



Durham E-Theses

The quantitative measurement of colour in minerals

Htein, Win

How to cite:

Htein, Win (1972) *The quantitative measurement of colour in minerals*, Durham theses, Durham University. Available at Durham E-Theses Online: <http://etheses.dur.ac.uk/8421/>

Use policy

The full-text may be used and/or reproduced, and given to third parties in any format or medium, without prior permission or charge, for personal research or study, educational, or not-for-profit purposes provided that:

- a full bibliographic reference is made to the original source
- a [link](#) is made to the metadata record in Durham E-Theses
- the full-text is not changed in any way

The full-text must not be sold in any format or medium without the formal permission of the copyright holders.

Please consult the [full Durham E-Theses policy](#) for further details.

Thesis of Win Htein

Page	to	R%	Page	to	R%	Page	to	R%
278		76	274		20	275		18
284		45	264		29	193		34
283		52	288		41	4	18	53
280		62	276		17	191		40
289		50	183		40			(hex)
83		22	192		33	286		45
71	82	18-24	245		32	1		99
181	182	41	142		52	142		53
295		35	195	200	25-31	290		42
273		21	266		20	89	119	18
286		50	145	146	47	141		55
240	242	27	85		19	222	227	28
306		12	188	189	40	216	221	28
28		34	281		59			(yellow)
296		33	305		0-12 E-36	298		26
268		27	258	263	25	284		54
164	178	47	156		58	305		14
122	139	13	271		42	293		37
298	299	29	289		44	238	239	24
141		50	148	150	42			46
19		50	161		53	287		31
82	153	34-38	235	237	25	32	35	31
41		30	87		17	162	163	60
40		53	84		21	301	302	17
13		32	300		24	53	63	28
7	308	0-7, E-22	157	159	53	42	52	33
74		34	88		18	265		28
		26	300		24	155		17
278		64	157		53	292		43
246	257	29	121		14	146		46
297		35	213	215	0-25 E-26	291		43
86		18	280		60			45
20	26	44	184	185	40	147		45
270	272	22	282		54	304		0-12 E-16
306		14	154		17			31
179	180	41	301		20	64		43
65	70	22	160		53	290		15
143	144	46	194		31	303		18
297		27	201	212	23-26	303		18
3		57	279		73	186	187	41
29	31	32	147		45	140		67
2		89	150	152	37-41			
277		20	190		37			
293		42	302		17			
			292		40			
			145		46			
			269		23			

Also on p. 288

THE QUANTITATIVE MEASUREMENT
OF COLOUR IN MINERALS

A THESIS SUBMITTED FOR THE DEGREE OF
DOCTOR OF PHILOSOPHY IN THE
UNIVERSITY OF DURHAM

by

WIN HTEIN,
B.Sc. (Rangoon), F.G.A. (London)

Graduate Society

December, 1972.

ABSTRACT

The present study has been directed towards investigating quantitative measurements of colours in minerals, which might serve as a readily accessible means of identification and understanding related properties.

A review of the concepts of colour measurement is described in the first part of the thesis. Attention has been paid to the improvement and application of methods of colour measurement in mineralogy and gemmology. Computational procedures and all the necessary computer programmes for calculating colour values are given with examples. For this purpose appropriate measurements of spectral reflectivity for opaque minerals, and spectral transmittance for non-opaque minerals and faceted gemstones are described.

Quantitative measurements of other colour properties - colour constancy, bireflectance and reflection pleochroism - are described and defined. A study of some colour problems in ore minerals is made. Tables of colour values and spectral reflectivity data for 130 ore minerals are given, and these values are also presented in colour diagrams. The uses of quantitative colour values and colour diagrams are explained with examples.

ACKNOWLEDGEMENT

The writer wishes to thank to the British Council, the Ministry of Overseas Development, and the Ministry of Education, Union of Burma, for the Colombo Plan Fellowship Award.

The writer wishes to thank to Professor G.M.Brown, of the Geology Department, for the facilities provided for this study .

The writer thanks Mr. R.Phillips, for suggesting this research topic, supervising the study and for critically reading the manuscript.

Special thanks are due to: Mr. E.A.Jobbins, of the London Geological Museum, for loan of gemstones out of his own collection; Mr. F.W.Smith, for translating the paper by Levy (1966); Dr. A.Peckett and Mr. R.Hardy, for analysing specimens; Dr.M.J.Reeves, for assistance in writing the computer programmes; and Mr. S.Gezci, of the Applied Physics Department, for his advice on the use of the Double Beam Grating Spectrograph.

The technical staff of the Geology Department and the Science Workshop are thanked for their kind help.

Finally the writer is indebted to the individuals and institutions, listed in References, for permission to redraw diagrams or copy tables and photos from the sources noted.

CONTENTS

ABSTRACT

ACKNOWLEDGEMENT

Chapter	Page
I. INTRODUCTION	1
II. LITERATURE REVIEW ON THEORY OF COLOUR MEASUREMENT	3
II.1 Introduction	3
II.2 Psychophysical Concepts of Colour	5
II.3 Attributes of Colour	7
II.4 The sensitivity of the Eye	9
Relative Luminous Efficiency Functions	9
Sensitivity to Differences of Hue and Saturation	12
II.5 Spectral Energy Distribution and Colour Sensation	15
II.6 Principles of Colour Measurement	17
II.7 The C.I.E. System of Colour Measurement	25
II.8 C.I.E. Standard Light Sources	28
II.9 Computation of Tristimulus Values and Chromaticity	
Coordinates	30
Weighted Ordinate Method	31
Selected Ordinate Method	32
Accuracy of two Methods	33
II.10 The 1931 C.I.E. Chromaticity Diagram	35
II.11 Dominant Wavelength and Excitation Purity	36
II.12 Colour Discrimination	38
III. SPECTRAL REFLECTIVITY MEASUREMENTS	46
III.1 Previous Work	46
III.2 Present Work	50
Specimen Preparation	50

	Apparatus and Technique	50
	Accuracy and Precision	53
	III.3 Spectral Reflectivity of Gemstones	54
IV.	POSSIBILITY OF TRANSMISSION MEASUREMENT THROUGH FACETTED GEMSTONES FOR COLOUR MEASUREMENT	58
	Theoretical Considerations	58
	Experiments and Results	63
v V.	COMPARISON OF TRANSMISSION MEASUREMENTS WITH A MICROSCOPE PHOTOMETER AND DOUBLE BEAM SPECTROGRAPH	67
VI.	METHODS OF COMPUTING COLOUR SPECIFICATIONS OF MINERALS	72
	VI.1 Computational Procedure for Tristimulus Values and Chromaticity Coordinates	72
	PL/1 Computer Programme	Appendix
	VI.2 Computational Procedure for Dominant and Complementary Wavelengths, and Excitation Purity	74
	PL/1 Computer Programme	Appendix
	PL/1 Computer Programme	Appendix
VII.	ACCURACY AND PRECISION OF COLOUR SPECIFICATIONS OF MINERALS	79
VIII.	MEASUREMENTS OF COLOUR CONSTANCY, BIREFLECTANCE AND REFLECTION PLEOCHROISM	83
	PL/1 Computer Programme	Appendix
IX.	A STUDY ON SOME COLOUR PROBLEMS IN ORE MINERALS	92
	Pure Gold, Pure Silver, Gold-Silver Alloy	93
	Germanite	95
	Bornite	96
	Pyrite	98
	Chalcopyrite	100
	Tetrahedrite and Tennantite	101
	Sphalerite	102

Galena	104
Colusite and Sulvanite	105
Stannite, Mawsonite, Idaite	106
Enargite, Luzonite-Stibiolumonite	108
Hematite powders	109
X. QUANTITATIVE COLOUR VALUES AND SPECTRAL REFLECTIVITIES OF SOME ORE MINERALS	Appendix
XI. QUANTITATIVE COLOUR VALUES, SPECTRAL TRANSMISSION CURVES AND A COLOUR DIAGRAM OF GEMSTONES	Appendix
XII. CONCLUSIONS AND SUGGESTIONS OF POSSIBLE FUTURE WORK	111
LIST OF TABLES	
LIST OF FIGURES	
LIST OF PLATES	
APPENDICES	
REFERENCES	
MINERALS IDENTIFICATION DIAGRAMS	

LIST OF TABLES	Page
II 1. Colour names and corresponding wavelength ranges	7
II 2. Relationships between the 1931 CIE- RGB System and the 1931 CIE- XYZ System	27 b
II 3. Standard chromaticity coordinates and colour-matching functions of equal-energy spectrum	27 c
II 4. Chromaticity coordinates of the C.I.E. standard sources	29 b
II 5. 1931 C.I.E. colour-matching functions weighted by relative spectral energy distribution of sources A,B & C	34 a
II 6. Selected ordinates for standard sources A,B & C	34 b
III 1. Accuracy and precision of reflectivity measurement	53 a
IV 1. Measured spectral R.I. & calculated values	66 a
IV 2. Measured and Calculation transmittance values	66 b
IV 3. Measured values of brown quartz	66 b
IV 4. Colour specifications derived from calculated and measured transmittance values	66 d
V 1. Spectral transmittance values of red spinel	71 a
V 2. " " " " brown quartz	71 b
V 3. " " " " red spinel	71 c
V 4. " " " " brown quartz	71 d
V 5. Quantitative colour values of red spinel	71 j
V 6. " " " " brown quartz	71 k
VI 1. Ratios to compute wavelengths and excitation purity	Appendix
VII 1. Measured reflectivity values and deviations at omitted wavelengths due to interpolation	82 a
VII 2. Variations in chromaticity coordinates, and precision	82 b
VIII 1. Differently changes in the colours of briatite and gallite when the illuminant is changed	91 a

LIST OF FIGURES (continued)	Page
II 16. Relative spectral distributions of the standard sources A,B,C & E	29 a
II 17. 1931 C.I.E. Chromaticity diagram with points of the standard sources and the locus of Planckian radiators	29 a
II 18. Derivation of the tristimulus specifications of an object colour by two methods	34 c
ii 19. 1931 C.I.E. standard chromaticity diagram superimposed on a colour plate	35 a
II 20 Graphical determination of dominant (and complementary) wavelength and excitation purity of a sample colour	37 a
II 21. Discrimination ellipses obtained by MacAdam	45 a
II 22. Measures of a discrimination ellipse	45 a
II 23. MacAdam's contour diagrams	45 b
II 24. " " "	45 c
II 25. " " "	45 d
III 1. Schematic diagram of the apparatus	51 a
III 2. A test of the stabilised voltage current with time	51 b
III 3. Total back reflection and dispersion of a ray of light passing through brilliant cut diamond	57 a
III 4. Spectral reflectivity curves of faceted gemstones	57 b
III 5. Hypothetical relationships between R.I. & R of non-opaque minerals	57 c
IV 1. Transmission, reflection and absorption of a parallel beam of light passing through a parallel sided plate	58 a
IV 2. Transmission of a combination of non-absorbing layers separated by air	59 a
IV 3. Transmission of a combination of non-absorbing layers in optical contact	59 a

LIST OF TABLES (continued)	Page
VIII 2. Example of quantitative values of pleochroism of some ore minerals	91 b
IX 1. Electron Microprobe analyses and colour values of bornite	97 b
IX 2. Electron Microprobe analyses, colour values and optical properties of Stannite-Idaite series	107 b
IX 3. XRF analysis of hematite specimens	110 a
XI 1. Mean transmittance values of some cut gemstones	Appendix
XI 2. Quantitative colour values of some cut gemstones	Appendix

LIST OF FIGURES

Page

II 1.	Spectral distributions of irradiance of direct sun light	10 a
II 2.	Relative spectral distributions of irradiance of three phases of daylight	10 a
II 3.	Standard relative luminous efficiency functions	10 b
II 4.	Typical wavelength discrimination curve	12 a
II 5.	Average purity discrimination as a function of wavelength	12 a
II 6.	Relation between spectral energy distribution and colour sensation	16 a
II 7.	A three dimensional vector diagram illustrating a colour space and a corresponding two dimensional colour plane	19 a
II 8.	Illustrating the additive mixture of the two colours producing a new colour	19 b
II 9.	A colour vector diagram with a colour plane of constant brightness	20 a
II 10.	Mean chromaticity coordinates of spectrum colours obtained by Guild	24 a
II 11.	Mean chromaticity coordinates of spectrum colours obtained by Wright	24 a
II 12.	Spectral distribution curves (colour-matching functions) for equal-energy spectrum in the RGB system	24 b
II 13.	Chromaticity diagram in the RGB system	24 b
II 14.	1931 C.I.E. colour-matching functions based on the XYZ primaries	27 a
II 15.	The C.I.E. standard chromaticity diagram in terms of the reference stimuli XYZ	27 a

LIST OF FIGURES (continued)	Page
IV 4. Transmission of a combination of non-absorbing layers of different R.I	59 a
IV 5. Transmission of a combination of an absorbing layer and non-absorbing layer in optical contact	60 a
IV 6A. Back reflection of a ray of light in passing through a facettted stone in air	60 a
IV 6B. A parallel transmitted ray of light in passing through a facettted stone in liquid	60 a
IV 7. Dispersion characters of solid and liquid	62 a
IV 8. Transmission curves (measured and calculated)	66 c
V 1. Spectral transmittance curves of red spinel	71 e
V 2. Spectral transmittance curves of brown quartz	71 f
V 3. Chromaticity diagram showing the colours of red spinel brown quartz	71 g
V 4. Spetral sensitivities of the two types of photomultipliers	71 h
VI 1. Explanation for computing complementary wavelength and excitation purity	78 a
VIII 1. Spectral distribution curve giving a green colour in daylight and changing to a red under tungsten light	85 a
VIII 2. Chromaticity diagram showing a poor colour sonstancy of a specimen	85 a
VIII 3. Changes of dominant wavelength and purity of some ore minerals in passing from source A to source C	91 c
VIII 4. Example of different effects of pleochroism under source A and source C	91 d
IX 1. Gold, Silver and Gold-Silver Alloy	94 a
IX 2. Germanite	95 a
IX 3. Bornite	97 a

LIST OF FIGURES (continued)

Page

IX 4.	Pyrite	99 a
IX 5.	Chalcopyrite	100 a
IX 6.	Tetrahedrite-Tennantite	101 a
IX 7.	Sphalerite	103 a
IX 8.	Galena	104 a
IX 9.	Colusite , Sulvanite	105 a
IX 10.	Stannite, Mawsonite, Idaite	107 a
IX 11.	Enargite, Luzonite-Stibiolumonite	108 a
IX 12.	Chromaticity diagram showing the colours of hematite powders	110 b
IX 13.	Enlarged chromaticity chart with colour points of hematite specimens and discrimination ellipse	110 c
XI 1.	Chromaticity diagram with colour points of some cut gemstones	Appendix
XI 2.	Spectral transmission curves of citrine & chrysoberyl	Appendix
XI 3.	" " " " synthetic rubies	Appendix
XI 4.	" " " " almandine & amethyst	Appendix
XII 1.	Showing marked variations in spectral reflectivity values quoted for specific minerals	Appendix
1 a.	Enlarged Chromaticity Diagram with the colour points of some Ore Minerals	Appendix
1 b.	Enlarged Chart of Tristimulus Value Y and Dominant (also complementary) Wavelength of some Ore Minerals	Appendix

LIST OF PLATES

Page

1 a.	Hue circuit of high saturation	7 a
1 b.	Red-to-Yellow series giving approximately equal subjective intervals of hue	7 a
2 a.	Polar arrangement of various saturations of several hues	8 a
2 b.	Grey-to-Green series giving approximately equal subjective intervals of saturation	8 a
3.	Additive combination of red, green and blue producing white	18 a
4.	1931 C.I.E. chromaticity diagram with approximate distribution of colours for daylight adaptation	35 a

~~#~~ 2. 285 or 276. They only
exist in some
in abundance

APPENDICES

Page

X. Tables of Quantitative Colour Values & Spectral
Reflectivity Data of some Ore Minerals

Isotropic Minerals Abbreviations for Figs. 1a & 1b.

(values from about full wavelength scale data)

Pure Silver		A.1
Gold-Silver Alloy		A.2
Pure Gold		A.3
Pyrite	Py	A.4-18
Cobaltite	Cob	A.19
Galena	Gal	A.20-26
Kutinaite	Kut	A.27
Cattierite	Cat	A.28
Goldfieldite	Goldf	A.29-31
Sulvanite	Sul	A.32-35
Colusite	Col	A.36-41
Tetrahedrite	Tet	A.42-52
Tennantite	Ten	A.53-63
Vaesite	Vae	A.64
Germanite	Ger	A.65-70
Bornite	Bor	A.71-82
Bixbyite	Bix	A.83
Magnetite	Mag	A.84
Jacobsite	Jac	A.85
Franklinite	Fran	A.86
Magnesioferrite	Magn	A.87
Marmatite	Marm	A.88
Sphalerite	Sph	A.89-120
Manganosite	Mang	A.121

APPENDICES (continued)

		Page
Chromite	Chr	A.122-139
(values from 4 wavelength data)		
Zvyagintsevite	Zvya	A.140
Copper	Copp	A.140
Sperrylite	Sperr	A.141
Clausthalite	Clau	A.141
Skutterudite	Skut	A.142
Hollingworthite	Holling	A.142
Ru-Hollingworthite	Ru-Holling	A.143
Gersdorffite	Gers	A.143-144
Polydymite	Pol	A.145
Irarsite	Ira	A.145-146
Tyrrelite	Tyr	A.146
Pentlandite	Pent	A.147
Ullmanite	Ull	A.147
Laurite	Lau	A.148-150
Petzite	Pet	A.150-152
Coloradoite	Colo	A.152-153
Murdochite	Mur	A.154
Titanomagnetite	Titanomag	A.155
Anisotropic Minerals		
(values from about full wavelength scale data)		
Kostovite	Kos	A.156
Marcasite	Marc	A.157-159
Niccolite	Nicc	A.160
Loellingite	Loe	A.161
Sylvanite	Syl	A.162-163
Chalcopyrite	Chalcopy	A.164-178
Geochronite	Geo	A.179-180

also 132

APPENDICES (continued)

		Page
Boulangerite	Bou	A.181-182
Heteromorphite	Hete	A.183
Meneghinite	Mene	A.184-185
Zinkenite	Zin	A.186-187
Jamesonite	Jam	A.188-189
Plagionite	Plag	A.190
Hexagonal Pyrrhotite	Pyrr	A.191
Hodrushite	Hod	A.192
Pyrolusite	Pyro	A.193
Alpha-MnO ₂ Nsutite	-Nsu	A.194
Idaite	Ida	A.195-200
Orange Bornite	Or-Bor	A.201-212
Mawsonite	Maws	A.213-215
Stannite Jaune	Stan-J	A.216-221
Stannite	Stan	A.222-227
Renierite	Ren	A.228-234
Luzonite	Luz	A.235-237
Stibiolumonite	Stibiolum	A.238-239
Briartite	Bri	A.240-242
Coronadite	Coro	A.243
Cryptomelane	Cryp	A.244
Hollandite	Holl	A.245
Enargite	Enar	A.246-257
Kosterite	Kost	A.258-263
Hematite	Hem	A.264
Ti-Hematite	Ti-Hem	A.265
Ilmanite	Ilm	A.266
Crednerite	Cred	A.267

APPENDICES (continued)

		Page
Chalcophanite	Chalcoph	A.268
Psilomelane	Psi	A.269
Gallite	Gall	A.270-272
Braunite	Brau	A.273
Hausmannite	Haus	A.274
Pyrochroite	Pyroch	A.275
Hetaerolite	Heta	A.276
Groutite	Grou	A.277
(values from 4 wavelength data)		
Antimony		A.278
Dyscrasite	Dysc	A.278
Paradocrasite	Par	A.279
Melonite	Melo	A.280
Bismuth	Bis	A.280
^{β} Kit ite	Kit	A.281
Rammelsbergite	Ramm	A.281-282
Millerite	Mill	A.282
Arsenopyrite	Ars	A.283
Stannopalladinite	Stannop	A.284
Arsenic	Arse	A.285 ⁺
Breithaupite	Brei	A.286
Schapbachite	Schapb	A.286
Stibnite	Stib	A.287
Hessite	Hess	A.288
Bismuthinite	Bism	A.289
Launayite	Laun	A.289
Veenite	Vee	A.290
Sorbyite	Sor	A.290

APPENDICES (continued)

		Page
Twinnite	Twinn	A.291
Tintinaite	Tint	A.292
Playfairite	Play	A.292
Sterryite	Ster	A.293
Guettardite	Guet	A.294 } 3
Bournonite	Bourn	A.295
Chalcocite	Chalc	A.296
Eskebornite	Eskeb	A.297
Getchellite	Get	A.297
Stannoidite	Stann	A.298
Cinnabar	Cin	A.298-299
Magnetoplumbite	Magneto	A.300
Rutile		A.300
Nb-Rutile		A.301
Tapiolite	Tap	A.301-302
Plattnerite	Plat	A.302
Wolframite	Wolf	A.303
Wodginite	Wod	A.303
Umangite	Uman	A.304
Starlingite	Star	A.305
Klockmannite	Kloc	A.305
Cassiterite	Cass	A.306
Geikielite	Gei	A.306
Covellite	Covel	A307-308
XI. Table XI 1. Mean Transmittance Values of some Cut Gemstones		A.309
XI Table XI 2. Quantitative Colour Values of some Cut Gemstones		A.310

APPENDICES (continued)

	Page
XI. Figure XI 1. Chromaticity Diagram with Colour Points of some Cut Gemstones	A.311
XI. Figures XI 2-4. Spectral Transmittance Curves of Gemstones	A.312-314
XII. Figure XII 1. Showing marked variations in Spectral Reflectivity values quoted for specific minerals	A.315
III.1 PL/1 Computer Programme "SPECTR" to compute Spectral R	A.316
VI 1a. PL/1 Computer Programme "REFLECT" to Compute Tristimulus Values & Chromaticity Coordinates	A.317-319
1b. Example of listing data for the Programme "REFLECT"	A.320-321
VI 2a. PL/1 Computer Programme "SLOPE" to compute ratios of slopes of constant dominant wavelength	A.322
2b. Example of listing data for the Programme "SLOPE"	A.323
VI 3. Tabular Ratios	A.324-330
VI 4a. PL/1 Computer Programme "CHROM" to compute Dominant & Complementary Wavelengths & Purities	A.331-333
4b. Example of listing data for the Programme "CHROM"	A.334
VIII 1. PL/1 Computer Programme "COLOR" to compute Normalized Spectral Distributions, Tristimulus Values & Coordinates of a Light Source	A.335-336

I.

INTRODUCTION

The colour of a mineral is one of the simplest qualitative properties that can be used as an aid to distinguishing it from other minerals, even though the same mineral may show different colours and different minerals may have the same colour.

The quantitative measurement of colour involves several theoretical and practical difficulties. Other physical properties that can be measured more easily have therefore been used for mineral identification and it is only relatively recently that an interest in the quantitative measurement of colour in mineralogy has developed.

One branch of mineralogy in which colour is of prime importance is gemmology, where small colour differences may greatly affect the relative value of two otherwise identical specimens. Once a gemstone has been faceted, transmitted light methods of determining its spectral transmission are difficult or impossible to apply, so it was decided that an investigation of the use of reflected light methods would be made.

At the same time, the application of similar methods to the measurement of colour in opaque minerals was also investigated, since it promised to be a useful additional method for the identification of these minerals.

It was found that the use of reflected light method in non-opaque minerals including gemstones was unfavourable for purposes of quantitative colour measurement.

After a theoretical study therefore, an alternative method was developed using a matching liquid to determine the spectral transmission of a faceted gemstone for colour comparison.

Finally, it was also decided to investigate some colour problems in minerals, such as changes in the observed colour of a mineral that occur as its grain size is reduced.

II. LITERATURE REVIEW ON THEORY OF COLOUR MEASUREMENT

II.1. INTRODUCTION

The art of coloring started many thousands of years ago, but the science of colour began in the last centuries. It was founded by Isaac Newton (1666) who separated the components of homogeneous white light (the sun light) into a sequence of coloured rays. In recent years modern methods of colour measurement have been developed. Concepts, methods and quantitative data have become standardized and used by colorimetrists and research workers.

Despite the fact that colour is an important physical property made use of in mineralogy, and particularly in gemmology, in various ways (e.g. specimen colour, colours seen under the microscope in polarised light, etc.), there has been comparatively little use or understanding of the concepts of quantitative colour measurement.

A summary of the necessary basic information has therefore been prepared from the literature, which is abundant. Further details if required may be found in the following useful books, and periodical papers listed in the reference section.

- (1) 'The Science of Color' by the Committee on Colorimetry of the Optical Society of America, 1968.
- (2) 'Color Science' by Gunter Wyszecki and W.S. Stiles, 1967.
- (3) 'The Measurement of Colour' by W.D. Wright, 1969.
- (4) 'Physical Aspects of Colour' by Dr. P.J. Bouma, 1947.

- (5) 'Color in Business, Science and Industry' by D.B. Judd & G. Wyszecki,
1963.
- (6) 'Sources of Color Science' selected and edited by D.L. MacAdam,
1970.

In 1931 the Commission Internationale de l'Eclairage (C.I.E.) first adopted a set of data to define the colour-matching characteristics of standard observers (that is of the average eye) and established a reference framework for the specifications of colours. Armed with this reference framework, specifications of various standard stimuli (light sources) and a standard chromaticity diagram were derived.

Any colour is then determined with reference to a standard stimulus or standard stimuli and expressed by three numbers termed tristimulus values. It can also be represented in the standard C.I.E. chromaticity diagram and characterised by numbers, namely dominant (or complementary) wavelength, excitation purity and relative brightness.

Since basic data representing the normal eye as standardised by the C.I.E. colour specifications are independent of the colour vision of any particular individual, they are reproducible from one laboratory to another.

II.2.

PSYCHOPHYSICAL CONCEPTS OF COLOUR

Man has been aware of colour since immemorial time . Aristotle first realized that light is necessary for colour vision. It is now well known that colours are seen because of light. The phenomena of light have been explained by wave (Maxwell's electromagnetic) and quantum theories. Both theories have been accepted, as light has a dual nature, possessing both wave and corpuscular properties.

As far as the measurement of colour is concerned the concepts of light and colour need to be defined by the procedures which are prescribed for their measurements.

The measurement of colour is not a purely physical practice. Physiology and in some cases (such as sensation, perception, recall , recognition, imagination, motivation, feeling, emotion, reasoning,etc.) psychology are involved in the measurement, since colour - matching experiments to obtain basic data were made by human observers.

When a beam of light strikes the retina of the eye the colour sensation is interpreted by the mind. Colour sensation which may be defined as the primary conscious response to excitation of the visual mechanism has three attributes, hue , saturation and brightness as interpreted by the eye and mind. Sensation is specifically the direct result of consciousness of the present stimulation of the sense organs, as distinguished from perception, which includes the combination of different sensations and utilization of past experience in recognizing the objects and facts from which the present stimulation arises.

Objects are very commonly recognized and classified according to the colour of the light they reflect. The colour of an object depends upon the effect of its selective absorption, upon the spectral distribution of the incident radiant energy and upon the psychophysical functions of human vision.

It is obvious that subjects of light and colour are linked to the sciences of physics, chemistry, physiology, and in some cases psychology. Light and colour, therefore, are defined as psychophysical concepts.

II 3.

ATTRIBUTES OF COLOUR

Light is characterised by its frequency, velocity and wavelength. Since the velocity is constant in a given medium, the frequency and wavelength are inversely proportional. However, wavelength is usually quoted in preference to frequency, as the numbers required are smaller. Table II 1 shows the colour names given to light of certain wavelength ranges in approximate divisions.

Table II 1

Wavelength (nm)	Colour name
380 - 450	Violet
450 - 490	Blue
490 - 560	Green
560 - 590	Yellow
590 - 630	Orange
630 - 780	Red

There is, of course, a continuous series of pure colours gradually passing from the one into the others. They are known as hues and illustrated, for example, in Plate 1.

Each hue (spectrum colour) has several different colours (having the same brightness) when it becomes progressively paler till it approach white. The progression from a spectrum colour (retaining its hue throughout towards white is said to be one of desaturation. The aspect in which the individual members of such series differ, namely the property

of being more or less white, we call their saturations. Plate 2 shows a green hue giving a gradual change of saturation. The saturation increases progressively from zero in the case of white sample (achromatic) to a maximum in the case of a highly saturated colour sample (chromatic).

A colour retaining its two properties, hue and saturation, will give rise to a slightly different sensation of colour when its brightness level is changed. In other words if two samples have the same hue and saturation but differ in brightness they will appear to be different colours.

Thus the colour sensation has three attributes, namely hue, saturation and brightness. Expressed differently, therefore, a colour sensation is completely determined by these three magnitudes .

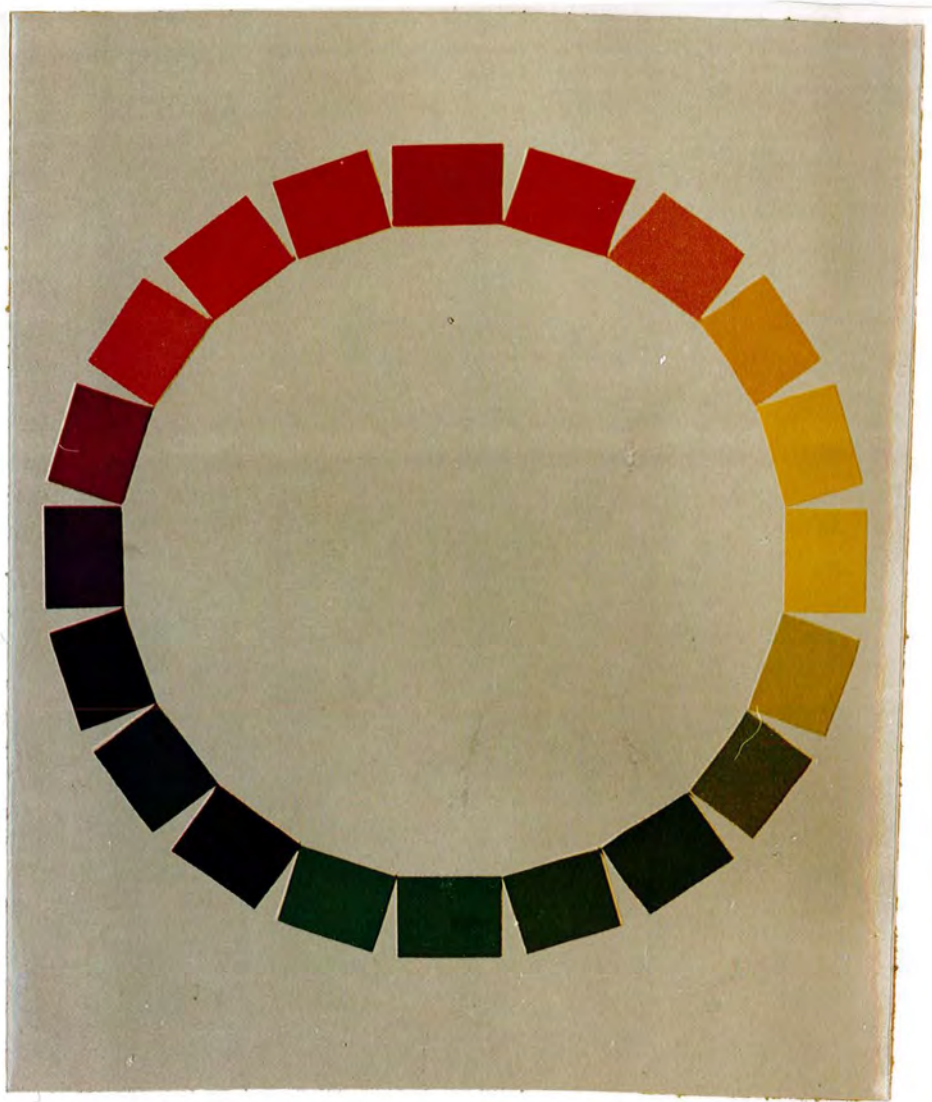


Plate 1 a. Hue circuit of high saturation, brightness varying to give the highest saturation of each hue.
(from Plate 19 in the 'Science of Color' published by the Committee on Colorimetry Optical Society of America, 1968)

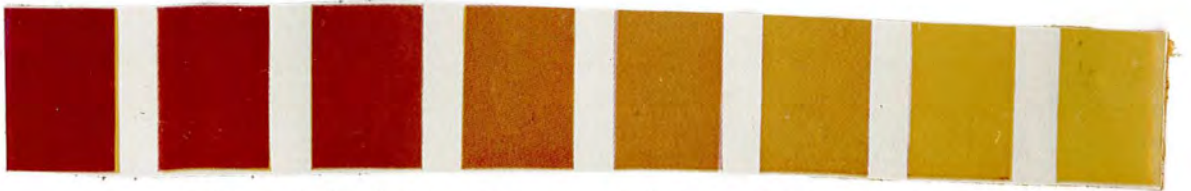


Plate 1 b. Red-to-Yellow series giving approximately equal subjective intervals of hue.
(from Plate 8 facing Page 102 in the 'Measurement of Colour' by W.D.Wright, 1969)

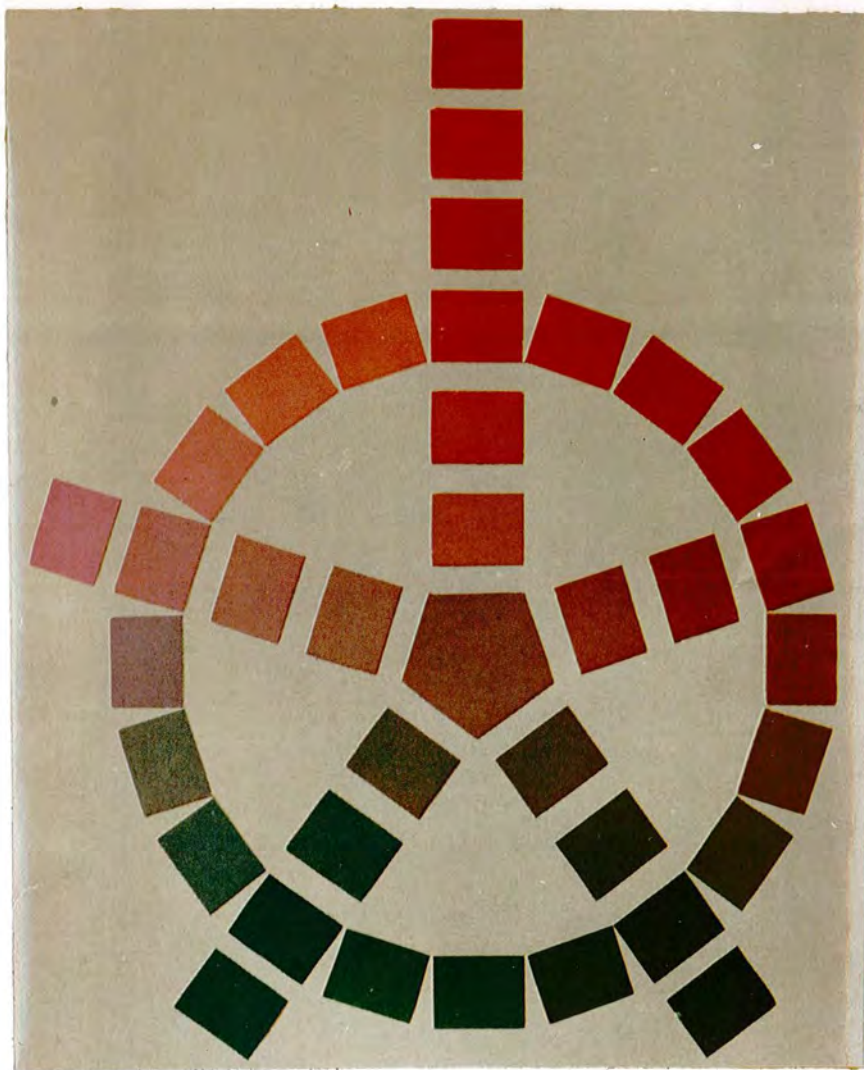


Plate 2 a. Polar arrangement of various saturations of several hues, all having the same brightness.

(from Plate 18 in the 'Science of Color' published by the Committee on Colorimetry Optical Society of America, 1968)

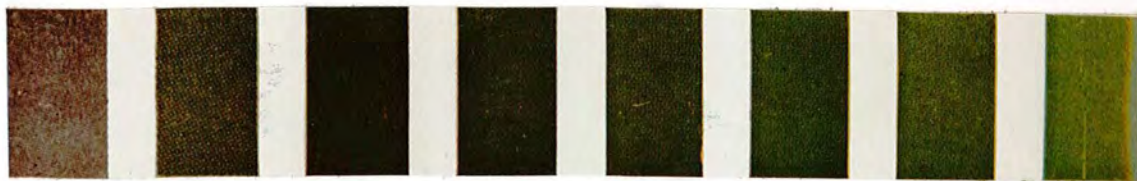


Plate 2b. Grey-to-Green series giving approximately equal subjective intervals of saturation.

(from Plate 8 facing Page 102 in the 'Measurement of Colour' by W.D.Wright, 1969)

II.4.

THE SENSITIVITY OF THE EYERelative Luminous Efficiency Functions

The most important natural light source is the sun. Many spectroradiometric measurements of the spectral energy distributions of the sun have been made and reported in the literature (Abbot et al 1923; Kimball 1928; Moon 1940; Taylor and Kerr 1941; MacAdam 1958 ; Henderson and Hodgkiss 1963; and Budde 1963). The significant spectral distribution data taken from Wyszecki and Stiles (1967) are shown in Figures II.1 and II.2.

Figure II.2 contains relative spectral irradiance distributions of daylight at correlated colour temperatures 5500, 6500 and 7500°K. They are considered as the most typical spectral distributions of irradiance produced by daylight at the earth's surface and are recommended as guides in the development of sources of artificial daylight.

The above data show that the spectrum of the sun radiates approximately equal amounts of energy per unit wavelength interval throughout the visible range. But these equal amounts of energy of different wavelengths do not produce visible sensation having equal brightnesses. In other words the brightness sensitivity of the eye is not equal for all wavelengths of the visible spectrum. The sensitivity of the eye to radiant energy depends upon the wavelength.

The conversion of light energy into nervous energy takes place in the light-sensitive retina of the eye. The retina contains two kinds of light sensitive elements, rods and cones. Rods are receptors

which function in the low level of illumination (scotopic vision) , whereas cones operate at high illumination levels (photopic vision). The luminance level of the normal photopic range is from about one to fifteen foot-lamberts* (Brown, 1951).

Important investigations of the spectral sensitivity of the eye were made by Gibson and Tyndall (1923), Crawford (1949), Judd and Wyszecki (1963), Wright (1946) and others. In such investigations spectral sensitivity was recorded subjectively by finding the energy required at each wavelength in the visible spectrum to produce a response of constant brightness; for cones, the observations were made under photopic conditions or by using the foveal area of the retina, while the rod sensitivity was obtained at near-threshold illumination using the extra-fovea (Wright, 1969). The sensitivity curve was then given by plotting the inverse of the energy obtained against wavelength.

Since only the relative sensitivity to the various wavelengths was of interest, all values were multiplied by a constant so that the maxima were equal to unity. These values thus obtained are termed the relative luminous efficiency functions.

Figure II.3 shows scotopic (V'_λ) and photopic (V_λ) curves as standardised in 1951 and 1924 respectively by the C.I.E. The rods have their maximum sensitivity at a wavelength of 507 nm and the cones at one of about 555 nm . The curves as a whole have been shifted with respect to each other.

* 1 foot-lambert = 0.0003426 candle per square centimeter (CCOSA 1968).

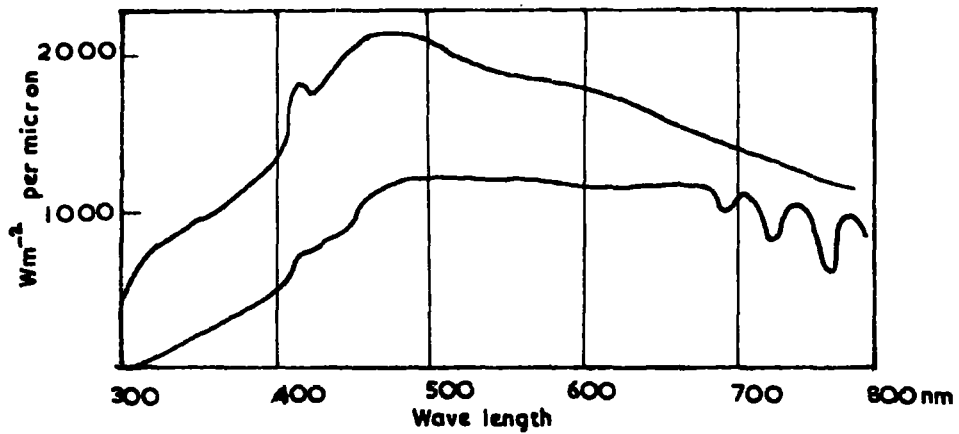


Figure 11.1. Spectral distributions of irradiance of direct sunlight in absolute units ($\text{Wm}^{-2} \mu\text{m}^{-1}$).
 upper curve: above atmosphere,
 lower curve: at sea level. (from Moon, 1940)

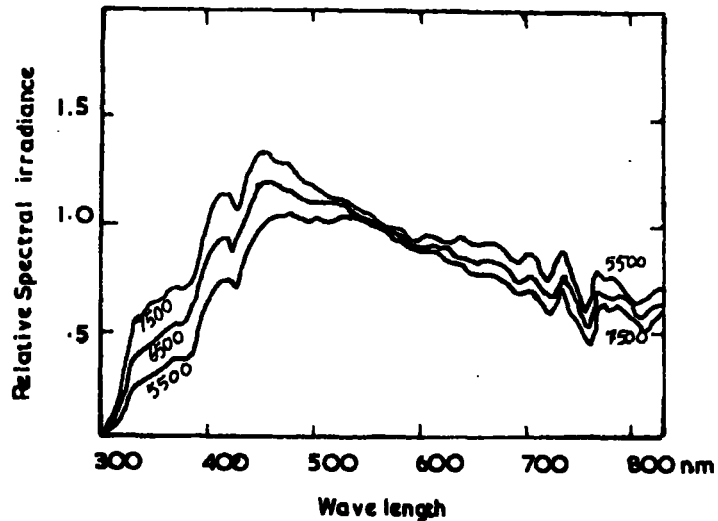


Figure 11.2. Relative spectral distributions of irradiance of direct sunlight, of correlated colour temperatures, 5000, 6500, 7500° K.
 (from Judd, Myszczek and MacAdam, 1964)

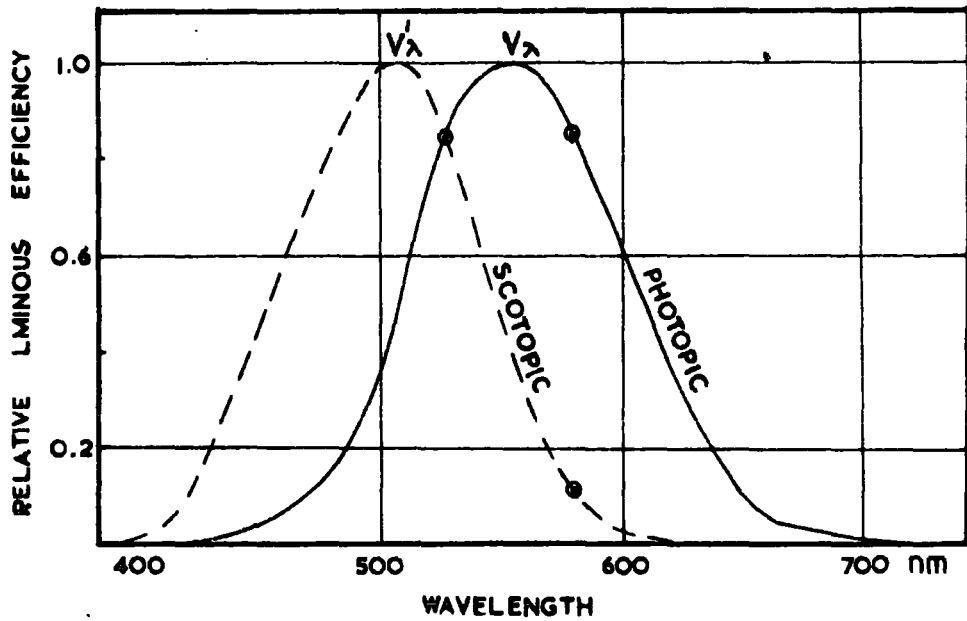


Figure 11.3. Standard relative luminous efficiency functions for photopic (V_λ) and scotopic (V'_λ) vision, (from Wyszecki and Stiles, 1967) and also showing the Perkinje effect.

A considerable difference between the two functions is that the relative brightness in different coloured objects alters when the quantity of light is greatly reduced. If one half of the field of vision, for instance, is strongly illuminated with yellow light of 581 nm and the other half with green light of 530 nm the eye sensitivity for these two wavelengths is equal. When the illumination levels are reduced by the same factor, the powers of the two halves have indeed remained equal but the eye sensitivity has become quite different for the two lights.

With these equal powers the green light has a much stronger apparent brightness than the yellow light (Figure II.3). This is known as the Purkinje effect. Expressed differently, in order to obtain the same apparent brightness we shall have to transmit much more energy per second of violet or red light to the eye than the green light.

It became necessary to select one of the relative efficiency functions to be used to obtain standard data specifying visual sensitivity for colour differences. When colours are observed at low luminance levels such as under twilight conditions or in areas of high density in colour transparencies, the scotopic function (V'_{λ}) would be useful to obtain data.

Normally, colour differences are observed under high illumination levels (i.e., higher than one foot-lambert) and therefore the photopic (V_{λ}) function was selected by the C.I.E. and used to derive the basic colour-matching functions for general purposes of colour measurements.

It was later verified from the experimental data by Brown (1951) that sensitivity to colour differences remains constant for a

normal observer until the field luminance drops below about one foot-lambert. Below this level the colour discrimination becomes poorer, slowly at first, and then rapidly.

Sensitivity To Differences Of Hue And Saturation

In the same way that the sensitivity of the eye to small differences of brightness varies, so there are sensitivities to small differences of hue (wavelength) and saturation (purity).

Hue discrimination was measured by setting just-noticeable differences of wavelengths. Starting with the two halves of the photometric field illuminated by light of the same wavelength λ and luminance (brightness), the wavelength of one half of the field was then gradually changed to $\lambda \pm d\lambda$, at the same time maintaining the brightness match between the two halves, until a just-noticeable chromaticness* difference was detected (Wright and Pitt, 1934; Bedford and Wyszecki, 1958). The average difference $d\lambda = [(+d\lambda) + (-d\lambda)]/2$ was obtained from several repeated measurements for each comparison wavelength and plotted against λ .

All curves have the same general appearance as shown in Figure II.4. It is regarded only as typical, since the differences between individuals are fairly large.

* Chromaticness = hue and saturation, taken together, expressive of the quality of colour sensation as distinguished from its intensity (CCOSA 1968).

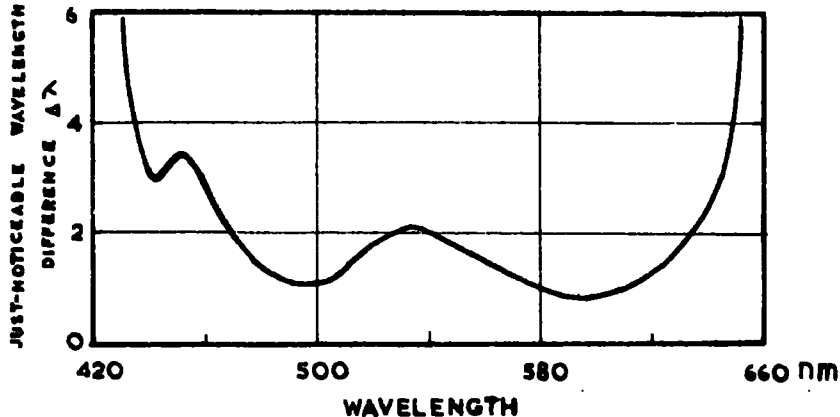


Figure II 4. Typical wavelength discrimination for a normal observer using a 2° field of view. (Wright and Pitt, 1934)

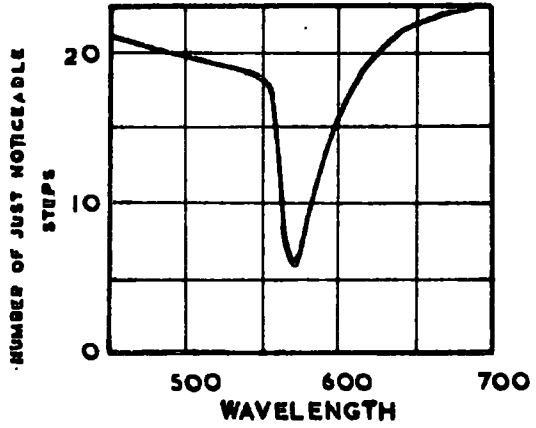


Figure II 5. Average purity discrimination as a function of wavelength (Martin et al, 1933, and Jones and Lowry, 1926)

Figure II.4 shows that the difference of wavelength which can just be detected by the normal eye varies irregularly from end of the spectrum to the other. A wavelength difference of about 1 nm will give rise to a just-noticeable chromaticness difference in the yellow-orange (590 nm) part of the spectrum and again in the blue-green (490 nm) under a 2-degree field of observation. Under large field conditions small differences of wavelength might be detectable in these regions of the spectrum (Wright, 1969). Hue discrimination rapidly becomes poor at both ends of the visible spectrum.

The sensitivity of the eye to differences of saturation is expressed by the number of just-noticeably different mixtures of white with light of any given wavelength. In the experiments additive mixtures of a monochromatic light of a given wavelength and a given achromatic (white) light were produced in two halves of a photometric field. The luminance of the both lights can be varied independently, thus providing a series of colours ranging from a pure white to a spectrum colour.

The observer was required to determine just-noticeable differences or equally differences of saturation by changing the colour of one half of the photometric field (Martin et al , 1933).

An alternative method was to make repeated colour matches and deduce the just-noticeable difference from the standard deviation of the matches (MacAdam , 1942).

In both cases it was found that the sensitivity to saturation was least. (i.e. the number of steps between white and the spectral radiation was

least) for wavelength in the yellow part (570 nm) of the spectrum, as shown in Figure II.5 (taken from Wyszecki and Stiles, 1967). This implies that a monochromatic light of wavelength 570 nm is in some sense less saturated than light of any other wavelength.

In conclusion, the photopic luminous efficiency functions V_λ play an important part in the derivation of colour-matching functions. The sensitivity of the eye to small differences of hue (wavelength) and saturation (purity) are of special interest and are major problems in measurements of colour discrimination, and are described in the later sections.

II.5. SPECTRAL ENERGY DISTRIBUTION AND COLOUR SENSATION

Experiments on radiant energy distribution of light sources showed that there is a relation between spectral composition of light and colour sensation. A white light sources, such as the sun, radiates more or less equal energy throughout the visible spectrum . In other words when the integrated white light is dispersed into distinguishable components an equal distribution of energy over the spectrum will result. If its spectral energies were plotted against wavelength a flat curve will be obtained. There is no region in which the energy is greater over other regions of the spectrum.

When the energy distribution at some wavelengths is lower the light in general appears coloured . A yellow stimulus of a source, for instance, is due to a high energy distribution at long wavelengths and low energy distribution in the shorter wavelength regions.

The spectral energy distribution (P_λ) is a purely numerical series as a function of wavelength. Wavelength and intensity of light are characteristics of the appropriate colour response. This is known as the colour stimulus (Q).

For object colours the same principle applies, because they act as secondary light sources. When radiant energy falls onto a surface of an object the spectral energy distribution (P_λ) of the incident light is altered after reflection by the surface of the object. The spectral energy distribution, the resultant of the combined

action of the incident light (P_λ) and reflection power (R_λ) of the object is equal to the colour stimulus of that object. that is,

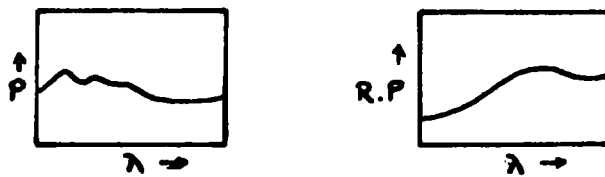
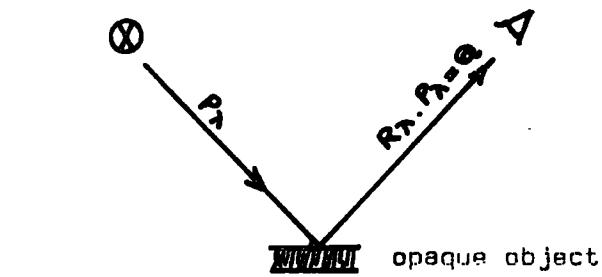
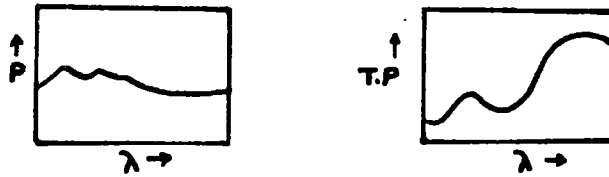
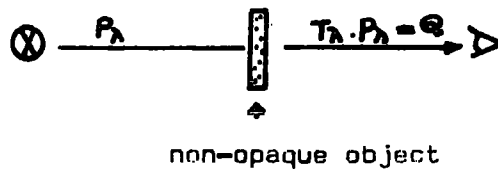
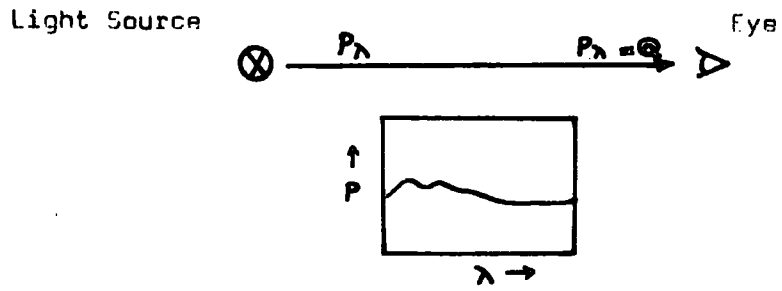
$$P_\lambda \cdot R_\lambda = Q.$$

This interrelations are the fundamental facts to define object colours and are illustrated in Figure II.6. Light sources in general having different energy distributions cause different colour response. It appears that different colour stimuli from an object caused by different light sources can be measured using the above general expression.

It can be assumed that the retina (cone) possesses three different types of light sensitive photo cells which are distinguished from one another by their spectral sensitivities. It can also be assumed that the first cell has its maximum sensitivity in the red part of the spectrum, the second in the green and the third in the blue. When a light beam stimulates the eye the three photo cells operate together simultaneously and form three magnitudes by which a colour sensation is determined.

The above assumption makes quantitative colour measurement possible by means of a photo-electric device functioning exactly the same as those of the retina. With such apparatus measurements of spectral reflectivity (or transmittance) on an object surface can be made and followed by appropriate calculation to derive the three magnitudes which will represent the attributes of colour sensation.

Figure II 6. Relation between spectral distribution and colour sensation.



II.6.

PRINCIPLES OF COLOUR MEASUREMENT

By experiments it was found that any colour could be produced by varying appropriate amounts of three suitable radiations (colours). The fundamental principles of colour measurement were first postulated by Grassmann (1854) and were familiar to Maxwell (1860 & 1872), Helmholtz (1866) and others.

Experimental checks were carried out later by Blottian (1947), Trezona (1954) and recently by Stiles (1963). Two significant investigations were made by Guild (1924 & 1931) and Wright (1928-1929) to determine the relative amounts of three specified radiations required to match the colours in the spectrum.

Their colour mixture data were used in 1931 by the Commission Internationale de l'Eclairage (C.I.E.) to standardize the colour-matching functions of a normal observer and to adopt a standard framework for colour specification.

One of the basic laws of colour theory would be expressed thus: By mixing three selected spectral colours in definite proportions any given colour sensation can be matched.

Grassmann's first law states that three primary colours can be selected so that it is impossible to match one of them by any mixture of the other two.

The primary colour stimuli employed in the colour-matching

experiments were saturated red, green and blue, thus satisfying the above law none of them can be produced by a mixture of the rest of two radiations. But with these primary colours a yellow can be produced by mixing the red and green, and a purple by mixing blue and red. All intermediate hues of successive degrees of desaturation can be produced by combining the stimuli in appropriate definite proportions. A white is obtained by mixing the correct proportions of all three stimuli and is demonstrated in Plate 3.

For a given colour Q , the colour equation can be conveniently expressed as

$$R(R) + G(G) + B(B) \equiv Q(Q) .$$

where R , G and B are amounts of the three primary colours (R), (G) and (B), and they are termed the tristimulus values required to produce the Q quantity of the (Q) colour. It is possible and convenient to represent colours in colour space by three dimensional vectors, and colour matches by linear equations between such vectors. The symbol Q , for instance, is a vector representing a colour and R, G, B are the vectors representing the amounts of three fixed primaries. A three dimensional vector diagram illustrating a colour space and a corresponding two dimensional colour plane are shown in Figure II.7.

Grassmann's third law says that if two colours Q_1 and Q_2 are produced by mixtures of three radiations, the colours together, when mixed additively, will be matched by the sum of the two mixtures similarly combined.

i.e., if

$$R_1 (R) + G_1 (G) + B_1 (B) \equiv Q_1 (Q) \quad \text{and}$$

$$R_2 (R) + G_2 (G) + B_2 (B) \equiv Q_2 (Q)$$

then

$$\frac{R_1 + R_2}{2}(R) + \frac{G_1 + G_2}{2}(G) + \frac{B_1 + B_2}{2}(B) = \frac{Q_1 + Q_2}{2}(Q) = Q_3 (Q)$$

The new colour Q_3 is thus obtained by the additive mixing of the two colours Q_1 and Q_2 and lies on the straight line connecting Q_1 and Q_2 , as shown in Figure II.8.

If m units of Q_1 are mixed with n units of Q_2 , $(m + n)$ units of a new colour Q_3 will be obtained. The colour Q_3 will lie at a distance on the line $Q_1 Q_2$ such that $Q_1 Q_3 / Q_3 Q_2 = n / m$.

For three or more colours the same principle applies. A convention was made to base the units of (R), (G) and (B) on a match on a white of some defined quality. For this purpose there was an assumption that white may be regarded as a colour (achromatic)* in which neither red nor green nor blue predominates. The standard white, therefore, could be defined by its colour temperature if it were a Plankian radiator (Black body), whereas the qualities of other colours (chromatic colours) could be defined by their dominant wavelength and purity corresponding, in an approximate way, to their hue and saturation respectively.

Based on such units the colour equation for the standard white (E) becomes:

$$1(E) = R(R) + G(G) + B(B)$$

* achromatic colour: colour that does not elicit hue.

white, grey, black are achromatic colours.

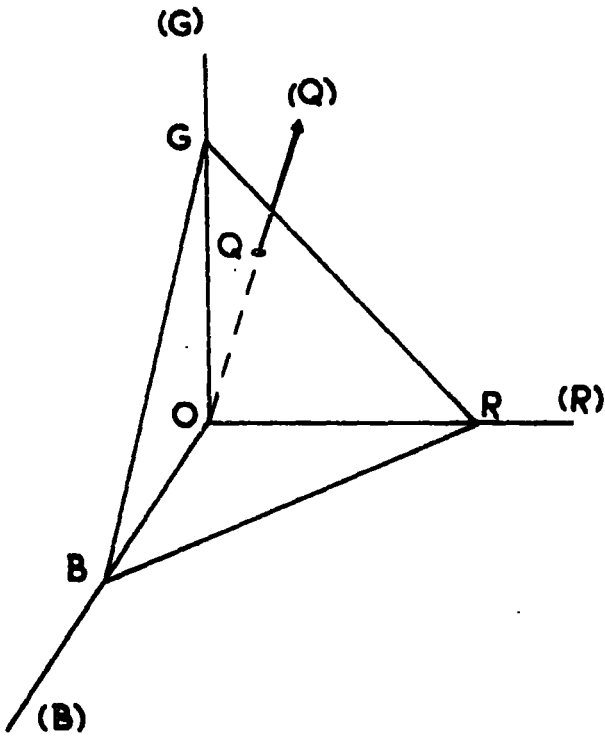


Figure 11 7. A three dimensional vector diagram illustrating a colour space and a corresponding two dimensional colour plane.

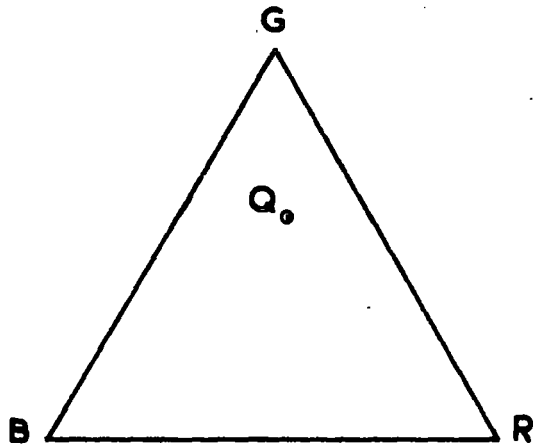




Plate 3. Additive combination of red, green and blue producing white. Partly overlapping mixture of red and green produces yellow, red and blue produces purple, blue and green produces blue-green.

(from Plate 6 in the 'Science of Color' by the Committee on Colorimetry Optical Society of America, 1968)

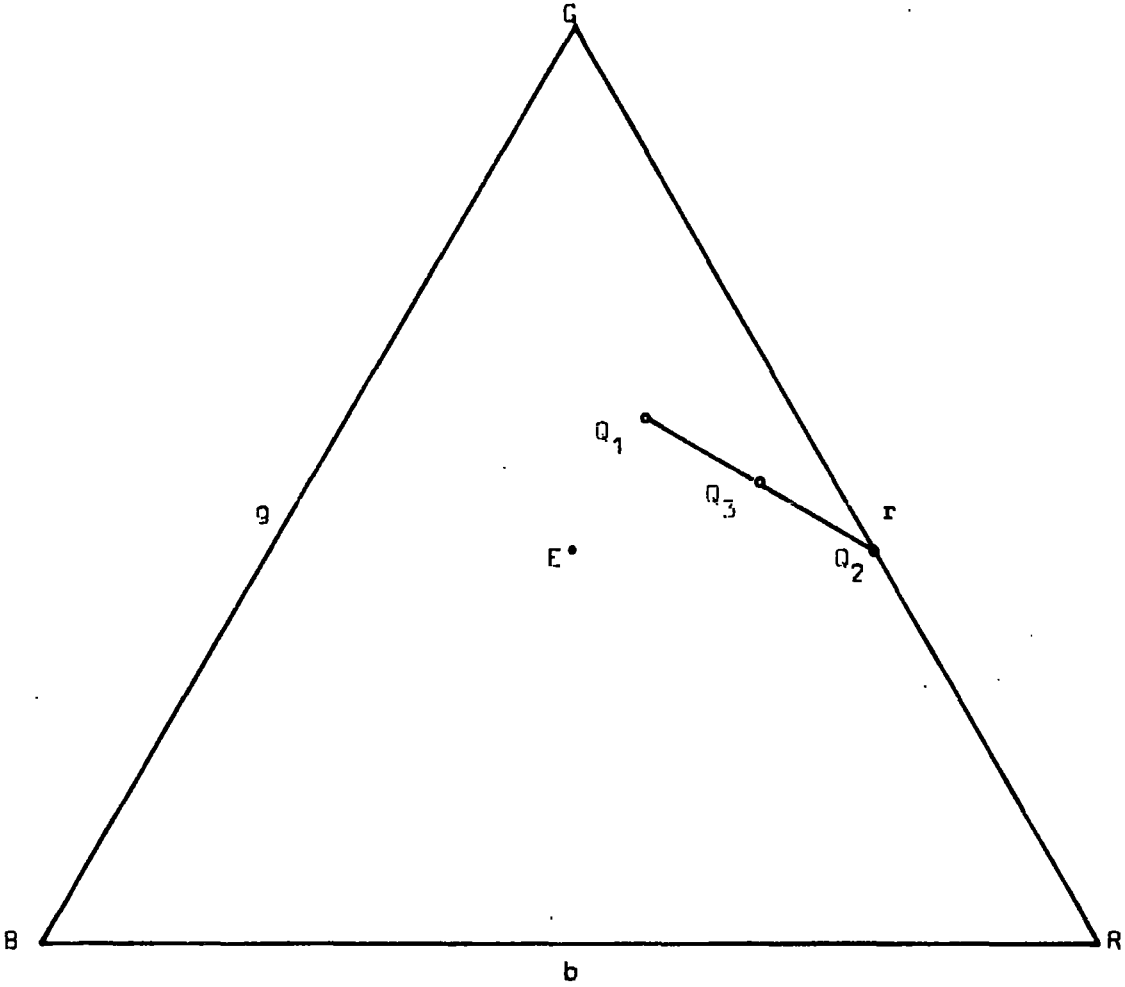


Figure II 8. Illustrating the additive mixture of the two colours Q_1 & Q_2 producing a new colour Q_3 .

$$r = \frac{R}{R + G + B} = 1 / 3$$

$$g = \frac{G}{R + G + B} = 1 / 3$$

$$b = \frac{B}{R + G + B} = 1 / 3 .$$

This particular white in fact is the centroid of the colour triangle. R, G, B are called tristimulus values and r, g, b are termed the chromaticity coordinates. By definition tristimulus values are components necessary in a three-colour mixture matching a sample colour.

Grassmann's second law states that if two different light spots give the same colour sensation they continue to do so if the brightnesses of both are increased or decreased by the same factor . It means that with any change in brightness of a colour, its tristimulus values increase proportionally.

All colours having the same hue and saturation but differing in brightness will be represented by one point in a colour plane. This is illustrated in Figure II.9. Therefore colours slightly different due to variation in brightness cannot be distinguished by the two-dimensional plane diagram.

Although the colour sensation is defined by three magnitudes only hue and saturation are taken into account in the colour plane, the brightness is ignored. Such a colour plane is called a chromaticity diagram. It is in fact a plane of constant brightness. The chromaticity (quality of hue and saturation) of a synthesized colour may be specified in terms of the proportions of each of two of the

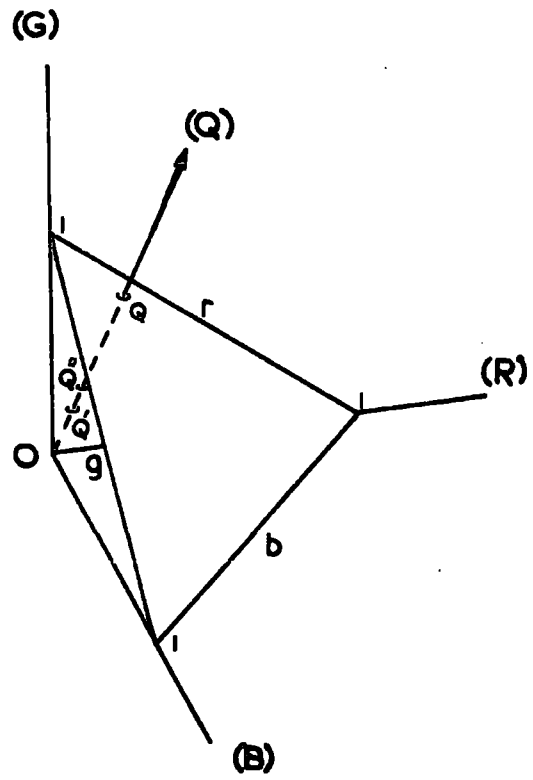


Figure 11 9. A vector diagram with a colour plane of constant brightness.

components in the trichromatic mixture.

From the above laws any additive mixture of the primary colours RGB will produce a colour point lying within the colour triangle formed by three spectral radiations RGB. Practically this additive colour-mixture system does not hold the fundamental rule which states that all spectrum colours must lie inside the colour triangle. Some of the spectrum colours lie slightly beyond the triangle RGB. They cannot be produced by positive mixtures of RGB. Therefore the fundamental law of additive colour-matching has a limitation in its application.

However the exceptions were eliminated by introducing subtractive amounts in the colour-matching. For example the colour of wavelength 510 nm cannot be produced by additive mixing of RGB but it can be matched in the following way expressed algebraically thus,

$$(510) + r(R) = g(G) + b(B).$$

$$\text{or} \quad (510) = -r(R) + g(G) + b(B).$$

Hence the spectrum colour (510) can be produced by a negative quantity of (R) plus two positive quantities of (G) and (B). In this manner the fundamental law holds universally.

Based on these fundamental criteria of colour-mixing theory, extensive experiments were made to specify the units of reference stimuli (R), (G) and (B) and their quantities required to match each wavelength colour in turn in the equal-energy spectrum.

To fulfill the above requirements important investigations were undertaken by Wright (1928-29 & 1930) and Guild (1931) independently, using very different instruments and with different groups of observers.

In Guild's experiment seven observers made colour matches through the spectrum. The instrumental stimuli were adjusted to be equal in a match on the N.P.L. white.

Wright used ten observers in his investigation and his reference stimuli were monochromatic radiations of wavelengths 650 (R), 530 (G) and 460 (B) nm. The units were adjusted so that equal amounts of the red and green stimuli were required in the match of a monochromatic yellow of wavelength 582.5 nm and equal amounts of the green and blue in the match on 494.0 nm.

Both sets of data obtained by Guild and Wright were transformed by the C.I.E. to the same reference stimuli, namely 700.0, 546.1 and 435.8 nm, with the units based on the N.P.L. white. The mean results of the two investigations were compared and it was found that they were in good agreement. Therefore the average values of the Guild-Wright investigations were accepted as basic data for further experiments. These data are representative of a normal eye and called the 1931 standard observer by definition.

In 1955 a redetermination of the colour-matching functions was made by Stiles (1955) and his data showed that no important errors were present in Guild-Wright data.

Figures II.10 and II.11 show the mean chromaticity coordinates of spectrum colours, obtained by Guild and Wright respectively. In both figures the coordinates add up to unity at each wavelength. But one or the other of the coordinates is always negative, because those colours indicated by negative coordinates cannot be matched by a positive

mixture of three components. In such case a match was established with a mixture of the sample light and one of the components. The amount of the component which is mixed with the sample light was taken as a negative quantity. It can be explained by the following expressions thus,

$$r(R) + g(G) = Q(Q) + b(B)$$

or
$$r(R) + g(G) - b(B) = Q(Q).$$

To obtain the amounts of (R), (G) and (B) in the equal-energy spectrum the relative luminous efficiency function (V_λ) was used (Figure II.3) instead of making direct colour matches on an equal-energy spectrum, because the light flux (luminance) at wavelength λ in the equal-energy spectrum is proportional to V_λ .

If the luminances are written as $V(R)$, $V(G)$, $V(B)$ the colour equation becomes

$$V(Q) = r.V(R) + g.V(G) + b.V(B).$$

the spectral colour with wavelength λ is

$$V(\lambda) = r V(R) + g V(G) + b V(B).$$

Since the light flux at wavelength (λ) in the equal-energy spectrum is proportional to V_λ , the unit quantities of the primaries in the equal-energy spectrum will be proportional to

$$\frac{V_\lambda}{V(\lambda)} = \frac{V_\lambda}{r V(R) + g V(G) + b V(B)} = m_\lambda$$

The value V_λ is already in existence (Figure II.3), the chromaticity coordinates, r_λ , g_λ , b_λ , and the reference stimuli, $V(R)$, $V(G)$, $V(B)$ are known experimentally. The colour-matching functions (also called distribution coefficients, or tristimulus values of the spectrum),

\bar{r}_λ , \bar{g}_λ , \bar{b}_λ , of the equal-energy spectrum are given by

$$\bar{r}_\lambda = m_\lambda r_\lambda$$

$$\bar{g}_\lambda = m_\lambda g_\lambda$$

$$\bar{b}_\lambda = m_\lambda b_\lambda .$$

Since only the relative values of the colour-matching functions are of interest, the coefficients for all wavelengths were multiplied by the same factor, and thus they were finally derived to satisfy the equation

$$V_\lambda = \text{a constant} [\bar{r}_\lambda V(R) + \bar{g}_\lambda V(G) + \bar{b}_\lambda V(B)]$$

Figure II.12 which was thus derived from Figures II. 11 and II.10 shows the spectral distribution curves for the equal-energy spectrum expressed in terms of the matching stimuli 700.0 nm, 546.1 nm and 435.8 nm with their units adjusted to be equal in a match on an equal-energy white.

Thus the spectrum colours were broken down into three parameters \bar{r}_λ , \bar{g}_λ and \bar{b}_λ measured in terms of the tristimulus values (or distribution coefficients).

These basic data had to be used to compute specifications of a given stimulus. The spectrum locus could be drawn by plotting the spectral chromaticity coordinates, giving the chromaticities of the spectrum colours. It is clear from Figure II.13 that part of the spectrum locus, unfortunately, lie outside the RGB triangle. Therefore it is not a suitable system for colour measurement purposes. The RGB system had to be transformed into a new reference system.

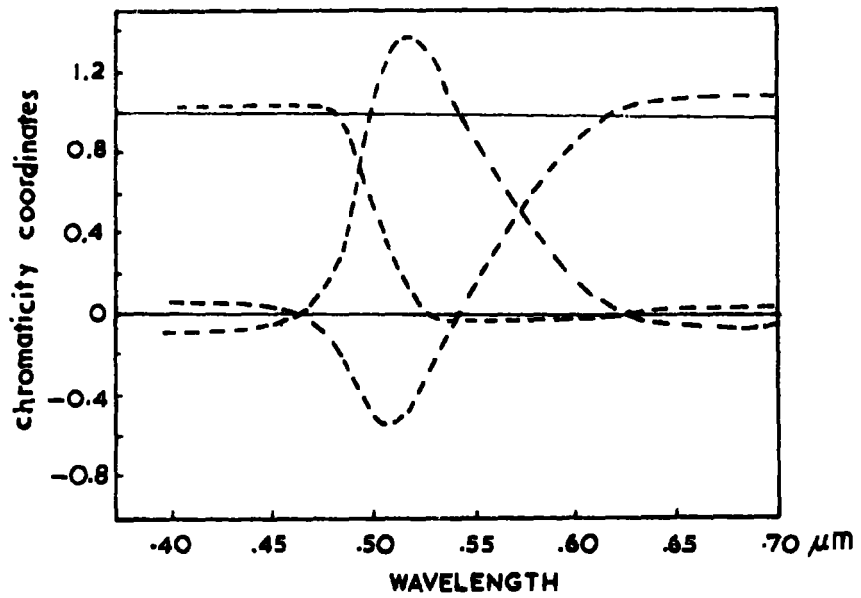


Figure II 10. Mean colour matching results obtained by Guild.
(from Guild, Phil. Trans., 1931)

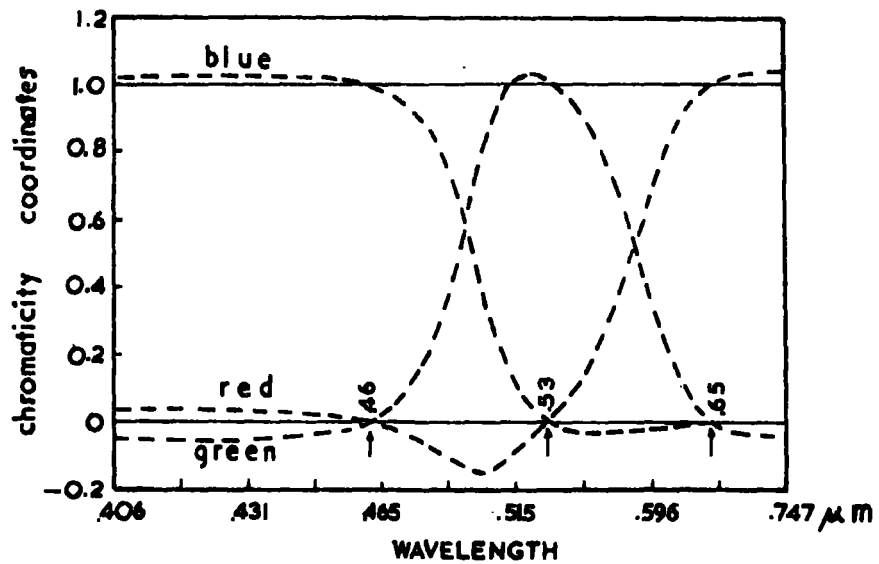


Figure II 11. Mean colour matching results obtained by Wright.
(from Wright, Trans. Opt. Soc., 1928-29)

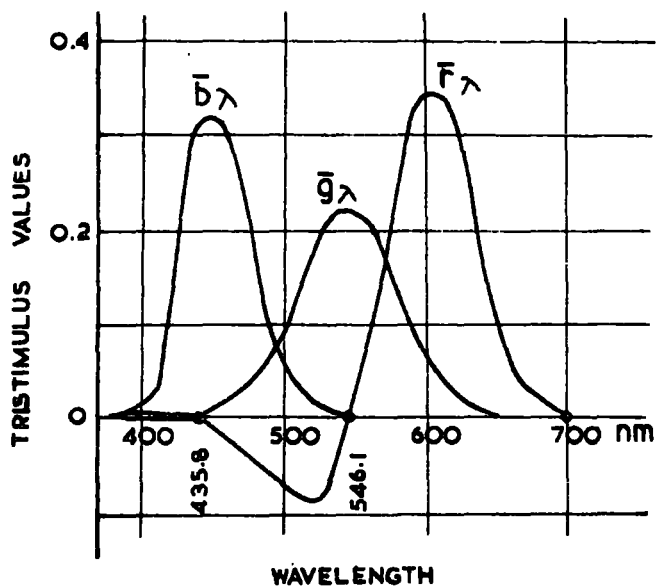


Figure II 12. Spectral distribution curves (colour-matching functions \bar{r}_λ , \bar{g}_λ , \bar{b}_λ) of equal energy spectrum in the primary system $R=700.0$ nm, $G=546.1$ nm, $B=435.8$ nm (from Wyszecki & Stiles, 1967)

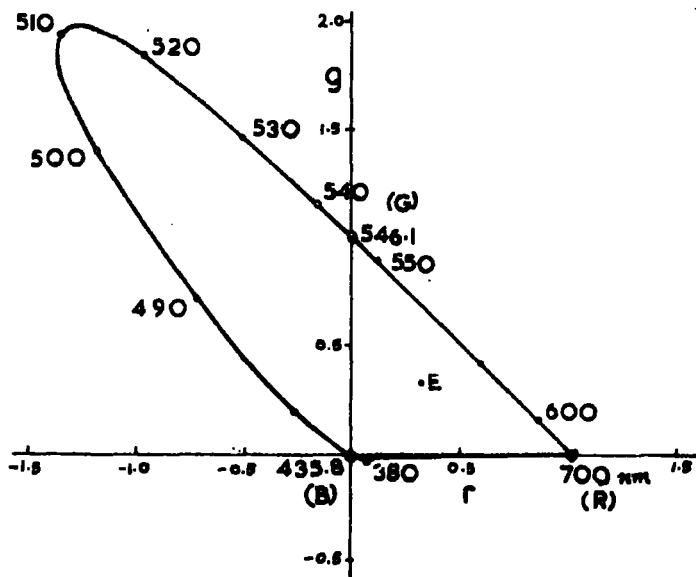


Figure II 13. Chromaticity diagram showing spectrum locus (380-700 nm) in the primary system R, G, B, equal-energy stimulus (E) (from Wyszecki & Stiles, 1967)

II.7. THE C.I.E. SYSTEM OF COLOUR MEASUREMENT

The principles of colorimetry were developed from the starting point of a colour-match with a visual trichromatic colorimeter. The whole of applied colorimetry is based on the primary RGB system. But the RGB system for the notation of colours in space and on a plane diagram is not convenient for practical use. The negative quantities of the chromaticity coordinates create difficulty in the sophisticated computations for colour specifications and cause the spectrum locus to lie outside the RGB colour triangle as shown in Figure II.13 .

The C.I.E. (Commission Internationale de l'Eclairage), a body which in 1931 took over the functions of the earlier 'Commission Internationale de Photometrie' was responsible for transforming the RGB system into one based on new primaries XYZ. This new system not only would make use of the experimental data underlying the RGB system but also would introduce changes chiefly intended to simplify computation in colorimetric calculations. Thus the new system was developed via a transformation of the RGB coordinates.

This is the transformation of Guild's and Wright's data by mathematical treatments into the new XYZ-CIE system so that the coordinates and tristimulus values of the spectrum are never negative. Details of the transformation equations, if required, from one system to another can be found in the text-books.

In this new system the colour-matching functions (\bar{r}_λ , \bar{g}_λ , \bar{b}_λ in the old RGB system) are denoted by the symbols \bar{x}_λ , \bar{y}_λ , \bar{z}_λ .

The 1931 C.I.E. colour-matching functions based on the XYZ primaries are presented by the curves in Figure II.14. Figure II.15 shows the CIE-XYZ standard chromaticity diagram in terms of the reference stimuli (X), (Y) and (Z) with unit based on an equal-energy (E) whose coordinates in the chromaticity diagram are $1/3$, $1/3$ and $1/3$. Table II.2 summarises the relationships between the 1931 CIE-RGB system and the 1931 CIE-XYZ system.

