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Aqua regia extractable selenium contents of some Scottish topsoils measured by ICP-MS and the relationship with mineral and organic soil components

**RUNNING TITLE:** Selenium in Scottish topsoils

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**Keywords**: aqua regia extraction; inductively coupled plasma - mass spectrometry; selenium; soil; soil carbon; soil minerals

#### Abstract

**BACKGROUND:** To provide information concerning the geographical distribution of selenium (Se) in the soils of Scotland, we analysed 47 arable soils selected on the basis of their parent rock, which were expected to have relatively high, low or unclassified Se contents. To investigate relationships between the actual minerals in the soils and the aqua-regia extractable Se content of the soil, soil minerals were quantified by X-ray diffraction.

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**RESULTS:** The aqua-regia extractable Se contents of the soils were between 0.19 and 1.46 mg kg<sup>-1</sup>. No simple correlation between the aqua-regia extractable Se contents of the soil and the parent rock classification estimated by soil survey was evident. Partial least squares analysis revealed that the aqua-regia extractable Se content of the soils was positively related to loss on ignition (LOI) and negatively related to the K–feldspar content with other mineral being less important.

**CONCLUSION:** The Se content of arable topsoils from Scotland is more related to LOI or carbon content with parent material being less important.

# INTRODUCTION

Following the discovery of Se as a factor against liver necrosis in rats <sup>1</sup> and of the role of Se in glutathione peroxidase <sup>2</sup> the element is widely recognised as being essential for human and animal health.<sup>3</sup> Possible links between Se status and the occurrence of health disorders in humans have been reviewed.<sup>4-6</sup> Intakes of Se by the UK population have declined from a mean of 60  $\mu$ g Se day <sup>-1</sup> in 1974 to 30 - 40  $\mu$ g Se day <sup>-1</sup> in recent times, primarily due to the replacement of American milling wheat with quantities of UK sourced grain with lower Se content.<sup>7-11</sup> Selenium intake by the population is also relatively low in Scotland <sup>12,13</sup> and there is increasing interest in the provision of information leading to a better understanding of Se supply, especially in situations where people and animals rely heavily on locally produced foods as a source of essential elements.

Selenium is a chalcophile element and concentrations of Se in evolved magmatic rocks, such as granite, are relatively low. During the weathering of rock, Se is accumulated in fine sediments.<sup>14,15</sup> Selenium can also become concentrated in organic

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matter and organic-rich sediments. Overall, the Se contents of soils formed from andesitic, basaltic or argillaceous materials are expected to be richer in Se than soils formed from granites, sandstones, rhyolites, mica schist or non-volcanic greywacke.<sup>16</sup> The influence of parent material on the Se content of soils has been demonstrated in England and Wales.<sup>17</sup> Scottish soils are distinguished from those in England and elsewhere in that they have geologically diverse parent material and often contain large amounts of organic C. Yet, relatively little is known <sup>16,18-20</sup> about the pattern of distribution of Se in the Scottish landscape.

Selenium can be measured by inductively coupled plasma – mass spectrometry (ICP-MS) with an Ar based plasma and the ICP-MS method has advantages over others with regard to ease of use and multi-element capability. Selenium has six naturally occurring isotopes; <sup>74</sup>Se (0.87 %), <sup>76</sup>Se (9.02 %), <sup>77</sup>Se (7.58 %), <sup>78</sup>Se (23.52 %),  ${}^{80}$ Se (49.82 %) and  ${}^{82}$ Se (9.19 %).<sup>21</sup> The use of  ${}^{76}$ Se,  ${}^{78}$ Se and  ${}^{80}$ Se suffer interferences from Ar dimmers e.g. <sup>40</sup>Ar<sub>2</sub><sup>+</sup>, whereas, the others may suffer from a range of mixed element ion interferences.<sup>22,23</sup> Cool plasma conditions reduce Ar<sub>2</sub>-based interference but are not suited for the analysis of Se, which has a large ionisation potential. Collision / reaction cell methodology using a mixture of H<sub>2</sub> and He has been developed to simultaneously remove <sup>40</sup>Ar<sup>40</sup>Ar<sup>+</sup> and <sup>40</sup>Ar<sup>35</sup>Cl<sup>+</sup> interferences in the determination of <sup>80</sup>Se and <sup>75</sup>As in chloride-rich matrices and mineral waters.<sup>24</sup> Aqua regia digests of sediments have been analysed by ICP-MS using <sup>82</sup>Se: interference from  ${}^{82}$ Kr<sup>+</sup> (Ar contaminant),  ${}^{34}$ S ${}^{16}$ O $_{3}$ <sup>+</sup> and  ${}^{66}$ Zn ${}^{16}$ O<sup>+</sup> were found to be insignificant or eliminated during blank subtraction. Bromine interference (<sup>81</sup>Br<sup>1</sup>H<sup>+</sup>) was eliminated through oxidation of Br<sup>-</sup> by NOCl, present in agua regia, to volatile Br<sub>2</sub>.<sup>25</sup> The use of H<sub>2</sub> in the reaction cell has advantages over other gases in that H<sub>2</sub> reacts little with Se and causes minimal scattering. Naturally occurring Ar gas is a mixture of three

isotopes; <sup>36</sup>Ar (0.34 %), <sup>38</sup>Ar (0.06 %), and <sup>40</sup>Ar (99.60 %).<sup>21</sup> For ICP-MS using an Ar based plasma, measurement of <sup>80</sup>Se with interference from <sup>40</sup>Ar<sub>2</sub><sup>+</sup> is best avoided. The use of the less abundant <sup>78</sup>Se with interference from <sup>38</sup>Ar<sup>40</sup>Ar<sup>+</sup> is preferred and is recommended by the manufacturer of the instrument we used. The use ICP-MS with H<sub>2</sub> in the reaction cell for the analysis of Se in nitric / perchloric acid digests of soil using both <sup>78</sup>Se and <sup>80</sup>Se has been reported <sup>11</sup> but no details related to the choice of isotopes for analytical purposes were presented.

