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1	Role of Microstructure in the Electron-Hole Interaction of Hybrid Lead-
2	Halide Perovskites

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- 16 Solar cells based on hybrid inorganic-organic halide perovskites have
- 17 demonstrated high power conversion efficiencies in a range of architectures. The
- existence and stability of bound electron-hole pairs in these materials, and their
- 19 role in the exceptional performance of optoelectronic devices, remains a
- 20 controversial issue. Here we demonstrate, through a combination of optical

21 spectroscopy and multiscale modeling as a function of the degree of polycrystallinity and temperature, that the electron-hole interaction is sensitive to 22 the microstructure of the material. The long-range order is disrupted by 23 polycrystalline disorder and the variations in electrostatic potential found for 24 smaller crystals suppress exciton formation, while larger crystals of the same 25 composition demonstrate an unambiguous excitonic state. We conclude that 26 fabrication procedures and morphology strongly influence perovskite behaviour, 27 with both free carrier and excitonic regimes possible, with strong implications for 28 optoelectronic devices. 29

30

31 Hybrid perovskites represent a new, disruptive technology in the field of optoelectronics. They have the potential to overcome the performance limits of current technologies and 32 achieving low cost and high integrability. Hybrid halide perovskite, e.g. CH₃NH₃PbX₃ [X 33 = CI, Br, or I], solar cells with power conversion efficiency exceeding 20% $^{1, 2}$ are 34 effectively challenging existing thin-film technologies. In addition, the incorporation of 35 hybrid perovskites in optical cavities as lasing materials ^{3,4,5} and in diode structures as 36 efficient light emitters⁶ demonstrates their flexibility and potential for technologically 37 relevant applications beyond photovoltaics. 38

Hybrid perovskites are usually deposited as polycrystalline thin-films with variable
mesoscale morphology depending on the growth conditions and the obtained grain size
ranges from tens to thousands of nm ^{7–9}. Over the last two years the impressive
improvement of photovoltaic performance has been driven by empirical evolution of the

device architecture and processing methodologies. However, there is a considerable 43 lack of understanding of material properties, both as pristine films and their embodiment 44 in a device. Early studies classified the working mechanism of perovskite based-solar 45 cells within a pure excitonic paradigm, ¹⁰ with bound electron and hole pairs being the 46 primary photoexcitation. More recent investigations have put forward a different 47 scenario, where photoexcitation leads mainly to the generation of free electrons and 48 holes, similar to the case of conventional polycrystalline inorganic semiconductors^{4,11–16}. 49 Recently, there have been several reports on the optical properties of hybrid perovskite 50 single crystals^{17–19}, which should represent a reliable reference state. However, even in 51 this case, a few discrepancies are evident, for example, Shi et al.¹⁷ report an optical 52 absorption spectrum with onset at 1.63eV, showing no sign of excitonic states¹⁷, while J. 53 Huang et al.¹⁸ show evidence of a strong excitonic peak at the onset of the external 54 guantum efficiency spectrum, at 1.51 eV^{18} . 55

In this report we describe the interplay between free carriers and excitons, based on 56 transient absorption (TA) spectroscopy and multi-scale numerical simulations. 57 CH₃NH₃PbI₃ (MAPbI₃) crystallized in a mesoporous scaffold, with small grain size (tens 58 of nm), does not support exciton states even at low temperature. In contrast, in 59 hundreds of nm-large domains, as formed by the deposition on flat substrate, free 60 carriers may thermalize and coalesce into the exciton state - depending on temperature 61 and excitation density. Similar behavior is also observed for the higher band gap 62 CH₃NH₃PbBr₃ (MAPbBr₃). Thus, a definitive classification – "excitonic" or "free carrier" 63 semiconductor – as well as a universal value for the exciton binding energy in 64 semiconductors presenting the same chemical composition, is not possible for hybrid 65

perovskites as both regimes are physically accessible by appropriately processing the 66 compounds. We rationalize this by introducing a model for dielectric screening in 67 perovskites. Such a screening is due to polarization of the medium, and can originate 68 from electronic as well as lattice displacements. Here we find it is strongly dependent on 69 the coherent long-range order in the lattice which can be disrupted by imperfections, 70 such as domain walls within crystals or surface defects. As the permanent dipoles 71 associated with the methylammonium cation are free to move within the inorganic cage 72 of the hybrid perovskites, they contribute to lattice polarization, screening the electron-73 74 hole Coulomb interaction.

We first consider a MAPbl₃ film fabricated by the two-step deposition method on 3 µm 75 thick alumina mesoporous scaffold, as one of the most established architecture for 76 efficient perovskite solar cells²⁰ (see Experimental methods for the details on sample 77 preparation). This sample provides simultaneous access to two distinct structural 78 79 morphologies: the crystalline phase grown within the scaffold (meso phase in the following), which, on average, limits the crystal size to the nanometer scale^{7,9} and the 80 81 thick capping layer on top of the scaffold, consisting of crystals up to hundreds of nm in 82 size (see Figures 1a and b, respectively and XRD analysis in Figure S1 in SI). The 83 temperature dependent optical absorption spectrum of such a sample is reported in Figure S2 of the Supplementary Information (SI) and does not show any excitonic 84 transition at its onset, similar to what we have previously reported¹². 85

In agreement with previous works^{4,11,21} the TA spectral evolution of the meso phase
upon photoexcitation above the band-edge can be accounted for by considering the
photo-induced charge-carrier dynamics, without invoking any excitonic contributions.

