Measurement of K X-ray fluorescence cross section of rare-earth element present in compounds

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Abstract : A new simple and direct method proposed by Jahagiradar *et al* to determine photoelectric cross section of elements at 123.6 keV by measuring K X-ray intensity has been used here to obtain the K X-ray fluorescence cross section. The K X-ray fluorescence cross section is determined by measuring the intensity of K X-ray radiations emitted by the rare-earth elements present in La₂O₃, Nd₂O₃, Sm₂(CO₃)₃.8H₂O, Eu₂(CO₃)₃, Gd₂(CO₃)₃.8H₂O, Tb₂(CO₃)₃ and Tb₂O₃ compounds at 123.6 keV by employing a NaI (T1) spectrometer system in a 2π geometry set up. The measured K X-ray fluorescence cross sections are compared with their corresponding theoretical values and a good agreement between them is obtained

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1. Introduction

Measurement of X-ray fluorescence cross section is important in both basic and applied research. The accurate knowledge of fluorescence cross section data makes XRF suitable for quantitative elemental analysis.

Krause *et al* [1] have estimated K and L X-ray fluorescence cross section for a wide range of elements. Bhan *et al* [2] have measured the absolute values of K X-ray fluorescence cross section by using the theoretical values of photoelectric cross sections, fluorescence yield, and X-ray emission rate, employing strong source of 5-8 mci strength. Jahagirdar *et al* [3] have measured K X-ray fluorescence cross section from which the total photoelectric cross section of elements in the region $42 \le Z \le 82$ are evaluated.

In the present investigation, a simple method proposed by Jahagiradar *et al* [3] is used to obtain the K X-ray fluorescence cross section of rare-earth elements present in La₂O₃, Nd₂O₃, Sm₂(CO₃)₃.8H₂O, Eu₂(CO₃)₃, Gd₂(CO₃.)₃8H₂O, Tb₂(CO₃)₃ and Tb₂O₃

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compounds having atomic numbers in the range $57 \le Z \le 65$ by taking lead (Z = 82) as a standard element. The K X-ray fluorescence cross section can be obtained by the measurement of K X-ray intensity. The intensity of K X-ray photons produced by irradiating rare-earth compounds with 123.6 keV gamma-rays using 2" × 1" NaI(T1) gamma-ray spectrometer system attached to 1K multichannel analyser (ORTEC) in a 2π geometrical configuration, is measured.

2. Basic relation

The K X-ray fluorescence cross section $\sigma_k(i)$ of an element *i* at an excitation energy ε_i is given by the equation [2-6]

$$\sigma_k(i) = I_k(i) / (I_0 G \varepsilon \beta t), \tag{1}$$

where

 $I_k(i)$ -the observed intensity of the K X-ray line of the element,

 I_0 -the intensity of the exciting radiation,

ɛ-the efficiency of the detector at the K X-ray energy,

G-is the geometrical factor accounting for the geometrical arrangement of detector and sample set up [7],

 β -the target attenuation correction factor for the incident as well as K X-ray radiations which is given by [8.9]

$$\boldsymbol{\beta} = \left\{ 1 - \exp\left[-(\boldsymbol{\mu}_{i} + \boldsymbol{\mu}_{e})t \right] \right\} / \left\{ (\boldsymbol{\mu}_{i} + \boldsymbol{\mu}_{e})t \right\},$$
(2)

where μ_t and μ_e are the mass attenuation coefficients of incident and K X-ray radiations in the target, t is the thickness of the target.

3. Experimental arrangement

In the present investigation, a Harshaw make $2" \times 1"$ NaI(T1) crystal attached to an RCA photomultiplier tube (6342 A) coupled to an 1K MCA (ORTEC) is used to measure K X-ray intensity I_k of the target in a 2π geometrical configuration.

The experimental arrangement, involving the detector, source and the target system is shown in Figure 1 and the details of its arrangement is given by Jahagirdar *et al* [3]. Owing to the poor resolution of NaI(T1) detector, the peaks of 122 keV and 136 keV photons from ⁵⁷Co source are not separated and hence the weighted average of these two energies equal to 123.6 keV is considered. The ⁵⁷Co radioactive isotope of about 10 μ ci strength standard gamma source, is used in this experiment.

