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# ROS-mediated Cytotoxicity and Macrophage Activation Induced by TiO<sub>2</sub> Nanoparticles with Different *in vitro* Non-Cellular Photocatalytic Activities

Traian Popescu<sup>1</sup>, Lidia Cremer<sup>2</sup>, Mihaela Tudor<sup>2</sup>, Andreea-Roxana Lupu<sup>2,3\*</sup>

<sup>1</sup>National Institute of Materials Physics, P.O. Box MG-7, 077125, Bucharest, Romania; <sup>2</sup>Immunomodulation Group, "Cantacuzino" National Research Institute, Bucharest, Romania; <sup>3</sup>Immunobiology Laboratory, Assay Development and Alternative Testing Team, "Victor Babes" National Institute of Pathology, Bucharest, Romania

## Abstract

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**Key words:** TiO<sub>2</sub> nanoparticles; macrophages; reactive oxygen species; inflammation; IL-6.

**\*Correspondence:** Andreea-Roxana Lupu, Immunomodulation Group, "Cantacuzino" National Research Institute, Bucharest, Romania. Tel: +40213192732. Email: [immunomod@cantacuzino.ro](mailto:immunomod@cantacuzino.ro), [ldreea@gmail.com](mailto:ldreea@gmail.com)

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**Competing Interests:** The author have declared that no competing interests exist.

**AIM:** The aim of the study described in the present paper was to assess several *in vitro* effects of TiO<sub>2</sub> nanoparticles with different colloidal and photocatalytic properties on RAW 264.7 macrophages.

**METHODS:** The cells were exposed to Degussa P25 titania and two other types of nanoparticles synthesized by a hydrothermal procedure in our laboratory: undoped and Fe<sup>3+</sup>-doped TiO<sub>2</sub>. Compared to Degussa P25, the hydrothermal nanomaterials were significantly less active in inducing cytotoxicity, production of intracellular reactive oxygen species (ROS) and release of pro-inflammatory cytokine interleukin-6 (IL-6). The induced effects were analysed with respect to nanoparticle size, surface charge, hydrophilicity, semiconductor bandgap energy and photocatalytic generation of ROS under non-cellular conditions.

**RESULTS:** The overall results indicated that TiO<sub>2</sub> nanoparticles with higher surface charge, hydrophilic surfaces and enhanced photocatalytic properties may preferentially induce macrophage cell damage and inflammation compared to other TiO<sub>2</sub> nanomaterials.

**CONCLUSION:** The present findings are relevant for studies regarding the evaluation of risks raised by self-cleaning technologies involving nanosized hydrophilic TiO<sub>2</sub> photocatalysts as well as development of synthesis methods optimized for producing biocompatible TiO<sub>2</sub> nanomaterials.

## Introduction

Titanium dioxide (TiO<sub>2</sub>, titania) is a multifunctional material with optical (high refractive index, absorption of ultraviolet (UV) radiation), photocatalytic [1-8] and photoconduction [9] properties that determine its use in a large variety of consumer products and technological applications. TiO<sub>2</sub> micro (MP) and nanoparticles (NP) are found in the composition of food, pharmaceutical and personal care products [10-14] (being thus ingested or having direct contact with the skin) or used as active agents (to enhance hydrophilicity or to induce bactericidal effects) in self-cleaning [15, 16] and antimicrobial technologies [17-19]. The extensive use of

engineered, highly reactive, TiO<sub>2</sub> nanomaterials (specifically designed for particular applications) raises increasing concern regarding their potential harmful effects to humans and environment.

Irrespective of the way they enter the human body, TiO<sub>2</sub> nanoparticles represent exogenous substances and trigger immune responses that involve specialized immune cells, especially macrophages, which actively internalize such foreign particles [20, 21]. The phagocytic activity of macrophage cells is often associated with generation of reactive oxygen species (ROS) and inflammation [21-23]. In the case of NP exposure, these two fundamental immunological processes are interdependent, the NP-induced ROS generation leading to activation of redox-dependent pro-

inflammatory mechanisms [24, 25]. Activated macrophages play a key role in promoting and maintaining inflammation through secretion of pro-inflammatory mediators including IL-6. However, the relationship between the structural and physicochemical properties of engineered nanoparticles and their ROS generation and pro-inflammatory effects in macrophage cells remains unclear.

