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Effects of Zn²⁺ on Ca²⁺ uptake by mitochondria and endoplasmic reticulum in permeabilized tilapia gill cells

P.M. Verbost, M.A. Salah El-Deen^{*}, P. Pelt, M.M.J.C. Bijvelds and S.E. Wendelaar Bonga Department of Animal Physiology, Faculty of Science, University of Nijmegen, Toernooiveld 1, NL-6525-ED Nijmegen, The Netherlands; *Water Research Center, Research Institute of Weed Control and Channel Maintenance, Kanater, El Khaireia, Kalubeia, Egypt

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Abbreviations: BSA, bovine serum albumin; EGTA, ethylene glycol bis-(β-aminoethyl ether)-N,N,N',N'tetraacetic acid; HEEDTA, N-(2-hydroxyethyl) ethylene diamine-N,N',N'-triacetic acid; HEPES, 4-(2hydroxyethyl)-1-piperazine ethane sulphonic acid; NTA, nitrilotriacetic acid

Abstract

An intracellular ATP-dependent Ca²⁺ pumping mechanism, distinct from mitochondrial Ca²⁺ accumulation, was identified within tilapia gill cells. Cell suspensions treated with 0.003% saponin, which selectively permeabilizes the plasma membrane, were used to characterize the Ca²⁺ sequestering mechanisms as endoplasmic reticulum and mitochondria and to determine the effect of Zn²⁺ on their Ca²⁺ storing activity. Of the Ca²⁺ taken up by the endoplasmic reticulum, 80% was released by IP₃ (10 µmol l⁻¹). The Ca²⁺ pump of the endoplasmic reticulum was 2.5 times less sensitive to Zn²⁺ (IC₅₀ = 0.05 nmol l⁻¹) than was the mitochondrial uptake mechanism (IC₅₀ = 0.20 nmol l⁻¹). The results indicate that Ca²⁺ is stored predominantly within the endoplasmic reticulum at 0.1 µmol l⁻¹ and that this storing capacity is seriously attenuated by nanomolar concentrations Zn²⁺.

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Introduction

Branchial epithelial cells are involved in both ion and gas exchange but are still similar to other cells in that they maintain resting levels of intracellular Ca^{2+} of about 100 nmol l⁻¹ (95 ± 15 nmol l⁻¹; Li *et al.* 1995). In cells in general a wide variety of actions are regulated by fluctuations in free Ca²⁺. Neurohumoral substances can induce rapid, transient increases in cytosolic Ca²⁺ either by stimulating Ca²⁺ influx *via* receptor operated Ca²⁺ channels in the cell membrane and/or by releasing Ca²⁺ from intracellular stores through inositol-1,4,5-triphosphate (IP₃) activated channels (Rasmussen *et al.* 1986; Berridge 1993).

In addition to serving a role as second messenger, Ca²⁺ is also transferred transepithelially during

branchial calcium absorption associated with the calcium homeostasis of the freshwater fish (reviewed by Fenwick 1989). It is well established that this uptake occurs via the ionocytes - mitochondria rich cells – in the gills and that the uptake occurs transcellularly (Perry and Flik 1988; McCormick et al. 1992; Marshall et al. 1995). This means that during transepithelial transport the transported calcium could affect the cytosolic Ca²⁺ concentration. The current model for active Ca2+ absorption involves passive entry across the apical membrane, diffusion through the cytosol, and ATP- and Na⁺-driven efflux across the basolateral plasma membrane (Flik et al. 1993; Verbost et al. 1994b). And little is known concerning the state of Ca²⁺ on its route through the cytosol. Active Ca²⁺ sequestration by intracellular organelles and

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buffering by calcium binding proteins occurs in rat intestinal epithelium (Wasserman and Fullmer 1982; Rubinoff and Nellans 1985; Van Corven *et al* 1987; Feher *et al* 1992) and this mechanism may be generally involved in the intracellular handling of Ca^{2+} during transit.

