

1 **Mollusk carbonate thermal behaviour and its implications in**  
2 **understanding prehistoric fire events in shell middens**

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26 Thermal-induced diagenesis; Pyrotechnology

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28

29 **Abstract**

30 Archaeological shell middens are particularly important for reconstructing prehistoric human

31 subsistence strategies. However, very little is known about shellfish processing, especially when related

32 to the use of fire for dietary and disposal purposes. To shed light on prehistoric food processing  
33 techniques, an experimental study was undertaken on modern gastropod shells (*Phorcus lineatus*). The  
34 shells were exposed to high temperatures (200-700 °C) to investigate subsequent mineralogy and macro-  
35 and microstructural changes. Afterwards, the three-pronged approach was applied to archaeological  
36 shells from Haua Fteah cave, Libya (*Phorcus turbinatus*) and from shell midden sites in the United Arab  
37 Emirates (*Anadara uropigimelana* and *Terebralia palustris*) to determine exposure temperatures.  
38 Results indicated that shells from the Haua Fteah were exposed to high temperatures (600 - 700 °C)  
39 during the Mesolithic period (c. 12.7 - 9 ka), whereas specimens from the Neolithic period (c. 8.5 - 5.4  
40 ka) were mainly exposed to lower temperatures (300 - 500 °C). The thermally-induced changes in *A.*  
41 *uropigimelana* and *T. palustris* shells from the South East Arabian archaeological sites were similar to  
42 those seen in *Phorcus* spp. suggesting a broad applicability of the experimental results at an interspecific  
43 level. Although heat significantly altered the appearance and mineralogy of the shells, <sup>14</sup>C<sub>AMS</sub> ages  
44 obtained on burnt shells fit within the expected age ranges for their associated archaeological contexts,  
45 indicating that robust radiocarbon ages may still be obtained from burnt shells. Our study indicates that  
46 the combination of microstructural and mineralogical observations can provide important information to  
47 infer shellfish processing strategies in prehistoric cultures and their change through time.

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## 51 1. Introduction

52 Shells grow incrementally throughout the lifetime of mollusks and function as protection and support  
53 structures. Shells also serve as excellent palaeoenvironmental archives (i.e. Jones, 1983; Schöne et al.,

54 2004; Butler et al., 2013), because they faithfully record the physical and chemical conditions of their  
55 ambient environment and temporal changes to these. Such information is stored in the form of  
56 geochemical and structural properties (Epstein, 1953; Goodwin et al., 2001; Schöne, 2008).  
57 Sclerochronology is the research field that studies the temporal context of shell chemical composition  
58 (i.e. stable isotopes and trace elements) and physical accretionary patterns to produce extremely highly  
59 resolved palaeoenvironmental reconstructions (Schöne et al., 2005; Miyaji et al., 2007; Milano et al.,  
60 2017; Oschmann, 2009). For example, shell oxygen isotope content ( $\delta^{18}\text{O}_{\text{shell}}$ ) is routinely used as  
61 paleothermometer (Schöne et al., 2005; Ferguson et al., 2011; Prendergast et al., 2013; Prendergast and  
62 Schöne, 2017).

63

64 A rapidly growing interest in the research field of sclerochronology supports the spread of its  
65 methodologies and approaches to different disciplines such as archaeology and environmental  
66 biomonitoring (Mannino and Thomas, 2002; Andrus, 2011; Steinhardt et al., 2016; Schöne and Krause,  
67 2016). The analysis of mollusk shell material is especially relevant within the framework of prehistoric  
68 archaeology. Shellfish have been an important dietary component since the emergence of anatomically  
69 modern humans (~ 300 kyr ago; de Lumley, 1966), due to their easy accessibility, reliable availability  
70 throughout the year and source of proteins and micronutrients essential for physical development  
71 (Erlandson, 2001; Broadhurst et al., 2002; Marean et al., 2007; Fa, 2008).

72

73 The application of sclerochronology substantially broadens the potential use of mollusk shells in  
74 archaeology. Besides being suitable palaeoenvironmental archives, the shells become a key tool for  
75 understanding the response of human behaviour to climatic changes. For instance,  $\delta^{18}\text{O}_{\text{shell}}$  is used to  
76 constrain the season of mollusk collection (Mannino et al., 2007; Burchell et al., 2013; Prendergast et

77 al., 2016). In turn, seasonality provides information on the seasonal mobility of hunter-gatherer societies  
78 and helps to identify whether sites were permanently or ephemerally occupied (Shackleton, 1973;  
79 Mannino and Thomas, 2002; Eerkens et al., 2013). However, the processes related to mollusk  
80 preparation, consumption and disposal are still largely unknown (Milano et al., 2016). In some cases, it  
81 has been observed a modification of the shell structure by removal of the gastropod apex with stone tool,  
82 thorn or canine tip to allow the mollusk to be sucked out (Girod, 2011). As for bivalves, the shells were  
83 sometimes scarred or wholesale smashed (Hammond, 2014). Evidence of pyrotechnology associated  
84 with mollusk shell middens suggests that a certain degree of heat exposure may have been involved in  
85 some cases (Erlandson et al., 1999; Berna and Goldberg, 2008; Taylor et al., 2011). However, few  
86 studies have addressed the reconstruction of such processes in the framework of prehistoric marine  
87 resource exploitation (Andrus and Crowe, 2002; Aldeias et al., 2016; Milano et al., 2016) and this study  
88 will be the first instance where this has been carried out.

89

90 The present study builds upon previous work by Milano et al. (2016) who found that significant  
91 structural and chemical changes occurred in modern mollusk shells when they were heated at  $\geq 300$  °C  
92 for 20 minutes or more. Here, we investigate the effects of a shorter heat exposure (5 minutes), since  
93 shellfish are generally thought to be processed for short periods of time. To achieve this aim, modern  
94 *Phorcus lineatus* are used as a calibration tool to understand the response to thermal treatments at the  
95 macro-, microstructural and geochemical level. These findings are then applied to archaeological  
96 specimens from the Haua Fteah, eastern Libya (*Phorcus turbinatus*) to test whether high temperatures  
97 exposure can be detected in archaeological shells of the same genus using the three-pronged approach  
98 developed in the experimental phase. Archaeological specimens from the United Arab Emirates  
99 (*Anadara uropigimelana* and *Terebralia palustris*) are used to understand the efficacy of this three-

100 pronged approach on a broader scale to other mollusk species. Furthermore, given the importance of  
101 shells for radiocarbon dating coastal sites and the abundance of burnt shells in middens (Douka et al.,  
102 2014; Lindauer et al., 2016), we present preliminary results on the influence of thermal alterations on  
103 shell radiocarbon dating results.

104

## 105 2. Materials and methods

106 A total of 41 shells were analysed (Table 1). Fifteen live-collected shells of *P. lineatus* were used for the  
107 experimental phase. The archaeological material consisted of eighteen specimens of *P. turbinatus*  
108 selected from key occupation contexts in the Mesolithic and Neolithic layers from the Haua Fteah cave  
109 in Libya, as well as six specimens of *A. uropigimelana* and two specimens of *T. palustris* from shell  
110 midden sites near Kalba in the United Arab Emirates (Fig. 1).

111

### 112 2.1 Shell material: *P. lineatus* and *P. turbinatus*

113 *Phorcus* spp. are rocky shore gastropods living in the intertidal zone along the Mediterranean (*P.*  
114 *turbinatus*: Menzies et al., 1992; Schembri et al., 2005; Mannino et al. 2008; Prendergast et al., 2013)  
115 and Eastern North Atlantic coasts (*P. lineatus*: Kendall, 1987; Donald et al., 2012; Gutiérrez-Zugasti et  
116 al., 2015). Both species are sensitive to seasonal environmental changes, specifically water temperature  
117 fluctuations. Furthermore, they are extremely abundant in prehistoric sites of this region (Mannino and  
118 Thomas, 2001; Colonese et al., 2011; Hunt et al. 2011; Bosch et al., 2015; Gutiérrez-Zugasti et al.,  
119 2015). Therefore, many previous studies have used their geochemical data for palaeoenvironmental  
120 reconstructions (Mannino et al., 2003; Colonese et al., 2009; Prendergast et al., 2016).

121

122 In the present study, *P. lineatus* (previously known as *Osilinus lineatus* and *Monodonta lineata*)  
123 was selected as a modern reference and *P. turbinatus* (previously known as *Osilinus turbinatus*,  
124 *Monodonta turbinata* and *Trochocochlea turbinata*) for the archaeological case study. Despite being  
125 two distinct species, they share similar shell shape, size and the same microstructural organization and  
126 mineralogy and they live within the same microenvironments on the Mediterranean and Atlantic coasts.  
127 Their shell consists of two aragonitic layers with specific organizations. The outer shell layer (oSL) is  
128 arranged in spherulitic prismatic microstructures whereas the inner shell layer (iSL) consists of nacre  
129 (Mannino et al., 2008; Milano et al., 2016). The minor phenotypic and taxonomically relevant  
130 differences between the two species in shell coloration and aperture size likely do not influence the  
131 structural behaviour of the shell in reaction to thermal stress.

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133

## 134 2.2 Shell material: *A. uropigimelana* and *T. palustris*

135 *A. uropigimelana* and *T. palustris* inhabit tropical mudflats. *A. uropigimelana* is a bivalve of the Arcidae  
136 family (ark clams), which is widely distributed in East Africa and the Central Pacific. It lives in shallow  
137 waters (1-8 m) in association with seagrass beds and can reach 45 years of age (Tebano and Paulay,  
138 2001; Petchey et al., 2013). *T. palustris*, also known as mud whelk, is one of the largest gastropods of  
139 mangrove-dominated habitats in the Pacific Ocean (Plaziat, 1984; Houbrick, 1991; Carlen and Olafsson,  
140 2002). It is typically found in the intertidal zone and its lifespan is currently unknown. Possibly the adult  
141 size is reached after about four years, when it moves from offshore environments into the mangrove  
142 settings (Nishihira et al., 2002). Both species are very abundant in the prehistoric record in the Arabian  
143 Sea region (Biagi et al., 1984; Beech and Kallweit, 2001; Gardner, 2005; Lindauer et al., 2017).

144

145 The shell of *A. uropigimelana* is entirely aragonitic and consists of three layers. The oSL is  
146 organized in branching crossed-lamellar structures, the middle shell layer (mSL) is formed of linear-  
147 crossed lamellae and the iSL is characterized by irregular complex-crossed lamellae. Likewise, the shell  
148 of *T. palustris* is aragonite and is organized in simple-crossed lamellar (oSL and iSL) and linear-crossed  
149 lamellar microstructures (mSL). The terminology follows the author's observations and the description  
150 from Carter et al. (2012).

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152

### 153 2.3 Modern and archaeological settings

154 Modern specimens of *P. lineatus* were collected from Praia do Tamariz, Portugal (38.7036 °N, -9.4038  
155 °E) on 10 August 2016. The shells were gathered from a rocky portion in the mid-intertidal zone just  
156 outside the estuary of the Tajo River on the Atlantic coast (Fig. 1). To minimize the effects of ontogeny  
157 on the stable isotope results, shells with similar sizes were selected. Maximum shell height ranged  
158 between 12 and 14.3 mm and the maximum shell width ranged between 13.3. and 15.7 mm. This size  
159 class represents the average shell size of *P. lineatus* adult population in the region (da Costa, 2015).

160 Immediately after collection, the specimens were frozen for about one hour and subsequently the soft  
161 tissues were removed.

162

163 The archaeological remains of *P. turbinatus* were collected from the Haua Fteah cave, Libya (Fig.  
164 1). The Haua Fteah cave was first excavated in 1951-1955 by Charles McBurney, who cut a 14 m-deep  
165 stepped trench from the level of the present cave floor, exposing a deep sequence of occupation dating  
166 from the last interglacial to the Holocene (McBurney, 1967). New excavations (ERC:TRANSNAP)  
167 coordinated by Prof. Graeme Barker between 2008-2015 reopened the McBurney trench and cut an

168 overlapping series of small trenches down its southern face (Barker et al., 2010, 2012; Farr et al., 2013).  
169 Eleven of the eighteen specimens used in this study were from Trench U, a small (0.75 m x 1 m) cutting  
170 in the upper 2 m of the sediments which exposed occupation levels of Neolithic character and age. The  
171 Neolithic phase in the cave has a modelled age of ca. 8.5 - 5.4 cal. BP (Douka et al., 2014). These shells  
172 were excavated in 2012 from contexts 742, 743 and 747. These sediments, especially context 747, were  
173 associated with evidence of extensive burning in form of frequent charcoal, visibly charred animal bone  
174 and shell, and lithic artefacts with thermoclastic fractures. Seven specimens were selected from Trench  
175 M, a 2 m x 1 m trench cut down the same side of the McBurney trench at ca. 2-8 m depth. The  
176 specimens were excavated in 2009 and 2010 from contexts 10,002, 10,005 and 10,006, which have been  
177 dated to the Early Holocene and can be ascribed to the Mesolithic Capsian phase of occupation defined  
178 by McBurney (1967). The sediments were similar in character to those of Trench U in terms of their  
179 plentiful evidence for firing events.

