

[Proceedings of the 3rd Pan American Materials Congress](#)

[Proceedings of the 3rd Pan American Materials Congress](#) pp 179-189| [Cite as](#)

# *Cassia fistula* Leaf-Extract Effect on Corrosion-Inhibition of Stainless-Steel in 0.5 M HCl

- [Authors](#)
- [Authors and affiliations](#)
- Olugbenga Adeshola Omotosho
- Joshua Olusegun Okeniyi
- Cleophas Akintoye Loto
- Abimbola Patricia Idowu Popoola
- Omokolade Babatunde Ajibola
- Adebani Samuel Ogiye

•  
1 .

Conference paper

**First Online:** 10 February 2017

- [2](#) Citations
- [2](#) Readers
- [1.5k](#) Downloads

Part of the [The Minerals, Metals & Materials Series](#) book series (MMMS)

## Abstract

This paper investigates *Cassia fistula* leaf-extract effects on the inhibition of stainless-steel corrosion in 0.5 M HCl. Measurements of corrosion rate were obtained through linear sweep voltammetry (LSV) technique, at the ambient temperature of 28 °C from stainless-steel specimens immersed in the acidic medium, containing different *Cassia fistula* leaf-extract concentrations. Results showed that inhibition effectiveness on stainless-steel corrosion increases with increasing concentration of the leaf-extract. The 10 g/L *Cassia fistula* leaf-extract, the highest concentration of the leaf-extract employed in the study, exhibited optimal inhibition efficiency  $\eta = 88.46\%$  on the corrosion of the stainless-steel metal. Adsorption isotherm modelling shows that the experimental data followed the Flory-Huggins isotherm with excellent model efficiency,  $r^2 = 90.27\%$ , and the Langmuir model with very good model efficiency,  $r^2 = 78.83\%$ . Other isotherm parameters indicate favourable adsorption and suggest physisorption as the prevalent mechanism of corrosion protection by the leaf-extract on stainless-steel in the acidic chloride environment.

## Keywords

Linear sweep voltammetry Corrosion rate *Cassia fistula* leaf-extract Stainless steel Inhibition efficiency Acidic chloride medium Adsorption isotherm modelling

This is a preview of subscription content, [log in](#) to check access.

## References

1. 1.

Finsgar, M., & Jackson, J. (2014). Application of corrosion inhibitors for steels in acidic media for the oil and gas industry: A review. *Corrosion Science*, 86, 17–41. [CrossRefGoogle Scholar](#)

2. 2.

Rajeev, P., Surendranathan, A. O., & Murthy, C. S. N. (2012). Corrosion mitigation of the oil well steels using organic inhibitors-A review. *Journal of Material and Environmental Science*, 3, 856–869. [Google Scholar](#)

3. 3.

Omotosho, O. A., Ajayi, O. O., Fayomi, O. S., & Ifepe, V. O. (2011). Assessing the deterioration behavior of mild steel in 2 M sulphuric acid using *Bambusa glauscescens*. *International Journal of Applied Engineering Research Dindigul*, 2, 85–97. [Google Scholar](#)

4. 4.

Okeniyi, J. O., Popoola, A. P. I., Loto, C. A., Omotosho, O. A., Okpala, S. O., & Ambrose, I. J. (2015). Effect of  $\text{NaNO}_2$  and  $\text{C}_6\text{H}_{15}\text{NO}_3$  synergistic admixtures on steel-rebar corrosion in concrete immersed in aggressive environments. *Advances in Materials Science and Engineering*, 2015, 11 p. [Google Scholar](#)

5. 5.

Okeniyi, J. O., Ambrose, I. J., Oladele, I. O., Loto, C. A., & Popoola, A. P. I. (2013). Electrochemical performance of sodium dichromate partial replacement models by triethanolamine admixtures on steel-rebar corrosion in concretes. *International Journal of Electrochemical Science*, 8, 10758–10771. [Google Scholar](#)

6. 6.

