Rare-metal granites as a potential source of critical metals: a geometallurgical case study

Quentin DEHAINE^{1,2,*}, Lev O. FILIPPOV¹, Hylke J. GLASS² and Gavyn ROLLINSON²

¹GeoRessources, Université de Lorraine, CNRS, UMR 7359, 2 rue du Doyen Marcel Roubault, TSA 70605, F54518,

Vandœuvre-lès-Nancy, France

² University of Exeter, Camborne School of Mines, Penryn, Cornwall, TR10 9FE, United Kingdom

*Corresponding author: O.Dehaine@exeter.ac.uk Phone: +44 (0) 1326 253602

Abstract

14 Because of their low grades in critical metals such as Light Rare Earth Elements (LREE) or Sn, rare-metal 15 granites are not considered as economic for metal recovery but, when altered, they are often exploited for their 16 industrial minerals. The St Austell rare-metal granite is well known for its world-class kaolin deposits which 17 formed as a result of the extensive weathering and alteration of the underlying granite. The St Austell granite body 18 is composed of several granite components, each having its own accessory minerals assemblage. As a result of the 19 kaolinisation process, some metal-bearing accessory minerals of the granite, such as monazite (LREE) or 20 cassiterite (Sn), are partially liberated from the gangue which allow their pre-concentration in the micaceous 21 residue which is considered as a potential source for critical metals recovery. A geometallurgical approach is 22 fundamental for this unconventional 'resource', not only because of its nature but also because the target metals 23 must be considered as by-products. Similarities with other similar rare-metal granites suggest that topaz granite is 24 the most prospective for disseminated magmatic Sn-Nb-Ta-REE mineralization. However, comparison of the 25 potentiality of 3 granite types *i.e.* biotite, topaz and tourmaline granites suggest that biotite granites is actually the 26 most prospective due to higher degree of kaolinisation of the biotite granite which favour pre-concentration of its 27 accessory mineral in the micaceous residue. In order to develop a geometallurgical framework for extraction of 28 kaolin and metals from the selected granite component, a field sampling campaign is performed. Core samples are 29 processed in the laboratory using a characterisation program that mimics the full-scale kaolin refining route. Two 30 main products are recovered through this program, viz. GT53MR180 (-180+53 µm) and R5P5 (-5 µm), which 31 correspond to a fine micaceous residue and a fine kaolin product respectively. These products are both analysed 32 routinely for major and minor trace elements by XRF and yields are recorded to indicate process performance. A 33 selected number of GT53MR180 samples are also being characterised in terms of particle size by laser light 34 scattering, geochemistry by ICP-MS, and mineralogy by QEMSCAN[®]. Comparison of characterisation results of 35 GT53MR180 samples and corresponding industrial residue samples shows a good correlation, suggesting that 36 sample analyses are representative for the in-situ deposit and the processing behaviour. Monazite is found to be 37 either fully liberated or fully locked from one sample to the other. Next, pilot-scale gravity concentration tests are 38 performed on micaceous residue samples. Characterisation of the processing products shows that monazite lost in 39 the tailings is mostly locked within tourmaline or micas and is fine grained. Then, predictive regression models 40 for spiral separation performance in terms of recovery, product grade and enrichment as a function of the feed 41 grade are developed for GT53MR180 LREE grade data. Finally, kaolin resources can be classified using 42 quantitative indicators such as yield of the R5P5 product and the iron oxides content which provides insight into 43 the kaolin quality in terms of whiteness. This geometallurgical classification can be used to delineate zones of 44 interest within the deposit. Although kaolin quality and recovery primarily inform extraction planning, zones 45 which are also of interest for metal recovery can be identified. The proposed model predicts whether the expected 46 LREE grade and recovery satisfy the by-product requirements. 47

48 Key words: kaolin residue, Light Rare Earths, critical metals, rare-metal granites, geometallurgy, QEMSCAN[®]. 49

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1. INTRODUCTION

56 Ensuring sustainability all along the raw materials value chain has been a growing concern for the extractive industry in recent years. While more efficient recycling of materials will be achieved in the 57 58 foreseeable future, as depicted by the concept of circular economy, recycling of some materials suffers 59 practical limitations which are not exempt from natural resources extraction with regards to our growing 60 societal needs. This is particularly true when it comes to critical metals which have a low recycling rate 61 and for which there is no substitute. These critical metals have been highlighted by a recent initiative by 62 the European Commission defining the criticality of a material based on three main indicators, *i.e.* the 63 supply risk, the economic importance and the environmental country risk. A list of 14 economically 64 important Critical Raw Materials (CRMs) was published in 2010, recently extended to 20 (European 65 Commission, 2014) which include Niobium (Nb) and the Light Rare Earth Elements (LREEs). For some of these metals, mined as by-product, the availability is also conditioned by the major metal of the ore 66 67 in which they occurred usually in low concentration (Mudd et al., 2016). Rare Earth Elements (REEs) 68 for instance, are mainly recovered as co-products/by-products of certain other minerals in many mining 69 operations (Kumari et al., 2015). As of 2018, there are no more active mines in the western world 70 operated exclusively for the recovery of Rare Earths. REE projects outside China, which re-opened or 71 opened after the 2011 spike in REE prices, such as the Mountain Pass mine and the Bear Lodge deposit 72 (United States), or the Mount Weld mine (Australia), have all sustained or suspended their production 73 or development due to relatively low REE prices (Imholte et al., 2018). This situation has stimulated 74 other countries to look for alternative REE resources and to develop their own REE industry, in 75 particular by looking into recycling of REE from End-of-Life products. However, much less attention 76 has been paid to historical landfilled stocks and freshly produced flows of REE-containing industrial 77 process residues which usually present much lower REE concentrations but are present in very large 78 volumes (Binnemans et al., 2013). Total amounts of REE locked in these residues are also very large 79 and may secure an independent source of REE as well as shielding resource-poor countries from export 80 quotas and price fluctuations (Binnemans et al., 2015). Industrial mineral wastes, such as those from the 81 kaolin industry in Cornwall, UK, have proven to be a potential source for REE, Sn and Nb recovery (Dehaine et al., 2017; Dehaine and Filippov, 2015a; Filippov et al., 2016; Scott et al., 1998). The kaolin 82

deposits are formed by weathering and alteration of granite rocks. As a consequence of the kaolinisation
process, accessory minerals of the granite are partially liberated from the gangue. Hence, during refining
of kaolin ore, spontaneous pre-concentration of the metal-bearing accessory minerals, such as monazite
hosting LREE (*i.e.*, cerium (Ce), lanthanum (La) and neodymium (Nd)), cassiterite hosting Sn or rutile
hosting Nb, occurs in the micaceous residue (Dehaine and Filippov, 2015a).

88 In order to convert such an unconventional resource into a potential REE resource, development of 89 innovative technologies and geometallurgical approaches an integrated approach to mine value chain 90 optimisation are is required on a strategic and tactical level. Geometallurgy is a discipline that seeks to 91 reduce technical risks through advanced ore characterisation which allows maximisation of resource 92 efficiency and minimisation of energy consumption and waste generation. This could be expressed in 93 the creation of a spatial predictive model which informs optimal extraction, processing and waste 94 disposal. combines geological and metallurgical information to create a spatially-based predictive model for mineral processing plants in order to reduce technical risk (Lamberg, 2011). This aims to 95 document The predictive model accounts for variability within the orebody -within the orebody and 96 97 quantify the impact of some geological, mineralogical, chemical and geotechnical factors that impacton 98 99 geometallurgical domains (Cabri et al., 2017). The database thus generated can be readily integrated 100 into a 3D model (Williams and Richardson, 2004). Thus, it is necessary to develop simple and effective 101 characterisation methods of the distinct processing products/residues. This is particularly true when it 102 comes to industrial minerals such as kaolin for which the final product is a mineral concentrate rather 103 than a mono-element product (Glass, 2016). Geometallurgy can be broadly split into two key levels: 104 strategic geometallurgy, which focuses on the whole orebody and long-term life-of-mine scale and 105 *tactical geometallurgy* which relates to the short- to medium-term scale during daily operation (McKay 106 et al., 2016).

107 Such a strategicn approach is fundamental for the unconventional 'resource' investigated in this 108 <u>feasibility</u> work, not only because of its nature but also because the target metals must be considered as 109 *by-products*. Applying a geometallurgical approach to this case study is rather challenging for several 110 reasons. First, there are no a priori certainties on the economic feasibility of metal by-product recovery 111 and there are no known specifications for a marketable product consisting of REE. Next, if the operation 112 is already in production, there is no processing on site for metal recovery from micaceous residues. This influences the development of a hypothetical geometallurgical model: most geometallurgical programs 113 114 are applied to more conventional deposits already in operation for which a significant amount there is no 115 historical process of data are available, with large numbers of nor a geometallurgical variables database 116 allowing the use of multivariate modelling (Boisvert et al., 2013). Finally, given the complexity of the 117 investigated deposits, with multiple granite-types each one having its own accessory mineral 118 assemblage, this approach could not consider the whole St Austell granite but will have to focus on a 119 specific granite type, ore-type or location. The representativeness of the results that will be obtained is 120 also a key feature for a potential application - Hence, there is a need to develop performance prediction 121 models to evaluate process efficiency and product characteristics (metal grades and recovery) therefore 122 assessing the critical metals by-product potentiality. The proposed approach applied to a typical 123 kaolinised rare-metal granite seek to feed the initial strategic/feasibility work on similar deposit with the objective of supporting tactical geometallurgy during kaolin production. 124 125 126 2. GEOLOGICAL BACKGROUND 127 The Cornubian granite province is well-known for its historically important deposits of tin and base 128 metals (Manning and Hill, 1990), and for its current international importance as a major kaolin producer. 129 130 2.1. The Cornubian Sn-W province 131 Tin is distributed on the continents in relatively narrow continent-sized belts (Schuiling, 1967). For 132 instance, most of tin deposits across Europe are related to Variscan granites. Figure 1 Figure 1 illustrates 133 the multi-scale spatial relationship between Variscan granites and Sn-W deposits from the Variscan belt 134 to the St Austell district. The major tin belt passes from the Cornubian massif (Redruth, Camborne, etc.) 135 in UK through the French Massif Central (Echassières) and continues towards Italy with tertiary tin 136 deposits (Elbe, Tuscany and Scilly). Other tin belts are also related to Variscan granites, like the one 137 which pass from Northern Portugal (Panasqueira) through Spain or the one around the Bohemian massif

138 (Cinovec-Altenberg). Most of these tin deposits are associated with Sn-W cupolas located at the top of

139 tardi- to post-orogenic plutons emplace at shallow depths (Jébrak and Marcoux, 2008).