With the new set of reference stimuli XYZ the transformed system has the following characteristics:

- (1) Tristimulus values and coordinates of all spectrum colours are now positive, so the triangle formed by the chromaticity points XYZ completely encloses the spectrum locus.
- (2) The units of (X), (Y) and (Z) are adjusted, as in the RGB system, to be equal in a match of an equal-energy white, so that the equal-energy white (E) lies in the middle of the chromaticity triangle.
- (3) The colour-matching function \bar{y}_λ is set to be identical with the luminous efficiency function V_λ . It follows that the tristimulus values Y of a given colour is equal to its relative reflection or transmission factor.
- (4) As in the RGB system, the areas under the curves \bar{x}_λ , \bar{y}_λ , \bar{z}_λ of the spectrum are equal (Figure II.10).

that is,
$$\int \bar{x}_\lambda d\lambda = \int \bar{y}_\lambda d\lambda = \int \bar{z}_\lambda d\lambda$$

- (5) The reference stimuli XYZ cannot be reproduced experimentally in the laboratory. They are hypothetical stimuli, but they have real merits as reference coordinates of the C.I.E. system.

All colours will be represented by points within the spectrum locus and the purple line connecting the red and blue ends. There is no colour outside the curve.

Standard chromaticity coordinates x , y , z and the colour-matching functions \bar{x}_λ , \bar{y}_λ , \bar{z}_λ , of the equal-energy spectrum are given in Table II.3 (from Wyszecki and Stiles, 1967).

These colour-matching functions are to be used in calculations for any light source when its spectral energy distributions are known.

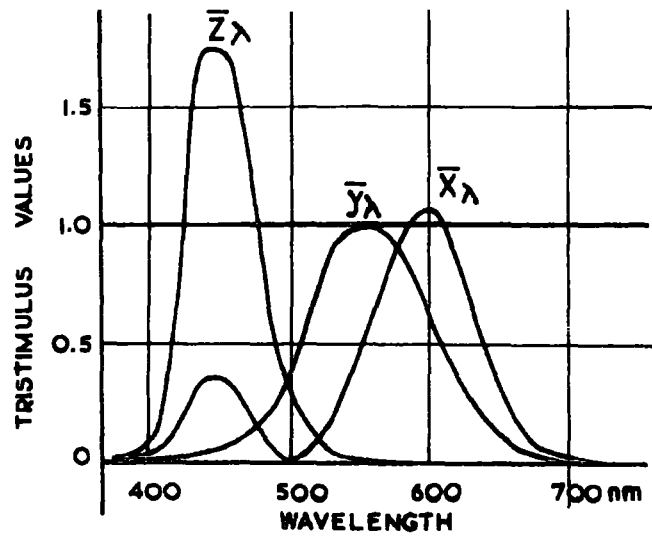


Figure 11 14. 1931 C.I.E. colour matching functions \bar{x}_λ , \bar{y}_λ , \bar{z}_λ .
 (from Wyszecki & Stiles, 1967)

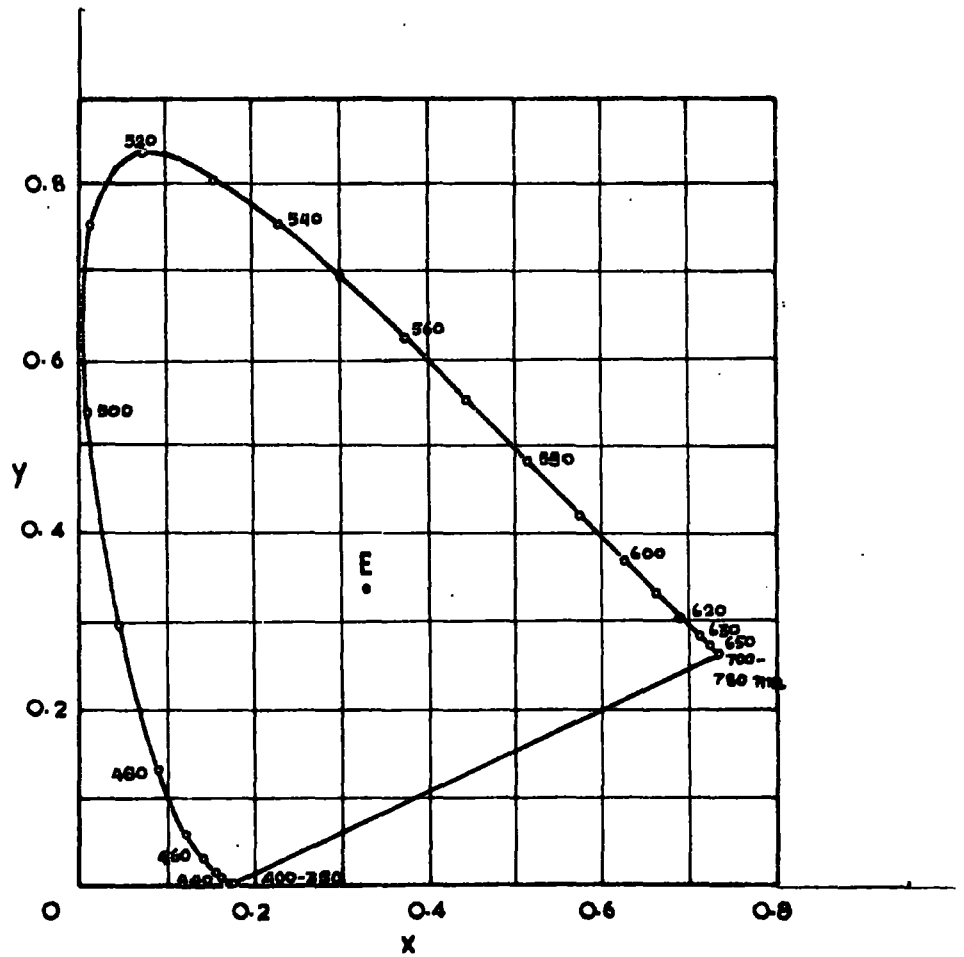


Figure 11 15. 1931 C.I.E. chromaticity diagram with spectrum locus, purple line, and the chromaticity point of equal-energy stimulus E (from Wyszecki & Stiles, 1967)

Table II 2.

Relationships between the 1931 C.I.E.-(R,G,B) System and the C.I.E.-(XYZ)
System of colour specification

Stimulus	R,G,B System (chromaticity coordinates)			X,Y,Z System (chromaticity coordinates)		
	r	g	b	x	y	z
(R) 700.0 nm	1	0	0	0.73467	0.26533	0.00000
(G) 546.1 nm	0	1	0	0.27376	0.71741	0.00883
(B) 435.8 nm	0	0	1	0.16658	0.00886	0.82456
Equal-energy source(E)	$\frac{1}{3}$	$\frac{1}{3}$	$\frac{1}{3}$	$\frac{1}{3}$	$\frac{1}{3}$	$\frac{1}{3}$
Source A	0.55255	0.32126	0.12619	0.44757	0.40745	0.14498
Source B	0.36230	0.34305	0.29465	0.34842	0.35161	0.29997
Source C	0.28226	0.33326	0.38448	0.31006	0.31616	0.37378

(from Wyszecki & Stiles 1967).

Table II 3. Chromaticity Coordinates and Color-Matching Functions
in 1931 CIE Colorimetric System (X, Y, Z)
($\lambda = 380 \cdots 780 \text{ nm}$; $\Delta\lambda = 5 \text{ nm}$)

Chromaticity Coordinates			Wavelength λ (nm)	Color-Matching Functions		
x_λ	y_λ	z_λ		\bar{x}_λ	\bar{y}_λ	\bar{z}_λ
0.1741	0.0050	0.8209	380	0.0014	0.0000	0.0065
0.1740	0.0050	0.8210	385	0.0022	0.0001	0.0105
0.1738	0.0049	0.8213	390	0.0042	0.0001	0.0201
0.1736	0.0049	0.8215	395	0.0076	0.0002	0.0362
0.1733	0.0048	0.8219	400	0.0143	0.0004	0.0679
0.1730	0.0048	0.8222	405	0.0232	0.0006	0.1102
0.1726	0.0048	0.8226	410	0.0435	0.0012	0.2074
0.1721	0.0048	0.8231	415	0.0776	0.0022	0.3713
0.1714	0.0051	0.8235	420	0.1344	0.0040	0.6456
0.1703	0.0058	0.8239	425	0.2148	0.0073	1.0391
0.1689	0.0069	0.8242	430	0.2839	0.0116	1.3856
0.1669	0.0086	0.8245	435	0.3285	0.0168	1.6230
0.1644	0.0109	0.8247	440	0.3483	0.0230	1.7471
0.1611	0.0138	0.8251	445	0.3481	0.0298	1.7826
0.1566	0.0177	0.8257	450	0.3362	0.0380	1.7721
0.1510	0.0227	0.8263	455	0.3187	0.0480	1.7441
0.1440	0.0297	0.8263	460	0.2908	0.0600	1.6692
0.1355	0.0399	0.8246	465	0.2511	0.0739	1.5281
0.1241	0.0578	0.8181	470	0.1954	0.0910	1.2876
0.1096	0.0868	0.8036	475	0.1421	0.1126	1.0419
0.0913	0.1327	0.7760	480	0.0956	0.1390	0.8130
0.0687	0.2007	0.7306	485	0.0580	0.1693	0.6162
0.0454	0.2950	0.6596	490	0.0320	0.2080	0.4652
0.0235	0.4127	0.5638	495	0.0147	0.2586	0.3533
0.0082	0.5384	0.4534	500	0.0049	0.3230	0.2720
0.0039	0.6548	0.3413	505	0.0024	0.4073	0.2123
0.0139	0.7502	0.2359	510	0.0093	0.5030	0.1582
0.0389	0.8120	0.1491	515	0.0291	0.6082	0.1117
0.0743	0.8338	0.0919	520	0.0633	0.7100	0.0782
0.1142	0.8262	0.0596	525	0.1096	0.7932	0.0573
0.1547	0.8059	0.0394	530	0.1655	0.8620	0.0422
0.1929	0.7816	0.0255	535	0.2257	0.9149	0.0298
0.2296	0.7543	0.0161	540	0.2904	0.9540	0.0203
0.2658	0.7243	0.0099	545	0.3597	0.9803	0.0134
0.3016	0.6923	0.0061	550	0.4334	0.9950	0.0087
0.3373	0.6589	0.0038	555	0.5121	1.0002	0.0057
0.3731	0.6245	0.0024	560	0.5945	0.9950	0.0039
0.4087	0.5896	0.0017	565	0.6784	0.9786	0.0027
0.4441	0.5547	0.0012	570	0.7621	0.9520	0.0021
0.4788	0.5202	0.0010	575	0.8425	0.9154	0.0018
0.5125	0.4866	0.0009	580	0.9163	0.8700	0.0017

Table II 3 (continued)

Chromaticity Coordinates			Wavelength λ (nm)	Color-Matching Functions		
x_λ	y_λ	z_λ		\bar{x}_λ	\bar{y}_λ	\bar{z}_λ
0.5125	0.4866	0.0009	580	0.9163	0.8700	0.0017
0.5448	0.4544	0.0008	585	0.9786	0.8163	0.0014
0.5752	0.4242	0.0006	590	1.0263	0.7570	0.0011
0.6029	0.3965	0.0006	595	1.0567	0.6949	0.0010
0.6270	0.3725	0.0005	600	1.0622	0.6310	0.0008
0.6482	0.3514	0.0004	605	1.0456	0.5668	0.0006
0.6658	0.3340	0.0002	610	1.0026	0.5030	0.0003
0.6801	0.3197	0.0002	615	0.9384	0.4412	0.0002
0.6915	0.3083	0.0002	620	0.8544	0.3810	0.0002
0.7006	0.2993	0.0001	625	0.7514	0.3210	0.0001
0.7079	0.2920	0.0001	630	0.6424	0.2650	0.0000
0.7140	0.2859	0.0001	635	0.5419	0.2170	0.0000
0.7190	0.2809	0.0001	640	0.4479	0.1750	0.0000
0.7230	0.2770	0.0000	645	0.3608	0.1382	0.0000
0.7260	0.2740	0.0000	650	0.2835	0.1070	0.0000
0.7283	0.2717	0.0000	655	0.2187	0.0816	0.0000
0.7300	0.2700	0.0000	660	0.1649	0.0610	0.0000
0.7311	0.2689	0.0000	665	0.1212	0.0446	0.0000
0.7320	0.2680	0.0000	670	0.0874	0.0320	0.0000
0.7327	0.2673	0.0000	675	0.0636	0.0232	0.0000
0.7334	0.2666	0.0000	680	0.0468	0.0170	0.0000
0.7340	0.2660	0.0000	685	0.0329	0.0119	0.0000
0.7344	0.2656	0.0000	690	0.0227	0.0082	0.0000
0.7346	0.2654	0.0000	695	0.0158	0.0057	0.0000
0.7347	0.2653	0.0000	700	0.0114	0.0041	0.0000
0.7347	0.2653	0.0000	705	0.0081	0.0029	0.0000
0.7347	0.2653	0.0000	710	0.0058	0.0021	0.0000
0.7347	0.2653	0.0000	715	0.0041	0.0015	0.0000
0.7347	0.2653	0.0000	720	0.0029	0.0010	0.0000
0.7347	0.2653	0.0000	725	0.0020	0.0007	0.0000
0.7347	0.2653	0.0000	730	0.0014	0.0005	0.0000
0.7347	0.2653	0.0000	735	0.0010	0.0004	0.0000
0.7347	0.2653	0.0000	740	0.0007	0.0003	0.0000
0.7347	0.2653	0.0000	745	0.0005	0.0002	0.0000
0.7347	0.2653	0.0000	750	0.0003	0.0001	0.0000
0.7347	0.2653	0.0000	755	0.0002	0.0001	0.0000
0.7347	0.2653	0.0000	760	0.0002	0.0001	0.0000
0.7347	0.2653	0.0000	765	0.0001	0.0000	0.0000
0.7347	0.2653	0.0000	770	0.0001	0.0000	0.0000
0.7347	0.2653	0.0000	775	0.0000	0.0000	0.0000
0.7347	0.2653	0.0000	780	0.0000	0.0000	0.0000
Totals				21.3713	21.3714	21.3715

II.8.

C.I.E. STANDARD LIGHT SOURCES

It is necessary to specify a particular kind of light source (or light sources) under which colours are observed and measured. For the purposes of general colour measurement three light sources have been recommended by the C.I.E. in 1931 since they are the most common of all kinds of light. These sources designated by the symbols A, B and C are representatives of the following types of radiant energy distribution.

A : Incandescent lamp light, at a colour temperature of 2854°K
(this provides a standard illuminant similar to the average tungsten-filament lamp).

B : Artificial sun light, correlated colour temperature of 4870°K .

C : Artificial average daylight, correlated colour temperature of 6770°K .

In the visible range the spectral emittance distributions of source A is equal to that of a Planckian radiator and the spectral irradiance from source A is found from the Planck radiation law. The locus of Planckian (Black-body) radiators is useful for determining the correlated colour temperature of a source, as shown in Figure II.17.

Sources B and C were obtained by combining source A with special filters, the Davis-Gibson filters. The relative spectral distributions of sources B and C were obtained by measuring spectrophotometrically and are represented by the curves in Figure II.16. Their

chromaticity coordinates are tabulated in Table II. 4 and are plotted in the C.I.E. chromaticity diagram of Figure II.17.

As a convenient and practical rule standard light sources are used as reference achromatic stimuli.

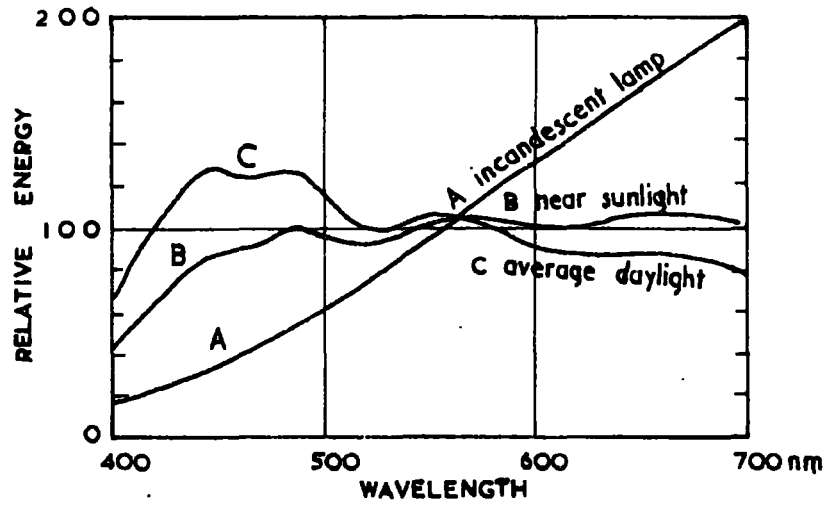


Figure 11.16. Relative spectral distributions of the C.I.E. standard sources A, B and C (from Judd & Wyszecki, 1965)

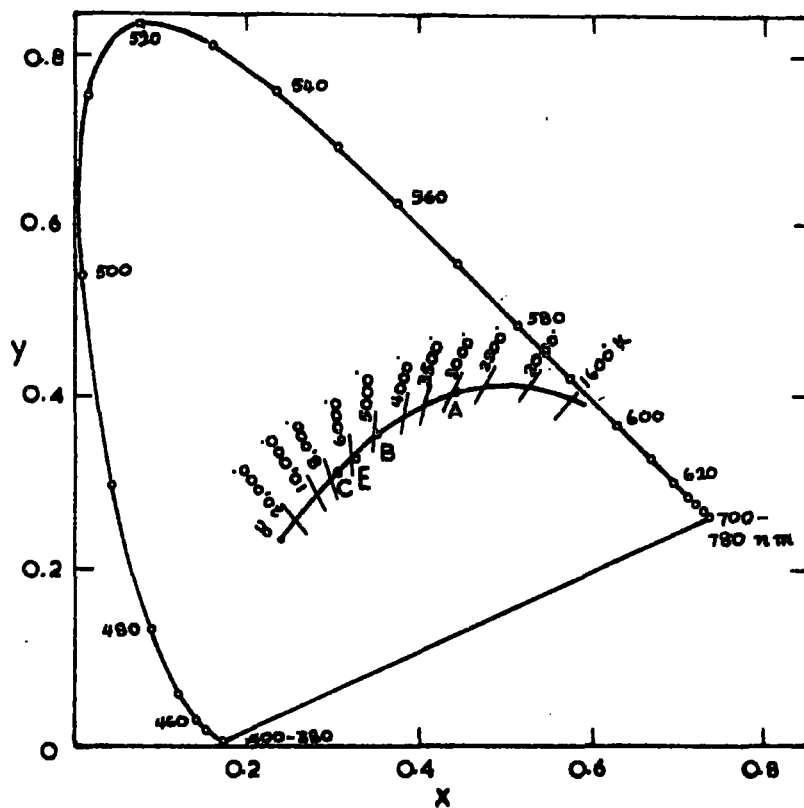


Figure 11.17. C.I.E. light sources A, B, C and the locus of Planckian radiators in the C.I.E. chromaticity diagram. Absolute colour temperatures are shown against the locus (from Judd & Wyszecki, 1965)

Table II.4.

CHROMATICITY COORDINATES OF THE 1931 C.I.E. STANDARD
LIGHT SOURCES

Source	x	y
A Incandescent Lamp	0.4476	0.4075
B Sunlight (artificial)	0.3485	0.3518
C Average Daylight (artificial)	0.3101	0.3163
E Equal-energy Stimulus	0.3333	0.3333

II.9. COMPUTATION OF TRISTIMULUS VALUES AND CHROMATICITY COORDINATES

The C.I.E. system of colour measurement is an internationally adopted method to specify the characteristics of a light source or an object colour in terms of its tristimulus values. For a desired light source specified by its spectral irradiance distributions ($S_\lambda d\lambda$) in the visible range (i.e. 380 nm to 780 nm) the tristimulus values XYZ are given by

$$\begin{aligned} X &= K \int S_\lambda \bar{x}_\lambda d\lambda , \\ Y &= K \int S_\lambda \bar{y}_\lambda d\lambda = 100.0 \\ Z &= K \int S_\lambda \bar{z}_\lambda d\lambda . \end{aligned} \quad (1)$$

where $K = \frac{100.0}{\int S_\lambda \bar{y}_\lambda d\lambda}$, a normalizing factor and so that $Y = 100.0$;

$d\lambda$ = wavelength interval.

The chromaticity coordinates are computed therefrom thus,

$$\begin{aligned} x &= \frac{X}{X + Y + Z} , \\ y &= \frac{Y}{X + Y + Z} , \\ z &= \frac{Z}{X + Y + Z} \end{aligned} \quad (2)$$

Spectral energy distributions, tristimulus values and chromaticity coordinates of the C.I.E. standard sources and many other sources can be found in the text-books, particularly in 'Color Science' by Wyszecki and Stiles, 1967.

To characterise an object colour irradiated by one of those sources the first requirement is to determine its spectral reflectance (r_λ) or transmittance (t_λ) by measurement, followed by calculation. Generally, spectral reflectance or transmittance is measured with a spectrophotometer and the C.I.E. tristimulus values are then calculated. Sometimes tristimulus values are measured directly with a photoelectric colorimeter.

The evaluation of the tristimulus values of an object colour with respect to a particular source can be done by either of the following methods.

(1) Weighted Ordinate Method.

This is merely a numerical method of integrating the values of spectral energy distributions ($S_\lambda \bar{x}_\lambda$, $S_\lambda \bar{y}_\lambda$, $S_\lambda \bar{z}_\lambda$) and spectral reflectance (r_λ) or transmittance (t_λ) at equal wavelength intervals.

The tristimulus values of an object colour are given by

$$\begin{aligned} X &= K \int r_\lambda S_\lambda \bar{x}_\lambda d\lambda \\ Y &= K \int r_\lambda S_\lambda \bar{y}_\lambda d\lambda \\ Z &= K \int r_\lambda S_\lambda \bar{z}_\lambda d\lambda . \end{aligned} \quad (3)$$

Since the products $S_\lambda \bar{x}_\lambda d\lambda$, etc. of a source are constants for all problems of computing object colours, tables of normalized products denoted by $H_\lambda \bar{x}_\lambda d\lambda$, $H_\lambda \bar{y}_\lambda d\lambda$ and $H_\lambda \bar{z}_\lambda d\lambda$ are convenient for the Weighted Ordinate Method of evaluating the integrals.

The normalized products to be used for standard sources A, B and C are listed in Table II.5 (from Wyszecki and Stiles, 1967). Using these Tables the tristimulus values of an object colour are given by

$$\begin{aligned} X &= \int R_{\lambda} H_{\lambda} \bar{x}_{\lambda} d\lambda \quad , \\ Y &= \int R_{\lambda} H_{\lambda} \bar{y}_{\lambda} d\lambda \quad , \\ Z &= \int R_{\lambda} H_{\lambda} \bar{z}_{\lambda} d\lambda \quad . \end{aligned} \tag{4}$$

In the same way for a transparent object T_{λ} replaces R_{λ} . The chromaticity coordinates are then calculated by the equations (2).

(2) Selected Ordinate Method

An alternative method for computing tristimulus specification of an object colour was proposed by Hardy and Pineo (1935). Their method is called the Selected Ordinate Method and is popular in practical colorimetry because of the simpler computation to be carried out. In this method numerous multiplications involved in the Weighted Ordinate Method are avoided and only summations of reflectance are required at selected wavelengths.

The selected wavelength intervals are chosen to be inversely proportional to $H_{\lambda} \bar{x}_{\lambda}$ for X, $H_{\lambda} \bar{y}_{\lambda}$ for Y and $H_{\lambda} \bar{z}_{\lambda}$ for Z so that where the distribution coefficients are large the wavelength intervals are small, and vice versa. The essential calculations for the tristimulus values XYZ of an object colour by the Selected Ordinate Method are performed simply by

- (1) adding the values of reflectance (or transmittance) at selected wavelengths , and
- (2) multiplying the three sums thus obtained by appropriate factors.

The thirty ordinates for standard sources A, B and C are listed in Table II.6 and the corresponding multiplication factors are also given at the bottom of the Table. The derivations of the tristimulus specification of an object by both methods are illustrated in Figure II. 18.

ACCURACY OF THE TWO METHODS

The methods used to compute tristimulus specifications have limits of accuracy. The accuracy, based on comparison between computed values of standard reflecting surfaces and filters, depends on the numbers of wavelength intervals employed in the Weighted Ordinate and Selected Ordinate methods (Kerf, 1957). The smaller the wavelength intervals in the computation the better the accuracy obtained.

According to Nickerson (1935) the Weighted Ordinate method with $d\lambda = 10$ nm and Selected Ordinate with $N = 30$ are sufficient for accuracy. The Selected Ordinate method $N = 10$ is rejected. Many authors think so for most purposes.

According to Kerf (1957) the Weighted Ordinate $d\lambda = 10$ nm and Selected Ordinate method $N = 100$ are equally accurate in average, but the error of both methods is greater than $1/2$ jps (just perceptible steps, or $= 1/10$ NBS units) in some cases. Kerf concludes that the

only method, which determines the tristimulus specifications of all the samples studied with an uncertainty smaller than $1/2$ jps, is the Weighted Ordinate method $d\lambda = 5$ nm ; and that the Selected Ordinate method $N = 30$ may only be applied to smooth curves of samples.

It is obvious that if accuracy is of prime importance the Weighted Ordinate method is preferable. With automatic computing machines becoming increasingly available to research workers the Selected Ordinate method is of less importance.

It is also found that the Weighted Ordinate method is more convenient to be used to derive weighted spectral energy distributions of any desired illumination . Therefore computer programs for the Weighted Ordinate method were developed and used in this present work.

Using the Weighted Ordinate method $d\lambda = 5$ nm consistently as a standard method the deviations in computed colour values would be only due to errors in measured reflectivity (or transmittance) or insufficient numbers of wavelength intervals used in the measurement of spectral reflectivity (or transmittance). Precision of colour specifications in this work is discussed in a later Chapter.

Table II 5. 1931 CIE Color-Matching Functions ($\bar{x}_\lambda, \bar{y}_\lambda, \bar{z}_\lambda$) Weighted by Relative Spectral Energy Distribution ($H_\lambda \Delta\lambda$) of CIE Source A ($\lambda = 380 \cdots 780 \text{ nm}; \Delta\lambda = 5 \text{ nm}$)

Wavelength λ (nm)	$\bar{x}_\lambda H_\lambda$	$\bar{y}_\lambda H_\lambda$	$\bar{z}_\lambda H_\lambda$	Wavelength λ (nm)	$\bar{x}_\lambda H_\lambda$	$\bar{y}_\lambda H_\lambda$	$\bar{z}_\lambda H_\lambda$
380	0.0006	0.0000	0.0029	580	4.8594	4.6139	0.0090
385	0.0011	0.0000	0.0053	585	5.3549	4.4668	0.0077
390	0.0024	0.0000	0.0113	590	5.7896	4.2704	0.0062
395	0.0047	0.0001	0.0224	595	6.1403	4.0379	0.0058
400	0.0097	0.0003	0.0463	600	6.3518	3.7733	0.0048
405	0.0174	0.0004	0.0825	605	6.4299	3.4855	0.0037
410	0.0356	0.0010	0.1699	610	6.3340	3.1780	0.0019
415	0.0694	0.0020	0.3319	615	6.0877	2.8622	0.0013
420	0.1308	0.0039	0.6283	620	5.6865	2.5358	0.0013
425	0.2269	0.0077	1.0974	625	5.1267	2.1901	0.0007
430	0.3246	0.0133	1.5840	630	4.4902	1.8523	0.0000
435	0.4055	0.0207	2.0036	635	3.8779	1.5529	0.0000
440	0.4632	0.0306	2.3236	640	3.2791	1.2812	0.0000
445	0.4976	0.0426	2.5484	645	2.7004	1.0344	0.0000
450	0.5155	0.0583	2.7173	650	2.1681	0.8183	0.0000
455	0.5230	0.0788	2.8621	655	1.7078	0.6372	0.0000
460	0.5097	0.1052	2.9254	660	1.3141	0.4861	0.0000
465	0.4690	0.1380	2.8539	665	0.9850	0.3625	0.0000
470	0.3882	0.1808	2.5581	670	0.7241	0.2651	0.0000
475	0.2998	0.2375	2.1979	675	0.5368	0.1958	0.0000
480	0.2138	0.3108	1.8179	680	0.4022	0.1461	0.0000
485	0.1372	0.4004	1.4575	685	0.2877	0.1041	0.0000
490	0.0799	0.5196	1.1622	690	0.2019	0.0729	0.0000
495	0.0387	0.6813	0.9308	695	0.1429	0.0515	0.0000
500	0.0136	0.8960	0.7545	700	0.1047	0.0377	0.0000
505	0.0070	1.1878	0.6191	705	0.0756	0.0271	0.0000
510	0.0285	1.5398	0.4843	710	0.0549	0.0199	0.0000
515	0.0934	1.9518	0.3585	715	0.0394	0.0144	0.0000
520	0.2127	2.3855	0.2627	720	0.0283	0.0097	0.0000
525	0.3849	2.7859	0.2012	725	0.0198	0.0069	0.0000
530	0.6069	3.1609	0.1547	730	0.0140	0.0050	0.0000
535	0.8631	3.4987	0.1140	735	0.0101	0.0041	0.0000
540	1.1567	3.7999	0.0809	740	0.0072	0.0031	0.0000
545	1.4904	4.0618	0.0555	745	0.0052	0.0021	0.0000
550	1.8660	4.2841	0.0375	750	0.0032	0.0010	0.0000
555	2.2887	4.4701	0.0255	755	0.0021	0.0010	0.0000
560	2.7550	4.6110	0.0181	760	0.0021	0.0010	0.0000
565	3.2564	4.6974	0.0130	765	0.0011	0.0000	0.0000
570	3.7853	4.7285	0.0104	770	0.0011	0.0000	0.0000
575	4.3259	4.7002	0.0092	775	0.0000	0.0000	0.0000
580	4.8594	4.6139	0.0090	780	0.0000	0.0000	0.0000
				Totals (X_A, Y_A, Z_A)	109.8472	100.0000	35.5824
				($\bar{x}_A, \bar{y}_A, \bar{z}_A$)	0.4476	0.4074	0.1450

(continued)

Table II 5. 1931 CIE Color-Matching Functions ($\bar{x}_\lambda, \bar{y}_\lambda, \bar{z}_\lambda$) Weighted by Relative Spectral Energy Distribution ($H_\lambda \Delta\lambda$) of CIE Source B
($\lambda = 380 \cdots 780 \text{ nm}$; $\Delta\lambda = 5 \text{ nm}$)

Wavelength λ (nm)	$\bar{x}_\lambda H_\lambda$	$\bar{y}_\lambda H_\lambda$	$\bar{z}_\lambda H_\lambda$	Wavelength λ (nm)	$\bar{x}_\lambda H_\lambda$	$\bar{y}_\lambda H_\lambda$	$\bar{z}_\lambda H_\lambda$
380	0.0015	0.0000	0.0070	580	4.4218	4.1984	0.0082
385	0.0028	0.0001	0.0135	585	4.6790	3.9030	0.0067
390	0.0063	0.0001	0.0301	590	4.8644	3.5880	0.0052
395	0.0131	0.0003	0.0626	595	4.9701	3.2684	0.0047
400	0.0282	0.0008	0.1340	600	4.9736	2.9546	0.0037
405	0.0517	0.0013	0.2455	605	4.8999	2.6561	0.0028
410	0.1083	0.0030	0.5163	610	4.7185	2.3672	0.0014
415	0.2139	0.0061	1.0236	615	4.4415	2.0882	0.0009
420	0.4058	0.0121	1.9495	620	4.0700	1.8149	0.0009
425	0.7017	0.0238	3.3944	625	3.6031	1.5392	0.0005
430	0.9916	0.0405	4.8394	630	3.1000	1.2788	0.0000
435	1.2134	0.0621	5.9951	635	2.6296	1.0530	0.0000
440	1.3446	0.0888	6.7448	640	2.1871	0.8545	0.0000
445	1.3878	0.1188	7.1067	645	1.7765	0.6804	0.0000
450	1.3718	0.1551	7.2308	650	1.4074	0.5312	0.0000
455	1.3229	0.1993	7.2399	655	1.0929	0.4078	0.0000
460	1.2269	0.2531	7.0422	660	0.8273	0.3060	0.0000
465	1.0807	0.3181	6.5769	665	0.6085	0.2239	0.0000
470	0.8589	0.4000	5.6599	670	0.4381	0.1604	0.0000
475	0.6365	0.5044	4.6670	675	0.3177	0.1159	0.0000
480	0.4348	0.6323	3.6980	680	0.2323	0.0844	0.0000
485	0.2667	0.7784	2.8332	685	0.1617	0.0585	0.0000
490	0.1475	0.9590	2.1449	690	0.1102	0.0398	0.0000
495	0.0672	1.1826	1.6156	695	0.0758	0.0273	0.0000
500	0.0221	1.4538	1.2242	700	0.0540	0.0194	0.0000
505	0.0106	1.7976	0.9370	705	0.0378	0.0135	0.0000
510	0.0403	2.1798	0.6856	710	0.0267	0.0097	0.0000
515	0.1246	2.6052	0.4785	715	0.0185	0.0068	0.0000
520	0.2707	3.0361	0.3344	720	0.0129	0.0044	0.0000
525	0.4735	3.4272	0.2476	725	0.0087	0.0030	0.0000
530	0.7291	3.7973	0.1859	730	0.0060	0.0021	0.0000
535	1.0186	4.1292	0.1345	735	0.0042	0.0017	0.0000
540	1.3445	4.4168	0.0940	740	0.0029	0.0012	0.0000
545	1.7042	4.6445	0.0635	745	0.0020	0.0008	0.0000
550	2.0915	4.8016	0.0420	750	0.0012	0.0004	0.0000
555	2.5006	4.8840	0.0278	755	0.0008	0.0004	0.0000
560	2.9200	4.8872	0.0192	760	0.0008	0.0004	0.0000
565	3.3360	4.8122	0.0133	765	0.0004	0.0000	0.0000
570	3.7359	4.6669	0.0103	770	0.0004	0.0000	0.0000
575	4.1019	4.4568	0.0088	775	0.0000	0.0000	0.0000
580	4.4218	4.1984	0.0082	780	0.0000	0.0000	0.0000
				Totals (X_B, Y_B, Z_B) (x_B, y_B, z_B)	99.0930 0.3484	100.0000 0.3516	85.3125 0.3000

(continued)

Table II 5. 1931 CIE Color-Matching Functions ($\bar{x}_\lambda, \bar{y}_\lambda, \bar{z}_\lambda$) Weighted by Relative Spectral Energy Distribution ($H_\lambda \Delta\lambda$) of CIE Source C
($\lambda = 380 \dots 780$ nm; $\Delta\lambda = 5$ nm)

Wavelength λ (nm)	$\bar{x}_\lambda H_\lambda$	$\bar{y}_\lambda H_\lambda$	$\bar{z}_\lambda H_\lambda$	Wavelength λ (nm)	$\bar{x}_\lambda H_\lambda$	$\bar{y}_\lambda H_\lambda$	$\bar{z}_\lambda H_\lambda$
380	0.0022	0.0000	0.0101	580	4.2084	3.9958	0.0078
385	0.0041	0.0002	0.0197	585	4.3859	3.6585	0.0063
390	0.0093	0.0002	0.0447	590	4.4920	3.3133	0.0048
395	0.0197	0.0005	0.0938	595	4.5265	2.9767	0.0043
400	0.0425	0.0012	0.2018	600	4.4745	2.6581	0.0034
405	0.0782	0.0020	0.3716	605	4.3617	2.3644	0.0025
410	0.1647	0.0045	0.7850	610	4.1622	2.0882	0.0013
415	0.3263	0.0092	1.5611	615	3.8863	1.8272	0.0008
420	0.6192	0.0184	2.9743	620	3.5349	1.5763	0.0008
425	1.0672	0.0363	5.1628	625	3.1074	1.3275	0.0004
430	1.4986	0.0612	7.3139	630	2.6548	1.0952	0.0000
435	1.8165	0.0929	8.9747	635	2.2358	0.8953	0.0000
440	1.9874	0.1312	9.9687	640	1.8468	0.7216	0.0000
445	2.0182	0.1728	10.3351	645	1.4909	0.5711	0.0000
450	1.9578	0.2213	10.3194	650	1.1743	0.4432	0.0000
455	1.8499	0.2786	10.1235	655	0.9058	0.3380	0.0000
460	1.6811	0.3469	9.6497	660	0.6807	0.2518	0.0000
465	1.4539	0.4279	8.8481	665	0.4965	0.1827	0.0000
470	1.1360	0.5291	7.4860	670	0.3542	0.1297	0.0000
475	0.8281	0.6562	6.0719	675	0.2548	0.0929	0.0000
480	0.5563	0.8088	4.7305	680	0.1846	0.0671	0.0000
485	0.3348	0.9773	3.5571	685	0.1270	0.0459	0.0000
490	0.1814	1.1790	2.6369	690	0.0855	0.0309	0.0000
495	0.0807	1.4197	1.9396	695	0.0581	0.0209	0.0000
500	0.0258	1.7004	1.4319	700	0.0408	0.0147	0.0000
505	0.0121	2.0462	1.0665	705	0.0283	0.0101	0.0000
510	0.0447	2.4165	0.7600	710	0.0197	0.0071	0.0000
515	0.1350	2.8223	0.5183	715	0.0136	0.0049	0.0000
520	0.2881	3.2309	0.3559	720	0.0093	0.0032	0.0000
525	0.4982	3.6052	0.2604	725	0.0062	0.0022	0.0000
530	0.7617	3.9671	0.1942	730	0.0042	0.0015	0.0000
535	1.0593	4.2941	0.1399	735	0.0029	0.0012	0.0000
540	1.3924	4.5742	0.0973	740	0.0020	0.0009	0.0000
545	1.7559	4.7853	0.0654	745	0.0014	0.0006	0.0000
550	2.1412	4.9157	0.0430	750	0.0008	0.0003	0.0000
555	2.5414	4.9636	0.0283	755	0.0005	0.0003	0.0000
560	2.9399	4.9204	0.0193	760	0.0005	0.0003	0.0000
565	3.3167	4.7844	0.0132	765	0.0003	0.0000	0.0000
570	3.6613	4.5736	0.0101	770	0.0003	0.0000	0.0000
575	3.9623	4.3051	0.0085	775	0.0000	0.0000	0.0000
580	4.2084	3.9958	0.0078	780	0.0000	0.0000	0.0000
Totals (X_C, Y_C, Z_C) (x_C, y_C, z_C)				98.0705	100.0000	118.2246	
				0.3101	0.3162	0.3737	

Table II 6.

Selected Ordinates for Computing 1931 C.I.E. Tristimulus Values of
Object Colours with Respect to C.I.E. Standard Sources A,B,C

Ordinate Number	Source A			Source B			Source C		
	X	Y	Z	X	Y	Z	X	Y	Z
1	444.0	487.8	416.4	428.1	472.3	414.8	424.4	465.9	414.1
2	516.9	507.7	424.9	442.1	494.5	422.9	435.5	489.4	422.2
3	544.0	517.3	429.4	454.1	505.7	427.1	443.9	500.4	426.3
4	554.2	524.1	432.9	468.1	513.5	430.3	452.1	508.7	429.4
5	561.4	529.8	436.0	527.8	519.6	433.0	461.2	515.1	432.0
6	567.1	534.8	438.7	543.3	524.8	435.4	474.0	520.6	434.3
7	572.0	539.4	441.3	551.9	529.4	437.7	531.2	525.4	436.5
8	576.3	543.7	443.7	558.5	533.7	439.9	544.3	529.8	438.6
9	580.2	547.8	446.0	564.0	537.7	442.0	552.4	533.9	440.6
10	583.9	551.7	448.3	568.8	541.5	444.0	558.7	537.7	442.5
11	587.2	555.4	450.5	573.1	545.1	446.0	564.1	541.4	444.4
12	590.5	559.1	452.6	577.1	548.7	448.0	568.9	544.9	446.3
13	593.5	562.7	454.7	580.9	552.1	450.0	573.2	548.4	448.2
14	596.5	566.3	456.8	584.5	555.5	451.9	577.3	551.8	450.1
15	599.4	569.8	458.8	588.0	559.0	453.9	581.3	555.1	452.1
16	602.3	573.3	460.8	591.4	562.4	455.8	585.0	558.5	454.0
17	605.2	576.9	462.9	594.7	565.8	457.8	588.7	561.9	455.9
18	608.0	580.5	464.9	598.1	569.3	459.8	592.4	565.3	457.9
19	610.9	584.1	467.0	601.4	572.9	461.8	596.0	568.9	459.9
20	613.8	587.9	469.2	604.7	576.7	463.9	599.6	572.5	462.0
21	616.9	591.8	471.6	608.1	580.6	466.1	603.3	576.4	464.1
22	620.0	595.9	474.1	611.6	584.7	468.4	607.0	580.5	466.3
23	623.3	600.1	476.8	615.3	589.1	470.8	610.9	584.8	468.7
24	626.9	604.7	479.9	619.1	593.9	473.6	615.0	589.6	471.4
25	630.8	609.7	483.4	623.3	599.1	476.6	619.4	594.8	474.3
26	635.3	615.2	487.5	628.0	605.0	480.2	624.2	600.8	477.7
27	640.5	621.5	492.7	633.4	611.8	484.5	629.8	607.7	481.8

Table II 6 (continued)

Ordinate Number	Source A			Source B			Source C		
	X	Y	Z	X	Y	Z	X	Y	Z
28	646.9	629.2	499.3	640.1	619.9	490.2	636.6	616.1	487.2
29	655.9	639.7	508.4	649.2	630.9	498.6	645.9	627.3	495.2
30	673.5	659.0	526.7	666.3	650.7	515.2	663.0	647.4	511.2

Factors: 0.03661 0.03333 0.01185 0.03303 0.03333 0.02842 0.03268 0.0333 0.03938

(from Committee on Colorimetry Optical Society of America, 1968)

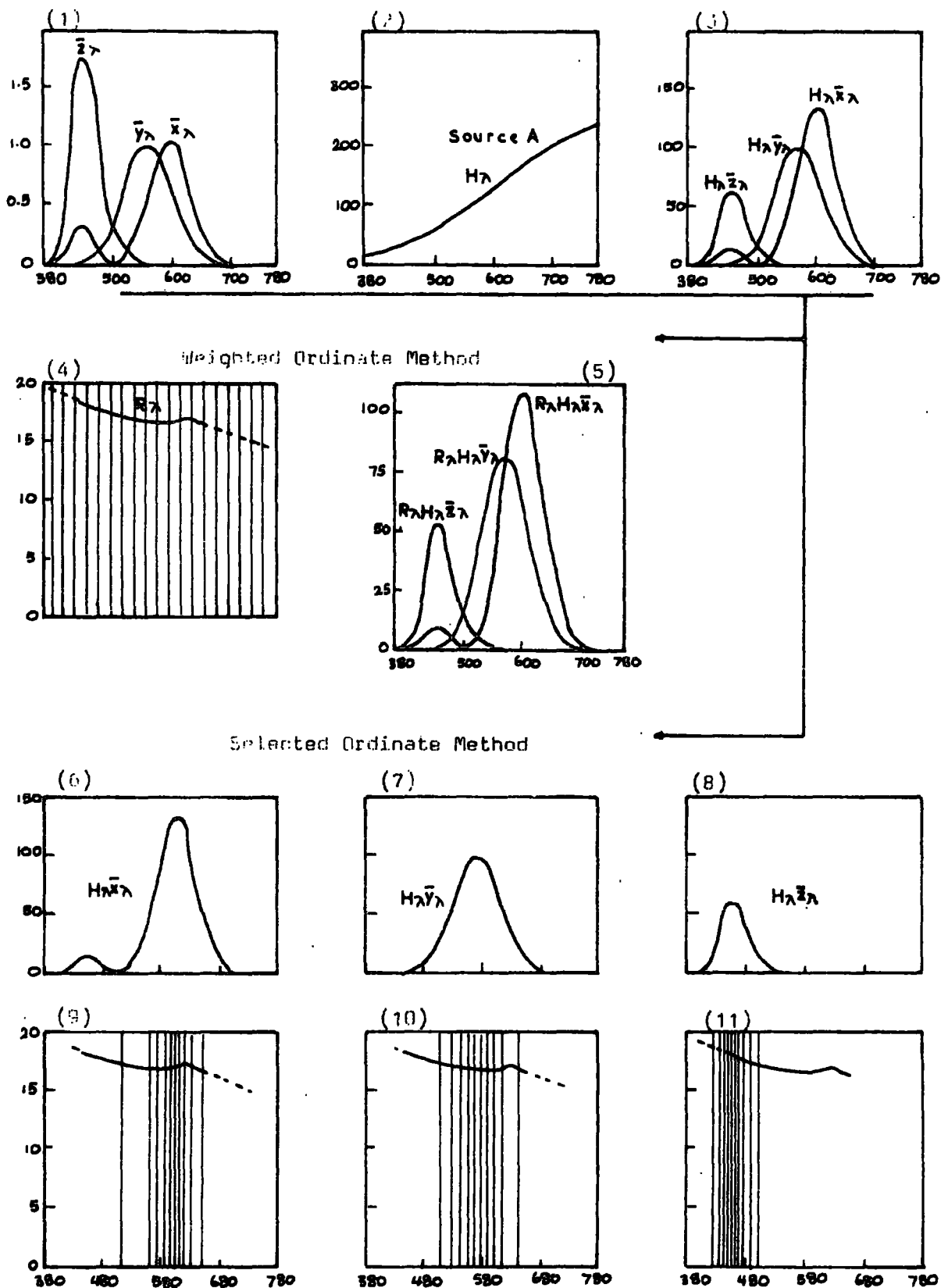


Figure II 18. Derivations of the tristimulus specifications of an object colour by Weighted Ordinate and Selected Ordinate methods.

Figure II 18. Illustration of the Computation of the Tristimulus Values of an Object Colour by the Weighted Ordinate and Selected Ordinate Methods.

- (1) Colour-matching functions (\bar{x}_λ , \bar{y}_λ , \bar{z}_λ).
- (2) Energy distributions of the C.I.E. source A (H_λ).
- (3) Products of (2) with the colour-matching functions ($H_\lambda \bar{x}_\lambda$, $H_\lambda \bar{y}_\lambda$, $H_\lambda \bar{z}_\lambda$).
- (4) Spectral reflectance (R_λ) of the object at equal wavelength intervals .
- (5) Products of (3) and (4). The areas under the three curves give the tristimulus values X,Y and Z of the object by the Weighted Ordinate method .
- (6)(7)(8) Selected ordinates derived from (3) for the C.I.E. source A.
- (9)(10)(11) Spectral reflectance of the object at selected ordinates. Products of the sums by the appropriate factors give the tristimulus values X,Y and Z of the object by the Selected Ordinate method.

II 10.

THE 1931 C.I.E. CHROMATICITY DIAGRAM

The 1931 C.I.E. chromaticity diagram with the spectral locus and the purple line is drawn on the x-y coordinates chart, as in Figure II 19. The chromaticity coordinates of the spectrum locus are included in Table II 3.

Chromaticity, the quality of a colour may be specified by a point in a plane diagram, the chromaticity diagram. This diagram represents the relative chromaticities of all colours in the same manner as a plane map represents the relative locations of various places on the earth. That is, the brightness of any colour is not taken into account in the diagram. The perfect white and black colours in fact will coincide at a point (achromatic point).

All colours represented by points on the spectrum locus are termed spectrum colours. All colours within the triangle defined by the points of two ends (380 nm and 780 nm) of the spectrum locus and the point of the achromatic stimulus are called non-spectral colours. Colours having chromaticities (chromatic points) represented within the curve and outside the non-spectral triangle are called spectral colours. There is no colour outside the spectrum locus and the purple line (the Science of Color by the Committee on Colorimetry of the Optical Society of America, 1968).

II 11.

DOMINANT WAVELENGTH AND EXCITATION PURITY

Although the quality of a colour can be completely specified by its tristimulus values X,Y,Z (known as the trichromatic system) these numerical values do not provide a suitable way to visualize the character of the colour. A given colour can also be characterised by additional numerical values in terms of the dominant wavelength and excitation purity (known as the monochromatic system).

It is in practice easier to get a general picture of a colour when it is expressed in terms of its dominant wavelength, purity and tristimulus Y value (brightness) with respect to a particular light source.

The dominant wavelength of a colour is the wavelength of spectrally pure radiant energy (spectrum colour) that, when additively mixed in suitable proportions with a specified white light yields a match with the given colour. The dominant wavelength of a colour correlates in an approximate way with the general term, hue. Thus, in general, colours of constant dominant wavelength would be said to have the same hue. The derivation of the dominant wavelength of a sample colour is shown in Figure II 20. That is the wavelength at the intersection point of the spectrum locus with the straight line drawn from the achromatic point through sample point.

If a sample point lies in the non-spectral triangle then the wavelength corresponding to the intersection is the complementary wavelength (λ_c) of the colour. The chromaticity of a spectral colour

is specified by its dominant wavelength (λ_d) and excitation purity (P_e), and the chromaticity of a non-spectral colour is specified by its complementary wavelength (λ_c) and excitation purity (P_e).

The determination of dominant wavelength from the chromaticity coordinates can also be carried out by computation, making use of tabulated ratios prepared by Judd (1933).

Radiant energy of a single wavelength is said to be spectrally pure. In terms of the chromaticity diagram, a dominant (or complementary) wavelength is constant for all points on any straight line passing through an achromatic point, whereas purity increases linearly with increasing distance from zero at the achromatic point to the maximum (1 or 100%) on the spectrum locus (or on the purple line).

The excitation purity (P_e) is defined as a ratio by the following equivalent expressions:

$$P_e = \frac{x_1 - x}{x_\lambda - x} = \frac{y_1 - y}{y_\lambda - y}$$

where x_1 , y_1 and x , y are coordinates of the sample point and specified achromatic point respectively; x_λ , y_λ are coordinates of the dominant or complementary wavelength at the intersection point.

When the straight line passing through the illuminant point and sample point is nearly horizontal the second formula will become less accurate and vice versa when the line is nearly vertical.

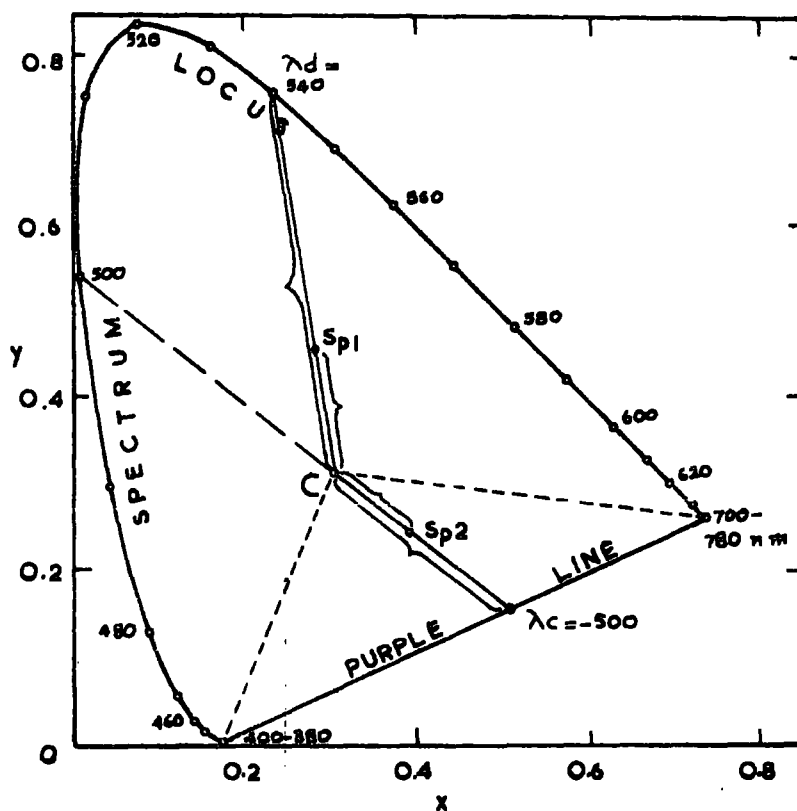


Figure 11.10. Graphical determination of dominant (or complementary) wavelength and excitation purity of a sample colour in the C.I.E. chromaticity diagram.

It has been shown that the standard C.I.E. system allows all colours to be specified quantitatively in terms of tristimulus specifications. As a consequence of the application of the system it is necessary to specify colour differences. The reason for this is the fact that small differences in XYZ figures do not give directly an indication of the visual effect of the colour difference.

There are limits to the precision and accuracy within which colours were matched by the standard observer, as described in previous sections, with a purely physical instrument. In other words there are limits within which colours are not discriminated to the visual sensation.

This means that the C.I.E. system is not uniform and not quite suitable for colour difference specification without additional standardization or transformation into a uniform scale. Therefore, there is need of an additional system for specifying colour differences.

A number of different methods of calculating colour differences have been tried by various workers : Judd (1935); Wright (1941); Nickerson (1944 & 1950); Balinkin (1941); Godlove (1951) ; MacAdam (1942 & 1943); and others.

All of these methods are based on measurements of intervals (distances) between colours. Three general approaches to the problem of establishing a uniform colour space have been proposed. The first

of these is a linear transformation of the C.I.E. chromaticity chart into a so called Uniform Chromaticity (UCS) diagram (Judd , 1935 ; MacAdam, 1937; Hunter, 1941; Committee on Colorimetry, 1963). Combination of this transformation with a lightness scale yields a colour difference formulation of limited usefulness.

The second approach, suggested by Nickerson (1936 & 1950), Balinkin (1941) and Godlove (1951), is the transformation of C.I.E. data into a so called uniform colour spacing as laid down in the Munsell Renotation and DIN systems. A colour difference specification can be defined in terms of hue, value and chroma (Munsell) or Farbton, Sättigung and Dunkelstufe (DIN). The transformation of C.I.E. data into these units can only be made by means of graphical representation.

The third approach which is the more promising and easiest way of handling the problem is to use the colour discrimination data, as obtained by MacAdam (1942 & 1943), since no system is sufficiently uniform and no simple set of transformation equations can distort the C.I.E. space into a uniform space. Various methods proposed by MacAdam, based on his discrimination ellipses have found wide acceptance in industrial tolerance specification.

The work of MacAdam was extended by Brown and MacAdam (1949) and Brown (1957). Many workers, Moon (1943), Davidson and Friede (1952), Brown (1951) and Friele (1961) analysed the MacAdam colour discrimination data and calculations based on these data were found to correlate with the visual results much better than do those based on other methods.

In this present work the MacAdam discrimination data were

used to compare nearly identical colours and colour differences of minerals. From the literature the necessary basic information is therefore summarised below.

MacAdam (1942) investigated visual sensitivity to colour differences by studying just equally noticeable differences of colour matches about a colour centre. Twenty-five representative colour centres scattered throughout the colour domain were used in colour matching to determine noticeability of colour differences at constant luminance.

Extensive tests with a colour discrimination apparatus have indicated that the just noticeable differences of colour are directly related to the corresponding standard deviations of colour matching. The standard deviation of colour matching has therefore been adopted as a satisfactory measure of the noticeability of colour differences.

The standard deviations (root mean square of individual deviations from the average setting) of MacAdam's colour matching were plotted on the C.I.E. chromaticity diagram and all were in the form of ellipses of varying size as shown in Figure II 21. These ellipses represent the noticeability of chromaticity variations in all directions from the chromaticities indicated at the centres of the ellipses. That is, these ellipses indicate the noticeabilities of conceivable combinations of purity and dominant wavelength differences.

Silberstein and MacAdam (1945) deduced from discrimination ellipses that if the colour matches were not confined to chromaticity variations (that is the luminance variations are also involved) the

surfaces in colour space representing the standard deviation of colour matching would be ellipsoids. Silberstein (1946) proved that the standard deviation figures were in the form of ellipsoids in colour space and gave formulae from which the coefficients of the ellipsoids could be determined.

The ellipses, therefore, are regarded as the constant luminance cross sections of the ellipsoids. MacAdam (1942) describes that C.I.E. chromaticity diagram appears ^{as} convenient as possible for the representation of relative chromaticity difference, in the manner ^{that} as a plane map represents the relative locations of various places on the earth.

Each ellipse is measured by the lengths of the principal semi-axes 'a' and 'b' and the angle of inclination θ of the major axis from the horizontal as shown in Figure II 22. These values of MacAdam's discrimination ellipses are tabulated in Table II 7.

The sizes, shapes and orientations of the ellipses vary throughout the chromaticity diagram, in a somewhat systematic manner. These trends encouraged the interpolation of ellipses representing colours equally noticeably different from any other fixed colours. MacAdam (1943) has prepared contour diagrams in order to evaluate colour difference specifications for most practical purposes. They are reproduced in Figures II 23, II 24, and II 25, and can be found in the original paper by MacAdam (1943) or in 'Color Science' by Wyszecki and Stiles (1967).

Each of the twenty-five ellipses representing the known standard deviations of colour matching, according to MacAdam, is

represented by the equation

$$g_{11} dx^2 + g_{12} dx \cdot dy + g_{22} dy^2 = 1.$$

where dx is the distance of the x coordinates of the centre of the ellipse and any point on the ellipse; dy is the difference of the y coordinates for the the same pair of points; and g_{11} , g_{12} and g_{22} indicated in the diagrams are constants for each ellipse.

These constants can be determined from the length dx_0 of the horizontal radius of the ellipse (for which $dy = 0$), the length dy_0 of the vertical radius of the ellipse (for which $dx = 0$), and the lengths dp and dq of the radii which are inclined at $\pm 45^\circ$ from the horizontal

$$\begin{aligned} g_{11} &= 1 / dx_0^2 \\ g_{12} &= 1 / dp^2 - 1 / dq^2 \\ g_{22} &= 1 / dy_0^2 . \end{aligned}$$

The orientation θ and lengths a and b of the ellipse may be derived from the values of the coefficients g_{11} , g_{12} , g_{22} indicated for that central colour

$$\tan 2\theta = 2g_{12} / (g_{11} - g_{22}).$$

$$\theta < 90^\circ \text{ when } g_{12} < 0 ,$$

$$\theta > 90^\circ \text{ when } g_{12} > 0 .$$

$$1 / a^2 = g_{22} + g_{12} \cot \theta$$

$$1 / b^2 = g_{11} - g_{12} \cot \theta$$

From these values an ellipse may be constructed. This ellipse is the equi-luminance cross section of the ellipsoid in colour space.

Every colour on the surface of the ellipsoid represents just equal noticeabilities of total colour difference from the central colour. Any colour which would plot within the ellipsoid would be visually indistinguishable from the colour plotted at the centre of the ellipsoid.

According to MacAdam the necessity of constructing such ellipses will be almost completely eliminated by the method which follows:

The ratio of any chromaticity difference colour matching is denoted by ds . Then

$$ds^2 = g_{11} dx^2 + 2 g_{12} dx \cdot dy + g_{22} dy^2.$$

In the calculation each of the numerical values, g_{11} , g_{12} , g_{22} for the region intermediate between the two chromaticities, is multiplied by 10,000.

For example, if two colours A and B are to be compared with each other the differences of their chromaticities and appropriate coefficients found from the contour diagrams are substituted in the above equation and say $ds = n$. This indicates that the two colours A and B exhibit a difference of chromaticity n times as great as the standard deviations of colour matching.

MacAdam (1943) described other methods of calculating colour differences for particular problems. Brown and MacAdam (1949) studied the visual sensitivities to combined chromaticity and luminance differences and determined the coefficients and axes of ellipsoids derived from standard deviations of colour matches. The cross sections

of all ellipsoids correlate well with results previously published by MacAdam (1942).

Thus the precision of colour matching is commonly described by colour-matching ellipses. To obtain colour discrimination specifications, numerous colour-matching experiments have been made later by several workers: Stiles (1946), Brown (1951, 1952 & 1956), MacAdam (1950), Wyszecki and Fielder (1971), Wyszecki (1972). For a given colour centre the ellipses of different observers are not in close agreement, but there is an overall resemblance and a general agreement between ellipses obtained by different investigators.

It has been found that the sizes, shapes and orientations of the ellipsoids depend upon many physical and psychophysical factors such as the luminance level, field size, surroundings, adaptation, portion of the retina, technique of observation, etc.

The effects of colour-matching field size, chromatic surround and luminance level on colour discrimination ellipsoids have been studied by Brown (1951 & 1952). The effect of luminance level is already described in section II 4.

From the work of Brown there are some interesting facts to be noted here. As a rule the colour of the field surrounding the matching field has a quite noticeable effect on colour discrimination when the matching field is small. Best discriminations obtained when the surround has the same chromaticity as the matching field. This agrees with results reported by Schonfelder (1933).

In general, large fields of view (10°) allow better colour

discrimination than fields covering only the foveal region (2°). This result is in agreement with observations by Judd (1930, Lobanova and Rautian (1949). The discrimination ellipsoids obtained for large-field and small-field viewing are similar in orientation and shape, but differ in size. The orientation and shape of the ellipsoids (2° matching field) are also affected by the colour of the surround.

However, at the present time the work of MacAdam, and Brown and MacAdam has found wide acceptance for colour discrimination purposes in many industrial and research problems.

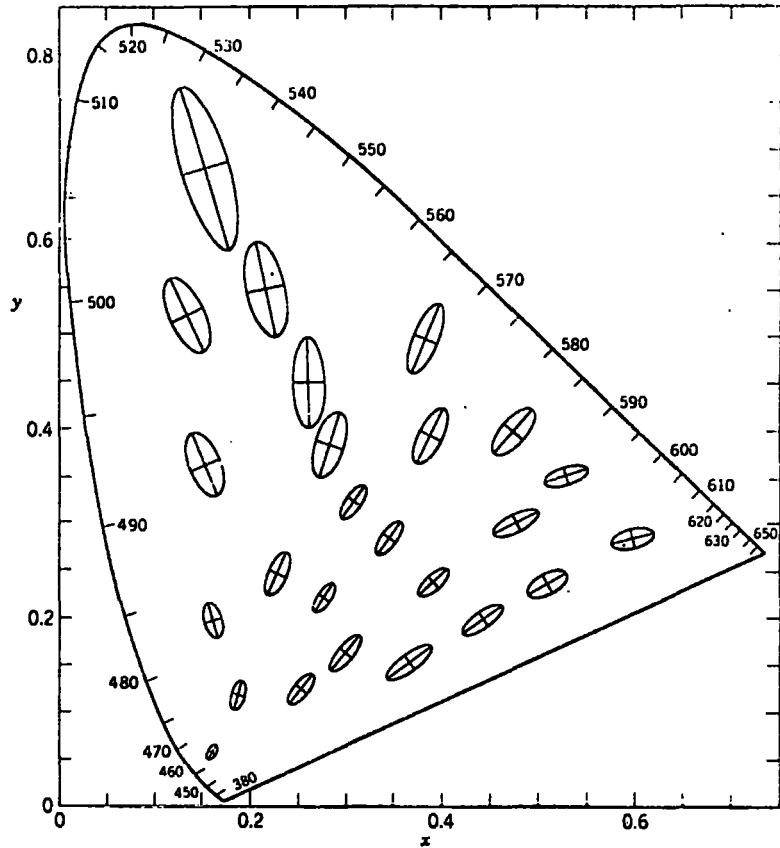


Figure II 21. MacAdam's discrimination ellipses
(ten times enlarged).
(from Wyszecki & Stiles 1967)

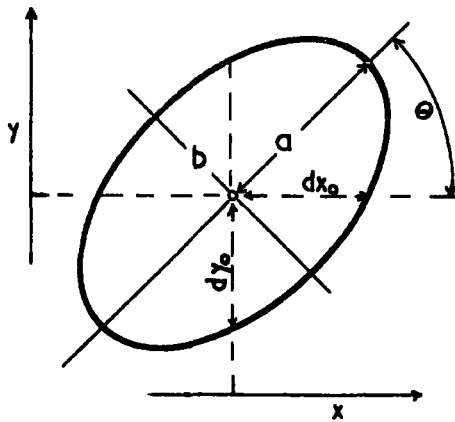


Figure II 22. Measures of a discrimination ellipse.

a = major semiaxis

b = minor semiaxis

dx_0 = horizontal radius (for which $dy = 0$)

dy_0 = vertical radius (for which $dx = 0$)

θ = angle in degrees

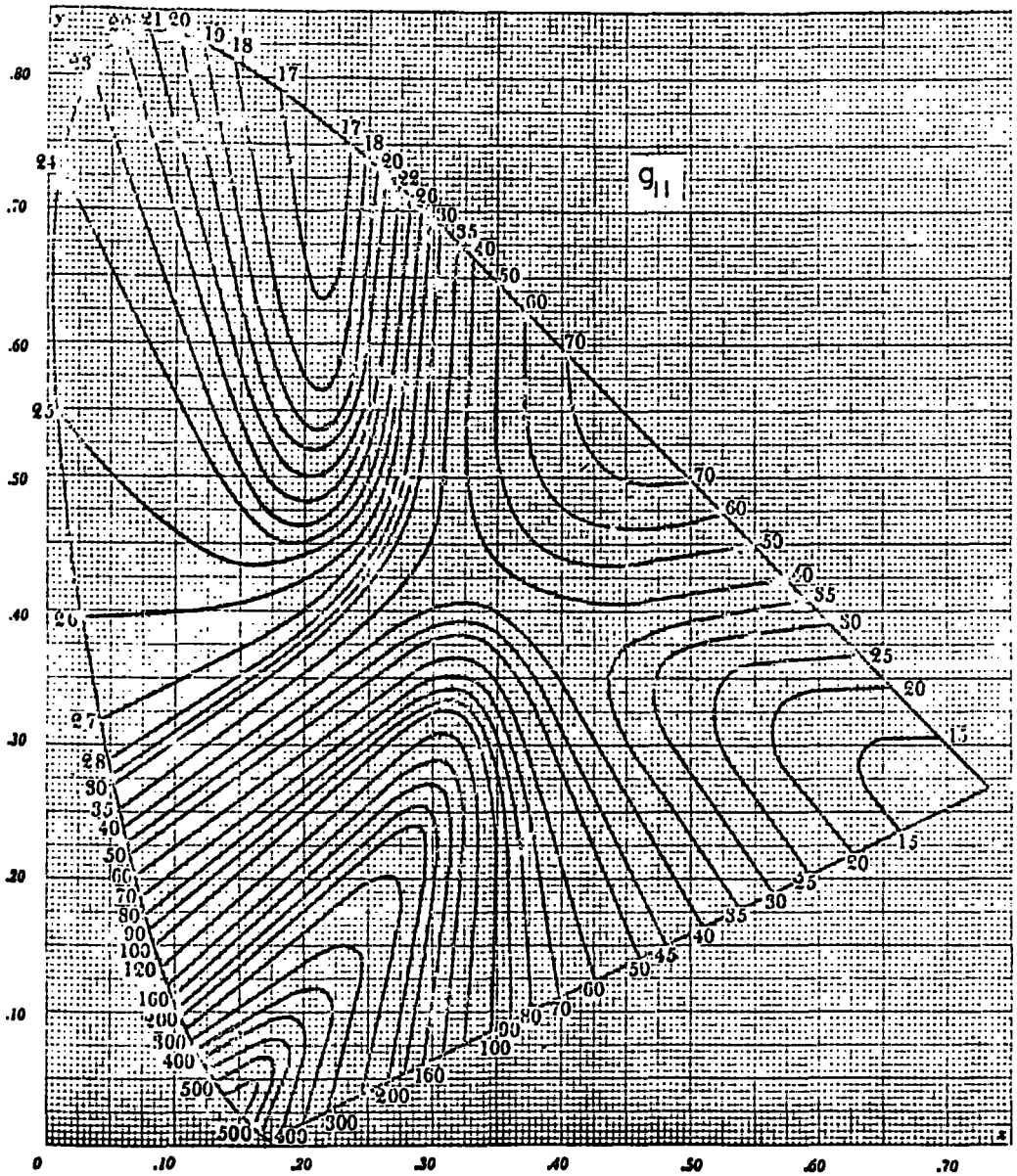


Figure II 23. Contour lines of constant metric coefficient g_{11} for various locations in the C.I.E. chromaticity diagram (from MacAdam 1943).

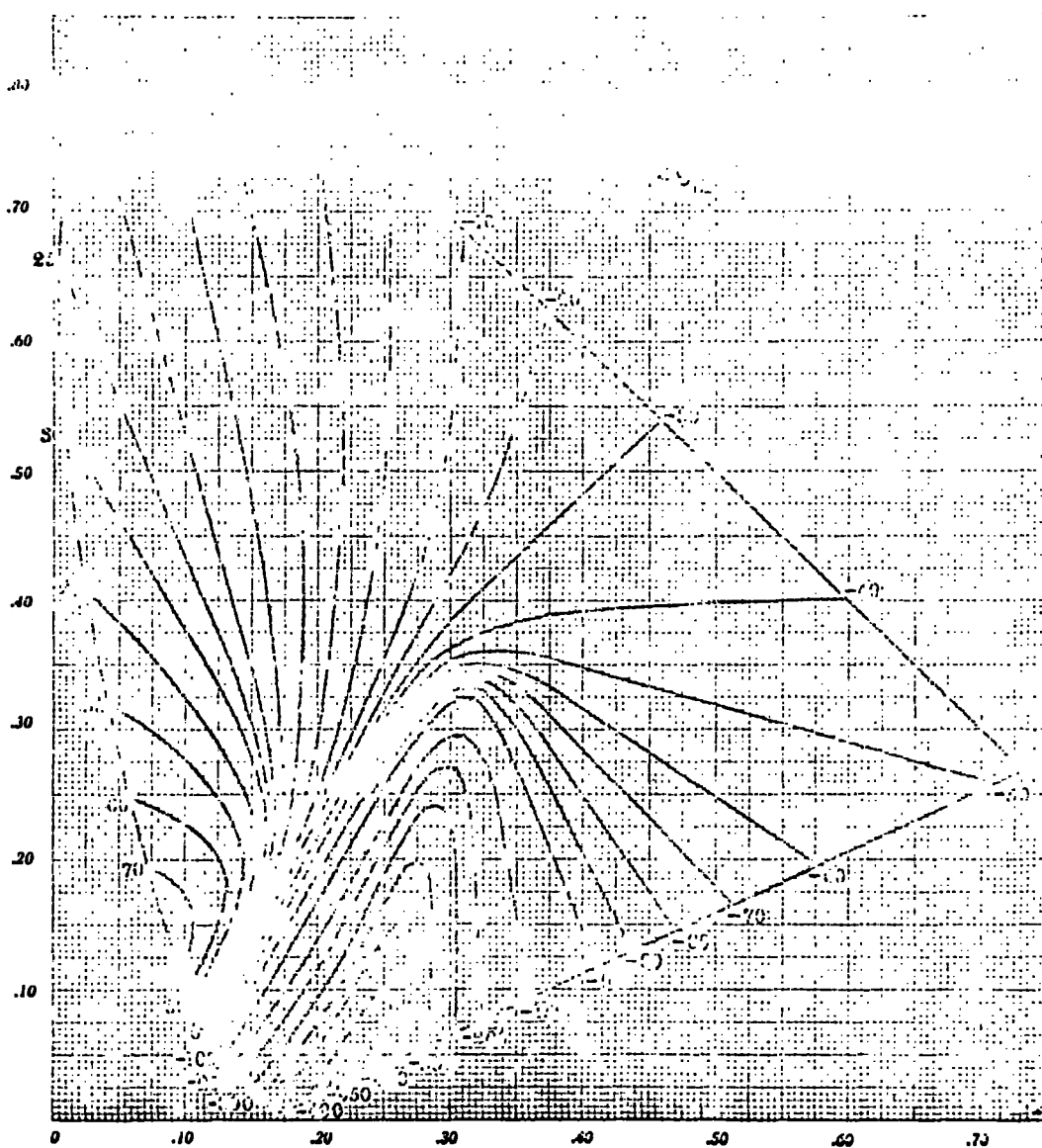


Figure II 24. Contour lines of constant metric coefficient $2g_{12}$ for various locations in the C.I.E. chromaticity diagram (from MacAdam 1943).

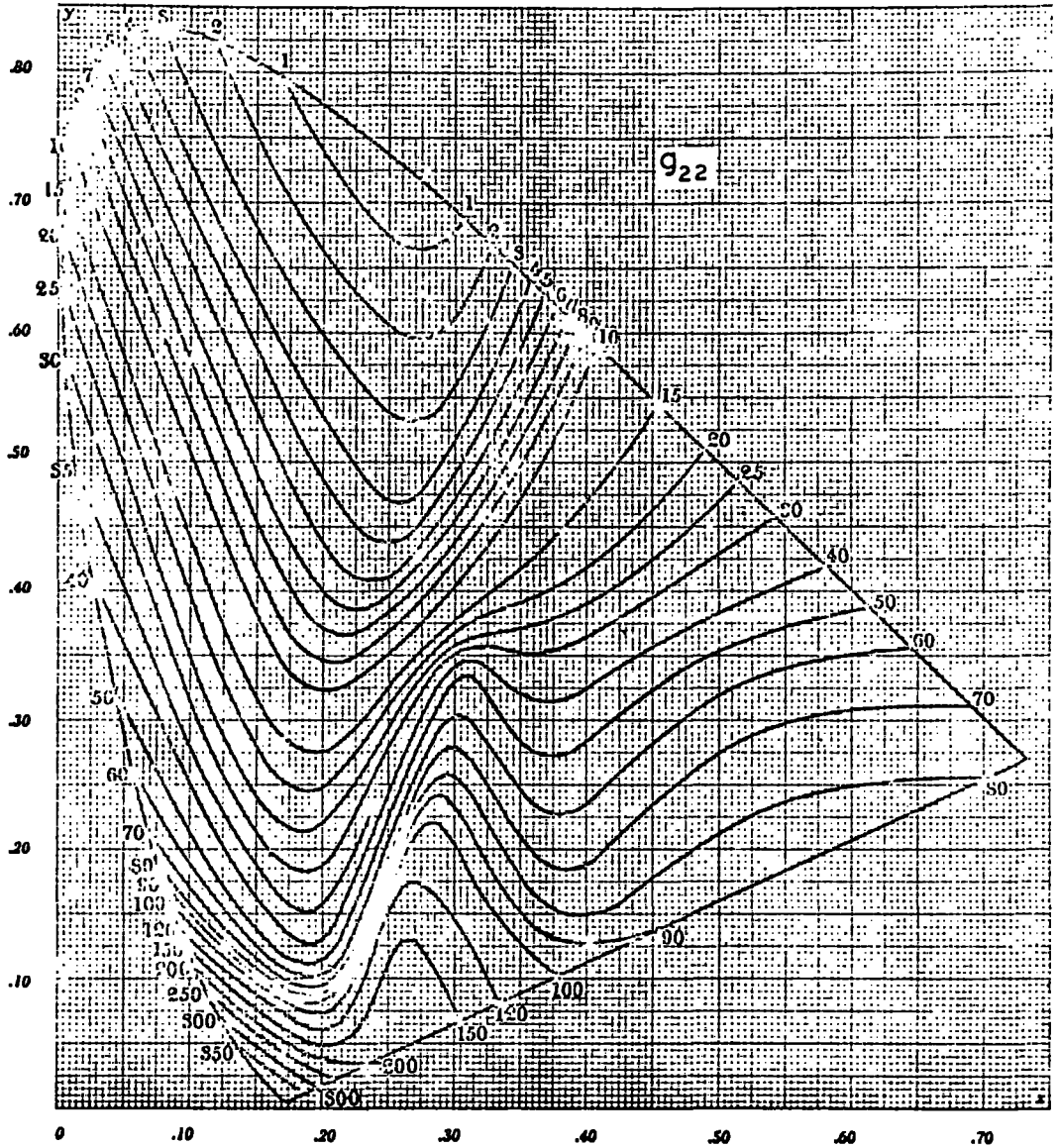


Figure II 25. Contour lines of constant metric coefficient g_{22} for various locations in the C.I.F. chromaticity diagram (from MacAdam 1943).

III. SPECTRAL REFLECTIVITY MEASUREMENTS

III 1. PREVIOUS WORK

The measurement of reflectivity is the only reliable quantitative method in ore microscopy since reflectivity is the most important property of a reflecting material. Significant contributions to the theory and practice of such measurements were made by Folinsbee (1949), Bowie and Taylor (1958), Gray and Millman (1960 & 1962) , Cameron (1961), Bowie (1962), Jones (1962), Nichol (1962), von Gehlen and Piller (1964), and Nichol and Phillips (1964), using photoelectric microphotometers. The photoelectric microphotometer replaces the visual photometer used by previous workers, Hallimond (1957) and Leonard (1960) and others.

The devices and technique of reflectivity measurement for the visible spectrum have been improved in recent years . Unfortunately there are still marked variations in reflectivity values quoted for specific minerals by different workers, as Cameron (1961) points out and ^{ca)}described by Nichol and Phillips (1964). Causes of error in reflectivity and in the calculation for refractive index and absorption coefficient have been discussed by Piller and von Gehlen (1964). Errors due to glare effects from optical parts of the microscope and correction procedures to reduce the errors have been described by Bowie and Henry (1964) and Piller (1965).

According to Piller and von Gehlen (1964) relative error of about $\pm 0.5\%$ to $\pm 1\%$ are unavoidable even under most favourable conditions. Cameron (1963) claims an accuracy of $\pm 0.2\%$.

An important method of the linearity test of a photomultiplier used for reflectivity measurement has been given by Phillips and Bradshaw (1965).

The interest in reflectivity measurement has been increased to some extent to study other physical properties as well as the chemical composition. By measuring reflectivities of a mineral in air and oil its refractive index (n) and absorption index (K) can be derived, provided that the measured values of reflectivity (R) are accurate (Cameron, 1961; von Gehlen and Piller, 1964; Bradshaw, Phillips and Smith, 1965).

Bowie and Taylor (1958) have proposed a system of ore mineral identification from two quantitative values of reflectivity and micro-hardness. On this basis an unknown mineral can be assigned to one of the five main groups of similar composition and structure (oxides; cobalt-nickel-iron; sulphides and arsenides; other sulphides; sulphosalts; and metals), and identification thereafter can be made with the aid of other observable properties such as colour, anisotropism, bireflection, etc.

Cameron (1963) has introduced the method of determination of optical symmetry from reflectivity measurements of randomly oriented grains of anisotropic ore minerals in monochromatic light (549 nm). From such measured data the recognition of R_w of uniaxial minerals and R_m , corresponding to $R\beta$, of biaxial minerals can be made. The optical signs are then classified by a suggested convention analogous to that used for transparent minerals. Cameron suggested that with an accuracy of $\pm 0.2\%$ the value of reflectivity corresponding to the ordinary ray (R_w) for any uniaxial mineral can be used as a prime basis of mineral

identification. For minerals sensibly of lower symmetry, a value of R_m , the reflectivity for Y direction, of biaxial minerals can also be determined and used in identification.

Cervelle et al (1968) and Vaasjoki (1969) pointed out an interesting optical phenomenon of reverse birefractance in a few ore minerals such as mawsonite and loellingite, showing the change of optic signs at some definite wavelength of the visible range. This new phenomenon has not been solved so far.

Vaughan (1969) and Demirsoy (1969) have made attempts to explain variations in reflectivity and composition in zoned bravoites. Engin (1969) showed a correlation between reflectivity and composition of chromite ores from Turkey. Cr_2O_3 has a positive correlation with measured reflectivity values at the 99.9 confidence level. That is reflectivity increases linearly with increasing of Cr_2O_3 weight percent.

Attempts have been made to correlate reflectivity and compositional variations of synthetic sphalerites by Akinçi (1970) and of synthetic tetrahedrite-tennantite series by Hall (1972).

Burns and Vaughan (1970) have shown the existence of correlation between reflectivity variations and effective number of electrons in pyrite type compounds and have explained this relation for other solid solution series of ore minerals.

Reflectivity data of most ore minerals can be found in the books by C. Schouten (1962) and by Ramdohr (1969). More important spectral reflectivity values for four wavelengths have been compiled

recently in ' International Tables for the Microscopic Determination of Crystalline Substances Absorbing in the Visible Light ' published by the Commission on Ore Microscopy (1970), and ' Tables for Microscopic Identification of Ore Minerals ' by Uytendogaardt and Burke (1971).

III 2.

PRESENT WORKSpecimen Preparation

The polishing procedure of specimens in this laboratory was according to that described in the paper by Nichol and Phillips (1964). For readily tarnishing minerals such as bornite, specimens were kept in a vacuum desiccator immediately after their final polishing.

Transparent, low-absorbing minerals were cut in such a way that the lower surfaces were non-parallel to the upper ones to reduce the back reflection. Facetted gemstones were measured without any preparation other than mounting and levelling in a piece of black plasticine and cleaning the surface thoroughly with very soft tissues.

Apparatus and Technique

The reflectivity apparatus in this laboratory was originally used and described by Nichol (1962), Nichol and Phillips (1964), Phillips and Bradshaw (1966), Burton (1967) who modified the photomultiplier to improve linearity and stability, Engin (1968) who replaced the galvanometer with the better digital voltmeter, and Tugal (1969).

Figure III 1 shows the general arrangement of the present reflectivity apparatus. To increase the accuracy and precision of results instrumental and operational improvements were made as follows:

Two specimen changer-stages which can be fixed one after the other on the rotating stage of the microscope were made. Polished sections of

a standard and specimen were levelled with plasticine on flat metal plates. They were fixed on the specimen changer-stages by means of magnets, after choosing areas free from imperfections. By this way whenever measurements were made the same area of standard and specimen was measured. These changer-stages are extremely useful to increase the accuracy and precision and also to save time in the measurement for various wavelengths.