Okeniyi, J. O., Omoniyi, O. M., Okpala, S. O., Loto, C. A., & Popoola, A. P. I. (2013). Effect of ethylenediaminetetraacetic disodium dihydrate and sodium nitrite admixtures on steel-rebar corrosion in concrete. *European Journal of Environmental and Civil Engineering*, 17, 398–416. [CrossRefGoogle Scholar](#)

7. 7.

Scendo, M., & Trela, J. (2013). Adenine as an effective corrosion inhibitor for stainless steel in chloride solution. *International Journal of Electrochemical Science*, 8, 9201–9221. [Google Scholar](#)

8. 8.

Omotosho, O. A., & Ajayi, O. O. (2012). Investigating the acid failure of aluminium alloy in 2 M hydrochloric acid using *Vernonia amygdalina*. *ITB Journal of Engineering Science*, 44, 77–92. [CrossRefGoogle Scholar](#)

9. 9.

Omotosho, O. A., Ajayi, O. O., Ajanaku, K. O., & Ifepe, V. O. (2012). Environment induced failure of mild steel in 2 M sulphuric acid using *Chromolaena odorata*. *Journal of Materials and Environmental Science*, 3, 66–75. [Google Scholar](#)

10.10.

Omotosho, O. A., Ajayi, O. O., Fayomi, O., & Ifepe, V. O. (2010). Evaluating the deterioration behaviour of mild steel in 2 M sulphuric acid in the presence of *Butyrospermum parkii*. *Asian Journal of Applied Science*, 5, 74–84. [CrossRefGoogle Scholar](#)

11.11.

Obot, I. B., & Obi-Egbedi, N. O. (2010). An interesting and efficient green corrosion inhibitor for aluminium from extracts of *Chromolaena odorata* L. in acidic solution. *Journal of Electrochemistry*, 40, 1977–1984. [CrossRefGoogle Scholar](#)

12.12.

Lebrini, M., Robert, F., & Roos, C. (2010). Inhibition effect of alkaloids extract from *annona squamosa* plant on the corrosion of C38 steel in normal hydrochloric acid medium. *International Journal of Electrochemical Science*, 5, 1678–1712. [Google Scholar](#)

13.13.

Raja, P. B., Rahim, A. A., Osman, H., & Awang, K. (2010). Inhibitory effect of *Kopsia singapurensis* extract on the corrosion behavior of mild steel in acid media. *Wuli Huaxue Xuebao/Acta Physico-Chimica Sinica*, 26, 2171–2176. [Google Scholar](#)

14.14.

Kasthuri, P. K., & Arulanantham, A. (2010). Eco-friendly extract of *Euphorbia hirta* as corrosion inhibitor on mild steel in sulphuric acid medium. *Asian Journal of Chemistry*, 22, 430–434. [Google Scholar](#)

15.15.

Rajalakshmi, R., Subhashini, S., Leelavathi, S., & Mary, R. F. (2008). Efficacy of sprouted seed extracts of *Phaseolus aureus* on the corrosion inhibition of mild steel in 1 M HCl. *Oriental Journal of Chemistry*, 24, 1085–1090. [Google Scholar](#)

16.16.

Okeniyi, J. O., Loto, C. A., Popoola, A. P. I., & Omotosho, O. A. (2015). Performance of *Rhizophora mangle* L. leaf-extract and sodium dichromate synergies on steel-reinforcement corrosion in 0.5 M H<sub>2</sub>SO<sub>4</sub>-immersed concrete. In: *Corrosion 2015 Conference & Expo*. Houston, TX: NACE International, Paper No. 5636. [Google Scholar](#)

17.17.

Omotosho, O. A., Ajayi, O. O., Fayomi, O., & Yussuff, O. (2012). Degradation evaluation of zinc in 2 M hydrochloric acid in the presence of *Bambusa bambos*. *Singapore Journal of Scientific Research*, 2, 14–24. [CrossRefGoogle Scholar](#)

18.18.