In this context, we have studied several *in vitro* effects of three types of TiO<sub>2</sub> nanoparticles on RAW 264.7 macrophages, a frequently used *in vitro* model for inflammation-related studies. The cells were exposed to Degussa P25 titania and two other types of TiO<sub>2</sub> nanoparticles synthesized by a hydrothermal procedure in our laboratory: undoped (HT) and Fe<sup>3+</sup>-doped (FeHT).

The tested nanomaterials were such chosen to offer the possibility to investigate the relevance of semiconductor bandgap energy, specific surface area, photocatalytic efficiency and suspension stability for the results obtained in biochemical tests regarding cytotoxicity, intracellular ROS production and inflammatory processes.

The inflammation marker considered in our study was the interleukin-6 (IL-6) cytokine. IL-6 is produced at the site of inflammation by a variety of cell types but its most important sources are macrophages and monocytes [26]. IL-6 may exhibit its pro-inflammatory action by either activating its near cells (paracrine action) to produce pro-inflammatory mediators or by acting upon its source cell (autocrine action) to promote its own release through a positive feedback mechanism. Therefore, IL-6 plays a key role in propagating chronic inflammation [27, 28].

The understanding of the relationship between structural and physicochemical nanoparticle properties and the basic immune response they induce under *in vitro* and *in vivo* conditions helps researchers develop synthesis methods optimized for producing biocompatible nanomaterials.

## Materials and Methods

### Materials synthesis

To prepare the undoped and iron-doped TiO<sub>2</sub> nanoparticles, TiCl<sub>3</sub> (solution 15 % in HCl 10 %, from Merck) and Fe<sub>2</sub>O<sub>3</sub> (RITVERC, 95.44 % <sup>57</sup>Fe Isotopic Enrichment) were used as starting reagents [29].

The Fe<sup>3+</sup> (1 at. %)-doped TiO<sub>2</sub> was synthesized by coprecipitation, NH<sub>4</sub>OH being dropwise added to a mixture of TiCl<sub>4</sub> (obtained by oxidizing TiCl<sub>3</sub> by air bubbling) and FeCl<sub>3</sub> (obtained by reacting Fe<sub>2</sub>O<sub>3</sub> to hydrochloric acid (4N)). The addition of NH<sub>4</sub>OH was stopped at pH=8. The obtained

precipitate was separated by filtration and washed with deionized water. The hydrothermal treatment underwent at 200 °C, for one hour, in a 50 cm<sup>3</sup> Teflon-lined autoclave.

The undoped TiO<sub>2</sub> was synthesized in a similar manner, using the titanium precursor only.

### Materials characterization

The physicochemical (structural, morphological, optical, colloidal) properties of the synthesized nanomaterials were characterized by X-ray diffraction (XRD), Mössbauer spectroscopy, transmission electron microscopy (TEM), Brunauer-Emmett-Teller (BET) nitrogen adsorption, dynamic light scattering (DLS), UV-Vis reflectance spectroscopy, diffuse reflectance infrared Fourier transform spectroscopy (DRIFT) and *energy-dispersive X-ray spectroscopy* (EDX). The detailed description of the used material characterization methods was presented in a previous paper [30].

### Determination of cell viability

The RAW 264.7 macrophages were seeded in 24-well culture plates, in volumes of 1 ml of DMEM-F12 culture medium (containing 10% FBS-complete culture medium), at a density of 1 x 10<sup>5</sup> cells/cm<sup>2</sup>. After 24 h of incubation (required for cell adhesion) the culture medium was removed and replaced with volumes of 1ml of fresh medium containing 1, 10, 50, 100, 200 µg of the tested TiO<sub>2</sub> nanoparticles (P25, HT, FeHT). After 24, 48 and 72 h respectively, the medium containing nanoparticles was discarded and volumes of 300 µl MTT solution (1 mg/ml MTT in PBS) were added to each well. The obtained samples were incubated for 2 hours at 37°C, 5% CO<sub>2</sub> and 90% relative humidity. The MTT solution was afterwards removed and volumes of 300 µl of DMSO (Aldrich) were added to each well in order to dissolve the formed formazan crystals. The absorbance of the purple formazan solution was determined at 540 nm using a Thermo Multiskan EX spectrophotometer.

The cell viability was expressed as percent versus control (untreated cells). Four technical replicates were used for each tested sample and a total number of three independent biological replications were conducted.

