

Sample preparation for food chemistry analysis

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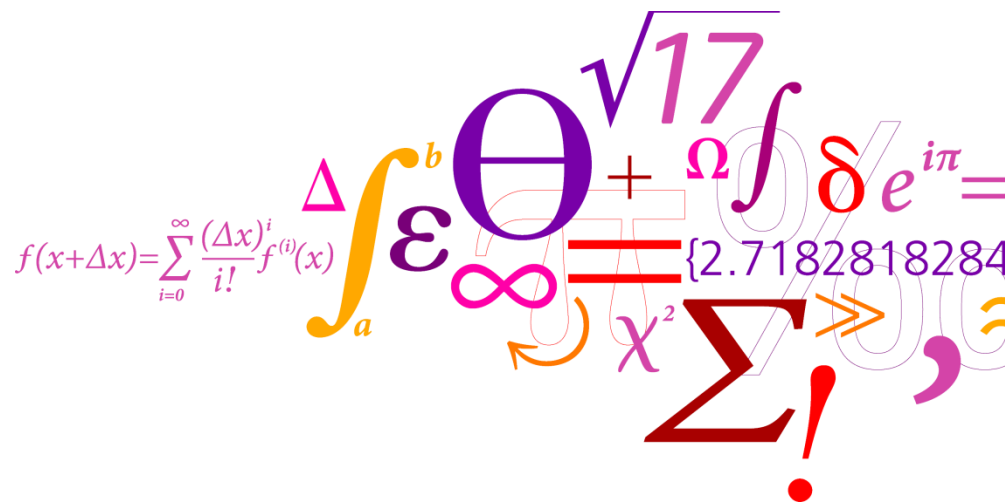
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Sample preparation for food chemistry analysis

Rie Romme Rasmussen

MSc, PhD, Chemist



National Food Institute

Focus on public health:

- human nutrition - food safety - food technology - environment and health -

Food products throughout the entire food chain:

- primary agricultural production
- industrial and home processing
- evaluating the impact on human health

Scientifically based risk assessments and advices

authorities and industry

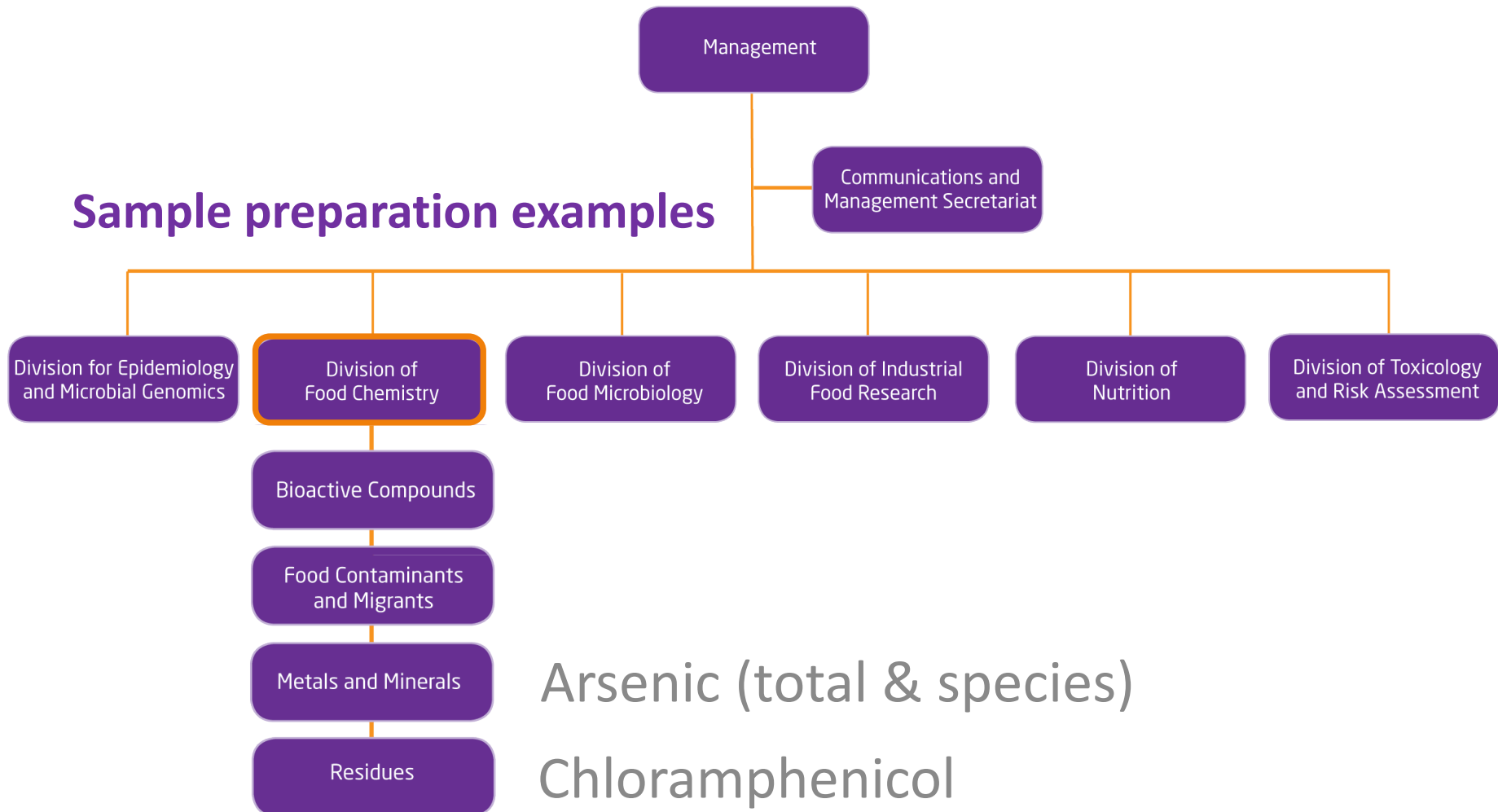
International and national reference laboratory

chemical and microbial food safety

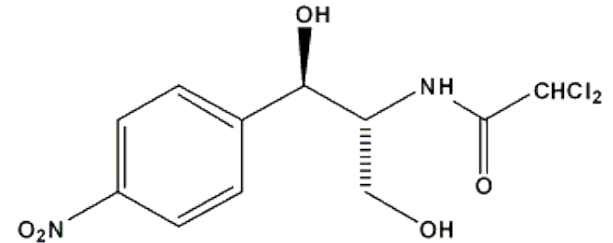
Part of the national food safety contingency plan