The microscope, photomultiplier tube, illuminator and continuous monochromator, all in optical alignment, were rigidly fixed by clamping to a wooden board to improve the mechanical stability.

All electrical units, which are voltage stabilizer (TSV 70), photomultiplier (EMI type 6094B), stabilized EHT power supply unit (type E2), digital voltmeter (DM 2005), were checked and found the following operational and conditional requirements for accurate measurement.

- (1) The calibrated digital voltmeter should be switched on at least 15 minutes before taking a reading.
- (2) The voltage stabilizer for the light source must be left switched on for about 2 hours at a required voltage before making any measurement. The result of a check is shown by a curve in Figure II 2.
- (3) The stabilized EHT power unit at a suitable voltage (depending on reflectivities of specimen and standards) range must be left switched on continuously for daily task so that a steady sensitivity of the photomultiplier was obtained.
- (4) The working room should be in a reasonably constant temperature (about 21°C) because the sensitivity of the

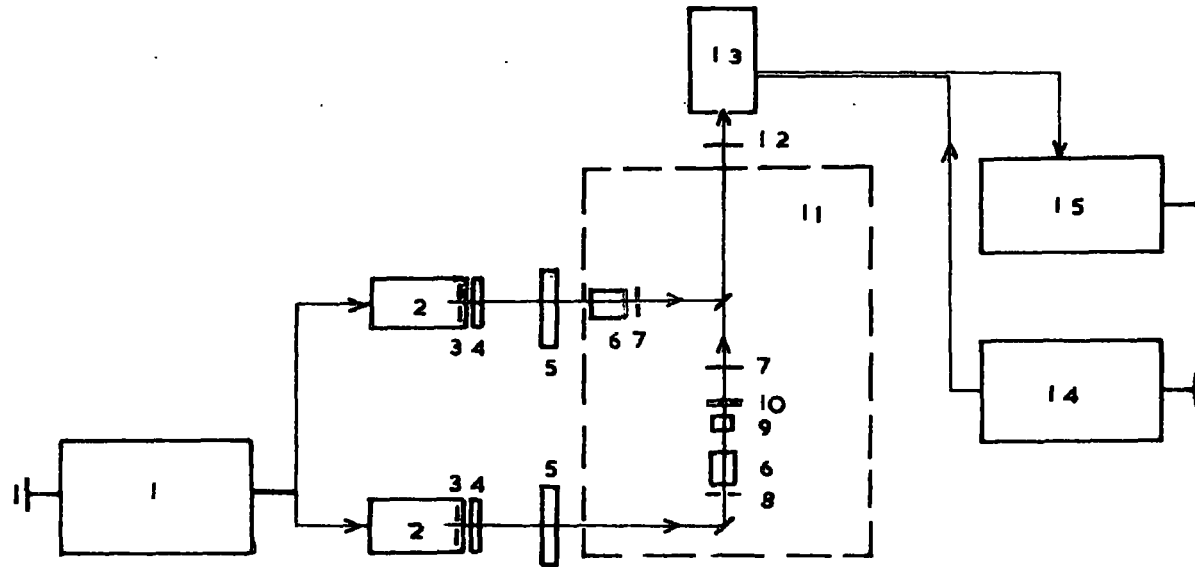
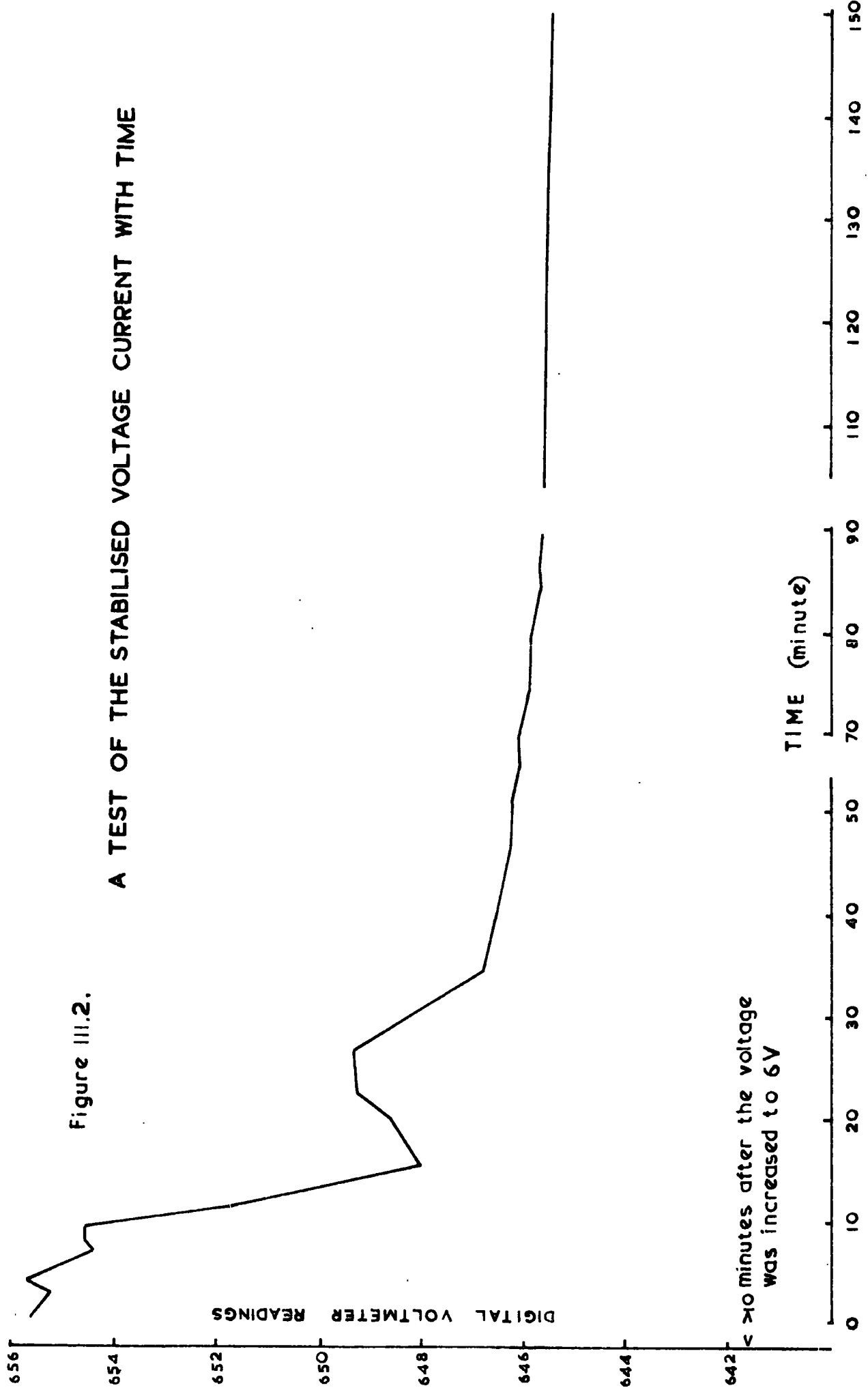


Figure III 1. Schematic diagram of the apparatus for reflectivity and transmittivity measurements.

- | | |
|--------------------------------|--|
| 1. Voltage stabiliser TSV 70 | 9. Condenser |
| 2. Illuminator | 10. Specimen |
| 3. Field diaphragm | 11. Microscope |
| 4. Heat absorbing glass filter | 12. Photometer stop |
| 5. Continuous monochromator | 13. Photomultiplier tube Type EMI 6094B |
| 6. Polarizer | 14. Stabilised EHT power supply unit Type E2 |
| 7 & 8. Measuring diaphragm | 15. Digital voltmeter DM 7999 |

Figure III.2.
A TEST OF THE STABILISED VOLTAGE CURRENT WITH TIME



>10 minutes after the voltage was increased to 6V

photomultiplier became unstable when the room was hot.

- (5) No stray light should be around the photomultiplier tube and the microscope.

A test of the linearity of the photomultiplier was made according to the method described by Bradshaw and Phillips (1965), and no significant deviation from linearity was found.

Spectral characteristics of the photomultiplier and continuous monochromator are according to the manufacturer's recommendations and are also given in Bradshaw's Ph.D. Thesis (1964).

When the above instrumental conditions were satisfied spectral reflectivity measurements were made using suitable standards. During operation care was taken in every case such as exact levelling polished surfaces, selecting measuring areas, exact focusing onto the surface and accurate setting wavelengths of the sliding monochromator.

At each wavelength three readings of the digital voltmeter were taken for standard, specimen and finally the black box. Measuring the same areas of the standard and of the specimen with the aid of the specimen changer-stages the procedure was repeated for various wavelengths.

Reflectivity value (in percent) corrected for the primary glare effect is computed by the following formula for a given wavelength.

$$R_{\lambda}(sp) = \frac{R_{\lambda}(st) \cdot V_{\lambda}(sp) - V_{\lambda}(b)}{V_{\lambda}(st) - V_{\lambda}(b)}$$

where $R_{\lambda}(sp)$, $R_{\lambda}(st)$ = reflectivity of specimen and standard, respectively, at wavelength λ .

$V_{\lambda}(sp)$, $V_{\lambda}(st)$, $V_{\lambda}(b)$ = digital voltmeter readings for specimen,

standard and black box, respectively, at wavelength λ .

When a large number of measurements is made it may be preferable to use a PL / 1 computer program, as given in Appendix III 1, to carryout the calculation.

Accuracy and Precision

The reflectivity apparatus of this laboratory and the technique applied provide accurate and high precision results with a small limit of error. Results of silicon standard (NPL) measured against carborundum standard (NPL) are given in Table III 1.

The accuracy and precision of reflectivity values for quantitative colour measurement are discussed in Chapter VI.

Table III 1.

Accuracy and Precision of Spectral Reflectivity Measurements

R% of Silicon (NPL) Measured Against Carborundum (NPL)

Si(NPL) Values	Wave- length	Measured Values				Mean R%	Mean deviation	Standard Deviation	Coeff.of Variation	% Error	Accuracy
		I	II	III	IV						
43.1	440	43.19	43.12	43.21	43.15	43.17	±.03	.035	.08	.16	.002
41.3	460	41.54	41.56	41.51	41.68	41.57	±.05	.065	.16	.65	.007
39.9	480	40.02	40.07	40.03	40.17	40.07	±.05	.059	.15	.42	.004
38.9	500	38.93	38.99	38.98	39.10	39.00	±.05	.062	.16	.26	.003
38.0	520	38.07	38.09	38.07	38.20	38.11	±.05	.054	.14	.29	.003
37.2	540	37.26	37.29	37.39	37.41	37.34	±.06	.064	.17	.37	.004
36.6	560	36.62	36.65	36.62	36.75	36.66	±.05	.053	.14	.16	.002
36.0	580	35.96	35.96	35.99	36.10	36.00	±.05	.058	.16	.00	.000
35.5	600	35.50	35.48	35.53	35.62	35.53	±.04	.054	.15	.08	.001
35.1	620	35.20	35.14	35.16	35.29	35.20	±.05	.058	.16	.28	.003
34.8	640	34.68	34.54	34.56	34.67	34.61	±.15	.063	.18	.55	.005
34.4	660	34.83	34.33	34.49	34.50	34.54	±.29	.182	.53	.41	.004

about 21° C room temperature.

III 3.

SPECTRAL REFLECTIVITY OF GEMSTONES

Gem varieties of natural minerals, artificial materials and some organic substances of beauty to mankind have been used as gemstones. The qualities of colour and reflection are the most striking and important properties of gemstones.

The colour of a gemstone is used as an aid for identification and also made use of for classifying into different qualities. The colour may be the most important property to assess the relative value of very similar stones, other things being equal.

To measure the colour quantitatively the first problem is to obtain the spectral reflection or transmission values of a gemstone. Once a gem mineral has been faceted, the transmitted light method of determining its spectral transmission is difficult or impossible to apply, because of its odd shape, therefore it was first decided to make an investigation of the use of reflected light method.

The reflectivity of an isotropic gemstone is given by Fresnel's equation

$$R \% = \frac{(n - N)^2 + k^2}{(n + N)^2 + k^2} \cdot 100.0$$

where n and k are the index of refraction and absorption coefficient, respectively, of the absorbing mineral. N is the refractive index of the surrounding medium (air $N = 1$), $k = nK$ and $k / \lambda_0 = K / \lambda$ where K is the absorption index, λ_0 is the wavelength in air and λ is the wavelength in mineral. The index of refraction and absorption

coefficient of a gemstone vary with wavelength and therefore reflectivity varies accordingly.

For uniaxial gemstones, the reflectivities R_o and R_e of the principal vibration directions are related in the same way to refractive indices n_o and n_e and the absorption coefficients k_o and k_e . Bireflectance of a gemstone is $R_o - R_e$ if it is optically negative or $R_e - R_o$ if it is optically positive.

Similarly, a biaxial gemstone has three principal reflectivities R_α , R_β and R_γ corresponding to $n_\alpha k_\alpha$, $n_\beta k_\beta$ and $n_\gamma k_\gamma$ respectively. The bireflectance is $R_\gamma - R_\alpha$. If R_β is nearer to R_α than R_γ the stone is said to be optically positive and if R_β approaches R_γ it is optically negative.

As in ore microscopy, the reflection principles for normally incident plane-polarised light are applied to gemstones and the reflectivity measurements are also applicable to polished surfaces of gemstones.

In an ideal brilliant cut stone the total back reflection occurs as shown in Figure III 3. A test showed that black plasticine absorbed all incident light, no detectable light reflected from it and so to reduce the back reflection faceted stones were mounted on pieces of black plasticine.

Inclusions and flaws are not uncommon in natural gemstones. Scattering of light due to such imperfections will cause error and unfortunately, this is unavoidable in the measurement.

However, an area as free as possible from optical imperfections should

be chosen and thus the conventional method of reflectivity measurement can be employed to faceted gemstones.

Spectral reflectivity measurements, as described in the previous section II 2, were made on polished surfaces of some ore minerals and faceted gemstones in the visible spectrum ranging from 440 nm to 660 nm at an interval of 20 nm. In all measurements a low-power objective (X 10) and a measuring area of 50 micron were used consistently. Measured reflectivity values of ore minerals and published data available in the literature were employed in the quantitative measurements of colour.

The spectral reflectivity values of faceted gemstones are presented by the curves in Figure II 4. For non-opaque minerals and gemstones there is no significant variation in reflectivity throughout the visible spectrum. Consequently, their spectral reflectivity values which are not related to the pronounced colours of the specimens are inapplicable for colour measurement.

The probable explanation for this is that reflection of normally incident light actually takes place at the top surface, perhaps a few micron thick layer of a non-opaque specimen. The amount of absorption of such a thin layer of low-absorbing material is negligible (i.e. $k \rightarrow 0$), and therefore it is not effective to reflectivity although measurement is made on thick specimens of pronounced colour.

It follows that reflectivity of a transparent (i.e. light can pass through a petrographic thin section with a thickness of 0.3 mm) mineral is completely dominated by its refractive index n rather than

the absorption coefficient k .

For these reasons the accurate measurements of reflectivity suggested that reflectivity values and refractive indices of non-opaque minerals are directly correlated as shown by the linear straight line in Figure III 5.

Therefore reflectivity measurement could provide an additional method for identification in gemmology, especially those stones whose refractive indices are high beyond the range of a normal refractometer.

Since the application of reflected light method to the measurement of colour in transparent gemstones was unfavourable, an investigation of the use of ^{the} transmitted method was made. The proper transmission measurement method requires a thin, parallel sided specimen. The most convenient thickness of a specimen depends on its amount of absorption (Tauc, 1967).

Such measurement is obviously impossible to apply to faceted gemstones. Therefore an alternative method of determining spectral transmission of faceted stones was considered and is described in the next Chapter.

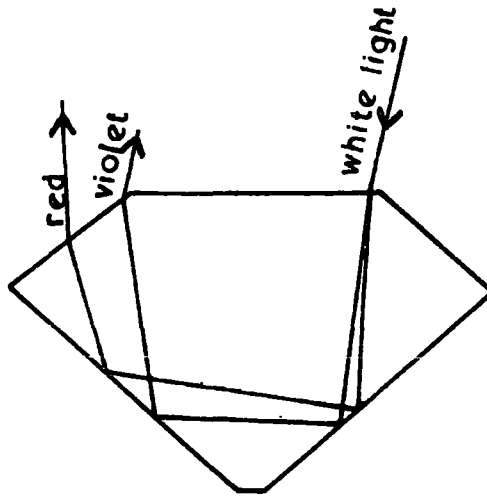


Figure III 3. Total back reflection and dispersion of a ray of light passing through brilliant cut diamond.

Figure III 4.

SPECTRAL REFLECTIVITY CURVES OF GEMSTONES

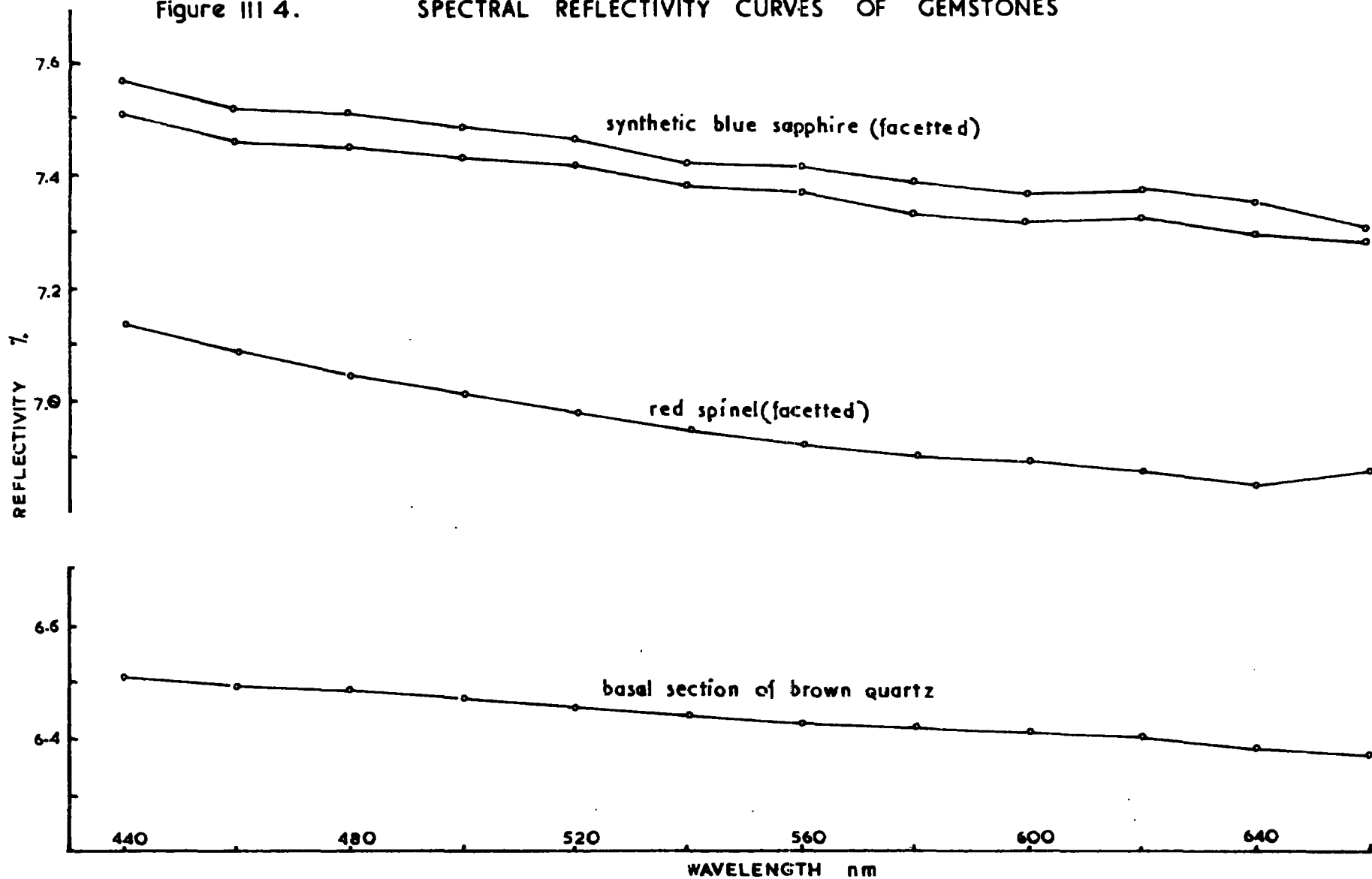
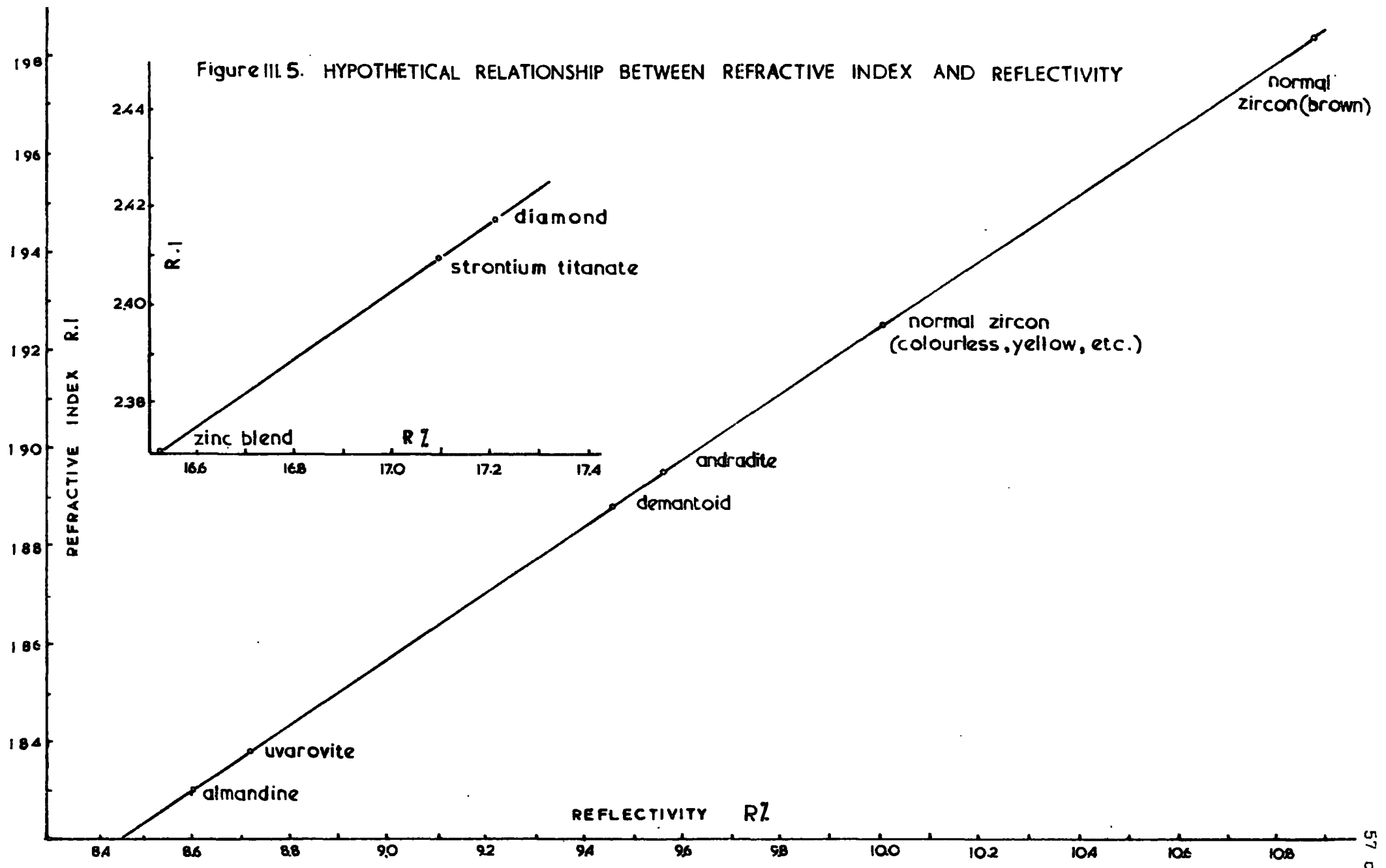


Figure III.5. HYPOTHETICAL RELATIONSHIP BETWEEN REFRACTIVE INDEX AND REFLECTIVITY



IV. POSSIBILITY OF TRANSMISSION MEASUREMENT THROUGH FACETTED
GEMSTONES FOR COLOUR MEASUREMENT

THEORETICAL CONSIDERATIONS

Consider a parallel beam of light incident normally on the surface of a parallel sided plate of thickness t , refractive index n_{λ_0} and absorption index K_{λ_0} . If the intensity is I_0 in the direction shown in Figure IV 1, let J_0 be the intensity of light travelling in the opposite direction due to reflection at the surface AB and transmission of components reflected from the surface CD. With similar meanings for other I_x and J_x we have the following relationships:

Equations for light leaving the surface AB

$$I_1 = R J_1 + (1 - R) I_0 ; \quad J_0 = R I_0 + (1 - R) J_1 ;$$

Equations for light leaving the surface CD

$$I_3 = R J_3 + (1 - R) I_2 ; \quad J_2 = R I_2 + (1 - R) J_3 ;$$

Equations for absorption in the medium

$$I_2 = I_1 \cdot e^{-mt} ; \quad J_1 = J_2 \cdot e^{-mt} .$$

where R is the reflection coefficient at normal incidence

$$R = \frac{(n_{\lambda_0} - N)^2 + n_{\lambda_0}^2 K_{\lambda_0}^2}{(n_{\lambda_0} + N)^2 + n_{\lambda_0}^2 K_{\lambda_0}^2} \quad (1)$$

$K = k/n$, N is the refractive index of the surrounding medium.

$$m_{\lambda_0} \text{ is the absorption modulus} = 4\pi n_{\lambda_0} K_{\lambda_0} / \lambda_0 = 4\pi k_{\lambda_0} / \lambda_0 \quad (2)$$

λ_0 is the wavelength of the light in vacuo.

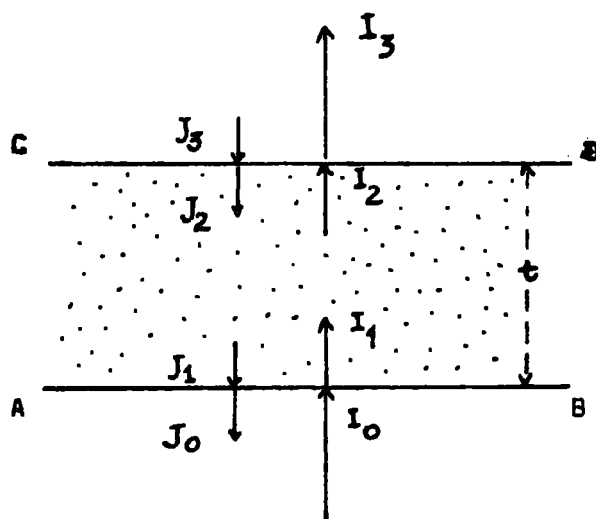


Figure IV. 1 Reflection and absorption of a parallel beam of normally incident light in passing through a parallel sided plate of an absorbing substance.

Assuming no reflection from the measuring instrument $J_3 = 0$.

$$\text{Hence } I_3 = (1 - R) I_2 ;$$

$$J_2 = R I_2 ;$$

$$J_1 = R I_2 \cdot e^{-mt} .$$

substitution gives

$$T_{\lambda_0} = I_3 / I_0 = \frac{(1 - R_{\lambda_0})^2}{e^{m\lambda_0 t} - R_{\lambda_0}^2 e^{-m\lambda_0 t}} \quad (3)$$

For non-absorbing substances it is usually assumed that $m = 0$, then equation (3) becomes

$$T = \frac{(1 - R)^2}{1 - R^2} \quad (4)$$

Thus for an isotropic absorbing and non-absorbing substance with two parallel surfaces on opposite sides the transmission of normally incident light is given by equations (3) and (4) respectively. From these equations, formulae for transmission of a combination of layers can be derived as follows:

(1) Transmission of a combination of non-absorbing layers separated by air as shown in Figure IV 2 is

$$T = \frac{(1 - R_1)^2 (1 - R_2)^2}{1 - R_1 \cdot R_2} \quad (5)$$

when both layers have the same refractive index, then

$R_1 = R_2 = R$, and hence

$$T = \frac{(1 - R)^2 \times 2}{1 - R^2}$$

(2) Transmission of a combination of non-absorbing layers in optical contact, when both layers have the same refractive index as shown

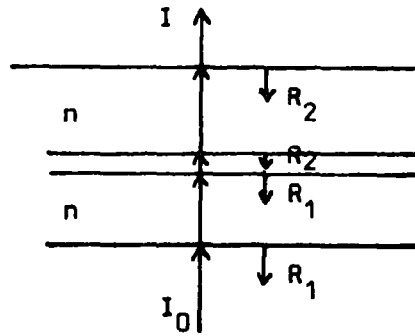


Figure IV.2 A combination of non-absorbing layers separated by Air, both layers have same refractive index 'n'

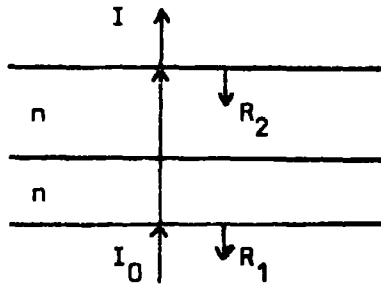


Figure IV.3 A combination of non-absorbing layers in optical contact, both layers have the same refractive index, 'n'

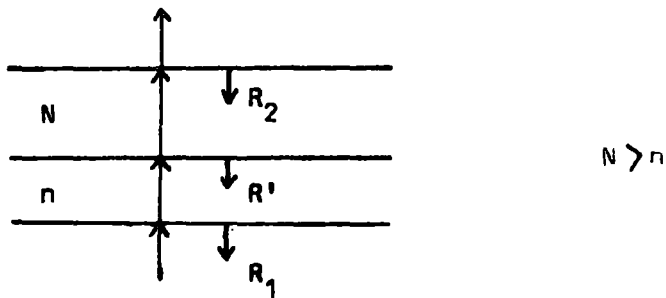


Figure IV. 4 A combination of non-absorbing layers in optical contact, refractive indices of layers are different.

in Figure IV 3, then

$$T = \frac{(1 - R_1)(1 - R_2)}{1 - R_1 \cdot R_2} = \frac{(1 - R)^2}{1 - R^2} = \frac{1 - R}{1 + R} \quad (6)$$

(3) Transmission of a combination of non-absorbing layers in optical contact when refractive indices of layers are different as shown in Figure IV 4, then

$$T = \frac{(1 - R_1)(1 - R') (1 - R_2)}{1 - R_1 \cdot R' \cdot R_2} \quad (7)$$

where

$$R_1 = (n - 1)^2 / (n + 1)^2$$

$$R' = (N - n)^2 / (N + n)^2 ; \quad N > n.$$

$$R_2 = (N - 1)^2 / (N + 1)^2 .$$

(4) Transmission of a combination of an absorbing layer and a non-absorbing layer in optical contact as in Figure IV 5

$$T = \frac{(1 - R_1)(1 - R') \mathcal{T} (1 - R_2)}{1 - R_1 \cdot R' \cdot R_2 \mathcal{T}^2} \quad (8)$$

where $\mathcal{T} = e^{-mt}$ (Bouguer's law), is called internal transmittance.

There is no parallel transmitted beam of light through a faceted gemstone since total back reflection takes place from inclined faces as shown in Figure IV 6. A parallel transmitted beam of light through such stones can be obtained when immersed in liquid of the same refractive index as shown in Figure IV 6B.

By this way transmission measurement is independent of

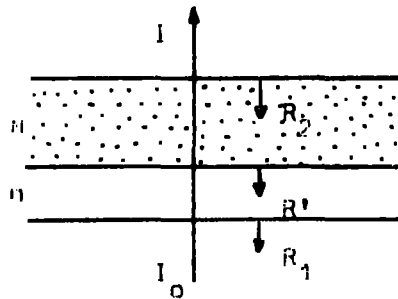


Figure IV 5. Transmission of a combination of an absorbing layer and a non-absorbing layer in optical contact.

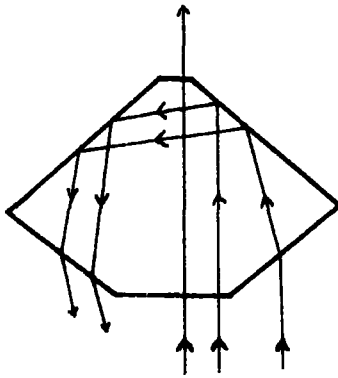


Figure IV 6 a. Back reflection of a ray of light in passing through a faceted stone in air.

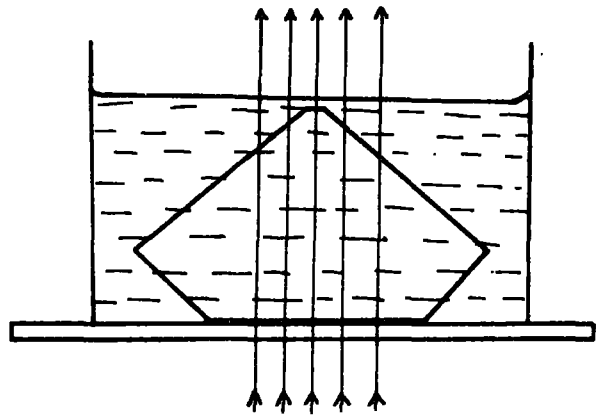


Figure IV 6 b. A parallel transmitted ray of light in passing through a faceted stone in oil of the same R.I.

i/ the shape of a gemstone and the difficulty of preparation for a thin parallel sided section is eliminated. The above equation, therefore, is also applicable for faceted gemstones (without spoling a stone) at a particular wavelength, where the refractive index of a stone and liquid is the same.

When measurement of spectral transmission based on this principle is made errors will be introduced in the experiment.

The probable errors considered are

- (1) due to the excess liquid over the stone and
- (2) due to higher dispersion of the liquid for other wavelengths, except the matching wavelength.

In the first case the error may be eliminated if the matching liquid is colourless. If the liquid has a negligible amount of absorption at thin layer over the stones the consequent error may also be regarded as negligible. However for accurate measurement colourless immersion liquids are preferable.

In the second case the amount of errors depends upon the dispersion character of a liquid used in the measurement. Unfortunately, the dispersion of immersion liquids are normally higher than that of solids as, for example, shown in Figure IV 7. (after Wahlstrom, 1966). The refractive index of solid and liquid will be matched at a particular wavelength, and so parallel transmitted beam of light through a faceted stone occurs.

i/ For other wavelengths in the visible spectrum the deviation of refractive indices of liquid from those of a solid will cause the divergence of the transmitted beam from parallelism. From the theoretical point of

view, the refractive index of liquid at every wavelength could be matched with that of the solid by adjusting the temperature, since the R.I. of liquid is much more sensitive to temperature than that of a solid. But in practice it will not be very convenient and costly to do it.

However calculation shows that the estimate of the error due to a difference of 0.01 n between liquid and solid is about 0.25% in transmission and due to a difference of 0.1 n is about 2.5% . If the second case is considered as a maximum error the method of transmission measurement in oil is reasonably acceptable.

The relative values of spectral transmission are of interest for colour measurement of a particular gemstone. If the dispersion of liquid is not too much stronger than that of the stone, and as long as the same liquid is used for different cut stones of the same mineral, their relative transmission values are comparable and therefore their colour values are also comparable to one another.

The equation for spectral transmission of an isotropic absorbing gemstone in oil (of the same n as that of the stone) may be written as

$$T_{\lambda} = \frac{(1 - R_1\lambda)(1 - R_2\lambda)}{1 - R_1\lambda \cdot R_2\lambda} \quad (9).$$

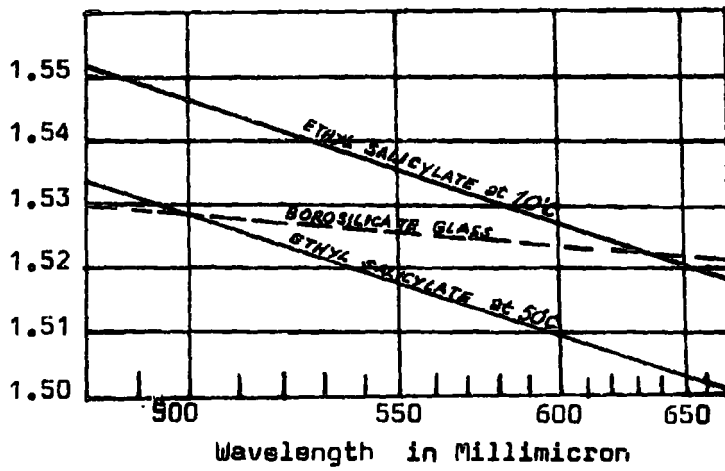


Figure IV.7 Dispersion curves plotted on a Hartman net and showing relationship among the index of refraction, wavelength, and temperature (after Wahlstrom, 1966)

EXPERIMENTS AND RESULTS

To establish the validity of the above considerations transmission measurements were made and measured values were compared with those obtained by calculation according to equation (7) for non-absorbing specimens.

For an absorbing mineral two methods of transmission measurement were applied according to the conditions for equation (8) and equation (9). The equation (9) is the same as equation (8) which is from equation (7), but the measuring methods are different. Hence the results obtained by two methods should be the same.

Measurements were made as follows.

The same reflectivity apparatus was used for transmitted light work, with some alterations of the equipment, as shown in the diagram III 2. In the microscope assembly an iris diaphragm below the polarizer, a measuring diaphragm, of about 80 micron in diameter, above the objective, a photomultiplier stop, of about 50 micron in diameter (the same as in reflectivity measurement), below the photomultiplier tube-window, and the condensing lenses just below the microscope stage were used. By using the condensing lenses the image of the light source can be brought up to the specimen surface.

A glass cell was made by joining one end of a glass tube to a surface of a polished petrographic glass slide with 'araldite'. The glass cell was thoroughly cleaned and placed on the microscope stage. The image of the light source was focused as in proper way. Intensities of the light source (I_0) at various wavelengths were read

on the digital voltmeter.

One side of a basal section of clear quartz (rock crystal) with polished parallel surfaces was wetted with a drop of Rayner's R.I. liquid of 1.54 n and was placed with gentle pressure on the glass slide in such a way that a thin film of the liquid served to make 'optical contact' with the specimen and the glass slide (i.e. the entire space between the specimen and the glass slide was completely filled without any air bubbles).

An area of the specimen as free from imperfection as possible was selected and the image of the light source was focused. So the distance between the surfaces of photo cells and the source image was always the same when the specimen was in or without the specimen in the plane polarized light.

Intensities of the emergent light passing through the glass slide and the specimen were read for various wavelengths. Therefore the spectral transmittance of the combination of the glass slide and clear quartz (w) in optical contact is

$$T = I / I_0 .$$

In order to compare the measured values with calculated results spectral refractive indices of the same glass slide and the quartz specimen were determined, at working temperature of about 20°C, with an Abbe Refractometer model B and a quartz monochromator. Direct readings on the scale for various wavelengths were corrected by using given calibration tables for the dispersion characteristics of the instrument as a function of the wavelength.

Fresnel reflectances of the quartz specimen, glass slide and quartz-glass interface were computed thereafter and substitutions of these values in the formula (7) gave spectral transmittances of the combination of the two non-absorbing layers at optical contact.

Measured spectral refractive indices and calculated Fresnel reflectances are tabulated in Table IV 1. Measured transmission values and calculated results of the combination of clear quartz and the glass slide at optical contact are given in Table IV 2. The measured values are slightly higher than the calculated values (N.B. according to Ayres, 1949 and Bloss, 1955, the relative errors increase considerably when transmission is lower than about 20% and higher than 60%).

Spectral transmission measurements of a combination of a basal section of brown quartz (cairnngorm) and the glass slide were made by two methods.

The first method was as described above for a combination of non-absorbing and absorbing layers (equation 8) in optical contact at interfaces. In the second method the brown quartz basal section was measured in the matching liquid (equation 9). The polished surface of the brown quartz specimen was thoroughly cleaned and wetted with the same Rayner R.I. liquid of 1.544 n. The specimen was placed on the glass in the cell as mentioned above and the cell was filled with the same liquid till it just covered the top of the specimen surface. Measurement, as close as possible to the previous area was made after focusing to the image of the light source.

The measured values by two methods are given in Table

IV 3. and are presented by curves in Figure IV 8. All calculated and measured transmission results were then computed for quantitative colour values. For the clear quartz specimen its chromaticity coordinates must be the same as that of the achromatic point since it is colourless.

For brown quartz its colour coordinate points derived from two transmission measurements must also be more or less the same, or at least fall within the limits of a discrimination threshold. These conditions were satisfied as evidences of the proof are given in Table IV 4.

Thus statistical and experimental results permit the use of matching liquid to determine the spectral transmission characteristics of faceted gemstones.

For an anisotropic gemstone the transmission T is according to the crystallographic orientation under examination. At a wavelength where the R.I. of the stone matches that of the liquid the image of the parallel transmitted beam is at the centre of the field of view and the boundary of the image is sharp. During a gentle rotation of the microscope stage and sliding the monochromator for other wavelengths the image of the emergent light slightly deviates from the centre of the field due to the different dispersion characteristics of liquid and the stone. In this case values of minimum and maximum transmissions were measured in plane polarised light at each of the two extinction positions. It was considered that it may be permissible to take the mean transmission values of an anisotropic stone for general colour measurement.

Table IV 1.

Wavelength nm	Measured Refractive Indices	
	Clear Quartz n_w	Glass Slide n
460	1.5517	1.524
480	1.5501	1.523
500	1.5488	1.522
520	1.5476	1.520
540	1.5465	1.519
560	1.5455	1.518
580	1.5446	1.517
600	1.5439	1.516
620	1.5431	1.515
640	1.5424	1.514

Calculated Fresnel Reflectances ($r = \frac{(n - N)^2}{(n + N)^2}$)

Wavelength nm	Clear Quartz r_w	Glass Slide r	Quartz-Glass interface r_i
460	0.04675	0.04310	0.000081
480	0.04653	0.04297	0.000078
500	0.04636	0.04284	0.000076
520	0.04620	0.04258	0.000081
540	0.04606	0.04245	0.000080
560	0.04592	0.04232	0.000081
580	0.04581	0.04219	0.000081
600	0.04571	0.04206	0.000083
620	0.04561	0.04193	0.000084
640	0.04552	0.04180	0.000086

Table IV 2. T % of a Combination of Clear Quartz and a Glass Slide
in Optical Contact

Wavelength nm	Calculated T_w %	Measured T %
460	91.21	91.41
480	91.24	91.83
500	91.27	91.90
520	91.31	92.02
540	91.34	92.36
560	91.36	92.66
580	91.39	92.69
600	91.41	92.99
620	91.43	93.12
640	91.53	93.36

Table IV 3. Measured T % of a Combination of Brown Quartz(w)
and a Glass Slide

Wavelength nm	in optical contact at interfaces	in matching oil n = 1.544
440	43.97	42.82
460	44.84	43.86
480	45.67	45.01
500	46.95	46.50
520	48.55	48.34
540	50.98	50.52
560	53.36	52.98
580	55.70	55.11
600	58.02	57.30
620	59.80	59.63
640	61.52	61.98
660	64.10	64.10

Figure IV 8.

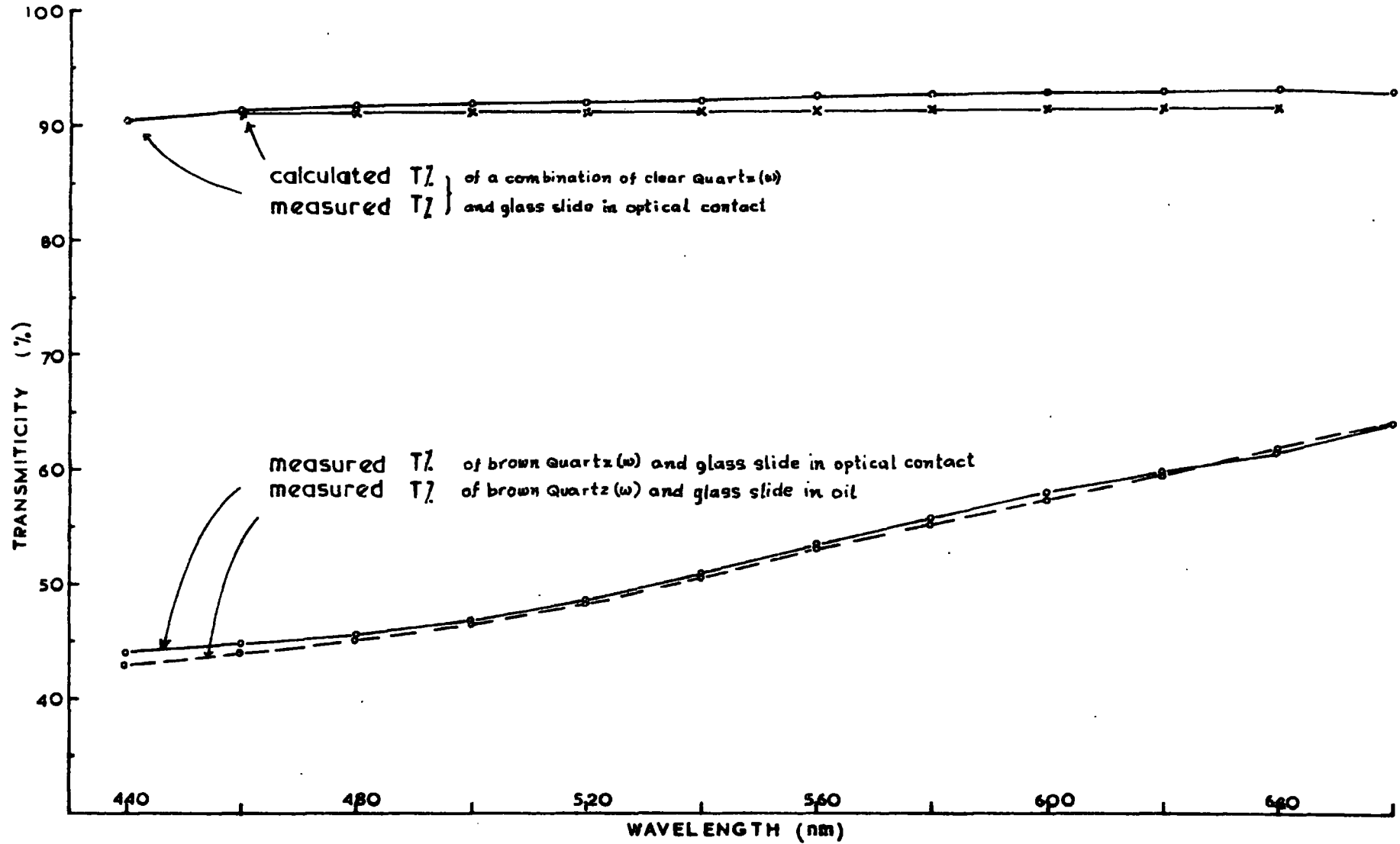


Table IV 4. Colour Specifications derived from Calculated and
Measured Transmittance Values

Source & Specimens	Chromaticity coordinates		Dominant wavelength d	Excitation purity Pe%
	x	y		
Illuminant A	0.4476	0.4075		
Clear Quartz(w) in air, calculated	0.4477	0.4075		
Clear Quartz-Glass in optical contact, calculated	0.4478	0.4075		
Clear Quartz-Glass in optical contact, measured	0.4491	0.4078		
Brown Quartz-Glass in optical contact, measured	0.4712	0.4090	591.42	17.39
Brown Quartz-Glass in matching oil, measured	0.4722	0.4094	591.10	18.31

V. COMPARISON OF TRANSMISSION MEASUREMENTS MADE WITH A MICROSCOPE
PHOTOMETER AND A GRATING SPECTROGRAPH

Spectral transmission measurements have been usually carried out with a recording grating spectrograph (or spectrophotometer). Such apparatus is not suitable to make measurement of very small specimens (< 2 mm in diameter) and specimens with non-parallel sided surfaces. Therefore, for non-opaque minerals and faceted gemstones, spectral transmission measurements were made with a microscope photometer (the same apparatus used for reflectivity measurement ; as described previously).

It was interesting to compare transmission measurements made with two different types of apparatus. For this purpose specimens were prepared as follows.

A red spinel crystal of gem quality, from Burma and a basal section of brown quartz (cairngorm) were embedded in synthetic resin discs. They were ground with medium grade silicon carbide on the first lap and with fine silicon carbide on the second lap. They were then polished with fast cutting alumina, with 6 micron diamond paste, with $\frac{1}{4}$ micron diamond paste and finally with 'finish polishing' alumina. The other sides of the embedded specimens were cut off and these second opposite surfaces of red spinel and brown quartz were ground and polished down to 1.5 mm and 1.0 mm in thickness respectively. Minute solid inclusions in both specimens were observed under the microscope with a low-power objective .

Measurements with the Microscope Photometer

Spectral intensity of the incident beam of light (I_0)

of about 80 micron in diameter was measured before and after measurement of a specimen, as described in the previous section.

The polished specimen was cleaned thoroughly with xylene and soft tissues. It was placed on the microscope stage and an area as clean and free from imperfection as possible was selected. Spectral intensity of the transmitted beam (I) of plane polarised light was recorded from wavelengths of 400 nm to 680 nm. The specimen was turned over to measure the second surface in the same way. The average of two measured I_0 was used to calculate spectral transmittance (in percent) of the specimen as

$$T \% = I / I_0 \times 100.0$$

The measured areas of the first surface and second surface of the specimen may not be the same. However, to estimate the precision of the measurement repeated experiments were made. The spectral transmittance values (%) of red spinel and brown quartz are given in Tables V 1 and V 2. The maximum variation of the average values of two measurements was less than 1 % (absolute) at any wavelength.

Measurements with an Optica Recording Grating Spectrograph

The optica recording grating spectrograph manufactured by Optica United Kingdom Limited has a double beam optical system and a working range of 185 nm to 3200 nm. It has a 600 lines/mm ruled grating with a dispersion of 16 Å/mm for the UV and visible regions (185 nm to 1000 nm).

Holders for a specimen and the reference beam of light

of 2 mm in diameter were used. The holders were specially made, as those supplied with the instrument are too large. The deflections of the potentiometers on the chart recorder were set to 0 and 100 over the visible spectrum by means of the 0% and 100% transmission controls, when the light beam was cut off and when the incident beam was passed through the reference holder respectively.

The same specimens were first cleaned thoroughly with xylene and one face of a specimen was stuck on the back of the specimen holder having a 2 mm hole. Measurement of the first surface was made from wavelengths of 400 nm to 700 nm under the following operational conditions.

Source: tungsten lamp(12V, 48W) from 400 to 600 nm
 " " " " with a red filter from
 600 to 700 nm.

Photomultiplier Type: RCA 1P 28 with a maximum response at about 340 nm.

Slit: automatic slit used to obtain a constant 100%
 transmission of the incident beam in air.

Time Constant: normal.

Gain: 6 (optimum condition).

Wavelength Scan Speed: 0.5 A°/sec.

Chart Speed: 2°/minute.

Accuracy of the wavelength on the chart was 5 ± 1 A°.

The measurement of the opposite second surface was repeated in the same way. Transmittance values (%) thus obtained are given in Tables: V 3 and V 4.

For comparison of the measurements made with two types

of different apparatus the spectral transmitticity (i.e. transmittance in %) curves of red spinel and brown quartz are shown in Figures V 1 and V 2.

From tabulated measured values and Figures V 1 and V 2 , it was seen that measurements with the spectrograph gave large variation in transmittance values of the first and second surfaces of the specimen. That is probably because of the different polishing qualities of opposite surfaces and imperfect parallelism of the section. Nevertheless, these large variations do not considerably affect quantitative colour measurements as shown in the chromaticity chart of Figure. V 3.

With the microscope photometer a specimen can be examined and areas of both surfaces can be selected as free from imperfection as possible. For this reason the variation in transmittance values of the first and second surfaces of the specimen was comparatively smaller. Another advantage of the use of the microscope photometer was that a very small specimen could be measured.

The differences in spectral transmittance values, within the experimental errors, obtained with two different apparatus may be due to the following major reasons.

- (1) Spectral sensitivities, over the visible spectrum, of the two types of photomultiplier are different as shown by curves in Figure V 4.

Results of colour matching experiments however, show that for different persons appreciable differences in light sensitivity of retinal photocells result in slightly different sensation of the same colour.

Likewise, two apparatus differing in sensitivity of light for each wavelength produced two different spectral responses resulting slightly different colours of same specimen.

- (2) Fields of illumination and measuring sizes were different, although the same type of sources were used.
- (3) There is a lower polarizer in the microscope photometer but no polarizer in the spectrograph.

Quantitative colour values were derived from the above measured transmittance values and results are given in Tables V 5 and V 6. Their colour points are plotted in the chromaticity diagrams of Figures V 3.

Table V. 1 .

TRANSMITTICITY(Transmittance in %)OF RED SPINEL (111) MEASURED WITHTHE MICROSCOPE PHOTOMETER.

(Thickness = 1.5 mm)

First Measurement				Second Measurement		
(nm)	Face 1	Face 2	Average	Face 1	Face 2	Average
400	9.7	10.1	9.9	10.6	10.6	10.6
410	11.0	11.1	11.05	11.5	11.0	11.25
420	13.3	13.3	13.3	13.9	12.6	13.25
430	17.1	17.1	17.1			
440	22.5	22.1	22.3	23.0	21.25	22.1
450	28.7	28.1	28.4			
460	34.0	33.0	33.5	34.4	32.15	33.3
470	37.4	36.0	36.7	37.8	35.4	36.6
480	37.9	36.4	37.15	38.6	36.1	37.35
490	35.9	34.3	35.1			
500	31.9	30.5	31.2	32.7	30.3	31.5
510	26.8	25.6	26.2			
520	22.2	21.3	21.75	23.0	21.2	22.1
530	18.5	18.0	18.25			
540	16.6	16.1	16.35	17.2	15.7	16.45
550	16.5	16.0	16.25	17.0	15.6	16.3
560	18.3	17.8	18.05	18.85	17.3	18.1
570	22.3	21.5	21.9			
580	28.15	27.0	27.6	28.4	26.2	27.3
590	35.3	33.6	34.45			
600	42.6	40.1	41.35	42.8	39.55	41.2
610	49.3	46.6	47.95			
620	55.1	51.7	53.4	55.0	50.9	52.95
630	59.9	55.9	57.9			
640	62.8	58.9	60.85	63.5	58.5	61.0
650	65.7	60.95	63.3			
660	66.7	63.0	64.85	65.6	62.55	64.1
670	68.5	64.0	66.25	67.9	62.9	65.4
680	67.8	64.9	66.35	68.1	66.0	67.05

Table V. 2 .

TRANSMITTICITY OF BROWN QUARTZ(W) MEASURED WITH THE MICROSCOPE PHOTOMETER.
 (thickness = 1.0 mm)

(nm)	Face 1	Face 2	Average
400	41.3	43.6	42.25
410	42.3	44.8	43.55
420	42.9	44.9	43.9
440	43.1	45.3	44.2
460	43.4	45.8	44.6
480	44.0	46.5	45.25
500	45.1	47.6	46.35
520	46.4	49.1	47.75
540	48.1	50.8	49.45
560	49.9	52.7	51.3
580	51.4	54.4	52.9
600	53.3	56.3	54.8
610	54.4	56.9	55.65
620	55.0	57.6	56.3
630	55.9	58.7	57.3
640	56.9	59.6	58.25
650	57.4	60.4	58.9
660	59.4	60.9	60.15
670	60.0	62.0	61.0
680	59.6	61.7	60.65

Table V 3.

TRANSMITTICITY OF RED SPINEL (111) MEASURED WITH THE OPTICA RECORDING

GRATING SPECTROGRAPH

(Thickness = 1.5 mm)

First Measurement				Second Measurement		
(nm)	Face 1	Face 2	Average	Face 1	Face 2	Average
400	5.6	3.25	4.4	3.3	2.3	2.8
410	8.0	4.7	6.35	4.6	4.0	4.3
420	10.0	5.1	7.55	6.0	5.0	5.5
430	15.7	8.9	12.3	10.0	8.6	9.3
440	21.7	17.4	19.55	17.5	15.5	16.5
450	38.1	24.9	31.5	25.4	22.4	23.9
460	45.5	30.0	37.75	31.0	27.1	29.05
470	49.6	32.9	41.25	33.6	30.0	32.0
480	49.0	32.5	40.75	34.2	30.0	32.1
490	44.3	29.4	36.85	31.0	27.1	29.05
500	36.9	24.3	30.6	26.0	22.8	24.4
510	28.5	19.0	23.75	20.0	18.2	19.1
520	21.4	14.3	17.85	15.0	13.5	14.25
530	17.0	11.4	14.2	12.0	10.7	11.35
540	15.7	10.5	13.1	11.2	9.9	10.55
550	17.0	11.4	14.2	11.5	10.2	10.85
560	21.5	14.6	18.05	14.8	13.2	14.0
570	29.0	20.0	24.5	20.0	18.0	19.0
580	39.6	27.5	33.55	28.0	25.0	26.5
590	50.8	35.3	43.05	36.7	32.0	34.35
600	60.8	42.6	51.7	44.4	39.3	41.85
610	69.0	48.3	58.65	50.4	45.4	47.9
620	74.3	52.0	63.15	55.0	49.3	52.15
630	78.3	54.8	66.55	58.3	52.1	55.2
640	80.4	56.7	68.55	60.2	54.0	57.1
650	82.2	57.6	69.9	61.7	55.5	58.6
660	83.3	59.0	71.15	63.0	57.0	60.0
670	84.3	59.7	72.0	64.2	58.0	61.1
680	84.6	60.4	72.5	64.6	58.4	61.5
690	83.7	60.0	71.85	64.5	58.2	61.35
700	83.9	60.5	72.2	64.5	58.8	61.65

Table V . 4 .

TRANSMITTICITY OF BROWN QUARTZ(W) MEASURED WITH OPTICA RECORDINGGRATING SPECTROGRAPH.

(Thickness = 1.0 mm)

(nm)	Face 1	Face 2	Average
400	26.4	41.8	34.1
410	26.6	42.0	34.3
420	26.9	42.0	34.45
430	27.0	42.2	34.6
440	27.1	42.2	34.65
450	27.3	42.4	34.85
460	27.5	42.4	34.95
470	27.7	42.5	35.1
480	28.0	42.7	35.35
490	28.3	43.2	35.75
500	28.6	43.7	36.15
510	29.3	44.2	36.75
520	29.9	44.8	37.35
530	30.5	45.6	38.05
540	31.4	46.5	38.95
550	32.0	47.5	39.75
560	32.8	48.7	40.75
570	33.2	49.4	41.3
580	33.8	50.3	42.05
590	34.6	51.0	42.8
600	35.4	52.1	43.75
610	36.1	54.5	45.3
620	36.9	55.4	46.15
630	37.5	56.5	47.0
640	38.2	57.5	47.85
650	38.5	58.7	48.6
660	39.8	59.9	49.85
670	40.8	60.9	50.85
680	41.6	61.8	51.7
690	42.8	62.9	52.85
700	43.9	64.0	53.95

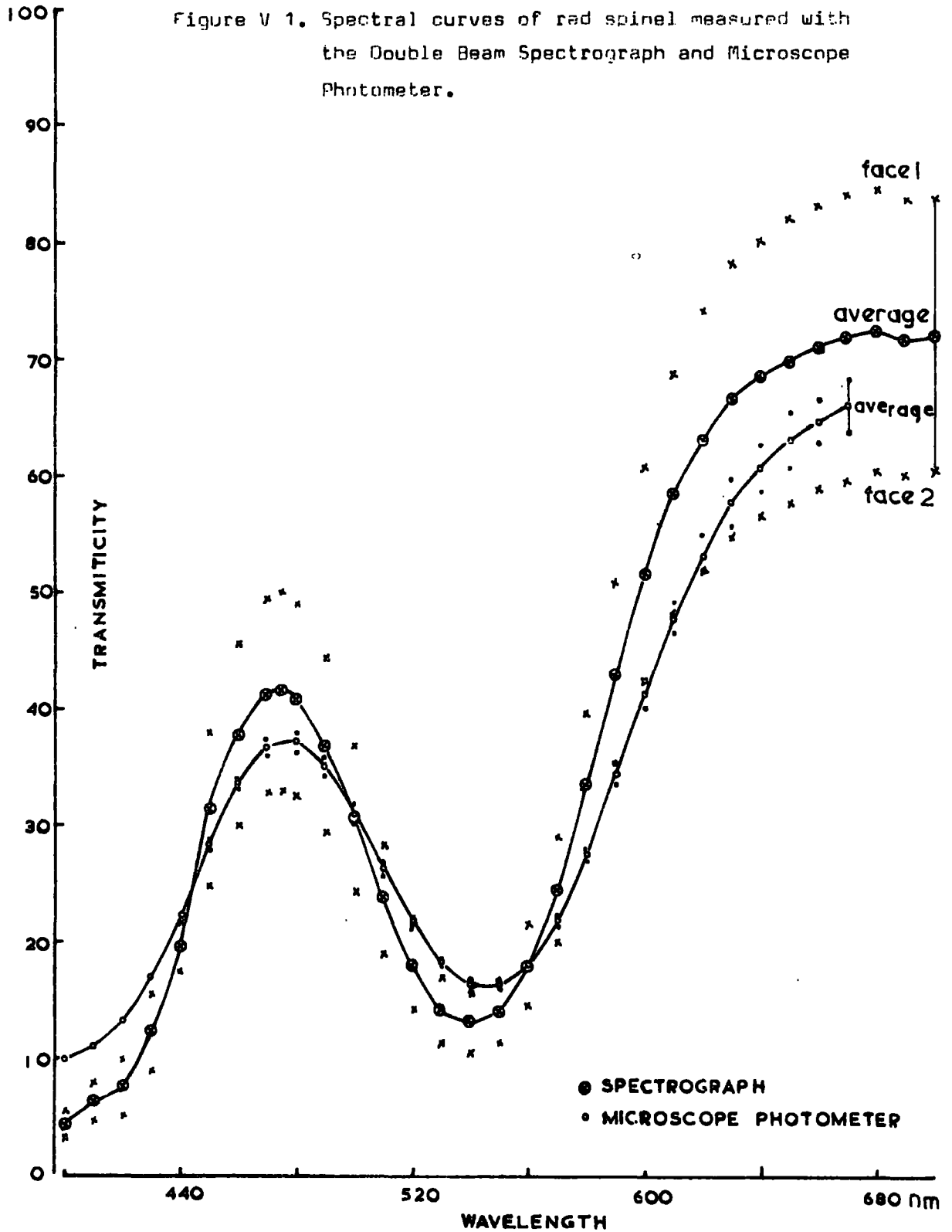


Figure V 2. Spectral transmittance curves of brown quartz measured with the Double Beam Spectrograph and Microscope Photometer.

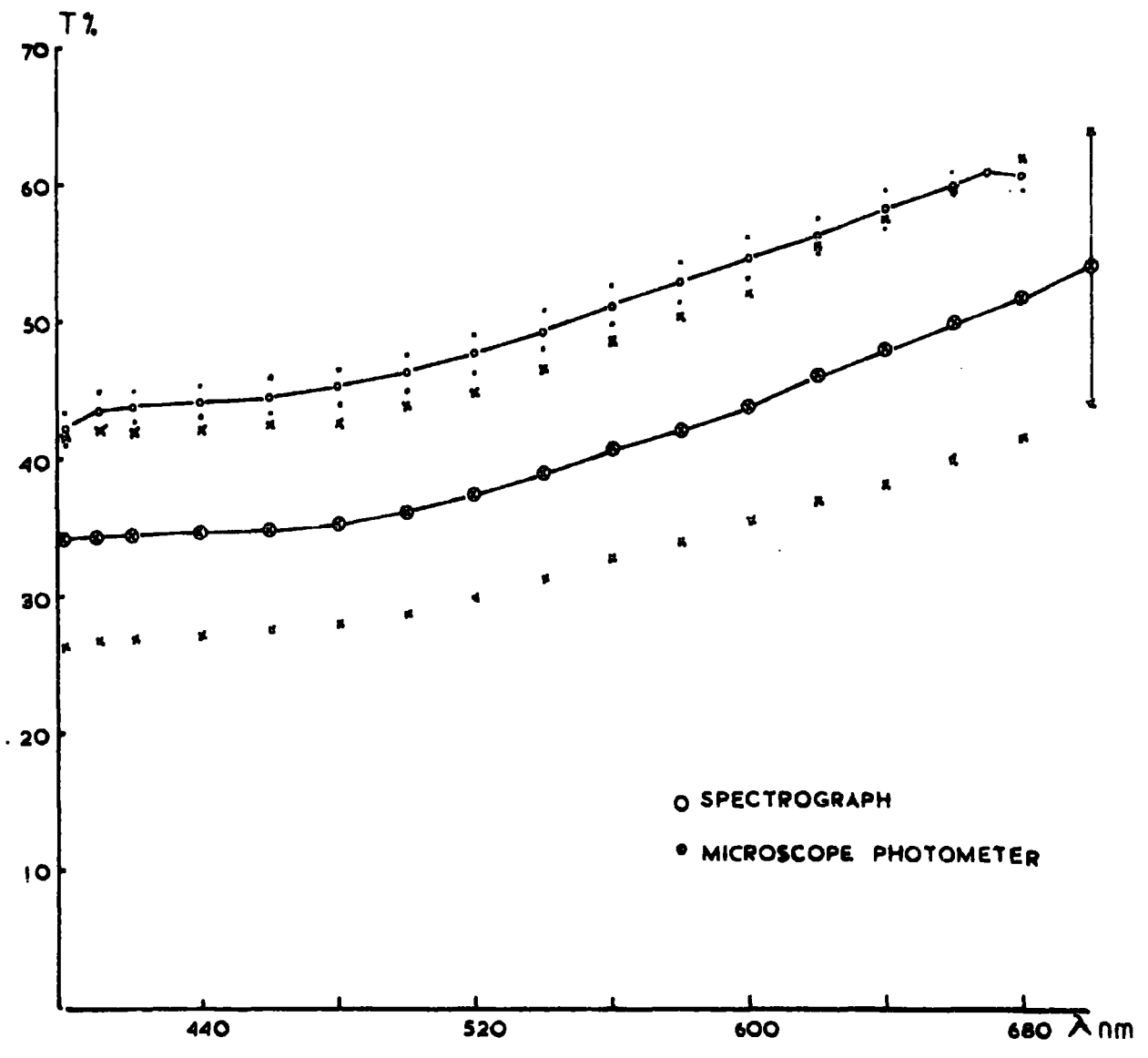


Figure V 4. Chromaticity diagram showing the colours of Red Spinel (1.5 mm) and Brown Quartz (w, 1.0 mm) under the C.I.E. source C. The differences in chromaticities are mainly due to two different types of apparatus used for transmission measurements.

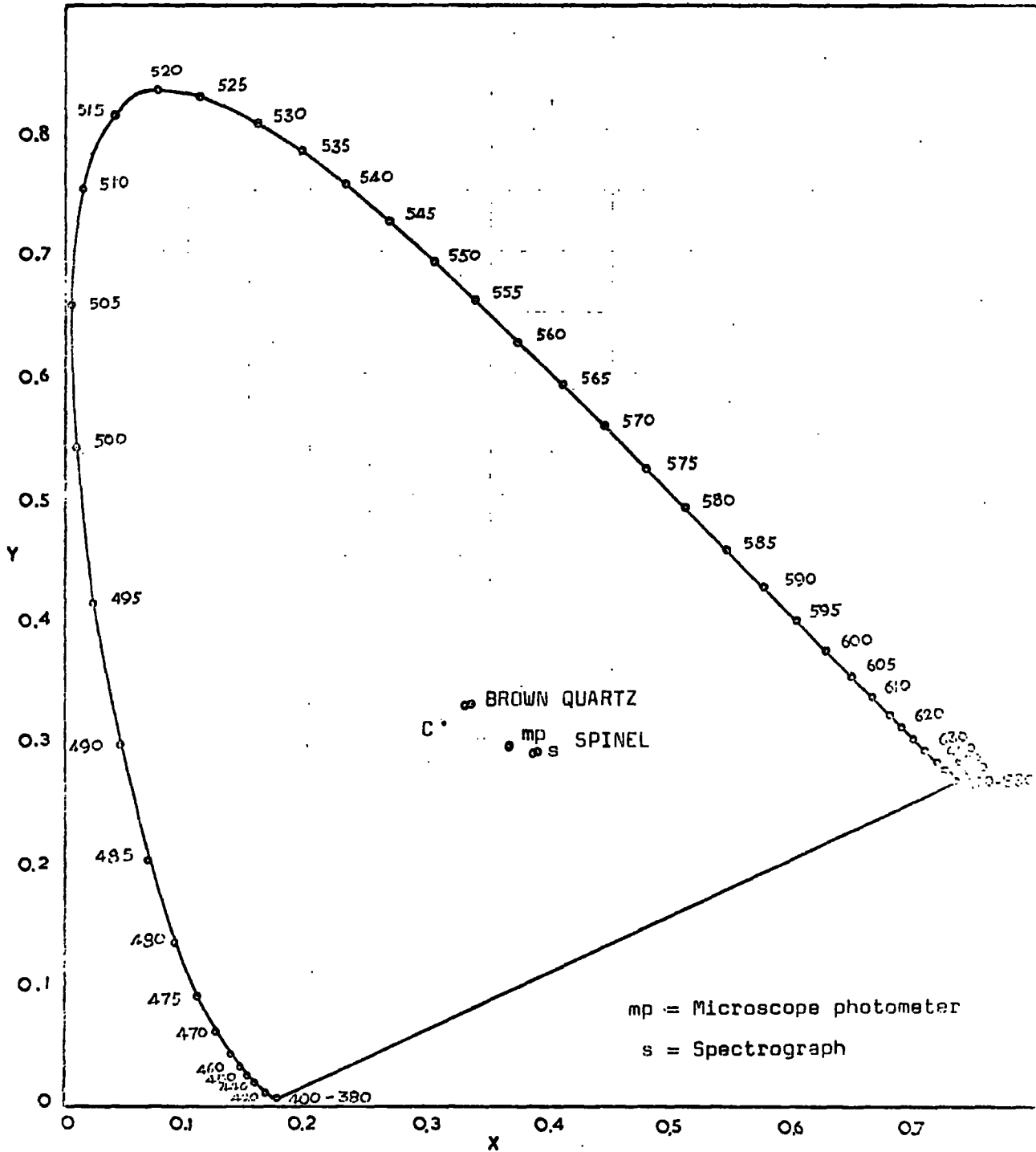


Figure V 4. Spectral sensitivities of the two types of photomultipliers.

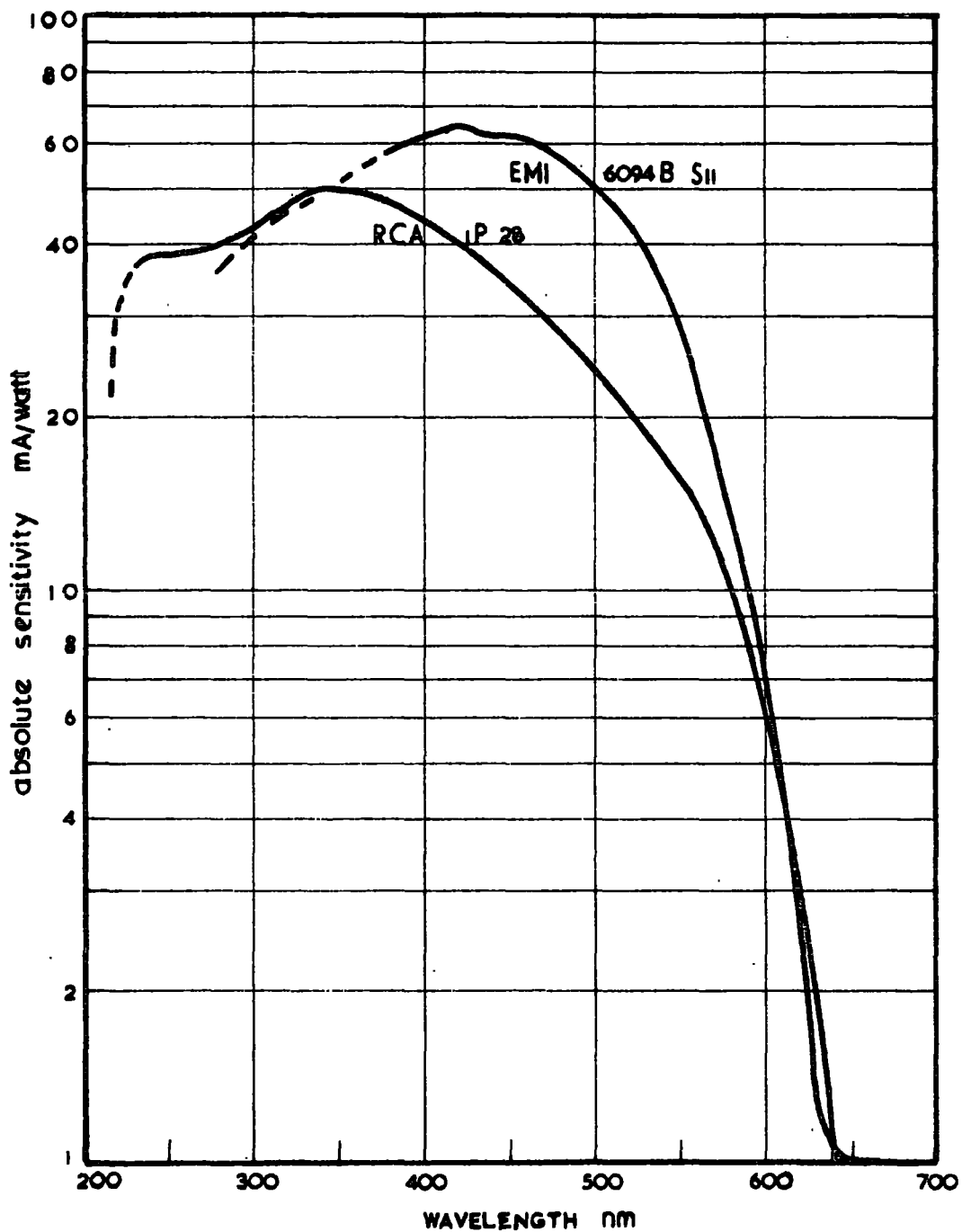


Table V/ 5

QUANTITATIVE COLOUR VALUES OF RED SPINEL (111)UNDER AVERAGE TUNGSTEN LAMP LIGHT

	CHROMATICITY COORDINATES		COMPLEMENTARY	EXCITATION
	x	y	WAVELENGTH(λ_c)	PURITY(Pe%)
T % Measurement with the Microscope Photometer.				
First measurement				
Face 1	0.5226	0.3602	- 506.51	29.81
Face 2	0.5196	0.3607	- 506.92	29.12
Average	0.5211	0.3604	- 506.73	29.48
Second measurement				
Face 1	0.5197	0.3621	- 506.49	28.63
Face 2	0.5195	0.3615	- 506.70	28.82
Average	0.5196	0.3618	- 506.60	28.72
T % Measurement with the Optica Recording Grating Spectrograph.				
First measurement				
Face 1	0.5349	0.3543	- 506.07	34.02
Face 2	0.5393	0.3541	- 505.51	34.84
Average	0.5367	0.3542	- 505.84	34.36
Second measurement				
Face 1	0.5410	0.3533	- 505.47	35.42
Face 2	0.5422	0.3536	- 505.25	35.51
Average	0.5416	0.3534	- 505.37	35.48

Table V 5 (continued)

QUANTITATIVE COLOUR VALUES OF RED SPINEL (111)UNDER AVERAGE DAYLIGHT.

	CHROMATICITY COORDINATES x	COORDINATES y	COMPLEMENTARY WAVELENGTH(λ_c)	EXCITATION PURITY(Pe%)
T % Measurement with the Microscope Photometer.				
First measurement				
Face 1	0.3671	0.2939	- 495.69	19.69
Face 2	0.3636	0.2926	- 496.30	19.56
Average	0.3654	0.2933	- 495.97	19.62
Second measurement				
Face 1	0.3646	0.2954	- 495.57	18.62
Face 2	0.3640	0.2944	- 495.85	18.91
Average	0.3643	0.2949	- 495.71	18.77
T % Measurement with the Optica Recording Grating Spectrograph.				
First measurement				
Face 1	0.3827	0.2903	- 495.26	24.06
Face 2	0.3887	0.2919	- 494.69	24.54
Average	0.3851	0.2909	- 495.03	24.27
Second measurement				
Face 1	0.3901	0.2920	- 494.60	24.76
Face 2	0.3919	0.2933	- 494.33	24.57
Average	0.3910	0.2926	- 494.47	24.69

Table V/6

QUANTITATIVE COLOUR VALUES OF BROWN QUARTZ (λ)
UNDER AVERAGE TUNGSTEN LIGHT.

	CHROMATICITY COORDINATES		DOMINANT	EXCITATION
	x	y	WAVELENGTH(λ_d)	PURITY(P_e %)

T % Measurement with the Microscope Photometer.

Face 1	0.4672	0.4082	592.06	14.06
Face 2	0.4665	0.4087	591.42	13.93
Average	0.4668	0.4085	591.68	13.99

T % Measurement with the Optica Recording Grating Spectrograph.

Face 1	0.4714	0.4085	591.91	17.18
Face 2	0.4701	0.4064	594.33	14.70
Average	0.4707	0.4073	593.18	15.36

Table V-6 (continued)

QUANTITATIVE COLOUR VALUES OF BROWN QUARTZ (w)
UNDER AVERAGE DAYLIGHT.

	CHROMATICITY COORDINATES		DOMINANT	EXCITATION
	x	y	WAVELENGTH(λ_d)	PURITY(Pe %)
T % Measurement with the Microscope Photometer.				
Face 1	0.3306	0.3292	584.07	8.96
Face 2	0.3303	0.3296	583.44	8.99
Average	0.3304	0.3294	583.72	8.96
T % Measurement with the Optica Recording Grating Spectrograph.				
Face 1	0.3355	0.3325	583.88	11.16
Face 2	0.3328	0.3281	586.72	9.25
Average	0.3339	0.3299	585.41	10.03

VI. METHODS OF COMPUTING COLOUR SPECIFICATIONS OF MINERALS

Description has already been made of the basic principles of colour measurement. In the past colour specification of objects have been usually carried out graphically in part, from spectral transmittance or spectral reflectance data (Hand Book of Colorimetry by Hardy, 1936; Piller, 1966). Since better accuracy and precision of colour values of minerals are preferable, the absolute mathematical approach is considered and computer programmes developed to evaluate all colour specifications. This has the advantage of eliminating personal errors and reducing the time required for calculation.

VI 1. Computational Procedure for Tristimulus values and Chromaticity Coordinates

Measurements of spectral reflectivity for opaque mineral and spectral transmittance for non-opaque minerals and cut gemstones could be made at various wavelengths only within the sensitivity range of the photomultiplier (i.e. from about 420 nm to 660 nm) since the colour-matching functions are very small below about 400 nm and above 680 nm, the reflectivity curve can be extended justifiably towards the two ends of the spectrum by extrapolation (see Figure II 14).

Therefore the values require for the calculations were then found by interpolation and extrapolation for wavelengths of 360 nm to 780 nm with an equal interval of 5 nm, by the following linear equation:

$$R_x = \frac{\lambda_x - \lambda_1}{\lambda_2 - \lambda_1} (R_2 - R_1) + R_1 \quad (1)$$

where R_x is a required reflectivity at the wavelength λ_x , λ_2 and λ_1 are longer and shorter successive wavelengths respectively, and R_2 and R_1 are the respective measured reflectivities.

The reflectivity values at 5 nm intervals are then multiplied by the corresponding normalized products of the colour-matching functions of the light source obtained from Table II 5. By definition, the tristimulus values XYZ of an object colour can be determined by evaluating the following integrals.

$$\begin{aligned} X &= \int R_\lambda H_\lambda \bar{x}_\lambda d\lambda \\ Y &= \int R_\lambda H_\lambda \bar{y}_\lambda d\lambda \\ Z &= \int R_\lambda H_\lambda \bar{z}_\lambda d\lambda \end{aligned} \quad (2)$$

Since the reflectivity (or transmittivity) values are in percent, the sum of the products for X is divided by 100.0, and similarly for Y and Z. Each integral extends over the visible spectrum at 5 nm intervals, that is an 81 - point multiplication and summation.

The chromaticity coordinates x , y , z are given by

$$x = \frac{X}{X + Y + Z}, \quad y = \frac{Y}{X + Y + Z}, \quad z = \frac{Z}{X + Y + Z} \quad (3)$$

PL / 1 Computer Programme

A PL/1 (Programming Language / One) programme written by the author and based on the above computational procedures (1), (2)

and (3), has the capacity to determine the required values from data for several samples at a time, for one or many different light sources.

The computer output includes measured reflectivity (or transmittivity) data (input), interpolated and extrapolated values and chromaticity coordinates (output). This programme and an example of listings of sample data are given in Appendices VI 1a and VI 1b. respectively .

VI 2. Computational Procedure for Dominant Wavelength, Complementary Wavelength and Excitation Purity

The dominant or complementary wavelength of a sample may be determined graphically from the standard chromaticity diagram (previously described). Judd (1933) gave a computational procedure to calculate the dominant wavelength and colorimetric purity* by linear interpolation from his tabular values given for the C.I.E. standard sources A,B,C and E. Judd's tabular values of slopes of the dominant wavelength lines can also be found in the book ' Color Science' by Wyszecki and Stiles (1967).

