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Synergistic toxicity of some sulfonamide mixtures on *Daphnia magna*

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1 **Synergistic toxicity of some sulfonamide mixtures on *Daphnia magna*.**

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8

9

10 **Abstract**

11

12 In livestock farming, sulfonamides (SAs) are used prophylactically and simultaneously  
13 in large numbers of animals. Therefore, traces of these compounds, alone or in  
14 combination, have been repeatedly detected in the environment. Synergistic interactions  
15 among chemicals in such mixtures represent an area of concern for the regulatory  
16 authorities. In this study, the acute toxic effects of binary and ternary mixtures of SAs  
17 were evaluated in *Daphnia magna*, in order to verify whether, based on their individual  
18 toxicity, they jointly exert a larger effect than would be predicted by individual actions  
19 alone. First, following the Concentration Addition (CA) principle, some preliminary  
20 observations were made by testing a number of drug combinations with an expected  
21 50% effect. Then, mixtures more recognised for their synergistic effect (four binary and  
22 two ternary) were assayed in a range of reducing concentrations. The data acquired were  
23 processed using CompuSyn software, which integrates the different shape of the curves  
24 obtained in calculating the Combination Index (CI) for the evaluation of synergistic  
25 effects. For binary mixtures, synergy was also evaluated using the curvilinear  
26 isobologram method for heterodynamic drugs. Results indicate that most of the selected  
27 mixtures exhibit a synergistic effect using the CI methodology. For binary mixtures,  
28 these findings were also confirmed by isobologram analysis. Detected synergies indicate

29 that the CA is not always precautionary as a reference model for the evaluation of the  
30 aquatic toxicity of SAs mixtures.

31

32

33 **Keywords:** *Veterinary Sulfonamides, Drug-Mixtures, Daphnids, Synergy, CompuSyn.*

34

## 35 **Introduction**

36

37 In livestock farming, antibacterial drugs are used not only for therapeutic treatment of  
38 infected animals, but also for the so-called ‘mass treatments’ involving, simultaneously,  
39 a large number of animals. The following are carried out: growth promoting treatments,  
40 characterized by small doses of antibacterial added daily to the food during much of the  
41 production cycle and aimed at increasing the productivity of the animals; prophylactic  
42 treatments, routinely scheduled at critical times of the breeding cycle (weaning, change  
43 of housing, transport, etc.); and metaphylactic treatments, implemented promptly at the  
44 onset of disease in one or more subject of the group and aimed at treating infection in  
45 those already sick and prevent it in the still healthy ones. In the EU, despite the ban of  
46 the use of antibiotics as growth promoters, there seems to be no significant decrease in  
47 the consumption of antibiotics in the veterinary sector, as they continue to be used  
48 systematically for "prophylactic" purposes, due to unsustainable agricultural practices  
49 (Bond and Jewel, 2014).

50 Sulfonamides (SAs) are the oldest antibacterial agents and remain among the most  
51 widely used active pharmaceutical ingredients in veterinary medicine (EMA, 2015),  
52 mainly because of low cost and relative efficacy in some common bacterial and  
53 protozoan diseases.

54 Many of the available Veterinary Medicinal Products (VMPs) containing SAs are  
55 marketed as a premix, to be added to feed, or as an oral solution to be added to water.  
56 Using these formulations, animals may be treated simultaneously for preventive  
57 purposes and this can result in a substantial environmental load of the drugs. SAs are

58 subject to weak metabolism in the body of livestock and thus are eliminated mainly as  
59 such, or in the form of active metabolites in the excreta (Białk-Bielińska et al., 2013).  
60 As manure and slurry from farms are usually employed for the fertilization of  
61 agricultural land, a contamination of soil with SAs residues is clearly expected. During  
62 rainy days these residues can, partially at least, be transferred from soil to surface water  
63 by runoff (Boxall et al., 2002). Furthermore, SAs may also be directly released to  
64 watercourses as their use is extended to aquaculture in various countries. It is a matter  
65 of fact that residues of SAs have been repeatedly detected in the aquatic environment  
66 (Boxall *et al.*, 2005; Perret *et al.*, 2006; Xu *et al.*, 2007; García-Galán *et al.*, 2009; Santos  
67 et al., 2010; Guedes-Alonso et al., 2013; Giang et al., 2015).

