

EPR spectra of spin labels in lipid bilayers

Hansgeorg Schindler and Joachim Seelig

Citation: *The Journal of Chemical Physics* **59**, 1841 (1973); doi: 10.1063/1.1680269

View online: <http://dx.doi.org/10.1063/1.1680269>

View Table of Contents: <http://aip.scitation.org/toc/jcp/59/4>

Published by the *American Institute of Physics*



**COMPLETELY
REDESIGNED!**

**PHYSICS
TODAY**

Physics Today Buyer's Guide
Search with a purpose.

EPR spectra of spin labels in lipid bilayers

Hansgeorg Schindler and Joachim Seelig*

Biocenter of the University of Basel, Department of Biophysical Chemistry, Klingelbergstrasse 70, CH 4056 Basel, Switzerland

(Received 26 March 1973)

Assuming a probability function $P(\theta) \propto \exp\{-q/RT \cos^2 \theta\}$ for the orientation of the chain segments in a lipid bilayer, a quantitative theory for the spin label spectra of such systems is developed. Only three parameters, namely two energy parameters q_3 and q_1 and the correlation time τ_{20} are required to yield perfect agreement between experimental and theoretical spectra for correlation times up to 3×10^{-9} sec. In calculating the EPR spectra pseudosecular and nonsecular contributions to the linewidth are included, since especially the former exert a distinct influence on the spectra. Also discussed are the effects of tilt angle and spread angle. The theory is applied to the anisotropic motion of the fatty acid spin label I (13.2) in a decanol-sodium decanoate bilayer. Between 44 and 8°C the correlation time is found to increase exponentially from 1.0×10^{-10} to 4.0×10^{-10} sec with an activation energy of 6.84 kcal mole⁻¹. The energy parameters q_3 and q_1 , which determine the order parameters S_3 and S_1 , remain constant in the same temperature range ($q_3 = -2.9$ kcal mole⁻¹; $q_1 = 1.7$ kcal mole⁻¹). The average orientation of the chain segment is perpendicular to the bilayer normal. Below 8°C a phase transition occurs and the chain segments become gradually tilted. At -8°C the correlation time has increased to 2.8×10^{-9} sec and the chains are now tilted by approximately 18° from the bilayer normal. From the correlation time the translational diffusion constant can also be deduced. At 21°C a value of $D_{\text{trans}} = 1.5 \times 10^{-6}$ cm² sec⁻¹ is obtained for the lipid molecules in the decanol-sodium decanoate bilayer. In vesicles of dimyristoyl-L- α -lecithin at 21°C the analysis of the spectrum yields a correlation time of 2.5×10^{-9} sec and a diffusion constant of the lipid molecules of $D_{\text{trans}} = 3.6 \times 10^{-8}$ cm² sec⁻¹.

I. INTRODUCTION

There is now considerable evidence that a substantial portion of biological membranes contains a lipid bilayer.¹ Magnetic resonance studies, in particular spin label investigations, have revealed that the membrane lipids are in a more or less "fluid" state. Probably the most important contributions to the membrane fluidity arise from fast rotations around the long molecular axes of the lipid molecules and from rotational isomerisations around carbon-carbon bonds. The latter process confers a distinct flexibility upon the fatty acid chains in a lipid bilayer as has been first demonstrated for very simple model systems.² The alkyl chains were found to be well-ordered in the vicinity of the polar interface, but the motion became increasingly more random towards the central part of the bilayer. Such flexibility gradients have subsequently been observed in phospholipid bilayers³ and in biological membranes.^{4,5} They pose an interesting packing problem which has been overcome by the proposition of bent fatty acid chains in phospholipid bilayers.^{6,7} Nevertheless, the physical description of the chain conformation in lipid bilayers and related to this the question of interaction energies between adjacent alkyl chains are still in an early stage of discussion.⁸⁻¹¹

Much could be gained in this situation if precise experimental parameters describing the behavior of the lipid molecules were available. This information can be obtained, in principle, from the

EPR signals of lipid spin labels, and the purpose of this report, therefore, is to reinvestigate the interpretation of such spectra. It will be shown in the following, that all molecular information is contained in three parameters, namely two energy parameters q_1 and q_3 , which determine the average position of a chain segment, and a dynamic parameter, the correlation time τ_{20} . These three parameters are sufficient to yield perfect agreement between experimental and simulated spectra for correlation times up to 3×10^{-9} sec. The theory will be applied to various experimental situations: We shall discuss (1) the effect of various types of macroscopic alignments of the bilayer phases, (2) the influence of the order parameter, and (3) the temperature dependence of the spectra. The latter process will also yield the activation energy of the motion. This report is essentially concerned with the theory of the spin label spectra. The statistical-mechanical interpretation of the q parameters together with results from other bilayer systems will be presented in a subsequent publication.

II. THEORY OF THE SPIN LABEL SPECTRA

The molecular motions in liquid crystals are highly anisotropic. The short-range order can be characterized by a director z' which defines the axis of rotational averaging. The average position of a molecule with respect to z' is then expressed in terms of order parameters S_i .¹² On a macroscopic scale, i. e., looking at a large sample, the z' axes of the microcrystalline regions are distrib-

uted at random, but under certain experimental conditions a macroscopic alignment of the z' axes can be achieved. In the case of nematic phases this is possible by orienting the system in a strong magnetic field. For lamellar smectic phases the substance may be pressed between two closely spaced quartz plates, resulting in a homogeneous planar orientation. A cylindrical alignment of the bilayer phases can be obtained by orienting the system in glass capillaries. In an oriented planar sample the directions of the z' axes may be either identical with the axis of macroscopic orientation N or they may be tilted by a certain angle. In simulating the spin label spectra one must also consider that the macroscopic ordering may not be perfect and that the N axis of a planar sample may be varied arbitrarily with respect to the direction of the external magnetic field.

McFarland and McConnell presented a theory which takes into account most of the above mentioned phenomena.⁶ We have refined and extended their treatment by introducing essentially the following modifications. (1) We include nonsecular and pseudosecular terms in the linewidth calculation. These terms yield contributions for very short or very long correlation times up to 10^{-10} and 3×10^{-9} sec, respectively. (2) The linewidth is expressed directly as a function of correlation times. (3) The order parameters are defined as functions of energy parameters. (4) The matrix of the order parameters is not necessarily assumed to be axial symmetric. A more detailed discussion of these and some other changes will be given in the following.