Our aims were: (1) to demonstrate that the analysis of the Se in aqua regia digests of soils could be satisfactorily achieved by ICP-MS using <sup>78</sup>Se with  $H_2$  as the reaction cell gas under routine, multi-element conditions, and (2), to investigate relationships between soil Se content of arable soils and other soil components, especially soil minerals and C.

#### MATERIAL AND METHODS

#### Soils

The information concerning the relative Se contents of different rock types<sup>16</sup> discussed in the introduction, was used to identify areas of soils (Fig. 1) with associations (i.e. parent material) expected to have relatively "high" or "low" Se status, the remainder being "unclassified" because there was insufficient information. To test the validity of the hypothesis that soils formed from these different parent materials would indeed have different Se contents, we selected 47 soils from an archive of 981 air-dried, 2-mm sieved topsoils, which had been taken across Scotland on a 10-km grid basis as part of a national inventory.<sup>26</sup> Our selection of samples was based firstly on a restriction to soils which had been ploughed (224 samples), and secondly, to soils from associations

with extensive coverage (Table 1) within which we selected examples of soils representing the major soil groups (alluvial soils, brown earths, podzols and gleys). The particle size distribution of the soil was determined by a hydrometer method after dispersion in 0.02 M sodium hydroxide. Loss on ignition (LOI) was determined by recording the weight loss after heating the soil at 900 °C for 2 h. Soil pH was measured in 0.01 M CaCl<sub>2</sub> using a 1:3 (mass to volume) soil to solution ratio. For the determination of Se and C contents, the soil was ground in an agate ball mill to a particle size < 150  $\mu$ m. The total C content of the soil was determined by combustion in a Hewlett-Packard CHN analyzer.

## Mineralogical composition of soils

To provide information about the mineralogical composition of the soils they were quantitatively analysed by X-ray diffraction using an X'Pert Pro diffractometer (PANalytical, Almelo, The Netherlands) with an X'Celerator position sensitive detector. Randomly orientated samples of the soil for X-ray analysis were prepared by spray-drying.<sup>27</sup> Analysis of the diffractgrams was performed by a full-pattern fitting reference intensity ratio method.<sup>28</sup>

#### Aqua regia extraction of soils

The soil extraction with aqua regia was based on the British standard reflux method, ISO 11466.<sup>29</sup> Each determination used 2 g of the finely-ground soil and high purity "Aristar" grade hydrochloric and nitric acids from VWR International Ltd, Poole, UK. A small amount of gold (III) chloride was added to each of the aqua regia digests to preserve them <sup>30</sup> for future Hg analysis (not reported here).

#### **ICP-MS analysis of soil extracts**

We used an Agilent 7500ce instrument with an octopole reaction cell employing H<sub>2</sub>, micro flow nebuliser and integrated autosampler to measure <sup>78</sup>Se<sup>+</sup>. Instrument performance was optimised using a solution containing 10  $\mu$ g l<sup>-1</sup> each of Li, Y, Tl and Ce. Maximum sensitivity was achieved by monitoring <sup>7</sup>Li, <sup>89</sup>Y and <sup>207</sup>Tl signals and minimum oxide formation by monitoring the <sup>140</sup>Ce:<sup>16</sup>O<sup>140</sup>Ce ratio. For calibration we used a 1g Se l<sup>-1</sup> solution diluted to provide a range of working standards containing up to 1 mg Se l<sup>-1</sup> in a matrix of 2.5 M hydrochloric acid and 1.4 M nitric acid. To minimise damage to the Ni skimmer cones, sample extracts and standard solutions were diluted by a factor of ten before entering the plasma. To correct for instrument instabilities a solution of 500  $\mu$ g Ge l<sup>-1</sup> was aspirated and the signal from <sup>72</sup>Ge monitored.

#### **Quality control**

To provide a check on the quality of the ICP-MS data for Se we analysed the aqua regia extracts by an independent method. An aliquot of the aqua regia digest containing up to 0.1  $\mu$ g Se was placed in a reflux tube, perchloric acid was added, and the mixture heated to remove nitric acid. The solution remaining was treated with hydrochloric acid to reduce Se(VI) to Se(IV) and the Se concentration of the extract determined fluorimetrically.<sup>31</sup>

To provide a check on the accuracy of the methods we used reference soils NCS DC 73387 and NCS DC 73389 from the Laboratory of the Government Chemist (LGC), Promochem, Teddington, UK. Additionally, we measured aqua regia extractable Se in a soil sample (C3a) also provided by LGC as part of their contaminated land proficiency testing scheme. Soil with certified aqua regia extractable Se content was

not available. The performance of the fluorimetric method was also assessed by analysis of IAEA-A-13 dried blood. The NCS reference soils and the dried blood were analysed in triplicate, whereas, reference soil C3a was analysed six-times.