68 For many years, the small crustacean *Daphnia magna* has been recognised as a keystone  
69 species in the food webs of many continental water bodies, and has served as an  
70 important model for ecotoxicological research (Seda and Petrussek, 2011). Impacts on  
71 daphnid populations may reverberate across the entire aquatic ecosystem as they are  
72 principal grazers of algae and primary forage for fish in lentic in-land ecosystems  
73 (Colbourne et al., 2011). The acute toxicity of SAs to *D. magna* is usually low ( $EC_{50s}$   
74  $>100 \text{ mg L}^{-1}$ ) with the notable exception of Sulfaguanidine ( $EC_{50}$   $6.2 \text{ mg L}^{-1}$ ) (Dalla  
75 Bona *et al.*, 2014). However, as SAs occur in natural environment not just as a single  
76 entity, but usually together with other compounds of the same family or the same type  
77 (Managaki *et al.*, 2007; Baran *et al.*, 2011; García-Galán *et al.*, 2011), it is of interest to  
78 evaluate the toxicity of their mixtures.

79 Here we adopted the concept of Concentration Addition (CA) to express the contribution  
80 of each chemical to the final mixture toxicity (Loewe and Muischnek, 1926). The  
81 concept is based on the assumption that all chemicals in a mixture act on the same  
82 biological target site and therefore could be viewed as being dilutions of each other, each  
83 having a different chemical potency (Cedergreen, 2014). Each chemical contribution to  
84 the overall toxicity of a mixture can be expressed as the quotient of its dose in the  
85 mixture and the dose of the same chemical alone that would be required to elicit the  
86 effect of the whole mixture. However, experimental data have often shown deviation

87 from this rule (Cedergreen, 2014), indicating more than additive interaction (an effect  
88 higher than expected, based on CA) or less than additive interaction (an effect lower  
89 than expected, based on CA).

90 To evaluate the CA deviations we prepared different binary and ternary mixtures  
91 containing SAs, with concentrations of each compound that, based on the CA concept,  
92 would be expected to result in a 50% immobilisation of *D. magna* after incubation for  
93 48h. The scope of this experiment was to provide only a preliminary assessment of the  
94 synergistic tendencies of the molecules studied to allow later selection of the most  
95 appropriate mixtures for further evaluation. Mixtures that showed a strong indication of  
96 interactions that were more than additive, were then tested using a range of reducing  
97 concentrations of the components, in an equi-toxicity concentration ratio design. These  
98 latter data were processed using CompuSyn software (Chou and Martin, 2005) to  
99 identify the EC<sub>50</sub> of each mixture, and to evaluate, more precisely, the interactions of its  
100 components (antagonism/synergy) at all effect levels. For binary mixtures, synergy was  
101 also evaluated using the curvilinear isobologram method proposed by Tallarida (2006)  
102 for heterodynamic drugs.

103

## 104 **Materials and Methods**

105

### 106 *Culture conditions*

107

108 Ehippia of *D. magna* were originally provided by ECOTOX (Milano, Italy). A single  
109 clone culture was selected based on the correct level of sensitivity to potassium  
110 dichromate (ISO, 1996) which was then rechecked periodically (every four months).

111 The subject organisms were maintained in Aachener Daphnien Medium (ADaM:  
112 hardness 193 mg CaCO<sub>3</sub>L<sup>-1</sup>; Klüttgen *et al.*, 1994a,b) at 20±1°C, with a photoperiod of  
113 16 h light (2.6 μE m<sup>-2</sup> s<sup>-1</sup>): 8 h dark. Their health status was optimal, and they did not  
114 show any sign of stress: mortality rate was ≤ 2% per week; reproduction rate was around  
115 10 neonates per day per individual; ehippia and/or males never appeared in the culture.