89 Note that the sample was photo-excited from the substrate side in order to selectively interrogate the perovskite phase grown in the scaffold. At room temperature (RT), the 90 bleaching signal around 1.67 eV shows a rise with time constant of 260 fs due to the 91 92 hot-carrier thermalization to the band edge and a broad photo-induced absorption (PA₁) band that forms for energies higher than 1.77 eV. Briefly, the PB band is assigned to the 93 band filling of the free carriers, while the PA₁ has been tentatively assigned in literature 94 to the change in the refractive index induced by the free carrier population^{4,11,21} (see 95 Figures S3 and S4 and the detailed discussion in the SI). In Figure 1c, by reducing the 96 97 temperature, still above the tetragonal-to-orthorhombic phase transition, we observe that the PB band red-shifts, gets narrower and gains intensity. The change in the line 98 shape is consistent with the lower thermal energy that reduces the homogeneous 99 broadening. The red shift can be simply related to the Varshni effect ²²(see discussion 100 in Figure S5 in the SI). It is worth underlining here that this sample *does not show* any 101 excitonic peak in the absorption spectrum even at 4K (see Figure S2b). 102

In Figure 1d we show the temperature dependent TA spectra of the same sample 103 illuminated from the capping side. Since pump photon penetration depth is comparable 104 to the thickness of the capping layer, we mainly excite the large crystals in the capping 105 phase, ⁴⁰ though some contribution from the smaller crystals within the scaffold can be 106 present (see Figure 1b). At RT we note that the PB band is broader and red-shifted 107 with respect to the PB of the meso phase of the film. The red shift is due to the 108 reduction of the band gap in the large crystals,¹³ while the broadening can be related to 109 contributions from the meso phase underneath. The most striking difference appears 110 when the sample is cooled down. In particular, at 170K the PB band is strongly red-111

112 shifted. Such a large red-shift does not follow the standard Varshni trend. Furthermore, the PA₁ band is simultaneously quenched and a new negative band appears, peaking 113 around 1.67eV. Note that this is not simply related to the low-temperature structural 114 phase transition that occurs below 170K^{12,23}, which would lead to a blue shift of the 115 whole spectrum as a consequence of a widening of the semiconductor band-gap (see 116 Figures S6 and S7 and the discussion in SI). The time evolution of the TA spectra at 117 170K is shown in Figure 1e. At 200 fs a positive band peaking close to 1.63 eV is 118 present. In about 1ps ($\tau \sim 260$ fs) a negative band peaking at 1.67eV forms along with 119 the red shift of the PB, that falls outside our experimental range. Such a behavior has 120 been well documented in semiconductors as a result of self-normalization of the exciton 121 122 energy-i.e. blue shift of the exciton absorption - due to exciton-exciton and excitoncarrier interaction^{24–28}. Thus, the negative band is the result of a modulation (in the 123 following we indicate it as MA*) and can be considered as a fingerprint for exciton 124 population^{24,26}. As previously asserted, large crystals show a clear excitonic transition 125 at the absorption edge that gains strength upon cooling ^{7,12,29}. Even considering the 126 lowest exciton binding energy reported so far in literature (i.e. 5meV^{14,16, 30}), one can 127 expect a decent exciton population at 170K, at the photoexcitation densities used (a 128 simple guideline to estimate exciton population fraction depending on the exciton 129 130 binding energy value, temperature and excitation density is presented in Figure S8). Accordingly, at 170K the formation of MA* is indicative of exciton formation upon carrier 131 thermalization. We estimate that the carrier coalescence into the bound excitonic state 132 133 occurs within 1 ps (see dynamics in Figure 1f), consistent with the similar phonon

134 assisted phenomenon that occur in the band relaxation. Note that in 2D hybrid perovskites the formation of MA* has also been reported, although slower^{24,28}. 135 In this specific morphology, the exciton population appears only upon temperature 136 reduction, implying that the exciton binding energy is insufficient to stabilize the exciton 137 population at RT. To broaden the perspective of our observation we fabricated 138 "cuboids-like" films of MAPbl₃²⁰, with controlled crystal dimension of either < 200 nm or 139 ~ 1 µm from visual inspection of SEM images in Figure 2a and 2b respectively (see 140 Figure S1 in SI for the XRD analysis). The UV-vis absorption spectra at RT are shown 141 in Figure S9 in the Supplementary Information. The TA spectrum of the film with < 200 142 nm crystal size (indicated in the legend of Figure 2c as "small crystals") closely 143 resembles the one obtained in the "meso phase" of the MAPbl₃ film at RT, showing that 144 the thermalized carriers stay free at the band edge - see TA spectra at 1ps, after carrier 145 146 thermalization, in Figure 2c and the entire spectral dynamics in Figure S10. No exciton feature is present. On the contrary, the sample made of ~1 µm large crystals exhibits, at 147 RT, different spectral features and dynamics. The TA spectrum forming in 1ps upon 148 149 photoexcitation above band-gap resembles the one of the capping layer at 170K (Figure 2c). In particular we highlight the presence of the MA* band even at RT, with a 150 formation time of about 270 fs (see inset in Figure 2d). Note that this sample keeps the 151 same TA spectral features even at 77K, where a clear stable excitonic state is also 152 present at the band edge (see UV-Vis spectra in Figure S6 in the SI), albeit shifted to 153 higher energies (about 95 meV, see TA spectrum in Figure S7 in the SI) as 154 consequence of a phase transition. In Figure 2c the red-shifted TA spectrum taken at 155 77K is reported (dashed line) for easier comparison. This provides additional support to 156