The same procedure was performed for cell samples containing TiO<sub>2</sub> (in the above mentioned concentrations) and 22mM Mannitol (used as ROS scavenger). Cell viability was determined as described above after 72 hours of treatment and expressed as percent versus control (untreated cells).

All results are represented as average values ± standard deviation (error bars).

### Determination of intracellular ROS production

The intracellular ROS production was determined for cells treated with TiO<sub>2</sub> alone (1, 10, 50, 100, 200 µg TiO<sub>2</sub>/ml of culture medium) or co-treated with TiO<sub>2</sub> and Mannitol (22 mM).

After 24 hours of incubation, the culture medium (containing the treating agents) was discarded and replaced with fresh medium containing DCFH-DA (0.2 µl of stock solution (2'-7'-diclorofluorescein-diacetat (DCFH-DA) 25 mg/mL in PBS) in 1 ml of culture medium).

The obtained samples were thermostat incubated for 30 minutes. After incubation, the culture medium containing DCFH-DA was discarded, the cells being detached with trypsin (0.25% trypsin and 0.53 mM EDTA solution) and suspended in PBS. The samples were centrifuged for 10 min at 1500 rpm and 4°C and the cell supernatant was removed. The cells were washed twice by centrifugation in 2 ml PBS to remove the excess of fluorescein and then resuspended in 500 µl PBS. Suspensions were homogenized and volumes of 100 µl of each sample were transferred to 96-well plates. Fluorescence signals were analysed using a Fluoroskan FL (Thermo) fluorimeter with 485 nm excitation and 530 nm emission wavelengths, respectively. Control samples, consisting of untreated cells (or cells treated with Mannitol alone), were subjected to the same procedure. Three identical samples (technical replicates) were used for each tested case and a total number of three independent biological replications were conducted. The results are expressed as percent versus control and represented as average values ± standard deviation (error bars).

#### *Quantification of IL-6 in culture supernatants*

The quantification of IL-6 in culture supernatants was performed using the Mouse IL-6 Quantikine ELISA Kit (R&D Systems).

The cultured RAW 264.7 macrophages were incubated with TiO<sub>2</sub> nanoparticles (P25, HT, FeHT) for 24 h. Culture supernatants were afterwards placed in ELISA plates coated with purified anti-(IL-6) antibodies (capture antibodies). The antigen (IL-6) present in each culture supernatant was bound by the capture antibody molecules. After two hours, the content of the plates was discarded and the plates were washed to remove the unbound antibody – antigen systems. A second anti-(IL-6) antibody, coupled with the horseradish peroxidase enzyme (antibody/enzyme conjugate), was added to the system to bind the antigen (which was already fixed to the bottom of the plate by the capture antibody). The content of the plates was discarded again and the plates were washed to remove all molecules that did not form *capture antibody – antigen (IL-6) – antibody/enzyme conjugate* chains. The substrate solution (1:1 mixture of stabilized hydrogen peroxide and stabilized chromogen peroxide

(tetramethylbenzidine)) was subsequently added to the plates and converted to a colored compound by the bound horseradish peroxidase enzyme.

The absorbance of the colored solution was proportional to the concentration of bound enzyme and IL-6 respectively.

Control samples consisting of untreated cells were subjected to the same, above described, procedure. The concentration of IL-6 in the sample was calculated based on a standard curve obtained by performing the above described protocol using known concentrations of IL-6.

Three identical samples (technical replicates) were used for each tested case and a total number of three independent biological replications were conducted. The results are expressed as percent versus control and represented as average values ± standard deviation (error bars).

#### *Non-cellular effects of TiO<sub>2</sub> nanoparticles on the fluorescence of fluorescein sodium salt*

Independent volumes of 1 ml of DMEM-F12 culture medium were mixed with volumes of 1 ml of TiO<sub>2</sub> aqueous suspension (100 µg/ml) corresponding to each of the tested materials (P25, HT and FeHT). The obtained samples were thermostat incubated for 24 hours. After incubation, the samples were centrifuged and the supernatants were discarded. Volumes of 1 ml of fluorescein sodium salt solution were added to each tube and vortexed. All samples were thermostat incubated for 30 minutes. The samples were centrifuged again and volumes of 200 µl of clear supernatant were collected, their fluorescence being measured and compared to the fluorescence of the untreated fluorescein solution.

