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Infuence of surface ocean density OPEN on planktonic foraminifera calcifcation

 \textsf{S} tergios D. Zarkogiannis \textsf{O}^1 , Assimina Antonarakou¹, Aradhna Tripati^{2,3}, George Kontakiotis \textsf{O}^1 , **P.Graham Mortyn4,5, HaraDrinia1 & MervynGreaves [6](http://orcid.org/0000-0001-8014-8627)**

This study provides evidence that ambient seawater density infuences calcifcation and may account for the observed planktonic foraminifera shell mass increase during glacial times. Volumes of weighed fossil *Globigerina bulloides* **shells were accurately determined using X-ray Computer Tomography and were combined with water density reconstructions from Mg/Ca and** *δ***18O measurements to estimate the buoyancy force exerted on each shell. After assessment of dissolution efects, the resulting relationship between shell mass and buoyancy suggests that heavier shells would need to be precipitated in glacial climates in order for these organisms to remain at their optimum living depth, and counterbalance the increased buoyant force of a denser, glacial ocean. Furthermore, the reanalysis of bibliographic data allowed the determination of a relationship between** *G. bulloides* **shell mass and ocean density, which** introduces implications of a negative feedback mechanism for the uptake of atmospheric CO₂ by the **oceans.**

Planktonic foraminifera adjust to the dynamic behaviour of the fluid in which they are immersed^{[1](#page-6-0)}. As with other zooplankton, the precipitation of relatively heavy calcitic tests, with a specifc gravity signifcantly greater than that of the ambient seawater, provides foraminifera a mechanism to counteract uplifing due to lighter cellular components and allows them to inhabit certain depths, which represent favourable ecological niches, by regulating their buoyancy at the expense of biochemical energy². But little is known about how this behaviour may have varied through time as the density of seawater has evolved in response to changes in the temperature and salinity structure of the oceans. Therefore we explore the novel hypothesis that downcore shell mass variations may in part refect a hydrostatic response to seawater density changes.

A number of authors have observed that planktonic foraminifera shell mass was higher during glacial stages and lower during interglacials^{3-[7](#page-6-3)}. The cause of this behaviour is debated. Since it was first shown in culture experiments that an increase in shell mass was induced by an increase in carbonate ion concentration⁸, a link between calcification efficiency and carbonate ion concentration ($[CO_3$ ⁼]) has been found in the physical environment^{[9](#page-7-1)}. A number of authors have shown that glacial shell mass of some planktonic foraminifera species from diferent regions, is correlated with proxy records for carbonate ion concentration $[CO_3^{-1}]$, which increases in seawater due to the decrease of glacial atmospheric $pCO_2^{3,4,7,10}$ $pCO_2^{3,4,7,10}$ $pCO_2^{3,4,7,10}$ $pCO_2^{3,4,7,10}$ $pCO_2^{3,4,7,10}$ $pCO_2^{3,4,7,10}$ $pCO_2^{3,4,7,10}$ $pCO_2^{3,4,7,10}$. This reasoning thus has been used to reconstruct past distribu-tions of carbonate ion concentration from shell mass^{[5,](#page-6-5)[11,](#page-7-3)12}.

Additional work has suggested that downcore variations in shell weight may not always be explained by $[CO₃$ ⁼]. For example, a causal relation between shell mass variations and $[CO₃$ ⁼] could not be verified for a number of down-core^{6,[13](#page-7-5)[,14](#page-7-6)} and contemporary biogenic material^{15-[18](#page-7-8)} investigations from different localities. Some authors have suggested that such attempts should be considered with caution⁶. Calcification temperature has been suggested to not always play a major role on shell mass¹⁰.