116 They were fed three times per week with *Scenedesmus dimorphus* ( $8 \times 10^5$  cells mL<sup>-1</sup>).  
117 The alga was cultured in 2L BBM (Bold Basal Medium) enriched with 3 g of sterilised  
118 poultry dung and suspended by bubbling filtered air. Before it was fed to the *Daphnia*  
119 culture, the chlorophyte was filtered through a 50 µm laboratory test sieve (Endecotts  
120 LTD, London, England), centrifuged at 3000 g for 10 min, resuspended in 25% BBM  
121 medium at a concentration of  $2 \times 10^8$  cells mL<sup>-1</sup> and stored at  $4 \pm 1$  °C.

122

### 123 *Chemicals*

124

125 Analytical grade compounds were purchased from Sigma–Aldrich (Milano, Italy) and  
126 were of the following minimum purity: Sulfadiazine [68-35-9] (SDZ) 99%,  
127 Sulfaguanidine [57-67-0] (SGD) 99%, Sulfamerazine [127-79-7] (SMA) 99%,  
128 Sulfadimethoxine [122-11-2] (SDM) 98%, Sulfamethazine[57-68-1] (SMZ) 99%,  
129 Sulfaquinoxaline [59-40-5] (SQO) 95%. The majority of these compounds have good  
130 water solubility at a slight alkaline pH (O’Neil, 2006; Białk-Bielińska et al., 2012),  
131 therefore for these compounds the preparation of their solutions in ADaM at  
132 concentrations equal to their individual EC<sub>50</sub> could be achieved by simple stirring at  
133 room temperature. In the cases of SQO and SDZ, complete solubilisation in ADaM was  
134 achieved by returning the pH of the medium to its original value (8.0) using 1 M NaOH  
135 (De Liguoro *et al.*, 2009, 2010)

136

### 137 *Assayed mixtures*

138

139 Drug mixtures for the immobilisation test were prepared taking into account the EC<sub>50</sub> of  
140 individual compounds. All the possible binary mixtures (15) of the 6 compounds were  
141 assayed. Ternary mixtures to be assayed were chosen on the basis of the results already  
142 obtained with binary mixtures. After preparing solutions of each single compound in  
143 ADaM medium, corresponding to its individual EC<sub>50</sub>; equal volumes of two or three of  
144 these solutions were mixed to generate the binary and ternary mixtures (Table 1). In this

145 way, based on the principle of CA, a 50% immobilisation effect would have been  
146 expected from each mixture after 48h incubation. Therefore, any detected effect >50%  
147 would have been considered as an indication of more than additive interaction.  
148 Similarly, any detected effect <50% would have been considered as an indication of less  
149 than additive interaction. In other words, for any number of additive agents the following  
150 equation holds:

151

$$\sum_{i=1}^N \frac{dA_i}{DA_i} = 1$$

153

154 where  $dA_i$  is the dose/concentration of  $A_i$  in a mixture that produces a specified effect,  
155 and  $DA_i$  is the dose/concentration of the single agent which on its own elicits the same  
156 effect as the mixture (Kortenkamp and Altenburger, 1998).

157 Given that the CA principle is rooted in the assumption of the constant relative potency  
158 of the drugs being combined (Tallarida, 2006) and that with the six SAs studied (Figure  
159 1), this was not the case; as already indicated (see Introduction section), the preliminary  
160 tests were introduced in order to obtain an indication of which mixtures showed  
161 synergistic tendencies. Mixtures with a strong effect (> 90%) were then further assayed  
162 in a range of reducing concentrations (Table 2). This in order to plot their concentration-  
163 response curves, derive the  $EC_{50}$ s, and proceed to a reliable analysis of the interactions  
164 between their components, using CompuSyn software (Chou and Martin, 2005). The  
165 CompuSyn program integrates the different shapes of the curves in the calculation of  
166 the Combination Index for the evaluation of synergy; in this way, the constant relative  
167 potency of the combined drugs is not a prerequisite. For combinations of two drugs (not  
168 for three drugs), we also addressed the question using the equations proposed by  
169 Tallarida (2006) for heterodynamic drugs, which allow the isobole of additivity to be  
170 represented as a region bounded by two well-defined curves.

