

1 A comparison of fresh and used aircraft oil  
2 for the identification of toxic substances  
3 linked to aerotoxic syndrome.

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10

11 **Abstract**

12 Fresh and used aircraft engine lubricants (Mobil Jet Oil II) were analysed using a Fourier Transform Ion  
13 Cyclotron Resonance Mass Spectrometer (FTICRMS) and comprehensive two dimensional gas  
14 chromatography with high resolution time of flight mass spectrometry (GCxGC-HRTOFMS). The  
15 composition of the fresh oil was established, with special focus to its tricresyl phosphate (TCP) content  
16 as this has formed the focus for most investigations into aerotoxic syndrome. The results showed that  
17 only four TCP isomers were present at detectable levels in the fresh oil: mmm-TCP, mmp-TCP, ppm-TCP

18 and ppp-TCP. The results indicate that the formulation of Mobile Jet Oil II does not contain the more  
19 toxic ortho substituted TCP isomers at concentrations above 0.0005%. The temperatures of jet engines  
20 during operation are greater than 200°C which creates the potential to alter the composition of the  
21 original oil and create other toxic compounds. The results show there may be a significant risk from  
22 alkylated cresyl phosphates, which were identified in the used oils at concentrations calculated in the  
23 range of 0.13 to 0.69%. w/w. Several xylenyl and ethylphenyl phosphates have been shown to exhibit a  
24 similar toxicity to ortho substituted TCP isomers which makes their discovery in used oil significant.  
25 These compounds should be included in future aircraft air quality studies and when assessing the risks  
26 and causes of aerotoxic syndrome.

27

## 28 **Key words**

29 Aerotoxic, multidimensional chromatography, high resolution mass spectrometry, cresyl phosphates,  
30 organophosphates

## 31 1. Introduction

32 Due to their widespread use as pesticides, plasticizers and flame retardants organophosphates are  
33 routinely detected in environmental samples. However, a specific concern has arisen in recent decades  
34 as aircraft crew have developed symptoms consistent with exposure to toxic fumes and  
35 organophosphates (Abou-Donia et al. 2013; Harrison and Mackenzie Ross, 2016; Liyasova et al. 2011;  
36 Payne, 2015). There have been reports of headaches, loss of balance, numbness and neurobehavioral  
37 abnormalities such as emotional instability, depression and cognitive dysfunction, including impaired  
38 short term memory, blurred vision and speech, altered coordination (de Ree et al. 2014; Abou-Donia et  
39 al. 2013). Organophosphates are not just used as pesticides on crops in fields but are also used in

40 aircraft as flame retardants, in engine oil and hydraulic fluids, and on material surfaces. This has resulted  
41 in their link to aerotoxic syndrome by Winder and Balouet (2002). The term Aerotoxic Syndrome was  
42 first published in 1999 by an international scientific team to describe the symptoms and exposure  
43 conditions reported by aircraft crew from Australia, US, Europe. Whilst aerotoxic syndrome has not  
44 been fully accepted as a medical syndrome (Wolkoff et al., 2016) it is commonly used to refer to air  
45 quality in aircraft and the associated exposure of crew and passengers to toxic compounds. Aerotoxic  
46 syndrome is understood to be caused by long term and repeated exposure to chemicals released from  
47 smoke and fume events in aircraft.

48 During a flight the air cabin pressure and temperature are maintained with outside air that is passed  
49 through the jet engine. Cabin air is recycled for 50%, whereas the flight deck air is comprised in most  
50 aircraft types of a continuous stream of bleed air (de Boer et al. 2015). Engine seals in use are known as  
51 “wet-seals”, an inherent design feature, whereby a thin film of oil prevents rotating surfaces to come  
52 into mechanical contact with each other. Pressure differentials over the seals cause a constant loss of oil  
53 and vapours into the core engine. These enter the inlet of the high pressure compressor and  
54 contaminate the bleed air which is taken downstream. Oil leaks can result in odd smells in the cabin and  
55 in more extreme cases smoke events. The tricresylphosphate chemical fingerprint from wipe samples  
56 taken from a cockpit showed a statistically significant correlation (p value 0.039) with used engine oil  
57 (Hourzager et al. 2013), indicating that this can be a significant source of exposure. Estimates of how  
58 often these events occur varies depending upon whether the information is sourced from regulatory  
59 authorities such as the UK Civil Aviation Authority (CAA), from airlines or from trade unions who  
60 represent aircrew (Harrison and Mackenzie Ross, 2016). Reported values for the frequency of  
61 smoke/fume events include 0.5% of flights (Murawski and Supplee, 2008), 0.05% (COT, 2007) and 0.02%  
62 (Shehadi et al., 2015). Lubricating oil is applied to gas turbines in aeroplanes as anti-wear agents.  
63 However, there have been several studies that have identified subchronic neurotoxicity of aviation oils

64 (Freudenthal et al. 1993; Daughtrey et al. 1996; Mackerer et al. 1999). Aircraft lubricating oils have  
65 remained relatively unchanged since the 1960s and are comprised of approximately 95% synthetic  
66 esters with 3% tri-cresyl phosphates and 1% phenyl- $\alpha$ -naphthalamines (Winder and Balouet, 2002). Tri-  
67 cresyl phosphates (TCPs) are a group of organophosphates which contain 10 structural isomers. The  
68 ortho substituted congeners are considered to be the most toxic and therefore the proportions of these  
69 compounds in oil have been reduced in recent decades (Craig and Barth. 1999). Most of the focus of  
70 investigations in aircraft air quality and aerotoxic syndrome has been focused on tri-ortho-cresyl  
71 phosphate (ooo-TCP or ToCP), however the mono-ortho and di-ortho isomers are also highly toxic with  
72 omp-TCP the most toxic (de Boer et al. 2015; De Nola et al., 2011).