National Food Institute

Approx. 400 employees in six divisions:



Chloramphenicol

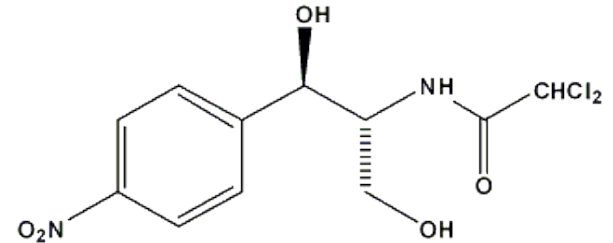


- Broad spectrum antibiotic
- Pharmaceutical prescribed for e.g., eye inflammation
- Severely toxic to bone marrow and the blood stem cells
- Banned in agricultural production in EU
- Analytical challenges; Chloramphenicol in honey
 - High sugar content
 - EU regulated at a low level (0.3 ng/g MRPL)
 - Verification method
 - GC-MS (4 ions required)
 - LC-MS-MS (1 precursor and 2 daughter ions required)

Chloramphenicol in honey

Sample preparation for GC-MS

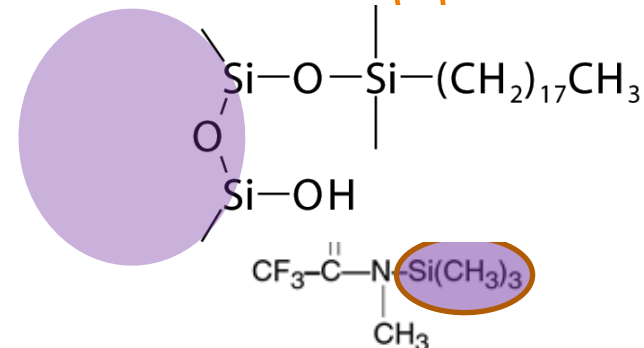
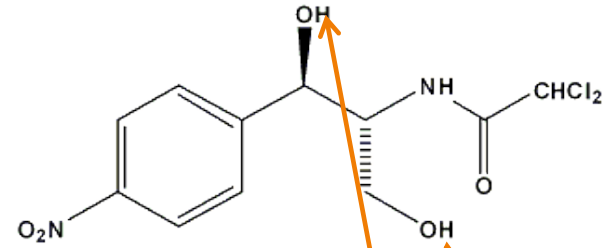
- Dissolve 1.5 g honey in 3 ml water
- Add internal standard (deuterium-labeled chloramphenicol)
- Load onto Extrelut column® (Merck)
- Elute with ethyl acetate
- Evaporate the ethyl acetate phase to dryness



Chloramphenicol in honey

Sample preparation for GC-MS

- Dissolve 1.5 g honey in 3 ml water
- Add internal standard (deuterium-labeled chloramphenicol)
- Load onto Extrelut column® (Merck)
- Elute with ethyl acetate
- Evaporate the ethyl acetate phase to dryness
- Re-dissolve with 0.3 ml methanol and 10 ml water
- **Load onto a C18 cartridge**
(conditioned with methanol and water)
- Wash
 - water
 - water/methanol (90/10)
- Elute: 5 ml methanol/water (55/45)
- Evaporate the eluate to dryness
- Derivate with MSTFA (1 hour at 60 °C)
- Evaporate to dryness and re-dissolve in 50 µl heptane



MSTFA
replaces labile H with
-Si(CH₃)₃

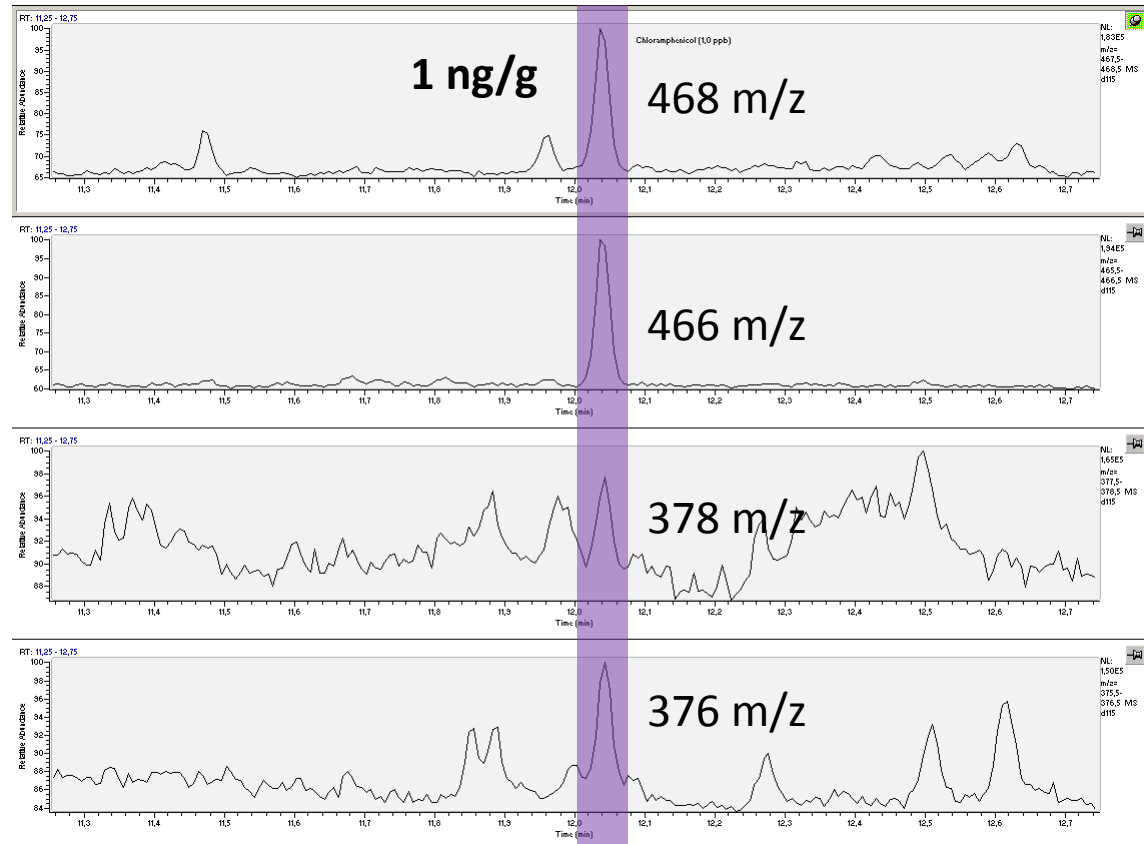
Chloramphenicol in honey

GC-MS

Temperature ramped separation on a 5% diphenyl column, negative chemical ionisation