¹National & Kapodistrian University of Athens, School of Earth Sciences, Faculty of Geology & Geoenvironment, Department of Historical Geology - Paleontology, Athens, Greece. ²Department of Earth, Planetary, and Space Sciences, Department of Atmospheric and Oceanic Sciences, Institute of the Environment and Sustainability, California Nanosystems Institute, University of California, Los Angeles, CA, USA. ³European Institute of Marine Sciences (IUEM) Université de Brest, UMR 6538, Domaines Océaniques, Rue Dumont D'Urville, and IFREMER, Plouzané, France. ⁴Institute of Environmental Science and Technology, Universitat Autònoma de Barcelona, Barcelona, Spain. ⁵Department of Geography, Universitat Autònoma de Barcelona, Barcelona, Spain. ⁶Department of Earth Sciences, University of Cambridge, Cambridge, UK. Correspondence and requests for materials should be addressed to S.D.Z. (email: stergiosz@geol.uoa.gr)

Figure 1. 1 million years of a shell preservation index for *Globigerina bulloides* from ODP Site 982 compared to Mg/Ca ratios. Bulloides Dissolution indeX (BDX') is solid line and circles represent Mg/Ca measurements. Increased BDX' values signify greater dissolution, with significant dissolution starting at values above 3. The results show no evidence for high dissolution, and also no systematic decrease in Mg/Ca values is observed with reduced preservation. Therefore any low-temperature bias to paleotemperature estimates from Mg/Ca due to preferential post-depositional dissolution of the Mg-rich portions of the foraminiferal calcite is likely to be minimal. The 1σ confidence interval for each plot is shown as solid grey area and bars, respectively.

Even for modern shells of the same species, in different sites almost identical in water $[CO_3$ ⁼] and temperature, an offset in shell mass has been observed¹⁹. Although the influence of nutrients has been hypothesized²⁰, yet another study^{[15](#page-7-7)} excluded nutrient availability and instead called upon optimum growth conditions as the cause for the observed shell mass variations, although subsequently size-normalized weight was found not to respond to optimum growth conditions¹⁷. Finally it has been suggested that changes in size-normalized shell weight of planktonic foraminifera refect mainly abiotic forcing since heavier shells in the Mediterranean were associated with high salinity waters²¹. Dissolution as the major parameter affecting recorded shell mass variation was excluded qualitatively in most of the studies above, while it was semi-quantitatively assessed only for a set of core-top samples 22 .

Since planktonic foraminifera are major contributors of calcium carbonate (CaCO₃) to the sea floor²³, the identifcation of the processes that afect their biomineralization is relevant to testing hypotheses on the origin of glacial-interglacial pCO_2 variations^{[24](#page-7-15)} and to predicting the future of marine calcifying organisms and ecosystems¹⁵. Here, the state of shell preservation is evaluated by semi-quantitative means, and the role of ocean density variations on shell mass and the subsequent change in the intensity of ocean buoyancy force during glacial-interglacial intervals are examined.

At North Atlantic ODP Site 982 we measured Mg/Ca to reconstruct temperature. Furthermore we measured the volumes of weighed *G. bulloides* shells at diferent intervals and calculated the buoyancy force exerted to them by the ocean. Reconstructed temperature values were combined with available oxygen isotope data to calculate seawater salinities. The proxy derived salinity and temperature estimates were used in the equation of state of seawater to derive past surface ocean density changes. We also evaluate shell weight and density data from published geochemical datasets for a number of Atlantic and Southwest Pacifc sites.

Results

Tests of *G. bulloides* from ODP Site 982 are very well preserved and exhibit evidence for only a minor increase in dissolution during glacial times (Fig. [1](#page-1-0)), in accordance with previous studies on dissolution impacts on planktic foraminifera from the Atlantic[25.](#page-7-16) Preservation events are recorded during stadials. No signifcant dissolution was inferred for the Mid-Brunhes interval (∼200–600 ka). In particular the assessment of the X-ray imprint of a number of specimens (Table [1\)](#page-2-0) suggests that calcite loss for most may be well below 15%.