The batholith of SW England was formed during the late Variscan orogeny in the Late Carboniferous to Early Permian (270-300 Ma) intruding deformed Devonian-Carboniferous terrestrial to marine sediments. The batholith has a 200-km-long extension (Willis-Richards and Jackson, 1989), comprising six major and several minors granite bodies. The major bodies, from east to west, are Dartmoor, Bodmin Moor, St Austell, Carnmenellis, Land's End and the Scilly Isles (<u>Figure 1Figure 1</u>b).

145 The outcrops are dominated by biotite-bearing monzogranite, with minor intrusions of tourmaline-146 bearing aplites, pegmatites and local fine-grained facies (Manning et al., 1996). In addition, there are 147 some occurrences of other granite types, resulting from multiple intrusive episodes, such as the lithium-148 mica granites or the topaz-bearing granite in the Tregonning-Godolphin Granite, and the St Austell 149 Granite (Manning et al., 1996; Manning and Exley, 1984; Manning and Hill, 1990). The batholith was 150 affected by several episodes of alteration including quartz-tourmaline veining associated with 151 greisening, intrusion of rhyolite dykes, guartz-hematite veining. Kaolinisation is the last alteration event, 152 which is believed to have a meteoric origin (Psyrillos et al., 1998; Sheppard, 1977), and was relatively 153 extensive in the western part of the St Austell pluton where the majority of the active kaolin pits are 154 located (Figure 1Figure 1b).

155 According to Willis-Richards & Jackson (1989) there is a close spatial association between the 156 batholith and the rich Sn, Cu mineralisation of the Cornubian Peninsula. Most of the ores of Sn, Cu, W, 157 Zn and As came from vein deposit type mineralisation, along or parallel to the axis of the batholith 158 (Moon, 2010). Within the St Austell granite, zones of intense kaolinisation are spatially related to 159 swarms of steeply dipping, quartz-tourmaline±cassiterite±wolframite sheeted veins with greisen 160 (quartz-muscovite-tourmaline±topaz±wolframite) alteration selvages, observable at Goonbarrow pit 161 (Bray and Spooner, 1983). These mineral lodes are located South, North, or to a lesser extent within the 162 St Austell granite kaolinised area (Figure 1 Figure 1 c and Figure 2 Figure 2).

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164 2.2. The St Austell rare-metal granite

Rare-metal granites are so-called because of their mineralisations being disseminated through the granites rather than concentrated in layers, lodes, etc. The St Austell rare-metal granite share numerous

167 characteristics with other granites of SW England, summarised in many reference papers and textbooks 168 (Floyd et al., 1993; Manning, 1996; Manning and Hill, 1990). These granites were originally described 169 by four different granite types, all observed in distinct SW England granites: biotite granite, lithium 170 mica granite, the tourmaline granites and topaz granite (Hill and Manning, 1987). Additional granite 171 type were added later based on field and textural observations within the St Austell granite, indicating 172 complex late-stage magmatic and hydrothermal processes (Manning et al., 1996). The St Austell rare-173 metal granite is composed of six major granite types, each unit being characterised by a specific mineral 174 assemblage and textures (Manning et al., 1996), see Figure 2Figure 2.

175 The *biotite granite* (BG) is the main lithology, which represent up to 70% of the outcrop of the St 176 Austell granite, exposed in the Western Area, at the eastern part of the pluton and intermittently within 177 the Central Area. It corresponds to the megacrystic biotite granite observed at Land's End and Dartmoor 178 granites (Manning et al., 1996; Manning and Exley, 1984), see Figure 1Figure 1b. It is coarse grained, 179 mainly composed of quartz, K-feldspar and micas with tourmaline and topaz as minor phases. Biotite is 180 the main mica, but muscovite is also present. The main accessory phases have been reported as rutile, 181 topaz, apatite, monazite, cassiterite, zircon and uraninite (Manning et al., 1996). The main host for the 182 LREE differs depending on the authors. Works of Jefferies, (1985) on Carnmenellis biotite granite 183 describe monazite as the main LREE-bearing mineral accounting for approximately 75 % of the total 184 LREE content. In contrast microprobe analyses on samples from Cornubian granites by Alderton et al. 185 (1980) show significant concentrations (up to 0.5%) of La, Ce, Nd and Sm in other phases, particularly 186 apatite, but also zircon and titanite.

The *lithium mica granite* (LMG) has a similar grain size and texture to the biotite granite (Manning et al., 1996). It is characterised by the presence of lithium mica (zinnwaldite) and plagioclase (albite component). The accessory minerals are present as inclusion in zinnwaldite (apatite, monazite, zircon and rutile) or in plagioclase (apatite, fluorite and secondary micas).

The tourmaline granites were subdivided by Manning et al. (1996) in a *Tourmaline granite* (TG), itself regrouping a globular quartz and medium coarse-grained equigranular facies, and in a *fine-grained tourmaline granite* (FGTG) with fine-grained (<0.5 mm) equigranular facies. The globular quartz facies is characterised by a considerable textural variation. It contains large quartz grains, phenocrysts of micropethite, zinnwaldite with a fine-grained groundmass of quartz, K-feldspar, plagioclase, zinnwaldite, tourmaline and topaz. The accessory phases are limited to apatite and rutile. On the other hand, the fine-grained facies is an equigranular rock with euhedral albite and zinnwaldite. Accessory minerals include monazite, apatite, zircon and rutile with some occurrences of arsenopyrite and cassiterite.

The *topaz granite* (TZG) is medium-grained, characterised by euhedral-subhedral fluorine-rich topaz, interstitial lithium mica, albite, plagioclase feldspar with perthitic orthoclase, and subhedral quartz (Manning and Hill, 1990). It contains a smaller proportion of accessory phase but a wider variety including sub-economic minerals (Manning et al., 1996). The accessory minerals are apatite, amblygonite (LiAlPO₄F), zircon, Nb-Ta oxides (columbite-tantalite and ilmenorutile) and uraninite. The mineralogy of the six granites is summarised in <u>Table 1Table 1</u>.

206 Apart from these magmatic disseminated mineralisations, Sn-W mineralisations associated with 207 guartz-tournaline sheeted veins previously described, are also present within the St Austell granite 208 (Bray and Spooner, 1983). Whereas two different types of mineralisations, a Sn breccia-stockwork in 209 the Treliver area, generally unenriched in base metals, and a E-W veins complex enriched in base metals 210 have been identified north of the St Austell pluton (Camm and Dominy, 1999; Camm and Moon, 2001). 211 In addition, Müller and Halls (2005) describe an intrusive tourmaline breccia in biotite granite at Wheal 212 Remfry where zoned rutile is the main host mineral for Sn (up to 1.88%) and exhibits high W (up to 213 1.95%) and Nb (up to 2.05%) domains.

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215 2.3. Similarities with rare-metal granites worldwide

The spatial relationship between critical metals mineralisations and Variscan granites previously discussed is confirmed by the numerous similarities between the St Austell rare-metal granite and several rare-metal granites worldwide. From all the magmatic units of the St Austell pluton, the topaz granite is the youngest, and the most geochemically different, relatively enriched in Li, Rb and with a much higher Nb/Zr ratio (Manning and Hill, 1990). Nonetheless, there are close similarities between the mineralogical assemblages described above and the accessory minerals assemblage of evolved granite elsewhere described as peraluminous high to intermediate phosphorus granites (Linnen and Cuney, 223 2005; Simons et al., 2017). All the St Austell granite types are felsic, weakly peraluminous granites 224 (Manning et al., 1996). The comparison of the P_2O_5/SiO_2 ratio of the St Austell granites with other rare-225 metal granites show that the St Austell granites belong to the intermediate phosphorous granite (IHP) 226 type (Figure 3Figure 3). The relatively low Zr and Th contents of St Austell granites, visible in Figure 227 <u>3Figure 3</u>b are characteristic of non-peralkaline rare-metal granites (Linnen and Cuney, 2005).

228 These granites are formed in a late-orogenic context and associated with orogenic belts (Figure 229 1Figure 1b) and are usually characterised by highly differentiated leucocratic two-micas granites, 230 corresponding to the S-type, rich in hydroxylated minerals (muscovite) and enriched in lithophile 231 elements such as Nb, Ta, Li, Be, P, F (Jébrak and Marcoux, 2008). Among these granites some display 232 similarities with the St Austell granite and are associated with Sn, W, and Ta-Nb disseminated 233 mineralisations (Černý et al., 2005). Some of this granites are associated to kaolin deposits which are 234 also known to display similarities with Cornish kaolin deposits (Wilson et al., 1997; Wilson and Jiranek, 235 1995).