73 There is evidence to indicate that aircraft crew and passengers have been exposed to organophosphates  
74 from traveling on aeroplanes and that it has resulted in neurotoxic effects (Abou-Donia et al. 2013;  
75 Liyasova et al. 2011). Abou-Donia et al. (2013) undertook a study of 34 flight crew members and the  
76 results suggest the possible development of neuronal injury and gliosis in flight crew members  
77 anecdotally exposed to cabin air emissions containing organophosphates. A symptom-free pilot was  
78 sampled before symptoms and then again afterward. This pilot developed clinical problems after flying  
79 for 45 hours in 10 days. Significant increases in autoantibodies were noted to most of the tested  
80 proteins in the serum of this pilot after exposure to air emissions. The levels of autoantibodies rose with  
81 worsening of his condition compared to the serum sample collected prior to exposure. After cessation of  
82 flying for a year, this pilot's clinical condition improved, and his serum autoantibodies against nervous  
83 system proteins decreased. Many crew members who have reported conditions consistent with  
84 organophosphate poisoning recover and return to their flight duties, however some staff have lost their  
85 jobs, and several have even passed away. When individuals are grounded they are no longer exposed to  
86 aircraft air their reported symptoms can gradually reduce. The underlying mechanism that caused the ill  
87 health may be an active auto immune reaction, set into motion by repeated low dose exposure to

88 organophosphates, causing actual damage to the Blood-Brain-Barrier and apoptosis inside the brain.  
89 Protein filaments of these decaying cells are able to re-enter the bloodstream, causing a secondary auto  
90 immune response. Memory B cells can recognise certain antigens for a long duration of time,  
91 comparable to a status after vaccination and can lead in rare instances to an anaphylactic shock (Banks  
92 et al. 2012), also known as “incapacitation”, or sudden heart failure due to lymphocytic myocarditis. UK  
93 Coroner Payne, recently issued a report to prevent further deaths following the death of a pilot, which  
94 was linked to OP poisoning (Abou-Donia et al., 2014; Payne, 2015). The post mortem investigations gave  
95 the cause of death of either pentobarbital toxicity or lymphocytic myocarditis, individually or in  
96 combination. Hair analysis had shown the use of pentobarbital during the 4 weeks prior to his death, as  
97 a form of self medication against severe headaches. Pentobarbital is not known for causing the  
98 widespread infiltration of T-lymphocytes in the brain, heart and neurological damage otherwise  
99 observed. Testing of samples taken both prior to and after death disclosed symptoms consistent with  
100 exposure to organophosphate compounds.

101 Due to the sporadic occurrence of smoke/fume incidents it is particularly difficult to obtain worst case  
102 air samples for analysis. No fume events were observed on any of the flights that were monitored by  
103 Crump et al., (2011) and de Ree et al., (2014). Although this is hardly surprising as they only involved  
104 analysis on 100 and 20 flights respectively. Using the estimates of the frequency of fume events from  
105 Murawski and Supplee, (2008), and Shehadi et al., (2015) a sample size of between 200 and 5000 flights  
106 would be required to identify one event. Interestingly nine smoke/odour events were recorded in 78  
107 samples taken by De Nola (2011). de Boer et al. (2015) used data from De Nola (2011) and Craig and  
108 Barth (1999) to calculate that even using worst-case scenarios they cannot explain a relation of TCP in  
109 flight deck air to the complaints of pilots and air crew, a similar conclusion was also reached by de Ree et  
110 al. (2014) and Schindler et al. (2013). Available data was reviewed by Ramsden (2013) who used jet oil  
111 consumption as a surrogate to measure chemical contamination in aircraft cabin air. Those results show

112 that the oil concentration in a fume event, in which visible smoke appears in the cabin, was estimated at  
113 50 mg/m<sup>3</sup>. The concentration of TCP was approximately 1.5 mg/m<sup>3</sup>.and using the ratio of ToCP to total  
114 TCP from Crump et al. (2011) resulted in a ToCP concentration of 0.5 mg/m<sup>3</sup>, which exceeded the short-  
115 term workplace exposure limit (15 minute reference period). The variability in these studies highlights  
116 the uncertainty in recording and reporting methodologies but also suggests that other possible  
117 explanations for the reported symptoms must be considered as the effects may not be due to TCP or  
118 ooo-TCP alone. The temperatures of jet engines during operation can vary, the oil may be heated to  
119 several hundred °C (Ramsden, 2013), although some parts of the engine (e.g., the combustion chamber)  
120 can get much hotter and exceed temperatures of 400°C. These temperatures have the potential to alter  
121 the composition of the original oil and create other toxic compounds such as trimethylolpropane  
122 phosphate (Winder and Balouet, 2002). Pyrolysis of the oil also has the potential to pose a health risk  
123 due to the generation of toxic asphyxiants such as carbon monoxide and hydrogen cyanide (Winder and  
124 Balouet, 2002). There is currently a large degree of uncertainty as to what compounds are produced and  
125 how toxic they are through inhalation in the vapour phase at high altitudes (de Boer et al. 2015).

126 In this study samples of fresh and used aircraft oil were analysed by Fourier Transform Ion Cyclotron  
127 Resonance Mass Spectrometry (FTICRMS) and comprehensive two dimensional gas chromatography  
128 with high resolution mass spectrometry (GCxGC-HRTOFMS), to characterise the composition of the oil  
129 and identify potentially toxic products that may be generated during use. Particular focus was given to  
130 organophosphates as these are the group of compounds most closely linked to aerotoxic syndrome.

131

## 132 2. Methodology

133 Three 10 mL samples of fresh and used Mobil II Jet oil were obtained from a maintenance facility of  
134 Falcon business jets in Europe. The samples were prepared for analysis by FTICRMS by diluting the oil by

135 a factor of 1:1000 with toluene. For analysis by GCxGC- HRTOFMS the FTICRMS samples were further  
136 diluted by a factor of 1:10 and 1:100. These samples were spiked with  $^{13}\text{C}_{18}$  labeled triphenyl phosphate  
137 (TPhP) for quantification (obtained from Wellington Laboratories).

