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Linking ozone pollution and climate change: The case for controlling methane

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[1] Methane (CH_4) emission controls are found to be a powerful lever for reducing both global warming and air pollution via decreases in background tropospheric ozone (O_3) . Reducing anthropogenic CH₄ emissions by 50% nearly halves the incidence of U.S. high-O₃ events and lowers global radiative forcing by 0.37 W m⁻² (0.30 W m⁻² from CH_4 , 0.07 W m⁻² from O_3) in a 3-D model of tropospheric chemistry. A 2030 simulation based upon IPCC A1 emissions projections shows a longer and more intense U.S. O₃ pollution season despite domestic emission reductions, indicating that intercontinental transport and a rising O₃ background should be considered when setting air quality goals. INDEX TERMS: 0325 Atmospheric Composition and Structure: Evolution of the atmosphere; 0345 Atmospheric Composition and Structure: Pollution—urban and regional (0305); 0368 Atmospheric Composition and Structure: Troposphereconstituent transport and chemistry. Citation: Fiore, A. M., D. J. Jacob, B. D. Field, D. G. Streets, S. D. Fernandes, and C. Jang, Linking ozone pollution and climate change: The case for controlling methane, Geophys. Res. Lett., 29(19), 1919, doi:10.1029/ 2002GL015601, 2002.

1. Introduction

[2] There is growing interest in linking air quality and climate change mitigation objectives in the design of emission control strategies. Tropospheric O_3 deserves particular attention as both the primary constituent of smog [*National Research Council (NRC)*, 1991] and a significant greenhouse gas [*Prather et al.*, 2001]. Ozone is produced in the troposphere by photochemical oxidation of volatile organic compounds (VOC) and carbon monoxide (CO) in the presence of nitrogen oxides (NO_x). While O_3 production on global and regional scales is sensitive to NO_x emissions from fossil fuel combustion [*NRC*, 1991; *Wang and Jacob*, 1998], reducing these emissions may increase greenhouse warming if the positive forcing from increased CH₄ concentrations offsets the negative forcing from decreased O_3 concentrations [*Wild et al.*, 2001].

[3] Methane is a known major source of the tropospheric O₃ background, but is not generally considered a precursor

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to regional O_3 pollution episodes in surface air because of its long lifetime (8–9 years). Recent recognition that intercontinental transport may contribute to these pollution episodes [*Jacob et al.*, 1999; *Yienger et al.*, 2000; *Wild and Akimoto*, 2001; *Fiore et al.*, 2002; *Li et al.*, 2002] raises the profile of CH₄. The present-day U.S. O_3 standard is based upon a 0.08 ppmv (8-hour average), not to be exceeded more than three times per year. If this standard becomes more stringent, as it is in European countries (55–65 ppbv), the relative contribution of the background component to exceedances of the standard will increase. We show that reductions in CH₄ emissions deserve consideration as a means to meet air quality standards while simultaneously lessening radiative forcing.

2. Model Description

[4] We apply GEOS-CHEM (v4.16), a 3-D global model of tropospheric O₃-NO_x-CO-VOC chemistry [*Bey et al.*, 2001], to investigate the response of U.S. pollution episodes and global O₃ and CH₄ to (1) 50% reductions in various anthropogenic precursor emissions relative to a 1995 base year and (2) projected 2030 emissions from the IPCC A1 and B1 scenarios [*Prather et al.*, 2001], which project relatively pessimistic and optimistic futures, respectively. All simulations were spun up for 6 months, long enough to remove the effects of initial O₃ concentrations on the results. Our simulations use assimilated observations of meteorological fields from NASA GEOS-1 with 20 vertical sigma layers and a 4° × 5° horizontal resolution; comparison with 2° × 2.5° resolution shows no significant bias [*Fiore et al.*, 2002].