A. Definition of Order Parameters

A Cartesian coordinate system is ascribed to the NO radical such, that the x axis is extended in the direction of the N-O bond and the z axis in the direction of the nitrogen $2p\pi$ orbital. In the case of the fatty acid spin labels used in this work, the z axis is practically identical with the direction of the extended hydrocarbon chain. The orientation of the NO group with respect to z' is then given by

$$\begin{aligned} x \cdot z' &= \cos\theta_1(t) = \sin\beta(t) \cos\gamma(t), \\ y \cdot z' &= \cos\theta_2(t) = \sin\beta(t) \sin\gamma(t), \\ z \cdot z' &= \cos\theta_3(t) = \cos\beta(t). \end{aligned} \quad (1)$$

(α , β , and γ are Eulerian angles, where α corresponds to a rotation around z' . The use of Eulerian angles simplifies coordinate transformations.) The time averaged position of the NO group can then be expressed in terms of order parameters 2S_i :

$$S_i = \frac{1}{2}(3\langle \cos^2\theta_i \rangle - 1), \quad i = 1, 2, 3, \quad (2)$$

with

$$\sum_{i=1}^3 S_i = 0. \quad (3)$$

The calculation of the line shape parameters requires knowledge of averages $\langle \cos^4\theta_i \rangle$, which are not accessible experimentally. Therefore, the following approximation will be made: From investigations of nematic liquid crystals it is known that the angular distribution of the molecules can be described by a probability function of the kind¹³⁻¹⁶

$$P(\theta) \propto \exp[-(q/RT) \cos^2\theta], \quad (4)$$

where θ is the angle between the long axis of the rodlike molecules and z' . It is assumed that this probability function can also be applied to the orientation of hydrocarbon chains in lipid bilayers. Since the chains are flexible the energy parameters q must vary as a function of the bond number n . It is further assumed that Eq. (4) not only describes the angular distribution of the long molecular axis but that similar relations hold for the remaining two axes. Combining Eqs. (2) and (4) we can thus write:

$$S_i = -\frac{1}{2} + \frac{3}{2} \frac{\int_0^\pi \cos^2\theta_i \exp[-(q_i/RT) \cos^2\theta_i] \sin\theta_i d\theta_i}{\int_0^\pi \exp[-(q_i/RT) \cos^2\theta_i] \sin\theta_i d\theta_i}. \quad (5)$$

As independent parameters we choose q_1 and q_3 , which can be determined from S_1 and S_3 . The third parameter is then fixed by Eq. (3). Knowing the q_i parameters other angular averages are easily obtained.

B. General Theory of the Spin Hamiltonian

To proceed the total spin Hamiltonian $\hbar\mathcal{H}(t)$ of the nitroxide radical is divided into a time-averaged part $\hbar\bar{\mathcal{H}}$ and a time-dependent part $\hbar\mathcal{H}_1(t)$:

$$\hbar\mathcal{H}(t) = \hbar\bar{\mathcal{H}} + \hbar\mathcal{H}_1(t) = \hbar\bar{\mathcal{H}} + \hbar\{\mathcal{H}(t) - \bar{\mathcal{H}}\}. \quad (6)$$

$\hbar\bar{\mathcal{H}}$ determines the position of the EPR lines, while the perturbation $\hbar\mathcal{H}_1(t)$ causes relaxation and line broadening. This has been discussed extensively in various texts.^{15, 17-20}

$\hbar\mathcal{H}(t)$ can be expressed as the sum of irreducible tensor operators $A_\lambda^{(l,m)}$:

$$\hbar\mathcal{H}(t) = \sum_{l,m,\lambda} F_\lambda^{(l,m)} A_\lambda^{(l,m)}(t). \quad (7)$$

The double prime signifies that the $A_\lambda^{(l,m)}(t)$ are referred to the molecular axes of the NO radical. A listing of the tensor operators $A_\lambda^{(l,m)}$ and the spatial factors $F_\lambda^{(l,m)}$ is given in Table I.

The transformation from the molecule fixed coordinates of the NO group to the space fixed coordinate system of the director z' is carried out

with rotation matrices $D_{m',m}^{(l)21}$:

$$\bar{h}\mathcal{H}(t) = \sum_{l,m,\lambda} F_{\lambda}^{(l,-m)} \sum_{\rho} D_{\rho,m}^{(l)} (\alpha\beta\gamma)(t) A_{\lambda}^{(l,\rho)}. \quad (8)$$

The time dependence is now contained in the rotation matrices. A second transformation is necessary in order to rotate the coordinate system of the director z' into the laboratory system defined by the direction of the external magnetic field H_0 :

$$\bar{h}\mathcal{H}(t) = \sum_{l,m,\lambda} F_{\lambda}^{(l,-m)} \sum_{\rho} D_{\rho,m}^{(l)} (\alpha\beta\gamma)(t) \times \sum_{\alpha} D_{\alpha,\rho}^{(l)} (\alpha'\beta'\gamma') A_{\lambda}^{(l,\alpha)}. \quad (9)$$

β' denotes the angle between the magnetic field H_0 and the director axis z' .

C. Time Averaged Spin Hamiltonian

The time averaged spin Hamiltonian must be invariant against rotations around the director axis z' . Therefore

$$\bar{h}\mathcal{H} = \sum_{l,m,\lambda} F_{\lambda}^{(l,-m)} \overline{D_{0,m}^{(l)}(\beta\gamma)(t)} \sum_{\alpha} D_{\alpha,0}^{(l)} (\alpha'\beta') A_{\lambda}^{(l,\alpha)}. \quad (10)$$

For $z' \parallel H_0$ ($\beta' = 0^\circ$) Eq. (10) reduces to

$$\bar{h}\mathcal{H}_{\parallel} = \sum_{l,m,\lambda} F_{\lambda}^{(l,-m)} D_{0,m}^{(l)} (\beta\gamma)(t) A_{\lambda}^{(l,0)}, \quad (11)$$

$$\bar{h}\mathcal{H}_{\parallel} = \beta_e H_0 g_{\parallel} S_z + \hbar I_z T_{\parallel} S_z, \quad (12)$$

with

$$g_{\parallel} = g + 2\Delta g S_3 + \frac{2}{3}\delta g(S_1 - S_2), \quad (13)$$

$$T_{\parallel} = a + 2\Delta a S_3 + \frac{2}{3}\delta a(S_1 - S_2). \quad (14)$$

In deriving Eqs. (13) and (14) the terms S^+I^+ have been neglected, since they lead only to small second-order effects.