our assignment of the TA spectra, and the correct prediction of a bleaching band just
outside the experimental range. In Figure 2d we show the details of the spectral
evolution of the sample with large crystals. Importantly, at longer time delays (>10 ps)
the MA* band reduces and the PB shifts to higher energies, towards the free carrier
bleach. This dynamic reflects the decay of the excitonic population that appears to be
shorter lived with respect to the free carrier population, as further confirmed by the TA
spectra in the ns time regime (see Figure S11 in the SI).

We have shown so far that different morphologies of MAPbl₃ thin films – with average 164 crystal size varying from tens to hundreds of nm may *i*) support only the free carrier 165 166 population, even at low temperature; *ii*) support an excitonic population upon temperature reduction; iii) sustain the formation of a fraction of short living excitons at 167 room temperature. Since the photo-excitation density used in the above three cases is 168 169 the same, this clearly indicates that the electron-hole interaction is modified by the degree of polycrystallinity in the film (please refer to Figure S8 for a simple visualization 170 of the variation of exciton population fraction as a function of exciton binding energy at a 171 given photoexcitation density). Thus, the exciton binding energy is not uniquely 172 determined by the chemical composition of the polycrystalline material but it can be 173 tuned in a range between a few to tens of meV^{14-16} . 174

To further generalize our observations we also consider thin films of MAPbBr₃. The halogen substitution induces a lowering of the valence band of the semiconductor and a blue shift of the optical gap, making the material appealing for a variety of applications such as high Voc solar cells^{31,32}, water splitting and light emitting devices^{3,6}. Seminal studies have suggested larger exciton binding energy for MAPbBr₃ with respect to

MAPbl₃³³. However, the optical spectra reported by some of the recent works^{32,34,35} do 180 not show any strong excitonic feature at RT. To verify the role of morphology also in this 181 system we prepared MAPbBr₃ thin films with average crystal dimensions much smaller 182 than 100 nm by growing them in an AI_2O_3 mesoporous scaffold and ~1 μ m (see SEM 183 images in Figure S12 of SI). Figure 3a and Figure 3b show the UV-vis spectra of such 184 samples while Figure 3c and Figure 3d show the photo-induced TA spectra when 185 exciting above band-gap at RT. Small crystals do not show any excitonic feature at the 186 on-set of the UV-vis absorption spectrum, at RT. In perfect agreement, the TA spectrum 187 resembles very much, in shape and dynamics, that from meso MAPbl₃, pointing to a 188 free carrier picture (refer to Figures S3 and S4 and the discussion in the SI). In contrast, 189 large crystals show a sharp excitonic feature at the onset of the absorption spectrum 190 (Figure 3b). In agreement, the TA spectra (Figure 3d) of the large MAPbBr₃ crystals 191 show the formation of a PB band at 2.34 eV that matches with the excitonic transition, 192 together with the appearance of the modulation feature, MA*, at 2.43 eV in the first ps (a 193 comparison of the 1ps spectra from the small and the large crystal is reported in the 194 inset of Figure 3d). This behavior indicates the formation of an exciton population upon 195 196 carrier thermalization which eventually recombines in hundreds of ps. The latter is clearly demonstrated by the fact that beyond 1ps the spectra lose their intensity but do 197 not change their spectral shape (see also dynamics in Figure S13 of the SI). 198

These results show that also for MAPbBr₃ it is not possible to asses a unique value for
the exciton binding energy, which will depend on the thin film morphology. Thus,
optoelectronic devices made of large MAPbBr₃ crystals, with a stable excitonic

population at RT, will work in a different manner with respect to those made of thin filmswith a higher degree of polycrystallinity.

The effect of the degree of polycrystallinity on the exciton binding energy can be 204 rationalized by considering the role of disorder in such hybrid systems, with particular 205 emphasis on the orientational order of the organic cation within the material. Large 206 perovskite crystals (~1 µm in size) show a cooperative ordered phase of the organic 207 cations which affects their rotational degrees of freedom^{9,36}. This is not intrinsically 208 related to the size of the crystal, but to the "quality" of the crystallization process. We 209 have shown recently that the crystallization process does affect the optoelectronic 210 properties through the modulation of the lattice strain^{8,13}. Raman analysis ^{9,29} on the 211 meso-phase of MAPbl₃ (see Figure S14 in the SI) suggests a more distorted structural 212 arrangement, thus dipoles in the small crystals might be more randomly oriented within 213 the inorganic cage. Of course, this can be induced by different factors, e.g. the 214 215 crystallization procedure, the presence of dangling bonds on the surface, or the influence of external agents. 216