* Colorimetric purity = spectral luminance purity, is the ratio of the luminance of the spectrally pure component of mixture with achromatic component, matching the colour, to the luminance of the colour itself.

(CCOSA , 1968)

Such ratios of constant dominant wavelength (also complementary) for any other light source are evaluated by

$$\begin{aligned}
 r &= (x - X) / (y - Y) \\
 &\quad \text{if absolute value } (x - X) < (y - Y) \\
 r &= (y - Y) / (x - X) \\
 &\quad \text{if absolute value } (y - Y) < (x - X).
 \end{aligned}
 \tag{4}$$

where x,y and X,Y are coordinates of the spectrum and of the light source respectively.

A PL / 1 computer programme to compute tabular ratios was prepared and is given in Appendix VI 2. Tabular ratios for sources A,B and C are listed in Table VI 1.

The computation of the dominant wavelength (λ_d) or complementary wavelength (λ_c) by means of the Tables of Appendix VI 3, is as follows:

Knowing the chromaticity coordinates (x,y) of a sample irradiated by a source whose coordinates are (X,Y) the ratio r is computed from (4) above. The ratio r will be found lying between two tabulated ratios r_1 and r_2 which correspond to dominant wavelengths λ_1 and λ_2 respectively. The required dominant wavelength (λ_d) of the sample was evaluated by linear interpolation within the intervals of 1 nm, therefore

$$\lambda_d = \frac{(r - r_1)}{(r_2 - r_1)} (\lambda_2 - \lambda_1) + \lambda_1 .
 \tag{5}$$

To compute the excitation purity (Pe) it is required to

find the coordinates of the dominant wavelength on the spectrum locus. The dominant wavelength of a sample lies between the tabulated wavelengths λ_1 and λ_2 . Their corresponding chromaticity coordinates (x_1, y_1) and (x_2, y_2) were read from Table II 3, and the required coordinates (x_d, y_d) of the calculated dominant wavelength (λ_d) were obtained by linear interpolation.

$$x_d = \frac{(\lambda_d - \lambda_1)}{(\lambda_2 - \lambda_1)} (x_2 - x_1) + x_1$$

$$y_d = \frac{(\lambda_d - \lambda_1)}{(\lambda_2 - \lambda_1)} (y_2 - y_1) + y_1$$
(6)

The excitation purity (Pe) of the sample colour was then computed from one of the following formulae:

$$Pe = \frac{x - X}{\lambda_d - X} = \frac{y - Y}{\lambda_d - Y}$$
(7)

These two expressions are equivalent, but if the straight line is approaching vertical the first expression is less accurate and vice versa when the straight line is nearly horizontal, and therefore the smaller value with the least rejection error was taken.

To Compute Excitation Purities of Non-Spectral Colours

Excitation purities of spectral colours always have positive signs, but one or both values (Pe) for the stimulus of a non-spectral colour will have a negative sign or signs when the above procedures are carried out. Hence the negative sign or signs of

excitation purity would indicate that the required wavelength was not a dominant wavelength (λ_d) but a complementary wavelength (λ_c) and that the specimen had a non-spectral colour. Therefore it was necessary to recalculate the excitation purity for the complementary wavelength.

In order to calculate the excitation purity of a non-spectral colour it is first required to find the coordinates of its complementary wavelength on the purple line. This could be done in two ways.

(1) Chromaticity coordinates of complementary wavelengths of maximum purity for the C.I.E. sources A, B and C can be obtained from tabulated values in the book 'Science of Color' by the Committee on Colorimetry, 1968. The calculated complementary wavelength was then read from that table and linear interpolation from tabulated coordinates would give the required coordinates. This method is less accurate because wavelengths are not in linear scale.

(2) The second method used in this present work was as follows:

The equation of the purple line is

$y = mx + c$, where

$$m = (y_m - y_n) / (x_m - x_n),$$

$$c = (x_m y_n - x_n y_m) / (x_m - x_n)$$

and x_m, y_m and x_n, y_n are coordinates of the two extreme ends of the spectrum locus.

The slope and intercept of the straight line passing

through the illuminant point X, Y and sample point x_1, y_1 as shown in Figure VI 1 are

$$m_1 = (Y - y_1) / (X - x_1) \quad (9)$$

$$c_1 = (Xy_1 - x_1Y) / (X - x_1)$$

The required coordinates x', y' of the complementary wavelength (λ_c) are involved at the intersection of the above two straight lines .

Hence simultaneous equations give

$$\begin{aligned} x' &= (c_1 - c) / (m - m_1) \\ y' &= m \frac{(c_1 - c)}{(m - m_1)} + c \end{aligned} \quad (10)$$

The excitation purity (P_e) of a non-spectral colour was then calculated by the formulae (4) in which x' and y' replace x_d and y_d respectively.

PL/1 Computer Programme

A computer programme written by Dr. Reeves and the writer, based on the above equations (4) to (10) , has the capacity to compute dominant and/ or complementary wavelengths, coordinates and excitation purities of large numbers of specimen data, and to produce input and output data. This programme and an example of listings of necessary data are given in Appendices VI 4 a and VI 4 b respectively.

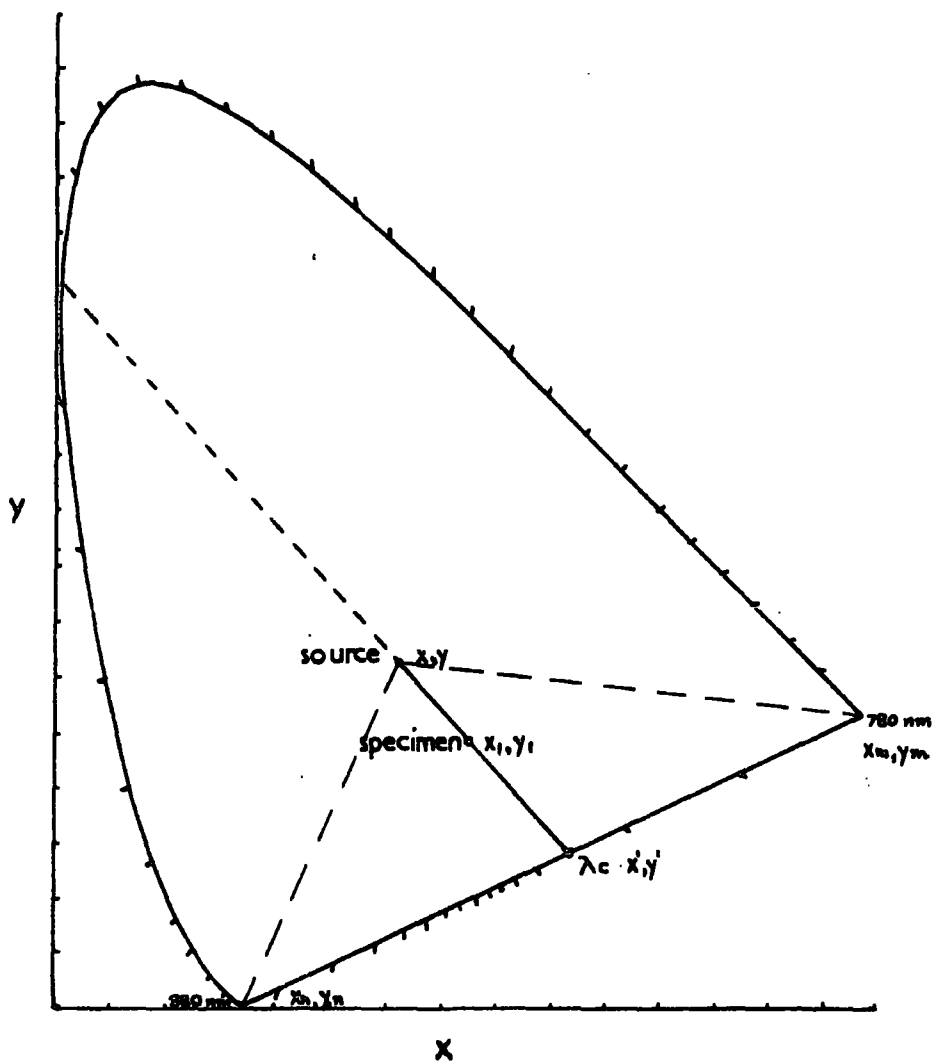


Figure VI 1. Explanation to compute complementary wavelength (λ_c) and excitation purity (P_e) of a non-spectral colour.

VII. ACCURACY AND PRECISION OF COLOUR SPECIFICATIONS
OF MINERALS

Spectral reflectivity data available from different papers were employed to evaluate colour values quantitatively. Different authors gave different accuracy and precision of reflectivity measurements. Piller and von Gehlen (1964) describe accurate measurements of reflectivity with relative errors about $\pm 0.5\%$ to $\pm 1\%$. Cameron (1963) claims an accuracy of $\pm 0.2\%$ at 549 nm (corresponding to a relative error of $\pm 1\%$ at a reflectivity of 20%).

Nichol and Phillips (1964) obtained a precision of $\pm 0.25\%$ at the reflectivity level of 18%. Rowie (1967) gives reflectivity data of some ore minerals with a precision of less than ± 0.5 absolute percent in white light. Levy (1966) obtained precision of about ± 0.2 to 0.6 from twenty measurements of spectral reflectivities.

It is to be pointed out that the precision and accuracy of spectral reflectivity measurements vary over the visible spectrum. This is the case to be considered seriously for reproducibilities of colour values of minerals, because calculations showed that a consistent error of 1% or 2% in every wavelength made no variation in calculated chromaticity coordinates, but small inconsistent error at some wavelengths only gave rise to differences in coordinates.

However, as the C.I.E. colorimetric system has limits of accuracy (threshold values), the accuracy and precision of colour values derived from spectral reflectivities can be estimated in terms of these threshold values.

The Weighted Ordinate method ($d\lambda = 5 \text{ nm}$) which is preferable to the Selected Ordinate method (previously described) was adopted as a standard method in this present work. It follows that the accuracy and precision depends on spectral reflectivity data and the numbers of wavelength intervals used in the measurements. Most of the reflectivity values given in the papers are only for a few wavelengths. Hence several interpolated values required to evaluate the integrals may deviate considerably from actual measurable values. This in turn depends upon the shape of the reflectivity curve.

In order to estimate the precision of quantitative colour values derived from available reflectivity data by the computer programmes it is necessary to estimate the true values. This was done by using reflectivity data with the smallest wavelength intervals. Then the precision of chromaticity values derived from large wavelength intervals can be compared to those obtained from small intervals.

A wide range of specimen data was selected to obtain strong, medium and weak colours with the chromaticity points scattered as regularly as possible around the standard illuminant points A and C.

Calculations for comparisons of results were carried out on the following basis. For a particular mineral measured values of reflectivity with smallest wavelength intervals were first used to derive the chromaticity coordinates. From the same series of measurement a few values of reflectivity were then selected and used to determine the effect of making measurements at wider wavelength intervals. So that deviations from the smallest intervals can be estimated as errors introduced in the colour values.

For example, pyrite has a comparatively medium colour saturation so originally the results of Demirsoy (1968), at 23 wavelengths, were used in the calculations and then values only 16, 12, 8 and 4 were used in turn, as shown in Table VII 1.

Table VII 1 shows the deviations of reflectivity values at omitted wavelengths from the actual measured values of 23 wavelengths, due to linear interpolation. The effective variations in the chromaticity coordinates due to such inconsistent deviations of spectral reflectivities over the spectrum are given in Table VII 2.

In the case of pyrite it was found that the chromaticity coordinates derived from the 23-point, 16-point and 12-point series were identical, although the interpolated reflectivities deviated from the actual measured values of the 23-point series. That is, the deviations ranging from 0.0 to ± 0.15 (see Table VII 2) in the visible spectrum indicate that the chromaticity values are not affected by errors of 0% to 0.3% in spectral reflectivities.

Due to linear interpolation the deviation of the 8-point and 4-point series from the actual measured values of the 23-point series ranged from 0.0 to ± 0.2 and ± 0.01 to ± 2.2 respectively (corresponding to relative errors of $\pm 0.4\%$ and $\pm 4\%$ for the upper limits).

Hence the chromaticity values of either the 23-point, 16-point or 12-point series can be taken as standard values for comparison and the deviations of the 8-point and 4-point values can be compared to the standard values, in terms of the MacAdam standard

deviation formula ($ds^2 = g_{11}dx^2 + 2g_{12}dx \cdot dy + g_{22}dy^2$; where the unit of $ds = 1$; previously described in Chapter II 12).

The chromaticity values obtained from the 8-point and 4-point series differed from the standard values in the fourth and third decimal places respectively, the results of calculations showed that the chromaticity point derived from the 8-point data was well within the threshold values, but the chromaticities of the 4-point data were only just within the threshold limit. It seems that the MacAdam's standard deviations are slightly large.

In the same way to estimate the accuracy and precision of colour values of minerals the calculated colour specifications and standard deviations for pyrite (Demirsoy, 1968), bornite (the writer), tennantite (Tugal, 1969), hematite (Demirsoy, 1968), covellite (von Gehlen and Piller, 1964) and cattierite (Demirsoy, 1968) are given in Table VII 2.

These considerations and results indicate that:

- (1) for most ore minerals 12- to 8-point measurements of reflectivity are sufficient for accuracy and precision,
- (2) for a strongly coloured mineral like covellite at least 12 measurements of reflectivity are necessary,
- (3) four-point data will only provide reasonably accurate values of colour within the thresholds when the reflectivity curve is fairly linear,
- (4) colour specifications derived from measured reflectivity (or transmitticity) values with regular approximate errors of 0 to + 2% over the visible spectrum are accurate within the small limits of the threshold values.

Table VII 1. Measured Reflectivity Values and Deviations at Omitted Wavelengths Due to Interpolation

Wave-length	23-point Series (measured)	16-Point Series	Deviation	12-Point Series	Deviation	8-Point Series	Deviation	4-Point Series	Deviation
440	40.9	40.9	-	40.9	-	40.9			+ .80
450	42.4	42.4	-		.00		.00		+ .40
460	43.9		.00	43.9		43.9		43.9	
470	45.5	45.5	-		+.05		+.05		- .50
480	47.2	47.2	-	47.2			.00		-1.10
490	48.7		+.15		+.15		+.15		-1.50
500	50.5	50.5	-	50.5	-	50.5	-		-2.20
510	50.9	50.9	-		+.10		+.15		-1.50
520	51.5		+.05	51.5	-		+.10		-1.00
530	52.2	52.2	-		-.10		-.05		-.60
540	52.7	52.7	-	52.7	-	52.7		52.7	-
550	53.2		.00		.00		-.10		- .10
560	53.7	53.7	-	53.7	-		-.20		- .20
570	54.1	54.1	-		-.10		-.20		-.20
580	54.3		.00	54.3	-	54.3	-	54.3	-
590	54.5	54.5	-		.00		.00		- .06
600	54.7	54.7	-	54.7	-	54.7	-		- .13
610	54.7		.00		.00		-.08		- .01
620	54.7	54.7	-	54.7	-		+.15		+ .15
630	54.9	54.9	-		-.05		+.03		+ .09
640	55.0		+.10	55.0	-	55.0			+ .13
650	55.3	55.3	-		-.10		-.10		- .04
660	55.4	55.4	-	55.4	-	55.4	-	55.4	-

Table VII 2. Variations in Chromaticity Coordinates Due to
Interpolation of Reflectivity Values

Specimen	Series	dY	dx	dy	dY/Y
Pyrite	23-Point	Compared to the 23-Point values			
(Demirsoy, 1968)	16-Point	0.009	0.000	0.000	0.0002
	12-Point	0.009	0.000	0.000	0.0002
	8-Point	0.025	0.000	0.0001	0.0005
	4-Point	0.362	0.0006	0.0020	0.0068
Bornite					
(the Writer)	12-Point	Compared to the 12-Point values			
	8-Point	0.079	0.0005	0.0002	0.004
	4-Point	0.138	0.0004	0.0048	0.007
Tennantite					
(Tugal, 1969)	12-Point	Compared to the 12-Point values			
	8-Point	0.033	0.0003	0.0001	0.001
	4-Point	0.050	0.0009	0.0017	0.002
Hematite(w)					
(Demirsoy, 1968)	23-Point	Compared to the 23-Point values			
	12-Point	0.009	0.0001	0.0001	0.0003
	8-Point	0.018	0.0000	0.0000	0.0006
	4-Point	0.035	0.0002	0.0002	0.001
Vovellite(w)					
(Gehlen & Piller, 1964)	14-Point	Compared to the 14-Point values			
	12-Point	0.008	0.0005	0.0001	0.001
	8-Point	0.019	0.0014	0.0004	0.003
	4-Point	0.260	0.0128	0.0030	0.038
Cattierite					
(Demirsoy, 1968)	23-Point	Compared to the 23-Point values			
	16-Point	0.000	0.0000	0.0000	0.000
	12-Point	0.009	0.0001	0.0001	0.0003
	8-Point	0.004	0.0002	0.0000	0.001

VIII. MEASUREMENTS OF COLOUR CONSTANCY, BIREFLECTANCE AND
REFLECTION PLEOCHROISM

Colour constancy refers to the substantial invariance of object-colour perceptions in the presence of changes in illumination (The Science of Color by Committee on Colorimetry, 1968). Discussion of the phenomenon of colour constancy is as old as the problem of matching surface colours in different illuminants. Two samples, for instance, will match under one illuminant, but they may not match under another. In some cases they will mis-match because the alteration in appearance may be so slight that it would pass undetected if the second sample were not available for comparison.

Each real object has a so-called object colour which is merely its capacity to modify the colour of the light incident upon it. This capacity depends essentially upon the spectral reflectance of the surface and is a more or less unique and constant characteristic of the given object.

Changes in appearance accompanying changes in illuminant may be too weak to be perceived. For instance, coal usually continues to look black, and snow white, in a wide range of illuminants, owing to the effect of colour constancy. They are said to have high degrees of colour constancy.

Changes in colour sensitivity are less frequently noticed although they, too, are continually occurring and may be quite large. The tendency in maintaining the colour constancy of object colours is, on the other hand, due to visual adaptation.

The effect of visual adaptation in colour perception may be illustrated as follows. The colour of daylight and the colour of tungsten light are very different when viewed side by side, but many ordinary objects in a room seem little changed in appearance when they are seen during the day and at night.

Another example of adaptation effect is well known in photographic colour processing. Whites in the original may consist of white surfaces illuminated by daylight, artificial light, or some other illuminant, and hence have varying energy distributions. These are all seen as more or less white, owing to the adaptation of the eye to the illuminant, but of course the sensitivities of the photographic material are unchangable, and therefore will give different reproductions for whites under different illuminants.

Thus process of visual adaptation could largely compensate for changes in sensitivity level of the eye and the colour of the illumination, and hence favours colour constancy. Nevertheless, the compensation is rarely quite complete and a high degree of colour constancy is rare (Wright, 1969; Committee on Colorimetry, 1968). On the other hand, if the illumination changes are too sudden or too great, or if the colour of the object instead of the illuminant is seen, adaptation could not prevent obvious changes in object colour perception due to changes in illuminant.

Changing the illuminant may change the brightness or relative spectral distributions of the light from the objects, and therefore they show changes in appearance. The degree of colour constancy will be of significance to particular light sources.

A classic example of a poor colour constancy can be

demonstrated by alexandrite (Cr-bearing chrysoberyl) in the field of mineralogy, particularly in gemmology. Alexandrite is green under daylight and changes very distinctly to a red under tungsten light, because it highly absorbs the light in the yellow and blue regions, gives a main peak in the green and a secondary one in the red. Its spectral diffuse reflectance or transmittance curve may be somewhat like the one as shown in Figure VIII 1. Unfortunately, no quantitative data for alexandrite was obtained because of lack of specimens. Its colour values calculated from the above predicted, general curve were plotted on the chromaticity diagram in Figure VIII 2 to show the obvious changes of colour in passing from daylight to tungsten light.

It was, therefore expected that some minerals may have characteristics changes in appearance accompanying the change of illuminant and so some minerals may be easily distinguished from others, which are similar under ordinary observation condition.

According to the literatures of ore microscopy briartite and gallite are difficult to distinguish, galena and sulvanite are similar. They can be distinguished from similar counterparts by observing changes differently in appearance when inserting and taking out a daylight filter in front of the tungsten lamplight. As shown in Table VIII 1 briatite shows very little change in appearance whereas gallite changes in colour noticeably when changing the illuminant from A to C. In the same way the changes in contrast enhancement of galena and sulvanite could be detected. Galena shows noticeably increasing saturation and brightness (additive effect), i.e. its appearance changes towards luminous bluish colour from bright grayish colour; whereas sulvanite performs slight decreasing saturation and brightness (subtractive effect) from gray-yellowish to faint yellowish colour.

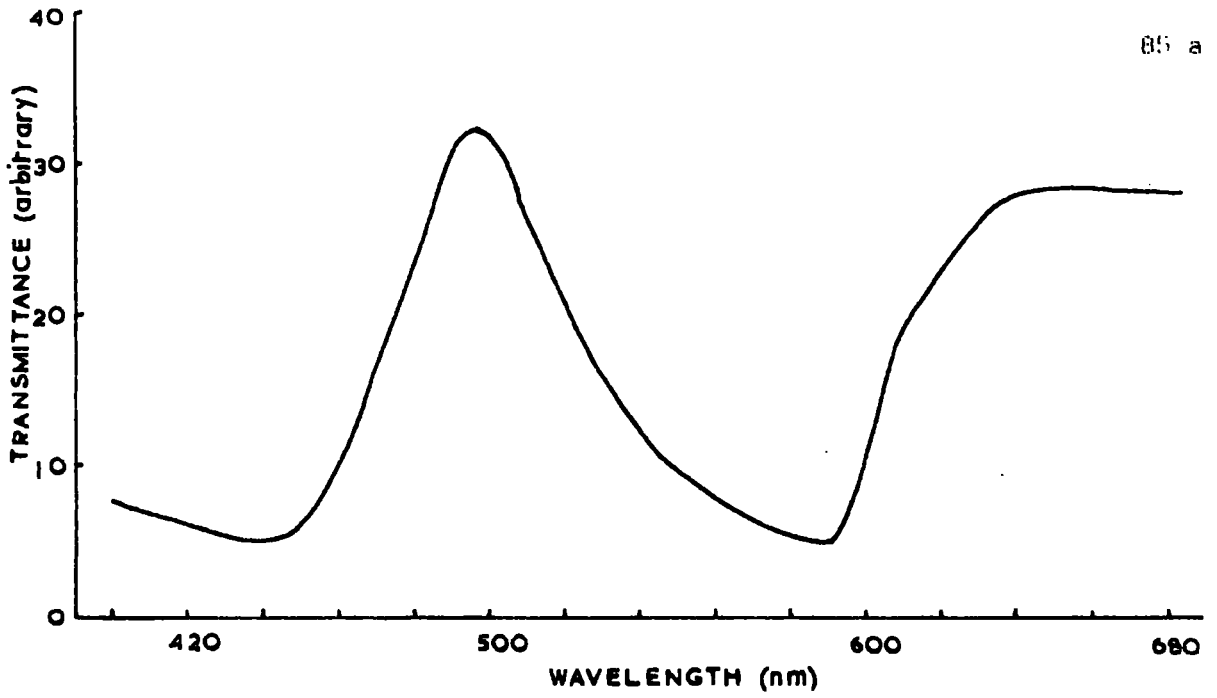


Figure VIII 1. A spectral distribution curve giving a green colour in daylight and changing to a red under tungsten light.

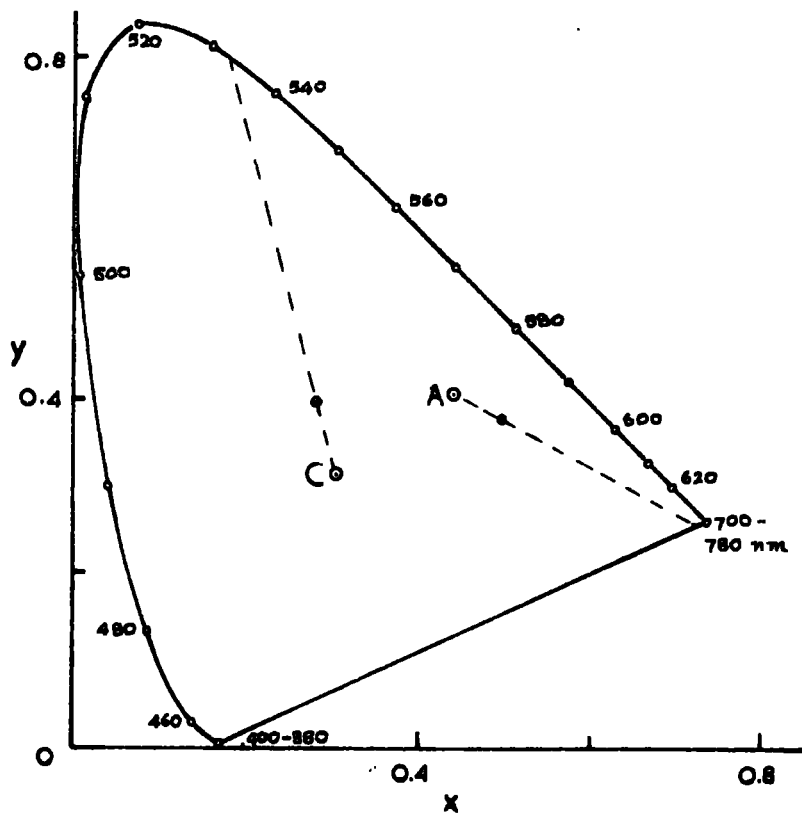


Figure VIII 2. Chromaticity diagram showing a poor colour constancy of a specimen of Figure VIII 1.

The basic quantitative colour values of some ore minerals for the standard light sources A (average tungsten light) and C (artificial average daylight) are tabulated in a later Chapter.

Isometric minerals have a single reflectivity in all directions in white light or in any monochromatic light and consequently they remain unchanged in chromaticness (hue and saturation) and visual brightness as the stage of the microscope is rotated under one light source. Many minerals of other crystal systems show distinct changes in appearance with rotation of the stage under ordinary observation conditions. Such changes are due to the variation of indices of reflection and absorption with different crystallographic orientations. The changes may be in brightness or chromaticness, or both.

The terms bireflectance and reflection pleochroism are here used in separate senses on the basis of quantitative values, although in the literature of ore microscopy these terms have been used with the same or similar meanings but without clear definitions. (particularly in the books published by Uytendogaardt and Burke, 1971 page VI; Ramdohr, 1969 page 297).

Bireflectance may be defined as the difference between the maximum and minimum reflectivities of principal crystallographic directions or as the difference of brightness Y values (= visual R), i.e. $dY = Y_{\max.} - Y_{\min.}$. Effective bireflectance to the visual sensation can be estimated with the threshold value of the brightness which is approximately $dY/Y = 1\%$.

Some sections of anisotropic minerals may show changes in brightness alone without changes noticeably in chromaticness, as the

stage of the microscope is turned . In other words the change in appearance of a polished section with rotation of the stage is due to the bireflectance, not due to changes in colour.

Reflection pleochroism may be defined as differences in chromaticity values (dx, dy) or as differences in dominant wavelengths (λ_d) and excitation purities (dPe) between the principal directions of a section. Effective reflection pleochroism can be estimated with MacAdam's threshold values.

Thus on the basis of quantitative colour values we can say that changes in appearance of anisotropic minerals with rotation of the stage are either due to the effect of bireflectance or due to reflection pleochroism, or due to both effects.

Pleochroism can then be defined quantitatively as differences between respective tristimulus values dX, dY, dZ for the principal directions of a section or expressed differently as the combined effects of bireflectance and reflection pleochroism.

For example, changes in colour of some ore minerals, as the microscope stage is rotated under one illuminant, are shown in Table VIII 2. Such quantitative values of bireflectance and reflection pleochroism may also help to direct the concentration of an observer in a particular case or cases.

Figure VIII 3 shows the changes of dominant wavelengths and excitation purities of some ore minerals in passing from tungsten light (source A) to artificial daylight (source C). In the Figure dots represent positions of the specimens under source A and the shifts are

indicated by arrow-heads when the illuminant is changed to source C .

From this Figure the following interesting points are noted.

- (1) Most of the ore minerals have hues in two major wavelength regions - i.e. yellow-orange part and purple-blue part.
- (2) The shifts of chromaticness, accompanying changes in illuminant, appear in a rather regular fashion. These regular shifts, in general, indicate that the colours of ore minerals having chromaticnesses in yellow-orange region become fainter when a daylight filter is interposed in front of the tungsten light, whereas the colours of other minerals having chromaticnesses in the purple-blue become stronger.
- (3) As consequences of the effect three significant characteristics in some cases were observed.

Firstly, in comparison, if two faint coloured minerals occur side by side there may be effective contrast enhancement with changes in illuminant.

For example, galena and briartite will show effective contrasts because the degree of colour constancy of galena is comparatively poorer than that of briartite. Similarly as shown in Table VIII 1, briartite and gallite which are difficult to distinguish from each other can be differentiated and recognised by observing contrasting degrees of colour constancy.

Secondly, some anisotropic minerals will show more effective pleochroism (bireflectance and reflection

pleochroism) under one illuminant or under another. For example, the reflection pleochroism of hexagonal pyrrhotite is stronger under illuminant A than under illuminant C. The reflection pleochroism of klockmannite is distinctly stronger under illuminant C than under illuminant A. These examples are illustrated in Figure VIII 4.

Thirdly, the colours of many minerals change correspondingly in passing from illuminant A to C, but their shifts are of the same magnitude, thus the contrast between the colours of many minerals remain the same.

The same effects were observed when attempts were made to obtain significant changes in colour under new light sources. These sources were combinations of source A and a neutral filter with 80 % transmission, and source A and a neutral filter with 20 % transmission characteristics.

The procedures for measurement of a new light source are described below.

First and foremost measurements of spectral transmittances of each neutral filter (F_λ) were made in the visible spectrum. The relative energy distributions of the C.I.E. source A ($H_\lambda d\lambda$) were multiplied by the 1931 C.I.E. colour-matching functions (\bar{x}_λ), similarly by (\bar{y}_λ) and (\bar{z}_λ) for wavelengths 380 nm to 780 nm with an equal interval of 5 nm.

The sums $\sum \bar{x}_\lambda H_\lambda d\lambda$, $\sum \bar{y}_\lambda H_\lambda d\lambda$, and $\sum \bar{z}_\lambda H_\lambda d\lambda$ were

multiplied by a normalizing factor K , where $K = 100.0 / \int \bar{y}_\lambda H_\lambda d\lambda$. Then the spectral transmittances F_λ were multiplied by corresponding products $\bar{x}_\lambda H_\lambda d\lambda$, etc. The normalizing factor K for the new light source was computed by

$$K = 100.0 / \int F_\lambda \bar{y}_\lambda H_\lambda d\lambda .$$

The tristimulus values of the new light source were

$$X_F = K \int F_\lambda \bar{x}_\lambda H_\lambda d\lambda ,$$

$$Y_F = K \int F_\lambda \bar{y}_\lambda H_\lambda d\lambda = 100.0 \quad \text{and}$$

$$Z_F = K \int F_\lambda \bar{z}_\lambda H_\lambda d\lambda .$$

The relative energy distributions or normalized energy distributions of the required light source were obtained thus

$$\text{for } X : \frac{\int (F_\lambda H_\lambda \bar{x}_\lambda d\lambda) X_F}{\int F_\lambda H_\lambda \bar{x}_\lambda}$$

$$\text{for } Y : \frac{\int (F_\lambda H_\lambda \bar{y}_\lambda d\lambda) Y_F}{\int F_\lambda H_\lambda \bar{y}_\lambda}$$

$$\text{for } Z : \frac{\int (F_\lambda H_\lambda \bar{z}_\lambda d\lambda) Z_F}{\int F_\lambda H_\lambda \bar{z}_\lambda}$$

These relative energy distributions were used in the calculation according to the C.I.E. method for the measurement of an object colour with respect to this new light source. The chromaticity coordinates of

the source were then calculated by the formulae given previously.

$$x = X_F / (X_F + Y_F + Z_F)$$

$$y = Y_F / (X_F + Y_F + Z_F) .$$

APL/1 computer program written by the author for IBM/360 to compute relative spectral distributions, tristimulus values and chromaticity coordinates of any light source whose spectral characteristics are known is given in Appendix VIII 1.

Colour measurements of a few ore minerals were made for the new light source and results were compared to those obtained previously for the standard light sources A and C. With changes in illuminant A to the new light sources or C to the new light sources all bases of comparison shifted correspondingly, but nothing looked significantly different compared to the shifts due to changes in illuminant A to C.

Therefore all quantitative colour values of some ore minerals are presented in Tables for the standard light sources A and C.

Table VIII 1.

Example: Degrees of colour constancy in passing from tungsten light (source A) to daylight (source C).

Mineral	$d\lambda$	$dPe\%$	dY	$dY/Y \%$
Briartite	5.8	-0.33	+0.09	0.3
Gallite	10.7	+2.03	+0.18	0.8
Galena	11.15	+2.92	+0.30	0.7
Sulvanite	11.44	-3.75	-0.36	-1.1

Table VIII 2.

Examples of Quantitative Values of Bireflectance and Reflection Pleochroism
under average tungsten light (source A).

Anisotropic mineral	Bireflectance dY	Reflection Pleochroism		Effective Pleochroism	
		dPe%	dλ	dY/Y	ds
Hematite (w - e)	3.7	0.5	1.1	12.7	1.0
Luzonite(max. - min.)	3.3	5.5	12.4	12.4	1.4
Mawsonite(max. - min.)	0.6	11.1	1.9	2.25	10.9
Chalcopyrite(max. - min.)	0.1	0.9	0.3	- 0.3	-0.9
Chalcophanite(max. - min.)	16.7	3.1	1.3	63.5	7.4
Niccolite(max. - min.)	2.2	2.9	0.1	4.2	7.7
Hodrushite(max. - min.)	1.2	0.0	4.0	3.6	2.1

Note: Pleochroism of hematite from (w) to (e) is mainly due to bireflectance. That is, the brightness difference between (w) and (e) directions is 12.7 times greater than just noticeable difference. Reflection pleochroism is just perceptible since ds = 1.

The brightness difference (dY) of chalcopyrite between maximum and minimum directions is below noticeable level. The saturation difference (dPe%) is just below noticeable difference. Therefore, its pleochroism (dY/Y & ds) may not be perceptible.

FIGURE VII 3 SHOWING CHANGES IN HUE (λ_d) AND SATURATION (P_e) OF ORE MINERALS IN PASSING FROM TUNGSTEN LIGHT TO DAYLIGHT

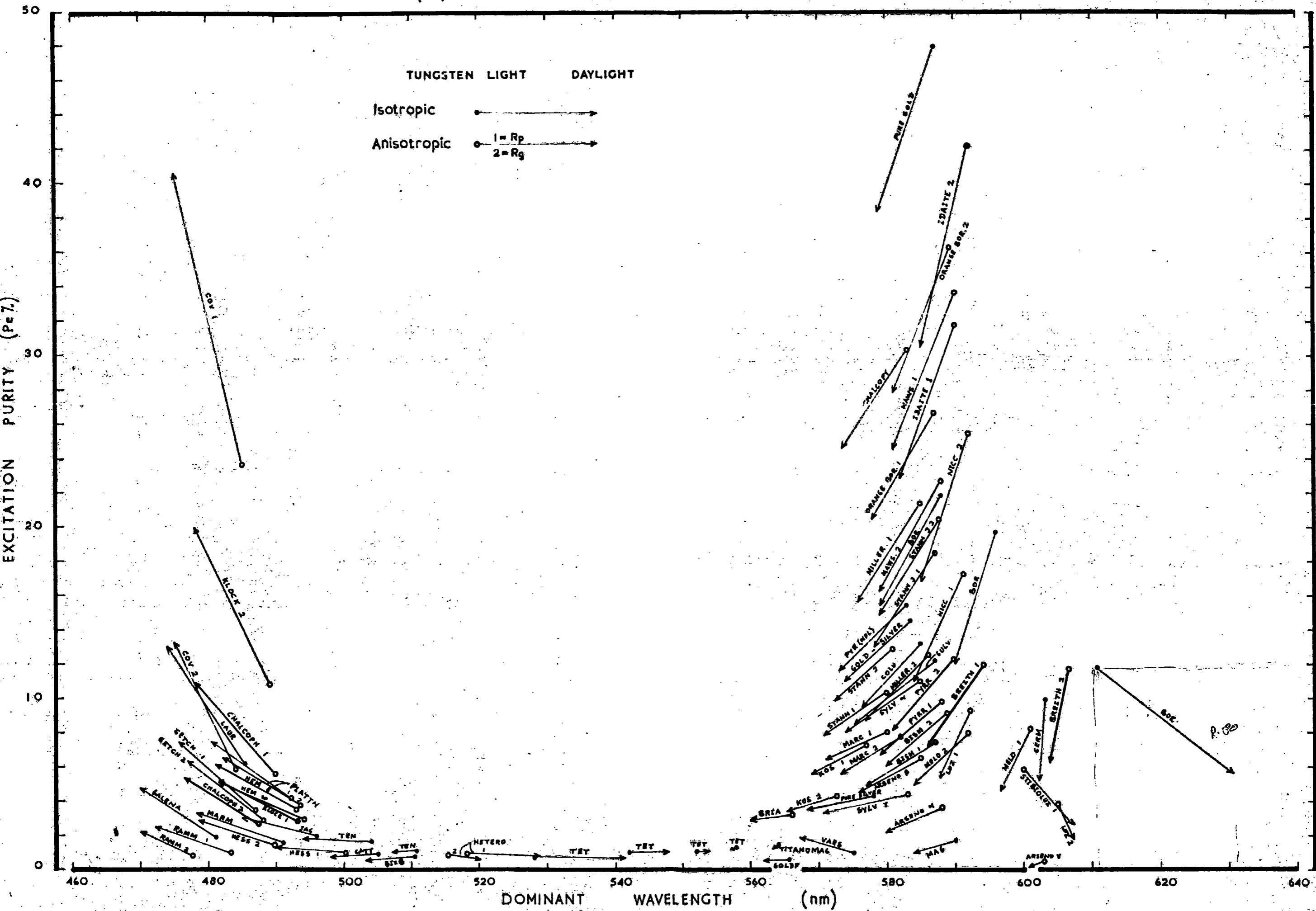
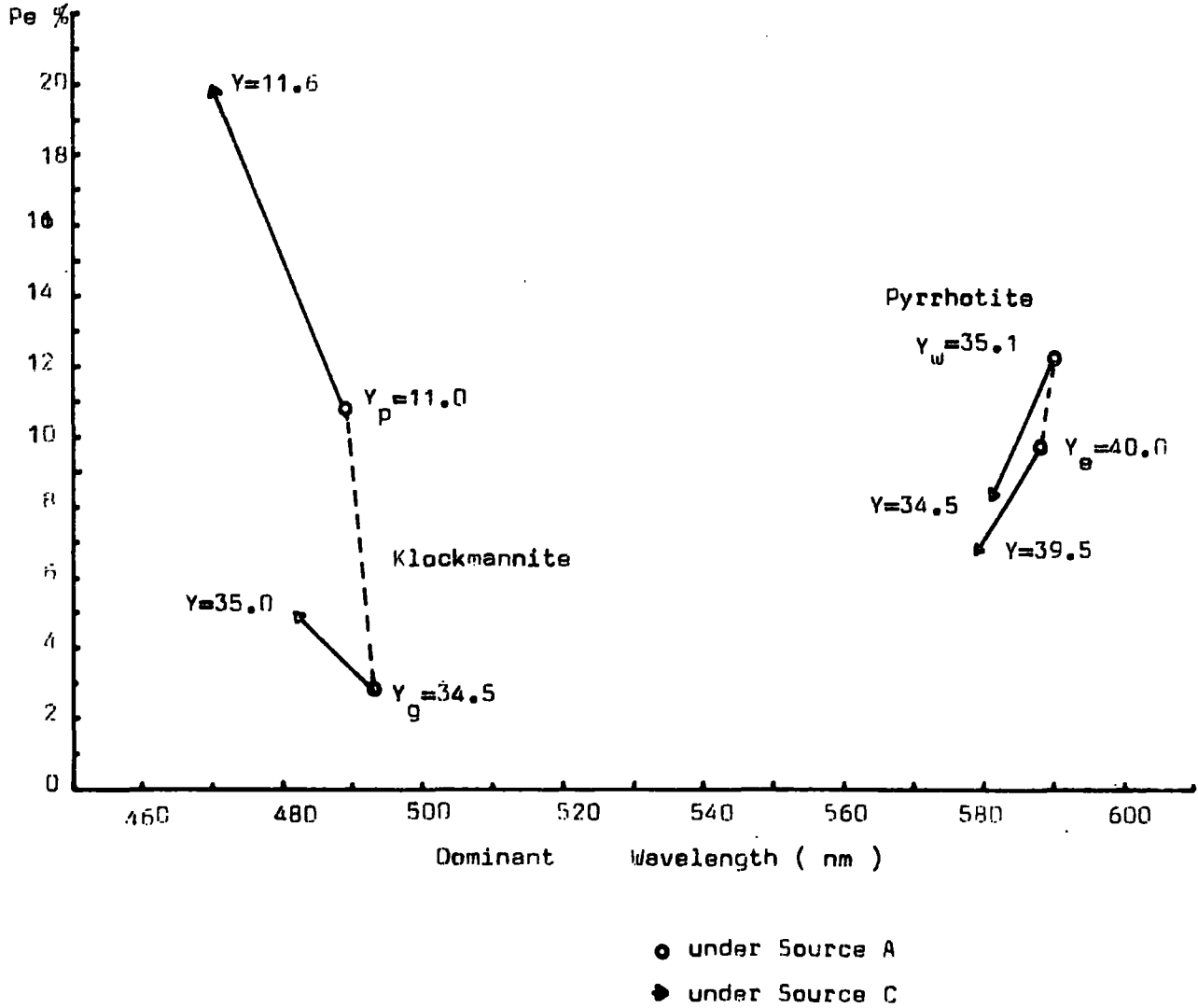


Figure VII 4. Example of different effects of Reflection Pleochroism
 in passing from Tungsten light (source A) to Daylight (source C)



IX. A STUDY ON SOME COLOUR PROBLEMS IN ORE MINERALS

The major aim of this present work lies in the development of quantitative colour measurement methods for minerals and the establishment of numerical colour values, spectral reflectivity data and graphical representation of the colours of some ore minerals.

On the basis of the quantitative values, attempts were made to understand the nature of the characteristic colour, colour variation, colour discrimination, colour constancy, reflection pleochroism and bireflectance of a mineral. Explanations of these properties have been given in the previous chapters.

Attempts were also made to understand the relationships between variations in colour quality and chemical composition and/ or physical properties. Unfortunately, in the majority of cases, insufficient chemical and other physical data are available for measured specimens, and therefore no promising correlation could be made, but general relations were observed.

The results of the analyses, described in the following pages should be consulted with the appropriate Figures and Tables which are compiled in Chapter X .

PURE GOLD, Au

PURE SILVER, Ag

GOLD - SILVER ALLOY, 50% Au + 50% Ag

The chromaticity coordinates of pure gold, pure silver and synthetic gold-silver alloy were plotted on the x-y chromaticity chart. A gentle curve passing through these points indicates the increase of excitation purity (saturation of hue) with increasing gold content in gold-silver alloys.

The chromaticity points of two natural sylvanite (AuAgTe_4) specimens from different localities are also shown in order to compare the positions of relative chromaticities with variable gold-silver content. The change in chromaticness of each sylvanite is due to two principal orientations (α and γ), and is almost parallel to the purity increasing direction of the gold-silver alloy.

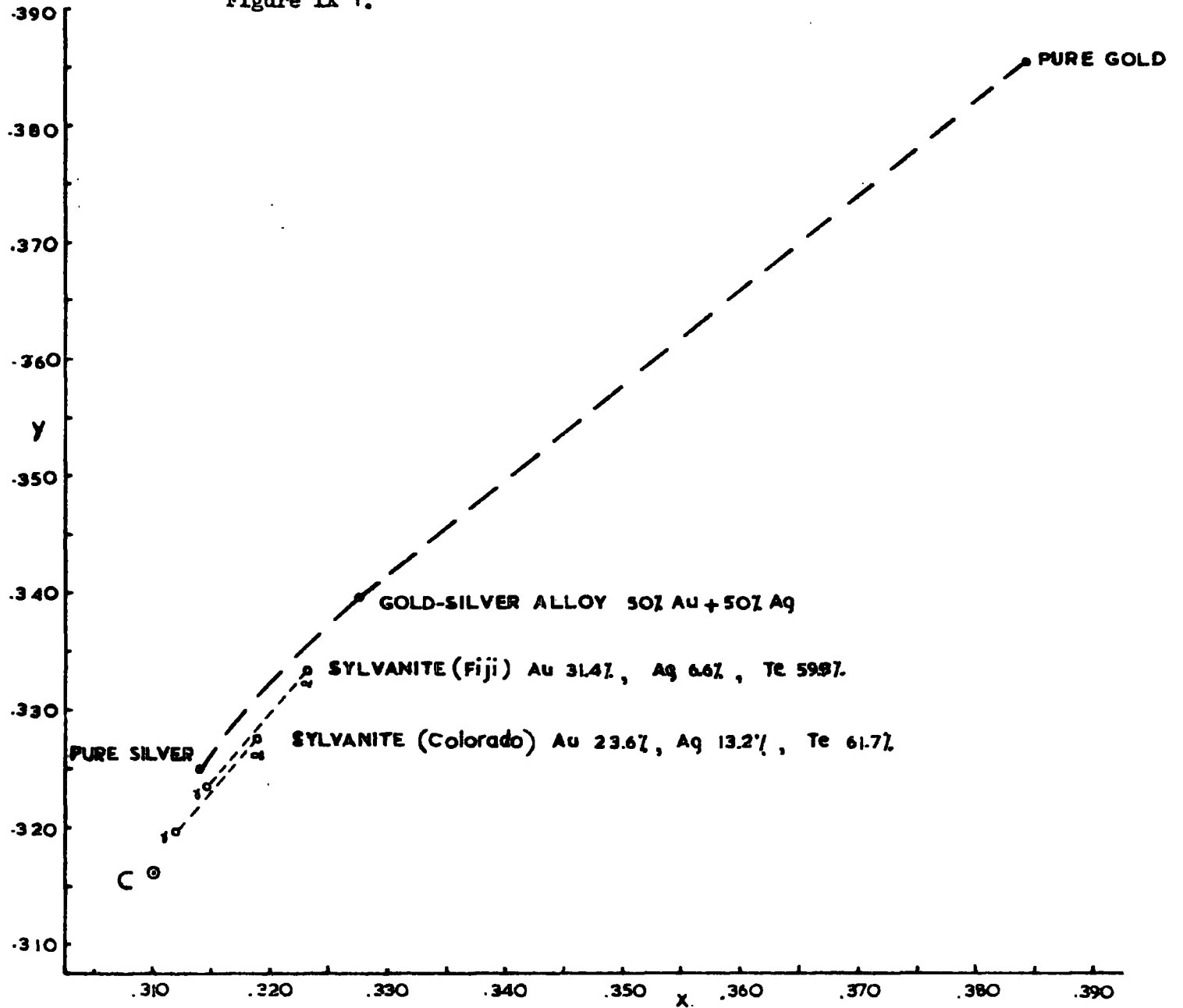
It is evident from the chromaticity chart that the higher purity of the sylvanite specimen from Fiji (the maximum purity point) suggest a higher gold content or conversely lower silver content than the specimen from Colorado, as quoted from their electron microprobe (Stumpfl, 1970) analyses on the chart.

These quantitative values of colour agree well with Eales (1961), who noted qualitatively that the colour of gold in polished sections is an index of its fineness. According to Eales nearly pure gold from some Southern Rhodesian mines has a golden colour with a distinct ruddy tint (approximately 950 fine) and with increasing proportion of alloys silver this colour changes through yellow to pale silvery yellow

(approximately 600 fine). Eales defined only four shades of colour qualitatively, to which all gold grains encountered could be referred.

It be of interest to determine colour coordinates of several gold-silver alloys having different finenesses because the above data suggest that their colour points will lie on the suggested gentle curve, showing an increase of purities with increasing gold contents and that all gold grains encountered could be compared with known standards.

Figure IX 1.



GERMANITE, $\text{Cu}_6 \text{Fe Ge S}_8$

Quantitative colour values of germanite specimens, as plotted on a chromaticity chart, are in good agreement with Levy (1966) who made measurements of spectral reflectivity on those specimens.

The chromaticity chart shows that the two specimens 1 and 2 from Bancairoun have the same colour as they are chemically homogeneous (Levy, 1966). The chromaticity points 3,4,5 and 6 scattering outside the discrimination ellipse represent the varying shades of colour of four different areas on a heterogeneous specimen from Tsumeb, as the differences are visually noticeable (Levy, 1966).

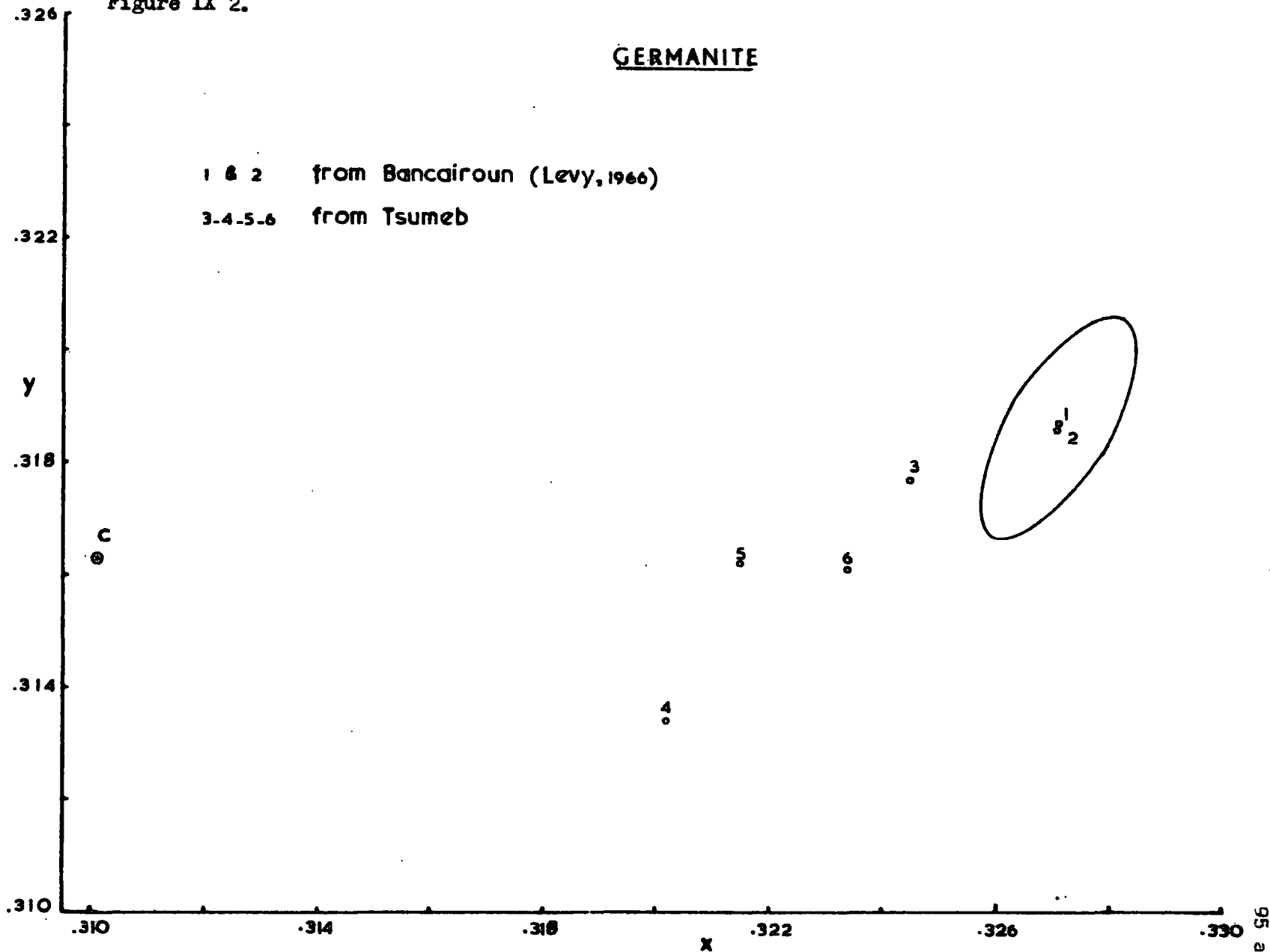
According to R. Phillips (personal communication), Springer (1969) and Levy (1966) the probable reason for the heterogeneous character of Tsumeb germanite is mainly due to the different contents of tungsten (tungsten-germanite). The colour of homogeneous, probably normal germanite is similar to that of bornite but slightly paler and higher in reflectivity (see in bornite).

Figure IX 2.

GERMANITE

1 & 2 from Bancairoun (Levy, 1966)

3-4-5-6 from Tsumeb



BORNITE, $\text{Cu}_5 \text{FeS}_4$

Freshly polished surfaces of some bornite specimens show noticeable colour differences. It was of interest to investigate the colour variation in bornites.

Spectral reflectivity measurements for colour evaluation were made on freshly polished surfaces of six specimens from five different localities, as described previously in Chapter III. In order to know the tarnishing effect with time, repeated reflectivity measurements were made. The values of the second measurement immediately after the first run, within about 40 minutes, showed no significant variation in reflectivity and colour values, but during the third run it was noted that the specimen surface tarnished quickly and consequently gave different values. Therefore the reflectivity values obtained on the first run were believed to be accurate for bornite.

Electron microprobe analyses of these specimens were made by Dr. Peckett (Geology Department, University of Durham) with a 'Cambridge Geoscan' microanalyser. The precision of the measurements was estimated at about $\pm 2\%$ of the measured value.

All analyses show that the measured specimens are nearly pure as all trace element contents are below the limits of detection. The contents of Ti, Cr, Co, Ni are approximately less than 180 ppm, As is less than 450 ppm and Sb is less than 350 ppm.

Figure IX 3, showing the colour distributions of measured

bornite specimens (also other bornite specimens from Turkey; Tugal, 1969), and the Table below indicate that there are slight variations in colour with slight differences in composition, but also shows that there is no linear correlation between colour and composition. The chemical formula of each specimen is given on the basis of 4 sulfur atoms per molecule.

Figure IX 3.

BORNITE

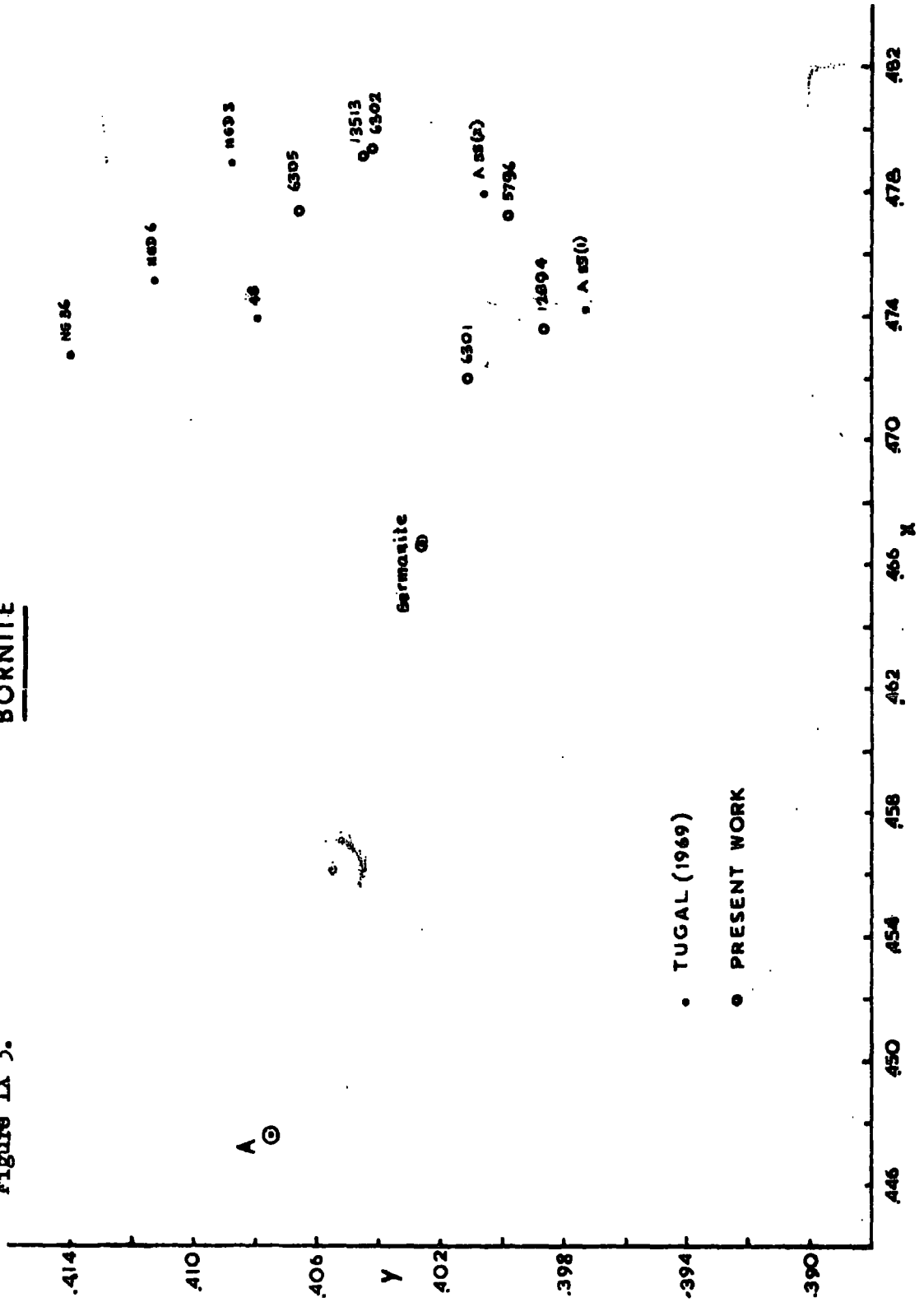


Table IX 1.

Electron Microprobe Analyses and Colour Values of Bornites

Specimen No.	6305	13513	6320	5796	12894	6301	
Locality	Korea	Australia	Arizona	Rhodesia	Rhodesia	New Jersey	
Cu	63.41	63.47	63.44	63.41	63.39	63.01	Wt. %
Fe	11.41	11.48	11.57	11.50	11.41	11.60	
S	25.49	24.82	24.88	25.70	24.88	25.50	
Sum	<u>100.31</u>	<u>99.77</u>	<u>99.89</u>	<u>100.61</u>	<u>99.68</u>	<u>100.11</u>	
Formula	$\text{Cu}_{5.03}\text{Fe}_{1.02}\text{S}_4$	$\text{Cu}_{5.17}\text{Fe}_{1.06}\text{S}_4$	$\text{Cu}_{5.14}\text{Fe}_{1.06}\text{S}_4$	$\text{Cu}_{4.98}\text{Fe}_{1.02}\text{S}_4$	$\text{Cu}_{5.14}\text{Fe}_{1.05}\text{S}_4$	$\text{Cu}_{4.98}\text{Fe}_{1.04}\text{S}_4$	
Y (brightness)	22.05	21.30	21.83	20.14	19.78	20.32	
d (nm)	593.8	595.8	596.1	603.7	610.6	603.8	
Pe %	19.74	19.74	19.69	15.16	11.82	12.46	

The colour values are with respect to the standard illuminant A.

PYRITE , Fe S₂

The colour distributions of 15 pyrite specimens are shown together with a discrimination ellipse on the chromaticity chart. All colour points are scattered within about 3 nanometers of the dominant wavelength, but the variation in purity (saturation) as a whole is surprisingly large, i.e. their colour qualities may be perceptibly whitest yellow to yellowish white.

Comparing the colour distributions with the threshold ellipse, the colour quality, for instance, of the R.S.M. pyrite is noticeably different from that of the A.E.D. pyrite.

Spectral reflectivities of the (100) face of pyrite (Demirsoy, 1968) are lower, by between 0.3 and 0.6 units, than those of the (111) face, but chromaticity coordinates are identical. Their relative brightness (= visual reflectivity) difference ($dY/Y\%$) is 1% which is just noticeable under an ore microscope.

Singh (1965) demonstrated the variations in spectral reflectivity of a group of pyrite specimens from different localities and the same effect is also described by Gray and Millman (1961) and others. According to Singh the specimens 1877, 1879 and 1883 have no significant variation in composition and contain no trace element greater than 0.1%. In the figure they all will lie within a discrimination ellipse centred on the specimen 1883.

The anisotropic pyrite NG 18 of Tugal (1969) has a slightly larger cell size (about 0.001 to 0.002 Å) than his other pyrite

specimens. It seems that this difference is not significant in colour quality as compared with his other pyrites.

Pyrite specimens have been employed as standards in comparative reflectometry but it is doubtful on the above evidence whether pyrite is a reliable standard for spectral reflectivity measurements.

It is not included amongst the standards recommended by the International Mineralogical Association Commission on Ore Microscopy (I.M.A.C.O.M.).

If the spectral reflectivity data given by different authors are assumed to be reliable with small, conventional errors, then appreciable chromaticity variations could be largely a function of chemistry.

Since pyrite may have slight substitutional and non-stoichiometric variation in composition, appreciable changes in optical properties of pyrites from different types of deposits may be due to the presence of a minor element or elements (Co, Ni, Cu, As, Zn), perhaps : greater than 0.1 %. Unfortunately, chemical data on measured specimens is not available in most cases.

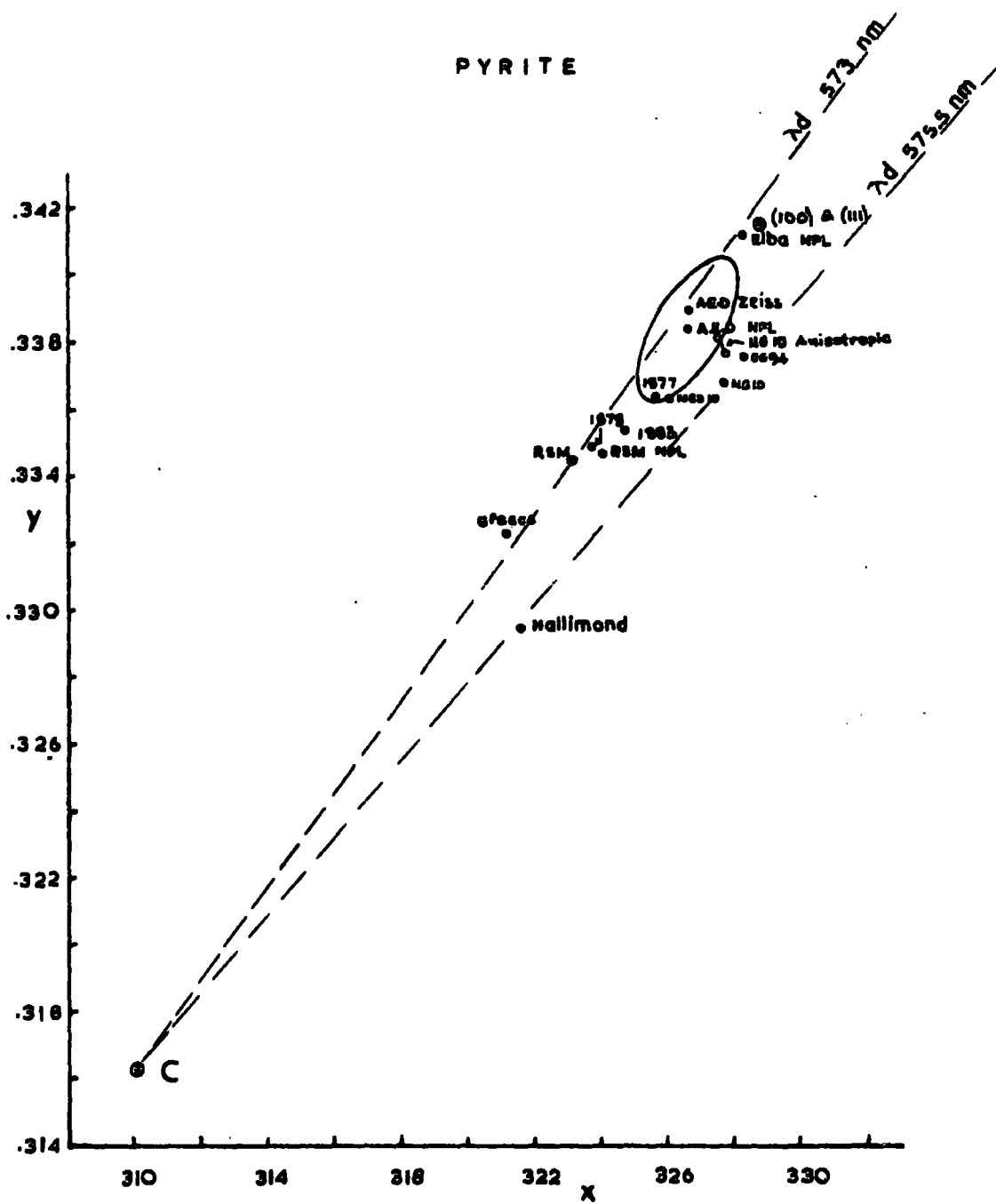


Figure IX 4.

CHALCOPYRITE , Cu Fe S₂

The chromaticity distribution of 15 chalcopyrite specimens from three mineral deposits of Turkey (Tugal, 1969) are presented in the x-y colour diagram of figure IX 5. Although the variation in chromaticness is not very large compared to the size of the discrimination ellipse , the brightness (visual reflectivity) differences from one mineral to another could be quite appreciably large, i.e. 47.8 - 43.6; 4.2 units.

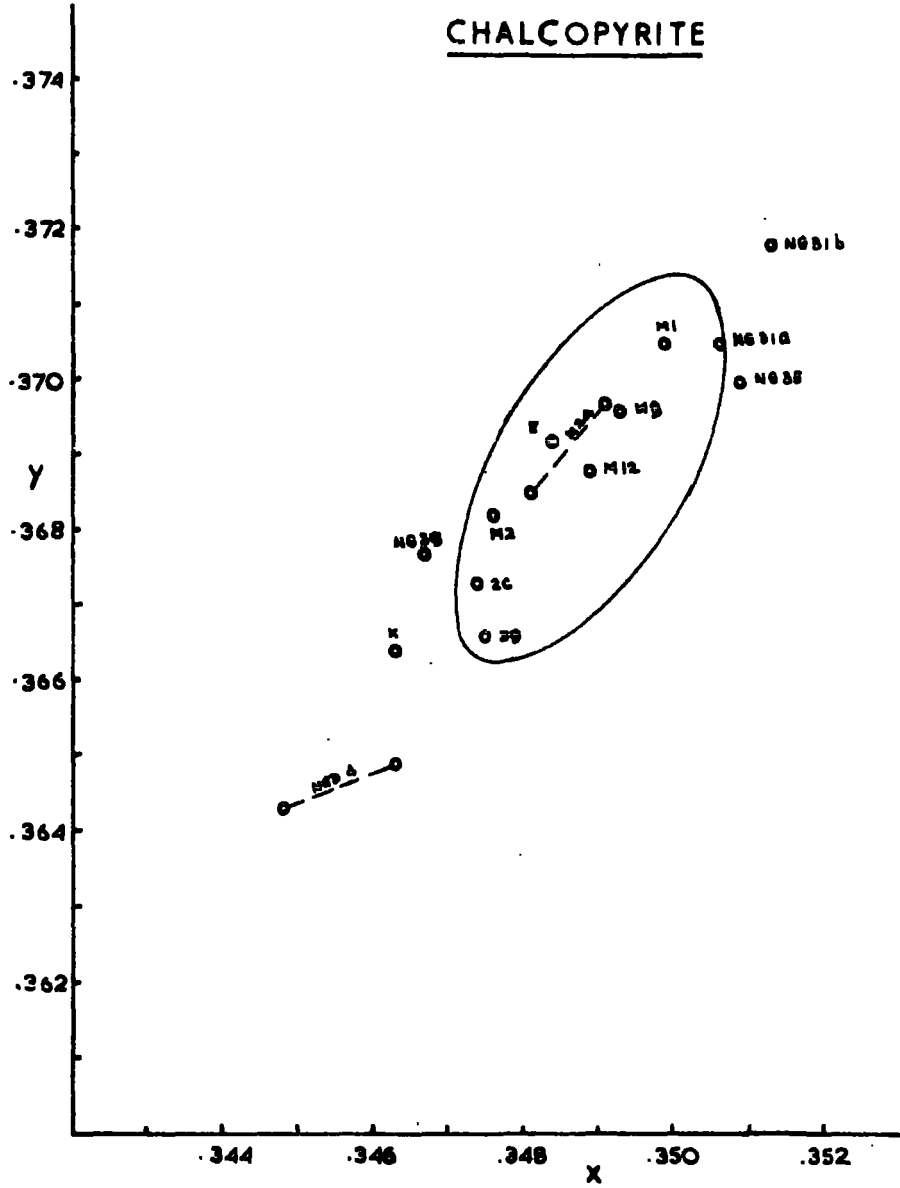


Figure IX 5.

TETRAHEDRITE, $\text{Cu}_3 \text{Sb S}_{3.25}$

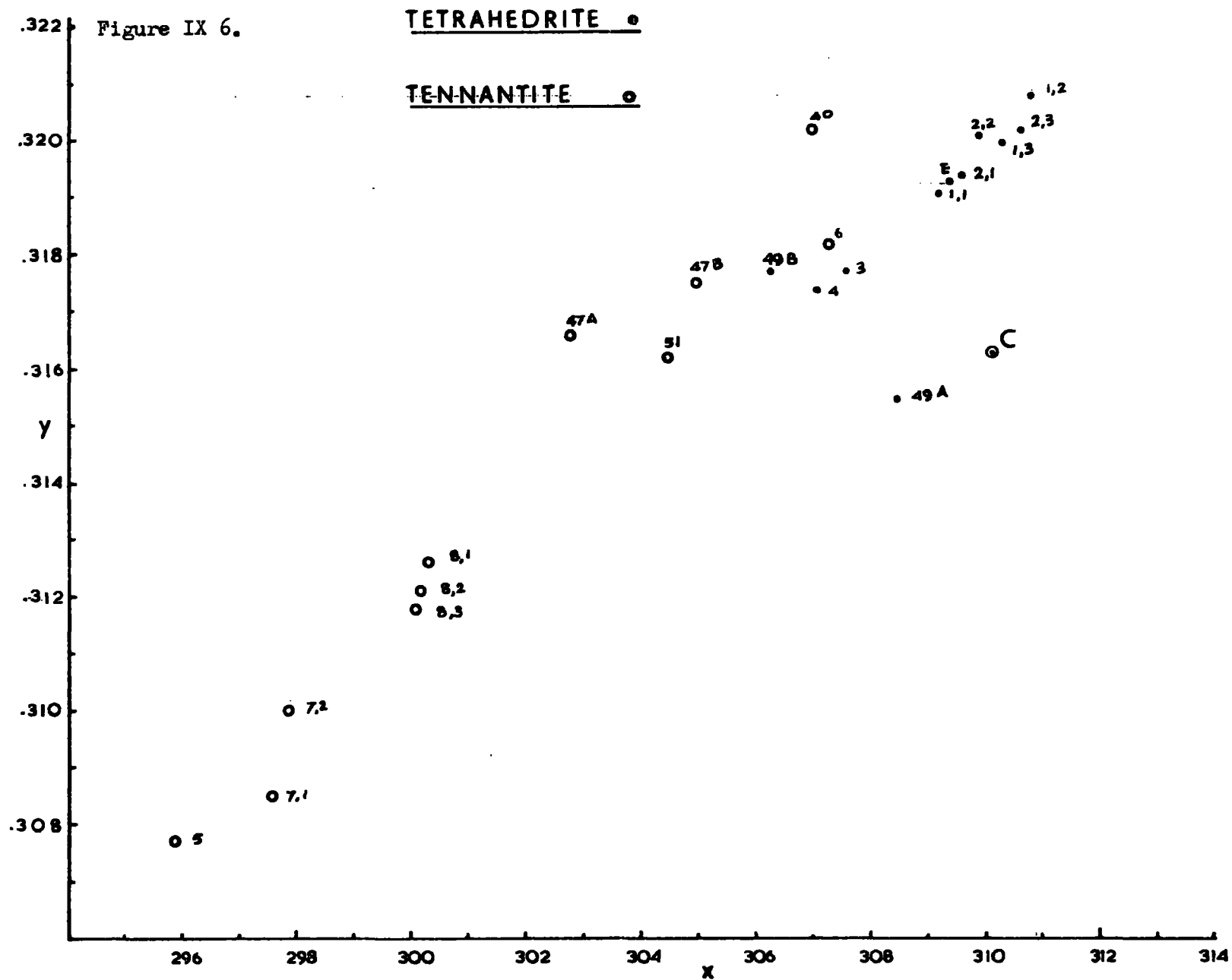
TENNANTITE , $\text{Cu}_3 \text{As S}_{3.25}$

Quantitative colour values show that tetrahedrite specimens have slight colour variation from greenish-yellowish grey to bluish-greenish grey and that tennantite specimens have comparatively larger variation in colour from greenish grey to bluish grey with higher saturation (purity).

Since the range of solid solution in the tetrahedrite-tennantite series is very large, their reflectivities generally vary correspondingly from 32.7 to 29.6 % in tetrahedrite and 31.0 to 26.5 % in tennantite.

The chromaticity coordinates of some members of synthesised tetrahedrite-tennantite series (Hall, 1971) are not directly comparable with those of natural tetrahedrite and tennantite, probably because of large variable substitutions in stoichiometric composition and weak colour values.

Hall (1971) concluded in his doctoral Thesis that with increasing As:Sb ratio there is a general decrease in cubic cell edge, increase in microhardness, decrease in reflectivity and increase in saturation of colour (purity). Unfortunately, it was not possible to support his conclusion for natural tetrahedrite and tennantite, because no chemical and physical data on measured specimens is available in the literature.



SPHALERITE, Zn S

Many investigations on cause of coloration in various sphalerite specimens have been reported. Since pure Zn sphalerites can be colourless, the change in colour of sphalerites, from a translucent yellow through brown to nearly opaque black, is usually considered to be a function of iron content (Dana's System of Mineralogy, 1944; Deer Howie and Zussman, 1969).

Roedder and Dwornik (1968) concluded, in a study on colour banding of some low-iron sphalerites (Pine Point) with the electron microprobe, that there is no correlation between iron content and colour. Graeser (1969) also reported that the coloration of low-iron sphalerites (from Binnatal) in various shades of colour from yellow to black is not due to the increase of iron content, but due to a strong influence of the Mn content: the yellow colour changes to brown and black whenever the Mn content exceeds a value of about 100 ppm. No report to support his view has appeared so far.

The quantitative colour values of 16 sphalerite specimens from Yugoslavia (Grafenauer et al, 1969) do not agree with Graeser's report and have no correlation with chemistry. Although there is a linear relationship between reflectivity and FeS content in synthesised sphalerites (Akinci, 1970), their quantitative colour values again do not correlate with chemistry or cell size.

The reason for lack of correlation with chemistry in this present work might be that the reflection colour of sphalerite is

very much less sensitive to its chemistry or other physical properties than the transmission colour.

The same reason will apply for other non-opaque minerals.

Therefore, quantitative colour measurements derived from spectral transmission data of sphalerite may be more suited to investigation in the colour problem.

GALENA , Pb S

All quantitative colour values of 7 galena specimens represent greyish white colours with slight differences in purity. The specimen M1 of Tugal (1969), being an average one, is taken as a reference origin of the discrimination ellipse to compare the chromaticity point with those of other specimens.

Figure IX 8 shows that all galena specimens have no difference in colour perception. The specimens of Tugal have the relative brightness difference (dY/Y) not greater than the threshold level, but Demirsoy's galena (100) has 5% greater than the reference.

Under a daylight illumination galena will show a relatively brighter and slightly stronger saturation, as quantitative values are given in Table VIII 1.

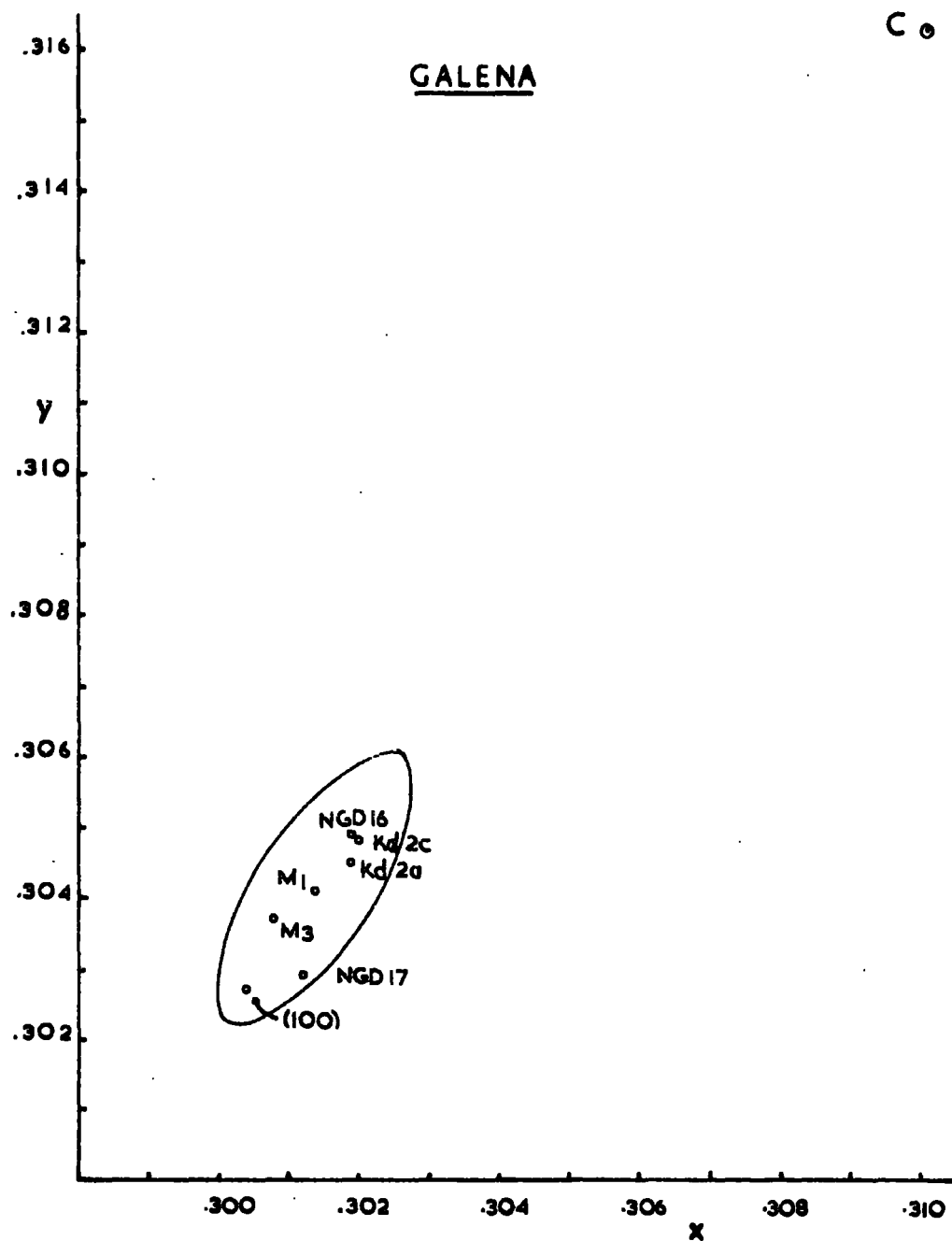


Figure IX B.

COLUSITE, $\text{Cu}_3 (\text{As}, \text{Sn}, \text{V}, \text{Fe}, \text{Sb}) \text{S}_4$

SULVANITE, $\text{Cu}_3 \text{V S}_4$

Colusite is very similar to sylvanite (rare) in reflectivity and colour quality as given in Tables and shown on the chromaticity chart of Figure IX 9. Sylvanite has a slightly higher reflectivity whereas colusite commonly shows different shades of zonal structure.

The different shades are not due to difference in chromaticness but due to differences in visual reflectivity (brightness). The chromaticity points of different zones lie within the discrimination ellipse, as shown in the diagram, but the colours of two zoned colusite specimens are different.