Required
identification
RT & 2 ion-ratios
not obtained

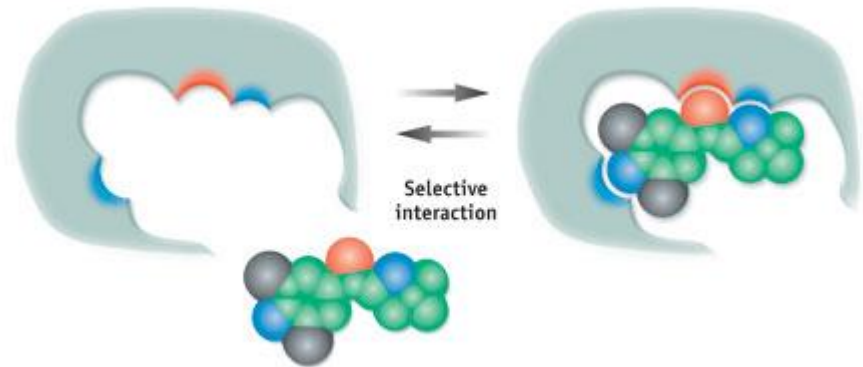
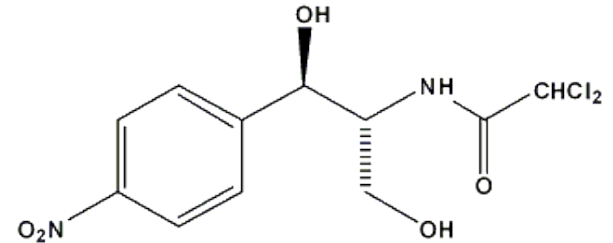
→ LC-MS/MS



Chloramphenicol in honey

Sample preparation for LC-MS/MS

- Dissolve 1.0 g honey in 1 ml water
- Add internal standard (deuterium-labeled chloramphenicol)
- Load onto a molecular imprinted polymers (MIP) cartridge (conditioned first with methanol and then with water)
- **Extensive wash**
 - 5% acetonitrile in 0.5% acetic acid
 - 1% ammonia
 - 20% acetonitrile in 1% ammonia
 - Water
 - 5% acetonitrile in 0.5% acetic acid
- Dry column with vacuum
- Elute with methanol
- Evaporate the eluate to dryness
- Re-dissolve in 500 μ l 10% methanol



Chloramphenicol in honey

LC-MS/MS

Kolonne: C-18 Inertsil ODS2 (2 x 150 mm, 3 μ m)

Gradient elution:

A: 10 % methanol / 90 % 0.001 M acetic acid

B: 100 % methanol

1 precursor and 2 daughter ions:

- 320.90 > 152.15 primary
- 320.90 > 257.15 secondary
- 325.93 > 157.18 internal standard



Detection capability = 10 times less than the required 0.3 ng/g

Arsenic in seafood

Arsenic is a metalloid present in soils, groundwater, surface water, air and **foods**

hydrogen 1 H 1.0079																	helium 2 He 4.0026						
lithium 3 Li 6.941	beryllium 4 Be 9.0122																	boron 5 B 10.811	carbon 6 C 12.011	nitrogen 7 N 14.007	oxygen 8 O 15.999	fluorine 9 F 18.998	neon 10 Ne 20.180
sodium 11 Na 22.990	magnesium 12 Mg 24.305																	aluminium 13 Al 26.982	silicon 14 Si 28.086	phosphorus 15 P 30.974	sulfur 16 S 32.065	chlorine 17 Cl 35.453	argon 18 Ar 39.948
potassium 19 K 39.098	calcium 20 Ca 40.078	scandium 21 Sc 44.956	titanium 22 Ti 47.867	vanadium 23 V 50.942	chromium 24 Cr 51.996	manganese 25 Mn 54.938	iron 26 Fe 55.845	cobalt 27 Co 58.933	nickel 28 Ni 58.693	copper 29 Cu 63.546	zinc 30 Zn 65.39	gallium 31 Ga 69.723	germanium 32 Ge 72.61	As 33 74.922	selenium 34 Se 78.96	bromine 35 Br 79.904	krypton 36 Kr 83.80						
rubidium 37 Rb 85.468	strontium 38 Sr 87.62	yttrium 39 Y 88.906	zirconium 40 Zr 91.224	niobium 41 Nb 92.906	molybdenum 42 Mo 95.94	technetium 43 Tc [98]	ruthenium 44 Ru 101.07	rhodium 45 Rh 102.91	palladium 46 Pd 106.42	silver 47 Ag 107.87	cadmium 48 Cd 112.41	indium 49 In 114.82	tin 50 Sn 118.71	antimony 51 Sb 121.76	tellurium 52 Te 127.60	iodine 53 I 126.90	xenon 54 Xe 131.29						
caesium 55 Cs 132.91	barium 56 Ba 137.33	57-70 *	lutetium 71 Lu 174.97	hafnium 72 Hf 178.49	tantalum 73 Ta 180.95	tungsten 74 W 183.84	rhenium 75 Re 186.21	osmium 76 Os 190.23	iridium 77 Ir 192.22	platinum 78 Pt 195.08	gold 79 Au 196.97	mercury 80 Hg 200.59	thallium 81 Tl 204.38	lead 82 Pb 207.2	bismuth 83 Bi 208.98	polonium 84 Po [209]	astatine 85 At [210]	radon 86 Rn [222]					
francium 87 Fr [223]	radium 88 Ra [226]	89-102 **	lawrencium 103 Lr [262]	rutherfordium 104 Rf [261]	dubnium 105 Db [262]	seaborgium 106 Sg [266]	bohrium 107 Bh [264]	hassium 108 Hs [269]	meitnerium 109 Mt [268]	ununnilium 110 Uun [271]	unununium 111 Uuu [272]	ununbium 112 Uub [277]	ununquadium 114 Uuq [289]										

* Lanthanide series

** Actinide series

lanthanum 57 La 138.91	cerium 58 Ce 140.12	praseodymium 59 Pr 140.91	neodymium 60 Nd 144.24	promethium 61 Pm [145]	samarium 62 Sm 150.36	europium 63 Eu 151.96	gadolinium 64 Gd 157.25	terbium 65 Tb 158.93	dysprosium 66 Dy 162.50	holmium 67 Ho 164.93	erbium 68 Er 167.26	thulium 69 Tm 168.93	ytterbium 70 Yb 173.04
actinium 89 Ac [227]	thorium 90 Th 232.04	protactinium 91 Pa 231.04	uranium 92 U 238.03	neptunium 93 Np [237]	plutonium 94 Pu [244]	americium 95 Am [243]	curium 96 Cm [247]	berkelium 97 Bk [247]	californium 98 Cf [251]	einsteinium 99 Es [252]	fermium 100 Fm [257]	mendelevium 101 Md [258]	nobelium 102 No [259]

Total arsenic

- Sample homogenisation
 - Ultra turrax
 - Kitchen mill

Contamination:

Dust and steel contain some trace elements. Apply titanium knives and a LAF bench.



