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Ink-Jet Printed Graphene Electronics

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We demonstrate ink-jet printing as a viable method for large area fabrication of graphene devices. We produce a graphene-based ink by liquid phase exfoliation of graphite in N-Methylpyrrolidone. We use it to print thin-film transistors, with mobilities up to~ $95 \text{cm}^2 \text{V}^{-1} \text{s}^{-1}$, as well as transparent and conductive patterns, with~80% transmittance and~ $30 \text{k}\Omega/\Box$ sheet resistance. This paves the way to all-printed, flexible and transparent graphene devices on arbitrary substrates.

I. INTRODUCTION

Flexible electronics is a rapidly expanding research area¹. Applications include touch screens², electronic paper (e-paper)^{3,4}, sensors⁵, radio frequency tags⁶, photovoltaic cells^{7,8}, and electronic textiles⁹. To date, it mainly relies on two fabrication strategies: one in which substrates bearing thousands of Field-effect Transistors (FETs) are bonded to plastic by transfer printing or pickand place methods¹⁰; another in which FETs are prepared directly on the target substrate by several coating, curing and lithographic steps^{1,11}. Rubber stamping¹², embossing¹³ and ink-jet printing^{14,15} reduce the number of such fabrication steps.

Ink-jet printing is one of the most promising techniques for large area fabrication of flexible plastic electronics¹⁵. A range of components can be printed, such as transistors^{13,15–18}, photovoltaic devices¹⁹, organic light emitting diodes (OLEDs)^{13,18,20}, and displays¹³. Inkjet printing is versatile¹⁸, involves a limited number of process steps²¹, is amenable for mass production, and can deposit controlled amounts of material 21 . Drop on demand^{21,22} ink-jet printing has progressed from printing text and graphics 21 , to a tool for rapid manufacturing 23 , being now an established technique to print Thin Film Transistor (TFT) based on organic conducting and semiconducting $inks^{5,15,24}$. However, their mobilities, $\mu < 0.5 \text{cm}^2 \text{V}^{-1} \text{s}^{-1}$,^{5,18} are still much lower than standard silicon technology. Several approaches aim to improve this, such as the use of polysilicon²⁵, zinc oxide nanoparticles²⁶ and carbon nanotubes $(CNTs)^{27-32}$. Metal nanoparticle inks are not stable in ordinary solvents, such as Deionized (DI) Water, Acetone, Isopropyl Alcohol, N-Methylpyrrolidone (NMP), Tetrahydrofuran^{18,33}. They need to be chemically modified in order to be dispersed¹⁸, using stabilizers, which usually degrade in a couple of years^{18,33}. Metal nanoparticles also tend to oxidize after printing^{18,33}. Inkjet printed CNT-TFTs have been reported with μ up to $50 \text{cm}^2 \text{ V}^{-1} \text{s}^{-1}$ and a ON/OFF ratio~ 10^3 .³²

Graphene is the two-dimensional (2d) building block for sp^2 carbon allotropes of every other dimensionality. It can be stacked into 3d graphite, rolled into 1d nanotubes, or wrapped into 0d fullerenes³⁴. It is at the centre of an ever expanding research area^{34–37}. Near-ballistic transport and high mobility make it an ideal material for nanoelectronics, especially for high frequency applications³⁸. Furthermore, its optical and mechanical properties are ideal for micro and nanomechanical systems, thin-film transistors, transparent and conductive composites and electrodes, and photonics^{34,37,39}. Graphene was isolated by micromechanical exfoliation of graphite⁴⁰. This technique is still the best in terms of purity, defects, mobility and optoelectronics properties. However, large scale production approaches are needed for widespread application. These encompass growth by chemical vapor deposition (CVD)^{41–46}, segregation by heat treatment of silicon carbide^{47–50} and metal substrates^{51–54}, liquid phase exfoliation (LPE)^{55–58}. Amongst these, LPE is ideally suited to produce printable inks.

Graphite can be exfoliated by chemical wet dispersion followed by ultrasonication, both in aqueous^{56,58} and non-aqueous solvents^{55,58}. Dispersions can be achieved by mild sonication in water with Sodium Deoxycholate, followed by sedimentation based-ultracentrifugation^{58,59}. Bile salt surfactants also allow the isolation of flakes with controlled thickness, when combined with density gradient ultracentrifugation (DGU)⁶⁰. Exfoliation of graphite intercalated compounds⁵⁷ and expandable graphite⁶¹ was also reported.

LPE was first achieved through sonication of graphite oxide⁶², following the Hummers method⁶³. The oxidation of graphite in the presence of acids and $oxidants^{64,65}$ disrupts the sp²-network and introduces hydroxyl or epoxide groups^{66,67}, with carboxylic or carbonyl groups attached to the $edge^{66,67}$. These make graphene oxide (GO) sheets readily dispersible in water 62,68 and several other solvents⁶⁹. Although large GO flakes can be produced, these are intrinsically defective 62,70 , and electrically insulating 62,66 . Despite several attempts 62,66 , reduced GO (RGO) does not fully regain the pristine graphene electrical conductivity 66,71 . It is thus important to distinguish between dispersion processed graphene flakes^{55–58}, retaining the electronic properties of graphene, and insulating GO dispersions^{62,71}. Several groups reported GO-based inks^{33,72,73}. Ref. 72 ink-jet printed RGO films for sensors applications, while Ref. 33 produced RGO-stabilized Cu nanoparticles as low temperature metal colloids, to replace standard metal nanoparticle inks, that require high temperature sintering postprocessing⁷⁴. Mobilities up to $90 \text{cm}^2 \text{V}^{-1} \text{s}^{-1}$ have been achieved for highly reduced GO films by ink-jet

printing⁷³, with an ON/OFF ratio up to $10.^{73}$

Here we produce a graphene-based ink and demonstrate its viability for printed electronics.