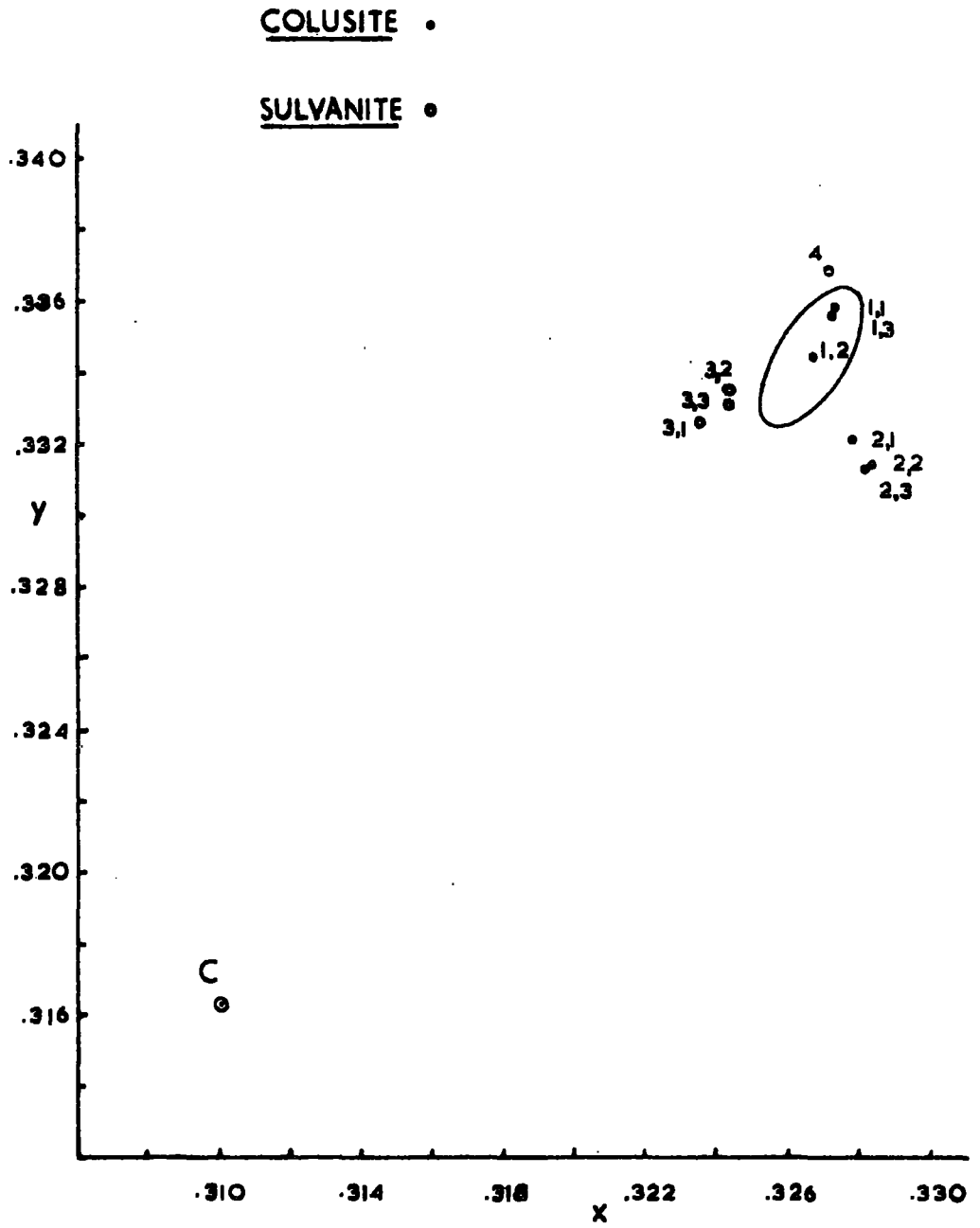


Figure IX 9.

STANNITE, $\text{Cu}_2 \text{Sn Fe S}_4$; Tetragonal

MAWSONITE, $\text{Cu}_{2+x} \text{Sn}_{1-x} \text{Fe S}_4$; Tetragonal

IDAITE , $\text{Cu}_3 \text{Fe S}_4$; probably Tetragonal

Several unnamed varieties of stannite have been reported (Ramdohr, 1944 & 1960). Ramdohr recognised four different stannites which he called 'Zinnkies I - IV'. Some of them have been named, but still noticeable deviations from ideal compositions of stannite and hexastannite (= stannite jaune of Levy, 1966; renamed as stannoidite by Kato, 1968) from different localities were found by electron microprobe analysis (Springer, 1968).

Mawsonite is considered as an intermediate member between stannite and idaite (Uytenbogaardt and Burke, 1971), but little is known about the relationships between stannite and idaite.

An investigation of the variations in colour and other optical properties of stannite, stannite jaune (hexastannite), mawsonite, orange bornite and idaite were made and observed as follows.

The chromaticity points, as shown in the x-y coordinate chart, show, in general, that excitation purity and combined effects of bireflectance and reflection pleochroism increase from stannite to idaite.

It seems that the colour saturation and reflection pleochroism is largely influenced by the Cu content. That is, as given in Table below, the above properties increase with increasing Cu content from stannite to idaite.

The electron microprobe analyses are average data as quoted below Table IX 2 which shows the general relationship between chemical composition and the optical properties.

For each member of the series the colour variation is also due to slight variation in chemistry and orientation of the measured sections. In Tables (Chapter X) and Figure IX 10 the spectral reflectivity data and respective colour values of 1-3-4 and 5-6-7 belong to the six different areas of the specimens no. 2 and no. 1 respectively (Levy, 1966). According to Levy (1966) the slight difference in reflectivity (as well as in colour) is perhaps due to the fact that the specimen no. 1 contains Zn whilst the specimen no. 2 contains no Zn.

It was seen from the Figure that the colour points of the so-called orange bornite lie inbetween mawsonite and idaite and its colour variation is large perhaps with large variation in composition. Orange bornite (13) from Peru has higher Cu and Fe contents than other orange bornites from Vaulry (Levy, 1966).

Similarly, the reason for large variations in reflectivity and colour of idaite are due to the different orientation and chemistry. Idaite (4-5-6) from Cerro Huemul, Argentina has higher Cu and Fe contents than those (1-2) from France.

Therefore, on above evidences the colour intensity in the series from stannite to idaite is strongly influenced by Cu content, i.e. the brownish yellow saturation increases with increasing Cu content. A slight variation in colour of a member of the series is probably due to the differences in content of Cu, Fe , perhaps Sn, and minor elements such as Zn, As.

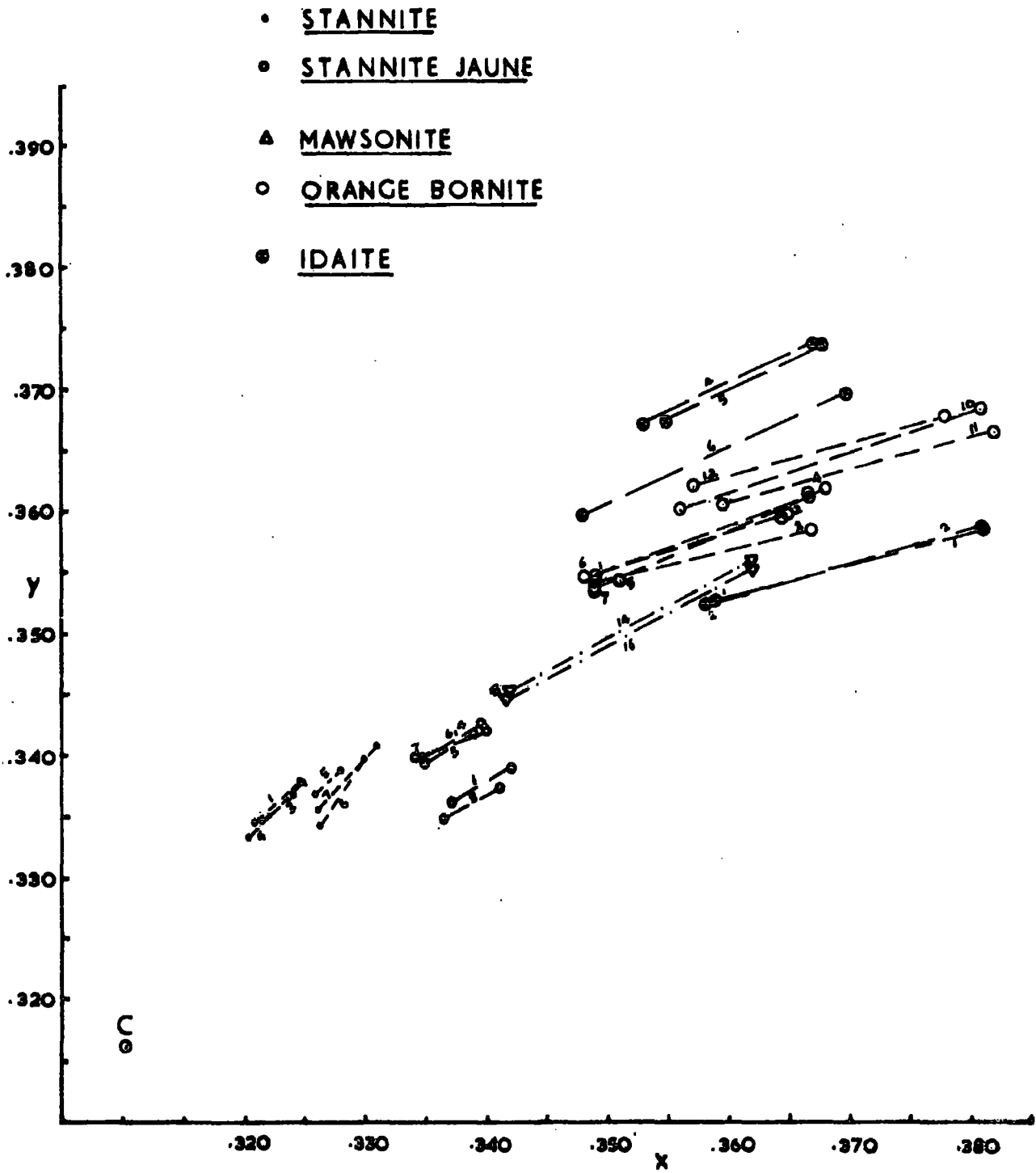


Figure IX 10.

Table IX 2. Electron Microprobe Analyses

	Stannite		Stannite Jaune		Hexastannite		Mawsonite		Orange Bornite		Idaite	
Cu	29.7	29.8	37.5	39.8	44.3	43.8	44.2	50.5				
Sn	27.1	26.7	20.1	16.4	12.5	12.6	15.1	-				
Fe	11.5	10.9	11.8	10.8	12.6	12.4	14.7	14.7				
Zn	1.7	3.2	2.7	1.7	-	-	-	-				
S	30.0	30.0	29.8	29.7	31.2	31.4	29.0	34.2				
	S	L	L	S	L	S	L	L	Ref.			
	(9)	(5)	(9)	(4)*	(4)	(5)	(7)	(7)	No. of Specimens			
<u>Optical Characters</u>												
	13.3-9.6		21.4 - 18.0		33.7-22.7	43.6-27.4	42.1-31.5	Pe % Max.-Min.				
	11.4		19.7		28.2	35.5	36.8	Mean Pe%				
	1.1		2.3		1.1	0.6	1.6	dY (Ymax.-Ymin.)				
	(+)		(+)		(-) from 420 to 540	(-) from 420 to 560	(-) from 420 to 580	Optic Sign				
					(+) from 560 to 640	(+) from 590 to 640	(+) from 600 to 640 nm					

S = Springer (1968)

L = Levy (9166)

* a specimen also contains Sb and As.

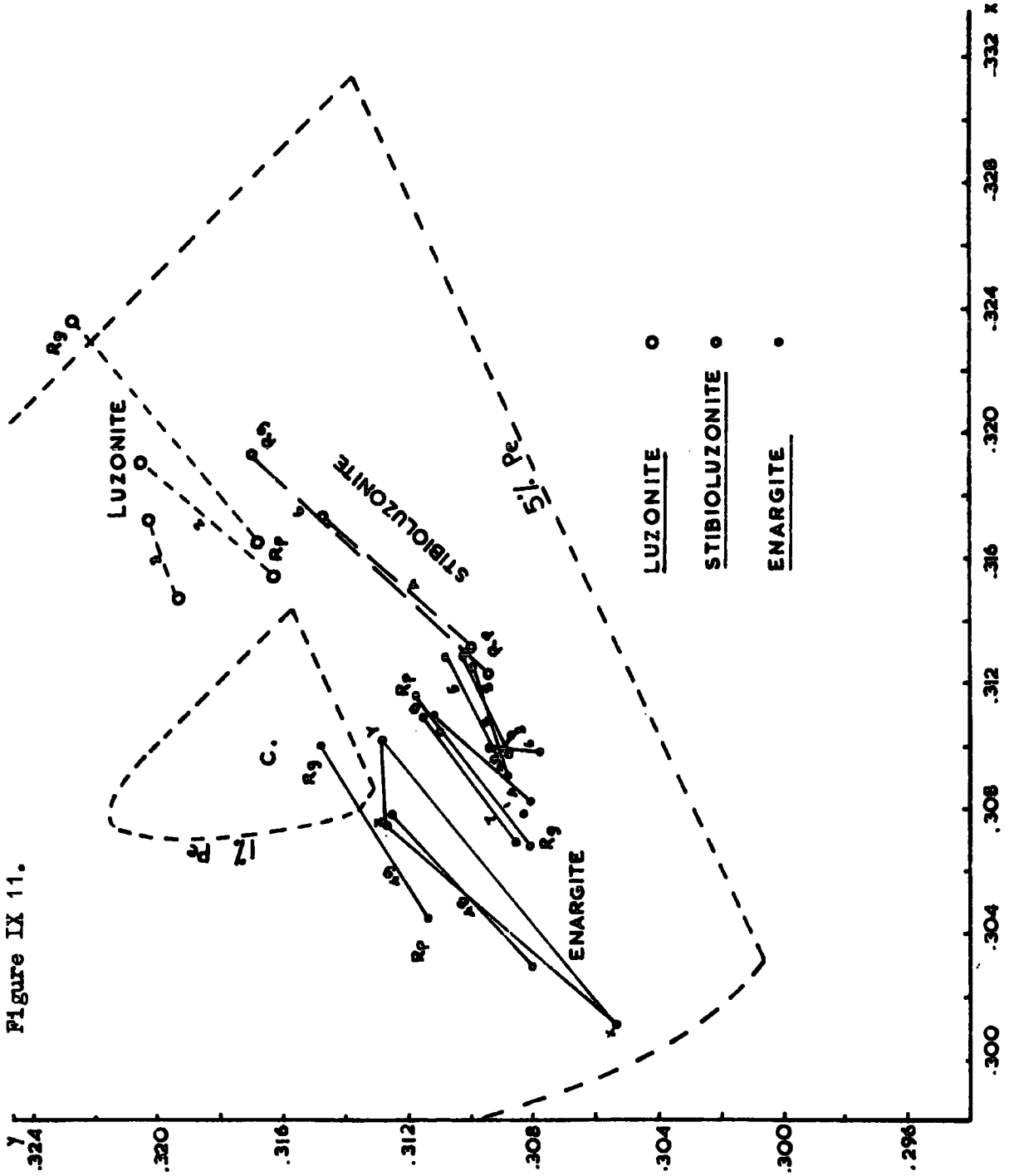
ENARGITE, $\text{Cu}_3 (\text{As,Sb}) \text{S}_4$; Orthorhombic

STIBIOLUZONITE-LUZONITE, $\text{Cu}_3 (\text{As,Sb}) \text{S}_4$; Tetragonal

The chromaticity variations of enargite and stibioluzonite are mainly due to the different orientations in the polished sections, whilst the measured luzonite specimens were very small and had many inclusions (Levy, 1966).

A portion of the chromaticity diagram shows the colour distributions of enargite, stibioluzonite and luzonite. In order to visualize the colour quality from part of the chromaticity diagram and their colour values, the qualitative colour descriptions are quoted below.

Enargite:	Pinkish grey to violet grey, bireflectance and reflection pleochroism distinct.
Stibioluzonite:	Pinkish without orange tints, bireflectance and reflection pleochroism distinct to moderate.
Luzonite:	Light pinkish orange, bireflectance and reflection pleochroism distinct to strong.



THE EFFECT OF GRAIN SIZE ON CHANGES OF HEMATITE COLOURS

A massive hematite specimen from the Lake District (England) and single hematite crystals from Sweden were ground down separately by means of a Tema disk mill to investigate the change of colour when the grain size is reduced.

Each powdered specimen was passed through successive nylon sieves. All separated samples were then pelletised (dry) with boric acid powder backings by means of hydraulic Ram Press.

Under the reflected light photometer microscope all pellets revealed brown to red earthy grains and fairly, evenly distributed shining tiny specular hematites. Three separate, smooth areas which will represent the overall reflection of each pellet were chosen and spectral reflectivity measurements were made on each area. The average results of each specimen were employed in quantitative colour measurement.

Some pellets having large grain sizes were reground down into sufficient fine powders for x-ray fluorescent analysis, because it was considered that the composition of the samples may change due to separation into different grain sizes.

X-ray fluorescent analysis was carried out on a Philips PW 1212 automatic sequential analyser for the following oxides. Si O_2 , $\text{Al}_2 \text{O}_3$ and $\text{Fe}_2 \text{O}_3$ line intensities of the elements were measured on 10 standards which had previously been analysed by wet chemical methods. Calibration curves for the elements were then obtained and from these the compositions of the unknown samples were determined by interpolation.

The analyses are given in the Table IX 3^{and} show that the compositions of the different samples of the massive hematite were not significantly different within the experimental error. The composition of the different samples of hematite crystals was therefore assumed to be the same as the original composition. Hence the change in colour of a specimen was mainly due to the change in grain size.

Figure IX 12 shows the effect of grain size on changes of colours of the massive hematite and hematite crystals. When the grain size decreases saturation (purity) of hematite colours distinctly increases together with slight change in hue (dominant wavelength). All sample points are also shown with the discrimination ellipse on an enlarged part of the chromaticity chart of Figure IX 13.

The changes of colour of the massive hematite were large at first and then decreased when the grain size was finer and finer. The dark red colour of the powders of hematite crystals changed little compared to the changes of the massive hematite powders.

For a hematite powdered specimen the variation in colour was due to the change of grain size and also the amount and nature of tiny specular hematites which were present in the powdered specimens.

Therefore the change of colour is probably a function of different light scattering (diffraction and reflection) from tiny spherical grains and crystalline hematites, together with the absorption of light incident upon the powdered specimens.

Table IX 3. X-Ray Fluorescent Analysis of Hematite

Specimen	Mesh of Nylon sieve	SiO ₂	Al ₂ O ₃	Fe ₂ O ₃	Total (Wt.%)
Crystals	191	0.73	0.43	98.10	99.26
Massive	25 52	0.80	0.31	95.77	96.88
	52 72	0.72	0.31	96.50	97.53
	72 100	0.72	0.33	95.53	96.57
	100 150	0.78	0.32	96.42	97.52
	150 200	0.79	0.33	96.04	97.16

Figure IX 12. Chromaticity diagram showing the colour sequence of hematite powders under the standard illuminants A and C. Saturation increases with decreasing grain size. Arrow indicates the starting chromaticity points.

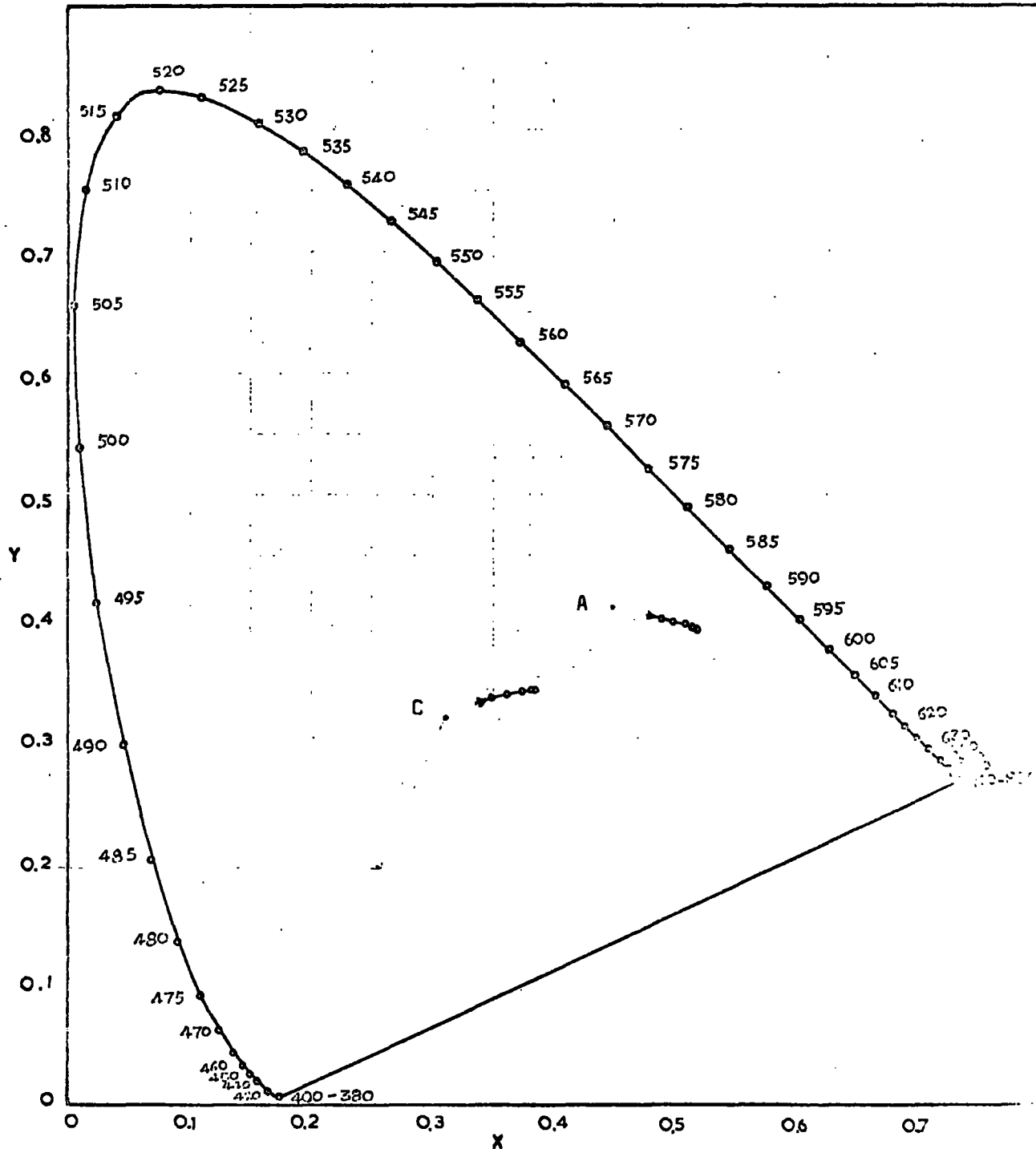
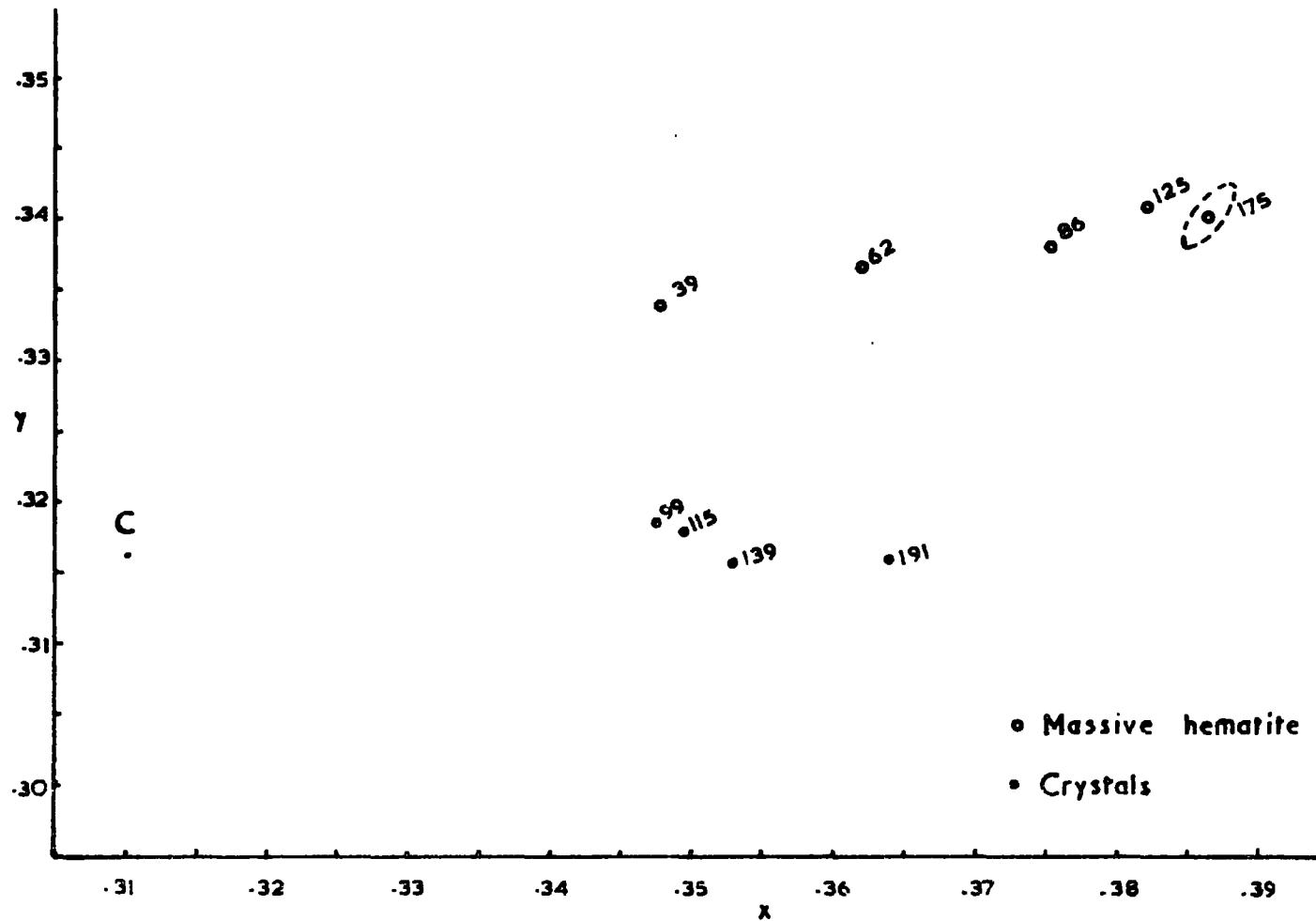


Figure IX 13. Enlarged part of Figure IX.12 to show the noticeably changes of the colours compared with a discrimination ellipse at its actual position.

The figures against the points are the average meshes of nylon sieves.



XII. CONCLUSIONS AND SUGGESTIONS OF POSSIBLE FUTURE WORK

Quantitative colour measurements have been widely applied, in many ways (e.g. the use of the visual quality of a colour for its own sake; the use of the colour of an object as an indicator of other properties which it may possess; etc.), in various laboratories and industries such as colour photography, colour television, signal glasses and road signs, agriculture and food, meteorology and astronomy, chemical tests, analysis of optical and visual phenomena, industries of automobile, dyeing, paint, pulp and paper, etc.

This project was undertaken to investigate the contribution to be made by concepts and methods of quantitative colour measurement to mineralogy and gemmology.

All necessary computer programmes were prepared for IBM/ 360 and used in this work to reduce personal errors and to save time, since numerous calculations are involved in the measurement of colour. (Chapter VI).

An investigation of the use of reflected light methods of determining spectral reflectivity values of faceted gemstones was made for quantitative colour measurement. (Chapter III 3). The same method was applied to the measurement of colour in ore minerals, since it promised to be a useful additional method for the identification of these minerals.

Many ore minerals, especially oxides, sulpho-salts and

sulphides have similar reflectivity in white light and some have overlapping values of reflectivity and microhardness. It has been realized that spectral reflectivity measurements, can be made use of in many cases for mineral identification.

Although reflectivity is the most important property of opaque minerals, it is still difficult to compare the measured values with available reflectivity data of a mineral for determination. As shown by curves in Figure XII 1 in the Appendix, spectral reflectivity values of many polished specimens of a mineral vary over the visible spectrum. Such variations may be within the experimental errors and/or due to slight variations in chemical and physical properties.

Unfortunately, such slight variation in properties is below detectable limits in many cases.

Therefore, it is difficult to interpret the significances of considerable variation, at a wavelength or at some wavelengths, of spectral reflectivities.

However, from such spectral reflectivity measurements, little work is required to obtain additional quantitative values, colour values, which may be more useful for mineral determination and which may help in explanation of the significant variations of spectral reflectivities of certain minerals (Chapter VIII and IX, Figures 1a and 1b).

The enlarged diagrams of Figures 1a and 1b may be used as an aid for mineral identification on the basis of quantitative colour values.

From the chromaticity diagram of Figure 1a, in general, the measured chromaticity point of a mineral can be compared with those of other minerals. All similar minerals may be eliminated by reading off the tristimulus values Y , which is the relative brightness (or visual reflectivity) at an appropriate dominant (or complementary) wavelength from Figure 1b.

The mineral identification thereafter can be made with the aid of other quantitative properties such as degrees of colour constancy, pleochroism (bireflectance and reflection pleochroism), etc. (Chapter VIII) and of other qualitative properties such as form, texture, mineral associations, etc.

Thus spectral reflectivity data and colour values together may form an additional system of mineral identification. Therefore the compilation of systematic data of spectral reflectivity and colour of ore minerals took the large volume of the present work.

Further more, a chromaticity diagram representing colour qualities of ore minerals will be useful especially in teaching ore mineralogy to students. The colour distributions of ore minerals can be seen in a such a diagram.

It is of interest to look at the chromaticity diagram of Figure 1a, because the colours of most ore minerals distribute across the diagram (i.e. within 40% constant purity curve) from the yellow-orange region through the illuminant point to the blue-purple region. The cause of this somewhat systematic and characteristic colour distribution poses a new problem.

Finally, from this work quantitative colour values may contribute to a useful means of representing colour characteristics of ore minerals in a colour diagram, comparing and distinguishing between similar coloured minerals, and to an understanding of relations between colour quality and other properties such as compositional variations, bonding and structure of minerals.

SUGGESTIONS

More information for other ore minerals is required to complete the chromaticity diagram.

An interference colour of a mineral can also be shown in a colour diagram if its spectral energy distributions are known and such measurement may reward further study.

It was considered that the transmission methods of measurement of faceted gemstones for colour measurement requires further development of accuracy and precision and eventually, standardisation of method. Further investigation could make use of internal transmitted light method, since such a method has already been used to qualitatively determine the absorption spectrum of gemstones (Chapter IV).

Further work will be worthwhile to investigate the measurement of fluorescent colour of particular minerals. The fluorescent colour quality of a mineral is very difficult to describe and yet is used as an important property for identification of some gemstones and minerals.

Finally, a suggestion is made to investigate the improvement of reflected light methods of observing polished ore minerals by using certain types of illumination under which effective colour constancy and/or different colour contrast of similar minerals may be easily observed.

X. QUANTITATIVE COLOUR VALUES AND SPECTRAL REFLECTIVITYDATA OF SOME ORE MINERALS

The quantitative colour values and spectral reflectivity data of about 340 specimens of 130 different ore minerals are compiled in the table forms.

In compiling these data minerals are divided into two sections, isotropic and anisotropic, according to the crystal system. The visual reflectivity (tristimulus value Y) has been chosen as the chief property in the sub-division, in order of increasing value. The colour values derived from 4-wavelength reflectivity data are given at the bottom of each sub-division in the same order.

All these quantitative data can also be used together with qualitative descriptions given in the literature of ore microscopy for mineral identification.

1. ISOTROPIC MINERALS

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y				Y		
PURE SILVER	A	0.4498	0.4115	577.93	4.31	99.25		436	89.70
AG	C	0.2142	0.2252	567.64	3.49	99.20		450	93.70
CUBIC								500	99.20
REF. FOR R: (SQUAID, 1965)								550	99.90 — X J. 1
REMARKS: LOC.								590	99.10
								590	99.10
								650	98.60

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMFL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
GCLD-SILVER	A	0.4603	0.4157	583.50	14.49	89.81	436	68.40
50%AL+50%AG	C	0.3275	0.3397	573.57	10.95	88.77	450	72.50
CUBIC							500	86.00
REF. FOR F: (SQUAID, 1965)							550	88.30
REMARKS: LCC.							590	92.10
							650	92.20

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y	WAVELENGTH		Y			
PURE GOLD	A	0.5004	0.4240	587.05	48.06	60.22		436	25.40
AU	C	0.3841	0.3852	578.60	38.33	56.66		450	26.90
CUBIC								500	37.40
REF. FOR R: (SQUAID, 1965)								550	59.00
REMARKS: LCC.								590	64.50
								650	74.20

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
PYRITE	A	0.4595	0.4157	583.14	13.95	54.01	440	42.30
FE S2	C	0.3267	0.3390	573.35	10.55	53.42	460	45.50
CUBIC							480	48.60
REF. FOR R: (SINGH, 1964)							500	51.00
REMARKS: LOC. FLBA. A.E.D. STD., ZEISS CALIBRATED							540	53.40
							580	54.70
							620	55.10
							660	55.80

48.1

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y				Y		
PYRITE	A	0.4603	0.4149	584.03	13.94	53.94		430	42.10
FE S2	C	0.3269	0.3376	574.35	10.22	53.28		470	46.20
CUBIC								490	49.50
REF. FOR R: NICHOL & PHILLIPS (1965)								520	52.00
REMARKS: LCC. RIC MARINA, ELBA								550	53.70
								580	54.40
								610	55.20
								650	56.20

47.5

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES X Y	DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
PYRITE	A	0.4571 0.4115	565.69	9.36	53.79	470	49.10
FE S2	C	0.3217 0.3295	575.82	6.66	53.27	490	50.90
CUBIC						520	52.60
REF. FOR R: FALLIMCND (1957)						550	52.70
REMARKS: LCC. RIC MARINA, ELPA						580	54.30
						610	55.10
						650	56.40
						670	55.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
PYRITE	A	0.4606	0.4167	532.98	15.41	53.86		400	27.10
FE S2	C	0.3283	0.3412	573.34	11.57	53.22		420	38.80
CUBIC								440	41.20
REF. FOR R: (COMMUNICATION WITH P.R. SIMPSON)								460	44.30
REMARKS: LCC. ELBA. N.F.L. CALIBRATED (ECWIE)								480	47.60
								500	50.30
								520	52.00
								540	53.20
								560	54.00
								580	54.70
								600	54.80
								620	55.00
								640	55.30
								660	55.70
								680	56.00
								700	56.00

45.9

A.7

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
PYRITE	A	0.4597	0.4154	583.44	13.87	53.54		400	37.80
FE S ₂	C	0.3267	0.3384	573.70	10.38	52.54		420	39.60
(UBIC								440	42.10
REF. FOR P: (SINGH, 1964)								460	45.10
REMARKS: LOC. ELBA. A.E.C. STD., N.P.L. CALIBRATED								480	48.10
								500	50.40
								546	53.20
								589	54.40
								620	54.70
								660	55.60
								700	55.80

46.6

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS		LAMBDA (NM)	REFLECTIVITY R
		COORDINATES X	Y			(VISUAL R)	Y		
PYRITE, NGD10	A	0.4594	0.4147	583.79	13.18	52.95		440	42.63
FE S2	C	0.3262	0.3363	574.62	9.69	52.32		460	45.34
CUBIC								480	47.68
REF. FOR R: (TUGAL, 1969)								500	48.94
REMARKS: LCC. TURKEY								520	50.42
								540	52.03
								560	53.33
								580	54.32
								600	53.47
								620	54.89
								640	53.47
								660	54.13

46.5.

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
PYRITE 1879	A	0.4576	0.4143	583.21	11.65	52.84		400	38.80
FE S2	C	0.3238	0.3349	573.44	8.67	52.36		420	40.70
CLBIC								440	43.20
REF. FOR R: (SINGH, 1964)								460	45.90
REMARKS: LCC. COLOMBIA, S. AMERICA								480	48.60
								500	50.20
								546	52.60
								589	53.60
								620	53.70
								660	54.30
								700	54.40

47.2

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
PYRITE	A	0.4571	0.4141	583.09	11.17	52.72		400	39.10
FE S2	C	0.3232	0.3345	573.14	8.40	52.28		420	40.90
CLBIC								440	43.40
REF. FOR R: (SINGH, 1964)								460	45.80
REMARKS: LCC. UNCERTAIN. R.S.M. STD. MEASURED AGAINST A.E.D. STD. (NPL)								480	48.80
								500	50.60
								546	52.50
								589	53.20
								620	53.70
								660	54.30
								700	54.30

47.3

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
PYRITE 1883 (100)	A	0.4586	0.4141	583.88	12.21	52.65		400	38.20
FE S2	C	0.3248	0.3354	574.03	9.07	52.11		420	40.30
CUBIC								440	42.60
REF. FOR R: (SINGH, 1964)								460	45.40
REMARKS: LCC. GRAY RIVER, NEWFOUNDLAND								480	48.30
								500	49.50
								546	52.30
								589	53.20
								620	53.90
								660	55.10
								700	55.20

45.9

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
PYRITE	A	0.4557	0.4133	582.92	9.65	52.48		400	38.20
FE S2	C	0.3212	0.3323	572.66	7.27	52.12		420	40.20
CLBIC								440	44.40
REF. FOR R: (SINGH, 1964)								460	47.10
REMARKS: LCC. KASSANDRA MINE, GREECE								480	49.40
								500	50.80
								546	52.30
								589	52.80
								620	53.30
								660	53.90

2.2

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (DOMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
FYRITE, NC10	A	0.4608	0.4145	584.53	14.01	51.88		440	41.37
FE S2	C	0.3277	0.3368	575.49	10.22	51.15		460	44.08
CUBIC								480	46.33
REF. FOR R: (TUGAL, 1969)								500	47.28
REMARKS: LCC. TURKEY								520	48.79
								540	50.51
								560	52.45
								580	53.35
								600	52.80
								620	53.54
								640	52.78
								660	54.13

45.2

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
PYRITE 1877	A	0.4592	0.4145	583.85	12.90	51.68		400	37.20
FE S2	C	0.3257	0.3364	574.14	9.58	51.11		420	38.70
CUBIC								440	41.40
REF. FOR R: (SINGH, 1964)								460	44.50
REMARKS: LCC. CHARLOTTE MINE, NORWAY								480	46.60
								500	48.70
								546	51.20
								589	52.60
								620	52.90
								660	53.50
								700	53.90

455

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
PYRITE	A	0.4581	0.4138	583.88	11.66	51.62		436	41.60
FE S2	C	0.3241	0.3347	573.87	8.70	51.11		460	45.10
CUBIC								480	47.40
REF. FOR R: (SINGH, 1964)								500	49.10
REMARKS: LCC. UNCERTAIN. R.S.M. STD. N.P.L. CALIBRATED								520	50.30
								546	51.20
								580	51.80
								610	52.80
								650	53.40
								670	53.60

46-3

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (DOMPL.) WAVELENGTH	PLURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
PYRITE, CG94	A	0.4616	0.4144	534.90	14.50	51.39		440	40.46
FE S2	C	0.3283	0.3376	575.43	10.60	50.65		460	43.59
CUBIC								480	45.79
REF. FOR R: (TUGAL, 1969)								500	47.09
REMARKS: LCC. TURKEY								520	48.49
								540	50.18
								560	51.60
								580	52.46
								600	52.29
								620	53.54
								640	52.78
								660	54.13

1:0

50 6

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PLURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY (R) (R)	
		X	Y			Y	LAMBDA (NM)	(R)	(R)
PYRITE NG18	A	0.4608	0.4148	584.31	14.22	52.15	440	41.25	41.25
FE S2	A	0.4605	0.4152	583.90	14.29	52.32	460	44.02	44.11
CUBIC	C	0.3278	0.3377	574.59	10.49	51.45	480	46.44	46.68
	C	0.3276	0.3381	574.59	10.54	51.65	500	48.16	48.39
REF. FOR R: TUGAL (1969)							520	49.48	49.74
REMARKS: LOC. TURKEY; ANISOTROPIC PYRITE							540	50.56	51.22
							560	52.37	52.63
							580	53.21	53.50
							600	53.50	53.50
							620	53.53	53.53
							640	53.90	53.90
							660	53.84	53.84

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES X	DOMINANT (COMPL.) WAVELENGTH Y	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
COBALTITE(100)	A	0.4556	0.4071	5.22	50.42	440	48.03
(CO,FE)AS ₂ S	C	0.3178	0.3198	3.00	49.90	460	47.43
CUBIC						480	47.44
REF. FOR R: (THE AUTHOR)						500	47.87
REMARKS: LOC. UNCERTAIN. SINGLE CRYSTAL						520	48.45
						540	49.16
						560	49.89
						580	50.67
						600	51.52
						620	52.19
						640	53.13
						660	53.50

Electron Probe Analysis:	Wt. %
Co	33.20
Fe	1.90
Ni	1.87
As	45.23
S	18.80
	<hr/>
	101.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (Cmpl.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y				Y		
GALENA	A	0.4403	0.4015	478.70	2.08	43.70		440	49.80
PP S	C	0.3004	0.3027	469.71	5.24	44.01		460	48.40
CUBIC								480	47.20
REF. FOR R: DEMIRSCY (1968)								500	45.90
REMARKS: LCC. UNKNOWN. MEASURED ON CLEAVAGE FACE								520	44.70
								540	43.90
								560	43.50
								580	43.20
								600	43.10
								620	43.30
								640	43.40
								660	44.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
GALENA M3	A	0.4402	0.4024	481.93	2.03	41.96	440	47.45
PB S	C	0.3008	0.3037	470.78	4.95	42.26	460	46.46
CLBIC							480	44.70
REF. FOR R: (TUGAL, 1969)							500	43.68
REMARKS: LCC. TURKEY							520	42.71
							520	42.71
							540	42.22
							560	42.08
							580	41.62
							600	41.33
							620	41.62
							640	41.41
							660	40.77

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
GALENA,NGC16	A	0.4411	0.4029	481.51	1.79	41.88	440	46.82
PB S	C	0.3019	0.3049	469.99	4.41	42.14	460	45.84
CUBIC							480	44.38
REF. FOR R: (TUGAL,1969)							500	43.24
REMARKS: LCC. TURKEY							520	42.41
							520	42.41
							540	42.02
							560	41.97
							580	41.80
							600	41.41
							620	41.57
							640	41.07
							660	40.77

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
GALENA, KCZA	A	0.4413	0.4026	475.84	1.77	41.80	440	46.93
PB S	C	0.3019	0.3045	468.86	4.48	42.04	460	45.45
CUBIC							480	44.36
REF. FOR R: (TUGAL, 1969)							500	43.22
REMARKS: LOC. TURKEY							520	42.49
							520	42.49
							540	41.91
							560	41.81
							580	41.38
							600	41.47
							620	41.62
							640	41.41
							660	40.77

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	WAVELENGTH (NM)	REFLECTIVITY R
		X	Y					
GALENA, MI	A	0.4408	0.4025	480.86	1.89	41.62	440	46.93
PBS	C	0.3014	0.3041	469.71	4.70	41.88	460	45.61
CUBIC							480	44.26
REF. FOR R: (TUGAL, 1969)							500	43.16
REMARKS: LOC. TURKEY							520	42.30
							520	42.30
							540	41.75
							560	41.76
							580	41.22
							600	41.01
							620	41.34
							640	41.41
							660	40.77

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
GALENA,NGD17	A	0.4409	0.4018	477.92	1.92	41.57	440	47.37
PE S	C	0.3012	0.3029	467.17	4.97	41.81	460	45.62
CUBIC							480	44.33
REF. FOR R: (TUGAL,1969)							500	43.11
REMARKS: LCC. TURKEY							520	41.86
							520	41.86
							540	41.72
							560	41.55
							580	41.31
							600	41.15
							620	41.46
							640	41.07
							660	40.77

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
GALENA, KC2C	A	0.4413	0.4028	480.61	1.76	41.55	440	46.50
PB S	C	0.3020	0.3048	469.31	4.40	41.79	460	45.22
CUBIC							480	44.06
REF. FOR R: (TUGAL, 1969)							500	43.01
REMARKS: LOC. TURKEY							520	42.15
							520	42.15
							540	41.66
							560	41.54
							580	41.38
							600	41.05
							620	41.34
							640	40.84
							660	40.77

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES X	Y	DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
KUTINAITE	A	0.4235	0.4059	492.86	3.38	41.50	420	47.20
CU2.07 AGC.84 AS	C	0.2968	0.3055	480.39	6.03	42.15	440	46.90
CUBIC							460	46.50
REF. FOR R: (J.HAK ET AL,1970)							480	45.80
REMARKS: LCC. CZECHOSLOVAKIA							500	44.20
Chemical Analysis:							500	44.20
							520	43.30
							540	42.80
							560	42.20
							580	41.40
							600	41.70
							620	39.30

	Wt. %
Cu	45.3
Ag	21.5
As	25.6
	<hr/>
	92.4

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			LAMBDA (NM)	Y	
CATTIERITE	A	0.4502	0.4061	-505.05	0.93	33.81	440	33.80
CC S2	C	0.3119	0.3154	-498.11	0.72	33.69	460	33.60
CUBIC							480	33.50
REF. FOR R: (DEMIRSCY,1968)							500	33.40
REMARKS: LOC. UNKNOWN							520	33.30
							520	33.30
							540	33.40
							560	33.60
							580	33.80
							600	33.90
							620	34.30
							640	34.70
							660	35.20

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			LAMBDA (NM)	Y	
GCLDFIELDITE 1,1	A	0.4475	0.4084	567.95	0.56	32.00	420	31.40
CUB(TE,SB)S4	C	0.3105	0.3179	561.58	0.54	32.02	440	31.20
CUBIC							460	32.00
REF. FOR R: (LEVY,1966)							480	31.80
REMARKS: LCC. GCLDFIELD,U.S.A.							500	32.20
							500	32.20
							520	32.00
							540	32.00
							560	32.20
							580	32.00
							600	31.80
							620	32.20
							640	31.80

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
GCLDFIELDITE 1,2	A	0.4477	0.4085	572.37	0.76	31.71	420	31.40
CUB(TE,SB)S4	C	0.3109	0.3180	567.89	0.67	32.29	440	31.60
CUBIC							460	32.20
REF. FOR R: (LEVY,1966)							480	32.00
REMARKS: LCC. GCLDFIELD,U.S.A.							500	32.20
							500	32.20
							520	32.20
							540	32.40
							560	32.40
							580	32.20
							600	32.40
							620	33.00
							640	31.80

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
GCLDFIELDITE 1,3	A	0.4460	0.4087	509.09	0.37	31.71	420	31.40
CU ₃ (TE,SB)S ₄	C	0.2093	0.3173	507.00	0.26	31.78	440	31.20
CUBIC							460	32.00
REF. FOR R: (LEVY,1966)							480	31.80
REMARKS: LCC. GCLDFIELD,U.S.A.							500	32.00
							500	32.00
							520	31.80
							540	32.00
							560	32.00
							580	31.60
							600	31.60
							620	31.80
							640	31.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
SULLVANITE 3,2	A	0.4605	0.4122	586.38	12.20	31.59		420	28.60
CL3 V S4	C	0.3244	0.3336	574.99	8.48	31.24		440	26.20
CUBIC								460	25.70
REF. FOR R: LEVY (1966)								480	27.90
REMARKS: LCC. MERCUR, UTAH, U.S.A.								500	31.70
								500	31.70
								520	32.20
								540	31.20
								560	30.20
								580	31.00
								600	31.20
								620	33.30
								640	34.80

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
SULFANITE 3,3	A	0.4607	0.4119	586.74	12.13	31.52	420	28.60
CUS V S4	C	0.3244	0.3332	575.30	8.37	31.16	440	26.20
ICUBIC							460	25.70
REF. FOR R: LEVY (1966)							480	27.90
REMARKS: LOC. MERCUR, UTAH, U.S.A.							500	31.60
							500	31.60
							520	32.20
							540	31.00
							560	30.00
							580	31.00
							600	31.20
							620	33.30
							640	34.80

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
SULFANITE 3,1	A	0.4600	0.4118	586.60	11.57	31.14	420	28.20
Cu ₃ V S ₄	C	0.3236	0.3327	574.93	8.02	30.82	440	26.20
CUBIC							460	25.40
REF. FOR R: LEVY (1966)							480	27.90
REMARKS: LOC. MERCUR, UTAH, U.S.A.							500	31.50
							500	31.50
							520	32.00
							540	30.60
							560	29.60
							580	30.40
							600	31.20
							620	33.00
							640	34.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
SULFANITE 4	A	0.4628	0.4129	586.49	14.28	29.89		420	25.30
CL3 V S4	C	0.3272	0.3369	575.04	10.12	29.50		440	23.80
CUBIC								460	24.00
REF. FOR R: LEVY (1966)								480	25.80
REMARKS: LCC. BURRA-BURRA, AUSTRALIA								500	30.00
								500	30.00
								520	30.60
								540	29.20
								560	28.30
								580	29.10
								600	30.50
								620	31.20
								640	33.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
CGLUSITE 1,1	A	0.4603	0.4148	584.10	13.88	30.65	420	23.60
CL3(AS,SN,V,FE,SB)S4	C	0.3274	0.3359	575.88	9.90	30.22	440	25.10
CUBIC							460	26.00
REF. FOR R: LEVY (1966)							480	26.50
REMARKS: LGC. ELTTE, MONTANA, CANADA							500	27.60
							500	27.60
							520	28.80
							540	29.90
							560	31.00
							580	31.60
							600	31.90
							620	31.50
							640	30.80

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS		REFLECTIVITY R
		COORDINATES X	Y			(VISUAL R) Y	LAMBDA (NM)	
CCLUSITE 1,3	A	0.4603	0.4147	584.18	13.81	30.57	420	23.60
CU3(AS,SN,V,FE,SB)S4	C	0.3272	0.3357	575.94	9.82	30.14	440	25.10
CUBIC							460	26.00
REF. FOR R: LEVY (1966)							480	26.50
REMARKS: LOC. ELTTE, MONTANA							500	27.40
							500	27.40
							520	28.80
							540	29.80
							560	31.00
							580	31.40
							600	31.80
							620	31.50
							640	30.80

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
CCLUSITE 1,2	A	0.4602	0.4139	584.75	13.18	30.01	420	23.40
CL3(AS,SN,V,FE,SB)S4 CUBIC	C	0.3268	0.3345	576.40	5.36	29.55	440	24.80
REF. FOR R: LEVY (1966)							460	25.80
REMARKS: LCC. BUTTE, MONTANA, CANADA							480	26.20
							500	27.10
							500	27.10
							520	28.30
							540	29.30
							560	30.20
							580	30.80
							600	31.10
							620	31.00
							640	30.50

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			LAMBDA (NM)	Y	
CCLUSITE 2,2	A	0.4634	0.4109	588.51	13.30	29.57	420	24.00
CUB(AS,SN,V,FE,SB)S4	C	0.3284	0.3314	580.28	8.96	29.02	440	24.80
CUBIC							460	25.40
REF. FOR R: LEVY (1966)							480	25.90
REMARKS: LOC. CHIZELIL, FRANCE							500	26.60
							500	26.60
							520	27.20
							540	28.20
							560	29.60
							580	30.20
							600	30.60
							620	31.60
							640	31.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	WAVELENGTH (NM)	REFLECTIVITY R
		X	Y					
CCLLSITE 2,3	A	0.4633	0.4108	588.60	13.17	29.53	420	24.00
CU3(AS,SN,V,FE,SB)S4	C	0.3282	0.3312	580.31	8.85	28.98	440	24.80
CUBIC							460	25.40
REF. FOR R: LEVY (1966)							480	25.90
REMARKS: LOC. CHIZEUIL, FRANCE							500	26.60
							500	26.60
							520	27.20
							540	28.30
							560	29.40
							580	30.10
							600	30.60
							620	31.60
							640	31.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMECA (NM)	REFLECTIVITY R
		X	Y			Y			
CELLSITE 2,1	A	0.4625	0.4117	587.49	13.24	29.50		420	24.00
Cu3(As,Sn,V,Fe,Sb)S4	C	0.3279	0.3321	579.26	9.71	28.99		440	24.80
CUBIC								460	25.00
REF. FOR R: LEVY (1966)								480	26.00
REMARKS: LCC. CHIZEUIL, FRANCE								500	26.80
								520	27.40
								540	28.40
								560	29.20
								580	30.10
								600	31.10
								620	31.30
								640	30.70

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
TETRAHEDRITE 1,2	A	0.4463	0.4109	558.12	1.48	32.67	420	30.80
CL3 SB S3.25	C	0.3108	0.3208	558.17	1.40	32.73	440	31.80
CUBIC							460	32.00
REF. FOR R: (LEVY,1966)							480	32.00
REMARKS: LCC. HORN-ALSEN							500	32.30
							500	32.30
							520	32.60
							540	33.00
							560	33.30
							580	33.20
							600	32.70
							620	31.40
							640	31.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (C/PL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
TETRAFERRITE 2,2	A	0.4456	0.4108	542.11	0.99	32.61	420	30.90
CL3 SB S3.25	C	0.3099	0.3201	548.34	0.98	32.70	440	32.00
CLBIC							460	32.20
REF. FOR R: (LEVY,1966)							480	32.20
REMARKS: LOC. BCUCJCUCCUN, ALGERIA							500	32.40
							500	32.40
							520	32.60
							540	33.10
							560	33.30
							580	33.10
							600	32.40
							620	31.10
							640	31.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y				Y		
TETRAECRITE 2,1	A	0.4454	0.4104	527.97	0.71	32.55		420	31.60
CU3 SB S3.25	C	0.3096	0.3194	541.56	0.72	32.64		440	32.00
CUBIC								460	32.20
REF. FOR R: (LEVY,1966)								480	32.20
REMARKS: LOC. BOUDJOLDCUN, ALGERIA								500	32.40
								500	32.40
								520	32.70
								540	33.10
								560	33.10
								580	32.90
								600	32.20
								620	31.60
								640	31.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y				Y		
TETRAECRITE 1,1	A	0.4451	0.4103	519.19	0.66	32.53		420	31.70
CU3 SB S3.25	C	0.3092	0.3191	529.70	0.57	32.64		440	32.00
CUBIC								460	32.40
REF. FOR R: (LEVY,1966)								480	32.20
REMARKS: LCC. HORNALSEN								500	32.50
								500	32.50
								520	32.80
								540	33.10
								560	33.10
								580	32.80
								600	32.10
								620	31.60
								640	31.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y				Y		
TETRAHECRITE 1,3	A	0.4462	0.4104	552.14	1.07	32.46		420	30.80
CL3 SB S3.25	C	0.3103	0.3200	553.84	1.06	32.52		440	31.80
CUBIC								460	32.00
REF. FOR R: (LEVY,1966)								480	32.00
REMARKS: LCC. FORNHAUSEN								500	32.20
								500	32.20
								520	32.50
								540	32.80
								560	33.00
								580	32.90
								600	32.40
								620	31.20
								640	31.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
TETRAHEDRITE 2,3	A	0.4464	0.4106	557.87	1.24	32.46	420	30.90
CUBIC	C	0.3106	0.3202	557.07	1.19	32.52	440	31.70
REF. FOR R: (LEVY, 1966)							460	32.00
REMARKS: LCC. ECUEJOUCCUN, ALGERIA							480	31.80
							500	32.00
							500	32.00
							520	32.40
							540	32.90
							560	33.00
							580	33.00
							600	32.40
							620	31.10
							640	31.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
TETRAHEDRITE 3	A	0.4437	0.4098	505.66	0.86	31.63	420	30.50
CU3 SB S3.25	C	0.3076	0.3177	497.76	0.84	31.79	440	31.70
CUBIC							460	32.20
REF. FOR R: (LEVY,1966)							480	31.80
REMARKS: LCC. FANATINA							500	32.00
							500	32.00
							520	31.90
							540	32.20
							560	32.20
							580	32.00
							600	31.20
							620	30.40
							640	30.30

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
TETRAFECRITE 4	A	0.4428	0.4100	504.27	1.08	30.80	420	29.20
CU3 SB S3.25	C	0.3071	0.2174	495.27	1.04	30.99	440	31.40
CUBIC							460	31.40
REF. FOR R: (LEVY,1966)							480	31.10
REMARKS: LOC. FAMATINA							500	31.00
							500	31.00
							520	31.40
							540	31.40
							560	31.30
							580	31.10
							600	30.40
							620	30.40
							640	29.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (DOMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
TETRAEDRITE, E	A	0.4450	0.4107	523.83	0.76	30.56	440	30.19
CL3 SB 53.25	C	0.3094	0.3193	536.39	0.65	30.66	460	29.93
CUBIC							480	30.09
REF. FOR R: (TUGAL, 1969)							500	30.47
REMARKS: LOC. TURKEY							520	30.80
							520	30.80
							540	31.01
							560	31.12
							580	30.94
							600	30.36
							620	29.91
							640	29.27
							660	28.08

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
TETRAFERRITE, 49A	A	0.4458	0.4076	495.74	0.42	29.97	440	30.30
CU ₂ SB 53.25	C	0.3085	0.3155	484.72	0.67	30.03	460	30.23
CUBIC							480	30.25
REF. FOR R: (TUGAL, 1969)							500	30.36
REMARKS: LCC. TURKEY							520	30.28
							520	30.28
							540	30.03
							560	30.07
							580	29.88
							600	29.87
							620	30.17
							640	29.55
							660	28.67

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
TETRAHEDRITE, 49B	A	0.4420	0.4106	504.92	1.26	29.33	440	29.58
CL3 SB 53.25	C	0.3063	0.3177	495.39	1.32	29.55	460	29.44
CLBIC							480	29.59
REF. FOR R: (TUGAL, 1969)							500	29.78
REMARKS: LCC. TURKEY							520	30.07
							520	30.07
							540	30.13
							560	30.01
							580	29.45
							600	28.78
							620	28.34
							640	27.36
							660	27.11

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
TENNANTITE 6	A	0.4433	0.4102	506.46	0.97	30.79		420	30.40
CUB AS S3.25	C	0.3073	0.3182	499.27	0.93	30.97		440	30.80
CUBIC								460	30.70
REF. FOR R: (LEVY,1966)								480	31.20
REMARKS: LCC. TSUMEB								500	31.50
								500	31.50
								520	31.70
								540	31.40
								560	31.00
								580	30.80
								600	30.80
								620	29.60
								640	29.30

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			LAMBDA (NM)	Y	
TENNANTITE 8,1	A	0.4358	0.4097	498.01	2.72	29.28	420	31.80
CL3 AS S3.25	C	0.3003	0.3126	486.30	3.96	29.70	440	30.80
CUBIC							460	31.40
REF. FOR R: (LEVY,1966)							480	30.70
REMARKS: LOC. FAURCN., SPECIMEN WITH SN							500	30.60
							500	30.60
							520	30.60
							540	30.70
							560	30.00
							580	29.20
							600	28.90
							620	27.40
							640	26.30

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
TENNANTITE 8,2	A	0.4360	0.4092	497.31	2.68	28.54	420	31.80
CUB AS S3.25	C	0.3002	0.3121	485.71	4.05	29.35	440	30.40
CUBIC							460	31.20
REF. FOR R: (LEVY,1966)							480	30.30
REMARKS: LCC. FAURCN., SPECIMEN WITH SN							500	30.40
							500	30.40
							520	30.30
							540	30.20
							560	29.60
							580	29.00
							600	28.30
							620	27.00
							640	26.30

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY		DOMINANT (COMPL.) WAVELENGTH	PLURITY (P%)	BRIGHTNESS		REFLECTIVITY R
		COORDINATES X	Y			(VISUAL R) Y	LAMBDA (NM)	
TENNANTITE,40	A	0.4420	0.4121	510.73	1.30	28.86	440	28.53
Cu3 AS 53.25	C	0.3070	0.3202	507.12	1.02	29.08	460	28.45
CUBIC							480	28.95
REF. FOR R: (TUGAL,1969)							500	29.25
REMARKS: LOC. TURKEY							520	29.72
							520	29.72
							540	29.78
							560	29.61
							580	29.04
							600	28.16
							620	27.72
							640	27.07
							660	26.14

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			LAMBDA (NM)	Y	
TENNANTITE 8,3	A	0.4360	0.4090	497.01	2.69	28.82	420	31.80
CU3 AS S3.25	C	0.3001	0.3118	485.37	4.11	29.23	440	30.40
CUBIC							460	31.00
REF. FOR R: (LEVY,1966)							480	30.30
REMARKS: LCC. HAURCN., SPECIMEN WITH SN							500	30.40
							500	30.40
							520	30.10
							540	30.10
							560	29.50
							580	28.80
							600	28.20
							620	26.80
							640	26.30

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
TENNANTITE, 47B	A	0.4404	0.4111	503.85	1.62	28.53	440	29.08
CUB AS 53.25	C	0.3050	0.3175	493.78	1.81	28.80	460	28.72
CUBIC							480	28.65
REF. FOR R: (TUGAL, 1969)							500	29.28
REMARKS: LCC. TURKEY							520	29.62
							520	29.62
							540	29.66
							560	29.09
							580	28.63
							600	27.84
							620	27.02
							640	26.50
							660	26.42

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
TENNANTITE, 47A	A	0.4380	0.4113	501.83	2.17	28.32	440	29.25
CUB AS 53.25	C	0.3028	0.3166	491.46	2.69	28.68	460	28.82
CUBIC							480	29.05
REF. FOR R: (TUGAL, 1969)							500	29.46
REMARKS: LOC. TURKEY							520	29.82
							520	29.82
							540	29.69
							560	28.91
							580	28.36
							600	27.36
							620	27.02
							640	25.84
							660	24.96

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
TENNANTITE, 51	A	0.4403	0.4102	501.34	1.65	28.14	440	28.85
CU3 AS 53.25	C	0.3045	0.3162	490.75	2.09	28.41	460	28.79
CUBIC							480	28.73
REF. FOR R: (TUGAL, 1969)							500	29.01
REMARKS: LOC. TURKEY							520	29.16
							520	29.16
							540	29.08
							560	28.77
							580	28.00
							600	27.66
							620	26.97
							640	26.23
							660	25.71

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y				Y		
TENNANTITE 7,2	A	0.4338	0.4086	496.15	3.22	27.22		420	30.50
CL3 AS S3.25	C	0.2979	0.3100	484.50	5.10	27.67		440	29.70
CUBIC								460	29.50
REF. FOR R: (LEVY,1966)								480	29.20
REMARKS: LCC. NIARI								500	28.90
								500	28.90
								520	28.90
								540	28.60
								560	27.90
								580	27.00
								600	26.20
								620	25.80
								640	24.60

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y				Y		
TENNANTITE 7,1	A	0.4339	0.4078	495.16	3.22	27.20		420	30.90
CUB AS S3.25	C	0.2976	0.3085	483.05	5.37	27.64		440	30.40
CUBIC								460	29.60
REF. FOR R: (LEVY,1966)								480	28.80
REMARKS: LCC. NIARI								500	29.00
								500	29.00
								520	28.90
								540	28.40
								560	27.80
								580	27.00
								600	26.60
								620	25.40
								640	24.60

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			LAMBDA (NM)	Y	
TENNANTITE 5	A	0.4319	0.4078	495.12	3.70	26.05	420	29.40
CL3 AS S3.25	C	0.2959	0.3077	483.30	6.08	26.53	440	29.30
CUBIC							460	28.80
REF. FOR R: (LEVY,1966)							480	28.40
REMARKS: LOC. SAN VINCENT							500	28.00
							500	28.00
							520	27.80
							540	27.30
							560	26.60
							580	26.10
							600	25.20
							620	24.20
							640	23.20

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
VAESITE	A	0.4467	0.4043	-575.05	1.01	30.89	440	32.50
NI S2	C	0.3076	0.3105	-566.90	1.87	30.89	460	31.90
CUBIC							480	31.60
REF. FOR R: (DEMIRSOY, 1968)							500	31.10
REMARKS: LOC. UNKNOWN							520	31.00
							540	30.70
							560	30.60
							580	30.70
							600	30.80
							620	31.10
							640	31.40
							660	31.90

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
GERMANITE,3	A	0.4645	0.4025	606.26	8.22	22.53		420	21.20
CU6 FE GE SE	C	0.3245	0.3177	605.55	4.23	22.02		440	21.20
CUBIC								460	21.00
REF. FOR R: (LEVY,1966)								480	20.60
REMARKS: LCC. TSUMEB								500	20.30
								500	20.30
								520	20.60
								540	21.00
								560	21.80
								580	22.60
								600	23.40
								620	24.60
								640	25.80

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS		REFLECTIVITY R
		COORDINATES X	Y			(VISUAL R) Y	LAMBDA (NM)	
GERMANITE, 6	A	0.4641	0.4016	612.34	7.33	22.27	420	21.20
CU6 FE GE S8	C	0.3234	0.3161	618.90	3.51	21.77	440	21.20
CUBIC							460	21.00
REF. FOR R: (LEVY, 1966)							480	20.70
REMARKS: LOC. TSUMEB							500	20.40
							500	20.40
							520	20.40
							540	20.80
							560	21.30
							580	22.40
							600	23.00
							620	24.00
							640	25.80

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS		LAMBDA (NM)	REFLECTIVITY R
		COORDINATES X	Y			(VISUAL R)	Y		
GERMANITE, 5	A	0.4616	0.4025	612.28	6.22	22.04		420	20.90
CU6 FE GE S8	C	0.3215	0.3162	617.83	3.01	21.62		440	21.20
CUBIC								460	21.00
REF. FOR R: (LEVY, 1966)								480	20.70
REMARKS: LOC. TSUMEB								500	20.40
								500	20.40
								520	20.40
								540	20.80
								560	21.20
								580	22.20
								600	22.80
								620	23.70
								640	24.80

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
GERMANITE, 4	A	0.4614	0.4008	657.58	4.90	21.60	420	21.20
Cu6 Fe Ge S8	C	0.3202	0.3134	-494.40	3.05	21.19	440	21.20
CUBIC							460	21.00
REF. FOR R: (LEVY, 1966)							480	20.60
REMARKS: LCC. TSUMEB							500	20.10
							500	20.10
							520	20.00
							540	20.40
							560	20.70
							580	21.60
							600	22.20
							620	23.10
							640	24.80

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
GERMANITE, 2	A	0.4666	0.4026	603.52	9.76	21.58	420	20.00
CU6 FE GE S8	C	0.3271	0.3186	602.68	5.17	21.02	440	20.10
CUBIC							460	19.60
REF. FOR R: (LEVY, 1966)							480	19.50
REMARKS: LOC. BANCAIRUAN. HOMOGENEOUS SPECIMEN							500	19.20
							500	19.20
							520	19.10
							540	19.90
							560	20.80
							580	21.80
							600	23.10
							620	24.00
							640	24.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y				Y		
GERMANITE,1	A	0.4668	0.4026	603.35	9.50	21.22		420	19.60
CU6 FE GE S8	C	0.3271	0.3187	602.29	5.20	20.67		440	19.80
CUBIC								460	19.30
REF. FOR R: (LEVY,1966)								480	19.00
REMARKS: LCC. BANCAIRCUN. HOMOGENEOUS SPECIMEN								500	18.90
								500	18.90
								520	18.90
								540	19.70
								560	20.30
								580	21.40
								600	22.60
								620	23.50
								640	24.20

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
BOURNITE, NG36	A	0.4727	0.4140	587.82	21.91	24.91		440	18.21
CUS FE S4	C	0.3403	0.3437	578.95	15.45	24.21		460	18.71
CUBIC								480	19.83
REF. FOR R: (TUGAL, 1969)								500	21.34
REMARKS: LCC. TURKEY								520	22.52
								520	22.52
								540	23.53
								560	24.50
								580	25.40
								600	26.18
								620	27.05
								640	27.45
								660	28.67

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			LAMBDA (NM)	Y	
BOURNITE, NGD3	A	0.4789	0.4088	591.92	22.58	22.88	440	17.45
CU5 FE S4	C	0.3443	0.3376	584.21	14.89	22.01	460	17.09
CUBIC							480	17.67
REF. FOR R: (TUGAL, 1969)							500	18.65
REMARKS: LOC. TURKEY							520	19.81
							520	19.81
							540	20.76
							560	22.08
							580	23.16
							600	24.75
							620	25.85
							640	27.47
							660	27.78

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			LAMBDA (NM)	Y	
BERNITE	A	0.4774	0.4066	593.78	19.74	22.05	440	17.55
CUS FE S4	C	0.3412	0.3327	586.53	12.74	21.22	460	17.42
CUBIC							480	17.53
REF. FOR R: THE AUTHOR							500	18.13
REMARKS: LCC. KCREA							520	18.97
							520	18.97
							540	20.00
							560	21.23
							580	22.35
							600	23.63
							620	25.06
							640	26.18
							660	27.43

Electron Probe Analysis: Wt. %
Cu 63.41
Fe 11.41
S 25.49

100.31

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
ECRNITE, NGC6	A	0.4751	0.4113	589.87	21.69	22.04		440	16.70
CU5 FE S4	C	0.3413	0.3397	581.62	14.65	21.32		460	16.60
CUBIC								480	17.26
REF. FOR R: (TLGAL, 1969)								500	18.48
REMARKS: LCC. TURKEY								520	19.41
								520	19.41
								540	20.65
								560	21.40
								580	22.49
								600	23.42
								620	24.17
								640	25.03
								660	26.46

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES X Y	DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
BCRNITE 6302	A	0.4793 0.4042	596.14	19.67	21.83	440	17.88
CU5 FE S4	C	0.3418 0.3293	589.90	11.98	20.94	460	17.37
CLBIC						480	17.20
REF. FOR R: THE AUTHOR						500	17.80
REMARKS: LCC. ARIZONA						520	18.50
Electron Probe Analysis:						520	18.50
		Wt. %				540	19.57
		Cu 63.44				560	20.68
		Fe 11.57				580	22.19
		S 24.88				600	23.60
		<u>99.89</u>				620	25.01
						640	26.74
						660	28.13

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
BCRNITE, 4E	A	0.4739	0.4080	592.47	18.56	21.52	440	17.37
CUS FE S4	C	0.3380	0.3328	584.93	11.91	20.82	460	17.16
CUBIC							480	17.42
REF. FOR R: (TUGAL, 1969)							500	18.18
REMARKS: LOC. TURKEY							520	18.92
							520	18.92
							540	19.89
							560	20.84
							580	21.84
							600	22.85
							620	24.16
							640	24.76
							660	25.71

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
BOURNITE 13513	A	0.4791	0.4045	555.83	19.74	21.30		440	17.52
CU5 FE S4	C	0.3417	0.3292	589.86	11.96	20.44		460	17.00
CUBIC								480	16.45
REF. FOR R: THE AUTHOR								500	17.38
REMARKS: LOC. AUSTRALIA								520	18.10
Electron Probe Analysis:								520	18.10
								540	19.08
								560	20.24
								580	21.63
								600	23.07
								620	24.37
								640	26.15
								660	27.12

Wt. %
Cu 63.47
Fe 11.48
S 24.82
99.77

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES X Y	DOMINANT (CCYPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
BOURNITE 6301	A	0.4720 0.4011	603.80	12.46	20.32	440	18.72
CUS FE S4	C	0.3317 0.3187	604.43	6.43	19.64	460	17.81
CUBIC						480	17.17
REF. FOR R: THE AUTHOR						500	17.28
REMARKS: LCC. NEW JERSEY						520	17.69
Electron Probe Analysis:						540	18.47
						560	19.35
						580	20.47
						600	21.69
						620	23.03
						640	24.04
						660	25.54

Wt. %
Cu 63.01
Fe 11.60
S 25.50

100.11

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES X Y	DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
BERNITE 5796	A	0.4772 0.3998	603.66	15.16	20.14	440	18.19
CUS FE S4	C	0.3366 0.3193	604.27	7.90	19.32	460	17.06
CUBIC						480	16.44
REF. FOR R: THE AUTHOR						500	16.51
REMARKS: LCC. RHODESIA						520	16.98
						520	16.98
Electron Probe Analysis:		Wt. %				540	17.85
		Cu 63.41				560	18.95
		Fe 11.50				580	20.43
		S 25.70				600	21.74
		<u> </u>				620	23.23
		100.61				640	24.86
						660	26.58

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
BCRNITE 12894	A	0.4736	0.3986	610.59	11.82	19.78	440	18.79
CU5 FE S4	C	0.3319	0.3145	631.04	5.46	19.06	460	17.49
CUBIC							480	16.69
REF. FOR R: THE AUTHOR							500	16.58
REMARKS: LOC. RHCDESIA							520	16.95
Electron Probe Analysis:							520	16.55
							540	17.66
							560	18.70
							580	19.95
							600	21.31
							620	22.65
							640	24.12
							660	25.44

Electron Probe Analysis: Wt. %
Cu 63.39
Fe 11.41
S 24.88

99.68

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
BOURNITE, A55 (2)	A	0.4779	0.4006	601.73	16.18	19.67		440	17.55
CUS FE S4	C	0.3379	0.3210	600.50	8.71	18.86		460	16.37
CUBIC								480	15.72
REF. FOR R: (TUGAL, 1969)								500	15.99
REMARKS: LCC. TURKEY								520	16.67
								540	17.48
								560	18.51
								580	19.83
								600	21.41
								620	22.90
								640	24.45
								660	25.27

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
BERNITE, A55(1)	A	0.4742	0.3973	615.87	11.34	18.62	440	18.02
CUS FE S4	C	0.3318	0.3128	600.98	6.75	17.92	460	16.54
CUBIC							480	15.67
REF. FOR R: (TUGAL, 1969)							500	15.65
REMARKS: LCC. TURKEY							520	15.94
							540	16.51
							560	17.48
							580	18.61
							600	20.17
							620	21.69
							640	23.20
							660	23.87

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CNFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
EIXBYITE	A	0.4445	0.4100	510.38	0.72	22.24	430	22.20
(MN,FE)2 O3	C	0.3084	0.3180	503.55	0.55	22.32	470	22.20
ICUBIC							490	22.00
REF. FOR R: (NICHOL & PHILLIPS, 1964)							520	22.50
REMARKS: LCC. UNKNOWN							550	22.70
							580	22.40
							610	21.80
							650	21.20

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
MAGNETITE	A	0.4497	0.4078	589.78	1.66	20.92	440	20.50
FE FE2 O4	C	0.3124	0.3178	583.58	1.02	20.64	460	20.50
CUBIC							480	20.60
REF. FOR R: (DEMIRSCY,1968)							500	20.60
REMARKS: LOC. UNKNOWN							520	20.60
							540	20.70
							560	21.00
							580	21.00
							600	21.00
							620	21.10
							640	21.10
							660	21.10

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
JACOBSITE	A	0.4394	0.4082	496.25	1.91	18.91		430	19.70
(MN,FE,MG)(FE,MN) ₂ O ₄	C	0.3029	0.3129	485.09	2.98	19.10		470	20.00
CUBIC								490	20.00
REF. FOR R: (NICHOL & PHILLIPS, 1964)								520	19.40
REMARKS: LCC. UNKNOWN								550	19.30
								580	19.00
								610	18.40
								650	17.50

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
FRANKLINITE	A	0.4410	0.4075	494.66	1.56	18.15	430	19.20
(Zn,Fe,Mn)(Fe,Mn) ₂ C4	C	0.3039	0.3122	482.54	2.69	18.29	470	18.90
CUBIC							490	18.80
REF. FOR R: (NICHOL & PHILLIPS, 1964)							520	18.60
REMARKS: LOC. UNKNOWN							550	18.40
							550	18.40
							580	18.20
							610	17.80
							650	17.10

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
MAGNETIC FERRITE	A	0.4354	0.4044	490.45	3.02	17.09	440	19.40
Mg Fe ₂ O ₄	C	0.2974	0.3047	478.87	5.93	17.33	460	19.10
CUBIC							480	18.70
REF. FOR R: (DEMIRSCY, 1968)							500	18.40
REMARKS: LCC. UNKNOWN							520	17.90
							520	17.90
							540	17.50
							560	17.30
							580	16.90
							600	16.60
							620	16.40
							640	16.20
							660	15.90

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES X Y		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
MARMATITE	A	0.4410	0.4055	489.58	1.65	17.63	430	19.00
ZN S	C	0.3031	0.3093	478.00	3.32	17.75	470	18.80
CLBIC							490	18.30
REF. FOR R: (NICHOL & PHILLIPS, 1964)							520	18.00
REMARKS: LCC. TREFCA, YUGOSLAVIA							550	17.80
							550	17.80
							580	17.60
							610	17.30
							650	17.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
SPHALERITE 15	A	0.4434	0.4044	480.80	1.17	19.19	440	20.72
ZN S	C	0.3049	0.3083	466.29	2.93	19.25	460	20.50
CUBIC							480	19.84
REF. FOR R: (GRAFENAUER ET AL, 1969)							500	19.30
REMARKS: LCC. NCVC BRDC, YUGOSLAVIA							540	19.22
							540	19.22
							580	19.25
							620	19.02
							660	18.78

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
SPHALERITE S	A	0.4402	0.4094	499.26	1.69	19.07	440	19.90
ZN S	C	0.3043	0.3142	485.50	2.33	19.23	460	19.71
CUBIC							480	19.51
REF. FOR R: (GRAFENALER ET AL, 1969)							500	19.18
REMARKS: LCC. LECE, YUGOSLAVIA							540	19.60
							540	19.60
							580	19.50
							620	17.90
							660	17.60

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y				Y		
SFHALERITE 14	A	0.4422	0.4040	482.84	1.46	18.99		440	20.85
ZN S	C	0.3035	0.3069	469.21	3.59	19.08		460	20.50
CUBIC								480	19.76
REF. FOR R: (GRAFENAUER ET AL, 1969)								500	19.22
REMARKS: LCC. STARI TRG, YUGOSLAVIA								540	19.11
								540	19.11
								580	19.02
								620	18.70
								660	18.56

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			LAMBDA (NM)	Y	
SPHALERITE 12	A	0.4403	0.4050	488.67	1.84	18.97	440	20.81
ZN S	C	0.3021	0.3077	476.21	3.91	19.12	460	20.40
CUBIC							480	20.02
REF. FOR R: (GRAFENAUER ET AL, 1969)							500	19.51
REMARKS: LOC. AJVALIJA , YUGOSLAVI							540	19.28
							540	19.28
							580	18.94
							620	18.48
							660	18.18

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	WAVELENGTH LAMBDA (NM)	REFLECTIVITY R
		X	Y					
SPHALERITE 11	A	0.4415	0.4042	485.10	1.61	18.95	440	20.88
ZN S	C	0.3027	0.3070	472.85	3.82	19.07	460	20.25
CUBIC							480	19.86
REF. FOR R: (GRAFENAUER ET AL, 1969)							500	19.58
REMARKS: LCC. KIZNICA, YUGOSLAVIA							540	19.07
							540	19.07
							580	18.90
							620	18.63
							660	18.51

MINERAL NAME		CHROMATICITY		DOMINANT		BRIGHTNESS		
COMPOSITION	SOURCE	COORDINATES		(COMPL.)	PURITY	(VISUAL R)	LAMBDA	REFLECTIVITY
CRYSTAL SYSTEM		X	Y	WAVELENGTH	(%)	Y	(NM)	R
SPHALERITE 16	A	0.4402	0.4035	485.10	1.95	18.94	440	21.28
ZN S	C	0.3012	0.3052	473.05	4.58	19.08	460	20.55
ICUBIC							480	19.92
REF. FOR R: (GRAFENALER ET AL, 1969)							500	19.76
REMARKS: LOC. JANJEVO, YUGOSLAVIA							540	19.10
							540	19.10
							580	18.80
							620	18.65
							660	18.30

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
SPHALERITE 13	A	0.4292	0.4042	467.94	2.14	18.62	440	20.80
ZN S	C	0.3007	0.3059	475.66	4.63	18.78	460	20.28
CUBIC							480	19.72
REF. FOR R: (GRAFENAUER ET AL, 1969)							500	19.42
REMARKS: LGC. SREBENICA, YUGOSLAVIA							540	18.92
							540	18.92
							580	18.51
							620	18.12
							660	17.85

MINERAL NAME COMPOSITION CRYSTAL SYSTEM		CHROMATICITY SOURCE COORDINATES X Y		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
SPHALERITE 10	A	0.4421	0.4047	485.73	1.44	18.56	440	20.32
ZN S	C	0.3035	0.3080	472.83	3.41	18.66	460	19.48
CLBIC							480	19.30
REF. FOR R: (GRAFENAUER ET AL, 1969)							500	19.08
REMARKS: LCC. SASE, YUGOSLAVIA							540	18.67
							540	18.67
							580	18.56
							620	18.24
							660	18.10

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
SPHALERITE 7	A	0.4431	0.4069	492.52	1.08	18.12		440	19.05
ZN S	C	0.3057	0.3121	478.22	2.08	18.20		460	18.88
CUBIC								480	18.40
REF. FOR R: (GRAFENAUER ET AL, 1969)								500	18.21
REMARKS: LCC. CKCSKA GCFA, YUGOSLAVIA								540	18.40
								540	18.40
								580	18.20
								620	17.72
								660	17.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			LAMBDA (NM)	Y	
SFHALERITE 8	A	0.4417	0.4053	488.31	1.49	17.93	440	19.41
ZN S	C	0.3034	0.3090	475.97	3.29	18.04	460	18.93
CUBIC							480	18.75
REF. FOR R: (GRAFENALER ET AL, 1969)							500	18.43
REMARKS: LCC. CCBREVC, YUGOSLAVIA							540	18.21
							540	18.21
							580	17.78
							620	17.63
							660	17.42

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
SPHALERITE 4	A	0.4437	0.4045	480.05	1.09	17.92		440	19.18
ZN S	C	0.3051	0.3090	468.33	2.75	17.98		460	19.02
CUBIC								480	18.70
REF. FOR R: (GRAFENAUER ET AL,1969)								500	18.15
REMARKS: LCC. PONOVICE, YUGOSLAVIA								540	17.92
								580	17.92
								620	17.80
								660	17.62