#### *Data analysis and representation*

Data statistical analysis and graphics were made using the SigmaPlot-11 software package. Depending on data normality, either one-way ANOVA or one-way ANOVA on ranks tests were performed, together with the *Student–Newman–Keuls* (SNK) post hoc test. Events with *p*-values < 0.05 were considered significant. Samples statistically different from controls were marked on figures with a (\*).

## Results

#### *Materials characteristics*

The structural and physicochemical characteristics of the three types of TiO<sub>2</sub> nanoparticles used in the present study were described and

discussed in detail in a previous paper [30]. Briefly, all three types of TiO<sub>2</sub> nanomaterials had average particle sizes between 10-30 nm (determined by TEM; HT and FeHT were smaller than Degussa P25) and similar shapes (no needle shaped particles). While the hydrothermal (HT and FeHT) samples had anatase structure, Degussa P25 TiO<sub>2</sub> was a mixture of anatase and rutile phases with anatase/rutile weight ratio of 85:15 (%). Specific surface areas (determined by BET nitrogen adsorption method) were 49 m<sup>2</sup>/g for Degussa P25, 130.62 m<sup>2</sup>/g for HT and 114.81 m<sup>2</sup>/g for FeHT.

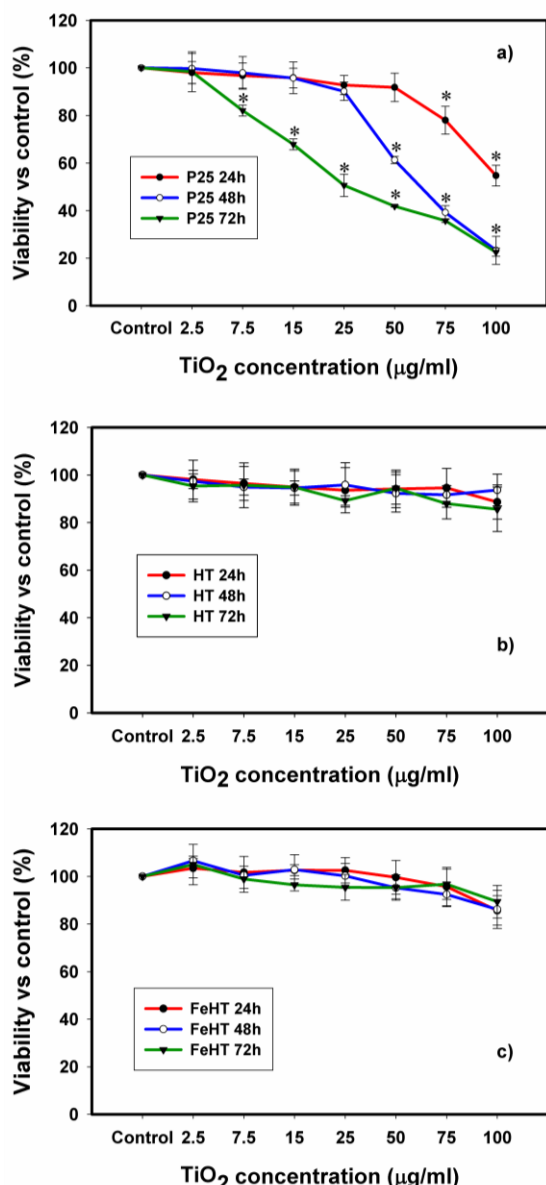


Figure 1: Viability of RAW 264.7 cells after 24, 48 and 72 h of treatment with: (a) Degussa P25, (b) HT and (c) FeHT; (\*) – significant differences with respect to control ( $p < 0.05$ )

All three types of nanomaterials acquired similar amounts of negative surface charge (Zeta potential) when dispersed in DMEM culture medium. The hydrophilic Degussa P25 had significantly higher colloidal stability in aqueous suspension compared to

HT and FeHT, this observation suggesting the hydrophobic character of the hydrothermal materials [31]. Fetal bovine serum (FBS) proteins had strong steric stabilization effects on P25 nanoparticles suspended in cell culture media, only insignificant such effects being observed for FeHT and HT. The semiconductor energy band gaps ( $E_g$ ) of Degussa P25 and HT were both around 3 eV and the iron-doped sample (FeHT) had  $E_g = 2.848$  eV.