Total arsenic

- Sample homogenisation
- Microwave assisted extraction
 - Strong acid (65 % v/v HNO₃)
 - High pressure and temperature

**Only small samples size
(organic material)**

**Glass can release arsenic.
Apply high quality quartz vessels.
Store samples and solutions in plastic
containers**



Total arsenic

Inductively Coupled Plasma Mass Spectrometry

→ Dilute conc. extract with water (to ~2 % v/v HNO₃)

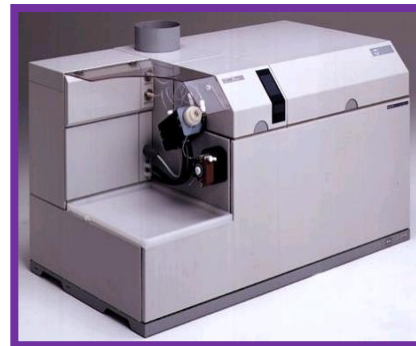
- Add internal standard mix (corrects for sensitivity drift)

☺ High sensitivity

☹ High purchase and running costs

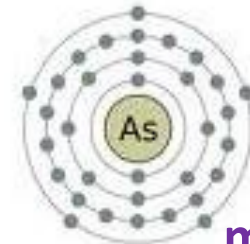
Inductively coupled
(argon) plasma
is 6000-8000 °K

- Desolvation
- Vaporisation
- Atomisation
- Ionisation



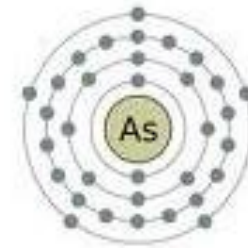
ICP-MS

Element specific
detection



m/z 75

Total arsenic



Hydride generation atomic absorption spectrometry

☺ Low purchase and running costs

- Continuous Flow Vapor Generation System
- Atomic Absorption Spectrometer

Instrument settings

- Electrical heated cell (900° C)
- **Element specific lamp for As**
- Wave length (193.7 nm)
- Slit width (0.5 nm)

Gaseous arsenic hydrides is transported by argon gas to the cell

→ Atomisation reaction

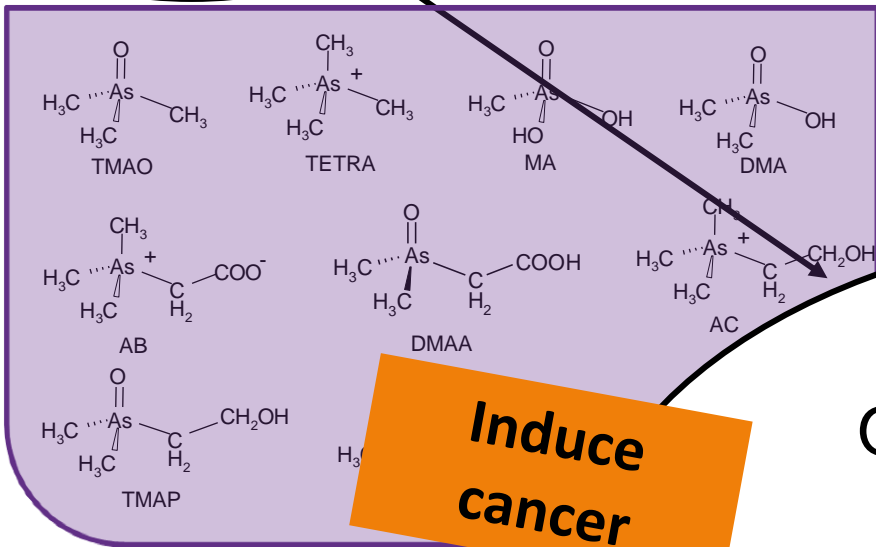
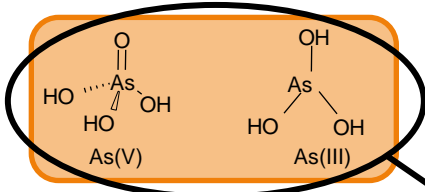
→ Atomic absorption of arsenic



