

Provided by Faculty of Chemistry Repository - Cherry

Tetrahedron Letters, Vol. 27, No. 49, pp 5981-5984, 1986 Printed in Great Britain

0040 - 4039 / 86 \$3.00 + .00Pergamon Journals Ltd.

FREE RADICAL CARBOCYCLIC RING RECONSTRUCTION

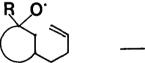
Živorad Čeković* and Radomir Saičić

Department of Chemistry, Faculty of Sciences, University of Belgrade, Studentski trg 16, 11001 Belgrade, Yugoslavia

Alkenyl radical generated by A-fragmentation of tertiary cyclohexyloxy radical with carbocyclic ring opening, possessing a suitably located olefinic double bond, undergoes to the intramolecular 5-exo-trigonal cyclization and a new carbocyclic ring was formed.

Among a variety of reactions **B**-fragmentation is one of the preferential mode of stabilisation of tertiary alkoxy radicals 1. Homopolar nature of **3**-cleavage reaction of alkoxy radicals is well-established and the carbonyl containing fragment and products derived from the alkyl radical fragment were produced $^{1-6}$. The rate of 6-cleavage of alkoxy radicals is independent of the radical precursors but is mainly dependent of the stability of the initialy formed carbon radicals 2,3 . Thereby, intramolecular cyclization of 5-hexenyl radicals to the cyclopentylmethyl radicals are well known reaction for the construction of carbocyclic 7,8 and heterocyclic rings.

However, homopolar scission of C-C bond in the alkoxy radical intermediates, followed by radical recombination reaction with a new C-C bond formation in the same molecule have not been investigated together as a sequence of reactions. This radical recombination reactions could involve one ring opening and the other ring closure with a considerable change of the carbon skeleton (Scheme 1.).







Scheme 1.

We wish to report a new approach to carbocyclic ring reconstruction involving a tertiary alkoxy radical fragmentation and intramolecular addition of arising alkenyl radical onto the suitably located olefinic double bond.

It was found that this free radical fragmentation-cyclization reaction involving a ring reconstruction can be achieved by ferrous ion induced decomposition of 1-glkyl- or 1-aryl-3-(3-alkenyl)-cyclohexyl hydroperoxides la- c^{10} . Two saturated ketones having a cyclopentane ring 20-c and 3a-c were obtained as a chief reaction products in 28-38% and 7-30% yields, respectively, in addition to the unsaturated ketones 4a-c (8-30%) and starting alcohols 5a-c (7-20% yields) (Scheme 2.).

In a typical experiment 2.48 g (13.5 mmole) of alkyl hydroperoxide $\underline{\text{la}}^{11}$ was treated with 3.75 g of powdered crystaline ferrous sulfate in acetic acid as a solvent in inert atmosphere. During 3 hrs hydroperoxide was completly reduced. The reaction mixture was worked up as it was described and products were separated and purified by $\underline{\text{glc}}$ and characterized by ir, nmr and mass spectra $\underline{\text{la}}^{12}$.

Scheme 2.

For example, by decomposition of 1-methyl-3-(3-butenyl)cyclohexyl hydroperoxide $\frac{1}{2}$ by ferrous ion 5-(3-methylcyclopentyl)-pentan-2-one $\frac{2}{2}$ (38%) and methyl (2-methyl 5- $\frac{1}{2}$ -propyl)-cyclopentyl ketone $\frac{3}{2}$ (30%) were obtained as a products of ring reconstruction reactions.