For $z' \perp H_0$ ($\beta' = 90^\circ$) the following result is obtained:

$$\bar{h}\mathcal{H}_{\perp} = \sum_{l,m,\lambda} F_{\lambda}^{(l,-m)} \overline{D_{0,m}^{(l)}(\beta\gamma)(t)} D_{0,0}^{(l)} (\beta' = 90^\circ) A_{\lambda}^{(l,0)}, \quad (15)$$

TABLE I. Irreducible tensor operators $A_{\lambda}^{(l,m)}$ and spatial factors $F_{\lambda}^{(l,-m)}$.

$A_A^{(0,0)}$	$[S_z I_z + \frac{1}{2}(S^+ I^+ + S^- I^-)]$	$F_A^{(0,0)} = ha$
$A_A^{(2,0)}$	$(\frac{2}{3})^{1/2} [S_z I_z - \frac{1}{2}(S^+ I^+ + S^- I^-)]$	$F_A^{(2,0)} = (\frac{2}{3})^{1/2} h\Delta a$
$A_A^{(2,\pm 1)}$	$\mp (S_z I^{\pm} + S^{\pm} I_z)$	$F_A^{(2,\pm 1)} = 0$
$A_A^{(2,\pm 2)}$	$S^{\pm} I^{\pm}$	$F_A^{(2,\pm 2)} = \frac{1}{2} h\delta a$
$A_G^{(0,0)}$	S_z	$F_G^{(0,0)} = I_0 \beta_e g$
$A_G^{(2,0)}$	$(\frac{2}{3})^{1/2} S_z$	$F_G^{(2,0)} = (\frac{2}{3})^{1/2} H_0 \beta_e \Delta g$
$A_G^{(2,\pm 1)}$	$\mp S^{\pm}$	$F_G^{(2,\pm 1)} = 0$
$A_G^{(2,\pm 2)}$	0	$F_G^{(2,\pm 2)} = \frac{1}{2} H_0 \beta_e \delta g$
$a = \frac{1}{2}(T_{xx} + T_{yy} + T_{zz})$	$g = \frac{1}{2}(g_{xx} + g_{yy} + g_{zz})$	
$\Delta a = \frac{1}{6}(2T_{zz} - T_{xx} - T_{yy})$	$\Delta g = \frac{1}{6}(2g_{zz} - g_{xx} - g_{yy})$	
$\delta a = \frac{1}{2}(T_{xx} - T_{yy})$	$\delta g = \frac{1}{2}(g_{xx} - g_{yy})$	

$$\bar{h}\mathcal{H}_{\perp} = \beta_e H_0 g_{\perp} S_z + \hbar I_z T_{\perp} S_z, \quad (16)$$

$$g_{\perp} = g - \Delta g S_3 - \frac{1}{3}\delta g(S_1 - S_2), \quad (17)$$

$$T_{\perp} = a - \Delta a S_3 - \frac{1}{3}\delta a(S_1 - S_2). \quad (18)$$

If β' lies between these two extremes, then

$$\bar{h}\mathcal{H}(\beta') = \beta_e H_0 g(\beta') S_z + \hbar I_z T(\beta') S_z, \quad (19)$$

with

$$g(\beta') = [\cos^2 \beta' g_{\parallel}^2 + \sin^2 \beta' g_{\perp}^2]^{1/2}, \quad (20)$$

$$T(\beta') = [\cos^2 \beta' T_{\parallel}^2 + \sin^2 \beta' T_{\perp}^2]^{1/2}. \quad (21)$$

Compared to Eq. (10) the angle α' is eliminated from Eq. (19). Insertion of Eqs. (13), (14), (17), (18) into (20) and (21) and in combination with (19) finally yields the effective spin Hamiltonian as a function of S_1 , S_2 , and β' .

The last problem in the theory of the static Hamiltonian is to include the effect of tilted fatty acid chains.^{6,7} A tilt angle can be detected only in an oriented sample of bilayers (i. e., planar or cylindrical orientation) but is averaged out in a random distribution. We therefore confine our discussion to a planar oriented sample. The normal on the bilayer planes is denoted with N . If ϑ is the tilt angle between N and the director axis z' and if Δ is the angle between N and the magnetic field H_0 , then β' , the angle between z' and H_0 , is found by the following relation:

$$\cos \beta' = \cos \vartheta \cos \Delta + \sin \vartheta \sin \Delta \cos \varphi. \quad (22)$$

All director axes need not have exactly the same tilt angle, instead they may be distributed according to a certain probability function $\rho(\vartheta)$. In our calculations we have used a distribution function suggested by McFarland and McConnell,⁶ but other choices are equally possible:

$$\rho(\vartheta) \sim \sin \vartheta \exp[-(\vartheta - \bar{\vartheta})^2 / 2\vartheta_0^2]. \quad (23)$$

$\bar{\vartheta}$ is the average angle of tilt and ϑ_0 is called the spread angle. In order to simulate the EPR spectra Eq. (22) is inserted into Eqs. (20) and (21). The angles ϑ and φ are then varied between $0 \leq \vartheta \leq \pi$ and $0 \leq \varphi \leq 2\pi$ and each spectrum for a given ϑ is weighted by Eq. (23).

D. Linewidth of the Spin Label Spectra

The linewidth of an electron spin resonance signal depends on both the molecular tumbling rate and the orientation of the radical with respect to the external magnetic field. In the case of the nitroxide radical (electron spin $\frac{1}{2}$) the linewidth parameter T_2 can be calculated according to Eq. (24)¹⁵:

$$T_2^{-1} = \int_0^{t \gg \tau_c} \langle M, \frac{1}{2} | [\mathcal{H}_1^*(t), [\mathcal{H}_1^*(t - \tau), S^*]] | -\frac{1}{2}, M \rangle d\tau, \quad (24)$$

TABLE II. Spectral densities.

