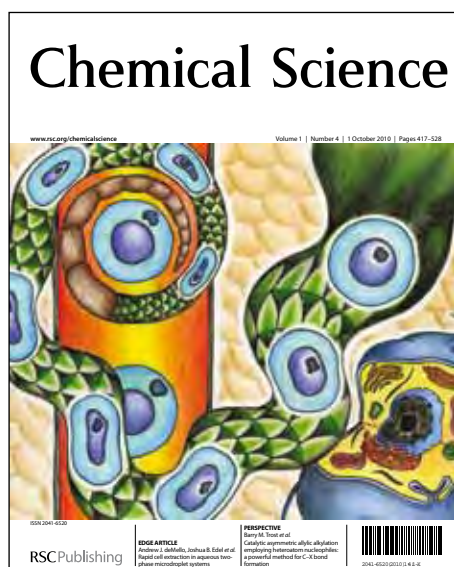


# Chemical Science

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**EDGE ARTICLE****Synthetic Dityrosine-linked  $\beta$ -Amyloid Dimers Form Stable, Soluble, Neurotoxic Oligomers**W. Mei Kok,<sup>a,b,c</sup> Jade M. Cottam,<sup>a,c</sup> Giuseppe D. Ciccotosto,<sup>b,c,d</sup> Luke A. Miles,<sup>c,e</sup> John A. Karas,<sup>c</sup> Denis B. Scanlon,<sup>a,c</sup> Blaine R. Roberts,<sup>d</sup> Michael W. Parker,<sup>c,e</sup> Roberto Cappai,<sup>b,c</sup> Kevin J. Barnham,<sup>c,df</sup> and Craig A. Hutton<sup>\*a,c</sup>

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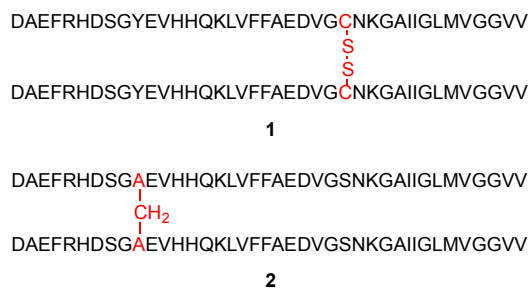
DOI: 10.1039/b000000x

Substantial evidence suggests that soluble oligomers of A $\beta$  are the neurotoxic form resulting in progression of Alzheimer's disease (AD). Tyrosine-10 has been identified as a pivotal residue in the neurotoxicity of A $\beta$  and dityrosine cross-linked A $\beta$  dimers have been proposed as the physiologically relevant A $\beta$  species linked to the progression of AD. We describe the synthesis and characterization of dityrosine-linked A $\beta$  dimers and demonstrate that, in contrast to other covalently-linked A $\beta$  dimers, dityrosine-linked A $\beta$  dimers form discrete, stable, soluble aggregates. Furthermore, dityrosine-linked A $\beta$  dimers display increased toxicity in a neuronal cell-line assay compared with the corresponding monomer, consistent with the hypothesis that dityrosine-linked A $\beta$  dimers are implicated in the progression of AD.

**Introduction**

Alzheimer's disease (AD) is the most common age related neurodegenerative disease, and is characterised by the deposition of extracellular amyloid plaques composed predominantly of  $\beta$ -amyloid (A $\beta$ ) peptides. Though the association of A $\beta$  with AD is well established, extensive evidence now suggests that soluble oligomers of A $\beta$ , particularly 'low-*n*' oligomers such as dimers or trimers, are the neurotoxic form of A $\beta$ , rather than amyloid plaque deposits.<sup>1-6</sup> Soluble A $\beta$  dimers extracted directly from the cerebral cortex of subjects with AD have been shown to disrupt the memory of a learned behaviour in normal rats.<sup>7</sup> Blood levels of A $\beta$  dimers have also been shown to be significantly raised in AD patients compared with healthy controls.<sup>8</sup> Further, it has been shown that insoluble amyloid plaque cores from the AD brain cortex do not inhibit long-term potentiation (LTP) unless they are solubilised to release A $\beta$  dimers,<sup>7</sup> further suggesting A $\beta$  dimers are synaptotoxic.