The linearity and stability of the spectrometer were checked with a precision pulse generator and the room temperature was maintained at $25 \pm 1^{\circ}$ C. The gamma ray spectrometer was calibrated using ²⁴¹Am (26, 59.6 keV), ¹⁴¹Ce (36.9, 145.4 keV) and ⁵⁷Co (123.6 keV weighted average) sources.

Measurement of K X-ray fluorescence cross section etc

The constant intensity is maintained throughout the experiment by fixing the source position and marking the orientation of the source. We have selected the live time mode because the strength of the source was so low (10 μ ci) and MCA showed no dead time losses even when it was placed on the face of the detector. In this method, we have used uniform circular pellets of 3 cm diameter of rare earth compounds prepared using a pellet making machine by applying a constant pressure of 400 kg/cm². The mass per unit area of each pellet was determined using an electrical balance and travelling microscope. The uniformity of each pellet was checked by measuring the transmitted intensity of the beam through different regions of the targets.

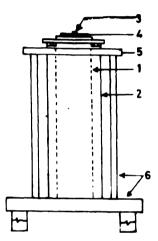


Figure 1. Experimental arrangement, involving the detector, source and the target system. (1) (2" × 1") Nal(TI) detector with pint, (2) Steel

cylinder, (3) Position of source, (4) Target, (5) Aluminium plate and (6) Iron stand.

First, we scanned the background counts due to cosmic rays and other radioactive sources. Next, the source spectrum was scanned by placing the source directly on the head of the detector. The entire area under this spectrum is the sum of background intensity and the intensity of the source. This spectrum is stored in first half (say H₁) of the memory; then the pellet of known thickness was sandwitched between the source and the detector and the spectra is recorded. The total area under the entire spectrum consists of scattered radiation. secondary photons along with transmitted intensity of the primary radiation. This spectrum was stored in second half (say H₂) of the memory. These two spectra were recorded during suitable interval of times so as to reduce the statistical error below 1%. The subtraction of first spectrum from the second spectrum in the X-ray energy region gives the pure characteristic K X-ray fluorescence spectrum. The total area under this X-ray peak gives the K X-ray fluorescence intensity I_k . This has been followed for rest of the targets of different thickness. It is observed that the intensity of K X-ray fluorescence initially increases as a function of thickness t and then decreases as the thickness of the target increases. We have taken the initial increasing portion, where the K X-ray production predominates over its attenuation for the evaluation of K X-ray fluorescence cross section σ_k .

The measured K X-ray intensity has been corrected using eq. (2) for self attenuation of fluorescent radiation in the target. The necessary theoretical mass attenuation coefficients are taken from Plechaty *et al* [10].

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We have used eq. (1) to find the K X-ray fluorescence cross section σ_k from the experimentally measured I_k . It is seen from eq. (1) that the quantity (I_k/β) is proportional to t and hence a plot of I_k/β vs t should be a straight line passing through origin and having slope equal to $\sigma_k I_0 G\epsilon$. Using the least square fit value of the slope, we have determined the value of σ_k . We have plotted in Figure 2, the corrected intensity (I_k/β) vs t for all the seven rare-earth compounds including lead. The solid lines are least square fit straight lines, all are passing through origin, and experimental points life very close to these lines.

SI. No I	Matenal used Lanthanum oxide	Chemical formula La ₂ O ₃	Element Z		Measured value of σ_k in cm ² /gm	Theoretical value of σ_k in cm ² /gm [1]
			La	57	0 726 ± 0 001	0714
2	Neodymuum oxide	Nd ₂ O ₃	Nd	60	0.835 ± 0 002	0 844
3	Samarium carbonate	Sm ₂ (CO ₃) ₃ 8H ₂ O	Sm	62	0.937 ± 0 001	0 923
4	Europium carbonate	Eu ₂ (CO ₃) ₃	Eu	63	0 956 ± 0 003	0.968
5	Gadolinium carbonate	Gd ₂ (CO ₃) ₃ . 8H ₂ O	Gd	64	0.970 ± 0 003	0 994
6	Terbium carbonate	Tb ₂ (CO ₃₎₃	Тъ	65	1.085 ± 0 006	1 045
7	Terbium oxide	Tb ₂ O ₃	ть	65	1.076 ± 0.008	1.045
8	Lead	Pb	РЬ	82	1.832 ± 0 014	1 815