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			LAMBDA (NM)	Y	
SPHALERITE 3	A	0.4420	0.4054	488.27	1.42	17.91	440	19.31
ZN S	C	0.3039	0.3052	474.99	3.09	18.01	460	19.10
CUBIC							480	18.51
REF. FOR R: (GRAFENAUER ET AL,1969)							500	18.15
REMARKS: LOC. REMSNIK, YUGOSLAVIA							540	18.20
							580	17.85
							620	17.58
							660	17.35

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
SPHALERITE 6	A	0.4438	0.4046	480.20	1.06	17.83	440	19.28
ZN S	C	0.3051	0.3087	466.79	2.80	17.89	460	18.68
CUBIC							480	18.32
REF. FOR R: (GRAFENAUER ET AL, 1969)							500	18.07
REMARKS: LCC. RUDNIK, YUGOSLAVIA							540	17.88
							580	17.80
							620	17.71
							660	17.60

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y		LAMBDA (NM)	REFLECTIVITY R
		X	Y						
SPHALERITE 1	A	0.4437	0.4047	481.27	1.08	17.82		440	19.08
ZN S	C	0.3051	0.3092	469.30	2.71	17.88		460	18.72
CUBIC								480	18.66
REF. FOR R: (GRAFENAUER ET AL,1969)								500	17.95
REMARKS: LOC. CEMERNICA, YUGOSLAVIA								540	17.90
								580	17.80
								620	17.70
								660	17.50

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES X	Y	DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
SPHALERITE M3	A	0.4415	0.4066	492.28	1.47	17.75	440	18.95
ZN S	C	0.3038	0.3110	479.97	2.88	17.88	460	18.50
CUBIC							480	18.37
REF. FOR R: (TUGAL, 1969)							500	18.24
REMARKS: LCC. TURKEY							520	18.40
							540	18.02
							560	17.83
							580	17.67
							600	17.47
							620	17.33
							640	17.18
							660	17.12

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS		REFLECTIVITY R
		COORDINATES X	Y			(VISUAL R) Y	LAMBDA (NM)	
SPHALERITE KD2A	A	0.4423	0.4061	489.17	1.08	17.38	440	18.32
ZN S	C	0.3054	0.3115	464.60	2.70	17.46	460	18.04
CUBIC							480	17.95
REF. FOR R: (TUGAL, 1969)							500	17.78
REMARKS: LCC. TURKEY							520	17.70
							520	17.70
							540	17.54
							560	17.38
							580	17.35
							600	17.11
							620	17.45
							640	17.25
							660	16.41

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
SPHALERITE MZ	A	0.4420	0.4066	492.05	1.36	17.34	440	18.39
ZN S	C	0.3046	0.3112	478.65	2.58	17.44	460	18.18
CUBIC							480	17.89
REF. FOR R: (TUGAL,1969)							500	17.65
REMARKS: LOC. TURKEY							520	17.70
							540	17.59
							560	17.39
							580	17.37
							600	17.24
							620	17.30
							640	16.45
							660	16.02

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y				Y		
SPHALERITE S	A	0.4415	0.4034	482.31	1.66	17.27		440	19.30
ZN S	C	0.3024	0.3054	469.40	4.17	17.36		460	18.48
CUBIC								480	18.20
REF. FOR R: (GRAFENAUER ET AL, 1969)								500	17.65
REMARKS: LCC. SUPLJA STENA, YUGOSLAVIA								540	17.40
								580	17.18
								620	17.38
								660	16.85

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
SPHALERITE 2	A	0.4437	0.4052	484.05	1.04	17.15	440	18.25
ZN S	C	0.3054	0.3102	472.01	2.46	17.26	460	18.06
CUBIC							480	17.88
REF. FOR R: (GRAFENALER ET AL, 1969)							500	17.45
REMARKS: LCC. MEZICA, YUGOSLAVIA							540	17.28
							580	17.15
							620	17.06
							660	16.86

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y				Y		
SPHALERITE 39B	A	0.4442	0.4071	492.75	0.82	17.05		440	17.74
ZN S	C	0.3068	0.3130	477.00	1.36	17.11		460	17.35
CUBIC								480	17.35
REF. FOR R: (TUGAL 1969)								500	17.24
REMARKS: LOC. TURKEY								520	17.20
								540	17.06
								560	17.09
								580	17.47
								600	16.77
								620	16.95
								640	16.18
								660	16.85

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (DOMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y				Y		
SPHALERITE 41	A	0.4422	0.4050	486.62	1.40	16.94		440	18.18
ZN S	C	0.3038	0.3092	475.30	3.13	17.04		460	18.06
CUBIC								480	17.76
REF. FOR R: (TUGAL 1969)								500	17.50
REMARKS: LCC. TURKEY								520	17.25
								540	17.08
								560	16.93
								580	16.93
								600	16.66
								620	16.80
								640	16.57
								660	16.54

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
SPHALERITE KD2C	A	0.4416	0.4054	468.73	1.51	16.90		440	18.22
ZN S	C	0.3033	0.3094	477.48	3.25	17.02		460	17.86
CUBIC								480	17.73
REF. FOR R: (TUGAL, 1969)								500	17.53
REMARKS: LOC. TURKEY								520	17.36
								540	17.09
								560	16.95
								580	16.83
								600	16.56
								620	16.58
								640	16.62
								660	16.41

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
SFHALERITE M1	A	0.4414	0.4058	490.10	1.54	16.81		440	17.98
ZN S	C	0.3035	0.3100	478.22	3.12	16.93		460	17.87
CLBIC								480	17.63
REF. FOR R: (TUGAL, 1969)								500	17.35
REMARKS: LCC. TURKEY								520	17.21
								540	17.06
								560	16.85
								580	16.70
								600	16.73
								620	16.63
								640	15.95
								660	16.02

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
SPHALERITE NG39B	A	0.4413	0.4057	489.89	1.57	16.73		440	17.97
ZN S	C	0.3033	0.3097	477.92	3.22	16.84		460	17.68
CUBIC								480	17.61
REF. FOR R: (TUGAL, 1969)								500	17.25
REMARKS: LCC. TURKEY								520	17.16
								540	16.94
								560	16.78
								580	16.69
								600	16.42
								620	16.54
								640	16.19
								660	16.02

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (DOMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
SPHALERITE NGD16A	A	0.4411	0.4051	488.38	1.65	16.69	440	18.20
ZN S	C	0.3028	0.3083	475.85	3.59	16.80	460	17.73
CUBIC							480	17.45
REF. FOR R: (TUGAL, 1969)							500	17.26
REMARKS: LCC. TURKEY							520	17.03
							540	16.88
							560	16.81
							580	16.64
							600	16.33
							620	16.67
							640	15.87
							660	16.02

MINERAL NAME LOCAL POSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (DOMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
SPHALERITE NG39A	A	0.4410	0.4055	489.58	1.65	16.62	440	17.98
ZN S	C	0.3029	0.3091	477.00	3.24	16.74	460	17.61
CUBIC							480	17.47
REF. FOR P: (TUGAL, 1969)							500	17.20
REMARKS: LCC. TURKEY							520	17.08
							540	16.86
							560	16.64
							580	16.58
							600	16.37
							620	16.32
							640	16.02
							660	16.02

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
SPHALERITE 40B	A	0.4418	0.4065	491.85	1.41	16.51	440	17.39
ZN S	C	0.3042	0.3116	480.62	2.66	16.63	460	17.41
CLBIC							480	17.22
REF. FOR R: (TUGAL 1969)							500	17.11
REMARKS: LCC. TURKEY							520	16.90
							540	16.72
							560	16.63
							580	16.43
							600	16.26
							620	16.32
							640	16.01
							660	15.67

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES X	DOMINANT (COMPL.) WAVELENGTH Y	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R	
SFHALERITE NG348	A	0.4436	0.4053	484.91	1.06	16.50	440	17.49
ZN S	C	0.3051	0.3106	475.08	2.49	16.57	460	17.26
CUBIC							480	17.13
REF. FOR R: (TUGAL, 1969)							500	16.94
REMARKS: LGC. TURKEY							520	16.83
							520	16.83
							540	16.57
							560	16.56
							580	16.34
							600	16.23
							620	16.39
							640	16.37
							660	16.54

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
SPHALERITE 4CA	A	0.4414	0.4057	489.81	1.54	16.47		440	17.62
ZN S	C	0.3034	0.3099	478.21	3.16	16.58		460	17.50
CUBIC								480	17.31
REF. FOR R: (TUGAL 1969)								500	17.13
REMARKS: LCC. TURKEY								520	16.90
								540	16.66
								560	16.48
								580	16.38
								600	16.32
								620	16.24
								640	15.94
								660	15.67

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CYFL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
SPHALERITE	A	0.4413	0.4052	488.45	1.59	16.35		440	17.70
ZN S	C	0.3031	0.3068	476.27	3.42	16.46		460	17.40
CUBIC								480	17.20
REF. FOR R: (DEMIRSCY, 1968)								500	16.90
REMARKS: LCC. UNKNOWN								520	16.70
								540	16.50
								560	16.40
								580	16.30
								600	16.20
								620	16.10
								630	16.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y				Y		
SPHALERITE 39A	A	0.4428	0.4048	484.66	1.27	16.04		440	17.16
ZN S	C	0.3043	0.3092	473.49	2.96	16.11		460	17.06
CUBIC								480	16.87
REF. FOR R: (TUGAL 1969)								500	16.45
REMARKS: LCC. TURKEY								520	16.19
								540	16.11
								560	16.10
								580	15.98
								600	15.88
								620	15.96
								640	15.47
								660	15.95

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS		REFLECTIVITY R
		COORDINATES				(VISUAL R)	LAMBDA (NM)	
		X	Y					
MANGANOSITE	A	0.4410	0.4064	491.95	1.60	14.07	430	15.00
MN C	C	0.2032	0.3108	480.61	3.12	14.18	470	14.90
CUBIC							490	14.60
REF. FOR R: (NICHOL & PHILLIPS, 1964)							520	14.40
REMARKS: LCC. UNKNOWN							550	14.40
							550	14.40
							580	13.90
							610	13.70
							650	13.70

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
CHROMITE A10	A	0.4414	0.4069	493.14	1.49	12.55		440	13.28
(FE,MG)(CR,AL,FE)2 O4	C	0.3041	0.3126	480.81	2.70	12.64		460	13.14
CUBIC								480	13.06
REF. FOR R: (ENGIN, 1969)								520	12.85
REMARKS: LCC. TURKEY								540	12.73
								560	12.66
								580	12.53
								600	12.45
								620	12.29
								640	12.15
								660	11.66

MINERAL NAME
 COMPOSITION
 CRYSTAL SYSTEM

CHROMATICITY COORDINATES
 X Y

DOMINANT (CMPL.) WAVELENGTH

PURITY (P%)

BRIGHTNESS (VISUAL R)
 Y

LAMBDA (NM)

REFLECTIVITY R

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES X	CHROMATICITY COORDINATES Y	DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
CHROMITE F119	A	0.4422	0.4060	490.03	1.34	12.47	440	13.33
(FE,MG)(CR,AL,FE)2 O4	C	0.3043	0.3105	477.00	2.75	12.55	460	13.04
CUBIC							480	12.91
REF. FOR P: (ENGIN, 1969)							520	12.74
REMARKS: LCC. TURKEY							540	12.62
							560	12.53
							580	12.44
							600	12.32
							620	12.35
							640	12.04
							660	11.98

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCHPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		WAVELENGTH (NM)	REFLECTIVITY R
		X	Y			X	Y		
CHROMITE ZU12	A	0.4424	0.4063	490.85	1.28	12.46		440	13.21
(FE,MG)(CR,AL,FE)2 O4	C	0.3046	0.3111	478.36	2.59	12.53		460	13.00
CUBIC								480	12.91
REF. FOR P: (ENGIN,1969)								520	12.69
REMARKS: LCC. TURKEY								540	12.61
								560	12.53
								580	12.40
								600	12.37
								620	12.24
								640	12.13
								660	11.97

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			LAMBDA (NM)	Y	
CHROMITE A17	A	0.4429	0.4056	487.73	1.20	12.42	440	13.17
(FE,MG)(CR,AL,FE)2 O4	C	0.3047	0.3106	476.53	2.62	12.49	460	13.12
CUBIC							480	12.88
REF. FOR R: (ENGIN,1969)							520	12.66
REMARKS: LOC. TURKEY							540	12.55
							560	12.45
							580	12.36
							600	12.22
							620	12.29
							640	12.29
							660	12.21

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES X	Y	DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
CHROMITE ZU30E	A	0.4421	0.4059	489.80	1.37	12.39	440	13.20
(FE,MG)(CR,AL,FE)2O4	C	0.3042	0.3105	477.53	2.78	12.46	460	13.04
CUBIC							480	12.86
REF. FOR R: (ENGIN,1969)							520	12.68
REMARKS: LOC. TURKEY							540	12.53
							560	12.43
							580	12.32
							600	12.21
							620	12.20
							640	12.32
							660	11.82

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
CHROMITE AU32	A	0.4412	0.4064	491.86	1.55	12.37		440	13.25
(FE,MG)(CR,AL,FE)2 O4	C	0.2036	0.3106	479.43	3.00	12.46		460	12.88
CLBIC								480	12.91
REF. FOR R: (ENGIN,1969)								520	12.68
REMARKS: LCC. TURKEY								540	12.54
								560	12.47
								580	12.35
								600	12.25
								620	12.00
								640	11.99
								660	11.76

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
CHROMITE A78	A	0.4424	0.4059	489.50	1.30	12.34		440	13.15
(FE,MG)(CR,AL,FE)2O4	C	0.3043	0.3105	477.00	2.85	12.41		460	12.92
CUBIC								480	12.80
REF. FOR R: (ENGIN, 1969)								520	12.60
REMARKS: LCC. TURKEY								540	12.50
								560	12.37
								580	12.29
								600	12.12
								620	12.16
								640	12.08
								660	12.08

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL P)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
CHROMITE F31	A	0.4414	0.4061	490.94	1.52	12.33	440	13.18
(Fe,Mg)(Cr,Al,Fe)2O4	C	0.3037	0.3103	478.49	3.71	12.42	460	13.01
CUBIC							480	12.86
REF. FOR R: (ENGIN,1969)							520	12.62
REMARKS: LOC. TURKEY							540	12.50
							560	12.40
							580	12.29
							600	12.20
							620	12.06
							640	12.05
							660	11.67

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
CHROMITE ZUS	A	0.4415	0.4070	493.35	1.46	12.31	440	12.94
(FE,MG)(CR,AL,FE) ₂ O ₄	C	0.3042	0.3120	481.59	2.61	12.29	460	12.94
CUBIC							480	12.82
REF. FOR R: (ENGIN,1969)							520	12.58
REMARKS: LOC. TURKEY							540	12.49
							560	12.42
							580	12.28
							600	12.16
							620	12.15
							640	11.72
							660	11.47

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CONF.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
CHROMITE AI	A	0.4421	0.4062	490.76	1.36	12.27		440	13.06
(FE,MG)(CR,AL,FE)2 O4	C	0.3044	0.3108	478.05	2.70	12.35		460	12.79
CUBIC								480	12.73
REF. FOR R: (ENGIN,1969)								520	12.55
REMARKS: LOC. TURKEY								540	12.43
								560	12.30
								580	12.21
								600	12.17
								620	12.15
								640	11.94
								660	11.58

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
CHROMITE F7C	A	0.4420	0.4067	492.36	1.35	12.18	440	12.81
(FE,MG)(CR,AL,FE)2 O4	C	0.3044	0.3118	480.72	2.57	12.26	460	12.77
CUBIC							480	12.69
REF. FOR R: (ENGIN,1969)							520	12.47
REMARKS: LOC. TURKEY							540	12.33
							560	12.24
							580	12.15
							600	12.08
							620	11.84
							640	11.92
							660	11.64

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
CHROMITE AU12	A	0.4417	0.4066	492.19	1.43	12.07		440	12.73
(FE,MG)(CR,AL,FE)2 O4	C	0.3042	0.3114	480.13	2.69	12.15		460	12.72
CUBIC								480	12.60
REF. FOR R: (ENGIN,1969)								520	12.35
REMARKS: LCC. TURKEY								540	12.21
								560	12.10
								580	12.12
								600	11.91
								620	11.94
								640	11.57
								660	11.33

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
CHROMITE ZU8	A	0.4427	0.4063	490.61	1.21	12.04	440	12.65
(FE,MG)(CR,AL,FE)2 O4	C	0.3049	0.3117	479.31	2.41	12.11	460	12.63
CUBIC							480	12.51
REF. FOR R: (ENGIN,1969)							520	12.31
REMARKS: LCC. TURKEY							540	12.15
							560	12.08
							580	11.99
							600	11.93
							620	11.81
							640	11.85
							660	11.65

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES X	CHROMATICITY COORDINATES Y	DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
CHROMITE ZG44	A	0.4424	0.4057	488.90	1.31	11.98	440	12.74
(FE,MG)(CR,AL,FE)2 O4	C	0.3043	0.3104	477.40	2.78	12.04	460	12.69
CUBIC							480	12.42
REF. FOR R: (ENGIN,1969)							520	12.23
REMARKS: LCC. TURKEY							540	12.10
							560	12.03
							580	11.90
							600	11.81
							620	11.78
							640	11.70
							660	11.75

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
CHROMITE F19	A	0.4424	0.4064	491.19	1.28	11.57	440	12.19
(FE,MG)(CR,AL,FE)2 O4	C	0.3046	0.3116	479.77	2.52	11.64	460	12.09
CUBIC							480	12.04
REF. FOR R: (ENGIN, 1969)							520	11.81
REMARKS: LCC. TURKEY							540	11.75
							560	11.62
							580	11.49
							600	11.41
							620	11.33
							640	11.30
							660	11.33

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		COORDINATES X	Y			X	Y		
CHROMITE Fe2	A	0.4417	0.4061	490.74	1.45	11.46		440	12.25
(Fe,Mg)(Cr,Al,Fe)2 O4	C	0.3039	0.3104	478.26	2.92	11.54		460	12.04
CBIC								480	11.92
REF. FOR R: (ENGIN,1969)								520	11.73
REMARKS: LCC. TURKEY								540	11.61
								560	11.51
								580	11.42
								600	11.32
								620	11.34
								640	11.21
								660	10.74

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
CHROMITE FeS	A	0.4428	0.4068	492.31	1.16	10.73	440	11.28
(Fe,Mg)(Cr,Al,Fe)2O4	C	0.3052	0.3122	480.03	2.24	10.79	460	11.06
CUBIC							480	11.05
REF. FOR R: (ENGIN, 1969)							520	11.00
REMARKS: LCC. TURKEY							540	10.88
							560	10.73
							580	10.69
							600	10.63
							620	10.70
							640	10.37
							660	10.13

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
CHROMITE F68	A	0.4430	0.4063	490.33	1.14	10.54		440	11.12
(FE,MG)(CR,AL,FE)2 O4	C	0.3051	0.3115	478.13	2.36	10.59		460	10.92
CUBIC								480	10.88
REF. FOR R: (ENGIN,1969)								520	10.74
REMARKS: LCC. TURKEY								540	10.64
								560	10.58
								580	10.49
								600	10.43
								620	10.34
								640	10.42
								660	10.23

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
ZVYAGINTSEVITE	A	0.4550	0.4102	586.37	7.00	67.18		450	61.40
(PD,PT)3(PD,SN)	C	0.3189	0.3253	577.47	4.74	66.63		550	66.80
CUBIC								580	67.60
								650	69.40

REMARKS: LOC: UNKNOWN

REF. FOR R L.VYALSOV, MOSCOW

COPPER	A	0.5055	0.4705	596.77	35.20	57.58		470	38.40
CU	C	0.3739	0.3412	590.56	23.78	53.18		546	42.60
CUBIC								589	68.20
								650	78.40

REMARKS: GEORGETOWN, USA. CU=99.7%

REF. FOR R CEFVELLE & CAYE, 1968

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
SPERRYLITE	A	0.4448	0.4793	506.77	0.63	54.91		460	55.00
PT AS2	C	0.3085	0.3173	498.58	0.54	55.10		540	55.50
CUBIC								580	55.50
								660	52.00

REMARKS: LOC: UNKNOWN

REF. FOR R L.VYALSOV, MOWSCOW

CLAUSTHALITE	A	0.4353	0.4016	486.30	3.19	49.08		470	55.40
PB SE	C	0.2960	0.3003	475.17	7.01	49.74		546	49.50
CUBIC								589	48.10
								650	47.10

REMARKS: LOC: EAGLE GROUP, CANADA

REF. FOR R BURKE, FREE UNIV., AMSTERDAM

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL. I WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
SKUTTERUDITE	A	0.4462	0.4072	491.14	0.34	53.20		470	53.80
(CO,FE,NI)AS ₃ -X	C	0.3086	0.3150	479.58	0.69	53.29		546	53.40
CUBIC								589	53.00
								650	52.90

REMARKS: LOC: JACHYMOV, CZECHOSLOVAKIA

REF. FOR R BURKE, FREE UNIV., AMSTERDAM

HOLLINGWORTHITE	A	0.4451	0.4197	512.14	0.59	51.71		460	51.50
(RH,PD)AS S	C	0.3089	0.3182	512.66	0.41	51.89		540	52.50
CUBIC								580	52.00
								640	50.00

REMARKS: LOC: UNKNOWN

REF. FOR R L. VYALSOV, MOSCOW

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
RU-HOLLINGWORTHITE	A	0.4459	0.4075	494.79	0.40	49.04		455	49.60
(RH,RU,PD)AS S	C	0.3085	0.3153	483.04	0.69	49.14		520	49.40
CUBIC								560	49.20
								650	48.40

REMARKS: LOC: UNKNOWN

REF. FOR R L.VYALSOV, MOSCOW

GERSDORFFITE	A	0.4490	0.4076	591.24	0.04	48.41		470	47.90
(NI,CO,FE)AS S	C	0.3116	0.3171	586.37	0.62	48.32		546	48.10
CUBIC								589	48.80
								650	48.50

REMARKS: LOC: SUDBURY, ONTARIO CANADA

REF. FOR R BURKE, FREE UNIV., AMSTERDAM

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
GERSDORFFITE	A	0.4490	0.4091	580.33	2.78	46.42		470	45.40
(NI,CO,FE)AS S	C	0.3123	0.3198	571.47	1.53	46.37		546	46.70
CUBIC								589	46.30
								650	46.60

REMARKS: LOC: UNKNCWN

REF. FOR R UYTENBOGAARDT & BURKE,1971

GERSDORFFITE	A	0.4488	0.4089	580.22	1.80	53.64		470	52.60
(NI,CO,FE)AS S	C	0.3120	0.3192	571.93	1.29	53.58		546	53.80
CUBIC								589	53.80
								650	53.40

REMARKS: LOC: UNKNCWN

REF. FOR R UYTENBOGAARDT & BURKE,1971

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL P) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
POLYDYMITE	A	0.4500	0.4786	590.86	9.34	47.12	470	42.90
NI3 S4	C	0.3230	0.3261	581.44	6.09	46.45	546	46.00
CUBIC							589	47.30
							650	51.70

REMARKS: LOC: DRY NICKEL MINE, S. RHODESIA

REF. FOR R BURKE, FREE UNIV., AMSTERDAM

IRARSITE	A	0.4426	0.4071	493.43	1.19	46.74	455	48.60
(IR, RU, PH, PT) AS S	C	0.3051	0.3127	481.72	2.21	47.02	520	47.80
CUBIC							560	47.00
							650	45.20

REMARKS: LOC: UNKNOWN

REF. FOR R L. VYALSOV, MOSCOW

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
IRARSITE	A	0.4436	0.4084	498.69	1.92	46.55		490	47.60
IR,RU,RH,PT)AS S	C	0.3767	0.3157	486.82	1.31	46.80		520	47.80
CUBIC								589	46.10
								650	45.20

REMARKS: LCC: UNKNCWN

REF. FOR R UYTENBOGAARDT & BURKE, 1971

TYRRELITE	A	0.4527	0.4087	588.18	4.37	46.01		470	44.00
(NI,CO,CU)3SE4	C	0.3158	0.3212	579.71	2.84	45.73		546	45.50
CUBIC								589	46.50
								650	47.10

REMARKS: EAGLE GROUP, CANADA

REF. FOR R BURKE, FREE UNIV., AMSTERDAM

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
PENTLANDITE	A	0.4647	0.4101	589.62	13.65	45.84	470	39.80
(NI,FE)9 S8	C	0.3290	0.3314	580.72	9.12	44.93	546	44.30
CUBIC							589	47.00
							650	50.50

REMARKS: LOC: FLAT NICKEL MINE,TELEMARK NORWAY

REF. FOR R BURKE, FREE UNIV.,AMSTERDAM

ULLMANITE	A	0.4460	0.4040	577.05	1.00	44.97	470	46.70
NI SB S	C	0.3067	0.3099	453.25	2.17	45.02	546	44.60
CUBIC							589	44.50
							650	46.10

REMARKS: LOC: EISENHARTER,SAXONY

REF. FOR R BURKE, FREE UNIV. ,AMSTERDAM

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
LAURITE (1) PURE	A	0.4294	0.4027	490.28	4.51	41.25		470	47.80
RU S2	C	0.2914	0.2989	478.60	8.77	42.13		546	42.50
CUBIC								589	40.30
								650	37.20

REMARKS: LOC: GOODNEWS BAY, ALASKA. RU=61.0; IR=1.0; S=38.0 WT.%

REF. FOR R (LEONARD ET AL, 1969)

LAUPITE (2) NUGGET	A	0.4315	0.4130	490.28	4.24	40.94		470	47.00
RU S2	C	0.2925	0.2999	478.58	8.26	41.74		546	42.10
CUBIC								589	40.00
								650	37.20

REMARKS: LOC: GOODNEWS BAY, ALASKA. RU=40.5; IR=13.0; S=31.1 WT.%

REF. FOR R (LEONARD ET AL, 1969)

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
LAURITE (3)	A	0.4243	0.3954	485.52	6.11	37.98		470	47.70
RU S2	C	0.2840	0.2860	474.69	13.08	38.95		546	38.50
CUBIC								589	36.40
								650	35.40

REMARKS: LOC: GOODNEWS BAY, ALASKA. RU=59.5; IR=2.0; S=36.7 WT.%

REF. FOR R (LEONARD ET AL, 1969)

LAURITE (4)	A	0.4274	0.3960	484.52	5.36	39.06		470	48.00
RU S2	C	0.2870	0.2880	473.47	11.80	39.90		546	39.00
CUBIC								589	38.30
								650	36.30

REMARKS: LOC: STILLWATER COMPLEX, MONTANA

REF. FOR R (LEONARD ET AL, 1969)

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
LAURITE (5)	A	0.4344	0.4037	489.85	3.29	41.00	470	45.80
RU S2	C	0.2962	0.3029	478.07	6.58	41.61	546	41.80
CUBIC							589	40.30
							650	38.20

REMARKS: LOC: BORNEO. RU=30.0; IR=20.0; S=27.0 WT.%

REF. FOR R (LEONARD ET AL, 1969)

PETZITE	A	0.4427	0.4052	486.50	1.27	40.32	460	42.60
AG3 AU TE2	C	0.3044	0.3096	474.42	2.87	40.52	540	40.50
CUBIC							580	40.20
							640	39.60

REMARKS: LOC: UNKNOWN

REF. FOR R L.VYALSOV, MOSCOW

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
PETZITE	A	0.4419	0.4074	494.49	1.35	38.55		460	40.10
AG3 AU TE2	C	0.3049	0.3127	482.29	2.31	38.91		540	39.10
CUBIC								580	38.70
								640	37.00

REMARKS: LOC: UNKNOWN

REF. FOR R L.VYALSOV, MOSCOW

PETZITE	A	0.4422	0.4054	488.02	1.37	38.10		460	40.30
AG3 AU TE2	C	0.3039	0.3098	476.65	3.00	38.33		540	38.50
CUBIC								580	37.80
								640	37.40

REMARKS: LOC: UNKNOWN

REF. FOR R P.BEISTEIN, MOSCOW

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
PETZITE	A	0.4276	0.3994	487.72	5.11	35.83		460	42.10
AG3 AU TE2	C	0.2884	0.2933	475.46	10.55	36.63		540	37.10
CUBIC								580	34.90
								640	33.30

REMARKS: LOC: UNKNOWN

REF. FOR R UYTENBOGAARDT & BURKE, 1971

COLORADOITE	A	0.4500	0.4078	590.12	1.87	37.89		460	37.10
HG TE	C	0.3132	0.3175	590.66	1.15	37.73		540	36.80
CUBIC								580	39.20
								640	37.40

REMARKS: LOC: UNKNOWN

REF. FOR R L.VYALSOV, MOSCOW

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
COLORADOITE	A	0.4504	0.4096	582.68	3.47	36.24		460	34.80
HG TE	C	0.3144	0.3207	577.46	2.32	36.10		540	35.50
CUBIC								580	37.50
								640	35.50

REMARKS: LOC: UNKNOWN

REF. FOR R P. BEISTEIN, MOSCOW

COLORADOITE	A	0.4433	0.4012	579.25	7.50	34.31		460	36.60
HG TE	C	0.3028	0.3037	577.37	5.34	34.43		540	33.90
CUBIC								580	33.90
								640	34.80

REMARKS: LOC: UNKNOWN

REF. FOR R UYTENBOGAARDT & BURKE, 1971

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
MURDOCHITE	A	0.4533	0.4787	588.59	4.78	17.31		470	16.50
CU6 PB 08	C	0.3163	0.3216	579.78	3.09	17.20		546	17.10
CUBIC								589	17.50
								650	17.80

REMARKS: LOC: TCHAH-KHCUNI MINE, IRAN

REF. FOR R BURKE, FREE UNIV., AMSTERDAM

MURDOCHITE	A	0.4534	0.4792	587.33	5.20	17.19		470	16.30
CU6 PB 08	C	0.3168	0.3224	578.90	3.43	17.07		546	17.00
CUBIC								589	17.40
								650	17.60

REMARKS: LOC: UNKNOWN

REF. FOR R UYTENBOGAARDT & BURKE, 1971

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y				Y		
TITANOMAGNETITE	A	0.4469	0.4193	563.76	1.47	16.93		470	16.60
FE1+X,FE2-2X,TIX O4	C	0.3112	0.3201	563.11	1.29	16.94		546	17.00
CUBIC								589	17.30
								650	16.00

REMARKS: LOC: UNKEL, RHENANIE. FE2 O3=84.5%; TI O2=15.3%

REF. FOR R CERVELLE, 1967

2. ANISOTROPIC MINERALS

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (Rp) (Rg)	
		X	Y			Y				
KCSTOVITE	A	0.4482	0.4130	572.52	4.24	53.22	γ_p	433	49.30	50.10
AL CU TE4	A	0.4508	0.4146	576.98	7.16	57.90	γ_g	465	49.70	53.70
PROBABLY MGNO.	C	0.3136	0.3255	565.54	3.41	53.26	γ_p	482	52.10	55.20
	C	0.3172	0.3302	568.84	5.64	57.79	γ_g	518	52.70	56.40
REF. FOR R: (TERZEIV,1966)								559	54.90	60.10
REMARKS: LOC. CHELCPECH,BULGARIA								589	53.00	57.90
								624	52.50	57.00
								668	49.30	55.20
								691	48.40	53.30

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMECA (NM)	REFLECTIVITY (Rp) (Rg)	
		X	Y			Y				
MARCASITE NGC 9(1)	A	0.4535	0.4127	581.77	7.70	50.84	Yp	440	44.98	45.92
FE S2	A	0.4534	0.4137	580.69	8.32	53.18	Yg	460	45.58	48.13
ORTHO-RHOMBIC	C	0.3189	0.3285	573.17	5.64	50.55	Yp	480	47.72	50.00
	C	0.3195	0.3306	571.98	6.36	52.91	Yg	500	48.53	51.15
REF. FOR R: (TUGAL, 1969)								520	49.54	51.88
REMARKS: LOC. TURKEY								540	50.56	52.86
								560	51.48	53.56
								580	51.63	54.88
								600	51.40	53.25
								620	50.97	52.41
								640	50.91	52.84
								660	50.17	51.60

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY		
		X	Y			Y	LAMBDA (NM)	(R _p)	(R _g)	
MARCASITE NGC 9(2)	A	0.4496	0.4121	576.81	4.58	49.55	Y _p	440	45.76	45.45
FE S2	A	0.4517	0.4128	579.67	6.52	51.06	Y _g	460	46.41	47.09
ORTHO RHOMBIC	C	0.3143	0.3251	568.05	3.49	49.51	Y _p	480	47.82	48.80
	C	0.3170	0.3280	570.59	4.99	50.90	Y _g	500	48.61	49.64
REF. FOR R: (TUGAL, 1969)								520	49.30	50.32
REMARKS: LCC, TURKEY								540	49.90	51.17
								560	50.52	51.72
								580	50.22	51.63
								600	48.16	51.24
								620	49.44	50.97
								640	49.62	50.27
								660	48.73	50.17

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y			(R _p)	(R _g)
MARCASITE KC 16	A	0.4532	0.4121	582.18	7.08	46.63	Y _p	440	42.01	46.20
FE S2	A	0.4514	0.4140	578.22	7.15	52.84	Y _g	460	42.56	48.13
ORTHOHOMBIC	C	0.3183	0.3269	574.10	5.05	46.37	Y _p	480	43.59	50.61
	C	0.3173	0.3300	569.18	5.61	52.74	Y _g	500	44.38	51.64
REF. FOR R: (TUGAL,1969)								520	45.23	52.35
REMARKS: LCC. TURKEY								540	46.29	53.16
								560	47.48	53.78
								580	47.71	53.78
								600	45.91	51.49
								620	47.70	53.10
								640	47.69	53.49
								660	45.41	50.91

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y				
NICCOLITE 1	A	0.4815	0.4134	589.22	27.58	51.29	Y _p	420	35.96	42.96
NI AS	A	0.4703	0.4113	589.34	18.36	54.52	Y _g	440	35.04	42.60
HEXAGONAL	C	0.3507	0.3477	581.18	19.32	49.22	Y _p	460	35.36	43.27
	C	0.3360	0.3477	580.99	12.39	53.06	Y _g	480	37.33	45.14
REF. FOR R: (THE AUTHCR)								500	39.94	46.79
REMARKS: LCC. ONTARIO, CANADA								520	43.30	48.93
Electron Probe Analysis:								540	47.01	51.45
								560	50.26	53.67
								580	53.08	55.58
								600	55.60	57.54
								620	57.64	59.10
								640	59.15	60.39

O. C. ... ?

Wt. %
 Ni 44.07
 As 55.12
 Sb 00.90

 100.09

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y				
LOCELLINGITE	A	0.4500	0.4110	579.05	4.10	50.20	Y _p	440	46.90	43.80
FE AS ₂	A	0.4584	0.4145	583.48	12.34	53.22	Y _g	480	48.30	47.40
ORTHOHOMBIC	C	0.3142	0.3236	569.99	3.06	50.12	Y _p	500	49.50	50.10
	C	0.3247	0.3352	574.08	8.99	52.66	Y _g	520	50.00	51.30
REF. FOR R: O. VAASJOKI (1969)								540	50.60	53.00
REMARKS: LCC. UNKNOWN								600	50.20	54.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _α) (R _β)	
		X	Y			Y			(R _α)	(R _β)
SYLVANITE, FIJI	A	0.4580	0.4130	584.55	11.02	50.71	Y _α	440	42.00	56.30
AU AG TE4	A	0.4518	0.4101	583.02	4.37	60.80	Y _β	460	44.60	57.50
MONOCLINIC	C	0.3233	0.3334	574.07	8.13	50.24	Y _α	480	47.00	59.10
	C	0.3148	0.3237	571.53	3.25	60.65	Y _β	500	48.40	60.60
REF. FOR R: (STUMFEL, 1970)								546	50.80	61.20
REMARKS: LGC. EMPEROR MINE, FIJI								589	50.00	59.50
Electron Probe Analysis: Wt. %								600	51.50	61.10
Au 31.4								620	52.30	61.60
Ag 6.6										
Te 59.9										

MINERAL NAME
 COMPOSITION
 CRYSTAL SYSTEM

CHROMATICITY
 COORDINATES
 X Y

DOMINANT
 (COMPL.)
 WAVELENGTH

PURITY
 (P%)

BRIGHTNESS
 (VISUAL R)
 Y

LAMBDA
 (NM)

REFLECTIVITY
 R

CHALCOPYRITE, E	A	0.4724	0.4262	582.65	30.16	47.62	440	26.22
CU FE S2	C	0.3484	0.3692	573.22	24.47	46.55	460	31.80
TETRAGONAL							480	37.00
REF. FOR R: (TUGAL, 1969)							500	41.23
REMARKS: LOC. TURKEY							520	44.05
							540	46.38
							560	48.41
							580	49.29
							600	49.52
							620	49.67
							640	49.09
							660	48.16

0.12.

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y				
CHALCOPYRITE 4	A	0.4706	0.4243	532.82	27.61	47.61	Y _p	440	27.63	27.63
CU FE S ₂	A	0.4720	0.4242	533.17	28.51	47.75	Y _g	460	32.66	33.26
TETRAGONAL	C	0.3448	0.3643	573.20	22.19	46.63	Y _p	480	37.75	38.47
	C	0.3463	0.3649	573.95	22.75	46.67	Y _g	500	41.62	42.05
REF. FOR R: TUGAL (1969)								520	44.21	44.18
REMARKS: LOC. TURKEY								540	46.50	46.55
								560	48.11	48.11
								580	49.23	49.14
								600	49.75	49.08
								620	49.53	49.84
								640	49.80	49.17
								660	50.17	48.73

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
CHALCOPYRITE, NG39	A	0.4711	0.4259	582.45	29.05	47.37		440	26.53
CU FE S2	C	0.3467	0.3677	573.00	23.62	46.38		460	32.33
TETRAGONAL								480	37.47
REF. FOR R: (TUGAL, 1969)								500	41.47
REMARKS: LOC. TURKEY								520	44.05
								520	44.05
								540	46.18
								560	47.95
								580	49.19
								600	49.35
								620	49.46
								640	49.09
								660	46.16

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
CHALCOPYRITE, M2	A	0.4721	0.4257	582.73	29.61	47.29	440	26.39
CU FE S2	C	0.3476	0.3682	573.19	23.99	46.25	460	31.74
TETRAGONAL							480	37.06
REF. FOR R: (TUGAL, 1969)							500	41.06
REMARKS: LOC. TURKEY							520	44.10
							520	44.10
							540	46.13
							560	47.77
							580	48.91
							600	49.05
							620	49.14
							640	49.62
							660	48.17

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
CHALCOPYRITE, 3%	A	0.4725	0.4248	583.07	29.28	47.22	440	26.68
CU FE S2	C	0.3475	0.3666	573.56	23.53	46.13	460	32.03
TETRAGONAL							480	37.26
REF. FOR R: (TUGAL, 1969)							500	40.89
REMARKS: LOC. TURKEY							520	43.59
							540	45.71
							560	47.72
							580	49.04
							600	49.19
							620	48.88
							640	48.60
							660	49.91

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
CHALCOPYRITE, M3	A	0.4733	0.4260	582.90	30.67	47.16	440	25.77
CU FE S2	C	0.3493	0.3696	573.42	24.82	46.06	460	31.32
TETRAGONAL							480	36.47
REF. FOR R: (TUGAL, 1969)							500	40.72
REMARKS: LOC. TURKEY							520	43.44
							520	43.44
							540	45.90
							560	47.77
							580	48.73
							600	49.20
							620	49.14
							640	49.62
							660	48.17

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (C.M.P.L.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
CHALCOPYRITE, M12	A	0.4731	0.4256	582.97	30.26	47.14		440	25.92
CU FE S2	C	0.3489	0.3688	573.48	24.50	46.05		460	31.57
TETRAGONAL								480	36.74
REF. FOR R: (TUGAL 1969)								500	40.76
REMARKS: LOC. TURKEY								520	43.40
								520	43.40
								540	45.90
								560	47.59
								580	48.82
								600	49.20
								620	49.14
								640	49.62
								660	48.17

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		X	Y					(R _g)	(R _p)
CHALCOPYRITE M24	A	0.4722	0.4260	582.67	29.88	46.59 Yp	440	25.92	26.08
CU FE S2	A	0.4730	0.4262	582.78	30.58	47.35 Yg	460	31.32	31.49
TETRAGONAL	C	0.3481	0.3685	573.29	24.21	45.94 Yp	480	36.58	36.68
	C	0.3491	0.3697	573.33	24.80	46.26 Yg	500	40.89	40.80
REF. FOR R: TUGAL (1969)							520	43.64	43.36
REMARKS: LCC. TURKEY							540	46.32	45.71
							560	47.89	47.65
							580	48.91	48.73
							600	49.51	49.20
							620	49.14	48.83
							640	49.62	48.98
							660	48.17	46.78

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
CHALCOPYRITE, K	A	0.4713	0.4252	582.71	28.71	46.87		440	26.67
CU FE S2	C	0.2462	0.3664	573.19	23.16	45.87		460	32.05
TETRAGONAL								480	37.21
REF. FOR R: (TUGAL, 1969)								500	40.92
REMARKS: LOC. TURKEY								520	43.53
								520	43.53
								540	45.74
								560	47.48
								580	48.52
								600	48.60
								620	48.48
								640	48.47
								660	48.16

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
CHALCOPYRITE, M1	A	0.4736	0.4262	582.90	31.02	46.77	440	25.26
CU FE S2	C	0.3499	0.3705	573.40	25.22	45.66	460	30.61
TETRAGONAL							480	36.23
REF. FOR R: (TUGAL, 1969)							500	40.42
REMARKS: LOC. TURKEY							520	43.06
							520	43.06
							540	45.50
							560	47.33
							580	48.32
							600	48.87
							620	48.81
							640	48.72
							660	48.17

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
CHALCOPYRITE NG31E	A	0.4748	0.4264	583.09	31.99	46.58		440	24.78
CU FE S2	C	0.2513	0.3718	573.54	25.95	45.42		460	29.94
(TETRAGONAL								480	35.64
REF. FOR R: (TUGAL, 1969)								500	39.88
REMARKS: LCC. TURKEY								520	42.73
								520	42.73
								540	45.02
								560	47.51
								580	48.03
								600	48.57
								620	48.33
								640	48.84
								660	49.54

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
CHALCOPYRITE, NG35	A	0.4749	0.4254	583.41	31.34	46.49	440	25.03
CU FE S2	C	0.3509	0.3700	573.85	25.36	45.22	460	30.69
TETRAGONAL							480	35.79
REF. FOR R: (TUGAL, 1969)							500	39.89
REMARKS: LOC. TURKEY							520	42.70
							520	42.70
							540	44.86
							560	46.74
							580	48.08
							600	48.75
							620	49.17
							640	49.09
							660	48.16

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
CHALCOPYRITE, 2C	A	0.4721	0.4253	582.85	29.34	46.48	440	26.08
Cu Fe S2	C	0.3474	0.3673	573.35	23.69	45.45	460	31.49
TETRAGONAL							480	36.63
REF. FOR R: (TUGAL, 1969)							500	40.43
REMARKS: LOC. TURKEY							520	43.03
							520	43.03
							540	45.29
							560	47.17
							580	47.76
							600	48.27
							620	48.83
							640	48.98
							660	46.78

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (%)	BRIGHTNESS		REFLECTIVITY R
		X	Y			(VISUAL R) Y	LAMBDA (NM)	
CHALCOPYRITE, NG31A	A	0.4745	0.4258	583.21	31.35	46.40	440	24.92
Cu Fe S2	C	0.3506	0.3705	573.63	25.41	45.25	460	30.51
TETRAAGONAL							480	35.69
REF. FOR R: (TUGAL, 1969)							500	39.88
REMARKS: LOC. TURKEY							520	42.61
							520	42.61
							540	44.98
							560	46.94
							580	47.86
							600	48.57
							620	48.05
							640	48.48
							660	49.54

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES X Y	DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
CHALCOPYRITE 43-46	A	0.4809 0.4288	583.55	37.86	43.60	440	19.97
CU FE S2	C	0.3609 0.3823	574.03	31.35	42.25	460	25.47
TETRACONAL						480	31.16
REF. FOR R: (TUGAL, 1969)						500	35.98
REMARKS: LCC. TURKEY						520	39.24
						520	39.24
						540	41.81
						560	43.77
						580	45.54
						600	46.17
						620	46.04
						640	46.61
						660	46.78

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CORNPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY		
		X	Y			Y	LAMPDA (NM)	(R _p)	(R _g)	
GROSSULARITE	A	0.4445	0.4134	549.02	2.73	37.14	Yp	450	35.50	39.00
ZEPHYRINE	A	0.4441	0.4132	542.25	1.74	40.67	Yg	472	36.00	39.50
MONOCLINIC	C	0.3101	0.3240	551.31	1.88	37.34	Yp	496	37.00	40.80
	C	0.3096	0.3237	547.48	1.88	40.91	Yg	527	38.30	42.00
REF. FOR R: NADEZDA N. MCZGICVA ET AL, IMM78 (1969)								550	38.00	40.60
REMARKS: LCC. UNKNOWN								579	37.70	40.20
								608	36.00	39.50
								640	35.00	38.30
								670	33.80	37.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
GEOCRONITE	A	0.4448	0.4151	559.80	3.37	40.59		470	39.00
27PB S 7 (SE, AS) 2 S3	C	0.3111	0.3276	555.39	3.22	40.83		520	42.00
MONOCLINIC								575	41.50
REF. FOR R: GRAY & MILLMAN (1962)								600	39.00
REMARKS: LCC. UNKNOWN; UNORIENTED SECTION.								700	37.90

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CNFL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		WAVELENGTH (NM)	REFLECTIVITY (Rp) (Rg)	
		X	Y			Y				
BCULANGERITE	A	0.4424	0.4130	527.56	1.24	36.46	Yp	450	35.30	39.70
SPB S 2SB? SB	A	0.4426	0.4122	528.66	1.20	40.36	Yg	472	36.00	39.80
MONOCLINIC	C	0.3087	0.3228	537.71	1.44	36.70	Yp	496	34.50	40.70
	C	0.3076	0.3208	514.63	0.91	40.66	Yg	527	37.00	42.50
REF. FOR R: NADEZDA N. MCZGICVA ET AL, IMM78(1969)								550	37.50	41.50
REMARKS: LCC. UNKNOWN; RP=NORMAL TO C, RG=PARALLEL TO C								579	36.50	40.40
								608	35.60	39.30
								640	34.30	38.00
								670	32.50	36.50

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CENL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
ECULANGERITE	A	0.4393	0.4130	507.19	1.88	36.49	470	37.00
SPB S 2SB2 S3	C	0.3046	0.3209	501.33	1.81	36.95	520	38.70
MONOCLINIC							575	37.00
REF. FOR R: GRAY & MILLMAN (1962)							600	34.50
REMARKS: LCC. UNKNOWN; UNORIENTED SECTION.							700	33.60

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y	Y		(R _p)	(R _g)
HETEROMORPHITE	A	0.4441	0.4110	515.21	0.86	36.21	Y _p	450	35.70	39.20
7PB S 4SB2 S3	A	0.4435	0.4120	518.42	1.06	40.17	Y _g	472	35.90	39.80
MONOCLINIC	C	0.3086	0.3198	521.81	0.68	36.39	Y _p	496	36.30	40.30
	C	0.3093	0.3213	526.79	0.59	40.42	Y _g	527	37.00	41.80
REF. FOR R: NADEZDA N. MCZGOWA ET AL, IMM7E(1969)								550	37.00	41.00
REMARKS: LCC. UNKNOWN								579	36.50	40.30
								608	35.40	39.10
								640	34.40	38.00
								670	34.20	37.70

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY		
		X	Y			Y	LAMBDA (NM)	(R _p)	(R _g)	
MENEGHINITE	A	0.4428	0.4083	497.66	2.11	35.48	Y _p	450	36.30	40.50
CU ₂ S.26PB S 7SB2 S3	A	0.4424	0.4090	497.82	2.18	39.76	Y _g	472	36.40	40.90
ORTHO-RHOMBIC	C	0.3058	0.3151	497.55	1.70	35.71	Y _p	494	36.70	41.00
	C	0.2057	0.3161	497.42	1.65	40.05	Y _g	527	36.40	41.00
REF. FOR R: NALFZEA M. NEZGOVA ET AL, IMM78 (1969)								550	36.00	40.50
REMARKS: LCC. UNKNOWN; RP=NORMAL TO C, RG=PARALLEL TO C								570	35.20	39.40
								608	34.80	39.00
								640	34.40	38.20
								670	34.20	38.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY %
		X	Y				Y		
MENEGHINITE	A	0.4372	0.4075	494.94	2.45	37.58		470	40.20
CU ₂ S.26PB S 7SB2 S3	C	0.2001	0.3110	484.32	4.20	38.08		520	39.90
ORTHOHRHOMBIC								575	37.00
REF. FOR R: GRAY & MILLMAN (1962)								600	36.00
REMARKS: LCC. UNKNOWN; UNORIENTED SECTION.								700	35.50

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL %)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y	Y		(R _p)	(R _g)
ZINKENITE	A	0.4416	0.4107	514.89	1.31	35.43	Y _p	450	35.60	38.40
FE S SR2S3	A	0.4417	0.4115	507.52	1.34	38.51	Y _g	472	36.00	38.90
HEXAGONAL	C	0.3062	0.3182	496.85	1.23	35.72	Y _p	496	36.30	39.20
	C	0.3064	0.3194	511.35	1.21	38.84	Y _g	527	36.50	40.10
REF. FOR R: NADEZDA N. MCZGOWA ET AL, IMM78 (1969)								550	36.50	39.70
REMARKS: LCC. UNKNOWN; RP=NORMAL TO C, RG=PARALLEL TO C								579	35.20	38.20
								608	34.80	37.70
								640	33.80	36.60
								670	31.00	33.70

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y	Y		(R _p)	(R _g)
ZINCFENITE	A	0.4427	0.4155	541.51	2.37	38.48	Y _p	470	37.50	38.50
FeS ₂ SE	A	0.4474	0.4165	570.01	5.71	40.95	Y _g	520	41.00	42.70
HEXAGONAL	C	0.2089	0.3277	545.21	2.80	38.85	Y _p	575	38.70	41.20
	C	0.3138	0.3215	561.36	5.79	41.12	Y _g	600	36.50	39.40
REF. FOR R: GRAY & MILLMAN (1962)								700	36.60	40.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CORNL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (Rp) (Rg)	
		X	Y			Y	Y			
JAMESONITE	A	0.4373	0.4214	500.50	2.73	34.95	Yp	450	36.30	39.60
4PR S.FE S SR2 SR	A	0.4356	0.4213	502.32	1.80	39.05	Yg	472	36.30	39.70
MONOCLINIC	C	0.3021	0.3168	497.72	2.95	35.44	Yp	496	36.50	40.40
	C	0.3040	0.3181	494.50	2.04	39.49	Yg	527	37.00	41.20
REF. FOR R: NADEZDA N. MCZGLOVA ET AL, IMM78(1969)								550	36.50	40.60
REMARKS: LCC. UNKNOWN; RP=NORMAL TO C, RG=PARALLEL TO C								579	34.70	38.70
								608	33.00	36.90
								640	32.50	36.30
								670	31.50	35.20

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		WAVELENGTH (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			X	Y		(R _p)	(R _g)
JAMESONITE	A	0.4344	0.4095	497.40	3.05	35.13	Y _p	470	37.70	29.70
4PB S FE S SR2 S3	A	0.4425	0.4120	512.20	1.20	39.76	Y _g	520	38.00	41.00
MONOCLINIC	C	0.2985	0.3127	487.25	4.61	35.75	Y _p	575	35.00	40.60
	C	0.3073	0.3207	531.49	0.96	40.06	Y _g	600	33.30	38.20
REF. FOR R: GRAY & MILLMAN (1962)								700	31.20	27.20

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL P) Y		WAVELENGTH (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y _p	Y _g		(R _p)	(R _g)
PLAGIOLITE	A	0.4425	0.4121	502.69	1.21	33.67	Y _p	450	33.00	37.20
5PR.S 4SR2 S3	A	0.4408	0.4113	505.02	1.53	37.03	Y _g	472	33.80	37.60
MONOCLINIC	C	0.3074	0.3214	505.85	1.72	33.94	Y _p	496	34.60	38.30
	C	0.3056	0.3186	497.25	1.53	37.37	Y _g	527	35.00	38.20
REF. FOR R: NADEZDA N. MCZGICVA ET AL, IMM78(1969)								550	34.50	38.00
REMARKS: LCC. UNKNOWN								579	33.60	37.50
								608	32.80	35.70
								640	31.60	34.50
								670	30.50	33.30

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _w) (R _E)	
		X	Y			X	Y		(R _w)	(R _E)
HEXAGONAL PYRROPHOTITE	A	0.4632	0.4097	599.82	12.34	35.14	Y _w	470	30.90	36.20
FE(1-X) S	A	0.4591	0.4102	599.19	9.94	40.04	Y _E	486	31.50	36.90
HEXAGONAL	C	0.3272	0.3299	580.79	8.24	34.50	Y _w	546	34.00	39.20
	C	0.3232	0.3281	579.05	6.68	39.52	Y _E	589	35.80	40.70
REF. FOR R: VON GEHLEN & PILLER (1964)								650	38.60	42.50
REMARKS: LCC. TREPČA, YUGOSLAVIA;								656	38.90	42.70

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y					(R _p)	(R _g)
HODRUSHITE	A	0.4561	0.4104	586.67	7.90	32.15 Y _p	420	26.20	27.00
CU(8.12)BI(11.54)FE(0.29)S22	A	0.4541	0.4124	582.66	7.90	33.35 Y _g	440	28.30	29.50
MONOCLINIC	C	0.3205	0.3273	576.81	5.74	31.86 Y _p	460	29.50	30.20
	C	0.3197	0.3289	573.89	5.96	33.14 Y _g	480	30.70	31.80
REF. FOR R: (KCDERA ET AL,1970)							500	31.50	32.40
REMARKS: LCC.							520	31.60	32.70
							540	31.00	32.90
							560	31.40	32.80
							580	32.00	33.40
							600	34.60	36.00
							620	33.40	34.10

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y				
PYROLUSITE	A	0.4441	0.4067	450.85	0.86	32.42	Y _p	430	33.30	34.00
BETA-MN C2	A	0.4446	0.4072	453.09	0.72	33.73	Y _g	470	33.80	35.00
TETRAGONAL	C	0.3064	0.3131	475.61	1.70	32.55	Y _p	490	33.20	34.50
	C	0.3072	0.3145	483.10	1.25	33.86	Y _g	520	32.70	34.00
REF. FOR R: (NICHOL & PHILLIPS, 1965)								550	32.70	34.00
REMARKS: LCC. UNKNOWN								580	32.40	33.70
								610	31.90	33.30
								650	32.00	33.20

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES X Y	DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
ALPHA-MN C2-NSUTITE	A	0.4411 0.4086	497.71	1.50	30.79	430	31.70
MN1-x MNx C2-2x(OH)2x, x=.06-.07	C	0.3046 0.3144	486.71	2.20	31.04	470	32.00
HEXAGONAL						490	31.80
REF. FOR R: (NICHOL & PHILLIPS, 1964)						520	31.50
REMARKS: LOC. UNKNOWN						550	31.50
						550	31.50
						580	30.80
						610	30.00
						650	29.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y					(R _p)	(R _g)
							400	15.60	
ICAITE 4	A	0.4888	0.4234	586.10	39.60	31.63 _p	420	15.50	16.40
Cu ₃ FeS ₄	A	0.4775	0.4242	584.26	32.32	32.35 _g	440	16.00	18.30
PROBABLY TETRAGONAL	C	0.3673	0.3737	577.92	30.35	30.20 _p	460	17.80	20.80
	C	0.3529	0.3674	575.16	25.20	31.38 _g	480	19.80	22.90
REF. FOR R: (LEVY, 1966)							500	22.90	26.10
REMARKS: LCC. CERRC-FUEMUL-MENDOZA, ARGENTINA							520	25.90	29.00
							540	29.20	31.20
							560	32.20	33.10
							580	33.40	33.40
							600	33.90	33.20
							620	35.10	34.80
							640	35.90	35.10
						660	607.4	36.00	
						680			

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y						
ICAITE 5	A	0.4886	0.4240	585.90	39.88	31.49 Y _p	420	15.30	16.20
CU3 FE S4	A	0.4791	0.4238	584.66	33.14	31.73 Y _g	440	15.80	17.70
PROBABLY TETRAGONAL	C	0.3677	0.3743	577.84	30.65	30.07 Y _p	460	17.60	20.00
	C	0.3549	0.3676	575.73	25.79	30.69 Y _g	480	19.70	22.40
REF. FOR R: (LEVY, 1966)							500	22.60	25.20
REMARKS: LGC. CERRIC-HUENUL-MENDOZA, ARGENTINA							520	25.50	28.00
							540	29.10	30.40
							560	31.60	31.90
							580	33.60	33.20
							600	34.50	33.60
							620	35.00	33.90
							640	34.90	34.20

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		X	Y					(Rp)	(Rg)
ICAITE 6	A	0.4920	0.4203	587.39	39.65	28.01 ^{Yp}	420	13.80	17.60
CU3 FE S4	A	0.4737	0.4220	584.29	28.16	29.65 ^{Yg}	440	14.70	18.10
PROBABLY TETRAGONAL	C	0.3696	0.3697	579.10	30.29	26.59 ^{Yp}	460	16.00	20.30
	C	0.3468	0.3598	575.26	21.52	28.86 ^{Yg}	480	17.40	22.00
REF. FOR R: (LEVY, 1966)							500	19.60	24.80
REMARKS: LOC. CERRC-HUEMUL-MENCOZA, ARGENTINA							520	22.30	26.80
							540	25.10	28.60
							560	28.10	30.20
							580	29.80	30.60
							600	30.60	30.60
							620	32.00	31.70
							640	32.50	31.60

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y					(R _p)	(R _g)
IDAITE 2	A	0.5062	0.4097	592.01	42.12	25.85 Y _p	420	14.40	15.40
CU3 FE S4	A	0.4877	0.4129	589.93	31.52	26.55 Y _g	440	14.50	16.70
PROBABLY TETRAGONAL	C	0.3812	0.3587	584.82	30.44	23.96 Y _p	460	14.40	17.90
	C	0.3583	0.3526	581.56	22.67	25.30 Y _g	480	14.90	18.90
REF. FOR R: (LEVY,1966)							500	16.00	20.10
REMARKS: LCC. BANCAIRCUN, FRANCE							520	18.20	21.60
							540	21.30	23.80
							560	24.30	25.70
							580	27.50	27.50
							600	29.70	29.10
							620	33.30	30.50
							640	33.70	31.90

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y			(R _p)	(R _g)
ICAITE 1	A	0.5065	0.4055	592.10	42.18	25.72	Y _p	420	14.40	15.20
Cu ₃ FeS ₄	A	0.4883	0.4127	590.06	31.78	26.48	Y _g	440	14.40	16.60
PROBABLY TETRAGONAL	C	0.3816	0.3584	584.99	30.47	23.82	Y _p	460	14.30	17.80
	C	0.3589	0.3527	581.70	22.86	25.20	Y _g	480	14.80	18.80
REF. FOR R: (LEVY, 1966)								500	16.00	19.90
REMARKS: LCC. BANCAIROUN, FRANCE								520	18.00	21.40
								540	21.00	23.60
								560	24.20	25.60
								580	27.40	27.60
								600	29.70	29.00
								620	33.20	30.40
								640	33.50	31.90

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y					(R _p)	(R _g)
ICAITE 3	A	0.4882	0.4164	588.45	34.30	25.56 Y _p	420	14.90	14.90
CU3 FE S4	A	0.4843	0.4150	587.03	33.43	26.25 Y _g	440	15.20	15.70
PROBABLY TETRAGONAL	C	0.3615	0.3587	580.28	25.16	24.34 Y _p	460	16.10	16.40
	C	0.3583	0.3611	578.62	24.95	25.16 Y _g	480	17.00	17.90
REF. FOR R: (LEVY,1966)							500	18.90	20.00
REMARKS: LCC. SAINT-VERAN							520	20.80	21.90
							540	22.90	24.20
							560	25.00	26.00
							580	27.00	27.80
							600	28.40	28.50
							620	28.90	28.80
							640	29.60	29.60

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _g) (R _p)	
		X	Y			Y			R _g	R _p
CRANCE BERNITE 6	A	0.4911	0.4153	589.12	35.55	27.10	Y _g	420	14.80	15.70
CU(2+X)SN(1-X)FE S4	A	0.4769	0.4179	586.50	27.52	26.88	Y _p	440	15.70	16.85
TETRAGONAL	C	0.3644	0.3598	580.69	26.24	25.72	Y _p	460	16.80	18.90
	C	0.3483	0.3550	577.69	20.43	26.03	Y _g	480	18.10	20.60
REF. FOR R: (LEVY,1966)								500	20.00	22.60
REMARKS: LCC. VALLRY								520	21.90	23.80
								540	24.10	25.20
								560	26.20	26.50
								580	28.20	27.60
								600	30.30	28.90
								620	31.00	29.80
								640	32.40	29.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	WAVELENGTH (NM)	REFLECTIVITY (R _g) (R _p)	
		X	Y					(R _g)	(R _p)
CRANGE BERNITE 1	A	0.4916	0.4150	589.28	35.68	27.07 Y _g	420	14.70	15.70
CU(2+X)SN(1-X)FE S4	A	0.4774	0.4177	586.66	27.72	26.93 Y _p	440	15.60	16.85
TETRAGONAL	C	0.3649	0.3598	580.82	26.37	25.68 Y _p	460	16.70	18.90
	C	0.3487	0.3550	577.94	20.46	26.07 Y _g	480	18.20	20.60
REF. FOR R: (LEVY, 1966)							500	20.10	22.60
REMARKS: LOC. VALLRY							520	21.80	23.80
							540	24.00	25.20
							560	26.00	26.50
							580	28.20	27.60
							600	30.40	29.00
							620	31.10	30.00
							640	32.50	29.60

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y					(R _p)	(R _g)
CRANGE BORNITE 4	A	0.4940	0.4150	589.44	27.34	26.63 Y _p	420	15.50	15.70
CU(2+X)SN(1-X)FE S4	A	0.4774	0.4177	586.66	27.72	26.93 Y _g	440	14.00	16.85
TETRAGONAL	C	0.3680	0.3623	580.81	27.88	25.21 Y _p	460	15.80	18.90
	C	0.3487	0.3550	577.94	20.46	26.07 Y _g	480	17.60	20.60
REF. FOR R: (LEVY,1966)							500	20.20	22.60
REMARKS: LCC. VALRY							520	21.20	23.80
							540	23.60	25.20
							560	25.20	26.50
							580	27.70	27.60
							600	29.60	29.00
							620	31.80	30.00
							640	32.50	29.60

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		X	Y					(R _p)	(R _g)
CRANGE BCRNITE 3	A	0.4941	0.4136	589.99	36.39	26.51 Y _p	420	15.70	16.60
CU(2+X)SN(1-X)FE S4	A	0.4774	0.4178	586.62	27.87	26.81 Y _g	440	15.10	16.85
TETRAGONAL	C	0.3671	0.3586	581.77	26.64	25.05 Y _p	460	16.00	18.70
	C	0.3486	0.3545	577.27	20.27	25.94 Y _g	480	17.40	20.15
REF. FGR R: (LEVY,1966)							500	19.35	22.20
REMARKS: LOC. VAULRY							520	21.00	23.15
							540	23.20	25.90
							560	25.30	26.20
							580	27.60	27.50
							600	29.70	28.50
							620	31.40	30.00
							640	32.50	29.60

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y				
CRANGE BCRNITE 5	A	0.4946	0.4145	585.67	37.42	26.50	Y _p	420	13.95	15.60
CU(2+X)SN(1-X)FE S4	A	0.4803	0.4160	587.80	28.57	26.67	Y _g	440	14.40	16.50
TETRAGONAL	C	0.3683	0.3622	580.91	27.93	25.06	Y _p	460	15.90	18.40
	C	0.3509	0.3545	578.52	21.19	25.72	Y _g	480	17.70	20.60
REF. FOR R: (LEVY,1966)								500	19.50	21.90
REMARKS: LCC. VALLRY								520	21.30	23.20
								540	23.30	24.80
								560	25.40	26.20
								580	27.30	27.10
								600	29.10	28.30
								620	31.40	30.00
								640	33.00	30.70

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y			(R _p)	(R _g)
CRANGE BERNITE 8	A	0.4940	0.4138	589.91	36.52	26.47	Y _p	420	15.70	16.60
CU(2+X)SN(1-X)FE S4	A	0.4776	0.4176	586.73	27.80	26.86	Y _g	440	15.00	16.80
TETRAGONAL	C	0.3672	0.3588	581.72	26.72	25.02	Y _p	460	16.00	18.70
	C	0.3488	0.3545	577.45	20.40	25.98	Y _g	480	17.30	20.30
REF. FOR R: (LEVY,1966)								500	19.30	22.30
REMARKS: LOC. VAULRY								520	21.00	23.70
								540	23.20	25.30
								560	25.20	26.30
								580	27.60	27.60
								600	29.60	28.80
								620	31.80	30.00
								640	32.30	29.60

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y			(R _p)	(R _g)
CRANGE BCRNITE 7	A	0.4926	0.4156	589.11	36.80	25.80	Y _p	420	13.90	16.60
CU(2+X)SN(1-X)FE S4	A	0.4783	0.4163	587.18	27.73	26.84	Y _g	440	14.20	16.80
TETRAGONAL	C	0.3671	0.3615	580.83	27.42	24.44	Y _p	460	15.70	18.70
	C	0.3491	0.3538	578.15	20.52	25.94	Y _g	480	17.30	20.50
REF. FOR R: (LEVY,1966)								500	18.70	22.40
REMARKS: LCC. VAULRY								520	20.50	23.40
								540	22.60	25.10
								560	24.90	26.40
								580	27.10	27.60
								600	29.10	28.70
								620	30.70	29.50
								640	30.50	30.10

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y							
CRANGE BCRNITE 2	A	0.4923	0.4156	589.09	36.59	25.78	Y _p	420	13.70	16.60
CU(2+X)SN(1-X)FE S4	A	0.4783	0.4168	587.18	27.73	26.84	Y _g	440	14.30	16.85
TETRAGONAL	C	0.3667	0.3615	580.74	27.31	24.44	Y _p	460	15.80	18.70
	C	0.3490	0.3537	578.15	20.47	25.54	Y _g	480	17.15	20.50
REF. FOR R: (LEVY,1966)								500	18.90	22.40
REMARKS: LCC. VAULRY								520	20.60	23.40
								540	22.70	25.10
								560	24.80	26.40
								580	27.00	27.60
								600	29.00	28.70
								620	30.70	29.50
								640	30.50	30.10

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _g) (R _p)	
		X	Y						
CRANGE BCRNITE 10	A	0.5040	0.4137	590.43	43.38	25.70 Y _g	420	11.35	13.40
CU(2+X)SN(1-X)FE S4	A	0.4831	0.4181	587.25	31.96	24.89 Y _p	440	12.60	14.40
TETRAGONAL	C	0.3809	0.3685	581.86	33.00	24.00 Y _g	460	14.10	16.20
	C	0.3561	0.3604	578.19	24.17	23.92 Y _p	480	15.70	18.40
REF. FOR R: (LEVY,1966)							500	17.40	19.80
REMARKS: LOC. VAULRY							520	19.45	21.40
							540	21.80	22.90
							560	24.20	24.40
							580	26.70	25.80
							600	29.30	26.80
							620	30.70	27.80
							640	33.50	28.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CC*FL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _g) (R _p)	
		X	Y					(R _g)	(R _p)
CRANGE BCRNITE 12	A	0.5019	0.4142	590.15	42.27	25.55 ^{Y_g}	420	11.40	13.40
CU(2+X)SN(1-X)FE S4	A	0.4836	0.4191	586.92	33.01	24.92 ^{Y_p}	440	13.00	14.20
TETRAGONAL	C	0.3782	0.3680	591.45	32.14	23.94 ^{Y_p}	460	14.00	16.00
	C	0.3572	0.3624	577.74	24.88	23.95 ^{Y_g}	480	15.90	17.60
REF. FOR R: (LEVY, 1966)							500	17.75	19.90
REMARKS: LOC. VALLRY							520	20.10	21.40
							540	21.80	23.20
							560	23.90	24.40
							580	26.50	25.80
							600	28.80	26.60
							620	30.70	28.20
							640	33.00	28.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y			R _p	R _g
CRANGE BORNITE 11	A	0.5055	0.4123	590.98	43.23	24.63	Y _p	420	11.35	13.40
CU(2+X)SN(1-X)FE S4	A	0.4871	0.4164	588.35	33.55	25.34	Y _g	440	12.10	14.40
TETRAGONAL	C	0.3819	0.3667	582.55	32.78	22.54	Y _p	460	13.60	16.20
	C	0.3555	0.3607	579.08	25.17	24.24	Y _g	480	14.85	18.25
REF. FOR R: (LEVY,1966)								500	16.65	19.90
REMARKS: LOC. VAULRY								520	18.40	21.50
								540	20.70	23.10
								560	23.00	24.60
								580	25.40	25.90
								600	28.40	27.40
								620	29.90	28.20
								640	32.50	30.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY (R _p) (R _g)		
		X	Y			Y	LAMBDA (NM)	(R _p)	(R _g)	
ORANGE BERNITE 13	A	0.5098	0.4082	592.67	43.57	23.57	Y _p	420	12.70	14.90
CU(2+X)SN(1-X)FE S4	A	0.4814	0.4132	589.31	27.37	24.18	Y _g	440	12.20	15.60
TETRAGONAL	C	0.3850	0.3604	584.99	31.91	21.76	Y _p	460	12.90	17.30
	C	0.3502	0.3499	580.06	19.77	23.27	Y _g	480	13.60	18.50
REF. FOR R: (LEVY,1966)								500	15.40	20.20
REMARKS: LOC. PERU								520	16.85	20.80
								540	18.90	22.20
								560	21.40	23.40
								580	24.60	24.60
								600	27.70	25.60
								620	30.70	27.60
								640	31.80	28.60

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _g) (R _p)	
		X	Y					(R _g) e	(R _p) o
MAKSCNITE 14	A	0.4904	0.4133	589.92	33.68	27.18 Y _g	420	14.90	17.10
CU ₂ +X SN _{1-X} FE S ₄ , .5<><1	A	0.4738	0.4141	587.92	22.72	26.10 Y _p	440	16.40	18.50
TETRAGONAL	C	0.3621	0.3555	581.40	24.57	25.81 Y _g ^F	460	17.50	19.90
	C	0.3419	0.3452	578.93	16.29	25.34 Y _p ^o	480	19.00	20.80
REF. FOR R: (LEVY, 1966)							500	20.50	22.40
REMARKS: LOC. MONT-LYELL							520	21.80	23.20
							540	24.10	24.50
							560	26.20	25.70
							580	28.20	26.60
							600	29.80	27.80
							620	31.90	28.50
							640	33.00	29.00

Should be 0.404

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY		
		X	Y			Y	LAMBDA (NM)	(R _g)	(R _p)	
MAWSONITE 16	A	0.4903	0.4132	589.95	33.52	26.64	Y _g	420	14.90	17.20
CU ₂ +X SN ₁ -X FE S ₄ , .5<X<1	A	0.4737	0.4136	588.08	22.44	26.00	Y _p	440	16.20	18.40
TETRAGONAL	C	0.3620	0.3553	581.55	24.38	25.30	Y _g F	460	17.20	20.00
	C	0.3416	0.3446	578.99	16.10	25.24	Y _p °	480	18.50	20.70
REF. FOR R: (LEVY, 1966)								500	20.20	22.30
REMARKS: MONT-LYELL								520	21.30	23.20
								540	23.60	24.50
								560	25.70	25.50
								580	27.70	26.40
								600	29.30	27.60
								620	31.40	28.50
							640	32.20	29.00	

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY	
		X	Y			Y			(R _p) e	(R _g) o
MAWSONITE 15	A	0.4739	0.4140	587.99	22.70	26.07	Y _p	420	17.30	17.10
CU ₂ +X SN ₁ -X FE S ₄ , .5<X<1	A	0.4738	0.4141	587.92	22.72	26.23	Y _g	440	18.30	18.50
TETRAGONAL	C	0.3420	0.3450	579.07	16.26	25.29	Y _p	460	20.00	20.10
	C	0.3418	0.3454	578.79	16.31	25.47	Y _g	480	20.80	20.90
REF. FOR R: (LEVY, 1966)								500	22.20	22.50
REMARKS: LOC. MONT-LYELL								520	23.20	23.50
								540	24.40	24.70
								560	25.70	25.80
								580	26.60	26.60
								600	27.70	27.80
								620	28.50	28.70
							640	29.00	29.30	

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y			(R _p)	(R _g)
STANNITE JAUNE,5	A	0.4712	0.4140	587.58	20.86	26.62	Y _p	420	20.50	21.20
CU(I+X)SN(1-X)FE S4	A	0.4684	0.4131	587.67	18.30	28.96	Y _g	440	19.30	22.00
TETRAGONAL	C	0.3389	0.3419	579.24	14.59	25.89	Y _p	460	20.50	23.30
	C	0.3349	0.3394	578.59	12.85	28.29	Y _g	480	21.20	24.40
REF. FOR R: (LEVY,1966)								500	22.60	25.20
REMARKS: LCC. VAULRY								520	23.90	26.70
								540	25.00	27.50
								560	26.60	28.90
								580	27.30	29.40
								600	28.20	30.00
								620	28.60	30.50
								640	29.00	31.90

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL 'R') Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y					(R _p)	(R _g)
STANNITE JAUNE,6	A	0.4724	0.4133	588.21	21.21	26.54 Yp	420	20.40	20.50
CU(I+X)SN(1-X)FE S4	A	0.4680	0.4137	587.17	18.44	27.58 Yg	440	19.40	20.80
TETRAGONAL	C	0.3356	0.3415	579.75	14.67	25.78 Yp	460	20.30	22.20
	C	0.3348	0.3399	578.24	12.96	26.95 Yg	480	21.00	22.90
REF. FOR R: (LEVY,1966)							500	22.50	24.00
REMARKS: LCC. VAULRY							520	23.80	25.40
							540	24.80	26.30
							560	26.40	27.60
							580	27.20	28.00
							600	28.00	28.80
							620	28.50	28.50
							640	29.50	30.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y			(R _p)	(R _g)
STANNITE JAUNE, 4	A	0.4720	0.4140	567.71	21.42	26.48	Y _p	420	20.00	20.80
CU(I+X)SN(1-X)FE S4	A	0.4680	0.4136	567.24	18.37	28.17	Y _g	440	19.20	21.30
TETRAGONAL	C	0.3356	0.3425	575.25	14.94	25.74	Y _p	460	20.20	22.70
	C	0.3348	0.3399	578.24	12.96	27.53	Y _g	480	21.05	23.40
REF. FOR R: (LEVY, 1966)								500	22.40	24.60
REMARKS: LOC. VAULRY								520	23.80	26.00
								540	24.80	26.80
								560	26.40	28.20
								580	27.10	28.60
								600	28.20	29.30
								620	28.40	29.60
								640	29.00	30.70

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		X	Y					(R _p)	(R _g)
STANNITE JAUNE,7	A	0.4724	0.4136	588.01	21.38	26.29 Y _p	420	20.00	20.70
CU(1+X)SN(1-X)FE S4	A	0.4673	0.4138	586.95	18.03	28.12 Y _g	440	19.00	21.20
TETRAGONAL	C	0.3401	0.3422	579.65	15.00	25.53 Y _p	460	20.00	22.90
	C	0.3344	0.3356	578.19	12.77	27.49 Y _g	480	21.00	23.50
REF. FOR R: (LEVY,1966)							500	22.20	24.50
REMARKS: LCC. VAULRY							520	23.50	25.90
							540	24.50	26.50
							560	26.00	28.40
							580	27.20	28.70
							600	28.10	29.40
							620	28.20	29.50
							640	28.80	30.20

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y					(R _p)	(R _g)
STANNITE JAUNE, 1	A	0.4762	0.4105	590.53	21.90	25.59 Y _p	420	20.20	21.00
CU(1+X)SN(1-X)FE S ₄	A	0.4725	0.4100	590.62	18.99	27.18 Y _g	440	18.80	20.90
TETRAGONAL	C	0.3421	0.3391	582.34	14.70	24.72 Y _p	460	19.55	21.90
	C	0.3373	0.3363	581.89	12.66	26.38 Y _g	480	20.00	22.20
REF. FOR R: (LEVY, 1966)							500	21.40	23.40
REMARKS: LCC. VAULRY							520	22.60	24.50
							540	23.30	25.20
							560	25.00	26.80
							580	26.20	27.40
							600	27.40	28.50
							620	27.90	29.00
							640	29.60	31.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY	
		X	Y			Y	LAMBDA (NM)	(R _p)	(R _g)
STANNITE JAUNE,3	A	0.4759	0.4095	591.25	21.00	25.38	Y _p 420	19.45	21.00
CU(I+X)SN(1-X)FE S4	A	0.4721	0.4092	591.28	18.15	27.55	Y _g 440	19.50	21.60
TETRAGONAL	C	0.3410	0.3374	582.93	13.95	24.52	Y _p 460	19.50	22.60
	C	0.3366	0.3348	582.63	12.07	26.75	Y _g 480	20.20	22.80
REF. FOR R: (LEVY,1966)							500	21.20	23.80
REMARKS: LCC. VAULRY							520	22.60	24.50
							540	23.00	25.60
							560	24.90	27.20
							580	25.60	27.80
							600	27.20	28.80
							620	27.70	29.50
							640	29.60	31.90

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y						
STANNITE,3	A	0.4568	0.4164	581.21	12.51	28.21 Y _p	420	22.20	22.20
CU ₂ SN FE S ₄	A	0.4543	0.4158	579.91	10.41	28.63 Y _g	440	22.80	24.00
TETRAGONAL	C	0.3243	0.3371	572.45	9.39	27.96 Y _p	460	24.00	25.20
	C	0.3215	0.3347	571.22	8.00	28.46 Y _g	480	25.20	26.20
REF. FOR R: (LEVY,1966)							500	26.20	27.20
REMARKS: LCC. SAINT-AGNES, CORNOUAILLES							520	27.20	27.80
							540	28.20	28.70
							560	28.70	29.00
							580	28.80	29.50
							600	28.70	29.00
							620	28.20	28.20
							640	28.00	28.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y				
STANNITE, 1	A	0.4571	0.4167	561.21	12.92	28.17	Y _p	420	22.20	22.20
CU ₂ SN FE S ₄	A	0.4541	0.4158	579.74	10.27	28.64	Y _g	440	22.50	24.00
TETRAGONAL	C	0.3247	0.3377	572.44	9.66	27.93	Y _p	460	24.00	25.40
	C	0.3212	0.3345	571.03	7.86	28.48	Y _g	480	24.80	26.20
REF. FOR R: (LEVY, 1966)								500	26.30	27.20
REMARKS: LOC. SAINT-AGNES, CORNCLAILLES								520	27.20	27.80
								540	28.20	28.80
								560	28.60	29.10
								580	28.80	29.40
								600	28.60	29.00
								620	28.20	28.20
								640	28.00	28.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y					(R _p)	(R _g)
STANNITE, 4	A	0.4575	0.4168	581.39	13.29	28.14 Y _p	420	22.20	23.50
CU ₂ SN FE S ₄	A	0.4537	0.4152	575.80	9.58	29.13 Y _g	440	22.50	24.70
TETRAGONAL	C	0.3252	0.3381	572.64	9.90	27.88 Y _p	460	23.60	25.80
	C	0.3203	0.3334	570.74	7.33	28.99 Y _g	480	24.80	26.70
REF. FOR R: (LEVY, 1966)							500	26.20	28.20
REMARKS: LOC. SAINT-AGNES, CCRNOUAILLES							520	27.20	28.60
							540	28.00	29.30
							560	28.60	29.40
							580	28.80	29.50
							600	28.60	29.60
							620	28.20	29.00
							640	28.00	28.60

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y						
STANNITE,6	A	0.4629	0.4153	584.73	16.02	28.01 Y _p	420	20.60	22.70
CU2 SN FE S4	A	0.4606	0.4130	585.67	12.83	28.84 Y _g	440	21.90	23.70
TETRAEDRAL	C	0.3299	0.3398	575.24	11.62	27.59 Y _p	460	22.70	24.30
	C	0.3264	0.3344	576.15	9.23	28.45 Y _g	480	24.40	26.10
REF. FOR R: (LEVY,1966)							500	25.60	27.20
REMARKS: LOC. S&INT-AGNES,CCRNQUAILLES							520	26.80	27.90
							540	27.50	27.40
							560	27.90	28.80
							580	28.20	29.20
							600	28.90	30.20
							620	28.90	29.70
							640	29.40	29.80

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		X	Y					(R _p)	(R _g)
STANNITE,7	A	0.4627	0.4154	584.92	16.65	27.94 Yp	420	20.70	22.80
CU2 SN FE S4	A	0.4599	0.4141	584.47	13.10	29.05 Yg	440	21.40	23.80
TETRAGONAL	C	0.3308	0.3408	575.28	12.13	27.50 Yp	460	22.30	24.30
	C	0.3262	0.3356	575.11	9.50	28.69 Yg	480	24.40	26.10
REF. FOR R: (LEVY,1966)							500	25.50	27.20
REMARKS: LOC. SAINT-AGNES,CCRNCLAILLES							520	26.50	27.90
							540	27.50	28.50
							560	27.90	28.90
							580	27.90	29.30
							600	28.90	30.30
							620	29.00	29.70
							640	29.50	29.80