$j(\lambda q, \lambda' \bar{q}) = \frac{1}{2} \int_{-\infty}^{+\infty} \exp(- \tau / \tau_{2m}) \cdot \langle M, \frac{1}{2} [A_{\lambda}^{*(2, a)}, [A_{\lambda}^{*(2, \bar{a})}, S^+]] -\frac{1}{2}, M \rangle d\tau$
$j(G0, G0) = \frac{8}{3} \tau_{2m}$
$j(G1, G\bar{1}) = 2 \tau_{2m} / \{1 + [\omega_0(\vartheta) + \omega_I(\vartheta)]^2 \tau_{2m}^2\} \approx 2 \tau_{2m} / [1 + \omega_0(\vartheta)^2 \tau_{2m}^2]$
$j(G\bar{1}, G1) = 0$
$j(G0, A0) = \frac{8}{3} M \tau_{2m}$
$j(G1, A\bar{1}) = 2 M \tau_{2m} / \{1 + [\omega_0(\vartheta) + \omega_I(\vartheta)]^2 \tau_{2m}^2\} \approx 2 M \tau_{2m} / [1 + \omega_0(\vartheta)^2 \tau_{2m}^2]$
$j(G\bar{1}, A1) = 0$
$j(A0, G0) = \frac{8}{3} M \tau_{2m}$
$j(A1, G\bar{1}) = 2 M \tau_{2m} / \{1 + [\omega_0(\vartheta) + \omega_I(\vartheta)]^2 \tau_{2m}^2\} \approx 2 M \tau_{2m} / [1 + \omega_0(\vartheta)^2 \tau_{2m}^2]$
$j(A\bar{1}, G1) = 0$
$j(A0, A0) = \frac{8}{3} M^2 \tau_{2m} + \frac{1}{3} [I(I+1) - M^2] \tau_{2m} / \{1 + [\omega_0(\vartheta) + \omega_I(\vartheta)(1+M)]^2 \tau_{2m}^2\}$ $\approx \frac{8}{3} M^2 \tau_{2m} + \frac{1}{3} [I(I+1) - M^2] \tau_{2m} / [1 + \omega_0(\vartheta)^2 \tau_{2m}^2]$
$j(A1, A\bar{1}) = 2 M^2 \tau_{2m} / \{1 + [\omega_0(\vartheta) + \omega_I(\vartheta)]^2 \tau_{2m}^2\} + [I(I+1) - M^2] \tau_{2m} / [1 + \omega_I(\vartheta)^2 \tau_{2m}^2]$ $\approx 2 M^2 \tau_{2m} / \{1 + [\omega_0(\vartheta)]^2 \tau_{2m}^2\} + [I(I+1) - M^2] \tau_{2m} / [1 + \omega_I(\vartheta)^2 \tau_{2m}^2]$
$j(A\bar{1}, A1) = [I(I+1) - M^2] \tau_{2m} / [1 + \omega_I(\vartheta)^2 \tau_{2m}^2]$
$j(A2, A\bar{2}) = 2 [I(I+1) - M^2] \tau_{2m} / [1 + (\omega_0(\vartheta) + \omega_I(\vartheta))^2 \tau_{2m}^2] \approx 2 [I(I+1) - M^2] \tau_{2m} / [1 + \omega_0(\vartheta)^2 \tau_{2m}^2]$
$j(A\bar{2}, A2) = 0$

bilayer with the following chemical composition: Decanol (42 wt%), sodium decanoate (28 wt%), and water (30 wt%). The synthesis of the spin label and its motion in this bilayer phase have been described elsewhere.^{2,8,9} The concentration of spin label was approximately 0.1% by weight.

The theoretical discussion has made it clear that for constant order parameters S_i and constant correlation times τ_{2m} the spectrum should still be influenced profoundly by the macroscopic ordering of the bilayers. We have therefore investigated three different types of bilayer orientations: (1) a *planar* arrangement, induced by introducing the sample between two optically flat quartz plates (spacing 0.01 mm). The magnetic field was then applied parallel and perpendicular to the normal of the bilayer surfaces. The corresponding spectra are shown in Fig. 1(a) and Fig. 2(a), respectively. (2) A *cylindrical* ordering, obtained by sucking the liquid crystalline phase into thin capillaries (with a diameter of approximately 0.04 mm). This situation bears some resemblance to the arrangement of the lipid bilayers in the myelin sheath of a nerve fiber. In order to increase the signal intensity, approximately 15 capillaries of 2 cm length were bundled together. This package of parallel glass fibers was mounted on a teflon rod and inserted into the resonance cavity so that the magnetic field was perpendicular to the cylinder axes. Figure 3(a) shows the experimental spectrum.

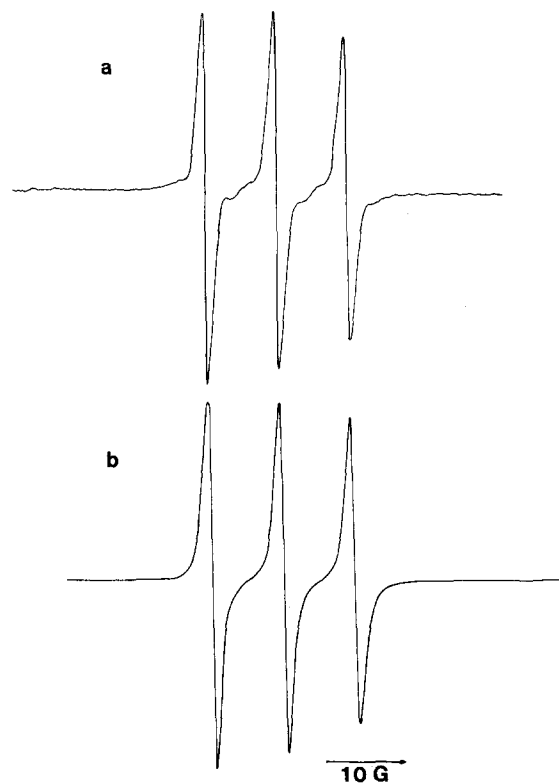


FIG. 2. Spin label I(13.2) in a *planar* oriented sample of bilayers at 21 °C. Magnetic field H_0 perpendicular to bilayer normal N . (a) experimental; (b) calculated spectrum. Same parameters as in Fig. 1.