The buoyancy force exerted on *G. bulloides* by surrounding seawater was estimated as a function of water density. Average buoyancy values calculated for diferent time intervals were plotted against the mass of contemporary shells and the results are shown in Fig. [2.](#page-2-1) Calculations of the weight of seawater displaced by the organism were performed with a constant gravitational acceleration $(9.82 \,\mathrm{m/s^2})$. We found that foraminifera shell mass correlates well ($R^2 = 0.58$, $p < 0.005$, $n = 10$) with the buoyant force exerted on them by the surrounding seawater suggesting that their shell mass increases with an increase in the force of buoyancy. The heavier shells are the most voluminous ones (Table [1](#page-2-0); $R^2 = 0.43$, $p < 0.05$, n = 10).

There is evidence of a relationship between *G. bulloides* shell mass and ocean density from the reanalysis of data from core NEAP 8K⁴ which we used to calculate paleodensities. Paleodensity estimates were plotted against *G. bulloides* shell normalized mass, from the 300–355μm size fraction, and the results are shown in Fig. [3.](#page-2-2) Te measurements correlate significantly ($R^2 = 0.56$, $p < 0.001$, $n = 79$) supporting the hypothesis that heavier shells precipitated in denser waters. The correlation between size normalized weight to salinity was $R^2 = 0.52$ ($p < 0.001$, $n=79$) and to temperature was $R^2=0.07$ ($p>0.1$, $n=79$), while temperature and salinity in this case were not correlated to each other $(p > 0.05, n = 79)$. The significant correlation of shell mass with seawater salinity, in contrast to temperature, supports previous findings²¹. This result implies that the salt concentration of ambient seawater greatly afects plankton shell mass, which could be due to the impact of salinity on density.

Table 1. Data summary of specimens used for buoyancy calculations. Cold climatic times (glacials, stadials) are shown in bold. MCD=Meters Composite Depth of ODP Site 982 samples, MIS=Marine Isotope Stage and $XDX = X-ray$ Dissolution Index³⁰.

Figure 2. Evidence for a signifcant correlation between the shell mass of *G. bulloides* from Site 982 and buoyancy force estimates derived from X-ray microtomographic volumetry and combined Mg/Ca - *δ*18O measurements of the same samples. Grey dots are values from interglacial or interstadial samples and black dots represent glacial or stadial samples (Table [1\)](#page-2-0). Te dashed lines around the regression line indicate the 95% level of confdence.

Figure 3. Evidence for a signifcant correlation between the mass of *G. bulloides* shells from the Eastern Atlantic and seawater paleodensity within two different size classes. The upper right (in blue) data are based on sizenormalized weight values for the 300-355 μm size fraction from core NEAP 8K in the Northeast Atlantic. The lower lef data (in red) are values from the 250–315 μm sieve fraction for cores MD02‐2594 and MD96‐2080 in the Southeast Atlantic. Ambient seawater paleodensity values are estimated from Mg/Ca and carbonate *δ*18O. Core top samples are denoted with bigger, black inflled symbols, while vertical lines represent modern *in-situ* annual average seawater density. The dashed lines around the regression line indicate the 95% level of confdence.

Figure 4. Synthesis of published shell mass data plotted against new estimates of seawater density. Data are for *G. bulloides* shells (300–355μm) and Mg/Ca - *δ*18O derived *in-situ* densities from N. Atlantic and S.W. Pacifc sites ($p < 0.001$, $n = 312$). The dashed lines around the regression line indicate the 95% level of confidence. Data are grouped based on marine isotopic stage (MIS). The correlation between size normalized weight to salinity or temperature alone was $R^2 = 0.44$ ($p < 0.001$, n = 315) and $R^2 = 0.03$ ($p > 0.1$, n = 315) respectively, while temperature and salinity were uncorrelated $(p > 0.05)$.