HG-AAS
total arsenic

Arsenic species

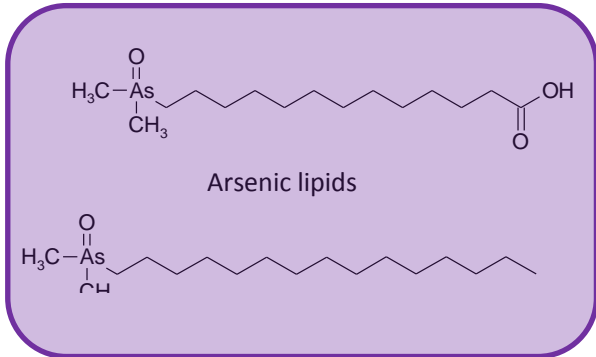
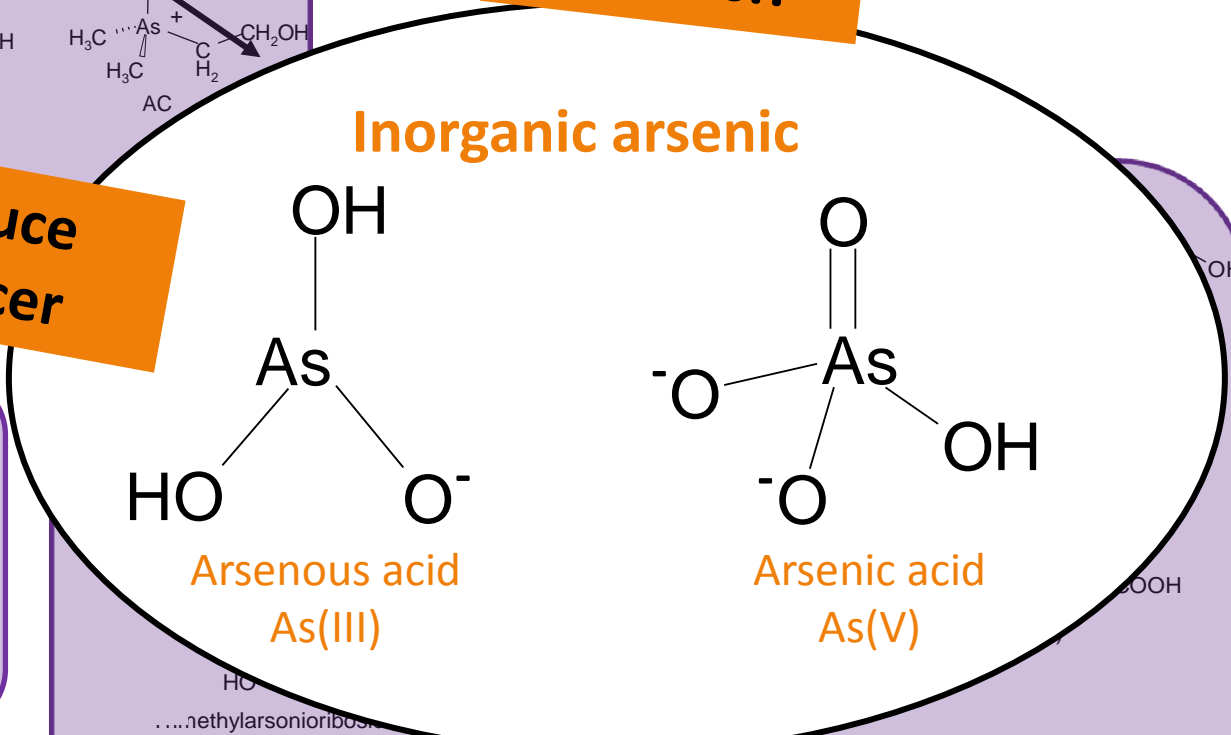
More than 50 different marine arsenic forms:



- inorganic species
- organic species

Most toxic form of arsenic!!

Induce cancer



Arsenic species

- Marine organisms can bioaccumulate arsenic (As)
- Major part of arsenic exists as organic arsenic in seafood
- As toxicity is species specific, inorganic arsenic (iAs) is most toxic
- iAs causes cancer & skin lesion
- Comprehensive As speciation data for food items missing
- Inorganic As and not total As in food should be determined for better risk assessment

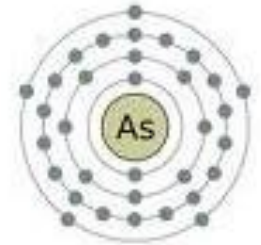
- **Sample extraction aim**

High extraction efficiency, no interconversions between inorganic As and organo As species

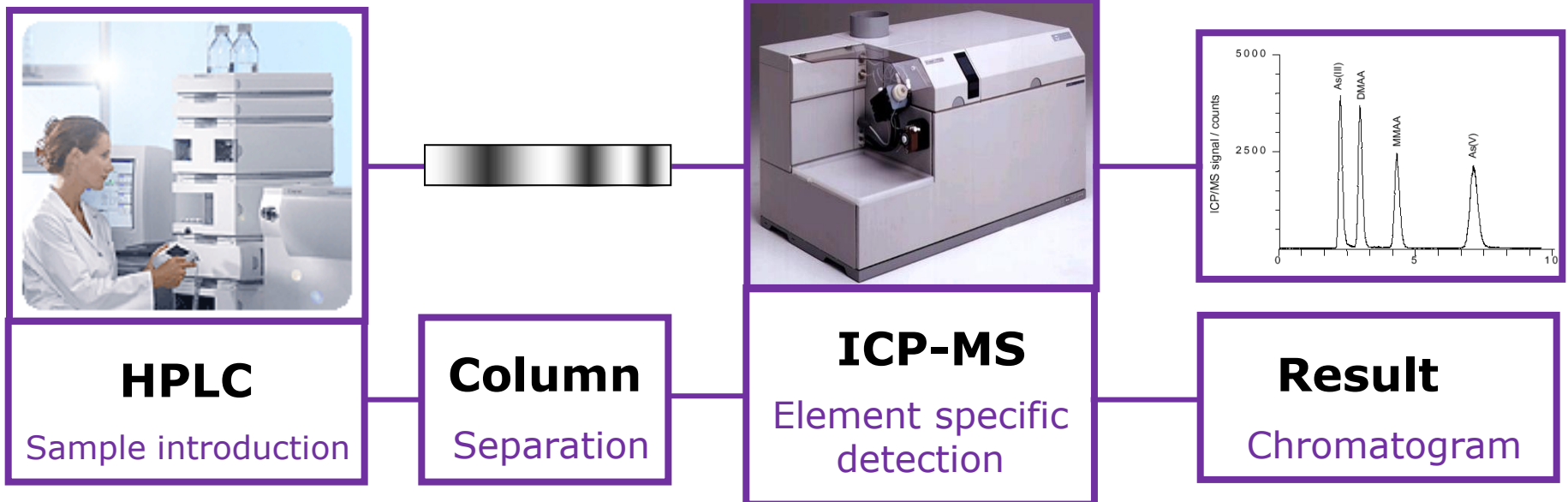
Arsenic species

HPLC-ICP-MS (anion exchange column)

- ☺ Species have different properties (pKa values)
- ☹ ICP-MS → high purchase and running costs