II. RESULTS AND DISCUSSION

III. INK REQUIREMENTS

A key property of inks viable for printing is their ability to generate droplets^{75,76}. Ink viscosity, η [mPa s], surface tension, γ [mJ m⁻²], density, ρ [g cm⁻³], and nozzle diameter, a [μ m], influence the spreading of the liquid drops^{75–77}. These parameters can be arranged into dimensionless figures of merit (FOM), such as the Reynolds (Re)^{75–77}, Weber (We)^{75–77}, and Ohnesorge (Oh)^{75–77} numbers: Re= $\frac{v\rho a}{\eta}$; We= $\frac{v^2\rho a}{\gamma}$, Oh= $\frac{\sqrt{We}}{Re} = \frac{\eta}{\sqrt{\gamma\rho a}}$, where v[m/s] is the drop velocity.

Refs. 75–77 suggested to use Z=1/Oh as the appropriate FOM to characterize drop formation, 1 < Z < 14 being required to get stable drop generation^{75,76}. For Z<1 the high viscosity prevents drop ejection^{75,76}, whereas at Z>14 the primary drop is accompanied by a number of satellite droplets^{75,76}. Moreover, when inks contain dispersed molecules or nano-particles, the latter should be smaller than the nozzle diameter, to prevent clogging^{21,23}. Refs. 23,78 suggested that the size of such molecules or particles should be at least 1/50 of the nozzle diameter, in order to exclude any printing instability, such as clustering of the particles at the nozzle edge, which may deviate the drop trajectory, or result in agglomerates that will eventually block the nozzle.

The ejected drop behavior on the substrate can be efficiently described by fluid dynamics. When a small liquid droplet is in contact with a flat surface, partial wetting results in a finite angle between the liquid and the substrate⁷⁹, known as contact angle, θ_C^{79-81} . The lower drop size limit is given by^{75,76} $s[\mu m] =$ $a\sqrt{\frac{We+12}{3(1-\cos\theta_C)+4We/Re^{1/2}}}.$ Thus, e.g., for a typical $a=50\mu m$, We=20, Re=58 and $\theta_C \sim 45^{\circ}$, we get $s \sim 85$ - $90\mu m$. The distance from the substrate must be optimized to guarantee both homogeneous printing and the highest resolution, barring any unusual jetting conditions, such as perturbations from the surrounding environment and diversion of the drop trajectory^{18,75,82}. Furthermore, a substrate very close to the nozzle causes secondary drops to scatter off during the impact of the primary drop^{18,83}, due to the initial drop jetting pressure, thus affecting the homogeneity of the final printed features⁸³. The final assembly of printed nano-particle inks depends on the substrate Surface Energy $(SE)^{21,23}$. as well as the ink viscosity and surface tension²¹.

When a drop of an ink containing dispersed particles evaporates on a surface it commonly leaves a dense, ringlike, deposit along its perimeter^{21,23}. This is the so-called "coffee ring effect"⁸⁴, i.e. a distortion of the drops during solvent drying due to the interplay of ink viscosity and solute transport via solvent motion (arising from surface tension interaction between solvent and substrate)^{18,84}. This is one of the most important phenomena affecting the homogeneity of ink-jet printed drops^{18,84}. In order to prevent this, it is necessary to "freeze" the drops geometry immediately after they form an homogeneous and continuous film on the substrate¹⁸.

Here we use an ink-jet printer with a nozzle diameter~ 50μ m, thus we need to have flakes less than 1μ m across. By tuning η , γ and ρ we will target a Z within the optimal range. We print on Si/SiO₂ (to probe the electrical properties of the ink) and borosilicate (Pyrex 7740-Polished Prime Grade) glass substrates (to test the ink as transparent conductor), both with a roughness $R_z < 15$ nm. Our aim is to obtain ink-jet printed drops on the substrate, with homogeneous flakes and uniform morphology, i.e. with roughness comparable to the substrate. We obtain this by varying the contact angle and optimizing the substrate wettability.

In order to reduce the coffee ring effect we need both a solvent with boiling point (T_c [°C]) and heat of vaporization (V_c [kJ/mol]) higher than water^{18,82,84}, and a substrate that promotes adhesion⁸⁵. Thus, we use NMP as solvent for two main reasons. First, it has higher boiling point (~202°C)⁸⁶ and heat of vaporization (54.5kJ/mol)⁸⁶, than water (~100°C and ~40kJ/mol). Second, NMP is the best solvent to get high-yield, surfactant-free exfoliation of graphite^{55,58}. We then test several surface treatments to optimize substrate adhesion. After printing, NMP is removed by thermal annealing at 170°C for 5 minutes.

