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1 **Bactericidal and virucidal mechanisms in the alkaline disinfection of compost using**
2 **calcium lime and ash**

3

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17

18 **Abstract**

19 In the present study, the bactericidal and virucidal mechanisms in the alkaline
20 disinfection of compost with calcium lime and ash were investigated. Two indicator
21 microorganisms, *Escherichia coli* and MS2 coliphage, were used as surrogates for enteric
22 pathogens. The alkaline-treated compost with calcium oxide (CaO) or ash resulted primarily
23 in damage to the outer membrane and enzyme activities of *E. coli*. The alkaline treatment of
24 compost also led to the infectivity loss of the coliphage because of the partial capsid damage
25 and RNA exteriorization due to a raised pH, which is proportional to the amount of alkaline
26 agents added. These results indicate that the alkaline treatment of compost using calcium
27 oxide and ash is effective and can contribute to the safe usage of compost from a mixing type
28 dry toilet.

29

30 **Keywords:**

31 alkaline disinfection, ash, calcium oxide, compost, dry toilet, pathogen

32

33 1. INTRODUCTION

34 A mixing type composting toilet using an organic matrix is one of the established
35 concepts of dry toilet. The mixture of human feces with organic matrix enhances aerobic
36 degradation of feces with little odor, decreases the fecal volume by water evaporation, and
37 contributes to internal heating generated by the activity of microorganisms. The dry toilet has
38 been recognized as an improved sanitation facility (WHO and UNICEF, 2008) and has
39 advantages for reducing the amount of water usage. It can be installed quickly and it saves
40 public investment, since it does not require large-scale infrastructure such as a water
41 distribution network and sewer piping system (Esrey, 1998; Lopez et al., 2002a). Furthermore,
42 compost produced in a dry toilet promotes plant growth by providing nutrients and
43 conditioning soil properties (Hijikata et al., 2011). These characteristics related to
44 installability, usefulness, and nutrient usability are attractive for low-income countries, and
45 rural areas, natural parks and emergency evacuation sites in developed countries, which often
46 have a lack of improved sanitation facilities, incomplete water and sewer services, a
47 restriction on wastewater discharge, and high demand for fertilizers.

48 From a hygienic perspective, feces should always be considered to contain pathogens
49 that cause gastrointestinal infections. Although the composting process in dry toilets reduces
50 concentrations of indicator microorganisms and pathogens to some extent (Sossou et al.,
51 2011; Sossou et al., 2014), the compost recovered from a dry toilet always has the potential to

52 include pathogens derived from feces of infected persons, which pose infectious disease risks
53 for users (Otaki et al., 2007). The highest infection risk most likely is posed during emptying
54 the compost and exchanging the matrix (Nakagawa et al., 2006; Schönning et al., 2007).
55 Potential hazard risks when the compost is removed from a composting toilet are caused by
56 direct and/or indirect ingestion of enteric pathogens originating from human feces, which
57 include bacteria, viruses, parasitic protozoa and helminthes (WHO, 2006). These pathogens
58 may cause diarrhea, fever and cramps. According to the global burden of diseases study,
59 diarrhea accounts for 89.5 million disability-adjusted life years (DALYs) and is still a large
60 contributor to the burden, accounting for 3.6% of the global DALYs in 2010 (Murray et al.,
61 2012). It is thus very critical to manage the infection risks of enteric pathogens associated
62 with the use of composting toilets.

63 WHO (2006) and Schönning et al. (2007) have recommended storing mature fecal
64 matter for 6-12 months to assure safety during handling. The recommendation mainly targets
65 at pit-type dry toilets such as the urine-diversion dry toilet (UDDT), which has a double
66 storage system. It may be thus difficult to apply this recommendation to the mixing type
67 composting toilets as it is, because of their structure comprised of a single reactor. Kazama et
68 al. (2011) showed that alkaline-treated compost with calcium lime was effective for the rapid
69 inactivation of bacteria and virus indicators in sawdust used in a composting toilet. Lime and
70 ash are often used in practical and scientific reports about the alkaline treatment of the UDDT,

71 and the potential for pathogen inactivation has been described (Nordin et al., 2009a;
72 Niwagaba et al., 2009). However, bactericidal and virucidal mechanisms in the alkaline
73 disinfection of compost from a mixing type composting toilet have not been well investigated.