The current model for mammalian epithelial cells suggests that the Ca²⁺ pumps in the plasma membrane and endoplasmic reticulum (ER) are of equal importance for maintaining low intracellular Ca²⁺ (Van Os *et al.* 1988). On the other hand, the mitochondria appear to perform little if any role in maintaining low intracellular Ca²⁺ concentration (Nicholls and Akerman 1982; Van Os et al. 1988) because they do not have a sufficiently high affinity for Ca²⁺. The mitochondria do have a large capacity to store Ca²⁺, but this can result in activation of various mitochondrial enzymes essential for the regulation of ATP production (Denton and McCormack 1985). The purpose of this study was: a) to test if in fish gill cells regulate intracellular Ca²⁺ in the same way as mammalian kidney cells and enterocytes and b) to determine the effects of zinc (Zn) on the Ca²⁺ sequestration by the mitochondrial and nonmitochondrial Ca²⁺ stores. There are two reasons for studying Zn²⁺. One is to learn about the mechanism of toxicity of this environmental pollutant and, more important for this study, Zn²⁺ can be used as a tool for studying intracellular Ca²⁺ as well as Ca²⁺ transport. Waterborne Zn²⁺ inhibits the branchial uptake of Ca²⁺ (Spry and Wood 1985; Hogstrand et al. 1994, 1995a). There is evidence to suggest that the inhibition by Zn^{2+} is caused by the inhibition of basolateral Ca²⁺ transport mechanism after entrance of Zn²⁺ into the cytosol. Concentrations of $Zn^{2+} > 100$ pmol l⁻¹ reduced the ATP-dependent Ca²⁺ transport in the basolateral membranes (Hogstrand et al. 1995b), the putative driving force uptake of Ca²⁺ (Flik *et al.* 1985a,b). However, nothing is known about the effects of Zn²⁺ on the sequestration of Ca^{2+} by intracellular stores. We set out to determine these effects with the idea that a disturbance of the intracellular Ca²⁺ homeostasis could also partly explain the reduction in Ca^{2+} uptake by Zn^{2+} .

Materials and methods

Holding conditions of fish

Male freshwater tilapia from laboratory stock, weighing around 150 g, were held in 100 l tanks. The aquaria were supplied with running tap water (0.7 mmol l⁻¹ Ca, 25°C) under a photoperiod of 12 h of light alternating with 12 h of darkness. Animals were fed Trouvit[®] fish pellets (Trouw & Co., Putten, The Netherlands) at the rate of 1.5% body weight per day.

Preparation of permeabilized gill cells

Gill cells were isolated as described by Verbost et al. (1994a). In brief, the branchial epithelium is scraped from the underlying cartilaginous filaments using the edge of a glass slide. The scrapings were then incubated for 20 min at room temperature in a lysis medium (9 parts 0.17 M NH₁Cl, 1 part 0.17 M Tris/HCl pH 7.4; Yust et al. 1976) which produced both blood cell lysis and tissue fractionation. Cells were suspended at the beginning and resuspended at the end of this incubation period by passing them 5 times through a 10 ml pipet (3 mm bore diameter). The lysis was optimal when 2.5 mg tissue scrapings were lysed in 10 ml lysis-medium (around 3.25×10^7 cells per 10 ml). The cells were filtered through 100 µm mesh nylon gauze to remove persistent cell clusters. The resulting suspension was centrifuged at 150 g for 5 min at 4°C in a swing out rotor (BS4402/A rotor, Heraeus). The cells were resuspended in lysis medium (1 mg of protein per ml) containing 135 mM KCl, 1.0 mM MgCl₂, 1.2 KH₂PO₄, 10 mM Hepes pH 7.4, and 30 μ g ml⁻¹ saponin and incubated for 5 min at 37°C. The isolated, permeabilized gill cells were washed twice in the final uptake medium (see below). Saponin treatment yielded 50 to 60% leaky cells. The degree of leakiness was estimated by uptake of trypan blue (0.5%). After saponin treatment cells were pelleted, washed and resuspended to a concentration of 4 mg ml⁻¹ in a Ca²⁺ uptake medium containing (in mmol l⁻¹) 120 KCl, 1 MgCl₂, 1.2 KH₂PO₄, 5 succinate, 5 pyruvate, 0.5 EGTA, 0.5 HEEDTA, 0.5 NTA, 25 Hepes/KOH (pH 7.1). Cells were used on the day of isolation. The protein content of the cell preparations was

determined with a commercial reagent kit (Bio-Rad) with BSA as a reference. This preparation comprises a mixture of gill epithelial cells with ionocytes (about 10%) and respiratory cells (around 90%).

Isolation of microsomes

Microsomes (the cell membrane fraction mainly consisting of endoplasmic reticulum membranes) were isolated as described by Vercesi et al. (1978). First, gill cells were isolated followed by erythrocyte lysis as described above (preparation of permeabilized gill cells). The following steps were all performed at 4°C. The cells were collected by centrifugation (5 min, $220 \times g$), resuspended in sucrose medium containing (in mmol 1⁻¹) 250 sucrose, 0.55 EGTA, 3 Hepes/Tris (pH 7.4) and subsequently homogenized in a douncer with loose fitting pestle (Braun Melsungen) and the homogenate was centrifuged (5 min, $600 \times g$). The resulting supernatant was centrifuged at $10,000 \times g$ (10) min). The pellet was resuspended in sucrose medium (as above) and centrifuged at $12,000 \times g$ (10) min). This final pellet was resuspended in Ca²⁺ uptake medium (same as given under preparation of permeabilized gill cells) to which mitochondrial inhibitors were added (1 mmol 1^{-1} NaN, and 5 µg ml⁻¹ oligomycin-B). Microsomes were used for the Ca^{2+} uptake studies on the day of isolation.