#### **Statistical analysis**

Statistical analysis was performed using Minitab 15. Unless otherwise noted, error terms presented are standard deviations. Partial least squares (PLS) analysis provides a non-mechanistic approach to investigate mathematical relationships between soil properties. The technique has most often been applied to spectroscopic data but has been used for prediction of soil properties from non-spectroscopic data and is particularly useful for analysis of data sets with collinear information and where there is a large number of variables.<sup>32</sup>

# RESULTS

#### Selenium content of inventory soils

The Scottish soils (n = 47) were found to have Se contents ranging from 0.19 to 1.46 mg kg<sup>-1</sup> measured by ICP-MS (Table 2). The mean value and median values were 0.630 and 0.577 mg Se kg<sup>-1</sup>, respectively. There was considerable variation between replicate values for soils with Se values near to the detection limit. A comparison of the ICP-MS results of analysis with those from fluorimetry is presented in Fig. 2. Statistical analysis of the data by the paired t-test led to the acceptance of the hypothesis that the mean difference between the results of the ICP-MS and those by fluorimetry was significantly different from zero (at p = 0.05) and that the ICP-MS method, overall, gave slightly greater values. There were no significant differences (p

= 0.05) between the mean values of Se in the parent material derived categories of high (0.630 mg Se kg<sup>-1</sup>), low (0.604 mg Se kg<sup>-1</sup>) and the unclassified (0.658 mg Se kg<sup>-1</sup>) soils. Similarly, there was no significant difference (p = 0.05) between Se concentration in soil divided into major soil groups; the average concentration of Se in alluvial soils was 0.603 (n = 5), brown earths 0.619 (n = 21), gleys 0.551 (n = 9) and podzols 0.720 (n = 12) mg kg<sup>-1</sup>. A more general significant link between the Se content of the soils and their C content ( $R^2 = 0.578$ ) or LOI, ( $R^2 = 0.700$ ) was apparent from a plot of soil C content or LOI against Se concentration in the soil (Fig. 3), both relationships being statistically significant (p = < 0.001).

#### Validation of analytical methodology to determine Se

The aqua regia extractable Se contents of certified reference soils were found to be  $1.21 \pm 0.25 \text{ mg kg}^{-1}$  for NCS DC 73387 (certified total  $1.34 \pm 0.17 \text{ mg kg}^{-1}$ ) and  $0.16 \pm 0.08 \text{ mg kg}^{-1}$  for NCS DC 73389 (certified total  $0.10 \pm 0.01 \text{ mg kg}^{-1}$ ). The value found for the aqua regia extractable Se content of the proficiency testing "Contest" soil was  $1.49 \pm 0.02 \text{ mg kg}^{-1}$  (assigned value  $1.49 \text{ mg Se kg}^{-1}$ ). The Se content of the IAEA dried blood determined by fluorimetry gave a value of  $0.22 \pm 0.009 \text{ mg kg}^{-1}$  (certified value  $0.24 \pm 0.009 \text{ mg kg}^{-1}$ ). The Se content of the aqua regia digests of the certified soils determined by fluorimetry were  $1.34 \pm 0.12 \text{ mg kg}^{-1}$  for NCS DC 73387 and  $0.08 \pm 0.00 \text{ mg kg}^{-1}$  for NCS DC 73389. The limit of detection for Se analysis by ICP-MS estimated as  $3 \times$  standard deviation of ten intraset blank readings was  $1.4 \text{ µg Se I}^{-1}$ , which is equivalent to the concentration of a digest derived from a soil containing  $0.07 \text{ mg Se kg}^{-1}$ .

#### Mineralogical composition determined by XRD

The minerals present in the soils included quartz with lesser amounts of K-feldspar, plagioclase and amphibole; for simplicity the minerals have been grouped together into classes (Table 2). The amount of clay-sized material in the soils measured by classical sedimentation methods presented in Table 1 was generally less than the amount of phyllosilicates measured by X-ray diffraction. This difference has been observed in many other samples analysed in the same way in our laboratory and is indicative of the presence of aggregates in the chemically dispersed samples.

#### **Correlations between Se and other soil components**

Partial least squares analysis of the data with the Se content of the soils determined by ICP-MS as the response variate and the mineralogical phases shown in Table 2 and the percentage LOI or C as the predictors indicated that the aqua regia extractable Se content of the soils was best explained (optimal with regard to cross validation) by a 3component model which had an  $R^2$  value of 0.79 and a predicted  $R^2$  value of 0.71 by cross validation. The PLS response plot indicated that the model fitted the data well and that there was no extreme leverage. The PLS coefficient plot (Fig. 4a) showed that the most important components were LOI (standardised regression coefficient 0.68) and K-feldspar (standardised regression coefficient - 0.26) with data for other mineral phases being less important and varying in sign. The loading plot (Fig. 4b) indicated that the influences of LOI and K-feldspar on Se content of the soil were relatively important and opposite in direction within component 1 of the model, which contains terms related to all the predictors.

We were unable measure S in the aqua regia digests because of isobaric interferences but found significant (p < 0.05), linear correlations with other elements including As (r = 0.426), Cd (r = 0.623), Cr (r = 0.347), Cu (r = 0.348), Fe (r = 0.348), P (r = 0.453) and Pb (r = 0.449).

The map (Fig. 5) shows the predicted distribution of aqua regia extractable Se in arable soils based on the relationship, aqua regia extractable soil S concentration (mg kg<sup>-1</sup>) =  $0.0381 \times \text{percent LOI} + 0.1655$  (Fig. 3b). The data for LOI was from the National Soils Inventory of Scotland, 10–km grid positions and was restricted to mineral soils (C content < 25 %).