74 In the present study, the inactivation mechanisms of surrogate microorganisms
75 (*Escherichia coli* and MS2 coliphage) in compost treated with varied quantities of alkaline
76 agents was investigated. Three types of growth medium were used for cultivating *E. coli* from
77 the alkaline-treated compost, which allows us to identify the most affected physiological
78 function of the surrogate for the bacterial pathogen. The plaque assay and the enzymatic
79 treatment using Protease K and RNase A coupled with reverse-transcription quantitative PCR
80 (ET-RTqPCR) were employed to identify the effect of the alkaline treatment on the infectivity
81 of MS2 coliphage. Based on the results from these disinfection tests, the efficacy of alkaline
82 disinfection of compost was discussed.

83

84

85 **2. Materials and Methods**

86

87 **2.1. Compost preparation**

88 Rice husks were used as a compost matrix and pig feces were used as a substitute for
89 human feces since the characteristics of pig fecal matter are similar to those of human feces

90 (Lopez et al., 2002b). Fresh pig feces (500-600 grams) were mixed into 20 L (1429 g-dry,
91 approximately a half volume of a general composting toilet reactor for one family) of the
92 matrix in a composting machine (Hitachi, BGD-120). Fresh pig feces (500-600 grams) were
93 put into the matrix in the composting machine everyday up to 38th day. At the end of the
94 operation, the fecal load ratio (total input of feces [g-dry] per initial matrix [g-dry]) was 1.71,
95 the fecal degradation (loss of the mass) ratio was 45.8%, and the water content was 48.5%,
96 which is comparable with compost from a mixing type dry toilet. The obtained compost was
97 stored in refrigerated room at 4 °C. The preparation process of the compost was depicted in
98 Fig. 1S.

99

100 2.2. Test microorganisms

101 *E. coli* NBRC 3301 and MS2 coliphage were used as a surrogate for pathogenic
102 bacteria and viruses, respectively. *E. coli* NBRC 3301 was incubated with 10 mL of 4% (w/w)
103 of Tryptic Soy Broth (Difco Laboratory Inc., USA) in a shaking water bath at 37 °C for 4-6 h,
104 and then the cultured *E. coli* was used as an inoculum for inactivation tests. *E. coli* NBRC
105 13965 was used as a host for MS2 coliphage. MS2 coliphage was co-incubated with
106 precultured *E. coli* NBRC 139651 in 10 mL of liquid LB medium in the shaking water bath at
107 37 °C for 20–24 h. The culture was centrifuged at 6000 rpm for 10 min at 4 °C and the
108 supernatant was filtered with a sterilized disposal filter (0.20 µm pore, DISMIC-25CS,

109 ADVANTEC). The filtered liquid was used as an MS2 coliphage inoculum for inactivation
110 tests.

111

112 2.3. Inactivation test of indicator microorganisms in the composting toilet

113 Calcium lime (CaO, Wako chemical, Japan) was used as the calcium oxide reagent,
114 according to the previous study (Kazama et al., 2011). Three types of ash, including wood ash,
115 grass ash made from rice straws, and mixed ash (a mixture of wood grasses, crushed oyster
116 shell and wood ash obtained from an Italian restaurant in Sapporo City, Japan) were also used
117 as alkaline agents. The volatile solid [% , w/w] of these alkaline agents was obtained from a
118 loss of weight by the incineration at 600 °C for 3 h. The chemical compositions of ash are
119 indicated in Table 1.

120 Stored compost (20 g) was transferred to a glass bottle and autoclaved at 121 °C for
121 15 min, and then the water content was adjusted to 50% using sterilized deionized water to
122 uniform the water content condition in the inactivation tests. After preincubation of the water
123 content-adjusted compost at 37 °C for 1 h, an adequate amount of CaO or ash and 1 g of
124 autoclaved pig feces were simultaneously added and mixed in the bottle, which can create a
125 sterile but very similar condition with actual compost from a mixing type dry toilet. The pH
126 value of compost mixture was measured by an electrode method using a suspension of
127 compost mixture and pure water at a ratio of 1:20 (w:w), which was shaken for 30 min. The