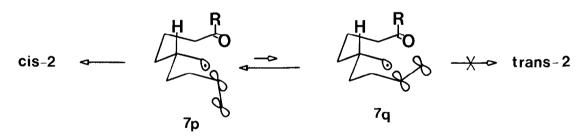
Tertiary 1-alkylcycloalkyloxy radicals, regardless of its precursors, undergo to the β -cleavage reaction and 5-carbonylpentyl radical was formed as an intermediate $^{4-6}$. By possessing an alkenyl group in position 3 of cyclohexane ring the alkoxy radical $\underline{6}$ is not symmetrical and two possibilities for $C_{\mathbf{A}}$ - $C_{\mathbf{A}}$ bond scission exist, thus two fragmented carbon centered radicals $\underline{7}$ and $\underline{8}$ were generated as intermediates (Scheme 3.). Ratio of ketones $\underline{2}$ and $\underline{3}$, derived from fragmented carbon

radicals $\underline{7}$ and $\underline{\delta}$, respectively, indicates that a small long range participation of olefinic double bond on the course of β -cleavage reaction exists. Participation of olefinic bond on the course of β -scission reaction can be explained by involving transition state $\underline{12}$. Although, a little lower yields of ketone $\underline{3}$ in respect to ketone 2 may be due to side reactions of radicals 8, 10 and 11.

The further reaction of these two primary, fragmented, carbon radicals $\underline{7}$ and $\underline{8}$ are considerably different. By possessing an olefinic bond in position 5 radical $\underline{7}$ undergoes almost exclusively to the intramolecular $\underline{5-\text{exo-trigonal}}$ cyclization, thus producing a cyclopentylmethyl radical $\underline{9}$. Termination process for radical $\underline{9}$ is hydrogen abstraction and ketone $\underline{2}$ was formed as a final product of radical recombination reactions. Products of $\underline{5-\text{endo-trigonal}}$ cyclization, coupling dicarbonyl compounds and hydrogen abstraction products, which could be derived from fragmented radical 7, were not observed.

However, fragmented primary carbon radical $\underline{8}$ undergoes to the 1,5-bydrogen abstraction from the methylene group adjacent to carbonyl group and a secondary radical $\underline{10}$ was generated (Scheme 3.). This radical rearrangement is energetically favourable for about 6 kcal/mole¹⁴. The fate of \mathbf{A} -carbonyl radical $\underline{10}$ with an elefinic bond in position 5, as it was expected, it undergoes to the intramolecular cyclization reaction and a new cyclopentylmethyl radical $\underline{11}$ arising which affords a ketone $\underline{3}$ as a final product $\underline{15}$.

Saturated ketone $\underline{2}$ was obtained as a pure $\underline{\text{cis}}$ -isomer, what is in agreement with greater stability of $\underline{\text{cis}}$ -1,3-dialkylcyclopentane derivatives 16 , as well as more favourable conformation of butenyl group in carbon radical $\underline{7p}$ (Scheme 4.), with quasi-endo-orientation of olefinic bond leading to $\underline{\text{cis}}$ -isomer $\underline{2}$. While the



Scheme 4.

conformation $\underline{7q}$ which is requested for \underline{trans} -isomer formation with quasi- \underline{exo} -orientation of alkenyl group is less favourable.

Investigation of other type of alkoxy radical precursors and other type of rings reconstructions are in progress.

References and Notes

- P. Gray and A. Williams, Chem. Rev., 59, 239 (1959); C. Walling and A. Padwa, J. Am. Chem. Soc., 83, 1593 (1963); C. Walling and B. B. Jacknow, J. Am. Chem. Soc., 82, 6108 (1960); F. D. Greene, M. L. Savitz, F. D. Osterholtz, H. H. Lau, W. N. Smith and P. M. Zanet, J. Org. Chem., 28, 55 (1963).
- 2. J. K. Kochi in Free Radicals, Ed. by J. K. Kochi, Vol. 2, Ch. 23, pp 683, Wiley Intersciences, New York, 1973; J. D. Bacha and J. K. Kochi, <u>J. Org. Chem.</u>, <u>30</u>, 3272 (1965).