138 FTICRMS analysis was performed using a Varian FTICRMS (Varian Inc., Walnut Creek, CA), consisting of a  
139 9.4 T superconducting magnet. Samples were directly infused at a rate of 5  $\mu\text{L}/\text{min}$  and ionised using  
140 atmospheric pressure photo ionisation (APPI). Mass spectra were obtained using arbitrary waveform  
141 excitation (90 v) and broadband detection from  $m/z$  150 to 1000 with a transient length of 2 seconds.  
142 The FTICRMS was operated at a resolving power of 300,000 (fwhm) at 368  $m/z$ . Mass accuracy was less  
143 than one ppm, achieved by internal mass calibration using the Agilent ESI calibration mix, which was  
144 added to the sample at a 1:1 ratio before infusion.

145 GCxGC-HRTOFMS analysis was performed using a Waters Xevo G2-XS QToF fitted with a 40 m x 0.18 mm  
146 x 0.18  $\mu\text{m}$  DB-5 HT GC column in the first dimension ( $^1\text{D}$ ) and 0.8 m x 0.15 mm x 0.15  $\mu\text{m}$  Rxi17 in the  
147 second dimension ( $^2\text{D}$ ), this was then connected to a 0.8 m x 1.8 mm Custom MXT tubing (sulfinert  
148 treated). The injector temperature was set at 280  $^{\circ}\text{C}$ , the initial oven temperature was held at 70  $^{\circ}\text{C}$  for  
149 three minutes then ramped at 10 $^{\circ}\text{C}$  a minute to 220  $^{\circ}\text{C}$ , then 2.5  $^{\circ}\text{C}$  a minute to 300  $^{\circ}\text{C}$  and held for 5  
150 minutes. The secondary oven was set at a 40  $^{\circ}\text{C}$  offset to the primary oven. The modulation period was  
151 set at 4 seconds with a hot pulse duration of 0.4 s and the transfer line temperature at 360  $^{\circ}\text{C}$ . The  
152 corona voltage was set at 5  $\mu\text{A}$ , the cone gas at a flow rate of 175 L/hr and auxiliary gas flow set at 100  
153 L/hr. Ionisation was undertaken using an atmospheric pressure chemical ionisation source at 150  $^{\circ}\text{C}$   
154 with the detector run in TOF mode using a scan window of 50 amu to 1200 amu with a scan time of 0.04  
155 (+ 0.015 interscan delay) seconds. The GCxGC-HRTOFMS was operated at a resolving power of >20,000  
156 (fwhm), internal mass calibration was performed by using a lock mass ion (355.0699) generated from a  
157 siloxane. Limits of detection for tri-cresyl phosphates were calculated by serial dilution of the calibration  
158 solutions, the lowest concentration detected with a S:N >10 was 0.0005%)

159

### 160 3. Results and discussion

#### 161 3.1 FTICRMS analysis

162 Three fresh oil and three used oil samples were analysed using the FTICRMS and compared. This was  
163 undertaken to identify the bulk compositions of the oil and identify potential differences between the  
164 fresh and used oil, based on exact mass elemental composition assignments. The data was interpreted  
165 by investigating the mass spectra and filtering the data using a Kendrick Mass defect plot, which was  
166 pioneered by Kendrick (1963) and Hughey (2001). These are created by converting the International  
167 Union of Pure and Applied Chemistry (IUPAC) mass scale ( $C = 12.000$  Da) to one in which  $CH_2 = 14.000$   
168 Da by using the following equation.

169 Equation 1. Kendrick mass = IUPAC mass  $\times$  (14/14.01565),

170 The exact mass is plotted against its mass defect (exact mass minus nominal mass). Using the Kendrick  
171 mass scale gives  $CH_2$  an exact mass of 14.0000, thus aligning series of hydrocarbons.

172 The results show that both oils were comprised of predominantly oxygen containing synthetic esters,  
173 with the  $O_6$  series being the most abundant. The data was filtered to identify potential compounds with  
174 a  $PO_4$  group. In both the used and the fresh oil TCP ( $C_{21}H_{21}PO_4$ ) could be clearly identified with a mass  
175 accuracy of less than 1 ppm (Figure 1). Three ions are displayed for TCP in the plot, these represent the  
176 molecular ion of TCP, the TCP +  $H^+$  adduct formed in the APPI source and a  $^{13}C$  TCP +  $H^+$  adduct.

177 As well as TCP, three  $PO_4$  containing compounds were consistently identified in the three used oil  
178 samples but were not identified in any of the fresh oil samples. The molecular formula for these  
179 compounds corresponded to  $C_{22}H_{23}PO_4$ ,  $C_{23}H_{25}PO_4$ , and  $C_{24}H_{27}PO_4$ , with a mass accuracy of 1 ppm  
180 indicating that these compounds are related to TCP but with the addition of a methyl group on one or all