m/z 75

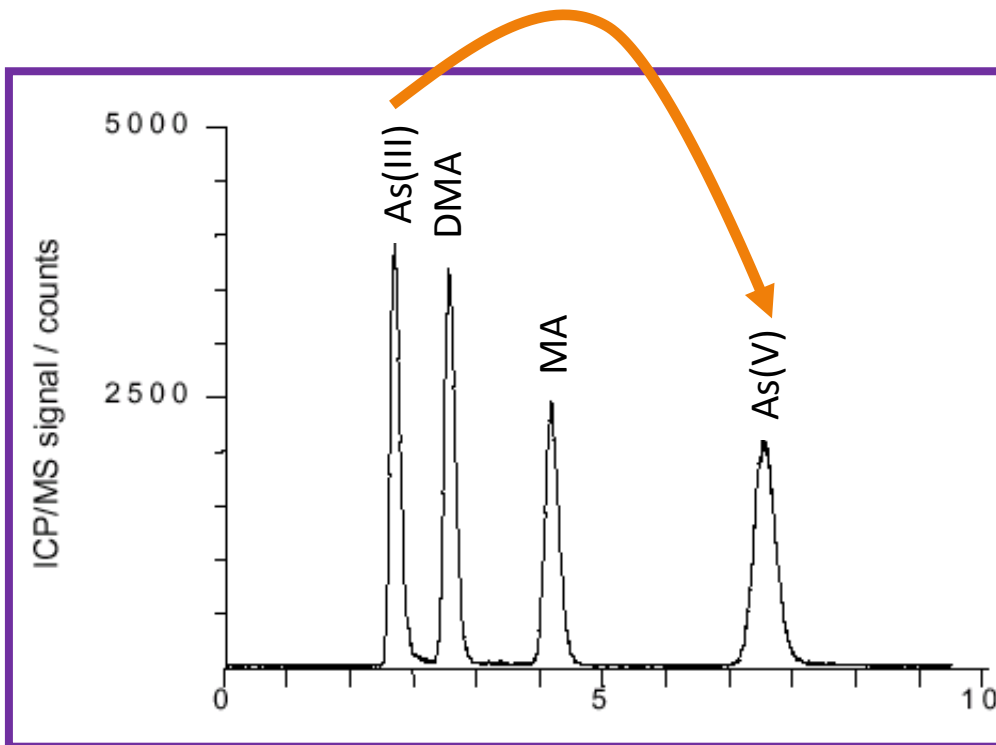


Arsenic species

- **Sample preparation aim**

Isolate inorganic arsenic in one fraction

→ Measure arsenic with the cheap HG-AAS detector



Trick

Convert As^{III} to As^{V}
for higher anion
properties **by an
oxidant**

SPE separation

by anion exchange

HPLC (anion exchange) -ICP-MS

Arsenic speciation analysis

speciation alternative: SPE, HG-AAS

0.2 g sample
+ 10 mL extractant
(0.06 M HCl, 3% H₂O₂)

μ-wave
extraction

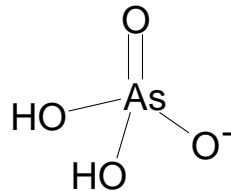
SPE
separation

HG-AAS
detection



25 minutes at 90° C

Centrifugation
10 min 2100 x g



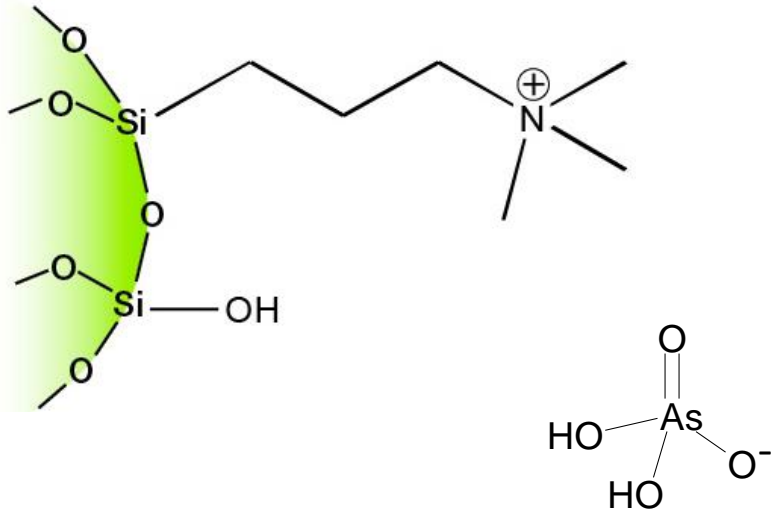
inorganic arsenic

hydride generation
atomic absorption
spectrometry

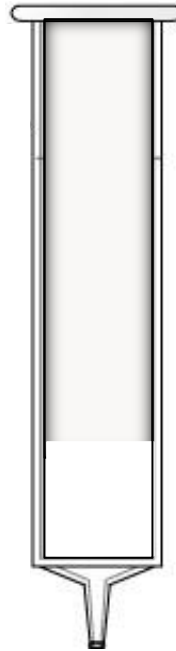
😊 low costs

SPE protocol

Separation of As species



Strong anion exchange SPE column
silica based



The **charge** of the arsenic species depends on pH

@ pH = 6 **iAs(V)** is **negatively charged**

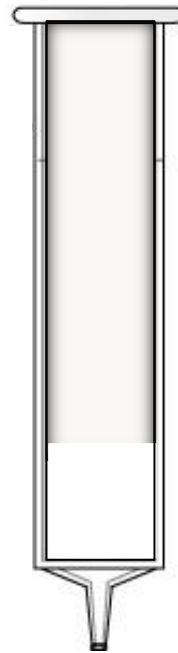
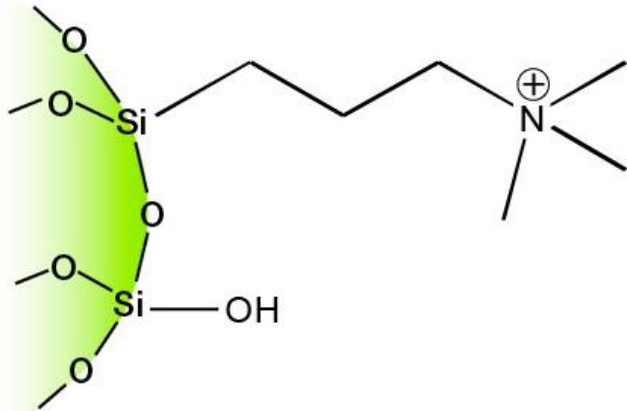
Load of pH buffered sample

Sequential elution

Separation of inorganic As from organo As species by SPE

SPE protocol

Separation of As species



Condition

100 % MeOH

Equilibrate

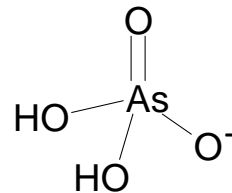
Buffer: 20mM (NH₄)₂CO₃, 0.03 M HCl
and 1.5% H₂O₂