180

181 The archaeological remains of *A. uropigimelana* and *T. palustris* were excavated from open sites  
182 on the Gulf of Oman coast in the United Arab Emirates (Fig. 1). Specimen KS Ana1, KSM UBL Ana2,  
183 KK1 Ana3 and KK1 Ana4 came from the shell midden KK1 situated near the mangroves at Kalba,  
184 along the northern coast of the Gulf of Oman (Lindauer et al., 2016). Specimen KS Ana1 was found on  
185 the beach nearby and it yielded a radiocarbon age of 7328-7113 cal. BP. It may have remained in the  
186 mangrove sediments or in the sabkha (salt flat) since Neolithic times or perhaps washed out recently  
187 from an archaeological midden where it was found on the beach. This would also explain its good  
188 preservation. The shell midden KK1 is at the outermost edge of the sabkha of the mangrove at Kalba  
189 and contains only material from the Neolithic, in two distinct phases. These are reflected in the shell <sup>14</sup>C  
190 ages that range from ca. 7146 - 6946 cal. BP at the base to ca. 6600 - 6450 cal. BP at the surface (1σ).



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192           Specimens K4 SL Ana3, K4 SL GA1, K4 BZ GT1 and K4 BZ UGT1 were excavated from site  
193 Kalba K4, lying close to the modern town of Kalba north of a mangrove area. This site has evidence of  
194 settlement structures that can be ascribed to the Bronze Age and Iron Age. Two distinct layers (MBZ 1  
195 and MBZ 2) situated close to an Iron Age mudbrick wall contain a large amount of shells, both burnt  
196 and unburnt, together with charcoal fragments. The lower layer (MBZ1) contained fewer shells than the  
197 immediately overlying layer MBZ2. Both layers are around 30 cm thick. Dating measurements show  
198 that both layers can be ascribed to the Middle Bronze Age (Lindauer et al., 2017). The shells of the  
199 lower MBZ1 have a mean age of ca. 3882 - 3727 cal. BP and the upper MBZ2 shells of ca. 3695 - 3641  
200 cal. BP. The results are in accordance with the archaeological data from previous excavations (Phillips  
201 and Mosseri-Marlio, 2002).

202

203

## 204 2.4 Experimental design

205 The experiment on modern shells followed a similar protocol to that described by Milano et al. (2016).  
206 However, in this case the shells were exposed to different temperatures for 5 min. The choice of this  
207 duration was twofold: (1) to identify potential changes in shell behaviour after short (5 min) heat  
208 exposure as compared with the medium (20 min) and long (60 min) heat exposures that were tested in  
209 the previous study (Milano et al. 2016) and (2) to attempt to reconstruct the temperatures at which  
210 archaeological shellfish were processed. Sparse ethnographic observations (e.g. Meehan, 1977, 1982) as  
211 well as modern traditions (e.g. Smith, 1978) suggest that the bivalves are processed for only few minutes  
212 before consumption.

213

214 The shells were heated at 200 °C, 300 °C, 400 °C, 500 °C, 600 °C and 700 °C in a high  
215 temperature tube furnace. Specimens were placed in the furnace when the desired temperature was  
216 stabilized to ensure the reproducibility of the experiment and to minimize the noise deriving by  
217 temperature fluctuations. In order to record potential weight loss, each shell was weighted before and  
218 after the roasting experiment to the nearest 1 mg.

219

220

## 221 2.5 Shell preparation

222 To study the specimens, modern and archaeological shells were fully embedded in Struers EpoFix resin  
223 and air-dried for 24 hours. This strategy prevented the shells from breaking during the following  
224 preparation steps. Each resin block was glued to a plexiglass cube and a low-speed precision saw  
225 (Buehler Isomet 1000) was used to cut two sections (ca. 2 mm thick) perpendicular to the growth  
226 direction. One section of each specimen was glued to a microscope slide with JB KWIK epoxy resin and  
227 was later used for geochemical analyses. The other section was used for SEM observations. All shell  
228 sections were ground on a Buehler Metaserv 2000 grinder-polisher machine with silicon carbide papers  
229 of different grit sizes (P320, P600, P1200, P2500). After each grinding step, the slabs were immersed in  
230 de-ionized water and ultrasonically rinsed for 2 minutes. A Buehler VerduTex cloth with a 3 µm  
231 diamond suspension was used to polish the sections. The sections used for SEM analysis were etched for  
232 5 s in 1 vol% HCl and bleached for 30 min in 6 vol% NaOCl, to remove the organic sheets masking the  
233 microstructures.

234

235

## 236 2.6 Macro-, microstructural and mineralogical analyses

237 Modern shells were photographed before and after the thermal treatments allowing a visual comparison  
238 at macroscopic scale. To identify thermal changes, the coloration of the shells was identified according  
239 to the Pantone colour system. The sections prepared for the SEM were analysed using a Zeiss Axio  
240 Imager.A1m stereomicroscope to record potential changes in the appearance of the two shell layers.  
241 Prior to the microstructural analysis, the samples were sputter-coated with a 2 nm-thick layer of  
242 platinum using a Leica EM ACE200 Vacuum Coater. The surface of the sections was studied with a 3<sup>rd</sup>  
243 generation LOT Quantum Design Phenom Pro desktop SEM with 10 kV accelerating voltage equipped  
244 with a backscatter electron detector. Shell mineralogy was investigated in each shell layer by using a  
245 Horiba Jobin Yvon LabRam800 spectrometer equipped with an Olympus BX41 optical microscope. The  
246 instrument employed a 532.21 nm laser wavelength, a 400 µm confocal hole, a grating with 1800  
247 grooves/mm, an entrance slit width of 100 µm and 50× long-distance objective lens. The integration  
248 time of each scan ranged between 3 and 5 s.

249

250

## 251 2.7 Stable isotope analysis

252 The stable isotope composition of the modern specimens was analysed on carbonate powder sampled  
253 after heating the shells. Carbonate material was obtained from each shell by microdrilling the inner layer  
254 (nacre) with a Rexim Minimo dental drill mounted to a stereomicroscope and equipped with a conical  
255 drill bit of 300 µm in diameter (Komet/Gebr. Brasseler GmbH & Co. KG, model no. H52 104 003). A  
256 Thermo Fisher MAT 253 gas source isotope ratio mass spectrometer in continuous flow mode coupled  
257 to a GasBench II at the Institute of Geosciences, University of Mainz was used to analyse the carbonate  
258 samples by phosphoric acid digestion (70 °C for 120 min). The isotope data were calibrated against a  
259 NBS-19 calibrated Carrara marble standard ( $\delta^{18}\text{O} = -1.91\text{‰}$ ). The average internal precision ( $1\sigma$ ) was

260 0.03‰ and the external reproducibility ( $1\sigma$ ) was 0.05‰. The  $\delta^{18}\text{O}$  profiles of the heated shells were  
261 compared to the  $\delta^{18}\text{O}$  profiles of the control specimens, which were not exposed to high-temperature  
262 treatments.

263

264

## 265 2.8 Radiocarbon dating

266 For radiocarbon dating, small chips were taken near the apex of three *P. turbinatus* specimens. In order  
267 to avoid any contamination from sedimentary calcium carbonate or a mixture of different shell layers,  
268 the external layer of the shells was physically removed. The fragments of nacre were further processed  
269 at the Curt-Engelhorn-Center for Archaeometry, Mannheim, Germany. The material was acidified in  
270 phosphoric acid at 70 °C by using an Autosampler system and measured using the MICADAS system  
271 (Kromer et al., 2013; Wacker et al., 2013). The results were calibrated with the Marine13 curve (Reimer  
272 et al., 2013) and corrected using the Mediterranean reservoir correction ( $\Delta R$ ) of  $58 \pm 85$   $^{14}\text{C}$  years  
273 (Reimer and McCormac, 2002) using OxCal v.4.3.2.

274

275

## 276 2.9 Statistical analyses

277 ANOVA, Mann-Whitney and Tukey's HSD post hoc tests were performed using the software PAST to  
278 test whether the 5-minute heat exposure had a significant effect on shell geochemistry as well as whether  
279 the different durations changed the oxygen isotopic signatures.

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## 283 3. Results

### 284 3.1 Experimental results on the effect of roasting on the shells

285 At macroscopic level, the 5-minute thermal treatment caused a similar effect as reported for *P.*  
286 *turbinatus* in the previous experiment by Milano et al. (2016), in which shells were roasted for 20  
287 minutes and 60 minutes. Visible alteration of the shell surface occurred at temperatures higher than 200  
288 °C. For instance, the typical background cream coloration (Pantone: 7497 C) alternated with pigmented  
289 blotches (438 C) started to change at 300 °C, when the shells acquired a dark coloration with a tendency  
290 toward a brown shade (background: 7531 C; blotches: 7518 C). At 400 and 500 °C, the shells became  
291 grey (background: 404 C; blotches: 417 C) and the outer shell layer started to detach. At 600 °C, the  
292 coloration only slightly changed (background: 416 C; blotches: 417 C) and the material brittleness  
293 increased, whereas at 700 °C, the coloration turned into a cream-white tone (background and blotches:  
294 401 C; Fig. 2A). Furthermore, the heat exposure induced a 14 to 25% weight loss, with a greater loss at  
295 higher temperatures (Fig. 2B).

296

297 At the microstructural level, roasting at 200 °C did not cause any visible change of the shell  
298 microstructure. First-order and second-order prisms (oSL) and platelets (iSL) of the heated samples did  
299 not appreciably differ from those of unheated samples. At 300 °C, the oSL did not show any visible  
300 alteration whereas the nacre platelets became slightly porous with nanometer sized holes appearing on  
301 their surfaces (Fig. 2C-D). At 400 °C, the single second-order prisms of the oSL were still recognizable.  
302 However, sheets of organic material emerged between the first-order prisms partially covering the shell  
303 surface. The platelets of the nacre started to fuse and large cracks developed on the surface. After  
304 heating at 500 °C, the oSL underwent a significant transformation with the formation of new irregular-  
305 shaped units surrounded by organic sheets. Meanwhile, the iSL surface was partially covered by organic

306 matter which emerged from the inter-lamellar voids that formed at 400 °C (Fig. 2C-D). The appearance  
307 of the oSL at 600 °C is very similar to the previous treatment (500 °C) with an enhanced presence of  
308 organic material. The platelets of the nacre showed a higher degree of fusion in every direction with  
309 large quantities of organics in between them. At 700 °C, the surface of the biomineral units of the oSL  
310 units became more compact with a decrease in porosity. At the same time, the agglomeration of the iSL  
311 platelets led to the formation of new boundaries delineating larger and irregular units (Fig. 2C-D).

312

313 The shell mineralogy was also affected by heat (Fig. 2C-D). Originally, the shells were fully  
314 aragonitic as indicated by Raman peaks at 152 cm<sup>-1</sup>, 180 cm<sup>-1</sup>, 207 cm<sup>-1</sup>, 707 cm<sup>-1</sup> and 1086 cm<sup>-1</sup>. The  
315 transition from aragonite into calcite started at 400 °C in the oSL and is indicated by an extra peak at  
316 281 cm<sup>-1</sup> (Fig. 2C). At this temperature, however, the iSL was still completely aragonitic. At 500 °C,  
317 both layers were transformed into calcite with peaks at 153 cm<sup>-1</sup>, 281 cm<sup>-1</sup>, 713 cm<sup>-1</sup> and 1086 cm<sup>-1</sup>.