[5] The coarse resolution precludes the model from capturing the local O_3 maxima that determine compliance with the national O_3 standard. These maxima, however, typically occur under regionally stagnant conditions that are conducive to the formation of elevated O_3 levels spanning large spatial scales (>600,000 km²) [*Logan*, 1989] resolved by the model. During the summer of 1995, these stagnation episodes occurred more frequently than would be predicted by climatological averages [*McNider et al.*, 1998]. We have previously shown that GEOS-CHEM captures these regional high- O_3 events, as well as the frequency distribution of O_3 at U.S. sites [*Fiore et al.*, 2002]. We compare here simulated June–August daily afternoon (1–5 p.m. local time) mean O_3

	Global				United States			
Species	Emissions 1995		Growth Factor ^b A1 2030		Emissions 1995		Growth Factor A1 2030	
	А	Ν	А	Ν	А	Ν	А	Ν
NO _x	27	15	1.8	1.0	6.6	0.6	0.8	1.0
CO	580	630	1.2	1.2	86	17	0.6	1.1
VOC ^c	58	450	1.6	1.0	12	33	0.7	1.0
CH4d	340	230	1.3 ^e		_	_	_	_

Table 1. Annual Emissions and Growth Factors in GEOS-CHEM^a

^a A = anthropogenic; N = natural (includes biomass burning). Emissions are in units of Tg N yr⁻¹ for NO_x, Tg CO for CO, Tg C for VOC and Tg CH_4 for CH_4 .

^bRatio of 2030 to 1995 emissions. Natural emissions remain constant except biomass burning in 2030 A1.

^cNonmethane volatile organic compounds; natural emissions include biogenic isoprene (404 Tg C), propene (12 Tg C), acetone (15 Tg C), biomass burning.

^d The contribution from U.S. emissions is not resolved because a uniform mixing ratio is applied (see text).

^eCalculated as described in the text.

concentrations with surface measurements at EPA AIRS sites averaged onto the GEOS-CHEM grid in the eastern U.S. (east of 97.5°W where the network is most dense). The correlation coefficient (r) between observed and simulated daily afternoon concentrations is 0.68. The percentages of total grid-square afternoons (92 summer afternoons times 27 grid-squares) in excess of 80 ppbv, 70 ppbv, and 60 ppbv, are 5%, 15%, and 36%, respectively, in the observations; the corresponding percentages in the model are 1%, 10%, and 37%, respectively. In this study, we adopt a threshold of 70 ppbv as a metric for polluted conditions since the model provides good statistics for the occurrence of events above this threshold.

[6] The GEOS-CHEM emissions are described in detail by *Bey et al.* [2001] and *Martin et al.* [2002]. We classify biomass burning emissions as natural, following *Fiore et al.* [2002]. The anthropogenic contributions to total global emissions for NO_x, CO, CH₄ and NMVOC in the base case 1995 simulation (Table 1) are 65%, 50%, 60% and 12%, respectively. The CH₄ concentration for 1995 is fixed at its uniform observed value of 1700 ppbv. This value implies a global emission of 570 Tg CH₄ yr⁻¹, computed from mass balance with the simulated global sink (oxidation by OH), plus a 4.9 ppbv yr⁻¹ tropospheric accumulation. Following *Prather et al.* [2001], we view 60% of CH₄ emissions as anthropogenic. We use the relationship between CH_4 emissions and concentrations (assuming an OH feedback factor of 1.6) determined by *Prather* [1996] to translate a 50% decrease in anthropogenic CH_4 emissions into a steady-state concentration of 1000 ppbv for the corresponding troposphere.

[7] Emissions for the A1 and B1 2030 scenarios were generated by scaling 1995 emissions by regional growth factors derived from emissions in the IMAGE socioeconomic model [IMAGE Team, 2001] for each anthropogenic source and biomass burning. The IMAGE model gives projections consistent with those of the IPCC reference models. Due to space restrictions, we focus here on the A1 scenario. Emissions increase globally (Table 1) from 1995 to 2030, but the distribution shifts. In the developed world (Europe, North America, Japan, etc.), anthropogenic NO_x emissions decline by 10%, but they increase by 130% in the developing world (South America, Africa, Asia, etc.). U.S. anthropogenic emissions of O₃ precursors decline by 20-40%. IMAGE projects a 43% increase in CH₄ emissions from 1995 to 2030, which we apply to the CH₄ concentration assuming constant OH. In the 2030 A1 simulation, CH₄ actually increases by 31% (as implied by mass balance) because OH concentrations decrease. Thus our CH₄ increase is conservative. The IPCC A1 scenario does not include aircraft NO_x emissions; we use projections from the Environmental Defense Fund [Henderson et al., 1999] to determine a growth factor of 2.26.