#### DISCUSSION

#### Aqua regia extraction

The range of values we obtained for the Se contents of the arable soils from Scotland (0.19 to 1.46; median 0.58 mg kg<sup>-1</sup>) is consistent with the range of previously reported values (0.11 to 1.59; median 0.69 mg kg<sup>-1</sup>).<sup>16,18,20</sup> Lesser values, ranging from 0.02 to 0.36 mg Se kg<sup>-1</sup> have been reported <sup>19</sup> for a group of ten Scottish soils analysed by molecular fluorescence but Se may have been lost during preliminary ashing of the soil. In comparison, soils from 34 farms in Wales and the Midlands of England were reported <sup>33</sup> to contain 0.2 to 1.8 mg Se kg<sup>-1</sup>, median 0.6 mg Se kg<sup>-1</sup>. Analysis of soil and stream sediments from England and Wales by the British Geological Survey have shown that the total Se ranged between 0.1 and 4 mg Se kg<sup>-1</sup> and that > 95 % of them contained < 1 mg Se kg<sup>-1</sup>.<sup>10</sup>

The data for the analysis of the standard reference or test materials agreed well with the given values providing confidence in the accuracy of the data. The considerable variation in Se concentration values measured by ICP-MS especially those for soils containing < 0.4 mg Se kg<sup>-1</sup> is a reflection of closeness of values to the limit of the method. The use of <sup>78</sup>Se (23.52 % natural abundance) was recommended by the

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instrument manufacturer for the analysis of Se and we used their standard operating conditions. Optimising the tune conditions for this mass alone and adjustment of  $H_2$  pressure may provide improvement. The nitric and hydrochloric acids used to prepare the aqua regia were of a high-purity, analytical grade but it is possible that further purification of the acids may result in a lower quantification limit for Se in soils.

A study involving sequential extraction of Se from four Scottish soils using 0.25 M potassium chloride (soluble fraction), 0.1 M potassium dihydrogen phosphate (ligandexchangeable), 4 M hydrochloric acid (acid extractable), potassium chlorate/ hydrochloric acid (oxidiseable fraction) and hydrofluoric acid (residual fraction) has shown that most of the Se (62%) was associated with the residual hydrofluoric acid soluble fraction.<sup>34</sup> It can be assumed that aqua regia will extract Se associated with the soluble, ligand exchangeable and oxidiseable fractions but the extent to which aqua regia extracts Se from the residual fraction is unknown. Near quantitative recovery of Se from reference soils RTC-CRM 023-050 (117 mg Se kg<sup>-1</sup>) and RTC-CRM 025-050 (518 mg Se kg<sup>-1</sup>) by aqua regia extraction has been reported.<sup>35</sup> However, these soils are from contaminated sites and the bulk of the Se may be present in easily accessible forms. The recovery of Se from certified reference materials [sediment 2704 (1.16 mg Se kg<sup>-1</sup>), sandy soil 2709 (1.56 mg Se kg<sup>-1</sup>), and contaminated soil 2711 (1.54 mg Se kg<sup>-1</sup>)] by open beaker agua regia digestion was  $112 \pm 24$ ,  $96 \pm 8.0$  and  $85 \pm 3.5$  %, respectively.<sup>36</sup> Variable recoveries of other aqua regia extractable elements from soils in Scotland are reported but refluxing is generally more effective than open digestion.<sup>37</sup> We have no information about the proportion of the total Se dissolved by aqua regia from Scottish soils, which tend to be relatively rich in organic matter. Boiling aqua regia effectively decomposes soil organic matter but dissolves only some mineral phases including iron oxides and iron-rich and trioctahedral phyllosilicates, which may

be linked with Se. Although the total amount of an element in soil can be a poor indicator of its bioavailability to plants, there are conflicting reports regarding the suitability of different extractants for the determination of plant available Se in soil.<sup>14</sup>

#### **Relationships between Se and other soil components**

No simple correlation between the aqua regia extractable Se contents of the soil and the parent material as assessed by traditional soil survey in the field <sup>26</sup> was evident. This could be explained by a lack of contrast in the nature of the parent materials, or a lack of resolution in the classification themselves. Positive relationships between the amounts of Se and soil organic matter have been reported.<sup>38,39</sup> For the soils studied by us, the C content of the soil was a more important factor in determining the aqua regia extractable Se content of soil than the parent rock type as assessed by soil surveyors in the field. There is also the possibility of inputs of Se from the atmosphere, fertilisers or organic matter, or by loss or gain of soil forming minerals by physical effect such as outwashing.<sup>40</sup> A comparatively high concentration of Se in Norwegian coastal areas was found to be associated with atmospheric inputs and a high content of soil organic matter.<sup>41</sup> Other studies have also shown that Se in soils is can be associated with organic matter although the mechanism of interaction and the chemical species present in the complex is unclear because of the presence of mixed organic matter and mineral phases.<sup>42</sup>

The C content of soils from the National Inventory of Scotland (10-km grid) follows a bimodal distribution reflecting the composition of mineral soils (mean 5.5 % C for 219 samples) and organic soils (mean 43.6 % C for 414 samples).<sup>43</sup> In the Scottish context, soils with a C content < 25% are considered to be mineral soils, whereas, soils with a C content of > 25% are considered to be organic soils. The soils studied here in

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relation to Se content fall into the mineral soil category and the Se content of organic soils such as peat or of soils with peaty surface horizons is not part of the study.