236 According to Scott et al., (1998), the Yichun granite in China is one of the closest rare-metal granite, 237 in terms of lithology. It comprises also several magmatic units including a biotite granite and a Li-mica 238 (lepidolite) granite such as the St Austell granite. The Li-micas granite is the most evolved magmatic 239 unit and display a similar accessory assemblage of amblygonite/montrebasite, apatite, fluorite, Mn-240 tantalite, microlite, Ta-cassiterite, ilmenite, monazite, zircon, sphalerite and topaz (Belkasmi et al., 241 2000; Yin et al., 1995). The Beauvoir topaz-lepidolite albite granite (Massif Central, France) is also very 242 similar to the St Austell topaz granite and display a Sn-Li-Ta-Nb-Be disseminated mineralisation with an accessory minerals assemblage composed of topaz, apatite, amblygonite, columbite-tantalite, 243 244 microlite, zircon, uraninite and sphalerite (Cuney et al., 1992; Raimbault et al., 1995). Further 245 similarities exist with the Podlesi granite in Czech Republic which is composed of three magmatic units, 246 including biotite granite, and two Li-mica granites. All granite type contain disseminated Ta-Nb-Ti-W-247 Sn minerals including rutile, cassiterite, Fe-columbite, ixiolite and ferberite (Breiter et al., 2007). The Younger Granites province of Nigeria displays also some similarities with the above-cited rare-metal 248 249 granite complexes, especially the subalkaline granites which are mineralised with rich deposits of 250 cassiterite, columbite, sphalerite, wolframite and galena and are the primary source of alluvial cassiterite

and columbite mineralisation (Ogunleye et al., 2006).

252 By analogy with the Beauvoir granite it is possible to suggest the most prospective granite type within 253 the St Austell pluton for disseminated magmatic mineralisation. Rare-element chemistry of the Beauvoir 254 granite shows that the concentration of most metallic elements including Nb, Ta, Sn and W increases with the Li contents from the bottom to the upper part of the granite body (Cuney et al., 1992). As 255 256 discussed above, the St Austell topaz granite exhibits the greatest enrichment in Li and is 257 petrogenetically distinct from other granite types (Manning and Hill, 1990) with mineralogical and 258 geochemical characteristics matching closely those of other rare metal granite worldwide. Thus, the St 259 Austell topaz granite is considered to be the most prospective for REE and Sn-Nb-Ta mineralisation. 260 The top of the Beauvoir Granite cupola is the most enriched part of the orebody in terms of rare-metals 261 due to a combination of magmatic process and interaction with meteoric water (Raimbault et al., 1995). 262 It is therefore the upper parts of the topaz granite that should have the greatest potential for disseminated 263 magmatic mineralization. The exact morphology and the intern organisation of the topaz granite is poorly constrained but Manning and Hill (1990) suggested that the individual outcrops of topaz granite 264 265 may be connected at depth. The field relations indicate that the Nanpean stock represents a position well 266 below the roof of the granite body and that the Hensbarrow stock represents the roof. Thus, from a pure metallogenic perspective, the Hensbarrow stock, which represents the upper part of the St Austell topaz 267 268 granite, is the most prospective for REE and Sn-Nb-Ta magmatic mineralisations.

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3. ST AUSTELL KAOLIN OPERATIONS

271 China clay, also called kaolin, is a commercial clay material mainly composed of kaolinite 272 $(Al_2Si_2O_5(OH)_4)$, a hydrated aluminosilicate clay mineral. Its main uses are for coating and filling in 273 paper industry (75% of world's production), the ceramics industry but it has also a wide variety of 274 speciality uses, *i.e.* for plastics, ink, paint, rubber, and pharmaceutical industry. The composition of the 275 granites has a significant influence on the quality of the kaolin ore (Scott et al., 2002). The commercial 276 value of a kaolin product is based on its whiteness and its fine particle size. Particle size affects fluidity, 277 strength, plasticity, colour, abrasiveness and ease of dispersion. Other important properties include its flat particle shape, which increases opacity or hiding power, its soft and non-abrasive texture, due to the
absence of coarser impurities, and its chemical inertness (Highley et al., 2009).

The UK is one of the world's largest producers and exporters of kaolin, after the USA, Brazil and China, supplying up to 1.36 million dry tonnes of clay per annum, in 2008 for the paper, ceramics, paints, plastics and rubber industries (Highley et al., 2009). Production is supplied from a large number of open pits scattered throughout the western part of the granite outcrop within a complex landscape of spoil heaps and mica lagoons as shown in <u>Figure 2Figure 2</u>. Kaolinite is a secondary mineral formed by weathering and hydrothermal alteration of the alumina silicates of host rock, mostly feldspar minerals, by acid aqueous fluids to form an aggregate of kaolinite and secondary micas.

287 Kaolin extraction has traditionally been done by hydraulic mining in which high-pressure jets of 288 water are used to disaggregate the weak, kaolinised granite and disperse the kaolinite particles, together 289 with the other components of the granite, into a slurry, leaving behind the less well decomposed material 290 in the form of coarse sand and gravel, locally called "stent". The resulting "clay stream" flows into the 291 pit sink and is then pumped to the in-pit processing loop More recently, 'dry' mining has been 292 introduced, allowing more selective extraction and improving yields (Highley et al., 2009). The plant is 293 fed with clay matrix transported directly from the pit by cut and carry operations without any washing. 294 Hence, the material fed into the plant is much coarser and more classification stages are required to 295 remove the coarsest materials (Ano, 2008). The first stages of kaolin ore processing consist of removing 296 the coarsest fractions of the clay matrix which are not valuable for kaolin recovery. Thus, the kaolin 297 refining plant can be seen as a classification process producing different size classified wastes and a fine 298 kaolin product (Figure 4Figure 4). The first sized-classified wastes *i.e.* crushed stones (-21+11 cm), 299 gravels (-110+8 mm) and sands (-8+0.5 mm), are removed by screening and are sent to stockpiles. The 300 last classification stage is realised by series of hydrocyclones. The finer particles (overflow) go to the 301 product stream for further refining whereas the coarser particles (underflow) are reported to the 302 micaceous residue where they are held in suspension and pumped to a tailing dam. Additional techniques 303 are used to improve the brightness (whiteness) and particle size of specific grades of clay. These include 304 blending, fine grinding, chemical reductive bleaching and/or the removal of iron-bearing impurities 305 using magnetic separation. Some kaolin products are also calcined at specific temperatures to give 306 different products.

The gravity processing route presented in Figure 4Figure 4 corresponds to an hypothetical critical metal by-product recovery route which have been tested on pilot-scale and described in Dehaine (2016). It starts by a screening of the micaceous residue to remove the coarser fraction (+180 μ m), which is not treated, and preparation of the feed for the gravity concentration (-180+53 μ m) and flotation routes (-53 µm). The proposed gravity concentration route comprise one spiral concentration step followed by a refining of the spiral concentrate by shaking table concentration. For details about the flotation route see Dehaine (2016) and Filippov et al. (2016).

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4. MATERIALS AND METHODS

316 4.1. Core samples

All the core sample data used in this work are from a sonic drilling campaign undertaken in 2011 by Imerys Minerals Ltd, UK. The dataset selected for this study covers only the Melbur-Virginia and Wheal Remfry pits (referred as pit A and pit B hereafter) located at the western part of the St Austell granite (see <u>Figure 2</u>Figure 2) and comprise around 1200 core sample collected every 3 meters over 100 drill holes.

322 *4.1.1. Core samples processing protocol*

323 The main objective of this sampling campaign was to assess kaolin resources and alternatively 324 critical metals resources, in terms of yields and chemical composition of the potential end-products 325 through a laboratory-scale sample processing program that mimics the full-scale kaolin refining route 326 (Figure 5Figure 5). Samples The whole core samples were crushed to 15 mm if required (50 mm top 327 size) in a jaw-crusher and then stirred/mixed with water to liberate any kaolin from the kaolinised 328 granite. At this stage, the obtained material is equivalent to the output of the washing barrel of the plant 329 (stream 5, Figure 4Figure 4). The slurry is then classified on a vibrating 180 µm screen as with the 330 classification stages realised by the bucket-wheel de-sander/ 660 mm cyclones to obtain a -180 μ m 331 material (stream 12, Figure 4Figure 4). This is followed by a screening of the -180 µm material on a 332 vibrating 53 µm screen as performed in the full-scale plant by primary cyclones. The -53 µm material 333 is then treated to remove the $-53+15 \,\mu m$ material by sedimentation. If required, a magnetic separation was then applied to the -15 µm material to improve brightness/colour as applied in the refining stages 334 of the industrial process. Finally, the -15+5 µm fractions are removed by sedimentation and a -5 µm 335 336 high quality kaolin product is recovered. In the end 3 products are recovered, viz. GT53-MR180 (-337 180+53 μ m), R15-P15 (-15+5 μ m) and R5-P5 (-5 μ m) corresponding to the -180+53 μ m fraction of the 338 micaceous residue fed to the gravity concentration route (stream 23, Figure 4Figure 4), a rough (-15 µm) kaolin product and a fine (-5 µm) kaolin product (stream 16, Figure 4Figure 4), respectively. Each 339 product is weighed at each step of the process, so that the yield of each product is known. All the final 340 341 products were then analysed by X-Ray fluorescence (XRF) for major elements whereas ProTrace XRF 342 analyses (analyses for trace elements) were only carried out on the -180+53 µm (GT53MR180) and the 343 -5 μ m (R5P5) fractions.

344 *4.1.2. Core sub-sampling*

345 In order to control the accuracy of the XRF data for the LREE and other metals a set of 30 346 GT53MR180 samples from the set of core samples covering the western part of the biotite granite, *i.e.* 347 kaolin pits A and B, have been selected to be analysed by ICP-MS for calibration and by OEMSCAN[®] 348 for mineral characterisation. All these samples correspond to relatively-high to high grade kaolin ore, 349 *i.e.* completely altered granite. The linear regression models for Nb and LREE prediction, obtained with 350 process samples in Dehaine and Filippov (2015), suggest that TiO₂ can be used as a "tracer" for Nb and 351 LREE. Moreover, TiO₂ is generally well quantified by XRF analysis. Hence, sub-samples have been selected among all the available core samples following a procedure based on their TiO₂ content to have 352 353 ato ensure that the obtained sub-set representative sub-set displays the same variability as the original 354 dataset (Dehaine, 2016). In the end, only 26 usable samples were recovered due to technical issues 355 (contamination, losses).