The organic cation has a permanent dipole moment, generating an electrostatic potential. If free to rotate, the dipoles will respond in a dielectric manner. This increases the low frequency (hundreds of GHz) dielectric constant up to 35 ^{37,38} in MAPbl₃. The Mott-Wannier exciton binding energy can be written as $E_b = \frac{m^* e^4}{\hbar^2 \mathcal{E}^2}$. If we take the optical frequency dielectric constant ($\varepsilon \sim 5$) this value is 45 meV, with an effective exciton Bohr radius of 4 nm. The binding energy is in very good agreement with the experimental value of ~50 meV ¹². However, this model is valid only if the Coulomb interaction 224 between the electron and hole is strong enough (and thus the kinetic energy of the small exciton high enough) that the slower lattice dielectric response does not screen 225 the interaction. If the exciton is less strongly bound, we must consider also the low 226 227 frequency component of the dielectric constant arising from the lattice contributions. This would result in the exciton sampling a higher dielectric constant, thus decreasing 228 its binding energy (to 2 meV) and increasing the size to 19 nm, eventually dissociating 229 it. In order to see how the different screening regimes are linked to the crystallization 230 process we consider the microscopic effect of disorder and temperature by sampling the 231 232 electrostatic potential resulting from simulating mono-crystalline and polycrystalline films. We describe the changes in electrostatic potential upon moving from a large grain 233 to microcrystalline structure by extending a Monte Carlo procedure based upon a model 234 Hamiltonian parameterized for MAPbl₃³⁹. The grain boundaries are induced by 235 incorporating inactive lattice sites ("point defects") in the simulation (at densities of 6% 236 and 10%), which enforces polycrystallinity within the simulation domains. The standard 237 deviation in the electrostatic potential is plotted as a function of temperature and defect 238 density (degree of polycrystallinity) in Figure 4 (top panel), while representative domain 239 structures, and associated electrostatic potentials, are shown in Figure 4a-c (bottom 240 panels). 241

For the mono-crystalline system, the standard deviation in the electrostatic potential
drops to zero with decreasing temperature. All the rotational disorder of the organic
cations is quenched leading to complete order and formation of fully twinned domains.
Disorder grows with temperature as would be expected from statistical mechanics,
generating increasing electrostatic potential variance. We note that this is in full

247 agreement with a recent work published by R. Nicholas and co-workers, which reports an increase in the exciton binding energy upon reduction of temperature¹⁶. At room 248 temperature, the electrostatic potential is fairly disordered, with a standard deviation of 249 250 163 meV, and the degree of polycrystallinity matters less at the level of defect density considered as all samples are thermally disordered. At lower temperatures however, the 251 variation in electrostatic potential is proportional to the degree of polycrystallinity and 252 does not disappear at 0K for the polycrystalline films (explicit tetragonal-orthorhombic 253 phase transitions are not treated by the model). The largest variation in electrostatic 254 potential occurs at grain boundaries, where the dipole twinning is disrupted (see Figure 255 4a-c bottom panels, dipoles alignment are represented in different morphologies and 256 temperatures). These simulations confirm an interesting trend, that the variance of the 257 258 electrostatic potential (i.e. local screening) can be controlled by the local order within the crystal. With larger, less defective, crystals the variance is minimized. Thus, electron-259 hole separation due to electrostatic disorder should be significant in small crystals 260 (countering the Coulomb attraction between electrons and holes) but weaker in large 261 crystals (allowing for Wannier exciton formation). 262

A single crystal sample should represent the ultimate case study for our model. However, as mentioned earlier, discrepancies can be found in literature with regards to the optical properties of such a sample^{17,18}. Indeed we have observed that a single crystal can show energetic dishomogeneity within surface and bulk phases with respect to the optical gap, following the same trend as the small and large crystallites (see Figure S15 in the SI). This can be understood from the presence of defects and fluctuations at the crystal termination. The surface of the single crystal is an extended 270 defect and it should be considered as comparable to the sample made of small crystallites. Due to the high absorbance of the single crystal, the absorption spectra 271 presented in literature have been measured through reflectivity^{17,18}, which is more 272 sensitive to the surface rather than the bulk of the semiconductor. Thus, according to 273 our model, we do not expect to see excitonic features. On the other hand, we have 274 noticed that the EQE spectra (which may be more sensitive to the bulk properties of the 275 semiconductor) of solar cells embodying single crystal of MAPbl₃ shows a defined 276 exciton like peak at the band edge in contrast to polycrystalline thin-film based 277 devices¹⁸. This observation suggests the strengthening of an excitonic transition, at 278 room temperature, in the bulk of the single crystal as we predict. 279 Thus we conclude that the sensitivity of the molecular order to the crystal quality, 280

defects, as well as induced strain and device history, implies that there is considerable

scope in the material processing to tune the nature and the dynamics of the

283 photophysical mechanisms characterizing each sample. Control of the dynamic

polarization effect, which can provide both free carrier and excitonic regimes for a single

material composition, may open up a plethora of novel optoelectronic applications.

286

287 **Experimental Methods:**

288 <u>Synthesis of the Precursor solutions:</u>

289 <u>Methylammonium iodide salt.</u> The precursor solution of perovskite was prepared
 290 following the well-established method reported in literature¹. Methylamine solution (33%)

wt. in absolute ethanol, Sigma-Aldrich) was reacted with hydroiodic acid (57% wt. in

water, Sigma-Aldrich), with excess methylamine in ethanol at 0°C. Crystallization of
CH₃NH₃I was achieved using a rotary evaporator; a white colored powder was formed,
indicating successful crystallization. The salt was washed twice in diethyl ether to
remove impurities.