A. Graphene-based printable ink

We prepare the graphene-based printable ink as follows. Graphite flakes (NGS Naturgraphit) are sonicated (Decon bath, 100W) in NMP for 9 hours. The unexfoliated flakes are left to settle for 10 mins after sonication. The decanted dispersions are then ultracentrifuged using a TH-641 swinging bucket rotor in a Sorvall WX-100 Ultra-centrifuge at 10,000 rpm (~15,000g) for an hour and filtered to remove flakes>1 μ m, that might clog the nozzle. The resulting ink is characterized by Optical Absorption Spectroscopy (OAS), High Resolution Transmission Electron Microscopy (HRTEM), Electron diffraction and Raman spectroscopy.

A Perkin-Elmer Lambda 950 spectrometer with 1nm resolution is used for OAS measurements. OAS can be used to estimate the concentration of graphene^{55,56,59} using the Beer-Lambert Law according to the relation $A = \alpha cl$, where A is the absorbance, l [m] is the light path length, c [g/L] the concentration of dispersed graphitic material and α [L g⁻¹ m⁻¹] the absorption coefficient. Fig.1 plots an OAS spectrum of our ink diluted to 10%. The ink is diluted to avoid strong scattering losses at higher concentrations, which could cause deviation of the measured value of A from the Beer-Lambert law. The



FIG. 1: Absorbance of graphene-ink. The inset is a picture of a vial of ink.



FIG. 2: a,b) HRTEM image and diffraction pattern of a dispersed SLG flake. c) Diffracted intensity along the dashed line in b. Lateral size distribution of d) SLGs, e) BLGs, f) FLGs.

spectrum in Fig.1 is mostly featureless, as expected due to the linear dispersion of the Dirac electrons^{37,39,87–90}, the peak in the UV region being a signature of the van Hove singularity in the graphene density of states⁸⁸. From $\alpha \sim 1390 \text{Lg}^{-1}\text{m}^{-1}$ at 660nm, as for Refs. 56,58, we estimate c $\sim 0.11 \pm 0.02 \text{g/L}$.

We disperse drops of our ink on Holey carbon Transmission electron microscopy (TEM) grids for analysis using a Tecnai T20 high resolution TEM, with an acceleration voltage of 200KV operating in phase contrast mode. Fig.2a is HRTEM image of a Single Laver Graphene (SLG) flake from the ink, while Fig.2b is a normal-incidence electron diffraction of the same flake of Fig.2a. It shows the expected sixfold symmetry $^{91-93}$. The peaks are labeled with the corresponding Miller-Bravais (hkil) indexes. For Few Layer Gaphene (FLG) flakes with Bernal (AB) stacking, the intensity ratio I_{1100}/I_{2110} is<1, while for SLG $I_{1010}/I_{2110}>1^{91,93}$. We use this to distinguish SLG from FLGs^{55,59}. Fig.2c plots the diffraction intensity measured along the line section through the $(1\overline{2}10)$, $(0\overline{1}10)$, $(\overline{1}010)$, $(\overline{2}110)$ axis, reported in Fig.2b. The inner peaks, $(0\overline{1}10)$ and $(\overline{1}010)$, are~1.5 times the outer ones, $(1\overline{2}10)$ and $(\overline{2}110)$, indicating that the flake is SLG^{91} . The analysis of the edges also gives a reliable information on the number of layers and can be used to investigate a large number of flakes⁹¹, from zoomed-in high resolution edge images^{55,94}. If SLG folds or several SLGs stack one on the other, selected area diffraction is used to distinguish contentious cases.

These combined analysis show that our ink mostly consists of SLGs, Bi-Layers (BLG) and FLGs, with lateral size $\sim 300-1000$ nm. We find that $\sim 35\%$ SLGs are larger than 300nm (Fig.2d); $\sim 40\%$ BLGs are larger than 350nm $(Fig.2e); \sim 55\%$ FLGs are larger than 450nm (Fig.2f). In particular, we have $\sim 33\%$ SLG with c ~ 0.11 g/L. Previous works on LPE of graphite in NMP reported up to $\sim 28\%$ SLG for $c \sim 0.18 \text{g/L}^{58}$ and $\sim 21\%$ for $c \sim 1.8 \text{g/L}^{94}$. Ref. 57 also reported exfoliation of intercalated graphite in NMP, with $\sim 20\%$ SLGs for $c \sim 0.01$ g/L. Thus, our ink has higher SLG yield with respect to previous works, but lower c than Ref.94. This higher c was achieved by long time (up to 460h) ultrasonication⁹⁴. However Ref. 94 reported defects and reduction of size as a result. Our combination of low-power sonication (<25W) and ultracentrifugation is ideal for high-yield of defect-free SLGs.