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4.2. Process samples for metal accounting and for gravity concentration testwork

358 The samples used for metal accounting were collected and aggregated from daily sampling over one 359 week from every output stream of three distinct kaolin refining plants processing ore from Virginia360 Melbur, Wheal- Martyn or Treviscoe pit. These samples are considered to be typical material from the 361 biotite, tourmaline and topaz granite respectively. For each location, the batch of samples were 362 composed of waste samples and a kaolin product, *i.e.* 5 to 6 representative samples each, depending of 363 the mining technique (dry or hydraulic mining). Each sample is a composite, aggregated from increments collected on-site from moving streams or belts, for details about the sampling procedure see 364 365 (Dehaine, (2016). To ensure an optimised sampling procedure in terms of frequency, sampling mode 366 and number of increments for all the critical metals, a multivariate variographic approach has been 367 developed to determine the required number of increments (Dehaine et al., 2016; Dehaine and Filippov, 368 2015b).

369 Process samples used for gravity concentration are screened (-180+53 µm) micaceous residues 370 corresponding to the spiral feed described in Figure 4Figure 4. Micaceous residue samples were 371 collected from the hydrocyclone underflow tank of the full-scale plant while treating material from the 372 same origin as the core samples, *i.e.* kaolin ore from pits A and B. Residue samples were collected using 373 a by-pass to divert the whole stream into the sample collector, following the protocol described in 374 Dehaine et al. (2016). The large micaceous residue samples collected were then screened in two 375 successive steps, using a vibrating Roto-Sieve apparatus model RS60 (Chauvin SA, France) in order to 376 recover the -180+53 µm size fraction of the residue for the pilot scale gravity concentration testwork. 377 The sieves opening have been selected based on the washability of the materials (Dehaine et al., 2017), 378 in order to improve the efficiency of the gravity concentration, but also to be able to use data from the 379 GT53MR180 core sample products s-to characterise the feed properties of the gravity route with the 380 objective of establishing process performance prediction models for core sample data.

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4.3. Pilot-scale gravity concentration testwork

The influence of the LREE grade in the feed on spiral concentration efficiency has been tested with a 5-turn MKIIA Reichert spiral (Mineral Deposit Limited, Australia) using the setup described in Dehaine et al. (2017) in scavenging configuration. Operating condition are set at 0.12 m³.h⁻¹ wash water flowrate and 15% solid pulp density to favour recovery (Dehaine et al., 2017), concentrate splitters positions are adjusted visually for the first test but remains fixed for all the others. Spiral concentrates are then processed using a No.13 Wilfley laboratory shaking table (Holman-Wilfley, United-Kingdom)
following the flowsheet presented in <u>Figure 4Figure 4</u>. The characterisation results for the shaking table
presented in this paper have been obtained using the shaking table products obtained from test No.4 in
Dehaine et al. (2017), which display the best separation performance.

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393 4.4. Particle size analysis

Particle size analysis was performed by laser light scattering using a Helium-Neon Laser Optical System Mastersizer 3000 (Malvern instruments Ltd.) coupled with a Hydro Extended Volume (EV) sample dispersion unit. Required amounts of sample were introduced in a beaker inside the dispersion unit, equipped with a dip-in pump and a stirrer, to reach a 20% obscuration level. The particle size distributions obtained are the average of five duplicate analyses.

399

400 **4.5. Chemical analysis**

401 Samples were crushed and riffled alternatively in accordance with the theory of sampling to obtain 402 representative subsamples suitable for whole rock analysis. Powdered samples (300 mg) are fused in Pt 403 crucibles along with ultra-pure LiBO₂ at 980 °C and dissolved in nitric acid. The chemical analysis of 404 the obtained solutions were carried out by Inductively Coupled Plasma Optical-Emission Spectrometry 405 (ICP-OES, Thermo Fischer ICap 6500) for major elements (Si, Al, Fe, Mn, Mg, Na, Ca, K, Ti, P) and 406 Mass Spectrometry (ICP-MS, Thermo Elemental X7) for 42 trace elements, following the routine 407 procedure of liquid chromatography ICP-MS described in Carignan et al. (2001), at the Service 408 d'Analyses des Roches et des Minéraux (SARM-CNRS, Nancy, France).

409

410 **4.6. Mineral analysis**

For mineralogical data samples were analysed using QEMSCAN[®]. Samples were randomly subsampledriffled, mixed with graphite powder to promote particle dispersion and reduce settling bias (Pascoe et al., 2007) and then prepared into 30 mm diameter epoxy resin blocks. The sample face of each block was then polished to a 1 micron finish using diamond solutions and carbon coated with 25 nm layer, prior to analysis. Samples were analysed using a QEMSCAN 4300 system at the Camborne 416 School of mines, University of Exeter, UK (Andersen et al., 2009; Anderson et al., 2014; Rollinson et 417 al., 2011) using PMA mode at a resolution of 5 microns, and a target of 5000 particles per sample (see 418 Goodall et al., 2005; Gottlieb et al., 2000; Pirrie and Rollinson, 2011, for detail on the PMA 419 measurement mode). As the samples were well mixed and carefully prepared, the 5000 particles 420 provided a representative sub-population of each sample. Data measurement used iMeasure 4.2 SR1 and 421 data processing used iDiscover software 4.2SR1 and 4.3. Data processing involved checking the data, 422 customizing the database to the match samples, adding and improving SIP (database) categories as 423 required, whilst considering the details of the technique outlined in Rollinson et al. (2011). Modal 424 mineralogy, mineral association, average grain size, liberation data and images were output for the 425 samples.

Additional textural observations were performed on a Hitachi FEG S4800 scanning electron
microscope at the *Service Commun de Microscopie Electronique et Microanalyses* (SCMEM-UL,
GeoRessources, Nancy, France) with an imaging voltage of 20 kV.

429

430 4.7. Methodology

431 The purpose of this work is to investigate and develop preliminary geometallurgical tools for by-432 product resource estimation of critical metals within the St Austell kaolin deposits with a focus on 433 LREE. The methodology used in this study follows three main steps which aim to populate the core 434 sample database: (1) to characterise the GT53MR180 core samples through QEMSCAN analysis, to 435 control, and calibrate if necessary, the trace element grades using duplicate ICP analyses (not presented 436 here, see Dehaine (2016) for details), then (2) to integrate metallurgical parameters through regression 437 models (e.g. spiral performance prediction models) applied to the GT53MR180 trace element data. 438 Finally (3), to use the core samples data along with the regression models to propose a simple preliminary geometallurgical classification scheme to help by-product resources estimation of critical 439 440 metals.

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5. RESULTS

444 **5.1.** Comparison of the granite types

Given the range and diversity of accessory minerals within the different granite types present at St Austell, it is likely that each granite type will have<u>a</u> the same<u>different</u> potential as for by-product recovery of critical metal. In order to define the most prospective granite type for by-product metal recovery, one must compare the metal content of all the output streams of the kaolin processing route and the associated metal accounting. This is done for three distinct kaolin refining plants processing ore from weathered biotite, tourmaline or topaz granite.

451 Previous studies on kaolin ore originating from biotite granite have shown that metals are pre-452 concentrated in the micaceous residue, as a consequence of the preferred separation of the heavy 453 minerals in the hydrocyclone underflow (Dehaine and Filippov, 2015a). Comparison of the critical metal 454 content of the output streams from the 3 investigated locations is presented in Table 2. Most of the 455 metals, with the exception of LREE, display a similar trend than observed for the biotite granite, with 456 an increasing grade from the coarsest waste to the micaceous residue which display the highest overall 457 grades. The LREE grade however is slightly higher in the product stream displaying the highest LREE 458 grade for topaz and tourmaline which means that LREE are more concentrated in the finer fractions of 459 the clay matrix.

In terms of distribution within the outputs streams, most of the metals are distributed in the coarsest streams (stent and sand) for topaz and tourmaline granite, or the micaceous residue for biotite granite (Figure 6Figure 6). Most of the metals display a less contrasted distribution for tourmaline and topaz granite. Indeed Sn, Nb and Ta are slightly more distributed in the sands then equally distributed between stent and the micaceous residue whereas the LREE show a more important proportion in the stent and product stream. This could be a consequence of the hydraulic mining processes as the stent fraction also contains fine material which sticks to the boulders or gravels.

The overall LREE and Sn content are significantly higher for the biotite granite than for the topaz and tourmaline granite. Nb and Ta content in the micaceous residue are of the same order of magnitude for the 3 locations, whereas the W content is three times lower for the topaz granite. This highlightsThis highlight strong differences in the metal content of the kaolin ore originated from different granite types. This is confirmed by the recalculated feed (clay matrix) contents which clearly shows higher LREE, Sn 472 and W grades in the biotite granite clay matrix while Nb-Ta grades are slightly higher for topaz granite 473 clay matrix (Table 2). This is somehow surprising as the St Austell topaz granite, covered by the 474 Treviscoe kaolin operations, was considered as one of the most prospective source of disseminated 475 magmatic mineralisations from a metallogenic perspective. In addition, the high Nb-Ta grades obtained 476 after a two-stage gravity concentration on hydrocyclone underflow samples presented in Scott et al. 477 (1998), suggested that micaceous residue from these areas may have a potential interest for metal 478 recovery. This could be explained by the variations in the intensity of kaolinisation within the St Austell 479 kaolin deposit. The kaolin ore from the topaz granite is lower grade (i.e. less weathered) than the kaolin 480 ore from biotite granite. A less kaolinised ore (and thus less liberated metal-bearing minerals) means 481 lower yields and larger proportions of coarse wastes.

It appears that, as a consequence the variations in the intensity of kaolinisation within the St Austell and differences in the mining technique, the micaceous residue from biotite granite, is the most prospective stream both in terms of metal content and distribution. In addition, the corresponding kaolin pits are among the largest pits located in the area associated with the more reliable extraction and processing method and the granularity of the micaceous residue doesn't require additional grinding prior to further processing.