171

172 *Toxicity tests*

173

174 Acute toxicity tests were performed according to the Guideline 202 ‘*Daphnia sp.*, Acute  
175 Immobilisation Test’ (OECD, 2004). The ADaM medium was used for Controls and the  
176 dilution of test compounds. Eight groups of 5 young daphnids (third brood neonates;  
177 <24 h) were exposed to each of the assayed mixtures (Table 1 and Table 2) or used as  
178 controls. The organisms were fed for about 1 h with 100% pure, dried Spirulina powder  
179 (15 mg in 100 mL ADaM) just before the start of the experiment, and then each group  
180 was incubated in a 20 mL glass vessel loosely covered with parafilm, and containing 10  
181 mL of the test solution, under the same conditions (light, temperature) used for culturing.  
182 Pre-feeding of the organisms is not deemed necessary by the test guideline, however in  
183 our experience is strongly advisable as it helps to sustain 100% survival in the control  
184 groups. The number of immobile daphnids recorded after 48h was the endpoint for effect  
185 calculation.

186

### 187 *Data Analysis*

188

189 Data were processed using CompuSyn software for Drug Combinations and General  
190 Dose-Effect Analysis (Chou and Martin, 2005). Raw data for the effects of both single  
191 drugs and mixtures were entered. CompuSyn fitted the data and provided the model  
192 parameters and the concentration-effect plots. The model parameters were the EC<sub>50</sub> and  
193 the shape value “*m*” of the Hill curve *f* as a function of the chemical concentration *x*, as  
194 given by:

195

196

$$f(x) = \frac{1}{1 + \left(\frac{EC_{50}}{x}\right)^m}$$

197

198 If the exponent *m* is greater than 1, the curve is sigmoidal, when it is equal to 1 the curve  
199 is hyperbolic (Chou and Martin, 2005). The most relevant aspect of CompuSyn is to  
200 provide the evaluation and the plots that report Combination Indices. The Combination



201 Index (CI) quantifies the dose-effect relationship on the basis of “mass-action law” to  
202 evaluate the effect of combination of chemicals (Chou and Martin, 2010). The CI index  
203 furnishes a value that quantitatively indicates synergism (CI < 1), additive effect (CI =  
204 1), and antagonism (CI > 1). Here we used this tool to evaluate possible synergy among  
205 the different compounds.

206 To further evaluate the possible synergy between pairs of chemicals, we completed an  
207 isobologram analysis at EC<sub>50</sub>. Since there is no obvious basis upon which to distinguish  
208 whether chemical A is contributing to chemical B or vice versa, the use of dose  
209 equivalence leads to not one but to two possible isoboles of additivity, depending on  
210 how the concept of dose equivalence is applied (Tallarida 2006). This means that rather  
211 than being a single straight line, the isobole becomes an area bordered by two curved  
212 lines. In particular, according to previous indications (Tallarida, 2006), when two  
213 compounds have two different shapes (or exponents) of the dose-effect curve, the  
214 signature of the synergy/antagonism of their mixture must be found outside the region  
215 bounded by two curves. In the EC<sub>50</sub> isobologram estimation, the equivalent doses were  
216 computed using two equations that describe the upper and lower bounds of the additivity  
217 area and are expressed as:

218

$$219 \quad b = B_{50} \left( 1 - \frac{a}{A_{50}} \right)^{q/p} ; \quad b = B_{50} \left( 1 - \left( \frac{a}{A_{50}} \right)^{q/p} \right)$$