Core NEAP 8K covers one glacial-interglacial cycle and the reconstructed seawater density values range from 26.4 to 29.2 Kg/m³. These values fall within the range of Atlantic intermediate water density reconstructions from benthic foraminifera across the last deglaciation^{26[,27](#page-7-19)}. The data from core NEAP 8K were plotted together with data of diferent size fractions from South Atlantic MD02‐2594 and MD96‐2080 core records and the results are shown in Fig. [3.](#page-2-2)

These data show that smaller *G. bulloides* shells, from the 250-315 μm size fraction, record lower seawater densities than shells from a larger size class that are heavier. Tis can be attributed to calcifcation at diferent depths during diferent ontogenetic life stages of the foraminifera, as has previously been proposed to explain shells of diferent sizes[28](#page-7-20). *G. bulloides* in its early life stages calcifes in shallower, less dense waters while during later stages of growth by chamber addition sinks into deeper and denser waters, suggesting as well, that larger and hence heavier shells must be precipitated in denser waters. The density values from the South Atlantic cores range from 24.3 to 26.1 Kg/m³ and are consistent with previous surface Atlantic water density estimates from planktonic foraminifera across the last deglaciation^{[29](#page-7-21)}.

To further test our hypothesis, we plotted all available published data (see methods) for *G. bulloides* from a restricted size class (Fig. [4\)](#page-3-0). Despite inconsistencies in data acquisition among diferent data sets, a signifcant relationship between shell mass and ambient paleo-seawater density $(R^2 = 0.51, p < 0.001, n = 312)$ is verified both at Atlantic and Pacifc sites for the past ~435 ky. Tus signifcant correlation between water density and shell masses of foraminiferal tests is observed for diferent size fractions, diferent localities and time spans.

All together, the volumes of individual *G. bulloides* tests were measured and the buoyancy force exerted on these organisms by the ocean, under different climatic regimes, has been calculated. The measurements were performed with the use of a X-ray microcomputed tomography scanner (XMCT). The accuracy of the instrument allows the assessment of shell volume variations under diferent past oceanographic conditions.

The preservation state of tests was assessed by examining their ultrastructures under the SEM using the Bulloides Dissolution Index (BDX^3) , and in some cases internal structures with XMCT using the X-ray Dissolution Index (XDX)³⁰, and they were found to be mostly well preserved. Dissolution is not believed to have altered the overall shell volume as any severe dissolution will initially affect the inner calcite layers 31 . In general the mass of the shells is inversely related with their preservation state. The shells were found to be relatively corroded during glacials, when their masses were found to be greater. The fact that the heavier shells were also the most corroded ones, together with the low BDX' and XDX values, rule out dissolution as the primary controlling factor of the observed shell mass pattern.

For the paleotemperature reconstructions derived from Mg/Ca, the well preserved foraminiferal calcite and the insignificant change of late Pleistocene seawater Mg/Ca concentrations^{[32](#page-7-23)} together with the nearly constant state of calcite saturation state of the ocean since the late Pliocene^{[33](#page-7-24)} significantly rule out processes that have affected skeletal Mg^{2+} record other than biomineralization temperature.

Discussion

A relationship that equates the biogenically precipitated shell calcite, at diferent ontogenetic stages, with ambient seawater density was identifed in Fig. [3](#page-2-2). *G. bulloides* in its early life stages calcifes in shallower, lighter waters and by chamber addition during growth sinks into deeper and denser waters. Because the density and viscosity of sea water are on the order of 10^3 and 10^2 times greater than the density and viscosity of air, respectively, foraminifera are much more strongly infuenced by buoyant and viscous forces than are comparably sized terrestrial organisms 34 .

The ability of planktonic foraminifera to inhabit specific depths in the surface ocean requires a means for species-specifc buoyancy adjustments and calcifcation control. Test formation is not an inert and solid state but rather a dissolution and reprecipitation process^{[35](#page-7-26)} that allows foraminifera to resorb and reallocate shell calcite during their life cycle^{36[,37](#page-7-28)}. The calcite shells provide foraminifera with the negative buoyancy needed to dive to certain depths. The substantial weight of the tests require some mechanism to sustain flotation and permit the foraminifera to adjust their position in the water column. The fibrillar bodies appear to be the most likely cyto-plasmic structures mediating buoyancy^{[38](#page-7-29)}. In addition to the fibrillar bodies, it is likely that buoyancy is enhanced by the presence of lipid droplets dispersed throughout the cytoplasm^{[39](#page-7-30)}.