318

319 Similar to our previous results (Milano et al. 2016), the shell oxygen isotope composition changed  
320 in response to the heat treatment. The  $\delta^{18}\text{O}_{\text{shell}}$  values of the control sample ranged between - 1.0 and +  
321 1.0 ‰. The oxygen isotope range is more negative than the range obtained by the control shells used in  
322 Milano et al. (2016). This offset reflects the difference of the local environments. The shells from  
323 Portugal used in the present study were collected from an area characterized by much larger freshwater  
324 input, and hence lower  $\delta^{18}\text{O}_{\text{shell}}$  values, than the specimens from Libya used in Milano et al. (2016). The  
325 oxygen isotope composition of shells processed at 200 °C and 300 °C for 5 mins ranged between - 0.5  
326 and + 0.7 ‰ (Fig. 3A). According to the Mann-Whitney *u*-test there is no statistical difference between  
327 these heated samples and the control shells (*p* > 0.05; Fig. 3B). The effect of the shell exposure at 400  
328 °C could not be tested, because the nacre layer was too thin to obtain sufficient carbonate powder. As

329 expected, higher temperatures (500 - 700 °C) had a significant influence on  $\delta^{18}\text{O}_{\text{shell}}$ . Average values  
330 were offset by  $-1.5\text{‰} \pm 0.2$  with respect to the average values of the control material. Minimum and  
331 maximum values were offset by  $-0.8\text{‰} \pm 0.1$  and  $-0.8\text{‰} \pm 0.7$ , respectively (Fig. 3A). The oxygen  
332 isotopic composition of the shells heated over 500 °C was statistically different from the lower  
333 temperature treatments and the control conditions (Mann-Whitney test  $p < 0.05$ ; Fig. 3B). A comparison  
334 between the current data and the results published in Milano et al. (2016) was used to determine the  
335 influence of the duration of the heating process on the shell. The offsets of the  $\delta^{18}\text{O}_{\text{shell}}$  values from the  
336 respective control values were calculated for the three duration treatments (5, 20 and 60 min). The  
337 offsets of minimum, average and maximum  $\delta^{18}\text{O}_{\text{shell}}$  values were statistically similar among all treatment  
338 durations (ANOVA  $p > 0.05$ ).

339

340

### 341 3.3 Archaeological shells: structural and mineralogical organization

342 The *P. turbinatus* specimens from the Haua Fteah presented inter-individual differences in their  
343 appearance at macroscopic scale (Fig. 4). All of the Mesolithic specimens (Trench M) showed a  
344 homogenous dark grey surface with areas of partial detachment of the oSL (i.e. M002\_1 in Fig. 4D,  
345 Supplementary material Fig. S1). The nacre was less uniform and lighter grey in colour. In general,  
346 these shells looked similar to those of the experiment roasted at 600 °C (Fig. 2A). Two specimens from  
347 the Neolithic context 747 (U747\_1 and U747\_2) were lighter grey and looked similar to shells heated at  
348 500 °C (Supplementary material Fig. S1). The third specimen from this layer (U747\_3) showed an  
349 iridescent iSL and an overall coloration likely related to lower heating temperatures (i.e., 300 °C;  
350 Supplementary material Fig. S1). All four Neolithic *P. turbinatus* specimens from context 743 were  
351 light brown in colour and showed iridescent nacre (i.e. U743\_3 in Fig. 4B). The same applied to the two

352 shells from context 742 (U742\_2 and U742\_4; Fig. 4A, Supplementary material Fig. S1), whereas  
353 U742\_1 and U742\_3 were grey as if subjected to higher temperatures (Fig. 4C, Supplementary material  
354 Fig. S1).

355

356 In good agreement with the observations of the macro appearance of the shells, microstructures  
357 varied greatly between individuals. The prisms in the oSLs of the Mesolithic specimens from were  
358 replaced by porous irregular units and the nacre platelets were fused together into larger agglomerates  
359 (Fig. 4D, Supplementary material Fig. S2). The integral microstructural reorganization and the  
360 subsequent absence of any of original secondary prisms (oSL) and platelets (iSL) recall the structural  
361 arrangement of the modern samples exposed to high temperatures (600 °C; Tab. 2). A similar  
362 architecture was observed in the Neolithic specimens U747\_1, U747\_2 and U742\_3. On the other hand,  
363 all shells from context 743 and specimens U747\_3 and U742\_4 showed well-defined prismatic  
364 structures in the oSL and slightly porous platelets in the iSL, which looked similar to the experimental  
365 shells heated at 300 °C (Fig. 4B, Supplementary material Fig. S2). Shell U742\_1 represented the only  
366 case identifiable as processed at a temperature of ca. 500 °C (Table 2). The prisms of the oSL appeared  
367 to have undergone major fusion at 400 °C (Fig. 4C). However, the single elongated prisms were still  
368 partially visible and they were not completely transformed as in the case of the material burned at 600  
369 °C. A similar pattern was observed in the nacre, where the agglomerations of platelets started to define  
370 new structural units (Fig. 4C). The microstructures of specimen U742\_2 were perfectly preserved and  
371 did not show any sign of thermal alteration (Fig. 4A, Table 2).

372

373 According to Raman spectroscopy, seven shells were entirely aragonitic and eleven were calcitic  
374 (Fig. 5). All Mesolithic specimens and four of the Neolithic specimens consisted of calcite (Fig. 5A),



375 whereas the remaining specimens from Trench U were aragonitic (Fig. 5B). No transition phase was  
376 detected and the two shell layers always shared the same mineralogy. Based on the results presented  
377 above, the calcitic shells were categorized as having been exposed to temperatures in excess of 500 °C  
378 (Tab. 2).

379

380 Similar to the *P. turbinatus*, some archaeological specimens of *A. uropigimelana* and *T. palustris*  
381 from the United Arab Emirates sites displayed an altered overall appearance (Fig. 6). Shells K4 SL  
382 Ana3, KSM UBL Ana2, KK1 Ana3 and KK1 Ana4 showed the characteristic homogenous creamy  
383 colour of unaltered *A. uropigimelana* (Fig. 6A; Supplementary material Fig. S3). A slightly darker  
384 shade, especially on the inner shell surface, was visible in KK1 Ana4. Two shells, KS Ana1 and K4 SL  
385 GA1, showed a grey shell surface. The first was characterized by a light tone, whereas the latter was  
386 darker with shaded areas on the inner shell surface clearly indicating heat exposure (Fig. 6A;  
387 Supplementary material Fig. S3). In the case of *T. palustris*, one specimen (K4 BZ UGT 1) had its  
388 surface coloration preserved, whereas a second specimen (K4 BZ GT 1) was very dark (Fig. 6B).

389

390 The well-preserved *A. uropigimelana* specimens K4 SL Ana3, KSM UBL Ana2, KS Ana1, KK1  
391 Ana3 and KK1 Ana4 showed well-defined crossed-lamellar structures in the oSL, linear-crossed  
392 lamellae in the mSL and irregular complex-crossed lamellae in the iSL. The excellent preservation state  
393 was also corroborated by the presence of organic microtubules across the shell layers (Fig. 6A;  
394 Supplementary material Fig. S3). Conversely, shell K4 SL GA1 displayed altered microstructures. The  
395 first order lamellae of the oSL and mSL were still visible, whereas the morphometric characteristics of  
396 the third order lamellae were changed. The units were fused together in large agglomerates with new  
397 boundaries (Fig. 6A). These features recall the oSL of *Phorcus* spp. heated at 600 °C. In the iSL, the

398 lamellae lost their individuality and elongated shape, forming compact clusters of material separated by  
399 organic sheets and cracks (Fig. 6A). The *T. palustris* specimen K4 BZ UGT 1 showed the typical simple  
400 and linear-crossed lamellar microstructures (Fig. 6B), whereas in specimen K4 BZ GT 1 lamellar  
401 biomineral units were transformed into large irregular-shaped assemblages in the oSL and iSL. In the  
402 mSL, small granular carbonate units with well-defined boundaries prevailed (Fig. 6B). Such alteration  
403 was likely related to high temperatures ( $\geq 500$  °C). However, in this case an accurate estimate of the  
404 processing temperature cannot be provided based on the microstructures, because no such architecture  
405 was previously observed in the mSL.

406

407 Typical aragonite Raman spectra occurred in all but two *A. uropigimelana* and *T. palustris*  
408 specimens. The mineralogy of K4 SL GA1 and K4 BZ GT 1 was entirely calcitic (Fig. 7A-B).

409

410

### 411 3.4 Radiocarbon dating of *P. turbinatus* from Haua Fteah

412 Specimen U742\_2, with no sign of heat exposure (Fig. 4A), was dated to  $7184 \pm 21$   $^{14}\text{C}$  yr BP calibrated  
413 to 7772 - 7438 cal. BP ( $2\sigma$ ). Specimen U742\_1, with signs of high temperature exposure (Fig. 4C), was  
414 dated to  $7300 \pm 21$   $^{14}\text{C}$  yr BP calibrated to 7901 - 7556 cal. BP. both these dates fit with existing  $^{14}\text{C}$   
415 dates for context U742. Specimen U742\_3 (Fig. 4B), with burning marks, was dated to  $6378 \pm 21$   $^{14}\text{C}$  yr  
416 BP calibrated to 7002 - 6558 cal. BP. These dates fit with the current chronostratigraphic understanding of  
417 Context U743, which is paired with U742 stratigraphically (Fig. 8). Associated  $^{14}\text{C}$  radiocarbon ages are  
418 available from other materials dated in Trench U are also included in this study for comparative  
419 purposes (Table 3). The relatively broad spread of dates from U742 fits sedimentological model for this  
420 part of the Haua Fteah deposition sequence and all sit within the existing age range for these context.

421 Context U742 is one of a group of chronologically identical contexts (with U743, U744, U746 and  
422 U747) that form a cross section of a debris flow that contains archaeological material that can be  
423 attributed to the Neolithic.

424

425

426

## 427 4. Discussion

### 428 4.1 A comparative approach to thermal-induced diagenesis in biogenic carbonates

429 Diagenesis encompasses all physical and chemical processes occurring to the tissues after the death of  
430 the animal. The present study focuses exclusively on thermally-induced diagenesis associated with heat  
431 exposure. High temperatures accelerate a process that naturally would require thousands of years. When  
432 dealing with archaeological remains it is important to differentiate between the two types of diagenetic  
433 processes to ensure a correct interpretation of the results. In order to do so, the whole shell assemblage  
434 needs to be taken into consideration. Specimens from a certain archaeological layer are subjected to  
435 similar environmental conditions after burial (i.e. depth, sediment composition and soil pH). Therefore  
436 they are likely to share similar preservation conditions. The overall preservation state indicates whether  
437 natural diagenesis altered the shells or not. Any deviation from the average preservation state is likely  
438 related to different processes. In our case, shells without evidence of burning are generally well-  
439 preserved suggesting that natural diagenesis did not occur (Prendergast et al., 2016). Any altered shell is  
440 likely to be related to heat-induced diagenesis. Although this approach allows a reasonable distinction  
441 between natural and artificial diagenesis, further studies on naturally-altered shell material are needed to  
442 characterize the structural and mineralogical key features relative to this type of diagenesis.

443

444 As previously demonstrated by Milano et al. (2016), the macroscopic and microscopic appearance  
445 of *Phorcus* spp. shells gradually altered as the processing temperatures increased. Although the thermal  
446 exposure time was reduced to 5 minutes in the present study, similar mineralogical and geochemical  
447 changes were observed. The only exception is the temperature at which the aragonite-to-calcite  
448 transformation of the oSL occurred, i.e., 400 °C in the present study and 300 °C in Milano et al. (2016).  
449 Based on this discrepancy, we suggest that the duration of the thermal exposure may play a role in the  
450 development rate of such taphonomic processes. However, time is only of secondary importance in  
451 comparison to the actual processing temperature. In good agreement with the most of our results,  
452 Aldeais et al. (2016) demonstrated that the mineralogy of *Cerastoderma edule* and *Scrobicularia plana*  
453 shells showed no differences after roasting for 5 and 20 minutes. In abiogenic and biogenic minerals, the  
454 aragonite-to-calcite and microstructural transformations have been previously observed to occur very  
455 rapidly at critical temperatures around 400 °C and 500°C, respectively (Lécuyer, 1996; Koga et al.,  
456 2013). As a consequence, it will most likely be impossible to reconstruct the processing duration based  
457 on the analysis of archaeological shells, but it is possible to reconstruct the temperatures at which they  
458 were exposed. Consequently, it may be possible to postulate the type of fire or fuel that created those  
459 temperatures.

460

461 *P. lineatus* shells lost weight after roasting, especially at temperatures above 300 °C (21.5 %  
462 weight loss). Heat-induced weight losses have previously been reported from biogenic and abiogenic  
463 minerals (Yoshioka and Kitano, 1985; Balmain et al., 1999; Bourrat et al., 2007). Thermal exposure  
464 induces a three-step release of water and organic content. At first, the endothermic reaction provokes  
465 dehydration through evaporation of the water incorporated in the mineral phase (Yoshioka and Kitano,  
466 1985; Perić et al., 1996; Huang et al., 2009). Then, at temperatures between 250 °C and 450 °C, the

467 organic matter in the shell starts to degrade (Bourrat et al., 2007; Huang et al., 2009). In *Phorcus* spp.,  
468 this temperature range conforms to the appearance of large organic sheets partially masking the  
469 carbonate microstructures (Fig. 2). The degradation intensifies at higher temperatures (> 500 °C) and is  
470 associated with weight losses up to 50 % (Balmain et al., 1999; Zolotoyabko and Pokroy, 2007; Huang  
471 et al., 2009; Villagran et al., 2011).