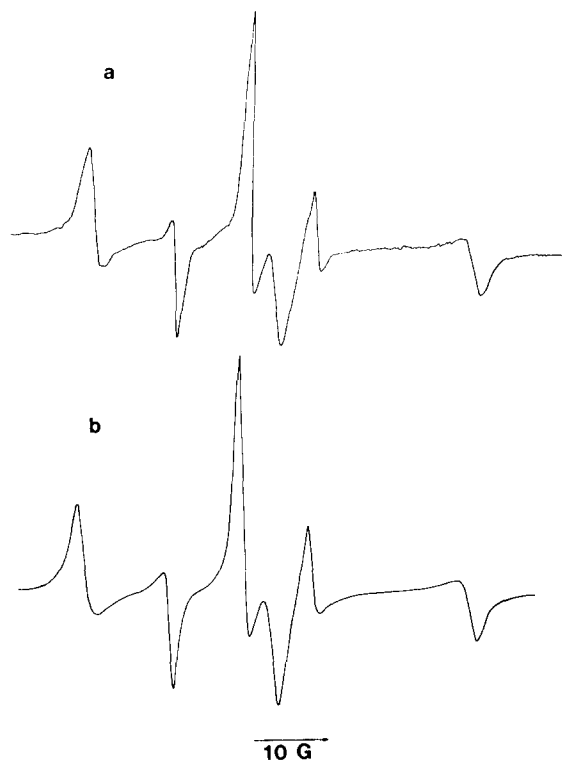


FIG. 3. Spin label I(13,2) in a *cylindrical* arrangement of bilayers at 21 °C. (a) experimental; (b) calculated spectrum. Same parameters as in Fig. 1.

(3) A *random* distribution of microcrystalline bilayer regions, obtained by filling the material into a sample tube of rather large diameter ($d \sim 4$ mm). In this case small contributions from cylindrical orientation on the glass surfaces can be neglected compared to the bulk material. The spectrum is given in Fig. 4(a).

The EPR measurements were performed with a Varian E-9 spectrometer (~ 9.35 GHz resonance frequency; 100 KHz modulation frequency) equipped with a variable temperature unit. The temperature was measured by means of a thermistor inserted into the center of the cavity. The absolute temperatures are accurate to $\pm 1^\circ$ C.

B. Simulated EPR Spectra

The simulation of the spin label spectra was based on Lorentzian line shapes. We have also tried Gaussian lines, but these yielded unsatisfying results.

The residual linewidth X was approximated by the following expression²²

$$X = b + c \cos^2 \beta' . \quad (31)$$

The influence of the parameters b and c on the spectrum was distinctly different from that of the order parameters or correlation times. There-

fore b and c could be fitted after the other parameters had been optimized.

No *a priori* predictions can be made about the ratio of the correlation times τ_{2m} . In the case of label I (13,2) the best fit of the experimental data was obtained by assigning $\tau_{22}/\tau_{20} = 1$.

The theoretical spectra depend strongly on the correct choice of the molecule fixed components of the g and T tensors. Since these are determined by the particular environment of the spin label, single crystal data must be corrected for the change of polarity in the bilayer membrane. In our simulation we have used the following parameters^{3,22-24}:

$$g_{xx} = 2.00872, \quad g_{yy} = 2.00616, \quad g_{zz} = 2.00270,$$

$$T_{xx} = 6.95 \text{ G}, \quad T_{yy} = 5.35 \text{ G}, \quad T_{zz} = 33.0 \text{ G} .$$

These numbers are within the limits of error of single crystal measurements. It should be pointed out here, that the spectra are especially sensitive to changes in $\delta a = \frac{1}{2}(T_{xx} - T_{yy})$.

Spectra calculated on the basis of the foregoing assumptions are shown in Figs. 1(b)–4(b). All spectra were computed with the parameters q_i and τ_{20} given in Fig. 1. In the cases of the random distribution and the planar oriented samples (both orientations) we have also measured the temperature dependence of the bilayer phase in the range of

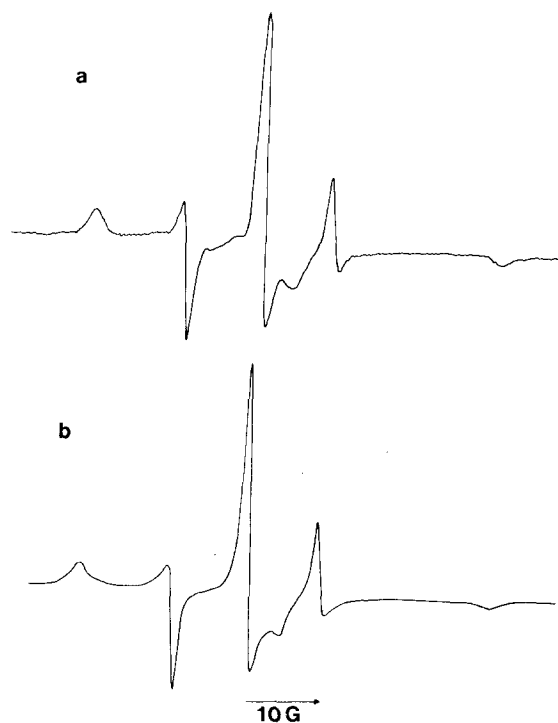


FIG. 4. Spin label I(13,2) in *random* distribution of bilayers at 21 °C. (a) experimental, (b) calculated spectrum. Same parameters as in Fig. 1.

TABLE III. Molecular parameters for the motion of spin label I(13, 2).