The behavior of foraminifera to regulate shell mass can also be found in their ability to change test porosity. Porosity affects the density of the test wall since it increases shell buoyancy in waters of low viscosity^{[40](#page-7-31)}. Different pore densities (number of pores/surface area) have been found between glacial and interglacial climates⁴¹. Field and laboratory work conducted on planktonic foraminifera indicate that temperature and salinity afect shell porosity. Warmer and less saline waters both produce lower density seawater where foraminifera construct less dense (more porous) shells as a buoyancy response^{[42](#page-7-33)[–44](#page-7-34)} or for metabolic reasons⁴⁵. Regardless of the actual phys-iological cause, porosity has be linked to relative seawater densities^{[46](#page-8-0)}.

For minor, temporary or short term migrations shallower into the water column, positive buoyancy in plankton may be regulated by alterations in the calcite to protoplasm ratio of the shell⁴⁷ or by the lower-density cytoplasmatic entities that can greatly affect their average living depth⁴⁸. The depth regulating mechanism proposed here may support major migrations over wide density ranges of the water column for the acquisition of a maximum depth during the foraminiferal life cycle, and especially during gametogenesis when additional calcite is secreted. This mechanism is probably present but likely less effective in phytoplankton and other smaller organisms where frictional drag and other surface area dependent forces predominate².

Provided that all other growth controlling factors such as $[CO_3^=]$, nutrient availability, temperature, salinity and others are sufficient for optimum growth, planktonic organisms will tend to attain a certain geometry rather than continuously calcifying passively as a function of resource availability. Their shell cannot get unlimitedly thick otherwise its pores would become too long such that gases could no longer difuse through them. In addition, if tests become too heavy, they might sink in the water column, beyond a critical pressure surface and habitat depth range. The influence of density in depth habitat determination has been manifested elsewhere for most planktonic foraminifera⁴⁸ since it was found that their calcification depths are tuned to particular density layers⁴⁹ or isopycnals⁵⁰.

Optimum habitat depth may be determined by carbonate ion concentration, nutrient availability, hydrography, and/or competition. It is possible to speculate in evolutionary terms, that planktonic calcifers may have had the advantage of waters with increased $[CO_3$ ⁼] at the times they needed them most. North Atlantic surface waters are, at present, supersaturated with respect to calcite⁵¹, as most surface ocean waters are⁵². In a glacial ocean, with increased carbonate ion concentration in the surroundings⁵³, organisms may need to spend less energy to gather CO_3^- for shell construction and can get heavier more quickly.

The data discussed above suggest that *G. bulloides* shell mass variations during glacial and interglacial times is related to the buoyant force exerted on organisms by surrounding seawater. During glacial times mean ocean density increases due to thermal contraction and storage of fresh water in ice sheets. Sea ice formation afects the salinity of the ocean, while continental ice deposition alters both ocean's salinity and volume. A denser ocean, with increased salt concentration, will exert greater buoyancy force and thus will tend to buoy plankton towards the sea surface. In order for organisms to counteract this increased buoyant force and maintain hydrostatic equilibrium, they would need to increase shell mass, which in turn would allow them to sink back to their preferred habitat as dictated by hydrography, competition, nutrient availability, and other factors.

Since foraminifera are forced to increase carbonate precipitation during glacial times, then the excess carbonate precipitation alone will decrease the ocean's total alkalinity or pH and will increase its $CO₂$ concentration $({\rm [CO₂)}^{54}$ $({\rm [CO₂)}^{54}$ $({\rm [CO₂)}^{54}$. On the contrary during terminations, when waters are fresher and lower in density, organisms would need to have a lower shell mass than their glacial form in order to maintain their optimal habitat depth within the water column. Such abiotically-driven reduction in planktonic calcifcation during terminations will decrease the [CO₂] in seawater, as a by-product of intracellular calcite formation and it may thus provide a mechanism for the ocean to counterbalance increased atmospheric *pCO*₂.

