however, their use is limited by the requirement of large
78 land spaces (Ghunmi et al., 2011).

79 Advanced oxidation processes (AOPs) have been presented as environmentally
80 friendly treatments for wastewater remediation; they achieve the successful
81 degradation of different contaminants of emerging concern (CECs) (Dominguez et al.,
82 2016; Rodríguez et al., 2016; Serra et al., 2011; Wankhade et al., 2013). AOPs are
83 based on the in situ generation of reactive oxidizing species (ROS), mainly hydroxyl
84 radicals ($\cdot\text{OH}$) (Fernández-Castro et al., 2014; Muñoz et al., 2006). Among them,
85 heterogeneous photocatalysis appears as an attractive emerging technology to treat
86 greywater because it avoids secondary pollution and works at ambient temperature

87 and pressure (Dominguez et al., 2016). As seen in Eq. 1, in this process a source of
88 appropriate light ($h\nu$) and a solid semiconductor material, the photocatalyst, are
89 necessary to promote the mineralization of the organic pollutant (Kumar and Bansal,
90 2013).

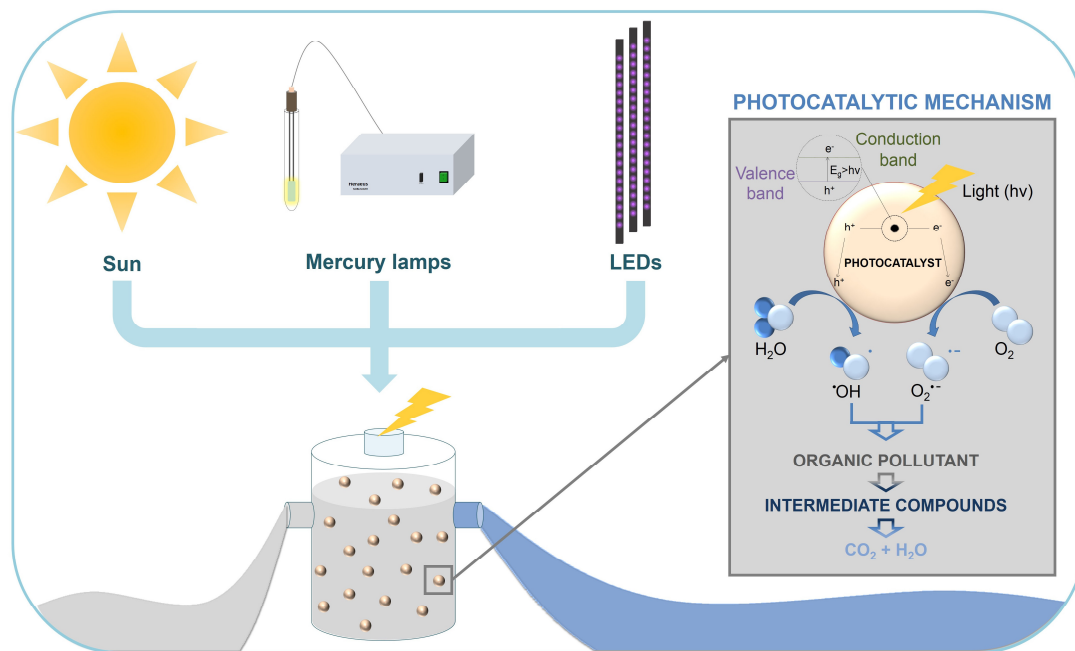
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94 Solar light is the most environmentally friendly light source (Fig. 1) and solar-assisted
95 photocatalysis has shown positive results over the last years in the removal of
96 emerging contaminants (Malato et al., 2016). However, several barriers still need to be
97 overcome for its full implementation worldwide (Spasiano et al., 2015). First, the solar
98 UV spectral irradiance reaching the Earth's surface is not homogeneous. Another
99 disadvantage already reported is that large areas might be required for the treatment
100 (Muñoz et al., 2006). Furthermore, the most commonly employed photocatalyst, TiO_2 ,
101 is only excited for radiations in the ultraviolet region (UV), which only represents about
102 3.00–4.00% of the solar spectrum (Spasiano et al., 2015). Thus, the effective
103 application of TiO_2 photocatalysts to the removal of recalcitrant compounds requires
104 artificial illumination (Ibhadon and Fitzpatrick, 2013). Hg lamps have known ample use
105 in laboratory studies; however, they have low efficiency in the transformation of energy
106 into light and short useful life, thus, making photocatalysis energy intensive. The use
107 of light emitting diodes (LEDs) provides a more energy efficient alternative with longer
108 useful life and lower price than the traditional photocatalytic mercury lamps (Song et
109 al., 2016). Besides, the use of solar photovoltaic panels as primary energy source
110 (Dominguez-Ramos et al., 2010) appears as the ultimate goal to convert
111 photocatalysis into a sustainable treatment.

112



113

114

Fig. 1. Light source alternatives in photocatalysis.

115

116 Another promising technical alternative to treat greywater consists in the use of
117 membrane biological reactors (MBR), which combine traditional activated sludge
118 biological treatment with membrane filtration (Atanasova et al., 2017; Chai et al., 2013;
119 Fountoulakis et al., 2016; Gander et al., 2000; Merz et al., 2007). This technology
120 provides high efficiencies in the removal of surfactants, good effluent quality, high
121 mixed liquor suspended solids concentrations, small space requirements, and reduced
122 sludge production (Chai et al., 2013; De Gisi et al., 2016; Dhouib et al., 2005; Gander
123 et al., 2000; Merz et al., 2007). Both, photocatalysis and MBR, have shown their
124 suitability for the treatment of greywater (Sánchez et al., 2010; Santasmasas et al.,
125 2013). Nevertheless, their deployment generates an environmental impact associated
126 with an intensive use of resources (chemicals and energy) and the construction of the
127 required infrastructures (Giménez et al., 2015; Rodríguez et al., 2016; Ortiz et al.,
128 2007). Thus, the application of the above-mentioned technologies should be preceded

129 not only by evaluation of the degradation and mineralization yield, but also by the
130 complete environmental assessment (Chatzisyneon et al., 2013; Giménez et al., 2015;
131 Rodríguez et al., 2016). In this sense, Life cycle assessment (LCA) appears as a
132 reliable methodology to define, evaluate, quantify and reduce the potential impacts of
133 the lifecycle stages (from 'cradle' to 'grave') of a product, activity or process
134 (Corominas et al., 2013; Margallo et al., 2014a; Serra et al., 2011), supporting the
135 environmental decision-making process (García-Herreó et al. 2017a). The inputs and
136 outputs of the system, such as energy, reagents, materials, emissions, waste, and
137 environmental impacts are quantified in LCA (Chong et al., 2010, Serra et al., 2011).
138 The implementation of the LCA tool in green chemistry processes supports the
139 development of more sustainable concepts based on the relationship between the
140 selection of compounds and process parameters and the resulting environmental
141 impacts (Kralisch et al., 2015). While LCA has been widely applied to MBR treatments
142 (Ortiz et al., 2007; Pretel et al., 2016; Zang et al., 2015), only scarce studies evaluating
143 the environmental performance of photocatalysis can be found in literature
144 (Chatzisyneon et al., 2013; Giménez et al., 2015; Muñoz et al., 2005). Furthermore, it
145 is worth remarking that most LCA studies applied to photocatalytic treatments are
146 performed in lab scale, which unquestionably limits the usefulness of the results
147 regarding the real large-scale application (Chatzisyneon et al., 2013; Giménez et al.,
148 2015; Muñoz et al., 2005).