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y					(R _p)	(R _g)
STANNITE, S	A	0.4606	0.4159	583.49	14.84	27.71 Y _p	420	21.60	21.10
CU ₂ SN FE S ₄	A	0.4591	0.4152	583.30	13.31	28.53 Y _g	440	21.60	23.20
TETRAGONAL	C	0.3283	0.3390	574.57	10.97	27.35 Y _p	460	22.60	24.00
	C	0.3261	0.3372	573.96	9.90	28.21 Y _g	480	24.50	25.70
REF. FOR R: (LEVY, 1966)							500	25.60	26.70
REMARKS: LOC. SAINT-AGNES, CERNCUAILLES							520	26.20	27.20
							540	27.40	28.20
							560	27.80	28.70
							580	28.00	29.00
							600	28.80	29.20
							620	29.20	29.20
							640	27.90	29.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _g) (R _p)	
		X	Y			Y			(R _g)	(R _p)
RENIERITE, 1 (RE>RD)	A	0.4846	0.4184	587.30	33.20	28.64	Y _g	420	17.30	17.60
CU(3-X)GE(X)FE S ₄ , X=<0.5)	A	0.4841	0.4150	588.88	30.49	27.77	Y _p	440	16.50	17.20
TETRAGONAL	C	0.3582	0.3559	578.96	24.60	27.42	Y _g	460	18.00	18.00
	C	0.3561	0.3550	580.00	22.72	26.38	Y _p	480	19.70	19.20
REF. FOR R: (LEVY, 1966)								500	21.50	20.80
REMARKS: LCC. KIPUSHI								520	23.80	22.90
								540	26.40	25.10
								560	28.40	27.20
								580	30.20	29.00
								600	31.00	30.20
								620	31.80	30.60
								640	32.50	31.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R_g) (R_p)	
		X	Y						
RENIERITE, ϵ (RE>RO)	A	0.4856	0.4179	587.61	33.55	28.44 Y_g	420	17.40	18.80
CU(3-X)GE(X)FE S ₄ , X=<0.5)	A	0.4843	0.4149	588.74	30.56	27.77 Y_p	440	17.30	18.50
TETRAGONAL	C	0.3592	0.3560	580.57	23.82	26.53 Y_g	460	17.80	18.30
	C	0.3550	0.3475	582.69	20.41	25.76 Y_p	480	18.50	18.50
REF. FOR R: (LEVY, 1966)							500	20.40	20.10
REMARKS: LOC. TSUMEB							520	22.60	22.00
							540	25.20	24.20
							560	27.30	26.30
							580	29.40	28.40
							600	31.00	29.90
							620	31.70	31.10
							640	31.60	31.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		X	Y					(R _g)	(R _p)
RENIERITE, 8 (RE>RO)	A	0.4869	0.4159	583.54	33.05	28.15 Y _g	420	17.60	17.60
CU(3-X)GE(X)FE S ₄ , X=<C.5)	A	0.4822	0.4138	585.07	28.34	26.03 Y _p	440	17.40	17.70
TETRAGONAL	C	0.3568	0.3546	580.37	22.80	26.76 Y _g	460	18.70	18.30
	C	0.3521	0.3482	581.44	19.82	24.94 Y _p	480	19.00	18.20
REF. FOR R: (LEVY, 1966)							500	21.00	19.70
REMARKS: LOC. M PASSA, CONGO							520	23.10	21.70
							540	25.40	23.70
							560	27.60	25.60
							580	29.40	27.20
							600	30.90	28.60
							620	31.40	29.40
							640	31.90	29.60

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (Cmpl.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		X	Y					(R _g)	(R _p)
RENIERITE, 4(R<R0)	A	0.4855	0.4174	587.74	33.14	27.93 Y _g	420	17.30	17.60
CL(3-X)GE(X)FE S4, X=<0.5)	A	0.4853	0.4125	589.97	29.56	27.02 Y _p	440	16.60	17.10
TETRAGONAL	C	0.3593	0.3556	579.38	24.81	27.19 Y _g	460	17.80	17.60
	C	0.3561	0.3541	580.34	22.48	25.90 Y _p	480	19.60	18.90
REF. FOR R: (LEVY, 1966)							500	21.20	20.50
REMARKS: LOC. KIPUSHI							520	23.60	22.20
							540	26.00	24.60
							560	28.00	26.60
							580	29.80	28.60
							600	31.40	30.10
							620	32.10	30.60
							640	32.20	30.30

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		X	Y					(R _p)	(R _g)
RENIERITE, 3(RE>RO)	A	0.4841	0.4163	588.03	31.38	27.56 Y _p	420	17.30	16.90
CL(3-X)GE(X)FE S ₄ , X=<0.5)	A	0.4864	0.4162	588.27	32.92	27.82 Y _g	440	16.60	16.30
TETRAGONAL	C	0.3587	0.3585	579.57	24.36	26.70 Y _p	460	17.50	17.00
	C	0.3555	0.3528	580.64	21.57	24.66 Y _g	480	19.40	18.30
REF. FOR R: (LEVY, 1966)							500	20.80	19.80
REMARKS: LCC. KIPUSHI							520	23.10	21.20
							540	25.40	23.40
							560	27.60	25.20
							580	29.40	27.00
							600	30.70	28.50
							620	31.20	28.80
							640	31.80	29.60

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y					(R _p)	(R _g)
RENIERITE, 7(Pe>Po)	A	0.4839	0.4161	588.16	31.12	27.07 ^{Yp}	420	17.40	18.40
CU(3-X)GE(X)FE S ₄ , X=<0.5)	A	0.4847	0.4159	588.33	31.54	27.99 ^{Yg}	440	17.30	18.30
TETRAGONAL	C	0.3597	0.3560	580.71	23.96	26.82 ^{Yp}	460	18.10	18.50
	C	0.3557	0.3515	581.21	21.68	26.54 ^{Yg}	480	18.80	18.90
REF. FOR R: (LEVY, 1966)							500	20.70	20.60
REMARKS: LOC. ISLMEB							520	22.90	22.70
							540	25.40	25.20
							560	27.50	27.20
							580	29.70	29.40
							600	31.20	30.90
							620	32.00	31.40
							640	32.50	31.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	WAVELENGTH (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y						
RENIERITE, S (RE>RO)	A	0.4844	0.4149	588.75	30.63	25.78 Y _p	420	17.40	18.40
CU(3-X)GE(X)FE S ₄ , X=<0.5)	A	0.4872	0.4157	588.65	33.12	28.15 Y _g	440	17.30	18.30
TETRAGONAL	C	0.3556	0.3562	580.57	24.01	26.88 Y _p	460	18.10	18.50
	C	0.3554	0.3516	581.08	21.62	26.55 Y _g	480	18.80	18.90
REF. FOR R: (LEVY, 1966)							500	20.80	20.70
REMARKS: LOC. TSUMEB							520	23.00	22.80
							540	25.50	25.20
							560	27.60	27.30
							580	29.70	29.30
							600	31.20	30.80
							620	32.00	31.40
							640	32.50	31.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		X	Y					(R _g)	(R _p)
LUZONITE 1	A	0.4607	0.4078	592.38	9.28	28.50 Y _g	420	27.60	25.00
CUB(AS,SB)S ₄	A	0.4552	0.4054	604.76	3.81	25.62 Y _p	440	26.80	25.30
TETRAGONAL	C	0.3236	0.3228	587.76	5.36	28.39 Y _g	460	25.80	24.60
	C	0.3165	0.3168	607.19	1.85	25.36 Y _p	480	25.20	24.30
REF. FOR R: (LEVY, 1966)							500	26.00	24.70
REMARKS: LCC. MANKAYAN, PHILIPPINES							520	26.70	24.50
							540	27.90	24.80
							560	28.50	25.20
							580	29.40	26.00
							600	30.60	25.80
							620	30.00	26.80
							640	30.50	27.20

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _g) (R _p)	
		X	Y					(R _g)	(R _p)
LUZONITE 2	A	0.4560	0.4081	591.24	6.24	28.42 Y _g	420	27.60	26.00
CU ₃ (AS,SB)S ₄	A	0.4543	0.4054	607.72	3.18	25.68 Y _p	440	27.20	25.30
TETRAGONAL	C	0.3191	0.3206	587.86	3.57	28.08 Y _g	460	26.40	24.80
	C	0.3155	0.3163	616.39	1.01	25.45 Y _p	480	25.60	24.30
REF. FOR R: (LEVY, 1966)							500	26.30	25.10
REMARKS: LGC. MANKAYAN, PHILIPPINES							520	26.90	24.70
							540	27.80	25.00
							560	28.20	25.30
							580	28.90	25.90
							600	29.60	25.80
							620	29.60	26.80
							640	29.00	27.20

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _g) (R _p)	
		X	Y					(R _g)	(R _p)
LUZONITE 3	A	0.4545	0.4079	591.54	5.06	26.54 Y _g	420	25.00	25.50
CL3(AS,SB)S4	A	0.4523	0.4079	590.93	3.52	25.84 Y _p	440	25.30	24.80
TETRAGONAL	C	0.3173	0.3204	585.46	3.03	26.30 Y _g	460	25.10	24.60
	C	0.3148	0.3194	583.42	2.09	25.65 Y _p	480	25.00	24.90
REF. FOR R: (LEVY, 1966)							500	25.20	25.10
REMARKS: LCC. MANKAYAN, PHILIPPINES							520	25.40	25.20
							540	26.10	25.60
							560	26.30	25.80
							580	26.80	26.00
							600	27.40	25.80
							620	27.20	26.20
							640	27.20	26.70

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _g) (R _p)	
		X	Y					(R _g)	(R _p)
STIBICLUZCNITE 5	A	0.4581	0.4054	600.25	5.81	27.07 Y _g	420	28.40	26.80
CL3(AS,SB)S4	A	0.4532	0.4017	-516.38	3.05	24.30 Y _p	440	26.30	25.40
TETRAGONAL	C	0.3194	0.3170	607.45	2.68	26.65 Y _g	460	25.20	24.40
	C	0.3124	0.3095	-528.43	3.17	24.09 Y _p	480	24.50	23.80
REF. FOR R: (LEVY, 1966)							500	24.70	23.70
REMARKS: LOC. FAMATINA, ARGENTINA							520	25.00	23.40
							540	26.20	23.70
							560	27.00	23.80
							580	27.40	24.00
							600	27.90	24.50
							620	27.90	25.30
							640	25.00	26.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		X	Y					(R _g)	(R _p)
STIBIOGLUCONITE 4	A	0.4565	0.4045	610.03	4.08	26.30 Y _g	420	27.90	26.30
CUB(AS,SB)S4	A	0.4540	0.4018	-512.41	3.15	24.59 Y _p	440	26.30	25.80
TETRAGONAL	C	0.3174	0.3148	-493.42	1.97	25.95 Y _g	460	25.00	24.60
	C	0.3132	0.3100	-517.83	3.12	24.34 Y _p	480	24.20	23.80
REF. FOR R: (LEVY,1966)							500	24.40	23.80
REMARKS: LCC. FAMATINA, ARGENTINA							520	24.50	23.50
							540	25.40	23.90
							560	26.40	24.20
							580	26.40	24.20
							600	26.70	25.00
							620	27.90	25.40
							640	28.10	26.80

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
BRIARTITE 9	A	0.4463	0.4134	564.77	3.21	26.89	420	24.30
CU ₂ (FE,ZN)S ₄	C	0.3117	0.3254	558.94	2.88	26.98	440	25.30
TETRAGONAL							460	25.40
REF. FOR R: (LEVY,1966)							480	26.10
REMARKS: LCC. KIFLSHI							500	26.90
							500	26.90
							520	27.60
							540	27.60
							560	27.20
							580	27.00
							600	26.50
							620	26.20
							640	25.80

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
BRIARTITE 10	A	0.4471	0.4140	566.79	4.14	26.87	420	23.30
CU ₂ (FE,ZN)GE S ₄	C	0.3129	0.3273	561.73	3.71	26.95	440	24.40
TETRAGONAL							460	25.50
REF. FOR R: (LEVY,1966)							480	26.10
REMARKS: LCC. KIPLSHI							500	26.80
							520	27.50
							540	27.50
							560	27.20
							580	27.00
							600	26.60
							620	26.20
							640	25.80

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y					(R _p)	(R _g)
CORONADITE	A	0.4379	0.4067	493.39	2.32	26.09 _{Yp}	430	28.20	34.40
PB<=2 MNB C6	A	0.4354	0.4077	495.07	2.87	31.48 _{Yg}	470	28.30	34.20
TETRAGONAL	C	0.3007	0.3096	481.82	4.14	26.39 _{Yp}	490	27.20	33.30
	C	0.2589	0.3096	483.40	4.79	31.94 _{Yg}	520	27.20	33.30
REF. FOR R: (NICHOL & PHILLIPS, 1965)							550	26.70	32.30
REMARKS: LCC. UNKNOWN							580	26.00	31.50
							610	25.00	30.00
							650	24.70	28.70

MINERAL NAME CCMPGSTITCN CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
CRYPTOMELANE	A	0.4371	0.4077	495.12	2.47	25.99		430	27.80
A<=2 B8 C16	C	0.3006	0.3109	483.81	4.03	26.32		470	28.00
TETR. & MCNO.								490	27.60
REF. FOR R: (NICHOL & PHILLIPS, 1964)								520	27.00
REMARKS: LOC. UNKNOWN								550	26.70
								550	26.70
								580	26.00
								610	25.00
								650	23.90

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y			(R _p)	(R _g)
HOLLANDITE	A	0.4397	0.4064	492.42	1.91	25.29	Y _p	430	27.00	33.90
EAK=2 R8 C16	A	0.4386	0.4067	497.08	2.09	31.78	Y _g	470	27.10	33.00
TETR.& MCNC.	C	0.3021	0.3103	481.29	3.56	25.53	Y _p	490	26.60	33.40
	C	0.3024	0.3126	484.98	3.19	32.12	Y _g	520	26.20	32.80
REF. FOR R: (NICHOL & PHILLIPS, 1965)								550	25.60	32.30
REMARKS: LOC. UNKNOWN								580	25.20	32.50
								610	24.50	30.20
								650	24.30	29.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY		
		X	Y			Y			R _x	R _y	R _z
ENARGITE	A	0.4408	0.4026	481.21	1.88	25.02	Y _x	440	27.7	26.7	29.8
CUB(AS,SB)S4	A	0.4497	0.4043	-537.34	1.52	26.30	Y _y	460	27.6	26.9	29.4
ORTHORHOMBIC	A	0.4454	0.4063	485.11	0.58	28.68	Y _z	480	27.2	26.8	29.1
	C	0.3011	0.3054	473.75	4.58	25.21	Y _x	500	26.5	26.5	28.9
	C	0.3103	0.3129	-547.99	1.41	26.23	Y _y	520	25.9	26.2	28.8
	C	0.3075	0.3127	470.11	1.40	28.73	Y _z	540	25.3	25.9	28.7
								560	24.8	25.9	28.7
								580	24.5	25.9	28.7
								600	24.4	26.3	28.6
								620	24.6	25.8	28.5
								640	25.3	27.7	28.4
								660	25.9	28.2	28.3

REF. FOR R: LOPEZ-SOLER ET AL (1970)

REMARKS: LOC. UTAH, U.S.A.: ORIENTED SECTION PARALLEL TO X- Y- Z-AXIS

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		X	Y					(R _p)	(R _g)
ENARGITE 7	A	0.4510	0.4032	-525.43	2.13	25.75 Y _p	420	26.70	29.10
CL3(AS,SB)S4	A	0.4469	0.4029	-573.22	1.55	26.56 Y _g	440	26.30	28.80
CRT-RHOMBIC	C	0.3110	0.3116	-535.45	2.06	25.62 Y _p	460	26.30	28.50
	C	0.3070	0.3086	-566.15	2.52	26.56 Y _g	480	26.00	27.40
REF. FOR R: (LEVY,1966)							500	25.60	27.50
REMARKS: LCC. PARAD							520	25.30	26.90
							540	25.20	26.90
							560	25.40	26.40
							580	25.40	26.90
							600	25.90	26.60
							620	26.20	27.30
							640	27.40	28.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y							
ENARGITE S	A	0.4520	0.4031	-580.00	2.34	25.62	Y _p	420	26.50	29.20
CL3(AS,SB)S4	A	0.4469	0.4026	-573.08	1.66	26.83	Y _g	440	26.20	28.60
ORTHOHOMBIC	C	0.3117	0.3118	-527.12	2.11	25.46	Y _p	460	26.20	28.50
	C	0.3068	0.3081	-566.15	2.69	26.82	Y _g	480	25.10	27.40
REF. FOR R: (LEVY,1966)								500	25.50	27.50
REMARKS: LCC. PARAC								520	25.20	26.80
								540	25.10	26.70
								560	25.30	26.20
								580	25.00	26.80
								600	25.70	26.40
								620	26.10	27.10
								640	27.70	28.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (Rp) (Rg)	
		X	Y					(Rp)	(Rg)
ENARGITE 1	A	0.4503	0.4032	-540.06	2.02	25.59 ^{Yp}	420	26.70	29.30
CU ₃ (AS,SB)S ₄	A	0.4481	0.4024	-568.27	1.94	26.91 ^{Yg}	440	26.40	28.60
ORTHO-RHOMBIC	C	0.3105	0.3110	-547.02	2.21	25.48 ^{Yp}	460	26.30	28.20
	C	0.3079	0.3083	-562.40	2.81	26.86 ^{Yg}	480	25.80	27.50
REF. FOR R: (LEVY, 1966)							500	25.40	27.20
REMARKS: LCC. FAURCN, PERU							520	25.20	26.70
							540	25.00	26.50
							560	25.20	26.50
							580	25.50	26.70
							600	25.70	26.70
							620	26.00	27.50
							640	27.00	28.20

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y			(R _p)	(R _g)
ENARGITE 4	A	0.4508	0.4031	-530.71	2.14	25.18	Y _p	420	26.20	29.40
CU ₃ (AS,SB)S ₄	A	0.4485	0.4023	-566.25	2.04	26.85	Y _g	440	25.90	28.70
ORTHO-RHOMBIC	C	0.3110	0.3112	-540.51	2.22	25.05	Y _p	460	25.70	28.00
	C	0.3083	0.3081	-560.53	2.97	26.77	Y _g	480	25.40	27.30
REF. FOR R: (LEVY,1966)								500	25.00	26.80
REMARKS: LOC. BOR, YUGOSLAVIA								520	24.80	26.60
								540	24.60	26.40
								560	24.60	26.40
								580	25.00	26.60
								600	25.50	27.10
								620	26.00	27.30
								640	26.50	28.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y			(R _p)	(R _g)
ENARGITE ε	A	0.4509	0.4011	-544.98	2.48	25.13	Y _p	420	27.20	28.90
CUB(AS,SB)S ₄	A	0.4507	0.4021	-549.77	2.88	27.14	Y _g	440	26.70	28.60
ORTHORHOMBIC	C	0.3099	0.3078	-551.92	2.76	24.98	Y _p	460	26.10	27.90
	C	0.3100	0.3094	-552.37	3.39	27.01	Y _g	480	25.50	27.30
REF. FOR R: (LEVY,1966)								500	24.90	27.20
REMARKS: LCC. BCR, YUGOSLAVIA								520	24.70	26.70
								540	24.50	26.70
								560	24.40	26.50
								580	24.90	26.80
								600	25.10	27.00
								620	26.40	27.90
								640	27.00	29.20

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY		
		X	Y			Y	LAMBDA (NM)	(R _p)	(R _g)	
ENARGITE 3	A	0.4510	0.4017	-543.90	2.68	24.82	Y _p	420	26.70	28.40
Cu ₃ (As, Sb) ₂ S ₄	A	0.4500	0.4024	-552.96	2.26	26.59	Y _g	440	26.20	28.10
ORTHO-RHOMBIC	C	0.3104	0.3087	-549.08	3.12	24.68	Y _p	460	25.60	27.40
	C	0.3096	0.3093	-554.64	2.73	26.48	Y _g	480	25.00	26.80
REF. FOR R: (LEVY, 1966)								500	24.60	26.70
REMARKS: LOC. HAURON, PERU								520	24.20	26.20
								540	24.30	26.20
								560	24.20	26.00
								580	24.60	26.30
								600	25.00	26.50
								620	25.80	27.40
								640	26.50	28.30

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y	Y		(R _p)	(R _g)
ENARGITE 5	A	0.4537	0.4021	-512.28	2.99	24.76	Y _p	420	26.20	29.00
CU ₃ (AS,SB) ₅ 4	A	0.4499	0.4021	-555.75	2.35	27.03	Y _g	440	25.70	28.40
ORTHO-RHOMBIC	C	0.3129	0.3108	-516.90	2.74	24.54	Y _p	460	24.70	28.10
	C	0.3094	0.3090	-555.71	2.81	26.93	Y _g	480	24.40	27.40
REF. FOR R: (LEVY, 1966)								500	24.10	27.10
REMARKS: LOC. BCR, YUGOSLAVIA								520	24.20	26.70
								540	24.00	26.70
								560	24.20	26.40
								580	24.50	26.50
								600	24.70	27.10
								620	25.70	27.90
								640	27.20	28.80

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y					(R _p)	(R _g)
ENARGITE 8	A	0.4537	0.4015	-514.83	3.21	24.70 Yp	420	26.40	29.20
CU3(AS,SB)S4	A	0.4495	0.4022	-559.28	2.24	26.99 Yg	440	25.90	28.30
ORTHO-RHOMBIC	C	0.3126	0.3099	-524.53	3.05	24.47 Yp	460	24.70	28.10
	C	0.3091	0.3089	-557.35	2.79	26.89 Yg	480	24.40	27.40
REF. FOR R: (LEVY, 1966)							500	24.10	27.20
REMARKS: LCC. PARAL							520	24.20	26.70
							540	23.80	26.50
							560	24.10	26.40
							580	24.40	26.50
							600	24.70	27.30
							620	25.70	27.70
							640	27.20	28.50

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y	Y		(R _p)	(R _g)
ENARGITE 48	A	0.4424	0.4036	480.51	1.45	24.70	Y _p	440	26.65	29.24
CL3(AS,SB)S4	A	0.4463	0.4059	461.18	0.43	28.17	Y _g	460	26.69	28.73
ORTHOCHROMIC	C	0.3030	0.3080	474.54	3.57	24.86	Y _p	480	26.42	28.37
	C	0.3079	0.3126	461.69	1.30	28.19	Y _g	500	26.01	28.42
REF. FOR R: (TUGAL,1969)								520	25.45	28.35
REMARKS: LCC. TURKEY								540	24.91	28.29
								560	24.51	27.94
								580	24.18	28.12
								600	24.16	28.11
								620	24.48	28.03
								640	24.76	28.11
								660	25.71	28.74

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y			(R _p)	(R _g)
ENARGITE 49	A	0.4429	0.4056	487.73	1.20	24.46	Y _p	440	25.68	26.01
CU ₃ (AS,SB)S ₄	A	0.4486	0.4059	-540.33	0.75	25.87	Y _g	460	25.73	26.09
ORTHORHOMBIC	C	0.3045	0.3114	479.46	2.59	24.61	Y _p	480	25.72	26.07
	C	0.3101	0.3149	-551.55	0.56	25.84	Y _g	500	25.55	26.12
REF. FOR R: (TUGAL,1969)								520	25.12	25.98
REMARKS: LOC. TURKEY								540	25.05	25.77
								560	24.17	25.63
								580	23.96	25.49
								600	24.04	25.78
								620	24.29	26.16
								640	24.74	26.70
								660	24.26	27.11

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y			(R _p)	(R _g)
ENARGITE 2	A	0.4540	0.4016	-513.20	3.22	23.68	Yp	420	25.20	28.00
CU ₃ (AS,SB)S ₄	A	0.4505	0.4018	-550.28	2.56	26.11	Yg	440	24.60	27.50
CRT+OR+OMBIC	C	0.3130	0.3102	-518.79	3.00	23.46	Yp	460	23.80	27.00
	C	0.3098	0.3088	-553.17	2.96	25.98	Yg	480	23.30	26.50
REF. FOR R: (LEVY, 1966)								500	23.00	26.00
REMARKS: LCC. FAURON, PERU								520	23.00	25.80
								540	23.00	25.60
								560	23.00	25.50
								580	23.40	25.60
								600	23.80	26.20
								620	24.80	27.00
								640	26.00	28.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
KCSTERITE 2,2	A	0.4502	0.4118	578.39	4.79	25.56	420	22.60
CU2 SN ZN S4	C	0.3149	0.3258	568.71	3.84	25.52	440	22.70
TETRAGONAL							460	24.60
REF. FOR R: (LEVY,1966)							480	24.70
REMARKS: LCC. VAULRY							500	25.20
							500	25.20
							520	25.70
							540	25.80
							560	25.50
							580	25.70
							600	25.50
							620	25.50
							640	25.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
KCSTERITE 1,1	A	0.4505	0.4122	578.49	5.27	25.55		420	22.10
CU2 SN ZN S4	C	0.3154	0.3266	568.70	4.24	25.52		440	22.50
TETRAGONAL								460	24.40
REF. FOR R: (LEVY,1966)								480	24.80
REMARKS: LCC. VAULRY								500	25.20
								500	25.20
								520	25.60
								540	25.80
								560	25.50
								580	25.80
								600	25.40
								620	25.50
								640	25.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
KCSTERITE 3,1	A	0.4502	0.4122	577.96	5.07	25.54	420	22.20
CU2 SN ZN S4	C	0.3151	0.3267	568.13	4.14	25.51	440	22.50
TETRAGONAL							460	24.40
REF. FOR R: (LEVY,1966)							480	24.90
REMARKS: LCC. VAULFY							500	25.20
							500	25.20
							520	25.60
							540	25.80
							560	25.60
							580	25.70
							600	25.30
							620	25.50
							640	25.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS		REFLECTIVITY R
		COORDINATES X	Y			(VISUAL R) Y	LAMBDA (NM)	
KCSTERITE 3,2	A	0.4494	0.4119	576.54	4.30	25.53	420	22.60
CL2 SN ZN S4	C	0.3142	0.3255	567.26	3.57	25.52	440	22.90
TETRAGONAL							460	24.60
REF. FOR R: (LEVY,1966)							480	24.80
REMARKS: LCC. VAULRY							500	25.40
							500	25.40
							520	25.60
							540	25.80
							560	25.60
							580	25.70
							600	25.40
							620	25.40
							640	25.20

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
KCSTERITE 2,1	A	0.4506	0.4121	578.78	5.28	25.52	420	22.20
CU2 SN ZN S4	C	0.3154	0.3266	568.93	4.19	25.48	440	22.50
TETRAGONAL							460	24.40
REF. FOR R: (LEVY,1966)							480	24.70
REMARKS: LOC. VALLRY							500	25.20
							500	25.20
							520	25.60
							540	25.70
							560	25.50
							580	25.70
							600	25.40
							620	25.50
							640	25.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
KCSTERITE 1,2	A	0.4490	0.4119	575.50	4.03	25.41	420	22.60
CU2 SN ZN S4	C	0.3138	0.3250	566.74	3.33	25.41	440	22.90
TETRAGONAL							460	24.60
REF. FOR R: (LEVY,1966)							480	24.80
REMARKS: LCC. VALLRY							500	25.20
							500	25.20
							520	25.50
							540	25.70
							560	25.40
							580	25.70
							600	25.40
							620	25.10
							640	24.90

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		X	Y					(R _E)	(R _W)
FEMATITE	A	0.4335	0.4061	493.10	3.37	25.40 Y _E	470	28.00	31.50
FE2 O3	A	0.4355	0.4071	494.23	2.87	29.12 Y _W	486	27.60	31.20
TRIGONAL	C	0.2965	0.3059	481.08	6.08	25.82 Y _E	546	26.20	30.00
	C	0.2988	0.3086	482.26	4.94	29.53 Y _W	589	25.10	28.80
REF. FOR R: (VCN GEHLEN & PILLER, 1965)							650	22.80	26.50
REMARKS: LOC. ELBA, ITALY							656	22.60	26.20

Chemical Analysis:	Wt.%
Fe ₂ O ₃	99.55
FeTiO ₃	- -
MnTiO ₃	0.04
MgTiO ₃	---
	99.59

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _g) (R _w)		
		X	Y			Y			R _g	R _w	
TI-FEMATITE	A	0.4357	0.4058	492.36	2.87	23.98	Y _g	470	26.10	29.80	
FE(Fe,Ti)O ₃	A	0.4362	0.4066	493.45	2.72	27.57	Y _w	486	25.70	29.50	
TRIGONAL	C	0.2983	0.3066	480.25	5.37	24.30	Y _g	546	24.60	28.30	
	C	0.2992	0.3084	481.65	4.82	27.94	Y _w	589	23.70	27.30	
REF. FOR R:	(VCN GEHLEN & PILLER, 1965)								650	22.00	25.20
REMARKS:	LCC. SNARUM, NORWAY								656	21.90	25.10

Chemical Analysis:	Wt.%
Fe ₂ O ₃	90.7
FeTiO ₃	7.8
MnTiO ₃	0.06
MgTiO ₃	0.6
	<hr/>
	99.16

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY (R _ε) (R _ω)	
		X	Y			Y	LAMBDA (NM)	(R _ε)	(R _ω)
ILMENITE	A	0.4504	0.4047	-515.00	1.49	17.32	Y _ε 470	17.30	20.60
FE Ti O ₃	A	0.4467	0.4054	-575.75	0.61	20.22	Y _ω 486	17.10	20.40
TRIGONAL	C	0.3114	0.3130	-524.29	1.57	17.24	Y _ε 546	17.00	20.10
	C	0.3082	0.3122	-567.71	1.30	20.22	Y _ω 589	17.40	20.20
REF. FOR R: (VON GEHLEN & PILLER, 1965)							650	18.00	20.40
REMARKS: LGC. ILMEN MTS., USSR							656	18.10	20.40

Chemical Analysis:	Wt. %
Fe ₂ O ₃	6.6
FeTiO ₃	80.5
MnTiO ₃	9.4
MgTiO ₃	3.0
	<hr/> 99.5

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y						
CREDNERITE	A	0.4373	0.4078	495.29	2.42	23.12 Y _p	430	24.70	35.80
CU MN C2	A	0.4348	0.4055	497.48	2.96	33.91 Y _g	470	24.90	36.80
MGNOCLINIC	C	0.3007	0.3110	483.87	3.98	23.41 Y _p	490	24.30	36.10
	C	0.2592	0.3125	486.67	4.37	34.46 Y _g	520	24.30	36.00
REF. FOR R: (NICHOL & PHILLIPS, 1965)							550	23.60	35.00
REMARKS: LCC. UNKNOWN							580	23.10	34.00
							610	22.30	32.20
							650	21.30	30.20

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY (R _p) (R _g)		
		X	Y			Y	LAMBDA (NM)			
CHALCOPHANITE	A	0.4379	0.4034	487.39	2.49	9.57	Yp	430	10.80	33.50
(Zn,Mn,Fe)MS C7.3F2 C	A	0.4252	0.4008	489.65	5.59	26.29	Yg	470	10.60	32.20
TRICLINIC	C	0.2992	0.3047	476.39	5.31	9.68	Yp	490	10.30	30.00
	C	0.2871	0.2942	473.12	10.88	26.96	Yg	520	10.00	28.50
REF. FOR R: (NICHOL & PHILLIPS, 1965)								550	9.60	27.30
REMARKS: LCC. UNKNOWN								580	9.40	25.70
								610	9.50	24.60
								650	9.10	23.50

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY R
		X	Y			Y			
PSILCMELANE	A	0.4375	0.4870	493.95	2.40	22.96		430	24.70
A3 X6 M8 C16	C	0.3005	0.3097	482.18	4.20	23.23		470	25.00
ORTHO RHOMBIC								490	24.10
REF. FOR R: (NICHOL & PHILLIPS, 1964)								520	23.80
REMARKS: LOC. UNKNOWN								550	23.50
								580	23.00
								610	22.10
								650	21.30

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY R
		X	Y			Y	LAMBDA (NM)	
GALLITE 1	A	0.4421	0.4051	487.13	1.41	21.65	420	23.60
CL GA S2	C	0.3036	0.3094	476.43	3.16	21.78	440	23.20
TETRAGONAL							460	23.00
REF. FOR R: (LEVY, 1966)							480	22.70
REMARKS: LCC. TSLMB							500	22.40
							500	22.40
							520	22.10
							540	22.00
							560	21.70
							580	21.40
							600	21.20
							620	21.20
							640	21.50

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY R
		X	Y					
GALLITE 3	A	0.4406	0.4050	488.60	1.77	21.55	420	23.90
CU GA S2	C	0.3022	0.3085	477.40	3.72	21.72	440	23.20
TETRAGONAL							460	23.20
REF. FOR R: (LEVY, 1966)							480	22.80
REMARKS: LCC. TSUMEB							500	22.70
							500	22.70
							520	22.20
							540	21.90
							560	21.50
							580	21.40
							600	21.10
							620	21.10
							640	21.10

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		WAVELENGTH (NM)	REFLECTIVITY R
		X	Y			X	Y		
GALLITE 2	A	0.4405	0.4049	489.43	1.80	21.54		420	23.90
CU GA S2	C	0.3020	0.3084	477.78	3.83	21.72		440	23.20
TETRAGONAL								460	23.20
REF. FOR R: (LEVY, 1966)								480	22.90
REMARKS: LOC. TSLMEB								500	22.70
								500	22.70
								520	22.40
								540	21.80
								560	21.40
								580	21.40
								600	21.10
								620	21.10
								640	21.10

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		COORDINATES X	Y					(Rp)	(Rg)
BRAUNITE	A	0.4391	0.4062	492.18	2.06	19.97 ^{Yp}	430	21.70	22.80
MN MN6(C8/SI O4)	A	0.4382	0.4064	492.80	2.26	20.88 ^{Yg}	470	21.50	22.50
TETRAGONAL	C	0.3015	0.3092	480.20	3.92	20.16 ^{Yp}	490	20.60	21.70
	C	0.3008	0.3091	480.95	4.17	21.10 ^{Yg}	520	20.70	21.70
REF. FOR R: (NICHOL & PHILLIPS, 1965)							550	20.40	21.40
REMARKS: LOC. UNKNOWN							580	19.80	20.70
							610	19.40	20.20
							650	19.00	19.70

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y		LAMBDA (NM)	REFLECTIVITY (Rp) (Rg)	
		X	Y							
HAUSMANNITE	A	0.4382	0.4073	494.43	2.23	16.85	Yp	430	18.00	21.70
(Mn,Fe)Mn2O4	A	0.4368	0.4073	494.47	2.56	19.93	Yg	470	18.10	21.40
TETRAGONAL	C	0.3013	0.3107	482.88	3.80	17.04	Yp	490	17.80	21.20
	C	0.3000	0.3099	482.92	4.35	20.19	Yg	520	17.50	20.80
REF. FOR R: (NICHOL & PHILLIPS, 1965)								550	17.20	20.50
REMARKS: LCC. UNKNOWN								580	16.80	19.80
								610	16.40	19.20
								650	15.60	18.40

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y	Y		R _p	R _g
PYROCHROITE	A	0.4388	0.4052	490.32	2.18	14.99	Y _p	430	16.40	19.10
MN(OH) ₂	A	0.4388	0.4067	493.24	2.10	17.59	Y _g	470	16.30	18.60
HEXAGONAL	C	0.3009	0.3077	478.52	4.32	15.13	Y _p	490	16.00	18.80
	C	0.3016	0.3100	481.42	3.78	17.78	Y _g	520	15.30	18.10
REF. FOR R: (NICHOL & PHILLIPS, 1965)								550	15.20	17.90
REMARKS: LCC. UNKNOWN								580	15.00	17.60
								610	14.60	17.10
								650	14.20	16.50

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y				
HETAERCLITE	A	0.4379	0.4065	493.03	2.32	13.17	Y _p	430	14.30	19.00
Zn Mn2 O4	A	0.4373	0.4062	492.64	2.48	17.22	Y _g	470	14.20	18.60
TETRAGONAL	C	0.3008	0.3092	481.10	4.16	13.32	Y _p	490	14.00	18.20
	C	0.3000	0.3064	480.83	4.54	17.43	Y _g	520	13.60	18.00
REF. FOR R: (NICHOL & PHILLIPS, 1965)								550	13.40	17.50
REMARKS: LOC. UNKNOWN								580	13.20	17.20
								610	12.80	16.70
								650	12.20	16.00

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		X	Y					(R _p)	(R _g)
GROUTITE	A	0.4415	0.4080	496.19	1.42	12.23 Y _p	430	12.80	21.40
ALPHA-MN OCH	A	0.4416	0.4059	490.23	1.45	19.82 Y _g	470	12.60	20.80
ORTHOROMBIC	C	0.3045	0.3135	484.72	2.33	12.32 Y _p	490	12.70	20.50
	C	0.3039	0.3099	476.94	2.99	19.94 Y _g	520	12.60	20.00
							550	12.40	20.00
REF. FOR R: (NICHOL & PHILLIPS, 1965)							580	12.20	20.00
REMARKS: LCC. UNKNOWN							610	12.00	19.50
							650	11.60	18.90

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y	Y		(R _p)	(R _g)
ANTIMONY	A	0.4426	0.4081	496.85	1.16	73.54	Y _p	470	75.50	77.30
SB	A	0.4434	0.4079	496.42	0.98	75.53	Y _g	546	74.70	76.60
TRIGONAL	C	0.3058	0.3143	485.18	1.78	73.98	Y _p	589	73.30	75.20
	C	0.3063	0.3145	485.07	1.57	75.92	Y _g	650	70.20	72.80

REF. FOR R: CERVELLE & CAYE (1968)

REMARKS: LCC: MEYMAC, FRANCE. SB=99%; AG=TRACES

CYSCRASITE	A	0.4505	0.4088	585.39	2.92	61.73	Y _p	470	59.90	62.60
AG3 SB	A	0.4536	0.4072	594.37	3.91	65.00	Y _g	546	61.50	64.00
CRT-CH-CMBIC	C	0.3136	0.3200	576.82	1.93	61.53	Y _p	589	62.20	65.10
	C	0.3157	0.3195	585.42	2.36	64.54	Y _g	650	62.10	68.50

REF. FOR R: ARAYA (1968)

REMARKS: LCC: UNKNOWN

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY		
		X	Y					R _p	R _m	R _g
PARACCCRASITE	A	0.4451	0.4074	494.11	0.59	67.50 Y _p	470	68.6	71.2	75.0
SB2 (SB,AS)2	A	0.4449	0.4078	496.70	0.63	70.22 Y _m	546	67.8	70.5	73.8
MONOCLINIC	A	0.4437	0.4076	495.23	0.92	73.27 Y _g	589	67.6	70.7	73.6
	C	0.3077	0.3145	481.29	1.07	67.69 Y _p	650	65.9	67.8	70.2
	C	0.3079	0.3150	483.50	0.94	70.42 Y _m				
	C	0.3066	0.3140	482.59	1.52	73.59 Y _g				

REF. FOR R: B.F. LEONARD & MEAD (1971)

REMARKS: LCC. BRCKEN HILL, NEW SOUTH WALES. A NEW MINERAL

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y		WAVELENGTH (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y	Y		(R _p)	(R _g)
MELCONITE	A	0.4586	0.4080	591.84	8.04	59.01	Y _p	470	54.50	56.30
NI TE2	A	0.4626	0.4043	600.93	8.17	60.88	Y _g	546	57.00	56.30
HEXAGONAL	C	0.3215	0.3236	583.83	5.02	58.22	Y _p	589	60.40	64.00
	C	0.3239	0.3196	595.64	4.58	59.66	Y _g	650	62.90	66.90

REF. FOR R: BURKE, FREE UNIV., AMSTERDAM

REMARKS: LOC: ROBB MONTRAY MINE, QUEBEC, CANADA

BISMUTH	A	0.4585	0.4097	583.74	5.08	58.55	Y _p	470	53.30	58.50
BI	A	0.4555	0.4102	586.67	7.35	63.07	Y _g	546	57.10	62.90
TRIGONAL	C	0.3224	0.3266	580.07	6.06	57.81	Y _p	589	59.60	62.90
	C	0.3192	0.3259	577.24	4.98	62.55	Y _g	650	61.60	66.40

REF. FOR R: CERVELLE & CAYE (1968)

REMARKS: LOC: ALLEMCNT, FRANCE. BI=96.8%; SB=2.3%; AS=C.2%

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CNPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		X	Y					(R _p)	(R _g)
KITKAITE	A	0.4581	0.4099	588.29	8.94	57.01 Y _p	470	52.00	54.70
NI TE SE	A	0.4576	0.4097	588.44	8.45	59.65 Y _g	546	55.80	58.30
HEXAGONAL	C	0.3220	0.3267	579.48	5.98	56.32 Y _p	589	58.10	61.00
	C	0.3214	0.3259	579.87	5.61	58.96 Y _g	650	60.00	62.30

REF. FOR R: BURKE, FREE UNIV., AMSTERDAM

REMARKS: LOC: KUUSAMC, FINLAND

RAMMESLBERGITE	A	0.4448	0.4051	477.74	0.81	56.90 Y _p	470	59.00	62.50
NI AS	A	0.4429	0.4051	482.83	1.00	59.92 Y _g	546	56.80	59.90
ORTHOCHROMBIC	C	0.3062	0.3109	470.11	2.09	57.06 Y _p	589	56.50	59.60
	C	0.3054	0.3103	472.44	2.44	60.15 Y _g	650	57.00	59.40

REF. FOR R: W. PETRUK, OTTAWA

REMARKS: LCC: EISLEBEN, GERMANY

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (Rp) (Rg)	
		X	Y						
RAMMELSBERGITE	A	0.4449	0.4058	483.24	0.73	56.47 ^{Yp}	470	58.20	61.80
NI AS	A	0.4437	0.4057	486.65	1.01	59.43 ^{Yg}	546	56.60	59.70
ORTHOHROMBIC	C	0.3066	0.3121	473.97	1.77	56.64 ^{Yp}	589	56.00	58.90
	C	0.3054	0.3111	475.66	2.32	59.69 ^{Yg}	650	56.40	58.80

REF. FOR R: W. PETRUK, OTTAWA

REMARKS: LCC: EISLEBEN, GERMANY

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (Rp) (Rg)	
		X	Y						
MILLERITE	A	0.4683	0.4175	584.99	21.29	55.24 ^{Ye}	470	43.60	43.90
NI S	A	0.4605	0.4127	585.89	12.56	50.17 ^{Yw}	546	54.30	49.40
HEXAGONAL	C	0.3377	0.3472	576.04	15.70	54.02 ^{Ye}	589	57.40	51.40
	C	0.3261	0.3331	576.92	8.80	49.46 ^{Yw}	650	58.40	52.40

REF. FOR R: BURKE, FREE UNIV., AMSTERDAM

REMARKS: LCC: LIBUSIN, KLADNO BOHEMIA

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY		
		X	Y			Y			R _α	R _β	R _γ
ARSENOPYRITE	A	0.4518	0.4085	588.14	3.60	52.68	Y _α	470	50.8	48.7	51.8
FE AS S	A	0.4539	0.4106	584.91	6.52	52.12	Y _β	546	52.2	51.8	51.9
MONOCLINIC	A	0.4468	0.4077	603.02	0.42	51.73	Y _γ	589	53.2	52.8	51.7
	C	0.3148	0.3203	579.85	2.33	52.43	Y _α	650	53.6	52.9	51.3
	C	0.3180	0.3251	576.11	4.48	51.77	Y _β				
	C	0.3095	0.3162	600.65	0.19	51.77	Y _γ				

REF. FOR R: P.R. SIMPSON (LONDON)

REMARKS: LCC. UNKNOWN

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y					(R _p)	(R _g)
STANNIC PALLADINITE	A	0.4601	0.4130	585.48	12.48	53.68 Yp	460	46.20	48.50
PC3 SN2	A	0.4606	0.4109	587.78	11.37	55.30 Yg	540	53.00	54.00
HEXAGONAL	C	0.3257	0.3336	576.17	8.83	52.98 Yp	580	54.00	55.50
	C	0.3249	0.3302	578.48	7.70	54.51 Yg	660	57.00	60.00

REF. FOR R: L. VYALSCV, MCSCCH

REMARKS: LCC: UNKNOWN

ARSENIC	A	0.4433	0.4061	489.17	1.08	44.28 Yp	470	46.00	46.70
AS	A	0.4428	0.4054	487.11	1.23	44.57 Yg	546	44.60	44.80
TRIGONAL (PSEUDOCUBIC)	C	0.3053	0.3116	477.65	2.27	44.49 Yp	589	43.50	44.20
	C	0.3045	0.3102	475.97	2.75	44.81 Yg	650	43.50	43.80

REF. FOR R: CERVELLE & CAYE (1968)

REMARKS: LCC: STE. MARIE AUX MINES, FRANCE. AS=98%; SB=2%

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	WAVELENGTH (NM)	REFLECTIVITY	
		X	Y					(R _E)	(R _W)
BREITHAUPITE	A	0.4719	0.4002	606.62	11.75	41.72 Y _E	470	37.40	45.50
NI SB	A	0.4658	0.4067	594.18	11.95	51.33 Y _W	546	36.90	48.20
HEXAGONAL	C	0.3312	0.3187	604.23	6.29	40.37 Y _E	589	43.70	53.00
	C	0.3283	0.3261	586.24	7.51	50.18 Y _W	650	51.00	58.00

REF. FOR R: BURKE, FREE UNIV., AMSTERDAM

REMARKS: LCC: CCBALT, ONTARIO CANADA

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	WAVELENGTH (NM)	R _P	R _G
		X	Y						
SCHAPBACHITE (MATILCITE)	A	0.4450	0.4075	494.79	0.61	43.49 Y _P	470	44.20	45.30
AG BI S2	A	0.4453	0.4073	493.31	0.55	44.60 Y _G	546	43.80	44.90
HEXAGONAL	C	0.3077	0.3147	482.46	1.04	43.63 Y _P	589	43.40	44.40
	C	0.3079	0.3147	481.61	0.97	44.72 Y _G	650	42.60	44.00

REF. FOR R: HARRIS & THORPE (1969)

REMARKS: LCC: CAMSELL RIVER, CANADA

This is the p 286

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY		
		X	Y			Y			R _α	R _β	R _γ
STIBNITE	A	0.4397	0.4049	439.11	1.58	41.40	Y _α	470	44.4	30.8	52.6
SR2 S3	A	0.4425	0.4090	459.93	1.16	30.13	Y _β	546	42.1	31.1	48.1
CATHOCROMEIC	A	0.4276	0.4039	431.72	4.87	45.27	Y _γ	589	40.4	29.3	42.1
	C	0.3013	0.3079	478.22	4.15	41.78	Y _α	650	40.4	29.3	42.1
	C	0.3057	0.3160	490.15	1.66	30.35	Y _β				
	C	0.2899	0.3005	480.83	9.08	46.39	Y _γ				

REF. FOR R: P.R. SIMFSCN(LCNDEN)

REMARKS: LCC. UNKNOW

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y	Y		(R _p)	(R _g)
FESSITE	A	0.4420	0.4057	489.25	1.40	38.17	Y _p	460	40.30	40.70
AG2 TE	A	0.4518	0.4062	500.22	0.95	41.43	Y _g	540	38.70	40.80
MONOCLINIC	C	0.3039	0.3103	478.01	2.94	38.41	Y _p	580	37.80	41.30
	C	0.3135	0.3167	489.54	1.29	41.20	Y _g	640	37.40	43.00

REF. FOR R: L.LOGINOVA, MOSCOW

REMARKS: LCC UNKNOWN

GEOCRONITE	A	0.4432	0.4076	495.18	1.03	37.92	Y _p	460	39.00	43.20
27PB S 7 (SE,AS) 2 S3	A	0.4408	0.4060	491.03	1.67	40.70	Y _g	540	38.40	41.30
MONOCLINIC	C	0.3061	0.3139	483.38	1.71	38.11	Y _p	580	38.00	40.60
	C	0.3030	0.3100	479.27	3.29	41.01	Y _g	650	36.50	39.00

REF. FOR R: L.VYALSOV, MOSCOW

REMARKS: LCC: UNKNOWN

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y			(R _p)	(R _g)
BISMUTHINITE	A	0.4425	0.4054	487.59	1.30	37.63	Y _p	470	39.50	50.60
BI2 S3	A	0.4428	0.4082	497.30	1.11	49.29	Y _g	546	37.80	50.20
ORTHORHOMBIC	C	0.3043	0.3100	475.03	2.84	37.84	Y _p	589	37.40	49.20
	C	0.3060	0.3146	485.83	1.67	49.68	Y _g	650	36.80	47.20

REF. FOR R: BURKE, FREE UNIV., AMSTERDAM

REMARKS: LCC: AVENTURA MINE, BOLIVIA

LAUNAYITE	A	0.4414	0.4055	489.23	1.55	36.56	Y _p	470	38.60	46.20
22PB S 13(SB,AS)2 S3	A	0.4389	0.4056	491.07	2.14	43.13	Y _g	546	36.90	43.80
MONOCLINIC	C	0.3033	0.3096	477.72	3.24	36.81	Y _p	589	36.20	42.70
	C	0.3011	0.3083	479.24	4.17	43.56	Y _g	650	35.50	40.90

REF. FOR R: JAMBOR (1967)

REMARKS: LCC: MAECC, ONTARIO CANADA

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		X	Y					(R _p)	(R _g)
VEENITE	A	0.4375	0.4063	492.77	2.43	36.75 Y _p	470	39.50	45.50
2PB S.(SB,AS)2 S3	A	0.4383	0.4059	491.86	2.26	42.42 Y _g	546	37.60	43.20
ORTHO RHOMBIC	C	0.3001	0.3086	481.01	4.48	37.18 Y _p	589	36.30	42.00
	C	0.3007	0.3084	479.98	4.30	42.87 Y _g	650	34.30	39.90

REF. FOR R: JAMBOR (1967)

REMARKS: LOC: MACOC, ONTARIO CANADA

SCRBYITE	A	0.4380	0.4060	492.13	2.32	36.32 Y _p	470	39.00	45.00
17PB S.11(SB,AS)2 S3	A	0.4382	0.4060	492.07	2.28	41.94 Y _g	546	37.00	43.00
MONOCLINIC	C	0.3005	0.3084	480.24	4.37	36.72 Y _p	589	36.00	41.00
	C	0.3005	0.3087	480.69	4.33	42.41 Y _g	650	34.00	40.00

REF. FOR R: JAMBOR (1967)

REMARKS: LOC: MACOC, ONTARIO CANADA

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y	Y			
TWINNITE	A	0.4394	0.4059	491.47	2.00	36.31	Y _p	470	38.70	45.60
FeS (SE, AS) 2 S3	A	0.4374	0.4055	493.45	2.49	42.15	Y _g	546	36.90	43.00
MONOCLINIC (FSELDG-CRTFG.)	C	0.3017	0.3090	479.55	3.87	36.65	Y _p	589	35.90	41.60
	C	0.2997	0.3074	479.75	4.77	42.65	Y _g	650	34.60	39.60

REF. FOR R: JANBGR (1967)

REMARKS: LOC: MACOC, ONTARIO CANADA

HETEROMORPHITE	A	0.4442	0.4109	515.31	0.83	36.17	Y _p	450	35.70	39.30
7PB S 4SB2 S3	A	0.4442	0.4112	518.08	0.88	39.98	Y _g	550	37.00	41.00
MONOCLINIC	C	0.3086	0.3196	520.07	0.64	36.34	Y _p	579	36.50	40.30
	C	0.3088	0.3202	528.79	0.79	40.17	Y _g	640	34.40	38.00

REF. FOR R: A. RAKTSHEEV, MCSGOW

REMARKS: LOC: UNKNOWN

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y		WAVELENGTH (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y							
TINTINAITE	A	0.4419	0.4062	490.90	1.40	35.71	Y _p	470	37.40	43.90
5PB S 4(BI)<SB)2 S3	A	0.4434	0.4071	493.17	1.00	42.61	Y _g	546	36.30	43.30
ORTHOCLINIC	C	0.3040	0.3111	479.80	2.80	35.95	Y _p	589	35.00	41.90
	C	0.3058	0.3134	482.36	1.88	42.84	Y _g	650	35.10	42.00

REF. FOR R: BURKE, FREE UNIV., AMSTERDAM

REMARKS: LOC: TINTINA, CANADA

PLAYFAIRITE	A	0.4390	0.4058	491.42	2.10	35.80	Y _p	470	38.30	42.30
16PB S.9(SB,AS)2 S3	A	0.4393	0.4059	491.51	2.03	39.64	Y _g	546	36.40	40.30
MONOCLINIC	C	0.3012	0.3066	479.60	4.10	36.16	Y _p	589	35.40	39.20
	C	0.3015	0.3089	479.68	3.95	40.02	Y _g	650	34.00	37.70

REF. FOR R: JAMBOR (1967)

REMARKS: LOC: MACOC, ONTARIO CANADA

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY		
		X	Y			Y	LAMBDA (NM)	(R _p)	(R _g)	
STERRYITE	A	0.4401	0.4061	491.61	1.83	35.46	Y _p	470	37.60	40.40
12PB S 5 (SB,AS)2 S3	A	0.4399	0.4062	491.91	1.87	38.09	Y _g	546	36.00	38.70
ORTHORHOMBIC	C	0.3023	0.3096	479.70	3.58	35.77	Y _p	589	35.10	37.70
	C	0.3022	0.3097	480.05	3.61	38.43	Y _g	650	33.90	36.30

REF. FOR R: JAMBOR (1967)

REMARKS: LCC: MACCC, ONTARIO CANADA

GUETTARDITE	A	0.4394	0.4065	492.71	1.97	34.22	Y _p	470	36.30	44.20
9PB S.8 (SB,AS)2 S3	A	0.4386	0.4059	491.76	2.19	41.26	Y _g	546	34.80	42.00
MONOCLINIC	C	0.3020	0.3099	480.71	3.65	34.54	Y _p	589	34.00	40.80
	C	0.3009	0.3085	479.87	4.21	41.69	Y _g	650	32.20	39.00

REF. FOR R: JAMBOR (1967)

REMARKS: LCC: MACCC, ONTARIO CANADA

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY		
		X	Y			Y			R _α	R _β	R _γ
ECURONITE	A	0.4405	0.4080	495.99	1.66	34.87	Y _α	470	36.3	35.5	37.6
2PB S CU2 S SB2 S3	A	0.4392	0.4062	492.15	2.03	33.34	Y _β	546	35.6	33.9	35.5
CARTHOROMBIC	A	0.4383	0.4057	491.49	2.27	35.04	Y _γ	589	34.7	32.9	34.9
	C	0.3038	0.3129	484.19	2.65	35.16	Y _α	650	32.7	31.6	32.7
	C	0.3016	0.3094	480.40	3.86	33.66	Y _β				
	C	0.3007	0.3079	479.17	4.36	35.40	Y _γ				

REF. FOR R: P.R. SIMPSON (LONDON)

REMARKS: LCC. PRIBRAM, USSR

Handwritten note: 100 mg. 1961. 1961

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY		
		X	Y			Y			R α	R β	R γ
CHALCOITE	A	0.4319	0.4039	490.88	3.86	32.43	γ	470	36.7	36.7	36.7
CU ₂ S	A	0.4323	0.4038	490.66	3.78	32.48	γ	546	33.3	33.3	33.1
CRTFCRFCMBIC	A	0.4326	0.4029	489.51	3.75	32.33	γ	589	31.7	31.8	31.5
	C	0.2941	0.3019	479.08	7.44	33.01	γ	650	29.6	29.7	30.2
	C	0.2944	0.3020	478.91	7.32	33.05	γ				
	C	0.2942	0.3009	477.99	7.53	32.88	γ				

REF. FOR R: P.R. SIMPSON (LONDON)

REMARKS: LCC. UNKNOWN

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		X	Y					(R _ε)	(R _ω)
ESKEBERNITE	A	0.4726	0.4130	588.44	21.14	25.66 Y _ε	470	20.40	30.60
CU FE SE2	A	0.4628	0.4122	587.09	13.80	35.39 Y _ω	546	24.40	34.60
HEXAGONAL	C	0.3398	0.3421	579.56	14.89	24.93 Y _ε	589	26.70	36.20
	C	0.3282	0.3338	574.61	10.89	34.79 Y _ω	650	29.00	37.90

REF. FOR R: BURKE, FREE UNIV., AMSTERDAM

REMARKS: LCC: EAGLE GROUP, CANADA

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY	
		X	Y					R _p	R _g
GETCHELLITE	A	0.4360	0.4031	488.19	2.94	25.55 Y _p	470	28.40	30.70
AS SB S3	A	0.4341	0.4016	487.12	3.47	27.01 Y _g	546	25.90	27.30
MONOCLINIC	C	0.2973	0.3029	476.68	6.20	25.88 Y _p	589	25.10	26.50
	C	0.2950	0.2996	475.67	7.44	27.41 Y _g	650	24.30	25.60

REF. FOR R: BURKE, FREE UNIV., AMSTERDAM

REMARKS: LCC: GETCHELL MINE, NEVADA, USA

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y					(R _p)	(R _g)
STANNICITE	A	0.4678	0.4137	587.13	18.30	25.08 Y _p	470	20.60	22.60
CUS ₂ SN(Fe,Zn) ₂ S ₈	A	0.4664	0.4132	587.18	16.99	27.12 Y _g	546	24.40	26.40
ORTHOCHRYSIC	C	0.2345	0.3401	577.86	12.90	24.52 Y _p	589	25.70	27.80
	C	0.3328	0.3382	578.10	11.97	26.55 Y _g	650	27.60	29.60

REF. FOR R: BURKE, FREE UNIV., AMSTERDAM

REMARKS: LCC: KONJG MINE, JAPAN

CINNABAR	A	0.4381	0.4031	486.62	2.46	24.71 Y _p	460	27.40	29.90
HG S	A	0.4386	0.4082	496.12	2.10	28.33 Y _g	540	25.00	29.30
TRIGONAL	C	0.2993	0.3040	475.10	5.38	24.96 Y _p	580	24.50	28.20
	C	0.3019	0.3123	484.88	3.40	28.64 Y _g	640	23.90	26.60

REF. FOR R: L.VYALSCV, MOSCOW

REMARKS: LCC: UNKNOWN

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y			(R _p)	(R _g)
CINNIBAR	A	0.4390	0.4038	487.24	2.21	23.88	Y _P	470	25.90	30.40
HG S	A	0.4403	0.4066	492.69	1.76	28.87	Y _g	546	23.90	29.10
TRIGONAL	C	0.3005	0.3054	475.16	4.78	24.09	Y _P	589	23.90	29.10
	C	0.3031	0.3103	497.73	3.22	29.10	Y _g	650	22.70	26.90

REF. FOR R: HABER, FREE UNIV., AMSTERDAM

REMARKS: LCC: ZIPS-GOMCRER, USSR

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	R _w	R _E
		X	Y			Y				
CINNABAR	A	0.4382	0.4043	488.91	2.36	24.42	Y _w	470	26.53	30.73
HG S	A	0.4364	0.4071	494.18	2.65	28.57	Y _E	546	24.68	29.51
TRIGONAL	C	0.2999	0.3059	477.32	4.89	24.67	Y _w	589	24.17	28.03
	C	0.2995	0.3093	482.55	4.61	28.95	Y _E	650	23.26	26.44

REF. FOR R: LOPEZ-SOLER ET AL (1970)

REMARKS: LCC: RIANC, LECN, SPAIN

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _E) (R _W)	
		X	Y					(R _E)	(R _W)
MAGNETFLUMBITE	A	0.4396	0.4064	452.45	1.93	21.48 Y _E	470	22.80	24.90
PE 0.6FE2 O3	A	0.4393	0.4065	452.74	2.00	23.44 Y _W	546	22.00	24.00
HEXAGONAL	C	0.3019	0.3101	481.20	3.66	21.69 Y _E	589	21.00	23.00
	C	0.3017	0.3101	481.46	3.73	23.68 Y _W	650	20.70	22.40

REF. FOR R: BURKE, FREE UNIV., AMSTERDAM

REMARKS: LCC: LONG BAN, SWEDEN

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	R _P	R _G
		X	Y					(R _P)	(R _G)
RUTILE	A	0.4395	0.4042	487.67	2.07	19.74 Y _P	470	21.30	24.70
TI O2	A	0.4400	0.4046	488.15	1.93	23.02 Y _G	546	19.90	23.20
TETRAGONAL	C	0.3010	0.3065	476.18	4.45	19.91 Y _P	589	19.50	22.80
	C	0.3016	0.3073	476.48	4.13	23.22 Y _G	650	19.10	22.25

REF. FOR R: HABER, FREE UNIV., AMSTERDAM

REMARKS: LCC: ZIPS-GOMCRER, USSR

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY	
		X	Y			Y	LAMBDA (NM)	(R)	(R)
NB-RUTILE(ILMENORUTILE) (FE,MN)X+Y(NE>TA)2X T12-(3X+Y)C4-Y	A	0.4415	0.4051	487.93	1.55	19.18	470	20.30	21.40
	A	0.4411	0.4046	486.95	1.67	20.09	546	19.30	20.20
TETRAGONAL	C	0.3032	0.3069	476.25	3.37	19.30	589	19.00	19.90
	C	0.3026	0.3080	475.66	3.70	20.23	650	18.70	19.60

REF. FOR R: BURKE, FREE UNIV.,AMSTERDAM

REMARKS: LOC: LEIRA,N.E.BRAZIL

TAPIOLITE	A	0.4444	0.4071	492.66	0.77	16.61	470	17.00	16.00
(FE,MN)TA2 O6	A	0.4420	0.4063	491.14	1.37	15.30	546	16.70	15.50
TETRAGONAL	C	0.3069	0.3136	479.93	1.46	16.67	589	16.60	15.10
	C	0.3041	0.3112	479.83	2.75	15.40	650	16.20	14.90

REF. FOR R: BURKE, FREE UNIV.,AMSTERDAM

REMARKS: LOC: N.E. BRAZIL

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY		
		X	Y			Y	LAMBDA (NM)	(R _E)	(R _W)	
TAPIOLITE	A	0.4476	0.4088	489.04	0.98	17.65	Y _E	470	17.50	16.40
(FE,MN)TA2 O6	A	0.4437	0.4062	489.04	0.98	15.84	Y _W	546	17.80	15.90
TETRAGONAL	C	0.3108	0.3183	564.67	0.73	17.70	Y _E	589	17.70	15.80
	C	0.3058	0.3118	476.69	2.08	15.91	Y _W	650	17.50	15.50

REF. FOR R: BURKE, FREE UNIV., AMSTERDAM

REMARKS: LCC: N.E. BRAZIL

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CCPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		REFLECTIVITY		
		X	Y			Y	LAMBDA (NM)	(R _E)	(R _W)	
PLATTNERITE	A	0.4320	0.4065	493.70	3.71	16.50	Y _P	470	18.30	19.50
PB O2	A	0.4307	0.4049	492.17	4.05	17.24	Y _G	546	17.20	17.90
TETRAGONAL	C	0.2953	0.3060	482.06	6.49	16.81	Y _P	589	16.10	16.80
	C	0.2935	0.3029	480.46	7.52	17.58	Y _G	650	14.80	15.50

REF. FOR R: BURKE, FREE UNIV., AMSTERDAM

REMARKS: LOC: TCHAF-KFCUNI MINE, ANARAK, IRAN

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _p) (R _g)	
		X	Y			Y	Y		(R _p)	(R _g)
WOLFRAMITE	A	0.4455	0.4069	507.27	0.48	15.78	Y _p	470	15.80	18.50
(Fe,Mn)WO ₄	A	0.4456	0.4068	506.92	0.45	16.46	Y _g	546	16.00	18.70
MONOCLINIC	C	0.3066	0.3174	501.46	0.43	15.82	Y _p	589	15.70	18.40
	C	0.3068	0.3172	499.44	0.43	18.51	Y _g	650	15.40	18.00

REF. FOR R: BURKE, FREE UNIV., AMSTERDAM

REMARKS: LOC: UNKNOWN

WGDGINITE	A	0.4438	0.4046	489.82	0.95	14.23	Y _p	470	14.70	15.80
(TA,NB,FE,MN,SN,ZR) ₁₆ O ₃₂	A	0.4457	0.4060	479.54	0.54	15.43	Y _g	546	14.30	15.40
MONOCLINIC	C	0.3060	0.3123	477.86	1.95	14.29	Y _p	589	14.20	15.40
	C	0.3076	0.3127	477.72	3.24	15.46	Y _g	650	13.90	15.40

REF. FOR R: BURKE, FREE UNIV., AMSTERDAM

REMARKS: LOC: SERICIZINHC, BRAZIL

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R)		WAVELENGTH (NM)	REFLECTIVITY (R _E) (R _W)	
		X	Y			Y	Y		(R _E)	(R _W)
LMANGITE	A	0.4309	0.4004	487.34	4.28	16.27	Ye	470	19.00	17.40
CU3 SE2	A	0.4533	0.3780	565.64	11.64	14.28	Yw	546	16.60	12.60
TETRAGONAL	C	0.2916	0.2965	476.47	8.99	16.57	Ye	589	15.70	12.80
	C	0.2997	0.2757	561.78	14.40	14.59	Yw	650	15.40	20.40

REF. FOR R: VON GELEN, FRANKFURT

REMARKS: LCC: SIERRA DE LMANGA, ARGENTINA

LMANGITE	A	0.4317	0.4003	486.82	4.10	16.00	Ye	470	18.60	15.80
CU3 SE2	A	0.4592	0.3798	556.16	12.00	13.60	Yw	546	16.00	12.00
TETRAGONAL	C	0.2926	0.2961	474.82	8.75	16.27	Ye	589	16.00	12.10
	C	0.2958	0.2814	556.86	12.76	13.34	Yw	650	14.60	20.10

REF. FOR R: EURKE, FREE UNIV., AMSTERDAM

REMARKS: LCC: EAGLE GROUP, LAKE ATHABASCA, CANADA

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _ε) (R _ω)	
		X	Y					(R _ε)	(R _ω)
STARINGITE	A	0.4427	0.4054	487.28	1.26	13.83 Y _ε	470	14.50	12.80
(Fe,Mn)0.5(Sn,Ti)4.5(Ta,Nb)0.12	A	0.4418	0.4047	486.24	1.51	12.07 Y _ω	546	13.90	12.10
HEXAGONAL	C	0.3044	0.3101	476.00	2.80	13.90 Y _ε	589	13.70	12.00
	C	0.3033	0.3084	474.67	3.41	12.15 Y _ω	650	13.60	11.80

REF. FOR R: BURKE, FREE UNIV., AMSTERDAM

REMARKS: LOC: SERICIZINCO, BRAZIL

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMFL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _ε) (R _ω)	
		X	Y					(R _ε)	(R _ω)
KLOCKMANNITE	A	0.4045	0.3933	489.10	10.81	10.99 Y _p	470	15.50	37.40
CL SE	A	0.4259	0.4065	493.33	2.80	34.50 Y _g	546	11.90	35.60
HEXAGONAL	C	0.2678	0.2754	478.02	20.04	11.55 Y _p	589	9.80	33.80
	C	0.2987	0.3081	481.72	5.04	34.98 Y _g	650	9.30	32.00

REF. FOR R: BURKE, FREE UNIV., AMSTERDAM

REMARKS: LOC: EAGLE GROUP, LAKE ATHABASCA, CANADA

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R)		LAMBDA (NM)	REFLECTIVITY (R _ε) (R _ω)	
		X	Y			Y _ε	Y _ω		(R _ε)	(R _ω)
CASSITERITE	A	0.4408	0.4054	489.48	1.70	12.26	Y _ε	470	13.00	11.30
SN O2	A	0.4421	0.4061	490.44	1.36	10.79	Y _ω	546	12.40	10.90
TETRAGONAL	C	0.3027	0.3091	477.85	3.50	12.35	Y _ε	589	12.10	10.70
	C	0.3042	0.3108	478.57	2.77	10.86	Y _ω	650	11.90	10.50

REF. FOR R: BURKE ET AL (1969)

REMARKS: LOC: UNKNOWN

GEIKIELITE	A	0.4422	0.4061	490.36	1.34	11.76	Y _ε	470	12.30	14.90
MG TI O3	A	0.4431	0.4068	492.14	1.09	14.40	Y _ω	546	11.90	14.50
TRIGONAL	C	0.3042	0.3110	479.10	2.74	11.83	Y _ε	589	11.60	14.40
	C	0.3055	0.3123	479.54	2.12	14.47	Y _ω	650	11.50	13.90

REF. FOR R: CERVELLE (1967)

REMARKS: LOC: CREST MORE, CALIFORNIA. MG O=29.4; FE C=3.8; TI O2=66.5

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (COMPL.) WAVELENGTH	PURITY (%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _g) (R _w)	
		X	Y					(R _g)	(R _w)
COVELLITE	A	0.4251	0.3944	484.25	5.99	23.06 Y _E	470	29.10	13.40
CU S	A	0.3581	0.3588	485.03	23.61	6.34 Y _w	546	23.70	7.15
HEXAGONAL	C	0.2837	0.2856	474.65	13.24	23.66 Y _E	589	21.15	4.20
	C	0.2284	0.2236	475.17	40.63	7.06 Y _w	650	23.00	5.90

REF. FOR R: SIMPSON; LONDON

REMARKS: LCC: UNKNOWN. S=32.6;CU=65.5;FE=0.05;SI=0.01;AL=0.03

COVELLITE	A	0.4195	0.3948	486.83	7.25	22.15 Y _E	470	28.70	13.60
CU S	A	0.3666	0.3504	481.57	22.30	6.30 Y _w	546	23.10	6.90
HEXAGONAL	C	0.2795	0.2836	476.31	14.92	22.87 Y _E	589	20.30	3.70
	C	0.2301	0.2177	473.33	40.97	6.96 Y _w	650	20.90	7.40

REF. FOR R: CERVELLE (1968)

REMARKS: LCC: BCR, YUGOSLAVIA. CU=66.2;FE=0.3;S=33.2

MINERAL NAME COMPOSITION CRYSTAL SYSTEM	SOURCE	CHROMATICITY COORDINATES		DOMINANT (CMPL.) WAVELENGTH	PURITY (P%)	BRIGHTNESS (VISUAL R) Y	LAMBDA (NM)	REFLECTIVITY (R _ω) (R _ε)	
		X	Y					(R _ω)	(R _ε)
COVELLITE	A	0.3673	0.3673	485.89	20.96	6.37 ^{Y_ω}	486	11.30	27.50
CU S	A	0.4248	0.3993	488.55	5.76	23.25 ^{Y_ε}	546	7.20	24.30
HEXAGONAL	C	0.2355	0.2343	475.79	36.70	7.01 ^{Y_ω}	589	4.50	21.70
	C	0.2857	0.2923	477.52	11.50	23.87 ^{Y_ε}	656	5.90	21.80

REF. FOR R: VON GEHLEN & PILLER (1965)

REMARKS: LCC. ALGHERO, SARDINIA, ITALY

Table XI 1. Mean Transmittance Values of Some Cut Gemstones

Wavelength (nm)	440	460	480	500	520	540	560	580	600	620	640	660	
Citrine	11.8	16.2	21.5	27.4	33.6	39.7	45.4	50.8	55.3	59.1	62.6	66.0	
Amethyst	33.9	38.7	38.7	36.2	33.4	31.9	32.0	33.7	36.5	40.2	44.1	47.5	
Chrysoberyl	39.2	42.9	43.4	42.5	47.3	56.0	66.5	75.0	80.5	83.5	84.8	86.2	Sp ₁
Chrysoberyl	9.2	11.8	12.6	13.1	17.4	25.8	36.9	48.5	57.4	63.1	67.0	70.1	Sp ₂
Spinel	0.0	0.1	0.1	0.1	0.1	0.6	2.3	6.6	13.8	21.9	29.9	37.9	
Syn. Ruby(1)	0.4	2.2	3.2	1.8	0.6	0.2	0.3	1.7	8.9	24.2	41.7	51.5	
Syn. Ruby(2)	6.0	8.6	6.5	4.7	4.1	4.5	6.8	13.7	29.0	48.1	61.6	68.0	
Almandine(1)	26.2	27.9	24.1	14.2	11.1	15.8	19.6	20.2	28.8	38.7	47.6	53.9	
Almandine(2)	31.8	33.6	29.9	18.8	15.0	20.1	24.1	24.2	31.7	40.3	47.6	51.3	

Matching Liquid

Di-iodo Methane R.I. = 1.74 for Chrysoberyl & Synthetic Ruby

Rayner = 1.70 for Spinel

Rayner = 1.55 for Citrine & Amethyst

Cargille = 1.78 for Almandine

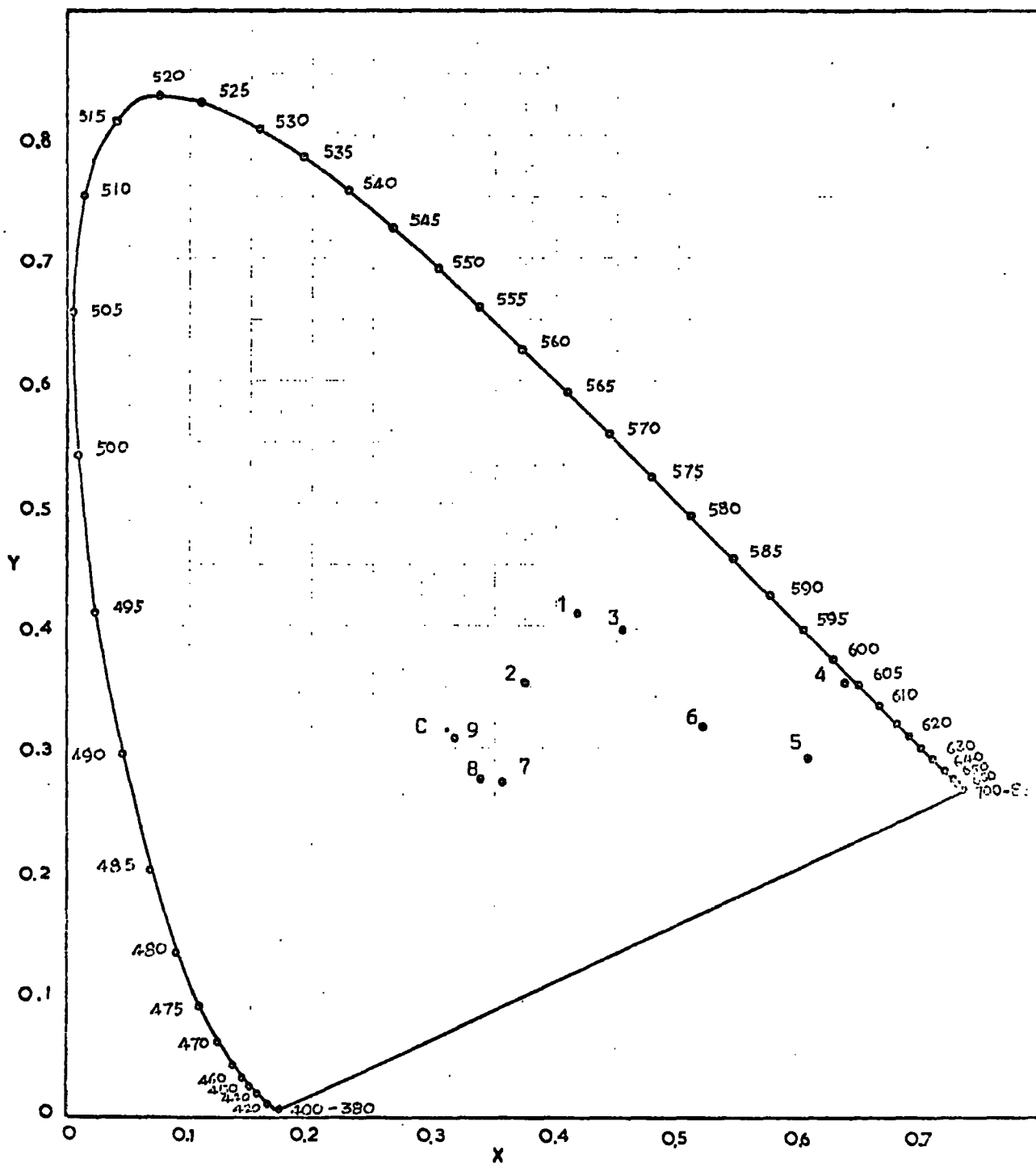
Table XI 2.

QUANTITATIVE COLOUR VALUES OF SOME CUT GEMSTONES.

	Chromaticity coordinates		Dominant or Complementary Wavelength	Excitation purity (P _e %)
	x	y		
Under Tungsten Light (Source A)				
Citrine	0.5203	0.4245	588.21	62.19
Chrysoberyl Sp1	0.5006	0.3940	591.88	38.31
Chrysoberyl Sp2	0.5556	0.4050	593.59	71.57
Spinel	0.6595	0.3397	608.32	99.65
Synthetic Ruby Sp1	0.6711	0.3064	633.26	84.55
Synthetic Ruby Sp2	0.6189	0.3417	615.97	72.92
Almandine Sp1	0.5163	0.3573	-508.66	29.80
Almandine Sp2	0.4969	0.3643	-512.05	23.99
Amethyst	0.4635	0.3940	-511.39	7.58
Under Average Daylight (Source C)				
Citrine	0.4.72	0.4100	579.46	53.86
Chrysoberyl Sp1	0.3745	0.3537	585.19	27.30
Chrysoberyl Sp2	0.4641	0.3955	586.86	62.54
Spinel	0.6407	0.3567	603.60	99.38
Synthetic Ruby Sp1	0.6063	0.2945	634.35	73.46
Synthetic Ruby Sp2	0.5197	0.3185	614.82	56.30
Almandine Sp1	0.3560	0.2731	-502.74	26.00
Almandine Sp2	0.3387	0.2746	-509.91	22.15
Amethyst	0.3186	0.3087	-502.11	4.65

* - sign indicates complementary wavelength.

Figure XI 1. Chromaticity diagram showing the colour distributions of some cut gemstones under average daylight (CIE source C).



Legen for Figure XI 1.

	Gemstones	Weight
1.	Citrine; brilliant cut,circular	1.22 carat
2.	Chrysoberyl Sp 1; mixed cut	0.58 "
3.	Chrysoberyl Sp 2; cushion-shaped cut	1.42 "
4.	Spinel; trap cut, 8-sided	1.30 "
5.	Synthetic ruby Sp 1; brilliant cut,circular	0.91 "
6.	Synthetic ruby Sp 2; mixed cut,oval	1.28 "
7.	Almandine Sp 1; cabochon with a concave base	0.99 "
8.	Almandine Sp 2; " " " " " "	0.91 "
9.	Amethyst; fancy cut, hart-shape	3.01 "

Figure XI 2.

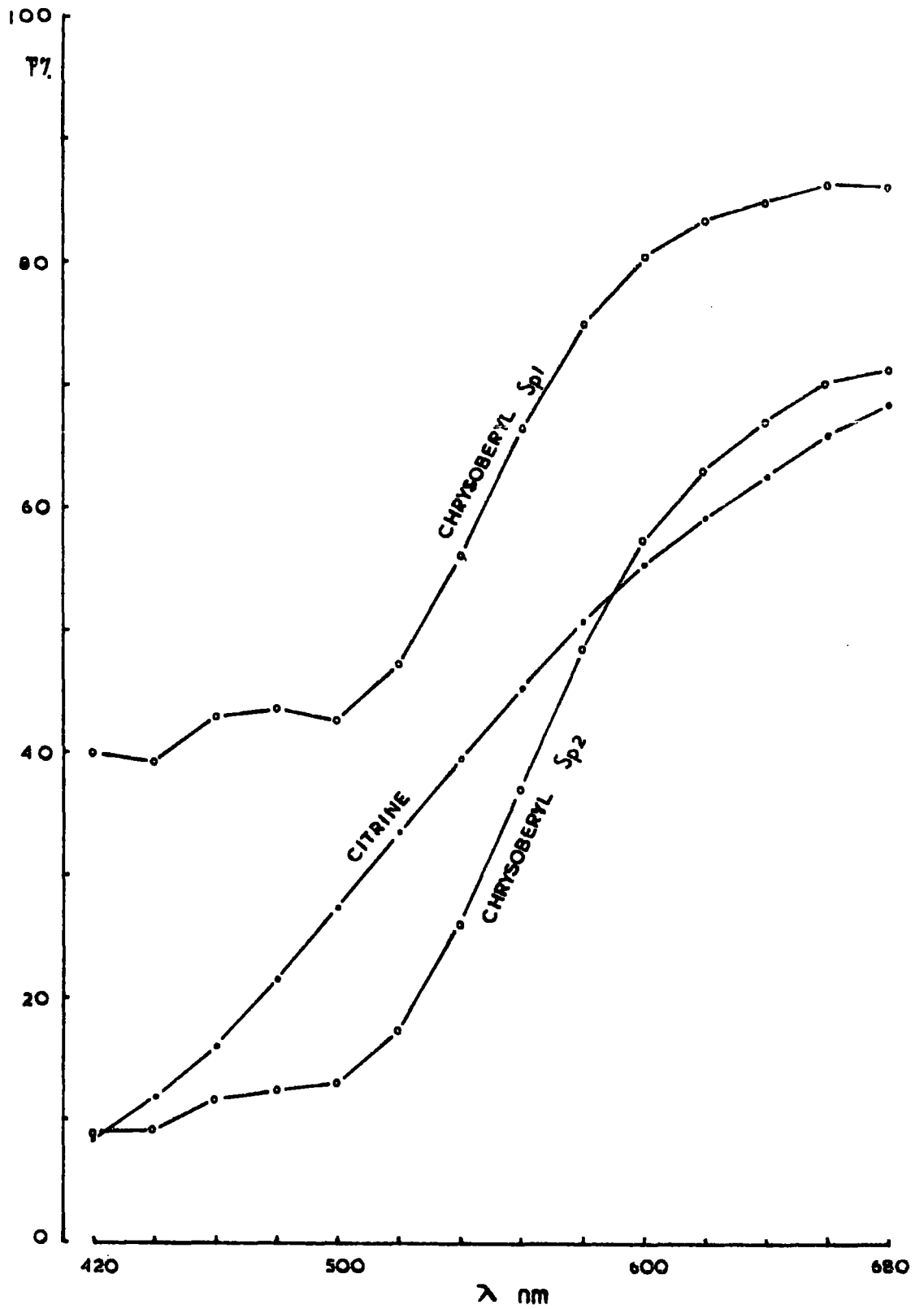


Figure XI 3.