Hepes/KOH (pH 7.1) and the suspension was filtered immediately (Schleicher and Schüll, GF92). The filters were then washed twice with 2 ml of stop solution, dissolved in scintillation fluid and counted for radioactivity. Total Ca²⁺ uptake was calculated from the radioactivity retained by the filter and is expressed as nmol mg⁻¹ protein.

Ruthenium red, an inhibitor of mitochondrial Ca^{2+} uptake (Moore 1971), was used (20 µmol l⁻¹). With other mitochondrial inhibitors as oligomycin-B (5 µmol l⁻¹) and sodium azide (5 mmol l⁻¹) very similar results were obtained (results not shown). Thapsigargin (1 µmol l⁻¹) was used to inhibit the Ca^{2+} uptake by the endoplasmic reticulum (Thastrup *et al.* 1990; Hovemadsen and Bers 1993). The inhibitors were added to the permeabilized cells 10 to 20 min before the start of the assay.

Data presentation

Data are expressed as the mean \pm SEM. For statistical analysis of the results a repeated measures ANOVA was applied (3 way; Zn*Ca*time). To compare individual means a two way ANOVA on the data from the 2 Ca concentrations was used. Differences were considered to be significant at p < 0.05.

Results

Ca²⁺ uptake in permeabilized cells and microsomes

ATP-dependent ⁴⁵Ca²⁺ uptake was measured as described by Van Corven *et al.* (1987) and Van de Put *et al.* (1991). The assay was performed at 28°C in the Ca²⁺ uptake medium (described above) but which also contained 10 units ml⁻¹ creatine kinase, 10 mmol l⁻¹) creatine phosphate, 1 mmol l⁻¹ ATP and 185 KBq ml^{-1 45}Ca²⁺. The free Mg²⁺ concentration (0.8 mmol l⁻¹) and the free Ca²⁺ concentration (0.8 mmol l⁻¹) and the free Ca²⁺ concentration (as indicated) were calculated according to Schoenmakers *et al.* (1992). ⁴⁵Ca²⁺ uptake was started by adding permeabilized cells or microsomes to prewarmed uptake medium. At certain time intervals 100 µl aliquots (40 µg protein) were quenched in 1 ml of ice-cold stop solution containing (in mmol l⁻¹) 150 KCl, 1 MgCl₂, 1 EGTA, 20 Ca²⁺ uptake in permeabilized cells

The time dependence of Ca^{2+} uptake by permeabilized gill cells at 0.1 µmol l⁻¹ is shown in Figure 1. Ca^{2+} uptake is stimulated by ATP, reaching a steady state level between 6 and 8 min. The uptake was not affected by a blocker of mitochondrial uptake (ruthenium red or oligomycin-B and sodium azide). Thapsigargin, a specific inhibitor of ER Ca^{2+} -ATPases, inhibited almost all ATP-dependent Ca^{2+} uptake. ATP-independent binding comprises about 30% of the total uptake.

The time dependence of Ca^{2+} uptake at 1.0 µmol $l^{-1}Ca^{2+}$ is shown in Figure 2. For the duration of the experiment (8 min) the uptake was linear with time and was almost completely inhibited by the mito-chondrial inhibitor ruthenium red. The uptake in the presence of mitochondrial inhibitors was fur-

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time (min)

Fig. 1. Time-dependent Ca^{2+} uptake by permeabilized tilapia gill cells at a free Ca^{2+} concentration of 0.1 µmol 1⁻¹. Solid circles represent uptake in the presence of ATP (control); this was compared with Ca²⁺ uptake in the presence of thapsigargin (1 μ mol l⁻¹) or ruthenium red (20 μ mol l⁻¹) and that the absence of ATP (-ATP). The inset shows the ATP-dependent Ca^{2+} uptake (same curves after subtraction of the -ATP values). Mean values of 4 experiments \pm SEM are shown.

ther reduced by the "ER inhibitor" thapsigargin to the level when ATP is omitted (this line is exactly the same as that for the -ATP condition in Fig. 2, and was left out for clarify).