Stable dispersions require the Gibbs free energy of mixing, ΔG_{mix} , to be zero or negative⁹⁵, where $\Delta G_{mix} =$ $\Delta H_{mix} - K \Delta S_{mix}$, K being the temperature, ΔH_{mix} the enthalpy of mixing and and ΔS_{mix} the entropy change in the mixing process^{55,95}. For graphene and nanotubes, ΔS_{mix} is small^{55,96}. Therefore, for dispersion and stabilization of graphene in solvents, ΔH_{mix} needs to be very small. This can be achieved by choosing a solvent whose surface energy is very close to that of graphene⁵⁵. The surface energy of NMP satisfies this requirement and allows efficient exfoliation of graphite. Graphite can also be efficiently exfoliated in water with the use of bile salt surfactants. Ref. 97 reported~20%SLGs and $c\sim 0.3$ g/L SLGs, while Ref. 59 reported~60% SLGs for $c \sim 0.012 \text{g/L}$. The yield can be increased up to $\sim 80\%$ by density gradient ultracentrifugation⁶⁰. The flake size of LPE graphene in water-surfactant dispersions is on average smaller($\sim 200 \text{nm}^{97}$, $\sim 30 \text{nm}^{59}$) than thus far reported for NMP($\sim 1 \mu \text{m}^{55,58}$). The viscosity at room temperature of NMP (1.7mPas^{86}) is higher than water ($\sim 1\text{mPas}^{86}$). Larger flakes in a higher viscosity medium (such as NMP) experience higher frictional force^{98,99} and sedimentation coefficient^{99,100}, making it more difficult for them to sediment during ultracentrifugation. This reduces the SLG yield in NMP compared to water.

The centrifuged dispersions are drop-cast onto a Si wafer with 300nm thermally grown SiO₂ (LDB Technologies ltd.) and annealed at 170°C to remove NMP. These samples are then used for Raman measurements, collected with a Renishaw 1000 at 457, 514.5 and 633nm and a $100 \times$ objective, with an incident power~1mW. Fig.3a plots a typical Raman spectrum of the ink at 514.5nm. Besides the G and 2D peaks, it shows significant D and D' intensities and the combination mode $D+D'\sim 2950 \text{ cm}^{-1}$. The G peak corresponds to the E_{2q} phonon at the Brillouin zone centre. The D peak is due to the breathing modes of sp^2 rings and requires a defect for its activation by double resonance $(DR)^{93,101,102}$. The 2D peak is the second order of the D peak. This is a single band in SLG⁹³, whereas it splits in four in BLG, reflecting the evolution of the band structure⁹³. The 2D peak is always seen, even when no D peak is present, since no defects are required for the activation of two phonons with the same momentum, one backscattering from the other⁹³. DR can also happen intra-valley, *i.e.* connecting two points on the same cone around **K** or $\mathbf{K}^{101-103}$. This gives the D' peak. The 2D' is the second order of the D' peak.

We assign the D and D' peaks to the edges of the sub-micrometer flakes $^{104},\,\,\rm rather$ than to the presence of a large amount of disorder within the flakes. This is further supported by the plot of the G peak dispersion, Disp(G)(Fig.3b)). This is defined as $Disp(G) = \Delta Pos(G) / \Delta \lambda_L$, where λ_L is the laser excitation wavelength. Disp(G) is generated from the linear fit the plot of the G peak position, Pos(G), as a function of the laser excitation wavelength. In disordered carbons Pos(G) increases as the excitation wavelength decreases, from IR to UV^{101} , thus Disp(G) increases with disorder^{101,105}. The full width at half maximum of the G peak, FWHM(G), always increases with disorder^{106,107}. Thus, combining the intensity ration of the D and G peaks, I(D)/I(G), with FWHM(G) and Disp(G) allows us to discriminate between disorder localized at the edges, and disorder in the bulk of the samples. In the latter case, to higher I(D)/I(G) would correspond higher FWHM(G) and Disp(G). Figs.4 a,b) show that Disp(G), I(D)/I(G)and FWHM(G) are not correlated, a clear indication that the major contribution to the D peak comes from the sample edges. Also, Disp(G) is nearly zero for all samples, compared to the values bigger than $0.1 \text{cm}^{-1}/\text{nm}$ expected for disordered carbons^{105,108}, another indication of the lack of large structural disorder within our flakes. The distribution of 2D peak positions, Pos(2D), shown in 3d, has two maxima~2692 and 2705 cm⁻¹, similar to FWHM(2D) (3e). This is consistent with the samples being a distribution of SLG, BLG and FLGs, but with a significant fraction of SLGs. We note that for the flakes



FIG. 3: a) Raman spectrum of graphene-ink deposited on Si/SiO₂. Distribution of b) Disp(G), c) I(D)/I(G), d) FWHM(G), e) Pos(2D), f) FWHM(2D), g) I(2D)/I(G).

with the smallest Pos(2D) and FWHM(2D), the ration of the 2D and G integrated areas, A(2D)/A(G), is at most 3.5, implying a doping of at least 10^{13} cm⁻².¹⁰⁹⁻¹¹¹

We now estimate η , ρ and γ for our ink, in order to check its viability for ink-jet printing. η can be evaluated as $\eta = \eta_0 (1+2.5\phi)^{82,112}$, where η_0 is the viscosity of the pure solvent and ϕ the volume fraction of particles in the dispersion. We assume $\eta_0 = \eta_{NMP} \sim 0.8$ mPas, the viscosity of pure NMP at $\sim 80^{\circ}$ C^{86,113} (the temperature of the drops ejected from our printer, as specified in Ref. 114). We take $\phi = 1 - \frac{Vol_{ink}}{Vol_{NMP}}$, where Vol_{NMP} [~0.972 mm³] is the volume of 1mg pure NMP and Vol_{ink} [~0.94 mm³] is the volume of 1mg of our ink, both measured by a micropipette (± 2 nL precision), at room temperature and pressure. We thus get $\phi \sim 0.03$, and $\eta \sim 0.96$ mPas. From