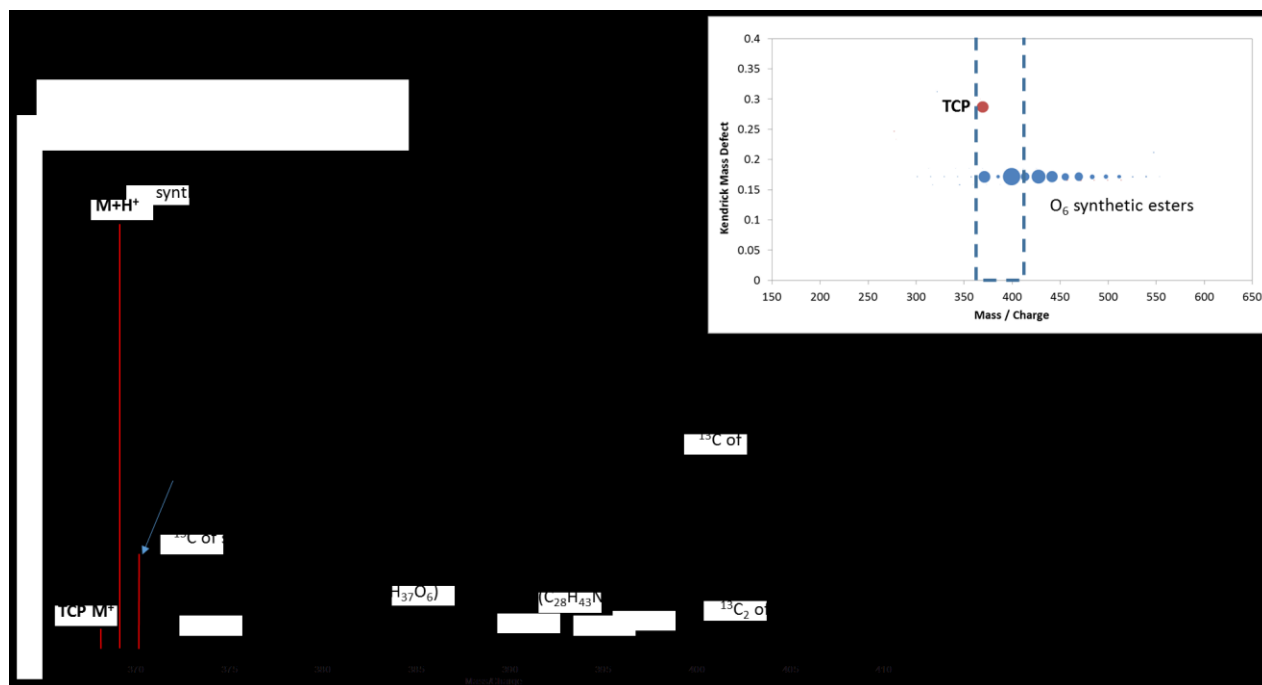
181 three of the cresyls. The structures are therefore hypothesised as monoxylenyl dicresyl phosphate,  
182 dixylenyl dicresyl phosphate and trixylenyl phosphate as xylenyl cresyl phosphates have been previously  
183 identified as potential contaminants in TCP solutions by Winder and Balouet (2002). Recently revised  
184 versions of the safety data sheets of some widely-used aviation engine oils also now report 0.1-1%  
185 trixylenyl phosphate (TXP) content (Exxon-Mobil, 2013). However, without the use of analytical  
186 standards we were unable to confirm that they are not another alkylated compound with the same  
187 molecular formula such as ethyl phenyl phosphates.

188 The discovery of alkylated cresyl phosphates in aircraft oil is a significant finding as the mono and di  
189 ortho ethyl phenyl phosphates and xylenyl phosphates have displayed a similar toxicity to ortho  
190 substituted TCP isomers (Bondey et al. 1960; Winder and Balouet 2002). Like TCP the ethyl phenyl  
191 phosphates and xylenyl phosphates have a toxicity that is position specific and so it is important to  
192 understand exactly which compounds are present. The analysis performed by FTICRMS was via direct  
193 infusion and whilst it provides an excellent mass accuracy it was not possible to identify how many  
194 different structural isomers were present. Therefore, further analysis was undertaken using GCxGC-  
195 HRTOFMS to identify the different TCP isomers and quantify the concentrations of TCP and the other  
196 alkylated cresyl phosphates detected in the used oil.