- 3. M. Lj. Mihailović and Ž. Čeković, Synthesis, 209 (1970); M. Lj. Mihailović, Ž. Čeković, Lj. Lorenç, in Organic Synthesis by Oxidation with Metal Compounds, Ed. by W. J. Mijs and C. R. H. I. de Jonge, Ch. 14, pp 772, Plenum Press, New York, 1986; K. Heusler and J. Kalvoda, Angew. Chem., Internat. Ed., 3, 525 (1964); Ž. Čeković and G. Djokić, Tetrahedron, 37, 4263 (1981).
- G. Sosnovsky and D. J. Rawlinson, in <u>Organic Peroxides</u>, Ed. by D. Swern, Vol. 2
 Ch. 2, Wiley Intersciences, New York, 1971; D. H. Hey, C. J. M. Stirling and
 G. H. Williams, <u>J. Chem. Soc.</u>, 1054 (1957); S. Choe and S. Tsutsumi, <u>Nippon</u>
 Kagaku Zasshi, 82, 52 (1961), Chem. Abstr. 56, 14096 (1962).
- P. Kabasakalian and E. R. Townley, <u>J. Org. Chem.</u>, <u>27</u>, 2918 (1962); F. Minisci,
 M. Cecera, R. Galli and R. Bernardi, Tetrahedron, 25, 2667 (1969).
- 6. G. I. Nikishin, N. I. Papustina, S. S. Spektor and E. I. Kaplan, <u>Izv. Akad. Nauk</u> SSSR, Ser. Him., 1548, 2383 (1979).
- 7. C. Walling, J. H. Cooley, A. R. Ponars and E. J. Racah, <u>J. Am. Chem. Soc.</u>, <u>88</u>, 5361 (1966); A. L. J. Beckwith and K. U. Ingold, <u>Free Radical Rearrangement</u>, in <u>Rearrangement in Ground and Excited States</u>, Vol. 1. Academic Press Inc., New York 1980; A. L. J. Beckwith, <u>Tetrahedron</u>, <u>37</u>, 3073 (1981); A. L. J. Beckwith and C. H. Schiesser, Tetrahedron, 41, 3925 (1985).
- 8. J. -M. Surzur, in Reactive Intermediates, Ed. by A. A. Abramovich, Vol. 2. Ch. 3
 Plenum Press, New York, 1981; S. -K. Chung and M. Newcomb, J. Am. Chem. Soc., 108
 240 (1986); E. J. Corey and S. G. Pyne, Tetrahedron Letters, 24, 2821 (1983).
- 9. J. Wilt, <u>Tetrahedron</u>, <u>41</u>, 3979 (1985); Y. Ueno, M. Chino and M. Okawa, <u>Tetrahedron</u> Letters, 23, 2575 (1982); J. -K. Choi and D. J. Hart, Tetrahedron, 41, 3959 (1989)
- 10. Prepared by reaction of alcohols <u>5a-c</u> with concentrated hydrogen peroxide (80%) and catalytic amount of sulfuric acid. Purity of alkyl hydroperoxides were from 88-95%. N. A. Milas and L. H. Perry, <u>J. Am. Chem. Soc.</u>, <u>68</u>, 1938 (1946); R. Criegee and H. Dietrich, <u>Ann.</u>, <u>560</u>, 135 (1948).
- 11. Mixtures of isomeric alkylhydroperoxides were used.
- 12. Ž. Čeković, Lj. Dimitrijević, G. Djokić and T. Srnić, <u>Tetrahedron</u>, <u>35</u>, 2021 (1979) Ž. Čeković and Lj. Dimitrijević, Glasnik Hem. <u>društva Beograd</u>, <u>46</u>, 79 (1981).
- 13. Yields were calculated in respect to the 88-95% purity of starting alkyl hydroperoxides la-c.
- 14. K. D. King, D. M. Golden and S. W. Benson, <u>J. Am. Chem. Soc.</u>, <u>92</u>, 5541 (1970);
 J. Grzechowiak, J. A. Kerr and H. F. Trotman-Dickenson, <u>J. Chem. Soc.</u>, 5080 (1969)
 R. K. Solly, D. M. Golden and S. W. Benson, <u>Int. J. Chem. Kinet.</u>, <u>2</u>, 11 (1970).
- 15. Intramolecular additions of carbon radicals bearing an electron-withdrawing group on the adjacent carbon atom are described. Ž. Čeković and R. Saičić, submited for publication in Tetrahedron Letters and references therein.
- 16. J. N. Haresnape, Chem. & Ind., 1091 (1953).

(Received in UK 10 October 1986)