FIG. 4: a) I(D)/I(G) as function of Disp(G), b) I(D)/I(G) as function of FWHM(G) measured on flakes of our ink deposited on Si/SiO₂

the same measurement we also obtain $\rho \sim 1.06 \text{gcm}^{-3}$ and derive $\gamma \sim 50 \text{mJ} \text{ m}^{-2}$ from tensiometer measurements. Given these parameters, and our nozzle diameter $\sim 50 \mu \text{m}$, we get $Z \sim \frac{\sqrt{\gamma \rho a}}{\eta} \sim 1.7$, which falls within the range suitable for printing^{75,76}, but close to the lower boundary of allowed Z,^{75–77} thus implying a lower probability of secondary drops ejection^{75,82,112}. However, high viscosity may generate nanoparticle re-aggregation¹¹².

B. Ink-jet printed features

The final layout of printed nano-particle inks depends on substrate $SE^{21,23}$, ink viscosity and surface tension²¹.

To investigate the influence of surface treatments, we print our ink on pristine, HMDS coated and O₂ plasma treated Si/SiO₂. A modified Epson Stylus 1500 ink-jet printer equipped with an Epson S020049 cartridge is used to print the dispersions under a constant nitrogen flow, followed by annealing at 170°C for 5 minutes to remove the NMP. The nozzle is placed~1mm above the substrate. HMDS is deposited by spin coating for 40s at 1000rpm, followed by annealing at 80°C for 2 min. Alternatively the substrates are cleaned by a RF O₂ plasma at 200W and 4×10^{-1} Torr for 2 min.

We use optical micrographs to visualize the ink-jet printed drops, Figs.5a,b,c. The bright green/blue color of the printed features is due to the use of dark field



FIG. 5: Dark field optical micrograph of inkjet printed drops on a) plasma cleaned, b) pristine and c) HMDS treated substrate. Scale: 20μ m. d) SEM micrograph of printed pattern.



FIG. 6: Images of water drops dispensed on a) pristine and b) HMDS teated $\rm Si/SiO_2$

imaging. These reveal that HMDS constrains the drops to 90μ m diameter (Fig.5c), smaller than on the other substrates (~100 μ m and ~150 μ m for pristine, Fig.5b, and plasma treated SiO₂, Fig.5a). As discussed above, we use NMP as solvent to reduce the coffee ring effect compared to low boiling point solvents (e.g. water, chloroform)^{18,82,84}. However, we still observe coffee-rings when printing on pristine SiO₂ (Fig.5b), while Fig.5c reveals a higher flake uniformity, and no coffee-rings on HMDS treated SiO₂. Fig.5d a representative printed pattern, showing the ability to fabricate complex layouts.

Thus, HMDS appears to prevent coffee-rings. To understand this, we measure the substrates SE and investigate the printed stripes morphology, before and after surface treatment. We perform contact angle measurements with a A KSV CAM200 system. The contact angle is measured by dispensing 1μ l DI water on the substrates. The surface tension is measured by the DuNouy-Padday technique¹⁴⁷. This consists in using a rod few millimeters diameter immersed in the dispersion, followed by pull out. The rod is attached to a scale or balance via a thin metal hook that measures the maximum pull force. This is recorded as the probe is first immersed 1mm into the solution and then slowly withdrawn from the interface. The contact angle, θ_C , depends on the liquid surface tension^{79–81} and the substrate critical surface tension^{79–81}, according to the Young's relation^{79,81,115}: $\gamma_{SV} - \gamma_{SL} - \gamma_{LV} \cos\theta_C = 0$, where γ_{SV} [mJ m⁻²] is the solidvapor surface tension, γ_{SL} is the solid-liquid surface tension and γ_{LV} is the liquid-vapor surface tension.

Figs.6a,b show ink drops printed onto pristine and

HMDS treated Si/SiO₂, with $\theta_C \sim 6^{\circ}$ and $\sim 65^{\circ}$, indicating that the pristine substrate SE is modified following HMDS treatment. γ_{LV} was measured $\sim 73 \text{mJ} \text{ m}^{-2}$ in Ref.116 for DI water, whereas $\gamma_{SV} \sim 116.5 \text{mJ} \text{ m}^{-2}$ and $\sim 40 \text{mJ} \text{ m}^{-2}$ were reported for pristine¹¹⁷ and HMDS treated¹¹⁸ Si/SiO₂. Consequently, $\gamma_{SL} \sim 43.9 \text{mJ} \text{ m}^{-2}$ and $\sim 9.1 \text{mJ} \text{ m}^{-2}$ for pristine and HMDS treated Si/SiO₂, respectively. A higher γ_{SL} implies a higher SE¹¹⁹. Indeed, our γ_{SL} correspond to SEs ~ 73.9 and $\sim 39.1 \text{mJ} \text{ m}^{-2}$ for pristine and HMDS treated Si/SiO₂. A small θ_C results in the drop rapid spreading on the substrate⁷⁹, as seen in pristine SiO₂. On the other hand, HMDS provides higher θ_C , since it lowers γ_{SL} (thus the substrate SE), therefore reducing the wettability^{80,120}.