### Cell viability

Important viability decreases (between 11-78 %) were observed among the cells exposed to P25 nanoparticles (Fig. 1). The cell killing effect of this material was most prominent after 72 hours of treatment. In this case, significant and concentration-dependent (direct proportionality) viability reductions were obtained for TiO<sub>2</sub> concentrations higher than 7.5 µg/mL (Fig. 1a). For shorter treatment times (48 h and 24 h), significant viability reductions were observed, only for TiO<sub>2</sub> concentrations higher than 25 µg/mL in the 48 h experiment and 50 µg/mL in the 24 h experiment. The hydrothermal TiO<sub>2</sub> samples (HT and FeHT) showed similar behavior, inducing no significant cell killing effects within the concentration range used in this study (Fig. 1b-c).

The occurrence of photocatalytic interferences between TiO<sub>2</sub> and MTT [32, 33] was tested for all nanomaterials and only weak effects were observed in the case of Degussa P25 (data not shown). This aspect was considered in data analysis in a manner similar to that described by Lupu and Popescu [32].

The cytotoxic effect of Degussa P25 TiO<sub>2</sub> was drastically reduced in the presence of the ROS (especially HO·) scavenger. In this case, cell viability was only slightly decreased (with 11-15 % with respect to untreated cells) (Fig. 2). Mannitol alone did not significantly influence cell viability compared with negative control (untreated cells) (data not shown).

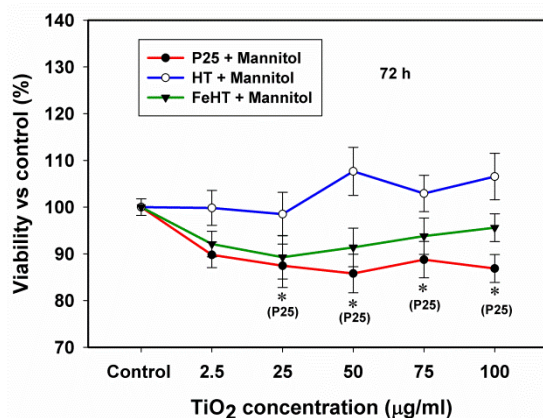


Figure 2: Viability of RAW 264.7 cells after 72 h of co-treatment with TiO<sub>2</sub> and Mannitol; (\*) – significant differences with respect to control ( $p < 0.05$ )

### Intracellular ROS production

The intracellular ROS production induced in macrophage cells treated with P25, HT and FeHT nanoparticles is displayed in Fig. 3a. The determined ROS production varied between 109-147% (with respect to control) and was directly proportional to the used TiO<sub>2</sub> concentration in all cases. All the tested nanomaterials induced significant intracellular ROS generation for concentrations higher than 25µg/ml. The effect of Degussa P25 was stronger compared to that of the hydrothermal samples (Fig. 3a). For concentrations higher than 50 µg/ml, the iron-doped titania induced slightly higher ROS production in the treated cells compared to HT.