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489 **5.2.** Core samples characterisation and validation of the geometallurgical program

One of the primary <u>objectiveobjectives</u> of this study is to validate the core sample processing program to ensure its data is reliable and can be used for geometallurgical modelling purpose. Hence, it is necessary to compare the properties of the products obtained from the core samples treatment in the laboratory with actual process samples collected from the plant. Since kaolin processing is mainly a size classification process, the particle size distribution will be investigated, followed by the metal grades and mineralogy of the samples which are both critical for the recovery of critical metals.

496 5.2.1. Particle size distribution

497 Particle size distribution of the spiral feed may significantly impact the performance of the separation 498 with the spiral concentrator. The influence of particle size on spiral performance in terms of heavy 499 mineral recovery has been investigated by Dehaine and Filippov (2016) who observed that for the 500 operating conditions at which the test have been performed, the variations of heavy minerals recovery 501 with particle size is limited and only decreases for particle sizes below 50 µm.

502 Particle size analysis of the GT53MR180 samples shows that the mean particle size distribution is very 503 close to the one of the spiral feeds (Figure 7Figure 6). This suggests that GT53MR180 samples are on 504 average representative, in terms of particle size, of the -180+53 µm size fraction of the micaceous 505 residue which is processed by the gravity concentration route. It can be observed that the variation in 506 particle size distribution of the GT53MR180 (-180+53 µm) fraction is relatively limited and is more 507 important in the -50+10 µm size range. These variations are still limited compared to the high variability 508 of the size distribution properties of the raw micaceous residue stream observed in Dehaine et al. (2016). 509 However, these results were obtained from the raw micaceous residue, and these variations may be less 510 visible on the screened spiral feed. In addition, the overall limited variations displayed by the core 511 samples may also be explained by the fact that the samples set is only sourced from highly kaolinised 512 ore and that more variations (toward the coarser fractions) should be observed with lower grade, *i.e.* less 513 altered, material. Variations in particle size distribution for the spiral feed are more limited and are 514 almost always included into the variation intervals of the GT53MR180 particle size distribution. These 515 variations, however, do not seem to affect the performance of the spiral as illustrated in Figure 7Figure 516 6 by the rather small variations of spiral concentrate particle size distribution. In addition, it can be seen 517 that the finer size ranges below 30-40 μ m, for which the spiral feed particle size variations are more 518 important, are poorly recovered in the spiral concentrate. This result is in agreement with the results 519 from size-recovery curves analyses presented in Dehaine and Filippov (2016), and suggests that 520 variations in particle sizes, at least for the finer size range is unlikely to influence the performance of 521 the separation.

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523 5.2.2. Whole rock geochemistry

The critical metals and REEs composition spectra for all <u>GT53MR180</u> core samples products are presented in <u>Figure 8</u>Figure 8ab along with the average raw micaceous residuea high grade spiral feed sample and corresponding fresh granite trace element composition for comparison. These spectrum display clear enrichments in certain critical metals, decreasing in the order Sn, Nb, Zr and to a lesser extent Ga (<u>Figure 8Figure 7</u>a) as well as an enrichment in LREE, decreasing in the order Ce, La, Nd
(<u>Figure 8Figure 7</u>b).

The average geochemical composition of the high-grade micaceous residuespiral feed sample shows

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531 a very similar pattern with however overall grades above around the maximum of those of the GT53MR180 samples for Nb, Ta, Zr, Y and REE. This correspondance suggests that MR180 samples 532 533 are on average representative, in terms of critical metal content, of the -180+53 µm size fraction of the 534 micaceous residue which is processed by the gravity concentration route. This is due to the fact that these 535 elements are known to be distributed in the finer size ranges of the micaceous residue (Dehaine and Filippov, 2015a) which are not recovered in the GT53MR180 samples (-180+53 µm). One sample (ref. 536 537 11069860) is highlighted in Figure 8 due to its significantly different REE pattern, displaying 538 an enrichment in Y and in Heavy Rare Earths Elements (HREEs), probably due to the presence of 539 xenotime in that sample (Jefferies, 1985).

540 Comparison with geochemical data from fresh St Austell biotite granite (Simons et al., 2017, 2016) 541 display a strong enrichment in Sn, Nb, Ta and W in the <u>GT53MR180</u> core samples. This suggests an 542 enrichment of these metals in the kaolinised granite whereas indium is on the contrary strongly depleted. 543 The chondrite-normalised plot (<u>Figure 8Figure 7</u>c) displays a LREE enriched pattern and a strong 544 anomaly in Eu in agreement with trace element geochemistry of the St Austell biotite granite established 545 in previous studies (Chappell and Hine, 2006; Simons et al., 2017, 2016).

546 Figure 9Figure 8a shows a positive linear trends line between Nb and TiO₂ ($R^2=0.70$). This 547 proportional relationship, also observed at microscopic scale and macroscopic scale in Dehaine and 548 Filippov (2015) suggest that Nb is most likely associated with a TiO₂-bearing mineral, *i.e.* rutile, 549 ilmenorutile or ilmenite. These results may appear in contradiction with the observations of Simons et 550 al. (2017) which suggest biotite (siderophyllite) is the main host for Nb in G3 (biotite) granites and 551 rutile/ilmenite in G4 (tourmaline) granites. While this may indicate a change in granite type at depth, as 552 biotite and tourmaline granites are known to be spatially associated, for example, crosscutting as sheets or forming gradational contacts (Simons et al., 2017), it may also point to the alteration of most brown 553 554 micas in highly kaolinised material which are almost completely weathered by the kaolinisation process (see Mueller et al. (1999), and mineralogy section hereafter). 555

556 Another interesting correlation is the strong correlation ($R^2>0.99$) between each LREE suggesting 557 constant relative proportions of LREE with a Ce/La ratio of 2.27, a Ce/Nd ratio of 2.41, a Ce/Pr ratio of 558 8.67 and a Ce/Sm ratio of 12.85 in all GT53MR180 samples (Figure 9Figure 8b). These correlations 559 and the corresponding ratios are close from the one determined by whole-rock analysis on process samples and by Energy-Dispersive X-ray Spectroscopy (EDS) analyses on monazite grains in Dehaine 560 561 and Filippov (2015). This other multi-scale correspondence of element-to-element ratios from the 562 macroscopic-deposit scale (core samples) to the microscopic scale (monazite grains) through the 563 mesoscopic scale (process samples) is a strong evidence that monazite is the only LREE-bearing mineral 564 in the St Austell biotite granite.

565 5.2.3. Mineral characteristics

The average modal mineralogy of the GT53MR180 core samples obtained by QEMSCAN[®] analysis 566 567 is presented in Table 1 Table 1. The mineralogy is largely dominated by quartz (48.7 wt %), followed by 568 fine secondary white-micas (muscovite) and tourmaline (schorl-dravite). Other minerals accounting for 569 less than 5 wt % each include remaining feldspars (orthoclase, microcline) and biotite (siderophyllite-570 annite), topaz, Ti-oxides (mostly rutile, ilmenorutile and ilmenite) and kaolinite. This is consistent with 571 the mineralogy of the biotite granite described in Manning et al. (1996) and in agreement with the mineralogical compositions of altered biotite granite (Mueller et al., 1999). The overall modal 572 573 mineralogy is very similar to the one of raw micaceous residues previously investigated, even if this size 574 fraction seem to contain more quartz and less biotite and feldspar than typical raw residues from biotite 575 granite (Dehaine et al., 2017; Dehaine and Filippov, 2015a).

In terms of accessory minerals hosting the critical metals, QEMSCAN[®] results confirmed cassiterite as the main host for Sn, rutile-ilmenorutile as the main host for Nb and monazite as the host mineral for LREE. These results confirm the observations of Dehaine and Filippov (2015), even if very little wolframite is observed in the samples due the overall low W grades.

580 Mineral association data obtained by QEMSCAN[®] quantifies which mineral grain is adjacent to or 581 touches another mineral grain in a particle. <u>Figure 10Figure 9</u> displays the monazite association data for 582 the <u>GT53MR180</u> samples based on QEMSCAN[®] results. On average, monazite grains mostly have no 583 association (36.9%), *i.e.* monazite boundary pixel occurs adjacent to a pixel of the background, which 584 means that most of the monazite grains surface seem to be free. The next major associations of monazite 585 is with tourmaline (17.3%) and micas (14.1%) and then by decreasing order Ti-oxides, zircon and quartz. 586 The average monazite liberation histogram suggests it is either locked (<10% free) or liberated 587 (>90% free) in the GT53MR180 samples (Figure 11Figure 10). However, the large error bars suggest a 588 large variability of liberation data for all core samples. This is because each sample contains a distinct 589 population of monazite. Indeed, samples can be roughly split into two categories, the ones in which 590 monazite is almost completely locked, representing 50% of the core samples, and the ones in which 591 monazite is almost completely liberated (see, for example, Figure 12Figure 11a), representing 47% of 592 the core samples. When locked, monazite is often included in tourmaline (Figure 12Figure 11b), 593 secondary muscovite (Figure 12Figure 11c) or rutile (Figure 12Figure 11d). This relates to the 594 observations made on fresh biotite granite in which monazite is mostly included, along with zircon, in 595 biotite and to a lesser extent in tourmaline resulting in metamictisation (Simons et al., 2016). The results 596 obtained for the GT53MR180 samples could be linked to the kaolinisation process, which is known to 597 affect brown micas from moderated de-ferruginisation in slightly altered granite to complete 598 decomposition in the completely altered biotite granite whereas tournaline remains poorly altered even 599 for the highest degree of kaolinisation (Mueller et al., 1999). Given the highly kaolinised ore considered 600 in this study, it is likely that monazite grains which were previously included in biotite have been 601 progressively liberated during the decomposition of biotite whereas those included in tourmaline 602 remained locked resulting in the above-mentioned liberation characteristics of monazite in the 603 GT53MR180 samples. It also worth mentioning that monazite has been found to be sometimes 604 associated with reticulated rutile (Figure 12Figure 11e). These ultrafine monazite grains display a needle 605 shape which has the same orientation that the rutile structure. It is not clear whether these needle-shaped 606 monazites just lay on the rutile surface or if they are actually associated with the rutile, but the presence 607 of intergrowths and the orientation of the monazite needles seems to suggest that there is an association 608 (Figure 12Figure 11f).