Load

Buffered sample: pH 5.0-7.5

Wash 0.5 M CH₃COOH

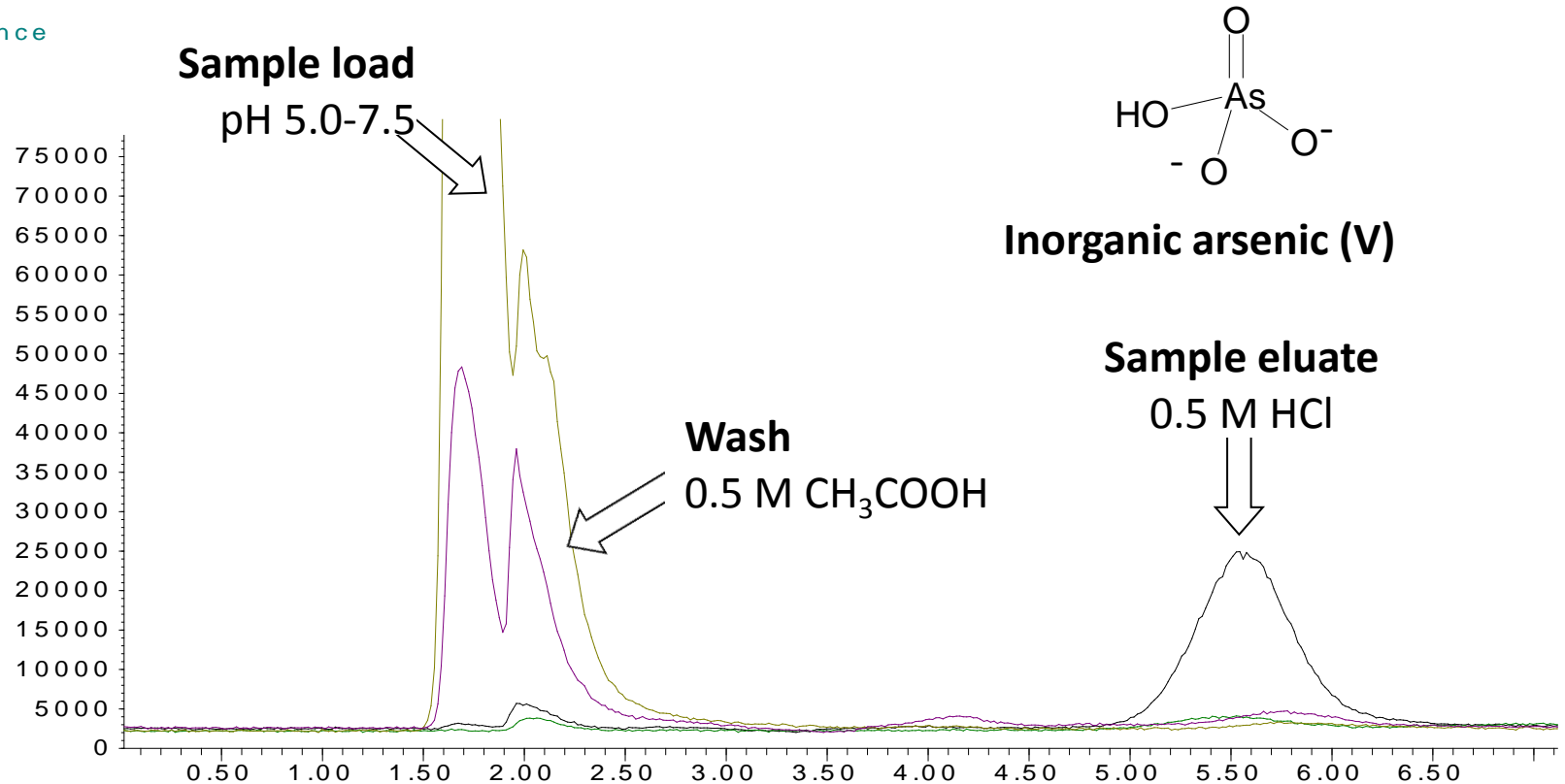
Elute 0.5 M HCl



Arsenic in SPE eluates

load, wash and sample fraction

Abundance



Time-->

Figure. HPLC-ICP-MS chromatogram of fish protein (TORT-2) 3 SPE fractions separated on an anion exchange column (ION-120 part nr. ANX-00-6550, 120x4.6 mm), 40 mM carbonate pH 10.3.

HG-AAS detection

arsenic in sample eluate

Pre-reduction: As(V) \rightarrow As(III)

- Mix sample eluate with KI and ascorbic acid, 3 M HCl
60 min incubation

- Add more 3 M HCl
Another 60 min incubation

Hydride generation reagents

HCl (4.7 M)

NaBH₄ / NaOH (0.5 % w/v)

Instrument settings:

Electrical heated cell (900° C)

Element specific lamp for As

Wave length (193.7 nm)

Slit width (0.5 nm)

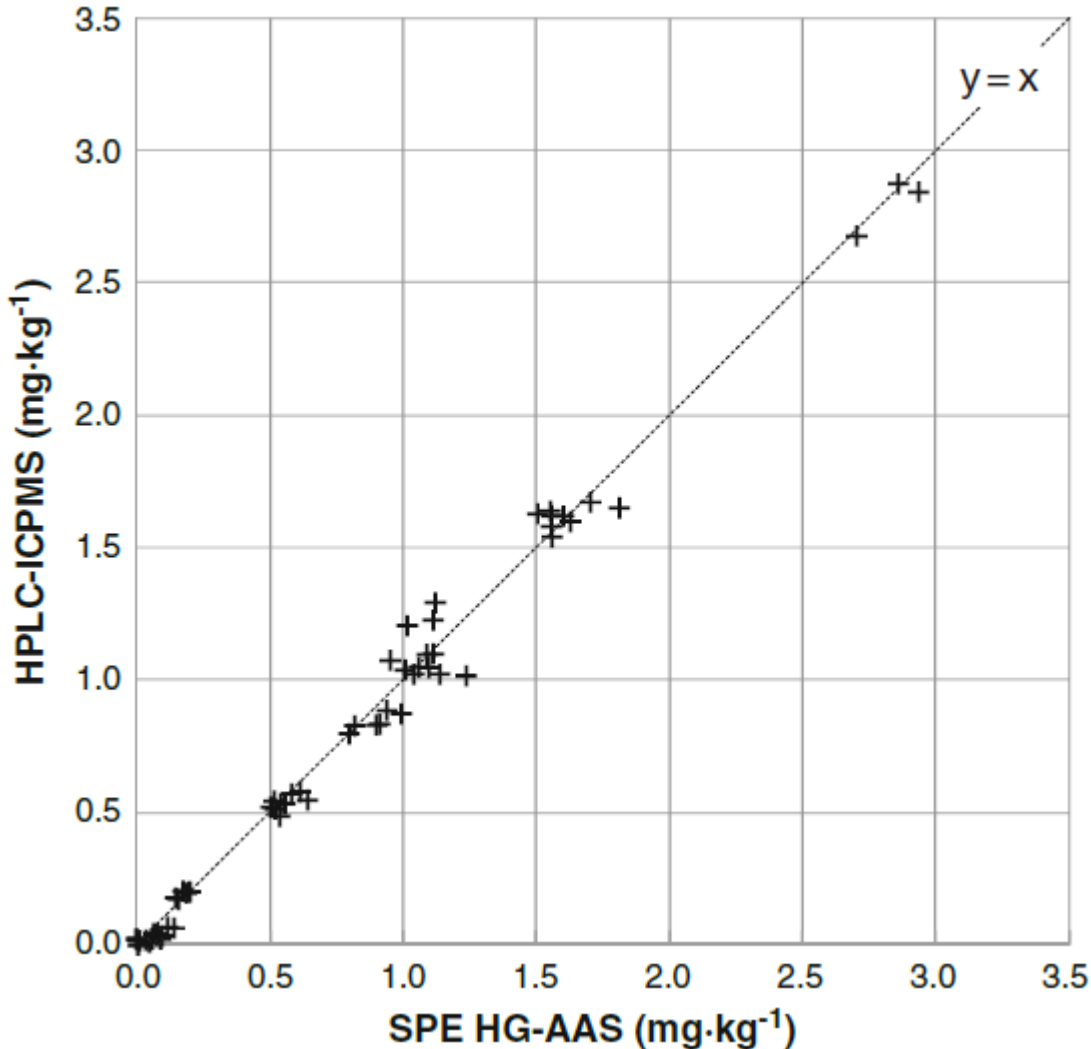
Hydride generation atomic absorption spectrometry