Synthetic A $\beta$  dimers have been shown to display different biological and biophysical properties than the corresponding monomers. Selkoe and co-workers have prepared disulfide-linked, modified A $\beta$  dimer **1** (Figure 1) and showed that this peptide dimer inhibits LTP in mice, whereas the monomer does not.<sup>7</sup> We have shown that alkyl-linked A $\beta$  dimer **2** fibrilises at a greater rate than the corresponding monomer.<sup>9</sup> However, these synthetic alkyl- and disulfide-linked A $\beta$  dimers clearly do not represent the physiologically relevant A $\beta$  dimers in AD.

**Figure 1.** Synthetic covalently-linked A $\beta$  dimers.

Tyrosine has been identified as an important amino acid for both the conformation and biological activity of A $\beta$ .<sup>10,11</sup> Tyrosine-10 has been proposed as a pivotal residue in the formation of A $\beta$  dimers through the formation of a dityrosine linkage, with variants of A $\beta$  lacking tyrosine at position 10 being non-toxic.<sup>12</sup> Further, evidence suggests dityrosine<sup>13</sup> and dityrosine-linked A $\beta$ 42 dimer levels<sup>8</sup> are elevated in AD patient brains and blood, respectively. Incubation of A $\beta$  under oxidative conditions induces both dityrosine formation and the SDS-resistant oligomerization of A $\beta$ , the latter feature being characteristic of neurotoxic, soluble A $\beta$  extracted from the AD brain.<sup>14,15</sup> Substantial evidence therefore points to dityrosine-linked A $\beta$  dimers being one of the neurotoxic species in AD. However, studies of the properties of dityrosine-linked A $\beta$  are limited due to the absence of a method to prepare such peptides as discrete species. Short dityrosine-linked A $\beta$  dimers have been prepared by oxidation with peroxidase or Cu<sup>2+</sup>/H<sub>2</sub>O<sub>2</sub>,<sup>16-18</sup> but the method is inefficient with A $\beta$  peptides longer than A $\beta$ 16.

Palladium-catalysed cross-linking of iodotyrosine-containing A $\beta$  fragments is similarly limited to production of short peptide dimers.<sup>19</sup>

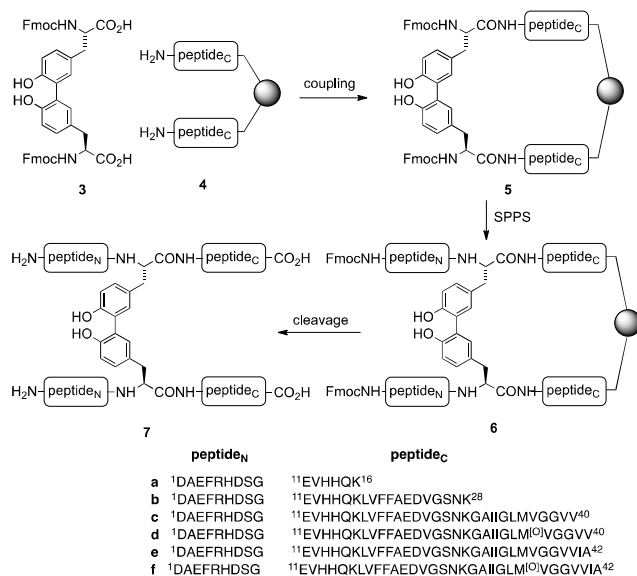
We herein disclose the first chemical synthesis of physiologically-relevant, dityrosine cross-linked A $\beta$  dimers. Our studies show that the nature of the cross-link has a dramatic effect on the properties of the corresponding A $\beta$  dimers. Specifically, dityrosine-linked A $\beta$  dimers are slow to fibrilise, forming *long-lived, soluble oligomeric aggregates*, and are significantly more neurotoxic than the corresponding monomeric peptides.