When ink-jet printing stripes, the inter-drop (i.e. centre to centre) distance is an important parameter¹²¹. For a large distance, individual drops are deposited on the substrate^{75,82,121}. As the inter-drop distance decreases, these merge into a line¹²¹. Thus, in order to obtain a continuous line we need an inter-drop distance smaller than the drop diameter¹²¹. On the other hand, Refs.82,112 reported that a very small inter-drop distance can result in particle aggregation on the substrate, thus a non-uniform stripe (i.e. irregular edges). We thus select an inter-drop distance suitable to have continuous lines, avoiding at the same time non-uniformities and irregular edges.

Figs.7a,b,c are optical images of printed stripes on pristine, O_2 plasma treated and HMDS treated Si/SiO₂, whereas Figs.7d,e,f plot the respective Atomic Force Microscope (AFM) topographies. The stripe in Fig.7a is~100-110 μ m wide, has an average thickness~70nm and an irregular flake distribution, with aggregation of flakes. That in Fig.7b is wider ($\sim 130-140 \mu m$), with aggregates at the edges, and an average thickness ~ 55 nm. The stripe in Fig.7c has a more uniform and regular distribution of flakes, $\sim 85-90 \mu m$ wide and $\sim 90 nm$ thick. The width narrows going from the O₂ plasma treated to the HMDS treated Si/SiO₂, due to the SE decrease. Figs.7d,e show stripes with voids and irregular flake distribution, with $R_z \sim 30-40$ nm. Fig.7f presents a more homogeneous network with $R_z \sim 15$ nm. Thus, R_z is lower when θ_C is higher, because the poor wettability of drops with higher θ_C reduces the stripe diameter (as shown in Figs.7a,b,c), confining the flakes onto a smaller area. The uniformity of stripes printed on the HMDS treated substrate corroborates the above considerations on the SE changes. In fact, the presence of silane in HMDS⁸⁵ promotes the adhesion of metallic particles to the substrate^{85,122}. Analogously, HMDS may promote the adhesion of graphene to the substrate, thus resulting in a uniform network.

Fig.8a compares a typical Raman spectrum of a flake in the ink, with a measurement on the first stripe and on a stripe 90 nm thick, after 30 printing repetitions. Figs.8b,c,d,e,f,g,9 compare the Pos(2D), FWHM(2D) and Disp(G) distributions. The data show that the first stripe has very similar characteristics to the ink, as expected. However, the spectra after 90 repetitions show a Pos(2D) and FWHM(2D) distribution more typical of



FIG. 7: Optical micrograph of ink-jet printed stripes on a) pristine, b) O_2 and c) HMDS treated substrates.d,e,f) AFM images of a,b,c



FIG. 8: a) Typical Raman spectrum of individual flakes in the ink, compared with spectra measured on the first stripe and on a stripe 90 nm thick. Pos(2D) and FWHM(2D) for b,c) ink; d,e) fist stripe; f,g) 90nm thick stripe



FIG. 9: Distribution of Disp(G) for a) ink; b) fist stripe; c) 90nm thick stripe

a multi-layer sample, having lost any direct signature of SLG. Note however that the 2D peak shape, even for the 90nm stripe, remains distinctly different from that of graphite. A similar aggregation of flakes was previously observed for thick films derived from graphene solutions⁵⁵. In all cases Disp(G) remains similar, and very low, again showing the lack of large amounts of defects within the flakes.

C. Transparent and conductive patterns

We now investigate the viability of our ink to print transparent and conductive patterns. We characterize the sheet resistance R_s [Ω/\Box] and Transmittance T [%] of our stripes when placed on a transparent substrate. We thus use pristine, O_2 and HMDS treated borosilicate glass, with $R_z < 15$ nm similar to SiO₂ on Si, but with T~99% (Pyrex 7740-Polished Prime Grade). T is measured on samples ink-jet printed on borosilicate glass (followed by annealing at 170°C for 5 mins to remove the NMP) by scanning a 514.5nm laser beam with 100 μ m steps. The transmitted beam is measured with a photodiode. A microscope equipped with 100× long distance objective focuses the laser to~2 μ m. The incident power is kept at~8mW. The transmitted power is measured by a Ophir Nova II power meter with 0.1 μ W resolution.