The improved linear correlation between the Se content and LOI of the soils ( $R^2 = 0.847$ ) compared to Se content and percentage C content of the soils ( $R^2 = 0.578$ ) may be explained by the inclusion of the weight loss due to structural dehydroxylation in the LOI measurement. For soil minerals, structural dehydroxylation on heating is largely a phenomenon associated with phyllosilicates, which on heating first loose interlayer water and then water from structural hydroxyl groups as the temperature is increased. Clay minerals are important for the fixation of Se<sup>44</sup> and the improved correlation may in part also reflect this phenomenon. However, the value of correlation coefficient between Se content and soil C or LOI (Fig 3) is dependent on the variation associated with the measurement of C or LOI. The values for C and LOI taken from the database of the Soil Survey of Scotland were single values and the associated error terms were not available.

The apparent link between Se and K-feldspar content of the soil is more difficult to explain. Nonetheless, feldspars are a major component of granite and other evolved igneous rocks. The negative link between Se and feldspar content could reflect the fact that rock with large K-feldspar contents are not of sedimentary origin and therefore less likely to contain S bearing minerals and associated Se components. Although we could not measure S in the aqua regia digests we did observe significant (p < 0.05), positive correlations with other elements such as Cu that are associated with S.<sup>14</sup>

For arable topsoils in Scotland it is possible to construct crude maps of Se distribution from soil carbon contents or LOI alone with mineralogical information adding improvements. At present this refinement is not possible because a complete set of quantitative mineralogical data for the arable soil of Scotland is not available.

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#### REFERENCES

- Schwarz K and Foltz M, Selenium as an integral part of Factor 3 against dietary necrotic liver degeneration. *J Am Chem Soc* 79: 3292-3293 (1957).
- Rotruck JT, Pope AL, Ganther HE, Swanson AB, Hafeman, DG and Hoekstra WG, Selenium: biochemical role as a component of glutathione peroxidase. <u>Science</u> 179: 588-590 (1973).
- Foster LH and Sumar S, Selenium in health and disease: A review. <u>*Crit Rev</u> <u>Food Sci Nutr</u> 37: 211-228 (1997).
  </u>*
- Combs GF (Jr.), Selenium in global food systems. <u>Br J Nutr</u> 85: 517-547 (2001).
- Beckett GJ and Arthur JR, Selenium and endocrine systems. <u>J Endocrinol</u> 184: 455-465 (2005).
- Rayman MP, Selenium in cancer prevention: a review of the evidence and mechanism of action. <u>*Proc Nutr Soc*</u> 64: 527-542 (2005).
- Ysart G, Miller P, Crews H, Robb P, Baxter M, de L'Argy C, Lofthouse S, Sargent C and Harrison N, Dietary exposure estimates of 30 elements from the UK Total Diet Study. *Food Addit Contam* 16: 391-403 (1999).

- World Health Organisation and Food and Agriculture Organization of the United Nations, <u>Human Vitamin and Mineral Requirements</u>, Chapter 10, Selenium. WHO, Rome, pp.194-216 (2002).
- Adams ML, Lombi E, Zhao FJ and McGrath SP, Evidence of low selenium concentrations in UK bread-making wheat grain. <u>J Sci Food Agric</u> 82: 1160-1165 (2002).
- Broadley MR, White PJ, Bryson RJ, Meacham MC, Bowen HC, Johnson SE, Hawkesford MJ, McGrath SP, Zhao F-J, Breward N, Harriman M and Tucker M, Biofortification of UK food crops with selenium. <u>*P Nutr Soc*</u> 65: 169-181 (2006).
- 11. Fan M-S, Zhao F-J, Poulton PR and McGrath SP, Historical changes in the concentrations of selenium in soil and wheat grain from the Broadbalk experiment over the last 160 years. <u>Sci Total Environ</u> 389: 532-538 (2008).
- Barclay MNI, MacPherson A and Dixon J, Selenium content of a range of UK foods. *J Food Comp Analysis* 8: 307-318 (1995).
- Shortt CT, Duthie GG, Robertson JD, Morrice PC, Nicol F and Arthur JR, Selenium status of a group of Scottish adults. *Eur J Clin Nutr* 51: 400-404 (1997).
- Adriano DC, <u>Trace Elements in Terrestrial Environments: Biochemistry</u>, <u>Bioavailability, and Risks of Metals</u>, Second Edition. Springer-Verlag, New York, pp. 707-758 (2001).
- Kabata-Pendias A and Pendias H, <u>Trace Elements in Soils and Plants</u>. CRC Press, Boca Raton, Florida, pp. 185-208 (1984).
- Ure AM and Berrow ML, The elemental constituents of soils, in <u>Environmental</u> <u>Chemistry, Volume 2</u>. A Specialist Periodical Report, The Royal Society of

Chemistry, London, pp. 94-204 (1982).

- 17. Thornton I, Kinniburgh DG, Abrahams P, Gaye C, Rundle S, Pullen G and Smith C, The influence of parent material on the selenium content of British soils. *J Sci Food Agric* 34: 50-51 (1983).
- Forbes S, Bound GP and West TS, Determination of selenium in soils and plants by differential pulse cathodic-stripping voltammetry. <u>*Talanta*</u> 26: 473-477 (1979).
- Ure AM, Bacon JR, Berrow ML and Watt JJ, The total trace element content of some Scottish soils by spark source mass spectrometry. <u>*Geoderma*</u> 22: 1-23 (1979).
- MacLeod F, McGaw BA and Shand CA, Stable isotope dilution mass spectrometry for determining total selenium levels in plants, soils and sewage sludges. <u>*Talanta*</u> 43: 1091-1098 (1996).
- Rosman KJR and Taylor PDP, Isotopic composition of the elements. <u>Pure Appl</u> Chem 70: 217-235 (1998).
- 22. May TW and Wiedmeyer RH, A table of polyatomic interferences in ICP-MS. <u>At Spectrosc</u> 19: 150-155 (1998).
- 23. Reyes LH, Gayón JMM, Alonso JIG and Sanz-Medel A, Determination of selenium in biological materials by isotope dilution analysis with an octapole reaction system ICP-MS. <u>J Anal At Spectrom</u> 18: 11-16 (2003).
- 24. Darrouzès J, Bueno M, Lespès G, Holeman M and Potin-Gautier M, Optimisation of ICPMS collision/reaction cell conditions for the simultaneous removal of argon based interferences of arsenic and selenium in water samples. <u>Talanta</u> 71: 2080-2084 (2007).
- 25. Pinho J, Canário J, Cesário R and Vale C, A rapid acid digestion method with