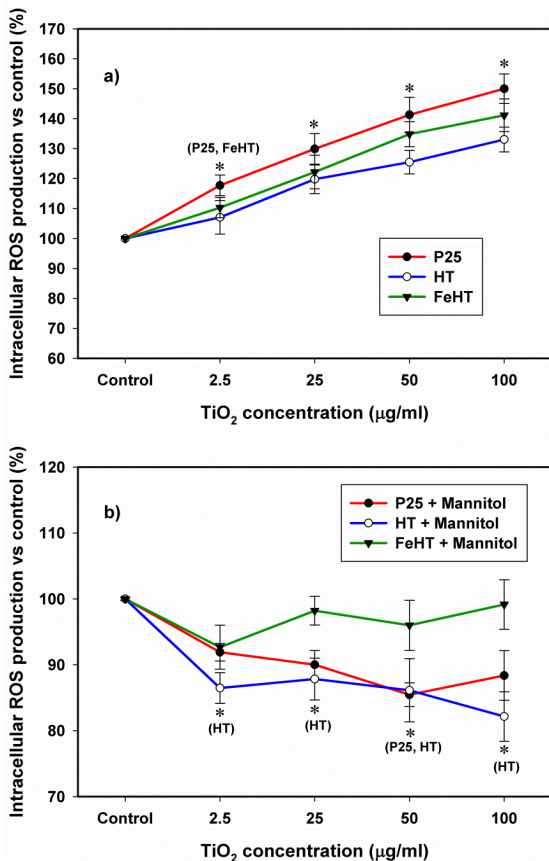


Figure 3: Intracellular ROS production in: a) RAW 264.7 cells treated with Degussa P25, HT and FeHT nanoparticles; b) RAW 264.7 cells co-treated with TiO<sub>2</sub> and Mannitol; (\*) – significant differences with respect to control ( $p < 0.05$ ); the (\*) symbols placed above particular data points refer to all materials unless otherwise specified in round brackets

In the case of co-treated cells (TiO<sub>2</sub> + Mannitol), the ROS scavenger had a strong effect on the intracellular ROS, the determined values being slightly lower compared to control (untreated cells) (Fig. 3b).

### IL-6 release

The concentrations of interleukin-6 (IL-6) released in culture supernatants by TiO<sub>2</sub> stimulated

RAW 264.7 macrophages are illustrated in Fig. 4. The release of the pro-inflammatory cytokine was significantly higher in Degussa P25-treated nanoparticles (increases between 15-200 % with respect to control), in a concentration-dependent (directly proportional) manner, for concentrations higher than 2.5 µg/ml. The hydrothermal nanomaterials proved to be less active in inducing the release of IL-6 by macrophages compared to Degussa P25. The iron-doped TiO<sub>2</sub> (FeHT) generated higher amounts of supernatant IL-6 than the undoped (HT) sample.

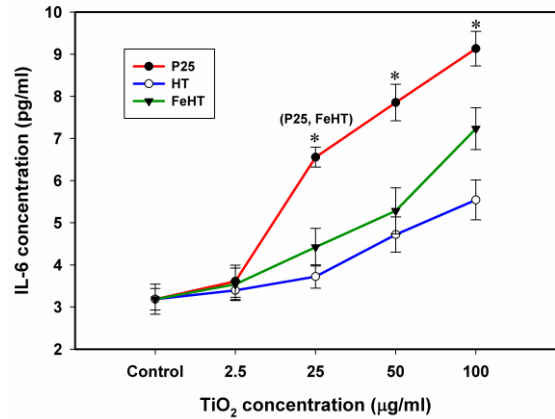


Figure 4: IL-6 release by RAW 264.7 cells treated with Degussa P25, HT and FeHT nanoparticles; (\*) – significant differences with respect to control ( $p < 0.05$ ); the (\*) symbols placed above particular data points refer to all materials unless otherwise specified in round brackets

### Photocatalytic effects under in vitro non-cellular conditions

The results regarding the effect of TiO<sub>2</sub> nanoparticles (P25, HT and FeHT) on the fluorescence of fluorescein sodium salt under *in vitro* non-cellular conditions are shown in Fig. 5.

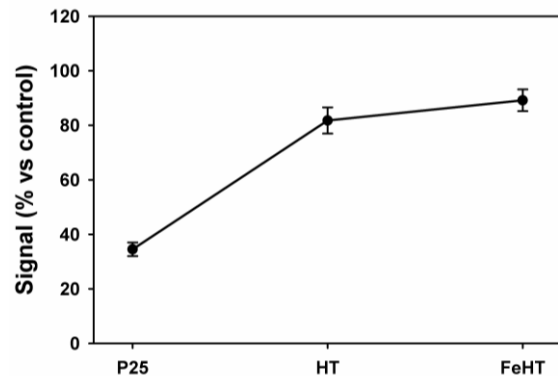


Figure 5: Decrease of fluorescence signal of fluorescein sodium salt induced by Degussa P25, HT and FeHT nanoparticles under non-cellular conditions; the control signal (untreated fluorescein sample) corresponds to 100 %

While severe reductions (on average 65% with respect to control) of the fluorescence signal

were induced by Degussa P25 nanoparticles, only weak effects were observed in the samples containing FeHT or HT. The undoped and iron-doped anatase nanoparticles led to average fluorescence reductions of 18% and 11%, respectively.

These results (qualitatively confirmed under UV irradiation at 312 nm (data not shown)), suggest a considerably higher photocatalytic ROS generation activity of the commercial TiO<sub>2</sub> compared to the hydrothermal samples.