	Temperature (°C)			
	44.2	21.5	8.4	-2.4
Correlation time $\tau_{20} \cdot 10^{10}$ (sec)	1.0 ± 0.2	2.3 ± 0.2	4.0 ± 0.2	28 ± 1
Energy parameters				
- q_3 (cal/mole)	2880	2895	2901	3016
+ q_1 (cal/mole)	1669	1684	1709	1390
Order parameters				
S_3	0.61	0.64	0.66	0.70
S_1	-0.25	-0.27	-0.28	-0.25
S_2	-0.36	-0.37	-0.38	-0.45
Tilt angle	0°	0°	0°	18°
Spread angle	7.0°	3.4°	5.0°	6.0°
Isotropic hyperfine splitting constant a (G)	15.0	15.0	15.0	15.24

- 8 to +44 °C. The agreement between the computer fitted and the experimental spectra was practically perfect over this temperature interval. The results of our theoretical analysis for four different temperatures are summarized in Table III.

IV. DISCUSSION

Inspection of Figs. 1-4 shows excellent agreement of the simulated spin label spectra with the experimental spectra. It should be emphasized, however, that these results can only be obtained if the time-averaged and the dynamic parts of the spin Hamiltonian are evaluated in their most general form. Since a concise discussion of the influence of the various parameters on the appearance of the EPR spectrum is rather difficult, only a representative example shall be given here. In Fig. 5 the spectra of label I(13, 2) have been calculated with and without the inclusion of pseudosecular terms. It is obvious that pseudosecular terms contribute essentially to the over-all linewidth. In the case of Fig. 5(b) they account for approximately 30% of the intensity of the low and high field lines. Other significant terms which cannot be neglected are the δa parameter or the order parameter S_1 .

Let us now consider the structural and dynamic information contained in the line shape parameters. The correlation times τ_{20} ($=\tau_{22}$) are given in the first row of Table III. To the best of our knowledge this is the first accurate measurement of these correlation times in a lipid bilayer. In the range of 44 to 8 °C, τ_{20} shows an exactly exponential temperature dependence with an activation energy of $E_a = 6.84$ kcal mole⁻¹. The activation energies for the rotation of small molecules in the liquid state lie in the range of 2 to 4 kcal mole⁻¹, but for *n*-decane and tridecane, which have some structural relationship to the lipids of this bilayer, activation energies of 4.9 and 5.5 kcal mole⁻¹, respectively,

have been found close to the melting points of these paraffins.²⁵ Since the flexibility gradient in lipid bilayers is caused by rotational isomerizations of carbon-carbon bonds, it may be speculated that the observed activation energy and also the equality $\tau_{20} = \tau_{22}$ are due to the same mechanism.

The viscosity in the bilayer as sensed by the spin label can be estimated according to

$$\eta = 3kT\tau_{20}/4\pi r^3. \quad (32)$$

At the position of the NO group the radius of the spin labeled hydrocarbon chain is approximately 3.6 Å. With $\tau_{20} = 2.3 \times 10^{-10}$ sec we obtain a viscosity of $\eta = 4.8 \times 10^{-2}$ P at 21 °C. The diffusion constant for translational diffusion can also be calculated. For a decanol molecule of roughly 10 Å length and 2.5 Å radius this calculation yields $D_{\text{trans}} = 1.5 \times 10^{-6}$ cm² sec⁻¹.

We also incorporated label I(13, 2) into vesicles of dimyristoyl-*L*- α -lecithin. At 21.5 °C (this is 2 °C below the transition temperature of the gel \rightarrow liquid crystal phase transition) we found a correlation time $\tau_{20} = 2.5 \times 10^{-9}$ sec, which yields a viscosity of $\eta = 0.5$ P and a translational diffusion constant for the lipid molecules of $D_{\text{trans}} = 3.6 \times 10^{-8}$

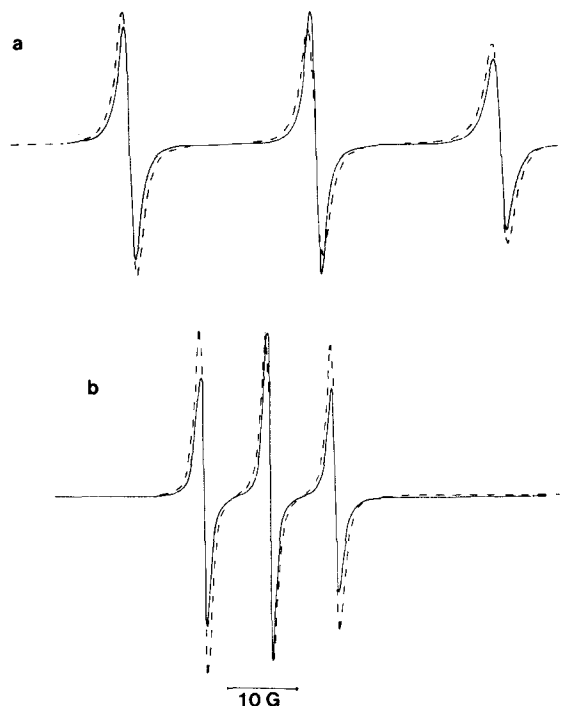


FIG. 5. Influence of the pseudosecular terms. Same parameters as in Fig. 1. Solid line: pseudosecular terms neglected. Broken line: pseudosecular terms included [corresponds to Fig. 1(b) and 2(b)]. Spectra calculated for (a) H_0 parallel to N ; (b) H_0 perpendicular to N .

$\text{cm}^2 \text{sec}^{-1}$. These results are in agreement with membrane viscosities and lateral diffusion constants as determined by entirely different methods.²⁸⁻²⁹

So far we have concentrated upon the dynamics of the bilayer molecules. Let us now discuss some structural aspects of the double layer. It should be pointed out that the line shape analysis confirms two experimental results already obtained previously.