Table 1. Values of the measured K X-ray fluorescence cross section $\sigma_k \ln \text{cm}^2/\text{gm}$ for 123.6 keV gamma rays for rare-earth compounds

We have determined the σ_k values of rare earth element present in compounds which are given in Table 1 along with the associated error. The theoretical cross sections which are calculated using log-log interpolation formula using theoretical cross sections of Krause et al [1] are also given for comparison. The two Tb compounds, namely Tb₂O₃ and Tb₂(Co₃)₃ with different chemical environment around the Tb atom in the molecule are used with the intention of knowing the effect of chemical environment in the K X-ray yield. The measured σ_k values in both cases are little higher than the theoretical value which may be mainly due to the errors involved in the measurement. Although these higher values do not reveal any information about the chemical environment, this effect needs to be studied in detail.

4. Results and discussion

The measured values of I_k/β are plotted against t in Figure 2 for all the rare-earth elements present in compounds. The solid lines are the least square fit lines and experimental points

lie very close to the solid straight lines. It is also seen that all the straight lines are passing through origin, indicating that the β correction factor applied for the self-attenuation of K X-ray in the increasing region of its production, is taking care of the necessary correction.

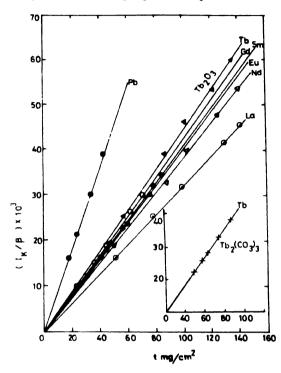


Figure 2. Variation of (I_k / β) as a function of thickness of seven rare-earth compounds including lead. To avoid overlapping of lines due to $Tb_2(CO_3)_3$ and Tb_2O_3 compounds a separate line due to $Tb_2(CO_3)_3$ is shown in the

Also it is seen that the inclination of each line is different, indicating the strong dependence of production of K X-rays on Z-value of the element. From the slopes of these straight lines the value of σ_k for all the elements have been obtained using eq. 1. The results of these σ_k values are tabulated in Table 1. In Table 1, column 2 to 5 give the details of rare-earth compounds used with the Z-values of the rare-earth elements present in each compound. Column 6 gives our measured values of σ_k along with their associated errors for elements in the range 57 $\leq Z \leq 82$ including lead. These σ_k values are the average values of the 6 independent measurements, performed on different days. Here, the σ_k value of lead being an high-Z element, is included to verify the reproducibility of the result and the accuracy of the method proposed by Jahagirdar et al [3]. The measured σ_k values are given at the end of Table 1 for completeness. The errors in each case are due to statistical error in I_k and I_o measurement, measurement of t, determination of β factor where theoretical values of μ_i and μ_e are used. The quoted error in μ_i and μ_e values [10] was found to be 2-3% in the energy range under consideration. The estimated error in β -factor was found to be less than 3% as μ_i and μ_e come under exponential term. The over all error in σ_k values in all cases was found to be less than 5%. The measured values of σ_k are compared with the corresponding theoretical values of Krause et al [1] given in column 7. It is seen that there

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is a close agreement between the measured values of σ_k with the corresponding theoretical values.

5. Conclusion

A simple, direct method proposed earlier for the determination of total photoelectric cross section by measuring K X-ray intensity has been successfully extended here to obtain the K X-ray fluorescence cross section of rare-earth elements present in their compounds having Z in the range $57 \le Z \le 82$, including lead. The σ_k value of each element is the average value of the six independent measurements made on different days. The error in each case is found to be less than 5% and a good agreement between the measured value with the theoretical value is seen, justifying the suitability of the method used. As far as our knowledge goes, the σ_k values of the above seven rare earth elements measured here are the first values at 123.6 keV. This data is very much required as it has important application in wide veriety of fields.

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