Thermo Scientific

HG-AAS
total arsenic
in eluate

Arsenic speciation analysis



Inorganic arsenic (As^V) by two different methods:

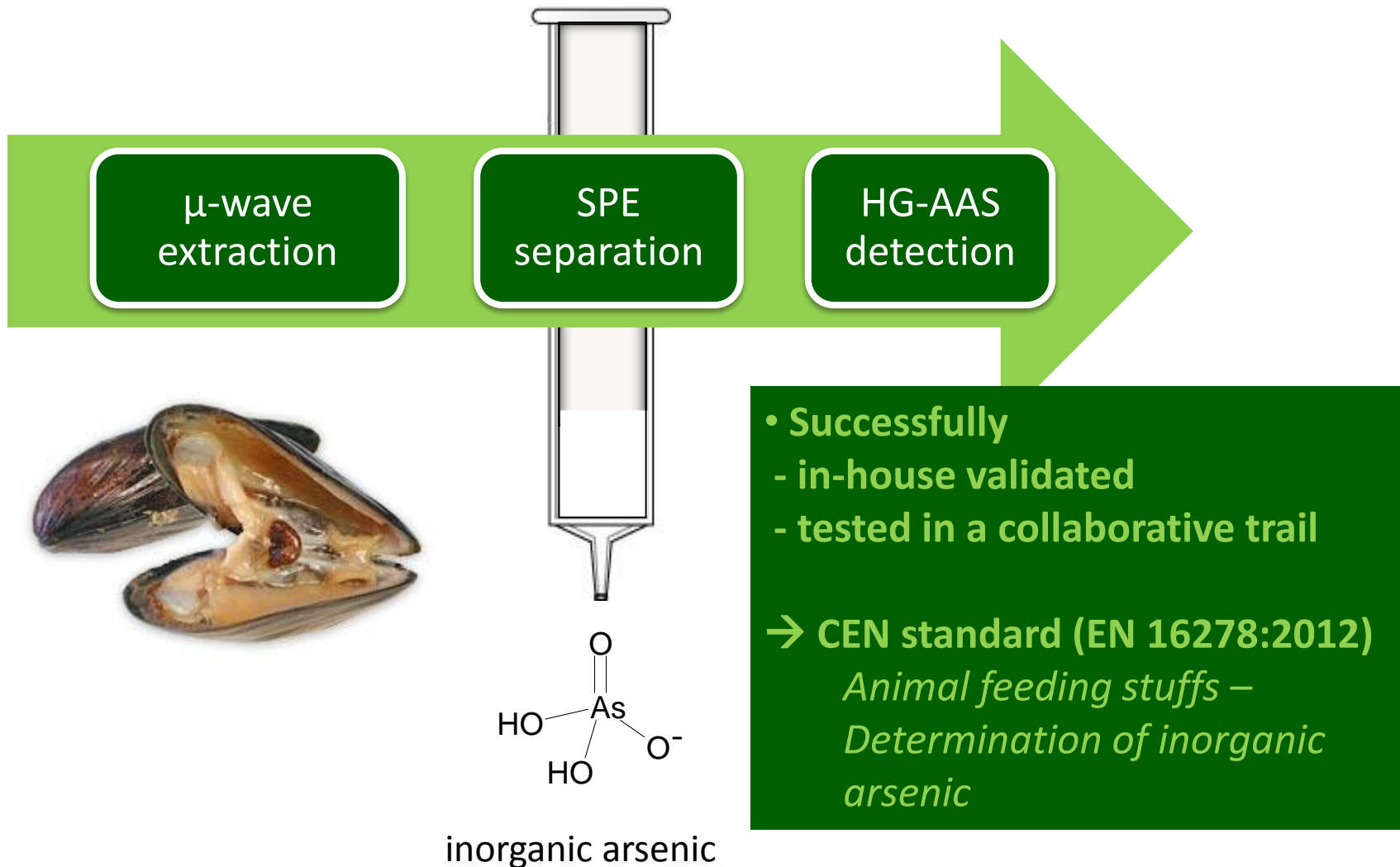
- HPLC-ICP-MS
- SPE HG-AAS

In total 72 blank, spiked and natural incurred marine samples were analysed.

No significant difference (P<5%)

Arsenic speciation analysis

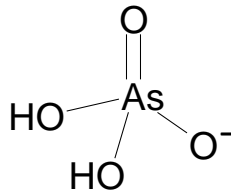
speciation alternative: SPE, HG-AAS



μ -wave
extraction

SPE
separation

HG-AAS
detection



inorganic arsenic

Sample preparation @ DTU Food

- Cartridges examples
 - **Extrelut**
Replacement for traditional liquid-liquid extraction, removed also water
 - **C18**
Retains apolar compounds. Removed only part of interfering compounds (chloramphenicol in honey by GC-MS)
 - **Molecular imprinted polymers (MIP)**
Compound specific, extensive wash possible, low detection capability obtained (chloramphenicol in honey by LC-MS/MS)
 - **Strong anion exchange**
Retain negatively charged compounds, sequential elution, inorganic arsenic isolated in one fraction

Thank you for your attention!