Conclusions

The analysis of well-preserved fossil *G. bulloides* shells from ODP Site 982 suggests a relationship between shell mass and ambient seawater density. We propose the hypothesis that this is driven by the need for buoyancy regulation. Changes in the density of waters in the upper water column, as a function of both temperature and salinity may explain some of the observed variation in planktonic foraminifera shell mass during glacial-interglacial transitions. Tis depth-regulating mechanism adds a new dimension to the debate over what causes downcore shell mass variations. Building on the carbonate ion hypothesis, we speculate these organisms would then have the advantage of heavier shells in glacial waters with higher $[CO_3^{-}]$. During terminations, abiotically-driven reduction in shell calcification promotes the mechanism of atmospheric $CO₂$ uptake by the ocean. Additional work will elucidate whether there are species-specifc shell mass-seawater density equations.

Methods

The present study was conducted on G. bulloides specimens from ODP Site 982 that were previously studied^{[3](#page-6-2)} and represent both cold and warm climatic phases. *G. bulloides* was investigated because it was present in all samples from Site 982, and due to the amount of published data already available for the species. Preservation of specimens was assessed via Scanning Electron Microscope (SEM) with the use of a semi-quantitative method, while the total volume of the shells during diferent time periods was determined with X-ray microcomputed tomography (XMCT) scannings. The determination of foraminifera test volume together with the ambient seawater density, reconstructed from bibliographic data, allowed, according to Archimedes' principle, the calculation of the buoyancy force of the ocean on the foraminifera shells. The estimation of buoyancy force involved ambient paleo-seawater density calculations, which were based on published *δ*18O data and new Mg/Ca measurements.

To further test the hypothesis, we examined all available published core datasets that include *G. bulloides* shell Mg/Ca, *δ*18O and mass measurements from diferent Atlantic and Pacifc regions. Data were combined to reconstruct ambient seawater densities, and reanalysed to produce graphs of seawater density versus shell mass (Fig. [4](#page-3-0)). Two-tailed regression analyses were performed using the Reduced Major Axis model, at n-1 degrees of freedom. *n* represents sample size. Standard Bonferroni corrections to signifcance levels were applied to correct for multiple comparisons.

Dissolution assessment. The dissolution degree of the ODP 982 specimens that were used in buoyancy calculations (Table [1](#page-2-0)), for which XMCT scans exist, was assessed by applying the XDX, an empirical dissolution index that evaluates the appearance of dissolution features in the tomographs^{[30](#page-7-17)}. The XMCT appearance of their internal structure resembles that of almost intact specimens implying in most of the cases less than 15% of calcite loss³¹. In the absence of XMCT scans for the rest of the record, the carbonate preservation state of samples from ODP Site 982 for the frst 1My was examined, with higher resolution during the late Quaternary, by applying the Bulloides Dissolution Index (BDX')⁵⁵. In order for the evaluation to be representative of each time slice and the result to be statistically signifcant, 20 of the previously weighed *G. bulloides* specimens from each downcore sample were assessed. The spiral side of the ultimate chamber of each test was investigated using a ZEISS DSM 940 A Scanning Electron Microscope at the Department of Geosciences, University of Bremen. The BDX' evaluates the corrosion of shell surface ultrastructure and thus provides a semi-quantitative measure of specimen dissolution^{[55,](#page-8-9)56}. These methods have the advantage of directly assessing the dissolution of the specimens rather than their general fragmentation, which is also afected by fragmentation through mechanical sieving or selective dissolution of the finer fragments. The results from the samples where overlapping in the dissolution assessment methods took place suggest that both methods are in good agreement.