220

221 where  $q$  and  $p$  are the exponents of the Hill curves ( $m$  in the previous equation) for the  
222 chemicals  $A$  and  $B$ , respectively (Tallarida, 2006).  $A_{50}$  and  $B_{50}$  refer to the EC<sub>50</sub> of each  
223 of the two chemicals, while  $a$  and  $b$  are the doses (or concentrations) of each chemical  
224  $A$  and  $B$ . When  $p = q$ , the dose equivalent for  $B$  collapses to a single straight line, as can  
225 be seen from the equation above. However, in the general case, the greater the difference  
226 between  $q$  and  $p$ , the farther from this diagonal the two isoboles are (Tallarida, 2006).  
227 Since we evaluated the Hill shape exponent with CompuSyn, and the program reports  
228 the associated error for the computed exponent, we highlighted in the isobologram the  
229 uncertainty of the computed exponents. Thus, in the isobologram we also included the

230 “worst-case isobole”, in which the larger exponent is increased by summing its error,  
231 and the smaller exponent is decreased by subtracting its error. In practice, if  $q > p$  and  $\varepsilon_q$   
232 and  $\varepsilon_p$  are the corresponding estimated errors, the worst-case isobole is computed with  
233 the highest possible ratio  $r = (q + \varepsilon_q) / (p - \varepsilon_p)$ . In this way, if the measured point of the  
234 combination dose exceeds this “fatter” isobole the indication of synergy (or antagonism)  
235 gains greater confidence.

236

237

## 238 **Results**

239

240 Raw data for acute toxicity of single compounds were already available from previous  
241 experiments run in our lab (Dalla Bona *et al.*, 2014) under the same conditions (T, light  
242 cycle, length of exposure, age of daphnids, feeding) used for mixture assays. The relative  
243 concentration-response curves and  $EC_{50s}$ , generated using CompuSyn software, are  
244 presented in Figure 1.

245 In all tests, validity criteria were fulfilled as control survival (mobility) was 100%, and  
246 the recorded values of water quality parameters, measured at the beginning and at the  
247 end of the test, were always within the following ranges: pH 7.9–8.1, dissolved oxygen  
248 7.70–8.40 mg L<sup>-1</sup>. Temperature stability ( $20 \pm 1^\circ\text{C}$ ) of the medium was guaranteed by the  
249 use of a refrigerated incubator. Six binary mixtures of the 15 assayed, gave indications  
250 of more than additive interaction (Figure 2). The following had more than 90% effect:  
251 SMA+SDZ (97.5%); SQO+SDM (92.5%); SGD+SDZ (92.5%); SDM+SGD (100%).  
252 These were re-assayed in a range of reducing concentrations (from 0.5 to 0.25  $EC_{50}$  of  
253 each component) under the same conditions used in the previous tests: their  
254 concentration-effect curves and  $EC_{50s}$  are shown in Figure 3 and compared to the effect  
255 curves predicted by CA. In general, their effects were confirmed to be synergic (Figure  
256 4); at high effect levels - in three cases out of four, the synergy was strong (Combination  
257 Index < 0.3; Chou and Martin, 2005). At the 50% effect level, synergy was also  
258 confirmed by applying the equations proposed by Tallarida (2006); however, with

259 SGD+SDM and SMA+SDZ the EC<sub>50</sub> fell just below the confidence limit of the  
260 additivity area (Figure 5).

261 Three ternary mixtures were tested, and all gave indications of greater than additive  
262 interaction (Figure 2). The following had greater than 90% effect: SDM+SGD+SDZ  
263 (100%); SMA+SGD+SDZ (100%). These were re-assayed in a range of reducing  
264 concentrations (from 0.33 to 0.165 EC<sub>50</sub> of each component): their concentration-effect  
265 curves and EC<sub>50</sub>s are shown in Figure 3 and compared to the effect curves predicted by  
266 CA. Their effects were confirmed to be synergic at all effect levels (Figure 4).