609 Characterisation results influence the recovery of monazite through the gravity concentration process
 610 presented in <u>Figure 4Figure 4</u>. The high proportions of tourmaline and micas in the <u>GT53MR180</u> (611 180+53 μm) fraction may be troublesome for the gravity concentration techniques investigated. Indeed,

tourmaline has an average density close to monazite (Table 3 Table 3) whereas fine phyllosilicates are known to behave erratically in gravity operations. Phyllosilicates are known to cause gravity processing problems, especially by increasing the slurry viscosity (Filippov et al., 2016; Sivamohan and Forssberg, 1985a). Monazite is mostly associated or included in tourmaline or muscovite and half of the samples only contain locked monazite. Therefore only half of the LREE content of the -180+53 μ m fraction of the micaceous residue is potentially recoverable, which could explain some of the low recoveries obtained in previous attempts to recover monazite from micaceous residues (Dehaine et al., 2017).

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620 5.3. Gravity concentration testwork on micaceous residue

621 Large process samples were collected on-site from the micaceous residue stream of the kaolin processing plant and prepared to correspond to the -180+53 µm feed of the pilot-scale gravity 622 623 concentration route described in Figure 4Figure 4. The samples were fed in a two-stage gravity 624 concentration route, starting by a spiral concentrator in scavenging configuration to maximise the 625 recovery. Spiral concentrates are then fed to a shaking table to increase the grade. Results from these 626 tests work are discussed from a mineral processing perspective process modelling and optimisation 627 perspective in Dehaine et al. (2017) and from process mineralogy and geometallurgical perspective in 628 the following sections.

629 5.3.1. Characterisation of monazite in gravity products

630 Monazite characterisation data in the different gravity concentration products obtained by QEMSCAN[®] are presented in terms of mineral association and mineral liberation in Figure 13Figure 12 631 632 and Figure 14Figure 13 respectively. It can be seen that most of the monazite recovered in the gravity 633 concentrates is liberated even if some monazite is, to a lesser extent, locked in tournaline or Ti-oxides. 634 The proportion of liberated monazite decreases in the middlings whereas the proportion of locked 635 monazite increases, in particular for the shaking table tailings. Most of the locked monazite in the 636 middlings is associated with tournaline which has an intermediate density (3.12 g.cm⁻³) between those 637 of the gangue minerals (2.60-2.82 g.cm⁻³) and the one of monazite or the other critical metal-bearing 638 minerals (\geq 4.25 g.cm⁻³). Other significant monazite mineral associations in the middlings include zircon, 639 micas and Ti-oxides. The monazite lost in the gravity tailings is mostly locked, even if a non-negligible 640 proportion is partially (\leq 80%) or fully (\leq 100%) liberated. These locked monazite grains appear to be 641 mostly locked within tourmaline in gravity tailings but results for the shaking table suggest that a non-642 negligible proportion of the monazite lost in the tailings is also associated with micas and kaolinite.

These results should be considered in parallel to the average monazite grain size in the various gravity concentration products (Figure 15Figure 14). It can be seen that coarse monazite tends to report to the gravity concentrates whereas ultra-fine (-10 μ m) monazite, either liberated or locked, reports to the tailings. This is consistent as recovery in the ultrafine particle size range is poor with most gravity processing techniques (Burt and Mills, 1984).

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649 5.3.2. Spiral performance prediction models

650 Spiral concentration is a gravity separation technique which differentiate minerals according to 651 particle size, density, and shape (Burt and Mills, 1984). In addition, some other important ore 652 characteristics such as feed grade, mineral associations or liberation degree may play an important role 653 on the separation performance (Sivamohan and Forssberg, 1985b). Most of these parameters can only be assessed by systematic mineralogical and textural analysis, for instance via QEMSCAN[®]. This type 654 655 of analysis is time consuming and expensive and, in this study, only a few samples were characterised 656 using this technique. However, the grade of the spiral feed is known as the feed material of the spiral 657 concentration tests consist in the -180+53 µm fraction of the micaceous residue which corresponds to the GT53MR180 samples analysed for trace elements, including LREE. Indeed, LREE content of the 658 659 GT53MR180 samples, calibrated using the ICP-MS results on duplicate samples, can be used as an 660 estimate of the LREE feed grade for the gravity concentration circuit. A multivariate analysis applied to 661 this LREE content and other important mineralogical properties of monazite (i.e., monazite liberation, 662 associations and grain size) allows the identification of the most important correlations between these 663 properties (Table 4Table 4). There is a strong positive correlation ($R^2=0.83$) between the average 664 monazite grain size and degree of liberation suggesting that the coarser is the monazite the more likely 665 it is liberated. In addition, monazite grain size and degree of liberation are both correlated to the LREE content with correlation coefficient of 0.73 and 0.65 respectively. All these properties are known to have 666 667 a positive influence on gravity concentration performance, *i.e.*, coarse and liberated monazite preferentially report to the concentrates (Figure 14Figure 13 and Figure 15Figure 14). The fact that there
 are all positively correlated is an indication that the LREE feed grade could potentially be used to predict
 spiral concentration performance.

The influence of the LREE grade in the feed material on the spiral concentration efficiency has been tested using the spiral setup in scavenging configuration. To reflect the representativeness of the spiral feeds in terms of grade, the LREE feed grade (G_F) is normalised using the minimum and maximum LREE contents found in the core sample database as follows:

$$G_F^* = \frac{G_F - \min}{\max - \min} \tag{1}$$

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 G_F^* can therefore vary between 0 and 1, these two extreme values corresponding to the case where 676 677 G_{F}^{*} equals respectively to the minimum or the maximum LREE content in the core sample database. 678 Figure 16Figure 15 shows the influence of the normalised LREE grade (G_F^*) in the feed material on the 679 performance of spiral concentration of LREE. The results show that the normalised feed grades are well 680 distributed over the [0;1] interval which suggests that they cover almost the whole range of LREE values 681 displayed in the GT53MR180 core samples data. As expected, spiral performance is found to improve 682 with increasing feed grade. Indeed, LREE recovery and concentrate grade increase with LREE content 683 in the feed, with a maximum recovery and concentrate grade of 55% and 1000 ppm respectively. 684 However, LREE enrichment degree reaches a plateau at 4.5 for LREE normalised feed grade higher 685 than 0.6-0.7 (~ 150 ppm LREE). The relationship between spiral performance indexes (y) and normalised feed grade shown in Figure 16Figure 15 exhibits clear trends that can be fitted by an 686 687 empirical model, using the equation:

688

689 The parameters *a*, *b* and *c* are fitted, and have no physical meaning. However, it is expected that they 690 would be linked to changes in mineralogy or liberation degree which consequently affects spiral 691 performance. The values of the fitted parameters are given in <u>Table 5</u>. Comparison between the

 $y = a + b \exp(c G^*_F)$

(2)

observed and predicted values shows a good agreement with high regression coefficient values (R²>0.9),
suggesting that the models can reasonably be used to estimate spiral performance (<u>Table 5</u><u>Table 5</u>). All
performance indexes trend lines suggest that below 0.2 LREE normalised feed grade (around 60 ppm
LREE), no separation should be expected (<u>Figure 16Figure 15</u>). Hence, this value could be considered
as the minimum grade of the <u>GT53MR180</u> fraction that should be considered for LREE recovery.

The aforementioned results suggest that the regression models proposed to predict spiral performance for LREE can reasonably been applied to <u>GT53MR180</u> core sample data. In addition, the range of LREE feed grade used to calibrate the models, as mentioned before, covers almost the whole range of <u>GT53MR180</u> LREE grades (from 0.25 to 0.95 normalised feed grade) therefore suggesting a good representativeness of the results. Hence, prediction of spiral performance using <u>a</u> core sample database can be realised using the following equation (with coefficient values given in Table 5Table 5):

$$y = \begin{cases} a + b \exp(c G_F), \text{ for } G^*_F \ge 0.2 \end{cases}$$
 (3)

$$\int 0 \qquad \qquad for \ G^*_F < 0.2 \tag{4}$$

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6. DISCUSSION AND CONCLUSION

706 <u>6.1. Geometallurgical classification proposal</u>

707 This study represents the first step in the development of an integrated geometallurgical model for 708 the optimised extraction St Austell kaolin deposits that takes into account of both kaolin and critical 709 metals from the St Austell deposits rare-metal graniteresources. Generally For extraction 710 purposesspeaking, two distinct approaches canould either be used to create a spatially-based predictive 711 geometallurgical model. The first one is based on geometallurgical testing which traditional approach 712 involves small scale laboratory geometallurgical tests which aim to measure the metallurgical response 713 of the samples in order to build predictive performance prediction models through simulation (Bulled 714 and Mcinnes, 2005). The Because the metallurgical response is then reported on an elemental basis, 715 therefore the geometallurgical variations in the deposit orebody are is only based on elemental grades 716 only but not onrather than on mineralogical information. The second A more recent approach, based on 717 particles, uses quantitative characterisation of particles and the minerals contained in theand particles characterisation data as parameters to develop <u>a</u>the geometallurgical program from data collection to
process simulation (Lamberg, 2011). The pProcess models established are based on minerals and can
use <u>as input mineralogicalore characteristics, inc-luding mineral abundance, grain size, associations,</u>
degree of liberation, grindability, recovery potential, etc., <u>as parametersinput for physical process</u>
<u>models and in studies and simulationss, such as These include mineral quantities abundances, minerals</u>
<u>grainsizes distribution, mineral associations, or liberation depending on their level of complexity (Lund et al., 2013)</u>.