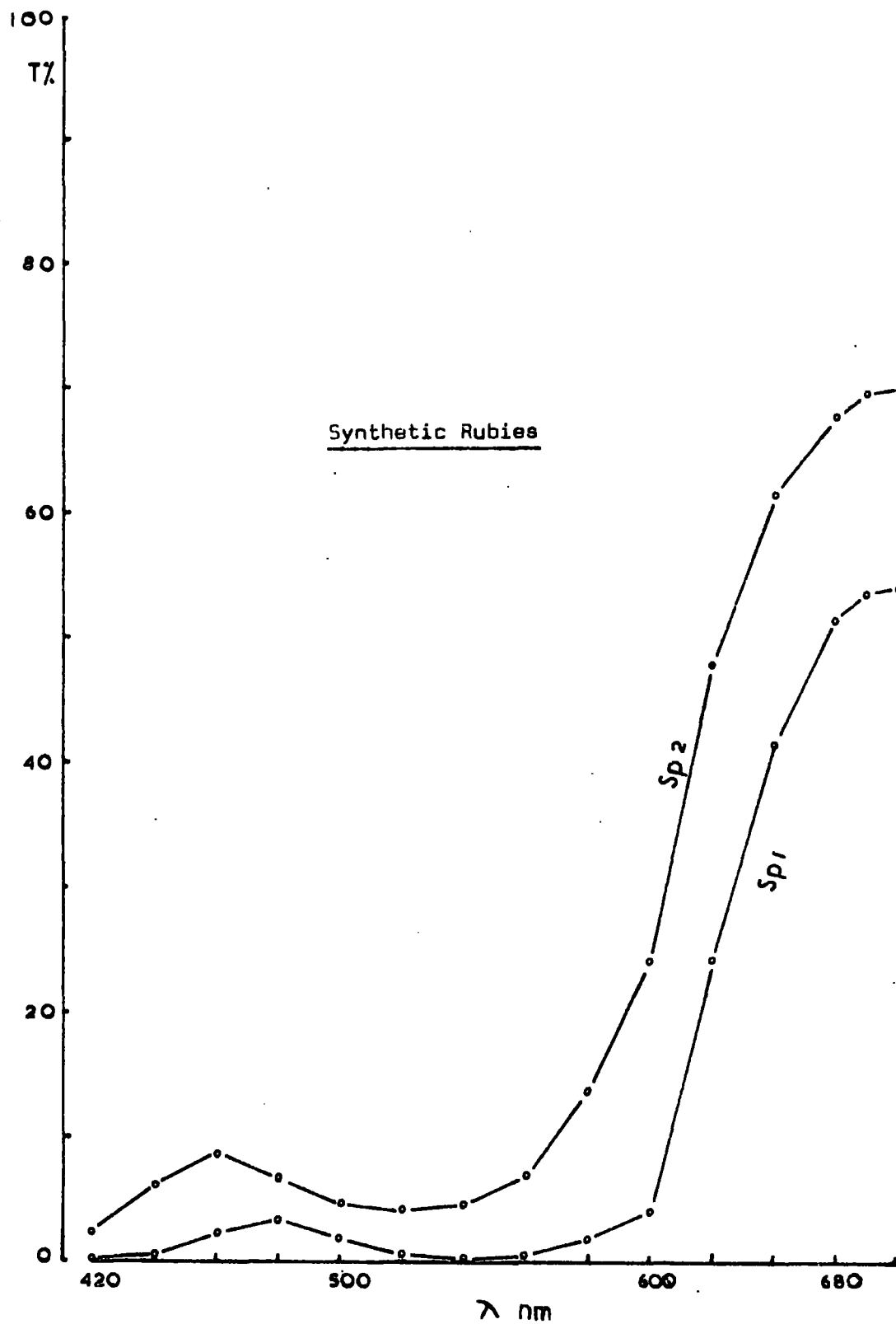
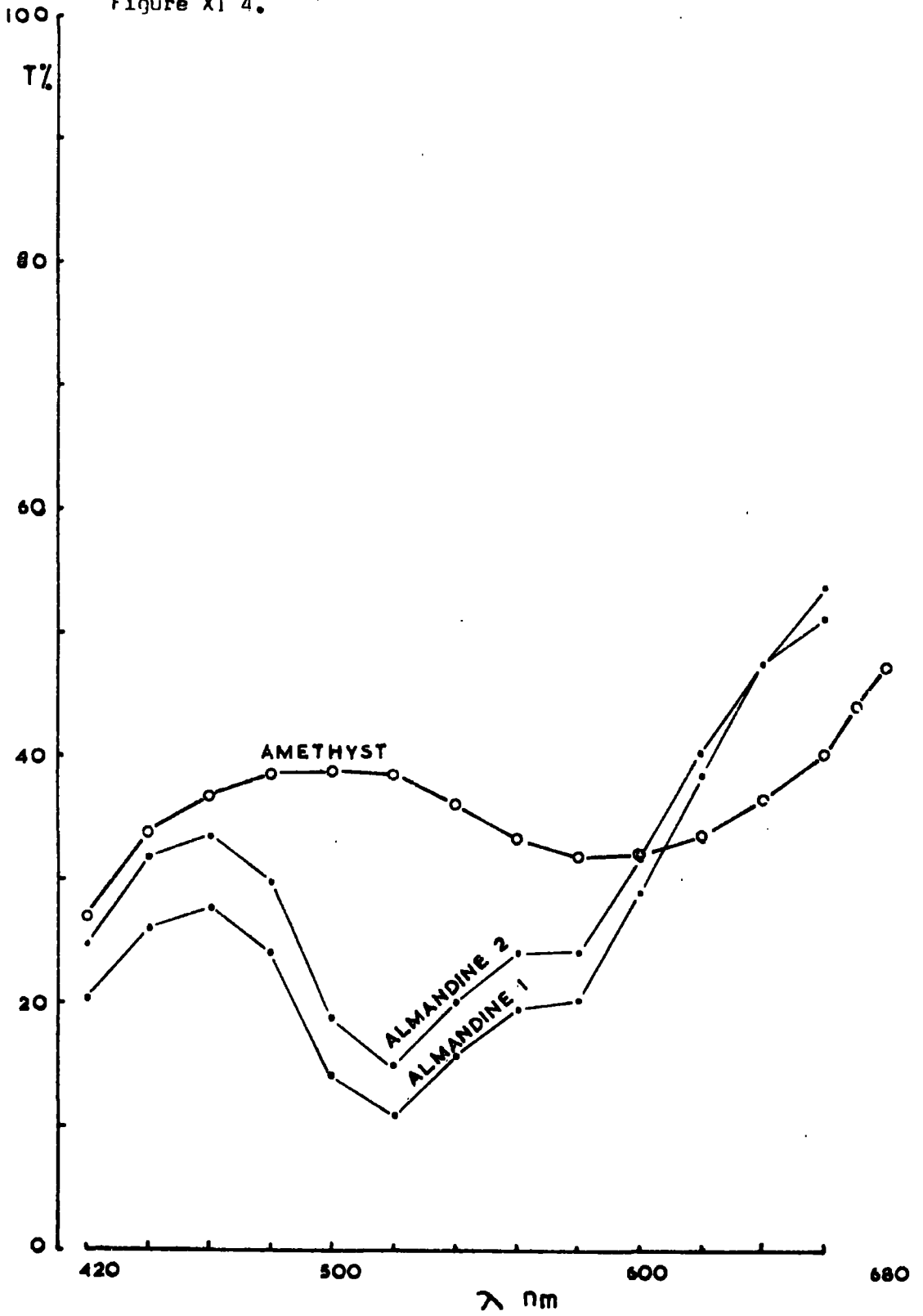


Figure XI 4.



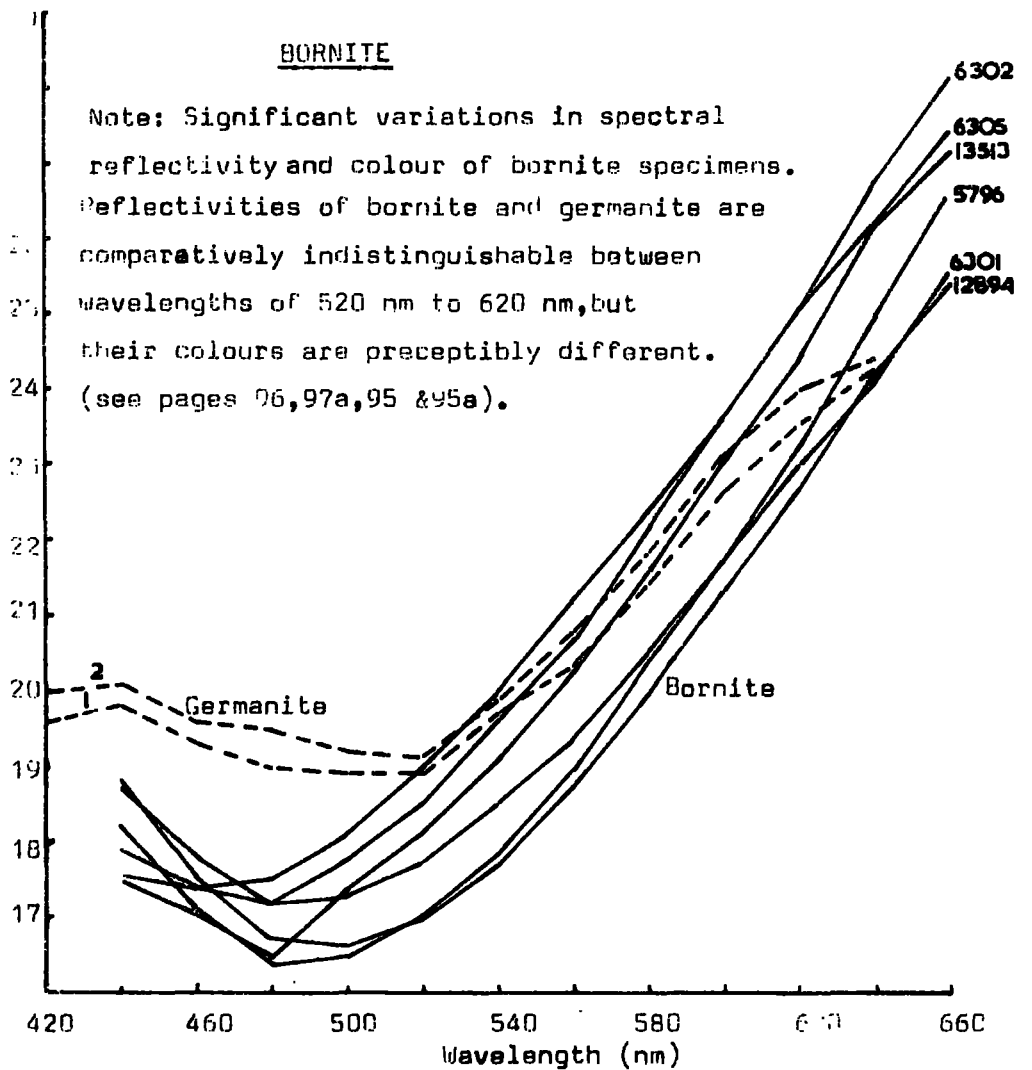
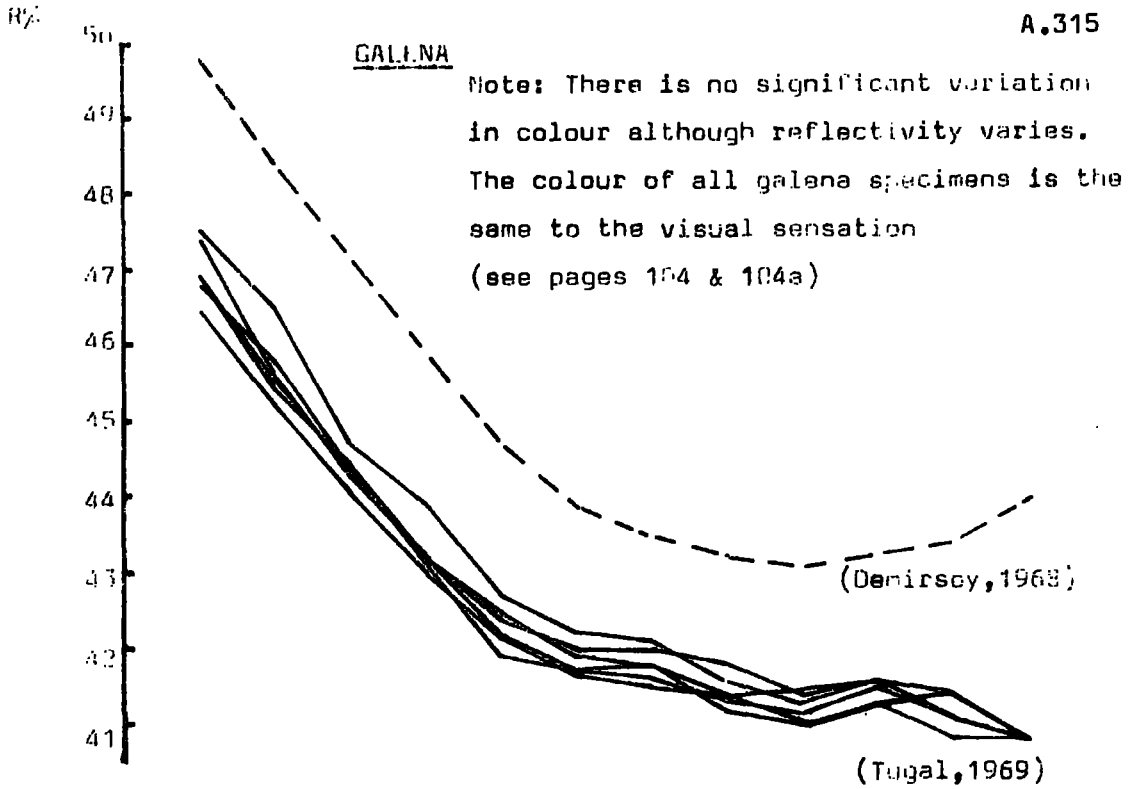


Figure XII 1. Showing marked variations in spectral reflectivity values quoted for specific minerals. (see pages 112 & 46).

```
CP:PROC OPTIONS(MAIN);
```

```

SPECR:PROC OPTIONS(MAIN);
  /*SPECR IS A PROGRAM TO COMPUTE SPECTRAL REFLECTIVITY OF
  /*MINERAL FROM GALVO READINGS.
  DCL(R(N),S(N),W(N),A(N),B(N),G(T),GG(T))CONTROLLED FLOAT;
  DCL(I,J,SUM,GG)FIXED BIN;
  DCL(M,TITLE)CHAR(50)VAR;
TAKE:GET LIST(TITLE);
  GET LIST(N);
  ALLOCATE R,S,W,A,B;
  DO I=1 TO N;
  GET LIST(S(I));
  END;
  GET LIST(T);
  GET LIST(W);
  PUT EDIT(TITLE,'SPECTRAL REFLECTIVITY')(SKIP(2),A,SKIP(2),COL
    (5.),A);
  PUT EDIT('WAVELENGTH NM',W)(SKIP(2),A,COL(22),(N)(F(5,1),X(3)))
    ;
START:GET LIST(N);
  ON ENDFILE(SCARDS)GOTO FIN;
  DO I=1 TO N;
  ALLOCATE G,GG;
  DO J=1 TO T;
  SUM=0;
  GET LIST(G(J));
  SUM=SUM+G(J);
  A(I)=SUM/T;
  END;
  DO J=1 TO T;
  SUM=0;
  GET LIST(GG(J));
  SUM=SUM+GG(J);
  B(I)=SUM/T;
  END;
  GET LIST(GO);
  R(I)=(A(I)-GO)/(B(I)-GO)*S(I);
  END;
  PUT EDIT(X,P)(SKIP(2),A,COL(22),(N)(F(5,2),X(3)));
  GOTO START;
FIN:END;

```

```
ECT:PROC OPTIONS(MAIN):
```

```
ST
```

```
REFLECT:PROC OPTIONS(MAIN);
  /*REFLECT IS A PROGRAM TO COMPUTE TRISTIMULUS VALVES AND
  /*CHROMATICITY COORDINATES OF OBJECT COLOURS BY WEIGHTED
  /*ORDINATE METHOD.
  DCL(K(M),R(M),A(N),B(N),H(N,C),D(N,C))CONTROLLED FLOAT;
  DCL(TOT(3),V(3),O(3),T,P)FLOAT;
  DCL(I,J,C,Z,M,N)FIXED BIN,(X,Y)CHAR(10)VAR;
  ON ENDFILE(SCARDS)GOTO LAST;
  /*READ IN MEASURED SPECTRAL REFLECTIVITY VALUES
RESFT:GET LIST(M,X);
  ALLOCATE K,R;
  GET LIST(K,R);
  PUT SKIP(4);
  PUT PAGE EDIT('MEASURED REFLECTIVITY','WAVELENGTH')(COL
    (49),A,SKIP(2),COL(18),A,SKIP);
  IF M<=6 THEN DO;
  1 PUT EDIT((K(I) DO I=1 TO M))(COL(38+7*I),F(3));
  1 PUT EDIT(X)(COL(18),A);
  1 PUT EDIT((R(I) DO I=1 TO M))(COL(38+7*I),F(5,2));
  1 FND;
  ELSE DO;
  1 PUT EDIT((K(I) DO I=1 TO 6))(COL(38+7*I),F(3));
  1 PUT EDIT(X)(COL(18),A);
  1 PUT EDIT((R(I) DO I=1 TO 6))(COL(38+7*I),F(5,2));
  1 IF M<=12 THEN DO;
  2 PUT SKIP(2);
  2 PUT EDIT((K(I) DO I=7 TO M))(COL(45+7*MOD(I-1,6)),F(3));
  2 PUT EDIT((R(I) DO I=7 TO M))(COL(45+7*MOD(I-1,6)),
    F(5,2));
  2 GOTO L1;
  2 END;
  1 IF (M>12)&(M<19) THEN DO;
  2 PUT SKIP(2);
  2 PUT EDIT((K(I) DO I=7 TO 12))(COL(45+7*MOD(I-1,6)),
    F(3));
  2 PUT EDIT((R(I) DO I=7 TO 12))(COL(45+7*MOD(I-1,6)),
    F(5,2));
  2 PUT SKIP(2);
  2 PUT EDIT((K(I) DO I=13 TO M))(COL(45+7*MOD(I-1,6)),
    F(3));
  2 PUT EDIT((R(I) DO I=13 TO M))(COL(45+7*MOD(I-1,6)),
    F(5,2));
  2 END;
  1 ELSE DO;
  2 PUT SKIP(2);
  2 PUT EDIT((K(I) DO I=7 TO 12))(COL(45+7*MOD(I-1,6)),
    F(3));
  2 PUT EDIT((R(I) DO I=7 TO 12))(COL(45+7*MOD(I-1,6)),
    F(5,2));
  2 PUT SKIP(2);
  2 PUT EDIT((K(I) DO I=13 TO 18))(COL(45+7*MOD(I-1,6)),
    F(3));
  2 PUT EDIT((R(I) DO I=13 TO 18))(COL(45+7*MOD(I-1,6)),
    F(5,2));
  2 PUT SKIP(2);
  2 PUT EDIT((K(I) DO I=19 TO M))(COL(45+7*MOD(I-1,6)),
```

```

          F(3));
      PUT EDIT((R(I) DO I=19 TO M))(COL(45+7*MOD(I-1,6)),
          F(5,2));
      END;
  END;
L1:GET LIST(N,C,Z);
  ALLOCATE A,B,H,D;
  GET LIST(A);
  DO I=1 TO N;
    IF A(I)<K(1) THEN
      DO;
        B(I)=R(1)+((A(I)-K(1))*(R(2)-R(1))/(K(2)-K(1)));
        GOTO KWINT;
      END;
    J=2;
  ZWGF:IF A(I)=K(J) THEN
    DO;
      B(I)=R(J);
      GOTO KWINT;
    END;
    IF A(I)<K(J) THEN
      DO;
        B(I)=R(J-1)+((A(I)-K(J-1))*(R(J)-R(J-1))/(K(J)-K(J-1)));
        GOTO KWINT;
      END;
    J=J+1;
    IF J<=M THEN
      GOTO ZWGF;
    IF A(I)>K(M) THEN
      DO;
        B(I)=R(M)+((A(I)-K(M))*(R(M)-R(M-1))/(K(M)-K(M-1)));
      END;
  KWINT:END;
  PLINK:TOT=:;
  SPLUD:GET LIST(P,Y);
  DO I=1 TO N;
    GET LIST(H(I,1),H(I,2),H(I,3));
    DO C=1 TO 3;
      D(I,C)=B(I)*H(I,C);
    END;
  END;
  PUT SKIP(3);
  PUT EDIT(X,'WAVELENGTH','REFLECTIVITY','ENERGY ',
    'DISTRIBUTION')(COL(25),A,SKIP(2),COL(25),A,
    COL(34),A,COL(58),A,A);
  DO I=1 TO N;
    PUT FDI(A(I),B(I),(D(I,C) DO C=1 TO 3))(COL(24),
    F(3),COL(37),F(6,3),COL(52),F(8,4),2(F(12,4)));
  DO C=1 TO 3;
    TOT(C)=TOT(C)+D(I,C);
  END;
  END;
  DO C=1 TO 3;
    V(C)=TOT(C)/P;

```

Appendix VI.1 a. (continued)

```

END:
T=SUM(V):
DO C=1 TO 2:
  Q(C)=V(C)/T:
END:
OUT EDIT(1SUMS,(EDIT(C) DO C=1 TO 2))(COL(10),A,COL(11),
      F(C,4),R(F(12,4))):
OUT EDIT(X,IXI,IXI,IXI,IXI,ITERIMINUM VALUE Q,(V(C) DO C=1 TO
      2),SCHEMATICITY COORDINATES,(Q(C) DO C=1 TO 2))
      (SKIP(1),COL(20),A,SKIP(2),COL(50),A,COL(60),A,
      COL(70),A,SKIP(2),COL(70),A,R(F(12,4)),SKIP(2),
      COL(20),A,R(F(12,4))):
Z=7 1:
IF Z=0 THEN
DO:
  ERSE A,B,K,D:
  GOTO RESET:
END:
ELSE GOTO OLONK:
LAST:END PEEECT:

```

Example of listing of the programme "REFLECT" of Appendix VI 1 a.

12 (number of wavelenghts used in the measurement of R% or T%)

'BORNITE SP. 6305(W.H, 1971)'

440 460 480 to 620 640 660

17.55 17.42 17.53 etc.

81 (number of weighted ordinates from 380 to 780 nm)

3 (for tristimulus values X,Y,Z)

2 (for two light sources)

380 385 390 to 770 775 780

'1931 CIE STANDARD SOURCE A'

0.0006 0.0000 0.0029

0.0011 0.0000 0.0053

0.0024 0.0000 0.0113

(colour-matching functions weighted by
relative spectral distribution of CIE
source A)

etc.

0.0011 0.0000 0.0000

0.0011 0.0000 0.0000

0.0000 0.0000 0.0000

0.0000 0.0000 0.0000

'1931 CIE STANDARD SOURCE C'

0.0022 0.0000 0.0101

0.0041 0.0002 0.0197

(colour-matching functions weighted by
relative spectral distribution of CIE
source C)

etc.

0.0003 0.0000 0.0000

0.0000 0.0000 0.0000

14

'BORNITE SP. NGD 6(TUGAL, 1969)'

440 460 480 to 660 680 700

16.70 16.60 17.26 etc.

81

3

2

380 385 390 395 to 770 775 780

'1931 CIE STANDARD SOURCE A'

0.0006 0.0000 0.0029

0.0011 0.0000 0.0053

etc.

0.0000 0.0000 0.0000

'1931 CIE STANDARD SOURCE C'

0.0022 0.0000 0.0101

0.0041 0.0002 0.0197

and so on.

'SOURCE A COMBINED WITH NEUTRAL FILTER(20%)(x=0.4473,y=0.4196)'

0.4473 0.4196

320 (number of wavelengths)

380 0.17411 0.00496

381 0.17409 0.00496

382 0.17407 0.00497

etc. (wavelength & chromaticity coordinates of
equal-energy spectrum)

697 0.73465 0.26535

698 0.73467 0.26533

699 0.73469 0.26531

Table VI 1. Slopes of Lines of Constant Dominant Wavelength for C.I.C. Standard Sources A,B,C and Source E (Equal-energy, Wavelength basis) (from Judd 1933)

Source A		Source B		Wave-length (nm)	Source C		Source E	
r ₁	r ₂	r ₁	r ₂		r ₁	r ₂	r ₁	r ₂
+0.67950		+0.50303		380	+0.43688		+0.48508	
0.67954		0.50307		381	0.43693		0.48513	
0.67957		0.50311		382	0.43698		0.48517	
0.67963		0.50319		383	0.43706		0.48525	
0.67968		0.50326		384	0.43714		0.48532	
+0.67972		+0.50330		385	+0.43719		+0.48537	
0.67980		0.50340		386	0.43731		0.48548	
0.67986		0.50347		387	0.43739		0.48555	
0.67991		0.50355		388	0.43747		0.48563	
0.68000		0.50365		389	0.43759		0.48574	
+0.68008		+0.50375		390	+0.43770		+0.48584	
0.68016		0.50385		391	0.43782		0.48595	
0.68024		0.50395		392	0.43793		0.48606	
0.68035		0.50408		393	0.43808		0.48620	
0.68046		0.50421		394	0.43822		0.48633	
+0.68052		+0.50430		395	+0.43832		+0.48643	
0.68066		0.50445		396	0.43850		0.48659	
0.68076		0.50458		397	0.43865		0.48673	
0.68087		0.50471		398	0.43879		0.48687	
0.68102		0.50489		399	0.43899		0.48705	
+0.68115		+0.50504		400	+0.43917		+0.48722	
0.68130		0.50522		401	0.43936		0.48740	
0.68143		0.50538		402	0.43954		0.48757	
0.68157		0.50553		403	0.43971		0.48774	
0.68171		0.50571		404	0.43991		0.48792	
+0.68189		+0.50591		405	+0.44013		+0.48813	
0.68202		0.50607		406	0.44031		0.48830	
0.68222		0.50630		407	0.44057		0.48854	
0.68241		0.50651		408	0.44081		0.48877	
0.68265		0.50679		409	0.44111		0.48906	
+0.6829		+0.5071		410	+0.4414		+0.4893	
0.6831		0.5074		411	0.4417		0.4897	
0.6834		0.5076		412	0.4421		0.4900	
0.6836		0.5079		413	0.4424		0.4903	
0.6839		0.5082		414	0.4427		0.4906	
+0.6841		+0.5085		415	+0.4430		+0.4909	
0.6846		0.5089		416	0.4435		0.4913	
0.6848		0.5092		417	0.4438		0.4916	
0.6855		0.5100		418	0.4446		0.4924	
0.6857		0.5102		419	0.4449		0.4927	
+0.6864		+0.5110		420	+0.4457		+0.4935	
0.6870		0.5117		421	0.4465		0.4942	
0.6877		0.5124		422	0.4473		0.4950	
0.6886		0.5133		423	0.4482		0.4959	
0.6892		0.5140		424	0.4490		0.4966	

Table VI 1 (continued)

A.325

r_1	r_2	r_1	r_2	r_1	r_2	r_1	r_2
+0.6903		+0.5152	425	+0.4502		+0.4979	
0.6918		0.5163	426	0.4515		0.4991	
0.6923		0.5172	427	0.4524		0.5000	
0.6933		0.5184	428	0.4537		0.5012	
0.6944		0.5196	429	0.4550		0.5024	
+0.6957		+0.5209	430	+0.4564		+0.5038	
0.6972		0.5225	431	0.4581		0.5055	
0.6988		0.5241	432	0.4598		0.5072	
0.7000		0.5254	433	0.4613		0.5086	
0.7020		0.5275	434	0.4635		0.5108	
+0.7037		+0.5293	435	+0.4654		+0.5126	
0.7056		0.5314	436	0.4676		0.5148	
0.7074		0.5332	437	0.4695		0.5167	
0.7095		0.5354	438	0.4719		0.5190	
0.7115		0.5375	439	0.4742		0.5212	
+0.7141		+0.5402	440	+0.4771		+0.5240	
0.7165		0.5428	441	0.4793		0.5267	
0.7191		0.5455	442	0.4827		0.5296	
0.7215		0.5481	443	0.4855		0.5323	
0.7244		0.5511	444	0.4888		0.5354	
+0.7277		+0.5546	445	+0.4926		+0.5391	
0.7310		0.5581	446	0.4964		0.5428	
0.7335		0.5617	447	0.5002		0.5465	
0.7382		0.5657	448	0.5045		0.5507	
0.7424		0.5702	449	0.5094		0.5555	
+0.7465		+0.5746	450	+0.5141		+0.5600	
0.7508		0.5791	451	0.5190		0.5648	
0.7556		0.5842	452	0.5244		0.5701	
0.7602		0.5891	453	0.5297		0.5753	
0.7655		0.5947	454	0.5358		0.5811	
+0.7708		+0.6003	455	+0.5419		+0.5871	
0.7766		0.6065	456	0.5486		0.5935	
0.7826		0.6129	457	0.5555		0.6003	
0.7894		0.6201	458	0.5633		0.6079	
0.7963		0.6273	459	0.5711		0.6155	
+0.8036		+0.6351	460	+0.5796		+0.6236	
0.8110		0.6429	461	0.5881		0.6319	
0.8192		0.6516	462	0.5975		0.6410	
0.8281		0.6611	463	0.6073		0.6510	
0.8362		0.6717	464	0.6192		0.6622	
+0.8490		+0.6831	465	+0.6317		+0.6743	
0.8610		0.6958	466	0.6455		0.6877	
0.8747		0.7103	467	0.6612		0.7030	
0.8899		0.7263	468	0.6788		0.7200	
0.9062		0.7435	469	0.6976		0.7382	
+0.9251		+0.7635	470	+0.7195		+0.7594	
0.9455		0.7852	471	0.7434		0.7825	
0.9682		0.8094	472	0.7702		0.8084	
0.9935	+1.0066	0.8364	473	0.8002		0.8372	
+1.0217	0.9788	0.8669	474	0.8342		0.8699	

Table VI 1 (continued)

r_1	r_2	r_1	r_2	r_1	r_2	r_1	r_2
	+0.913	+0.9018		475	+0.8736	+0.9075	
	0.9168	0.9121		476	0.9193	0.9510	+1.0515
	0.8812	0.9879	+1.0122	477	0.9719	+1.0009	-0.9991
	0.8179	+1.0405	0.9611	478	+1.0328		0.9549
	0.8167		0.9076	479		0.9050	0.8883
	+0.7713		+0.8515	480		+0.8391	+0.8290
	0.7296		0.7927	481		0.7705	0.7670
	0.6511		0.7322	482		0.7602	0.7033
	0.6110		0.6695	483		0.6277	0.6374
	0.5914		0.6056	484		0.5543	0.5704
	+0.5158		0.5397	485	+0.4789		+0.5013
	0.4953		0.4717	486	+0.4015		+0.4302
	0.4433		0.4023	487	+0.3227		+0.3577
	0.3899		0.3315	488	+0.2428		+0.2838
	0.3353		0.2596	489	+0.1619		+0.2089
	+0.2797		+0.1871	490	+0.0805		+0.1373
	+0.2224		+0.1127	491	-0.0126		+0.0560
	+0.1638		+0.0371	492	-0.0869		-0.0225
	+0.1051		-0.0382	493	-0.1706		-0.1008
	+0.0464		-0.1131	494	-0.2537		-0.1765
	-0.0123		-0.1877	495	-0.3367		-0.2530
	-0.0498		-0.2619	496	-0.4185		0.3329
	-0.1287		-0.3350	497	0.4993		0.4687
	-0.1860		-0.4074	498	0.5793		0.4838
	-0.2423		-0.4784	499	0.6579		0.5574
	-0.2979		-0.5486	500	-0.7357		-0.6304
	0.3519		0.6169	501	0.8114		0.7013
	0.4050		0.6842	502	0.8863		0.7714
	0.4569		0.7504	503	-1.0415	0.9601	0.8403
	0.5075		0.8153	504	0.9681	-1.0530	0.9081
	-0.5571		-0.8796	505	-0.9010		-1.0252
	0.6062	-1.0601	0.9433	506	0.8490		0.9594
	0.6539	0.9939	-1.0061	507	0.8012		0.9021
	0.7006	0.9359		508	0.7567		0.8516
	0.7459	0.8850		509	0.7178		0.8068
	-0.7902	-0.8396		510	-0.6826		-0.7666
	0.8329	0.7992		511	0.6507		0.7304
	0.8742	0.7629		512	0.6216		0.6977
	0.9143	0.7298		513	0.5947		0.6677
	0.9530	0.6998		514	0.5699		0.6403
-1.0104	-0.9897	-0.6726		515	-0.5471		-0.6153
0.9767	-1.0239	0.6483		516	0.5263		0.5928
0.9673		0.6262		517	0.5072		0.5722
0.9206		0.6057		518	0.4890		0.5528
0.8969		0.5865		519	0.4718		0.5367
-0.8757		-0.5688		520	-0.4557		-0.5178
0.8568		0.5522		521	0.4403		0.5019
0.8399		0.5368		522	0.4258		0.4870
0.8244		0.5221		523	0.4117		0.4726
0.8101		0.5079		524	0.3979		0.4587

Table VI 1 (continued)

r_1	r_2	r_1	r_2	r_1	r_2	r_1	r_2
-0.7083		-0.4958		525	-0.3852	-0.3563	
0.7073		0.4902		526	0.3775	0.3513	
0.7064		0.4844		527	0.3712	0.3477	
0.7055		0.4784		528	0.3659	0.3445	
0.7047		0.4733		529	0.3606	0.3413	
-0.7037		-0.4677		530	-0.3574	-0.3376	
0.7028		0.4617		531	0.3543	0.3352	
0.7019		0.4568		532	0.3513	0.3323	
0.7021		0.4579		533	0.3467	0.3394	
0.6913		0.4749		534	0.2650	0.3264	
-0.6908		-0.4619		535	-0.2519	-0.3135	
0.6899		0.4590		536	0.2486	0.3065	
0.6890		0.4457		537	0.2352	0.2872	
0.6881		0.4233		538	0.2111	0.2737	
0.6389		0.4888		539	0.1977	0.2602	
-0.6386		-0.2953		540	-0.1828	-0.2566	
0.6379		0.2812		541	0.1694	0.2525	
0.6073		0.2671		542	0.1548	0.2484	
0.5962		0.2523		543	0.1399	0.2344	
0.5851		0.2473		544	0.1243	0.1884	
-0.5739		-0.2220		545	-0.1074	-0.1729	
0.5625		0.2063		546	-0.0926	-0.1573	
0.5504		0.1899		547	-0.0789	-0.1417	
0.5387		0.1730		548	-0.0656	-0.1260	
0.5257		0.1558		549	-0.0440	-0.0967	
-0.5136		-0.1377		550	-0.0224	-0.0680	
0.4989		-0.1189		551	-0.0033	-0.0490	
0.4849		-0.0996		552	+0.0159	-0.0306	
0.4703		-0.0792		553	+0.0365	-0.0161	
0.4547		-0.0583		554	+0.0575	-0.0096	
-0.4387		-0.0365		555	+0.0744	-0.0011	
0.4217		-0.0133		556	0.1026	+0.0071	
0.4036		+0.0109		557	0.1295	+0.0167	
0.3847		+0.0359		558	0.1512	+0.0333	
0.3647		+0.0626		559	0.1704	+0.1094	
-0.3453		+0.0902		560	+0.1777	+0.1364	
0.3259		0.1193		561	0.2327	0.1667	
0.2775		0.1803		562	0.2627	0.1949	
0.2165		0.1926		563	0.2971	0.2261	
0.2433		0.2168		564	0.3264	0.2591	
-0.2134		+0.2330		565	+0.2608	+0.2104	
-0.1614		0.2915		566	0.2969	0.2407	
-0.1469		0.3323		567	0.3350	0.2701	
-0.1092		0.3757		568	0.3752	0.3007	
-0.0681		0.4221		569	0.5177	0.3704	
-0.0236		+0.4709		570	+0.5400	+0.4000	
+0.0167		0.5127		571	0.6600	0.4700	
+0.0760		0.5768		572	0.7087	0.6000	
+0.1377		0.6394		573	0.7119	0.6564	
+0.2033		0.7039		574	0.7679	0.7154	

Table VI 1 (continued)

r_1	r_2	r_1	r_2		r_1	r_2	r_1	r_2
+0.2768		+0.7733		575	+0.8276		+0.7788	
0.3888		0.8479		576	0.8964		0.8456	
0.4571		0.9290	+1.0764	577	0.9580	+1.0139	0.9180	
0.5574		+1.0162	0.9841	578	+1.0294	0.9716	0.9952	+1.0041
0.6791			0.8996	579		0.9039	+1.0788	0.9269
+0.8205			+0.8276	580		+0.8114		+0.8554
0.9862	+1.0170		0.7521	581		0.7833		0.7894
+1.1818	0.5192		0.6877	582		0.7295		0.7289
	0.7053		0.6285	583		0.6793		0.6729
	0.5653		0.5737	584		0.6322		0.6267
	+0.4825		+0.5232	585		+0.5881		+0.5724
	0.4936		0.4765	586		0.5175		0.5276
	0.3187		0.4332	587		0.5000		0.4857
	0.2463		0.3925	588		0.4721		0.4601
	0.1889		0.3552	589		0.4392		0.4161
	+0.1509		+0.3198	590		+0.4070		+0.3711
	+0.0917		0.2869	591		0.3769		0.3403
	+0.0261		0.2566	592		0.3490		0.3146
	-0.0371		0.2277	593		0.3222		0.2862
	-0.0880		0.2011	594		0.2976		0.2579
	-0.0708		+0.1761	595		+0.2739		+0.2311
	-0.1001		0.1530	596		0.2521		0.2112
	-0.1276		0.1316	597		0.2318		0.1899
	-0.1516		0.1114	598		0.2125		0.1666
	-0.1744		0.0923	599		0.1943		0.1408
	-0.1951		+0.0747	600		+0.1773		+0.1332
	0.2148		+0.0576	601		0.1609		0.1161
	0.2336		+0.0418	602		0.1455		0.1002
	0.2507		+0.0266	603		0.1306		0.0847
	0.2654		+0.0122	604		0.1167		0.0701
	-0.2797		-0.0010	605		+0.1038		+0.0572
	0.2926		-0.0132	606		0.0918		-0.0469
	0.3051		-0.0251	607		0.0802		-0.0329
	0.3166		-0.0360	608		0.0693		-0.0216
	0.3271		-0.0462	609		0.0593		+0.0117
	-0.3368		-0.0558	610		+0.0503		+0.0018
	0.3461		0.0649	611		+0.0421		-0.0095
	0.3549		0.0736	612		+0.0342		-0.0162
	0.3628		0.0815	613		+0.0271		-0.0233
	0.3703		0.0891	614		+0.0206		-0.0320
	-0.3776		-0.0965	615		+0.0142		-0.0395
	0.3845		0.1033	616		+0.0081		0.0464
	0.3902		0.1094	617		-0.0017		0.0526
	0.3961		0.1154	618		-0.0098		0.0588
	0.4016		0.1211	619		-0.0156		0.0646
	-0.4067		-0.1265	620		-0.0210		-0.0701
	0.4111		0.1313	621		0.0258		0.0750
	0.4157		0.1361	622		0.0306		0.0798
	0.4199		0.1405	623		0.0351		0.0844
	0.4238		0.1447	624		0.0394		0.0886

Table VI 1 (continued)

r_1	r_2	r_1	r_2	r_1	r_2	r_1	r_2
-0.1277		-0.1155	625	-0.0635		-0.0929	
0.1313		0.1527	626	0.0171		0.0901	
0.1336		0.1562	627	0.0511		0.1165	
0.1357		0.1596	628	0.0544		0.1035	
0.1369		0.1631	629	0.0580		0.1074	
-0.1317		-0.1661	630	-0.0611		-0.1105	
0.1355		0.1691	631	0.0641		0.1136	
0.1392		0.1721	632	0.0672		0.1167	
0.1417		0.1748	633	0.0700		0.1195	
0.1442		0.1776	634	0.0727		0.1223	
-0.1465		-0.1809	635	-0.0753		-0.1248	
0.1487		0.1825	636	0.0775		0.1273	
0.1467		0.1847	637	0.0800		0.1296	
0.1467		0.1869	638	0.0823		0.1319	
0.1447		0.1891	639	0.0846		0.1341	
-0.1465		-0.1911	640	-0.0866		-0.1362	
0.1482		0.1931	641	0.0886		0.1382	
0.1496		0.1947	642	0.0903		0.1399	
0.1712		0.1965	643	0.0921		0.1417	
0.1725		0.1980	644	0.0936		0.1432	
-0.1739		-0.1995	645	-0.0952		-0.1448	
0.1752		0.2010	646	0.0967		0.1463	
0.1763		0.2022	647	0.0980		0.1476	
0.1775		0.2035	648	0.0993		0.1489	
0.1786		0.2048	649	0.1006		0.1502	
-0.1795		-0.2058	650	-0.1017		-0.1513	
0.1805		0.2069	651	0.1028		0.1524	
0.1814		0.2079	652	0.1039		0.1535	
0.1821		0.2088	653	0.1047		0.1543	
0.1831		0.2098	654	0.1058		0.1554	
-0.1838		-0.2106	655	-0.1066		-0.1562	
0.1845		0.2115	656	0.1075		0.1571	
0.1851		0.2121	657	0.1081		0.1577	
0.1858		0.2129	658	0.1090		0.1586	
0.1864		0.2135	659	0.1096		0.1592	
-0.1869		-0.2142	660	-0.1103		-0.1599	
0.1873		0.2146	661	0.1107		0.1603	
0.1878		0.2152	662	0.1113		0.1609	
0.1882		0.2156	663	0.1117		0.1613	
0.1885		0.1260	664	0.1122		0.1618	
-0.1889		-0.2164	665	-0.1126		-0.1622	
0.1892		0.2167	666	0.1130		0.1626	
0.1896		0.2171	667	0.1134		0.1630	
0.1899		0.2174	668	0.1139		0.1634	
0.1901		0.2178	669	0.1141		0.1637	
-0.1905		-0.2183	670	-0.1145		-0.1641	
0.1907		0.2185	671	0.1147		0.1643	
0.1910		0.2189	672	0.1151		0.1647	
0.1912		0.2191	673	0.1153		0.1649	
0.1916		0.2195	674	0.1157		0.1653	

Table VI 1 (continued)

r_1	r_2	r_1	r_2	r_1	r_2	r_1	r_2
-0.4918		-0.2197	675	-0.1159		-0.1655	
0.4921		0.2201	676	0.1164		0.1660	
0.4923		0.2203	677	0.1166		0.1662	
0.4925		0.2205	678	0.1168		0.1664	
0.4928		0.2209	679	0.1172		0.1668	
-0.49300		-0.22110	680	-0.11741		-0.16700	
0.49321		0.22134	681	0.11766		0.16725	
0.49343		0.22158	682	0.11791		0.16750	
0.49362		0.22180	683	0.11814		0.16773	
0.49382		0.22203	684	0.11837		0.16796	
-0.49401		-0.22225	685	-0.11860		-0.16819	
0.49419		0.22245	686	0.11881		0.16839	
0.49435		0.22263	687	0.11899		0.16858	
0.49451		0.22281	688	0.11918		0.16877	
0.49465		0.22297	689	0.11935		0.16893	
-0.49477		-0.22311	690	-0.11949		-0.16908	
0.49488		0.22324	691	0.11962		0.16920	
0.49496		0.22334	692	0.11972		0.16931	
0.49503		0.22342	693	0.11980		0.16939	
0.49510		0.22350	694	0.11987		0.16947	
-0.49514		-0.22354	695	-0.11993		-0.16951	
0.49519		0.22360	696	0.11999		0.16957	
0.49521		0.22362	697	0.12001		0.16960	
0.49523		0.22364	698	0.12003		0.16962	
-0.49525		-0.22366	699	-0.12005		-0.16964	

(see pages 74 to 76, Chapter VI)

```

M:PROC OPTIONS(MAIN):

```

```

EST

```

```

CHROM:PROC OPTIONS(MAIN):
  /*CHROM IS A PROGRAM TO CALCULATE DOMINANT WAVELENGTH,
  /*CHROMATICITY COORDINATES AND EXCITATION PURITY PERCENTAGE.
  DCL(X(M),Y(M),T(N,2),W(N),C(N),D(N),P(M),Z(M),E(M),F(M),P(M),
    Q(M),V(M),G(M),H(M))CONTROLLED FLOAT;
  DCL(A,B,IND,SX,SY,PX,PY,VV,GG,HH)FLOAT;
  DCL(M,N,I,J,K)FIXED BIN;
  DCL(COL(M))CONTROLLED FIXED BIN;
  DCL(NAME CHAR(10))VAR;
  DCL(TITLE(M)CHAR(50))CONTROLLED VAR;
  ON UNDERFLOW PUT DATA;
  GET LIST(NAME);
  PUT PAGE EDIT(NAME)(COLUMN(20),A);
  /* READ IN NUMBERS OF MINERAL SAMPLES
  GET LIST(M);
  /* READ IN CHROMATICITY COORDINATES OF STANDARD SOURCE
  GET LIST(A,B);
  /* ALLOCATE ARRAY STORAGE
  ALLOCATE X,Y,R,Z,E,F,P,Q,TITLE,COL;
  /* READ IN CHROMATICITY COORDINATES OF MINERAL SAMPLES AND
  /* TITLES
  PUT EDIT('CHROMATICITY COORDINATES')(COLUMN(6),A);
  PUT EDIT('SAMPLE','X','Y')(COLUMN(20),A,COLUMN(66),A,COLUMN(76)
    ,A);
  DO I=1 TO M;
  GET LIST(TITLE(I));
  GET LIST(X(I),Y(I));
  PUT EDIT(TITLE(I),X(I),Y(I))(SKIP(2),COLUMN(20),A,COLUMN(63),F
    (6,4),COLUMN(76),F(6,4));
  END;
  /* COMPUTE COORDINATE RATIOS
  DO I=1 TO M;
  IF ABS(X(I)-A)<ABS(Y(I)-B) THEN DO;
  R(I)=(X(I)-A)/(Y(I)-B);
  COL(I)=1;
  END;
  ELSE DO;
  R(I)=(Y(I)-B)/(X(I)-A);
  COL(I)=2;
  END;
  PUT SKIP EDIT(I,R(I))(X(5),F(2),X(5),F(7,3));
  END;
  /* READ IN NUMBER OF TABULATED COORDINATE RATIOS,WAVELENGTH
  /* VALUES, AND CHROMATICITY COORDINATES
  GET LIST(N);
  /* ALLOCATE ARRAY STORAGE
  ALLOCATE T,W,C,D;
  /* READ IN WAVELENGTHS ETC
  DO J=1 TO N;
  GET LIST(T(J,1),T(J,2),W(J),C(J),D(J));
  END;
  /* COMPUTE WAVELENGTH OF CALCULATED MINERAL COORDINATE RATIOS
  /* FROM TABULATED VALUES
  Z=0;
  DO I=1 TO M;
  K=COL(I);

```

ST

```

1      DO J=1 TO N-1;
2      IF T(J,K)>R(I)& T(J+1,K)<R(I) THEN GOTO L1;
2      ELSE IF T(J,K)<R(I)& T(J+1,K)>R(I) THEN GOTO L1;
2      ELSE GOTO L2;
2      L1:Z(I)=(R(I)-T(J,K))/(T(J+1,K)-T(J,K));
2      Z(I)=Z(I)+W(J);
2      L2:IF T(J,K)=R(I) THEN Z(I)=W(J);
2      END;
1      PUT SKIP EDIT(I,Z(I))(X(5),F(2),X(5),F(7,3));
1      END;
1      DO I=1 TO M;
1      IF Z(I)=0 THEN
1      DO;
2      DO J=1 TO M;
3      TITLE(J)=TITLE(J+1);
3      Z(J)=Z(J+1);
3      X(J)=X(J+1);
3      Y(J)=Y(J+1);
3      END;
2      END;
1      END;
1      J=0;
1      DO I=1 TO M;
1      IF Z(I)=0 THEN
1      J=J+1;
1      END;
1      M=M-J;
1      /* COMPUTE DOMINANT WAVELENGTH COORDINATES */
1      DO I=1 TO M;
2      IF ABS(Z(I)-TRUNC(Z(I)))>1 THEN DO;
2      F(I)=Z(I)-TRUNC(Z(I));
2      J=TRUNC(Z(I))-W(1)+1;
2      E(I)=(F(I)*(C(J+1)-C(J))+C(J));
2      F(I)=Z(I)-TRUNC(Z(I));
2      F(I)=F(I)*(D(J+1)-D(J))+D(J);
2      END;
1      IF ABS(Z(I)-TRUNC(Z(I)))=0 THEN DO;
2      E(I)=C(J);
2      F(I)=D(J);
2      END;
1      END;
1      /* COMPUTE EXCITATION PURITY PERCENTAGE */
1      IND=0;
1      L3:DO I=1 TO M;
1      P(I)=(X(I)-A)*100/(F(I)-A);
1      Q(I)=(Y(I)-B)*100/(F(I)-B);
1      PUT SKIP EDIT(I,P(I),Q(I),E(I),F(I))(X(5),F(2),X(5),F(7,3),
1      X(5),F(7,3),X(5),F(7,3),X(5),F(7,3));
1      END;
1      IF IND=1 THEN GOTO L4;
1      GET LIST(SX,SY,PX,PY);
1      ALLOCATE V,G,H;
1      DO I=1 TO M;
1      IF (P(I)>0) & (Q(I)>0) THEN GOTO L5;

```

```

1      ELSE DO;
2          /*COMPUTE COORDINATES OF COMPLEMENTARY WAVLENGTHS
3          VV= SX-PX;
4          GG=(SY-PY)/VV;
5          HH=(SX*PY-PX*SY)/VV;
6          V(I)=A-X(I);
7          G(I)=(B-Y(I))/V(I);
8          H(I)=(A*Y(I)-B*X(I))/V(I);
9          E(I)=(H(I)-HH)/(GG-G(I));
10         F(I)=GG*E(I)+HH;
11     END;
12 L5:END;
13     IND=1;
14     GOTO L3;
15     /* PRINT OUT RESULTS
16 L4:PUT PAGE;
17     PUT SKIP(6);
18     PUT EDIT('WEIGHTED CRDINATE METHOD')(COL(1),A);
19     PUT EDIT(NAME)(COL(1),A);
20     PUT SKIP(2);
21     PUT EDIT('WAVELENGTH', 'COORDINATES', 'EXCITATION PURITY')
22         (COL(5),A,COL(6),A,COL(8),A);
23     PUT EDIT('NM', 'X', 'Y', 'P', 'Q')(COL(54),A,COL(6),A,COL(7),A,COL
24         (9),A,COL(10),A);
25     PUT SKIP(2);
26     DO I=1 TO M;
27     PUT EDIT(TITLL(I),Z(I),E(I),F(I),P(I),Q(I))(SKIP(2),COL(1),A,
28         COL(52),F(6,2),COL(66),F(6,4),COL(76),F(6,4),COL(9),
29         F(6,2),COL(10),F(6,2));
30     END;
31     PUT SKIP(5);
32     PUT EDIT('PROBLEM COMPLETED')(COLUMN(3),A);
33     END CHROM;

```

Example of listing of the programme "CHROM" of Appendix VI 4 a.

'1931 CIE STANDARD SOURCE A (0.4476, 0.4075)'

10 (number of specimens)

0.4476 0.4075

'BORNITE SP. 6305(W.H, 1971)'

0.4774 0.4066

'BORNITE SP. NGD 6(TUGAL, 1969)'

0.4751 0.4113

and so on

'GERMANITE SP. 1(LEVY, 1966)'

0.4666 0.4026

184 (number of wavelengths from 516 to 699 nm)

-0.9769 1 516 0.04533 0.81939

-0.9473 1 517 0.05218 0.82516

etc.

+0.9862 1 581 0.51906 0.48003

1 +0.8462 582 0.52560 0.47353

1 +0.7053 583 0.53207 0.46709

(see Table VI 1)

etc.

1 -0.49525 699 0.73469 0.26531

0.17411 0.00496 0.73469 0.26531 (chromaticity coordinates of 380 nm
and 830 nm of the spectrum)

FOR: PROC OPTIONS(MAIN):

```

COLOR: PROC OPTIONS(MAIN);
  DCL(W(N),T(N),E(N),F(N,3),P(N,3),PP(N,3),CMF(N,3))CONTROLLED
    FLOAT;
  DCL(S(3),TOT(3),TOTAL(3),X(3),Y(3),A(3),B(3),K(3),NK,Q,V,R(2),
    NF,YY(3))FLOAT;
  DCL(I,J,N)FIXED BIN(31),(TITLE)CHAR(100)VAR;
  ON ENDFILE(SCARDS)GOTO DONE;
BEGIN:GET LIST(TITLE,N);
  ALLOCATE W,T,E,F,P,PP,CMF;
  GET LIST(T,E);
  L1:DO I=1 TO N;
    GET LIST(W(I),F(I,1),F(I,2),F(I,3));
  L2:DO J=1 TO 3;
    P(I,J)=E(I)*F(I,J);
    PP(I,J)=T(I)*P(I,J);
  END L2;
  END L1;
  PUT PAGE EDIT('FILTER','WAVELENGTH','SOURCE-ENERGY',
    '1931 CIE CM FUNCTIONS','PRODUCTS','PRODUCTS',
    'T','H','X','Y','Z','XH','YH','ZH','XHT','YHT',
    'ZHT')(COL(2),A,COL(9),A,COL(26),A,COL(34),A,COL
    (73),A,COL(103),A,SKIP,COL(4),A,COL(25),A,COL(34)
    ,A,COL(44),A,COL(54),A,COL(65),A,COL(76),A,COL(87)
    ,A,COL(96),A,COL(106),A,COL(116),A);
  PUT SKIP(2);
  L3:DO I=1 TO N;
    PUT EDIT(T(I),W(I),E(I),(F(I,J) DO J=1 TO 3),
      (P(I,J) DO J=1 TO 3),(PP(I,J) DO J=1 TO 3))
      (COL(3),F(4,2),COL(13),F(3),COL(21),F(6,2),
      3(F(10,4)),3(F(11,2)),3(F(10,2)));
  END L3;
  S=0; TOT=0;
  L4:DO I=1 TO N;
  L5:DO J=1 TO 3;
    S(J)=S(J)+P(I,J);
    TOT(J)=TOT(J)+PP(I,J);
  END L5;
  END L4;
  PUT EDIT('SUMS',(S(J) DO J=1 TO 3),(TOT(J) DO J=1 TO 3))
    (SKIP(2),COL(55),A,F(9,2),2(F(11,2)),3(F(9,2)));
  /*COMPUTE NORMALIZED TRISTIMULUS VALUES AND COORDINATES OF */
  /*THE SOURCE */
  L6:DO J=1 TO 3;
    K(J)=100.0/S(J);
    IF J=2 THEN
      GOTO BUR;
  END L6;
  BUR:DO;
    NK=K(J);
  END;
  L7:DO J=1 TO 3;
    X(J)=NK*S(J);
    Y(J)=NK*TOT(J);
  END L7;
  Q=SUM(X);
  V=SUM(Y);

```

Appendix VIII 1. (continued)

A.336

```

/*CHROMATICITY COORDINATES OF SOURCE AND FILTER.
L8:DO J=1 TO 3;
    A(J)=X(J)/Q;
    B(J)=Y(J)/V;
END L8;
PUT EDIT('NORMALIZING FACTOR K = ',NK,'X',Y,'Z',
        'NORMALIZED TRISTIMULUS', '(SOURCE)', (X(J) DO J=1 TO
        3), '(FILTER)', (Y(J) DO J=1 TO 3)) (SKIP(2), COL(2), A,
        COL(44), F(7,5), SKIP, COL(59), A, COL(69), A, COL(79), A,
        SKIP, COL(2), A, COL(46), A, 3(F(11,4)), SKIP(2), COL(46),
        A, 3(F(11,4)));
PUT EDIT('CHROMATICITY COORDINATES', '(SOURCE)', (A(J) DO J=1
        TO 3), '(FILTER)', (B(J) DO J=1 TO 3)) (SKIP(2), COL(2),
        A, COL(46), A, 3(F(10,4)), SKIP(2), COL(46), A, 3(F(10,4))
        );
L9:DO J=1 TO 3;
    R(J)=100.0/TOT(J);
    IF J=2 THEN
        GOTO DUR;
END L9;
DUR:DO;
    NF=R(J);
END;
/*COMPUTE NORMALIZED TRISTIMULUS VALUES OF FILTER
L10:DO J=1 TO 3;
    YY(J)=NF*TOT(J);
END L10;
/*COMPUTE NORMALIZED COLOUR-MATCHING FUNCTIONS OF FILTER
/*COMBINED WITH CIE SOURCE.
DO I=1 TO N;
DO J=1 TO 3;
    CMF(I,J)=YY(J)*PP(I,J)/TOT(J);
END;
END;
PUT PAGE EDIT(TITLE,'WAVELENGTH', 'COLOUR MATCHING FUNCTIONS')
        (COL(10), A, SKIP(2), COL(15), A, COL(35), A);
TOTAL=0;
PUT SKIP(2);
L11:DO I=1 TO N;
    PUT EDIT(W(I), (CMF(I,J) DO J=1 TO 3)) (COL(19), F(3),
        3(F(15,4)));
L12:DO J=1 TO 3;
    TOTAL(J)=TOTAL(J)+CMF(I,J);
END L12;
END L11;
PUT EDIT('SUMS', (TOTAL(J) DO J=1 TO 3)) (SKIP(2), COL(18), A,
        3(F(14,4)));
DUNE:END COLOR;

```


REFERENCES

- Abbot, C.G; Fowle, F.E & Aldrich, L.B (1923) The Distribution of Energy in the Spectra of Sun and Stars. Smithsonian Misc.Collection, 74, No.7, Publ.No. 2714.
- Abney, W. de W. (1913) Researches in Colour Vision. Longmans Green & Co., Inc. New York.
- Akinci, O.T (1970) The Effect of Iron Substitution On the Cell Size, Reflectivity and Colour of Sphalerite. Unpub. M.Sc. Theis, Univ. of Durham.
- Aksenov, V.S; Kosyak, Ye.A. Mergenov, Shk; & Rafikov, T.K (1968) A new bismuth telluride, Bi_2Te_5 . Dokl.Acad.Sci.USSR, Earth Sci.Sect., 181, p 113-115 (1969).
- Anderson, B.W. (1964) Gem Testing. 7th. edition, A Heywood Book, Temple Press Books Ltd., London.
- Bates, F & Douglas, M.L (1967) Programming Language/One. Prentice-Hall, Inc., Englewood Cliffs, New Jersey.
- Birch, J & Wright, W.D. (1961) Color Discrimination. Physics in Med. & Biol., 6, 3.
- Bloss, F.D (1966) An Introduction to the Methods of Optical Crystallography. Holt, Rinehart & Winston. New York.
- Brett, R & Yund, R.A (1964) Sulfur-rich bornites. Am.Mineral., 49, p 1084-1098.
- Bridgeman, T (1961) The Evaluation of Tristimulus Values from Spectral Reflectances. Die Farbe, 10 Band, Heft 5/6, p 243-254
- Borodayev, Yu.S; Mozgova, N.N & Seuderova, V.M. (1968) Chalcostibite from Tereksay (Kirgiz SSR). Dokl.Acad.Se.USSR, Earth Sci.Sect., 178, p 118-121.
- Bowie, S.H.U. (1967) Microscopy: Reflected Light - in Physical Methods in Determinative Mineralogy - edited by Zussman, p 103-159.

- Bowie,S.H.U. (1967) Minutes of I.M.A/Commission on Ore Microscopy - C.O.M. - Meetings held in Cambridge on 31st.Aug. & 3rd. Sept. 1966. Econ.Geol. 62, p 278-282.
- Bowie,S.H.U. & Henry,N.F.M. (1963-64) Quantitative Measurements with the Reflecting Polarizing Microscope. Trans.Instn.Min.Metall.,73,(Bull.Instn. Min.Metall. No.689,p 467-478, 1964).
- Bowie,S.H.U. & Taylor,K. (1958) A System of Ore Mineral Identification. Mining Mag.,London, 99,p 265 ,337.
- Boynton,R.M. (1960) Theory of Color Vision. J.Opt.Soc.Am., 50, p 929-944.
- " (1963) Contributions of threshold measurements to color-discrimination theory. J.Opt.Soc.Am., 53, 165
- Brown,W.R.J. (1951) The influence of luminance level on visual sensitivity to color differences. J.Opt.Soc.Am., 41, 684.
- " (1952) The effect of field size and chromatic surroundings on colour discrimination. J.Opt.Soc.Am.,42,837.
- " (1957) Color Discrimination of Twelve Observers. J.Opt.Soc.Am., 47, p 137-143.
- Brown,W.R.J. & MacAdam,D.L. (1949) Visual Sensitivities to Combined Chromaticity and Luminance Differences. J.Opt.Soc.Am., 39, 808.
- Bouma,P.J. (1947) Physical Aspects of Colour. N.V.Philips'Gloeilampenfabrieken, Eindhoven(The Netherlands).
- Bradshaw,P.M.D. (1965):Part 1. An Absolute Method for the Measurement of Reflectivity. Unpub. Ph.D.Thesis, Univ. of Durham.
- Burton,C.J. (1969) Further Developments in Measurement of Reflectivity at Normal Incidence. Unpub. M.Sc. Thesis, Univ. of Durham.

- Burns, R.G. & Vaughan, D.T. (1970) Interpretation of the Reflectivity Behaviour of Ore Minerals. Am.Mineral. 55, p 1576-1586.
- Cameron, E.N. (1963) Optical Symmetry from Reflectivity Measurements. Am.Mineral., 48 p 1070-1079.
- Cameron, E.N. (1966) Ore Microscopy. Second Printing John Wiley & Sons, Inc. New York. London.
- Cervelle et al (1968) Sur l'inversion du signe de la bireflectance de certains mineraux absorbants cas de la mawsonite. Bull.Soc.francaise Min.et.Crist., 91.
- Commission On Ore Microscopy of the International Mineralogical Association (1970) International Tables for the Microscopic Determination of Crystalline Substances Absorbing in Visual Light. Provisional Issue - Barcelona.
- Committee on Colorimetry of the Optical Society of America. (1968) The Science of Color. Published by Opti.Soc.Am., Washington, D.C. 20037.
- Companion, A.L. (1964) Chemical Bonding. McGraw Hill, Inc., New York.
- Crawford, B.H. (1949) The Scotopic Visibility Function. Proc.Phys.Soc., B 62, 321.
- Davis, R. (1931) A Correlated Color Temperature for Illuminants. Bureau of Standards Journal of Research. 7, 659.
- Davidson, H.R. & Friede, E. (1952) The Size of Acceptable Color Differences. J.Opt.Soc.A., 43, 7, p 581-589.
- Davidson, H.R. (1951) Visual sensitivity to surface color differences. J.Opt.Soc.Am., 41, 104.
- " (1951) Calculation of Color Differences from Visual Sensitivity Ellipsoids. J.Opt.Soc.Am., 41, 12, p 1052-1056.
- " (1955) Use of Charts for Rapid Calculation of Color Differences. J.Opt.Soc.Am., 45, 617.

- Deer, Howie & Zussman (1967) An Introduction to the Rock Forming Minerals. Second Impression, Longmans, Green and Co. Ltd., London.
- De Kerf, J.L.F. (1958) Accuracy of Tristimulus Computations. *J. Opt. Soc. Am.* 48, 5, p 334-338.
- Demirsoy, S. (1968) Untersuchungen über den Einfluss der chemischen Zusammensetzung auf die spektralen Reflexionsfunktionen und Mikroemdrückharten. Thesis Aachen.
- Eales, H.V. (1967) Reflectivity of Gold & Silver Alloys. *Econ. Geol.*, 62, p 412-420.
- Eales, H.V. (1964) Mineralogy and Petrology of the Empress nickel-copper deposit, Southern Rhodesia. *Trans. Geol. Soc. S. Africa.* 67, p 173-201.
- Eales, H.V. (1961) Fineness of Gold in Some Southern Rhodesian Mines. *Trans. Inst. Min. Metall.*, 71, p 196- (Bull. Inst. Min. Metall., No. 660, Nov. p 49-73.
- Engin, T. (1969) The Geology and Geochemistry of Ultrabasic Rocks and Associated Chromite Deposits of the Andizlik-Zimparalik Area, Fethiye, Southwest Turkey. Unpub. Ph.D. Thesis, University of Durham.
- Fleischer, M. (1960) Studies of the Manganese Oxide Minerals. III Psilomelane. *Am. Mineral.*, 45, p 176-187.
- Fleischer, M. (1964) Manganese Oxide Minerals. VIII. Hollandite. *Advan. Frontiers Geol. Geophys.*, Hyderabad, 221-232.
- Folinsbee, R.E. (1949) Determination of Reflectivity of the Ore Minerals. *Econ. Geol.*, V 44, p 425-436.
- Frenzel, G. (1959) Ein neues Mineral: Idait. *Neues Jahrb. Min., Monatshefte.*, 142
- Friele, L.F.C. (1961) Analysis of the Brown and Brown-MacAdam Colour Discrimination Data. *Farbe*, 10, 193.

- Friele, L.F.C. (1965) Further analysis of colour discrimination data. Proc.Int.Colour Meeting (Lacarne, 1965).
- Fukuda, T & Sugiyama, Y. (1961) The Measurement of Daylight Fluorescent Color. Farbe, 10, 73.
- Gaines, R.V. (1969) Luzonite, farnatinitite, and some related minerals. Am.Mineral., 42, p 766-779.
- Gamyanin, G.N. (1968) Bismuth sulfotellurides from Northern Yakutia. Dokl.Acad.Sci. USSR, Earth Science Sect., 178, p 121-124.
- Gibson, K.S. & Tyndall, E.P.T. (1923) Visibility of Radiant Energy. Bull.Bureau of Standards. 19, 131.
- Goncharenko, A.M & Fedorov, F.I. (1958) Surfaces of refraction and absorption of absorbing crystals. Kristallografiya, 3, p 592-598.
- Grassmann, H.G. (1854) Theory of Compound Colors. Philosophical Magazine, 4, no.7, p 254-264.
- Gray, I.M. & Millman, A.P. (1964) Reflection Characteristics of Ore Minerals. Econ.Geol., 59, 1394.
- Gray, I.M. & Millman, A.P. (1962) Reflection Characteristics of Ore Minerals. Econ.Mineral., V 57, p 325-349.
- Grisafe, D.A. & Hummel, F.A. (1970) Crystal Chemistry and Colour in Apatites containing Co, Ni, and Rare-Earth Ions. Am.Mineral., V 55, p 1121-1145.
- Guild, J.A. (1931) The Colorimetric Properties of the Spectrum. Phil.Trans.Roy.Soc.(London), A, 230, 149.
- Hak, J; Johan, Z & Skinner, B.J. (1970) Kutinaite: a new copper-silver arsenide (Czechoslovakia). Am.Mineral., V 55, p 1083-1087.
- Hall, A.J. (1971) The Mineralogy of some Synthetic Sulphosalts. Unpub. Ph.D.Thesis, University of Durham.
- Hallimond, A.F. (1970) The Polarizing Microscope. 3rd.edition, Vickers Ltd., Vickers Instruments, York, England.
- Hallimond, A.F. (1957) Direct Measurement of Standard Reflectances with the Microphotometer. Min.Mag., V 13, p 487.

- Hallimond, A.F. & Bowie, S.H.U. (1964) On the Reflectivity of Pyrite. *Min.Mag.*, London, 111, 385.
- Hardy, A.C. (1936) *Handbook of Colorimetry*. Massachusetts Institute of Technology, Cambridge, Mass. The Technology Press.
- Helson, H. & GROVE, J. (1947) Changes in Hue, Lightness, and Saturation of Surface Colors in passing from Daylight to Incandescent Lamp Light. *J.Opt.Soc.Am.*, 37, p 387-395.
- Henderson, S.T. & Hodgkiss, D. (1963) The Spectral Energy Distribution of Daylight. *Brit.J.Appl.Phys.*, 14 125.
- Harcourt, G.A. (1937) The distinction Between enargite and famatinite (luzonite). *Am.Mineral.*, 22, p 517-525.
- Harris, D.C. & Thorpe, R.I. (1969) New observations on matildite (schapbachite). *Canadian Mineral.* 9, p 655-662.
- Hunt, R.W.G. (1950) The Effects of Daylight and Tungsten light Adaptation on Color Perception. *J.Opt.Soc.Am.*, 40 362.
- Hunter, R.S. (1941) *Photoelectric Tristimulus Colorimetry*. Symposium on color-its specification and use in evaluating the appearance of materials. Philadelphia American Soc. for Testing Materials, p 61
- Isaac Newton (1704) *New Theory about Light and Colors*. Philosophical Transactions of the Royal Society of London, 80, 3075-3087. Opticks by Isaac Newton. published by Smith & Walford, London. (1704). A reprint is available; Dover publications, Inc., New York, 1952.
- Jsmbor, J.L. (1967) New lead sulfantimonides from Madoc, Ontario. Part 1, 9, p 7-24; Part 2, 9. p 191-213. Mineral description. *Can.Mineral.*
- " (1969) Dadsonite (minerals Q & Qm), a new lead sulphantimonide. *Min.Mag.*, 37, p 437-441.
- Jenkins, R. & De Vries, J.L. (1967) *Practical X-Ray Spectrometry*. N.V. Philips' Gloeilampenfabrieken, Eindhoven, The Netherlands.

- Judd,D.B. (1913) The 1931 I.C.I. Standard Observer and Coordinate System for Colorimetry. *J.Opt.Soc.Am.*, 23, p 359-374.
- Judd,D.B. & Wyszecki,G.W. (1963) Color in Business, Science, and Industry. 2nd.edition.New York. John Wiley and Sons.
- Kimball.H.H. (1928) The Distribution of Energy in the Visible Spectrum of Sunlight, Skylight and the Total Daylight. *CIE Proc.* 1928,p 501, Easton,Pennsylvania, Chemical Publishing C.
- Kodera,M; Kupcik,V & Makovicky,E.(1970) Hoderushite - a new sulphosalt. *Min.Mag.*, 37, 290,p 641-648.
- Larsen,E.S. & Berman,H. (1964) The Microscopic Determination of the Nonopaque Minerals. 2nd.edition. Geological Survey Bull. 848, United States Government Printing Office, Washington:1934.
- Leonard,B.F; Mead,C.W. & Finney,J.J.(1971) Paradocrosite, $Sb_2(Sb,As)_2$, a new mineral. *Am.Mineral.*,56,p 1127-1146.
- Leonard,B.F: Desborough,G.A. & Norman J.P. (1969) Ore Microscopy and Chemical Composition of some Laurites. *Am.Mineral.*, 54, p 1330-1346.
- Levy,C. (1966) Contribution a la Mineralogie des Sulfures de cuivre du type Cu_3xS_4 . Premiere these, Presentee a la Faculte des Sciences de l'Universite de Paris, Publie dans les Memoires di Bureau de Recherches Geologiques et Minieres.
- Long,J.V.P. (1967) Electron Probe Microanalysis, in Physical Methods in Determinative Mineralogy. Edited by J.Zussman. Academic Press.
- Lopez-Soler,A. & Bosch-Figueroa,J.M.(1970) Optical Characteristics of enargite. *Trans.Inst.Mining Metall. Sect.B Appl.Earth Sci.*) 79 p 249-251.
- MacAdam,D.L. (1970) Sources of Color Science. The MIT Press.Cambridge, Massachusetts, & London.
- " (1942) Visual Sensitivities to Color Differences in Daylight. *J.Opt.Soc.Am.*, 32, 247.
- " (1943) Specification of small chromaticity differences.

- differences. J.Opt.Soc.Am., 33, 18.
- MacAdam, D.L. (1950) Loci of Constant Hue and Brightness Determined with Various Surrounding Colors. J.Opt.Soc.Am., 40, 589.
- " (1951) Influence of Surround Color. J.Mot.Pict.Telv.Eng., 57, 197.
- Markham, N.L. & Lawrence, L.J. (1965) Mawsonite, a new Cu-Fe-Sn Sulfide from Mt. Lyell, Tasmania and Tingha, New South Wales. Am.Mineral., 50, p 900-908.
- Martin, L.C.; Warburton, F.L. & Morgan, W.J. (1933) Determination of the sensitiveness of the eye to differences in the saturation of colours. Med.Res.Council, Rep.188, London.
- Maxwell, J.C. (1860) Theory of Compound Colors and the Relations of the Colors of the Spectrum. Proceedings of the Royal Soc. of London, 10, 404-409; Scientific papers, V 1, 410-444.
- Moon, P. (1940) Proposed Standard Solar-radiation Curves for engineering use. J.Franklin Inst., 230, 583.
- Moss, T.S. (1959) Optical Properties of Semi-Conductors. London Butterworths Scientific Pub.
- Mozgova, N.N; Borodaev, Yu.S; Rakcheev, A.D. & Borishamskaya, S.S. (1969) On the Diagnostics of Lead Sulphantimonides. Institution of Mining & Metallurgy, Trans./Sect. B, 78 p 857-64.
- Nichol, I. (1962) A Study of some Opaque Manganese Minerals. Unpub. Ph.D.Thesis, University of Durham.
- Nickerson, D. & Stultz, K.F. (1944) Color tolerance specification. J.Opt.Soc.Am., 34, 550.
- Nichol, I. & Phillips, R. (1965) Measurements of Spectral Reflectivity of Manganese Oxides. Min.Mag., No.269.
- Pauly, H. et al. (1969) Study of Minerals and Artificial Materials in Polished Sections on the Micro scale (third Annual Regional Conference, 2-3 May, 1969). Reprinted from Meddelelser fra Dansk Geolodisk Forening.

- Petruk, W; Harris, D.C & Stewart, J.M. (1969) Langisite, a new mineral, and the rare minerals cobalt, pentlandite, siegenite, parderite and bravoite from the Langis mine, Cobalt-Gowganda area, Ontario. *Canadian Mineral.*, 9, p 597-616.
- Phillips, R. & Bradshaw, P.M.D. (1966) A Test of the Linearity of a Photomultiplier used for Reflectivity Measurement. *Min.Mag.*, V 35, p 756-758.
- Piller, H. (1966) Colour Measurements in Ore Microscopy. *Mineralium Deposita.* 1, 175-192.
- Piller, H & vonGehlen, K. (1964) On Errors of Reflectivity Measurements and of Calculations of Refractive index n and Absorption Coefficient k . *Am.Mineral.*, V 49, p 867-882.
- Piller, H. (1967) Influence of Light Reflection at the Objective in the Quantitative Measurement of Reflectivity with the Microscope. *Min.Mag., London*, V 36, p 242-259.
- Physical Society's Colour Group. (1948) Report on Colour Terminology. Physical Society(London), p 56.
- Ramdohr, P. (1969) The Ore Minerals and Their Intergrowths. Pergamon Press. Oxford. London, Edinburgh, New York. Toronto, Sydney.
- Ramdohr, P. (1964) Remarks on Quantitative Measurements of Reflectivity. *Econ.Geol.*, 59, 502-9.
- Roedder, E. & Dwornik, E.J. (1968) Sphalerite Color Banding: lack of Correlation with Iron Content, Pine Point, Northwest Territories, Canada. *Am.Mineral.*, 53, p 1523-1529.
- Sclar, C.B. & Geuer, B.H. (1957) The Paragenetic Relationship of Germanite and Renierite from Tsumeb, South West Africa. *Econ.Geol.*, 52, p 612-631.
- Short, M.N. (1968) Microscopic Determination of the Ore Minerals. *Geol. Survey Bulletin* 914, 2nd. edition 5th. printing. United States Government Printing Office, Washington: 1940.
- Schouten, C. (1962) Determination Tables for Ore Microscopy. Elsevier Publishing Co., Amsterdam. New York.

- Silberstein, L & MacAdam, D.L. (1945) The Distribution of Color Matchings Around a Color Center. *J. Opt. Soc. Am.*, 35, 32.
- Santokh Singh, D. (1964) Measurement of Spectral Reflectivity with the Reichert Microphotometer. *Trans. Inst. Min. & Metall.*, V 74, p 901-916.
- Springer, G. (1969 b) Compositional Variations in Enargite and Luzonite. *Mineral. Deposita*, 4, p 72-74.
- " (1969 d) Microanalytical Investigations into Germanite, Renierite, Briartite and Gallite. *Neues Jahrb. Mineral., Monatsh.*, 435-441.
- Stumpfl, E.F. (1970) New Electron Probe and Optical Data on Gold Tellurides. *Am. Mineral.*, 55, p 808-814.
- " (1968) New Data on Natural Phases in the System Ag-Te. *Am. Mineral.*, 53, 1513, 1522.
- Squair, H. (1964) A Reflectometric Method of Determining the Silver Content of Natural Gold Alloys. *Trans. Inst. Min. Metall.*, V 74, p 917-931.
- Symposium on Color Tolerance. London (1958) *Farbe*, 7, 164, 1958.
- Tansel, T. (1970) On the Mineralogy of Some Base Metal Ores from Ireland. Unpub. M.Sc. Thesis, University of Durham.
- Taylor, A.H. & Kerr, G.P. (1941) The Distribution of Energy in the Visible Spectrum of Daylight. *J. Opt. Soc. Am.*, 31, 3.
- Terziev, G. (1966) Kostovite, an Au-Cu telluride from Bulgaria. *Am. Mineral.*, 51, p 29-36.
- Topping, J. (1966) Errors of Observation and Their Treatment. 3rd. edition, Chapman & Hall Ltd., London.
- Tugal, H.T (1969) The Pyritic Sulphide Deposits of the Lahanos Mine Area, Eastern Black Sea Region, Turkey. Unpub. Ph.D. Thesis, Univ. of Durham.
- Tyndall, E.P.T. (1933) Chromaticity Sensibility to Wavelength Difference as a Function of Purity. *J. Opt. Soc. Am.*, V 23, p 15-24.

- Uytenbogaardt, W. & Burke, E.A.J. (1971) Tables for Macroscopic Identification of Ore Minerals. 2nd. revised edition, Elsevier Publishing Co., Amsterdam. London. New York.
- von Helmholtz, H.L.F. (1924) Physiological Optics. translated by J.P.C. Southall. Published by the Opt. Soc. Am.; Reprint: Dover Publications, Inc., New York, 1962.
- von Gehlen, K & Piller, H (1965) On the Optics of Covellite. Mining Mag., 113, p 438-445.
- " " (1964) Optics of Hexagonal Pyrrhotite, Fe_9S_{10} . Min. Mag. (London), V 35, p 335-346.
- " " (1965) Zur Optik von Hamatit und Ilmenit. N. Jb. Miner. Mh. H 4, p 97-108.
- Wahlstrom, E.E. (1966) Optical Crystallography. 3rd. edition, 4th. printing. John Wiley & Sons, Inc.
- Webster, R. (1970) Gems: Their Sources Description and Identification. 2nd. edition. Butterworth & Co., Ltd., London.
- Wright, W.D. (1928-29) A Re-determination of the Trichromatic Coefficients of the Spectral Colours. Trans. of the Opt. Soc. V xxx No. 4.
- " (1929-30) A Re-determination of the Mixture Curves of the Spectrum. Trans. Opt. Soc., 31, 201.
- " (1969) The Measurement of Colour. 4th. edition. Adam Hilger Ltd., London.
- " & Pitt, F.H.G. (1934) Hue Discrimination in Normal Colour-Vision. Proc. Phys. Soc. (London), 46, 459.
- Wright, W.D. (1927-28) A Trichromatic Colorimeter with Spectral Stimuli. Trans. Opt. Soc., 29, 225.
- " (1941) The Sensitivity of the Eye to Small Colour Differences. Proc. Phys. Soc. (London), 53, 93.
- Wyszecki, G. & Stiles, W.S. (1967) Color Science. Concepts and Methods, Quantitative Data and Formulas. John Wiley & Sons, Inc. New York. London.
- " & Fielder, G.H. (1971) Color-Difference Matches. J. Opt. Soc. Am., V 61, No. 11, 1501.

- Wyszecki, G. & Fielder, G.H. (1971) Color-Difference Matches. J. Opt. Soc. Am., V 61, No. 9, 1135.
- Yund, R.A. (1963) Crystal data for synthetic $\text{Cu}_{5.5x}\text{Fe}_{x}\text{S}_{6.5x}$ (Idaite). Am Mineral., 48, p 672-676.
- Zussman, J. (1967) Physical Methods in Determinative Mineralogy. Edited by Zussman. Academic Press, London & New York.