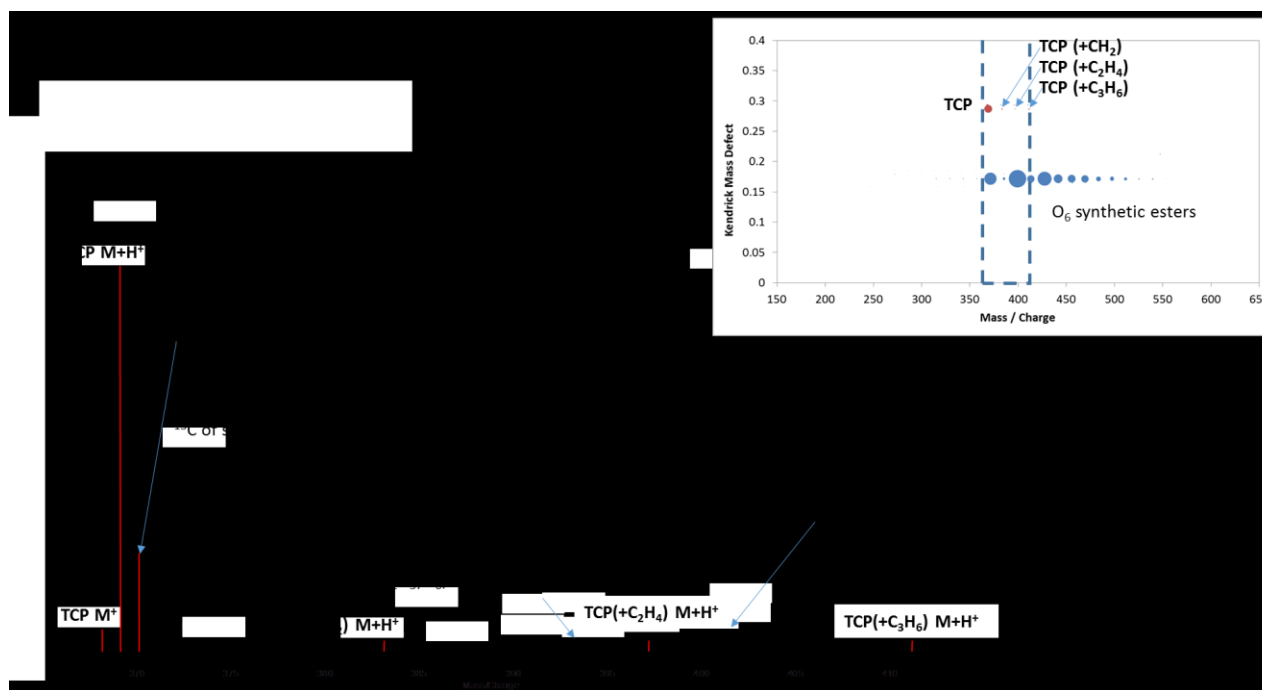
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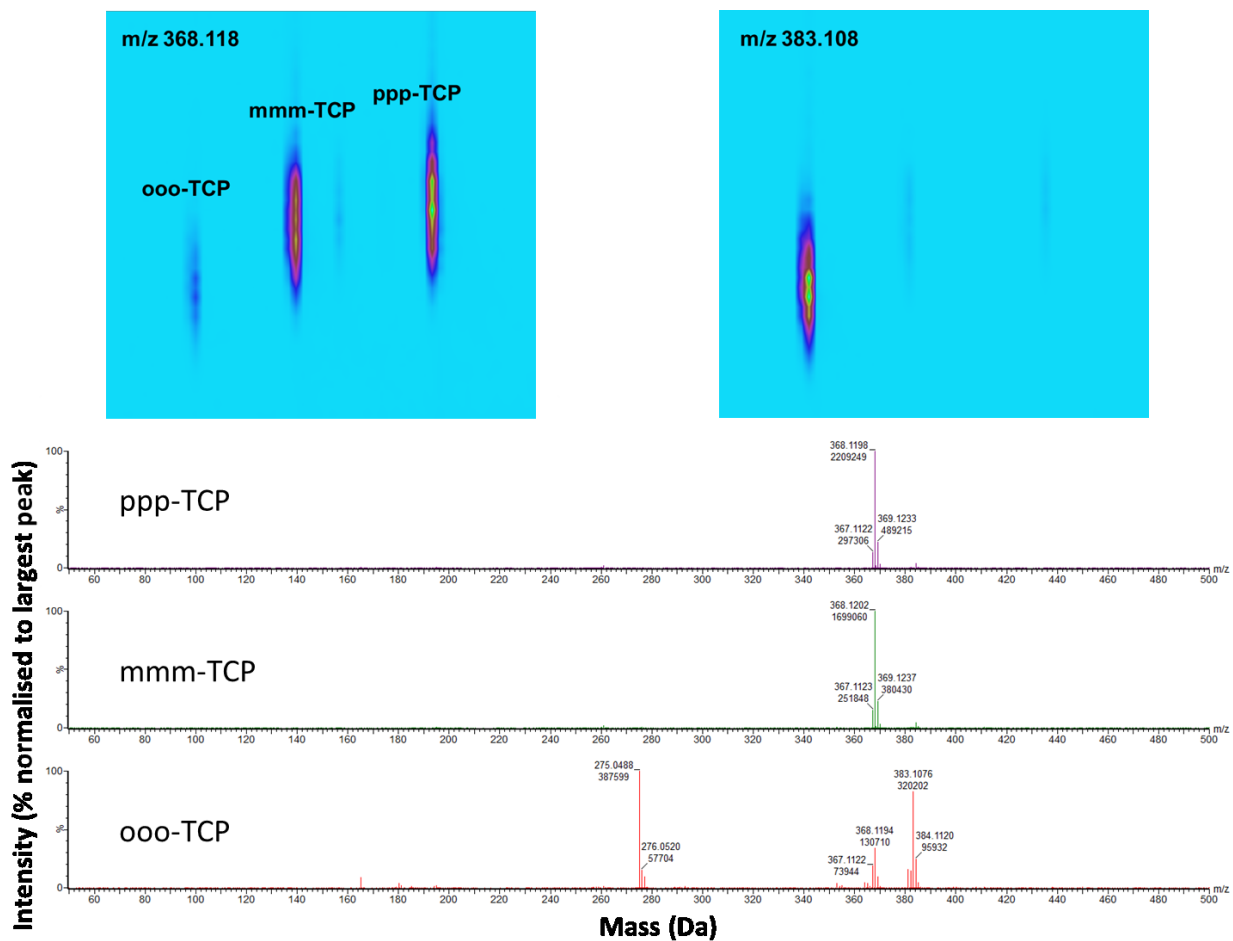
201 **Figure 1.** Kendrick mass defect plot and selected mass spectra (M/Z 365 to 413- represented by the blue  
202 box in the mass defect plot) for a fresh and used oil sample. Circles on the mass defect plot are sized to

203 reflect the intensity of each ion recorded, and the data filtered to display potential PO<sub>4</sub> (red) and O<sub>6</sub>  
204 (blue) containing ions. These ions are annotated on the corresponding mass spectra.

## 205 3.2 GCxGC-HRTOFMS analysis

### 206 3.2.1 TCP quantification

207 A calibration series was produced by using <sup>13</sup>C<sub>18</sub> labeled triphenyl phosphate (TPhP) and a native stock  
208 solution of ooo-TCP, mmm-TCP and ppp-TCP (all obtained from Wellington Laboratories). These were  
209 diluted to produce calibration solutions ranging from 0.5 to 1000 pg μL<sup>-1</sup>. A mass of 368.1177 Da was  
210 selected for quantification of TCP isomers as this was the dominant molecular ion [M<sup>++</sup>] that was  
211 expected to be generated through ionisation in the APCI source. However, when the calibration  
212 solutions were analysed the results showed generation of an [M-H+O]<sup>+</sup> ion at m/z 383.108. This ion-  
213 molecule reaction greatly favoured the ortho-substituted TCP, which along with a fragment at m/z  
214 275.049 resulted in a significant decrease in the abundance of the molecular ion and dominance of the  
215 oxygen containing ion at m/z 383.108 for ooo-TCP (Figure 2). The formation of the [M-H+O]<sup>+</sup> ion at m/z  
216 383.108 likely corresponds to the addition of oxygen from O<sub>2</sub> from residual air ion the ion source. The  
217 ion at m/z 275.049 may be generated by the loss of one of the side groups (M-C<sub>7</sub>H<sub>9</sub>) by double hydrogen  
218 transfer. These proposed ions are <3ppm of the theoretical values. The ion-molecule reaction involving  
219 O<sub>2</sub> appears to be structure specific. Although a mechanism is not yet known, similar ion molecule  
220 reactions between radical cations and O<sub>2</sub> have been observed (Jobst et al. 2009).