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ICP-MS detection for the determination of selenium in dry sediments. <u>Anal</u> <u>Chim Acta</u> **551:** 207-212 (2005).

- 26. Macaulay Institute for Soil Research, <u>Organization and Methods of the</u> <u>1:250000 Soil Survey of Scotland</u>. The Macaulay Institute for Soil Research, Aberdeen, UK (1984).
- 27. Hillier S, Use of an air brush to spray dry samples for X-ray powder diffraction.*Clay Miner* 34: 127-135 (1999).
- 28. Omotoso O, McCarty DK, Hillier S and Kleeberg R, Some successful approaches to quantitative mineral analysis as revealed by the 3rd Reynolds Cup contest. <u>*Clays Clay Miner*</u> 54: 748-760 (2006).
- 29. ISO (International Organization for Standardization), ISO 1146:1995, Soil Quality, <u>Extraction of Trace Elements Soluble in Aqua Regia</u>, BSI, London (1995).
- 30. Zhu X and Alexandratos SD, Determination of trace levels of mercury in aqueous solutions by inductively coupled plasma atomic emission spectrometry: Elimination of the 'memory effect'. *Microchem J* 86: 37-41 (2007).
- Olson OE, Fluorometric analysis of selenium in plants. J Assoc Official Anal <u>Chemists</u> 52: 627-634 (1969).
- 32. Rigol A, Camps M, de Juan A, Rauret G and Vidal M, Multivariate soft-modeling to predict radiocesium soil-to-plant transfer. *Environ Sci Technol* 42: 4029-4036 (2008).
- 33. Archer FC, Trace elements in soils in England and Wales, in <u>Reference Book</u> <u>326, Inorganic Pollution and Agriculture</u>, Proceedings of a conference organised by the Agricultural Development and Advisory Service, April 1977, Ministry of Agriculture, Fisheries and Food, HMSO, London, pp. 184-190

(1980).

- 34. MacLeod F, McGaw BA and Shand CA, Sequential extraction of selenium from four Scottish soils and a sewage sludge. <u>Commun Soil Sci Plant Anal</u> 29: 523-534 (1998).
- 35. González-Nieto J, López-Sanchez JF and Rubio R, Comparison of chemical modifiers for selenium determination in soil aqua regia extracts by ZETAAS. <u>*Talanta*</u> 69: 1118-1122 (2006).
- 36. Chen M and Ma LQ, Comparison of three aqua regia digestion methods for twenty Florida soils. <u>Soil Sci Soc Am J</u> 65: 491-499 (2001).
- Berrow ML and Stein WM, Extraction of metals from soils and sewage sludges by refluxing with aqua regia. <u>Analyst</u> 108: 277-285 (1983).
- 38. Johnsson L, <u>Selenium in Swedish Soils. Factors Influencing Soil Content and</u> <u>Plant Uptake</u>. PhD Thesis, Swedish University of Agricultural Sciences, Department of Soil Sciences, Uppsala, Sweden (1992).
- Eriksson J, Andersson A and Andersson R, <u>Current Status of Swedish Arable</u> <u>Soils</u>. Report 4778, Naturvårdsverket Förlag, Stockholm (1997).
- 40. Haygarth PM, Global importance and global cycling of selenium, in <u>Selenium</u> <u>in the Environment</u>, ed. by Frankenberger WT (Jr.) and Benson S. Marcel Dekker, New York, pp. 1-27 (1994).
- 41. Wu XF and Låg J, Selenium in Norwegian farmland soils, <u>Acta Agr Scand</u> 38: 271-276 (1988).
- 42. Coppin F, Chabroullet C and Martin-Garin A, Selenite interactions with some particulate organic and mineral fractions isolated from a natural grassland soil.
   *Eur J Soil Sci* 60: 369-376 (2009).
- 43. Langan SJ, Paterson E and Taylor AG, The Scottish soil resource: current

status and future priorities for management, in Soils, Sustainability and the Natural Heritage, ed. by Taylor AG, Gordon JE and Usher MB. Scottish .bu. .s.<u>1 Soil Sci</u> 40: 64 Natural Heritage, HMSO, Edinburgh, pp. 69-91 (1996). 44. Christensen BT, Bertelsen F and Gissel-Nielsen G, Selenite fixation by soil 