149 Within these premises, this work provides an LCA study to assess and compare the
150 environmental impacts generated in the treatment of greywater by photocatalysis,
151 photovoltaic solar-driven photocatalysis, and MBR. It will also identify the
152 environmental bottlenecks in order to address the main technological challenges for
153 greywater reuse.

154

155 **2. Methodology**

156 LCA is carried out according to the requirements of the ISO 14040 and ISO 14044
157 international standards (ISO, 2006a; ISO, 2006b). Therefore, LCA is applied in the
158 following stages: definition of the goal and scope of the study, development of the life
159 cycle inventory (LCI), life cycle impact assessment (LCIA) and results interpretation.

160

161 2.1. Goal and scope

162 This research aims to assess the environmental sustainability of three alternatives for
163 greywater treatment, photocatalysis, photovoltaic solar-driven photocatalysis, and
164 MBR. It provides an appropriate framework to evaluate the opportunities for process
165 success leading also to the identification of hot-spots, which are the stages with the
166 highest environmental impact. The purpose of the system is to treat greywater with
167 high degree of removal of SDBS, allowing its reuse for toilet flushing and garden
168 irrigation. SDBS has been selected as target pollutant due to its environmental
169 persistence and because the treatment is applied to hotel laundry greywater, where
170 SDBS is a key component. Thus, the functional unit is defined on the basis of the same
171 treated volume of greywater and the same amount of SDBS removed. In order to
172 establish the amount of SDBS removed, a minimum threshold accomplished by the
173 three scenarios within a given treatment time has to be selected (Muñoz et al., 2005).
174 Therefore, 1.00 m³ of treated greywater with 90.0% reduction of the SDBS initial
175 concentration is designated as functional unit. All the mass and energy inputs and
176 outputs will be referred to this unit. The use of a similar functional unit that considers
177 the same treated water volume and a fixed reduction level of the contaminant has been
178 previously reported in literature. For instance, Muñoz et al. (2005) defined as functional

179 unit the removal of 15.0% DOC from 1.00 m³ kraft pulp mill wastewater, and Serra et
180 al. (2011) selected as functional unit the removal of 93.0% total organic carbon in 250
181 mL of wastewater with 500 mg L⁻¹ of α -methyl-phenylglycine.

182 The study is carried out from a 'cradle to gate' pathway, considering the extraction,
183 production, and transportation of raw materials, the greywater treatment and the
184 management of generated waste. This approach is developed for three scenarios
185 based on photocatalytic, photovoltaic solar photocatalytic, and MBR technologies.

186 *Scenario 1 (Sc. 1)*, photocatalytic technology: photocatalytic studies were performed
187 in laboratory to obtain kinetic data, and after modeling the process, scale-up was
188 carried out. The commercial photocatalyst used is TiO₂ Aeroxide[®] P25 (Evonik
189 Industries). One g L⁻¹ of TiO₂ was added to the effluent and kept for 0.50 h premixing
190 in the dark to reach adsorption equilibrium before the photocatalytic treatment started.
191 The photocatalyst loading was selected after the results attained in preceding works
192 (Dominguez et al., 2016). The photocatalytic reactor (APRIA Systems S.L. Photolab
193 LED/160) is constituted of 1.00 L jacketed annular reactor, 5.00 L mixing tank and 40
194 LEDs LZ1-00U600 (LED Engin). LEDs emit in a wavelength between 365 nm and 370
195 nm, being the total electrical power between 1.00 W and 100 W. A fan (San Ace 80,
196 Sanyo Denki) is used to keep LEDs temperature in the suitable range (20.0 °C - 30.0
197 °C) to keep constant radiation over time and high lamp lifetime. SDBS concentration is
198 quantified by means of an UV-1800 spectrophotometer (Shimadzu) at 223 nm. The
199 waste TiO₂ obtained after the photocatalytic treatment is sent to a municipal landfill.

200 *Scenario 2 (Sc. 2)*, photovoltaic solar-driven photocatalysis: the photocatalytic studies
201 detailed in Sc. 1 were used for process scale-up as well. Since the existing photovoltaic
202 panels have different materials and processing requirements that leads to diverse
203 emission profiles, a global average share of different photovoltaic panels is considered;

204 these include mono-silicon 47.7%, multi-silicon 38.3%, cadmium-telluride 6.4%,
205 amorphous-silicon 5.10%, ribbon-silicon 1.50%, and copper-indium-gallium-diselenide
206 1.00%.

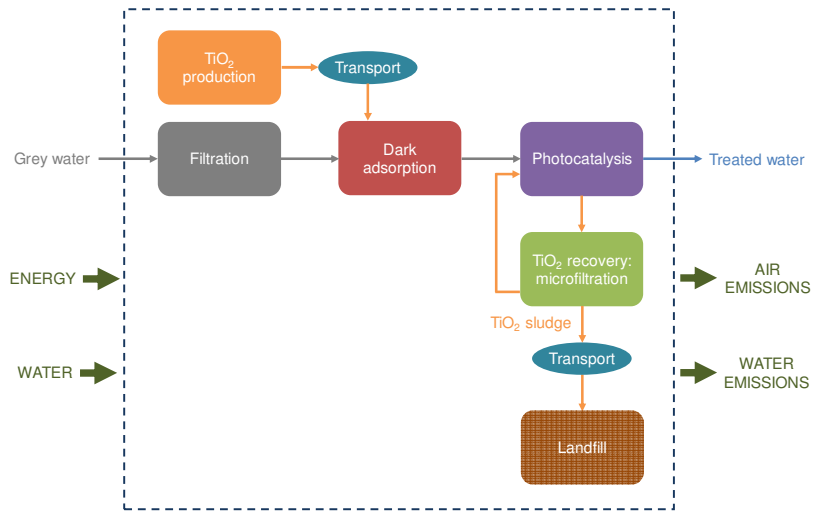
207 *Scenario 3 (Sc. 3)*, MBR technology: all the data have been collected from literature.
208 The selected MBR has a submerged configuration in order to reduce energy
209 consumption (Khan et al., 2016). The membrane is a flat sheet ultrafiltration
210 polyethersulfone membrane with 50 nm and a permeate flux of 19.2 L m⁻² h⁻¹
211 (Santasmás et al., 2013). The hydraulic retention time (HRT) is estimated as 25.3 h
212 (Santasmás et al., 2013). It is assumed that the sludge retention time (SRT) is 35
213 days (Gori et al., 2010) because high SRTs cause endogenous respiration in the
214 biomass reducing the sludge production (Gander et al., 2000). For the biomass
215 conditions an average mixed liquor total suspended solids (MLTSS) of 8.00 g L⁻¹ is
216 taken as reference (Gori et al., 2010). The sludge is supposed to be treated by
217 incineration and then deposited in a municipal landfill, which is one of the most
218 common processes in the wastewater area. However, an alternative option consisting
219 on sludge compost-stabilization for its land application has been also analyzed (Sc.
220 3b).

221 Fig. 2 shows the flow diagram and the system boundaries considered for each
222 treatment. The main system flows are the energy inputs, water, manufacture of the
223 reagents used in each treatment (extraction of resources, manufacture, and transport)
224 and their outputs to the environment. The systems boundaries for Sc. 1 and Sc. 2 are
225 the same since the only difference between both scenarios is the method to obtain the
226 required energy, being the electricity grid in Sc. 1 and renewable energy in Sc. 2.

227 It is to be highlighted that in order to simplify the LCA application, the infrastructure
228 related to the three greywater treatments has not been considered (Giménez et al.,

229 2015). Moreover, the contribution of the infrastructure to the impacts of these
 230 processes is typically negligible owing to the long lifetimes of the considered industrial
 231 installations and because its impact is insignificant compared to the impact produced
 232 by the operation phase (Garcia-Herrero et al., 2017a; Hospido et al., 2012).
 233

a).