221  
 222 **Figure 2.** Mass spectra generated from ppp-, mmm- and ooo-TCP using APCI, and the corresponding SIMs  
 223 from the M+ ion and M+O ion (+/- 0.05 da).

224  
 225 There are 10 different structural isomers of TCP, however Figure 3 (a&e) shows that only four TCP  
 226 isomers were identified at detectable levels in the fresh and used oil; mmm-TCP, mmp-TCP, ppm-TCP  
 227 and ppp-TCP. Quantification of mmp-TCP and ppm-TCP isomers was performed using the calibration  
 228 curves produced for mmm-TCP and ppp-TCP respectively. Houtzager et al. (2013) reported that ooo-TCP  
 229 was not identified in air and wipe samples taken from an aircraft, however ooo-TCP had been identified  
 230 in aircraft by Crump et al. (2011) and Rosenberger et al. (2013). The mmm-TCP, mmp-TCP, ppm-TCP and

231 ppp-TCP isomers were present in the fresh oil samples at concentrations of 0.69 % (+/- 0.01, 1 $\sigma$ ), 1.70 %  
232 (+/- 0.27, 1 $\sigma$ ), 1.34 % (+/- 0.11, 1 $\sigma$ ) and 0.51 % (+/- 0.06, 1 $\sigma$ ) respectively, equating to approximately 4.25  
233 % total TCP (+/- 0.42, 1 $\sigma$ ), which is consistent with the manufacturer's specifications (1-3 %). This was  
234 greater than the concentrations found in the used oil which were 0.50 % (+/- 0.10, 1 $\sigma$ ), 1.14 % (+/- 0.09,  
235 1 $\sigma$ ), 0.87 % (+/- 0.10, 1 $\sigma$ ) and 0.24 % (+/- 0.02, 1 $\sigma$ ) respectively, equating to 2.75 % total TCP (+/- 0.27,  
236 1 $\sigma$ ). The results are similar to those reported by Hecker et al. (2014) where total TCP concentrations in  
237 Mobil II jet oil were 5.23%. Hecker et al. (2014) reported a slightly lower TCP concentration in BP 2380  
238 fresh oil (4.7%) compared to used oil (5.1%). However, in this study the concentration in the used oil was  
239 less than the fresh oil. This indicates that the TCP concentration in different oils can vary, and that TCP  
240 may be lost during use (potentially to bleed air) or modified and converted into other compounds.

241 The non-detection of ooo-TCP (< 0.0005%) in our study significantly contrasts with earlier investigations  
242 where the ooo-TCP represented between 10 and 60% of all TCP isomers in cabin air (Ramsden, 2013;  
243 Rosenberger et al., 2013). Whilst this study cannot discount the presence of ooo-TCP below  
244 concentrations of 0.0005% the initial results indicate that the oil is not the source of ooo-TCP in cabin  
245 air. However one potential explanation for the absence of ooo-TCP in the oil but its presence in air  
246 samples is the catalysis of meta and para isomers (by a palladium catalyst) which can generate ortho-  
247 isomers (Imbert et al, 1997). The catalyst is used in units to decompose ozone and is often located after  
248 the engine and upstream of the air conditioning pack. The authors are currently performing laboratory  
249 testing to validate this hypothesis.

250

### 251 3.2 Identification of other toxic contaminants of concern

252 The results indicate that the formulation of Mobile II jet oil does not contain the more toxic ortho  
253 substituted TCP isomers at detectable concentrations. However, there are several other toxic

254 components that may be present in jet oil. The absence of ortho containing TCP isomers does not  
 255 necessarily mean that the oil does not pose a significant risk. Table 2 contains a list of potential  
 256 compounds of concern that were screened for in the fresh and used oil. These compounds were  
 257 selected from a literature search of organophosphates and other toxic compounds found in lubricating  
 258 oils.

259 **Table 2.** Summary of screened analytes in the triplicate fresh and used oil samples

	CAS #	[M+] m/z	Formula	Concentration in oil (%)					
				Fresh oil 1	Fresh oil 2	Fresh oil 3	Used oil 1	Used oil 2	Used oil 3
ooo-TCP	1330-78-5	368.118	C <sub>21</sub> H <sub>21</sub> PO <sub>4</sub>						
oom-TCP		368.118	C <sub>21</sub> H <sub>21</sub> PO <sub>4</sub>						
oop-TCP		368.118	C <sub>21</sub> H <sub>21</sub> PO <sub>4</sub>						
omm-TCP		368.118	C <sub>21</sub> H <sub>21</sub> PO <sub>4</sub>						
omp-TCP		368.118	C <sub>21</sub> H <sub>21</sub> PO <sub>4</sub>						
mmm-TCP		368.118	C <sub>21</sub> H <sub>21</sub> PO <sub>4</sub>	0.68	0.70	0.70	0.40	0.52	0.59
opp-TCP		368.118	C <sub>21</sub> H <sub>21</sub> PO <sub>4</sub>						
mmp-TCP		368.118	C <sub>21</sub> H <sub>21</sub> PO <sub>4</sub>	1.51	2.01	1.58	1.05	1.16	1.22
mpp-TCP		368.118	C <sub>21</sub> H <sub>21</sub> PO <sub>4</sub>	1.21	1.42	1.39	0.78	0.97	0.87
ppp-TCP		368.118	C <sub>21</sub> H <sub>21</sub> PO <sub>4</sub>	0.45	0.55	0.53	0.22	0.26	0.24
xylenyl dicresyl phosphate	Not identified	382.133	C <sub>22</sub> H <sub>23</sub> PO <sub>4</sub>	0.001 – 0.004	0.001 – 0.004	0.001 – 0.004	0.04 – 0.21x	0.06 – 0.30x	0.05 – 0.28x
dixylenyl monocresyl phosphate	Not identified	396.149	C <sub>23</sub> H <sub>25</sub> PO <sub>4</sub>				0.02 – 0.13	0.03 – 0.17	0.03 – 0.15
trixylenyl phosphate	121-06-2	410.165	C <sub>24</sub> H <sub>27</sub> PO <sub>4</sub>	0.009 – 0.003	0.009 – 0.003	0.009 – 0.003	0.04 – 0.23	0.06 – 0.31	0.05 – 0.29
dibutylphenyl phenyl phosphate	65652-41-7	438.196	C <sub>26</sub> H <sub>31</sub> PO <sub>4</sub>						
tributylphenyl phosphate	78-33-1	494.259	C <sub>30</sub> H <sub>39</sub> PO <sub>4</sub>						
tributyl phosphate	126-73-8	266.165	C <sub>12</sub> H <sub>27</sub> PO <sub>4</sub>						
trimethyl phosphate	512-56-1	140.024	C <sub>3</sub> H <sub>9</sub> PO <sub>4</sub>						
cresyl diphenyl phosphate	26444-49-5	340.086	C <sub>19</sub> H <sub>17</sub> PO <sub>4</sub>						