(1) The order parameters S_1 and S_2 are not equal for label I(13.2),² and (2) the ordering of the lipid molecules in the bilayer is sensitive to changes in temperature.⁸ Molecular motions within the hydrocarbon chain increase gradually as the temperature increases, leading to smaller order parameters. However, the line shape analysis also sheds some new light on the flexibility problem. It is found in our investigation that within the range of existence of the liquid crystalline bilayer (45 to 8 °C) the temperature behavior of the order parameters S_3 and S_1 can be explained by *constant* energy parameters q_3 and q_1 . This gives us confidence not only that the probability function (5) is a reasonable

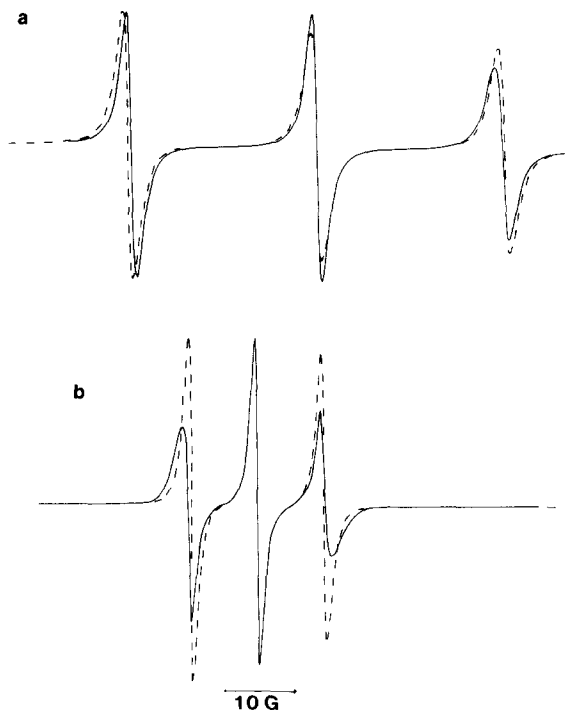


FIG. 6. Influence of the tilt angle. Same parameters as in Fig. 1. Solid line: $\bar{\beta}=10^\circ$, $\vartheta_0=5^\circ$. Broken line: $\bar{\beta}=0^\circ$, $\vartheta_0=3.4^\circ$ [corresponds to Figs. 1(b) and 2(b)]. Spectra calculated for (a) H_0 parallel to N ; (b) H_0 perpendicular to N .

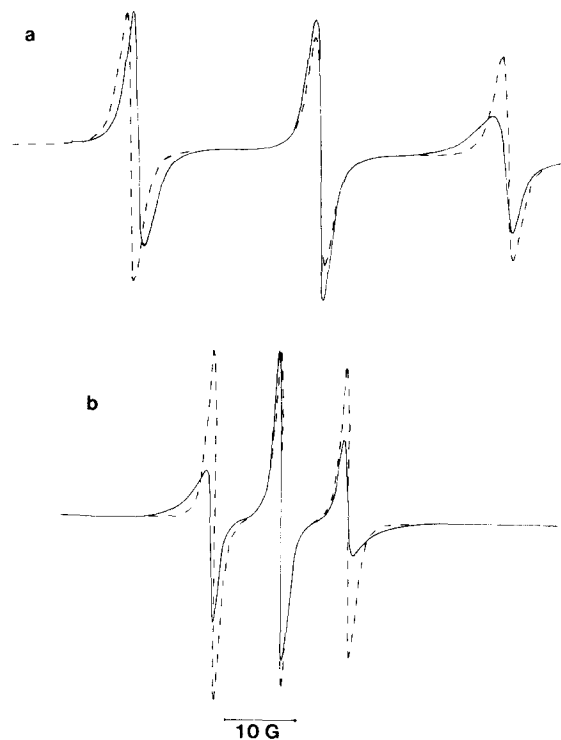


FIG. 7. Influence of the spread angle. Same parameters as in Fig. 1. Solid line: $\vartheta_0=15^\circ$. Broken line: $\vartheta_0=3.4^\circ$ [corresponds to Figs. 1(b) and 2(b)]. Spectra calculated for (a) H_0 parallel to N ; (b) H_0 perpendicular to N .

choice for lipid bilayers but also that the very precise knowledge of the energy parameters q may eventually open a way to a detailed statistical-mechanical understanding of bilayer structures. Even though this problem requires additional work, the description of the spin label spectra over a range of temperatures with only three parameters (q_1 , q_3 , and τ_{20}) is in itself a satisfying result.

The influence of tilt and spread angle on the spin label spectra of *planar* oriented bilayers is illustrated in Fig. 6 and Fig. 7, respectively. Both parameters cause asymmetric broadening of the outer lines and reduce their amplitude relative to the central line. Furthermore, the positions of the outer lines in the spectrum are shifted. If the magnetic field H_0 is parallel (perpendicular) to the axis N the calculated splittings in Fig. 6 and 7 are smaller (larger) compared to those calculated without tilt or spread angle. The latter spectra, which correspond to Figs. 1(b) and 2(b) are also included in Figs. 6 and 7. Comparing Figs. 1, 6, and 7 it is obvious that the occurrence of a tilt *or* spread angle in a lipid bilayer can be detected without difficulties. However, a very careful line shape analysis is required in order to distinguish between

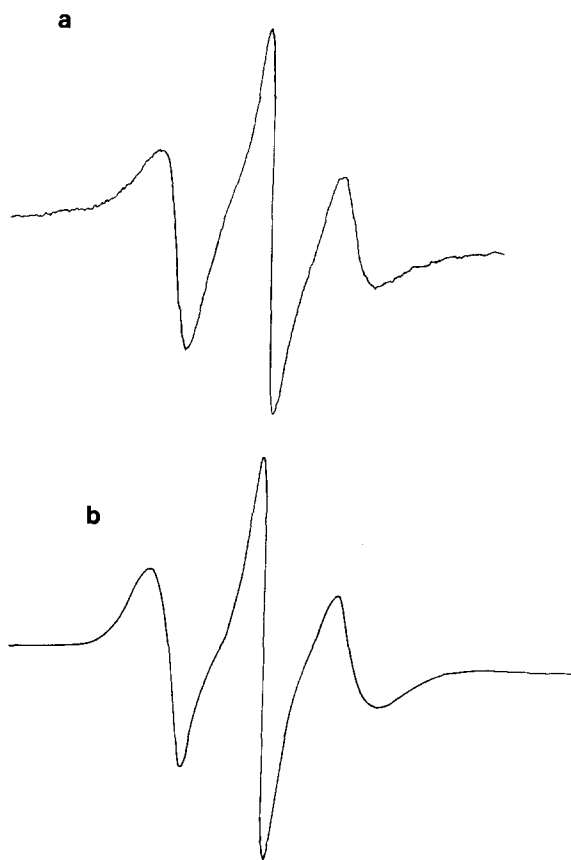


FIG. 8. Spin label I(13.2) in a planar oriented sample of bilayers at -8°C . Magnetic field H_0 perpendicular to bilayer normal N . (a) experimental spectrum, (b) calculated spectrum with $q_3 = 3.016$ kcal/mole, $q_1 = -1.390$ kcal/mole $\tau_{20} = 2.8 \cdot 10^{-9}$ sec, $\bar{\beta} = 18^{\circ}$, $\beta_0 = 6^{\circ}$. Residual linewidth: $b = 0.7$ G, $c = 0.8$ G.

the two parameters, since the spectral changes caused by a tilt or spread angle are quite similar.