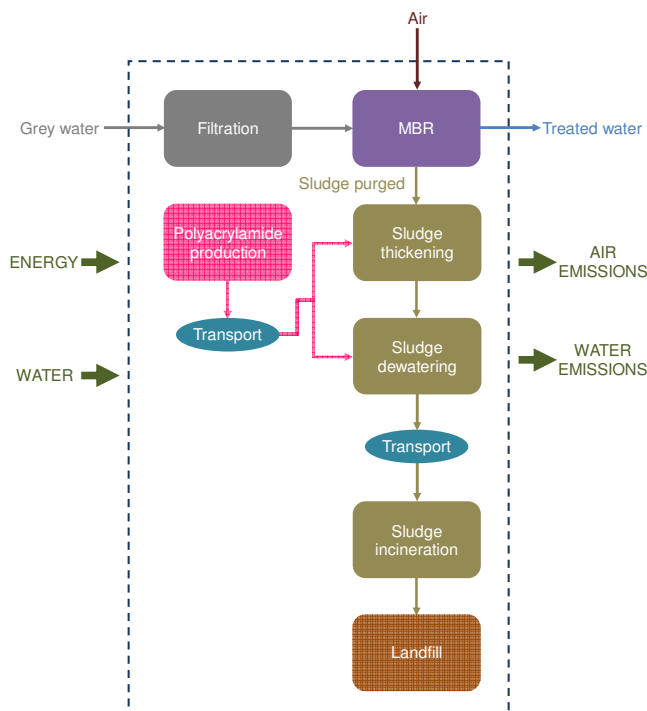


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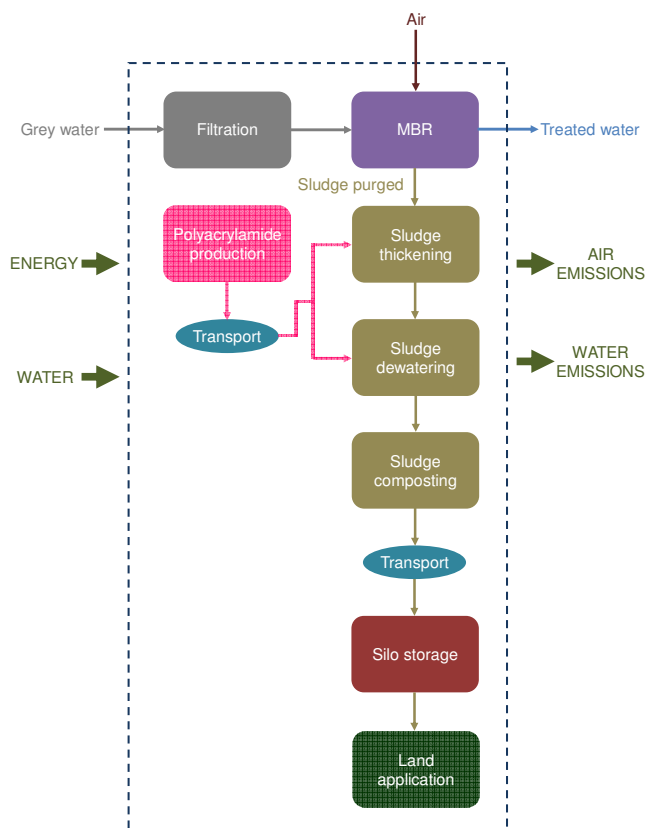
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b).



237

c).



238

239 **Fig. 2. System boundaries for greywater treatment in a). Sc. 1 (photocatalysis)**
240 **and Sc 2. (photovoltaic solar-driven photocatalysis), b). Sc. 3 (MBR with sludge**
241 **incineration) and c). Sc. 3b (MBR with sludge composting).**

242

243 The three scenarios are multi-functional processes, in which greywater treatment is
244 the main function, and the recovery of energy in the landfill site and in the incinerator
245 are additional functions. Furthermore, a modification of Sc. 3 has been set out in the
246 MBR variation assessment. In this case (Sc. 3b), after composting, the sludge is used
247 as fertilizer, adding a new function to the system. In these systems, the environmental
248 burdens associated with a particular process must be partitioned over the various
249 functional flows of that process (Margallo et al., 2014b).

250 According to the ISO recommendation, this work solved the existence of additional
251 functions gaining credit by the reduction of the emissions related to the co-products.
252 That is to say, the impact of the co-product manufacturing is subtracted from the

253 original systems. In this case, for energy and material valorization, the 'avoided'
254 emissions of conventional production of electricity and fertilizer were subtracted from
255 those produced during waste treatment.

256 This procedure requires identifying the type of material substituted or displaced. In Sc.
257 3, the energy mix is the substituted process, whereas in Sc. 3b the displaced fertilizer
258 is ammonium sulfate.

259

260 2.2. Life cycle inventory

261 The mass and energy flows considered within the scope of the work are recorded in
262 the life cycle inventory (LCI), which collects the most relevant input and output data for
263 the scenarios under study in separate unit processes. In this work the data are taken
264 either from fieldwork (Dominguez et al., 2016) or from literature; the sources and quality
265 of the LCI per functional unit are depicted in Table 1, and detailed in Table 2. The
266 natural resources consumption and the environmental burdens associated to the
267 systems can be estimated from these values.

268 The main hypothesis assumed in the inventory phase of the LCA can be summarized
269 as follows:

- 270 - For the process scale-up and estimation of energy consumption, reagents, and
271 waste, both scenarios are assumed to be implemented in a hotel laundry to treat
272 greywater with 50.0 mg L⁻¹ of SDBS.
- 273 - The treatment is assumed to be carried out in Santander, Cantabria, Spain, in
274 a hotel of 75 guests.
- 275 - It is assumed that each guest produces 1 kg of laundry per day, including 2 bed
276 sheets, 1 pillow slip and 1 towel (Filimonau et al., 2011), and that 13 L of fresh
277 water are required to wash 1 kg of laundry (Máša et al., 2013).

- 278 - The photocatalytic treatment works in batch mode, 20.6 h day⁻¹, all year round.
279 This time has been extrapolated from the results previously obtained at
280 laboratory scale by the authors (Dominguez et al., 2016).
- 281 - The MBR works in continuous mode, with a hydraulic retention time (HRT) of
282 25.6 h. This value has been estimated taking into account data taken from
283 literature (Santasmayas et al., 2013).
- 284 - To improve data quality and consider the local idiosyncrasy in Sc. 1 and Sc. 3,
285 the electricity mixed provided by the PE database is adapted to the
286 characteristics of the Spanish mix of 2016, which contains 40.5% of renewable
287 sources and 35.3% of fossil fuel based sources.
- 288 - The energy employed in Sc. 2 is taken from photovoltaic solar panels.
- 289 - The electricity consumption corresponds to a treatment time required to remove
290 90.0% of the initial SDBS concentration, being 19.5 h for photocatalysis and
291 25.6 h for MBR. Moreover, in the case of the photocatalytic treatment an
292 additional time of 0.50 h has been considered for dark adsorption of the
293 photocatalyst, 0.14 h for pumping the greywater to the system and 1.00 h for
294 pumping the treated water during the TiO₂ separation step.
- 295 - A photocatalyst recovery stage by means of microfiltration membranes has
296 been taken into account in the case of the photocatalytic treatment (Rivero et
297 al., 2006). It is assumed that the TiO₂ is fully recovered and it can be reused 10
298 times in a closed cycle (Muñoz et al., 2006). Then it is disposed of in landfill; it
299 should be transported along 32.8 km by a 28 tones Euro 4 truck.
- 300 - TiO₂ is delivered to the consumer after transport by a Euro 4 truck with a
301 maximum total capacity of 28 tones along 1596 km from the production plant of

302 Evonik Industries in Frankfurt, Germany (Evonik Industries, 2017; Muñoz et al.,
303 2005).

304 - The manufacturing of the membranes is considered as part of the infrastructure
305 and, therefore, it is not considered in this work.