128 pH of compost mixture was adjusted to pH 9.5, 10.0, 10.5 and 11 by adding the alkaline
129 agents. *E. coli* inoculum (0.5 mL, approximately 10^8 CFU/mL) was spiked into the compost
130 mixture and shaken by hand for 1 min, which gave the experimentally highest *E. coli*
131 concentration, enabling us to follow the inactivation profile very efficiently. The compost
132 mixture was incubated at 37 °C, the optimum temperature for the *E. coli* growth, which may
133 attenuate the disinfection efficiency of alkaline treatments. One gram of the compost mixture
134 was sampled in an appropriate time (0-8 h) and *E. coli* was extracted by suspending it in 40
135 mL of 3% (w/v) beef extraction solution adjusted to pH 9.5 with an NaOH solution (Otaki et
136 al., 2002). The previous study reported that the total recovery efficiency using this method is
137 70-100% (Otaki et al., 2002). After dilution (10^1 - 10^4 folds) with an autoclaved phosphate
138 buffer (pH 7.5 adjusted with the NaOH solution), 1 mL of each extract was separately
139 inoculated on three types of agar media: Tryptic Soy Agar (TSA, Difco Laboratory Inc., USA),
140 Desoxycholate Agar (DESO, Eiken Kagaku Inc., Japan), and Compact Dry EC (C-EC,
141 Nissui Inc., Japan). TSA is a non-selective agar that contains two peptones (casein and soy) as
142 nutrition for microorganisms. The agar can isolate and cultivate a wide variety of organisms
143 that metabolize the peptones and grow. DESO is a selective agar that inhibits to grow
144 gram-positive bacteria. The agar contains general growth requirements with sodium
145 desoxycholate, sodium citrates, lactose, and neutral red. Sodium desoxycholate, a surfactant,
146 lyses gram-positive bacteria because they do not have outer membrane unlike gram-negative

147 bacteria. Bacteria that ferment lactose produce acid and form red colonies in the presence of
148 neutral red. Bacteria that do not ferment lactose form colorless colonies. C-EC, a selective
149 agar for *E.coli* and coliform, includes general growth requirements with two chromogenic
150 substrates: X-GLUC and Magenta-GAL. The chromogenic substrates are hydrolyzed with
151 enzymes: β -glucuronidase and β -galactosidase that *E.coli* produces during fermenting lactose
152 and make colonies blue or blue purple color. These agar media were incubated at 37°C for 24
153 h and the colonies formed were counted.

154 The alkaline-treated compost identical with that used in the *E .coli* inactivation test
155 was also prepared for the inactivation test of MS2 coliphage. Five milliliters of MS2
156 coliphage inoculum were spiked into 20 g of the compost and incubated at 37°C. One gram of
157 the incubated compost mixture was sampled in an appropriate time (0-8 h) and suspended in
158 40 mL of the autoclaved phosphate buffer for 3 min. The suspension was diluted 10^1 - 10^4 fold
159 with autoclaved liquid LB culture. The diluted liquid was used for the plaque assay method
160 with a two-layer LB agar media (Adams 1985) and for enzyme treatment qPCR (ET-qPCR)
161 detection with proteinase K and RNase A (Pecson et al., 2009), respectively. In the plaque
162 assay, 25 ml of the diluted liquid were inoculated on the bottom layer agar. A top layer
163 medium (10-20 mL) including host *E. coli* cells (approximately 10^8 CFU per an agar plate)
164 was pulled on the bottom agar. The two-layer agar was incubated at 37 °C for 20-24 h and the
165 plaques formed were counted. In ET-qPCR detection, the 10-fold diluted liquid was applied to

166 the enzyme pretreatments, in which only RNase A (ET1-qPCR), a combination of proteinase
167 K with RNase A (ET2-qPCR), or no enzyme treatment (qPCR) was employed. These enzyme
168 treatments were performed according to Nuanualsuwan and Cliver (2003) with a slight
169 modification. For the ET2-qPCR reaction, 33 μL of proteinase K (20 U/min) and 7 μL of
170 Tris-EDTA buffer (TE buffer) were mixed into 250 μL of the diluted liquid. The mixture was
171 reacted in a shaking incubator at 37°C for 1 h. Then, 10 μL of 10 ng/ μL RNase A were added
172 to the mixture and incubated at 37 °C for 1 h. For the ET1-qPCR, 40 μL of the TE buffer were
173 mixed in the diluted liquid as a substitute for the proteinase K solution, and then the same
174 operation with ET2-qPCR was conducted. For the qPCR, 40 μL of TE buffer and 10 μL of TE
175 buffer were used as a substitute for the proteinase K solution and RNase A solution,
176 respectively. After the enzyme pretreatment, viral RNA was extracted with QIAamp Viral
177 RNA mini Kit (QIAGEN, Japan) according to the manufacturer's protocol. The extracted
178 RNA was applied to a reverse transcription reaction (RT) with PrimeScript RT reagent Kit
179 (Takara Bio, Japan) to synthesize complementary DNA (cDNA). For the quantitative PCR, a
180 25 μL reaction mixture, containing 5 μL cDNA, 12.5 μL Premix Ex Taq, 1 μL of forward and
181 reverse primer, 0.5 μL Taqman probe, 0.5 μL Rox Reference Dye and 4.5 μL sterilized pure
182 water, was prepared. The PCR thermal cycling conditions were 15 min at 55 °C for 15 min
183 and 95 °C for 15 sec at the beginning; 95 °C for 15 min and 60 °C for 1 min repeated for 45
184 cycles; and 60 °C for 1 min at the end, using the ABI 7500 Real Time PCR system (Applied

185 Biosystemes, USA). The sequence of PCR primers (Blaise-Boisseau et al., 2010) and the
186 probe are listed in Table S1.