3. Impacts on Global Chemistry and Climate

[8] Table 2 summarizes the important results from our simulations. We find that the CH₄ lifetime is longer when NO_x emissions are decreased and shorter when VOC or CO emissions are decreased, consistent with our understanding of the effect of these emissions on global OH [*Wang and Jacob*, 1998]. The 2030 A1 simulation yields a CH₄ lifetime that is 10% longer than in 1995. The global O₃ production efficiency (OPE), the number of tropospheric O₃ molecules produced per molecule of NO_x emitted [*Liu et al.*, 1987], increases with decreasing NO_x emissions and decreases with increasing VOC or CO concentrations (Table 2), again as expected [*Wang and Jacob*, 1998]. The OPE decreases in the A1 simulation, due to the large increase in global fossil fuel NO_x emissions.

[9] We find that 50% reductions in anthropogenic CH_4 emissions have more influence on the tropospheric O_3

Selected Diagnostics Base Case 1995 50% CH 50% NO 50% VOC 50% NO_x & VOC 50% CO 50% All A1 2030 294 300 394 Tropospheric O₃ burden (Tg) 321 317 297 317 269 global CH₄ conc.^b (ppbv) 1700 1000 1867 1685 1846 1643 1040 2431 **OPE**^c 33 30 43 32 42 32 38 27 CH₄ lifetime^d 8.5 9.4 8.3 9.3 7.2 8.5 9.3 7.5 U.S. summer afternoon surface O3 (ppbv) Mean 51 48 42 50 41 50 38 55 23 21 23 23 18 29 Background 21 21

Table 2. Impacts of Perturbations to Anthropogenic Emissions^a

^aEmissions from biomass burning are considered natural [Fiore et al., 2002].

^bConcentrations are specified for the base case, A1, and 50% CH_4 simulations. For the other simulations, concentrations are derived from the model computed OH concentrations by assuming the same CH_4 source as in the 1995 base case simulation.

^cOzone Production Efficiency (OPE) is defined as the ratio of O₃ (actually odd oxygen) production to total NO_x emissions, following the procedure used by *Wang and Jacob* [1998].

^d Total atmospheric burden of CH₄ (4760 Tg in GEOS-CHEM) divided by tropospheric loss of CH₄ by reaction with OH.

^e Background is defined as O₃ produced outside of the North American boundary layer [Fiore et al., 2002] (see text).



Figure 1. Response of radiative forcing and U.S. air quality to (1) 50% global decreases in anthropogenic emissions relative to 1995, (2) IPCC A1 and B1 emissions projections for 2030. (upper panel) Global radiative forcing relative to 1995 as calculated from changes in CH₄ and O₃ concentrations. (lower panel) Number of model grid-square days in the U.S. in summer with mean afternoon (1–5 p.m.) surface O₃ concentrations in excess of 70 ppbv.

burden (Table 2) than 50% reductions in anthropogenic NO_x emissions. This might appear to be inconsistent with the O_3 source inferred from scaling global NO_x emissions by the OPEs in Table 2. But, anthropogenic NO_x emissions have low OPEs due to chemical nonlinearity and titration by NO_x [*Liang et al.*, 1998; *Kasibhatla et al.*, 1998]. Thus, anthropogenic NO_x emissions are less effective than natural (in particular from lightning) in contributing to the global O_3 budget, whereas the homogeneity of CH₄ permits anthropogenic and natural CH₄ emissions to be equally effective.

[10] Climatic implications of an atmospheric perturbation can be assessed using the standard concept of radiative forcing [*Ramaswamy et al.*, 2001] which describes the instantaneous global change in the radiative balance of the Earth system resulting from the perturbation. We calculate the change in radiative forcing from CH₄ directly from the global change in concentration [*Ramaswamy et al.*, 2001], while for O₃ we use a relationship of 0.034 W m⁻² per Dobson Unit change in the mean tropospheric column [*Mickley et al.*, 1999; *Ramaswamy et al.*, 2001]. Results are shown in the top panel of Figure 1. A 50% reduction in anthropogenic CH₄ emissions yields the largest decrease in radiative forcing (-0.37 Wm^{-2} : -0.30 Wm^{-2} from CH₄ and -0.07 Wm^{-2} from O₃). The radiative effect of 50% reductions in anthropogenic NO_x emissions is neutral, as the positive forcing due to decreased O₃ is balanced by the negative forcing from increased CH₄ [*Fuglestvedt et al.*, 1999]. Both 2030 simulations show positive forcing from 1995 to 2030 (A1: 0.44 W m⁻², B1: 0.19 W m⁻²).

