Mutual interaction of phenolic compounds and microbiota: Metabolism of complex phenolic apigenin *C*- and kaempferol *O*-derivatives by human fecal samples

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16 **ABSTRACT**

Human colonic bacteria have an important impact on the biotransformation of 17 flavonoid glycosides and their conversion can result in the formation of bioactive 18 compounds. However, information about the microbial conversion of complex 19 glycosylated flavonoids and the impact on the gut microbiota are still limited. In this 20 study, in vitro fermentations with selected flavonoid O- and C-glycosides and three 21 different fecal samples were performed. As a result, all flavonoid glycosides were 22 metabolized via their adjycones vielding smaller substances. Main metabolites were 23 3-(4-hydroxyphenyl)propionic acid, 3-phenylpropionic acid, and phenylacetic acid. 24 Differences in the metabolite formation due to different time courses between the 25 donors were determined. Therefore, from all fermentations, the ones with a specific 26 donor were always slower resulting in a lower number of metabolites compared to the 27 others. Exemplarily, tiliroside was totally degraded from 0h (105 \pm 13.2 μ M) within the 28 29 first 24h, while in the fermentations with fecal samples from other donors, tiliroside $(107 \pm 52.7 \mu M \text{ at 0h})$ was not detected after 7h anymore. In general, fermentation 30 rates of C-glycosides were slower compared to the fermentation rates of O-31 glycosides. The O-glycoside tiliroside was degraded within 4h while the gut 32 microbiota converted the C-glycoside vitexin within 13h. However, significant 33 changes (p < 0.05) in the microbiota composition and short chain fatty acid levels as 34 products of carbohydrate fermentation were not detected between incubations with 35 different phenolic compounds. Therefore, microbiota diversity was not affected and a 36 significant prebiotic effect of phenolic compounds cannot be assigned to flavonoid 37 glycosides in food-relevant concentrations. 38

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41 **KEYWORDS**

- 42 Flavonoid glycosides, metabolism, *in vitro* fermentation, human gut microbiota, short
- 43 chain fatty acids

45 **INTRODUCTION**

Flavonoids are a class of secondary plant metabolites which are found ubiquitously in 46 the plant kingdom. Their chemical structure is characterized by two aromatic rings 47 with at least one hydroxyl group.¹ Flavonoids can be classified into several 48 subgroups depending on the constitution of the heterocyclic C-ring.² The chemical 49 structures range from very simple substances to guite complex flavonoids consisting 50 of several phenolic rings with a wide range of substituents. In plants, they are 51 frequently conjugated with different small organic molecules (preferentially sugars 52 and organic acids), affecting their water-solubility and functionality.³ Only 53 occasionally they are present in plants as non-conjugated flavonoid aglycones.¹ 54 Beside the very frequent O-glycosylation of flavonoids, flavonoid glycosides can also 55 occur as C-glycosides.⁴ The most common sources of flavonoid C-glycosides are 56 swiss chard, tomato, lemons, and some cereals such as maize, wheat, and rice.^{5, 6} 57

When consumed with plant foods, flavonoid glycosides are hypothesized to have 58 health beneficial properties.⁷ Antimutagenic, anticarcinogenic, antiviral, antibacterial, 59 and antiinflammatory properties have been described.⁸⁻¹¹ Even though studies have 60 mainly focused on the more common O-glycosides, it is assumed that C-glycosides 61 would have better therapeutic properties due to their enhanced stability over the 62 respective aglycone and O-glycosides.¹² But especially for C-glycosides, more 63 investigations on resorption, metabolism, and health beneficial effects have to be 64 carried out.¹² However, uptake/bioavailability of flavonoids is a crucial factor for 65 potential bioactivities but this is still discussed controversially.¹³ It is highly dependent 66 on the chemical structure and can therefore differ between different compounds. 67 Moreover, bioactivity is influenced by the resorption rate, intestinal, and hepatic 68 metabolism, and the subsequent distribution in the organism.¹⁴ Although different 69

pathways have been identified as being responsible for an enhanced resorption, it has to be concluded that in most cases resorption rates of intact flavonoids are not very high.^{15, 16} Consequently, the most important metabolic transformation of flavonoid glycosides takes place in the colon, where they undergo significant degradation by the gut microbiota.

75 In the human colon more than 500 different bacterial species are found, whereby the composition of the microbiome is highly diverse and unique for every individual.^{7, 17} 76 Some of the bacterial species can catalyze (flavonoid) O- and C-deglycosylation, 77 demethylation, dehydroxylation, ester cleavage, reduction of carbon-carbon double 78 bond, isomerization, ring fission, and decarboxylation.¹⁸ However, not all bacteria are 79 able to carry out every degradation step.¹⁹ For example, *Enterococcus casseliflavus* 80 is only able to cleave and ferment the sugar moiety from different quercetin-O-81 glycosides to short chain fatty acids (SCFA), while the aglycone is not degraded any 82 further. In contrast, Eubacterium ramulus is capable of converting also the 83 aglycone.²⁰ Additionally, it is possible that some kind of cross-feeding effects due to 84 the degradation of phenolic compounds occur.²¹ It means that one strain gives good 85 growth on the primary substrate such as the phenolic compound, which is 86 metabolized to a product being a secondary substrate for another strain. This might 87 enhance growth of further microorganisms and consequently influence composition 88 of the colonic microbiota.22 89

Although the structures of *O*- and *C*-glycosides are very different from a chemical point of view, the conversion of *C*-glucosides also showed a dependence on the presence of specific bacteria for degradation steps such as the cleavage of the C-C bond. For example, *Eubacterium cellulosolvens* is not able to deglycosylate the *C*-

glycosides vitexin, while it is degradable by the intestinal *Lachnospiraceae* strain
 CG19-1.²³

When looking at fermentation experiments with fecal samples, it is possible to 96 investigate at least two ways of action: On the one hand, the microbial transformation 97 of flavonoids to bioactive products is of interest, whereas on the other hand, the 98 influence of phenolic compounds and their metabolites on the composition of the gut 99 microbiota is an important aspect, too (two-way phenolic - microbiota interaction).²⁴ In 100 general, combining metagenomics and metabolomics studies will help to better 101 understand both types of interactions.²⁵ However, information about the microbial 102 transformation of more complex, highly glycosylated flavonoids, and evidence for 103 their effect on the gut microbiota is still not sufficiently investigated. In some studies, 104 it has been suggested that more complex polyphenols have an even higher effect on 105 the microbiome than simple structures, because a larger variety of potentially 106 bioactive metabolites can be formed.²⁶ Therefore, such structures might also 107 influence the diversity of the microbiota which is hypothesized to correlate with the 108 health of an individual.²⁷ 109

The aim of this study was to investigate the degradation of selected structurally 110 related, highly glycosylated flavonoids and their influence on gut microbiota 111 composition. Special focus was set on the comparative assessment of group specific 112 differences between O- and C-glycosides. For this purpose, different in vitro 113 fermentations with a limited number of human fecal samples were performed. 114 Breakdown products were analyzed using HPLC-ESI-MS/MS analysis. To assess 115 116 whether phenolic exposure influenced the microbiota, gPCR and Illumina sequencing were carried out to investigate potential compositional changes and SCFA production 117 was determined to detect major functional changes. 118

119 MATERIAL AND METHODS

Chemicals. Apigenin (AP, 4',5,7-trihydroxyflavone), kaempferol (K, 3,4',5,7-120 tetrahydroxyflavone), kaempferol-3,4'-O-diglucoside-7-O-rhamnoside (K-DG-R), 121 tiliroside (T, kaempferol-3-(6"-trans-p-coumaroyl)glucoside), vitexin (V, apigenin-8-C-122 glucoside), and vitexin-2"-O-rhamnoside (V-R) were purchased from Phytolab GmbH 123 & Co. KG (Vestenbergsgreuth, Germany). Vitexin-2"-O-glucoside (V-G), and vitexin-124 2"-O-xyloside (V-X) were isolated from swiss chard. The isolation method is 125 described in the supplementary information. 126

127 The chemical structures of the substrates used in this study are shown in Figure 1. All O-glycosides are based on the aglycone structure of kaempferol, whereas the 128 structures of the C-glycosides are related to the aglycone apigenin. Both aglycones 129 130 differ only in a hydroxyl group at C3-position of the basic skeleton. While tiliroside has only one sugar moiety being bound via a C-O bond at the C3-position and further 131 esterification with p-coumaric acid, kaempferol-3,4'-O-diglucoside-7-O-rhamnoside is 132 O-glycosylated with three sugar moieties at different positions. In contrast, vitexin is 133 the simplest C-glucoside of apigenin, bound via a C-C bond at the C8-position. The 134 different vitexin derivatives have a second sugar moiety at the C2"-position. These 135 bonds are C-O bonds. Consequently, the vitexin derivatives combine O- and C-136 137 glycosidic bonds in one structure.

In vitro degradation of phenolic substrates by fecal microbiota. This study was part of a set of experiments already described by Vollmer, et al. ²⁸ in which three independent *in vitro* fermentations with fecal microbiota were performed. Fecal samples were donated by three different, healthy donors (donor A, B, and C) without a history of gastrointestinal disorders or any antibiotics consumption for at least three month prior to the fermentation experiment. Number of samples and fermentation

strategy were similar to related fermentation experiments already described in the 144 literature.^{21, 29-31} The preparation of the fecal suspension was conducted according to 145 Vollmer, Schröter, Esders, Farguharson, Neugart, Duncan, Schreiner, Louis, Maul 146 and Rohn²⁸. In vitro fermentations were performed with eight different flavonoids in 147 triplicate at an initial pH value of 6.5 in a final volume of 10 mL. Based on preliminary 148 experiments with similar levels of carbon source(s) it was estimated that during 149 fermentation the pH drops by 0.5-1 units. The volume of 10 mL consisted of 9.4 mL 150 fermentation medium which contained several minerals, supplements, and 151 carbohydrates to allow the gut bacteria to grow, 100 µL phenolic substrate (final 152 concentration 200 µM, pre-dissolved in DMSO), 14 µL vitamin solution, and 500 µL 153 freshly prepared fecal suspension (0.2% final fecal concentration). The preparation 154 and ingredients of the fermentation media and vitamin solution are described 155 elsewhere.28 156

157 After adding the fecal suspension to the medium, the samples were incubated on a rotator (Stuart SB3, Bibby Scientific, Stone, UK) for 48 h at 37 °C and 25 rpm. Two 158 aliquots of 750 µL were collected initially (0h), and after 3 h, 7 h, 24 h, and 48 h while 159 flushing with CO₂. One aliquot was frozen with liquid nitrogen and stored at -80 °C 160 until HPLC-ESI-MS/MS analysis. The second aliquot was used for investigations of 161 the microbiota and SCFA analyses. For that, the aliquot was separated into 162 supernatant (SCFA analysis) and cell pellet (DNA analysis) by centrifuging.28 163 Supernatant was stored at -20 °C until SCFA analysis and cell pellet dissolved in 164 buffer at -80 °C. Incubation parameters of fecal fermentation and sample treatments 165 were already described by Vollmer, Schröter, Esders, Farguharson, Neugart, 166 Duncan, Schreiner, Louis, Maul and Rohn²⁸. 167

In addition to fecal suspensions with phenolic substrates, control samples for every fermentation without any substrates were conducted. This was done to determine the metabolite formation caused by compounds endogenously present in the fecal samples. For that, 100 μ L DMSO were used instead of the substrate ('control samples I'). For the estimation of the interactions of the substrates with the medium, 100 μ L substrate were combined with 9.9 mL medium and 14 μ L vitamin solution and were treated the same way as the samples ('control samples II').