Methylammonium bromide salt. The solution was prepared as reported elsewhere⁶. The 296 solution was prepared by adding methylamine solution (33% wt. in absolute ethanol, 297 Sigma-Aldrich) and hydrobromic acid (48% wt. in water, Sigma-Aldrich) to 100 ml of 298 absolute ethanol. The reaction mixture was stirred at 0°C. The solvent was removed by 299 rotary evaporation. The obtained white crystals were washed with anhydrous diethyl 300 301 ether and recrystallized in ethanol. The perovskite precursor solution was prepared by mixing CH₃NH₃Br and PbBr₂ in a 1:1 molar ratio in anhydrous N,N-dimethylformamide 302 to give concentrations of 20% and 5% wt. 303

304 Preparation of samples for spectroscopy:

All the samples were prepared in a controlled nitrogen atmosphere either on glass microscope slides or on mesoporous Al_2O_3 .

307 <u>Mesoporous Al_2O_3 </u>: a commercial alumina nanoparticles dispersion (20% wt. in IPA, 308 nanoparticles average size around 50 nm, *Sigma-Aldrich*) was spin-coated at 2000 rpm 309 to form a ~3 µm thick mesoporous layer. These were then dried at 150°C for 30 minutes 310 in air and 10 minutes under inert atmosphere.

311 $CH_3NH_3PbI_3$ deposition method on mesoporous AI_2O_3 : A hot (70°C) solution of PbI2 in 312 DMF (0.5M) was spin coated at 2000 rpm for 60 seconds and subsequently annealed at 313 70°C for 30 minutes. After letting it cool down to room temperature, the substrate was dipped at room temperature in a CH₃NH₃I solution (0.063 M) in anhydrous IPA for 2
 minutes. Samples are finally rinsed in anhydrous IPA to remove the excess of
 unreacted CH₃NH₃I .

317 <u>Glass</u>: the substrates were cleaned (two cycles of water, acetone and IPA in an
 318 ultrasonic bath for 10 minutes each) followed by an oxygen plasma treatment for 10
 319 minutes.

<u>CH₃NH₃PbI₃ deposition method on glass:</u> A hot (70°C) solution of PbI₂ in DMF (1M) 320 was spin coated at 2000 rpm for 60 sec in order to obtain a 300 nm thick layer. A 321 subsequent annealing at 70°C for 30 min was required to obtain the PbI₂ thin film. After 322 letting it cool down to room temperature, the substrate was dipped in a CH_3NH_3I 323 324 solution in anhydrous IPA for 2 minutes. To obtain samples with different crystal sizes, the concentration and temperature of CH₃NH₃I was varied. To obtain films with ~100 325 nm crystals (Figure 2a), the concentration was set at 0.063 M, and the dipping was 326 327 performed at room temperature, while the concentration was reduced to 0.045 M and the bath was warmed up to 70°C to obtain crystals above 1µm large (Figure 2b). 328 Samples are finally rinsed in anhydrous IPA to remove the excess of unreacted 329 CH₃NH₃I. 330

331 <u>*CH*₃*NH*₃*PbBr*₃ one step-deposition method: CH₃*NH*₃Br and PbBr₂ were both dissolved 332 in DMF (concentration of 20% wt. for the deposition on glass substrate and 10%wt. for 333 deposition in the mesoporous alumina scaffold). The solution was spin coated at 3000 334 rpm for 60 seconds. A subsequent annealing at 100°C for 15 min is required to obtain 335 the perovskite thin film.</u>

336 Scanning Electron Microscopy

High Resolution Scanning Electron Microscopy (HRSEM) was used for Figures 1a and
1d. The samples were sticked on aluminum stubs with ultra smooth double-sided
adhesive tape, made of conductive carbon, specific for UHV systems and then coated
with a 15 nm layer of conductive amorphous carbon. HRSEM observation was carried
out using a JEOL JSM 7500FA scanning electron microscope, equipped with a cold
field emission gun (single crystal tungsten <310> emitter, ultimate resolution of 1 nm)
and operating at 10 kV.

The SEM images shown in Figures 2 have been collected by using an high vacuum

tungsten filament commercial Jeol 6010-LV, with a working bias of 20 kV.

346 <u>Ultraviolet–visible absorption</u>

347 Absorption spectra have been recorded using a UV-VIS-NIR spectrophotometer

348 (PerkinElmer Lambda 1050 model) with a spectral range from 200 nm to 2000 nm, with

a resolution of about 1 nm.