472

473 In the present study, temperature was increased stepwise by 100 °C steps to systematically  
474 investigate structural and mineralogical changes of the shells. The outer shell layer of *P. lineatus* started  
475 to transform from aragonite to calcite at 400 °C. This process was completed at 500 °C. In the case of  
476 the nacre, no such broad transition phase was detected. Conversion to calcite took place at 500 °C, at  
477 which point the iSL was entirely calcitic. According to previous studies, however, polymorphic  
478 transition of mollusk shells already started at temperatures of 250 - 350 °C, and conversion to calcite  
479 was completed between 400 °C and 500 °C (Lécuyer, 1996; Maritan et al., 2007; Aldeias et al., 2016).  
480 Conversion temperatures seem to vary between species. As shown by Aldeias et al. (2016), the  
481 mineralogy of *C. edule* and *S. plana* reacts differently to heating. In *C. edule* polymorphic conversion  
482 occurs between 250 and 500 °C, whereas in *S. plana* the transformation only starts at 350 °C and is  
483 completed at 400 °C. The observed differences in polymorphic conversion temperatures may be  
484 explained by specific structural properties of the materials. Crystal morphology and crystallography  
485 were previously identified as potential factors controlling mineralogical transformations (Koga et al.,  
486 2013). In abiotic systems, aragonite crystals split and released water trapped within them. Furthermore,  
487 calcite nucleation occurs preferably along mineral cracks and twin boundaries (Bischoff and Fyfe, 1968;  
488 Liu and Yund, 1993). For these reasons, crystal characteristics and positions of boundaries influence the  
489 conversion reaction (Koga et al., 2013). The architectural diversity among mollusk species and shell

490 layers may be the origin of the variable reaction of different biominerals to thermal stress (Kobayashi  
491 and Samata, 2006). Aragonitic mineral units tend to lose density and acquire nanopores on their  
492 surfaces, while a network of cracks forms in the material (Perdikouri et al., 2011; Villagran et al., 2011;  
493 Gomez-Villalba et al., 2012). These features were also observed in the present study in *P. lineatus*  
494 specimens roasted at 400 °C, i.e., before full conversion into calcite.

495

496         Given the complexity in the thermal behaviour of biogenic materials, the three-pronged approach  
497 of mineralogical, macro- and microscopic observations is essential to better understand the thermally-  
498 induced diagenetic processes. The identification of specific changes in the shell can be used as indicator  
499 of shell processing temperatures, although possibly with some differences from species to species.

500

501

## 502 4.2 Implications of heat exposure for palaeoenvironmental reconstructions

503 Heat exposure can induce an exchange of oxygen isotopes with the surrounding environment, drastically  
504 altering initial shell isotopic composition (Andrus and Crowe, 2002; Milano et al., 2016). Temperatures  
505 higher than 300 °C are observed to cause a significant decrease in  $\delta^{18}\text{O}_{\text{shell}}$  (Milano et al., 2016). Our  
506 results are in good agreement with the previous findings suggesting that the use of burnt mollusk shells  
507 may have important implications for palaeoenvironmental reconstructions. A decrease in  $\delta^{18}\text{O}_{\text{shell}}$   
508 translates into an overestimation of the reconstructed water temperature, which can jeopardize the  
509 reliability of related reconstructions. Furthermore, burning at high temperatures can influence the results  
510 of palaeoseasonality studies by incorrectly estimating the season(s) of mollusk collection, which in turn  
511 may affect the interpretation of hunter-gatherer mobility and settlement use.

512

513 It is recommended to discard shells with signs of thermal exposure for isotopic analyses. However,  
514 it has to be noted that not all heating intensities imply isotopic alteration. Low temperatures (< 300 °C)  
515 do not induce significant changes in the  $\delta^{18}\text{O}_{\text{shell}}$ . Shells exposed to such temperatures can still be used to  
516 accurately estimate palaeotemperatures. However, a careful analysis is needed to differentiate between  
517 the different heat exposures and to ensure an optimal sample selection for isotopic studies. It is also  
518 advisable to avoid any form of pre-treatment involving heating, which was commonly adopted in the  
519 past as method for organic matter removal (Mook, 1971).

520

521

### 522 4.3 Prehistoric fire events and shell middens

523 Different types of fire events can be associated to shell middens, according to their nature and use. The  
524 first evidence for controlled fire is dated to around 300,000-250,000 years ago and it is related to a  
525 change in hominin habits, especially concerning diet (James et al., 1989; Shahack-Gross et al., 2014;  
526 Gowlett, 2016). Among the main benefits of using fire for dietary purposes are the maximization of  
527 energy gain and food digestibility and the development of social abilities such as storing, sharing and  
528 processing food (Ragir, 2000; Wrangham and Conklin-Brittain, 2003; Wrangham, 2009; Wrangham and  
529 Carmody, 2010). Several ethnographic studies suggest that cooking may play an important role in  
530 shellfish consumption (Percy, 1608; Meehan, 1977, 1982). Cooking is generally known to facilitate the  
531 extraction of the edible portion. In fact, the heat weakens the muscle attachment of the mollusks causing  
532 the valves to open (bivalves) and the flesh to detach more easily. In Australia and North America  
533 mollusks were placed over or underneath fire, coals or hot stones (Kroeber and Barrett, 1960;  
534 Beaglehole, 1974; Bailey, 1977). In other cases, they were cooked in earth ovens (Meehan, 1982;  
535 Waselkov, 1987; Thoms, 2008). Additionally, steaming was adopted as an alternative technique by

536 some ethnic groups such as the Maori in New Zealand and the Quileute and the Tlingit in the Pacific  
537 Northwest (Best, 1924; Reagan, 1934; Newton and Moss, 1984). Generally, the cooking processes were  
538 rather fast, possibly lasting only a few minutes (Meehan, 1977; Thoms, 2008). After being cooked,  
539 mollusk soft tissues were sometimes dried or smoked to enable storage for longer periods of time and  
540 for trading purposes (Greengo, 1952; Moss, 1993). In addition to food preparation, fire could have being  
541 used for trash disposal (Ford, 1989). Ethnographic observations by Meehan (1982) describe occasional  
542 burning of food debris by igniting grass on top of large mounds of disposed shells. However, in other  
543 cases, the shells were arranged in heaps and covered with sand, without involving any fire (Bird and  
544 Bird, 1997). Alternatively, Mougne et al. (2014) suggested the shell refuses to be directly thrown into a  
545 fire.

546 To contextualize our results in the light of the ethnographic records, the experimental work by  
547 Aldeias et al. (2016) simulating different firing techniques offers a valuable source of information. The  
548 authors showed that when a fire was set on top or underneath a pile of shells, there was a great degree of  
549 heterogeneity in the shell mineralogical response. The heat diffusion affected differently the single  
550 specimens according to their position in respect to the heat source, resulting in a mixture of aragonitic  
551 and calcitic shells. Our results indicate that the distribution of specific burning-related features is  
552 generally shared among the specimens of the same layer suggesting a homogenous exposure to heat. An  
553 exception to this trend is represented by the shells from the Neolithic layer U742, which will be  
554 discussed further below. Such uniform heat exposure likely implies heating of relatively small quantities  
555 of shells with a similar position in respect to the fire whereas large amounts of material, characteristic of  
556 disposal mounds, would induce a significant spatial heat gradient and diverse thermal response.  
557 Furthermore, concerning the food disposal by throwing the shells into the fire, this involves direct  
558 contact with extreme temperatures. Under these circumstances, the shells would become extremely



559 brittle and prone to a rapid disintegration with limited chances of fossil ‘survival’ (Milano et al., 2016).  
560 In this perspective, our data conform better with firing events associated to food processing rather than  
561 food disposal. However, further analysis on additional material is needed before unequivocally  
562 discriminate between the two different types of fire.

563

564 Estimation of shellfish heating temperatures from the Haua Fteah suggests a clear distinction  
565 between the two archaeological contexts studied here. The seven Mesolithic shells all appear to have  
566 been exposed to temperatures between 600 and 700 °C. These temperatures can be reached when the  
567 shellfish are situated in close proximity to the heat source without being directly in contact with it.  
568 Temperatures between 633 and 781 °C were recorded when experimentally roasting the mollusks on a  
569 surface with fuel and fire on top of them (Aldeias et al., 2016). Such temperatures fit well with our  
570 observations on the shell remains. However, it has to be considered that roasting temperatures also  
571 largely depend on the type of fuel used. In the experiments by Aldeias et al. (2016), pine wood and pine  
572 needles were adopted, whereas grass material is known to burn at lower temperatures (max. 430 °C;  
573 Wolf et al., 2013). In the light of these observations and our own results, shellfish during the Mesolithic  
574 phase at the Haua Fteah were probably processed underneath woody fires.

575

576 In contrast, the Neolithic shells from the Haua Fteah layers U747 and U743 have features pointing  
577 to lower temperature exposure: ca 71 % were heated at temperatures around 300 °C. The drastic increase  
578 in remains exposed to lower temperatures in the Neolithic phase can be explained by changes in the  
579 shellfish processing. A potential interpretation is related to the appearance of the first pottery during the  
580 Neolithic, which could have been used to prepare shellfish (McBurney, 1967; Douka et al., 2014;  
581 Prendergast et al., 2016). Although the use of vessels for processing marine resources has not been yet

582 attested around the Mediterranean during the Neolithic (Debono Spiteri and Craig, 2014), it has been  
583 previously observed in other regions (Craig et al., 2011; Taché and Craig, 2015). An experiment  
584 conducted by Maggetti et al. (2011) showed that clay pots exposed to fire can reach temperatures of  
585 around 300 °C after ca. 10 minutes, supporting our observations in the shells of layers U747 and U743.  
586 A second interpretation of the lower temperature exposure during the Neolithic encompasses similar  
587 heating processes with the adoption of substantially different types of fuel. As mentioned above, the  
588 usage of grasses as fuel could explain the lower burning temperatures experienced by the Neolithic  
589 shells.

590

591 As an exception to the general trend observed in the results, the shells from unit U742 present  
592 heterogeneous burning features related to temperatures from 300 °C to 600 °C, with one shell without  
593 any burning sign. On the basis of our observations two hypotheses can be discussed. Different burning  
594 degrees may indicate the presence of a thermal gradient related with the use of fire for disposal  
595 purposes. Alternatively, the shell may reveal changes in cooking processes in relation to the type of fuel  
596 used and/or cooking vessels adoption.

597

598 In the case of the archaeological samples from the United Arab Emirates, both species showed  
599 evidences related to heating at 500 - 600 °C. Archaeological sites in this area dated to the late Bronze  
600 Age or Classic Wadi Suq period (2000 - 1500 BC) are characterized by a large amount of pottery  
601 remains such as beakers and spouted globular jars (Carter, 1997). In this region, food processing with  
602 pots seems to have appeared during the Bronze Age, whereas roasting on fire was the preferred method  
603 in earlier times (Händel, 2013). However, the small sample size considered in the present study does not

604 allow us a more precise interpretation. In order to better understand the evolution of shellfish processing  
605 in the area, additional specimens are needed.

606

607 Together with intentional fires, natural burning events can occur in shell middens. A distinction  
608 between intentional and unintentional firing events can hardly be determined on the basis of shell  
609 thermal response. However, as in the case of disposal by directly throwing the shells into fire, long-  
610 lasting uncontrolled fires would lead to the complete disappearance of the shell remains within the  
611 midden. In our case, the preservation of large fragments and whole shells suggests that the heat exposure  
612 was likely not related to long-lasting natural fires. However, rapid fires cannot fully be discharge as  
613 potential burning sources.