306 - In the MBR treatment the membrane-cleaning step is based on air scouring
307 avoiding backwashing cycles or the use of chemicals (Liberman et al., 2016).

308 - The data used for the sludge treatment are recompiled from literature (Hospido
309 et al., 2005; Suh and Rousseaux, 2002). The sludge is thickened and dewatered
310 on-site; the addition of polyacrylamide is required in both stages. Then, it is
311 transported by a 28 tones Euro 4 truck along 32.8 km to an incineration plant
312 located in a landfill site placed in Meruelo, Cantabria, Spain, where it is treated
313 and disposed of (Suh and Rousseaux, 2002).

314 - The polyacrylamide is transported by a 28 tones Euro 4 truck along 722 km after
315 its manufacture in a plant of Derypol, S.A. in Les Franqueses del Vallés, Spain
316 (Derypol, 2017).

317

318 2.3. Life cycle impact assessment

319 The life cycle impact assessment (LCIA) calculates environmental indicators from the
320 LCI data. It implies further classification and characterization of these indicators,
321 including their additional and non-mandatory normalization and weighting (Garcia-
322 Herrero et al., 2017b). In this work, the software selected for the modeling of the two
323 treatments under study is the LCA software GaBi 6.0 and the database of PE
324 International (PE International, 2016).

325 The Environmental Sustainability Assessment (ESA) method followed in this work was
326 initially developed by Irabien et al. (2009). Accordingly, a first classification stage is

327 performed in which the inventory data are organized in different impact categories.
328 Then, the possible impact of each resource consumption or emission is estimated
329 using a characterization factor (CF) (Garcia-Herrero et al., 2017b).
330 To conduct the environmental assessment, the two main indicators considered are the
331 natural resources consumption (NRs) and the environmental burdens (EBs). The
332 consumption of energy ($X_{1,1}$), materials ($X_{1,2}$), and water ($X_{1,3}$) are considered within
333 the NRs and the primary burdens to air ($X_{2,1}$), water ($X_{2,2}$), and land ($X_{2,3}$) are included
334 in the EBs. These indicators are based on the environmental sustainability metrics
335 established by the Institution of Chemical Engineers (IChemE, 2002). Specifically, the
336 EBs are classified in 12 impact categories. The atmospheric burdens are atmospheric
337 acidification (AA), global warming (GW), human health effects (HHE), photochemical
338 ozone formation (POF), and stratospheric ozone depletion (SOD). The impact
339 categories for the water burdens are aquatic acidification (AqA), aquatic oxygen
340 demand (AOD), ecotoxicity to aquatic life (metals to seawater) (MEco), ecotoxicity to
341 aquatic life (other substances) (NMEco), and eutrophication (EU) (García et al., 2013).
342 For the land burdens the categories are given by the amount of hazardous and non-
343 hazardous waste produced and its management (Margallo et al., 2014a).
344 Since the environmental sustainability indicators employed in this study are expressed
345 in different units depending on the environmental impact category considered, their
346 normalization is recommended. Therefore, with the purpose of conducting a
347 comparison in a common basis, dimensionless impacts indicators are required
348 (Garcia-Herrero et al., 2017a). The NRs are normalized regarding the natural resource
349 with the highest impact and the EBs regarding the threshold values specified in the
350 European Pollutant Release and Transfer Register (E-PRTR, 2006).

351

352 Eq. (2) and Eq. (3) show the calculations used to normalize the NRs and EBs:

353
$$X_{1,i}^* = \frac{X_{1,i}}{X_{1,i}^{ref}} \quad (2)$$

354
$$X_{2,j,k}^* = \frac{X_{2,j,k}}{X_{2,j,k}^{ref}} \quad (3)$$

355 where “i” represents the NRs indicators (energy, materials, and water), “j” symbolizes
356 the environmental compartments (air, water, and land) and “k” designates the
357 environmental impacts to the corresponding compartment.

358 Then, $X_{1,i}$ represents the consumption of each NRs, $X_{1,i}^*$ is the normalized $X_{1,i}$, $X_{1,i}^{ref}$ is
359 the reference natural resource, $X_{2,j,k}$ designates the environmental burdens to the
360 corresponding compartment, $X_{2,j,k}^*$ is the normalized $X_{2,j,k}$, and $X_{2,j,k}^{ref}$ is the reference
361 environmental burden.

362 After normalization, a weighting stage is developed. This procedure ranks the different
363 impact categories taking into account their relative importance (EC JCR, 2010). Thus,
364 the normalized NRs and EBs variables are aggregated as shown in Eq. (4) and Eq.
365 (5):

366
$$X_1 = \sum_{i=1}^{i=n} \alpha_{1,i} \cdot X_{1,i}^* \quad n \in [1, 3] \quad (4)$$

367
$$X_{2,j} = \sum_{k=1}^{k=m} \beta_{2,j,k} \cdot X_{2,j,k}^* \quad m \in [1, 5] \text{ if } 1 \leq j \leq 2 \wedge m \in [1, 2] \text{ if } j = 3 \quad (5)$$

368 where $\alpha_{1,i}$ is the weighting factor for the NRs and $\beta_{2,j,k}$ is the weighting factor for the
369 EBs.

370 In this work it is considered that the three natural resources are equally important, then
371 $\alpha_{1,i}$ is 1/3 for each i. This assumption is taken as it is the best way to obtain a single

372 indicator that allows comparison of the three proposed greywater treatments (Margallo
373 et al., 2014a).

374

375 **3. Results and discussion**

376 3.1. Natural resources

377 The consumption of NRs, including energy ($X_{1,1}$), materials ($X_{1,2}$), and water ($X_{1,3}$), is
378 analyzed for all the scenarios. The results are normalized regarding the natural
379 resource with the highest impact, which is water for the three scenarios (Table 3).

380 The energy ($X_{1,1}$) embraces the consumption of electricity, steam, diesel, and natural
381 gas. Sc. 2 is the most energy intensive, bringing the total energy demand close to 1304
382 MJ, while in Sc. 1 and Sc. 3 the energy demand is 450 MJ and 162 MJ, respectively.

383 As it can be observed in Table 4, 99.5% of the energy consumed in Sc. 1 and 99.84%
384 of the one required by Sc. 2 is demanded by the photocatalytic process. This is mainly
385 due to the intensive energy demand of the light source, which represents the main hot-
386 spot of the system. Therefore, the influence to $X_{1,1}$ of cleaning water and transport,
387 production, consumption, and end of life of TiO_2 are below 0.50% in the three scenarios
388 and, thus, it can be considered negligible. It has to be highlighted that in Sc. 3 the $X_{1,1}$
389 takes negative values in the sludge treatment stage due to the fact that during
390 incineration thermal energy is produced.