The fermentation period in the present study was set to 48 h based on other *in vitro* fermentation experiments on phenolic compounds described in the literature. ^{19, 32-34} Some of the fermentation experiments ended already after 24 h, but there, mainly simple phenolic substances were investigated. Due to the assumption that the microorganisms need longer for the degradation process of complex phenolic compounds the fermentation period was extended to 48 h.

Sample preparation and HPLC-ESI-MS/MS analysis for quantification. The substrates and their metabolites were extracted with ethyl acetate from 400 μ L of the fermentation mixture from every time point. The sample preparation and extraction were carried out according to Vollmer, Schröter, Esders, Farquharson, Neugart, Duncan, Schreiner, Louis, Maul and Rohn²⁸.

HPLC analyses of the substrates and their metabolites were performed using a 1260
Infinity Series system from Agilent Technologies Deutschland GmbH & Co. KG
(Waldbronn, Germany). The system consisted of a binary pump, an online-degasser,
an autosampler, and a column oven. The separation of the substances and their
metabolites was carried out using a Kinetex 5 u EVO C18 100 A (150 x 2.1 mm)
column equipped with an EVO C18 pre-column (both from Phenomenex Inc.,
Aschaffenburg, Germany). The chromatographic separation took place at 20 °C. The

mobile phase was 0.1% formic acid in water (A) and 0.1% formic acid in acetonitrile (B) and the following gradient elution was used: 5% B (0-1 min), 11% B (1-2 min), 14% B (2-4 min), 50% B (4-15 min), 95% B (15-18 min), 95% B (18-22 min), 5% B (22-23 min) and 5% B (23-32 min). The flow rate was set to 300 μ L/min and the injection volume was 5 μ L.

For detecting the substrates and their metabolites, mass analyses were performed in 198 negative ion mode on an API2000 triple quadrupole MS/MS system (AB Sciex 199 Germany GmbH, Darmstadt, Germany) equipped with an ESI interface. The following 200 mass spectrometer settings were used: electrospray voltage = -4500 V, temperature 201 = 450 °C, curtain gas = 1.7 bar, ion source gas 1 = 2.1 bar, ion source gas 2 = 5.2 202 bar and collisions gas = 0.5 bar (all nitrogen). For each substance, the optimum 203 settings of the declustering potential, entrance potential, collision energy, collision 204 exit potential, and the characteristic fragments were also determined. 205

For quantification, an external matrix calibration (0-200 μ M) for every donor was used (**Supplementary Table 1**). The matrix consisted of the fermentation medium and the fecal sample. The data obtained were analyzed with the software Analyst[®] 1.5.2 from AB Sciex Germany GmbH (Darmstadt, Germany).

Beside the substrates used, the following metabolites were included in the method: 210 benzoic acid, 4-hydroxybenzoic acid, caffeic acid, p-coumaric acid (p-CA), 3-(3,4-211 dihydroxyphenyl)propionic acid, 3-(3,4-dihydroxyphenyl)acetic acid, 3,4-212 dihydroxytoluene, ferulic acid, 3-(3-hydroxyphenyl)propionic acid, 3-(4-213 hydroxyphenyl)propionic acid (4-HPPA), 3-phenylpropionic acid (3-PPA), 3-214 hydroxyphenylacetic acid (PAA), 4-hydroxyphenylacetic acid, phenylacetic acid 215 (PAA), hippuric acid, and kaempferol-3-O-glucoside (K-G). 216

Analysis of short chain fatty acids. Determination of short chain fatty acids (SCFA) 217 including acetate, propionate, butyrate, iso-butyrate, formate, but also of further 218 organic acids such as lactate, succinate, valerate, and iso-valerate was conducted 219 with 500 µL of the supernatants from the 48 h samples and from one 0 h sample per 220 substrate. The sample preparation was based on the method developed by 221 Richardson, et al.³⁵ followed by gas chromatography analysis using a Hewlett-222 Packard gas chromatograph fitted with a fused silica capillary column with helium as 223 a carrier gas. Calculations were done with external standards, 2-ethylbutyrate was 224 used as internal standard. The mean of the SCFA concentration resulting from the 225 fermentations with the different substrates were set in relation to the SCFA 226 concentration from the 'control samples I'. 227

DNA extractions, fluorimetric DNA quantification and gPCR analysis. DNA 228 extractions were done with 400 µL of the fecal suspension from every donor and with 229 the cell pellets from the 48 h samples using the ${\sf FastDNA}^{{\rm \tiny (\!\!R\!)}}$ spin kit for soil (MP 230 Biomedicals, Illkirch, France). Extraction, fluorimetric DNA guantification and gPCR 231 analysis were performed according to the protocol described by Vollmer, Schröter, 232 Esders, Farquharson, Neugart, Duncan, Schreiner, Louis, Maul and Rohn²⁸. 233 Universal primers against total bacteria and specific primers against *Bifidobacterium* 234 spp., Bacteroidetes, Ruminococcaceae, Lachnospiraceae, F. prausnitzii, Blautia spp., 235 and the Roseburia/Eubacterium rectale group, were used.²⁸ 236

Illumina sequencing of 16S rRNA genes. The Illumina sequencing of selected compounds was performed at the ZIEL – Institute for Food & Health of the TU Munich with DNA extracts of the fecal suspension from every donor (initial composition), the 48 h sample of tiliroside from all three donors, and the 48 h samples of vitexin, vitexin-2"-O-rhamnoside and kaempferol-3,4'-O-diglucoside-7-O-

rhamnoside from donor A and C. The method details are described in the supplementary information. 429,958 initial sequence reads were filtered as described in supplementary information, which resulted in 245,816 final reads (11,698-19,887 per sample). Those were assigned to 151 operational taxonomic units (OTUs) at \geq 97% sequence identity (**Supplementary Table 2**). Sequencing data generated during this study are available in the SRA database under SRA accession SRP117249 at http://www.ncbi.nlm.nih.gov/sra/SRP117249.

Statistical analysis. Statistical analysis was performed to investigate whether 249 differences in SCFA levels or in the microbiota composition resulting from the 250 substrate fermentations were significant or not compared to the 'control samples I'. 251 For that, data were analyzed using IBM SPSS Statistics 22 (Ehningen, Germany). All 252 data were tested for normality. With the normal distributed data investigations for 253 significant differences were carried out with an independent t-test. Non-normal 254 distributed values were analyzed with the non-parametric MANN-WHITNEY-U-test. For 255 significance, a confidence level of 95% (p < 0.05) was used. In addition, the qPCR, 256 SCFA, and sequence data were summarized by Principal Component Analysis 257 (PCA). All data were standardized before applying PCA. The analysis was performed 258 in R (R Core Team (2012). R: A language and environment for statistical computing. 259 R Foundation for Statistical Computing, Vienna, Austria. http://www.R-project.org). 260 For correlation analysis between qPCR and sequencing data, OTUs were assigned 261 to the corresponding gPCR assays based on their taxonomy (Supplementary Table 262 **2**). 263

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266 **RESULTS AND DISCUSSION**

In the present study, the degradation of selected, structurally related, highly glycosylated flavonoids were determined in three independent *in vitro* fermentations in order to estimate metabolite formation and changes in the microbiota composition. Here, it was of interest to determine if the gut microbiota is able to convert the more complex flavonoids to lower molecular metabolites, or if they are not or only partially degraded. For comparison and as a control, *in vitro* fermentations with the corresponding aglycones were also performed.

In vitro fermentations of the flavonol kaempferol and its O-glycosides tiliroside and kaempferol-3,4'-O-diglucoside-7-O-rhamnoside.

Kaempferol. When starting the fermentation experiments with the comparatively 276 simple aglycone structure kaempferol (K), there was only a slow degradation by the 277 fecal sample from donor A. K was still present at 48 h and only small amounts of the 278 metabolite 3-(4-hydroxyphenyl)propionic acid (4-HPPA) were detected (Table 1). 279 However, K was fully metabolized by 7 h and 24 h, respectively, with the fecal 280 sample from donors B and C and 4-HPPA was detected at 7 h at its highest 281 concentration in both cases. During the fermentation with the fecal sample from 282 donor B no further metabolites were present after 24 h. By contrast, the metabolite 283 phenylacetic acid (PAA) was detected in the fermentation with the fecal sample from 284 donor C. In the HPLC-ESI-MS/MS method used, the detector sensitivity for the 285 metabolite 3-phenylpropionic acid (3-PPA) and PAA was rather low. It cannot be 286 judged whether e.g., PAA was only formed to lower amounts by donor B unlike for 287 the other two donors, because responsible microorganisms are missing or the 288 amount was simply lower than the limit of detection (LOD) of the method. Therefore, 289 it is possible that the formation of 3-PPA and PAA could not be detected in the 290

fermentation B. Nevertheless, the results obtained are in agreement with results described in the literature, where mainly 4-hydroxyphenylacetic acid, the precursor for PAA, which was also detected in the present study, has been described as an important phenolic metabolite (**Figure 2**).^{7, 36}

Tiliroside. In all three in vitro fermentation experiments, tiliroside (T) was metabolized 295 significantly by the gut microbiota. Figure 3 displays its degradation steps and 296 metabolite formation. The detected metabolites were kaempferol-3-O-glucoside (K-297 G), kaempferol (K), p-coumaric acid (p-CA), 3-(4-hydroxyphenyl)propionic acid (4-298 HPPA), and 3-phenylpropionic acid (3-PPA), whereby the latter was not detected in 299 the fermentation with the fecal sample from donor A. In the fermentations of donor B 300 301 and C, the bacteria already cleaved the ester bond between the sugar moiety and p-CA within short time resulting in the formation of K-G at 0 h, immediately after 302 inoculation with the fecal samples (Table 1, Figure 3). An interaction or breakdown 303 304 due to the fermentation media can be excluded, because K-G was not detected in the 'control samples II' (substrate + medium, data not shown). The detection of K-G 305 leads to the assumption that the microbiota did not mainly first cleave the bond 306 between the kaempferol-moiety and the sugar substituent in T, and T is mostly 307 degraded to its aglycone K via the intermediate product K-G. However, the 308 metabolite K-G was only detected in very low concentrations in fermentation A at 0 h, 309 3 h, and 7 h. Instead, a significant increase of K and p-CA was present within the first 310 7 h. This may be due to the cleavage of the bond between K and glucose and the 311 312 release of *p*-CA taking place at the same time intervals. Subsequently, the aglycone was transformed to smaller phenolic acids deriving from the A- and B-ring during the 313 breakdown of the heterocyclic flavonoid C-ring.²⁵ Furthermore, the microbiota seems 314 315 also to be able to use the free p-CA as a substrateas already described in some

studies where different strains were able to convert *p*-CA. As a result, further small metabolites such as 4-HPPA are formed (**Figure 2**).^{37, 38} Therefore, the formation of 4-HPPA derives from the degradation of the aglycone structure and *p*-CA, being a structural part in the chemical structure T.