Fig.10a shows that for our stripes the experimentally measured thickness (t) increases linearly as a function of printing repetitions, with a slope defined by the surface treatment. Fig.10b plots the four-probe measured R_s as a function of t. For large t, R_s settles to~34,~500,~10⁵k Ω/\Box for HMDS treated, pristine and O_2 treated glass, respectively. For t<20nm, R_s increases for all substrates. For a thin film, $R_s =$ $(\sigma t)^{-1}$, where σ [S/m] its conductivity¹²³. Thus, from Fig.10b and $\sigma = (\mathbf{R}_s t)^{-1}$, we get the data in Fig.10c. σ is constant for t>20nm, in the case of HMDS treated, pristine and plasma treated glass, with an average $\sim 10^2$, ~ 30 , $\sim 10^{-1}$ S/m, respectively. Thus, stripes on HMDS treated glass have an higher σ combined with a more regular network of flakes, compared to the other two substrates. When t<20nm, σ decreases for all substrates. A similar trend was reported for CNT films on SiO₂ (produced by vacuum filtration)^{124,125}, ink-jet printed CNT patterns on $SiO_2^{29,30}$, graphene films on SiO₂,^{126,127} and Polyethylene-terephthalate(PET),^{126,127} as well as Ag nanowire films, produced by vacuum filtration on SiO_2^{126} . Refs. 124–127 explained this decrease of σ for small t, due to percolation.

The percolation theory¹²⁸ predicts σ , for a network of conductive particles, to scale as¹²⁸:

$$\sigma \propto (X - X_c)^{\beta} \tag{1}$$

where $X \ [\mu g/mm^2]$ is the concentration of conductive particles per unit area, $X_c \ [\mu g/mm^2]$ is the critical concentration of flakes corresponding to the percolation threshold and β is the percolation exponent. Eq.1 can be rewritten in terms of t, rather than X^{124} as:

$$\sigma \propto (t - t_c)^{\epsilon} \tag{2}$$

where t_c is the critical thickness and ϵ is the percolation exponent. Fig.10c shows two regimes for σ as a function of t: a percolative linear behavior for t<20nm and a constant σ_{bulk} for t>20nm. This can be explained considering that our films stop behaving like bulk materials below a critical thickness (t_{min}) .

The exponent ϵ can be estimated by a linear fit of the \log_{10} plot of σ vs t, in the percolation region (t<20nm), Fig.11. We get $\epsilon \sim 4$ for stripes on HMDS treated and pristine glass, while $\epsilon \sim 3$ for O₂ treated glass. These values indicate percolation, as reported by Refs.126,129–131 for networks with various geometries. ϵ is expected to increase with particle size^{130,131} and decrease with $X_c^{130,131}$. Assuming a similar particle size, since the same ink is used for all cases, we deduce that $\epsilon \sim 4$ points to a bigger X_c than $\epsilon \sim 3$. This indicates formation of a more uniform network on HMDS treated and pristine glass compared to O₂ treated glass.

We also determine the minimum concentration necessary to achieve the bulk conductivity regime. To do so, we assume $X \gg X_C$, because the bulk regime needs a tight network of interconnected flakes^{126,129,132}. Given our $c \sim 0.11$ g/L, volume per printed drop~10nL¹¹⁴, and a dried drop size on the three substrates of~90,100,130 μ m, we estimate $X \sim 4 \times 10^{-2}$,~ 10^{-2} and~ $0.7 \times 10^{-2} \mu$ g/mm² for stripes printed on HMDS, pristine and plasma treated



FIG. 10: a) Thickness as a function of printing repetitions. b, c) R_s and σ as a function of stripe thickness. d) T as a function of R_s for HMDS coated (red dots), O_2 plasma treated (green triangles) and pristine (black squares) substrates



FIG. 11: Conductivity as a function of thickness, in logarithmic scale, for stripes printed on HMDS treated (red dots), O_2 treated (green triangles) and pristine (black squares) substrates. Lines are fits in the percolation regime

glass, respectively. Consequently, from Eq.1, σ for stripes on HMDS treated glass ($\sim 10^2$ S/m) is higher than on pristine (~ 40 S/m) and plasma treated glass(~ 0.1 S/m).

Fig.10d shows T as a function of R_s . The dashed lines

are a plot of the relation $T = \left(1 + \frac{Z_0 \ G_0}{2R_s \sigma_{bulk}}\right)^{-2}$ expected for stripes with σ_{bulk} conductivity, where $Z_0 = 377\Omega$ is the free-space impedance, $G_0 \sim 6 \times 10^{-5} \Omega^{-1}$ is the universal optical conductance of graphene¹³³. The solid lines are a plot of $T = \left[1 + \frac{1}{\Pi} \left(\frac{Z_0}{R_s}\right)^{1/(\epsilon+1)}\right]^{-2}$ expected in the percolative regime¹²⁶, where Π is the percolative Figure of Merit $\Pi = 2 \left[\frac{\sigma_{bulk}/G_0}{(Z_0 \ t_{min} \ G_0)^{\epsilon}}\right]^{1/(\epsilon+1)}$. Our experimental T deviates from the dashed lines for T>75%. We assign this to the percolative regime where σ_{DC} deviates from a bulk-like behavior. Also in this case, printing on HMDS treated glass gives the highest T for a given R_s .