Effects of IP, on the uptake of Ca²⁺ in permeabilized cells and microsomes

time (min)

Fig. 2. Time-dependent Ca^{2+} uptake by permeabilized tilapia gill cells at a free Ca^{2+} concentration of 1.0 µmol l⁻¹. Solid circles represent uptake in the presence of ATP (control); this was compared with Ca²⁺ uptake in the presence of thapsigargin (1 μ mol l⁻¹) or ruthenium red (20 μ mol l⁻¹) and that in the absence of ATP (-ATP). The inset shows the ATP-dependent Ca²⁺ uptake (same curves after subtraction of the -ATP values). Mean values of 5 experiments \pm SEM are shown.

Effects of Zn²⁺ on Ca²⁺ uptake in permeabilized cells at 0.1 $\mu M Ca^{2+}$

There was not significant difference between the ATP-dependent uptake in the presence of 0.1 and 0.5 nmol 1^{-1} Zn²⁺ compared to the control (Fig. 4). With 1 nmol l⁻¹ Zn²⁺ in the medium, uptake was inhibited significantly at all time points. The uptake was linear with time for up to 2 min and applying the initial ATP-dependent uptake rates from 1 min time points an IC₅₀ value of 0.50 nmol l^{-1} Zn²⁺ was calculated.

From the Ca²⁺ taken up ATP-dependently during a 10 min period by permeabilized cells, about 80% is released within 1 min following addition of IP, (10 μ mol l⁻¹): from 2.85 ± 0.36 to 0.57 ± 0.22 nmol Ca²⁺ per mg protein in permeabilized cells (Fig. 3A). Heparin (10 μ mol l⁻¹) completely prevented the IP₃ effect (control); this was done by adding a mixture of IP, and heparin. With the microsomes a similar result was obtained. There was an immediate release of Ca²⁺ following addition of IP₃ (64%, from 2.8 \pm 0.2 to 1.0 \pm 0.1 nmol Ca²⁺ per mg protein microsomes) which could be blocked by heparin (by adding IP, and heparin together = control; Fig. 3B).

as functioning as a calcium sink. From the initial ATP-dependent uptake rates at the 2 min time

Effects of Zn^{2+} on Ca^{2+} uptake in permeabilized cells at 1.0 $\mu M Ca^{2+}$

The (mitochondrial) uptake in the presence of 0.1 nmol l⁻¹ Zn²⁺ was not significantly different from the control (Fig. 5). However, at 0.5 or 1.0 nmol 1^{-1} Zn^{2+} uptake was inhibited at all time points (Fig. 5). The uptake was linear with time for the complete time period in this experiment (8 min) and confirms the classic characteristic of the mitochondria

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Fig. 3. A. Effects of IP, (added at 10 min, indicated by arrow) or a mixture of heparin and IP, (control) on the release of ATP-



dependently stored Ca^{2+} in permeabilized tilapia gill cells. B. Effects of IP₃ (added at 10 min, indicated by arrow) or a mixture of heparin and IP₃ (control) on the release of ATP-dependently stored Ca^{2+} in microsomes from tilapia gill cells.

points an IC₅₀ value of 0.20 μ mol l⁻¹ Zn²⁺ was calculated.

Discussion

Our results show that in gill cells the ER is of major importance for Ca²⁺-sequestration at physiological intracellular levels (0.1 µM). Mitochondria only function as Ca²⁺ buffer at high intracellular Ca²⁺ levels (around 1 μ M). This is congruent with the current models for Ca²⁺ buffering that are know for various cells in higher vertebrates (Joseph et al. 1983; Kreutter and Rasmussen 1984). Although the complete kinetics of the Ca²⁺ uptake were not determined, the results demonstrate that the affinity for Ca²⁺ of the mitochondria is lower than that of the ER, but that the capacity of the mitochondrial Ca²⁺ uptake is much higher than that of the ER. Thus, the ER will be more important for buffering intracellular Ca²⁺ at normal cell-physiological Ca²⁺ levels, comparable, in this respect, to the function of the Ca²⁺ pump in the plasma membrane. The maximum Ca²⁺ release from the intracellular, inositol 1,4,5-triphosphate(IP₃)-sensitive Ca²⁺ pool amounts to about 80% of the ATP-dependent, intracellularly stored ⁴⁵Ca²⁺. The remaining 20% of the accumulated ⁴⁵Ca²⁺ is insensitive to IP₃ and appears to be stored in an IP₃-insensitive part of the ER. The remaining activity was not

Fig. 4. Effects of Zn^{2+} on the ATP-dependent Ca^{2+} uptake in permeabilized tilapia gill cells at a free Ca^{2+} concentration of 0.1 µmol l⁻¹, representing the uptake by ER. Mean values of 4 experiments ± SEM are shown.