187

188 2.4. Inactivation rate constant

189 The inactivation rate constant was calculated using the following Chick model:

190

$$191 \quad \text{Log} (N / N_0) = - k t \quad (1),$$

192

193 where N is a concentration of the microorganism at a time t [number/g-compost mixture], N_0
194 is an initial concentration of the microorganism [number/g-compost mixture], k is an
195 inactivation rate constant expressed by a common logarithm [1/h] and t is the incubation time
196 in the compost mixture [h]. The microorganism concentration (N) corresponded to
197 CFU/g-compost mixture for the *E. coli*, PFU/g-compost mixture for the plaque assay of the
198 MS2 coliphage and cDNA copy number/g-compost mixture for the ETs-qPCR of the MS2
199 coliphage.

200

201 2.5. Statistical analysis

202 All statistical and regression tests were performed using IBM SPSS Statistics 21
203 software.

204

205

206 **3. Results**

207

208 3.1. Alkaline treatment of compost with CaO and ash

209 The pH of compost mixture was increased as the addition of CaO (Fig. 1A) and ash

210 (Fig. 1B). The regression lines for each alkaline agent were obtained as follows:

211

212
$$pH_{CaO} = 76.03 W_{CaO} + 8.89 (R^2 = 0.902) \quad (2),$$

213
$$pH_{mixed\ ash} = 0.32 W_{mixed\ ash} + 9.59 (R^2 = 0.855) \quad (3),$$

214
$$pH_{wood\ ash} = 0.80 W_{wood\ ash} + 9.80 (R^2 = 0.906) \quad (4),$$

215
$$pH_{grass\ ash} = 0.41 W_{grass\ ash} + 9.71 (R^2 = 0.852) \quad (5),$$

216

217 where pH_{agents} is the pH of compost mixture adjusted by each alkaline agent and W_{agents} is the

218 weight of the alkaline agent added [g/g-dry compost mixture]. The slope values of the

219 regression lines indicated that the pH increase with ash was more moderate than that with

220 CaO. Among three ash types, the alkalization capacity of wood ash was higher than those of

221 the other two ash types.

222

223 3.2. Alkaline-treated compost and inactivation rate of indicator microorganisms

224 Time-dependent changes of *E. coli* on the TSA media and MS2 coliphage on the
 225 plaque assay in the alkaline-treated compost are shown in Fig. 2. The actual pH of compost
 226 mixture was measured after the time-dependent test and the value variation was within 0.1.
 227 The inactivation rate values of *E. coli* (Fig. 2A) and MS2 coliphage (Fig. 2B) were influenced
 228 by the pH of compost mixture. The type of alkaline agent also affected the inactivation rate at
 229 the identical pH value. The mixed ash gave higher inactivation rate of *E. coli* than CaO at pH
 230 10 (Fig. 2A), while the addition of CaO led to the higher inactivation of MS2 coliphage than
 231 the mixed ash at the identical pH value (Fig. 2B). The relationship between the inactivation
 232 rate constant (k) and the pH of compost mixture was well fitted with a quadratic curve model
 233 in the present experimental conditions ($p < 0.05$) for both *E. coli* on TSA (Fig. 3A) and MS2
 234 coliphage on the plaque assay (Fig. 3B). The model gave the following equations:

235

$$236 \quad k_{E.coli} = 1.444 pH_{CaO}^2 - 26.384 pH_{CaO} + 120.76 \quad (R^2 = 0.961) \quad (6),$$

$$237 \quad k_{E.coli} = 1.673 pH_{mixed\ ash}^2 - 27.97 pH_{mixed\ ash} + 115.84 \quad (R^2 = 0.92) \quad (7),$$

$$238 \quad k_{MS2} = 0.655 pH_{CaO}^2 - 11.894 pH_{CaO} + 53.96 \quad (R^2 = 0.981) \quad (8),$$

$$239 \quad k_{MS2} = 0.195 pH_{mixed\ ash}^2 - 3.5743 pH_{mixed\ ash} + 16.41 \quad (R^2 = 0.862) \quad (9),$$

240

241 where $k_{E.coli}$ and k_{MS2} are the inactivation rate constant for *E. coli* [1/h] and MS2 coliphage

242 [1/h], respectively. These equations obviously show that the alkaline treatment of compost
243 with the mixed ash causes a faster inactivation of *E. coli* than with CaO at the identical pH of
244 compost mixture (Fig. 3A), but the opposite result (faster inactivation with CaO than mixed
245 ash) was obtained for MS2 coliphage (Fig. 3B).