cresyl saligenin phosphate	1222-87-3	276.055	C <sub>14</sub> H <sub>13</sub> PO <sub>4</sub>						
triethylphosphate	78-40-0	182.071	C <sub>6</sub> H <sub>15</sub> PO <sub>4</sub>						
Trimethylopropane phosphate	1005-93-2	213.053	C <sub>6</sub> H <sub>13</sub> PO <sub>6</sub>						
tetraethyl pyrophosphate	107-49-3	290.068	C <sub>8</sub> H <sub>20</sub> P <sub>2</sub> O <sub>7</sub>						
triphenyl phosphorothionate	597-82-0	342.048	C <sub>18</sub> H <sub>15</sub> PSO <sub>3</sub>						
N-phenyl-1-naphthalamine	90-30-2	219.105	C <sub>16</sub> H <sub>13</sub> N	x	x	x	x	x	x
dioctyldiphenylamine	68411-46-1	393.339	C <sub>28</sub> H <sub>43</sub> N	x	x	x	x	x	x
dinaphthylamine	532-18-3	269.120	C <sub>20</sub> H <sub>15</sub> N						
naphthylamine	134-32-7	143.074	C <sub>10</sub> H <sub>9</sub> N						
naphthol	90-15-3	144.058	C <sub>10</sub> H <sub>8</sub> O						

260 X = present in the sample with S:N greater than 10:1 but not quantified as no specific internal standard  
261 or calibration series were used

262

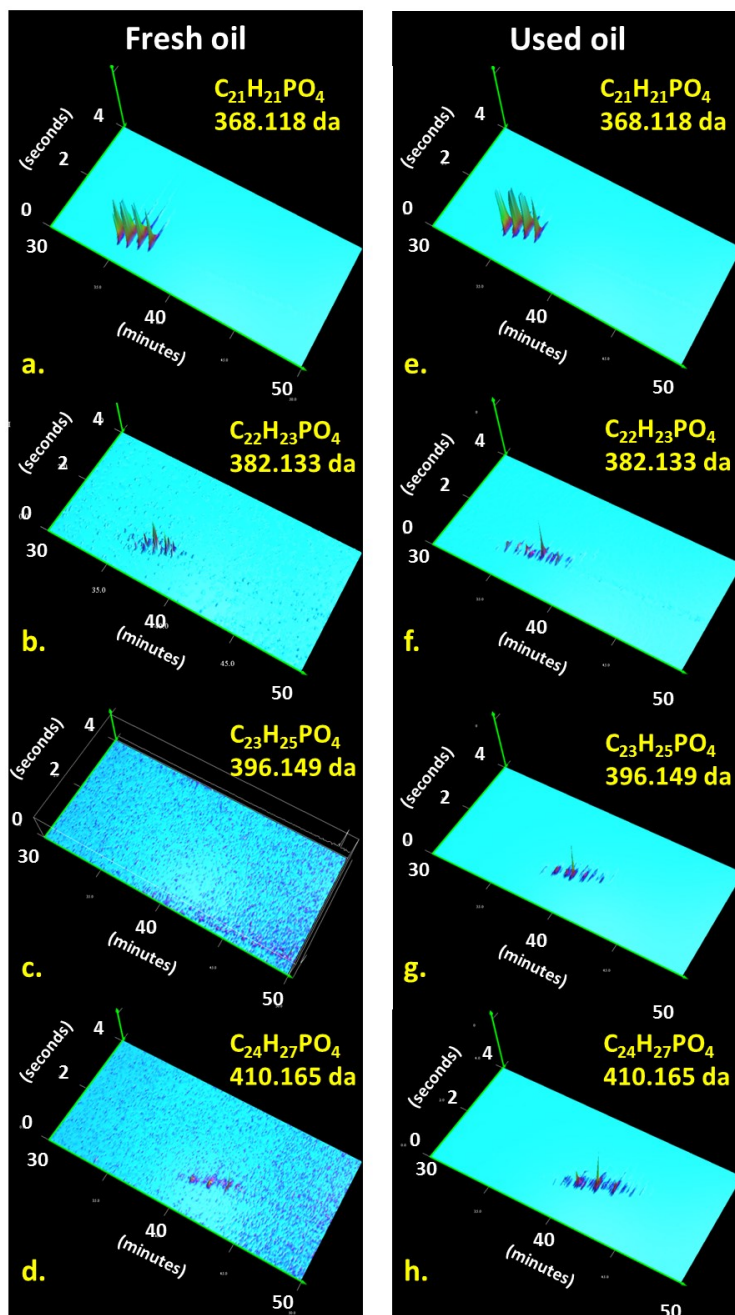
263 The GCxGC-HRTOFMS analysis confirmed the presence of the same group of three alkylated cresyl  
264 phosphates that were previously identified by FTICRMS for the used oil (Figure 1). The limits of detection  
265 by HRTOFMS were lower than the FTICR which enabled the detection of mono and tri xylenyl  
266 phosphates in the fresh oil extract in concentrations slightly greater than the limit of detection (0.0005%  
267 in the oil), however no dixylenyl phosphates were detected.