Determination of past seawater density. The combination of Mg/Ca ratios of foraminiferal shell calcite with *δ*¹⁸O isotope measurements has been used here to reconstruct ambient water salinities^{[57–](#page-8-11)[59](#page-8-12)}. Mg/Ca thermometry for core ODP Site 982 was calculated using Anand *et al*. [60](#page-8-13) equation, which is calibrated for temperatures lower than about 10° C and thus more reliable for reconstructing glacial N. Atlantic temperatures. The use of the same equation is mentioned for the rest of published cores present here except from the temperate latitude cores MD02‐2594 and MD96‐2080 where temperature reconstructions were based on the Mashiotta *et al*. [61](#page-8-14) equation because it yields more accurate results for core-top samples at warmer temperatures.

*G. bulloides δ*¹⁸O values were adjusted for a 'vital effect' using the published offset of 0.52‰^{[62](#page-8-15)}. *δ*¹⁸O_{sw} was calculated using a published paleotemperature equation⁶³ and a VPDB-to-SMOW δ^{18} O conversion of 0.27‰⁶⁴. Salinity estimates were derived from $\delta^{18}O_{sw}$ values using the modern regional salinity- $\delta^{18}O$ relationships^{[65](#page-8-18)}, after correcting for global ocean salinity and ice volume changes $66,67$ $66,67$ $66,67$, assuming that these relationships remain constant throughout time⁶⁸.

Error propagation in the conversion of Mg/Ca to temperature. The data used in this study are mainly published data, from diferent laboratories, analysed afer diferent cleaning procedures and precise replicate errors were not published in most cases. For samples from sites other than ODP Site 982 the replicate error (*σ*_{Mg/Ca}=0.7 °C) and the error of the calibration curve (σ _{calib} = 1.1 °C) gives a 1*σ* uncertainty in the temperature estimate of each sample of \pm 1.2 °C. NEAP 8K is quoted with a typical paleotemperature estimate error of \pm 1 °C and for cores ODP 1123, CHAT 1K, CHAT 16K the estimated error is on average ± 1.3 °C. For ODP core 982 the replicate error based on two splits of eight samples is $\sigma_{\text{Mg/Ca}}$ = 0.08 mmol/mol.

In-situ seawater densities for particular depths were calculated using the equation of state of seawater⁶⁹ from the temperature and salinity estimates. The Mg/Ca ratios from the present study were combined with published *δ*18O isotope measurements[70](#page-8-23) to calculate densities at ODP Site 982 for diferent time slices. Seawater densities were calculated from available Mg/Ca and *δ*18O measurements on foraminifera tests from the 300–355 μm size fraction for core NEAP 8K in the North Atlantic^{[4](#page-6-4)}, ODP 1123, CHAT 1K, CHAT 16K in the Southwest Pacific^{[71](#page-8-24)} and from the 250–315 μm size fraction from combined nearby MD02‐2594 and MD96‐2080 core records in the South Atlanti[c72](#page-8-25). Together cores MD02‐2594 and MD96‐2080 yielded data for ∼200 kyrs, while the NEAP 8K core covers the last deglacial cycle⁷³. For the S. Atlantic shells of the 250–315 μ m sieve fraction densities were calculated frst for a depth of 75m, to avoid forcing of the regression with *a priori* lower densities when shallower water depths were used in the equation of state of seawater. Subsequently the regression was recalculated for a more realistic living depth of 25 $m⁷⁴$ $m⁷⁴$ $m⁷⁴$ for the smaller foraminifera, which gave a slightly better correlation and is presented in Fig. [3](#page-2-2).