## Results and Discussion

### Synthesis of dityrosine-linked A $\beta$ dimers

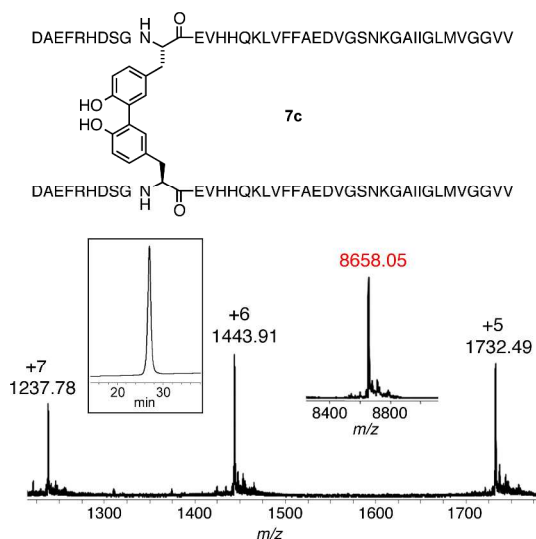
Synthesis of dityrosine-linked A $\beta$  dimers **7a–f** proceeded through modification and optimization of our method developed for the preparation of diaminopimelate- (DAP)-linked peptide dimers,<sup>9</sup> wherein an Fmoc-protected bis amino acid (such as **3**) was coupled to two contiguous resin-bound peptides **4** to give bis-ligated adduct **5**. Peptide extension through standard Fmoc-based SPPS to give **6** was followed by cleavage from the resin to give the peptide dimer **7** (Scheme 1). Modification of the protocol was necessary due to the presence of unprotected side-chain phenolic groups in Fmoc-dityrosine **3**. Accordingly, coupling of Fmoc-dityrosine **3**<sup>20,21</sup> was performed under non-basic conditions, employing a mixture of DIC and HOBt in place of the standard HBTU/DIPEA coupling conditions. Optimization of the dityrosine coupling step was investigated through synthesis of the A $\beta$ 16 dimer **7a**, by varying the ratio of Fmoc-dityrosine **3** and coupling agent. Coupling of 1 equiv of dityrosine **3** to resin-bound peptide **4a** gave the corresponding mono-coupled adduct **8a** as the major product through ligation of a single acid group of **3** to the resin. Reducing the amount of dityrosine **3** to 0.5 equiv (1 equiv acid per peptide *N*-terminal amine) similarly yielded the mono-coupled adduct **8a** as the major product, with some bis-adduct **7a**. The use of 0.25 equiv of dityrosine **3** yielded the bis-coupled adduct **7a** (along with unreacted A $\beta$ (11–16) peptide), without the generation of the corresponding mono-coupled adduct **8a**. Subsequent coupling of a further 0.25 equiv of dityrosine **3** increased the amount of bis-coupled adduct **7a**. Accordingly, optimised conditions involved 2–3 sequential couplings of 0.25 equiv of Fmoc-dityrosine **3**. Purification was performed under standard conditions by preparative scale RP-HPLC to give the A $\beta$ 16 dityrosine-linked dimer **7a** in >95% purity (see SI).

Preparation of the A $\beta$ 28 dityrosine-linked dimer **7b** proceeded according to method for the A $\beta$ 16 dimer **7a**, except that a two-step purification protocol was required: RP-HPLC followed by size exclusion chromatography afforded the A $\beta$ 28 dimer **7b** in >95% purity (see SI).



Scheme 1. SPPS of dityrosine-linked peptide dimers.

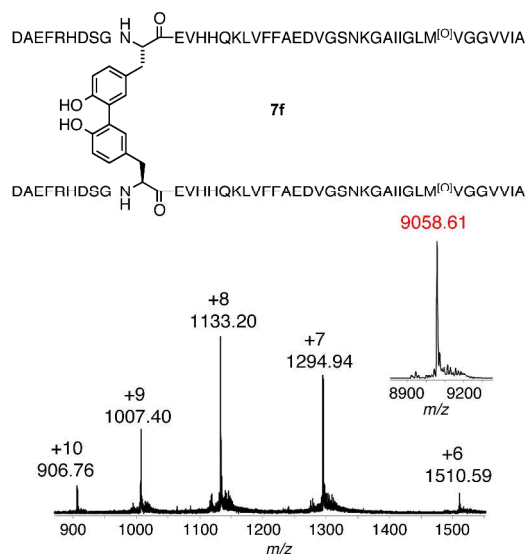
Preparation of the A $\beta$ 40 dimer **7c** was achieved following the same method developed for the syntheses of the shorter A $\beta$  peptide dimers. Mass spectrometric analysis of the crude peptide revealed the appearance of the expected molecular ion  $[M+6H]^{6+}$  at *m/z* 1443.9, and an additional +6-charged ion at 1446.4, corresponding to a byproduct 16 amu greater than the dimer **7c**. The  $[M+16]$  species presumably arises from oxidation of the methionine residue at position 35 to the methionine sulfoxide (Met<sup>(O)</sup>).<sup>24,25</sup> Treatment of the crude peptide with TMS-Br and thioanisole for 15 minutes resulted in reduction of the oxidised peptide **7cd** to give clean **7c**.<sup>22</sup> An optimised procedure was developed in which addition of TMS-Br and thioanisole to the cleavage mixture resulted in successful reduction of the peptide, thus enabling preparation of the A $\beta$ 40 peptide dimer **7c** free from oxidised version **7cd** with no additional steps. Purification of the A $\beta$ 40 dimer **7c** was achieved through a two-step RP-HPLC protocol, providing the peptide dimer **7c** in >95% purity (Figure 2). Alternatively, direct incorporation of a Met<sup>(O)</sup> residue into the peptide sequence gave the fully-oxidised A $\beta$ 40-Met<sup>(O)</sup>35 dimer **7d** via the same protocol.