In the sodium decanoate-decanol bilayer, zero tilt angle and only a small spread angle are found in the range of 44 to 8°C . The axis of motional averaging is therefore perpendicular to the bilayer surface. This situation changes if the temperature is lowered below 8°C . The spin label spectra then indicate that the liquid crystalline bilayer is transformed into a gel like structure with stiff hydrocarbon chains. This phase transition expresses itself not only in the energy parameters q_i and the significant increase of the correlation time τ_{20} , but it is also found that the fatty acid chains become tilted with respect to the bilayer normal. The tilt angle increases gradually with decreasing temperature. At -8°C the line shape analysis yields a tilt angle of 18° . The corresponding spectra are shown in Fig. 8. Since the correlation time is already at lower limit for the applica-

bility of our theory we cannot decide if the tilt angle further increases at even lower temperatures. Besides the spin label data we have also obtained calorimetric evidence for the observed phase transition.

In concluding, it should be emphasized that the theory of the spin label spectra presented here makes no assumptions about the specific motion of the spin label. In particular the order parameters should not be confused with any random walk model.³⁰ The order parameters are an unambiguous and most general description for the averaging of tensorial quantities and need not further molecular foundation.

The purpose of this report is to present a general theory for the simulation of spin label spectra in liquid crystalline bilayers. Only a few examples have been given to illustrate the applicability of the method. The results of a systematic study of phospholipids and other membrane models by means of a variety of different spin labeled fatty acids will be published elsewhere.

ACKNOWLEDGMENTS

The authors are much indebted to Dr. B. Kowert for valuable comments and to Dr. C. Scandella for critically reading the manuscript.

*To whom correspondence should be addressed.

¹D. Branton and D. N. Deamer, *Membrane Structure* (Springer, New York, 1972).

²J. Seelig, *J. Am. Chem. Soc.* **92**, 3881 (1970).

³W. L. Hubbell and H. M. McConnell, *J. Am. Chem. Soc.* **93**, 314 (1971).

⁴S. Rottem, W. L. Hubbell, L. Hayflick and H. M. McConnell, *Biochim. Biophys. Acta* **219**, 104 (1970).

⁵J. Seelig and W. Hasselbach, *Eur. J. Biochem.* **21**, 17 (1971).

⁶B. G. McFarland and H. M. McConnell, *Proc. Natl. Acad. Sci. USA* **68**, 1274 (1971).

⁷H. M. McConnell and B. G. McFarland, *Ann. N.Y. Acad. Sci.* **195**, 207 (1972).

⁸J. Seelig, *J. Am. Chem. Soc.* **93**, 5017 (1971).

⁹J. Seelig, H. Limacher, and P. Bader, *J. Am. Chem. Soc.* **94**, 6364 (1972).

¹⁰H. Limacher and J. Seelig, *Angew. Chem.* **84**, 950 (1972); *Angew. Chem. (Int. Ed. Engl.)* **11**, 920 (1972).

¹¹J. Seelig, F. Axel, and H. Limacher, "Molecular Architecture of Bilayer Membranes," *Ann. N.Y. Acad. Sci.* (to be published).

¹²A. Saupe, *Z. Naturforsch.* **19a**, 161 (1964).

¹³H. Lippmann, *Ann. Phys. (Leipz.)* **2**, 287 (1958).

¹⁴G. Meier and A. Saupe, *Molec. Crystals* **1**, 515 (1966).

¹⁵S. H. Glarum and J. H. Marshall, *J. Chem. Phys.* **46**, 55 (1967).

¹⁶P. Diehl and C. F. Schwerdtfeger, *Mol. Phys.* **17**, 423 (1969).

¹⁷J. H. Freed and G. K. Fraenkel, *J. Chem. Phys.* **39**, 326 (1963).

¹⁸J. H. Freed, *J. Chem. Phys.* **41**, 2077 (1964).

- ¹⁹S. H. Glarum and J. H. Marshall, *J. Chem. Phys.* **44**, 2884 (1966).
- ²⁰G. R. Luckhurst and A. Sanson, *Mol. Phys.* **24**, 1297 (1972).
- ²¹M. E. Rose, *Elementary Theory of Angular Momentum* (Wiley, New York, 1957).
- ²²S. A. Goldman, G. V. Bruno, C. F. Polnaszek, and J. H. Freed, *J. Chem. Phys.* **56**, 726 (1972).
- ²³C. L. Hamilton and H. M. McConnell, *Structural Chemistry and Molecular Biology*, edited by A. Rich and N. Davidson (Freeman, San Francisco, 1968).
- ²⁴L. J. Libertini and O. H. Griffith, *J. Chem. Phys.* **53**, 1359 (1970).
- ²⁵A. L. Buchachenko, A. M. Wasserman, and A. L. Kovarskii, *Int. J. Chem. Kinet.* **1**, 361 (1969).
- ²⁶R. A. Cone, *Nat. New Biol.* **236**, 39 (1972).
- ²⁷P. Deveaux and H. M. McConnell, *J. Am. Chem. Soc.* **94**, 4475 (1972).
- ²⁸E. Sackmann and H. Träuble, *J. Am. Chem. Soc.* **94**, 4482 (1972); *J. Am. Chem. Soc.* **94**, 4492 (1972); *J. Am. Chem. Soc.* **94**, 4499 (1972).
- ²⁹C. J. Scandella, P. Deveaux, and H. M. McConnell, *Proc. Natl. Acad. Sci. USA* **69**, 2056 (1972).
- ³⁰P. Jost, L. J. Libertini, V. C. Hebert, and O. H. Griffith, *J. Mol. Biol.* **59**, 77 (1971).