While K was metabolized only slowly in the fermentation of T with the fecal sample 320 from donor A and was still present in a very high concentration at 48 h, p-CA was 321 degraded continuously up to 24 h. This resulted in the formation of 4-HPPA 322 corresponding to the reduction of the double bond, which is a very typical reaction in 323 the microbial transformation of cinnamic acid derivatives.³⁹ At 48 h, 4-HPPA showed 324 a similar concentration compared to the 24 h samples indicating that no significant 325 further degradation of 4-HPPA was carried out by the microbiota from the fecal 326 sample of donor A. In the fermentations with the fecal samples from donors B and C, 327 4-HPPA was already detected at 7 h at its highest concentration. Subsequently, 4-328 329 HPPA was fully metabolized to 3-PPA by a dehydroxylation at the C4-position. This is in accordance with a study published by Scheline ⁴⁰, who found that the reduction of 330 the double bond by fecal microorganisms greatly exceeds dehydroxylation reactions. 331 Differences between the fermentations B and C can be found in the concentration of 332 the intermediary metabolites p-CA and 4-HPPA. Both had their concentration 333 maximum at 7 h. While the concentration of p-CA in fermentation B was much lower 334 than the concentration of 4-HPPA, the concentration of both metabolites was almost 335 identical in fermentation C. It is well known, that many anaerobic bacteria are not 336 able to carry out every degradation step and therefore metabolize aromatic 337 compounds not completely.¹⁹ Consequently, it might be possible that the microbiota 338 composition of donor B was better adapted for carrying out reduction reactions than 339 340 the microbiota of donor C.

Taking the results of the *in vitro* degradation of the reference compound K into account (**Table 1**), only 4-HPPA and PAA were detected. Therefore, it can be assumed that *p*-CA in the fermentations with T was mainly detected because of being a structural part of T. As already mentioned, *p*-CA is also the precursor for the formation of 4-HPPA (**Figure 2**). Therefore, the amount of 4-HPPA in all fermentations with T as substrate was much higher than the concentration of 4-HPPA in the fermentations with the substrate K alone.

Kaempferol-3,4'-O-diglucoside-7-O-rhamnoside. Compared to the fermentation of T, 348 where several metabolites were identified, a lower number of phenolic metabolites 349 was determined in the fermentation of the more complex kaempferol-3,4'-O-350 diglucoside-7-O-rhamnoside (K-DG-R, Table 1). Due to the fact that K-DG-R itself 351 was not stable in the ion source and therefore, could not be detected with the 352 analytical method used, the degradation was investigated only by the formation and 353 degradation of K-G and K. In the 'control samples II', where the substrate was 354 incubated with the medium, neither K-G nor K were detected (data not shown). 355 Consequently, the formation of the different metabolites resulted only from the 356 microbial degradation of K-DG-R. For quantification, only the metabolite K-G could be 357 included in the method, due to the lack of reference compounds. However, 358 additionally qTOF analyses were carried out and the tentative identification of the 359 metabolites kaempferol-3,4'-O-diglucoside, kaempferol-3-O-glucoside-7-O-360 rhamnoside, or kaempferol-4'-O-glucoside-7-O-rhamnoside, and kaempferol-7-O-361 362 rhamnoside was possible. The metabolites were detected at the same fermentation times as K-G (data not shown). Furthermore, the metabolites K-G (0-7 h), K (3-48 h), 363 and 4-HPPA (48 h) for the fermentation with the fecal sample from donor A and K-G 364 365 (0-3 h), K (3 h), and PAA (24-48 h) in the fermentations with the fecal sample from donors B and C were detected and quantified. In conclusion, as already shown for
 the substrate T, K-DG-R was degraded to phenolic acids via the consecutive release
 of the sugar moieties and the aglycone.

In comparison to the reference compound K, similar metabolites were detected. But, in the fermentation B and C of K-DG-R only the phenolic acid PAA but not the metabolite 4-HPPA was present compared to the fermentation of K. It is possible that the formation of PAA went via 4-HPPA which was not detected in the fermentations B and C, probably resulting from the time intervals, when the samples were taken (**Table 1, Figure 2**).

In vitro fermentation of the flavone apigenin and its C-glycosides vitexin, 375 vitexin-2"-O-rhamnoside, vitexin-2"-O-glucoside and vitexin-2"-O-xyloside. In 376 the *in vitro* fermentations of apigenin (AP) and its C-glycosidic derivatives, all three 377 human fecal microbiomes were able to convert the substrates yielding smaller 378 phenolic acids. Analogously to the fermentation of the phenolic compounds 379 mentioned above, differences between the donors in the metabolite formation were 380 determined. Furthermore, the fermentations with the fecal sample from donors B and 381 382 C showed the formation of smaller molecular metabolites than the fermentation with the fecal sample from donor A. Also, the same metabolites, 4-HPPA, PAA, and 3-383 PPA were detected, but the intermediate products differed between the initial 384 substrate. 385

<u>Apigenin</u>. In the fermentation with the fecal sample from donor A, AP was still detected at 48 h, while in the fermentation with the fecal sample from donors B and C, there were only low concentrations of AP at 0 h. Consequently, as AP was not present in the 'control samples II' (data not shown), an interaction between the substrate and the medium can be excluded, and it appears that it was already

partially degraded. In general, the formed metabolites were 4-HPPA (A: 24 h, 48 h; B:
7 h, C: 3 h, 7 h) and PAA (A: 48 h; B: 48 h, C: 24 h, 48 h) for all three fermentations
and additionally 3-PPA (B: 24 h, 48 h, C: 24 h, 48 h) for the fermentations with the
fecal sample from donors B and C (**Table 2**).

Vitexin. The fermentations with the substrate vitexin (V) showed the same phenolic 395 acid metabolites as the fermentation of AP, with different maxima: 4-HPPA (A: 24 h, 396 48 h; B: 24 h, 48 h), PAA (A: 48 h; B: 24 h, 48 h, C: 24 h, 48 h), and 3-PPA (B: 24 h, 397 48 h; C. 24 h, 48 h) (Table 2). V itself was degraded within the first 24 h in the 398 fermentations with the fecal samples from donors B and C but only disappeared after 399 48 h in the fermentation with the fecal sample from donor A. Additionally, the 400 metabolite AP was detected at 48 h in the fermentation with the fecal sample from 401 donor A. Therefore, the degradation of V seems to take place via the formation of the 402 aglycone structure and the glucose moiety was primarily cleaved by the bacteria. Due 403 to faster metabolism, AP was probably not detected in the fermentations with the 404 fecal samples from donors B and C. The metabolites observed are in agreement with 405 descriptions in the literature, where for example the rod-shaped Gram-positive 406 Lachnospiraceae strain CG19-1, was able to convert the substrate V.²³ In contrast, 407 two Lactococcus species and one Enterococcus species were not able to convert V, 408 as described by Kim, et al. ⁴¹. 409

<u>Vitexin derivatives</u>. In the fermentations with the different vitexin derivatives, the concentrations of vitexin-2"-O-glucoside (V-G), vitexin-2"-O-rhamnoside (V-R), and vitexin-2"-O-xyloside (V-X) were very low or initial substances not even detectable in the fermentation samples. In contrast to V, those compounds have a second sugar moiety bound which increases their water solubility. It is assumed that the vitexin derivatives still remain in the aqueous fermentation medium when extracting the

substrates and metabolites with ethyl acetate. Due to the fact, that the 'control samples II' (substrate + medium) did not show any phenolic breakdown products and substrate concentrations in the 'control samples II' and fermentation samples were very similar, the formed metabolites can be attributed to the degradation of the corresponding initial substances (data not shown). Consequently, the degradation of the vitexin derivatives could only be displayed by their metabolite formation.

In all *in vitro* fermentations with the different vitexin derivatives, the degradation went 422 via the intermediate product V indicating that the first step was cleavage of the O-423 glycoside bond by the microbiota. On the basis of the formation of V, it was possible 424 to compare the release of the second glycoside moiety. It can be seen that in 425 fermentation B and C the release of the glucose moiety was guicker than the release 426 of rhamnose, or xylose (Table 2). Figure 4 shows exemplarily the V formation within 427 the in vitro fermentation with the fecal sample from donor B for the three different 428 vitexin derivatives investigated. Graphs were normalized to 100%. This effect may be 429 due to the fact that glucoside units are more common in nature and more 430 microorganisms are adapted to utilize glucose linked to secondary plant phenolics 431 than other sugars.⁴² Despite the intermediate metabolite V, the main end products of 432 the fermentation of the vitexin derivatives were 4-HPPA, PAA, and only 3-PPA in one 433 case, which is in accordance with the metabolites resulting from the degradation of V 434 (Table 2). 435

The aglycone AP was not detected in the fermentations of the different vitexin derivatives, but similar metabolites were detected compared to the fermentation of pure AP (**Table 2**). Due to the fact, that the metabolite V was present and V itself appears to be degraded to phenolic acids via the intermediate product AP, it was

440 assumed that the formation of the phenolic acids went via the aglycone structure as441 well. However, this was not detectable in the present study.

It is recognizable that the recovery rates of the initial substrates in some 442 fermentations (i. e., V-G) were very low compared to the amount of the phenolic 443 substrates in the fermentation mixture. When looking at the 'control samples II', 444 where the phenolic compounds were co-incubated with the medium, it is notably that 445 the concentrations of the phenolic compounds were not higher and very similar 446 compared to the ones in the fermentation mixtures (data not shown). Therefore, one 447 possible reason for the low concentrations is interactions between the compounds 448 and the medium. It is also possible that the compounds precipitate or are adsorbed 449 without metabolism by the microbial biomass, the solid components of the media or 450 the inoculum. Furthermore, due to their different hydrophilicity, it is possible that more 451 hydrophilic substrates still remain in the aqueous medium when extracting with ethyl 452 acetate. 453

Before starting the fermentation experiments, different extraction procedures were investigated and optimized for an overall mixture of compounds in buffered aqueous media, similar to the one used in the present study, by looking at the recovery rates after the extraction (data not shown). Due to the fact that the substrates and metabolites used in this study show very different polarities and solubilities the methodological approach was a compromise for covering substrates (comparatively more hydrophobic) and metabolites (comparatively more hydrophilic).

gPCR analysis, Illumina sequencing and SCFA production. Overall microbiota 461 composition and activity after 48 h of incubation were assessed by qPCR for different 462 bacterial (total bacteria, Bifidobacterium Bacteroidetes, 463 groups spp., Ruminococcaceae, Lachnospiraceae, F. prausnitzii, Blautia spp., the 464 and

Roseburia/Eubacterium rectale group) and determination of short chain fatty acid 465 production in order to investigate a possible effect of the phenolic substrates on the 466 microbiota. Significant differences (p < 0.05) between the 'control samples I' (medium 467 + fecal sample without any substrate) and the different fermentations after 48 h of 468 incubation were not determined for any of the bacterial groups or SCFA 469 (Supplementary Table 3). The PCA (Figure 5) based on the relative gPCR and 470 SCFA results shows that the different samples, even after 48 h of incubation, 471 clustered by donor and not by the different phenolic substrates used. Therefore, the 472 different phenolic compounds did not have a major effect on the microbiota. 473 474 Furthermore, it is assumed that the SCFA in this study were mainly formed from the carbohydrates present in the fermentation media and fermenting the sugar moiety of 475 the flavonoid glycosides did not lead to a significant difference because of the quite 476 477 low concentrations of these compounds in the fermentation mixture.

