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PREE RADICAL ANNULATION OF CYCLOPENTANE RING

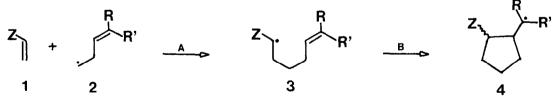
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<u>Abstract</u>: Homoallyl radical, e.g. $\underline{2}$, reacts with an electron-deficient olefinic bond $\underline{1}$ with a new C-C bond forming and arising of a 5-hexenyl radical $\underline{3}$ which further undergoes to $\underline{5-\text{exo-trigonal}}$ cyclization with a cyclopentane ring (4) annulation.

Regioselective intermolecular addition of an alkyl radicals onto the electron-deficient olefinic bond become an useful reaction for carbon-carbon bond formation 1 . This general reaction is independent of the radical precursors and of the Michael acceptors. On the other hand intramolecular cyclization of 5-hexenyl radical to the cyclopentylmethyl radical is well known reaction for the construction of carbocyclic $^{2-7}$, and heterocyclic rings $^{8-10}$. This reaction is stereoelectronically controlled and also proceeds regardless of the alkenyl radical precursors 3,8 . Alkenyl radical cyclization has been systematically investigated and applied in the synthesis of variety cyclic and polycyclic organic molecules 6,7,9 and natural products $^{10-12}$.

We conceived to connect Giese's intermolecular radical conjugated addition (step A, Scheme 1.) and Walling-Beckwith's intramolecular 5-hexenyl radical cyclization (step B) into one sequence of radical chain reactions, in order to close a cyclopentane ring from two unsaturated molecules.



Scheme 1.

We found that annulation of cyclopentane ring can be achieved by $tri-\underline{n}-butyltin$ hydride reduction of homoallyl halides $\underline{5}$ (X = I or Br) in an excess of electron-deficient olefinic compounds 1 (Z = CN, COOEt, COCH₃, CHO) (Scheme 2.).

Annulation of cyclopentane ring by addition of homoallyl radical $\underline{2}$, as a three carbon block, onto the conjugated olefinic bond $\underline{1}$, as a radicophilic two carbon block, represent a two C-C bond forming reaction, i.e. 2+3 stepwise cycloaddition reaction. Cyclopentane ring annulation was observed in the $\boldsymbol{\beta}$ -acetylenic radical addition onto the conjugated olefinic bond $\underline{13}$.

$$Z + R' \qquad Bu_3SnH + Z_3R + Z_3R + Z_4R + Z_5R + Z_$$

Scheme 2.

The cyclization reactions were generally performed by treating of homoall halides 5 (11 mmole) with tri-n-butyltin hydride (12 mmole), ten times excess of acrylonitrile or other electron-deficient olefinic compounds (120 mmole) and AJBN as an initiator in benzene or other solvent with refluxing in an inert atmosphere. Products were isolated $^{1.4}$ and separated by column (SiO $_{\!\!\!2}$) or gas chromatography and characterized by ir, H nmr and mass spectra (Table 1).

Table 1. Annulation of cyclopentane ring. Products and their yields (%)

	Starting compounds				Products ^a		Method	Time
Exp.	1		<u>5</u>		<u>6</u>	<u>7</u>		hrs
	Z	R	R`	X				
i.	CN	Ph	Ph	I	23 ^b	-	С	48
ii.	CN	Ph	ſΡħ	Br	16 (70) ^d	-	e, f	45 (30)
iii.	COOEt	Ph	Ρh	I	25 (39) ^g	-	е	48
iv.	COOEt	Ph	Ρh	Br	(45) ^g	-	е	48
v .	CN	Me	Ph	Ih	52 ¹ (67)	_	c, j	40 (12)
vi.	CN	Me	Me	I	-	44 ^k (65) ^g	С	4
vii.	COOEt	Me	Me	I	-	38 ¹ (75) ^g	f	3
viii.	COCH3	Me	Me	I	_	25 ^m (75) ^g	С	4
ix.	СНО	Me	Me	I	-	60 ^g , m	С	1.5
х.	CN	Н	Н	Br	-	30 (58) ^g	f	5