246

247 3.3. Inactivation mechanism of indicator microorganisms

248 The inactivation rate constants of *E. coli* in three cultivation media were compared
249 (Fig. 4A). The value on TSA was lower than those on DESO and C-EC which was very clear
250 at pH 11. There was no significant difference between the values of inactivation rate constant
251 on DESO and C-EC at the compost pH of 10 to 11. These results suggests that the alkaline
252 treatment of compost with CaO led to the damage on the outer membrane and specific
253 enzymes before damaging the nucleic acid and/or metabolism of *E. coli*. Very similar tendency
254 of *E. coli* inactivation was observed in the case of alkaline-treated compost with ash, which
255 suggested that the similar inactivation mechanism was exerted on *E. coli* when ash were
256 mixed with compost.

257 In the case of MS2 coliphage (Fig. 4B), the inactivation rate measured with the
258 plaque assay was 10-13 times higher than that measured with ET-PCR in all CaO-treated
259 compost. In the compost mixture at pH 10 and 10.5, the inactivation rate measured with
260 ET2-PCR and ET1-PCR were 1.9-2.1 times higher than that with qPCR. These results suggest

261 that the alkaline treatment of compost led MS2 primarily to reduce the infectivity, and then
262 caused capsid damage and partial RNA exteriorization accompanied by a pH increase. At pH
263 11, on the other hand, there was no significant difference in the inactivation rate among
264 ET2-PCR, ET1-PCR and qPCR, which implies that the capsid damage and the degradation of
265 the exteriorized RNA progressed rapidly at this pH. The pattern of the inactivation rate
266 constants in CaO and mixed ash was the same in the compost mixture at pH 10, meaning that
267 there was no clear difference in the inactivation mechanism between alkaline agents at the
268 identical pH of compost mixture.

269

270

271 **4. Discussion**

272 In the present study, the inactivation rate constant values of *E. coli* and MS2
273 coliphage in compost from a mixing type dry toilet, which was treated by CaO or ash, were
274 obtained. The alkaline-treated compost mixture with CaO or ash resulted primarily in damage
275 to the outer membrane and enzyme activities of *E. coli*. The alkaline treatment of compost
276 also led to the infectivity loss of the coliphage because of the partial capsid damage and RNA
277 exteriorization due to a raised pH, which is proportional to the amount of alkaline agents
278 added.

279 Among the three types of ash used in this study, the wood ash efficiently raised the

280 compost pH (Fig. 2B). The increase of pH is considered to be dependent on the burning
281 temperature and chemical composition of the ash, mainly the content of carbonates and
282 hydroxides. According to Etiegni and Campbell (1991), carbonates and bicarbonates
283 predominate when the combustion temperature is below 500 °C whereas oxides become more
284 prevalent with temperatures over 1000 °C. Cations play the role of counter ion for these bases
285 and determine the dissociation constant. In the present study, the lowest volatile solid
286 concentration of wood ash (Table 1) possibly indicated that the ash had been burned at a high
287 temperature, which might imply a relatively higher content of oxides. The commercial mixed
288 ash used in this study contained crushed oyster shell to supply mineral divalent cations and
289 moderate the pH increase for agricultural purposes, according to the manufacturer's
290 instructions. The addition of crushed oyster shell must be the reason for the lower alkalinity of
291 the mixed ash. It is important to analyze chemical compositions of ash prior to the usage for
292 the alkaline treatment of compost.

293 Damaged components of *E. coli* in the alkalinized compost were estimated by a
294 comparison of inactivation rate constants (k) in the three media (Kazama et al., 2011; Sossou
295 et al., 2014). TSA is a nonselective agar that contains digested proteins and can detect bacteria
296 that can metabolize these proteins. The higher k value on the TSA indicates that there were
297 damages on the nucleic acid and/or enzymes involved in the protein metabolism. DESO is a
298 selective agar and can detect bacteria that can metabolize lactose and peptone in the presence