478 Additionally, Illumina sequencing was performed with selected samples to investigate whether specific bacteria were affected by the presence of the phenolic substrates 479 that may not have been detected by the qPCR analysis. Correlation between qPCR 480 data and corresponding sequence data overall showed good agreement 481 (Supplementary Figure 1). Small variations are likely due to slight differences in 482 qPCR primer specificity and the presence of un- or misassigned sequences in the 483 sequencing data. Figure 6 shows the results of the Illumina sequencing at genus 484 level. The distribution on the phylum and family level is displayed in the supporting 485 information (Supplementary Figures 2 and 3). 486

When looking at the microbiota composition, it is recognizable that the composition of the fecal sample from donors B and C were quite similar and in line with a microbiota composition to be found in healthy humans dominated by Firmicutes and

Bacteroidetes.⁴³ The composition of the fecal sample from donor A on the other hand 490 was very different, which was confirmed by PCA at OTU level (≥97% sequence 491 identity, Supplementary Figure 4). In particular, it had a very high abundance of 492 Firmicutes, but low Bacteroidetes (Supplementary Figure 2). This initial fecal 493 sample also showed the biggest change in microbiota composition investigated by 494 Illumina sequencing after 48 h of incubation with all substrates (Supplementary 495 Figure 4), with a big increase in Bacteroidetes and Proteobacteria at the expense of 496 Firmicutes at the phylum level (Supplementary Figure 2). Only a few genera 497 498 showed a major increase in relative abundance after incubation (Bacteroides, *Clostridium* XIVa, Acidaminococcus, Parabacteroides, Veillonella, unknown 499 Burkholderiales and Escherichia/Shigella; Figure 6). Microbiota compositions 500 between samples that were incubated with different phenolic substrates for 48 h were 501 very similar within the donors (Figure 6, Supplementary Figure 4). This is in 502 agreement with the results from the qPCR and SCFA analysis. Usually, α-diversity is 503 used to express the mean variation of species to be found in a certain microbiome. In 504 this study, α -diversity was similar in all three fecal samples and remained high after 505 incubation with the fecal sample from donor C, but was lower after incubation with the 506 other two donors, particularly with donor A (Supplementary Table 4). β-Diversity 507 could not be calculated due to the limited number of samples. PCA analysis and a 508 dendogram were used for a visual presentation (Supplementary Figure 4). 509

In a study published by Duda-Chodak ⁴³, it was concluded that flavonoid aglycones may inhibit growth of some intestinal bacteria, consequently leading to a modulation of the whole intestinal microbiome. Especially in the fermentations with the fecal sample from donor A, where the microbiota composition changes most between 0 h and 48 h, the aglycone was always present for longer than in the fermentations with

the fecal sample from donors B and C. Consequently, it might be possible that this is 515 a reason for the bigger differences. However, gPCR and SCFA data (Figure 5) 516 indicate that the control samples without phenolic substances had a similar 517 microbiota composition and activity. The incubations of fecal microbiota from donor A 518 showed a microbiota shift that may indicate exposure to oxygen as oxygen tolerant 519 bacteria were stimulated (in particular Escherichia/Shigella). While this is principally 520 possible, there was no indication that this actually happened in the corresponding 521 incubations. Anaerobic conditions were checked with the addition of resazurin (0.1%) 522 to the fermentation media. Resazurin is a redox indicator that changes from colorless 523 524 via pink into blue when oxygen is present. In all three fermentations no color change was detected, suggesting that the microbiota shift seen was due to unusual initial 525 composition of fecal sample A rather than an experimental mistake. 526

For decades, it is controversially discussed that manipulation of the composition by 527 prebiotics might help to improve health.⁴⁴ A high amount of bifidobacteria in the gut is 528 often associated with health promoting effects. In a review published by Duenas, 529 Munoz-Gonzalez, Cueva, Jimenez-Giron, Sanchez-Patan, Santos-Buelga, Moreno-530 Arribas and Bartolome²⁴ studies are described, where a stimulation of the growth of 531 beneficial bacteria, such as bifidobacteria, was caused by polyphenols, thus, exerting 532 prebiotic-like effects. In the present study, an increase in the amount of bifidobacteria 533 between the initial fecal samples and the fermentation samples after 48 h were 534 detected for nearly all substrates tested. However, a significant difference of the 535 amount of bifidobacteria between the 'control samples I' and the fermentation 536 samples based on the qPCR results were not detected (Supplementary Table 3). 537 So, changes are probably caused by the fermentation media or reaction conditions. 538

539 During the fermentation of carbohydrates, where SCFA often result as main products, 540 specific gases are produced, as well. In this context, it could be also possible to 541 cluster the bacteria based on their gas release depending on the specific 542 glycosylated phenolic compounds. For that purpose, gas sensing is an alternative 543 technology for measuring gases from *in vitro* fecal sample fermentation.⁴⁵ However, 544 this was not possible to test within this study.

Interindividual comparison of the phenolic conversion rates of the different 545 donors. For all substrates tested, phenolic conversion rate of donor A was slower 546 than the fermentation rates with fecal samples of donors B and C, which were in turn 547 quite similar. For evaluating kinetics of the degradation, D₅₀ (degradation₅₀) values of 548 all substrates were used. These represent the time point, when 50% of the substrate 549 was totally degraded by the gut microbiota (Tables 1 and 2). It was not possible to 550 estimate such values for the metabolites, because their formation and a possible 551 follow-up degradation of a metabolite can take place in parallel. In general, 552 deglycosylation reactions occurred more quickly than the breakdown of the aglycone 553 structures. Consequently, based on the conversion rates, the metabolite profiles and 554 the corresponding time courses differed between the different donors and 555 fermentations, with the fecal samples from donor A showing in general less 556 metabolites. These differences are probably caused by the individual microbial 557 compositions in the microbiota from each donor. In the study published by Justesen, 558 Arrigoni, Larsen and Amado³⁴, a major decrease of rutin was observed at first after 8 559 560 h of incubation. They concluded that it might be possible that other compounds being in the media or feces are more preferred substrates (such as carbohydrates) and are 561 easier to ferment for the bacteria and thus, that the microorganisms probably needed 562 46 563 an adaptation period. Moreover, Cassidy and Minihane describe wide

interindividual variability in the bioconversion of flavonoids being attributed to specific
 enterotypes, suggesting that individuals may be either weak or strong flavonoid
 converters.

When comparing the composition of the microbiota after 48 h of incubation with the 567 substrate T in the present study, it is noticeably that Firmicutes were more dominant 568 in the fecal sample from donor B and C. In conjunction with above, the microbial 569 profile of the fermentation A showed more members of the Escherichia/Shigella 570 cluster (Figure 6). So far, more strains from the Firmicutes have been identified to be 571 responsible for flavonoid conversion than species from Proteobacteria where 572 Escherichia/Shigella belongs to.¹⁸ For example, three *Ruminococcaceae* species and 573 different Lachnospiraceae species were described in the mentioned review as being 574 part of the degradation of some flavonoid structures, whereby the latter were mainly 575 found to carry out deglycosylation reactions. Looking at the family level of the Illumina 576 577 sequencing results of the present study (Supplementary Figure 3), the amount of Ruminococcaceae and Lachnospiraceae were much higher in donors B and C than 578 in donor A. Furthermore, some species within the Lachnospiraceae family were also 579 identified being able to catalyze reduction reactions, e.g., of p-CA.^{37, 38} The 580 sequencing data do not allow resolution at species level, but it is possible that donors 581 B and C carry more bacteria that are able to perform deglycosylations and reduction 582 reactions and therefore result in a faster conversion rate than donor A. 583

584 **Comparison of the degradation of C- and O-glycosides by human fecal** 585 **microbiota.** When comparing the time courses of the deglycosylations of *O*- and *C*-586 glycosides, the present study showed that *O*-glycosides are metabolized faster than 587 *C*-glycosides, while the degradation rates of the two aglycone structures are very 588 similar within a fermentation (**Tables 1 and 2**). Looking at the fermentations of T and

V with the fecal sample from donor B, exemplarily, the concentration of the C-589 glycoside V remained nearly constant up to 7 h and V was then fully metabolized 590 (**Table 2**), whereas the O-glycoside T was already totally degraded within the first 7h 591 (Table 1, Figure 3B). Additionally, with regard to the different vitexin derivatives, it 592 was shown that the release of the O-glycoside was faster than the degradation of the 593 intermediate product V, where the sugar is bound C-glycosidically (Table 2). This 594 may be due to the fact that O-glycosides occur more frequently in the nature than C-595 glycosides and that microorganisms are more adapted to cleave the C-O-bond 596 providing more binding energy. Furthermore, it can be assumed that after the 597 consumption of C- and O-glycosides they reach different areas of the gut due to a 598 described enhanced stability of C-glycosides over the O-glycosides. Based on the 599 different prevailing conditions which are found for several intestinal areas, it might be 600 601 also possible that C- and O-glycosides in humans are converted by different microorganisms. Braune and Blaut²³ investigated the potential of the 602 603 deglycosylations of different flavonoid C- and O-glycosides by Eubacterium cellulosolvens which was only able to cleave the C-glycoside bond in two out of 604 seven compounds tested. In their study, both aglycones were then not further 605 degraded to phenolic acids. In contrast, the incubation of E. cellulosolvens with six 606 different O-glycosides showed that the microorganism was able to deglycosylate five 607 substrates.²³ 608

In all *in vitro* fermentations with the different flavonoid *C*- and *O*-glycosides, more than one degradation step was identified. As a consequence, not only deglycosylations occurred, but also a transformation of the aglycones to smaller phenolic compounds took place. It is described in the literature that the hydrolysis of (flavonoid) glycosides to their aglycones results in potentially more bioactive

metabolites compared to the initial compounds. However, a further microbial 614 degradation of the aglycone to smaller metabolites can lead to the formation of more 615 or less active compounds.⁴⁷ The fact that microorganisms are often not able to carry 616 out all degradation steps was already described for the small phenolic acid caffeic 617 acid by Peppercorn and Goldman⁴⁸, in 1971. This seems to be valid for the 618 degradation of flavonoids as well. For example, in the study published by Braune and 619 Blaut²³, *E. cellulosolvens* was only able to carry out deglycosylation reactions, 620 resulting in the aglycone structure as the only metabolite. However, in the substrates 621 metabolized by *E. cellulosolvens*, the glucose moiety was bound at the C6-position. 622 When glucose was present at the C8-position (e.g., vitexin) degradation did not 623 occur. Braune and Blaut further showed that the intestinal Lachnospiraceae strain 624 CG19-1, in contrast to E. cellulosolvens, was able to convert six out of the seven 625 626 tested C-glycosides via the aglycone structure to small phenolic acids, indicating deglycosylation, ring fission, and dehydroxylation reactions.²³ Investigations done by 627 Nakamura, et al.⁴⁹ on the cleavage of the *C*-glycosyl bond of puerarin (an isoflavone 628 glucoside) by a strain called PUE showed only a conversion to its aglycone daidzein. 629 Smaller phenolic acids were not detected.⁴⁹ In this context, it is not clear whether the 630 degradation steps of the substrates used in the present study are carried out by only 631 one or even more microorganisms. 632

In summary, all substrates used in the present *in vitro* study were converted yielding lower molecular weight phenolic acids. Deglycosylation reactions were carried out by the microbiota first, followed by the further breakdown of the aglycone structure. Comparing the donors with each other, similar metabolites were detected, even though different time courses for metabolite formation were observed. With regard to

the microbial composition or activity (SCFA), an influence of the flavonoid glycosideswas not detected.