#### Table 1. Properties of the soils. Apart from the predicted Se status, the data is from the

National Soils Inventory of Scotland

Soil association, parent material, and land area of Scotland	Predicted Se status <sup>a</sup>	Soil No.	Major soil group	Sand-Silt- Clay (%) <sup>b</sup>	Carbon (%)	pH (CaCl <sub>2</sub> )
Darleith, drifts derived from basaltic rocks, 3.53 % <sup>c</sup>	Н	1	Brown earth	55-28-10	4.94	5.31
		2	Gley	40-36-24	2.3	5.57
		3	Podzol	56-27-17	10.5	4.22
Ettrick, drifts derived from lower Paleozoic greywackes	Н	4	Brown earth	55-30-8	6.17	5.6
and shales, 9.26 %		5	Brown earth	52-41-7	5.99	5.31
		6	Brown earth	70-27-3	3.54	6.28
		7	Gley	50-37-13	1.72	5.55
Foudland, drifts derived from slates, phyllites and other	Н	8	Brown earth	54-38-8	4.42	4.44
weakly metamorphosed argillaceous rocks, 3.25 %		9	Podzol	62-24-14	4.18	5.37
		10	Podzol	60-24-16	6.88	5.11
Sourhope, drifts derived from Old Red Sandstone	Н	11	Brown earth	NA	15.5	5.12
intermediate lavas, 1.71 %		12	Brown earth	65-24-6	3.67	5.33
		13	Brown earth	49-31-20	3.55	5.35
		14	Brown earth	59-35-6	3.99	4.88
		15	Gley	65-20-12	1.64	6.51
		16	Podzol	88-11-2	10.5	3.9
Balrownie, drifts derived from sandstones of Lower	L	17	Brown earth	70-26-4	3.66	4.94
Old Red Sandstone age, often water-modified, 1.83 %	$\overline{}$	18	Brown earth	NA	16.9	3.73
		19	Brown earth	70-23-7	2.69	5.42
		20	Brown earth	72-17-11	2.92	5.12
		21	Glev	70-16-13	3.25	5.27
Countesswells drifts derived from granites and granitic	L	21	Podzol	70-10-13	2 72	5.27
rocks. 5.75 % <sup>c</sup>		22	Podzol	73-13-14	4.06	5.27
Corby fluioglacial and raised heach sands and gravels	I	23	Glev	86-9-5	11.3	4 34
derived from acid rocks 3 08 % <sup>c</sup>	L	25	Podzol	36-56-8	3.26	5 18
derived from deld focks, 5.00 %		25	Podzol	30-30-8 88-8-4	5.20	3.10
Strichan drifts derived from grangeable schiets and	т	20	Brown earth	72 24 4	1.15	5.5
strongly metamorphosed argillaceous schists of the	L	28	Brown earth	72-24-4	3.41	1.2
Dalradian Series 7.98 %		20	Glev	71-21-0	3.41	4.2 5.11
		29	Podzol	62 24 15	5.05	5.65
		21	Podzol	02-24-15	0.10	J.0J 4 1
		22	Podzol	91-0-3	9.50	4.1
	UC	32 22	Podzol	87-9-4	9.52	5.0 5.06
Alluvial, recent riverine and lacustrine alluvial deposits,	UC	33	Alluvial	87-9-4	2.34	5.06
1.05 %		34	Alluvial	79-16-4	2.83	5.43
		35	Alluvial	51-37-12	1.05	5.76
		36	Alluvial	64-30-3	1./3	6.49
	UC	37	Alluvial	44-35-9	12.1	5.35
Deecastle, drifts derived from Dalradian limestones and	UC	38	Brown earth	59-34-7	6.02	3.95
calc-silicate focks, 0.19 %	110	39	Podzol	74-19-7	6.42	3.94
Hindsward, drifts derived from Carboniferous	UC	40	Brown earth	63-20-17	2.24	5.73
sediments and basic igneous rocks, 0.24 %		41	Gley	57-21-23	5.1	3.93
Kilmarnock, drifts derived from igneous and	UC	42	Brown earth	57-30-7	4.65	5.33
Sedimentary rocks of Carboniferous age, 0.40 % Peterhead, drifts derived from Old Red Sandstone Sediments with igneous metamorphic rocks and conglomerate cobbles, 0.18 %	UC	43	Brown earth	61-20-14	4.68	4.99
Rowanhill, drifts derived from Carboniferous	UC	44	Brown earth	75-20-5	3.58	6.17
sandstones, shales and limestones, $3.04 \%$ <sup>c</sup>		45	Gley	NA	10.8	4.51
Tarves, drifts derived from intermediate rocks or mixed	UC	46	Brown earth	63-20-12	4.27	4.71
acid and basic rocks, both metamorphic and igneous, $2.07 \%$		47	Gley	64-21-15	3.12	5.8

 $^{a}$ H = high, L =low, UC = unclassified.

<sup>b</sup> Sand 2000 - 20, silt 20 - 2, clay < 2  $\mu$ m.

<sup>c</sup> Includes some other soils with similar parent material.

NA = Not available.