299 of desoxycholic acid. Gram-negative bacteria with an incomplete outer membrane and
300 gram-positive bacteria are unable to grow in the presence of desoxycholic acid due to its
301 surface-active effect. The higher k value in the DESO indicates that there were damages in the
302 outer membrane structure. C-EC is a selective agar for *E. coli* and coliform. The agar can
303 detect bacteria that can produce β -glucuronidase and β -galactosidase and metabolize the
304 carbon sources. The higher k value in the C-EC indicates that there were damages in specific
305 enzymatic activities. These results suggest that alkaline-treated compost with CaO and ash
306 result primarily in damages on the outer membrane and the enzyme activities of *E. coli* and
307 then on the nucleic acid and/or metabolism. This result seems to be inconsistent with results
308 reported by Kazama et al. (2011), who also observed the damage mechanism in
309 alkaline-treated sawdust compost with CaO by the same method. This may be related to the
310 type of matrix (Sossou et al., 2014) or the ammonium concentrations in the compost mixture.
311 Low pH sawdust compost (6.96-7.51), in which ammonium was probably nitrified or
312 volatilized, was used in Kazama et al. (2011). Several reports have revealed a positive
313 correlation between the ammonium concentration and the inactivation rate of pathogenic
314 bacteria (Vinneras et al., 2003; Nordin, 2009b). The *E. coli* disinfection efficiency is affected
315 by the composting process; particularly, the ammonium ion generated through the composting
316 process is supposed to cause the damage on the outer membrane and enzyme activities
317 (Sossou et al., 2014).

318 Damaged components of the MS2 coliphage were also estimated by a comparison of
319 results from the plaque assay, ET-qPCR and qPCR as a model of enteric viruses, according to
320 Pecson et al. (2009). The k value in the infectivity titer indicates the total inactivation
321 efficiency, whereas the higher k value obtained by ETs-qPCR using both the protease and
322 RNase (ET2-qPCR) may show the extent of denaturation of the coliphage capsid, since
323 proteinase K digests the damaged capsid (Nuanualsuwan and Cliver, 2002) but is unable to
324 digest the intact capsid (Nuanualsuwan and Cliver, 2003). The k value obtained by ETs-qPCR
325 using only RNase (ET1-qPCR) reflects the level of RNA exteriorization. The infectivity of
326 enteric viruses requires the functional integrity of both viral RNA and the capsid. The capsid
327 must be sufficiently intact in terms of (i) protecting the genome from degradation and (ii)
328 interacting with host cell receptors (Pecson et al., 2009). At the protein structural level,
329 conformational changes in the capsid, which may affect viral stability and attachment
330 efficiency to the host cell receptor, directly leads to the loss of viral infectivity (Nuanualsuwan
331 and Cliver 2003). MS2 coliphage has a positive-sense, ssRNA genome of 3569 nt that
332 encodes just 4 proteins: maturation, coat, lysis and replicase. Its capsid comprises 90 coat
333 protein dimers arranged in a $T = 3$ icosahedral lattice (Toropova et al., 2011). Each virion also
334 incorporates a single copy of the maturation protein, which binds the virion to the side of the
335 bacterial F-pilus, as a receptor of host *E. coli*, and is therefore an essential protein for
336 infectivity (Toropova et al., 2011). Therefore, it was estimated based on the obtained results

337 that (i) the alkaline-treated compost primarily induced conformational change of the
338 maturation protein on the capsid surface, which led to the reduction of infectivity; and (ii) an
339 increase of the compost pH to 10-10.5 by the alkaline agents then accelerated the
340 conformation change of the coat protein, that caused the partial capsid damage and RNA
341 exteriorization; and (iii) 90 coat protein dimers were deformed and genomic RNA was
342 exteriorized and degraded by bulk RNase in the compost mixture.

343 Multiple parameters of compost characteristics, including temperature and water
344 content, affect the fate of pathogens in compost and also the efficacy of the alkaline
345 disinfection. In this sense, this study represents the disinfection efficiency of surrogate
346 microorganisms (*E. coli* and MS2 coliphage) under a limited but possible condition of
347 compost. It is important in the further study to assess the efficacy of alkaline disinfection
348 under varied environmental conditions to ensure the safe use of compost from a mixing type
349 dry toilet. The effect of microflora on the fate of pathogens in compost is also not negligible.
350 Under the real situation, there is a need to consider the microbial competitive exclusion
351 processes in small bioreactors, where non-pathogenic environmental bacteria are predominant
352 (Evans et al., 2009; Spinks et al., 2006). In this study, the compost mixture before the
353 inoculation of indicator microorganisms was sterile, which means that the effect of microflora
354 in compost on the disinfection efficiency was excluded. This means that an additional effect
355 of the microbial competitive exclusion with the alkaline treatment on the pathogen reduction

356 in compost may be expected in the real operation of a mixing type dry toilet.

357 Indicator microorganisms have pros and cons as surrogates for pathogens (Wu et al.,
358 2011). We employed *E. coli* and MS2 coliphage in this study, which are commonly the first
359 choice as reduction indicators in disinfection practices (Chren et al., 2014; Mondal et al.,
360 2015). However, the disinfection efficacy may depend on bacterial and viral species,
361 particularly when some microbial species have a capability to specifically bind with biosolids
362 (Sano et al., 2004). The comparison of the efficacy of alkaline disinfection on multiple
363 microorganisms is a next research interest for the safe use of compost from a mixing type dry
364 toilet.