4. Impacts on U.S. O₃ Pollution

[11] We next examine how summertime (June-August) surface air quality over the U.S. responds to the perturbed emissions, focusing on afternoons (1-5 p.m. local time) when O₃ concentrations typically peak. We define background O₃ as that produced outside of the North American boundary layer (surface to 700 hPa). It is diagnosed by applying production and loss rates archived from the full chemical simulation to drive an off-line simulation where O_3 is divided into individual tagged tracers which are produced in different regions of the atmosphere [Fiore et al., 2002]. Table 2 shows mean and background O₃ concentrations simulated on summer afternoons in surface air over the U.S. Summer afternoon O₃ concentrations respond most strongly to the 50% reductions in anthropogenic NO_x emissions, but reductions in CH₄ emissions also have a large effect. The decreases in NO_x and CH₄ emissions, however, are equally effective at lowering background O₃ concentrations (2 ppbv in the mean). Decreases in CO or VOC emissions have little impact. For the A1 2030 simulation, mean afternoon surface O₃ increases by 4 ppbv while background concentrations rise by 6 ppbv. Even in the more optimistic B1 scenario, higher global CH₄ emissions contribute to elevating U.S. background O₃ levels by 2 ppbv, partially offsetting air quality gains achieved via domestic emissions controls.

[12] We use a threshold of 70 ppbv as a metric to gauge changes in the frequency of O_3 pollution events in our simulations. Figure 1 (bottom panel) shows the number of U.S. grid-square days in June–August 1995 where simulated afternoon average (1–5 p.m. local time) O_3 levels exceed 70 ppbv. Reducing anthropogenic NO_x emissions by 50% nearly eliminates the occurrence of grid-square days in excess of 70 ppbv (Figure 1). Because of subgrid variability in O_3 , this result does not mean that local exceedances of 70 ppbv would be as drastically reduced, but it does point to significant improvement in air quality. When anthropogenic CH₄ emissions are reduced by 50%, the incidence of O_3 concentrations in excess of 70 ppbv (Figure 1) declines by 45% (for an 80 ppbv threshold that statistic is 54%).

[13] The U.S. has aggressive emission controls to abate future O_3 pollution. Although fossil fuel emissions in the U.S. for NO_x, CO, and VOC decline in the IPCC A1 2030 scenario relative to 1995 by 27%, 45%, and 30%, respectively, we find that the number of grid-square summer days over the U.S. with $O_3 > 70$ ppbv increases relative to 1995. Efforts to improve U.S. air quality are thus thwarted by the rise in global background O_3 levels due to increased emissions outside U.S. borders. These results are consistent with the modeling study of *Collins et al.* [2000] who found



Figure 2. Number of U.S. model grid-square days per month with afternoon (1-5 p.m.) O₃ concentrations in surface air above a 70 ppbv threshold for the 1995 base case (white bars) and 2030 A1 (black bars) simulations. The A1 simulation reveals a longer O₃ pollution season. Results are similar for an 80 ppbv O₃ threshold.

that European efforts to improve air quality via domestic emissions reductions may be offset by a rise in emissions from developing nations by 2015. These simulations underscore the need to consider regional air quality in a global context.

[14] Another adverse impact of rising global emissions on U.S. air quality is a longer U.S. O_3 pollution season, as diagnosed by exceedances of a 70 ppbv threshold. Intercontinental transport makes a larger contribution to U.S. O_3 pollution in spring and fall than in summer because of the longer O_3 lifetime [*Jacob et al.*, 1999]. Figure 2 shows the larger numbers of exceedances of the 70 ppbv threshold in March through November in the A1 2030 simulation; results are similar for an 80 ppbv threshold. Relatively small changes in mean U.S. O_3 concentrations resulting from a higher global O_3 background are thus sufficient to extend the U.S. O_3 pollution season into April–May.

5. Conclusions

[15] Our global 3-D model analysis shows that reducing CH₄ emissions enables a simultaneous pursuit of O₃ air quality and climate change mitigation objectives. Whereas reductions in NO_x emissions achieve localized decreases in surface O₃ concentrations, reductions in CH₄ emissions lower the global O₃ background and improve surface air quality everywhere. Simulation of a 2030 (IPCC A1) scenario where anthropogenic U.S. emissions of O₃ precursors decrease but global emissions (including CH₄) increase indicates a greater incidence of O₃ pollution episodes and a longer U.S. O₃ season, stressing the need for a global perspective in the design of future regional pollution control strategies.

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