391 Within the materials ($X_{1,2}$), TiO_2 is considered for Sc. 1 and Sc. 2 while air and
392 polyacrylamide are taken into account for Sc. 3. Nevertheless, it is necessary to assess
393 not only the amount of materials but also the toxicity and environmental impacts of
394 their production and consumption. This point will be analyzed in the next section by
395 means of the study of the environmental burdens. The results show that the demand
396 of materials associated to the primary energy transformation is the major contributing

397 factor to this indicator. The consumption of material resources is significantly higher in
398 Sc. 3, 2481 kg, than in Sc. 1, 77.1 kg, and Sc. 2, 24.1 kg. The main reason behind this
399 result lies on the high demand of air required by the MBR, implying high consume of
400 materials for the energy required in the aeration process. This behavior was also
401 previously reported in literature, where the energy consumption required for the
402 aeration is the parameter that has the most significant influence in the environmental
403 performance of biological reactors (De Feo and Ferrara, 2017). In Sc. 1, although the
404 consumption of materials in the photocatalysis represents 92.4% of the indicator,
405 cleaning water and TiO₂ production have contributions of 6.66% and 0.79%,
406 respectively. In the case of Sc. 2, the intake of materials in the photocatalysis
407 diminishes to 69.2%, while the TiO₂ production increases to 11.0% and the cleaning
408 water to 19.5%. In Sc. 3, the production of polyacrylamide, the sludge treatment and
409 the transport have a contribution below 0.01% to X_{1,2}, because the aeration required
410 by the MBR has a contribution near 100%.

411 Despite the fact that water consumptions for the reagents production and for cleaning
412 are included within the indicator X_{1,3}, the hot-spot is the water required for the primary
413 energy transformation. According to Table 4, the value ranges from 96.84% for Sc. 2
414 to 100% for Sc. 3, being the contribution of other stages to the indicator X_{1,3} minimal.
415 Consistent with the results, Sc. 3 has the greatest global consumption of NRs (X₁ =
416 0.57), displaying a value 1.62 times higher than Sc. 1 (X₁ = 0.35) and 1.46 times higher
417 than Sc. 2. This behavior is mainly due to the high energy consumption in the aeration
418 (2072 m³ m⁻³ greywater).

419

420

421

422 3.2. Environmental burdens

423 The environmental burdens to air and water are estimated following the methodology
424 explained above. Before the normalization process, global warming represents the
425 highest impact in all the scenarios. The main reason is the emission of greenhouse
426 gases during energy production (CO₂, CO, etc.), the consumption of coal and energy
427 in the manufacture of reagents (CH₄, CO, CO₂, NO_x, N₂O), diesel consumption and
428 production and landfill emissions (NO_x, N₂O), and the transport of reagents and wastes
429 (NO_x, N₂O). It is worth noticing that in the energy consumption for the Sc. 1 and Sc. 3,
430 the electricity grid mix selected might have an important impact on the quantity of
431 greenhouse gas emissions and in derived results (De Feo and Ferrara, 2017).
432 Therefore, as it was previously specified, the Spanish mix of 2016 is selected for both
433 scenarios. The smallest score for this environmental burden is obtained in Sc. 2 (2.14
434 kg CO₂ eq.) being almost 6-fold smaller than in Sc. 1 (12.7 kg CO₂ eq.) and 2-fold
435 smaller than in Sc. 3 (4.42 kg CO₂ eq.). Regarding the aquatic indicators, the EU has
436 the highest impact on the three scenarios before the normalization owing to the
437 emissions of nitrogen, ammonia, phosphate, and chemical oxygen demand during
438 energy production.

439 Table 5 shows the EBs to air and water normalized using the European threshold
440 values (E-PRTR, 2006). After normalization, the HHE and POF become the most
441 important categories among air metrics for the three scenarios. The principal reason is
442 that, although GW has the highest air impact, when it is referenced to its threshold
443 value (1.00·10⁸ kg CO₂ eq.) the normalized results are reduced by 8 orders of
444 magnitude. Nonetheless, lower thresholds for HHE and POF are used as reference
445 (1000 kg benzene eq. and 1000 kg. ethylene eq., respectively). In the case of water

446 impacts, there are no significant differences after the normalization process because
447 the threshold values are lower than those in the air categories.

448 The EBs to air in Sc. 2 are smaller than in the other two scenarios for all the indicators
449 with the exception of the HHE. This high contribution to human toxicity in Sc. 2 is due
450 mainly to the extraction of raw materials and the manufacturing of components for the
451 photovoltaic solar panels fabrication. For instance, regarding the copper part of the
452 cables, electric components, and electronic devices, the toxicity is frequently related to
453 the mining and processing of the raw metal, particularly to the disposal of sulfidic ore
454 tailings (Corona et al., 2017). Nevertheless, the development of photovoltaic panels
455 that do not require toxic elements such as cadmium or rare elements like tellurium is
456 under study (Tsang et al., 2016), which will diminish the influence of the HHE in the
457 photovoltaic solar-driven photocatalysis in the future.

458 Regarding EBs to water, all the indicators are slightly smaller in Sc. 2 than in the other
459 scenarios. Sc. 3 shows the highest total aquatic EBs due to its high NMeCo value,
460 behavior mostly associated with the disposal of sludge incineration wastes (Pretel et
461 al., 2016). Additionally, the total EBs of Sc. 1 are slightly higher due to the high energy
462 demand of the light source in photocatalysis. However, LEDs have been evolving
463 rapidly over the last few years (Song et al., 2016), the development of energy efficient
464 LEDs with the same intensity of radiation but less electricity demand seems feasible.
465 Thus, an extraordinary environmental progress of the photocatalytic treatment seems
466 feasible within the upcoming years.

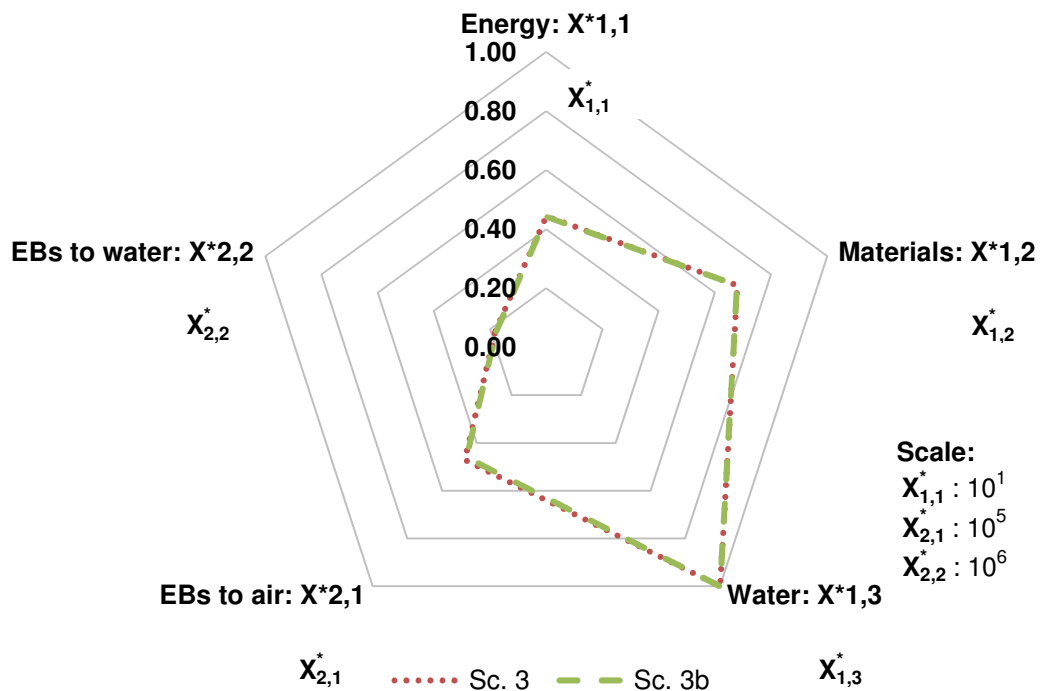
467

468 3.3. MBR variation assessment

469 Since the EBs to water in Sc. 3 are slightly higher than in Sc. 1 and Sc. 2, due to the
470 landfilling of the sludge, a variation in the MBR has been performed in order to assess

471 the environmental impact of an alternative sludge. Therefore, a new scenario, Sc. 3b,
 472 is considered. In this process, the sludge is thickened and dewatered on-site following
 473 the same procedure as in Sc. 3. Nevertheless, after the dewatering process, the sludge
 474 is stabilized by composting and then transported and stored for several days before its
 475 use in land stabilization. All the data used for the analysis are taken from literature
 476 (Hospido et al., 2005; Suh and Rousseaux, 2002). The results obtained are shown in
 477 Fig. 3. Although a reduction in the NRs and EBs is expected in Sc. 3b, both alternatives
 478 have similar environmental performance. This is because in the MBR what causes
 479 greater consumption of resources and generation of impacts is the energy used in the
 480 aeration of the reactor and, thus, the loads avoided, both by incineration and by
 481 composting, are minimal compared to aeration.