Al-Turkustani, A. M. (2010). Effect of *Ajowan* seeds as safe inhibitor on the corrosion of steel in 2.0 M sulfuric acid. *Modern Applied Science*, 4, 52–61. [Google Scholar](#)

19.19.

Ating, E. I., Umoren, S. A., Udousoro, I. I., Ebenso, E. E., & Udoh, A. P. (2010). Leaves extract of *Ananas sativum* as green corrosion inhibitor for aluminium in hydrochloric acid solutions. *Green Chemistry Letters and Reviews*, 3, 61–68. [CrossRefGoogle Scholar](#)

20.20.

Gunasekaran, G., & Chauhan, L. R. (2007). Corrosion inhibition of mild steel by plant extract in dilute HCl medium. *Corrosion Science*, 49, 1143–1161. [CrossRefGoogle Scholar](#)

21.21.

Jain, T., Chowdhary, R., Arora, P., & Mathur, S. P. (2005). Corrosion inhibition of aluminum in hydrochloric acid solutions by peepal (*Ficus religiosa*) extracts. *Bulletin of Electrochemistry*, 21, 23–27. [Google Scholar](#)

22.22.

Omotosho, O. A., Okeniyi, J. O., Obi, E. I., Sonoiki, O. O., Oladipupo, S. I., & Oshin, T. M. (2016). Inhibition of Stainless Steel Corrosion in 0.5 M H<sub>2</sub>SO<sub>4</sub> in the presence of C<sub>6</sub>H<sub>5</sub>NH<sub>2</sub>. In: *TMS2016: 145 annual meeting & exhibition: Supplemental proceedings*(pp. 465–472). Cham, Switzerland: Springer International Publishing AG. doi: [10.1007/978-3-319-48254-5\\_56](https://doi.org/10.1007/978-3-319-48254-5_56).

23.23.

Herle, R., Shetty, P., Shetty, S. D., & Kini, U. A. (2011). Corrosion inhibition of 304SS in hydrochloric acid solution by N-Furfuryl N'-Phenyl Thiourea. *Portugaliae Electrochimica Acta*, 29, 69–78. [Google Scholar](#)

24.24.

Satpati, A. K., & Ravindran, P. V. (2008). Electrochemical study of the inhibition of corrosion of stainless steel by 1,2,3-benzotriazole in acidic media. *Materials Chemistry and Physics*, 109, 352–359. [Google Scholar](#)

25.25.

Silva, A. B., Agostinho, S. M. L., Barcia, O. E., Cordeiro, G. G. O., & D'Elia, E. (2006). The effect of cysteine on the corrosion of 304L stainless steel in sulphuric acid. *Journal of Corrosion Science*, 48, 3668–3674. [CrossRefGoogle Scholar](#)

26.26.

Galal, A., Atta, N. F., & Al-Hassan, M. H. S. (2005). Effect of some thiophene derivatives on the electrochemical behavior of AISI 316 austenitic stainless steel in acidic solutions containing chloride ions II, effect of temperature and surface studies. *Materials Chemistry and Physics*, 89, 28–37. [CrossRefGoogle Scholar](#)

27.27.

Refaey, S. A. M., Taha, F., & Abd El-Malak, A. M. (2004). Inhibition of stainless steel pitting corrosion in acidic medium by 2-mercaptobenzoxazole. *Applied Surface Science*, 236, 175–185. [Google Scholar](#)

28.28.

Abdallah, M. (2003). Corrosion behavior of 304 stainless steel in sulphuric acid solutions and its inhibition by some substituted pyrazolones. *Materials Chemistry and Physics*, 82, 786–792. [CrossRefGoogle Scholar](#)

29.29.