When looking at the results of the present study, the degradation of the selected 640 phenolic compounds showed more or less the same course of degradation that 641 resulted in very similar metabolite profiles, as well. Only the fermentation rates were 642 different. Therefore, we did not expect very different degradation pathways and more 643 outcomes when fermenting the substrates with fecal samples of additional 644 volunteers. The degradation pathways might only be significantly different, when 645 using fecal samples of volunteers having a known disorder (e.g., by taking antibiotics) 646 or significantly different dietary habits. It is known that one donor of the present study 647 was a vegetarian, but significant differences in metabolite formation were not 648 detected, at all. Also, seasonable differences might be possible, too, but the main aim 649 of the study was to monitor the degradation profile with regard to the (complex) 650 phenolic compounds selected. Therefore, more fermentations with more fecal 651 samples were not carried out. When comparing with *in vitro* fermentation experiments 652 described in the literature it is obvious that the use of one to four different fecal 653 samples is usual.^{21, 29-31} 654

In conclusion, flavonoid glycosides can be metabolized by the human gut 655 microorganisms when consumed in concentrations found in typical diets (~20 656 mg/day), but a prebiotic effect of the phenolic compounds seems not to be 657 achievable. However, in a review by Etxeberria, et al. ⁵⁰, *in vitro* and *in vivo* studies 658 are described which showed an influence of phenolic compounds on the microbiota 659 660 composition. The effects were often only significant when the concentrations were artificially high. This does not mean that a certain influence of the microbiome is not 661 possible, as not all species present in the gut can be analyzed comprehensively to 662

date. But, it is also possible that a substrate on its own is not able to influence the gut 663 microbiota and the described effects on the microbiota composition are a result of the 664 interaction of different phenolic compounds and other food components being 665 present in the food matrix such as proteins. Besides other research groups, Ozdal, et 666 al. ⁵¹ described that a polyphenol-protein interaction results in changes in the 667 structural, functional, and nutritional properties of both compounds. This interaction is 668 influenced by several parameters such as pH, temperature, and the chemical 669 structure of the compounds. Therefore, interactive effects of the polyphenols with 670 further compounds might influence their metabolism by the microbiota to the 671 672 individual strength/binding mechanism of the polyphenol-protein interaction, when consuming whole foods. 673

Microbial transformation products might affect microbial composition and might 674 therefore lead to different consequences for the host and its health.^{21, 52} Such an 675 676 effect could be more pronounced after consumption of C-glycosides, as their slower cleavage leads to a prolonged exposure to the intact flavonoid. Moreover, an impact 677 on deeper regions of the colon seems to be possible due to the enhanced stability. 678 With regard to diversity, it can be hypothesized that the more complex the substrates 679 (for gastrointestinal fermentation) are, more different metabolites can be formed likely 680 resulting in a higher variation of gut microbial composition. Although this research 681 topic is now studied for more than two decades, a final proof of a health-beneficial 682 effect of isolated phenolic compounds is still missing. In the future, research has to 683 684 be extended towards synergisms/interactions with other food compounds and all what we would call food matrix. 685

686

Abbreviations. AP, apigenin; p-CA, *p*-coumaric acid; 4-HPPA, 687 3-(4-K, kaempferol; K-DG-R, kaempferol-3.4'-O-688 hydroxyphenyl)propionic acid; diglucoside-7-O-rhamnoside; K-G, kaempferol-3-O-glucoside; LOD, limit of detection; 689 PAA, phenylacetic acid; PCA, Principal Component Analysis; 3-PPA, 3-690 phenylpropionic acid; SCFA, short chain fatty acids; T, tiliroside; V, vitexin; V-G, 691 vitexin-2"-O-glucoside; V-R, vitexin-2"-O-rhamnoside; V-X, vitexin-2"-O-xyloside. 692

693

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702 SUPPORTING INFORMATION

Isolation method of vitexin-2"-O-glucoside and vitexin-2"-O-xyloside (material and
methods), Illumina sequencing of 16S rRNA genes (material and methods), figures
about the relative abundance of OTUs grouped at phylum and family level (results),
PCA with the sequencing data (results), tables about the SCFA production,
microbiota profile and Illumina sequencing operational taxonomic units (results). This
material is available free of charge via the Internet at http://pubs.acs.org

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845 **LIST OF FIGURE CAPTIONS**

Figure 1: Structures of the compounds used for the *in vitro* fermentations.

Figure 2: Overview of the degradation pattern of the kaempferol-O-derivates und 847 848 apigenin-C-derivatives used in the in vitro study (AP, apigenin; p-CA, p-coumaric acid; Glc, glucose; 4-HPAA, 3-(4-hydroxyphenyl)acetic acid; 4-HPPA, 3-(4-849 hvdroxvphenvl)propionic acid: K, kaempferol; K-DG-R. kaempferol-3.4'-O-850 diglucoside-7-O-rhamnoside; K-G, kaempferol-3-O-glucoside; PAA, phenylacetic 851 acid; 3-PPA, 3-phenylpropionic acid; Rha, rhamnose; T, tiliroside; V, vitexin; V-G, 852 vitexin-2"-O-glucoside; V-R, vitexin-2"-O-rhamnoside; V-X, vitexin-2"-O-xyloside; - - -853 > = pathway is not totally identified). 854

Figure 3: *In vitro* degradation and metabolite formation of tiliroside with three different donors (A, B, and C). Data represent the mean and standard deviation of triplicates (T, tiliroside; K-G, kaempferol-3-*O*-glucoside; K, kaempferol; 4-HPPA, 3-(4hydroxyphenyl)propionic acid; *p*-CA, *p*-coumaric acid; 3-PPA, 3-phenylpropionic acid).

Figure 4: Comparison of vitexin formation within the *in vitro* fermentations with different vitexin derivatives of donor B. Data are normalized to 100%.

Figure 5: Principal component analysis with the percentage data for net SCFA and production microbiota composition (qPCR) after 48 h of incubation. Colors are coded by different donor (donor A: black, donor B: red, donor C: blue).

Figure 6: Relative abundance of OTUs obtained by Illumina sequencing of the three different initial fecal samples (donor A, B and C) and after 48 h of incubation with different substrates. OTUs are grouped together at the genus level.

TABLES

Table 1: Concentration of the substrates and metabolites within the *in vitro* degradation of K, T, and K-DG-R with the three different donors (A, B, and C). Data represent the mean and standard deviation of triplicates (D₅₀, time point when 50% of the substrate was degraded).

		Concentration [µM]					
Compound	Donor	0 h	3 h	7 h	24 h	48 h	
Kaempferol	A	232 ± 16.0	201 ± 15.8	179 ± 51.4	170 ± 8.32	34.0 ± 20.7	33 h 32 min
	В	68.9 ± 13.4	61.1 ± 32.9	<lod< td=""><td><lod< td=""><td><lod< td=""><td>4 h 45 min</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>4 h 45 min</td></lod<></td></lod<>	<lod< td=""><td>4 h 45 min</td></lod<>	4 h 45 min
	С	27.5 ± 5.31	16.9 ± 4.53	8.28 ± 11.8	<lod< td=""><td><lod< td=""><td>4 h 28 min</td></lod<></td></lod<>	<lod< td=""><td>4 h 28 min</td></lod<>	4 h 28 min
K metabolites							
4-HPPA	А	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>32.8 ± 0.0100</td><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td>32.8 ± 0.0100</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>32.8 ± 0.0100</td><td></td></lod<></td></lod<>	<lod< td=""><td>32.8 ± 0.0100</td><td></td></lod<>	32.8 ± 0.0100	
	В	<lod< td=""><td><lod< td=""><td>19.4*</td><td>5.97*</td><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td>19.4*</td><td>5.97*</td><td><lod< td=""><td></td></lod<></td></lod<>	19.4*	5.97*	<lod< td=""><td></td></lod<>	
	С	<lod< td=""><td><lod< td=""><td>13.8 ± 5.91**</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td>13.8 ± 5.91**</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	13.8 ± 5.91**	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
PAA	А	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td>13.7 ± 4.91</td><td>21.1 ± 18.6</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>13.7 ± 4.91</td><td>21.1 ± 18.6</td><td></td></lod<></td></lod<>	<lod< td=""><td>13.7 ± 4.91</td><td>21.1 ± 18.6</td><td></td></lod<>	13.7 ± 4.91	21.1 ± 18.6	
Tiliroside	А	105 ± 13.2	84.4 ± 6.65	54.6 ± 13.0	<lod< td=""><td><lod< td=""><td>7 h 17 min</td></lod<></td></lod<>	<lod< td=""><td>7 h 17 min</td></lod<>	7 h 17 min
	В	183 ± 48.5	125 ± 35.1	<lod< td=""><td><lod< td=""><td><lod< td=""><td>4 h 4 min</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>4 h 4 min</td></lod<></td></lod<>	<lod< td=""><td>4 h 4 min</td></lod<>	4 h 4 min
	С	107 ± 52.7	75.7 ± 38.2	<lod< td=""><td><lod< td=""><td><lod< td=""><td>3 h</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>3 h</td></lod<></td></lod<>	<lod< td=""><td>3 h</td></lod<>	3 h
T metabolites							
K-G	А	5.85 ± 2.41	7.16 ± 0.522	8.91 ± 2.62	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	В	91.4 ± 28.1	22.1 ± 10.5	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	31.7 ± 6.36	16.8 ± 3.82	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
K	A	32.3 ± 17.7	95.3 ± 21.2	145 ± 19.9	89.6 ± 0.719	79.4 ± 27.2	
	В	5.24 ± 0.176	39.8 ± 8.89	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	4.53 ± 2.33	8.70 ± 4.45	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
p-CA	A	19.5 ± 2.61	31.3 ± 12.2	123 ± 19.5	17.1 ± 24.4	3.02 ± 0.776	
	В	3.97 ± 0.453	5.86 ± 1.46	12.7 ± 8.36	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	8.14 ± 1.56	19.4 ± 4.35	49.1 ± 9.71	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	