482



483

484 **Fig. 3. NRs and EBs dimensionless variables for Sc. 3 (MBR with sludge**
 485 **incineration) and Sc. 3b (MBR with sludge composting).**

486

487 The EBs for both scenarios are detailed in Table 6. Sc. 3 shows slightly higher EBs
488 than Sc. 3b. This trend is observed for all the indicators but for the water aquatic
489 ecotoxicity, due to the presence of heavy metals in the sludge applied to agricultural
490 fields in Sc. 3b. However, it has to be remarked that the presence of heavy metals in
491 air is more important than in the aquatic medium because they have more possibilities
492 to directly contact human beings. Regarding the air burdens, Sc. 3 shows a higher
493 global warming indicator as a result of the greenhouse gases emissions from the
494 incineration step. It has to be highlighted that the human toxicity is the indicator with
495 the highest contribution to the EBs in Sc. 3, owing to the heavy metals present in the
496 gaseous effluent generated during the incineration of the sludge (Suh and Rousseaux,
497 2002). Furthermore, in Sc. 3 the stratospheric ozone depletion also shows a high value
498 due to the landfill gas emissions originated when the incinerated sludge is landfilled.
499 Thus, taking all this into account, Sc. 3b can be considered the best alternative for the
500 MBR treatment of greywater.

501

502 **4. Conclusions**

503 This work provides technological and environmental decision criteria to use clean,
504 safe, and renewable solar energy for the treatment of greywater under a circular
505 economy of water. The LCA methodology is applied to evaluate the environmental
506 impacts of three greywater treatment alternatives, photocatalysis, photovoltaic solar-
507 driven photocatalysis, and MBR. The analysis shows that photovoltaic photocatalysis
508 driven by solar energy is the most sustainable scenario from the environmental point
509 of view. The variable that contributes mostly to the use of natural resources and the
510 generation of environmental burdens is energy consumption. This is due to the high
511 energy requirements of the light source, which is the main bottleneck of photovoltaic

512 solar-driven photocatalysis and photocatalysis scenarios. Therefore, this study
513 determines the main hot-spot of an emerging technology such as photocatalysis. The
514 analysis and the results allow to promote the deployment of the technology through its
515 combination with photovoltaic solar energy. This can be considered as the first step in
516 establishing the best available techniques for greywater reuse.

517 Despite the higher consumption of natural resources observed in the MBR_r, due to the
518 high air consumption, their EBs are lower than in the photocatalysis scenario.
519 However, due to the landfill of the sludge, the EBs to water in the MBR scenario are
520 slightly higher than in the photocatalysis and photovoltaic solar photocatalysis
521 scenarios.

522 Taking into account the environmental assessment of the greywater reuse process
523 through the scenarios considered, future technological challenges have to be
524 addressed under an environmentally friendly framework. Energy consumption could
525 be optimized to a large extent to avoid the excess of energy applied and, therefore, to
526 allow the process to operate in a sustainable manner.

527 In this context, despite the potential of photocatalysis for greywater treatment, there
528 are still some key technological issues related to its application that have to be solved,
529 with the high energy demand being the main one. Thus, the development of more
530 energy efficient light sources is being studied. In order to reduce their environmental
531 burdens, the most desirable scenario would be the use of solar light. Nonetheless,
532 further research is needed to overcome some important issues like the development
533 of photocatalysts that are active under visible light, which could help to implement solar
534 photocatalysis for the treatment of greywater.

535 Thus, to achieve a sustainable greywater treatment, future discussion including
536 technical and economic evaluations should be performed in order to complement the
537 LCA study.

538

539

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546

547 **References**

548 [1] Abu-Ghunmi, D., Abu-Ghunmi, L., Kayal, B., Bino, A., 2016. Circular economy
549 and the opportunity cost of not ‘closing the loop’ of water industry: the case of
550 Jordan. *J. Clean. Prod.* 131, 228–236.

551 [2] Atanasova, N., Dalmau, M., Comas, J., Poch, M., Rodriguez-Roda, I., Buttiglieri,
552 G., 2017. Optimized MBR for greywater reuse systems in hotel facilities. *J.*
553 *Environ. Manage.* 193, 503–511.

554 [3] Chai, H., Bao, Y., Lin, H., 2013. Engineering applications on reclaimed water
555 treatment and reuse of hotel’s high grade gray water. *Adv. Mat. Res.* 610–613,
556 2391–2396.

557 [4] Chatzisyneon, E., Foteinis, S., Mantzavinos, D., Tsoutsos, T., 2013. Life cycle
558 assessment of advanced oxidation processes for olive mill wastewater
559 treatment. *J. Clean. Prod.* 54, 229–234.

- 560 [5] Chong, M.N., Jin, B., Chow, C.W.K., Saint, C., 2010. Recent developments in
561 photocatalytic water treatment technology: a review. *Water Res.* 44,
562 2997–3027.
- 563 [6] Corominas, Ll., Foley, J., Guest, J.S., Hospido, A., Larsen, H.F., Morera, S.,
564 Shaw, A., 2013. Life cycle assessment applied to wastewater treatment: state
565 of the art. *Water Res.* 47, 5480–5492.
- 566 [7] Corona, B., Escudero, L., Quéméré, G., Luque-Heredia, I., San Miguel, G.,
567 2017. Energy and environmental life cycle assessment of a high concentration
568 photovoltaic power plant in Morocco. *Int. J. Life Cycle Assess.* 22, 364–373.
- 569 [8] De Feo, G., Ferrara, C., 2017. A procedure for evaluating the most
570 environmentally sound alternative between two on-site small-scale wastewater
571 treatment systems. *J. Clean. Prod.* 164, 124–136.
- 572 [9] De Gisi, S., Casella, P., Notarnicola, M., Farina, R., 2016. Grey water in
573 buildings: a mini-review of guidelines, technologies and case studies. *Civ. Eng.*
574 *Environ. Syst.* 33, 35–54.
- 575 [10] Derypol. <http://www.derypol.com/> (accessed 03.02.2017).
- 576 [11] Dhouib, A., Hdiji, N., Hassaïri, I., Sayadi, S., 2005. Large scale application of
577 membrane bioreactor technology for the treatment and reuse of an anionic
578 surfactant wastewater. *Process Biochem.* 40, 2715–2720.
- 579 [12] Dominguez, S., Rivero, M.J., Gomez, P., Ibañez, R., Ortiz, I., 2016. Kinetic
580 modeling and energy evaluation of sodium dodecylbenzenesulfonate
581 photocatalytic degradation in a new LED reactor. *J. Ind. Eng. Chem.* 37,
582 237–242.
- 583 [13] Dominguez-Ramos, A., Held, M., Aldaco, R., Fischer, M., Irabien, A., 2010.
584 Prospective CO₂ emissions from energy supplying systems: photovoltaic