614

615

#### 616 4.4 Heat exposure and radiocarbon dating robustness

617 Previous studies on shell-tempered pottery showed that these materials provide consistent  $^{14}\text{C}_{\text{AMS}}$   
618 dates (Rick and Lowery, 2013). To obtain resistant ceramics, shell-tempered pottery is typically fired at  
619 300 - 600 °C and - after cooling - fired again to 500 - 600 °C (Kingery et al., 1976; Herbert, 2008). Such  
620 temperatures do not seem to affect radiocarbon ages on the shell inclusions, which fit well with dates  
621 from charcoal and bone fragments of the same context (Rick and Lowery, 2013). Shells analysed in the  
622 present study did not differ from already available ages of the same stratigraphic context. According to  
623 the structural and mineralogical analyses, two of the samples were burned at 500 °C and 600 °C,  
624 whereas the third one was not burned. The relatively homogeneous AMS dates suggest that heating does  
625 not affect radiocarbon ages, even when polymorphic transformation has occurred. This may be due to  
626 the fact that the extremely small quantity of carbon remaining in the material (ca 0.1 wt%) is still

627 enough to allow AMS dating (Lanting et al., 2001). Furthermore, physical alterations of the biomineral  
628 such as structural unit enlargement and recrystallization seem to provide a barrier protecting the  
629 remaining carbon (van Strydonck et al., 2005).

#### 630 631 632 4.5 Methods to identify thermal exposure on shell remains

633 Our results show that the overall appearance of shells alone is an insufficient approach to identify  
634 heating-related changes (Milano et al., 2016). Although useful in cases of exposure to high temperatures  
635 with evident burning marks, examination by the naked eye has important limitations when it comes to  
636 lower temperatures. For instance, based on the outer shell layer, specimen U742\_2 seemed to have been  
637 exposed to 300 °C. However, according to more detailed analysis using SEM it turned out to be  
638 untreated.

639         The analysis of the mineralogical response to heat is valuable when combined with other methods.  
640 However, it may prove ambiguous when used alone. The archaeological material analyzed for this study  
641 did not show any carbonate polymorph transition phase. Therefore, Raman spectroscopy could only help  
642 to distinguish between temperatures below and above 500 °C. Our current and previous results indicate  
643 that a combined approach of optical microscopy, SEM and Raman spectroscopy is the most suitable  
644 solution to identify alterations related to shellfish processing. The combined use to these techniques will  
645 provide more robust data on heating exposure, whether the aim of the research is to preclude heated  
646 shells for palaeotemperature reconstructions (Milano et al., 2016) or to examine the food processing in  
647 more detail. Although the experimental phase was specifically designed for *Phorcus* spp., the  
648 observations showed similarities with the thermal behaviour of *A. uropigimelana* and *T. palustris*. Such  
649 analogy in the responses to heat suggests that temperature estimation can be achieved among different

650 mollusk species even without a specific foregoing experimental phase. Results of our findings can be  
651 used to answer a broad spectrum of different questions in the framework of shell midden research and  
652 shell-tempered pottery studies. Further studies on calcite mollusk shells are needed to complement the  
653 existing data on aragonite shells. This will improve the understanding of the biomineralized tissue  
654 thermal response in the case of a more stable calcium carbonate phase.

655         Although the methodology developed offers a powerful toolbox to reconstruct shell thermal  
656 exposure, care must be paid when interpreting the results in the archaeological context. As discussed, the  
657 preservation conditions of the shells (i.e., entire shells/small fragments; homogenous/heterogeneous  
658 burning features) can offer insights into the type of fire event. However, these parameters cannot resolve  
659 with certainty the exact nature of the fire (intentional/natural; cooking-related/disposal-related).  
660 Furthermore, it is important to mention that, given the irreversible nature of the structural and  
661 mineralogical thermal response, multiple burning events cannot be discerned. As for any study on burnt  
662 remains, the signature preserved is related to the highest temperature experienced by the shell material.

663

664

665

## 666 5. Conclusions

667         The experimental component of this study has confirmed that shellfish processing methods  
668 encompassing temperatures greater than or equal to 300 °C do significantly alter mollusk shell material,  
669 but that the duration of the heating does not appear to significantly affect the behaviour of the shell to  
670 thermal treatment. According to these findings, structural and chemical changes occur as soon as the  
671 shell is exposed to high temperatures. The results also confirm that using burnt shells for  
672 palaeoenvironmental reconstructions can produce incorrect temperature estimations and affect the

673 interpretation of environmental and human-related conditions. One proviso to these findings is that the  
674 experiments were conducted on fresh shell from which the flesh had been removed when the live  
675 mollusks were collected, and further experimental work is needed to establish what impact the presence  
676 of body fluids might have on the results. However, burned shells remain a valuable source of  
677 information. The experimental changes observed in the modern shells at macro-, microstructural and  
678 mineralogical scales have allowed us to determine a set of criteria to detect heating temperatures that can  
679 easily be applied to archaeological shells. Analysis of archaeological shells of *P. turbinatus* from the  
680 Haa Fteah cave in coastal northeast Libya revealed that during the Mesolithic the shells were uniformly  
681 exposed to high temperatures (shells in close proximity to the heat source) whereas during the Neolithic  
682 lower temperatures were recorded. Although, the use of fire for non-dietary purposes cannot be  
683 completely ruled out, the trend in our results can be interpreted as a temporal change in shellfish  
684 preparation.

685         Analysis of archaeological shells of *A. uropigimelana* and *T. palustris* from Neolithic and Bronze  
686 Age shell midden sites on the coast of the United Arab Emirates found that some specimens did not  
687 show any evidence of burning while others showed indications of high-temperature exposure. Although  
688 burning affected the shell structure and mineralogy, it does not appear to have a significant influence on  
689 radiocarbon ages suggesting the possibility of using burnt remains for dating purposes. The fact that the  
690 thermal reaction of different species is similar provides confidence in applying the techniques outlined  
691 in this study at an interspecific level.

692

693

694

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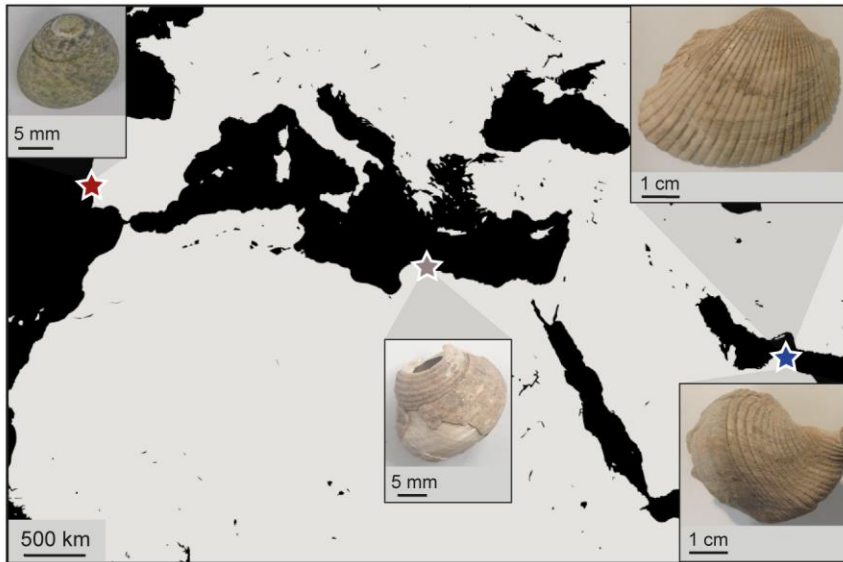
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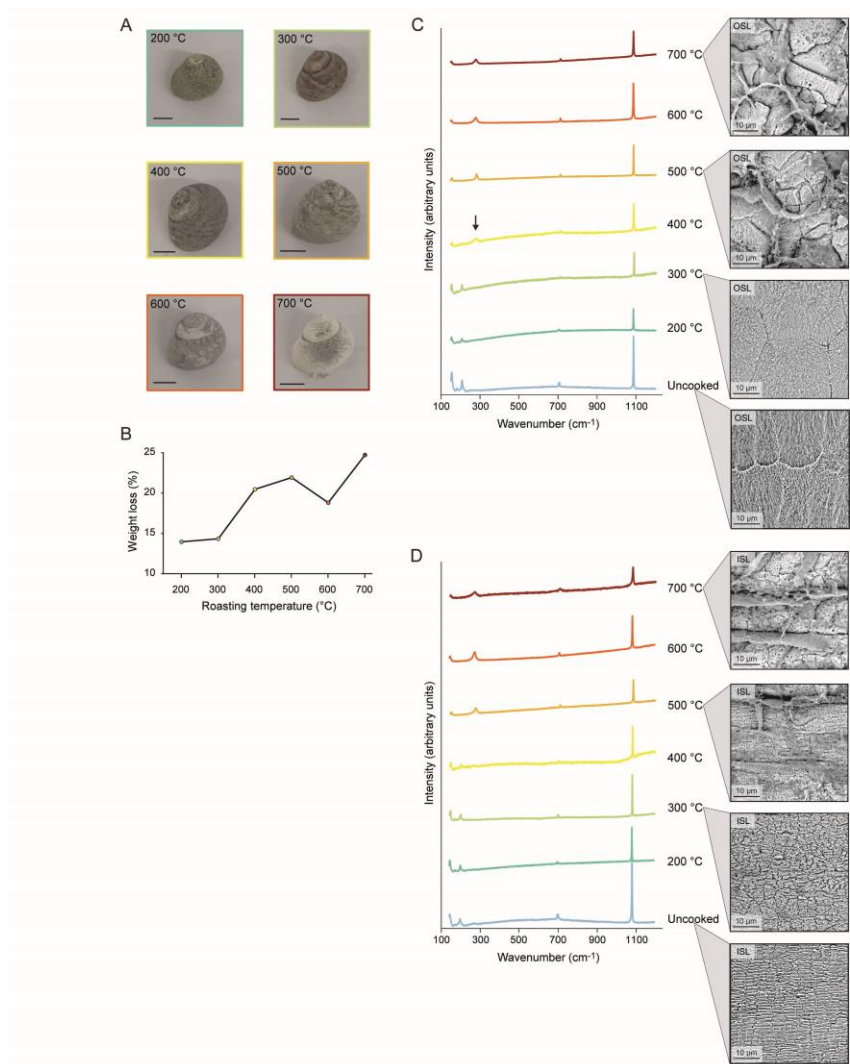
1025 Figure and tables

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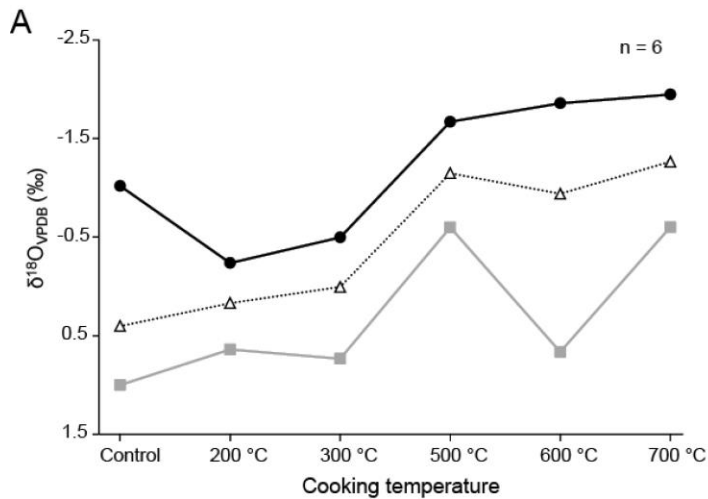
1027 **Fig. 1.** Map showing the localities encompassed in the study. Red star: Praia do Tamariz, Portugal,  
1028 where the modern *Phorcus lineatus* were collected. Grey star: Haua Fteah cave, Libya, where the  
1029 archaeological *Phorcus turbinatus* were excavated. Blue star: sites in the UEA from which the  
1030 archaeological remains of *Anadara uropigimelana* and *Terebralia palustris* were excavated.





1031

1032 **Fig. 2.** Effects of the 5-minute heating experiment on *P. lineatus* shell. (A) Overall shell appearance  
 1033 after the thermal exposure at different temperatures. Scale bars = 5 mm. (B) Shell weight loss in  
 1034 response to heat. (C) Effect of heat on the outer shell layer (OSL) mineralogy and microstructures. The  
 1035 black arrow shows the extra peak in the 400 °C Raman spectrum indicating the transition phase from  
 1036 aragonite into calcite. (D) Thermal response of the inner shell layer (ISL) mineralogy and  
 1037 microstructures.