**Figure 2.** ESI-MS of Aβ40 dityrosine dimer **7c** with deconvoluted MS and HPLC (inset) (calcd.  $m/z$  8658.1).

5 With the successful preparation of the Aβ40 dimers **7c** and **7d**, the synthesis of the full-length Aβ42 dimer was next investigated. Aβ42 is known to display significantly greater propensity to aggregate than Aβ40.<sup>26</sup> Coupling of Fmoc-dityrosine **3** to the Aβ(11–42) resin-bound peptide **4e** was not successful; only  
10 unreacted Aβ(11–42) peptide was detected by mass spectrometric analysis. Further investigations with alternative coupling agents (e.g., HATU, triphosgene) were also unsuccessful. Presumably, on-resin peptide aggregation occludes the *N*-termini of  
15 contiguous peptide chains such that coupling to both acid groups of Fmoc-dityrosine **3** is prevented.<sup>27</sup>

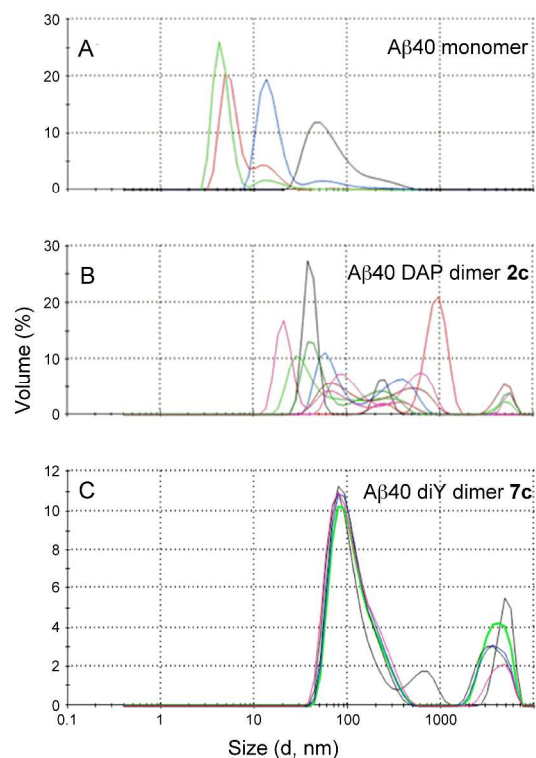
The presence of Met<sup>[O]</sup>35 in Aβ peptides has been shown to stabilise the random structure and decrease the rate of β-sheet production due to the increased polarity of the Met<sup>[O]</sup> side-chain.<sup>28</sup> Thus the incorporation of an oxidised methionine was  
20 anticipated to result in a more soluble peptide with less hydrophobicity-induced aggregation. Indeed, coupling of Fmoc-dityrosine **3** to resin-bound Aβ(11–42)Met<sup>[O]</sup>35 peptide **4f** was found to be successful, where coupling to Aβ(11–42) peptide **4e** was not. Mass spectrometric analysis of the cleavage mixture  
25 revealed the presence of both the mono-coupled product **8f** and the desired Aβ42-Met<sup>[O]</sup>35 dityrosine-linked dimer **7f** (see SI). However, the Aβ42 dityrosine-linked dimer **7f** was extremely prone to aggregation resulting in purification of this peptide being much more difficult than for the Aβ40 dimers **7c** and **7d**, with  
30 standard conditions failing to separate the Aβ42-Met<sup>[O]</sup>35 dimer **7f** from the mono-coupled product **8f**. After considerable experimentation, purification of the oxidised Aβ42 dimer **7f** was achieved by size-exclusion chromatography eluting with 70% formic acid in water. Formic acid has previously been used to  
35 prevent aggregation of Aβ peptide solutions.<sup>29,30</sup> However, formation of formate esters of Aβ occurred if the peptide was left in the eluting solvent for >30 min.<sup>31</sup> Accordingly, immediate removal of the formic acid by elution through a hydrophilic-lipophilic balance (HLB) cartridge was necessary, furnishing  
40 small amounts of the dimer **7f** (Figure 3).