725 The geometallurgical approach described depicted in this paper is, in some aspects, unconventional, and borrows aspects from both does not really fall into one of the aforementioned approaches but rather 726 727 borrows elements aspects from both of them. Firstly, this study is based on a systematic lab-scale core 728 sample processing program which mimics the full-scale kaolin processing plant. This is an unusual 729 approach which allows obtaining generates sub-samples out of core samples. Rather than performing 730 specific metallurgical tests referring to one particular unit operation, processing of these samples which have yields similar characteristics than the corresponding similar products and residues as produced by 731 732 the full-scale process., rather than performing specific metallurgical tests referring to one particular unit 733 operation. NextIn this case, if the elemental-based final spiral performance prediction models are only 734 elemental-based, are*i.e.* based on LREE grade-only and, do consider the mineralogy of the orebody has 735 nonetheless been taken into accountindirectly. HoweverIndeed, the fact that LREE is only hosted inby monazite-is the only host mineral for the LREE, and a positive correlation is observed between that the 736 737 overall the LREE grade grade is positively correlated to and mineralogical factors such as liberation or 738 monazite grain size (Table 4Table 4), which influence favouring the recovery, such as liberation or 739 monazite grain size (Table 4), may This which explains why the process performance canould be 740 predicted using only the LREE feed grade.

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743 6.1.1.1. Geometallurgical classification proposal

In order to develop <u>A</u>^a geometallurgical ore type classification for critical metals recovery <u>must</u>
 <u>consider</u>, one should first consider the kaolin resources in which as metals are regarded as a potential

by-products, <u>.</u> In other words, <u>i.e.</u> to be considered for metal recovery, the ore <u>is considered for metal</u>
recovery when it is extracted for the primary <u>must primarily valuable for the main</u>-commodity, <u>i.e.</u>
kaolin, <u>to be considered for metal recovery</u>. When developing an ore-type classification for
geometallurgical purposes it is also important to capture essential elements which are relevant in terms
of metallurgical and commercial <u>significancemeaning</u>.

751 A geometallurgical ore -type classification scheme proposal for the kaolin resources is presented in 752 Figure 17Figure 16a. It is based on the core sample data related to the R5P5 sub-samples which is an 753 indicator of the fine kaolin product. The first classification criterion for the kaolin resources is the yield 754 of the R5P5 product, *i.e.* the weight ratio of the R5P5 product over the feed. The R5P5 product is almost 755 entirely composed of kaolinite plus ancillary minerals. Hence, the yield of the R5P5 product gives an 756 indication of the future kaolin recovery, which is one of the main metallurgical response for the kaolin 757 refining operation considered. It is closely related to the classical geological ore classification based on 758 the degree of kaolinisation of the granite rock using visual inspection of texture preservation and 759 feldspars alteration, as the more kaolinised is the granite the higher will be the kaolin yield once it is 760 processed. The next classification criterion is the FeO content of the R5P5 products, which reflect the 761 brightness/whiteness of the kaolin product (Bertolino et al., 2010). This is a key property of the kaolin 762 product as it drives theirs commercial end-use and will define to which industry it will be directed. In 763 the end, five different ore grades are defined from the Grade I, nearly fresh to partly kaolinised granite 764 (R5P5 yield<10%), to the Grade V, fully kaolinised granite (R5P5 yield>15%) and low iron content 765 (FeO<0.4%), material.

Classifying the kaolin ore in terms of its potential for critical metals recovery could be performed using the process performance prediction models. Figure 17Figure 16b shows an example of the distribution of the predicted LREE recovery after the spiral concentration step for two kaolin pits by applying the empirical models presented in equations (3)(3) and (4)(4) to the corresponding GT53MR180 core sample data. One could use this chart to classify the resources in distinct recovery classes but also to delimitate the potential resources itself using the minimum feed grade suggested by the models (equations (4)(4)).

775 **6.2.** Application to core sample

776 The application of the aforementioned geometallurgical classification schemes to an actual drill core 777 is presented in Figure 18Figure 17. Applying the spiral performance prediction models to GT53MR180 778 data allows evaluation of the potential separation performance in terms of concentrate grade or recovery. 779 This data, along with data pertaining to the R5P5 product (*i.e.*, yield and FeO grade), could then be used 780 to classify both kaolin and critical metals resources using the corresponding geometallurgical 781 classification schemes. These domains could be used to first define the zones of interest for kaolin and 782 then to establish if these zones are also of interest for metal recovery, therefore targeting the zones of 783 interest for metals recovery as a by-product of kaolin.

784 While this approach-preliminary work is promising and inexpensive, it would benefit from being 785 validated by comparing the predicted performances with those obtained after processing micaceous 786 residue samples collected when refining a block of ore for which the GT53MR180 grades are known. 787 This may be very difficult for several technical reasons including ore traceability, contamination, large 788 sample collection, etc. In most geometallurgical approaches, process performance prediction models are 789 built using production data comprising a large dataset collected after months of production. Hence 790 implementation on site of a primary concentration process (spiral concentrators for instance) seems to 791 be the next logical step. Indeed, it would give the opportunity to collect enough data for developing 792 similar process performance prediction models and could also allow assessing the effect of the ore 793 variability in terms of metal grades, alteration grade and lithology (granite type) which is fundamental 794 for the development implementation of a proper tactical geometallurgical model approach. In addition, 795 it would allow recovering directly some primary concentrate samples for metallurgical testing, therefore 796 removing the need for large residue samples collection. In the meantime, while no actual critical metal 797 recovery is applied on site, this inexpensive approach could still be used to store the residues properly 798 according to their potential for metals recovery.

799

800 6.3. On the commercial potential of kaolinised granites for metal recovery

801 Assessing the commercial potential of kaolin residues for metal recovery is not an easy task even

802 after all the work performed on this case study (Dehaine et al., 2017; Dehaine and Filippov, 2015a; 803 Filippov et al., 2016; Scott et al., 1998). Some of the kaolinised granites mentioned in section 2.3 have 804 been considered as potential sources for Nb, Ta, Sn or W and some of them are actually mined for these 805 metals. For instance, the Yichun granite is the largest Ta, Nb and Li raw material producer in China 806 (Schwartz, 1992; Yichun Tantalum Co., 2005) and the Beauvoir granite is known to produce cassiterite 807 (800 g/t Sn) and columbite (190 g/t Ta plus 120 g/t Nb) as by-products of kaolin production (Negroni, 808 2015; Pohl, 2011). There are also some mining operations such as the Greenbushes pegmatite complex, 809 (Australia) or the Manono-Kitotolo deposit (DRC) for which the situation is reversed and where Sn, Ta, 810 Nb or Li are primary products but kaolin is regarded as a by-product (Laznicka, 2006).

811 From a resources point of view, the metal grades within the kaolinised rare-metal granites are low, 812 most of them are actually well below the usual cut-off grades, but the volumes are large and the 813 mineralisation is disseminated within the ore which is already exploited for the kaolin. The potentiality 814 is also dependent on the granite type from which the kaolin ore originate and the variability in terms of 815 intensity of kaolinisation as well as the extraction method. Therefore, the St Austell metal granite which 816 is considered as one of the most prospective sources of disseminated magmatic mineralisations from a 817 metallogenic perspective is not necessarily the most prospective in terms of by-product recovery of 818 metals from kaolin production.

819 Whilst previous work on St Austell kaolin residues has demonstrated that heavy minerals can be 820 recovered and concentrated from kaolin residues using a simple processing route with a limited number 821 of operations, the metal grades obtained are still relatively low and further processing steps will be 822 required before reaching an acceptable concentrate grade (Dehaine et al., 2017). The effectiveness of 823 the heavy mineral recovery for this case study is accounted to the complexity of the considered material 824 which is low grade, fine grained and contains high proportions of medium density minerals and clay 825 minerals. However, high enrichment ratios (10-60) and reasonable recoveries, at least for LREE (\sim 60%), 826 have been obtained in a few separation steps without any additional grinding required.

From an environmental point of view, only benefits could arise from a potential processing of micaceous residue as the proposed processing route has a low environmental footprint with low energy consumption (gravity concentration) and environment<u>-al</u>-friendly chemical reagents (dispersion, flotation). In addition, processing the micaceous residue will result in a reduced waste production (and
even more if historical tailing dams are considered) and a better waste management strategy (wastes
sorted by size fractions).

833 From an economic point of view, the arguments are more balanced. Indeed, the mineral concentrates 834 obtained by beneficiating the kaolin residues is rather atypical, with no equivalent on the market and 835 thus there is no reference for the terms of a potential contract (Negroni, 2015). Another limitation is the 836 application of penalties due to the presence of several metal-bearing minerals that would impair the 837 value of the concentrate. The more restrictive issue facing LREE production is associated with the 838 potential radioactive activity of the concentrate due to the presence of Th in monazite (Zhu et al., 2015). 839 This means that a limitation in terms of LREE concentration should be accepted above which some 840 special dispositions controls must be taken (dry processing, dust control, etc.). The annual income 841 cannot be estimated but an assessment of the annual production, considering only the grade V ore from the sole kaolin pits investigated in this study, suggests that around 15.5 t/y LREE could be recovered in 842 843 metal pre-concentrates (Dehaine, 2016). Although the current concentrate grade is low, the annual 844 tonnage is high and total capital investment for the metal recovery process is low (Dehaine, 2016). With 845 demand for Nd and Ce expected to grow in the future due to their use in permanent magnets (Kingsnorth, 2014b), metal recovery from kaolin residue is worth investigating for other kaolinised rare-metal 846 847 granites worldwide.

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FIGURES



Figure 1. Spatial relationship at distinct geological scales between Variscan granites and tin or tungsten deposits.
(a) Tin belts in Europe and their spatial connection to the Variscan belt modified from Schuiling (1967). Main tin and tungsten deposits are from the PROMINE project (BRGM, 2011). (b) Outline geological map of South-West England peninsula showing the location of the major plutons of the Cornubian batholith, the kaolinised areas and the active kaolin pits, modified from Černý et al. (2005). (c) Distribution of the felsite dykes (elvans) and Sn-W mineral lodes around and within the St Austell rare-metal granite after Bray and Spooner (1983).