stored by mitochondria as the results clearly show the absence of mitochondrial uptake under these test conditions. The fact that the simultaneous addition of the IP, receptor antagonist heparin completely blocked Ca²⁺ release by IP₃ indicates that this release is not an artifact of the addition, but is really mediated by the IP, receptor. In microsomes, a rough ER membrane fraction, the same induction of Ca²⁺ release by IP, was observed and confirms the ER origin of the Ca^{2+} . Zinc (Zn^{2+}) exerted a strong inhibition of the Ca²⁺-uptake by both the ER and mitochondria, the mitochondrial uptake being 2.5 times more sensitive than the uptake by ER. But this does not mean that the mitochondria form an earlier target for Zn²⁺ than the ER because the low prevailing Ca²⁺ concentration in the cytosol would dictate negligible mitochondrial Ca²⁺ uptake under these conditions. The situation in which mitochondrial Ca²⁺ uptake exceeds that of the ER will be close to one where the cell is dying. This is perhaps not a very interesting condition from a physiological point of view, although the high sensitivity of the mitochondria to Zn²⁺ may be very detrimental when cytosolic Ca^{2+} is high in the presence of, and perhaps because of the presence of, Zn²⁺. It is important to note that the mechanisms of Ca²⁺ uptake are completely different for the two Ca²⁺ stores. ER utilise a Ca²⁺-pump whereas mitochondria take Ca²⁺ up *via* a uniporter driven by the proton gradient across the inner membrane (Alberts et al. 1994). These differences could be the



time (min)

bition of ER Ca²⁺ pumps by Zn²⁺ could enhance the inhibition of the plasma membrane Ca²⁺ pump and cause a more widespread Ca²⁺ signal in the cell. It should not be forgotten that in this study we worked with preparations of gill epithelial cells, in which respiratory cells will predominate by about 10-fold over ionocytes (mitochondria rich cells). Since the ionocytes are believed to be responsible for the transcellular uptake of Ca²⁺ (Fenwick 1989; McCormick et al. 1992; Marshall et al. 1995), it is not possible to say that we are only dealing with events in calcium transporting cells. This study rather suggests that waterborne Zn²⁺ may lead to an increase in intracellular Ca²⁺ in gill epithelial cells via inhibition of Ca^{2+} sequestration by ER (and eventually mitochondria). In ionocytes this could result in reduced transport of Ca²⁺. For respiratory cells nothing is known at present about the possible effects a rise in cytosolic Ca²⁺ could have on cell function.

Fig. 5. Effects of Zn^{2+} on the ATP-dependent Ca^{2+} uptake in permeabilized tilapia gill cells at a free Ca^{2+} concentration of 1.0 µmol 1⁻¹, representing the uptake by mitochondria. Mean values of 4 experiments ± SEM are shown.

basis for the dissimilar sensitivity to Zn^{2+} of the ER and mitochondria.

How can we fit these data into a complete picture of what will happen when a fish is exposed to water containing Zn²⁺? At sublethal concentrations in the micromolar range, Zn²⁺ impairs the branchial influx of Ca²⁺ resulting in hypocalcemia (Spry and Wood 1985). Zn^{2+} has 10 times the affinity for Ca²⁺-uptake sites than Ca²⁺ and clearly outcompetes Ca²⁺ (Spry and Wood 1989; Hogstrand et al. 1994, 1995a). Recent studies indicated that Zn²⁺ enters the gill cells (Wicklund et al. 1992; Hogstrand et al. 1995b) and this has resulted in the hypothesis that Zn²⁺ inhibits transcellular Ca²⁺ uptake by inhibiting the Ca²⁺ pump in the basolateral membrane. A similar mechanism was suggested for the inhibition of Ca²⁺ uptake by Cd²⁺ (Verbos *et* al. 1988, 1989). The effects of Zn^{2+} (free Zn levels in a metal-buffered assay media) on the basolateral Ca²⁺ pump has, to the best of our knowledge, only been studied in trout (Hogstrand et al. 1995b). There, it caused a mixed inhibition of the Ca²⁺ transporter. At 0.5 nmol l⁻¹ Zn²⁺, the affinity was reduced 8-fold and the V_{max} was decreased by a factor of 2. The results in this study show that the intracellular sequestration of Ca²⁺ is a putative target of Zn²⁺ entering the cell in addition to the plasma membrane Ca²⁺-pump. Inhibition of the Ca²⁺ sequestration will result in a rise in cytosolic Ca²⁺ which will in turn lead to a reduction of Ca²⁺ uptake (Marshall et al. 1995). In this way, the inhi-

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