350 <u>Femtosecond transient absorption set-up:</u>

In a typical pump-probe experiment, the system under study is photoexcited by a short pump pulse (~120 fs) and its subsequent dynamical evolution is detected by measuring the transmission changes ΔT of a delayed probe pulse as a function of pump-probe delay and probe wavelength . The signal is given by the differential transmission $\Delta T/T =$ [(Tpump on-Tpump off)/Tpump off]. The system is driven by a mode-locked Ti:Sapphire oscillator (Coherent Micra-18) operating at 80 MHz was used as a fundamental 357 broadband source. This provided pulses with durations of ~20 fs and a central wavelength of 800 nm. A grating based pulse stretcher (Coherent 9040) was used to 358 temporally expand the pulses before amplification in a 250 kHz actively Q-switched 359 360 Ti:Sapphire based regenerative amplifier (Coherent RegA 9000). The amplified pulses were subsequently temporally compressed in a grating based compressor (Coherent 361 9040), resulting in pulses with temporal widths of \sim 35 fs and energies of \sim 6 μ J. A thin 362 beam splitter is used to split the amplified output into pump and probe beams. The 363 pump beam is input into a two-pass BBO-based collinear OPA (Coherent 9450), 364 allowing spectral conversion to any desired wavelength in the 480-750 nm wavelength 365 range with resulting temporal broadening to ~120 fs. The probe beam was used for 366 super-continuum generation within a sapphire plate, leading to probe pulses with 367 significant continuous spectral content from 480-780 nm and temporal widths of ~100 fs. 368 Both pump and probe pulses were focused and spatially overlapped in the sample 369 space, with the temporal delay between them given by an optical retro-reflective delay 370 line located on the pump arm of the system. Great care was taken to ensure the spot 371 size of the probe beam was significantly smaller than that of the pump beam. The 372 resulting probe signal typically measured in transmission is coupled into an Acton 373 SP2300i imaging spectrograph and the dispersed signal was measured by a custom 374 (Stresing) silicon based CCD linear array. The minimum detectable signal is $\Delta T/T \sim 10^{-5}$. 375 The pump beam energy density used in the experiment is kept deliberately low (pump 376 fluence less than 1 μ J/cm², which results in excitation densities in the order of 10¹⁷ cm⁻ 377 ³). All the measurements were taken with the samples in a vacuum chamber to prevent 378 379 any influence from oxygen or sample degradation. The temperature-dependent

380 experiments were carried out using a continuous flow static exchange gas cryostat (Oxford Instruments). The cryostat consist of three chambers, one inside the other. The 381 sample is housed inside the internal chamber filled with gaseous nitrogen. The 382 cryogenic liquid (N_2) is fluxed inside the second chamber allowing temperature control 383 of the N₂ atmosphere of the sample chamber. Eventually a third chamber is evacuated 384 (~ 10^{-5} - 10^{-6} mbar) in order to assure thermal isolation from the external ambient. A 385 sensor close to the sample has been mounted in order to accurately monitor the sample 386 387 temperature.

388 <u>Theoretical Simulations</u>

The Starrynight (molecular ferroelectric simulation) code³⁹ was adapted to model 389 defective domains. Simulations were carried out in two dimensions with a 25 meV 390 interaction between near-neighbour dipoles, no cage-strain term, a 3 unit-cell cut-off for 391 dipole interactions and periodic boundary conditions on a two dimensional 250x250 392 393 grid. Strain (ordering) terms would be required to fully describe the tetragonalorthorhombic phase transition and are expected to increase the order-disorder transition 394 temperature of a phase, and so the effective temperature reported in Figure 4 may be 395 considerably higher. The initial dipole orientation was random. 10⁵ Monte-Carlo moves 396 were attempted per site, with a Metropolis algorithm. The electrostatic potential variation 397 was calculated from the sampled dipole orientation at equilibrium with a 10 unit-cell cut-398 off. 399

400 **References**:

Zhou, H. *et al.* Interface engineering of highly efficient perovskite solar cells.
 Science 345, 542–546 (2014).

- 403 2. Grätzel, M. The light and shade of perovskite solar cells. *Nat. Mater.* 13, 838–842 (2014).
- Xing, G., Mathews, N., Lim, S. & Yantara, N. Low-temperature solution-processed
 wavelength-tunable perovskites for lasing. *Nat. Mater.* 13, 476–480 (2014).
- 407 4. Deschler, F. *et al.* High Photoluminescence Efficiency and Optically Pumped
 408 Lasing in Solution-Processed Mixed Halide Perovskite Semiconductors. *J. Phys.*409 *Chem. Lett.* 5, 1421–1426 (2014).
- 5. Zhu, H. *et al.* Lead halide perovskite nanowire lasers with low lasing thresholds and high quality factors. *Nat. Mater.* **14**, 636-642 (2015).
- 412 6. Tan, Z.-K. *et al.* Bright light-emitting diodes based on organometal halide
 413 perovskite. *Nat. Nanotechnol.* 9, 687 692 (2014).
- 414 7. Ball, J. M., Lee, M. M., Hey, A. & Snaith, H. J. Low-temperature processed meso415 superstructured to thin-film perovskite solar cells. *Energy Environ. Sci.* 6, 1739
 416 (2013).
- 8. De Bastiani, M., D'Innocenzo, V., Stranks, S. D., Snaith, H. J. & Petrozza, A. Role
 of the crystallization substrate on the photoluminescence properties of organolead mixed halides perovskites. *APL Mater.* 2, 081509 (2014).
- 420 9. Grancini, G. *et al.* The Impact of the Crystallization Processes on the Structural
 421 and Optical Properties of Hybrid Perovskite Films for Photovoltaics. *J. Phys.*422 *Chem. Lett.* 5, 3836–3842 (2014).
- Marchioro, A. *et al.* Unravelling the mechanism of photoinduced charge transfer
 processes in lead iodide perovskite solar cells. *Nat. Photonics* 8, 250–255 (2014).
- 425 11. Manser, J. S. & Kamat, P. V. Band filling with free charge carriers in organometal
 426 halide perovskites. *Nat. Photonics* 8, 737 743 (2014).
- 427 12. D'Innocenzo, V. *et al.* Excitons versus free charges in organo-lead tri-halide
 428 perovskites. *Nat. Commun.* 5, 3486 (2014).
- D'Innocenzo, V., Srimath Kandada, A. R., De Bastiani, M., Gandini, M. &
 Petrozza, A. Tuning the light emission properties by band gap engineering in
 hybrid lead-halide perovskite. *J. Am. Chem. Soc.* **136**, 17730–17733 (2014).
- Lin, Q., Armin, A., Nagiri, R. C. R., Burn, P. L. & Meredith, P. Electro-optics of perovskite solar cells. *Nat. Photonics* (2014). doi:10.1038/nphoton.2014.284

434 15. Saba, M. *et al.* Correlated electron-hole plasma in organometal perovskites. *Nat.* 435 *Commun.* 5, 5049 (2014).