Figure 2. Geological map of the St Austell granite showing the distribution of the various granite types from
Manning et al. (1996), the major kaolin resources and the micaceous residue dams after British Geological Survey
Geological Map Data (British Geological Survey, 1997), as well as the main Sn-W mineralisation after Bray and
Spooner (1983).



Figure 3. Trace element characteristics of peraluminous high phosphorous (PHP), peraluminous intermediate
phosphorous (IHP), peraluminous low phosphorous (PLP) and peralkaline granites (PLK). (a) Silica-phosphorus
diagram. (b) Zirconium-thorium diagram. B series: Beauvoir granite, Massif Central; YCH: Yichun granite,
Jiangxi, China. Abbreviations for St Austell granites are given in the text. Based on data from Manning et al.,
(1996) and the review of Raimbault et al. (1995).



Figure 4. Simplified schematic of the kaolin ore processing and the by-product gravity processing route proposal.





1076 W) within the output streams of kaolin refining process for biotite, topaz and tourmaline granites.





1079Figure 76. Comparison of particle size distribution between GT53MR180 samples and corresponding1080process samples (spiral feed). Solid lines represent the average and dashed lines the lower and upper fence.



Figure <u>87</u>. Geochemical composition spectra of the <u>GT53MR180</u> core samples products compared to the <u>a</u>-raw micaceous residuespiral feed sample, obtained by ICP-MS. (a) Critical metals contents, plus Sn, Ta, Y and Zr. (b) REEs content. (c) Chondrite normalised REEs content using the values of McDonough and Sun (1995). Sample 1085 11069860 is highlighted due to its significantly different REE pattern, "micaceous residueSpiral feed" refers to the average of 5 micaceous reside highest grade process samples fed to the spiral and "biotite granite" corresponds to the average REE data of 3 fresh St Austell G3 (biotite) granite from Simons et al. (2016).



1090Figure 28. Binary linear correlation plots and trend lines For Nb/TiO2 and LREE using GT53MR180 samples1091ICP-MS data. (a) Good correlation ($R^2=0.70$) suggesting a proportional relationship between TiO2 and Nb. (b)1092Strong correlations ($R^2>0.99$) suggesting a proportional relationship between each LREE, similar to the one1093observed on process samples and at the microscopic scale on monazite grains in Dehaine and Filippov (2015).



Figure <u>10</u>9. Box-plots summarising monazite mineral association data in decreasing order for the

GT53MR180 samples.





Figure <u>1140</u>. Average monazite liberation for all <u>GT53MR180</u> samples (top), for locked monazite only samples (middle) and for liberated monazite only samples (bottom).



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Figure 1211. Scanning electron micrograph in back-scattered electron mode illustrating the various habitus of monazite in the micaceous residue. (a) Liberated monazite (Mnz) grain, (b) Monazite included in a zoned tourmaline (Tur) with rutile (Rt) inclusions, (c) Weathered biotite (Bt) with inclusion of monazite, (d) Monazite included in Nb-rich rutiles, (e) Monazite associated with reticulated rutile and (f) Zoomed image showing a needleshaped monazite oriented along the rutile structure.



I 109Figure 1312. Average mineral association for monazite in the gravity concentration products. F: Feed, SC: Spiral1110Concentrate, SM: Spiral Middlings, ST: Spiral Tailings, TC1: Table Concentrate 1, TC2: Table Concentrate 2,

1111 TM: Table Middlings, TT: Table Tailings.





Figure 1413. Monazite liberation data for the gravity concentration products. (a) False colour QEMSCAN[®] monazite liberation grid showing typical examples of spiral concentrates (top), spiral middlings (middle) or spiral tailings (bottom), only the particles containing monazite are shown. (b) Average monazite liberation data for all gravity concentration products (spiral and shaking table) including concentrates (top), middlings (middle) and tailings (bottom).





Figure <u>1746</u>. Proposal for geometallurgical classification schemes. (a) Geometallurgical kaolin ore-type classification scheme using <u>R5P5</u> product yield and FeO content, threshold values are based on Negroni (2015).
(b) Histograms of LREE recovery (R) for two kaolin deposits and corresponding box-plots that could be used to classify the ore into distinct categories (*i.e.*, C1: R≥60%, C2: 60%>R≥50%, etc.).



Figure 1817. LREE by-product potential evaluation using process performance prediction models and geometallurgical classifications schemes applied to drill cores. (Left) Core sample data related to the final kaolin product (R5P5) and to the spiral feed (GT53MR180), (middle) Application of spiral performance prediction models for LREE recovery, dashed lines represent 95% confidence intervals and (right) Geometallurgical classification of kaolin and LREE resources allowing to target the area of high potential for the recovery of LREE 1141 as by-product of kaolin.

TABLES

- 1144 **Table 1.** Mineralogy of the six major granite types of the St Austell rare-metal granite (Manning et al., 1996;
- 1145 Simons et al., 2016).

Granite Type	Major minerals	Minor minerals	Accessory minerals
Biotite Granite (BG)	Quartz, K-feldspar, plagioclase, biotite, muscovite	Tourmaline	Rutile, topaz, apatite, monazite, uraninite, zircon, cassiterite, cordierite
Lithium-mica Granite (LMG)	Quartz, K-feldspar, plagioclase (albite), lithium-mica (zinnwaldite, lepidolite)	Tourmaline, topaz	Apatite, monazite, zircon, rutile, fluorite
Tourmaline Granite (TG)	Quartz, K-feldspar, plagioclase, lithium-mica (zinnwaldite)	Tourmaline, topaz	Apatite, rutile
Fine-grained Tourmaline Granite (FGTG)	Quartz, K-feldspar, plagioclase (albite), lithium-mica (zinnwaldite)	Tourmaline, topaz	Monazite, apatite, zircon, rutile, arsenopyrite, cassiterite
Topaz Granite (TZG)	Quartz, K-feldspar (orthoclase), plagioclase, lithium-mica (zinnwaldite, lepidolite), muscovite	Topaz, fluorite, tourmaline	Apatite, amblygonite, zircon, columbite-tantalite, ilmenorutile, fluorite, uraninite

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1147 Table 2. Mean LREE, Sn, Nb, Ta and W contents in the output streams of the kaolin plants and the estimated clay

1148 matrix content obtained by metallurgical accounting for biotite, topaz and tourmaline granite.

Material	Mass Balance	LREE	Sn	Nb	Та	W
	wt.%	ppm	ppm	ppm	ppm	ppm
Biotite granite						
Matrix	100	79.1	68.7	39.9	11.4	30.8
Stone	4	50	38.3	26.2	7.6	15.7
Gravel	8	35.4	48.5	18.4	5.2	14.9
Sand	48	41.2	51.4	32.9	10	17.5
Micaceous residue	20	168.7	139.4	93.7	25.9	92.9
Product	20	103.7	53.5	14.2	3.6	10.3
Topaz granite						
Matrix	100	11.1	20.6	53.1	23.6	-
Stent	36	9	13.1	42	17.9	18
Sand	40.5	5.5	20.6	32.9	23	-
Micaceous residue	13.5	15.9	38.9	97.21	45.6	30.3
Product	10	34.3	21.4	38.84	15.8	15.5
Tourmaline granite						
Matrix	100	49	49.6	42.3	10.5	20.27
Stent	36	55	21	28	2	9.4
Sand	40.5	29.6	60.7	44.1	5	8
Micaceous residue	13.5	46.5	87.2	80.2	32	93
Product	10	109.9	56.8	55.1	34	10.9

Table 3. Average mineralogical composition of the 26 GT53MR180 core samples obtained by QEMSCAN[®] along

1151 with density data (Anthony et al., 2001).

Mineral	Average (wt %)	Std Dev.	Density (g.cm ⁻³)
Quartz	48.47	12.16	2.60
Feldspar (orthoclase, microcline)	3.18	7.41	2.60
Kaolinite	1.12	1.03	2.60
Muscovite	21.99	10.15	2.82
Biotite (siderophyllite-annite)	1.72	1.28	3.09
Tourmaline (schorl-dravite)	20.97	10.69	3.12
Topaz	1.28	1.27	3.55
Ti-Oxides	1.07	0.48	4.25
Zircon	0.04	0.05	4.65
Monazite	0.04	0.03	5.15
Cassiterite	0.06	0.08	6.90
Others	0.03	0.21	-

Table 4. Multivariate analysis highlighting the main correlations between LREE grade and some mineralogical1155properties of the GT53MR180 samples (2 outliers deleted). Significant correlations (≥ 0.65 or (-0.65 \geq) are1156highlighted in italic. Abbreviations: LREE: Calibrated LREE content/feed grade, NA: No association, A-Tour.:1157association with tourmaline, A-Micas: association with micas, A-Ti: association with Ti-oxides, Size: average1158grain size, %Locked: liberation $\leq 10\%$ and %Liberated: liberation $\geq 90\%$.

	LREE	NA	A-Tour.	A-Micas	A-Ti	Size	%Locked	%Liberated
LREE	1.00	0.07	-0.08	-0.38	0.02	0.73	-0.60	0.65
NA	0.07	1.00	-0.62	-0.33	-0.15	0.04	-0.08	0.05
A-Tour.	-0.08	-0.62	1.00	-0.17	-0.41	0.08	0.04	0.05
A-Micas	-0.38	-0.33	-0.17	1.00	0.69	-0.32	0.52	-0.56
A-Ti	0.02	-0.15	-0.41	0.69	1.00	0.09	0.09	-0.20
Size	0.73	0.04	0.08	-0.32	0.09	1.00	-0.80	0.83
%Locked	-0.60	-0.08	0.04	0.52	0.09	-0.80	1.00	-0.96
%Liberated	0.65	0.05	0.05	-0.56	-0.20	0.83	-0.96	1.00

Table 5. Values for the fitted parameters a, b and c in equation (2)(2) of the models shown in Figure 16Figure 15.

Performance index\Coefficient	a	b	с	R ²
Recovery	0.60	-4.43	-11.06	0.94
Grade	1921.17	-2386.91	-1.06	0.99
Enrichment	4.85	-14.51	-6.19	0.96