Propagation of errors in the calculation of in-situ density. The error in the δ^{18} O of seawater ($\sigma_{\delta w}$) is a combination of the error in the measurement of δ^{18} O of *G. bulloides* ($\sigma_{\delta c}$ = 0.081‰) and the error in the Mg/Ca-derived pelagic temperature (σ_T =0.192‰): $\sigma_{\delta w} = \sqrt{\sigma_T^2+\sigma_{\delta c}^2}$, consequently $\sigma_{\delta w}$ =0.208‰. The total error in the ice-volume– corrected δ^{18} O of seawater is $\sigma_{\delta w\text{-ice}} = \sqrt{\sigma_{\delta w}^2 + \sigma_{SL}^2} = 0.231\%$. By assuming a linear relationship between $\sigma_{\delta w\text{-ice}}$ and salinity⁶⁵, the partial differential equations of ref.²⁷ yields an error for salinity σ_S = 0.63 psu and for *in-situ* density $\sigma_t = 1.73$ Kg/m³.

Determination of shell volume and buoyancy force. In order to determine the foraminifera shell volume, X-ray computed tomography (CT) was used. The scannings were performed with a Skyscan 1072 desktop XMCT scanner at the Department of Earth Sciences, University of Cambridge. The scanner uses a point Xray source to create a series of radiographs of a sample as it rotates. Cross‐sectional slices ("tomographs") were reconstructed using Skyscan's own software that uses the Feldkamp cone-beam algorithm⁷⁵. The reconstructed

tomographs were subsequently processed with CT data visualization sofware for entire shell volume determination (*Vforam*). In each tomograph the interior of the foraminifera imprint, together with the calcite imprint itself were segmented as a single object. By summing up all cross sections the data visualization sofware was able to calculate the overall volume (i.e. calcite and cavities) of each shell (Fig. [5\)](#page-6-7).

The available CT data set consisted of scans for foraminifera from 12 different down-core time intervals representing (approximately equal) glacial and interglacial stages of the last three climatic cycles. All samples were scanned under the same conditions: anode voltage was set at 100 kV and the X-ray tube current at 0.98 μA. By processing 1024 images per sample a voxel size of ∼1.8 μm was achieved. In total 191 shells were reconstructed from the 12 available down-core time interval samples. On average 16 tests from each sample were scanned and a mean *G. bulloides* shell volume V_{form} was calculated for each time interval (Table [1\)](#page-2-0). The total volume V_{form} approximates the volume that a living foraminifer would occupy in the water column or the volume, *Vdis.water*, of seawater that it displaces. The buoyant force exerted on the shells was calculated from equation ([1](#page-6-8)) using the estimated water density (ρ) and the results are shown in Table [1.](#page-2-0) The buoyant force (in Newton, N) exerted on a foraminifer (*A*) in equilibrium during foating at its optimum depth equals the weight of the water it displaces (*W_{displaced}*). The weight of the water that a single foraminifer (without spines) displaces equals it mass ($m_{dis, water}$) times the acceleration of gravity g (m/s), or is the product of its volume times water density times g. There is an underestimation of the shell volume, both because fossil spines usually break and also because their exact efect on buoyancy is not well understood, but the geochemical signal of the core shell will always refect ambient seawater properties. Given that $V_{\text{form}} = V_{\text{dis.water}}$ from above, we get:

$$
A = W_{displaced} = m_{dis.water} * g = \rho * V_{dis.water} * g = \rho * V_{form} * g \tag{1}
$$

Error propagation in the determination of shell volume and buoyancy force. The error in the calculation of the buoyancy force of the ocean (σ_A) is a combination of the error in volume estimate of *G. bulloides* shells (σ_{Vform} = 13.64%) and the error of combined Mg/Ca - δ¹⁸O derived *in-situ* density (σ_t = 6.27%): $\sigma_A = \sqrt{\sigma_{Vform}^2 + \sigma_t^2}$, consequently $\sigma_A = 15$ %, which is almost equal to the size interval of the sieve fraction used.

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Author Contributions

S.D.Z. conceived the project and the methodology. A.T. performed Mg/Ca data collection and M.G. performed Mg/Ca data collection and curation. S.D.Z. wrote the manuscript with the help of A.T., G.K., P.G.M. and M.G. under the supervision of A.A., P.G.M. and H.D.

Additional Information

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