- 436 16. Miyata, A. *et al.* Direct Measurement of the Exciton Binding Energy and Effective
 437 Masses for Charge carriers in an Organic-Inorganic Tri-halide Perovskite. *Nat.*438 *Phys.* **11**, 582-587 (2015).
- 439 17. Shi, D. *et al.* Low trap-state density and long carrier diffusion in organolead
 440 trihalide perovskite single crystals. *Science* **347**, 519–522 (2015).
- 18. Dong, Q. *et al.* Electron-hole diffusion lengths > 175 μm in solution-grown
 CH3NH3Pbl3 single crystals. *Science* 347, 967 970 (2015).
- 19. Nie, W. *et al.* High-efficiency solution-processed perovskite solar cells with
 millimeter-scale grains. *Science* 347, 522–525 (2015).
- 445 20. Im, J.-H., Jang, I.-H., Pellet, N., Grätzel, M. & Park, N.-G. Growth of CH3NH3PbI3
 446 cuboids with controlled size for high-efficiency perovskite solar cells. *Nat.*447 *Nanotechnol.* 9, 927–932 (2014).
- 448 21. Xing, G. *et al.* Long-range balanced electron- and hole-transport lengths in 449 organic-inorganic CH3NH3PbI3. *Science* **342**, 344–7 (2013).
- 450 22. Varshni, Y. P. Temperature dependence of the energy gap in semiconductors.
 451 *Physica* 34, 149–154 (1967).
- 452 23. Onoda-Yamamuro, N., Matsuo, T. & Suga, H. Calorimetric and IR spectroscopic
 453 studies of phase transitions in methylammonium trihalogenoplumbates (II)[†]. J.
 454 Phys. Chem. Solids **51**, 1383–1395 (1990).
- Shimizu, M., Fujisawa, J.-I. & Ishi-Hayase, J. Influence of dielectric confinement
 on excitonic nonlinearity in inorganic-organic layered semiconductors. *Phys. Rev.*B71, 205306 (2005).
- 458 25. Hulin, D. *et al.* Well-size dependence of exciton blue shift in GaAs multiple-459 quantum-well structures. *Phys. Rev. B* **33**, 4389–4391 (1986).
- Peyghambarian, N. *et al.* Blue shift of the exciton resonance due to excitonexciton interactions in a multiple-quantum-well structure. *Phys. Rev. Lett.* 53,
 2433–2436 (1984).
- Schmitt-Rink, S., Chemla, D. & Miller, D. Theory of transient excitonic optical
 nonlinearities in semiconductor quantum-well structures. *Phys. Rev. B* 32, 6601–
 6609 (1985).
- Wu, X., Trinh, M. T. & Zhu, X. Excitonic Many-Body Interactions in TwoDimensional Lead Iodide Perovskite Quantum Wells. *J. Phys. Chem. C* (2015).
 doi:10.1021/acs.jpcc.5b00148

- 469 29. Quarti, C., Grancini, G. & Mosconi, E. The Raman Spectrum of the CH3NH3PbI3
 470 Hybrid Perovskite: Interplay of Theory and Experiment. *J. Phys. Chem. Lett.* 5,
 471 279–284 (2014).
- 472 30. Even, J. *et al.* Solid-State Physics Perspective on Hybrid Perovskite
 473 Semiconductors. *J. Phys. Chem. C* **119**, 10161-10177 (2015).
- 474 31. Heo, J. H., Song, D. H. & Im, S. H. Planar CH 3 NH 3 PbBr 3 Hybrid Solar Cells
 475 with 10.4% Power Conversion Efficiency, Fabricated by Controlled Crystallization
 476 in the Spin-Coating Process. *Adv. Mater.* 26, 8179–8183 (2014).
- 477 32. Edri, E., Kirmayer, S., Cahen, D. & Hodes, G. High Open-Circuit Voltage Solar
 478 Cells Based on Organic–Inorganic Lead Bromide Perovskite. *J. Phys. Chem. Lett.*479 4, 897–902 (2013).
- 480 33. Tanaka, K. *et al.* Comparative study on the excitons in lead-halide-based
 481 perovskite-type crystals CH3NH3PbBr3 CH3NH3PbI3. *Solid State Commun.* **127**,
 482 619–623 (2003).
- 483 34. Hoke, E. T. *et al.* Reversible photo-induced trap formation in mixed-halide hybrid 484 perovskites for photovoltaics. *Chem. Sci.* **6**, 613–617 (2014).
- 35. Sadhanala, A. & Deschler, F. Preparation of Single-Phase Films of CH3NH3Pb
 (I1-x Br x) 3 with Sharp Optical Band Edges. *J. Phys. Chem. Lett.* 5, 2501–2505
 (2014).
- Mosconi, E., Quarti, C., Ivanovska, T., Ruani, G. & De Angelis, F. Structural and
 electronic properties of organo-halide lead perovskites: a combined IRspectroscopy and ab initio molecular dynamics investigation. *Phys. Chem. Chem. Phys.* 16, 16137–44 (2014).
- 492 37. Wasylishen, R., Knop, O. & Macdonald, J. Cation rotation in methylammonium
 493 lead halides. *Solid State Commun.* 56, 581–582 (1985).
- 494 38. Poglitsch, A. & Weber, D. Dynamic disorder in
 495 methylammoniumtrihalogenoplumbates (II) observed by millimeter-wave
 496 spectroscopy. *J. Chem. Phys.* 87, 6373 (1987).
- 497 39. Frost, J. M., Butler, K. T. & Walsh, A. Molecular ferroelectric contributions to
 498 anomalous hysteresis in hybrid perovskite solar cells. *APL Mater.* 2, 081506
 499 (2014).
- The pump-probe signal is due to the photo-excited volume, while the unexcited
 regions (however large) do not contribute. Transmission of the probe through the
 unexcited volume of the sample is cancelled out in the transmission difference
 signal.