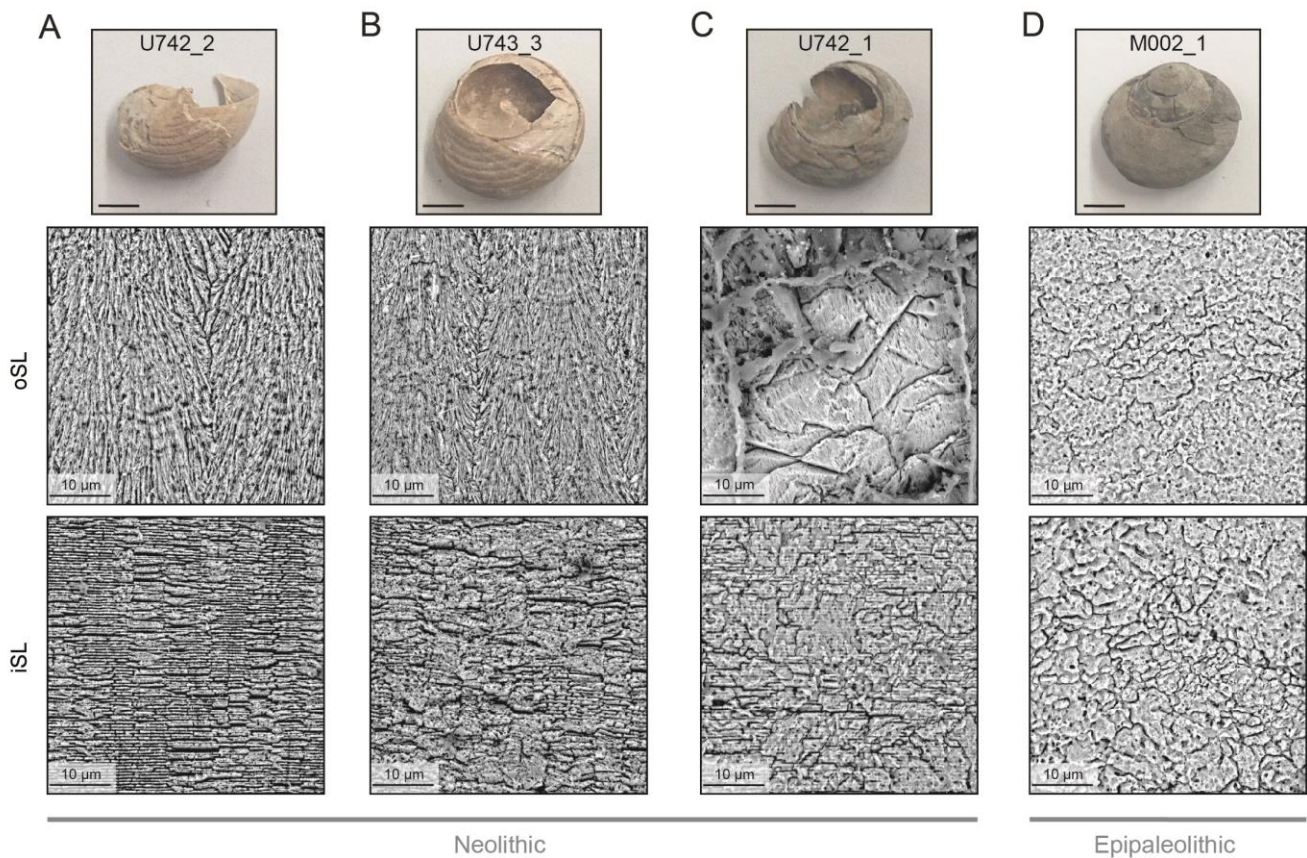


**B**

Cooking temperature	$\delta^{18}\text{O}_{\text{shell}}$ (p value)		
	5 min	20 min	60 min
100 °C	NA	Green	
200 °C	Green	NA	NA
300 °C	Green	Yellow	Red
500 °C	Red		
600 °C	NA		NA
700 °C	Red		

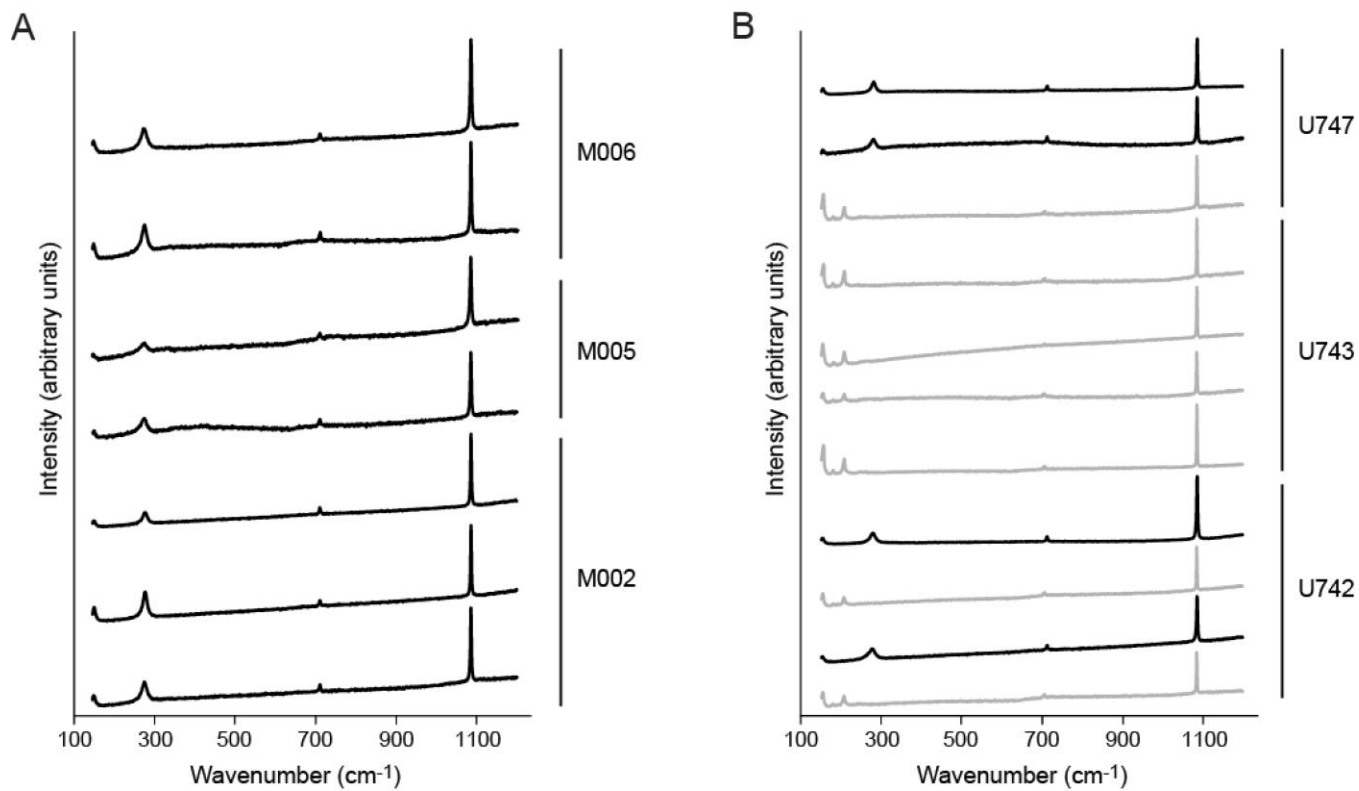
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1039 **Fig. 3.** Thermal behaviour of shell oxygen isotopes. (A) Oxygen isotope composition after roasting at  
 1040 the different temperatures. Gray squares=maximum  $\delta^{18}\text{O}_{\text{shell}}$ ; open triangles=average  $\delta^{18}\text{O}_{\text{shell}}$ ; black  
 1041 circles=minimum  $\delta^{18}\text{O}_{\text{shell}}$ . (B) The table refers to the different cooking durations tested in the present  
 1042 study and the previous study by Milano et al. (2016). Green cells indicate statistical similarity between  
 1043 isotope values of the heated and control shells ( $p > 0.05$ ). Red cells indicate statistical difference ( $p <$   
 1044  $0.05$ ). The yellow cell relates to a disagreement in the case of the specimen roasted at 300 °C for 20  
 1045 minutes. Its isotopic signature was statistically similar to the one of the control shell but different to the  
 1046 second control sample.



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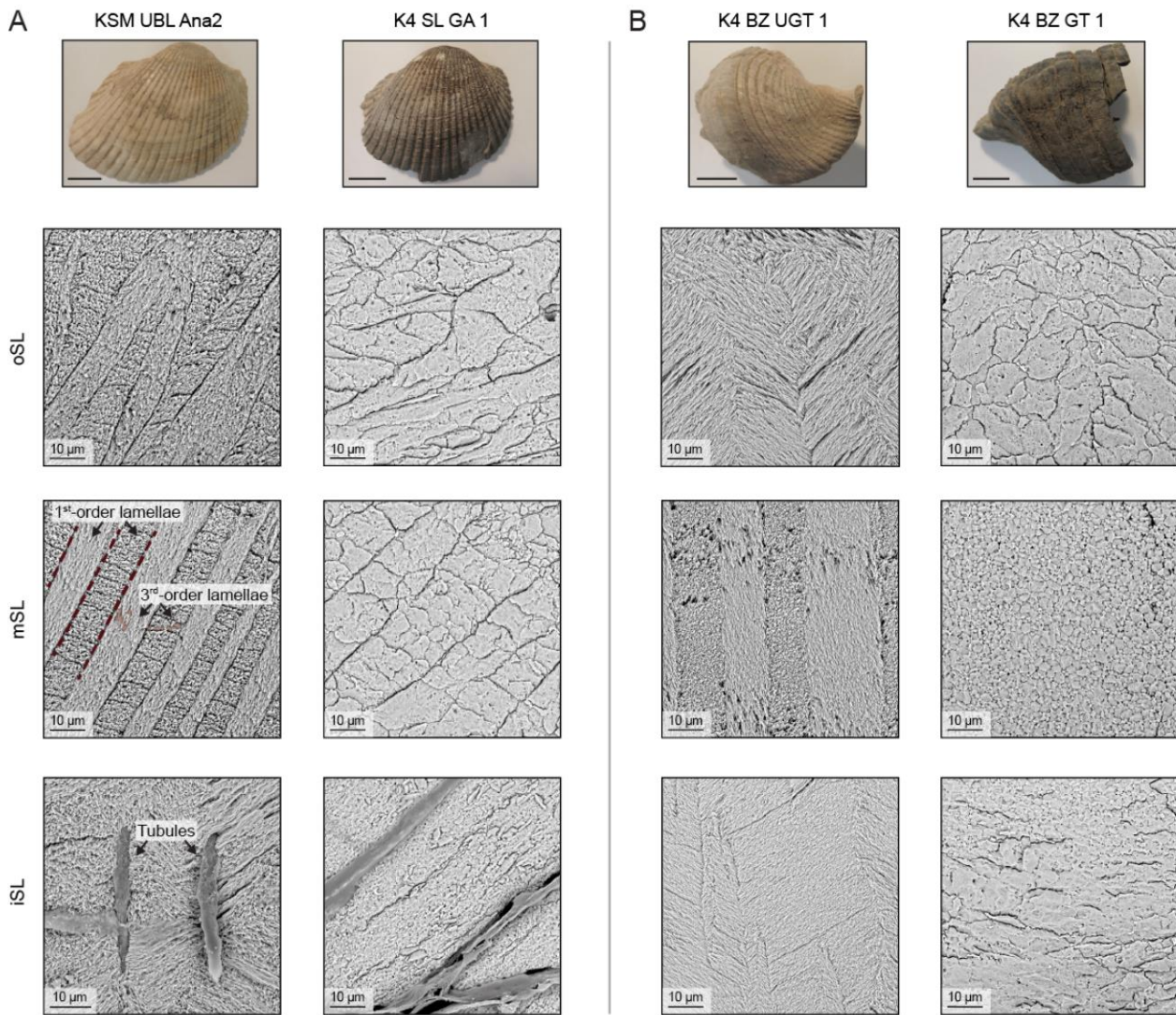
1048 **Fig. 4.** Macroscopic appearance and microstructural organization of *P. turbinatus* shells from (A-C)  
 1049 Neolithic contexts and (D) Mesolithic contexts in the Haua Fteah cave. The first row of SEM images  
 1050 refers to the prismatic microstructures of the OSL. The second row displays the nacre platelets of the  
 1051 ISL. The visible changes in structural organization denote possible exposure to heating processes. Scale  
 1052 bars if not otherwise indicated = 5 mm.



1053

1054 **Fig. 5.** Raman spectra of *P. turbinatus* ISLs from (A) Neolithic contexts and (B) Mesolithic contexts in  
 1055 the Haua Fteah cave. All Mesolithic specimens and four Neolithic specimens were calcitic (black lines);  
 1056 the rest of the Neolithic specimens were aragonitic (grey lines).