#### Table 2. Average Se content of soils determined by ICP-MS and the mineralogical

composition of soil

Soil No. <sup>a</sup>	Se (mg kg <sup>-1</sup> )	Quartz (%)	K-feldspar (%)	Plagioclase (%)	Amphibole (%)	Fe-oxides <sup>b</sup> (%)	Other minerals $^{\circ}$	Phyllo- silicates
1	$0.77 \pm 0.03$	22	83	4.6	0.7	5.6	2.2	(70)
2	$0.77 \pm 0.03$ 0.51 ± 0.13	22	10.5	11.8	0.7	5.0 4.2	1.5	20
2	$0.31 \pm 0.13$	2/	4 1	94	1.1	4.2 8.0	11.9	20
3	$1.00 \pm 0.10$	5 40	4.1	2. <del>4</del> 2.7	1.5	1.0	1.9	20
4	$0.80 \pm 0.08$	49	0.9	2.7	1.1	1.0	1.2	34
5	$0.77 \pm 0.03$	41	1.1	9.0 5.5	1.0	1.5	1.9	29
0	$0.73 \pm 0.03$	45	9.0	5.5	0.8	2.1	1.1	20
/	$0.29 \pm 0.06$	49	4.0	0.0	0.8	1.0	1.2	27
8	$0.75 \pm 0.11$	35	2.8	12.0	0.7	2.9	1.7	35
9	$0.58 \pm 0.07$	49	5.8	10.2	1.5	1.4	1.5	19
10	$0.71 \pm 0.10$	28	4.7	8.3	5.1	3.6	2.6	27
11	$0.96 \pm 0.04$	14	9.1	20.1	2.5	3.5	2.2	20
12	$0.36 \pm 0.18$	45	9.6	10.8	1.9	2.2	1.0	15
13	$0.20 \pm 0.14$	12	1.7	26.7	1.5	2.7	2.9	31
14	$0.42 \pm 0.10$	38	4.3	14.7	0.8	2.3	1.4	23
15	$0.27 \pm 0.04$	46	19.8	4.9	1.1	0.8	0.8	14
16	$0.90 \pm 0.17$	34	7.4	17.6	4.9	1.5	3.4	14
17	$0.64 \pm 0.07$	42	5.4	19.5	1.1	0.8	1.4	16
18	$1.25 \pm 0.03$	38	3.1	14.6	1.6	2.1	1.1	15
19	$0.38 \pm 0.05$	45	5.9	17.5	1.1	1.1	1.8	18
20	$0.33 \pm 0.12$	48	7.1	14.9	1.6	1.0	2.2	13
21	$0.27 \pm 0.02$	60	7.5	5.5	1.2	0.8	0.6	15
22	$0.42 \pm 0.13$	36	14.6	25.8	2.8	1.2	1.3	11
23	$0.52 \pm 0.14$	33	10.3	24.9	2.4	1.0	1.5	16
24	$0.48 \pm 0.14$	37	11.8	22.4	1.5	0.1	0.9	8
25	$0.49 \pm 0.12$	51	7.0	9.2	2.8	1.0	1.3	15
26	$0.59\pm0.02$	41	7.2	17.2	6.7	1.5	1.6	15
27	$0.21 \pm 0.10$	47	7.6	10.7	1.3	1.0	0.5	26
28	$0.81 \pm 0.05$	39	2.3	15.5	2.9	2.8	2.9	28
29	$0.54 \pm 0.11$	51	8.5	13.2	2.9	1.8	3.4	11
30	$0.98 \pm 0.12$	45	4.2	5.0	1.0	3.0	1.2	25
31	$0.70 \pm 0.06$	37	3.0	15.8	4.4	1.3	3.3	19
32	$1.06 \pm 0.09$	25	2.5	17.2	8.7	5.8	3.1	20
33	$0.19 \pm 0.16$	61	13.0	12.4	0.7	0.5	0.7	6
34	$0.47 \pm 0.04$	33	2.7	20.1	1.1	2.8	2.7	26
35	$0.37 \pm 0.05$	39	8.3	14.5	1.4	1.2	1.6	22
36	$0.54 \pm 0.07$	46	4.1	13.1	0.6	1.5	1.4	21
37	$1.46 \pm 0.09$	30	3.4	7.8	0.4	1.5	0.7	27
38	$0.92 \pm 0.24$	33	2.8	10.8	6.2	1.6	1.8	31
39	$0.64 \pm 0.01$	31	3.0	15.1	3.9	2.3	3.3	27
40	$0.40 \pm 0.06$	47	7.2	9.8	1.1	2.2	2.1	18
41	$1.08 \pm 0.09$	34	2.6	5.8	0.4	0.5	1.0	33
42	$0.62 \pm 0.07$	42	6.3	7.5	1.5	4.1	2.0	20
43	$0.47 \pm 0.09$	44	8.2	11.9	1.1	1.8	1.4	19
44	$0.67 \pm 0.02$	56	4.6	6.3	1.0	2.8	2.4	14
45	$1.28 \pm 0.02$	46	3.0	4.7	0.4	2.3	0.6	16
46	$0.54 \pm 0.02$	33	9.6	18.9	2.7	3.7	3.4	19
10	5.51 ± 0.00	33	5 1	10.5	1.6	2.4	1.0	17

<sup>b</sup> Sum of goethite, hematite and magnetite. <sup>c</sup> Sum of pyroxene, garnet, anatase and ilmenite.

<sup>d</sup> Sum of chlorite, saponite / trioctahedral smectite, muscovite, illite, illite / smectite, biotite, vermiculite, hydrobiotite, kaolinite, halloysite and Fe-smectite.

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# 1 Figure captions

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3	Fig. 1. Map of Scotland showing areas predicted to have soils with relatively high, low
4	or unclassified Se contents, and the soil sampling locations.
5	Fig. 2. Comparison of values for the aqua regia extractable Se concentrations of soils

- 6 measured by ICP-MS and fluorimetry.
- 7 **Fig. 3**. Relationship between the aqua regia extractable Se concentrations of the soils
- 8 measured by ICP-MS and (a) the C contents of the soils, and (b) the LOI of the

# 9 soils.

- Fig. 4. Partial least squares analyses with aqua regia extractable Se concentrations of
   the soil as the response and various mineral and LOI as predictors: (a) standard
   coefficient plot; and (b) loading plot.
- Fig. 5. Map of Scotland showing the predicted values of aqua regia extractable Se in
  arable soils at the National Soils Inventory of Scotland, 10-km grid positions.
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