4-HPPA	А	<lod< th=""><th><lod< th=""><th><lod< th=""><th>123 ± 3.56</th><th>121 ± 12.4</th><th></th></lod<></th></lod<></th></lod<>	<lod< th=""><th><lod< th=""><th>123 ± 3.56</th><th>121 ± 12.4</th><th></th></lod<></th></lod<>	<lod< th=""><th>123 ± 3.56</th><th>121 ± 12.4</th><th></th></lod<>	123 ± 3.56	121 ± 12.4	
	В	<lod< td=""><td><lod< td=""><td>99.9 ± 53.2</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td>99.9 ± 53.2</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	99.9 ± 53.2	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	<lod< td=""><td><lod< td=""><td>46.9 ± 19.1</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td>46.9 ± 19.1</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	46.9 ± 19.1	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
3-PPA	А	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td>27.6 ± 26.8**</td><td>5.36 ± 9.28**</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>27.6 ± 26.8**</td><td>5.36 ± 9.28**</td><td></td></lod<></td></lod<>	<lod< td=""><td>27.6 ± 26.8**</td><td>5.36 ± 9.28**</td><td></td></lod<>	27.6 ± 26.8**	5.36 ± 9.28**	
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td>91.8 ± 13.3**</td><td>38.0 ± 65.8**</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>91.8 ± 13.3**</td><td>38.0 ± 65.8**</td><td></td></lod<></td></lod<>	<lod< td=""><td>91.8 ± 13.3**</td><td>38.0 ± 65.8**</td><td></td></lod<>	91.8 ± 13.3**	38.0 ± 65.8**	
K-DG-R	А	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
K-DG-R metabolites							
K-G	А	8.77 ± 1.94	36.6 ± 13.2	10.3 ± 1.17	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	В	35.8 ± 6.55	88.7 ± 61.2	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	3.92 ± 0.976	30.7 ± 24.9	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
К	А	13.0 ± 7.57	131 ± 12.1	170 ± 14.4	151 ± 40.8	6.60 ± 1.65	
	В	<lod< td=""><td>54.9 ± 0.701</td><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	54.9 ± 0.701	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	<lod< td=""><td>13.3 ± 5.25</td><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	13.3 ± 5.25	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
4-HPPA	А	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>33.5 ± 2.88</td><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td>33.5 ± 2.88</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>33.5 ± 2.88</td><td></td></lod<></td></lod<>	<lod< td=""><td>33.5 ± 2.88</td><td></td></lod<>	33.5 ± 2.88	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
PAA	А	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td>21.8 ± 23.4</td><td>15.5 ± 19.9</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>21.8 ± 23.4</td><td>15.5 ± 19.9</td><td></td></lod<></td></lod<>	<lod< td=""><td>21.8 ± 23.4</td><td>15.5 ± 19.9</td><td></td></lod<>	21.8 ± 23.4	15.5 ± 19.9	
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td>3.05 ± 7.74</td><td>18.3 ± 10.1</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>3.05 ± 7.74</td><td>18.3 ± 10.1</td><td></td></lod<></td></lod<>	<lod< td=""><td>3.05 ± 7.74</td><td>18.3 ± 10.1</td><td></td></lod<>	3.05 ± 7.74	18.3 ± 10.1	

<LOD (limit of detection)</p>
* the metabolite was only detected in one sample out of the triplicate fermentation
** the metabolite was only detected in two samples out of the triplicate fermentation

Table 2: Concentrations of the substrates and metabolites within the *in vitro* degradation of AP, V, V-G, V-R and V-X with the three different donors (A, B, and C). Data represent the mean and standard deviation of triplicates (D₅₀, time point when 50% of the substrate was degraded).

		Concentration [µM]					
Compound	Donor	0 h	3 h	7 h	24 h	48 h	
Apigenin	А	106 ± 16.0	97.9 ± 7.57	98.6 ± 11.7	65.4 ± 26.9	34.0 ± 20.7	33 h 29 min
	В	22.9 ± 14.5	15.4 ± 3.02	<lod< td=""><td><lod< td=""><td><lod< td=""><td>4 h 35 min</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>4 h 35 min</td></lod<></td></lod<>	<lod< td=""><td>4 h 35 min</td></lod<>	4 h 35 min
	С	24.1 ± 2.66	25.3 ± 6.55	<lod< td=""><td><lod< td=""><td><lod< td=""><td>5 h 6 min</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>5 h 6 min</td></lod<></td></lod<>	<lod< td=""><td>5 h 6 min</td></lod<>	5 h 6 min
AP metabolites							
4-HPPA	А	<lod< td=""><td><lod< td=""><td><lod< td=""><td>38.7 ± 9.56</td><td>173 ± 55.5</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>38.7 ± 9.56</td><td>173 ± 55.5</td><td></td></lod<></td></lod<>	<lod< td=""><td>38.7 ± 9.56</td><td>173 ± 55.5</td><td></td></lod<>	38.7 ± 9.56	173 ± 55.5	
	В	<lod< td=""><td><lod< td=""><td>187 ± 66.2</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td>187 ± 66.2</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	187 ± 66.2	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	<lod< td=""><td>32.9 ± 18.6</td><td>124 ± 7.95</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	32.9 ± 18.6	124 ± 7.95	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
3-PPA	А	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td>8.01 ± 1.55</td><td>49.8 ± 35.1</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>8.01 ± 1.55</td><td>49.8 ± 35.1</td><td></td></lod<></td></lod<>	<lod< td=""><td>8.01 ± 1.55</td><td>49.8 ± 35.1</td><td></td></lod<>	8.01 ± 1.55	49.8 ± 35.1	
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td>143 ± 24.1</td><td>214 ± 39.4**</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>143 ± 24.1</td><td>214 ± 39.4**</td><td></td></lod<></td></lod<>	<lod< td=""><td>143 ± 24.1</td><td>214 ± 39.4**</td><td></td></lod<>	143 ± 24.1	214 ± 39.4**	
PAA	А	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>13.8 ± 12.1</td><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td>13.8 ± 12.1</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>13.8 ± 12.1</td><td></td></lod<></td></lod<>	<lod< td=""><td>13.8 ± 12.1</td><td></td></lod<>	13.8 ± 12.1	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>24.4 ± 21.3</td><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td>24.4 ± 21.3</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>24.4 ± 21.3</td><td></td></lod<></td></lod<>	<lod< td=""><td>24.4 ± 21.3</td><td></td></lod<>	24.4 ± 21.3	
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td>52.4 ± 14.9</td><td>50.4 ± 7.64</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>52.4 ± 14.9</td><td>50.4 ± 7.64</td><td></td></lod<></td></lod<>	<lod< td=""><td>52.4 ± 14.9</td><td>50.4 ± 7.64</td><td></td></lod<>	52.4 ± 14.9	50.4 ± 7.64	
Vitexin	А	113 ± 23.1	117 ± 27.3	108 ± 11.6	112 ± 1.04	<lod< td=""><td>35 h 52 min</td></lod<>	35 h 52 min
	В	147 ± 21.9	96.6 ± 31.1	115 ± 17.8	<lod< td=""><td><lod< td=""><td>13 h 8 min</td></lod<></td></lod<>	<lod< td=""><td>13 h 8 min</td></lod<>	13 h 8 min
	С	131 ± 4.53	121*	128 ± 4.85	<lod< td=""><td><lod< td=""><td>15 h 17 min</td></lod<></td></lod<>	<lod< td=""><td>15 h 17 min</td></lod<>	15 h 17 min
V metabolites							
AP	А	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>39.4 ± 30.5</td><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td>39.4 ± 30.5</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>39.4 ± 30.5</td><td></td></lod<></td></lod<>	<lod< td=""><td>39.4 ± 30.5</td><td></td></lod<>	39.4 ± 30.5	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
4-HPPA	А	<lod< td=""><td><lod< td=""><td><lod< td=""><td>22.6 ± 0.961</td><td>102 ± 32.6</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>22.6 ± 0.961</td><td>102 ± 32.6</td><td></td></lod<></td></lod<>	<lod< td=""><td>22.6 ± 0.961</td><td>102 ± 32.6</td><td></td></lod<>	22.6 ± 0.961	102 ± 32.6	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td>123 ± 6.76</td><td>80.2 ± 17.9</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>123 ± 6.76</td><td>80.2 ± 17.9</td><td></td></lod<></td></lod<>	<lod< td=""><td>123 ± 6.76</td><td>80.2 ± 17.9</td><td></td></lod<>	123 ± 6.76	80.2 ± 17.9	
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
3-PPA	А	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td>41.1 ± 27.0**</td><td>25.7 ± 4.90*</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>41.1 ± 27.0**</td><td>25.7 ± 4.90*</td><td></td></lod<></td></lod<>	<lod< td=""><td>41.1 ± 27.0**</td><td>25.7 ± 4.90*</td><td></td></lod<>	41.1 ± 27.0**	25.7 ± 4.90*	