267 Predicted No-Effect Concentrations (PNECs), for individual compounds in mixtures,  
268 obtained by applying an Assessment Factor of 1000 to the EC<sub>50</sub>s (CVMP/VICH/790/03),  
269 were always > 40 µg L<sup>-1</sup>, with the exception of SGD (> 1 µg L<sup>-1</sup>).

270

## 271 **Discussion**

272

273 Following an in-depth study on the hydrolysis of SAs in aqueous solutions, Białk-  
274 Bielińska *et al.* (2012) concluded that under typical environmental conditions (pH and  
275 temperature) SAs are hydrolytically stable with a long half-life, and that all could be  
276 assumed to be hydrolytically stable at pH 9 and 25°C for least 1 year. Moreover, in  
277 previous experiments with *D. magna* (De Liguoro *et al.*, 2009, 2010) it was verified,  
278 using HPLC analysis, that the 48 h level of decline of the above mentioned compounds  
279 under the conditions used in the tests (pH 8.0; 20°C) was between 0 and 13%. Based on  
280 the CRED (Moermond *et al.*, 2016), in acute toxicity tests with stable substances,  
281 nominal concentrations without further measurements are acceptable. Furthermore,  
282 Guideline 202 ‘*Daphnia sp.*, Acute Immobilisation Test’ (OECD, 2004) states that if the  
283 concentration of the test substance has been maintained throughout the test within ± 20  
284 per cent of the nominal initial concentration, the results can be based on the nominal  
285 values. Thus, in the present study, the use of HPLC analysis was rendered redundant and  
286 consequent undesirable excess use of solvents was avoided, with test results being based  
287 on nominal concentrations.

288 The various SAs evaluated in this study share the same mechanism of action and cellular  
289 target (Eguchi *et al.*, 2004); consequently, their combinations should follow the CA  
290 principle (Cedergreen, 2014). The 15 (preliminary) binary tests, where all possible pairs  
291 were tested, showed 9 cases of less than additive interaction (Combination Index >1),  
292 and 6 cases of more than additive interaction (Combination Index <1). The 3  
293 (preliminary) ternary tests, chosen, based on binary test results, all showed more than  
294 additive interaction. In some cases the deviation from the rule of CA, in one way or  
295 another, was strong (Figure 2). The CA principle, however, is rooted in the assumption  
296 of a constant relative potency of the drugs being combined. In other words, the Hill  
297 coefficients (Faust *et al.*, 2003), which respectively describe their concentration-effect  
298 relations, should be equal (Tallarida, 2006). With the studied SAs this assumption did  
299 not hold true (Figure 1). Therefore, the positive results of the preliminary tests were  
300 taken only as a possible indication of synergy, rather than definitive proof. The more  
301 promising mixtures were then re-assayed in a range of five concentrations, spreading  
302 experimental data points below and above the EC<sub>50</sub> value, as suggested by Chou for the  
303 use of the CompuSyn application (Chou and Martin, 2005). At the 50% effect level,  
304 synergy was confirmed for all four binary mixtures by applying the equations of  
305 Tallarida. To further check the synergetic effects, worst-case isoboles were also included  
306 by adding the computed error to the larger exponent and subtracting the computed error  
307 from the lowest exponent. Figure 5 shows that all the binary mixtures, passed this more  
308 restrictive test, indicating significant synergy around the EC<sub>50</sub> combined dose.