Abdallah, M. (2002). Rhodanine azosulpha drugs as corrosion inhibitors for corrosion of 304 stainless steel in hydrochloric acid solution. *Journal of Corrosion Science*, 44, 717–728. [CrossRefGoogle Scholar](#)

30.30.

Olsson, C. O. A., & Landolt, D. (2003). Passive films on stainless steels—chemistry, structure and growth. *Electrochimica Acta*, 48, 1093–1104. [CrossRefGoogle Scholar](#)

31.31.

Bera, S., Rangarajan, S., & Narasimhan, S. V. (2002). Electrochemical passivation of iron alloys and the film characterization by XPS. *Corrosion Science*, 42, 1709–1724. [CrossRefGoogle Scholar](#)

32.32.

Bahorun, T., Neergheen, V. S., & Aruoma, O. I. (2005). Phytochemical constituents of *Cassia fistula*. *African Journal of Biotechnology*, 4, 1530–1540. [Google Scholar](#)

33.33.

ASTM D2688-94 R99. (2005). *Standard test methods for corrosivity of water in the absence of heat transfer (weight loss methods)*. West Conshohocken: ASTM International, PA. [Google Scholar](#)

34.34.

Okeniyi, J. O., Loto, C. A., & Popoola, A. P. I. (2014). Electrochemical performance of *Anthocleista djalonensis* on steel-reinforcement corrosion in concrete immersed in saline/marine simulating-environment. *Transactions of the Indian Institute of Metals*, 67, 959–969. [CrossRefGoogle Scholar](#)

35.35.

Okeniyi, J. O., Loto, C. A., & Popoola, A. P. I. (2014). Electrochemical performance of *Phyllanthus muellerianus* on the corrosion of concrete steel-reinforcement in industrial/microbial simulating-environment. *Portugaliae Electrochimica Acta*, 32, 199–211. [CrossRefGoogle Scholar](#)

36.36.

Hameurlaine, S., Gherraf, N., Benmnine, A., & Zellagui, A. (2010). Inhibition effect of methanolic extract of *Atractylis serratuloides* on the corrosion of mild steel in H<sub>2</sub>SO<sub>4</sub> medium. *Journal of Chemical and Pharmaceutical Research*, 2, 819–825. [Google Scholar](#)

37.37.

Okeniyi, J. O., Oladele, I. O., Omoniyi, O. M., Loto, C. A., & Popoola, A. P. I. (2015). Inhibition and compressive-strength performance of Na<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> and C<sub>10</sub>H<sub>14</sub>N<sub>2</sub>Na<sub>2</sub>O<sub>8</sub>·2H<sub>2</sub>O in steel-reinforced concrete in corrosive environments. *Canadian Journal of Civil Engineering*, 42, 408–416. [CrossRefGoogle Scholar](#)

38.38.

Okeniyi, J. O., Oladele, I. O., Ambrose, I. J., Okpala, S. O., Omoniyi, O. M., Loto, C. A., et al. (2013). Analysis of inhibition of concrete steel-rebar corrosion by Na<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> concentrations: Implications for conflicting reports on inhibitor effectiveness. *Journal of Central South University*, 20, 3697–3714. [CrossRefGoogle Scholar](#)

39.39.

Canmet, S. P. (2008). Electrochemical polarization techniques for corrosion monitoring. In: L. Yang (Ed.), *Techniques for corrosion monitoring* (pp. 49–85). Cambridge: Woodhead Publishing Limited. [Google Scholar](#)

40.40.

Okeniyi, J. O., Loto, C. A., & Popoola, A. P. I. (2015). Corrosion inhibition of concrete steel-reinforcement in saline/marine simulating-environment by *Rhizophora mangle* L. *Solid State Phenomena*, 227, 185–189. [CrossRefGoogle Scholar](#)

41.41.

Okeniyi, J. O., Loto, C. A., & Popoola, A. P. I. (2014). *Morinda lucida* effects on steel-reinforced concrete in 3.5% NaCl: Implications for corrosion-protection of wind-energy structures in saline/marine environments. *Energy Procedia*, 50, 421–428. [CrossRefGoogle Scholar](#)

42.42.