## Discussion

Altogether, the above described results reveal the enhanced biological reactivity (at least with respect to macrophage cells) of Degussa P25 TiO<sub>2</sub> compared to the hydrothermal materials (HT and FeHT).

While P25 titania shows a strong macrophage killing capacity, the HT and FeHT materials appear inert with respect to macrophage viability (Fig. 1). The observed cytotoxic effects reflect specific physicochemical characteristics of the studied TiO<sub>2</sub> nanomaterials that dictate their interactions with macrophage cells and their environment. Compared to the hydrophilic Degussa P25 TiO<sub>2</sub>, the hydrothermal nanomaterials used in the present study have smaller particle sizes, higher curvature, lower surface charge in aqueous environments and similar surface charge in culture medium, as well as hydrophobic surfaces [30]. These characteristics suggest a potentially higher affinity of FeHT and HT nanoparticles for protein adsorption compared to Degussa P25 [34-40], which may contribute to the reduction of toxicity of the hydrothermal TiO<sub>2</sub> nanomaterials.

The severe reduction of cytotoxicity and intracellular ROS production in the cells co-treated with TiO<sub>2</sub> and Mannitol suggests the involvement of a ROS-mediated action mechanism of TiO<sub>2</sub> nanoparticles on RAW 264.7 macrophages.

Hypothetically, one triggering factor for such mechanism could be the capacity of the studied nanomaterials to photogenerate ROS on their surfaces under relevant *in vitro* conditions. Such ROS may induce membrane damage and trigger production of intracellular ROS followed by subsequent cellular processes [41, 42]

Our results clearly showed that Degussa P25 TiO<sub>2</sub> generated significantly higher levels of ROS compared to FeHT and HT under non-cellular conditions (Fig. 5). Although the Fe<sup>3+</sup>-doped TiO<sub>2</sub> had a reduced energy bandgap (resulting in enhanced charge carrier photogeneration under visible light conditions) compared to the other two samples, this

characteristic was not sufficient to confer higher reactivity to this nanomaterial. The reason for this is the dependency of the photocatalytic process on other factors such as charge recombination rates (influenced by the amount and localization of Fe<sup>3+</sup> ions into the host TiO<sub>2</sub> matrix [43]) and specific physicochemical characteristics of the semiconductor-liquid interface.

Despite the known occurrence of TiO<sub>2</sub> induced photocatalytic processes under *in vitro* non-cellular conditions [32, 33] and the observed correlation between the *in vitro* activity of P25 TiO<sub>2</sub> and its non-cellular ROS generation capacity (Fig. 5), the present data is not sufficient to discuss the involvement of a macrophage activation and killing mechanism triggered by photocatalytic processes. This hypothesis remains to be further investigated.

Regarding IL-6 production, many recent studies have identified concentrations of hundreds of pg/ml IL-6 for activated macrophages [44 - 48]. In this context, although we obtained significant differences between treated cells and control in the case of P25, the very low concentrations of IL-6 corresponding to all the samples tested in our study have no biological relevance. Thus, we can't conclude that P25, HT or FeHT influenced the pro-inflammatory IL-6 production.

Summarizing, our study suggests that TiO<sub>2</sub> nanoparticles with hydrophilic surfaces, high surface charge and enhanced photocatalytic properties may preferentially induce macrophage cell activation and damage compared to other TiO<sub>2</sub> nanomaterials. The present findings help in establishing biocompatibility criteria for TiO<sub>2</sub> nanomaterials.

In conclusion, the conducted study has revealed relevant information regarding the generation of intracellular ROS, inflammation and cytotoxicity in macrophage cell cultures exposed to TiO<sub>2</sub> nanoparticles with known physicochemical properties. The undoped and Fe<sup>3+</sup>-doped TiO<sub>2</sub> nanoparticles synthesized by a hydrothermal procedure in our laboratory exhibited reduced pro-oxidative and pro-inflammatory effects and low cytotoxicity compared to the commercial hydrophilic Degussa P25 photocatalyst. The enhanced biocompatibility of the hydrothermal TiO<sub>2</sub> materials was supported by their specific characteristics (particle size, surface curvature, surface charge, hydrophobicity) and reduced capacity to generate ROS under non-cellular conditions. The hydrothermal method can thus be employed to synthesize low toxicity TiO<sub>2</sub> nanomaterials.

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