309 For a more comprehensive evaluation of the drug interactions, we used the CompuSyn  
310 program, which allows the evaluation of synergy at all effect levels, both for binary and  
311 ternary mixtures. Detected synergies (Figure 4) indicate that the concept of CA is not  
312 always precautionary as a reference model for the evaluation of the aquatic toxicity of  
313 SA mixtures. Interestingly, SDZ that is the only SA licensed for aquaculture in EU, and  
314 therefore more prone than other compounds to the contamination of the aquatic  
315 environment, was frequently involved in synergic interactions. It should be noted,  
316 however, that synergies were generally stronger when immobilisation percentages were

317 very high (Figure 4), i.e. when relatively high concentrations of SAs were mixed. This  
318 means that at the very low concentrations usually encountered in the natural  
319 environment, SA synergies may be of more limited relevance to *D. magna*. More  
320 generally, calculated PNECs for single components of each mixture indicate that the  
321 currently reported level of SA contamination ( $<1 \mu\text{g L}^{-1}$ ) should have no impact on the  
322 freshwater environment. Nevertheless, it would be of interest to assess the effects of the  
323 selected SA mixtures in the chronic *D. magna* Reproduction Test, which is indeed far  
324 more sensitive than the acute immobilisation test and allows the estimation of NOEC  
325 for PNEC calculation.

326 The authors think that some experiments with similar SAs mixtures, on more sensitive  
327 species, such as cyanobacteria, would complement the present work. Indeed,  
328 cyanobacteria are generally considered to be the most sensitive aquatic organism to  
329 antibacterials; however, it has also been shown that green algae are more sensitive than  
330 daphnids to the toxicity of some selected SAs, with NOEC values in the range 0.02-1  
331  $\text{mg L}^{-1}$  (Eguchi *et al.*, 2004; De Liguoro *et al.*, 2010). In the natural environment, some  
332 cascade effect on daphnids may also be expected, as green algae are their basic food  
333 resource. Such an effect could not be highlighted by the acute immobilisation test where,  
334 in order to avoid any nutritional variation among the experiments, daphnids were fed  
335 only with a calibrated quantity of dried spirulina. Possible synergic interactions with  
336 Trimethoprim (TMP) or Pyrimethamine (PMT), two SA potentiators frequently  
337 included in VMPs, should be taken into consideration in addition. For instance, Eguchi  
338 *et al.* (2004) showed that pairing SMZ, SDZ and SDM with TMP or PMT strongly  
339 enhanced their algal growth inhibition effects. Overall, the indications of synergy  
340 between SAs observed during these tests open the way for a range of new experiments  
341 to further deepen our understanding of this phenomenon.

342

## 343 **Conclusions**

344

345 A range of methods is available for the evaluation of the synergistic interactions of drug  
346 mixtures. As suggested by Fouquier and Guedj (2015), in the absence of a reference  
347 methodology appropriate for all situations, the evaluation of the impacts of various drug  
348 combinations may be facilitated by the collective use of different approaches. Here,  
349 binary mixtures of veterinary SAs were evaluated using two different models, and both  
350 generally confirmed the possibility of synergistic interactions among these compounds.  
351 Whilst their combined acute toxicity to *D. magna* still seems too low to represent a real  
352 threat in the natural environment, future studies with SAs mixtures should focus on the  
353 possible chronic harm to daphnids and to other, more sensitive, aquatic organisms.

354

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356

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360

361

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446

447 **Figure Captions**

448

449 Figure 1. Concentration-effect curves of single SAs in *D. magna* immobilisation test (48h).  $r$ =correlation  
450 coefficient;  $m$ =exponent of the Hill-curve that defines the curve slope. Vertical error bars show standard  
451 deviation (4 vessels, each with 5 daphnids). SDM, sulfadimethoxine; SGD, sulfaguanidine; SDZ, sulfadiazine;  
452 SMA, sulfamerazine; SQO, sulfaquinoxaline; SMZ, sulfamethazine.

453 Figure 2. Effect percentage of binary and ternary mixtures of SAs in *D. magna* immobilisation test (48h);  
454 based on the CA principle a 50% effect was to be expected. However, deviations from this rule may also be  
455 the consequence of the inconstant relative potency of the drugs being combined (Tallarida, 2006) Horizontal  
456 error bars show standard deviation (8 vessels, each with 5 daphnids). SMA, sulfamerazine; SGD,  
457 sulfaguanidine; SDZ, sulfadiazine; SDM, sulfadimethoxine; SQO, sulfaquinoxaline; SMZ, sulfamethazine.