Kumar, S., Ladha, D. G., Jha, P. C., & Shah, N. K. (2013). Theoretical study of Chloro-N-(4-methoxybenzylidene)aniline derivatives as corrosion inhibitors for zinc in hydrochloric acid. *International Journal of Corrosion*, 2013, 10 p. [Google Scholar](#)

43.43.

Okeniyi, J. O., Loto, C. A., & Popoola, A. P. I. (2015). Evaluation and analyses of *Rhizophora mangle* L. leaf-extract corrosion-mechanism on reinforcing steel in concrete immersed in industrial/microbial simulating-environment. *Journal of Applied Sciences*, 15, 1083–1092. [CrossRefGoogle Scholar](#)

44.44.

Anejjar, A., Salghi, R., Zarrouk, A., Benali, O., Zarrok, H., Hammouti, B., et al. (2014). Inhibition of carbon steel corrosion in 1 M HCl medium by potassium thiocyanate. *Journal of the Association of Arab Universities for Basic and Applied Sciences*, 15, 21–27. [CrossRefGoogle Scholar](#)

45.45.

Okeniyi, J. O. (2016).  $C_{10}H_{18}N_2Na_2O_{10}$  inhibition and adsorption mechanism on concrete steel-reinforcement corrosion in corrosive environments. *Journal of the Association of Arab Universities for Basic and Applied Sciences*, 20, 39–48. [CrossRefGoogle Scholar](#)

46.46.

Foo, K. Y., & Hameed, B. H. (2010). Insights into the modeling of adsorption isotherm systems. *Chemical Engineering Journal*, 156, 2–10. [CrossRefGoogle Scholar](#)

47.47.

Vijayaraghavan, K., Padmesh, T. V. N., Palanivelu, K., & Velan, M. (2006). Biosorption of nickel(II) ions onto *Sargassum wightii*: Application of two-

parameter and three-parameter isotherm models. *Journal of Hazardous Materials, B133*, 304–308. [CrossRefGoogle Scholar](#)

48.48.

Okeniyi, J. O., Loto, C. A., & Popoola, A. P. I. (2015). Investigating the corrosion mechanism of *Morinda lucida* leaf extract admixtures on concrete steel rebar in saline/marine simulating environment. *International Journal of Electrochemical Science, 10*, 9893–9906. [Google Scholar](#)

49.49.

Tao, Z. H., Zhang, S. T., Li, W. H., & Hou, B. R. (2009). Corrosion inhibition of mild steel in acidic solution by some oxo-triazole derivatives. *Corrosion Science, 51*, 2588–2595. [CrossRefGoogle Scholar](#)

50.50.

Ferreira, E. S., Giacomelli, C. F., Gicomelli, F. C., & Spinelli, A. (2004). Evaluation of the inhibitor effect of l-ascorbic acid on the corrosion of mild steel. *Materials Chemistry and Physics, 83*, 129–134. [CrossRefGoogle Scholar](#)

51.51.

Okeniyi, J. O., Ambrose, I. J., Okpala, S. O., Omoniyi, O. M., Oladele, I. O., Loto, C. A., et al. (2014). Probability density fittings of corrosion test-data: Implications on  $C_6H_{15}NO_3$  effectiveness on concrete steel-rebar corrosion. *Sadhana, 39*, 731–764. [CrossRefGoogle Scholar](#)

52.52.

Coffey, R., Dorai-Raj, S., O'Flaherty, V., Cormican, M., & Cummins, E. (2013). Modelling of pathogen indicator organisms in a small-scale agricultural catchment using SWAT. *Human and Ecological Risk Assessment: An International Journal, 19*, 232–253. [CrossRefGoogle Scholar](#)

## Copyright information

© The Minerals, Metals & Materials Society 2017