**Figure 3.** ESI-MS and deconvoluted MS (inset) of Aβ42 dityrosine-linked dimer **7f** (calcd.  $m/z$  9058.2).

### Particle Size Analysis by Dynamic Light Scattering

We have previously shown that alkyl-linked Aβ dimers exhibit formation of large, polydisperse aggregates soon after dissolution, similar to the aggregation of the corresponding Aβ  
50 monomers.<sup>9</sup> The aggregation properties of the dityrosine-linked Aβ dimers **7a–f** were investigated using dynamic light scattering (DLS).



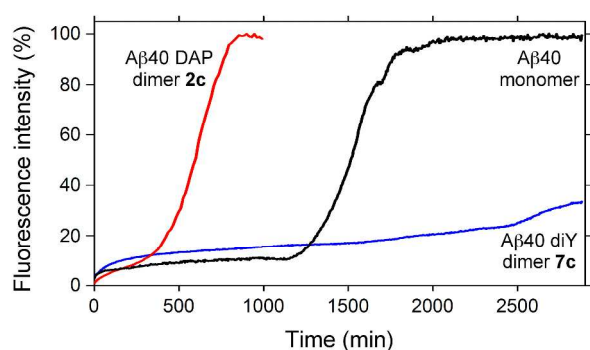
**Figure 4.** Particle size analysis by DLS: repeat measurements shown for (A) Aβ40 monomer; (B) Aβ40 DAP dimer **2c**; (C) Aβ40 dityrosine dimer **7c**.

The dityrosine-linked dimers **7a** and **7b** display a propensity towards the formation of large, soluble aggregates (see SI), in stark contrast to the corresponding monomer and alkyl-linked dimer **2a**, which both display size distribution profiles consistent with monomeric states at the same concentration.<sup>9</sup>

The dityrosine-linked A $\beta$ 40 dimer **7c** was also observed to reproducibly form high-order aggregates, with the majority of particles of *ca.* 80 nm diameter (Figure 4C). The narrow dispersity of the dityrosine-linked A $\beta$ 40 dimer **7c** aggregates contrasts with both the A $\beta$ 40 monomer and the corresponding DAP-linked dimer **2c**, which display a high polydispersity in formation of aggregates (Figure 4A/B).<sup>9</sup> The A $\beta$ 42-Met<sup>[O]35</sup> dimer **7f** was found to be so prone to aggregation that consistent preparations of solutions of this peptide were difficult to reproduce. Accordingly, the A $\beta$ 42-Met<sup>[O]35</sup> dimer **7f** was not subject to these or subsequent assays. Nevertheless, the DLS studies show that the presence of a dityrosine cross-link in A $\beta$  peptide dimers specifically promotes the formation of high-order, soluble, oligomeric forms of A $\beta$ .

### 20 Fibrilization Kinetics using ThT Assay

The effect of the dityrosine cross-link on the formation of amyloid fibrils was next investigated. We have previously shown that DAP-linked A $\beta$ 40 dimer undergoes fibrilization with a considerably reduced lag-time compared with the A $\beta$ 40 monomer, consistent with the dimer representing the first stage of nucleation toward the fibrilization of the peptide. Intriguingly, the dityrosine-linked A $\beta$ 40 dimer **7c** exhibited a greatly *decelerated* amyloid formation process, with no fibrilization detected after two days by ThT assay (Figure 5: diY-dimer **7c**, blue; DAP-dimer **2c**, red; monomer, black). The combination of DLS and ThT studies show that whilst the dityrosine-linked dimers form high order aggregates, these do not transform readily to amyloid fibrils; i.e. they form *long-lived, soluble aggregates*.