458 Figure 3. Concentration-effect curves of binary (a,b,c,d) and ternary (e,f) mixtures of SAs in *D. magna*  
459 immobilisation test (48h). Vertical error bars show standard deviation (8 vessels, each with 5 daphnids).  
460 Dashed lines are the concentration-effect curves predicted by Concentration Addition principle. SDM,  
461 sulfadimethoxine; SGD, sulfaguanidine; SDZ, sulfadiazine; SMA, sulfamerazine; SQO, sulfaquinoxaline.

462 Figure 4. Graphic representations obtained from the CompuSyn Report for SAs binary (a,b,c,d) and ternary  
463 mixtures (e,f) assayed on *D. magna*: Combination Index <1 indicates synergic interaction. SDM,  
464 sulfadimethoxine; SGD, sulfaguanidine; SDZ, sulfadiazine; SMA, sulfamerazine; SQO, sulfaquinoxaline.

465 Figure 5. Isobolograms of SAs binary mixtures assayed on *D. magna*. For compounds with a variable potency  
466 ratio, synergy is detected only if the  $EC_{50}$  of the mixture lies below the region of the plane bounded by the  
467 two curves of additivity for a 50% effect (Tallarida, 2006). Dotted lines represent curves of additivity. Dashed  
468 lines are their confidence limits based on Hill coefficient variability. SGD, sulfaguanidine; SDM,  
469 sulfadimethoxine; SDZ, sulfadiazine; SMA, sulfamerazine; SQO, sulfaquinoxaline.

**Table 1.** Preliminary assays of binary (B) and ternary (T) mixtures of SAs in *D. magna* immobilisation test.

Mixture	Sulfadimethoxine (mg/L)	Sulfaguanidine (mg/L)	Sulfadiazine (mg/L)	Sulfaquinoxaline (mg/L)	Sulfamethazine (mg/L)	Sulfamerazine (mg/L)
B1	132.8	3				
B2	132.8		95.4			
B3	132.8			69		
B4	132.8				104.4	
B5	132.8					102.5
B6		3	95.4			
B7		3		69		
B8		3			104.4	
B9		3				102.5
B10			95.4	69		
B11			95.4		104.4	
B12			95.4			102.5
B13				69	104.4	
B14				69		102.5
B15					104.4	102.5
T1	88.5	2	63.6			
T2	88.5	2		46		
T3		2	63.6			68.3

**Table 2.** Assays in a range of reducing concentrations of selected binary and ternary mixtures of SAs in *D. magna* immobilisation test.

Combination	SAs	Assayed concentrations (mg/L)				
Binary	Sulfadimethoxine	132.8	116.2	99.6	83.0	66.4
	Sulfaguanidine	3.0	2.6	2.3	1.9	1.5
Binary	Sulfamerazine	102.5	89.7	76.9	64.1	51.3
	Sulfadiazine	95.4	83.5	71.6	59.6	47.7
Binary	Sulfaguanidine	3.0	2.6	2.3	1.9	1.5
	Sulfadiazine	95.4	83.5	71.6	59.6	47.7
Binary	Sulfaquinoxaline	69.0	60.4	51.8	43.1	34.5
	Sulfadimethoxine	132.8	116.2	99.6	83.0	66.4
Ternary	Sulfadimethoxine	88.5	77.5	66.4	55.4	44.3
	Sulfaguanidine	2.0	1.8	1.5	1.3	1.0
	Sulfadiazine	63.6	55.7	47.7	39.8	31.8
Ternary	Sulfaguanidine	2.0	1.8	1.5	1.3	1.0
	Sulfadiazine	63.6	55.7	47.7	39.8	31.8
	Sulfamerazine	68.3	59.8	51.2	42.7	34.2