365

366

367 **5. Conclusions**

368 The alkaline treatment of compost with CaO and ash resulted primarily in damage to
369 the outer membrane and enzyme activities of *E. coli* and thus damage to the nucleic acid
370 and/or metabolism. The alkaline treatment also led MS2 coliphage primarily to reduce the
371 infectivity and secondarily to cause partial capsid damage and RNA exteriorization. These
372 results indicate that the alkaline treatment can effectively contribute to the disinfection and
373 safe use of compost from a mixing type dry toilet.

374

375

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379

380

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465

466 **Figure captions**

467

468 Fig. 1. Compost pH response to the addition of alkaline agents. A, CaO; B, ash type.

469

470 Fig. 2. Inactivation profiles of surrogate microorganisms in alkaline-treated compost. A,

471 *Escherichia coli*; B, MS2 coliphage.

472

473 Fig. 3. The relationship between inactivation rate constant and compost pH. A, *Escherichia*

474 *coli*; B, MS2 coliphage.

475

476 Fig. 4. Inactivation rate constant of surrogate microorganisms in the alkaline treatment of

477 compost. A, *Escherichia coli*; B, MS2 coliphage.

478

479

480 Table 1. Chemical composition of ash (average in triplicates \pm standard deviation)

	Volatile solids [%]	Cations [mg/g-ash]					
		K	Ca	Mg	Fe	Mn	Zn
mixed	1.7 \pm	173.7 \pm	80.7 \pm	49.5 \pm	10.6 \pm	1.2 \pm	0.5 \pm
ash	0.0	13.1	5.2	2.4	1.0	0.1	0.1
wood	0.2 \pm	112.2 \pm	294.7 \pm	9.7 \pm	1.5 \pm	1.1 \pm	0.1 \pm
ash	0.0	3.1	17.5	0.3	0.1	0.0	0.0
grass ash	5.8 \pm	101.1 \pm	14.3 \pm	10.3 \pm	8.7 \pm	1.0 \pm	0.1 \pm
	0.0	4.9	1.1	0.7	0.7	0.1	0.1