268 In all three used oil samples, 10 monoxilylenyl dicresyl phosphate isomers, 7 dixylenyl monocresyl  
269 phosphate isomers and 10 trixylenyl phosphate isomers were detected (with S:N >10). Concentrations  
270 for the sum of monoxilylenyl dicresyl phosphate isomers were calculated at between 0.05 and 0.26 %, for  
271 the sum of the dixylenyl monocresyl phosphate isomers at between 0.03 and 0.15 % and the sum of the  
272 trixylenyl phosphate isomers between 0.05 and 0.28 %. In all three fresh oil samples, 4 monoxilylenyl  
273 dicresyl phosphate isomers, 0 dixylenyl monocresyl phosphate isomers and 3 trixylenyl phosphate  
274 isomers were detected (with S:N >10). Concentrations for the sum of monoxilylenyl dicresyl phosphate

275 isomers were calculated at between 0.001 and 0.004 %, for and the sum of the trixylenyl phosphate  
276 isomers between 0.001 and 0.003 %. The range in potential concentrations is based on 'best' and 'worst'  
277 case calculations using calibration data from either ppp-TCP or ooo-TCP. Further analysis should be  
278 undertaken to confirm the structural identity of the xylenyl phosphates and other alkylated cresyl  
279 phosphates. Several standards are currently commercially available; however this task would be greatly  
280 aided by a comprehensive set of standards. Several alkylated cresyl phosphates have been shown to  
281 have a comparable toxicity to ortho substituted TCP (Winder and Balouet, 2002; Bondey et al. 1960)  
282 therefore, even though they are present in lower concentrations than the TCP isomers they may well  
283 pose a significant risk to human health. Further research should be undertaken to identify and  
284 accurately quantify the different xylenyl phosphates isomers as these results indicate that they should  
285 be included in further studies in aircraft air quality assessments. N-phenyl-1-naphthalamine and  
286 dioctyldiphenylamine were also identified in both the used and fresh oil samples, this has been  
287 previously identified in oil at approximately 1% (Winder and Balouet, 2002) and alkylated  
288 diphenylamines are noted at 1-5% on the MSDS for of Mobile jet oil II. These compounds should also be  
289 included in further studies.

290 Another potentially important source of organophosphate exposure that warrants further investigation  
291 is from flame retardants being released from fabrics, foams and plastics in the fittings and upholstery in  
292 the cabin. Schindler et al. (2012) found metabolite levels of flame retardants such as tributyl phosphate  
293 (TNBP), tris-(2-chloroethyl) phosphate (TCEP) and triphenyl phosphate (TPHP) (DBP 0.28 µg/l; BCEP 0.33  
294 µg/l; DPP 1.1 µg/l) in urine at levels significantly higher than in unexposed persons from the general  
295 population. None of the samples contained o-TCP metabolites above the limit of detection (LOD 0.5  
296 µg/l). Only one sample contained metabolites of m- and p-tricresyl phosphates with levels near the LOD.  
297 When assessing the risks in cabin air it is clear that assessments should not just consider ooo-TCP but  
298 investigate other compounds that may be present in oil and consider other pollutant pathways.





300

301 **Figure 3.** Selected ion chromatograms for the fresh and used oil samples displaying tricresyl phosphate  
 302 isomers (a. & e.), monoxylenyl dicresyl phosphate isomers (b. & f.), dixylenyl monocresyl phosphate  
 303 isomers (c. & g.) and xylenyl phosphates (d. & h.)

304

## 305 4. Conclusions

306 Flying is an important form of transportation and, for some, a rewarding past time. Although, aircraft  
307 crew are exposed to greater levels of cosmic radiation, VOCs and ozone, it is exposure to  
308 organophosphates that has been most closely linked to aerotoxic syndrome. The majority of studies on  
309 aerotoxic syndrome have focused on TCP and specifically tri-ortho-cresyl phosphate (ooo-TCP). This  
310 paper presents the findings of a wider screening method performed by FTICR MS and GCxGC-HRTOFMS  
311 to assess the presence of other organophosphates in fresh and used engine oil.

312 The results show that the formulation of Mobile II jet oil does not contain the more toxic ortho  
313 substituted TCP isomers at detectable concentrations. However, there may still be a significant risk from  
314 alkylated cresyl phosphates (xylenyl or ethylphenyl phosphates) which were identified in the used oils at  
315 concentrations calculated in the range of 0.13 to 0.69%. Several xylenyl and ethylphenyl phosphates  
316 have been shown to exhibit a similar toxicity to ortho substituted TCP isomers which makes there  
317 discovery in used oil significant. These compounds have not been analysed or accounted for in many of  
318 the previous exposure and air quality studies which may therefore have underestimated the actual risks  
319 from organophosphates.

320 More research is needed to further understand the problem of aerotoxic syndrome and establish if  
321 protective measures are necessary to ensure the health of future flight crews and passengers. These  
322 studies should include not only targeted analysis of suspected contaminants of concern such as tri-  
323 ortho-cresyl phosphate and N-phenyl-1-naphthalamine but also include non-targeted screening for  
324 other potential contaminants such as xylenyl phosphates generated during oil use. Future research  
325 should also include more detailed sampling of different matrices such as used oil, bleed air vapour, cabin  
326 air, fabric and of air crew. We can only fully understand the risks from aircraft oil when we understand,

327 a) what toxic compounds are in the oil, b) what is in the air, and c) what crew members have been  
328 exposed to. This paper indicates that the oil is not as safe as previously thought and so further research  
329 should be undertaken to characterise what is in the air and to measure what those adversely effected  
330 have been exposed to.

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