- a. Isolated yields after column chromatograph.
- b. Mixture of cis- and trans-isomers in ratio of 1 : 2.
- c. $\frac{5}{7}$ (Exp. i.) $\frac{1}{10}$ SnH/1 $\frac{1}{1}$ 1/15 eq., AIBN, C₆H₆, reflux. d. $\frac{5}{7}$ ield calculated on the converted bromide $\frac{5}{5}$ (Exp. ii.).
- e. Toluene was used as a solvent.
- f. <u>5</u> (Exp. ii.) 1 eq./Bu₃SnH 0.25 eq./NaBH_u1.5 eq./<u>1</u> 10 eq./<u>t</u>-BuOH¹².
- g. Gas chromatography yields.
- h. Mixture of Z and E isomers was used.
- i. Mixture containing trans-isomer and two cis-enantiomers was separated by gc, (10% OV-275 on Chromosorb) and has a ratio 2.8:1:157. j. 5 (Exp. v.)/Bu SnH/1 1/1/12 eq., Et 20/h ν /25 c k. Mixture of two somers in ratio of 2:1.

- 1. Ratio of two isomers is 0.7 : 1. m. Ratio of isomers is 3 : 1.

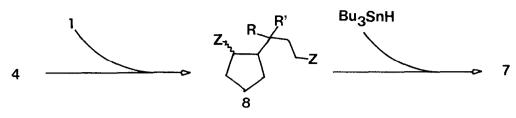
For example, reaction of 4-phenyl-3-penten-1-yl iodide 5 (Exp. v.) with tri-n-butyltin hydride and acrylonitrile, under above conditions, afforded as a major $\frac{1}{2}$ product the mixture of cis- and trans-2-(1-phenylethyl)-cyclopentyl cynide 6 (52%), together with the minor by product 2-phenyl-2-pentene and starting iodide 5.

The first step in this chain cycloaddition reaction is production of a tri- \underline{n} -butyltin radical, and in subsequent reaction with alkenyl halides $\underline{5}$ homoallyl radical $\underline{2}$ is generated. Although the homoallyl radical $\underline{2}$ may be in an equilibrium with cyclopropylmethyl radical $\underline{1}^6$, intermolecular conjugated addition onto the electron-deficient olefinic bond $\underline{1}$ is predominant reaction when olefinic compounds were used in an excess (Scheme 3.). Only minor amount of buten derivatives (e.g. $\underline{5}$, X = H) as a reduction products (up to 8%), were detected. This C-C bond forming reaction generates an adduct radical $\underline{3}$ which is much less nucleophilic because of the presence of nitrile or other electron-withdrawing group. Possessing an appropriate

Scheme 3.

located olefinic bond (e.g. i δ -position) radical $\underline{3}$ rather undergoes to the $\underline{5}$ -exotingonal intramolecular addition to close a cyclopentane ring $\underline{4}$ than to the $\underline{5}$ -endocyclization or hydrogen abstraction from organotin hydride, since cyclohexane derivative or unsaturated nitrile have not been detected in the reaction mixtures. The fate of cyclopentylmethyr radical $\underline{4}$ generated after ring closure reaction, i.e. second C-C bond formation, depends of the substituents on the radical centre and reaction conditions. More stable tertiary radicals $\underline{4}$ (R = R = Ph and R = Ph, R = Me, Expts. i - v) rather undergoe to abstraction of hydrogen from tin hydride thus affording a saturated cyclopentane derivatives $\underline{6}$ than to addition reactions. However, when tertiary radicals are not benzylic, e.g. $\underline{4}$ (R = R = Me, Expts. vi - ix) or is primary radical, e.g. $\underline{4}$ (R = R = H, Exp. x) they rather undergo to the intermolecular addition onto the olefinic compounds $\underline{1}$ thus producing a new $\underline{4}$ -substituted radical $\underline{8}$ (Scheme 4. Table 1.). Termination reaction for such a radical is hydrogen abstraction thus giving a difuctional cyclopentane derivative of type $\underline{7}$ as a final products of the three C-C bond formation proces $\underline{17}$.

The other possibilities of free radical two steps cyclopentane ring annulation are in progres.



Scheme 4.

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