C A B C A C A	$\begin{array}{c} < \text{LOD} \\ 0.500 \pm 0.200 \\ < \text{LOD} \\ < \text{LOD} \\ < \text{LOD} \\ 0.824 \pm 0.0560 \\ < \text{LOD} \\ < $	$\begin{array}{c} < LOD \\ \end{array}$ $\begin{array}{c} 2.02 \pm 0.652 \\ < LOD \\ < LOD \\ < LOD \\ \end{array}$ $\begin{array}{c} 0.351 \pm 0.0137 \\ 7.02 \pm 1.01 \\ 2.29 \pm 0.578 \\ < LOD \\ $	<lod .n.d="" 0.127="" 0.443="" 0.519="" 0.550="" 0.714="" 1.23="" 10.3="" 12.3="" 14.0="" 2.09="" 2.34="" 2.45="" 3.46="" 3.52="" 3.79**="" 4.65="" 6.70*="" <="" <lod="" pre="" ±=""></lod>	131 ± 30.7 <pre><lod< pre=""> 43.8 ± 10.2</lod<></pre> 47.9 ± 19.8 1.22 ± 0.765 <lod< pre=""> 1.29 ± 0.612 <lod< pre=""> 1.29 ± 0.612 <lod< pre=""> 25.9* <lod </lod 25.9* <lod </lod 38.3 ± 10.0 35.1 ± 29.0 13.7 ± 1.66 <lod< th=""><th>142 ± 35.9 10.9 ± 4.79 47.0 ± 10.2 47.3 ± 9.75 0.769 ± 0.152 $< LOD$ $< LOD$ $< LOD$ $37.5 \pm 14.4^{**}$ $< LOD$ $7.37 \pm 1.30^{**}$ 32.2 ± 3.6 19.6 ± 1.41 $< LOD$ $< LOD$</th><th>27 h 50 min 1 h 29 min - 36 h 3 min 5 h 55 min 15 h 29 min</th></lod<></lod<></lod<></lod<>	142 ± 35.9 10.9 ± 4.79 47.0 ± 10.2 47.3 ± 9.75 0.769 ± 0.152 $< LOD$ $< LOD$ $< LOD$ $37.5 \pm 14.4^{**}$ $< LOD$ $7.37 \pm 1.30^{**}$ 32.2 ± 3.6 19.6 ± 1.41 $< LOD$ $< LOD$	27 h 50 min 1 h 29 min - 36 h 3 min 5 h 55 min 15 h 29 min
A B C A B C A B C A B C A B C A B C	$\begin{array}{c} < LOD \\ < LOD \\ < LOD \\ < LOD \\ \end{array}$ $\begin{array}{c} 1.84 \pm 0.342 \\ 0.500 \pm 0.200 \\ < LOD \\$		$\begin{array}{c} 6.70^{*} \\ < LOD \\ < LOD \\ \end{array}$ $\begin{array}{c} 2.34 \pm 0.519 \\ < LOD \\ < LOD \\ < LOD \\ \end{array}$ $\begin{array}{c} 0.714 \pm 0.127 \\ 3.52 \pm 2.09 \\ 2.45 \pm 1.23 \\ < LOD \\ 14.0 \pm 3.79^{**} \\ < LOD \\ < LOD \\ < LOD \\ < LOD \\ - LOD \\ .n.d \\ \end{array}$ $\begin{array}{c} 12.3 \pm 3.46 \\ 4.65 \pm 0.443 \\ 10.3 \pm 0.550 \end{array}$	$$	10.9 ± 4.79 47.0 ± 10.2 47.3 ± 9.75 0.769 ± 0.152 $< LOD$ $< LOD$ $< LOD$ $37.5 \pm 14.4^{**}$ $< LOD$ $7.37 \pm 1.30^{**}$ 32.2 ± 3.6 19.6 ± 1.41 $< LOD$ $< LOD$	27 h 50 min 1 h 29 min - 36 h 3 min 5 h 55 min 15 h 29 min
B C A C A	$\begin{array}{c} < LOD \\ < LOD \\ < LOD \\ \end{array}$ $\begin{array}{c} 1.84 \pm 0.342 \\ 0.500 \pm 0.200 \\ < LOD \\ < LOD \\ < LOD \\ 0.824 \pm 0.0560 \\ < LOD $		$$	$43.8 \pm 10.2 \\ 47.9 \pm 19.8 \\1.22 \pm 0.765 \\ < LOD \\ < LOD \\1.29 \pm 0.612 \\ < LOD \\ < LOD \\ 25.9^* \\ < LOD \\ < LOD \\ < LOD \\ 38.3 \pm 10.0 \\35.1 \pm 29.0 \\13.7 \pm 1.66 \\ < LOD \\$	$47.0 \pm 10.2 \\ 47.3 \pm 9.75 \\ 0.769 \pm 0.152 \\ $	27 h 50 min 1 h 29 min - - 36 h 3 min 5 h 55 min 15 h 29 min
C A B C A C A	$ 1.84 \pm 0.342 0.500 \pm 0.200 0.824 \pm 0.0560 <2LOD <1.0D <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02 <1.02$	<lod< math=""> 2.02 ± 0.652 <math display="block"><lod< math=""> 2.02 ± 0.652 <math display="block"><lod< math=""> 0.351 ± 0.0137 7.02 ± 1.01 2.29 ± 0.578 <math display="block"><lod< math=""> $8.95 ± 1.38$ 9.71 ± 0.424 7.51 ± 1.40</lod<></math></lod<></math></lod<></math></lod<></math></lod<></math></lod<></math></lod<></math></lod<></math></lod<></math></lod<>	<lod< math=""> 2.34 ± 0.519 <math display="block"><lod< math=""> 2.00 <math display="block"><lod< math=""> 0.714 ± 0.127 3.52 ± 2.09 2.45 ± 1.23 <math display="block"><lod< math=""> 14.0 ± 3.79** <math display="block"><lod< math=""> <math display="block"><lod< math=""> <math display="block"><lod< math=""> $.n.d$ 12.3 ± 3.46 4.65 ± 0.443 10.3 ± 0.550</lod<></math></lod<></math></lod<></math></lod<></math></lod<></math></lod<></math></lod<>	47.9 ± 19.8 1.22 ± 0.765 	47.3 ± 9.75 0.769 ± 0.152 $< LOD$ $< LOD$ $< LOD$ $< LOD$ $37.5 \pm 14.4^{**}$ $< LOD$ $7.37 \pm 1.30^{**}$ 32.2 ± 3.6 19.6 ± 1.41 $< LOD$ $< LOD$	27 h 50 min 1 h 29 min - - 36 h 3 min 5 h 55 min 15 h 29 min
A B C A B C A B C A B C A B C A B C	$\begin{array}{c} 1.84 \pm 0.342 \\ 0.500 \pm 0.200 \\ < LOD \\ < LOD \\ 0.824 \pm 0.0560 \\ < LOD \\$	$2.02 \pm 0.652 \\ < LOD \\ < LOD \\ 0.351 \pm 0.0137 \\ 7.02 \pm 1.01 \\ 2.29 \pm 0.578 \\ < LOD \\ < S.95 \pm 1.38 \\ 9.71 \pm 0.424 \\ 7.51 \pm 1.40 \\ \end{cases}$	2.34 ± 0.519 <lod <lod 0.714 ± 0.127 3.52 ± 2.09 2.45 ± 1.23 <lod $14.0 \pm 3.79^{**}$ <lod <lod <lod .n.d 12.3 ± 3.46 4.65 ± 0.443 10.3 ± 0.550</lod </lod </lod </lod </lod </lod 	$1.22 \pm 0.765 \\ < LOD \\ < LOD \\ 1.29 \pm 0.612 \\ < LOD \\ < LOD \\ 25.9^* \\ < LOD \\ < LOD \\ < LOD \\ < LOD \\ 38.3 \pm 10.0 \\ 35.1 \pm 29.0 \\ 13.7 \pm 1.66 \\ < LOD \\ < LOD \\ = 1.66 \\ < LOD \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0.00 \\ < 0$	$\begin{array}{c} 0.769 \pm 0.152 \\ < LOD \\ < LOD \\ < LOD \\ < LOD \\ 37.5 \pm 14.4^{**} \\ < LOD \\ 37.5 \pm 13.0^{**} \\ 32.2 \pm 3.6 \\ 19.6 \pm 1.41 \\ < LOD $	27 h 50 min 1 h 29 min - - 36 h 3 min 5 h 55 min 15 h 29 min
B C A B C A B C A B C A B C A B C	$\begin{array}{c} 0.500 \pm 0.200 \\ < \text{LOD} \\ < \text{LOD} \\ 0.824 \pm 0.0560 \\ < \text{LOD} \\ < $		$$		<lod <lod <lod <lod 37.5 ± 14.4** <lod 7.37 ± 1.30** 32.2 ± 3.6 19.6 ± 1.41 <lod <lod< td=""><td>1 h 29 min - 36 h 3 min 5 h 55 min 15 h 29 min</td></lod<></lod </lod </lod </lod </lod </lod 	1 h 29 min - 36 h 3 min 5 h 55 min 15 h 29 min
C A B C A B C A B C A B C A B C	<lod <lod <lod 0.824 ± 0.0560 <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <lod <ldd <ldd <ldd <ldd <ldd <ldd <ldd <ldd <ldd <ldd <ldd <ldd <ldd <ldd <ldd <ldd <ldd <ldd <ldd <ldd <ldd <ldd 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A B C A B C A B C A B C A B C		$\begin{array}{c} 0.351 \pm 0.0137 \\ 7.02 \pm 1.01 \\ 2.29 \pm 0.578 \\ < LOD \\ & < LOD \\ & \\ 8.95 \pm 1.38 \\ 9.71 \pm 0.424 \\ 7.51 \pm 1.40 \end{array}$	$\begin{array}{c} 0.714 \pm 0.127 \\ 3.52 \pm 2.09 \\ 2.45 \pm 1.23 \\ < LOD \\ 14.0 \pm 3.79^{**} \\ < LOD \\ < LOD \\ < LOD \\ .n.d \\ 12.3 \pm 3.46 \\ 4.65 \pm 0.443 \\ 10.3 \pm 0.550 \end{array}$	$1.29 \pm 0.612 \\ < LOD \\ 25.9^{*} \\ < LOD \\ < LOD \\ < LOD \\ 38.3 \pm 10.0 \\ 35.1 \pm 29.0 \\ 13.7 \pm 1.66 \\ < LOD \\ = 1.00 \\ < LOD \\ $	<lod <lod 37.5 ± 14.4** <lod <lod 7.37 ± 1.30** 32.2 ± 3.6 19.6 ± 1.41 <lod <lod< td=""><td>36 h 3 min 5 h 55 min 15 h 29 min</td></lod<></lod </lod </lod </lod </lod 	36 h 3 min 5 h 55 min 15 h 29 min
A B C A B C A B C A B C A B C		$\begin{array}{c} 0.351 \pm 0.0137 \\ 7.02 \pm 1.01 \\ 2.29 \pm 0.578 \\ < LOD \\ & \\ 8.95 \pm 1.38 \\ 9.71 \pm 0.424 \\ 7.51 \pm 1.40 \end{array}$	$\begin{array}{c} 0.714 \pm 0.127 \\ 3.52 \pm 2.09 \\ 2.45 \pm 1.23 \\ < LOD \\ 14.0 \pm 3.79^{**} \\ < LOD \\ < LOD \\ < LOD \\ < LOD \\ .n.d \\ 12.3 \pm 3.46 \\ 4.65 \pm 0.443 \\ 10.3 \pm 0.550 \end{array}$	$1.29 \pm 0.612 \\ < LOD \\ 25.9^* \\ < LOD \\ < LOD \\ < LOD \\ < LOD \\ 38.3 \pm 10.0 \\ 35.1 \pm 29.0 \\ 13.7 \pm 1.66 \\ < LOD \\ < LOD \\ - 1.00 \\ - 1.0$	<lod <lod 37.5 ± 14.4** <lod <lod 7.37 ± 1.30** 32.2 ± 3.6 19.6 ± 1.41 <lod <lod< td=""><td>36 h 3 min 5 h 55 min 15 h 29 min</td></lod<></lod </lod </lod </lod </lod 	36 h 3 min 5 h 55 min 15 h 29 min
B C A B C A B C A B C		7.02 ± 1.01 2.29 ± 0.578 $ 8.95 \pm 1.38 9.71 \pm 0.424 7.51 \pm 1.40$	3.52 ± 2.09 2.45 ± 1.23 $ 14.0 \pm 3.79^{**} .n.d 12.3 \pm 3.46 4.65 \pm 0.443 10.3 \pm 0.550$	<lod <lod 25.9* <lod <lod 38.3 ± 10.0 35.1 ± 29.0 13.7 ± 1.66 <lod< td=""><td><lod <lod 37.5 ± 14.4** <lod <lod 7.37 ± 1.30** 32.2 ± 3.6 19.6 ± 1.41 <lod <lod< td=""><td>36 h 3 min 5 h 55 min 15 h 29 min</td></lod<></lod </lod </lod </lod </lod </td></lod<></lod </lod </lod </lod 	<lod <lod 37.5 ± 14.4** <lod <lod 7.37 ± 1.30** 32.2 ± 3.6 19.6 ± 1.41 <lod <lod< td=""><td>36 h 3 min 5 h 55 min 15 h 29 min</td></lod<></lod </lod </lod </lod </lod 	36 h 3 min 5 h 55 min 15 h 29 min
C A B C A B C A B C	$\begin{array}{c} 0.824 \pm 0.0560 \\ < LOD \\ \\ 8.43 \pm 0.623 \\ 12.0 \pm 5.17 \\ 5.58 \pm 1.43 \end{array}$	$\begin{array}{r} 2.29 \pm 0.578 \\ < LOD \\ \\ 8.95 \pm 1.38 \\ 9.71 \pm 0.424 \\ 7.51 \pm 1.40 \end{array}$	$2.45 \pm 1.23 \\ $	<lod 25.9* <lod <lod 38.3 ± 10.0 35.1 ± 29.0 13.7 ± 1.66 <lod< td=""><td><lod 37.5 ± 14.4** <lod 7.37 ± 1.30** 32.2 ± 3.6 19.6 ± 1.41 <lod <lod< td=""><td>36 h 3 min 5 h 55 min 15 h 29 min</td></lod<></lod </lod </lod </td></lod<></lod </lod </lod 	<lod 37.5 ± 14.4** <lod 7.37 ± 1.30** 32.2 ± 3.6 19.6 ± 1.41 <lod <lod< td=""><td>36 h 3 min 5 h 55 min 15 h 29 min</td></lod<></lod </lod </lod 	36 h 3 min 5 h 55 min 15 h 29 min
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A B C	8.43 ± 0.623 12.0 ± 5.17 5.58 ± 1.43	8.95 ± 1.38 9.71 ± 0.424 7.51 ± 1.40	12.3 ± 3.46 4.65 ± 0.443 10.3 ± 0.550	13.7 ± 1.66 <lod< td=""><td><lod <lod< td=""><td>36 h 3 min 5 h 55 min 15 h 29 min</td></lod<></lod </td></lod<>	<lod <lod< td=""><td>36 h 3 min 5 h 55 min 15 h 29 min</td></lod<></lod 	36 h 3 min 5 h 55 min 15 h 29 min
B C	12.0 ± 5.17 5.58 ± 1.43	9.71 ± 0.424 7.51 ± 1.40	4.65 ± 0.443	<lod< td=""><td><lod< td=""><td>5 h 55 min 15 h 29 min</td></lod<></td></lod<>	<lod< td=""><td>5 h 55 min 15 h 29 min</td></lod<>	5 h 55 min 15 h 29 min
C	5.58 ± 1.43	7.51 ± 1.40	10.3 ± 0.550			15 h 20 min
Ŭ	0.00 ± 1.10	1.01 ± 1.10	1 1 1 . 1 1 1 1 . 1. 1. 1.	<1 00		
			10.0 ± 0.000		LOD	10112011111
Δ			2 29 + 0 851	10 7 + 7 45		
R		1.65 ± 0.415	68 8 + 19 8			
C	0.845 ± 0.154	1.00 ± 0.410 1.41 ± 0.475	7.82 ± 1.30	0.748 ± 0.063		
Δ					28.2*	
R						
C				12 2*		
Δ				10.0 23.2 ± 0**	130 8 ± /8 1	
R				23.2 ± 0 230 ± 25 7	160 ± 40.1	
C				200 ± 20.1 20 2*	109 ± 39.9 10 ∩*	
Δ						
R						
C					<lod 226*</lod 	
U.			<lud 1 01*</lud 		∠30 107*	
Λ	<lod< td=""><td></td><td>4.04</td><td><luu 52.4 + 7.24</luu </td><td></td><td></td></lod<>		4.04	<luu 52.4 + 7.24</luu 		
A			<lud< td=""><td>52.4 ± 1.21</td><td>00.0 ± 22.5</td><td></td></lud<>	52.4 ± 1.21	00.0 ± 22.5	
A B	<lod< td=""><td><lod< td=""><td></td><td>160,110</td><td>44 4 . 24 0</td><td></td></lod<></td></lod<>	<lod< td=""><td></td><td>160,110</td><td>44 4 . 24 0</td><td></td></lod<>		160,110	44 4 . 24 0	
	C A B C A	C <lod A <lod B <lod C <lod A <lod< td=""><td>C <lod <lod<br="">A <lod <lod<br="">B <lod <lod<br="">C <lod <lod<br="">A <lod <lod<br="">B <lod <lod<="" td=""><td>C<lod< th=""><lod< th=""><lod< th="">A<lod< td=""><lod< td=""><lod< td="">B<lod< td=""><lod< td=""><lod< td="">C<lod< td=""><lod< td=""><lod< td="">A<lod< td=""><lod< td="">4.84*B<lod< td=""><lod< td=""><lod< td=""></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></td><td>$\begin{array}{c ccccccccccccccccccccccccccccccccccc$</td><td>$\begin{array}{c c c c c c c c c c c c c c c c c c c$</td></lod></lod></lod></lod></lod></lod></td></lod<></lod </lod </lod </lod 	C <lod <lod<br="">A <lod <lod<br="">B <lod <lod<br="">C <lod <lod<br="">A <lod <lod<br="">B <lod <lod<="" td=""><td>C<lod< th=""><lod< th=""><lod< th="">A<lod< td=""><lod< td=""><lod< td="">B<lod< td=""><lod< td=""><lod< td="">C<lod< td=""><lod< td=""><lod< td="">A<lod< td=""><lod< td="">4.84*B<lod< td=""><lod< td=""><lod< td=""></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></td><td>$\begin{array}{c ccccccccccccccccccccccccccccccccccc$</td><td>$\begin{array}{c c c c c c c c c c c c c c c c c c c$</td></lod></lod></lod></lod></lod></lod>	C <lod< th=""><lod< th=""><lod< th="">A<lod< td=""><lod< td=""><lod< td="">B<lod< td=""><lod< td=""><lod< td="">C<lod< td=""><lod< td=""><lod< td="">A<lod< td=""><lod< td="">4.84*B<lod< td=""><lod< td=""><lod< td=""></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<></lod<>	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$ \begin{array}{c c c c c c c c c c c c c c c c c c c $