- 585 systems and conventional grid within Spanish frame conditions. *Int. J. Life Cycle*
586 *Assess.* 15, 557–566.
- 587 [14] EC JCR, 2010. *ILCD Handbook: general guide for life cycle assessment -*
588 *provisions and action steps.* Publications office of the European Union,
589 Luxembourg.
- 590 [15] E-PRTR, 2006. Regulation (EC) No 166/2006 of the European Parliament and
591 of the Council of 18 January 2006 concerning the establishment of a European
592 pollutant release and transfer register and amending council directives
593 91/689/EEC and 96/61/EC.
- 594 [16] European Commission 2017a. https://ec.europa.eu/commission/index_en
595 (accessed 10.02.2017).
- 596 [17] European Commission 2017b. [http://ec.europa.eu/environment/water/reuse-](http://ec.europa.eu/environment/water/reuse-actions.htm)
597 [actions.htm](http://ec.europa.eu/environment/water/reuse-actions.htm) (accessed 10.02.2017).
- 598 [18] Evonik Industries. <http://corporate.evonik.com> (accessed 03.02.2017).
- 599 [19] Fernández-Castro, P., Vallejo, M., San Román, M.F., Ortiz, I., 2015. Insight
600 on the fundamentals of advanced oxidation processes. Role and review of the
601 determination methods of reactive oxygen species. *J. Chem. Technol. Biot.* 90,
602 796–820.
- 603 [20] Filimonau, V., Dickinson, J., Robbins, D., Huijbregts, M.A.J., 2011. Reviewing
604 the carbon footprint analysis of hotels: Life Cycle Energy Analysis (LCEA) as a
605 holistic method for carbon impact appraisal of tourist accommodation. *J. Clean.*
606 *Prod.* 19, 1917–1930.
- 607 [21] Fountoulakis, M.S., Markakis, N., Petousi, I., Manios, T., 2016. Single house
608 on-site grey water treatment using a submerged membrane bioreactor for toilet
609 flushing. *Sci. Total Environ.* 551–552, 706–711.

- 610 [22] Frischknecht, R., Jungbluth, N., Althaus, H.J., Doka, G., Heck, T., Hellweg,
611 S., Hischer, R., Nemecek, T., Rebitzer, G., Spielmann, M., Wernet, G., 2007.
612 Overview and methodology. Ecoinvent Report No 1, Swiss Centre for Life Cycle
613 Inventories, Dübendorf.
- 614 [23] Gabarró, J., Batchellí, L., Balaguer, M.D., Puig, S., Colprim, J., 2013. Grey
615 water treatment at a sports centre for reuse in irrigation: a case study. Environ.
616 Technol. 34, 1385–1392.
- 617 [24] Gander, M., Jefferson, B., Judd, S., 2000. Aerobic MBRs for domestic
618 wastewater treatment: a review with cost considerations. Sep. Purif. Technol.
619 18, 119–130.
- 620 [25] García, V., Margallo, M., Aldaco, R., Urriaga, A., Irabien, A., 2013.
621 Environmental sustainability assessment of an innovative Cr (III) passivation
622 process. ACS Sustainable Chem. Eng. 1, 481–487.
- 623 [26] Garcia-Herrero, I., Laso, J., Margallo, M., Bala, A., Gazulla, C. Fullana-i-
624 Palmer, P., Vázquez-Rowe, I., Irabien, A., Aldaco, R., 2017a. Incorporating
625 linear programming and life cycle thinking into environmental sustainability
626 decision-making: a case study on anchovy canning industry. Clean Techn.
627 Environ. Policy. 19(7), 1897–1912.
- 628 [27] Garcia-Herrero, I., Margallo, M., Onandía, R., Aldaco, R., Irabien, A., 2017b.
629 Environmental challenges of the chlor-alkali production: seeking answers from
630 a life cycle approach. Sci. Total Environ. 580, 147–157.
- 631 [28] Ghunmi, L.A., Zeeman, G., Fayyad, M., Van Lier, J.B., 2011. Grey water
632 treatment systems: a review. Crit. Rev. Env. Sci. Tec. 41, 657–698.

- 633 [29] Giménez, J., Bayarri, B., González, Ó., Malato, S., Peral, J., Esplugas, S.,
634 2015. Advanced oxidation processes at laboratory scale: environmental and
635 economic impacts. *ACS Sustainable Chem. Eng.* 3, 3188–3196.
- 636 [30] Gori, R., Cammilli, L., Petrovic, M., Gonzalez, S., Barceló, D., Lubello, C.,
637 Malpei, F., 2010. Fate of surfactants in membrane bioreactors and conventional
638 activated sludge plants. *Environ. Sci. Technol.* 44, 8223–8229.
- 639 [31] Guo, T., Englehardt, J., Wu, T., 2014. Review of cost versus scale: water and
640 wastewater treatment and reuse processes. *Water Sci. Technol.* 69, 223–234.
- 641 [32] Holloway, R.W., Miller-Robbie, L., Patel, M., Stokes, J.R., Munakata-Marr, J.,
642 Dadakis, J., Cath, T.Y., 2016. Life-cycle assessment of two potable water reuse
643 technologies: MF/RO/UV-AOP treatment and hybrid osmotic membrane
644 bioreactors. *J. Membrane Sci.* 507, 165–178.
- 645 [33] Hospido, A., Moreira, M.T., Martín, M., Rigola, M., Feijoo, G., 2005.
646 Environmental evaluation of different treatment processes for sludge from urban
647 wastewater treatments: anaerobic digestion versus thermal processes. *Int. J.*
648 *LCA* 10, 336–345.
- 649 [34] Hospido, I. Sanchez, G. Rodriguez-Garcia, A. Iglesias, D. Buntner, R. Reif,
650 M.T. Moreira, G. Feijoo, 2012. Are all membrane reactors equal from an
651 environmental point of view?. *Desalination* 285, 263–270.
- 652 [35] Ibadon, A.O., Fitzpatrick, P., 2013. Heterogeneous photocatalysis: recent
653 advances and applications. *Catalysts* 3, 189–218.
- 654 [36] IChemE, 2002. The sustainability metrics: sustainable development progress
655 metrics recommended for use in the process industry. Retrieved February. 20
656 p. 2011.

- 657 [37] Irabien, A., Aldaco, R., Dominguez-Ramos, A., 2009. Environmental
658 sustainability normalization of industrial processes. *Comput. Aided Chem. Eng.*
659 26, 1105–1109.
- 660 [38] ISO, 2006a. ISO 14040: environmental management - Life cycle assessment
661 – Principles and framework.
- 662 [39] ISO, 2006b. ISO 14044: environmental management - Life cycle assessment
663 – Requirements and guidelines.
- 664 [40] Khan, S.J., Hankins, N.P., Shen, L.-C., 2016. Submerged and attached
665 growth membrane bioreactors and forward osmosis membrane bioreactors for
666 wastewater treatment, in: Singh, R., Hankins, N.P. (Eds.), *Emerging membrane
667 technology for sustainable water treatment*, Elsevier B.V, pp. 277–296.
- 668 [41] Kralisch, D., Ott, D., Gericke, D., 2015. Rules and benefits of life cycle
669 assessment in green chemical process and synthesis design: a tutorial review.
670 *Green Chem.* 17, 123–145.
- 671 [42] Kumar, J., Bansal, A., 2013. Photocatalysis by nanoparticles of titanium
672 dioxide for drinking water purification: a conceptual and state-of-art review.
673 *Mater. Sci. Forum* 764, 130–150.
- 674 [43] Liberman, N., Shandalov, S., Forgacs, C., Oron, G., Brenner, A., 2016. Use
675 of MBR to sustain active biomass for treatment of low organic load grey water.
676 *Clean Techn. Environ. Policy* 18, 1219–1224.
- 677 [44] Malato, S., Maldonado, M.I., Fernández-Ibáñez, P., Oller, I., Polo, I., Sánchez-
678 Moreno, R., 2016. Decontamination and disinfection of water by solar
679 photocatalysis: the pilot plants of the Plataforma Solar de Almeria. *Mat. Sci.
680 Semicon. Proc.* 42, 15–23.