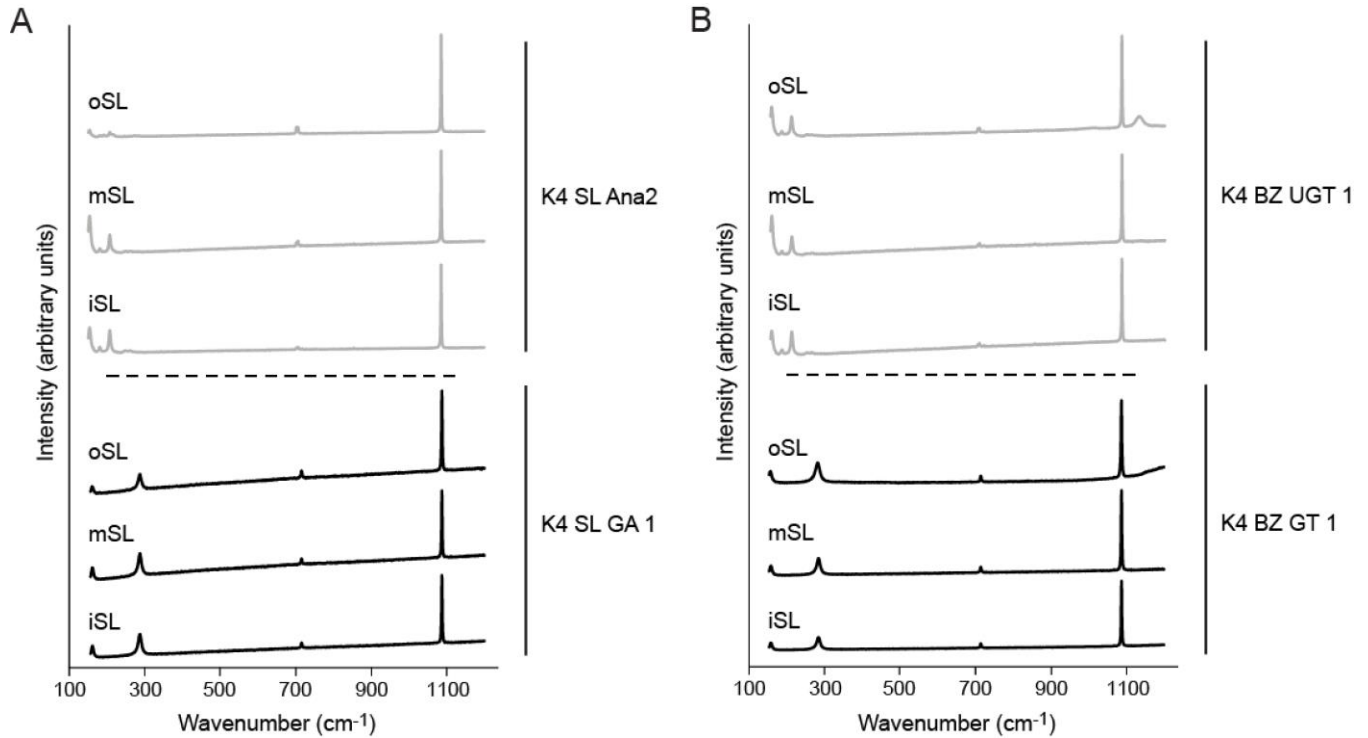


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1058 **Fig. 6.** Macroscopic appearance and microstructural organization of the archaeological shells from the  
 1059 United Arab Emirates. (A) Whole shell and SEM images of two *A. uropigimelana* specimens. Shell  
 1060 KSM UBL Ana2 showed the typical microstructural architecture of the species. Highlights on the SEM  
 1061 images indicate the first and third order units of the crossed-lamellar structures as well as the organic  
 1062 microtubules perforating the shell material. Specimen K4 SL GA 1 shows the alteration of the overall  
 1063 color and microstructures, possible related to heat treatment. (B) Whole shell and SEM images of two *T.*  
 1064 *palustris* specimens with regular (K4 BZ UGT 1) and altered microstructures and appearance (K4 BZ  
 1065 UGT 1). Scale bars if not otherwise indicated = 1 cm.

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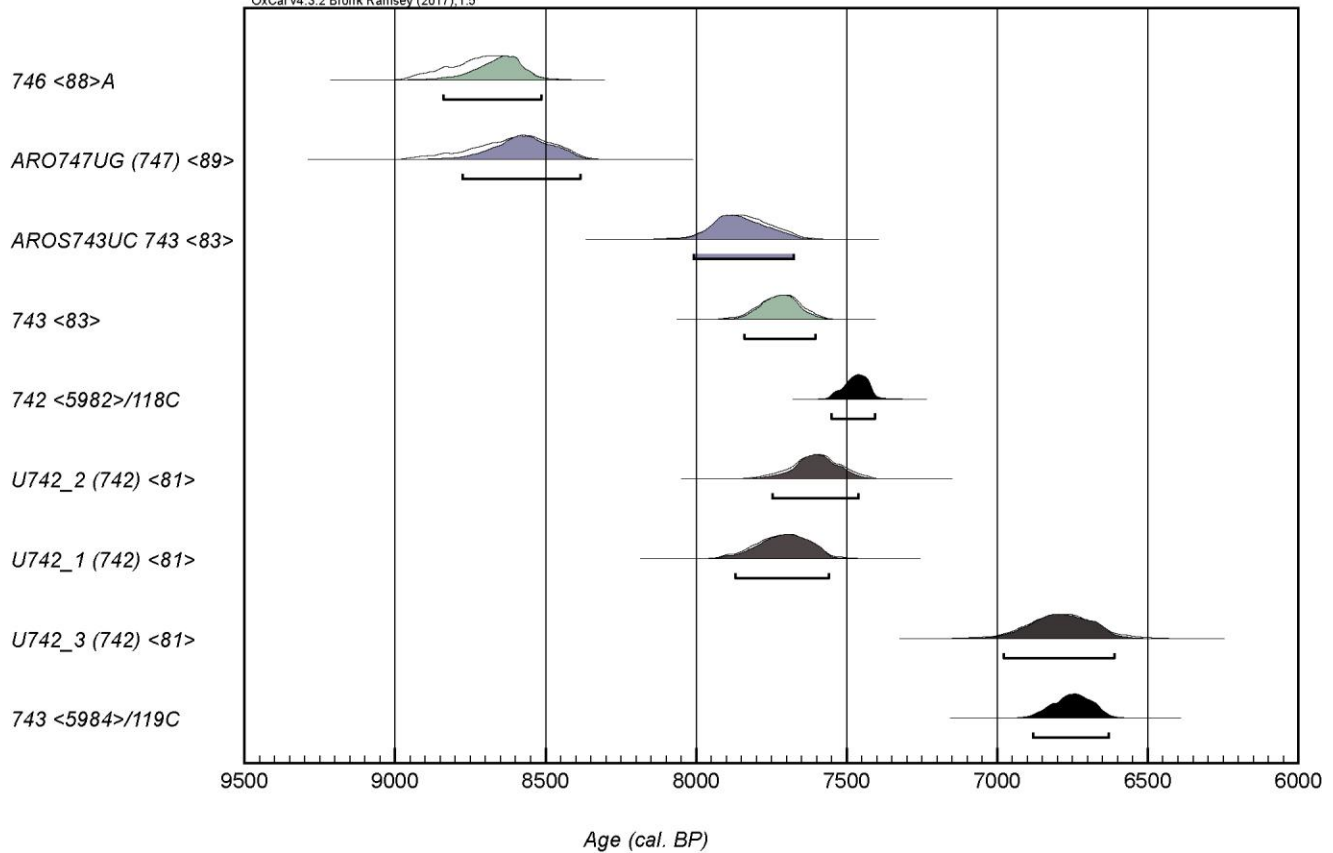
1067 **Fig. 7.** Raman spectra of (A) *A. uropigimelana* and (B) *T. palustris* from the United Arab Emirates  
1068 archaeological sites. Specimens K4 SL Ana2 and K4 BZ UGT 1 preserve their aragonitic OSL, MSL  
1069 and ISL (grey lines). Specimens K4 SL GA 1 and K4 BZ GT 1 display calcite in all shell layers (black  
1070 lines).



1071

1072 **Fig. 8.** Distribution of calibrated <sup>14</sup>C dates of Haua Fteah Trench U. Color-coding corresponds to the  
1073 different materials used in the dating. Green = *H. Melanostoma* shells; blue = well-preserved *P.*  
1074 *turbinatus*; grey = *P. turbinatus* from this study; black = charcoal.

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1077

1078 **Table 1.** List of studied specimens and details on their provenance.

ID	Species	Provenance	Period
A001	<i>P. lineatus</i>	Portugal	Modern
A001_M	<i>P. lineatus</i>	Portugal	Modern
A001_MR	<i>P. lineatus</i>	Portugal	Modern
A200	<i>P. lineatus</i>	Portugal	Modern
A200_M	<i>P. lineatus</i>	Portugal	Modern
A300	<i>P. lineatus</i>	Portugal	Modern
A300_M	<i>P. lineatus</i>	Portugal	Modern
A400	<i>P. lineatus</i>	Portugal	Modern
A400_M	<i>P. lineatus</i>	Portugal	Modern
A500	<i>P. lineatus</i>	Portugal	Modern
A500_M	<i>P. lineatus</i>	Portugal	Modern
A600	<i>P. lineatus</i>	Portugal	Modern
A600_M	<i>P. lineatus</i>	Portugal	Modern
A700	<i>P. lineatus</i>	Portugal	Modern
A700_M	<i>P. lineatus</i>	Portugal	Modern
M006_1	<i>P. turbinatus</i>	Haua Fteah, Libya	Capsian

M006_2	<i>P. turbinatus</i>	Haua Fteah, Libya	Capsian
M005_1	<i>P. turbinatus</i>	Haua Fteah, Libya	Capsian
M005_2	<i>P. turbinatus</i>	Haua Fteah, Libya	Capsian
M002_1	<i>P. turbinatus</i>	Haua Fteah, Libya	Capsian
M002_2	<i>P. turbinatus</i>	Haua Fteah, Libya	Capsian
M002_3	<i>P. turbinatus</i>	Haua Fteah, Libya	Capsian
U747_1	<i>P. turbinatus</i>	Haua Fteah, Libya	Neolithic
U747_2	<i>P. turbinatus</i>	Haua Fteah, Libya	Neolithic
U747_3	<i>P. turbinatus</i>	Haua Fteah, Libya	Neolithic
U743_1	<i>P. turbinatus</i>	Haua Fteah, Libya	Neolithic
U743_2	<i>P. turbinatus</i>	Haua Fteah, Libya	Neolithic
U743_3	<i>P. turbinatus</i>	Haua Fteah, Libya	Neolithic
U743_4	<i>P. turbinatus</i>	Haua Fteah, Libya	Neolithic
U742_1	<i>P. turbinatus</i>	Haua Fteah, Libya	Neolithic
U742_2	<i>P. turbinatus</i>	Haua Fteah, Libya	Neolithic
U742_3	<i>P. turbinatus</i>	Haua Fteah, Libya	Neolithic
U742_4	<i>P. turbinatus</i>	Haua Fteah, Libya	Neolithic
K4 SL Ana3	A. <i>uropigimelana</i>	Oman Sea	Bronze Age
K4 SL GA1	A. <i>uropigimelana</i>	Oman Sea	Bronze Age
K4 BZ GT1	<i>T. palustris</i>	Oman Sea	Bronze Age
K4 BZ UGT1	<i>T. palustris</i>	Oman Sea	Bronze Age
KSM UBL Ana2	A. <i>uropigimelana</i>	Oman Sea	Neolithic
KS Ana 1	A. <i>uropigimelana</i>	Oman Sea	Neolithic
KK1 Ana3	A. <i>uropigimelana</i>	Oman Sea	Neolithic



KK1 Ana4

A.  
*uropigimelana*

Oman Sea

Neolithic

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**Table 2.** Reconstruction of shell exposure temperature by using shell overall appearance, microstructures and mineralogy as proxies in archeological specimens of *P. turbinatus*.