Vitexin-xyloside***	А	3.99 ± 0.934	2.94 ± 0.194	3.88 ± 0.314	<lod< th=""><th><lod< th=""><th>15 h 16 min</th></lod<></th></lod<>	<lod< th=""><th>15 h 16 min</th></lod<>	15 h 16 min
-	В	3.23 ± 0.962	3.48 ± 0.367	4.62 ± 0.559	<lod< th=""><th><lod< th=""><th>15 h 36 min</th></lod<></th></lod<>	<lod< th=""><th>15 h 36 min</th></lod<>	15 h 36 min
	С	1.37 ± 0.336	1.19 ± 0.218	<lod< th=""><th><lod< th=""><th><lod< th=""><th>4 h 39 min</th></lod<></th></lod<></th></lod<>	<lod< th=""><th><lod< th=""><th>4 h 39 min</th></lod<></th></lod<>	<lod< th=""><th>4 h 39 min</th></lod<>	4 h 39 min
V-X metabolites							
V	А	1.60 ± 0.284	6.20 ± 0.593	14.2 ± 2.05	57.5 ± 5.07	<lod< th=""><th></th></lod<>	
	В	<lod< th=""><th>0.746 ± 0.093</th><th>6.96 ± 2.07</th><th><lod< th=""><th><lod< th=""><th></th></lod<></th></lod<></th></lod<>	0.746 ± 0.093	6.96 ± 2.07	<lod< th=""><th><lod< th=""><th></th></lod<></th></lod<>	<lod< th=""><th></th></lod<>	
	С	0.949 ± 0.034	2.81 ± 1.03	14.7 ± 5.78	0.710 ± 0.008	<lod< th=""><th></th></lod<>	
4-HPPA	А	<lod< th=""><th><lod< th=""><th><lod< th=""><th>24.6*</th><th>60.9 ± 7.46</th><th></th></lod<></th></lod<></th></lod<>	<lod< th=""><th><lod< th=""><th>24.6*</th><th>60.9 ± 7.46</th><th></th></lod<></th></lod<>	<lod< th=""><th>24.6*</th><th>60.9 ± 7.46</th><th></th></lod<>	24.6*	60.9 ± 7.46	
	В	<lod< th=""><th><lod< th=""><th><lod< th=""><th>36.8 ± 9.47**</th><th>32.7 ± 30.2**</th><th></th></lod<></th></lod<></th></lod<>	<lod< th=""><th><lod< th=""><th>36.8 ± 9.47**</th><th>32.7 ± 30.2**</th><th></th></lod<></th></lod<>	<lod< th=""><th>36.8 ± 9.47**</th><th>32.7 ± 30.2**</th><th></th></lod<>	36.8 ± 9.47**	32.7 ± 30.2**	
	С	<lod< th=""><th><lod< th=""><th><lod< th=""><th><lod< th=""><th><lod< th=""><th></th></lod<></th></lod<></th></lod<></th></lod<></th></lod<>	<lod< th=""><th><lod< th=""><th><lod< th=""><th><lod< th=""><th></th></lod<></th></lod<></th></lod<></th></lod<>	<lod< th=""><th><lod< th=""><th><lod< th=""><th></th></lod<></th></lod<></th></lod<>	<lod< th=""><th><lod< th=""><th></th></lod<></th></lod<>	<lod< th=""><th></th></lod<>	
PAA	А	<lod< th=""><th><lod< th=""><th><lod< th=""><th><lod< th=""><th>9.41 ± 2.47</th><th></th></lod<></th></lod<></th></lod<></th></lod<>	<lod< th=""><th><lod< th=""><th><lod< th=""><th>9.41 ± 2.47</th><th></th></lod<></th></lod<></th></lod<>	<lod< th=""><th><lod< th=""><th>9.41 ± 2.47</th><th></th></lod<></th></lod<>	<lod< th=""><th>9.41 ± 2.47</th><th></th></lod<>	9.41 ± 2.47	
	В	<lod< th=""><th><lod< th=""><th><lod< th=""><th>54.0 ± 10.1</th><th>65.4 ± 20.7</th><th></th></lod<></th></lod<></th></lod<>	<lod< th=""><th><lod< th=""><th>54.0 ± 10.1</th><th>65.4 ± 20.7</th><th></th></lod<></th></lod<>	<lod< th=""><th>54.0 ± 10.1</th><th>65.4 ± 20.7</th><th></th></lod<>	54.0 ± 10.1	65.4 ± 20.7	
	С	<lod< th=""><th><lod< th=""><th><lod< th=""><th>42.2 ± 13.0</th><th>38.9 ± 13.3</th><th></th></lod<></th></lod<></th></lod<>	<lod< th=""><th><lod< th=""><th>42.2 ± 13.0</th><th>38.9 ± 13.3</th><th></th></lod<></th></lod<>	<lod< th=""><th>42.2 ± 13.0</th><th>38.9 ± 13.3</th><th></th></lod<>	42.2 ± 13.0	38.9 ± 13.3	

<LOD (limit of detection)</p>
* the metabolite was only detected in one sample out of the triplicate fermentation
** the metabolite was only detected in two samples out of the triplicate fermentation
*** Vitexin-xyloside was quantified via Vitexin-rhamnoside

FIGURE GRAPHICS

Figure 1



Kaempferol: R₁ & R₂ & R₃ = OH Kaempferol-3,4'-O-diglucoside-7-O-rhamnoside: R₁ & R₂ = glucosyl, R₃ = rhamnosyl Tiliroside: R₁ = OH, R₂ = glucosyl-6''-trans-p-coumaroyl, R₃ = OH





Apigenin

Vitexin: $R_1 = OH$ Vitexin-2"-O-glucoside: $R_1 = glucosyl$ Vitexin-2"-O-rhamnoside: $R_1 = rhamnosyl$ Vitexin-2"-O-xyloside: $R_1 = xylosyl$





















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