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514

515 *Author contributions:*

GG, ARSK and AJB performed the Transient Absorption Measurements; MG and MDB
prepared the samples and characterized them by SEM; SM performed XRD and SEM
characterization. GG, ARSK, GL and AP analyzed the optical spectroscopy data; JMF
and AW performed the multi-scale modelling and analyzed the results; the manuscript
was written through contributions of all authors; AP supervised the project.

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522 Figure Captions

523

524	Figure 1	Temperature-de	pendent transient absor	ption spectra (of MAPbl₃ meso
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525 **phase and capping layer.** SEM images of (a) the "meso-phase" and of (b), the capping

layer of the 3μ m-thick MAPbl₃ sample. Note that (a) is showing a zoom-in of figure (b). 526 (c) Temperature dependence of the TA spectra at 1ps pump-probe delay of the meso 527 phase; (d), Temperature dependence of the TA spectra at 1ps pump-probe delay of the 528 "capping layer". (e), TA spectral evolution between 200fs and 1ns at 170 K of the 529 530 capping layer. (f) dynamics probed at 1.72 eV (MA* band) of the capping layer at 170K compared to the dynamics probed at 1.64eV (PB band) of the same sample at RT. For 531 532 all the TA measurements, the excitation wavelength is at 2.38eV with an excitation density of approximately 5×10^{17} cm⁻³. 533

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Figure 2| Photo-induced excited population of MAPbI₃ as a function of the crystal 535 size. SEM images of perovskite films with average crystal dimension of: (a), < 200 nm 536 and (b), ~ 1µm. Scale bar: 2µm. (c), TA spectra, at room temperature, at 1 ps pump-537 probe delay of the two samples (red squares and blue circles, respectively) along with 538 the TA spectrum (dashed line) of sample (b) taken at 77K, red shifted by 95meV. (d) TA 539 spectra at different time delays at RT from the sample shown in (b). The inset shows the 540 dynamics probed at 1.63 eV and 1.66 eV. For all the TA measurements, the excitation 541 wavelength is at 2.38eV with an excitation density of approximately 5×10^{17} cm⁻³. 542

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544 Figure 3 Photo-induced excited population of MAPbBr₃ as a function of the

crystal size. Uv-vis absorption spectra from MAPbBr3 films with average crystallite dimension (a) << 100 nm and (b), \sim 1µm. TA spectra at different time delays at RT from

the sample with (c) small and (d) large crystallites (see SEM images in Figure S12 of

SI). In the inset of Figure 3d the comparison of the TA spectra at 1 ps of small and large MAPbBr₃ crystals. For all the TA measurements, the excitation wavelength is at 3.1 eV with an excitation density of approximately 1×10^{17} cm⁻³

551

552 Figure 4| Multi-scale numerical simulations of dipole alignment in

553 methylammonium lead iodide.

(Top) Standard deviation of electrostatic potential in Starrynight simulations of 250x250 554 interacting dipoles (150 x 150 nm crystallite), as a function of temperature. The standard 555 556 variation in electrostatic for defect free perovskite crystals (blue) decays reduces to zero with a decrease in temperature. The 6% point defects (orange) and 10% point defect 557 (red) trends show that there is non-vanishing disorder in the electrostatic potential, even 558 559 at zero temperature. Below the figure we show small excerpts (25x25) of the simulation showing both dipole alignment (top) by pixel hue, and the resulting electrostatic 560 potential (bottom). (a) Pure domains at zero Kelvin are highly ordered in a columnar 561 antiferroelectric alignment leading to a smooth electrostatic potential, whereas defective 562 crystals (b) at zero Kelvin contain electrostatic potential disorder as a result of frustrated 563 alignment of the domains at the point defects. Room temperature domains (c) show that 564 the thermal disorder at room temperature leads to a mostly paraelectric phase, with 565 considerable electrostatic potential variance. (d) Schematic representation of electron-566 567 hole interaction driven by electrostatic potential fluctuations. In samples where there is considerable electrostatic potential variation, the exciton will not be stable. 568

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