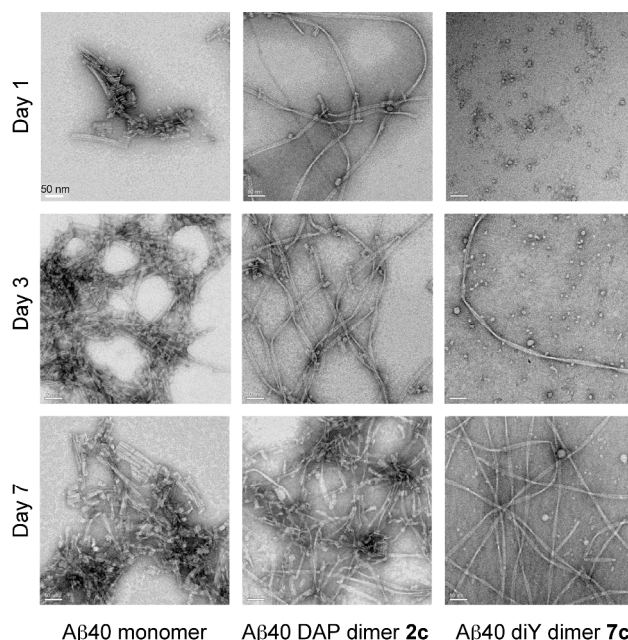
35 **Figure 5.** ThT assay of fibril formation of A $\beta$ 40 monomer (black), A $\beta$ 40 DAP-linked dimer **2c** (red) and A $\beta$ 40 dityrosine-linked dimer **7c** (blue); Relative fluorescence intensity; excitation 444 nm, emission 480 nm.

### Electron Microscopy

40 In order to corroborate the DLS and ThT assay results, the A $\beta$  monomer and dimers were aged over 7 days and analysed by electron microscopy. The microscopy results are in close accord with the ThT assays, with the DAP-linked A $\beta$ 40 dimer **2c** showing extensive fibril formation after one day, while the A $\beta$ 40

45 monomer requires 2–3 days for extensive fibril formation.

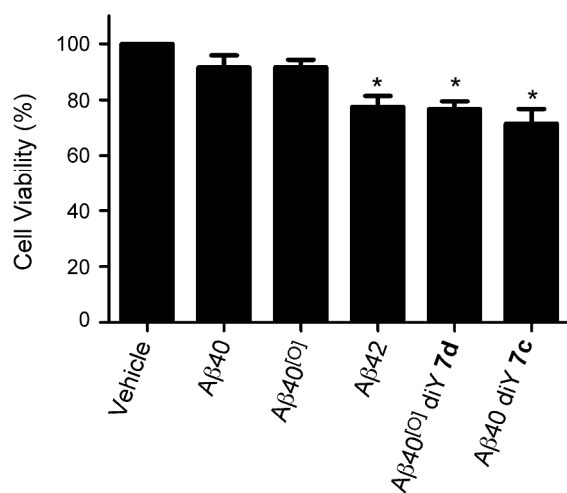
In contrast, the dityrosine-linked A $\beta$ 40 dimer **7c** shows dramatically reduced fibril formation (Figure 6). After one day, no fibrils are seen; a high concentration of globular aggregates is observed. After 2–3 days, few fibrils are observed, with the 50 globular aggregates predominating. Only after 7 days are significant numbers of fibrils observed, though substantial quantities of globular aggregates remain. The combination of DLS, ThT and EM findings are all consistent with the dityrosine-linked A $\beta$  dimers forming high-order aggregates that do not 55 transform readily to amyloid fibrils.



**Figure 6.** TEM of A $\beta$ 40 monomer (left), A $\beta$ 40 DAP-linked dimer **2c** (middle) and A $\beta$ 40 dityrosine-linked dimer **7c** (right), after aging for 1, 3 60 and 7 days.

### Toxicology

Cell-based assays were undertaken to determine the neurotoxicity of the different A $\beta$  monomer and dimer species. Neuroblastoma 65 SH-SY5Y cells<sup>23</sup> were treated with A $\beta$  monomer species A $\beta$ 42, A $\beta$ 40 and A $\beta$ 40-Met<sup>[O]35</sup>, and the A $\beta$ 40 and A $\beta$ 40-Met<sup>[O]35</sup> dityrosine-linked dimers **7c** and **7d**. The peptides were added to the culture medium at 6  $\mu$ M effective concentration of the A $\beta$  monomer (that is, the monomers were added at 6  $\mu$ M and the 70 dimers were added at 3  $\mu$ M). At 6  $\mu$ M the A $\beta$ 40 monomers (containing either Met35 or Met<sup>[O]35</sup>) both have a weak effect on the neuronal cell viability, reducing viability by <10% (Figure 7). In contrast, the A $\beta$ 40 dityrosine-linked dimers **7c** and **7d** cause a significant decrease in cell viability (25–30% reduction), 75 comparable to or greater than that of the of the A $\beta$ 42 monomer. These results clearly demonstrate that the dityrosine-linked A $\beta$  dimers are significantly more toxic than the corresponding monomers.