- 681 [45] March, J.G., M. Gual, F. Orozco, 2004. Experiences on greywater re-use for
682 toilet flushing in a hotel (Mallorca Island, Spain). *Desalination* 215, 37–43.
- 683 [46] Margallo, M., Aldaco, R., Irabien, A., Carrillo, V., Fischer, M., Bala, A., Fullana,
684 P., 2014b. Life cycle assessment modelling of waste-to-energy incineration in
685 Spain and Portugal. *Waste Manag. Res.* 32, 492–499.
- 686 [47] Margallo, M., Dominguez-Ramos, A., Aldaco, R., Bala, A., Fullana, P., Irabien,
687 A., 2014a. Environmental sustainability assessment in the process industry: a
688 case study of waste-to-energy plants in Spain. *Resour. Conserv. Recy.* 93,
689 144–155.
- 690 [48] Margallo, M., Taddei, M.B.M., Hernández-Pellón, A., Aldaco, R., Irabien, Á.,
691 2015. Environmental sustainability assessment of the management of municipal
692 solid waste incineration residues: a review of the current situation. *Clean Techn.*
693 *Environ. Policy* 17(5), 1333–1353.
- 694 [49] Máša, V, Bobák, P. Kuba, P., Stehlík P., 2013. Analysis of energy efficient and
695 environmentally friendly technologies in professional laundry service. *Clean*
696 *Techn. Environ. Policy* 15, 445–457.
- 697 [50] Merz, C., Scheumann, R., El Hamouri, B., Kraume, M., 2007. Membrane
698 bioreactor technology for the treatment of greywater from a sports and leisure
699 club. *Desalination* 215, 37–43.
- 700 [51] Muñoz, I., 2003. Life cycle assessment as a tool for green chemistry:
701 application to kraft pulp industrial wastewater treatment by different advanced
702 oxidation processes. Universitat Autònoma de Barcelona.
- 703 [52] Muñoz, I., Peral, J., Ayllón, J.A., Malato, S., Passarinho, P., Domènech, X.,
704 2006. Life cycle assessment of a coupled solar photocatalytic–biological
705 process for wastewater treatment. *Water Res.* 40, 3533–3540.

- 706 [53] Muñoz, I., Rieradevall, J., Torrades, F., Peral, J., Domènech, X., 2005.
707 Environmental assessment of different solar driven advanced oxidation
708 processes. *Sol. Energy* 79, 369–375.
- 709 [54] Ortiz, I., Mosquera-Corral, A., Lema, J.M., Esplugas, S., 2015. Advanced
710 technologies for water treatment and reuse. *AIChE J.* 61, 3146–3158.
- 711 [55] Ortiz, M., Raluy, R.G., Serra, L., Uche, J., 2007. Life cycle assessment of
712 water treatment technologies: wastewater and water-reuse in a small town.
713 *Desalination* 204, 121–131.
- 714 [56] PE International, 2016. Gabi 6 Software and Database on Life Cycle
715 Assessment. Leinfelden-Echterdingen, Germany.
- 716 [57] Pretel, R., Robles, A., Ruano, M.V., Seco, A., Ferrer, J., 2016. Economic and
717 environmental sustainability of submerged anaerobic MBR-based (AnMBR-
718 based) technology as compared to aerobic-based technologies for moderate-
719 /high-loaded urban wastewater treatment. *J. Environ. Manage.* 166, 45–54.
- 720 [58] Real Decreto 1620/2007, de 7 de diciembre, por el que se establece el
721 régimen jurídico de la reutilización de las aguas depuradas, BOE no. 294, 2007.
- 722 [59] Rivero, M.J., Parsons, S.A., Jeffrey, P., Pidou, M., Jefferson, B., 2006.
723 Membrane chemical reactor (MCR) combining photocatalysis and microfiltration
724 for grey water treatment. *Water Sci. Technol.* 53, 173–180
- 725 [60] Rodríguez, R., Espada, J.J., Pariente, M.I., Melero, J.A., Martínez, F., Molina,
726 R., 2016. Comparative life cycle assessment (LCA) study of heterogeneous and
727 homogenous Fenton processes for the treatment of pharmaceutical
728 wastewater. *J. Clean. Prod.* 124, 21–29.
- 729 [61] Sanchez, M., Rivero, M.J., Ortiz, I., 2010. Photocatalytic oxidation of grey
730 water over titanium dioxide suspensions. *Desalination* 262, 141–146.

- 731 [62] Sanchez, M., Rivero, M.J., Ortiz, I., 2011. Kinetics of
732 dodecylbenzenesulphonate mineralisation by TiO₂ photocatalysis. *Appl. Catal.*
733 *B-Environ.* 101, 515–521.
- 734 [63] Santasmasas, C., Rovira, M., Clarens, F., Valderrama, C., 2013. Grey water
735 reclamation by decentralized MBR prototype. *Resour. Conserv. Recy.* 72,
736 102–107.
- 737 [64] Serra, A., Brillas, E., Domènech, X., Peral, J., 2011. Treatment of
738 biorecalcitrant α -methylphenylglycine aqueous solutions with a solar photo-
739 Fenton-aerobic biological coupling: biodegradability and environmental impact
740 assessment. *Chem. Eng. J.* 172, 654–664.
- 741 [65] Song, K., Mohseni, M., Taghipour, F., 2016. Application of ultraviolet light-
742 emitting diodes (UV-LEDs) for water disinfection: a review. *Water Res.* 94,
743 341–349.
- 744 [66] Spasiano, D., Marotta, R., Malato, S., Fernandez-Ibañez, P., Di Somma, I.,
745 2015. Solar photocatalysis: materials, reactors, some commercial, and pre-
746 industrialized applications. a comprehensive approach. *Appl. Catal. B-Environ.*
747 170–171, 90–123.
- 748 [67] Suárez-Ojeda, M.E., Kim, J., Carrera, J., Metcalfe, I.S., Font, J., 2007.
749 Catalytic and non-catalytic wet air oxidation of sodium dodecylbenzene
750 sulfonate: kinetics and biodegradability enhancement. *J. Hazard. Mater.* 144,
751 655–662.
- 752 [68] Suh, Y.-J. and Rousseaux, P., 2002. An LCA of alternative wastewater sludge
753 treatment scenarios. *Resour. Conserv. Recy.* 35, 191–200.
- 754 [69] Tsang, M.P., Sonnemann, G.W., Bassani, D.M., 2016. Life-cycle assessment
755 of cradle-to-grave opportunities and environmental impacts of organic
756 photovoltaic solar panels compared to conventional technologies. *Sol. Energ.*
757 *Mat. Sol. C.* 156, 37–48.

- 758 [70] Wankhade, A.V., Gaikwad, G.S., Dhonde, M.G., Khaty, N.T., Thakare, S.R.,
759 2013. Removal of organic pollutant from water by heterogenous photocatalysis:
760 a review. Res. J. Chem. Environ. 17 (1), 84–94.
- 761 [71] Wilcox, J., Nasiri, F., Bell, S., Rahaman, M.S., 2016. Urban water reuse: a
762 triple bottom line assessment framework and review. Sustainable Cities and
763 Society 27, 448–456.
- 764 [72] Zang, Y., Li, Y., Wang, C., Zhang, W., Xiong, W., 2015. Towards more
765 accurate life cycle assessment of biological wastewater treatment plants: a
766 review. J. Clean. Prod. 107, 676–692.