481

Fig. 1

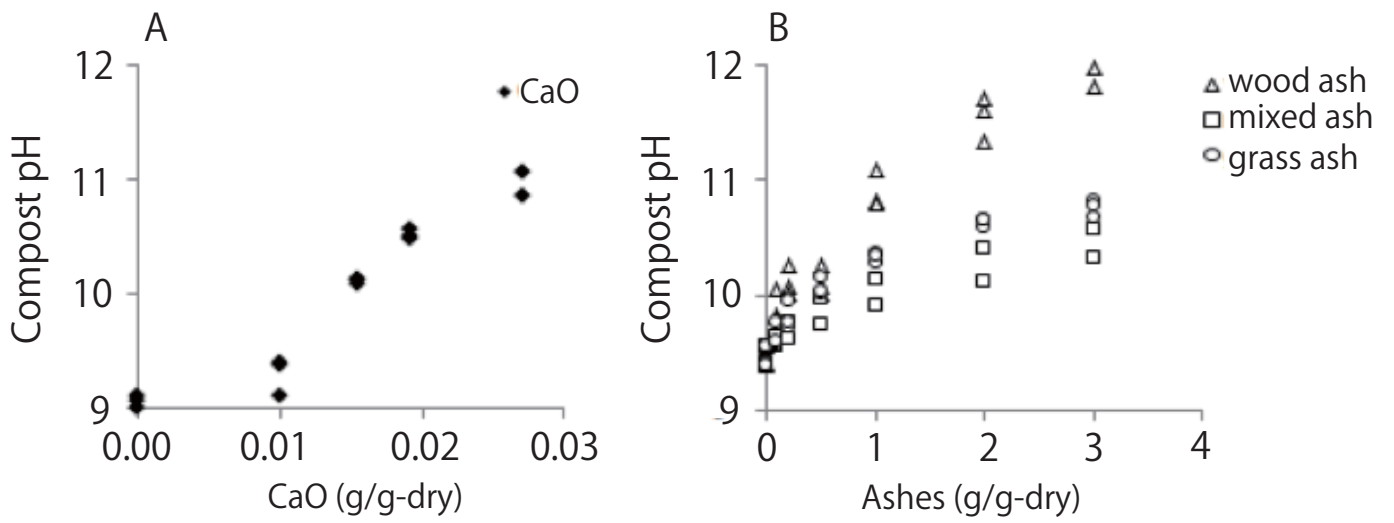


Fig. 2

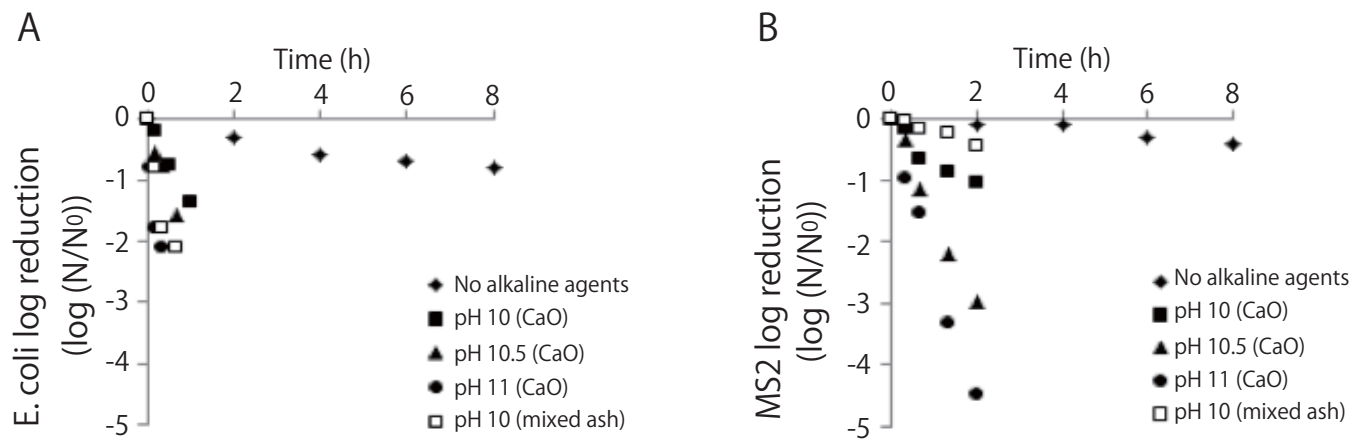


Fig. 3

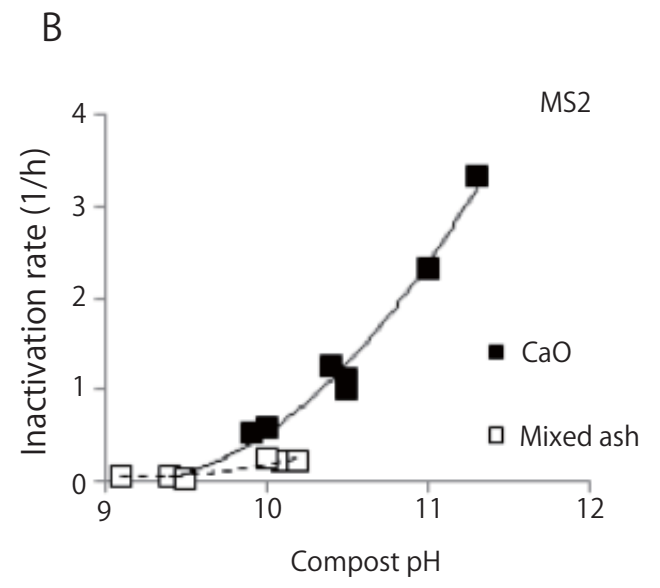
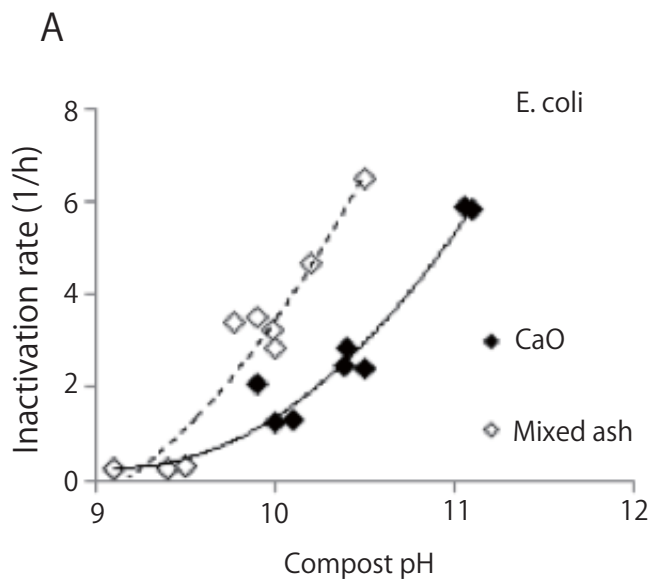


Fig. 4