**Figure 7.** SH-SY5Y neuronal cell viability: cells were treated with 6  $\mu\text{M}$  A $\beta$  peptide for 4 days and cell viability determined as a percentage of the untreated vehicle controls. \*,  $p < 0.001$  vs vehicle.

## 5 Conclusions

In conclusion, we have synthesised for the first time the dityrosine-linked A $\beta$ 4 peptide dimers proposed as physiologically-relevant A $\beta$  species present in the brains of AD patients. We have demonstrated that in contrast to other covalently-linked A $\beta$  dimers, the dityrosine-linked dimers display decreased rates of fibrilization and instead form discrete, stable, soluble aggregates. Further, dityrosine-linked A $\beta$  dimers display increased toxicity in a neuronal cell-line assay compared with the corresponding monomers. These biophysical and cell-based assays all provide corroborating evidence that dityrosine-linked A $\beta$  dimers form long-lived, soluble oligomers that are highly neurotoxic – consistent with the hypothesis that dityrosine-linked A $\beta$  dimers are one of the species responsible for the progression of AD. Further studies of the neurotoxicity of A $\beta$  dimers will be reported in due course.

## Materials and Methods

### Peptide Synthesis and Purification

Dityrosine was prepared according to our reported procedure,<sup>20,21</sup> then converted to Fmoc-protected dityrosine **3** with Fmoc-OSu under standard conditions. Incorporation of Fmoc-dityrosine into the solid phase peptide synthesis (SPPS) strategy for preparation of peptide dimers was performed on Fmoc amino acid loaded PEG-PS resin, using a modified version of our reported method for preparation of DAP-linked A $\beta$  dimers **2**,<sup>9</sup> employing DIC/HOBt in place of HBTU/DIPEA. For the A $\beta$ 40 dimer **7c**, cleavage of the peptide was performed in the presence of TMSBr/thioanisole to reduce formation of methionine sulfoxides (*vide infra*).<sup>22</sup> A $\beta$ 16, A $\beta$ 28 and A $\beta$ 40 dimers **7a–d** were purified by the reported methods. Purification of the A $\beta$ 42 Met35<sup>[O]</sup> dimer **7f** was performed by elution through two connecting Superdex<sup>TM</sup> 75 size exclusion columns (13  $\mu\text{M}$ , 10 mm  $\times$  300 mm) with 70% formic acid/30% water with a flow rate of 0.7 mL/min for 35 minutes. Dimer containing fractions were immediately eluted through an HLB Oasis cartridge to remove formic acid with 2% ammonium hydroxide in 90% methanol.

### Dynamic Light Scattering

Lyophilised peptides were dissolved on ice under a range of buffer conditions including water alone, 20 mM HEPES pH 7.0, and 1 $\times$  phosphate buffered saline (PBS; 137 mM NaCl, 10 mM Na<sub>2</sub>HPO<sub>4</sub>, 2.7 mM KCl, 1.5 mM KH<sub>2</sub>PO<sub>4</sub>, pH 7.4). Samples were subject to centrifugation (15,000 g, 10 min, 4  $^{\circ}\text{C}$ ) to remove particulates immediately prior to analysis. DLS measurements were made with a Malvern Instruments Zetasizer Nano ZS instrument. Size distribution profiles were measured repeatedly over the first 30 minutes following dissolution. All samples demonstrated some degree of polydispersity and no discernible difference was observed between buffer conditions. Peptide solutions were prepared at 0.5 mg.mL<sup>-1</sup>: A $\beta$ 16 monomer, 0.26 mM; A $\beta$ 16 dimers **2a**, **7a**, 0.13 mM; A $\beta$ 28 monomer, 0.15 mM; A $\beta$ 28 dimers **2b**, **7b**, 0.08 mM; A $\beta$ 40 monomer, 0.11 mM; A $\beta$ 40 dimers **2c**, **7c**, 0.06 mM.