Shell ID	Reconstructed cooking temperature		
	Macroscale appearance	Microstructures	Mineralogy
M006_1	600 °C	600 °C	≥ 500 °C
M006_2	600 °C	600 °C	≥ 500 °C
M005_1	600 °C	600 °C	≥ 500 °C
M005_2	600 °C	600 °C	≥ 500 °C
M002_1	600 °C	700 °C	≥ 500 °C
M002_2	600 °C	600 °C	≥ 500 °C
M002_3	600 °C	700 °C	≥ 500 °C
U747_1	500 °C	600 °C	≥ 500 °C
U747_2	500 °C	600 °C	≥ 500 °C
U747_3	500 °C	300 °C	< 500 °C
U743_1	300 °C	300 °C	< 500 °C
U743_2	300 °C	300 °C	< 500 °C
U743_3	300 °C	300 °C	< 500 °C
U743_4	300 °C	300 °C	< 500 °C
U742_1	400 °C	500 °C	≥ 500 °C
U742_2	Not cooked	Not cooked	< 500 °C
U742_3	500 °C	600 °C	≥ 500 °C
U742_4	500 °C	300 °C	< 500 °C

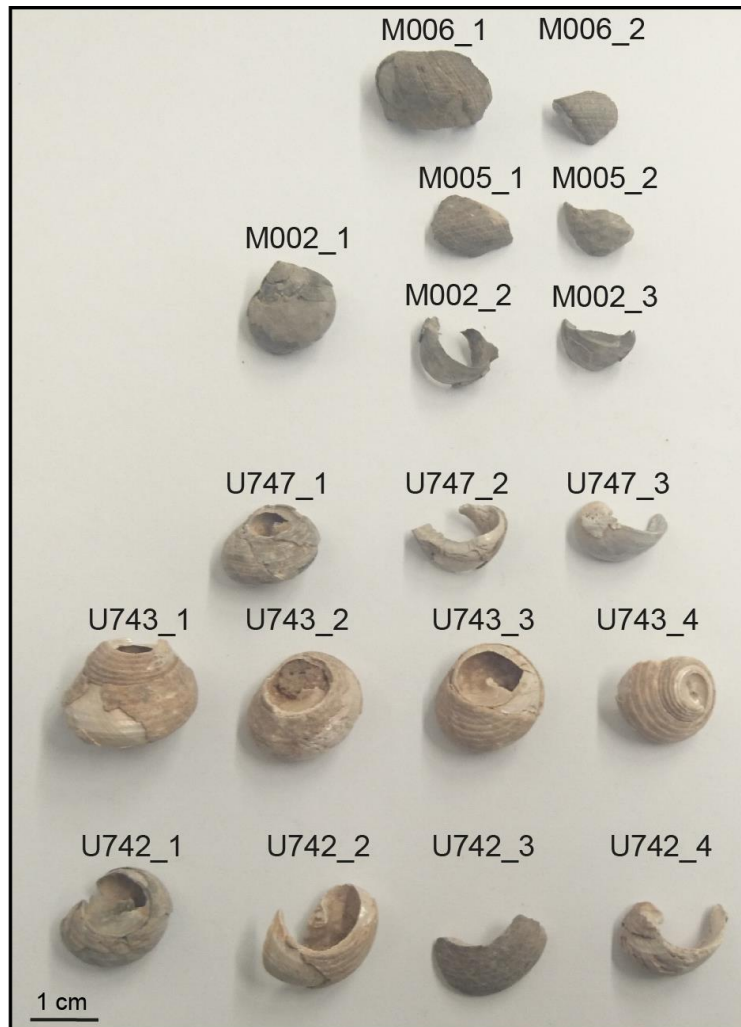
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1086 **Table 3.** Trench U (Haua Fteah) <sup>14</sup>C dates. Symbol <sup>\*1</sup> indicates dates corrected using reservoir offset of 58 ± 85  
 1087 <sup>14</sup>C years (Reimer and McCormac, 2002). Symbol <sup>\*2</sup> indicates dates corrected using reservoir offset of 476 ± 48  
 1088 <sup>14</sup>C years (Hill et al., 2017).  
 1089

Sample ID	Lab Code	Material	Age ( <sup>14</sup> C yr)	±	Age (cal. BP) (unmodelled)	Age (cal. BP) (modelled)
U742_1	MAMS 30538 U742-1	<i>P. Turbinatus</i>	7300	21	7901 - 7556 <sup>*1</sup>	7869 - 7560 <sup>*1</sup>
U742_2	MAMS 30538 U742-2	<i>P. Turbinatus</i>	7184	21	7772 - 7438 <sup>*1</sup>	7747 - 7461 <sup>*1</sup>
U742_3	MAMS 30538 U742-3	<i>P. Turbinatus</i>	6378	21	7002 - 6558 <sup>*1</sup>	6978 - 6610 <sup>*1</sup>
ARO747UG	OxA-27391	<i>P. Turbinatus</i>	8171	37	8891 - 8381 <sup>*1</sup>	8776 - 8385 <sup>*1</sup>
ARO747UC	OxA-27490	<i>P. Turbinatus</i>	7447	35	8024 - 7652 <sup>*1</sup>	8008 - 7675 <sup>*1</sup>
742 <5982>/118C	UBA-28873	Charcoal	6965	39	7551 - 7405	7551 - 7405
743<5984>/119C	UBA-28874	Charcoal	6291	50	6883 - 6627	6880 - 6630
746 <88> A	UBA-25915	<i>H. Melanostoma</i>	8670	39	8930 - 8534 <sup>*2</sup>	8840 - 8531 <sup>*2</sup>
743 <83>	UBA-25912	<i>H. Melanostoma</i>	7732	37	7850 - 7590 <sup>*2</sup>	7840 - 7605 <sup>*2</sup>

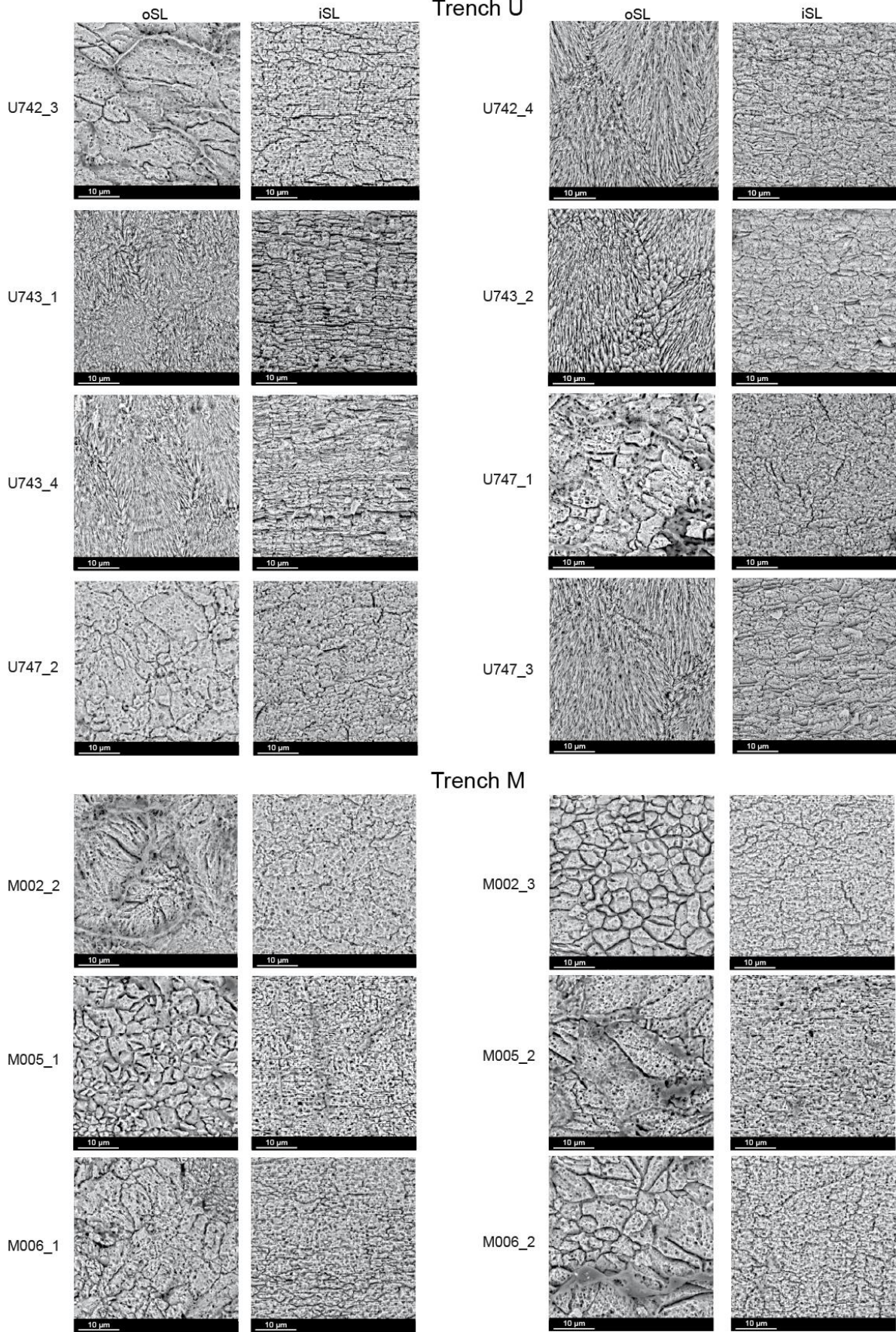
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1091 **Supplementary material**



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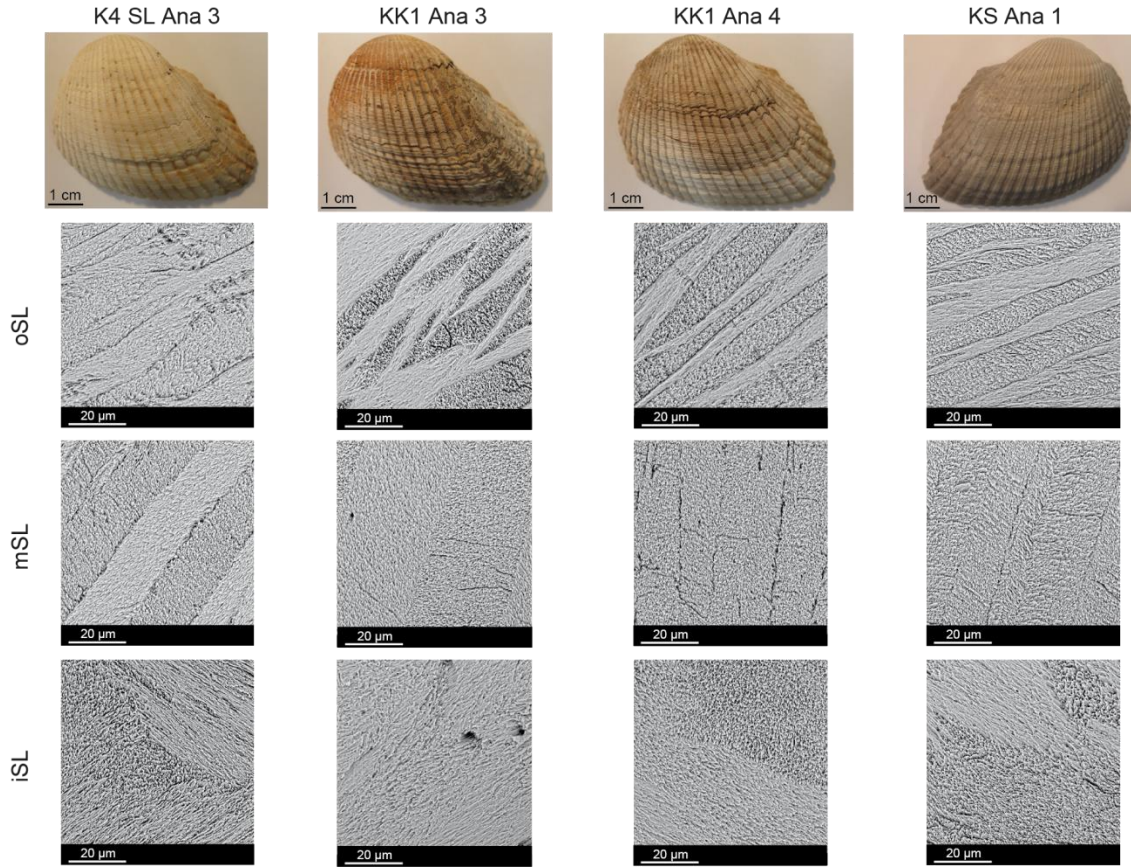
**Fig. S1.** Overall appearance of all *P. turbinatus* excavated from Haua Fteah cave and used in the present study. The upper seven specimens were excavated from Trench M (Mesolithic). The rest of the specimens were excavated from Trench U (Neolithic).





1099

1100 **Fig. S2.** SEM images of the microstructural organization of *P. turbinatus* from Haua Fteah. Trench U = Neolithic  
1101 context. Trench M = Mesolithic context.  
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1104

1105 **Fig. S3.** Overall appearance and microstructural organization of *A. uropigimelana* from Southeast Arabia.

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