D. Ink jet printed devices

Ink-jet printed TFTs based on organic semiconducting polymers have been widely investigated^{15,134,135}. The current state of the art devices have μ ranging from 0.01 to~0.5cm²V⁻¹s⁻¹, with ON/OFF ratios up to 10⁵.^{134–136} Several Inkjet printed TFTs using various carbon nanomaterials have been reported. For example, fullerene-based TFTs were discussed in Refs. 137,138, with μ up to 0.01cm²V⁻¹s⁻¹ and an ON/OFF ratio<10. TFTs printed from CNT-based inks have been presented



FIG. 12: a) Ink on Si/SiO_2 .b) Cr-Au pads define the source and drain contacts. c) A layer of Poly[5,5'-bis(3-dodecyl-2thienyl)-2,2'-bithiophene] (PQT-12) is printed on top

by several groups^{27–29,31,32}. The highest μ thus far is~50cm²V⁻¹s⁻¹ combined with an ON/OFF ratio 10³, but measured at 10⁻⁶ Torr and 100K³². Ink-jet printed TFTs from GO-based inks were discussed in Refs. 72,73, with μ up to~90cm²V⁻¹s⁻¹ for an ON/OFF ratio of 10 (measured at room conditions), after GO reduction.

We print our TFTs as for Fig.12a, and contact them with chromium-gold source and drain pads (Fig.12b). The transfer characteristics are measured (at room conditions) at different drain voltages (V_d=-2,-4,-8V). μ is derived from $\mu = \frac{L}{W C_i V_d} \frac{dI_d}{dV_g}$, where L [μ m] and W [μ m] are the channel length and width respectively, C_i is the gate dielectric capacitance $(\sim 10 \text{nF}/\text{cm}^2)^{139}$. We get $\mu \sim 95 \text{cm}^2 \text{V}^{-1} \text{s}^{-1}$ for an ON/OFF ratio~10 at V_d=-2V, comparable to that reported in Ref. 73 for ink-jet printed RGO TFTs. μ in our devices is almost four orders of magnitude higher than printed fullerene-based TFTs^{137,138} (for the same ON/OFF ratio) and more than two orders higher than ink-jet printed $CNTs^{27,29}$ (for a ON/OFF ratio of 10). However, the ON/OFF ratio in our TFTs is lower than the state of the art for CNTs (but measured at 10^{-6} Torr and 100K) at similar μ^{32} . We note that ink-jet printed electronics requires high μ at room conditions^{11,18}. So far CNT ink-jet printed devices measured at room conditions have μ no larger than $\sim 1 \text{cm}^2 \text{V}^{-1} \text{s}^{-1}$ (at ON/OFF ~ 10)²⁹, which is two orders of magnitude smaller than our jet printed TFTs.

Organic semiconducting inks^{134–136} suffer from low μ , limited by variable range hopping between the isolated polymer chains¹⁴⁰. The overall charge conduction in crystalline organic semiconducting thin films is determined by both intra-chain and inter-chain charge transport¹⁴¹. The former is much faster than inter-chain hopping^{140,141}. Many groups have tried to improve interchain hopping^{27,28,142,143}. Ref. 142 proposed a chemical modification of the semiconducting organic ink by electron acceptors, while embedding Au nano-particles in the semiconducting organic ink was proposed by Ref. 143. Embedding CNTs in the semiconducting ink^{27,28} allowed us to get $\mu \sim 0.07 cm^2 V^{-1} s^{-1}$ at room conditions.

We combine our graphene-ink with one of the most commonly used organic polymers for ink-jet printing, Poly[5,5'-bis(3-dodecyl-2-thienyl)-2,2'-bithiophene] (PQT-12)¹³⁴⁻¹³⁶ in order to investigate its viability as interchain hopping enhancer (similarly to Au nanopar-



FIG. 13: a) Output and b) transfer characteristics of an inkjet printed graphene/PQT TFT.

ticles and CNTs). PQT-12 is widely used due to the higher environmental stability (up to 300 days at room conditions¹⁴⁴), with respect to other organic semiconducting inks^{143,144}. Graphene can bridge the polymer chains, allowing a more efficient charge transport.

We fabricate a graphene/PQT-12 TFT following the steps shown in Figs.12a,b,c. Fig.13a plots its output characteristics at V_g =-2,-5,-20 V. For each V_g , V_d is swept from 0 to -30 V in steps of 2V. At V_d =-2V, we get $\mu \sim 0.17 \text{cm}^2 \text{V}^{-1} \text{s}^{-1}$ and an ON/OFF ratio~4×10⁵. This μ is about ten times that of ink-jet printed CNTs/PQT-12 TFTs^{27,28} at ON/OFF~10⁵. When compared to pure organic semiconducting polymers, our μ is ~20 times higher than ink-jet printed PQT-12^{135,136}, and twice the highest reported μ for ink-jet printed TFTs made of pure (Poly(2,5-bis(3-tetradecyllthiophen-2-yl)thieno[3,2-b]thiophene)^{18,143,145,146}. Thus, the combination of our graphene-ink with organic semiconducting inks is promising for high performance printed electronics.

IV. CONCLUSIONS

We demonstrated ink-jet printing of graphene. Liquid phase exfoliated graphene is an ideal and low cost material for the fabrication of transparent conductive inks. Our graphene-ink was used to print TFTs with μ up to~95cm²V⁻¹s⁻¹. It was also combined with PQT-12 to fabricate devices with $\mu \sim 0.2 \text{cm}^2 \text{V}^{-1} \text{s}^{-1}$ and ON/OFF ratios~4×10⁵. This demonstrates the viability of graphene-inks for flexible and transparent electronics. We acknowledge funding from the Royal Society Brian Mercer Award for Innovation, the ERC

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