### ThT assay for fibril formation

Dry peptide was dissolved in 1,1,1,3,3,3-hexafluoro-2-propanol (HFIP) at a concentration of 1.0 mg.mL<sup>-1</sup>, then aliquotted and dried under vacuum and stored at  $-80^{\circ}\text{C}$ . Peptide concentration was determined using absorbance at 214 nm and extinction coefficients of 91462 M<sup>-1</sup>cm<sup>-1</sup> for A $\beta$ 40, 239706 M<sup>-1</sup>cm<sup>-1</sup> for A $\beta$ 40 DAP-dimer **2c** and 197244 M<sup>-1</sup>cm<sup>-1</sup> for A $\beta$ 40 dityrosine-dimer **7c**, as determined by amino acid analysis. Aliquots of dried peptide were dissolved in 20 mM NaOH then diluted in deionised water and phosphate buffer (100 mM potassium phosphate, pH 7.4) at a v/v/v ratio of 2:7:1. All solutions were sonicated at 0  $^{\circ}\text{C}$  for 10 min and filtered (20  $\mu\text{m}$ ) to ensure pre-formed aggregates were removed. Final concentrations: A $\beta$ 40 monomer, 14  $\mu\text{M}$ ; A $\beta$ 40 dimers **2c**, **7c**, 7  $\mu\text{M}$ ; Thioflavin-T (ThT), 28  $\mu\text{M}$ ; in 1 $\times$  PBS to a final volume of 600  $\mu\text{L}$ . ThT-induced fluorescence was measured using a fluorescence spectrophotometer fitted with a peltier-driven temperature controller and a multi-cell holder. Each sample was incubated at 37  $^{\circ}\text{C}$ . Excitation was at 444 nm with fluorescence emission measured at 480 nm. Readings were taken every 60 s for the first 15 min, then every 15 min for the next 885 min. Slit widths were 5 nm for both excitation and emission.

### Electron Microscopy

A 3.5  $\mu\text{L}$  aliquot of the sample solution used for the ThT assay was adsorbed onto a carbon-coated film mounted on 300-mesh copper grid. Prior to adsorption, the grids were rendered hydrophilic by glow discharge in a reduced atmosphere of air for 10 s. After 30 s adsorption, samples were blotted and negatively stained with 1.5% aqueous uranyl acetate. The transmission electron microscope was operated at 200 kV, with images acquired digitally.

### Neuronal cell line toxicity assays

Dry peptide was weighed and dissolved in HFIP to monomerise the peptide, then was then dispensed into small amounts and dried using a speed-vac and stored at  $-80^{\circ}\text{C}$  until use. A $\beta$  peptides were dissolved in 20 mM NaOH, then diluted in water followed by the addition of 1 $\times$  PBS in a ratio of 2:7:1, as previously described.<sup>1</sup> Peptide concentrations were determined from the absorbance value at 214 nm, using the calculated molar extinction coefficient values of 75887 M<sup>-1</sup>cm<sup>-1</sup> for A $\beta$ 42, 91462

$M^{-1}cm^{-1}$  for A $\beta$ 40 and A $\beta$ 40Met<sup>[O]</sup>, and 197,244 for A $\beta$ 40 and A $\beta$ 40Met<sup>[O]</sup> dityrosine dimers **7c** and **7d**.

SH-SY5Y human neuroblastoma cells (American Type Culture Collection, Rockville, MD, USA) were grown in DMEM supplemented with 10% (v/v) heat-inactivated fetal calf serum (FCS), 1 U.mL<sup>-1</sup> of penicillin, 1  $\mu$ g.mL<sup>-1</sup> of streptomycin, and 2 mM glutamate (Gibco BRL; Invitrogen, Victoria, Australia) and maintained at 37 °C and 5% CO<sub>2</sub> as previously described.<sup>23</sup> To obtain differentiated cells, 20,000 cells were plated per well of a 48-well plate and allowed to adhere for 24 h. Differentiation was started in DMEM supplemented with 1.5% FCS and 10  $\mu$ M retinoic acid (RA). Fresh medium, containing RA and 1.5% FCS, was applied to the cells every 1–3 days. Experiments were typically performed on cells differentiated for at least 14 days and displaying a differentiated neuronal phenotype, including extensive neurites and branching, as evidenced by light microscopy. To determine the cell toxicity of the A $\beta$  peptides, differentiated cells were treated with 0 and 6  $\mu$ M effective concentration of A $\beta$  peptides for 4 days in DMEM/1.5% FCS. The cell Counting Kit-8 (Ausppep, Australia) was used to determine the cell viability at the end of the treatment. The measured cell viability values were normalised to vehicle-treated control groups (set to 100%) and each treatment was done in at least triplicate ( $n = 3-7$ ). Results are shown as mean  $\pm$  S.E.M. The data were analysed by two-tailed Student's *t*-test. Values of  $p < 0.05$  were considered significant.

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## Notes and references

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† Electronic Supplementary Information (ESI) available: Experimental procedures, including the synthesis of **3**, preparation and purification of peptides **7a-d,f**, and protocols for DLS, ThT, EM and cell viability experiments. See DOI: 10.1039/

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